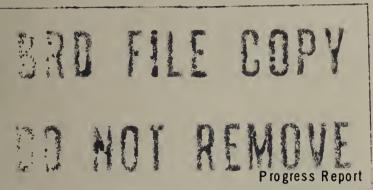
NATIONAL BUREAU OF STANDARDS REPORT

10 075



April 1 through June 30, 1969

DEVELOPMENT OF METHODS OF TEST
FOR QUALITY CONTROL OF PORCELAIN ENAMELS



U.S. DEPARTMENT OF COMMERCE NATIONAL BUREAU OF STANDARDS

NATIONAL BUREAU OF STANDARDS

The National Bureau of Standards was established by an act of Congress March 3, 1901. Today, in addition to serving as the Nation's central measurement laboratory, the Bureau is a principal focal point in the Federal Government for assuring maximum application of the physical and engineering sciences to the advancement of technology in industry and commerce. To this end the Bureau conducts research and provides central national services in four broad program areas. These are: (1) basic measurements and standards, (2) materials measurements and standards, (3) technological measurements and standards, and (4) transfer of technology.

The Bureau comprises the Institute for Basic Standards, the Institute for Materials Research, the Institute for Applied Technology, the Center for Radiation Research, the Center for Computer Sciences and Technology, and the Office for Information Programs.

THE INSTITUTE FOR BASIC STANDARDS provides the central basis within the United States of a complete and consistent system of physical measurement; coordinates that system with measurement systems of other nations; and furnishes essential services leading to accurate and uniform physical measurements throughout the Nation's scientific community, industry, and commerce. The Institute consists of an Office of Measurement Services and the following technical divisions:

Applied Mathematics—Electricity—Metrology—Mechanics—Heat—Atomic and Molecular Physics—Radio Physics ²—Radio Engineering ²—Time and Frequency ²—Astrophysics ²—Cryogenics.²

THE INSTITUTE FOR MATERIALS RESEARCH conducts materials research leading to improved methods of measurement standards, and data on the properties of well-characterized materials needed by industry, commerce, educational institutions, and Government; develops, produces, and distributes standard reference materials; relates the physical and chemical properties of materials to their behavior and their interaction with their environments; and provides advisory and research services to other Government agencies. The Institute consists of an Office of Standard Reference Materials and the following divisions:

Analytical Chemistry—Polymers—Metallurgy—Inorganic Materials—Physical Chemistry. THE INSTITUTE FOR APPLIED TECHNOLOGY provides technical services to promote the use of available technology and to facilitate technological innovation in industry and Government; cooperates with public and private organizations in the development of technological standards, and test methodologies; and provides advisory and research services for Federal, state, and local government agencies. The Institute consists of the following technical divisions and offices:

Engineering Standards—Weights and Measures — Invention and Innovation — Vehicle Systems Research—Product Evaluation—Building Research—Instrument Shops—Measurement Engineering—Electronic Technology—Technical Analysis.

THE CENTER FOR RADIATION RESEARCH engages in research, measurement, and application of radiation to the solution of Bureau mission problems and the problems of other agencies and institutions. The Center consists of the following divisions:

Reactor Radiation—Linac Radiation—Nuclear Radiation—Applied Radiation.

THE CENTER FOR COMPUTER SCIENCES AND TECHNOLOGY conducts research and provides technical services designed to aid Government agencies in the selection, acquisition, and effective use of automatic data processing equipment; and serves as the principal focus for the development of Federal standards for automatic data processing equipment, techniques, and computer languages. The Center consists of the following offices and divisions:

Information Processing Standards—Computer Information — Computer Services — Systems Development—Information Processing Technology.

THE OFFICE FOR INFORMATION PROGRAMS promotes optimum dissemination and accessibility of scientific information generated within NBS and other agencies of the Federal government; promotes the development of the National Standard Reference Data System and a system of information analysis centers dealing with the broader aspects of the National Measurement System, and provides appropriate services to ensure that the NBS staff has optimum accessibility to the scientific information of the world. The Office consists of the following organizational units:

Office of Standard Reference Data—Clearinghouse for Federal Scientific and Technical Information "—Office of Technical Information and Publications—Library—Office of Public Information—Office of International Relations.

Headquarters and Laboratories at Gaithersburg, Maryland, unless otherwise noted; mailing address Washington, D.C. 20234.

² Located at Boulder, Colorado 80302.

³ Located at 5285 Port Royal Road, Springfield, Virginia 22151.



NATIONAL BUREAU OF STANDARDS REPORT

NBS PROJECT

NBS REPORT

421.04-12-4212270

August 1, 1969

10 075

Progress Report
April 1 through June 30, 1969

DEVELOPMENT OF METHODS OF TEST FOR QUALITY CONTROL OF PORCELAIN ENAMELS

BY
M. D. Burdick and M. A. Baker

Porcelain Enamel Institute Research Associateship
National Bureau of Standards

IMPORTANT NOTICE

NATIONAL BUREAU OF STAP for use within the Government. Be and review. For this reason, the p whole or in part, is not authorize Bureau of Standards, Washington, the Report has been specifically pr

Approved for public release by the director of the National Institute of Standards and Technology (NIST) on October 9, 2015

accounting documents intended ibjected to additional evaluation isting of this Report, either in Office of the Director, National the Government agency for which lies for its own use.



U.S. DEPARTMENT OF COMMERCE NATIONAL BUREAU OF STANDARDS



SUMMARY

Preliminary tests were made on porcelain enamel cover coats direct-to-steel to determine whether a double cantilever direct-pulling test could be adapted to measure the adherence of these porcelain enamels to their substrates.

Work was completed on the development of a Continuity of Coating test method for general purpose enamels.

A cupric chloride accelerated test method is described, which appears to give a more specific indication of color changes found for porcelain enamels on aluminum during three years of natural weathering, than does the boiling acid solubility.

The Double Cantilever Approach to the Adherence of Porcelain Enamel Cover Coats Direct-to-Steel

INTRODUCTION

Fracture mechanics is a relatively new discipline which relates the fracturing behavior of flawed bodies to applied loads. By its concepts, one defines a quantity, the fracture toughness, which has a similar importance in brittle failure that yield strength has in ductile failure. In a ductile material the load at which deviations from elasticity occur is associated with the yield point. In a cracked body the load that causes a stationary crack to propagate rapidly is associated with the fracture toughness. Hence the fracture toughness of a material is seen to be a measure of its ability to resist extension of a pre-existing crack. The parameter, fracture toughness, may be defined as the critical strain energy release rate

$$G_{c} = \frac{P^{2}}{2b} \qquad \left(\frac{d^{1/M}}{dc}\right) \qquad \dots \qquad (1)$$

in units of pounds per inch.

where P is the load at which a crack propagates.

b is the specimen width

1/M is the reciprocal slope of the loadseparation curve, and
c is the existing crack length before
propagation.

The double cantilever method has previously been used to study many metals, tranclucent crystals, transparent glass, flawed concrete, brittle plastic materials and electrical porcelain. The appended bibliography refers to some of these applications. This investigation was initiated to explore whether the fracture toughness of porcelain enameled cover coats direct-to-steel can be used to evaluate the force required to strip an enamel coating from its substrate. A pre-existing crack will tend to advance through the weakest plane within the coating-substrate system. Thus the fracture toughness value will evaluate the adherence of the coating only if the separation occurs between the coating and substrate.

RESULTS AND DISCUSSION

The experimental set-up includes a tensile testing machine with appropriate hardware to apply load to a specimen assembly. A specimen assembly consists of a direct-on specimen between two alluminum alloy bars, one of which is epoxied to the slightly sandblasted enamel surface, and the other is attached with the same adhesive to the bottom surface of the steel substrate.

A linear variable differential transformer, L.V.D.T., is mounted to measure and record small changes in separation between the cantilevers. A typical geometry for the specimen assembly is shown in Figure 1.

It can be seen by referring to Figure 1 that in the initial situation the force P acts through moment arms c, (equal to the effective cantilever length) and strain energy builds up at the leading specimen edge. When the strain energy exceeds a critical value, a crack will propagate through the weakest path. As the crack increases in length so does the effective cantilever length. The increased deflection of the lengthened cantilevers results in a decreased load on the specimen assembly and crack propagation will be temporarily arrested if the rate of cross head movement is slow enough.

A. Estimation of the Crack Length, c.

For transparent materials measurements of each crack length can easily be made during the tests. For opaque materials, in which the crack lengths cannot be readily observed, previous investigators have used a relationship between the reciprocal slope of the load-separation curve and the crack length to establish the latter quantity.

Calibration bars of height 2h plus specimen thickness were prepared from the aluminum alloy material. Saw cuts of various lengths were used to simulate a range of crack lengths. The reciprocal slope of the load-separation curves of these calibration bars, shown in Figure 2, served to establish the relationships plotted in Figures 3 and 4. The crack length within specimen assembles employing similar alloy cantilevers can then be estimated from the 1/M of a load-separation curve using the relation shown in Figure 3.

The relationship between 1/M and c was obtained from the equation of the straight line of the straight line of the log-log plot.

$$\log 1/M = 1.4917 + 2.0487 \log c$$

$$\log c = \frac{\log 1/M - 1.4917}{2.0487} ... (3)$$

The value of c was obtained directly from its logarithm. In addition a polymomial, in terms of crack length, c, was fitted to the experimentaly determined values of 1/M, the relationship between 1/M and c was assumed to be

$$1/M = A + Bc^2 + Cc^3 + Dc^4$$
 (4)

where A, B, C, and D are constants.

$$\frac{d(1/M)}{dc} = 2Bc + 3Cc^{2} + 4Dc^{3} \qquad ... (5)$$

B. Calculations of Fracture Toughness, Gc

A specimen bank has been established which includes three groups of direct-to-steel cover coats, each of supposedly differnt adherence values. Three degrees of pickling were employed in the preparation of these specimens, varying from normal pickling, through an intermediate degree to an insufficient or zero amount.

Several trials of the double cantilever method outlined above, have been made. These were made primarily to develop the details of the experimental techniques and also to give preliminary evaluation to the same enamel on the same substrate after different degrees of pickling. Enamel B was selected to give a comparison between normal and, in this case, no pickling. The results calculated from Equation (1) are given in Table 1. Eight crack propagations are given for the normally pickled system leading to eight values of Gc with a mean value of 1.6 pounds per inch (of width). Lesser numbers of crack events were usable when the pickling was inadequate. Only those cracks were used for which the slope of the load-separation curve diminished from the previous value. It can be seen that the values of Gc for the unpickled specimens had a mean value of 0.03 pounds per inch, or less than two percent of the value for the normally pickled substrate systems.

C. The Mode of Failure Observed in the Trials

The unpickled specimens failed in such a way that the glass and oxide layers were cleanly separated from the bright metal substrate for the full width of the cantilevers (o.5 inch). The weakest adherence was between the oxide layer (black) and the bright substrate. The values of 0.03 pounds per inch were an evaluation of the adherence sought.

The coating of the normally pickled speciment was not stripped cleanly from the substrate. Approximately one half of the coating was removed exposing the black oxide layer but practically no bright metal. The fracture toughness, here evaluated, corresponded to the adherence of the glass layer to the oxide layer. One can assume this to be the interface of poorest adherence. The irregularity in glass removal may possibly be attributed to insufficient adhesive strength of the epoxy to the glass layer. Also this irregularity was undoubtedly reflected in the fairly large variability in the Gc values calculated for this enamel system.

PLANS FOR NEXT REPORT PERIOD

- 1. Evaluate the fracture toughness of several other enamel systems which differ only in pickling pretreatment.
- Seek improved epoxy adhesives or better technique in their use.

3. A portable adherence measuring device manufactured by the Gardner Laboratory for use primarily with organic coatings is available. This is a self-contained force measuring device. It will be tried on cover coats direct-to-steel prepared after various pretreatments.



CONTINUITY OF COATING

During this report period, a second dc high-voltage apparatus was borrowed to determine whether two dc instruments would give more consistant results than the four ac instruments did. It was found that the air gap - voltage curves for the dc instruments were different but the "critical overvoltage" remained constant. Therefore, the original test precedure was modified so only the air-gap voltage needs to be determined for each test instrument. Once this is determined, the same electrical stress may be added to any enamel by adding a given overvoltage to the air gap voltage for the instrument being used. If ac instruments are used, the air-gap voltages and the overvoltage must be converted to peak voltages by multiplying by 1.414.

These changes were incorporated into the papers describing the development of this test and the test procedure. Both of these papers have cleared the editorial review at the National Bureau' of Standards and have been submitted for publication to the American Ceramic Society and PEI.

WEATHERING OF PORCELAIN ENAMELS

The inspection of the enamels on aluminum included in the 1966 Exposure Test of Porcelain Enamels on Aluminum was completed during the quarter ending March 31, 1969. Upon completion of this inspection, it was noted that the boiling acid solubility test did not correlate well with the color retention of many of the enamels exposed at Kure Beach. Therefore some time was devoted toward the development of a new test that might correlate better with actual results than the boiling acid solubility test. The test developed utilized the same apparatus as the boiling acid solubility test except hat 50 ml of a 12.5 percent solution of cupric chloride is substituted for 150 ml of 6 percent citric acid. The cupric chloride is boiled on the specimen surface for one hour, poured out, and the specimen surface is rinsed with 10 ml of a one percent solution of oxalic acid to remove any green copper stains that may have formed. The specimens are then removed from the clamping devise, washed with a mild detergent solution, rinsed with tap water, distilled water and alcohol and allowed to dry in a near vertical position. When the specimens are dry, the color change between the tested and untested portions of the specimen is measured and is used as a measure of the enamel's weather resistance. This test produces color changes in the enamels which are similar to those occurring during natural weathering. Figures 6 and 7 compare the color retentions at Kure Beach and Los Angeles vs boiling acid and cupric chloride test results. A better correlation exists between the cupric chloride specimens and color retention than between the boiling acid solubility treated specimens and color retention. It is felt that the cupric chloride test may be a better test for predicting the weatherability of these enamels but it is also felt that exposure data for longer periods of time are needed to determine whether these same color changes will occur at the milder sites. If the large color changes are confined to Kure Beach, then more work will have to be done to determine the cause of the severity at this site.

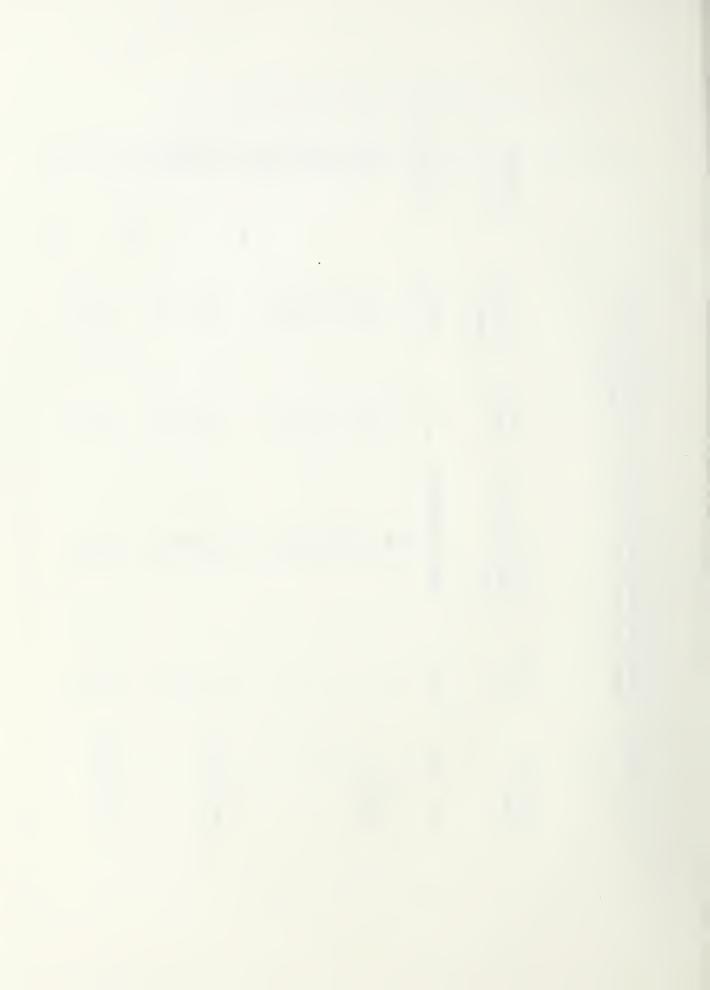
BIBLIOGRAPHY

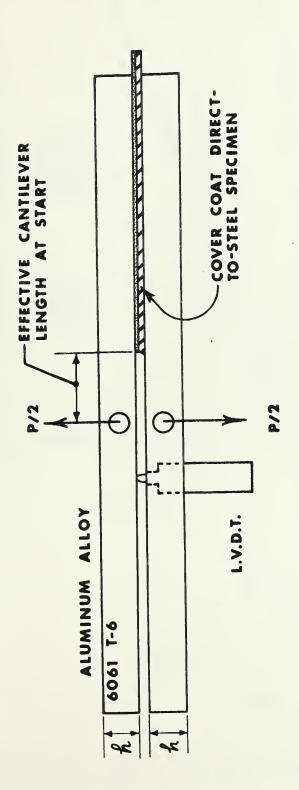
- J. J. Gilman, "Direct Measurements fo the Surface Energies of Crystals", J. Appl. Phys., 31 (12) 2208 (1960).
- J. J. Gilman, Fracture, p193-221, Technology Press of MIT, Cambridge, Mass, J Wiley & Sons, New York (1959).
- A. R. C. Westwood & T. T. Hitch, "Surface Energies of (100) Potassium Chloride", J. Appl. Phys., 34 3085 (1963).
- A. R. C. Westwood & D. L. Goldheim, "Cleavage Surface Energy of (100) Magnesium Oxide", J. Appl. Phys. 34 (11) 3335 (1963).
- D. A. Shockey & G. W. Groves, "Effect of Water on Toughness of MgO Crystals", J. Am. Ceram. Soc., 51, (6) 299 (1968).
- D. A. Shockey & G. W. Groves, "Origin of Water-Induced Toughning in MgO Crystals", J. Am. Ceram. Soc., <u>52</u>, (2) 82 (1969).
- S. M. Weiderhorn, "Fracture Surface Energy of Soda-Lime Glass", Materials Science Research Vol 3, Edited by W. W. Kriegel and Hayne Palmour III, Plenum Press, New York, N. Y. (1966).
- S. M. Weiderhorn, A. H. Shorb and R. L. Moses, "Critical Analysis of the Theory of the Double Cantilever Method of Measuring Fracture Surface Energies", J. Appl. Phys. 39 (3) 1569 (1968).
- S. M. Weiderhorn, "Fracture Surface Energy of Glass", J. Am. Ceram. Soc., <u>52</u> (2) 99 (1969).
- J. P. Berry, "Determination of Fracture Surface Energies by the Cleavage Technique", J. Appl. Phys., 34 (1) 62 (1963).
- H. Fukui & N. Higuchi, "Toughness Parameter of Electrical Porcelain Materials", J. Am. Ceram. Soc., (5) 583 (1969).
- S. Mostovoy & E. J. Ripling, "Fracture Toughness of an Epoxy System" J. Appl. Polymer Sci., 10, 1351 (1966).

- G. R. Irwin, J. A. Kies, and H. L. Smith, "Fracture Strengths Relative to Onset and Arrest of Crack Propagation", Proc. A.S.T.M. 58 640, (1958).
- M. F. Kaplan, Crack Propagation and the Fracture of Concrete", J. Am. Concrete Inst., Title No 58-28, p. 591, Nov. (1961).

Values Obtained for the Fracture Toughness, G for a Porcelain Enamel Cover Coat Direct- to-Steel. Table 1.

Fracture Toughness, G	Pounds/inch	,	1.15	2 .05	1.46	1,61	1.73	1.75	2.06	mean 1.61	0.0320	.0271	0261	•0246	0.0219	mean 0.0263	0°0309	•0303	•0299	mean 0.0304
Load to Propagate Crack	Pounds	0	82.7	79.0	45.1	37.8	33.7	29 •0	26.3	Œ	6.13	4 •85	4.13	3,34	2.86	E	11.8	9.37	4.93	E
Crack Length	In.	o c	2.48	3,32	4.34	66° 7	5 •49	6.02	6.71		4.58	5 • 03	5.49	•	6.50		2.83	3,38	5.13	
Slope of Load-Deflection Curve	Pounds/microinch	x 10 ³	5.01	2.75	1.59	1.19	86.0	0.81	0.65		1.42	1.17	86.0	0.78	69°0		3.83	2.66	1.12	
Crack No•		c	v w	4	2	9	7	œ	6		4	5	9	œ	6		2	က	7	
Direct-On Ename1		ρ	(Normal	Pickle)							В	(Not Pickled)					В	(Not Pickled)		

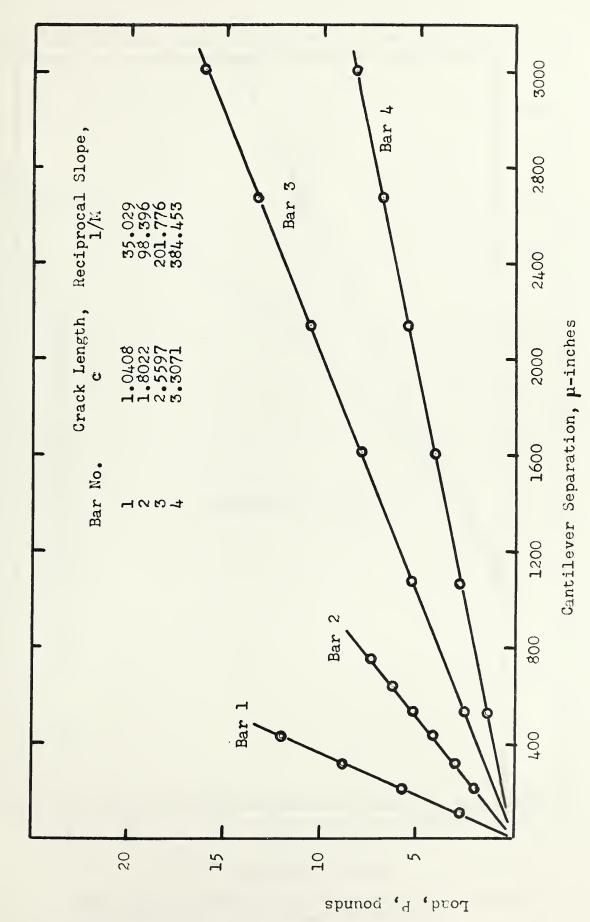




DOUBLE CANTILEVER SPECIMEN ASSEMBLY

FIGURE





Load-Separation Curves for Calibration Bars Figure 2.

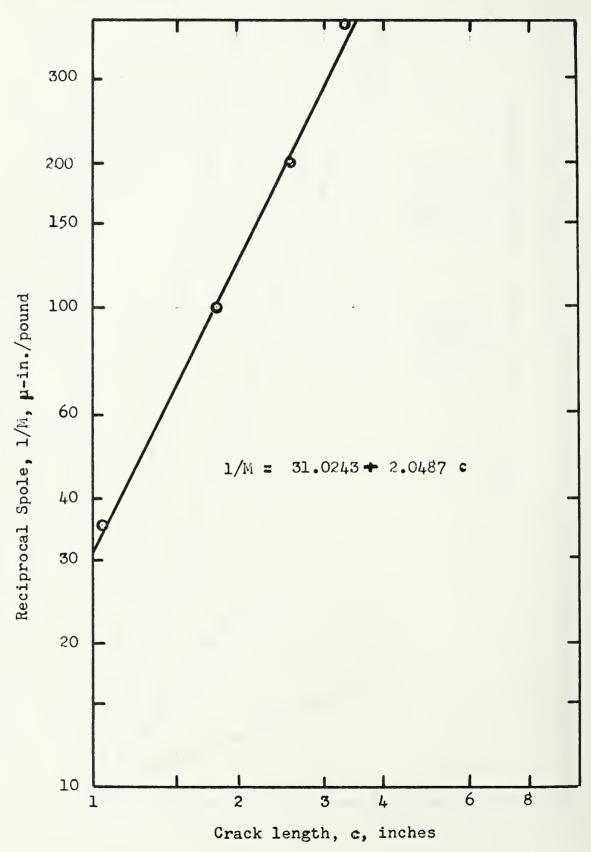


Figure 3. Log-Log Plot of Reciprocal Slope, 1/M and Crack Length for Calibration Bars.

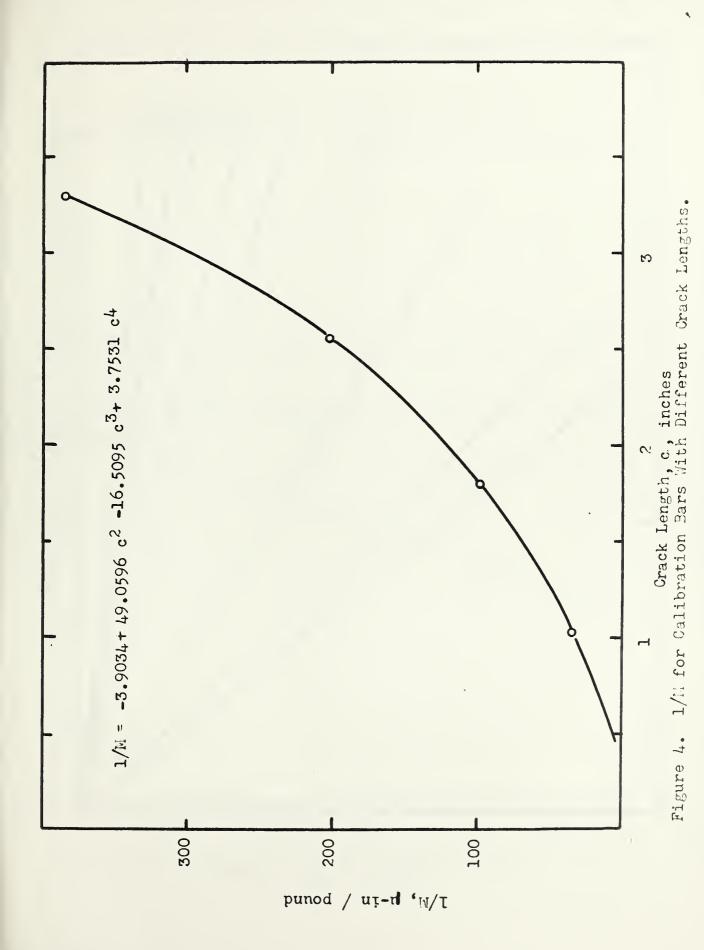




Figure 5. Superimposed Load-Jeparation Curves for Cracks 2 thru 9 of a Specimen.

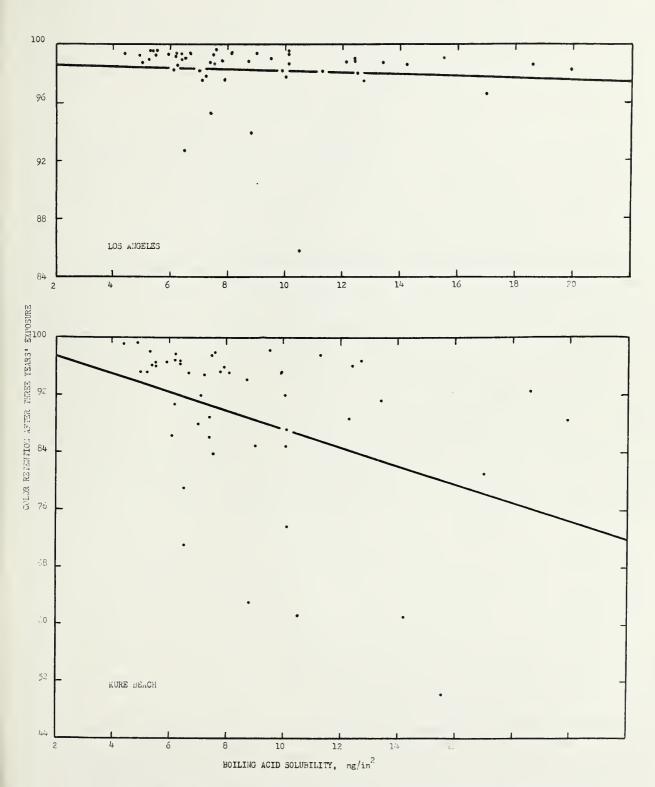


Figure 0. Correlation between the Boiling Acid Solubility Test and Color detention of Engrels Exposed Chree Var. at Los angeles and Kure Beach.

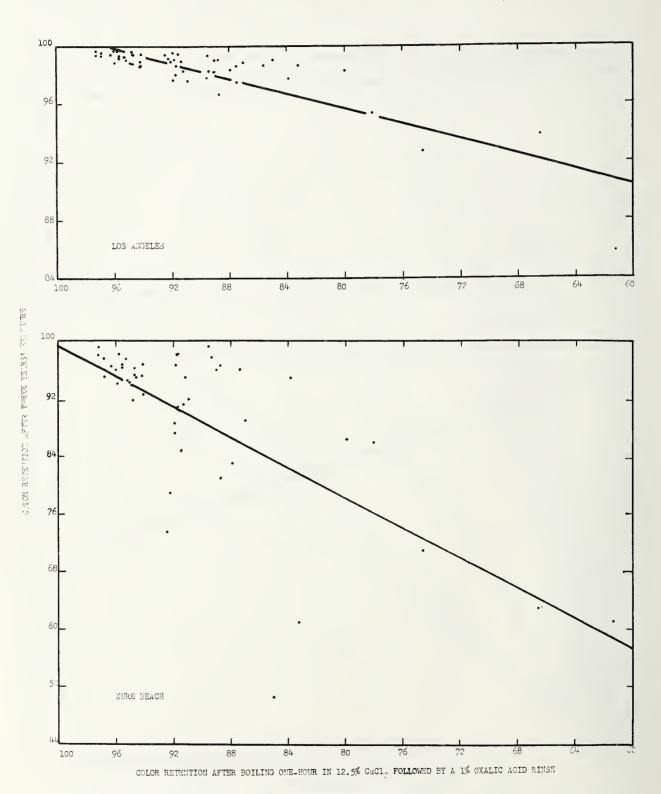


Figure 7. Correlation between the Cupric Chloride Treatment and the Color Retention of Engmels Exposed Three Years at Los Angeles and Kure Beach.





ì