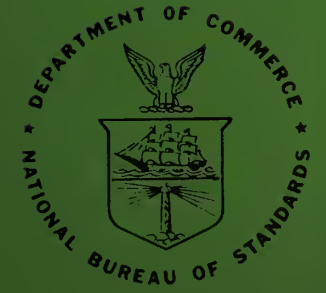
NBS TECHNICAL NOTE

417

Spectral Emission Properties of NBS Standard Phosphor Samples Under Photo-Excitation



U.S. DEPARTMENT OF COMMERCE
National Bureau of Standards

THE NATIONAL BUREAU OF STANDARDS

The National Bureau of Standards[1] provides measurement and technical information services essential to the efficiency and effectiveness of the work of the Nation's scientists and engineers. The Bureau serves also as a focal point in the Federal Government for assuring maximum application of the physical and engineering sciences to the advancement of technology in industry and commerce. To accomplish this mission, the Bureau is organized into three institutes covering broad program areas of research and services:

THE INSTITUTE FOR BASIC STANDARDS . . . provides the central basis within the United States for a complete and consistent system of physical measurements, coordinates that system with the measurement systems of other nations, and furnishes essential services leading to accurate and uniform physical measurements throughout the Nation's scientific community, industry, and commerce. This Institute comprises a series of divisions, each serving a classical subject matter area:

—Applied Mathematics—Electricity—Metrology—Mechanics—Heat—Atomic Physics—Physical Chemistry—Radiation Physics—Laboratory Astrophysics[2]—Radio Standards Laboratory,[2] which includes Radio Standards Physics and Radio Standards Engineering—Office of Standard Reference Data.

THE INSTITUTE FOR MATERIALS RESEARCH . . . conducts materials research and provides associated materials services including mainly reference materials and data on the properties of materials. Beyond its direct interest to the Nation's scientists and engineers, this Institute yields services which are essential to the advancement of technology in industry and commerce. This Institute is organized primarily by technical fields:

—Analytical Chemistry—Metallurgy—Reactor Radiations—Polymers—Inorganic Materials—Cryogenics[2]—Office of Standard Reference Materials.

THE INSTITUTE FOR APPLIED TECHNOLOGY . . . provides technical services to promote the use of available technology and to facilitate technological innovation in industry and government. The principal elements of this Institute are:

—Building Research—Electronic Instrumentation—Technical Analysis—Center for Computer Sciences and Technology—Textile and Apparel Technology Center—Office of Weights and Measures—Office of Engineering Standards Services—Office of Invention and Innovation—Office of Vehicle Systems Research—Clearinghouse for Federal Scientific and Technical Information[3]—Materials Evaluation Laboratory—NBS/GSA Testing Laboratory.

---

[1] Headquarters and Laboratories at Gaithersburg, Maryland, unless otherwise noted; mailing address Washington, D. C., 20234.

[2] Located at Boulder, Colorado, 80302.

[3] Located at 5285 Port Royal Road, Springfield, Virginia 22151.

UNITED STATES DEPARTMENT OF COMMERCE
C. R. Smith, Secretary
NATIONAL BUREAU OF STANDARDS • A. V. Astin, Director

TECHNICAL NOTE 417

ISSUED MARCH 1968

Spectral Emission Properties of NBS Standard Phosphor Samples Under Photo-Excitation

Carl F. Shelton

Metrology Division
Institute for Basic Standards
National Bureau of Standards
Washington, D.C. 20234

For sale by the Superintendent of Documents, U.S. Government Printing Office, Washington, D.C., 20402
Price 25 cents.

Contents

SPECTRAL EMISSION PROPERTIES OF NBS STANDARD PHOSPHOR SAMPLES UNDER PHOTO-EXCITATION

Carl F. Shelton*

The photo-excitation spectral emission properties of 10 of the 14 NBS standard phosphor samples have been determined. Pressed tablets of the phosphors were excited by radiation from a mercury arc lamp, passing through a narrow band-pass filter to obtain either 2537Å or 3650Å excitation. The measurement system is described, correction of the data is discussed, and the spectral emission data are presented. Relative quantum efficiencies were calculated. The results are compared with measurements reported by two other laboratories.

Key Words: Phosphors, photo-excitation, photoluminescence, spectral emission, spectral radiometry, standard phosphor samples.

1. Introduction

The spectral emission properties of ten NBS standard phosphor samples under ultraviolet excitation have been measured. The phosphors and the exciting radiation used are listed in Table I. The relative quantum efficiencies of the 2537Å excited phosphors were calculated relative to magnesium tungstate from the data obtained.

The purpose of this report is to describe the measurement technique used, to discuss a computer program written to reduce the raw data, and to present the results obtained.

2. Measurement Technique

Measurements of the relative spectral emission of the phosphor samples under photo-excitation were made with the equipment shown in Figure 1. The phosphor powder sample to be measured was placed in a small cup made of aluminum (1.25 inches in diameter and approximately 0.04 inch deep) and pressed level by using a spatula. This phosphor plaque was then placed on a turret sample holder mounted horizontally in front of the monochromator which was mounted vertically.

A 4-watt low-pressure mercury-arc lamp (germicidal) was used as source for 2537Å excitation with a filter (Corning No. 9863) in front of the lamp to block the mercury lines in the visible region. A 100-watt high-pressure mercury-arc lamp with a narrow-band-pass filter centered at 3650Å, in front

---

*Research Associate from the International Business Machines Corporation at the National Bureau of Standards, 1967.

TABLE I

NBS Standard Sample Phosphors

Sample No.	Measurement Excitation (Å)	Phosphor Description	Use
1020	3650	Zinc Sulfide ZnS:Ag	Blue Component of P-4 Cathode Ray Tube (CRT) Phosphor
1021	2537	Zinc Silicate Zn_2SiO_4:Mn	P-1 CRT Phosphor
1022	3650	Zinc Sulfide ZnS:Cu	P-2 CRT Phosphor
1023	3650	Zinc-Cadmium Sulfide ZnCdS:Ag	Yellow Component of P-4 CRT Phosphor
1024	3650	Zinc-Cadmium Sulfide ZnCdS:Cu	Orange Component of P-14 CRT Phosphor
1025	----(1)	Zinc Phosphate $Zn_3(PO_4)_2$:Mn	Red Component of P-22 CRT Phosphor
1026	2537	Calcium Tungstate $CaWO_4$:Pb	CRT, Lamps
1027	2537	Magnesium Tungstate $MgWO_4$	CRT, Lamps
1028	2537	Zinc Silicate Zn_2SiO_4:Mn	Lamps
1029	2537	Calcium Silicate $CaSiO_3$:Pb,Mn	Lamps
1030	----(2)	Magnesium Arsenate $(MgO)_x(As_2O_5)_y$:Mn	Lamps
1031	2537	Calcium Halophosphate $3Ca_3(PO_4).Ca(F,Cl)$:Sb,Mn	Lamps
1032	----(3)	Barium Silicate $BaSi_2O_5$:Pb	Lamps (UV)
1033	----(3)	Calcium Phosphate $Ca_3(PO_4)_2$:Tl	Lamps (UV)

(1) Would not emit under 2537Å or 3650Å excitation
(2) Fine emission structure not resolved with 5Å pass band of monochromator used; results not reported
(3) Not measured.

of the lamp, was used as a source of 3650Å excitation. The output of the monochromator was measured by using a photomultiplier tube (PMT) with an S-20 photocathode (Dumont EM2433), an amplifier or a picoammeter, and a digital voltmeter as shown in Figure 1.

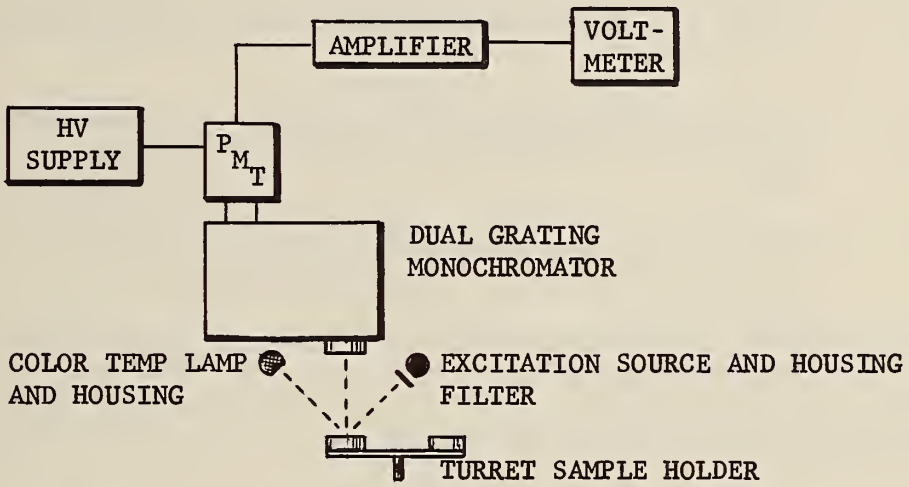


Figure 1 Spectral Measurement Equipment

In order to calibrate the monochromator and PMT combination, its relative spectral sensitivity was measured by using an NBS color temperature standard operated at a color temperature of 2854K. A USP grade barium sulfate (standard reflector) plaque was placed on the turret sample holder so that it could be rotated into position in front of the entrance slit of the monochromator. The spectral radiant flux of the color temperature standard, as diffusely reflected by the plaque, was used to calibrate the equipment.

A measurement of the phosphor sample consisted of setting the monochromator at a particular wavelength (scale readable to three places), and recording the digital voltage reading; first, with the color temperature lamp and the $BaSO_4$ plaque; and second, with the excitation source and the phosphor sample in front of the entrance slit. Both lamps were housed and shuttered such that they could be left on continuously during a data run and used independently. The wavelength scale on the monochromator was verified by using the mercury lines in the low pressure excitation source.

3. Computer Correction of Data

A computer program was used to calculate the relative spectral sensitivity of the instrument for each run and to apply calibration corrections to the data obtained in order to obtain the relative spectral distribution of the phosphor photo-luminescence for each data run. The corrected phosphor relative radiant energy distribution curves for several independent runs were then averaged over the number of runs to provide an average curve for each phosphor measured.

The following quantities are defined:

N ≡ Number of data runs for each phosphor

λ_j ≡ Wavelength value: $j = 1, \ldots, 20$

$S_i(\lambda_j)$ ≡ Relative sensitivity of instrument for the i-th data run $i = 1, \ldots, N$

$C_i(\lambda_j)$ ≡ Digital voltage reading when using color temperature lamp and $BaSO_4$ plaque for the i-th data run at the j-th wavelength value

$L(\lambda_j)$ ≡ Relative spectral luminance of $BaSO_4$ plaque when irradiated by color temperature standard (2854K).

$D_i(\lambda_j)$ ≡ Digital voltage reading when using excitation source and phosphor plaque.

Thus, since $C_1(\lambda_j)$ is by definition

$$C_1(\lambda_j) = S_i(\lambda_j)\ L(\lambda_j), \tag{1}$$

the relative spectral sensitivity for the i-th data run is therefore

$$S_i(\lambda_j) = \frac{C_i(\lambda_j)}{L(\lambda_j)} \tag{2}$$

also by definition, $$D_i(\lambda_j) = S_i(\lambda_j)\ P_i(\lambda_j) \tag{3}$$

where:

$P_i(\lambda_j)$ ≡ Relative radiant energy distribution of the phosphor for i-th data run.

Therefore, $$P_i(\lambda_j) = \frac{D_i(\lambda_j)}{S_i(\lambda_j)} = \frac{D_i(\lambda_j)\ L(\lambda_j)}{C_i(\lambda_j)} \tag{4}$$

and the normalized distribution curve is given by,

$$NP_i(\lambda_j) = \frac{P_i(\lambda_j)}{P_{i_{Max}}} \tag{5}$$

where:

$P_{i_{Max}}$ ≡ Maximum value of $P_i(\lambda_j)$ for $j = 1,\ldots,20$

and the average relative energy distribution over the N data runs for each phosphor is finally

$$P_{ave}(\lambda_j) = \frac{\sum_{i=1}^{N} NP_i(\lambda_j)}{N} \tag{6}$$

3.1. Results

The relative energy distribution curves obtained for the ten phosphors measured are shown in figures 2 to 11. Each figure is followed by its computer output data, in Tables 2 to 11, showing the tabular values for each curve. The average curve for each phosphor is plotted and the tabular values are given on each curve. The tabular values shown are 95% confidence interval estimates of the true mean, calculated as:

$$\bar{x} \pm \tau_n w$$

where: w = range (maximum value minus minimum value)
n = sample size
$\bar{x}$ = computed mean of sample

and values of τ_n are given in Table 8c(1), page 408, of Ref. 5.

The relative energy distribution curves obtained for Sample Nos. 1020, 1022, 1023 and 1024 with photo-excitation are compared with data obtained with cathode-ray excitation by Bril (Ref. 2). Sample 1022 is a P-2 phosphor and the JEDEC P-2 data (Ref 3) is compared with the results obtained with photo-excitation.

The relative energy distribution curves obtained for Sample Nos. 1021 and 1026 with 2537 Å excitation are compared with results obtained by Bril (Ref 1) also with 2537 Å excitation. While Sample No. 1021 compares very closely with Bril's data, Sample No. 1026 shows some differences. Sample Nos. 1026 and 1029 have been compared with data obtained in 1961 by Dr. Frank J. Studer at the Nela Park Laboratory of the General Electric Company (now at NBS). The two sets of data are in close agreement. Since Sample No. 1021 is a P-1 phosphor, the JEDEC P-1 curve is also plotted from Ref. 3.

The difference shown in comparing the results obtained with photo- and cathode-ray excitation are the same order of magnitude as the differences noted above obtained for Sample No. 1026 from data obtained with only photo-excitation, e.g., compare results for Sample No. 1024 where the results of photo and cathode ray excitation are shown, with Sample No. 1026. One might therefore hypothesize that the spectral emission curves are independent of the excitation used, but this hypothesis requires further investigation.

The tabular results show the repeatability of the measurements. In general, the results obtained with the low-pressure mercury-arc lamp (2537 Å excitation) are more repeatable than those obtained with the high-pressure lamp (3650Å excitation) because of greater fluctuations of the output of the latter lamp.

Other measurement problems and sources of uncertainty include the low resolution of the wavelength scale on the monochromator (readable to only three places), the incomplete blocking of the mercury lines by the filter (preventing accurate readings near 4000 Å with 2537 Å excitation source), and the variations in the physical repositioning of the sample with the turret mount between data points.

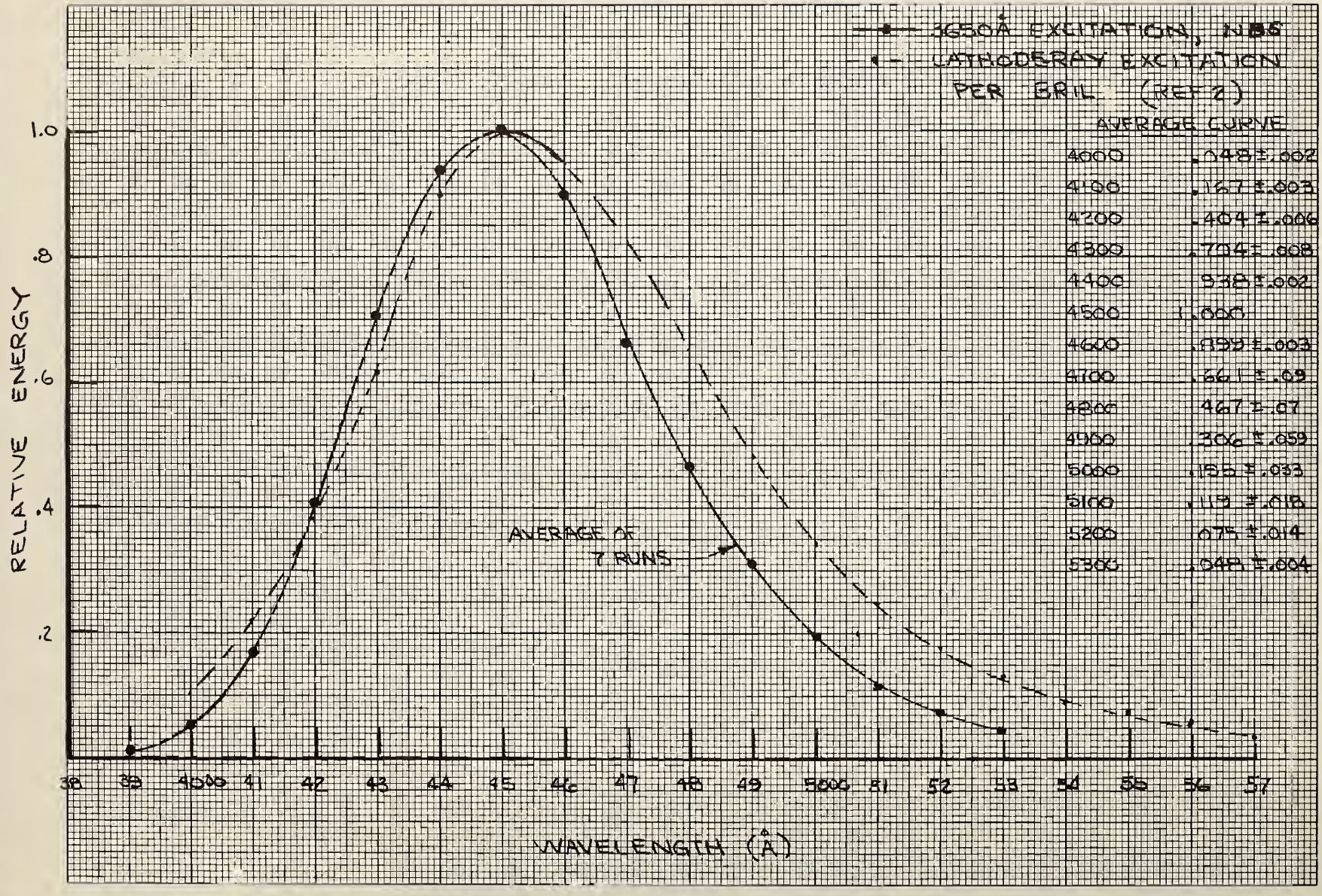


Figure 2. Relative Spectral Energy of Phosphor NBS 1020, Zinc Sulfide, ZnS:Ag

Table 2. Computer Printout of Relative Spectral Energy of Phosphor NBS 1020, Zinc Sulfide, ZnS:Ag

WAVE LENGTH	**NORMALIZED CURVES**							AVERAGE
3800	0.007	0.006	0.007	0.006	0.006	0.006	0.006	0.006
3900	0.011	0.010	0.011	0.012	0.012	0.012	0.012	0.011
4000	0.044	0.045	0.047	0.050	0.050	0.050	0.050	0.048
4100	0.162	0.164	0.163	0.172	0.170	0.171	0.168	0.167
4200	0.399	0.402	0.395	0.405	0.412	0.410	0.406	0.404
4300	0.703	0.690	0.699	0.707	0.713	0.713	0.705	0.704
4400	0.935	0.936	0.936	0.940	0.940	0.941	0.938	0.938
4500	1.000	1.000	1.000	1.000	1.000	1.000	1.000	1.000
4600	0.897	0.897	0.899	0.897	0.907	0.899	0.899	0.899
4700	0.704	0.438	0.709	0.699	0.687	0.694	0.695	0.661
4800	0.491	0.296	0.490	0.489	0.496	0.506	0.498	0.467
4900	0.329	0.158	0.334	0.325	0.336	0.331	0.332	0.306
5000	0.211	0.112	0.209	0.206	0.208	0.209	0.208	0.195
5100	0.124	0.075	0.130	0.129	0.119	0.129	0.130	0.119
5200	0.076	0.044	0.086	0.080	0.079	0.079	0.079	0.075
5300	0.048	0.038	0.051	0.048	0.049	0.049	0.050	0.048

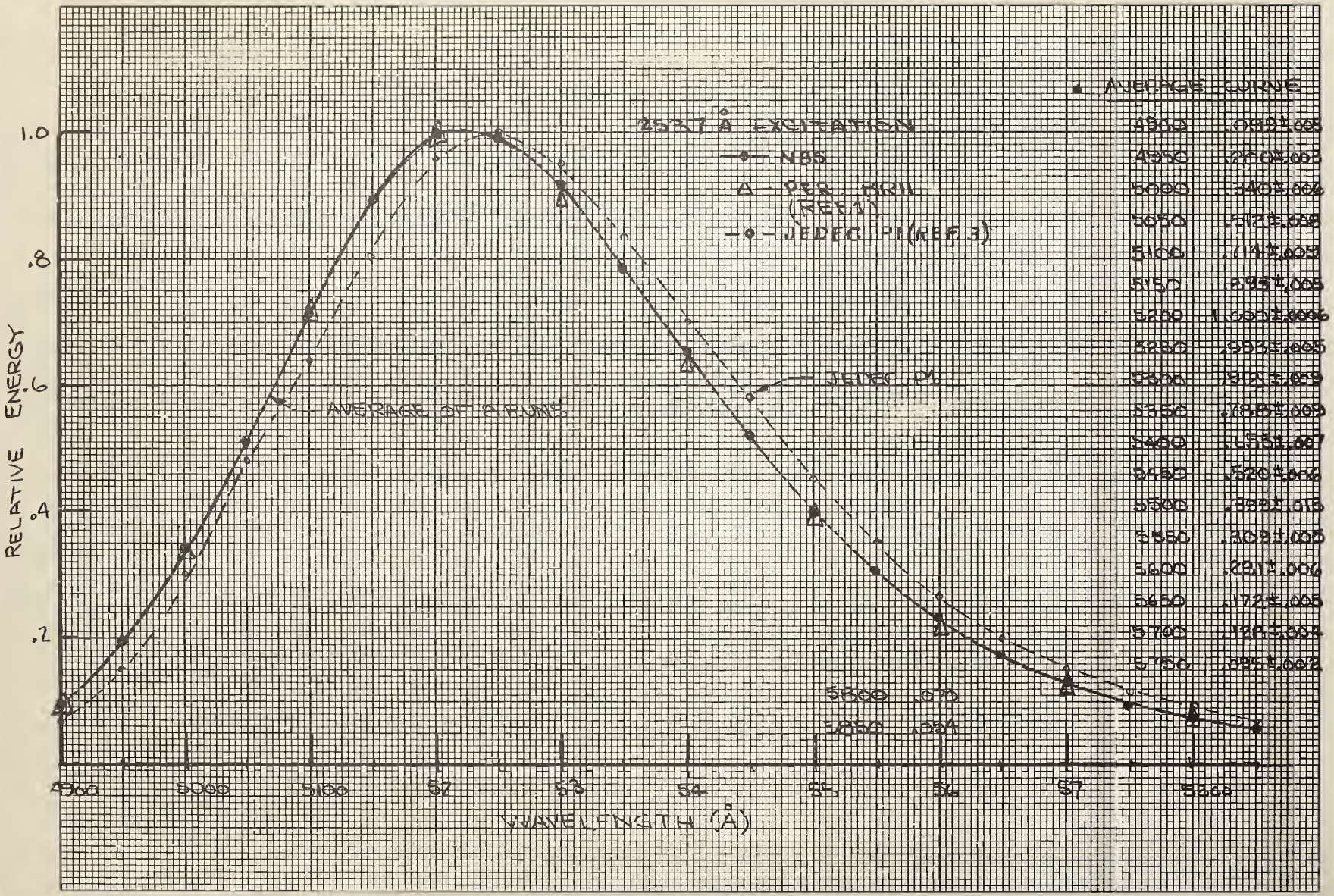


Figure 3. Relative Spectral Energy of Phosphor NBS 1021, Zinc Silicate, $ZnSiO_4$:Mn

Table 3. Computer Printout of Relative Spectral Energy of Phosphor NBS 1021, Zinc Silicate, $ZnSiO_4$:Mn

WAVE LENGTH	**NORMALIZED CURVES**							AVERAGE
4900	0.100	0.098	0.097	0.102	0.104	0.099	0.087	0.105	0.099
4950	0.200	0.203	0.200	0.204	0.198	0.198	0.193	0.202	0.200
5000	0.347	0.342	0.329	0.342	0.345	0.333	0.332	0.349	0.340
5050	0.498	0.515	0.504	0.515	0.526	0.514	0.513	0.515	0.512
5100	0.720	0.709	0.696	0.711	0.729	0.707	0.710	0.725	0.714
5150	0.890	0.899	0.884	0.900	0.899	0.892	0.898	0.901	0.895
5200	1.000	0.998	1.000	1.000	1.000	1.000	1.000	1.000	1.000
5250	0.994	1.000	0.982	0.991	0.998	0.994	0.994	0.993	0.993
5300	0.916	0.931	0.903	0.918	0.927	0.934	0.909	0.910	0.918
5350	0.793	0.805	0.772	0.787	0.789	0.805	0.774	0.779	0.788
5400	0.647	0.663	0.644	0.655	0.653	0.670	0.644	0.646	0.653
5450	0.521	0.525	0.518	0.526	0.532	0.524	0.504	0.510	0.520
5500	0.398	0.409	0.388	0.400	0.418	0.415	0.367	0.399	0.399
5550	0.308	0.312	0.303	0.310	0.317	0.315	0.298	0.307	0.309
5600	0.230	0.231	0.230	0.233	0.238	0.236	0.218	0.230	0.231
5650	0.174	0.176	0.171	0.178	0.175	0.174	0.160	0.167	0.172
5700	0.133	0.131	0.125	0.130	0.132	0.132	0.118	0.121	0.128
5750	0.094	0.094	0.094	0.098	0.100	0.091	0.092	0.092	0.095
5800	0.073	0.071	0.067	0.070	0.072	0.071	0.068	0.066	0.070
5850	0.052	0.054	0.055	0.056	0.057	0.057	0.049	0.053	0.054

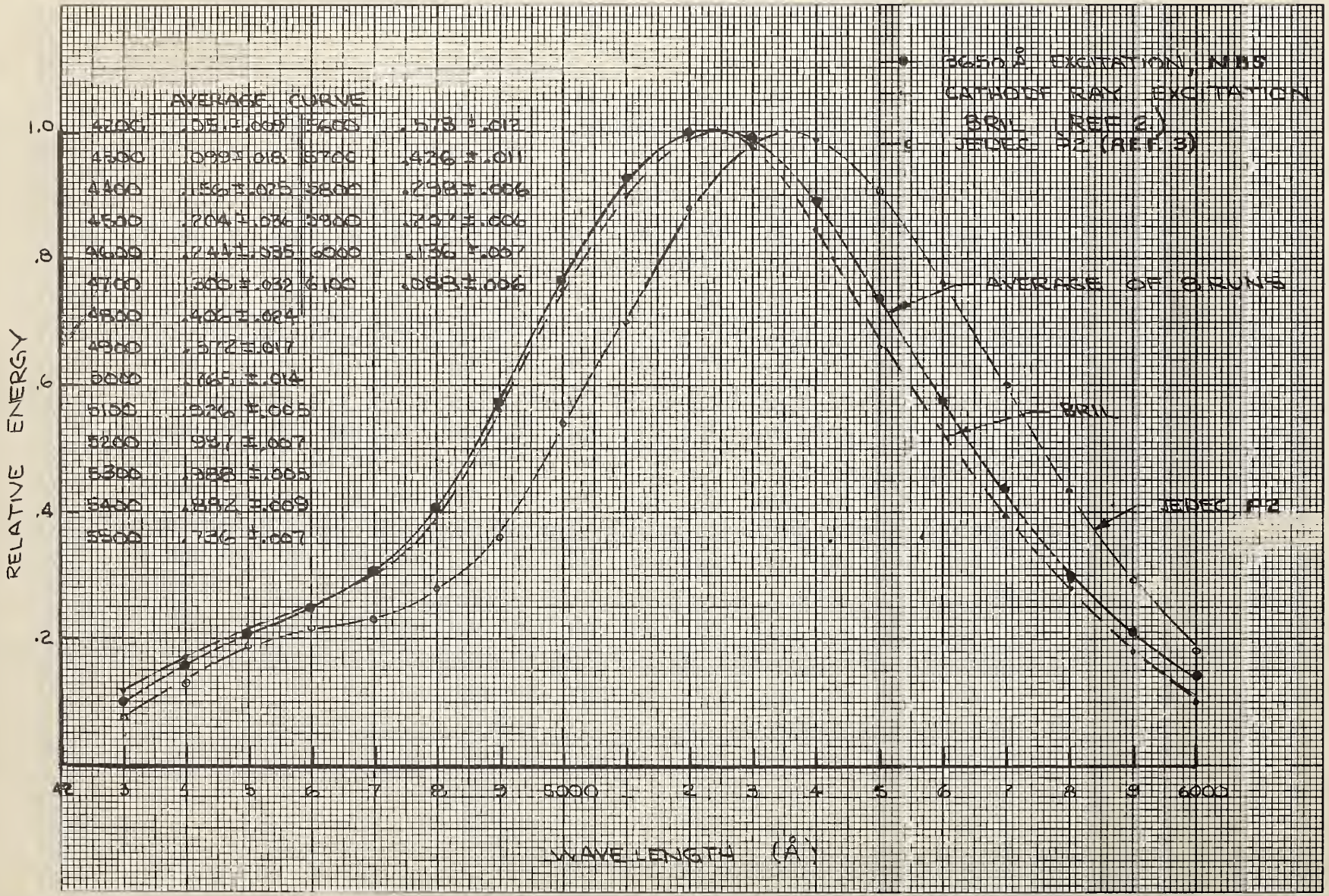


Figure 4. Relative Spectral Energy of Phosphor NBS 1022, Zinc Sulfide, ZnS:Cu

Table 4. Computer Printout of Relative Spectral Energy of Phosphor NBS 1022, Zinc Sulfide, ZnS:Cu

WAVE LENGTH	**NORMALIZED CURVES**							AVERAGE
420C	0.039	0.035	0.038	0.036	0.064	0.065	0.066	0.065	0.051
430C	0.074	0.070	0.068	0.069	0.128	0.129	0.128	0.128	0.099
440C	0.113	0.113	0.114	0.113	0.200	0.200	0.199	0.200	0.156
450C	0.150	0.139	0.150	0.154	0.259	0.263	0.258	0.258	0.204
460C	0.187	0.187	0.183	0.194	0.298	0.303	0.298	0.299	0.244
470C	0.250	0.239	0.260	0.257	0.347	0.351	0.347	0.347	0.300
480C	0.373	0.365	0.366	0.379	0.439	0.447	0.440	0.437	0.406
490C	0.554	0.542	0.543	0.547	0.593	0.602	0.594	0.598	0.572
500C	0.762	0.738	0.739	0.760	0.775	0.786	0.778	0.780	0.765
510C	0.936	0.919	0.919	0.927	0.927	0.933	0.921	0.924	0.926
5200	0.974	1.000	1.000	1.000	1.000	1.000	1.000	1.000	0.997
530C	1.000	0.993	0.988	0.996	0.975	0.991	0.981	0.983	0.988
540C	0.873	0.903	0.878	0.900	0.871	0.882	0.876	0.877	0.882
5500	0.747	0.748	0.740	0.741	0.730	0.732	0.723	0.727	0.736
560C	0.592	0.553	0.576	0.594	0.561	0.573	0.564	0.567	0.573
570C	0.451	0.414	0.430	0.433	0.418	0.428	0.417	0.420	0.426
580C	0.310	0.288	0.309	0.287	0.296	0.298	0.298	0.298	0.298
590C	0.205	0.213	0.213	0.219	0.199	0.203	0.200	0.201	0.207
600C	0.150	0.133	0.135	0.125	0.134	0.134	0.141	0.134	0.136
6100	0.097	0.076	0.092	0.085	0.087	0.089	0.087	0.088	0.088

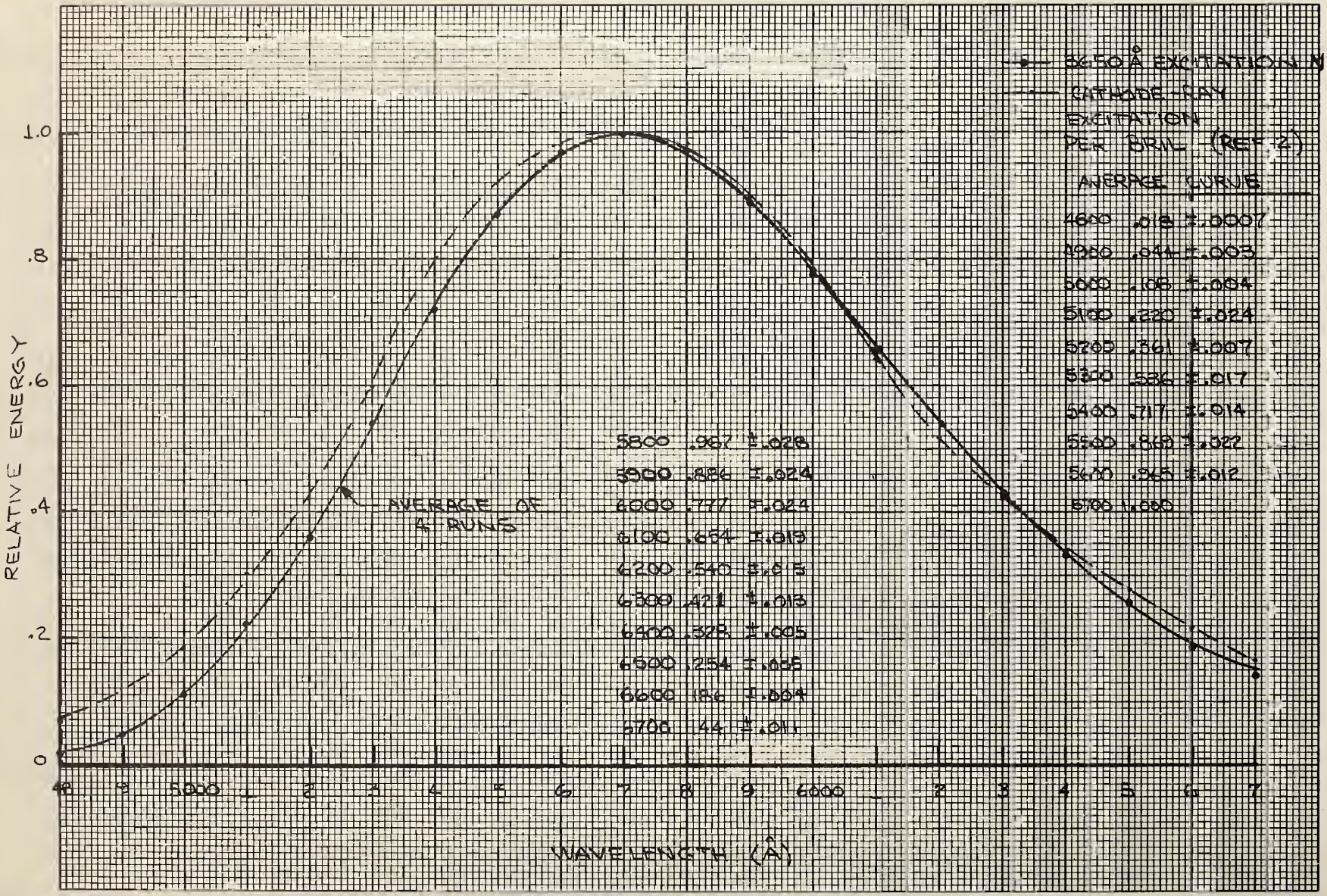


Figure 5. Relative Spectral Energy of Phosphor NBS 1023, Zinc Cadmium Sulfide, ZnCdS:Ag

Table 5. Computer Printout of Relative Spectral Energy of Phosphor NBS 1023, Zinc Cadmium Sulfide, ZnCdS:Ag

WAVE LENGTH	**NORMALIZED CURVES**				AVERAGE
480C	0.017	0.018	0.018	0.018	0.018
490C	0.045	0.045	0.042	0.046	0.044
500C	0.109	0.105	0.109	0.110	0.108
510C	0.244	0.212	0.211	0.212	0.220
520C	0.361	0.356	0.359	0.366	0.361
530C	0.525	0.531	0.540	0.549	0.536
540C	0.713	0.709	0.716	0.729	0.717
550C	0.856	0.868	0.867	0.886	0.869
560C	0.957	0.966	0.964	0.974	0.965
570C	1.000	1.000	1.000	1.000	1.000
580C	0.960	0.953	0.962	0.992	0.967
590C	0.879	0.875	0.881	0.908	0.886
600C	0.760	0.774	0.781	0.794	0.777
610C	0.646	0.642	0.657	0.668	0.654
620C	0.528	0.549	0.540	0.542	0.540
630C	0.408	0.424	0.423	0.426	0.421
640C	0.332	0.325	0.327	0.328	0.328
650C	0.257	0.254	0.256	0.250	0.254
660C	0.185	0.183	0.187	0.189	0.186
670C	0.133	0.146	0.148	0.148	0.144

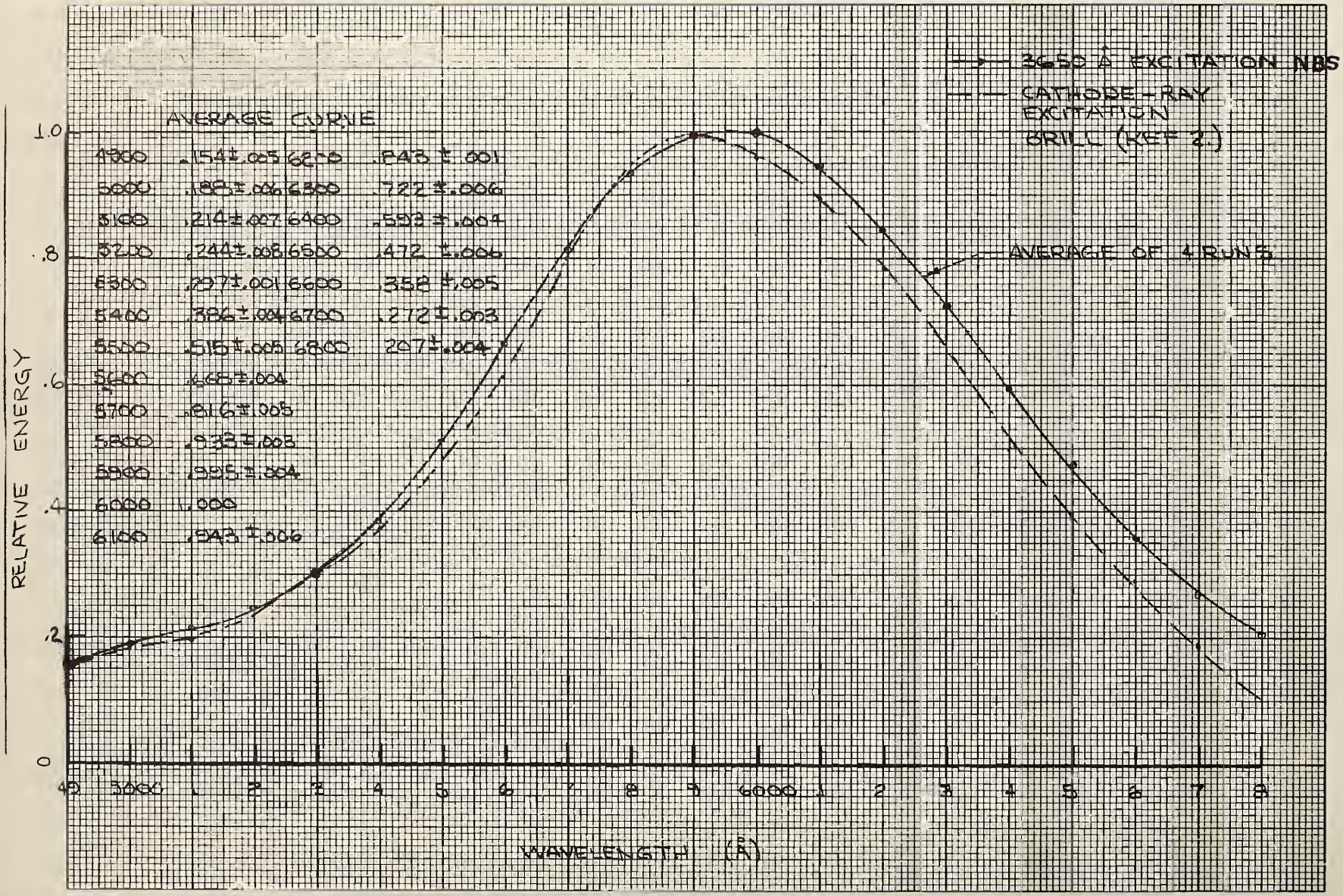


Figure 6. Relative Spectral Energy of Phosphor NBS 1024, Zinc Cadmium Sulfide, ZnCdS:Cu

Table 6. Computer Printout of Relative Spectral Energy of Phosphor NBS 1024, Zinc Cadmium Sulfide, ZnCdS:Cu

WAVE LENGTH	**NORMALIZED CURVES**				AVERAGE
4900	0.158	0.151	0.153	0.154	0.154
5000	0.192	0.183	0.188	0.188	0.188
5100	0.218	0.208	0.215	0.216	0.214
5200	0.247	0.236	0.246	0.246	0.244
5300	0.298	0.296	0.298	0.297	0.297
5400	0.389	0.385	0.384	0.384	0.386
5500	0.515	0.518	0.511	0.515	0.515
5600	0.667	0.672	0.667	0.667	0.668
5700	0.817	0.817	0.812	0.819	0.816
5800	0.932	0.933	0.930	0.936	0.933
5900	0.994	0.998	0.992	0.998	0.995
6000	1.000	1.000	1.000	1.000	1.000
6100	0.939	0.945	0.939	0.948	0.943
6200	0.843	0.842	0.843	0.844	0.843
6300	0.717	0.723	0.725	0.722	0.722
6400	0.590	0.596	0.591	0.593	0.593
6500	0.473	0.474	0.467	0.476	0.472
6600	0.354	0.360	0.356	0.361	0.358
6700	0.273	0.272	0.270	0.274	0.272
6800	0.206	0.209	0.204	0.208	0.207

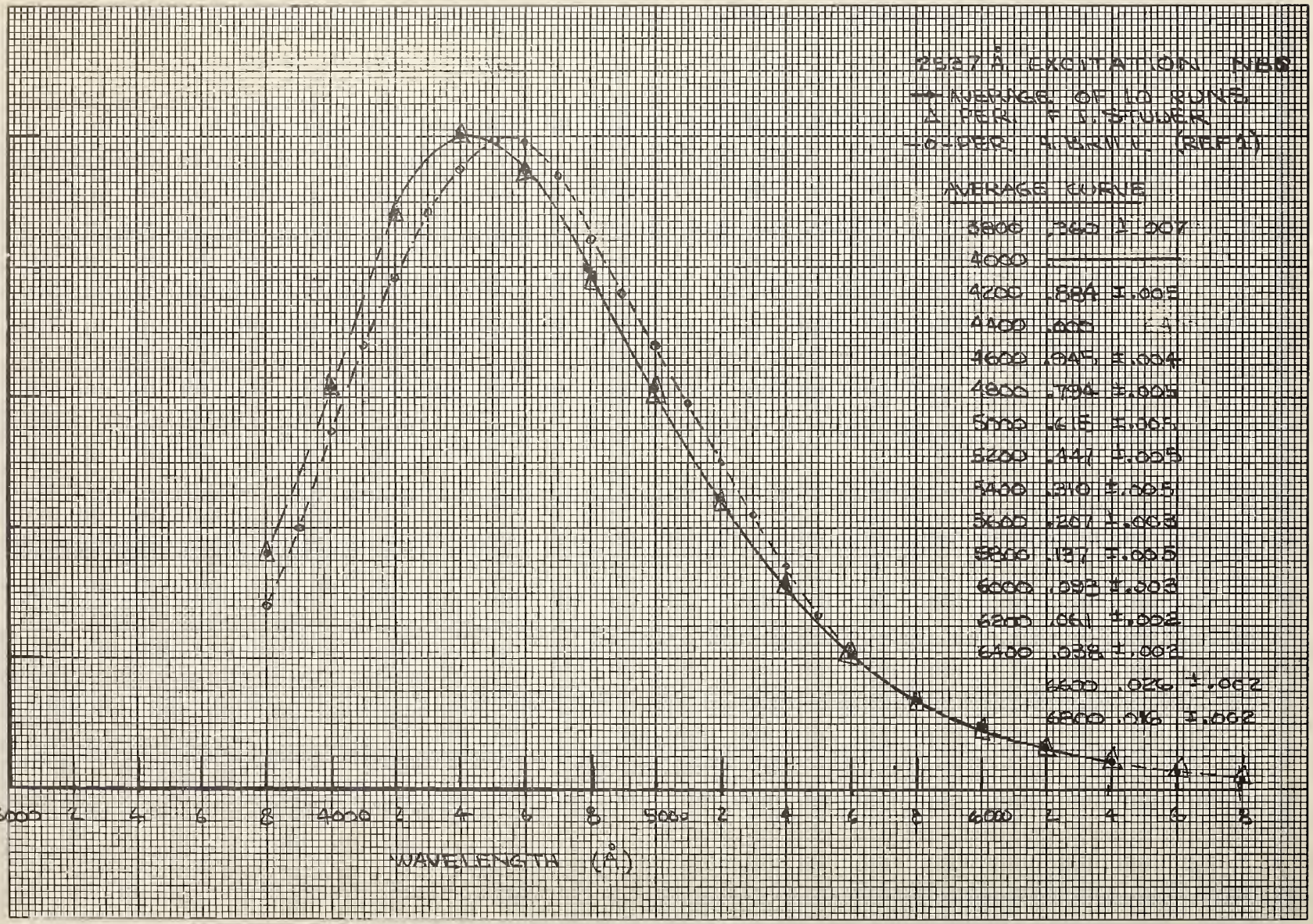


Figure 7. Relative Spectral Energy of Phosphor NBS 1026, Calcium Tungstate, $CaWO_4$:Pb

Table 7. Computer Printout of Relative Spectral Energy of Phosphor NBS 1026, Calcium Tungstate, $CaWO_4$:Pb, (Average energy at 4000 Å is interpolated because of interference of nearby mercury lines in measurements)

WAVE LENGTH	**NORMALIZED CURVES**										AVERAGE
380C	0.355	0.382	0.371	0.368	0.355	0.352	0.356	0.355	0.354	0.354	0.360
400C	~~0.681~~	~~0.860~~	~~0.704~~	~~0.688~~	~~0.694~~	~~0.689~~	~~0.689~~	~~0.686~~	~~0.691~~	~~0.698~~	≈0.620 ~~0.700~~
420C	0.892	0.895	0.887	0.882	0.887	0.886	0.880	0.878	0.880	0.875	0.884
440C	1.000	1.000	1.000	1.000	1.000	1.000	1.000	1.000	1.000	1.000	1.000
460C	0.935	0.947	0.945	0.949	0.947	0.940	0.949	0.951	0.944	0.946	0.945
480C	0.782	0.795	0.780	0.802	0.792	0.802	0.794	0.799	0.802	0.795	0.794
500C	0.606	0.606	0.604	0.617	0.613	0.622	0.617	0.621	0.616	0.624	0.615
520C	0.440	0.441	0.431	0.446	0.447	0.459	0.447	0.451	0.453	0.452	0.447
540C	0.306	0.302	0.295	0.311	0.306	0.311	0.314	0.316	0.317	0.317	0.310
560C	0.202	0.204	0.201	0.209	0.207	0.207	0.209	0.210	0.212	0.212	0.207
580C	0.135	0.135	0.125	0.138	0.139	0.146	0.146	0.130	0.149	0.130	0.137
600C	0.088	0.089	0.085	0.089	0.091	0.098	0.096	0.097	0.097	0.098	0.093
620C	0.057	0.056	0.058	0.056	0.063	0.062	0.063	0.064	0.064	0.064	0.061
640C	0.037	0.039	0.037	0.039	0.040						0.038
660C	0.024	0.028	0.026	0.027	0.024						0.026
680C	0.015	0.018	0.015	0.015	0.018						0.016

Table 8. Computer Printout of Relative Spectral Energy of Phosphor NBS 1027, Magnesium Tungstate, $MgWO_4$ (Average energy at 4000 Å is interpolated because of interference of nearby mercury lines in measurements)

WAVE LENGTH	**NORMALIZED CURVES**									AVERAGE
380C	0.092	0.099	0.106	0.104	0.093	0.094	0.097	0.096	0.097	0.098
400C	~~0.292~~	~~0.328~~	~~0.318~~	~~0.317~~	~~0.902~~	~~0.310~~	~~0.286~~	~~0.306~~	~~0.340~~	≈0.250 ~~0.311~~
420C	0.500	0.490	0.490	0.487	0.496	0.474	0.481	0.479	0.484	0.487
440C	0.761	0.754	0.753	0.749	0.772	0.744	0.750	0.745	0.744	0.752
460C	0.947	0.944	0.938	0.941	0.957	0.934	0.934	0.938	0.943	0.942
480C	1.000	1.000	1.000	1.000	1.000	1.000	1.000	1.000	1.000	1.000
5000	0.941	0.944	0.934	0.936	0.933	0.927	0.952	0.940	0.933	0.938
520C	0.802	0.802	0.801	0.802	0.797	0.802	0.805	0.805	0.806	0.803
540C	0.643	0.634	0.644	0.630	0.621	0.652	0.649	0.655	0.637	0.641
560C	0.474	0.486	0.474	0.472	0.465	0.480	0.491	0.490	0.496	0.481
5800	0.341	0.350	0.346	0.339	0.332	0.344	0.344	0.344	0.347	0.343
600C	0.239	0.245	0.245	0.242	0.227	0.250	0.250	0.249	0.253	0.244
620C	0.160	0.167	0.166	0.162	0.152	0.163	0.164	0.163	0.166	0.163
640C	0.112	0.105	0.111	0.111	0.089	0.135	0.136	0.136	0.137	0.119
660C	0.072	0.071	0.072	0.074	0.057	0.077	0.076	0.077	0.077	0.073
680C	0.047	0.050	0.050	0.051	0.038					0.049

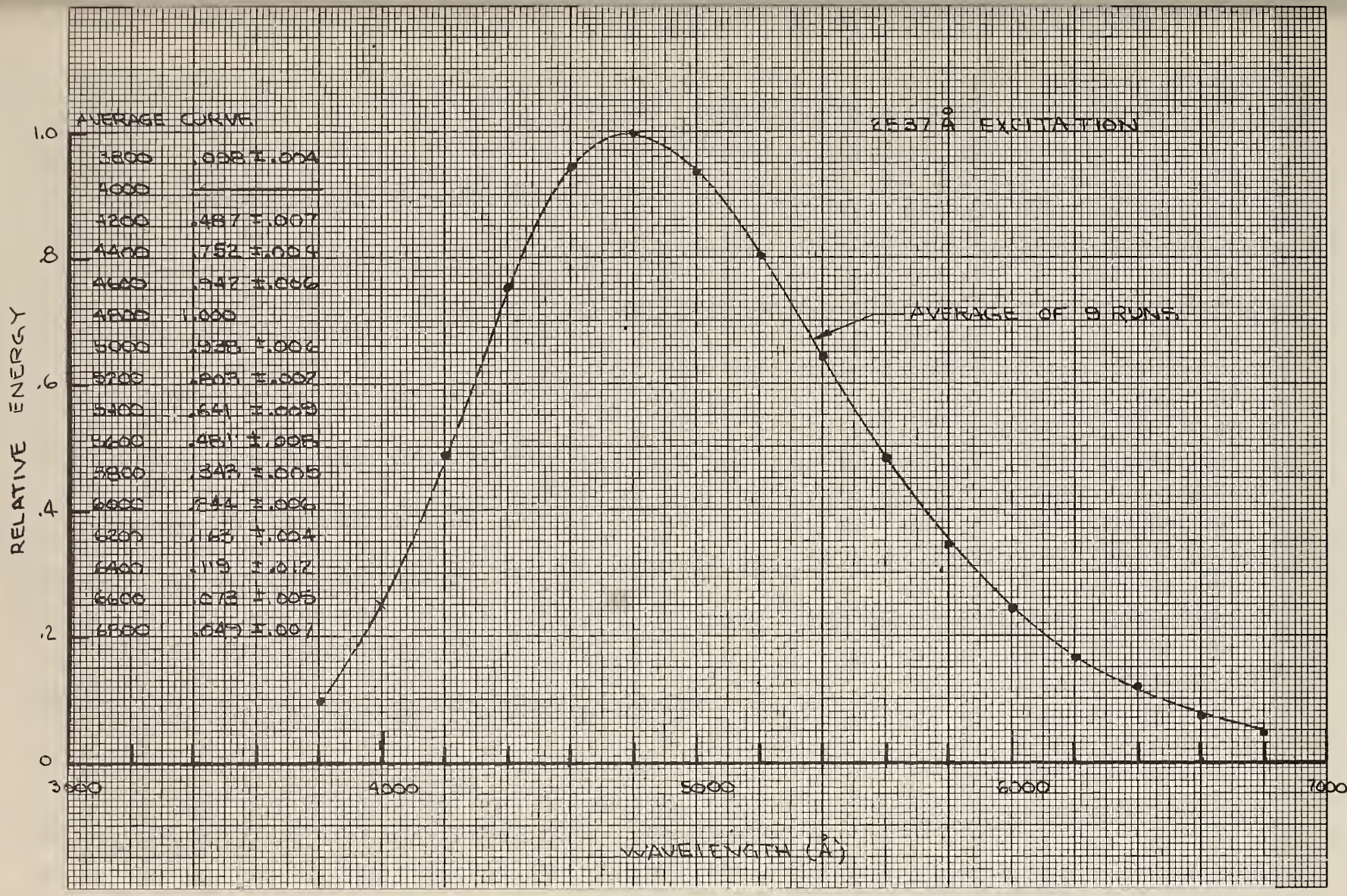


Figure 8. Relative Spectral Energy of Phosphor NBS 1027, Magnesium Tungstate, $MgWO_4$

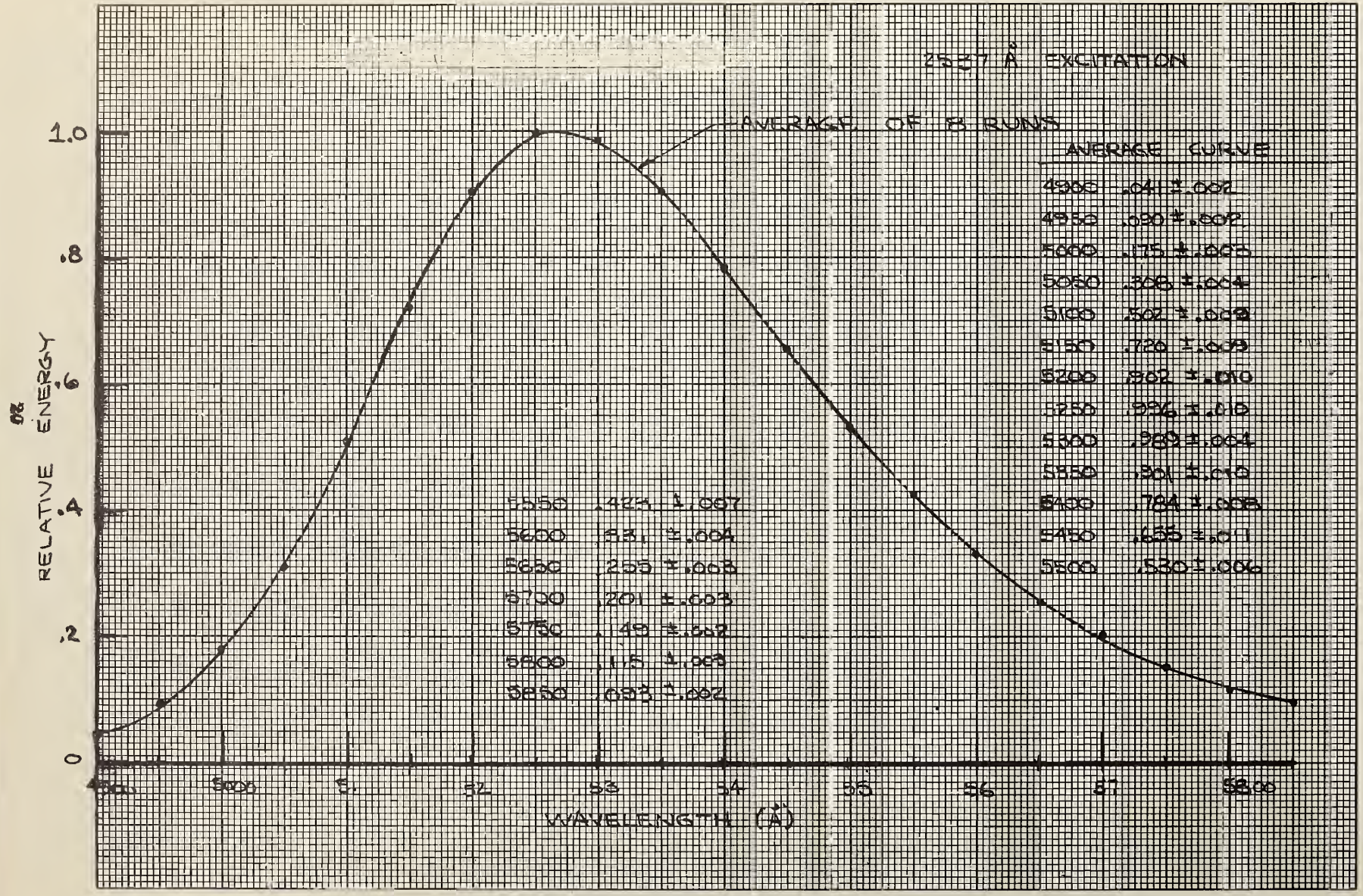


Figure 9. Relative Spectral Energy of Phosphor NBS 1028, Zinc Silicate, Zn_2SiO_4:Mn

Table 9. Computer Printout of Relative Spectral Energy of Phosphor NBS 1028, Zinc Silicate, Zn_2SiO_4:Mn

WAVE LENGTH	**NORMALIZED CURVES**							AVERAGE
490C	0.038	C.042	0.C39	0.045	0.041	0.042	0.043	0.038	0.041
495C	C.088	C.093	0.C93	C.091	0.091	0.085	0.094	0.088	0.090
500C	0.174	C.175	0.172	C.180	0.180	0.172	0.179	0.170	0.175
505C	0.3C6	C.3C9	0.3C5	C.316	0.316	0.306	0.308	0.301	0.308
510C	C.513	C.5C1	0.5C4	0.519	0.496	0.504	0.488	0.487	0.502
515C	0.716	0.723	0.721	0.725	0.725	0.727	0.723	0.696	0.720
520C	C.910	C.9C2	0.9CC	0.913	0.906	0.915	0.894	0.874	0.902
525C	1.000	1.000	1.000	1.000	1.000	1.000	1.000	0.965	0.996
530C	0.989	C.991	0.992	0.981	0.982	0.990	0.987	1.000	0.989
535C	0.9C9	C.91C	0.9C7	0.892	0.895	0.911	0.905	0.878	0.901
540C	C.791	0.789	0.795	0.773	0.781	0.786	0.787	0.766	0.784
545C	0.652	C.671	0.656	0.634	0.656	0.659	0.662	0.647	0.655
550C	0.525	C.537	0.539	0.523	0.527	0.529	0.537	0.518	0.530
555C	0.423	C.421	0.434	0.411	0.420	0.430	0.426	0.415	0.423
560C	0.331	0.338	0.335	C.321	0.333	0.331	0.334	0.326	0.331
565C	0.255	C.257	0.258	0.246	0.258	0.259	0.256	0.250	0.255
570C	0.201	C.2C3	0.2C3	0.197	0.200	0.205	0.201	0.196	0.201
575C	0.153	C.148	0.148	0.148	0.152	0.149	0.151	0.147	0.149
580C	0.116	0.117	0.117	0.108	0.115	0.118	0.112	0.116	0.115
585C	0.095	C.C88	0.C96	0.088	0.094	0.097	0.090	0.095	0.093

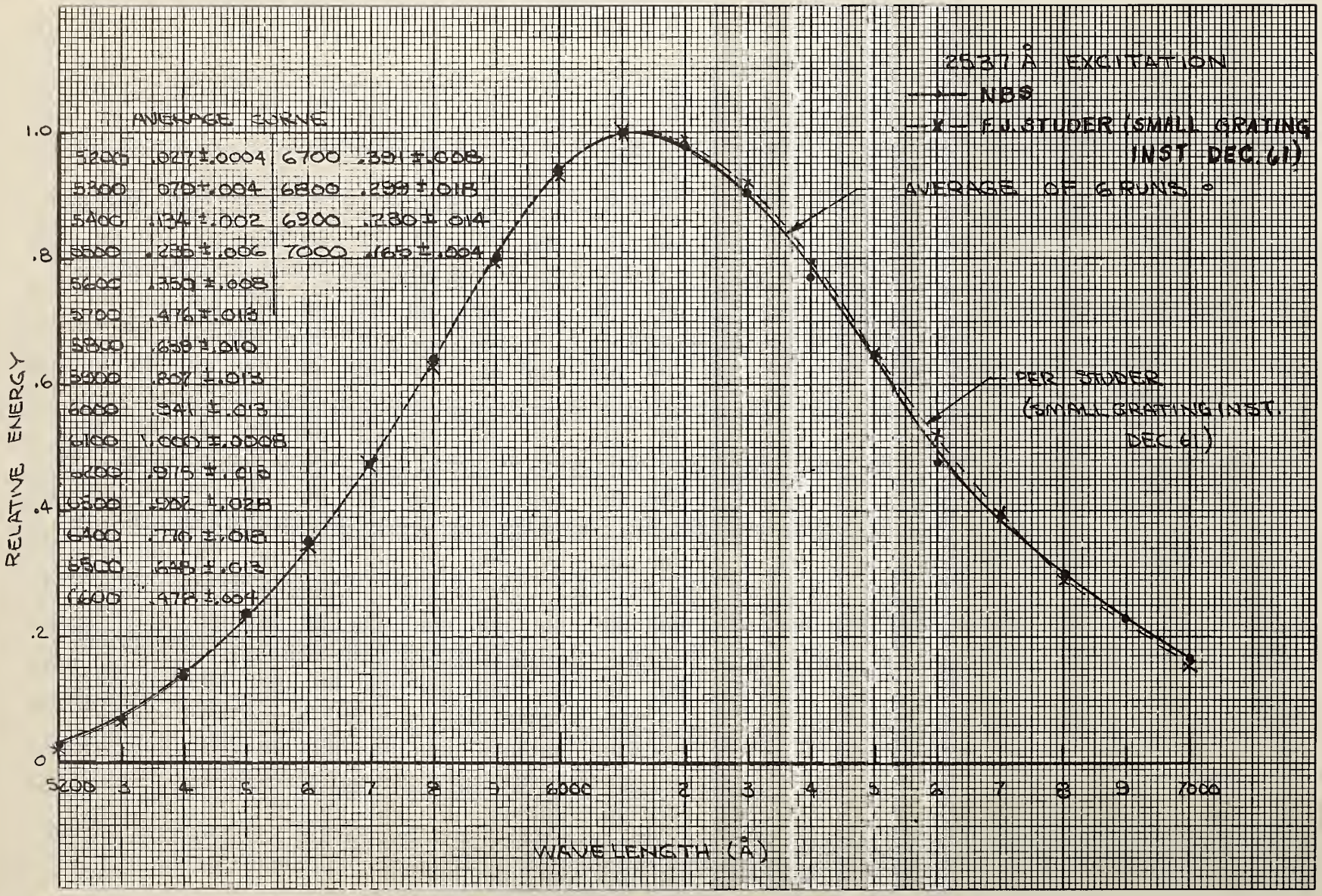


Figure 10. Relative Spectral Energy of Phosphor NBS 1029, Calcium Silicate, $CaSiO_3$:Pb,Mn

Table 10. Computer Printout of Relative Spectral Energy of Phosphor NBS 1029, Calcium Silicate, $CaSiO_3$:Pb,Mn

WAVE LENGTH	**NORMALIZED CURVES**						AVERAGE
520C	0.026	0.027	0.027	0.027	0.027	0.027	0.027
530C	0.071	0.074	0.074	0.063	0.074	0.063	0.070
540C	0.130	0.135	0.135	0.135	0.135	0.135	0.134
550C	0.233	0.241	0.241	0.243	0.227	0.227	0.235
560C	0.348	0.360	0.362	0.345	0.344	0.342	0.350
570C	0.454	0.487	0.487	0.470	0.470	0.486	0.476
580C	0.642	0.648	0.648	0.648	0.625	0.624	0.639
590C	0.811	0.816	0.819	0.819	0.787	0.792	0.807
600C	0.929	0.961	0.961	0.932	0.931	0.934	0.941
610C	0.998	1.000	1.000	1.000	1.000	1.000	1.000
620C	1.000	0.998	0.961	0.965	0.962	0.962	0.975
630C	0.871	0.940	0.901	0.901	0.900	0.901	0.902
640C	0.736	0.766	0.780	0.779	0.778	0.780	0.770
650C	0.627	0.657	0.660	0.657	0.633	0.653	0.648
660C	0.476	0.479	0.475	0.475	0.485	0.479	0.478
670C	0.377	0.396	0.393	0.393	0.395	0.395	0.391
680C	0.298	0.307	0.308	0.264	0.310	0.308	0.299
690C	0.234	0.240	0.210	0.241	0.244	0.213	0.230
700C	0.158	0.165	0.167	0.167	0.166	0.164	0.165
710C	0.0	0.0	0.0	0.0	0.111	0.164	0.046

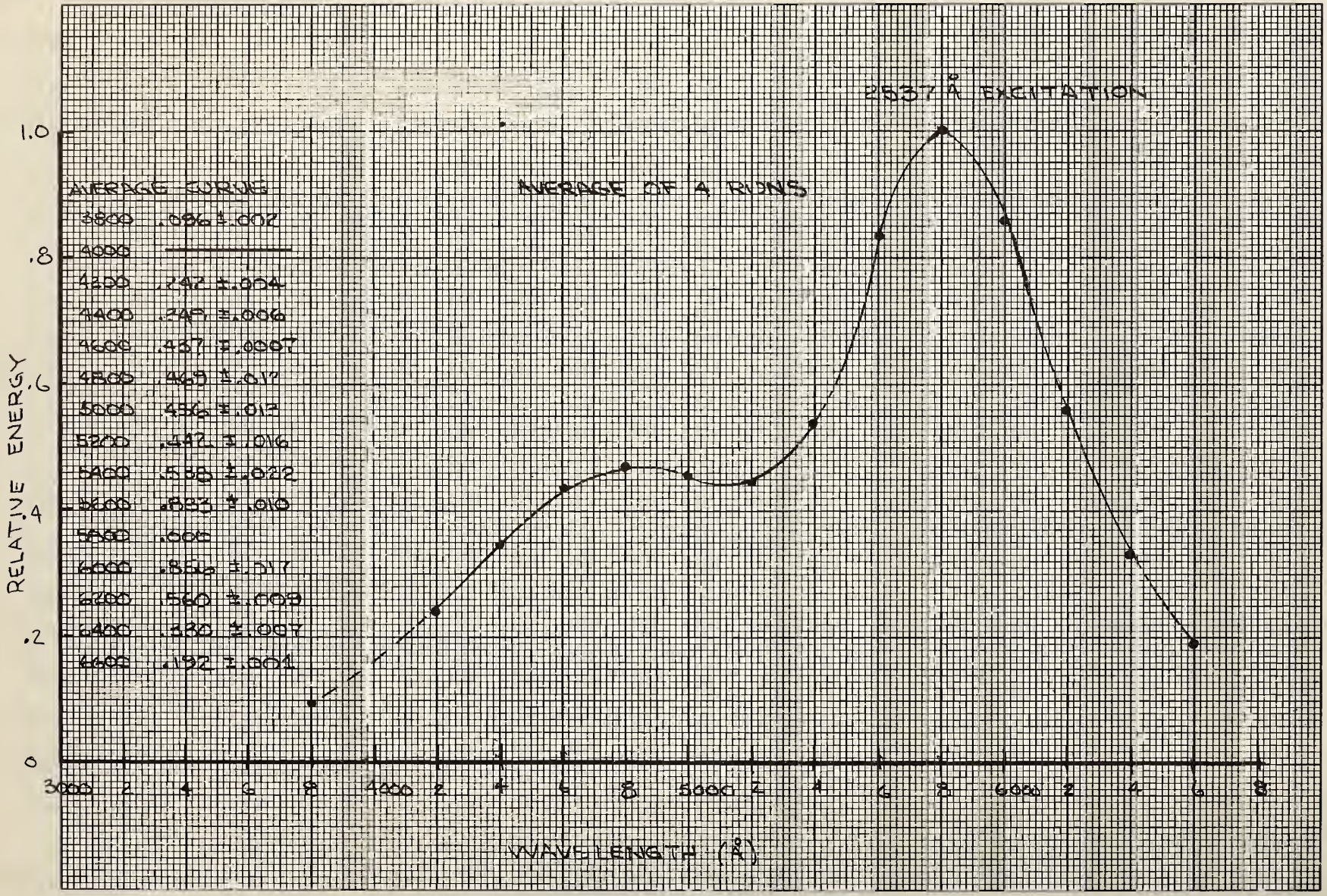


Figure 11. Relative Spectral Energy of Phosphor NBS 1031, Calcium Halophosphate, $3Ca_3(PO_4)_2.Ca(F,Cl):Pb,Mn$

Table 11. Computer Printout of Relative Spectral Energy of Phosphor NBS 1031, Calcium Halophosphate, $3Ca_3(PO_4)_2.C_a(F,Cl):Pb,Mn$ (Average energy at 4000 A is interpolated because of interference of nearby mercury lines in measurements)

WAVE LENGTH	**NORMALIZED CURVES**				AVERAGE
3800	0.097	0.096	0.094	0.097	0.096
4000	~~0.237~~	~~0.264~~	~~0.256~~	~~0.242~~	≈0.169 ~~0.250~~
4200	0.244	0.244	0.238	0.242	0.242
4400	0.350	0.350	0.343	0.352	0.349
4600	0.437	0.438	0.437	0.437	0.437
4800	0.471	0.472	0.459	0.476	0.469
5000	0.449	0.467	0.450	0.458	0.456
5200	0.448	0.450	0.428	0.441	0.442
5400	0.526	0.557	0.528	0.540	0.538
5600	0.829	0.832	0.829	0.843	0.833
5800	1.000	1.000	1.000	1.000	1.000
6000	0.851	0.853	0.849	0.873	0.856
6200	0.556	0.557	0.559	0.569	0.560
6400	0.327	0.328	0.328	0.337	0.330
6600	0.191	0.193	0.192	0.193	0.192

The 45°-0° spectral reflectance factor of the $BaSO_4$ plaque was assumed constant over the wavelength range of interest for each phosphor and the relative spectral emission for a blackbody at 2854°K was used for $L(\lambda_j)$ in all calculations involved in correcting the data. The absolute spectral reflectance of USP grade (unpublished data) $BaSO_4$ is shown in Figure 12. The 45°-0° spectral reflectance factor of the plaque was assumed to be equal to its spectral reflectance.

4. Relative Quantum Efficiency

The relative quantum efficiencies of the 2537 Å excited phosphor samples have been calculated from the scale factors used with a picoammeter during the measurements. The method of calculation is as follows: define

$$f \equiv \text{scale factor}$$

then the calibrated spectral energy distribution is

$$E(\lambda) = f\, P\,(\lambda)$$

where $P(\lambda)$ is the average curve obtained for each phosphor. Thus, since

$$E = h\upsilon = \frac{hc}{\lambda} ,$$

the number of quanta of luminescence is given by,

$$Q = f \int \frac{P(\lambda)}{\frac{hc}{\lambda}}\, d\lambda = \frac{f}{hc} \int \lambda P(\lambda) d\lambda .$$

Since the calculated quantum efficiencies of the samples will be referred to $MgWO_4$, we compute the number of quanta emitted by it as

$$Q_{MgWO_4} = \frac{f_{MgWO_4}}{hc} \int \lambda\, P_{MgWO_4}\,(\lambda) d\lambda .$$

Finally the relative quantum efficiency of each sample is given by,

$$\epsilon \equiv \text{Relative Q.E.} = \frac{Q}{Q_{MgWO_4}} = \frac{f \int \lambda P(\lambda) d\lambda}{f_{MgWO_4} \int \lambda P(\lambda) d\lambda}$$

or for ease of calculation from the data,

$$\epsilon = \frac{f \sum_{j=1}^{20} \lambda_j\, P(\lambda_j)}{f_{MgWO_4} \sum_{j=1}^{20} \lambda_j P(\lambda_j)}$$

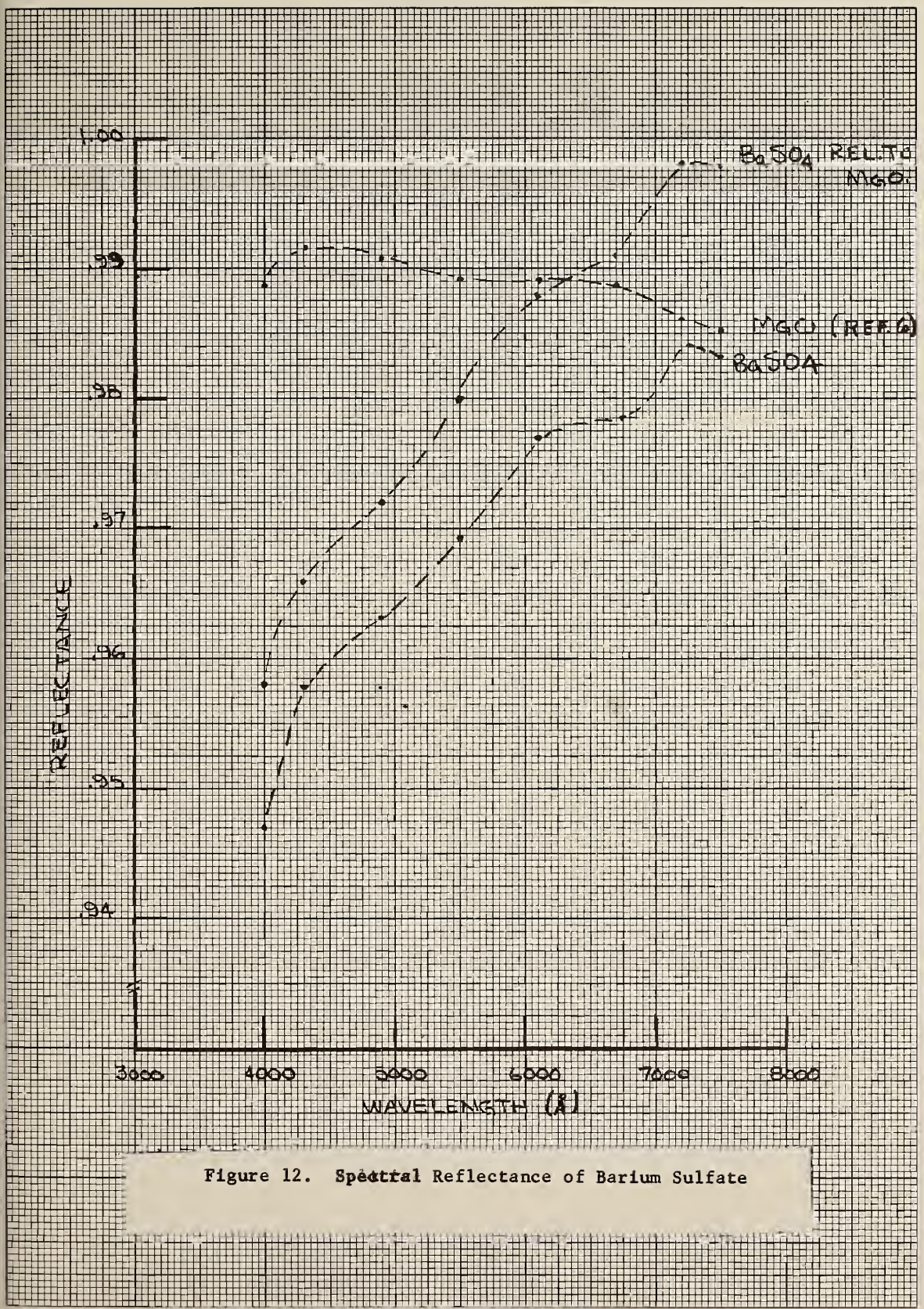


Figure 12. Spectral Reflectance of Barium Sulfate

The following results were obtained with the quantum efficiency of $MgWO_4$ normalized to unity.

RELATIVE QUANTUM EFFICIENCIES
OF 2537Å EXCITED SAMPLES

Sample No.	ε	Per Bril (Ref 1)
1021	.66	.83
1026	.86	.89
1027 (M_gWO_4)	1.00	1.00
1028	.86	.81
1029	.85	.81
1031	.82	.84

REFERENCES

1. A. Bril and W. Hoekstra, "Efficiencies of Phosphors for Short-Wave Ultra-Violet Excitation," Philip Res. Repts. 16, 356-370, (1961).

2. A. Bril, "Absolute Efficiencies of Phosphors with Ultra-Violet and Cathode-Ray Excitation," Kallman and Sprunch, Luminescence of Organics and Inorganic Materials, Wiley, 1962.

3. "Opitcal Characteristics of Cathode Ray Tube Screens," JEDEC Publication No. 16A, JEDEC Electron Tube Council, January 1966.

4. F. J. Studer, "Measurement of Quantum Efficiency" General Electric Research Laboratory Memo Report, November P-207, November 1959.

5. Dixon and Massey, "Introduction to Statistical Analysis", McGraw-Hill, 1957 (page 408).

6. Goebel, Caldwell, and Hammond, Journal of the Optical Society of America 56, 783-788, (1966).

NBS TECHNICAL PUBLICATIONS

PERIODICALS

JOURNAL OF RESEARCH reports National Bureau of Standards research and development in physics, mathematics, chemistry, and engineering. Comprehensive scientific papers give complete details of the work, including laboratory data, experimental procedures, and theoretical and mathematical analyses. Illustrated with photographs, drawings, and charts.

Published in three sections, available separately:

• Physics and Chemistry

Papers of interest primarily to scientists working in these fields. This section covers a broad range of physical and chemical research, with major emphasis on standards of physical measurement, fundamental constants, and properties of matter. Issued six times a year. Annual subscription: Domestic, $5.00; foreign, $6.00*.

• Mathematical Sciences

Studies and compilations designed mainly for the mathematician and theoretical physicist. Topics in mathematical statistics, theory of experiment design, numerical analysis, theoretical physics and chemistry, logical design and programming of computers and computer systems. Short numerical tables. Issued quarterly. Annual subscription: Domestic, $2.25; foreign, $2.75*.

• Engineering and Instrumentation

Reporting results of interest chiefly to the engineer and the applied scientist. This section includes many of the new developments in instrumentation resulting from the Bureau's work in physical measurement, data processing, and development of test methods. It will also cover some of the work in acoustics, applied mechanics, building research, and cryogenic engineering. Issued quarterly. Annual subscription: Domestic, $2.75; foreign, $3.50*.

TECHNICAL NEWS BULLETIN

The best single source of information concerning the Bureau's research, developmental, cooperative and publication activities, this monthly publication is designed for the industry-oriented individual whose daily work involves intimate contact with science and technology—*for engineers, chemists, physicists, research managers, product-development managers, and company executives.* Annual subscription: Domestic, $1.50; foreign, $2.25*.

*Difference in price is due to extra cost of foreign mailing.

NONPERIODICALS

Applied Mathematics Series. Mathematical tables, manuals, and studies.

Building Science Series. Research results, test methods, and performance criteria of building materials, components, systems, and structures.

Handbooks. Recommended codes of engineering and industrial practice (including safety codes) developed in cooperation with interested industries, professional organizations, and regulatory bodies.

Special Publications. Proceedings of NBS conferences, bibliographies, annual reports, wall charts, pamphlets, etc.

Monographs. Major contributions to the technical literature on various subjects related to the Bureau's scientific and technical activities.

National Standard Reference Data Series. NSRDS provides quantitative data on the physical and chemical properties of materials, compiled from the world's literature and critically evaluated.

Product Standards. Provide requirements for sizes, types, quality and methods for testing various industrial products. These standards are developed cooperatively with interested Government and industry groups and provide the basis for common understanding of product characteristics for both buyers and sellers. Their use is voluntary.

Technical Notes. This series consists of communications and reports (covering both other agency and NBS-sponsored work) of limited or transitory interest.

CLEARINGHOUSE

The Clearinghouse for Federal Scientific and Technical Information, operated by NBS, supplies unclassified information related to Government-generated science and technology in defense, space, atomic energy, and other national programs. For further information on Clearinghouse services, write:

Clearinghouse
U.S. Department of Commerce
Springfield, Virginia 22151

Order NBS publications from:
Superintendent of Documents
Government Printing Office
Washington, D.C. 20402

U.S. DEPARTMENT OF COMMERCE
WASHINGTON, D.C. 20230

OFFICIAL BUSINESS

POSTAGE AND FEES PAID
U.S. DEPARTMENT OF COMMERCE