



Codata recommended values of
the fundamental constants of
physics and chemistry (2008)

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2008

**2006 CODATA RECOMMENDED VALUES OF THE FUNDAMENTAL
CONSTANTS OF PHYSICS AND CHEMISTRY** NIST SP 959 (Aug/2008)

Values from: P. J. Mohr, B. N. Taylor, and D. B. Newell, *Rev. Mod. Phys.* **80**, 633 (2008) and *J. Phys. Chem. Ref. Data* **37**, 1187 (2008). The number in parentheses is the one-sigma (1σ) uncertainty in the last two digits of the given value.

Quantity	Symbol	Numerical value	Unit
speed of light in vacuum	c, c_0	299 792 458 (exact)	m s^{-1}
magnetic constant	μ_0	$4\pi \times 10^{-7}$ (exact)	N A^{-2}
electric constant $1/\mu_0 c^2$	ϵ_0	$8.854 187 817 \dots \times 10^{-12}$	F m^{-1}
Newtonian constant of gravitation	G	$6.674 28(67) \times 10^{-11}$	$\text{m}^3 \text{kg}^{-1} \text{s}^{-2}$
Planck constant	h	$6.626 068 96(33) \times 10^{-34}$	J s
$h/2\pi$	\hbar	$1.054 571 628(53) \times 10^{-34}$	J s
elementary charge	e	$1.602 176 487(40) \times 10^{-19}$	C
fine-structure constant $e^2/4\pi\epsilon_0\hbar c$	α	$7.297 352 5376(50) \times 10^{-3}$	
inverse fine-structure constant	α^{-1}	137.035 999 679(94)	
Rydberg constant $\alpha^2 m_e c/2h$	R_∞	10 973 731.568 527(73)	m^{-1}
Bohr radius $\alpha/4\pi R_\infty$	a_0	$0.529 177 208 59(36) \times 10^{-10}$	m
Bohr magneton $eh/2m_e$	μ_B	$927.400 915(23) \times 10^{-26}$	J T^{-1}

Quantity	Symbol	Numerical value	Unit
electron mass	m_e	$9.109\,382\,15(45) \times 10^{-31}$	kg
proton mass	m_p	$1.672\,621\,637(83) \times 10^{-27}$	kg
proton-electron mass ratio	m_p/m_e	1836.152 672 47(80)	
Avogadro constant	N_A, L	$6.022\,141\,79(30) \times 10^{23}$	mol ⁻¹
Faraday constant $N_A e$	F	96 485.3399(24)	C mol ⁻¹
molar gas constant	R	8.314 472(15)	J mol ⁻¹ K ⁻¹
Boltzmann constant R/N_A	k	$1.380\,6504(24) \times 10^{-23}$	J K ⁻¹
Stefan-Boltzmann const. $\pi^2 k^4/60h^3 c^2$	σ	$5.670\,400(40) \times 10^{-8}$	W m ⁻² K ⁻⁴
magnetic flux quantum $h/2e$	Φ_0	$2.067\,833\,667(52) \times 10^{-15}$	Wb
Josephson constant $2e/h$	K_J	$483\,597.891(12) \times 10^9$	Hz V ⁻¹
von Klitzing constant h/e^2	R_K	25 812.807 557(18)	Ω
electron volt (e/C) J	eV	$1.602\,176\,487(40) \times 10^{-19}$	J
(unified) atomic mass unit $\frac{1}{12}m(^{12}\text{C})$	u	$1.660\,538\,782(83) \times 10^{-27}$	kg

A more extensive listing of constants is available in the references given above and on the NIST Physics Laboratory Web site physics.nist.gov/constants.

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