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REPORT OF  
INSPECTION OF PLASTER WORK AT  
ALASKA NATIVE SERVICE HOSPITAL  
AND TESTS FOR QUALITY AND  
CHARACTERISTICS OF PLASTER MATERIALS

By

Nolan D. Mitchell

Requested by the  
Bureau of Indian Affairs



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REPORT OF INSPECTION OF PLASTER WORK  
AT ALASKA NATIVE SERVICE HOSPITAL AND  
TESTS FOR QUALITY AND CHARACTERISTICS  
OF PLASTER MATERIALS

1. INTRODUCTION

Pursuant to a request from the Bureau of Indian Affairs, Department of the Interior, dated February 13, 1953, an inspection was made to determine the condition of plaster in the main hospital building and the quarters buildings for nurses and help at the Alaska Native Service Hospital, Anchorage, Alaska, on February 21, 22, 23, 1953. Participating in the inspection were Messrs. Lansing S. Wells, William C. Cullen, and Nolan D. Mitchell of the National Bureau of Standards. The report of this inspection and related tests supplement a letter of this Bureau dated June 16, 1952, and a report 9.8/G-10015 dated September 17, 1952, both signed by Lansing S. Wells.

Among representatives of the Department of Interior, Bureau of Indian Affairs, who were contacted at the job were:

Mr. John J. Finley, Chief, Building & Utilities Branch, Wash., D.C.  
Mr. L. T. Burn, Architectural Engineer, Bldg. & Util. Br., " "  
Mr. H. T. McCurdy, Chief, Mech. Engineers, " " " " "  
Mr. J. C. Helfrich, Electrical Engineer, " " " " "  
Mr. Harry Halverson, Area Chief,  
Bureau Indian Office, Juneau, Alaska  
Mr. V. P. Reimer, Project Engineer,  
Alaska Native Service Hospital, Anchorage, Alaska

2. INSPECTION OF QUARTERS BUILDING

The first inspections were made in the quarters building because the construction there was nearing completion. Fifteen of twenty rooms and the corridors in the help's quarters section were found to have cracks in the plaster. Cracks were observed in rooms 200, 201, 207, 209, 212, 214, 215, 217, 303, 309, 311, 321, 325, 329, 331, and in the corridor wall in the vicinity of room 303. A number of cracks extended from the corners of electrical outlet boxes. Some of the ceiling cracks extended from electrical outlets toward the walls. The direction of the larger wall cracks was usually more or less vertical. Many of these more prominent cracks were intersected by narrower cracks extending laterally from the main cracks. Many of the thin cracks were in random directions in a pattern known as map cracking.





Not only were there cracks in the plaster but a number of the ceramic wall tiles in the wainscots of bathrooms and toilet rooms were found to be broken, and in a number of places wall tiles were found to be loose from the backing of portland cement plaster on metal lath. Wall tiles were removed from the wainscot in one location, and equally spaced horizontal shrinkage cracks in the cement plaster base were observed in the height of the wainscot.

### 3. INSPECTION OF MAIN HOSPITAL BUILDING

The inspection of the Hospital Building to determine the condition of the plaster included visits to approximately sixty percent of the total number of rooms and all corridors. A careful inspection of the condition of the concrete of the exterior walls was also made to determine whether defects of concrete work might have contributed to the failure of the plaster. The plaster on metal lath for corridor and room partitions, without exception, was found to be cracked or had been cracked and repaired. A large proportion of the plaster surfaces had been painted, but cracking appeared to have continued subsequent to the application of the paint. There were few boxes for outlets for electrical services where cracks were not found to radiate from one or more corners. The general cracking was of the pattern known as map cracking. The spacing of cracks ranged from a few inches to about three feet.

Specimens of plaster for laboratory tests and analyses were cut from rooms and corridor as shown in table 1. The 15- by 15-inch specimens as cut from partitions for tests to determine the stability of the plaster when subjected to changes in relative humidity included the metal lath as well as the plaster. Those for analysis to determine the proportions of plaster and aggregates were approximately 10 inches square and did not include lath or plaster keys.

Not only were there cracks in the plaster but a number of the ceiling tiles in the hallway of the building were missing. The ceiling tiles were missing in the hallway of the building and in a number of places will also be found in the hallway of the building. The ceiling tiles were missing in the hallway of the building and in a number of places will also be found in the hallway of the building. The ceiling tiles were missing in the hallway of the building and in a number of places will also be found in the hallway of the building.

### 3. PROPERTIES OF THE BUILDING

The building is a three-story structure with a flat roof. The exterior walls are made of brick and are in good condition. The interior walls are made of plaster and are in poor condition. The ceiling is made of tiles and is in poor condition. The floor is made of concrete and is in good condition. The building is located in a residential area and is surrounded by other buildings. The building is owned by a private individual and is used as a residence. The building is in need of major repairs and renovation. The estimated cost of repairs and renovation is \$100,000. The building is a good investment and is expected to increase in value over time. The building is a good example of a well-maintained residential building. The building is a good example of a well-maintained residential building. The building is a good example of a well-maintained residential building.

Condition of plaster and masonry work was checked. The plaster was found to be in poor condition and was recommended for repair. The masonry work was found to be in good condition and was recommended for no further action. The condition of the plaster and masonry work was checked. The plaster was found to be in poor condition and was recommended for repair. The masonry work was found to be in good condition and was recommended for no further action. The condition of the plaster and masonry work was checked. The plaster was found to be in poor condition and was recommended for repair. The masonry work was found to be in good condition and was recommended for no further action.



Table 1. Source or location of specimens of lath and plaster for laboratory tests

Story	Wing	Room	Tests
Basement	South	Corridor	Analysis and stability
Do	do	12	Analysis
First	East	106	Analysis and stability
Do	West	182	Stability
Second	North	245	Analysis
Third	East	316	Analysis
Do	West	318	Stability
Fourth	South	438	Analysis
Fifth	West	504	Analysis
Do	South	523	Stability
Do	do	{Closet 523}	Stability

TABLE 1. Summary of results of the  
1954-55 Survey of the Fishery of the

Species	Number of Fish	Weight (lb)	Length (in)
Atlantic Salmon	15	14	22
Atlantic Salmon	10	11	18
Atlantic Salmon	12	13	20
Atlantic Salmon	11	12	19
Atlantic Salmon	13	14	21
Atlantic Salmon	14	15	22
Atlantic Salmon	15	16	23
Atlantic Salmon	16	17	24
Atlantic Salmon	17	18	25
Atlantic Salmon	18	19	26
Atlantic Salmon	19	20	27
Atlantic Salmon	20	21	28
Atlantic Salmon	21	22	29
Atlantic Salmon	22	23	30
Atlantic Salmon	23	24	31
Atlantic Salmon	24	25	32
Atlantic Salmon	25	26	33
Atlantic Salmon	26	27	34
Atlantic Salmon	27	28	35
Atlantic Salmon	28	29	36
Atlantic Salmon	29	30	37
Atlantic Salmon	30	31	38
Atlantic Salmon	31	32	39
Atlantic Salmon	32	33	40
Atlantic Salmon	33	34	41
Atlantic Salmon	34	35	42
Atlantic Salmon	35	36	43
Atlantic Salmon	36	37	44
Atlantic Salmon	37	38	45
Atlantic Salmon	38	39	46
Atlantic Salmon	39	40	47
Atlantic Salmon	40	41	48
Atlantic Salmon	41	42	49
Atlantic Salmon	42	43	50
Atlantic Salmon	43	44	51
Atlantic Salmon	44	45	52
Atlantic Salmon	45	46	53
Atlantic Salmon	46	47	54
Atlantic Salmon	47	48	55
Atlantic Salmon	48	49	56
Atlantic Salmon	49	50	57
Atlantic Salmon	50	51	58
Atlantic Salmon	51	52	59
Atlantic Salmon	52	53	60
Atlantic Salmon	53	54	61
Atlantic Salmon	54	55	62
Atlantic Salmon	55	56	63
Atlantic Salmon	56	57	64
Atlantic Salmon	57	58	65
Atlantic Salmon	58	59	66
Atlantic Salmon	59	60	67
Atlantic Salmon	60	61	68
Atlantic Salmon	61	62	69
Atlantic Salmon	62	63	70
Atlantic Salmon	63	64	71
Atlantic Salmon	64	65	72
Atlantic Salmon	65	66	73
Atlantic Salmon	66	67	74
Atlantic Salmon	67	68	75
Atlantic Salmon	68	69	76
Atlantic Salmon	69	70	77
Atlantic Salmon	70	71	78
Atlantic Salmon	71	72	79
Atlantic Salmon	72	73	80
Atlantic Salmon	73	74	81
Atlantic Salmon	74	75	82
Atlantic Salmon	75	76	83
Atlantic Salmon	76	77	84
Atlantic Salmon	77	78	85
Atlantic Salmon	78	79	86
Atlantic Salmon	79	80	87
Atlantic Salmon	80	81	88
Atlantic Salmon	81	82	89
Atlantic Salmon	82	83	90
Atlantic Salmon	83	84	91
Atlantic Salmon	84	85	92
Atlantic Salmon	85	86	93
Atlantic Salmon	86	87	94
Atlantic Salmon	87	88	95
Atlantic Salmon	88	89	96
Atlantic Salmon	89	90	97
Atlantic Salmon	90	91	98
Atlantic Salmon	91	92	99
Atlantic Salmon	92	93	100

#### 4. MATERIALS FOR AUXILIARY TESTS

Bags of materials were secured from the same sources from which plaster, lime, and aggregates for the building plaster came. The gypsum cement plaster, gauging plaster, and finishing lime were bought from Pioneer Sand and Gravel Co., 901 Fairview North, Seattle, Washington. The plaster supplied by this concern was from the Sigurd, Utah, plant of United States Gypsum Co., and the lime was "Miracle" brand from U. S. Lime Products Corporation, Henderson, Nevada. The vermiculite aggregate was from the Portland, Oregon, plant of Vermiculite Northwest, Spokane, Washington, and the sand was obtained by Mr. V. P. Reimer, Project Engineer, from Anchorage, Alaska, who also forwarded retained samples of plastering materials from his office.

#### 5. CHARACTERISTICS OF PLASTERS

Characteristics of the plasters and hydrated lime are given in tables 2 and 3. These materials conform to applicable Federal specifications. Analyses of plaster materials found in specimens of the set plasters are given in tables 4, 5, and 6. The proportions of gypsum cement and aggregates in the set-plaster samples were determined by dissolving the gypsum from weighed amounts of dried set plaster in ammonium acetate solution, then washing and weighing the residue after it had been dried at 70° C. The gypsum cement plaster was found to be 96.15 percent soluble; therefore, when the weight of the soluble portion of the sample of set plaster was divided by 0.96, the quotient gave the weight of the gypsum plaster from which the percent of gypsum plaster in the mix was computed. The ratio of sand to vermiculite in the aggregates was determined from the weights of the two constituents. It was found that the sand used in the plaster had a fairly definite proportion of a magnetic constituent. By the use of a magnet a large part of the magnetic material could be extracted from the mixture of aggregates. It was reasoned that the ratio of magnetic material extracted from the mixed aggregate to that of magnetic material extracted from the sand would be the ratio of sand to that of the combined aggregates. The proportions of sand and vermiculite and the number of cubic feet of each per 100-lb. bag of gypsum plaster were computed on that basis.







Table 2. Characteristics of gypsum plasters

[By Federal Specification SS-P-402, except as otherwise stated]

Plaster sample designation	Type	Time of set	Tensile strength	Compressive strength	CaSO <sub>4</sub> .1/2H <sub>2</sub> O
		hrs	psi	psi	
U. S. Gypsum, fibered <sup>1</sup>	N	24	190	---	90.9
U. S. Gypsum, not fibered <sup>2</sup>	N	10 1/2	225	---	90.3
U. S. Gypsum, medium set gauging <sup>2</sup>	G	2 1/4	540	---	93.0
U. S. Gypsum, fibered <sup>2</sup>	N	7 1/2	240	3 1190	92.8
Do (with Alaska sand)	N	5 1/4	185	3 1030	
Do (Fed. Spec. SS-P-401)	N	8	350	3 1930	
Do (Fed. Spec. SS-P-401, but with 150 g needle for consistency)	N	---	365	(avg. 2 lots) 3 1880	
U. S. Gypsum gauging <sup>2</sup>	G	2	540	2470	

<sup>1</sup> Sample of plaster approved for project.

<sup>2</sup> Sample purchased from dealer who supplied material for project. All of 25 g of gauging plaster passed through No. 100 sieve.

<sup>3</sup> The cubes of the four plaster mixes in this group exhibited volumetric shrinkage as follows: 2.0 percent; 0.6 percent; 2.2 percent (avg. of 2 lots); and 3.0 percent, respectively.

TABLE 2. Summary of results

(Data obtained from 1950-1951 field observations)

Year	Number of birds	Number of eggs	Number of chicks	Survival rate	Notes
1950	100	100	100	1.00	
1951	100	100	100	1.00	
1952	100	100	100	1.00	
1953	100	100	100	1.00	
1954	100	100	100	1.00	
1955	100	100	100	1.00	
1956	100	100	100	1.00	
1957	100	100	100	1.00	
1958	100	100	100	1.00	
1959	100	100	100	1.00	
1960	100	100	100	1.00	
1961	100	100	100	1.00	
1962	100	100	100	1.00	
1963	100	100	100	1.00	
1964	100	100	100	1.00	
1965	100	100	100	1.00	
1966	100	100	100	1.00	
1967	100	100	100	1.00	
1968	100	100	100	1.00	
1969	100	100	100	1.00	
1970	100	100	100	1.00	
1971	100	100	100	1.00	
1972	100	100	100	1.00	
1973	100	100	100	1.00	
1974	100	100	100	1.00	
1975	100	100	100	1.00	
1976	100	100	100	1.00	
1977	100	100	100	1.00	
1978	100	100	100	1.00	
1979	100	100	100	1.00	
1980	100	100	100	1.00	
1981	100	100	100	1.00	
1982	100	100	100	1.00	
1983	100	100	100	1.00	
1984	100	100	100	1.00	
1985	100	100	100	1.00	
1986	100	100	100	1.00	
1987	100	100	100	1.00	
1988	100	100	100	1.00	
1989	100	100	100	1.00	
1990	100	100	100	1.00	
1991	100	100	100	1.00	
1992	100	100	100	1.00	
1993	100	100	100	1.00	
1994	100	100	100	1.00	
1995	100	100	100	1.00	
1996	100	100	100	1.00	
1997	100	100	100	1.00	
1998	100	100	100	1.00	
1999	100	100	100	1.00	
2000	100	100	100	1.00	

✓ Data of birds banded in 1950.  
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 ✓ Data of birds banded in 2000.

Age, Alaska

	CO <sub>2</sub> (non-volatile basis)	MgO ÷ CaO + MgO	Total MgO hydrated	CaO + MgO unhydrated	autoclave expansion
0.1	0.70	40.9	97.3	0.61	---
21					
(3)					
(4)	94	40.5	97.9	.62	---
	2.59	39.6	97.3	.78	0.03

✓





Table 3. Characteristics of hydrate lime used in white-cast plaster for Alaska Native Service projects, 1960-1962, Alaska (basis: room temperature)

Sample designation	Fineness through		Elasticity	Oxide content				Compound composition					Loss on drying (nonvolatile basis)	Loss on ignition (nonvolatile basis)	CaO ÷ MgO	Total MgO hydrated	CaO ÷ MgO unhydrated	Autoclave expansion	
	#20	#100		CO <sub>2</sub>	H <sub>2</sub> O	SiO <sub>2</sub> + R <sub>2</sub> O <sub>3</sub>	CaO	MgO	CaCO <sub>3</sub>	Ca(OH) <sub>2</sub>	Mg(OH) <sub>2</sub>	MgO							SiO <sub>2</sub> + R <sub>2</sub> O <sub>3</sub>
U.S. Lime Products Corp. "Miracle" colomitic, prescribed (small can)	100.0	95.1	500	0.50	26.47	1.85	42.87	24.68	1.14	55.80	41.77	0.81	0.85	99.3	0.70	40.9	97.3	0.81	---
Do (medium can)	100.0	93.2	530	.69	26.51	.83	42.71	24.44	1.57	56.66	41.70	.62	.63	99.0	.94	42.5	97.9	.62	---
Do ✓	100.0	93.2	430	1.85	25.50	.69	43.32	24.39	4.21	54.12	38.95	.78	.69	98.7	2.50	38.6	97.3	.78	0.03

↓ Lime purchased from dealer who supplied materials for the project. All other samples were retained approved samples from the project.

MEMORANDUM FOR THE RECORD

DATE: 10/15/54

NO.	NAME	ADDRESS	CITY	STATE	ZIP	REMARKS
1	Mr. J. H. Smith	123 Main St.	Chicago	Ill.	60601	
2	Mr. W. R. Jones	456 Elm St.	New York	N.Y.	10001	
3	Mr. T. G. White	789 Oak St.	Los Angeles	Calif.	90001	
4	Mr. S. L. Black	101 Pine St.	San Francisco	Calif.	94101	
5	Mr. K. M. Green	202 Cedar St.	Philadelphia	Penn.	19101	
6	Mr. N. P. Brown	303 Birch St.	Washington	D.C.	20001	
7	Mr. Q. R. Gold	404 Spruce St.	Portland	Ore.	97201	
8	Mr. U. S. Silver	505 Ash St.	Seattle	Wash.	98101	
9	Mr. V. T. Lead	606 Hickory St.	Denver	Colo.	80201	
10	Mr. X. Y. Zinc	707 Walnut St.	St. Louis	Mo.	63101	
11	Mr. Z. W. Tin	808 Chestnut St.	Cincinnati	Ohio	45201	
12	Mr. A. B. Copper	909 Locust St.	Kansas City	Mo.	64101	
13	Mr. C. D. Iron	1010 Olive St.	St. Paul	Minn.	55101	
14	Mr. E. F. Nickel	1111 Madison St.	Minneapolis	Minn.	55401	
15	Mr. G. H. Cobalt	1212 Broadway	New York	N.Y.	10001	
16	Mr. I. J. Cadmium	1313 Wall St.	New York	N.Y.	10001	
17	Mr. K. L. Mercury	1414 Nassau St.	New York	N.Y.	10001	
18	Mr. M. N. Bismuth	1515 Broadway	New York	N.Y.	10001	
19	Mr. O. P. Antimony	1616 Broadway	New York	N.Y.	10001	
20	Mr. Q. R. Arsenic	1717 Broadway	New York	N.Y.	10001	
21	Mr. S. T. Selenium	1818 Broadway	New York	N.Y.	10001	
22	Mr. U. V. Tellurium	1919 Broadway	New York	N.Y.	10001	
23	Mr. W. X. Polonium	2020 Broadway	New York	N.Y.	10001	
24	Mr. Y. Z. Astatine	2121 Broadway	New York	N.Y.	10001	
25	Mr. A. B. Francium	2222 Broadway	New York	N.Y.	10001	
26	Mr. C. D. Radium	2323 Broadway	New York	N.Y.	10001	
27	Mr. E. F. Actinium	2424 Broadway	New York	N.Y.	10001	
28	Mr. G. H. Thorium	2525 Broadway	New York	N.Y.	10001	
29	Mr. I. J. Protactinium	2626 Broadway	New York	N.Y.	10001	
30	Mr. K. L. Uranium	2727 Broadway	New York	N.Y.	10001	
31	Mr. M. N. Neptunium	2828 Broadway	New York	N.Y.	10001	
32	Mr. O. P. Plutonium	2929 Broadway	New York	N.Y.	10001	
33	Mr. Q. R. Americium	3030 Broadway	New York	N.Y.	10001	
34	Mr. S. T. Curium	3131 Broadway	New York	N.Y.	10001	
35	Mr. U. V. Berkelium	3232 Broadway	New York	N.Y.	10001	
36	Mr. W. X. Californium	3333 Broadway	New York	N.Y.	10001	
37	Mr. Y. Z. Einsteinium	3434 Broadway	New York	N.Y.	10001	
38	Mr. A. B. Fermium	3535 Broadway	New York	N.Y.	10001	
39	Mr. C. D. Mendelevium	3636 Broadway	New York	N.Y.	10001	
40	Mr. E. F. Nobelium	3737 Broadway	New York	N.Y.	10001	
41	Mr. G. H. Lawrencium	3838 Broadway	New York	N.Y.	10001	
42	Mr. I. J. Rutherfordium	3939 Broadway	New York	N.Y.	10001	
43	Mr. K. L. Dubnium	4040 Broadway	New York	N.Y.	10001	
44	Mr. M. N. Seaborgium	4141 Broadway	New York	N.Y.	10001	
45	Mr. O. P. Bohrium	4242 Broadway	New York	N.Y.	10001	
46	Mr. Q. R. Hassium	4343 Broadway	New York	N.Y.	10001	
47	Mr. S. T. Meitnerium	4444 Broadway	New York	N.Y.	10001	
48	Mr. U. V. Darmstadtium	4545 Broadway	New York	N.Y.	10001	
49	Mr. W. X. Roentgenium	4646 Broadway	New York	N.Y.	10001	
50	Mr. Y. Z. Copernicium	4747 Broadway	New York	N.Y.	10001	

Prepared by: J. H. Smith

Table 4. Analysis of base-coat plaster for proportion of aggregates by ammonium acetate method  
(Basis: Material dried at 50° C)

Sample designation	Weight of sample	Total insoluble	Total insoluble	Magnetic content extracted
	g	g	%	g
U.S. G. "Red Top" fibered gypsum plaster <sup>1</sup>	3.9945	0.1490	3.73	---
Do	5.0030	.1925	3.85	---
Do	5.0037	.1948	3.89	---
Do	5.0032	.1992	3.98	---
Do	5.0033	.1901	3.80	---
			(Avg. 3.85)	
Vermiculite <sup>2</sup>	4.9975	5.0072	100.19 (Soluble: none)	0.0000
Alaska sand <sup>3</sup>	19.9500	19.8635	99.57 (Soluble: 0.43)	.7920 (Avg. of 6 determinations)
Base-coat plaster <sup>4</sup>				
B hall	9.9905	4.6338	46.78	0.1565
Do	10.0025	4.5300	45.29	.1580
Average	9.9965	4.5819	46.04	0.1568
N 12	9.9875	3.2155	32.20	.0318
Do	10.0010	3.4143	34.14	.0320
Average	9.9943	3.3149	33.17	0.0319
E 106	9.9860	2.5108	25.14	---
Do	10.0005	2.6432	26.43	0.0000
Do	10.0015	2.6422	26.42	.0000
Average	9.9996	2.5991	26.00	0.0000
N 245	9.9900	3.4605	34.64	0.0500
Do	10.0007	3.5407	35.40	.0375
Average	9.9954	3.5006	35.02	0.0478
E 316	9.9925	5.4580	54.99	0.1240
Do	10.0005	4.5940	45.94	.1250
Do	10.0000	4.6195	46.20	.1265
Average	9.9978	4.8905	49.04	0.1252
S 438	9.9900	4.4018	44.06	0.0915
Do	10.0002	4.4860	44.86	.0915
Average	9.9951	4.4439	44.46	0.0915
W 504	9.9800	3.4795	34.86	0.0650
Do	10.0005	3.7210	37.21	.0560
Average	9.9903	3.6003	36.04	0.0605

<sup>1</sup> Sample of plaster from dealer who supplied plaster for project.

<sup>2</sup> Sample of vermiculite aggregate from Vermiculite Northwest who supplied aggregate for project.

<sup>3</sup> Sample of sand supplied by Project Engineer as representative of material used on project.

<sup>4</sup> Base-coat plaster samples taken from walls of hospital building. Letters signify, respectively: B, basement; N, North; E, East; S, South; W, West, and figures denote the room number as given on architectural plans. The proportion of aggregates were determined from analyses of these samples.







e, Alaska

Sample designation	gO e. ed	Ratio, line putty to CaSO <sub>4</sub> .1/2H <sub>2</sub> O in original mix	Apparent MgO in lime used	Thickness of the white coat
B hall	11	3.5+	41.4	<u>in.</u> < 1/64 - 1/16
N 12	10	0.95+	40.0	1/16 - 3/32
E 106	11	2.0-	38.7	1/32 - 1/16
N 245	10	1.3-	✓ 47.9	1/32
E 316	10	3.3-	39.6	1/32 - 1/16
S 438	12	1.4-	39.8	1/16
W 504	4	0.81-	✓ 51.7	< 1/64 - 1/16

1/ Letters signify

2/ Samples designated from the base coat on the assumption that the white coat consists of mineral containing a considerable percentage of MgO) nothing other than

3/ The excess H<sub>2</sub>O

12. Medicines for the treatment of the following diseases  
 (Listed in the order of their importance)  
 (Listed in the order of their importance)

Medicine	Indication	Dose	Frequency	Remarks
1. Aspirin	Headache, fever	0.5g	4 times a day	General analgesic and antipyretic
2. Paracetamol	Headache, fever	0.5g	4 times a day	Analgesic and antipyretic
3. Ibuprofen	Headache, fever, inflammation	0.2g	4 times a day	Analgesic, antipyretic, anti-inflammatory
4. Diclofenac	Headache, fever, inflammation	0.1g	4 times a day	Analgesic, antipyretic, anti-inflammatory
5. Morphine	Severe pain	2-5mg	As needed	Strong analgesic
6. Codeine	Moderate to severe pain	15-30mg	4 times a day	Mild analgesic
7. Tramadol	Moderate to severe pain	50-100mg	4 times a day	Mild analgesic
8. Paracetamol + Codeine	Moderate to severe pain	500mg + 30mg	4 times a day	Mild analgesic
9. Paracetamol + Ibuprofen	Moderate to severe pain	500mg + 200mg	4 times a day	Mild analgesic
10. Paracetamol + Diclofenac	Moderate to severe pain	500mg + 75mg	4 times a day	Mild analgesic
11. Paracetamol + Morphine	Moderate to severe pain	500mg + 2-5mg	4 times a day	Mild analgesic
12. Paracetamol + Codeine + Morphine	Moderate to severe pain	500mg + 30mg + 2-5mg	4 times a day	Mild analgesic
13. Paracetamol + Codeine + Ibuprofen	Moderate to severe pain	500mg + 30mg + 200mg	4 times a day	Mild analgesic
14. Paracetamol + Codeine + Diclofenac	Moderate to severe pain	500mg + 30mg + 75mg	4 times a day	Mild analgesic
15. Paracetamol + Codeine + Morphine + Ibuprofen	Moderate to severe pain	500mg + 30mg + 2-5mg + 200mg	4 times a day	Mild analgesic
16. Paracetamol + Codeine + Morphine + Diclofenac	Moderate to severe pain	500mg + 30mg + 2-5mg + 75mg	4 times a day	Mild analgesic
17. Paracetamol + Codeine + Morphine + Ibuprofen + Diclofenac	Moderate to severe pain	500mg + 30mg + 2-5mg + 200mg + 75mg	4 times a day	Mild analgesic

Table 4. Analysis of base-coat plaster for proportion of aggregates by ammonium acetate method  
(Basis: Material dried at 50° C)

Sample designation	Weight of sample	Total insoluble	Total insoluble	Magnetic content extracted
U. S. C. "Red Ton" fibered gypsum plaster <sup>1</sup>	g	g	%	g
	3.9945	0.1490	3.73	---
Do	5.0030	.1925	3.85	---
Do	5.0037	.1948	3.89	---
Do	5.0032	.1992	3.98	---
Do	5.0033	.1901	3.80	---
			(Ave. 3.85)	
Vermiculite <sup>2</sup>	4.9975	5.0072	100.19 (Soluble: none)	0.0000
Alaska sand <sup>3</sup>	19.9500	19.8635	99.57 (Soluble: 0.43)	.7920 (Avg. of 6 determinations)
Base-coat plaster <sup>4</sup>				
B hall	9.9905	4.6335	46.78	0.1565
Do	10.0025	4.5300	45.29	.1580
Average	9.9965	4.5819	46.04	0.1568
N 12	9.9875	3.2155	32.20	.0318
Do	10.0010	3.4143	34.14	.0320
Average	9.9943	3.3149	33.17	0.0319
N 106	9.9860	2.5108	25.14	---
Do	10.0005	2.6432	26.43	0.0000
Do	10.0015	2.6422	26.42	.0000
Average	9.9996	2.5991	26.00	0.0000
N 245	9.9900	3.4605	34.64	0.0500
Do	10.0007	3.5107	35.10	.0775
Average	9.9954	3.5006	35.02	0.0478
E 316	9.9925	5.4580	54.99	0.1240
Do	10.0005	4.5940	45.94	.1250
Do	10.0000	4.6195	46.20	.1265
Average	9.9978	4.8775	47.04	0.1252
S 438	9.9900	4.4018	44.06	0.0915
Do	10.0002	4.4860	44.86	.0915
Average	9.9951	4.4439	44.46	0.0915
W 504	9.9800	3.4795	34.86	0.0650
Do	10.0005	3.7210	37.21	.0560
Average	9.9903	3.6003	36.04	0.0605

<sup>1</sup> Sample of plaster from dealer who supplied plaster for project.

<sup>2</sup> Sample of vermiculite aggregate from Vermiculite Northwest who supplied aggregate for project.

<sup>3</sup> Sample of sand supplied by Project Engineer as representative of material used on project.

<sup>4</sup> Base-coat plaster samples taken from walls of hospital building. Letters signify, respectively: B, basement; N, North; E, East; S, South; W, West, and figures denote the room number as given on architectural plans. The proportion of aggregates were determined from analyses of these samples.

Table 5. Analyses of white-coat plaster finish from walls of Alaska Native Service Hospital, Barrow, Alaska  
(basis: material dried at 50°C)

Sample designation	oxide content						compound composition							Total MgO of lime hydrated	ratio, lime putty to CaSO <sub>4</sub> .1/2H <sub>2</sub> O in original mix	apparent MgO in lime used	Thickness of the white coat
	SiO <sub>2</sub>	Fe <sub>2</sub> O <sub>3</sub> (combine.)	Al <sub>2</sub> O <sub>3</sub>	CaO	MgO	SiO <sub>2</sub>	CaSO <sub>4</sub> .2H <sub>2</sub> O	CaCO <sub>3</sub>	Ca(OH) <sub>2</sub>	Mg(OH) <sub>2</sub>	MgO	H <sub>2</sub> O excess	SiO <sub>2</sub> + H <sub>2</sub> O <sub>3</sub>				
B hall	11.87	21.16	1.12	35.62	18.02	12.49	26.86	26.99	15.52	26.07	3.72	0.00	1.12	79.4	3.5+	41.4	< 1/64 - 1/16
N 12	10.02	16.47	1.21	34.95	10.57	27.23	59.55	22.79	4.11	10.42	3.37	0.00	1.21	68.1	0.95+	40.0	1/16 - 3/32
E 106	11.52	17.54	0.60	36.87	14.89	14.99	40.84	26.20	11.74	19.88	1.15	0.00	0.60	92.3	2.0-	38.7	1/32 - 1/16
N 245	10.02	18.27	0.61	32.70	14.83	23.69	50.94	22.79	4.41	21.16	0.20	0.00	0.61	98.7	1.3-	<sup>2</sup> 47.9	1/32
E 316	10.62	23.34	2.18	34.58	16.86	12.70	27.31	24.15	16.06	24.39	0.00	<sup>3</sup> 6.19	<sup>3</sup> 2.18	100.0	3.3-	39.6	1/32 - 1/16
S 438	12.22	17.47	1.09	34.51	12.44	22.38	48.13	27.27	5.00	18.00	0.00	<sup>3</sup> 0.62	<sup>3</sup> 1.09	100.0	1.4-	39.8	1/16
W 504	4.41	21.01	0.70	32.24	12.77	29.03	62.43	10.03	8.31	18.47	0.00	<sup>3</sup> .22	<sup>3</sup> 0.70	100.0	0.81-	<sup>2</sup> 51.7	< 1/64 - 1/16

<sup>1</sup> Letters signify, respectively: B, basement; N, north; E, east; S, south; W, west, and figures denote the room number as given on architectural plans.

<sup>2</sup> Samples designated N 245 and W 504, respectively, apparently show abnormally high percentages of MgO in the lime used. Possibly some vermiculite (a mineral containing a considerable percentage of MgO) from the base coat was carried mechanically into the white coat during the trowelling. If so, this would upset the validity of the calculations based on the assumption that the white coat consists of nothing other than lime and gypsum.

<sup>3</sup> The excess H<sub>2</sub>O may be attributed to unidentified minerals under SiO<sub>2</sub> + H<sub>2</sub>O<sub>3</sub>.



Table 6. Plaster aggregates in set gypsum plaster

(Computed from weights given in table 4)

Sample designation	Proportion of sand in aggregate	Aggregates per 100-lb bag of gypsum cement plaster	
		Vermiculite (10 lb/ft <sup>3</sup> )	Sand (100 lb/ft <sup>3</sup> )
B hall	95.0	1.0	0.70
B 12	36.2	3.2	.12
E 106	0.0	2.0	.00
N 245	35.0	3.1	.17
E 316	46.5	3.9	.50
S 438	73.0	2.0	.53
W 504	45.3	2.7	.23

Approved for Release by NSA on 05-08-2014 pursuant to E.O. 13526

TABLE 1. *Estimated values of the parameters of the model*

Year	Estimated values of the parameters		Standard error	t-value
	$\alpha$	$\beta$		
1970	0.15	0.85	0.05	3.00
1971	0.16	0.84	0.05	3.20
1972	0.17	0.83	0.05	3.40
1973	0.18	0.82	0.05	3.60
1974	0.19	0.81	0.05	3.80
1975	0.20	0.80	0.05	4.00
1976	0.21	0.79	0.05	4.20

## 6. STABILITY OF LATH AND PLASTER FROM THE HOSPITAL BUILDING

Five of the 15- by 15-inch specimens of lath and plaster cut from partitions of the basement, first, third, and fifth stories of the hospital building were sawn into 3 inch wide strips and subjected alternately to high and low or low and high humidity at nearly constant temperature. Changes in length between gage points 10 inches apart on 1/4-inch diameter thin brass discs cemented to the front and back surfaces of the plaster were measured by means of an optical comparator which was calibrated against a 10-inch invar standard bar at the time each set of readings was made. The changes in length shown by the differences of readings of the comparator indicated an initial shrinkage of many specimens followed by expansion when placed in an atmosphere having high humidity and shrinkage when in a container having low humidity.

The changes of length of the specimens in terms of percentage of the 10-inch gage length are shown in figures 1 and 2. These are typical of the behavior of the twenty-two specimens subjected to the tests. In some instances there appeared to be evidence that the expansion of the plaster caused flexing of the lath, thus producing marked or anomalous differences in the lengthening or shortening of the gage length of the specimens during the period in which it was subjected to a constant humidity. Some of these anomalies have been attributed to cracking of the base-coat plaster or both the base-coat plaster and the white-coat finish, thus allowing the metal lath, already strained by the shrinking plaster, to regain its length to the extent that the strain from shrinkage of the plaster had been relieved.





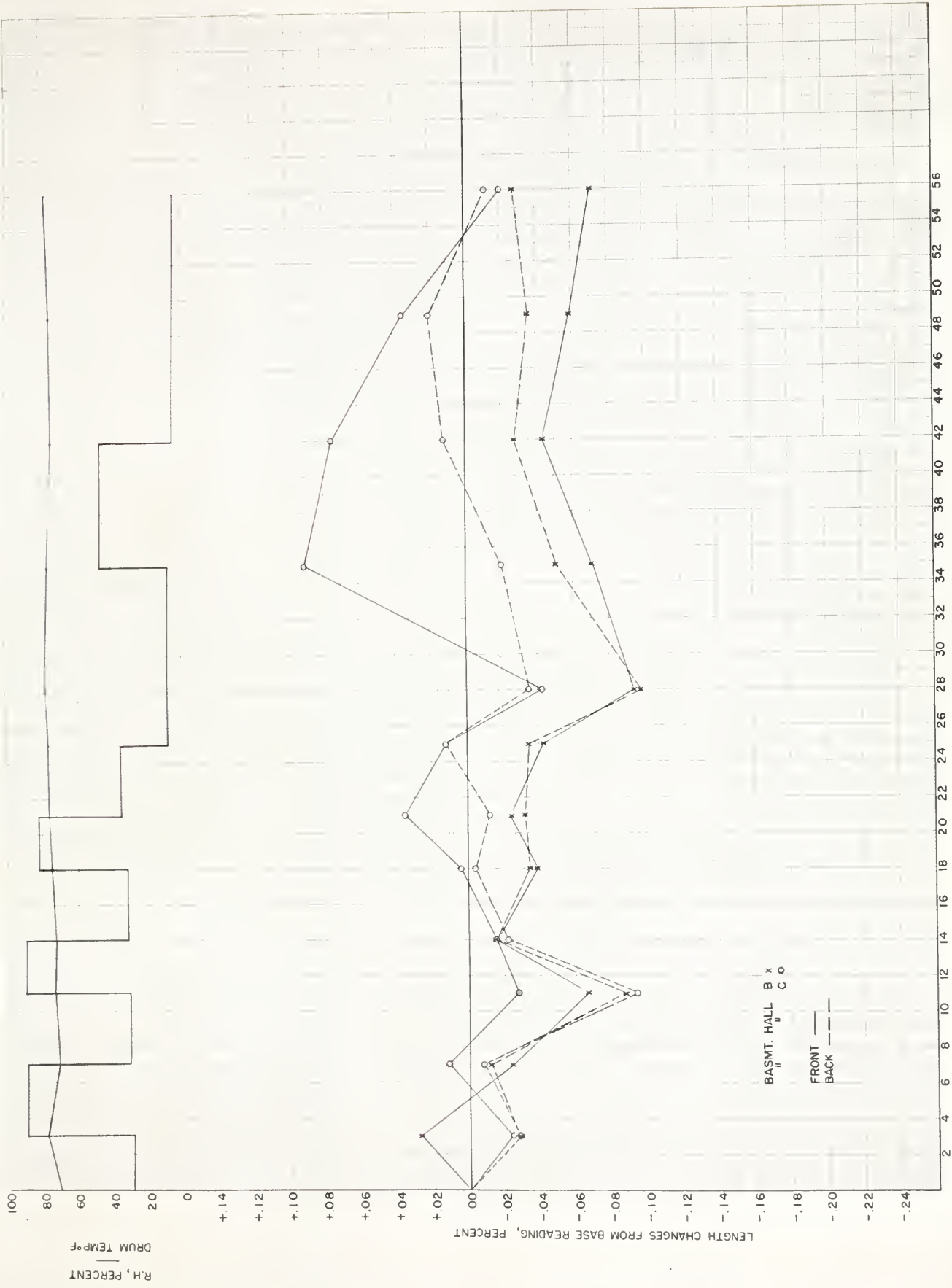
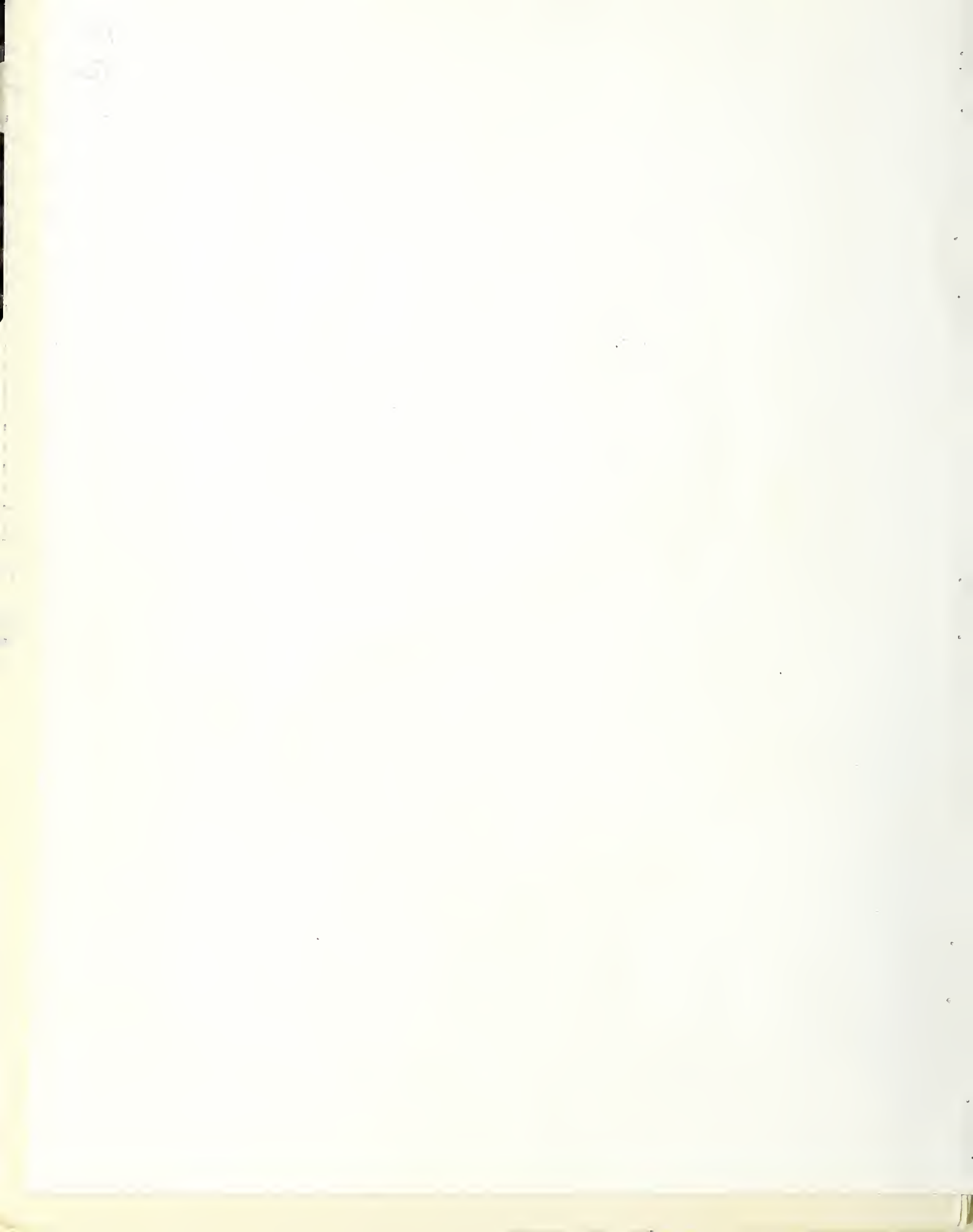


Figure 1. Changes in length of specimens of lath and plaster from basement corridor in percentage of a 10-in. gage length when subjected to changes in humidity.



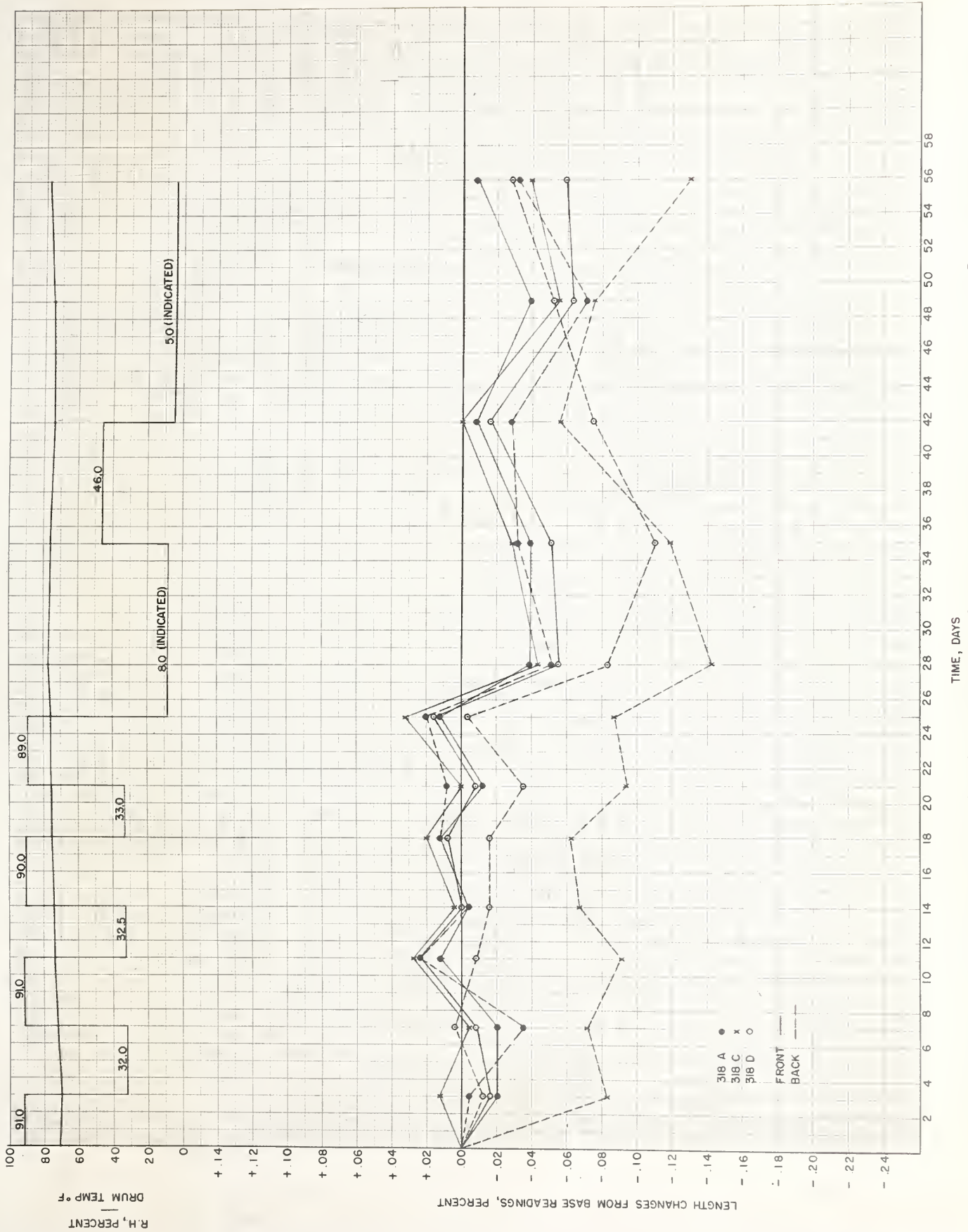


Figure 2. Changes in length of specimens of lath and plaster from room 318 in percentage of a 10-in. gage length when subjected to changes in humidity.





## 7. BEHAVIOR OF MOLDED PLASTER BARS

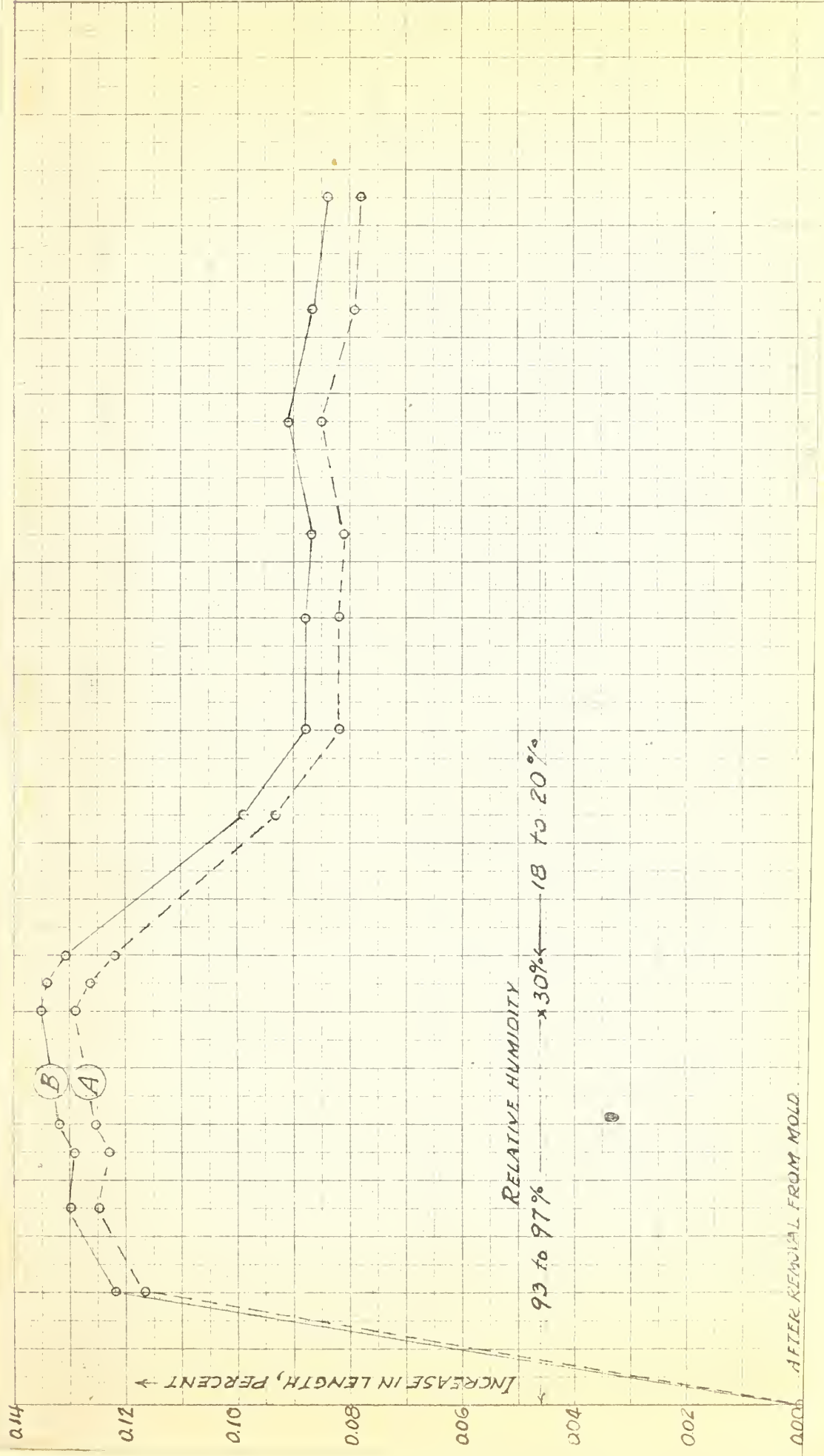
Two test bars of plaster designated A and B were cast in steel molds in order to study the behavior of the typical plaster mix when subjected to changing conditions of humidity. The bars were cast in molds to give an effective length of 10 inches between steel plugs cast into the ends of the bars. The bars were made from plaster having the composition: 20 parts fibered-gypsum cement, 8 parts vermiculite, and 5 parts Alaska sand by weight. Bar B had 10 percent more water in the mix than did bar A. The 1- by 7/8-inch bars had a 1/16 inch thick white finish applied to one face after they had been removed from the molds and measured for expansion. The changes in length of the two bars are shown in figure 3. On release from the constraint of the molds, the bars expanded 0.02 and 0.03 percent of their length at once, and after having lain horizontally without other restraint than their own weight for 4 days in a chamber at 70° F and at 93 to 97 percent relative humidity, the bars were found to have increased in length by 0.117 and 0.122 percent. During the next 10 days, at the same humidities, increases of 0.012 and 0.013 percent in length were observed, making the total expansion from the beginning of storage at high humidity 0.129 and 0.135 percent, respectively. The bars were then placed in a chamber held at 70° F and at 30 to 40 percent relative humidity for 2 days. The shrinkage during these 2 days amounted to not more than 0.007 percent. During the next 8 days, however, after storage in the chamber held at 18 to 20 percent relative humidity, the shrinkage amounted to 0.040 and 0.043 percent for specimens A and B, respectively. The shrinkage during the next 19 days, under the same humidity conditions, amounted to 0.004 percent. From the point of maximum expansion the shrinkage at the end of 29 days' storage in chambers in which the relative humidity was lowered by stages to 18 percent had reached 0.052 percent and appeared to be continuing.

The first part of the report discusses the changes in the composition of the total assets of the Federal Reserve banks during the period from 1913 to 1933. It shows that the total assets of the Federal Reserve banks increased from \$1,000,000,000 in 1913 to \$10,000,000,000 in 1933. This increase was due to a number of factors, including the increase in the amount of currency in circulation, the increase in the amount of government securities held, and the increase in the amount of other assets held.

The second part of the report discusses the changes in the composition of the liabilities of the Federal Reserve banks during the same period. It shows that the total liabilities of the Federal Reserve banks increased from \$1,000,000,000 in 1913 to \$10,000,000,000 in 1933. This increase was due to a number of factors, including the increase in the amount of currency in circulation, the increase in the amount of government securities held, and the increase in the amount of other liabilities.

The third part of the report discusses the changes in the composition of the assets and liabilities of the Federal Reserve banks during the period from 1913 to 1933. It shows that the total assets and liabilities of the Federal Reserve banks increased from \$1,000,000,000 in 1913 to \$10,000,000,000 in 1933. This increase was due to a number of factors, including the increase in the amount of currency in circulation, the increase in the amount of government securities held, and the increase in the amount of other assets and liabilities.

The fourth part of the report discusses the changes in the composition of the assets and liabilities of the Federal Reserve banks during the period from 1913 to 1933. It shows that the total assets and liabilities of the Federal Reserve banks increased from \$1,000,000,000 in 1913 to \$10,000,000,000 in 1933. This increase was due to a number of factors, including the increase in the amount of currency in circulation, the increase in the amount of government securities held, and the increase in the amount of other assets and liabilities.







## 8. FLEXURE OF LATH AND PLASTER

Twelve specimens for tests to determine the amount of flexure of metal lath and plaster resulting from expansion or shrinkage of the plaster under different conditions of humidity were prepared. These were made by attaching 5 1/2 inch wide by 36 inch long strips of 3.4 lb. expanded metal lath to 1/4-inch diameter steel uprights mounted on a 6 by 1/8 by 36 inch long steel base plate. The uprights were spaced 16 inches on centers in two rows of three each. Two of the uprights of each row were welded in place while those at one end were fitted into holes in the base, so as to be removable. The lath strips were wired to the uprights so that one end of the lath would be free to flex after the removable upright had been removed.

Each coat of the base-coat plaster applied was made up of 20 lb. gypsum cement plaster, 8 lb. vermiculite, and 5 lb. of Alaska sand. After the batch had been well mixed it was halved. To one half, 14 lb. of water was added to give a relatively stiff mixture; to the other, 15 1/2 lb. of water was added to produce a comparatively soft mixture. The plaster coats were applied to the lath strips on two successive days to 11/16-inch thickness as determined by wood grounds, the drier mix being applied to the lath on the right-hand side of the base. After storage under wet burlap overnight, the 12 specimens were transferred, four to each of three humidity chambers. The relative humidities in the three chambers into which the specimens were placed were 52, 72, and 92 percent.

The large amount of water in the specimens increased for a time the relative humidity above those fixed by the saturated salt solutions intended to keep the humidity at a constant level. After 3 days' storage of the specimens, a white-coat finish consisting of 4 parts lime putty to 1 part gauging plaster was applied to the set base-coat plasters to thicknesses ranging from 1/16 to 1/8 inches. The specimens were returned to the humidity chambers as soon as the white-coat finish had been applied.



In the 60-hour interval between the application of the brown-coat plaster and the white-coat finish, the free ends of the metal lath strips made with the drier mix of plaster were deflected from their original positions by expansion of the plaster an average of 0.16 inch and those to which the wetter plaster was applied an average of 0.18 inch. After the application of the white-coat plaster the specimens continued to expand, thus causing increased deflections. A pair of specimens having plaster of the two degrees of wetting continued to deflect after 46 days in the chamber with high humidity. The specimen with the drier mix, and without white coat, has deflected 0.47 inch and the specimen having the wetter mix, with white coat, has deflected 0.14 inch. When placed in chambers conditioned to low humidity, the plaster shrank, thus causing the strips to be deflected in the opposite direction. The four specimens which have been in a chamber, the relative humidity of which has been lowered from 55 to 18 percent, have shrunk to cause deflections of the free ends ranging from 0.197 to 0.276 inch. The shrinkage continues.







## 9. CONCLUSIONS

From the observations made during the inspections of the buildings and the subsequent as well as previous laboratory tests which indicated a high degree of dimensional instability of the plaster when subjected to the range of relative humidities not in excess of those encountered in buildings, it has been concluded that the cracking has resulted from this instability. The relative effects of shrinkage of the fibered gypsum cement plaster as revealed by the volumetric shrinkage of cubes of the neat and sanded plasters, items 4, 5, and 6, table 2, and the expansion and contraction of the plaster bars having both sand and vermiculite as the aggregate in causing the plaster to crack have not yet been determined. Doubtless both effects have played a part in causing cracks in the plaster. The measurements of change of length of specimens of the set plaster on metal lath by means of the comparator have shown that when subjected to large variations in relative humidity the base-coat plasters usually crack, and in a few cases the white-coat finish cracked also. Some of the 3 inch wide strips cut from the 15- by 15-inch specimens of lath and plaster taken from partitions in the hospital building exhibited decided curvature, which increased as the tests proceeded, indicating that the strength of the plaster was sufficient to cause progressive bending of the expanded metal lath. Some of these strips cracked during the tests, presumably when the strength of the plaster was not sufficient to cause further bending of the metal lath.

Although the studies have not been completed, we can, with reasonable assurance, say that the cracking of the plaster has resulted from dimensional instability of the combination of lath and weak plaster. Dimensional changes have doubtless been aggravated by the humidity conditions prevailing within the building during and subsequent to the plastering operations.

It is quite probable that cracking of the plaster on the metal lath partitions will continue through several seasons unless some process for stabilizing the plaster against excessive expansion and contraction with the changes in humidity to which it is normally subjected can be applied.

None of the contents of this report are released for publication or for use in sales promotion.

For the Director

Nolan D. Mitchell  
Consultant, Building Technology Division

The following information was obtained from the investigation of the activities of the Communist Party, U.S.A., in the State of New York, during the period from 1945 to 1950. The information was obtained from the files of the State Security Council, New York, and is being furnished to you for your information.

The Communist Party, U.S.A., is a political party which is organized and controlled by the Communist Party, U.S.S.R., and the Communist Party, C.P.S.U. The Communist Party, U.S.A., is a political party which is organized and controlled by the Communist Party, U.S.S.R., and the Communist Party, C.P.S.U. The Communist Party, U.S.A., is a political party which is organized and controlled by the Communist Party, U.S.S.R., and the Communist Party, C.P.S.U.

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For the Director

William J. Brennan  
 Chairman, Select Committee on Assassinations

**STAPLES**

