

A11102 145880

NAT'L INST OF STANDARDS & TECH R.I.C.



A11102145880

Moore, Charlotte Emm/Atomic energy level
QC100 .U573 V35:1:1971 C.1 NBS-PUB-C 197

DOMESTIC
PUBLICATION



NSRDS—NBS 35

Volume I

1H-23V



ATOMIC ENERGY LEVELS

As Derived From the Analyses
of Optical Spectra

U.S.
DEPARTMENT
OF
COMMERCE

National
Bureau
of
Standards

**Author's Note on the Reprinting of
Atomic Energy Levels: Volume I, 1949; Volume II,
1952; Volume III, 1958, Circular National Bureau of Standards 467**

Although twelve years have elapsed since the publication of Volume III, there is a continuing steady demand for these Volumes. The data they contain on atomic spectra cover all elements in the Periodic Table except the two groups of rare-earths: the lanthanides ($Z = 58-71$) and the actinides ($Z = 90-?$). Similar data for these spectra will be handled in a forthcoming Volume IV, now in course of preparation by W. C. Martin and his colleagues.

One of the rewarding aspects of these compilations has been the stimulation they have provided to further research on the analyses of atomic spectra. Gaps in the knowledge of spectra, and needs for investigation of additional spectra are immediately apparent in this comprehensive compendium.

Many additional spectra have been studied and numerous extended analyses have been published that supersede the material contained in Circular 467. A bibliography in the National Bureau of Standards Special Publication 306, Sections 1 to 4 (1968-1969), provides later reference material on individual spectra. It will be some years, however, before the entire set of Volumes will be superseded. The existing supply of these books is low. In order to meet the steady flow of requests, it has been decided to reissue the three Volumes as part of the National Standard Reference Data System. They are reprinted here as NSRDS-NBS 35, Volumes I, II, III.

The first Volume, issued in 1952, is in great demand, and more seriously in need of extensive revision than are the others. As new analyses appear for spectra of the lighter elements, the lists of revised energy levels, together with revised Multiplet Tables, are being published by the National Bureau of Standards under the title "Selected Tables of Atomic Spectra, Atomic Energy Levels and Multiplet Tables," as Sections of NSRDS-NBS 3. Section 1 contains these data for the spectra Si II, Si III, Si IV; Section 2 for Si I; Section 3 for C I, C II, C III, C IV, C V, C VI. Similar data on the nitrogen spectra of higher ionization will be presented in Section 4. A number of other spectra are partially completed for inclusion in this Series.

Wherever the individual spectra in Volume I have been revised and reported in the NSRDS-NBS 3 Series, indication of this fact is clearly stated for each spectrum, in this reprinted issue.[†] Readers are urged to use the revised material for the spectra thus marked and to take note of further revisions of selected spectra as they appear in this series.

Washington, D.C.
November 30, 1970

Charlotte E. Moore

Abstract

ATOMIC ENERGY LEVELS, VOLUME I. ^1H to ^{23}V

This series of three volumes is a critical compilation of atomic energy levels prepared at the National Bureau of Standards from the analyses of optical spectra. Volume I contains data on the spectra of hydrogen, deuterium, tritium, helium, lithium, beryllium, boron, carbon, nitrogen, oxygen, fluorine, neon, sodium, magnesium, aluminum, silicon, phosphorus, sulfur, chlorine, argon, potassium, calcium, scandium, titanium, and vanadium (^1H to ^{23}V). Volume II covers the spectra of chromium, manganese, iron, cobalt, nickel, copper, zinc, gallium, germanium, arsenic, selenium, bromine, krypton, rubidium, strontium, yttrium, zirconium, and niobium (^{24}Cr to ^{41}Nb). Volume III includes the spectra of molybdenum, technetium, ruthenium, rhodium, palladium, silver, cadmium, indium, tin, antimony, tellurium, iodine, xenon, cesium, barium, lanthanum; hafnium, tantalum, tungsten, rhenium, osmium, iridium, platinum, gold, mercury, thallium, lead, bismuth, polonium, radon, radium, and actinium (^{42}Mo to ^{57}La ; ^{72}Hf to ^{89}Ac).

Key words: Energy levels; H-V.

[†]EDITORIAL NOTE: See revision note on pages 21, 24, 26, 29, 30, 31, 144, 147, 148, and 150, Volume I.

Library of Congress Catalog Card Number 75-609945

AUG 29 1972

165762

AC 100
10573
1035
V. 1
1971
212

UNITED STATES DEPARTMENT OF COMMERCE • Maurice H. Stans, *Secretary*
NATIONAL BUREAU OF STANDARDS • Lewis M. Branscomb, *Director*

ATOMIC ENERGY LEVELS

As Derived From the Analyses of Optical Spectra

Volume I

The Spectra of Hydrogen, Deuterium, Tritium, Helium,
Lithium, Beryllium, Boron, Carbon, Nitrogen, Oxygen,
Fluorine, Neon, Sodium, Magnesium, Aluminum, Silicon,
Phosphorus, Sulfur, Chlorine, Argon, Potassium, Calcium
Scandium, Titanium, and Vanadium

BY CHARLOTTE E. MOORE



NSRDS-NBS 35

Nat. Stand. Ref. Data Ser., Nat. Bur. Stand. (U.S.), 35/V.I 359 pages (Dec. 1971)
CODEN: NSRDA

Reissued December 1971

Reprint of NBS Circular 467, Volume I.
See author's note opposite title page

For sale by the Superintendent of Documents, U.S. Government Printing Office, Washington, D.C. 20402
(Order by SD Catalog No. C13.48:35/V.I), Price \$5
Stock Number 0303-0949

Preface

The present volume is the first of a series being prepared at the National Bureau of Standards as part of a general program on atomic energy levels derived from observations of optical spectra. This program can be traced back to 1924 when the Division of Physical Sciences of the National Research Council created a Committee on Line Spectra of the Elements. The general plan was to encourage and contribute to the structural analysis of atomic spectra and eventually to publish the results in a series of monographs. For twenty years the lure of complex spectra gave emphasis to analysis rather than to compilation and publication of Committee Reports.

In 1932 an extremely timely and useful book entitled "Atomic Energy States as Derived from the Analyses of Optical Spectra" was published by Robert F. Bacher and Samuel Goudsmit. That book set a precedent for omitting experimental details (wavelengths, intensities, Zeeman patterns, etc.) and summarized the terms then known for 231 spectra of 69 elements. Now structure has been recognized in more than 460 spectra, representing 83 elements, and the earlier analyses have in almost all cases been greatly extended.

The accumulation of spectroscopic data is now too vast for publication in a reasonable number of monographs, but the energy levels derived from them are so important for physics, chemistry, and astronomy that a revision of "Bacher and Goudsmit" is urgently needed; it can probably be condensed into three or four volumes. In the spring of 1946 it was determined that neither Bacher nor Goudsmit contemplated such a revision, and it was decided to undertake this at the National Bureau of Standards. Details of this project were discussed at a meeting of the National Research Council Committee on Line Spectra of the Elements, called by the Chairman, Henry Norris Russell, and held in Washington in May 1946. It was then decided to send to interested workers in various fields a questionnaire regarding the most useful form of presentation of the data on atomic energy levels. The present form represents the majority vote resulting from that inquiry.

It was originally planned to issue sections in pamphlet form as the manuscript was completed, and to assemble the sections into volumes of about 400 pages each. Section 1 has been published separately.

This volume comprises the first three sections of Circular 467 of the National Bureau of Standards as follows:

Section 1. The Spectra of Hydrogen, Deuterium, Tritium, Helium, Lithium, Beryllium, Boron, Carbon, Nitrogen, Oxygen, and Fluorine. (Pages 1 to 75.)

Section 2. The Spectra of Neon, Sodium, Magnesium, Aluminum, Silicon, Phosphorus, Sulfur, and Chlorine. (Pages 76 to 210.)

Section 3. The Spectra of Argon, Potassium, Calcium, Scandium, Titanium, and Vanadium. (Pages 211 to 309.)

It has since been decided not to publish sections 2 and 3 separately because they are simultaneously in press and complete Volume I.

The manuscript has been prepared by Charlotte E. Moore under the direction of William F. Meggers, Chief of the Spectroscopy Section of the Atomic and Molecular Physics Division. Sincere appreciation is hereby expressed for the cordial cooperation of the National Research Council Committee on Line Spectra of the Elements, and for the heretofore unpublished contributions of many spectroscopists. Because the current volumes of Atomic Energy Levels disclose many gaps in our knowledge in addition to some uncertainties and occasional irregularities, it seems certain that they will inspire further researches in experimental and theoretical spectroscopy, and thus in turn advance the specialized subjects of atomic and nuclear physics.

E. U. CONDON, *Director*.

WASHINGTON, D. C., June 1948.

Contents

	Page		Page
Preface.....	II	7. Tables of Predicted and Observed Arrays of Terms—Continued	
1. Introduction.....	VII	7.3. Arrays of predicted terms of the sequences Mg I through Al (tables 12 to 18).....	XVI
2. Scope of the Present Tables.....	VII	7.4. Arrays of predicted levels of the Ne I and Al sequences (tables 11 and 18).....	XVI
3. Nomenclature (atomic energy levels, spectroscopic terms, multiplets).....	VIII	7.5. Arrays of predicted terms of the sequences Ca I through V I (tables 19 to 22).....	XVI
4. Arrangement.....	VIII	8. The Periodic Table.....	XVII
4.1. Headings, remarks.....	VIII	8.1. The chemical elements by atomic number, ionization potentials (table 23).....	XVII
4.2. References.....	IX	8.2. The chemical elements by chemical symbol (table 24).....	XVII
4.3. Reference symbols.....	IX	8.3. The periodic system (table 25).....	XVII
5. Spectroscopic Notation.....	IX	8.4. Index—iso-electronic sequences (table 26).....	XVII
5.1. Series spectra.....	X	9. Future Investigations.....	XVII
5.2. Inert gases.....	XI	9.1. Need for further analysis.....	XVII
5.3. Complex spectra.....	XII	9.2. Term intervals.....	XVIII
6. Columns of the Table.....	XII	9.3. Series spectra—Rydberg denominators.....	XVIII
6.1. Author.....	XII	9.4. Observed Zeeman patterns.....	XVIII
6.2. Configuration.....	XII	9.5. Energy or Grotrian diagrams.....	XVIII
6.3. Designation.....	XII	10. Acknowledgments.....	XIX
6.4. Inner quantum number <i>J</i>	XIII		
6.5. Atomic energy level.....	XIII		
6.6. Interval.....	XIII		
6.7. Observed <i>g</i> -value (tables 1 to 4, Landé <i>g</i> -values).....	XIV		
7. Tables of Predicted and Observed Arrays of Terms.....	XV		
7.1. Shells.....	XV		
7.2. Arrays of predicted terms of the sequences Be I through Ne I (tables 5 to 11).....	XV		

List of Tables

Table	Subject	Page	Table	Subject	Page
1 to 4	Landé <i>g</i> -values	XX to XXVII		PREDICTED TERMS—continued	
	PREDICTED TERMS		16	S I	XXXIV
5	Be I	XXVIII	17	Cl I	XXXIV
6	B I	XXVIII	18	A I	XXXV
7	C I	XXIX	19	Ca I	XXXV
8	N I	XXIX	20	Sc I	XXXVI
9	O I	XXX	21	Ti I	XXXVII
10	F I	XXXI	22	V I	XXXVIII
11	Ne I	XXXI	23	Ionization Potentials	XL
12	Mg I	XXXII	24	Chemical Symbols	XLI
13	Al I	XXXII	25	The Periodic System	XLII
14	Si I	XXXIII	26	Index—Iso-electronic Sequences	XLIII
15	P I	XXXIII			

Index to Spectra

Element	Z	Spectrum	Page	Element	Z	Spectrum	Page
Hydrogen	1	H	1	Neon	10	Ne I.....	76
		D	3			Ne II.....	81
		T	3			Ne III.....	83
Helium	2	He I.....	4			Ne IV.....	84
		He II.....	6			Ne V.....	86
Lithium	3	Li I.....	8			Ne VI.....	88
		Li II.....	10	Sodium	11	Na I.....	89
		Li III.....	11			Na II.....	91
Beryllium	4	Be I.....	12			Na III.....	93
		Be II.....	14			Na IV.....	95
		Be III.....	14			Na V.....	96
		Be IV.....	15			Na VI.....	98
Boron	5	B I.....	16			Na VII.....	100
		B II.....	17			Na VIII.....	103
		B III.....	19			Na IX.....	105
		B IV.....	19	Magnesium	12	Mg I.....	106
		B V.....	20			Mg II.....	108
Carbon	6	C I.....	21			Mg III.....	109
		C II.....	24			Mg IV.....	111
		C III.....	26			Mg V.....	113
		C IV.....	29			Mg VI.....	114
		C V.....	30			Mg VII.....	117
		C VI.....	31			Mg VIII.....	119
Nitrogen	7	N I.....	32			Mg IX.....	121
		N II.....	35			Mg X.....	122
		N III.....	38			Mg XI.....	123
		N IV.....	40	Aluminum	13	Al I.....	124
		N V.....	42			Al II.....	126
		N VI.....	43			Al III.....	129
		N VII.....	44			Al IV.....	130
Oxygen	8	O I.....	45			Al V.....	131
		O II.....	47			Al VI.....	133
		O III.....	50			Al VII.....	135
		O IV.....	53			Al VIII.....	136
		O V.....	56			Al IX.....	138
		O VI.....	58			Al X.....	140
		O VII.....	59			Al XI.....	142
		O VIII.....	59			Al XII.....	143
Fluorine	9	F I.....	60	Silicon	14	Si I.....	144
		F II.....	62			Si II.....	147
		F III.....	64			Si III.....	148
		F IV.....	66			Si IV.....	150
		F V.....	69			Si V.....	151
		F VI.....	71			Si VI.....	152
		F VII.....	74			Si VII.....	154
		F VIII.....	75			Si VIII.....	156
						Si IX.....	157
						Si X.....	159
						Si XI.....	160
						Si XII.....	162

Index to Spectra—Continued

v

Element	Z	Spectrum	Page	Element	Z	Spectrum	Page
Phosphorus	15	P I.....	163	Potassium— (Continued)	19	K IX.....	239
		P II.....	164			K X.....	239
		P III.....	166			K XI.....	241
		P IV.....	168	Calcium	20	Ca I.....	242
		P V.....	169			Ca II.....	245
		P VI.....	170			Ca III.....	247
		P VII.....	171			Ca IV.....	248
		P VIII.....	173			Ca V.....	249
		P IX.....	174			Ca VI.....	251
		P X.....	176			Ca VII.....	252
		P XI.....	177			Ca VIII.....	253
		P XII.....	179			Ca IX.....	254
		P XIII.....	180			Ca X.....	255
						Ca XI.....	255
						Ca XII.....	257
Sulfur	16	S I.....	181			Ca XIII.....	258
		S II.....	183			Ca XV.....	258
		S III.....	185	Scandium	21	Sc I.....	259
		S IV.....	187			Sc II.....	262
		S V.....	188			Sc III.....	263
		S VI.....	189			Sc IV.....	264
		S VII.....	190			Sc V.....	265
		S VIII.....	191			Sc VI.....	266
		S IX.....	193			Sc VII.....	267
		S X.....	194			Sc VIII.....	268
		S XII.....	194			Sc IX.....	269
						Sc X.....	270
						Sc XI.....	271
Chlorine	17	Cl I.....	195			Sc XII.....	272
		Cl II.....	197	Titanium	22	Ti I.....	273
		Cl III.....	199			Ti II.....	279
		Cl IV.....	201			Ti III.....	281
		Cl V.....	202			Ti IV.....	283
		Cl VI.....	204			Ti V.....	284
		Cl VII.....	205			Ti VI.....	285
		Cl VIII.....	206			Ti VII.....	286
		Cl IX.....	207			Ti VIII.....	287
		Cl X.....	209			Ti IX.....	288
		Cl XI.....	210			Ti X.....	288
						Ti XI.....	289
Argon	18	A I.....	211			Ti XII.....	289
		A II.....	216			Ti XIII.....	290
		A III.....	218	Vanadium	23	V I.....	291
		A IV.....	220			V II.....	298
		A V.....	222			V III.....	301
		A VI.....	223			V IV.....	303
		A VII.....	224			V V.....	304
		A VIII.....	224			V VI.....	304
		A IX.....	225			V VII.....	305
		A X.....	226			V VIII.....	306
		A XI.....	226			V IX.....	306
		A XIV.....	226			V XI.....	307
						V XII.....	307
Potassium	19	K I.....	227			V XIII.....	308
		K II.....	230			V XIV.....	309
		K III.....	231				
		K IV.....	233				
		K V.....	234				
		K VI.....	236				
		K VII.....	237				
		K VIII.....	238				

1. Introduction

Since the publication in 1932 by Bacher and Goudsmit of their book "Atomic Energy States,"¹ the number of energy levels determined from the analyses of optical spectra has increased by a factor of perhaps 4 or 5 and yet no critical compendium of these data exists. In order to meet this need, the present compilation has been undertaken at the National Bureau of Standards.

A handbook of "Atomic Energy Levels" is an indispensable tool for workers in many fields of science today. For the spectroscopist it reveals the gaps in our knowledge of atomic spectra—both those spectra that are incompletely analyzed because of insufficient observations and those that have not yet been observed. For the theoretical as well as the experimental investigators, the detailed comparison of data on related spectra, uniformly arranged, is a useful guide in the study of series, intervals, electron configurations, and many other related problems of atomic structure.

Many interesting spectroscopic problems also arise in connection with microwave spectroscopy, with ultraviolet solar spectra observed from rockets, with infrared spectra observed with a sensitive detector, and in general with types of observation that have developed comparatively recently. If the analysis of a spectrum is complete the positions of the lines can be calculated from the known energy levels, including in many cases those of lines in the far infrared or ultraviolet. The present term tables are now being used in connection with some problems of this sort.

The needs of the nuclear as well as the atomic physicist, of the chemist interested in atomic structure, of the astrophysicist interested in the study of stellar structure and cosmical abundances, and of those in many other fields of science all provide the inspiration for this work.

2. Scope of the Present Tables

Ten of the fourteen members of the National Research Council Committee on Line Spectra of the Elements attended the meeting held in Washington in May 1946, to consider details of this program. Two members who were unable to attend, I. S. Bowen and R. A. Sawyer, made personal visits to the Bureau before the meeting for this purpose. A number of other spectroscopists, including B. Edlén, have also been consulted in private conference.

On the recommendation of the committee a questionnaire regarding details of arrangement of the tables was sent to 94 interested workers in various fields of science. Sixty-one replied to this inquiry. The scope, uses, and format of the book have been discussed at length and the general form adopted is a direct outgrowth of these conferences and recommendations.

The cordial collaboration of those who have been contacted is gratifying. The Chairman of the Committee, H. N. Russell, has read all of the manuscript, provided much material, and made many helpful suggestions. The writer has had the benefit of his broad experience with spectroscopic problems. The committee and others as well are giving their wholehearted support to this program.

Requests to extend the scope of the tables have been seriously considered. It was finally decided to include only the energy levels derived from observations of atomic spectra, exclusive of hyperfine structure ascribed to atomic nuclei (with the exception of H, D). With full

appreciation of the importance of critical data on nuclear and X-ray spectra, on isotopes, and on other subjects related to atomic structure the present policy was adopted for several reasons. The usefulness of the tables might well be vitiated by the inclusion of too many kinds of data. The critical editing of the enormous amount of literature entailed by extending the program would of necessity delay by years the publication of data on any one phase of the subject. Finally, the preparation of the volumes of "Atomic Energy Levels" is an appropriate sequel to the work on the revised edition of "A Multiplet Table of Astrophysical Interest,"² hereinafter referred to as RMT.³ These two types of tables used in conjunction with each other provide a condensed and unified picture of many atomic spectra—the one containing the energy levels and term designations used to compile the multiplets and excitation potentials recorded in the other.

In view of the limitations imposed here, reference is made under the relevant spectra to the excellent summary and bibliography of data on hyperfine structure by Meggers, in his paper entitled "Spectroscopy, Past, Present, and Future."⁴ In addition, selected later papers on hyperfine structure and isotope shifts are listed for certain spectra. The reader is warned, however, that the individual references on these subjects included here are highly selected and that the present book is inadequate for workers in these fields.

² Princeton Univ. Obs. Contr. No. 20 (1945).

³ This edition is limited to lines of wavelength longer than 3000 Å. Along with the tabulation of energy levels, the writer is also preparing an ultraviolet extension to the Revised Multiplet Table.

⁴ J. Opt. Soc. Am. **38**, 431 (1946).

¹ McGraw-Hill Book Co., Inc., New York, N. Y., and London (1932).

3. Nomenclature

(Atomic Energy Levels, Spectroscopic Terms, Multiplets)

Briefly summarized, the atoms of a gas or vapor, when excited by radiation, absorb certain wavelengths corresponding to transitions of their outer electrons from lower energy levels to higher ones. When the transitions are from higher to lower energy levels the lines are emitted. Each chemical element can emit as many atomic spectra as it has electrons. If, for example, a sample of pure vanadium is placed in an electric arc and light from the arc is observed through a spectroscope, a complex array of spectral lines of various intensities appears. Most of these lines are produced by neutral vanadium atoms and are characteristic of the first (or arc) spectrum of vanadium, VI.

If vanadium atoms are excited by an electric spark instead of an arc the higher energy of the spark will cause a large proportion of them to lose an electron. The atoms with one less electron in turn exhibit their own characteristic array of spectral lines, i. e., the second spectrum of vanadium, VII. Similarly, with suitable sources of excitation, spectra of higher ionization can be observed corresponding to the loss of 2, 3, etc., electrons, the total number possible being equal to the atomic number of the element in question, in the case of vanadium, 23. To date, however, nothing is known about the vanadium spectra beyond VXIV. The present volume contains the energy levels of all atomic and ionic spectra in which structure has been recognized, for the 23 chemical elements hydrogen through vanadium, H, HeI, HeII, LiI, . . . VXIV, and includes 206 spectra.

The wavelengths, or positions of the lines observed in a given spectrum are carefully measured, and estimated intensities of the lines recorded. The *wavelengths* are then converted into *wave numbers in vacuo* from standard tables.⁵ By studying differences among the *wave numbers* of the observed lines the *energy levels* can be found, since each spectral line is produced by a transition between two such *levels*. From a careful study of groups of lines that have similar characteristics, such as intensity behavior when produced at different temperatures in the laboratory, the levels involved in the production of the lines are grouped to form spectroscopic *terms*. The terms result from definite configurations and motions of the outer electrons of the atom and are explained by a well-established theory of spectral structure.⁶ For any given electron configuration the array of terms to be expected in a given spectrum can be predicted from the quantum theory. Conversely, the energy levels and the terms formed from them furnish fundamental information, based on observation, concerning the outer electrons of the atom. The energy levels are, therefore, important constants of nature.

A group of related lines produced by transitions between two complex terms was first called a *multiplet* by M. A. Catalán in 1922.⁷ The Multiplet Tables mentioned above (RMT, sec. 2) give the observed wavelengths of the lines that form the leading multiplets of many different spectra.

4. Arrangement

An attempt has been made to follow the general plan adopted by Bacher and Goudsmit in 1932, but some major changes have been introduced. In the present work the elements are arranged in order of increasing atomic number rather than in the alphabetical order of their chemical symbols. The tables on pages XL and XLI should facilitate cross reference to the earlier book. For a given element the arc spectrum is followed by the successive spark spectra in order of increasing stage of ionization, as was done previously. Gaps occurring in the run of spark spectra for a given element indicate that structure has not yet been recognized in the missing spectrum.

Contrary to the earlier arrangement, in the present compilation the energy levels of *all* spectra are listed upward from the ground state *zero*. Absolute values are not given, but can be found for series spectra by consulting the references to the analysis or by subtracting the tabulated values from the limit quoted for a given spectrum.

4.1. Headings, Remarks

For each spectrum descriptive remarks which are self-explanatory, are preceded by headings as follows: Those on the left give (1) the number of electrons in the atom, and, except for arc spectra, the isoelectronic sequence to which the spectrum belongs (see sec. 8.4); (2) the ground state of the atom with its complete electron configuration; (3) the absolute value of the ground level in cm^{-1} , i. e., the limit referred to the ground state of the ion of next higher ionization. The headings on the right give (1) the atomic number Z and (2) the ionization potential in electron volts obtained by multiplying the limit quoted

⁵ H. Kayser, *Tabelle der Schwingungszahlen*, Revised Edition (Edwards Brothers, Inc., Ann Arbor, Mich., 1944).

⁶ F. Hund, *Linienpektren und Periodisches System der Elemente* (Julius Springer, Berlin, 1927).

⁷ Phil. Trans. Roy. Soc. London (A) 223, 127 (1922); Rev. Acad. Madrid 25, 20 (1922).

on the left by the factor 0.00012395, which was recommended by Birge in 1941.^{8,9}

In the remarks the word "author" refers to the investigator who has worked on the analysis of the spectrum, in contrast to the word "writer," which applies to the present compiler of these data.

4.2. References

In 1914 W. F. Meggers started a card catalog of all literature references on the description and analysis of atomic spectra, which has been carefully kept up to date and is doubtless the most complete of its kind in existence today. This catalog, together with the valuable and extensive collection of spectroscopic reprints of Meggers and Kiess, furnish the basic material requisite to the present program.

Following the descriptive remarks, literature references are given for each spectrum. It is not the purpose of this book to list all references to the analysis of each spectrum. The writer has attempted to make a careful appraisal of the literature and to list all the references needed to cover the complete analysis, including, of course, those used in the present work, and those giving the classified lines, energy or Grotrian diagrams, and observed g -values. A few selected references to hyperfine structure and isotope shift are also included, as mentioned in sec. 2.

In many spectra important regularities have been found by an author whose name does not appear in the references quoted here. This occurs when later and more complete papers include the earlier results and references. For example, Bowen and Millikan first discussed a number of the spectra described in Edlén's Monograph,¹⁰ but only the later reference is listed. Full recognition should be given to all such contributors in spite of the arbitrary limitations imposed here.

4.3 Reference Symbols

Most of the literature references are followed by letters in parentheses, which describe the scope and content of the paper, as follows:

I P	Ionization potential.
T	Terms.
C L	Classified lines.
G D	Grotrian diagram.
E D	Energy diagram.
Z E	Zeeman effect.
I S	Isotope shift.
hfs	Hyperfine structure.

Several of these topics are frequently discussed in one paper, in which case all the symbols that are applicable are mentioned with the reference. If, for example, the symbols (I P) (T) (C L) follow a reference, it signifies that the paper gives an ionization potential, terms, and classified lines.

In a few selected cases, self-explanatory descriptions follow the reference, as, for example, in C I "(Solar data)."

Some papers are described in abstracts or letters to the editor in the Physical Review. These are indicated by (A) or (L) preceding the date in the reference, as is customary, but they should not be confused with the above symbols.

References for which no symbol is given are described in the remarks on the spectrum. Many of these are theoretical in character, as for example, the one to Racah's paper (see Ne I) which deals with $j\ell$ -coupling in the spectra of the Ne I type (sec. 5.2). Symbols have been omitted in general from references that are specialized in character as compared with those that can be more concisely described by the array of letters given above.

5. Spectroscopic Notation

Some details of spectrum analysis should perhaps be mentioned in order to explain the plan of presentation of spectroscopic data adopted here. According to the quantum theory each energy level is defined by an inner quantum number commonly known as J . The terms (groups of related levels) have multiplicities which are all odd (1, 3, 5, 7, . . .) or all even (2, 4, 6, 8, . . .) in a given spectrum. For terms of odd multiplicity the J -values are always integers, 0, 1, 2, 3, . . .; for those of even multiplicity the J -values are odd multiples of the fraction $\frac{1}{2}$, denoted as $\frac{1}{2}$, $1\frac{1}{2}$, $2\frac{1}{2}$, $3\frac{1}{2}$, etc. Terms are further de-

fined by azimuthal quantum numbers L that have for terms labeled S, P, D, F, G, H, I, K, etc., the values 0, 1, 2, 3, 4, 5, 6, 7, etc., respectively.

A term of a given kind and multiplicity consists of a definite number of energy levels whose inner quantum numbers are stipulated by the quantum theory. For example, an "S" term of multiplicity three has only one level with J -value equal to 1. This term is designated as 3S_1 . A "D" term of multiplicity four consists of four levels whose J -values are $3\frac{1}{2}$, $2\frac{1}{2}$, $1\frac{1}{2}$, $\frac{1}{2}$, respectively, designated as $^4D_{3\frac{1}{2}}$, $^4D_{2\frac{1}{2}}$, $^4D_{1\frac{1}{2}}$, $^4D_{\frac{1}{2}}$. Tables giving the J -values of terms of each multiplicity are discussed in sec. 6.7.

The designation is further described by two other quantities discussed in sec. 5.1 and sec. 5.3: (1) a prefix that

⁸ Rev. Mod. Phys. **13**, No. 4, 233 (1941).

⁹ The discrepancies between the ionization potentials in this book and those given by the writer in the RMT are, in general, due to the use of the older factor, 0.00012345, in calculating data for the Multiplet Tables.

¹⁰ Nova Acta Reg. Soc. Sci. Uppsala (IV) **9**, No. 6 (1934).

serves to distinguish terms of the same type and multiplicity from each other and which, for simpler spectra, gives information about the electron configuration, and (2) a superscript "o" denoting that a term belongs to the odd set (sec. 5.1). The complete multiplet designation of any spectral line includes all of these quantities: multiplicity, azimuthal quantum number, and inner quantum numbers for both the lower and higher energy levels involved in the production of the line.

The lines normally observed in a spectrum, i. e., the permitted lines, do not result from differences among the levels of each term and every other term, but from differences (called combinations) between two sets of terms, one "even" and one "odd." Permitted lines are further restricted by the rules governing the J -values. Only those J -value combinations between even and odd terms for which J changes by 0 or ± 1 are permitted, and normally no combinations occur between levels with $J=0$. Under special conditions "Forbidden" lines are observed, in which case these selection rules for odd and even terms and for J -values do not hold.

A relatively limited number of terms can thus account for a complex array of spectral lines. It is obviously desirable to describe these terms by a uniform notation that defines the quantum properties as completely as possible, and is also adaptable to all the varieties of spectra that have been and are likely to be observed.

A general scheme of notation was outlined in 1929,¹¹ which has been widely used. This scheme has been interpreted so freely by various investigators that a serious lack of uniformity has resulted in the literature. When this question arose in connection with the RMT the writer did not anticipate the present project, which is far wider in scope. She did, however, attempt to introduce uniformity and, in order to avoid further confusion, she has adopted here the notation of the RMT with only slight modifications. It is admittedly far from ideal, but is perhaps justifiable if it serves only to stimulate serious consideration of the question and the general adoption of a more satisfactory scheme.

The "Designation" (sec. 6.3) adopted for the less complex spectra that exhibit conspicuous series differs from that used for the more complex spectra that do not.

5.1. Series Spectra

For many elements the spectra become more complex as the degree of ionization decreases. The terms of each spark spectrum are the parent terms or "limits" of the series of terms in the spectrum of next lower degree of ionization. The term arrays resulting from the addition of s , p , d , f , etc., electrons to each limit are well known from theory (sec. 7). Consequently, for the simpler spectra the electron configurations of the observed terms

can be assigned without ambiguity by a study of the limits in the spectrum of next higher degree of ionization.

The spectrum of Ovi may be used as an illustration. Here the lowest term of Ovii, $1s^2 1S$, is so much lower than any other that no other limit need be considered. The addition of a "running" s , p , d , f , etc., electron to this state produces a series of doublet S , P° , D , F° , etc., terms in Ovi. In this case the electrons and terms are of the same type. The ground term of Ovi is $1s^2 (1S) 2s^2 S$, the next term is $1s^2 (1S) 2p^2 P^\circ$, etc., where $(1S)$ signifies the parent term or limit in Ovii. The "Designations" adopted for these terms are $2s^2 S$, $2p^2 P^\circ$, etc.¹² The number "2" in the prefix $2s$, etc., denotes the total quantum number, which depends on the shell occupied by the outer electrons giving rise to the term (see sec. 7). This number increases by unity for the series terms of a given type, as for example, for the series $2s^2 S$, $3s^2 S$, $4s^2 S$, etc.

An additional electron is effective in the production of the spectrum of Ov. The configuration $1s^2 2s^2$ gives the ground term $1S$, designated here as $2s^2 1S$; and $1s^2 2p^2$ gives the terms $2p^2 3P$, $2p^2 1D$ and $2p^2 1S$. The spectrum of Ov is more complex because, in addition, there are two low terms in Ovi, both of which are important parent terms or "limits" giving rise to terms in Ov. The addition of running electrons to these limits gives, among others, the following theoretical or predicted array of terms:

Ovi		Ov		
Config.	Limit	Added Electron	Config.	Terms
$1s^2 2s$	$2S$	$3s$	$1s^2 2s(2S)3s$	$\begin{Bmatrix} 3S \\ 1S \end{Bmatrix}$
"	"	$2p$	$1s^2 2s(2S)2p$	$\begin{Bmatrix} 3P^\circ \\ 1P^\circ \end{Bmatrix}$
"	"	$3d$	$1s^2 2s(2S)3d$	$\begin{Bmatrix} 3D \\ 1D \end{Bmatrix}$
$1s^2 2p$	$2P^\circ$	$3s$	$1s^2 2p(2P^\circ)3s$	$\begin{Bmatrix} 3P^\circ \\ 1P^\circ \end{Bmatrix}$
"	"	$3p$	$1s^2 2p(2P^\circ)3p$	$\begin{Bmatrix} 3S & 3P & 3D \\ 1S & 1P & 1D \end{Bmatrix}$
"	"	$3d$	$1s^2 2p(2P^\circ)3d$	$\begin{Bmatrix} 3P^\circ & 3D^\circ & 3F^\circ \\ 1P^\circ & 1D^\circ & 1F^\circ \end{Bmatrix}$

Terms are "odd" (denoted by the superscript "o") when the configuration contains an odd number of p , f , h , etc. electrons, $3P^\circ$, for example. In the case of Ov, since one limit is even and the other one odd, no ambiguity occurs if a designation consisting of the running electron and term is used for terms from both limits, i. e., for terms from $2S$ in Ovi, $3s^3 S$, $3s^1 S$, $2p^3 P^\circ$, $2p^1 P^\circ$,

¹¹ H. N. Russell, A. G. Shenstone, and L. A. Turner, Phys. Rev. **33**, 900 (1929).

¹² In the RMT the notation $2^2 S$, $2^2 P^\circ$, etc. was used for series of this kind when the term and running electron were of the same type.

$3d^3D$, $3d^1D$; and for terms from $^2P^\circ$ in OVI , $3s^3P^\circ$, $3s^1P^\circ$, $3p^3S$, $3p^3P$, . . . $3d^1F^\circ$. This notation has been adopted for those spectra that have two low limits, one even and one odd.

When two or more of the effective limits are all even or all odd, an addition to this notation is required. The limit terms are always listed in the term arrays (sec. 7) from lowest to highest, i. e., according to increasing value of the terms, starting from zero. In Ov the ground term is 2S and the next higher is $^2P^\circ$. Consequently, 2S is listed first in the above array and in the one on page 57. For terms from the lowest of a group of limits the running electron is used as described above. For those from the *next higher* or *second* limit a prime is affixed to the running electron, for those from the third limit a double prime, etc. The use of primes is well illustrated by the term arrays: (1) of Orv , p. 55, where the lowest limit is even and the next odd, in which case primes are first introduced for the *third* limit; and (2) that of OII , p. 50, where the primes are used for the *second* limit, since the two lowest limits are even.

With the exception of the spectra of the inert gas type (sec. 5.2), the notation giving the running electron with primes affixed as described above has been used for the spectra of all isoelectronic sequences through K and for the spectrum of CaI . The rest of the CaI sequence and the ScI , TiI , and $V I$ sequences have the notation adopted for complex spectra (secs. 5.3 and 7.5).

5.2. Inert Gases

The first spectra of the inert gases form a special class of series spectra that must be discussed separately. In these neutral atoms the last electron required to close the different shells is added. Terms are not definitely distinguishable for many types of higher series members owing to the departure from LS -coupling, and the J -values of the components of the limit term must be indicated. A detailed account of the theory of the couplings of various types will not be attempted here. Briefly summarized, when LS -coupling does not hold, jI - or jj -coupling becomes important, the Landé g -values (tables 1 to 4), (sec. 6.7) do not hold, and levels are grouped by pairs rather than by terms. For further details, special treatises on the subject should be consulted.¹³

The present volume contains two sequences of this type: $Ne I$ and $Ar I$. In these spectra the last of the six p -electrons is added and completes these shells.

¹³ E. Back and A. Landé, *Zeemaneffekt und Multiplettstruktur der Spektrallinien*, (Julius Springer, Berlin, 1925).

F. Hund, *Linienpektren und Periodisches System der Elemente* (Julius Springer, Berlin, 1927).

R. F. Bacher and S. Goudsmit, *Atomic Energy States* (McGraw-Hill Book Co., Inc., New York, N. Y. and London, 1932).

H. E. White, *Introduction to Atomic Spectra* (McGraw-Hill Book Co., Inc., New York, N. Y., and London, 1934).

E. U. Condon and G. H. Shortley, *The Theory of Atomic Spectra* (The Macmillan Co., New York, N. Y.; The University Press, Cambridge, Eng., 1935).

As stated in the remarks for NeI , Edlén suggested that a pair-coupling notation be adopted for NeI -like spectra to take into account the departure from LS -coupling. The jI -coupling notation in the general form suggested by Racah¹⁴ has, consequently, been adopted, on Shortley's suggestion. Shortley has also prepared a detailed array of the theoretical arrangement of the pairs, for the writer to use as a guide in preparing the tables of spectra of this type.

A few general remarks will suffice to explain the general plan of presentation. All levels from a given configuration are in one group. The groups are listed in order of increasing value of the smallest level in each group. Within a group the levels are paired and the pairs form two subgroups, each of which has as a limit one of the two components of a $^2P^\circ$ term, $^2P_{1/2}^\circ$ and $^2P_{3/2}^\circ$, the former being the lower. Within the subgroup members of a pair are listed together in order of increasing value of the lower member, unless they are widely separated numerically, in which case the lower pairs precede the higher member of the wide pair. Each pair consists of two levels whose J -values are known from theory, and differ by only one unit. The designation of the pair gives the running electron, followed by the mean value of the two quantum numbers given in brackets. As usual, a prime is used with the running electron to indicate the higher limit.

The spectrum of $Ne I$ may be used as an illustration. The pairs from the $3s$ -configuration form one group. The next group in order of increasing numerical value of the lowest member is $3p$, the next is $4s$, etc. Within the $3s$ group one pair having $J=2, 1$, respectively, has the limit ($^2P_{1/2}^\circ$) in $Ne II$, and is designated as $3s[1\frac{1}{2}]^\circ$, where the "o" has the usual meaning. The second pair in the $3s$ group has the higher limit ($^2P_{3/2}^\circ$) in $Ne II$ and J -values 0 and 1, respectively. The designation is, therefore, $3s'[\frac{1}{2}]^\circ$. In the group having the $3p$ -configuration the components of pair 1, 0 are widely separated, 148259 and 150919, respectively. They are listed separately in numerical order within the subgroup having the limit ($^2P_{1/2}^\circ$), each member being labeled $3p[\frac{1}{2}]$. Then follows the related subgroup $3p'[1\frac{1}{2}]$, etc., with the pairs listed in increasing order.

The spectra to which the pair-coupling applies are listed under the $Ne I$ and $Ar I$ isoelectronic sequences in table 26.

For convenience of cross reference to Bacher and Goudsmit's book and to other publications, the Paschen notation for these spectra has been retained in column 1. Unfortunately, the jI -coupling notation was not used in the RMT, but it is hoped that the style adopted there can be translated into the present form by means of the table on page xvii of that Contribution.¹⁵

¹⁴ Phys. Rev. **61**, 537 (L) (1942).

¹⁵ A *Multiplet Table of Astrophysical Interest*, Princeton Univ. Obs. Contr. No. 20 (1945).

5.3. Complex spectra

In the majority of complex spectra the terms are so numerous that it is impracticable to designate them by their configurations. For these spectra the prefixes, *a*, *b*, *c*, *d* are assigned to the low terms of each type (even or odd) and *z*, *y*, *x*, etc., to those that combine with them (odd or even). The high terms of the same type as the low ones start with the prefix *e* and continue through *f*, *g*, etc.

This notation for complex spectra is first used for Sc II in the present volume. It is also used for all subsequent spectra of the Ca I sequence and for the spectra of the Sc I, Ti I, and V I sequences. These spectra are discussed further in sec. 7.5.

In many complex spectra it is impossible to group all known levels into spectroscopic terms. Miscellaneous levels are assigned numbers, and the superscript "°" if they belong to the odd set.

6. Columns of the Table

The data on atomic energy levels are presented in a maximum number of seven columns in the tables. These columns may be described as follows, although the numbers on the left serve only as a guide to the order of presentation, since all are not needed for every spectrum.

Column	Description	Tabular Entry
1	Author	Edlén, Paschen, Author
2	Configuration	Config.
3	Designation	Desig.
4	Inner Quantum Number	<i>J</i>
5	Atomic Energy Level	Level
6	Interval	Interval
7	Observed <i>g</i> -value	Obs. <i>g</i>

6.1. Author

Column one gives the notation used in individual papers on the analysis of certain spectra. For many spectra discussed by Edlén, i. e., mostly spectra of the light elements, the heading "Edlén" is used to indicate his notation.

As stated above, the heading "Paschen" is given for spectra of the inert gas type, meaning that the column contains Paschen's notation.

Frequently "Author" or "Authors" and, occasionally, initials are used as a heading. This is explained in the remarks and references for the spectrum in question.

This column is used only when necessary to enable the reader to translate the notation in the literature into that adopted in the "Designation" column for the sake of uniformity. It is omitted for the simpler spectra and for those in which no ambiguity can occur in the interpretation of the individual papers on analysis.

6.2. Configuration

Column two gives the electron configuration. For the simpler spectra, where only one limit term is involved, the limit is not repeated in the configuration for each term.

Similarly, the electrons in closed shells are given only when necessary. For example, in Li I, p. 9, all terms have the limit (¹S) in Li II, and two electrons form the closed 1s shell. The complete configuration of the ground term 2s²S is 1s²(¹S)2s, here called 2s for brevity. Similarly, for the next term, 2p ²P°, it is 1s²(¹S)2p, called 2p, etc. For each spectrum, any electrons not mentioned in the configuration column may be found in the heading giving the ground state.

In more complex spectra, all electrons and limits needed to explain the terms are given, the limit terms being in parentheses, as usual. In C II, p. 24, for example, the term at 116537.88 has the limit (¹S) in C III, as indicated by the configuration 2s²(¹S)3s. The rules governing the use of primes for terms from different limits have been described in detail in sec. 5.1.

The *J*-value indicating the component of the limit term responsible for certain terms or levels is of considerable theoretical interest. Many papers discuss this question. No attempt has been made to list here the *J*-values for the limit terms except in the case of inert gas spectra (sec. 5.2).

6.3. Designation

The designation column has been explained in sec. 5. Spectra have been divided into three classes and a uniform designation adopted for each class. For series spectra, the running electron without or with primes is given as a prefix to the term. For inert gas spectra the *jl*-coupling notation of the related pairs of levels is used. For complex spectra the prefixes *a*, *b*, . . . *e*, *f*; *z*, *y*, *x*, . . . are given.

Miscellaneous levels are assigned numbers and odd levels are indicated throughout by the symbol "°."

Other miscellaneous designations, which are usually self-explanatory, are also used. In F I, p. 60, for example, the type of notation adopted by Edlén for miscellaneous levels from the 3*d* and 4*d* configurations, 3*d* X₂, etc., has been retained. Edlén remarks that it is impos-

sible to assign term designations to these levels because of the departure from *LS*-coupling.

6.4. Inner Quantum Number *J*

This column gives the inner quantum number *J* for each level when known, or the quantum numbers of all components of a term if the term is unresolved into its component levels. For brevity the end quantum numbers of a term are frequently given for unresolved terms. For example, the term of F II, p. 63, at 264610 is an unresolved ⁵F term. A ⁵F term consists of 5 components with *J*-values of 5, 4, 3, 2, 1, respectively. They are denoted as "5 to 1" in the column headed *J*. The *J*-values for terms of the various types, S, P, D, etc., and multiplicities are given in tables 1 and 2. A blank in this column indicates that the author has not defined the *J*-value. In sec. 6, following, *J*-values are discussed further.

As a rule, *J*-values can be determined from the observed combinations. In the spectra of Ne I and Ar I, however, Shortley has suggested that special care be taken to indicate those that are verified by observation in the case of levels produced by *f*-electrons, since some pairs overlap and some are unresolved. As an aid in the theoretical interpretation of these spectra, the *J*-values that are derived from the observed combinations involving *f*-electrons are entered in italics in the tables.

6.5. Atomic Energy Level

This column gives the atomic energy levels of the individual spectra, odd levels being in italics throughout. With the exception of H-like spectra they are, in general, *observed* values.¹⁶ In a number of spectra extrapolated values estimated from isoelectronic sequence data are also included to supplement incomplete observational results. Brackets are used to denote extrapolated values.

For every spectrum the levels are listed from the ground state zero, i. e., *absolute values* are not given in these tables. The levels are grouped by terms, or by pairs in the case of the inert gas spectra (sec. 5.2). The terms are listed in order of increasing numerical value of the smallest level in each. Miscellaneous levels are given in proper numerical order between terms. For unresolved levels the effective mean value of the components is given. For terms in which only certain components have been observed, those levels that are known are listed with the known *J*-value, and blanks occur in the table opposite the *J*-values of the missing members.

The value of the limit referred to the *ground state of the atom of next higher stage of ionization*, i. e., the limit

¹⁶ For spectra of the H sequence the values calculated by J. E. Mack from the series formula are given, as is explained in the remarks.

giving the *principal ionization potential*, is entered in bold face in the table. In spectra having terms with negative absolute values, the limit appears in the correct numerical place in the table and is followed by higher terms. More often, it appears at the end of the table, following a row of leaders which indicate that many high terms have not yet been found. The value of the limit given in the heading is repeated in the table, throughout. Two limits are given for Ne I- and Ar I-like spectra, when the absolute values of both components of the limit term ²P_{1/2, 3/2}^o are known, the lower limit being in bold-face type (see sec. 5.2).

The selection of the numerical value of the limit adopted here is frequently arbitrary, and those who are seriously interested in the best value should consult the references. The length and type of the series, the series formula used, the type of extrapolation, and many other factors affect the accuracy of the limit. The remarks contain relevant details regarding the evaluation of the limit. Higher limits, if any, can be calculated by the addition of the appropriate term values of the succeeding spectrum to the limit quoted here.

In many spectra no intersystem combinations connecting the terms of different multiplicity within a spectrum, have been observed. For these spectra a constant correction, *x*, which may be either positive or negative, must, therefore, be applied to the terms of one multiplicity, and a different constant *y* to those of another in spectra where terms of three multiplicities have been detected, in order to put all terms on the same scale. In the tables the entries "+*x*" and "+*y*" follow the levels of all such sets of terms.

If long series have been observed the relative positions of the terms of different multiplicity can be determined accurately from the series limits, and the correction *x* is small.¹⁷ In many cases series are short or lacking and the error may be considerable. Estimated relative positions of terms have, however, frequently been used in order to place all terms in the order that is approximately correct. The remarks on the spectrum and the use of brackets to denote estimated values should suffice to explain the procedure in the individual cases.

The uncertainty *x* is also occasionally used to indicate groups of detached terms that have not yet been connected by observation with the rest of the spectrum, but whose multiplicity is the same as that of terms that are known. This is true for a group of terms of Sc I, for example (p. 260).

6.6. Interval

The term intervals in this column are, with a very few exceptions, the differences between the level values of the

¹⁷ In a few spectra *x* has been omitted for this reason, as noted in the remarks.

components of terms in the preceding column. If, for a given term, the level of smallest J has the smallest numerical value, and this succession holds for all components from the lowest to the highest, the intervals are positive and the term is *normal*. On the contrary, if the level of smallest numerical value has the largest J , etc., throughout the term, the intervals are negative and the term is *inverted*. The general run of intervals is positive or negative in a given spectrum according to whether the shell of outer electrons is less than or greater than half full (see sec. 7.1), although many exceptions to this general rule occur.

If some components of a term are missing, the order in which the J -values are listed is governed either by the foregoing rules concerning the shell, or by the behavior of other series members of the same type within the spectrum or the sequence.

The J -values are always given either in increasing or decreasing order for a term, even if the term may be partially inverted. For example, a 3P term has its J -values listed either in the order 2, 1, 0 or 0, 1, 2 even if this arrangement causes the levels to be given out of numerical order. For such terms the signs of the intervals call attention to the irregularity, since both positive and negative intervals occur whenever the term is partially inverted. The term $3d\ ^5D$ of O III, p. 52, starting with the value 398135.0, is a term of this kind.

Estimated intervals are in brackets and are explained in the remarks.

6.7 Observed g -Value (Tables 1 to 4, Landé g -Values)

When a spectrum is observed in a magnetic field of suitable strength most lines are broken up into groups of related components arranged in definite patterns. The separations of the components are proportional to the magnetic field strength and to magnetic splitting factors (g -values) characteristic of the atomic energy levels. The g -values can be derived from a study of the observed patterns. These determine the multiplicity and the azimuthal and inner quantum numbers of the individual atomic energy levels. The theoretical g -values are well known for the individual levels of terms of all types. Consequently Zeeman patterns furnish one of the most reliable criteria for the correct interpretation of a complex spectrum.

Details of the theoretical and experimental aspects of this important subject will not be given here. Back and Landé, Bacher and Goudsmit, H. E. White,¹⁸ and many others discuss it.

Observed g -values are given in the last column of the tables. There is a surprising scarcity of reliable data on observed Zeeman patterns among the spectra of the light elements. The first entries in the table are for N I.

Some papers state that the analysis is confirmed by the observed Zeeman effect but give no details. The general policy is to list here only those references that give observed g -values or sufficient data from which to calculate them. The accuracy of the Zeeman material varies greatly and depends on such factors as the determination of the magnetic field used for the observational data, the resolving power of the spectroscope, the interpretation of the observed effect, and many others. As a result the listed g -values vary greatly in accuracy.

For spectra in which LS -coupling holds the observed values agree well with the Landé theoretical g -values. Because of their importance in spectrum analysis, these theoretical values are given in tables 1, 2, 3, and 4. Table 1 contains J - and g -values for terms of types S, P, D . . . Q of odd multiplicity, i. e., singlet, triplet, quintet, . . . undecet terms. For example, the theoretical g -value of a 3F_4 level is 1.250; that of a 7I_6 level is 1.143. Since the data are identical for odd and even terms alike, one table suffices for both sets of terms. Table 2 gives similar data for terms of even multiplicity: doublets, quartets, . . . decets.

For the convenience of those who are analyzing spectra, the theoretical g -values are also given in order of increasing numerical value followed by the designation of the level or levels for each g , for terms of odd multiplicity in table 3; and for those of even multiplicity in table 4. These g -values are quoted from the "Tables of Theoretical Zeeman Effects" by Kiess and Meggers,¹⁹ supplemented by their unpublished data for terms of multiplicity greater than eight.²⁰ Their tables give also the theoretical Zeeman patterns for practically all of the multiplet designations that have been observed within the range of multiplicity they cover.

Tables of theoretical g -values for jj -coupling may be found in papers by J. B. Green and his collaborators.²¹

Finally, the date of completion of the manuscript of each spectrum is given at the end of the table of energy levels of the spectrum.

¹⁸ E. Back and A. Landé, *Zeemaneffekt und Multiplettstruktur der Spektrallinien* (Julius Springer, Berlin, 1925).

F. Hund, *Linienpektren und Periodisches System der Elemente* (Julius Springer, Berlin, 1927).

R. F. Bacher and S. Goudsmit, *Atomic Energy States* (McGraw-Hill Book Co., Inc., New York, N. Y. and London, 1932).

H. E. White, *Introduction to Atomic Spectra* (McGraw-Hill Book Co., Inc., New York, N. Y., and London, 1934).

E. U. Condon and G. H. Shortley, *The Theory of Atomic Spectra* (The Macmillan Co., New York, N. Y.; The University Press, Cambridge, Eng., 1935).

¹⁹ Bur. Std. J. Res. 1, 641, RP23 (1928).

²⁰ They have extended their tables of theoretical Landé g -values to include all types of terms and multiplicities (up to ^{11}Q) that are likely to be needed, in order that tables 1 to 4 may be complete. The writer is indebted to them for this useful contribution.

²¹ Phys. Rev. 52, 736 (1937); 54, 876 (1938); 58, 1094 (1940); 59, 72 (1941); 64, 151 (1943).

7. Tables of Predicted and Observed Arrays of Terms

With the exception of the simpler spectra and of those for which the analysis is seriously incomplete, arrays of observed terms are given following the individual tables of energy levels, the first being that of Be I, p. 13.

As stated above, the arrays of terms to be expected for a given configuration are well known from theory. A comparison of the terms observed in a given spectrum with those predicted reveals at once the completeness of the analysis. To facilitate this comparison, arrays of predicted terms arranged similarly to those of the observed terms are included here.

7.1. Shells

In the discussion of notation (sec. 5) reference was made to the "shells" of electrons and their importance in the production of spectroscopic terms. A clear description of these shells is quoted from White,²² p. 80: "The various electrons are classified under so-called *shells* of electrons. *All electrons belonging to the same shell are characterized by the same total quantum number n*"

"The shells $n=1, 2, 3, 4, \dots$ are sometimes called (from x-ray spectra) the *K, L, M, N, . . .* shells, respectively."

"The electrons in any shell n are further divided into *subshells* so that *electrons belonging to the same subshell have the same azimuthal quantum number l* . Electrons for which $l=0, 1, 2, 3, \dots$ are called *s, p, d, f, . . .* electrons, respectively, . . .". For example, *2s* is used to specify one electron with $l=0$ and in the shell $n=2$.

No shell can contain more than 2 type-*s* electrons starting with $n=1$, 6 type-*p* electrons starting with $n=2$, 10 type-*d* electrons starting with $n=3$, or 14 type-*f* electrons starting with $n=4$, etc. The successive periods 1 to 7 in the periodic system (sec. 8.3) can, therefore, contain only 2, 8, 8, 18, 18, 32, and 32 elements, respectively. These are consequences of Pauli's exclusion principle.

This is illustrated in the following brief tabular excerpt from White's complete Table of Electron Configurations:

Shell		K $n=1$	L $n=2$	
		$l=0$	$l=0$	$l=1$
1	H	1s		
2	He	1s ²		
3	Li	1s ²	2s	
4	Be	1s ²	2s ²	
5	B	1s ²	2s ²	2p
6	C	1s ²	2s ²	2p ²
-----	---	---	-----	-----

²² H. E. White, *Introduction to Atomic Spectra* (McGraw-Hill Book Co., Inc., New York, N. Y., and London, 1934).

Superscripts denote the number of electrons of a given type. Where no superscript is given unity is understood. He I, for example, has two electrons of the type 1s, as indicated by 1s² in the above array. These similar electrons are known as *equivalent* electrons. The terms produced by *equivalent* and *nonequivalent* electrons and detailed discussions of Pauli's exclusion principle may be found in many standard treatises on atomic spectra.²³

All spectra having the same shells of electrons are similar. An *isoelectronic sequence* consists of spectra of different elements having the same shells of electrons. Each *arc* spectrum sets the pattern for the sequence, so far as the effective electrons are concerned. For example, the spectra of Be I, B II, C III, etc., form an isoelectronic sequence for which Be I, the *arc* spectrum of beryllium, sets the pattern, i. e., the Be I isoelectronic sequence. In B II, the first spark spectrum of B, the boron atoms have lost the outer electron, 2p. This spectrum, therefore, resembles that of Be I having two 2s electrons (denoted by 2s²) outside the closed shell 1s². Similarly the carbon atoms have lost both outer 2p-electrons when the spectrum of C III is observed. This spectrum thus belongs in the same sequence. An array of predicted terms of each arc spectrum suffices, therefore, for all spectra of the sequence, as, for example, Be I.

No arrays are given for spectra of the H, He I, Li I, and similar sequences. Since only 1s, 1s², and 2s electrons are involved, the arrays of predicted and observed terms are simple.

7.2. Arrays of Predicted Terms of the Sequences Be I Through Ne I (Tables 5 to 11)

Starting with Be I, predicted arrays of terms of the isoelectronic sequences from Be through Ne are given in the following tables (pages xxviii to xxxi):

Table	Sequence
5	Be I
6	B I
7	C I
8	N I
9	O I
10	F I
11	Ne I

In all of these tables the closed shells are indicated immediately under the heading "Config." ("1s²+" for this group of spectra). The tables are divided into two sections. The upper half gives the terms from equivalent

²³ H. N. Russell, *Phys. Rev.* **29**, 782 (1927).

R. C. Gibbs, D. T. Wilbur, and H. E. White, *Phys. Rev.* **29**, 790 (1927).

F. Hund, *Linienpektren und Periodisches System der Elemente* (Julius Springer, Berlin, 1927).

C. L. B. Shudeman, *J. Franklin Inst.* **224**, 501 (1937). (Terms from equivalent *g, h*, and *i* electrons.)

electrons and, for simpler spectra, the first low series members. The lower half indicates the series to be expected from the various limit terms (sec. 5.1), with the running electron denoted as nx , where n is the total quantum number, and x the type of electron, s, p, d, f, \dots , etc.

The quantities n and x are indicated in the headings, nx ($n \geq 3$), etc., above the columns of the tables and are evaluated in the arrays of observed terms of the separate spectra of the sequence. For example, the ns^3S series of BeI , p. 13, with the configuration $2s(2S)nx$ has been observed from $n=3$ through $n=8$.

Many more terms can be predicted than are likely to be observed. The present tables are designed to contain enough predicted terms to suffice for all terms thus far observed in any spectrum of the sequence.

7.3. Arrays of Predicted Terms of the Sequences MgI Through Al , (Tables 12 to 18)

Starting with MgI , arrays of predicted terms of the isoelectronic sequences from Mg through Al are given in the following tables (pages xxxii to xxxv):

Table	Sequence
12	MgI
13	AlI
14	SiI
15	Pi
16	Si
17	ClI
18	Al

A comparison of these tables with the set described above, tables 5 to 11, shows that the same terms are predicted for spectra having the same numbers and types of electrons outside the closed shells. Beginning with table 12, the closed shells are $1s^2 2s^2 2p^6$ (entered directly under the heading "Config." in the tables). The total quantum number n of the running electron is one unit larger, but the term arrays are identical for similar spectra in the two sets of tables. For example, tables 5 and 12, 6 and 13, etc., are alike, except for the total quantum numbers and for the number of predicted terms included, which is governed by the terms that have been observed within the sequence.

7.4. Arrays of Predicted Levels of the NeI and Ar Sequences (Tables 11 and 18)

These tables give both predicted terms (LS -coupling) and predicted pairs of levels (jl -coupling) sec. 5.2. In the arrays of predicted and observed pairs of levels for these spectra, the pairs are listed in the general order of increasing value of the lower member of the pair, as suggested by Shortley. As some spectra in this sequence are

of an intermediate type, more nearly LS -coupling, this order is not always obeyed numerically among the observed levels, but is retained in these tables for uniformity.

Similarly, in all of these tables (5 to 18, inclusive) and the corresponding arrays of observed terms, the limit terms are listed in the general order of increasing numerical value with primes added to indicate higher limits, as described in section 5.1.

7.5. Arrays of Predicted Terms of the Sequences CaI Through V_I (Tables 19 to 22)

Brief mention has been made of the special notation adopted for complex spectra (sec. 5.3). An examination of the tables for the sequences CaI , ScI , TiI , and V_I , tables 19 to 22, inclusive, reveals the rapid increase in the number of terms after d electrons are included in the structure of the unexcited atom. The use of primes is retained to indicate the different limits in CaI and in table 19. For $ScII$ and subsequent spectra in the sequence, the notation for complex spectra is introduced (see below). Since the limits are carefully specified, no difficulty should arise in comparing the arrays of observed and predicted terms in this sequence.

For the configurations involving equivalent electrons, listed in the upper section of each array, Pauli's principle restricts the array of resulting terms, and the latter cannot be unequivocally assigned to specific limits.

When only s and p electrons appear in the low configurations the ground state is always to be found in the upper section, but in the lower, when d electrons are present in a configuration involving one s electron. Examples among arc spectra may be found in table 23, and others occur for singly ionized atoms.

Beginning with the ScI sequence terms from eight limits must be considered. For this reason, a simple type of prefix a, b, c, \dots, z, y, x , etc., is adopted for the terms from the different limits. In the TiI group 15 limits must be handled, and in V_I the number increases to 22. For these complex spectra the limits in the tables of predicted terms are tabulated in order of increasing numerical value of the terms in the arc spectrum of the sequence, TiI for example. The same order does not necessarily apply to the other spectra in the sequence. In the arrays of observed terms the prefixes a, b , etc., of the *limit* terms are given in order to avoid confusion in comparing the different sets of tables.

As the complexity of the spectra increases there is a serious overlapping of families of terms from the various limits. The assignment of electron configurations is ambiguous in many cases. Beginning with TiI , a number of question marks and colons appear in the arrays of observed terms, denoting the uncertainty of many suggested configurations.

8. The Periodic Table

8.1. The Chemical Elements by Atomic Number, Ionization Potentials (Table 23)

In the present work the elements are handled in order of increasing atomic number and they are listed in this order in table 23. Column one gives this number, Z ; column two, the name of the element; and column three, the chemical symbol. Columns four and five give, respectively, the principal ionization potential and configuration of the ground state of the neutral atom. For elements with $Z > 23$, i. e., for those beyond the range of the present volume, these data are taken from table 1, columns 5 and 9, respectively, of the key to the Periodic Chart of the Atoms revised in 1947 by Meggers.²⁴ Additional data on the ground states of the rare earths are given in his paper on this subject.²⁵

8.2. The Chemical Elements by Chemical Symbol (Table 24)

Bacher and Goudsmit arranged the spectra in the alphabetical order of the chemical symbol of the element. Table 24 gives the elements in this order, with the chemical symbol in column one followed by the name of the element in column 2 and the atomic number in column 3.

8.3. The Periodic System (Table 25)

The Periodic System in table 25 is arranged in the form suggested by Catalán, who generously furnished an unpublished copy for inclusion here.

8.4. Index—Isoelectronic Sequences (Table 26)

This table contains the index to the data in Volume I of this work, the spectra from H through V. In the left

margin the atomic number is given, followed by the chemical symbol. Across the top the successive stages of ionization appear, I denoting arc spectra, II first spark spectra, III second spark spectra, etc. The numbers in the table indicate the pages on which the individual spectra may be found. For example, F VIII is on page 75.

In this table, isoelectronic spectra appear on the diagonals. Every other diagonal is printed in bold face type in order to emphasize the spectra of each sequence. For example, Si x belongs to the O I sequence, printed in bold-face along the diagonal. Similarly, Mg vi can be traced to N I along the diagonal not printed in bold face. Blanks occur for spectra that have not yet been analyzed.

No sequences are carried beyond V in this volume, but they will be continued in later volumes and indicated in tables arranged similarly to this one. The sequences started in Volume I but not completed there are listed below. The last spectrum in each sequence for which any data on analysis are known is indicated.

Sequence	Spectrum	Sequence	Spectrum
Ne I	Co xviii	Cl I	Ni xii
Na I	Cu xix	Ar I	Fe ix
Mg I	Co xvi	K I	Fe viii
Al I	Ni xvi	Ca I	Ni ix
Si I	Ni xv	Sc I	Ni viii
P I	(V ix) ¹	Ti I	Ni vii
S I	Ni xiii	V I	Cu vii

¹ This sequence is completed in the present volume.

9. Future Investigations

9.1. Need for Further Analysis

During the course of this compilation many interesting problems have presented themselves. The gaps in the sequences call attention to some spectra in which no structure has as yet been recognized. Within the sequences these gaps include the following spectra: Ne vii, viii, ix; Na x; S xi; Cl xii, xiii; Ar xii, xiii; K xii, xiii, xiv; Ca xiv; and V x. If, in addition, Fe ix and Ne x could be observed, the spectra of all possible stages of ionization would be represented for these two elements.

A careful study of the configurations in which a $3d$ electron becomes effective, is desirable. In the Fe I se-

quence the terms with $3d$ and $4d$ electrons for Na iii, Mg iv, Al v, and Si vi should be verified, as there are marked irregularities along this sequence.

In Si I the $3d\ ^3D^o$ term is lower than $3p\ ^3D^o$, but the reverse is true for the rest of the sequence.

In the P I sequence the configuration assignments of terms in which $3p^4$, $3d$, and $4s$ electrons are involved, should be examined along the sequence. More observations are also needed to verify the extensive extrapolations from K v on.

Similar remarks apply to some spectra of the Cl I sequence, particularly to Ca iv, where various authors disagree on the interpretation. Analogous terms along this sequence are strikingly irregular as regards both position and intervals. Many such irregularities could be pointed out. It is hoped that the present work will stimulate further study along these lines.

²⁴ W. M. Welch Scientific Co., 1515 Sedgwick St., Chicago 10, Ill., U. S. A. (Chart and key, \$7.50; key, \$1.00). For Mn I and Mo I Catalán's revised values are quoted. The data on Te I are from Meggers.

²⁵ Electron Configurations of "Rare-Earth" Elements, Science 105, 514, No. 2733 (May 16, 1947).

The arrays of observed terms enable one to detect a number of conspicuous missing terms whose positions can be estimated by analogy with neighboring related terms. For example, Russell²⁶ has suggested that the $3d'''^2G$ term in O IV might be found. To quote him "It should give a strong combination with $3p'''^2F^\circ$, lying in the violet or near ultraviolet." Similarly, the absence of the $3d^2F$ term of Cl III is conspicuous. Russell has also commented on the incompleteness of the analyses of Si III and S IV.

In He I the term $11s^1S$ is missing from the series. In Mg I Shortley has called attention to the fact that the triplets are higher than the singlets, an anomaly that appears to be unexplained.

The general need for further analysis can perhaps best be visualized by a comparison of the arrays of observed and predicted terms of the various spectra. This procedure enables the user to grade each analysis for himself. For spectra whose energy levels are not yet tabulated for this program it is recommended that he consult the existing surveys of spectrum analysis.²⁷

Perhaps the most urgent needs of the astrophysicist are extensions to the work on the second and third spark spectra in the first long period (except for Fe III, which is well known). Many spectra of the heavier elements are incompletely analyzed and much work remains to be done on the highly complex spectra of the rare earths.

9.2. Term Intervals

A careful examination of the term intervals within a spectrum and in related spectra affords a useful check on the correctness of the analysis. In regular terms the intervals are roughly proportional to the larger J -values of the term, and term separations of similar terms usually increase smoothly along the sequence. Enough data are presented here for an extensive survey of this subject. The theoretical as well as observational aspects of this topic and its important relation to configuration assignments need not be emphasized to workers on spectrum analysis.

9.3. Series Spectra—Rydberg Denominators

Requests have been made for a tabulation of absolute term values and Rydberg denominators of the series members of each spectrum in which series have been detected, including the J -values of the limit terms. The need for a critical compilation of this material is fully appreciated. It is felt, however, that such a project can best be handled

in a program restricted to the study of series in atomic spectra. Standard treatises such as Fowler's *Report on Series in Line Spectra* and Paschen-Götze's *Seriengesetze der Linienspektren*, the paper by Catalán and Poggio,²⁸ etc., together with other references included under the separate spectra should provide some data for those who are interested.

9.4. Observed Zeeman Patterns

A glance at the data on Zeeman effect in this volume alone, reveals a glaring need of further observations. The first entry of g -values occurs in the spectrum of N I. An outstanding example may be found in Ti I. The best observed g -values obtainable from existing data are given, and they serve remarkably well to confirm the analysis. For Ti, and also for other elements, however, Harrison²⁹ has made extensive observations that doubtless show many excellently resolved patterns and would yield precise observed g -values, but his data for a number of complex spectra have not yet been utilized. A wealth of information is in store for future study in this field.

9.5. Energy or Grotrian Diagrams

There have been urgent requests to prepare a homogeneous set of energy diagrams to accompany these tables. This topic is handled very inadequately here. If the individual authors have included either an energy level diagram or a Grotrian diagram,³⁰ this fact is indicated by the symbol (E D) or (G D) following the references. If not, recourse to general references such as Grotrian's classical publication³¹ or White's *Introduction to Atomic Spectra*³² must be had. Readers are warned that the existing diagrams are far from uniform in style and scale and that many of them are not up to date, i. e., they do not represent the analysis as given in the tables. In many cases, the most notable being probably that of Al, the writer has been unable to locate diagrams representing the analysis.

The present work would be seriously delayed by the inclusion of diagrams, but the energy levels as recorded here furnish the requisite material for such a project.

Only a few of the many interesting subjects for future investigation have been touched upon. If this work provides the inspiration and stimulus for at least some of them, it will have been justified.

²⁸ Zeit. Phys. 102, 461 (1936).

²⁹ Reports on Progress in Physics 8, 228 (1941).

³⁰ In energy diagrams only the positions of the levels or terms are indicated. In Grotrian diagrams lines indicating observed combinations connect the terms.

³¹ *Graphische Darstellung der Spektren von Atomen und Ionen mit ein, zwei und drei Valenzelektronen*, Part II (Julius Springer, Berlin, 1928).

³² H. E. White, *Introduction to Atomic Spectra* (McGraw-Hill Book Co., Inc., New York, N. Y., and London, 1934).

²⁶ Letter (Aug. 1947).

²⁷ W. F. Meggers, J. Opt. Soc. Am. 36, 433 (1946); C. E. Moore, RMT (1945).

10. Acknowledgments

Many scientific workers and many institutions at home and abroad are represented in this work. The cordial collaboration and generous supply of unpublished material have been extremely gratifying.

Members of the National Research Council Committee on Line Spectra of the Elements have given enthusiastic support to the program. The chairman, H. N. Russell, has placed at the disposal of the writer the large collection of spectroscopic data accumulated at Princeton from the time the committee was formed in 1924. He has furnished unpublished analyses (Ca I, Sc I, Ti I, Ti II) and read all of the manuscript. Throughout the work he has been a valued and keenly interested consultant.

This undertaking has been made possible by the enthusiastic support of E. U. Condon, Director of the Bureau of Standards, and W. F. Meggers, Chief of the Spectroscopy Section. The personal interest taken by Dr. Condon has been a continual source of encouragement. The careful supervision and valued suggestions of Meggers, based on his wide experience and expert judgment, greatly enhance the value of this Circular. C. C. Kiess has also been ever ready to give the writer unpublished material (Ni, OI) and authoritative and helpful suggestions on many important and troublesome questions. Other members of the Committee who have responded generously with data and stimulated further research for this program are J. E. Mack, who calculated all of the data on the spectra of the H sequence especially for inclusion here; and A. G. Shenstone, who submitted important unpublished results on C I, and Ca II.

The most extensive contributions in manuscript form have come from Sweden, from B. Edlén and his colleagues. The writer had the benefit of a conference with Edlén during his visit to Washington shortly after this project had been started. From that time he has continuously supplied unpublished analyses and valuable comments as each section of the book was being prepared. His contributions include data on selected spectra from Be through O, on all the spectra of F, and complete term arrays of the arc spectra of Ne, S, and A. It has also been possible to include the spectra of higher ionization of Al, Si, and S only because E. Ferner submitted his unpublished manuscript on these spectra. H. A. Robinson supplied his material on the spectra P VI through P XIII

together with comments on related spectra of Ne through Si; and K. Lidén furnished his data on F I.

The writer has had much helpful advice from G. Shortley on spectra of the Ne I and Ar sequences. M. A. Catalán of the University of Madrid has been a most helpful consultant throughout his entire stay in the United States. He calculated the g -values of Sc I, Sc II and Ti II for inclusion here.

Manuscripts by H. R. Kratz (K I), by K. W. Meissner, L. G. Mundie and P. Stelson (Li I), by E. R. Thackeray (Na I), by W. E. Lamb, Jr., and R. C. Retherford (H), by H. E. Clearman, Jr., (B I) and by F. Rohrlach (Ti I); and a reprint on Ni sent from Japan by T. Takamine have been submitted especially for use in connection with this program. The writer has attempted to record her gratitude to each one in the pages of the book itself.

No project of this kind can be completed without the cooperation of experts in many lines. One of the greatest rewards has been the pleasure afforded by these contacts. Miss Sarah A. Jones, Librarian at the Bureau, and her competent staff deserve special mention for the splendid assistance they have so willingly given in locating hundreds of references. Mrs. Isabel D. Murray has also provided much expert technical assistance.

The details of publication of spectroscopic data such as those included here present a most taxing and difficult problem; one which has been ably and efficiently handled by Publications Section of the Bureau, the Department of Commerce, and the Government Printing Office. The painstaking care, cordial cooperation, and skill of J. L. Mathusa and his staff in the Publications Section of the Bureau are lasting contributions that can be fully appreciated only by the many users of this Circular. In the Department of Commerce, V. Vasco, and, in the Government Printing Office, H. D. Merold, have been equally cooperative. The book reflects their personal interest and skill and those of all whose services they have enlisted.

It is a pleasure to the writer to record here her appreciation of the enormous amount of assistance all have so graciously given her.

She is also extremely grateful to her husband, B. W. Sitterly, for his many helpful suggestions and cordial cooperation throughout this work.

TABLE 1. LANDÉ g -VALUES

Term	Multiplicity											
	Singlets		Triplets		Quintets		Septets		Nonets		Undecets	
	1		3		5		7		9		11	
	<i>J</i>	<i>g</i>	<i>J</i>	<i>g</i>	<i>J</i>	<i>g</i>	<i>J</i>	<i>g</i>	<i>J</i>	<i>g</i>	<i>J</i>	<i>g</i>
S	0	0/0	1	2. 000	2	2. 000	3	2. 000	4	2. 000	5	2. 000
P	1	1. 000	2	1. 500	3	1. 667	4	1. 750	5	1. 800	6	1. 833
			1	1. 500	2	1. 833	3	1. 917	4	1. 950	5	1. 967
			0	0/0	1	2. 500	2	2. 333	3	2. 250	4	2. 200
D	2	1. 000	3	1. 333	4	1. 500	5	1. 600	6	1. 667	7	1. 714
			2	1. 167	3	1. 500	4	1. 650	5	1. 733	6	1. 786
			1	0. 500	2	1. 500	3	1. 750	4	1. 850	5	1. 900
					1	1. 500	2	2. 000	3	2. 083	4	2. 100
					0	0/0	1	3. 000	2	2. 667	3	2. 500
F	3	1. 000	4	1. 250	5	1. 400	6	1. 500	7	1. 571	8	1. 625
			3	1. 083	4	1. 350	5	1. 500	6	1. 595	7	1. 661
			2	0. 667	3	1. 250	4	1. 500	5	1. 633	6	1. 714
					2	1. 000	3	1. 500	4	1. 700	5	1. 800
					1	0. 000	2	1. 500	3	1. 833	4	1. 950
							1	1. 500	2	2. 167	3	2. 250
							0	0/0	1	3. 500	2	3. 000
G	4	1. 000	5	1. 200	6	1. 333	7	1. 429	8	1. 500	9	1. 556
			4	1. 050	5	1. 267	6	1. 405	7	1. 500	8	1. 569
			3	0. 750	4	1. 150	5	1. 367	6	1. 500	7	1. 589
					3	0. 917	4	1. 300	5	1. 500	6	1. 619
					2	0. 333	3	1. 167	4	1. 500	5	1. 667
							2	0. 833	3	1. 500	4	1. 750
							1	-0. 500	2	1. 500	3	1. 917
									1	1. 500	2	2. 333
									0	0/0	1	4. 000
H	5	1. 000	6	1. 167	7	1. 286	8	1. 375	9	1. 444	10	1. 500
			5	1. 033	6	1. 214	7	1. 339	8	1. 431	9	1. 500
			4	0. 800	5	1. 100	6	1. 286	7	1. 411	8	1. 500
					4	0. 900	5	1. 200	6	1. 381	7	1. 500
					3	0. 500	4	1. 050	5	1. 333	6	1. 500
							3	0. 750	4	1. 250	5	1. 500
							2	0. 000	3	1. 083	4	1. 500
									2	0. 667	3	1. 500
									1	-1. 000	2	1. 500
											1	1. 500
											0	0/0
I	6	1. 000	7	1. 143	8	1. 250	9	1. 333	10	1. 400	11	1. 455
			6	1. 024	7	1. 179	8	1. 292	9	1. 378	10	1. 445
			5	0. 833	6	1. 071	7	1. 232	8	1. 347	9	1. 433
					5	0. 900	6	1. 143	7	1. 304	8	1. 417
					4	0. 600	5	1. 000	6	1. 238	7	1. 393
							4	0. 750	5	1. 133	6	1. 357
							3	0. 250	4	0. 950	5	1. 300
									3	0. 583	4	1. 200
									2	-0. 333	3	1. 000
											2	0. 500
											1	-1. 500
K	7	1. 000	8	1. 125	9	1. 222	10	1. 300	11	1. 364	12	1. 417
			7	1. 018	8	1. 153	9	1. 256	10	1. 336	11	1. 402
			6	0. 857	7	1. 054	8	1. 194	9	1. 300	10	1. 382
					6	0. 905	7	1. 107	8	1. 250	9	1. 356
					5	0. 667	6	0. 976	7	1. 179	8	1. 319
							5	0. 767	6	1. 071	7	1. 268
							4	0. 400	5	0. 900	6	1. 191
									4	0. 600	5	1. 067
									3	0. 000	4	0. 850
											3	0. 417
											2	-0. 667

TABLE 1. LANDÉ g -VALUES—Continued

Term	Multiplicity											
	Singlets 1		Triplets 3		Quintets 5		Septets 7		Nonets 9		Undecets 11	
	<i>J</i>	<i>g</i>	<i>J</i>	<i>g</i>	<i>J</i>	<i>g</i>	<i>J</i>	<i>g</i>	<i>J</i>	<i>g</i>	<i>J</i>	<i>g</i>
L	8	1. 000	9 8 7	1. 111 1. 014 0. 875	10 9 8 7 6	1. 200 1. 133 1. 042 0. 911 0. 714	11 10 9 8 7 6 5	1. 273 1. 227 1. 167 1. 083 0. 964 0. 786 0. 500	12 11 10 9 8 7 6 5 4	1. 333 1. 303 1. 264 1. 201 1. 139 1. 036 0. 881 0. 633 0. 200	13 12 11 10 9 8 7 6 5 4 3	1. 385 1. 365 1. 341 1. 309 1. 267 1. 208 1. 125 1. 000 0. 800 0. 450 -0. 250
M	9	1. 000	10 9 8	1. 100 1. 011 0. 889	11 10 9 8 7	1. 182 1. 118 1. 033 0. 917 0. 750	12 11 10 9 8 7 6	1. 250 1. 205 1. 145 1. 067 0. 958 0. 804 0. 571	13 12 11 10 9 8 7 6 5	1. 308 1. 276 1. 235 1. 182 1. 111 1. 014 0. 875 0. 667 0. 333	14 13 12 11 10 9 8 7 6 5 4	1. 357 1. 335 1. 308 1. 273 1. 227 1. 167 1. 083 0. 964 0. 786 0. 500 0. 000
N	10	1. 000	11 10 9	1. 091 1. 009 0. 900	12 11 10 9 8	1. 167 1. 106 1. 027 0. 902 0. 778	13 12 11 10 9 8 7	1. 231 1. 186 1. 129 1. 055 0. 906 0. 819 0. 625	14 13 12 11 10 9 8 7 6	1. 236 1. 253 1. 212 1. 159 1. 091 1. 000 0. 875 0. 696 0. 429	15 14 13 12 11 10 9 8 7 6 5	1. 333 1. 310 1. 280 1. 244 1. 197 1. 136 1. 056 0. 944 0. 786 0. 548 0. 167
O	11	1. 000	12 11 10	1. 083 1. 008 0. 909	13 12 11 10 9	1. 154 1. 096 1. 023 0. 927 0. 800	14 13 12 11 10 9 8	1. 214 1. 170 1. 115 1. 045 0. 955 0. 833 0. 667	15 14 13 12 11 10 9 8 7	1. 267 1. 233 1. 192 1. 141 1. 076 0. 991 0. 878 0. 722 0. 500	16 15 14 13 12 11 10 9 8 7 6	1. 312 1. 288 1. 257 1. 220 1. 173 1. 114 1. 036 0. 933 0. 792 0. 589 0. 286
Q	12	1. 000	13 12 11	1. 077 1. 006 0. 917	14 13 12 11 10	1. 143 1. 088 1. 019 0. 932 0. 818	15 14 13 12 11 10 9	1. 200 1. 157 1. 104 1. 038 0. 955 0. 845 0. 700	16 15 14 13 12 11 10 9 8	1. 250 1. 217 1. 176 1. 126 1. 064 0. 985 0. 882 0. 744 0. 556	17 16 15 14 13 12 11 10 9 8 7	1. 294 1. 268 1. 238 1. 200 1. 154 1. 096 1. 023 0. 927 0. 800 0. 625 0. 375

TABLE 2. LANDÉ q -VALUES

Term	Multiplicity									
	Doublets		Quartets		Sextets		Octets		Decets	
	2		4		6		8		10	
	<i>J</i>	<i>g</i>	<i>J</i>	<i>g</i>	<i>J</i>	<i>g</i>	<i>J</i>	<i>g</i>	<i>J</i>	<i>g</i>
S	½	2. 000	1½	2. 000	2½	2. 000	3½	2. 000	4½	2. 000
P	1½	1. 333	2½	1. 600	3½	1. 714	4½	1. 778	5½	1. 818
	½	0. 667	1½	1. 733	2½	1. 886	3½	1. 937	4½	1. 960
			½	2. 667	1½	2. 400	2½	2. 286	3½	2. 222
D	2½	1. 200	3½	1. 429	4½	1. 556	5½	1. 636	6½	1. 692
	1½	0. 800	2½	1. 371	3½	1. 587	4½	1. 697	5½	1. 762
			1½	1. 200	2½	1. 657	3½	1. 809	4½	1. 879
			½	0. 000	1½	1. 867	2½	2. 057	3½	2. 095
F					½	3. 333	1½	2. 800	2½	2. 572
	3½	1. 143	4½	1. 333	5½	1. 455	6½	1. 538	7½	1. 600
	2½	0. 857	3½	1. 238	4½	1. 434	5½	1. 552	6½	1. 631
			2½	1. 029	3½	1. 397	4½	1. 576	5½	1. 678
			1½	0. 400	2½	1. 314	3½	1. 619	4½	1. 758
					1½	1. 067	2½	1. 714	3½	1. 905
					½	—0. 667	1½	2. 000	2½	2. 229
G							½	4. 000	1½	3. 200
	4½	1. 111	5½	1. 273	6½	1. 385	7½	1. 467	8½	1. 529
	3½	0. 889	4½	1. 172	5½	1. 343	6½	1. 456	7½	1. 537
			3½	0. 984	4½	1. 273	5½	1. 441	6½	1. 549
			2½	0. 571	3½	1. 143	4½	1. 414	5½	1. 566
					2½	0. 857	3½	1. 365	4½	1. 596
					1½	0. 000	2½	1. 257	3½	1. 651
							1½	0. 933	2½	1. 772
							½	—1. 333	1½	2. 133
									½	4. 667
H	5½	1. 091	6½	1. 231	7½	1. 333	8½	1. 412	9½	1. 474
	4½	0. 909	5½	1. 133	6½	1. 282	7½	1. 388	8½	1. 467
			4½	0. 970	5½	1. 203	6½	1. 354	7½	1. 459
			3½	0. 667	4½	1. 071	5½	1. 301	6½	1. 446
					3½	0. 825	4½	1. 212	5½	1. 427
					2½	0. 286	3½	1. 048	4½	1. 394
							2½	0. 686	3½	1. 333
							1½	—0. 400	2½	1. 200
I									1½	0. 800
									½	—2. 000
	6½	1. 077	7½	1. 200	8½	1. 294	9½	1. 368	10½	1. 429
	5½	0. 923	6½	1. 108	7½	1. 239	8½	1. 337	9½	1. 414
			5½	0. 965	6½	1. 159	7½	1. 294	8½	1. 393
			4½	0. 727	5½	1. 035	6½	1. 231	7½	1. 365
					4½	0. 828	5½	1. 133	6½	1. 323
					3½	0. 444	4½	0. 970	5½	1. 259
							3½	0. 667	4½	1. 152
K							2½	0. 000	3½	0. 952
									2½	0. 514
									1½	—0. 800
	7½	1. 067	8½	1. 176	9½	1. 263	10½	1. 333	11½	1. 391
	6½	0. 933	7½	1. 090	8½	1. 207	9½	1. 298	10½	1. 371
			6½	0. 964	7½	1. 129	8½	1. 251	9½	1. 343
			5½	0. 769	6½	1. 015	7½	1. 184	8½	1. 307
					5½	0. 839	6½	1. 087	7½	1. 255
					4½	0. 545	5½	0. 937	6½	1. 179
							4½	0. 687	5½	1. 063
						3½	0. 222	4½	0. 869	
								3½	0. 508	
								2½	—0. 286	

TABLE 3. LANDÉ g -VALUES—TERMS OF ODD MULTIPLICITY IN ORDER OF INCREASING g

g	Desig.	g	Desig.	g	Desig.	g	Desig.
-1.500	$^{11}\text{I}_1$	0.744	$^9\text{Q}_9$	0.955	$^7\text{O}_{10}$ $^7\text{Q}_{11}$	1.076	$^9\text{O}_{11}$
-1.000	$^9\text{H}_1$	0.750	$^3\text{G}_3$ $^5\text{M}_7$ $^7\text{H}_3$	0.958	$^7\text{M}_8$	1.077	$^3\text{Q}_{13}$
-0.667	$^{11}\text{K}_2$		$^7\text{I}_4$	0.964	$^7\text{L}_7$ $^{11}\text{M}_7$	1.083	$^3\text{F}_3$ $^3\text{O}_{12}$ $^7\text{L}_3$
-0.500	$^7\text{G}_1$	0.767	$^7\text{K}_5$	0.976	$^7\text{K}_6$		$^9\text{H}_3$ $^{11}\text{M}_8$
-0.333	$^9\text{I}_2$	0.778	$^5\text{N}_8$	0.985	$^9\text{Q}_{11}$	1.088	$^5\text{Q}_{13}$
-0.250	$^{11}\text{L}_3$	0.786	$^7\text{L}_6$ $^{11}\text{M}_6$ $^{11}\text{N}_7$	0.991	$^9\text{O}_{10}$	1.091	$^3\text{N}_{11}$ $^9\text{N}_{19}$
0.000	$^5\text{F}_1$ $^7\text{H}_2$ $^9\text{K}_3$	0.792	$^{11}\text{O}_8$	1.000	$^1\text{P}_1$ $^1\text{D}_2$ $^1\text{F}_3$	1.096	$^5\text{O}_{12}$ $^{11}\text{Q}_{12}$
	$^{11}\text{M}_4$	0.800	$^3\text{H}_4$ $^5\text{O}_9$ $^{11}\text{L}_5$		$^1\text{G}_4$ $^1\text{H}_5$ $^1\text{I}_6$	1.100	$^3\text{M}_{10}$ $^5\text{H}_5$
0.167	$^{11}\text{N}_5$		$^{11}\text{Q}_9$		$^1\text{K}_7$ $^1\text{L}_8$ $^1\text{M}_9$	1.104	$^7\text{Q}_{13}$
0.200	$^9\text{L}_4$	0.804	$^7\text{M}_7$		$^1\text{N}_{10}$ $^1\text{O}_{11}$ $^1\text{Q}_{12}$	1.106	$^5\text{N}_{11}$
0.250	$^7\text{I}_3$	0.818	$^5\text{Q}_{10}$		$^5\text{F}_2$ $^7\text{I}_5$ $^9\text{N}_9$	1.107	$^7\text{K}_7$
0.286	$^{11}\text{O}_6$	0.819	$^7\text{N}_8$		$^{11}\text{I}_3$ $^{11}\text{L}_6$	1.111	$^3\text{L}_9$ $^9\text{M}_9$
0.333	$^5\text{G}_2$ $^9\text{M}_5$	0.833	$^3\text{I}_5$ $^7\text{G}_2$ $^7\text{O}_9$	1.006	$^3\text{Q}_{12}$	1.114	$^{11}\text{O}_{11}$
0.375	$^{11}\text{Q}_7$	0.845	$^7\text{Q}_{10}$	1.008	$^3\text{O}_{11}$	1.115	$^7\text{O}_{12}$
0.400	$^7\text{K}_4$	0.850	$^{11}\text{K}_4$	1.009	$^3\text{N}_{10}$	1.118	$^5\text{M}_{10}$
0.417	$^{11}\text{K}_3$	0.857	$^3\text{K}_6$	1.011	$^3\text{M}_9$	1.125	$^3\text{K}_8$ $^{11}\text{L}_7$
0.429	$^9\text{N}_6$	0.875	$^3\text{L}_7$ $^9\text{M}_7$ $^9\text{N}_8$	1.014	$^3\text{L}_8$ $^9\text{M}_8$	1.126	$^9\text{Q}_{13}$
0.450	$^{11}\text{L}_4$	0.878	$^9\text{O}_9$	1.018	$^3\text{K}_7$	1.129	$^7\text{N}_{11}$
0.500	$^3\text{D}_1$ $^5\text{H}_3$ $^7\text{L}_5$	0.881	$^9\text{L}_6$	1.019	$^5\text{Q}_{12}$	1.133	$^5\text{L}_9$ $^9\text{I}_5$
	$^9\text{O}_7$ $^{11}\text{I}_2$ $^{11}\text{M}_5$	0.882	$^9\text{Q}_{10}$	1.023	$^5\text{O}_{11}$ $^{11}\text{Q}_{11}$	1.136	$^{11}\text{N}_{10}$
0.548	$^{11}\text{N}_6$	0.889	$^3\text{M}_8$	1.024	$^3\text{I}_6$	1.139	$^9\text{L}_8$
0.556	$^9\text{Q}_8$	0.900	$^3\text{N}_9$ $^5\text{H}_4$ $^5\text{I}_5$	1.027	$^5\text{N}_{10}$	1.141	$^9\text{O}_{12}$
0.571	$^7\text{M}_6$		$^9\text{K}_5$	1.033	$^3\text{H}_5$ $^5\text{M}_9$	1.143	$^3\text{I}_7$ $^5\text{Q}_{11}$ $^7\text{I}_6$
0.583	$^9\text{I}_3$	0.902	$^5\text{N}_9$	1.036	$^9\text{L}_7$ $^{11}\text{O}_{10}$	1.145	$^7\text{M}_{10}$
0.589	$^{11}\text{O}_7$	0.905	$^5\text{K}_6$	1.038	$^7\text{Q}_{12}$	1.150	$^5\text{G}_4$
0.600	$^5\text{L}_4$ $^9\text{K}_4$	0.906	$^7\text{N}_9$	1.042	$^5\text{L}_8$	1.153	$^5\text{K}_8$
0.625	$^7\text{N}_7$ $^{11}\text{Q}_8$	0.909	$^3\text{O}_{10}$	1.045	$^7\text{O}_{11}$	1.154	$^5\text{O}_{13}$ $^{11}\text{Q}_{12}$
0.633	$^9\text{L}_5$	0.911	$^5\text{L}_7$	1.050	$^3\text{G}_4$ $^7\text{H}_4$	1.157	$^7\text{Q}_{14}$
0.667	$^3\text{F}_2$ $^5\text{K}_5$ $^7\text{O}_8$	0.917	$^3\text{Q}_{11}$ $^5\text{G}_3$ $^5\text{M}_8$	1.054	$^5\text{K}_7$	1.159	$^9\text{N}_{11}$
	$^9\text{H}_2$ $^9\text{M}_6$	0.927	$^5\text{O}_{10}$ $^{11}\text{Q}_{10}$	1.055	$^7\text{N}_{10}$	1.167	$^3\text{D}_2$ $^3\text{H}_6$ $^5\text{N}_{12}$
0.696	$^9\text{N}_7$	0.932	$^5\text{Q}_{11}$	1.056	$^{11}\text{N}_9$		$^7\text{G}_3$ $^7\text{L}_9$ $^{11}\text{M}_9$
0.700	$^7\text{Q}_9$	0.933	$^{11}\text{O}_9$	1.064	$^9\text{Q}_{12}$	1.170	$^7\text{O}_{13}$
0.714	$^5\text{L}_6$	0.944	$^{11}\text{N}_8$	1.067	$^7\text{M}_9$ $^{11}\text{K}_5$	1.173	$^{11}\text{O}_{12}$
0.722	$^9\text{O}_8$	0.950	$^9\text{I}_4$	1.071	$^5\text{I}_6$ $^9\text{K}_6$	1.176	$^9\text{Q}_{14}$

TABLE 3. LANDÉ g -VALUES—TERMS OF ODD MULTIPLICITY IN ORDER OF INCREASING g —Continued

g	Desig.	g	Desig.	g	Desig.	g	Desig.
1. 179	5I_7 9K_7	1. 268	$^{11}K_7$ $^{11}Q_{16}$	1. 381	9H_6	1. 625	$^{11}F_8$
1. 182	$^5M_{11}$ $^9M_{10}$	1. 273	$^7L_{11}$ $^{11}M_{11}$	1. 382	$^{11}K_{10}$	1. 633	9F_5
1. 186	$^7N_{12}$	1. 276	$^9M_{12}$	1. 385	$^{11}L_{13}$	1. 650	7D_4
1. 191	$^{11}K_6$	1. 280	$^{11}N_{13}$	1. 393	$^{11}I_7$	1. 661	$^{11}F_7$
1. 192	$^9O_{13}$	1. 286	5H_7 7H_6 $^9N_{14}$	1. 400	5F_5 $^9I_{10}$	1. 667	5P_3 9D_6 $^{11}G_5$
1. 194	7K_8	1. 288	$^{11}O_{15}$	1. 402	$^{11}K_{11}$	1. 700	9F_4
1. 197	$^{11}N_{11}$	1. 292	7I_8	1. 405	7G_6	1. 714	$^{11}D_7$ $^{11}F_6$
1. 200	3G_5 $^5L_{10}$ 7H_5	1. 294	$^{11}Q_{17}$	1. 411	9H_7	1. 733	9D_5
	$^7Q_{15}$ $^{11}I_4$ $^{11}Q_{14}$	1. 300	7G_4 $^7K_{10}$ 9K_9	1. 417	$^{11}I_8$ $^{11}K_{12}$	1. 750	7P_4 7D_3 $^{11}G_4$
1. 201	9L_9		$^{11}I_5$	1. 429	7G_7	1. 786	$^{11}D_6$
1. 205	$^7M_{11}$	1. 303	$^9L_{11}$	1. 431	9H_8	1. 800	9P_5 $^{11}F_5$
1. 208	$^{11}L_8$	1. 304	9I_7	1. 433	$^{11}I_9$	1. 833	5P_2 9F_3 $^{11}P_6$
1. 212	$^9N_{12}$	1. 308	$^9M_{13}$ $^{11}M_{12}$	1. 444	9H_9	1. 850	9D_4
1. 214	5H_6 $^7O_{14}$	1. 309	$^{11}L_{10}$	1. 445	$^{11}I_{10}$	1. 900	$^{11}D_5$
1. 217	$^9Q_{15}$	1. 310	$^{11}N_{14}$	1. 455	$^{11}I_{11}$	1. 917	7P_3 $^{11}G_3$
1. 220	$^{11}O_{13}$	1. 312	$^{11}O_{16}$	1. 500	2P_2 2P_1 5D_4	1. 950	9P_4 $^{11}F_4$
1. 222	5K_9	1. 319	$^{11}K_8$		5D_3 5D_2 5D_1	1. 967	$^{11}P_5$
1. 227	$^7L_{10}$ $^{11}M_{10}$	1. 333	3D_3 5G_6 7I_9		7F_6 7F_5 7F_4	2. 000	3S_1 5S_2 7S_3
1. 231	$^7N_{13}$		9H_5 $^9L_{12}$ $^{11}N_{15}$		7F_3 7F_2 7F_1		9S_4 $^{11}S_5$ 7D_2
1. 232	7I_7	1. 335	$^{11}M_{13}$		9G_8 9G_7 9G_6	2. 083	9D_3
1. 233	$^9O_{14}$	1. 336	$^9K_{10}$		9G_5 9G_4 9G_3	2. 100	$^{11}D_4$
1. 235	$^9M_{11}$	1. 339	7H_7		9G_2 9G_1 $^{11}H_{10}$	2. 167	9F_2
1. 238	9I_6 $^{11}Q_{15}$	1. 341	$^{11}L_{11}$		$^{11}H_9$ $^{11}H_8$ $^{11}H_7$	2. 200	$^{11}P_4$
1. 244	$^{11}N_{12}$	1. 347	9I_8		$^{11}H_6$ $^{11}H_5$ $^{11}H_4$	2. 250	9P_3 $^{11}F_3$
1. 250	3F_4 5F_3 5I_8	1. 350	5F_4		$^{11}H_3$ $^{11}H_2$ $^{11}H_1$	2. 333	7P_2 $^{11}G_2$
	$^7M_{12}$ 9H_4 9K_8	1. 356	$^{11}K_9$	1. 556	$^{11}G_9$	2. 500	5P_1 $^{11}D_3$
	$^9Q_{16}$	1. 357	$^{11}I_6$ $^{11}M_{14}$	1. 569	$^{11}G_8$	2. 667	9D_2
1. 253	$^9N_{13}$	1. 364	$^9K_{11}$	1. 571	9F_7	3. 000	7D_1 $^{11}F_2$
1. 256	7K_9	1. 365	$^{11}L_{12}$	1. 589	$^{11}G_7$	3. 500	9F_1
1. 257	$^{11}O_{14}$	1. 367	7G_5	1. 595	9F_6	4. 000	$^{11}G_1$
1. 264	$^9L_{10}$	1. 375	7H_8	1. 600	7D_5		
1. 267	5G_5 $^9O_{15}$ $^{11}L_9$	1. 378	9I_9	1. 619	$^{11}G_6$		

TABLE 4. LANDÉ g -VALUES FOR TERMS OF EVEN MULTIPLICITY IN ORDER OF INCREASING g

g	Desig.	g	Desig.	g	Desig.	g	Desig.
-2.000	$^{10}H_{3/2}$	0.713	$^8L_{5/2}$	0.937	$^8K_{5/2}$	1.059	$^2L_{6/2}$ $^8L_{7/2}$
-1.333	$^8G_{3/2}$	0.727	$^4I_{3/2}$	0.941	$^2L_{7/2}$	1.060	$^4N_{10/2}$
-0.800	$^{10}I_{1/2}$	0.737	$^6O_{3/2}$	0.947	$^2M_{8/2}$	1.063	$^{10}K_{5/2}$
-0.667	$^6F_{3/2}$	0.738	$^8M_{6/2}$	0.952	$^2N_{9/2}$ $^{10}I_{3/2}$	1.064	$^6Q_{12/2}$
-0.400	$^8H_{13/2}$	0.761	$^8N_{7/2}$	0.957	$^2O_{10/2}$ $^{10}Q_{10/2}$	1.067	$^2K_{7/2}$ $^6F_{13/2}$
-0.286	$^{10}K_{2/2}$	0.762	$^6Q_{9/2}$	0.960	$^2Q_{11/2}$	1.068	$^4M_{9/2}$
0.000	$^4D_{3/2}$ $^6G_{13/2}$ $^8I_{23/2}$	0.769	$^4K_{5/2}$	0.962	$^{10}O_{9/2}$	1.071	$^6H_{43/2}$ $^6O_{13/2}$
	$^{10}L_{3/2}$	0.780	$^8O_{3/2}$	0.964	$^4K_{6/2}$	1.073	$^{10}N_{9/2}$
0.182	$^{10}M_{43/2}$	0.797	$^8Q_{9/2}$	0.965	$^4I_{53/2}$ $^4L_{73/2}$	1.077	$^2I_{63/2}$ $^4L_{83/2}$
0.222	$^8K_{33/2}$	0.800	$^2D_{13/2}$ $^4L_{63/2}$ $^{10}H_{13/2}$	0.966	$^4M_{83/2}$	1.081	$^6N_{103/2}$
0.286	$^6H_{23/2}$	0.824	$^4M_{73/2}$	0.967	$^4N_{93/2}$	1.084	$^8Q_{123/2}$
0.308	$^{10}N_{53/2}$	0.825	$^6H_{33/2}$	0.969	$^4O_{103/2}$	1.087	$^8K_{63/2}$
0.364	$^8L_{43/2}$	0.828	$^6I_{43/2}$	0.970	$^4H_{43/2}$ $^4Q_{113/2}$ $^8I_{43/2}$	1.090	$^4K_{73/2}$
0.400	$^4F_{13/2}$ $^{10}O_{63/2}$	0.831	$^{10}M_{63/2}$ $^{10}N_{73/2}$	0.972	$^{10}N_{83/2}$	1.091	$^2H_{53/2}$
0.444	$^6I_{33/2}$	0.836	$^{10}O_{83/2}$	0.984	$^4G_{33/2}$	1.093	$^6M_{93/2}$
0.462	$^8M_{53/2}$	0.839	$^6K_{53/2}$ $^{10}L_{63/2}$	0.988	$^6Q_{113/2}$ $^{10}M_{73/2}$	1.096	$^8O_{113/2}$ $^{10}M_{83/2}$
0.471	$^{10}Q_{73/2}$	0.842	$^4N_{83/2}$ $^{10}Q_{93/2}$	0.990	$^6O_{103/2}$	1.108	$^4I_{63/2}$ $^6L_{83/2}$
0.508	$^{10}K_{33/2}$	0.851	$^6L_{63/2}$	0.992	$^6N_{93/2}$	1.110	$^8N_{103/2}$
0.514	$^{10}I_{23/2}$	0.857	$^2F_{23/2}$ $^4O_{93/2}$ $^6G_{23/2}$	0.997	$^6M_{83/2}$	1.111	$^2G_{43/2}$ $^4Q_{133/2}$ $^{10}Q_{123/2}$
0.533	$^8N_{63/2}$	0.863	$^6M_{73/2}$	1.004	$^6L_{73/2}$	1.120	$^4O_{123/2}$
0.545	$^6K_{43/2}$ $^{10}L_{43/2}$	0.869	$^{10}K_{43/2}$	1.012	$^8Q_{113/2}$	1.124	$^6Q_{133/2}$
0.571	$^4G_{23/2}$	0.870	$^4Q_{103/2}$	1.015	$^6K_{63/2}$ $^{10}L_{63/2}$	1.127	$^{10}O_{113/2}$
0.587	$^{10}M_{53/2}$	0.873	$^6N_{83/2}$	1.019	$^8O_{103/2}$	1.128	$^8M_{93/2}$
0.588	$^8O_{73/2}$	0.882	$^6O_{93/2}$	1.028	$^8N_{93/2}$	1.129	$^6K_{73/2}$ $^{10}L_{73/2}$
0.615	$^6L_{63/2}$	0.889	$^2G_{33/2}$	1.029	$^4F_{23/2}$	1.130	$^4N_{113/2}$
0.626	$^{10}N_{63/2}$	0.890	$^6Q_{103/2}$	1.035	$^6I_{53/2}$	1.133	$^4H_{53/2}$ $^8I_{53/2}$
0.632	$^8Q_{83/2}$	0.909	$^2H_{43/2}$	1.040	$^2Q_{123/2}$ $^8M_{83/2}$	1.135	$^6O_{123/2}$
0.659	$^{10}O_{73/2}$	0.916	$^8N_{83/2}$	1.043	$^2O_{113/2}$ $^{10}Q_{113/2}$	1.142	$^8Q_{133/2}$
0.667	$^2P_{3/2}$ $^4H_{33/2}$ $^6M_{63/2}$	0.917	$^8O_{93/2}$	1.048	$^2N_{103/2}$ $^8H_{33/2}$	1.143	$^2F_{33/2}$ $^4M_{103/2}$ $^6G_{33/2}$
	$^8I_{33/2}$	0.918	$^8M_{73/2}$	1.049	$^4Q_{123/2}$	1.147	$^{10}N_{103/2}$
0.686	$^8H_{23/2}$	0.919	$^8Q_{103/2}$	1.053	$^2M_{93/2}$	1.148	$^6N_{113/2}$
0.687	$^8K_{43/2}$ $^{10}Q_{83/2}$	0.923	$^2I_{53/2}$ $^8L_{63/2}$	1.054	$^4O_{113/2}$	1.152	$^8L_{83/2}$ $^{10}L_{43/2}$
0.706	$^6N_{73/2}$	0.933	$^2K_{63/2}$ $^8G_{13/2}$	1.056	$^{10}O_{103/2}$	1.156	$^8O_{123/2}$

TABLE 4. LANDÉ g -VALUES FOR TERMS OF EVEN MULTIPLICITY IN ORDER OF INCREASING g —Continued

g	Desig.	g	Desig.	g	Desig.	g	Desig.
1. 158	$4L_{9\frac{1}{2}}$	1. 255	$10K_{7\frac{1}{2}}$	1. 394	$10H_{4\frac{1}{2}}$	1. 714	$6P_{3\frac{1}{2}}$ $8F_{2\frac{1}{2}}$
1. 159	$6I_{6\frac{1}{2}}$	1. 257	$8G_{2\frac{1}{2}}$	1. 397	$6F_{3\frac{1}{2}}$	1. 733	$4P_{1\frac{1}{2}}$
1. 164	$6M_{10\frac{1}{2}}$	1. 259	$8N_{13\frac{1}{2}}$ $10I_{5\frac{1}{2}}$	1. 412	$8H_{8\frac{1}{2}}$	1. 758	$10F_{4\frac{1}{2}}$
1. 165	$10Q_{13\frac{1}{2}}$	1. 261	$10O_{14\frac{1}{2}}$	1. 414	$8G_{4\frac{1}{2}}$ $10I_{9\frac{1}{2}}$	1. 762	$10D_{5\frac{1}{2}}$
1. 172	$4G_{4\frac{1}{2}}$ $6Q_{14\frac{1}{2}}$ $8N_{11\frac{1}{2}}$	1. 263	$6K_{9\frac{1}{2}}$ $10L_{9\frac{1}{2}}$	1. 427	$10H_{5\frac{1}{2}}$	1. 772	$10G_{2\frac{1}{2}}$
1. 173	$10M_{9\frac{1}{2}}$	1. 267	$8L_{10\frac{1}{2}}$	1. 429	$4D_{3\frac{1}{2}}$ $10I_{10\frac{1}{2}}$	1. 778	$8P_{4\frac{1}{2}}$
1. 176	$4K_{8\frac{1}{2}}$	1. 273	$4G_{5\frac{1}{2}}$ $6G_{4\frac{1}{2}}$ $10M_{11\frac{1}{2}}$	1. 434	$6F_{4\frac{1}{2}}$	1. 809	$8D_{3\frac{1}{2}}$
1. 179	$10K_{6\frac{1}{2}}$		$10Q_{16\frac{1}{2}}$	1. 441	$8G_{5\frac{1}{2}}$	1. 818	$10P_{5\frac{1}{2}}$
1. 182	$10O_{12\frac{1}{2}}$	1. 280	$8M_{12\frac{1}{2}}$	1. 446	$10H_{6\frac{1}{2}}$	1. 867	$6D_{1\frac{1}{2}}$
1. 183	$6L_{9\frac{1}{2}}$	1. 282	$6H_{6\frac{1}{2}}$ $10N_{13\frac{1}{2}}$	1. 455	$6F_{5\frac{1}{2}}$	1. 879	$10D_{4\frac{1}{2}}$
1. 184	$8K_{7\frac{1}{2}}$	1. 290	$10O_{15\frac{1}{2}}$	1. 456	$8G_{6\frac{1}{2}}$	1. 886	$6P_{2\frac{1}{2}}$
1. 185	$6O_{13\frac{1}{2}}$	1. 294	$6I_{8\frac{1}{2}}$ $8I_{7\frac{1}{2}}$	1. 459	$10H_{7\frac{1}{2}}$	1. 905	$10F_{3\frac{1}{2}}$
1. 188	$8Q_{14\frac{1}{2}}$	1. 298	$8K_{9\frac{1}{2}}$	1. 467	$8G_{7\frac{1}{2}}$ $10H_{8\frac{1}{2}}$	1. 937	$8P_{3\frac{1}{2}}$
1. 193	$8M_{10\frac{1}{2}}$	1. 301	$8H_{5\frac{1}{2}}$	1. 474	$10H_{9\frac{1}{2}}$	1. 960	$10P_{4\frac{1}{2}}$
1. 200	$2D_{2\frac{1}{2}}$ $4D_{1\frac{1}{2}}$ $4I_{7\frac{1}{2}}$	1. 304	$8L_{11\frac{1}{2}}$ $10L_{10\frac{1}{2}}$	1. 529	$10G_{8\frac{1}{2}}$	2. 000	$2S_{\frac{1}{2}}$ $4S_{1\frac{1}{2}}$ $6S_{2\frac{1}{2}}$
	$6N_{12\frac{1}{2}}$ $10H_{2\frac{1}{2}}$	1. 307	$10K_{8\frac{1}{2}}$ $10M_{12\frac{1}{2}}$	1. 537	$10G_{7\frac{1}{2}}$		$8S_{3\frac{1}{2}}$ $8F_{1\frac{1}{2}}$ $10S_{4\frac{1}{2}}$
1. 203	$6H_{5\frac{1}{2}}$ $8O_{13\frac{1}{2}}$ $10N_{11\frac{1}{2}}$	1. 310	$10N_{14\frac{1}{2}}$	1. 538	$8F_{6\frac{1}{2}}$	2. 057	$8D_{2\frac{1}{2}}$
1. 207	$6K_{8\frac{1}{2}}$ $10L_{8\frac{1}{2}}$	1. 314	$6F_{2\frac{1}{2}}$	1. 549	$10G_{6\frac{1}{2}}$	2. 095	$10D_{3\frac{1}{2}}$
1. 208	$10Q_{14\frac{1}{2}}$	1. 323	$10I_{6\frac{1}{2}}$	1. 552	$8F_{5\frac{1}{2}}$	2. 133	$10G_{1\frac{1}{2}}$
1. 212	$8H_{4\frac{1}{2}}$	1. 333	$2P_{1\frac{1}{2}}$ $4F_{4\frac{1}{2}}$ $6H_{7\frac{1}{2}}$	1. 556	$6D_{4\frac{1}{2}}$	2. 222	$10P_{3\frac{1}{2}}$
1. 217	$6M_{11\frac{1}{2}}$		$8K_{10\frac{1}{2}}$ $10H_{3\frac{1}{2}}$ $10M_{13\frac{1}{2}}$	1. 566	$10G_{5\frac{1}{2}}$	2. 229	$10F_{2\frac{1}{2}}$
1. 218	$8L_{9\frac{1}{2}}$	1. 336	$10L_{11\frac{1}{2}}$	1. 576	$8F_{4\frac{1}{2}}$	2. 286	$8P_{2\frac{1}{2}}$
1. 221	$8N_{12\frac{1}{2}}$	1. 337	$8I_{8\frac{1}{2}}$	1. 587	$6D_{3\frac{1}{2}}$	2. 400	$6P_{1\frac{1}{2}}$
1. 226	$8Q_{15\frac{1}{2}}$ $10O_{13\frac{1}{2}}$	1. 343	$6G_{5\frac{1}{2}}$ $10K_{9\frac{1}{2}}$	1. 596	$10G_{4\frac{1}{2}}$	2. 572	$10D_{2\frac{1}{2}}$
1. 230	$10M_{10\frac{1}{2}}$	1. 354	$8H_{6\frac{1}{2}}$	1. 600	$4P_{2\frac{1}{2}}$ $10F_{7\frac{1}{2}}$	2. 667	$4P_{\frac{1}{2}}$
1. 231	$4H_{6\frac{1}{2}}$ $8I_{6\frac{1}{2}}$	1. 360	$10L_{12\frac{1}{2}}$	1. 619	$8F_{3\frac{1}{2}}$	2. 800	$8D_{1\frac{1}{2}}$
1. 238	$4F_{3\frac{1}{2}}$ $6L_{10\frac{1}{2}}$	1. 365	$8G_{3\frac{1}{2}}$ $10I_{7\frac{1}{2}}$	1. 631	$10F_{6\frac{1}{2}}$	3. 200	$10F_{1\frac{1}{2}}$
1. 239	$6I_{7\frac{1}{2}}$	1. 368	$8I_{9\frac{1}{2}}$	1. 636	$8D_{5\frac{1}{2}}$	3. 333	$6D_{\frac{1}{2}}$
1. 241	$8O_{14\frac{1}{2}}$	1. 371	$4D_{2\frac{1}{2}}$ $10K_{10\frac{1}{2}}$	1. 651	$10G_{3\frac{1}{2}}$	4. 000	$8F_{\frac{1}{2}}$
1. 242	$8M_{11\frac{1}{2}}$	1. 385	$6G_{6\frac{1}{2}}$	1. 657	$6D_{2\frac{1}{2}}$	4. 667	$10G_{\frac{1}{2}}$
1. 243	$10Q_{15\frac{1}{2}}$	1. 388	$8H_{7\frac{1}{2}}$	1. 678	$10F_{5\frac{1}{2}}$		
1. 247	$10N_{12\frac{1}{2}}$	1. 391	$10K_{11\frac{1}{2}}$	1. 692	$10D_{6\frac{1}{2}}$		
1. 251	$8K_{8\frac{1}{2}}$	1. 393	$10I_{8\frac{1}{2}}$	1. 697	$8D_{4\frac{1}{2}}$		

TABLE 5. PREDICTED TERMS OF THE Be I ISOELECTRONIC SEQUENCE

Config. $1s^2+$	Predicted Terms														
$2s^2$	1S														
$2s(2S)2p$	{ $^3P^{\circ}$ $^1P^{\circ}$														
$2p^2$	{ 1S 3P 1D														
	$ns\ (n\geq 3)$		$np\ (n\geq 3)$		$nd\ (n\geq 3)$		$nf\ (n\geq 4)$		$ng\ (n\geq 5)$					
$2s(2S)nx$	{ 3S 1S		$^3P^{\circ}$ $^1P^{\circ}$		3D 1D		$^3F^{\circ}$ $^1F^{\circ}$		3G 1G					
$2p(2P^{\circ})nx$	{ $^3P^{\circ}$ $^1P^{\circ}$		3S 1S	3P 1P	3D 1D	$^3P^{\circ}$ $^1P^{\circ}$	$^3D^{\circ}$ $^1D^{\circ}$	$^3F^{\circ}$ $^1F^{\circ}$	3D 1D	3F 1F	3G 1G	$^3F^{\circ}$ $^1F^{\circ}$	$^3G^{\circ}$ $^1G^{\circ}$	$^3H^{\circ}$ $^1H^{\circ}$

TABLE 6. PREDICTED TERMS OF THE B I ISOELECTRONIC SEQUENCE

Config. 1s ² +	Predicted Terms											
2s ² (¹ S)2p	² P ^o											
2s 2p ²	{	² S	⁴ P ² P		² D							
2p ³			{	⁴ S ^o	² P ^o		² D ^o					
		ns (n ≥ 3)			np (n ≥ 3)			nd (n ≥ 3)				
2s ² (¹ S)nx	² S		² P ^o			² D						
2s 2p(³ P ^o)nx	{	⁴ P ^o ² P ^o		⁴ S ² S	⁴ P ² P	⁴ D ² D	⁴ P ^o ² P ^o		⁴ D ^o ² D ^o	⁴ F ^o ² F ^o		
2s 2p(¹ P ^o)nx'		² P ^o		² S	² P	² D	² P ^o ² D ^o		² F ^o			
2p ² (³ P)nx''	{	⁴ P ² P		⁴ S ^o ² S ^o	⁴ P ^o ² P ^o	⁴ D ^o ² D ^o	⁴ P ² P		⁴ D ² D	⁴ F ² F		
2p ² (¹ D)nx'''				² D		² P ^o	² D ^o	² F ^o	² S	² P	² D	² F
2p ² (¹ S)nx ^{IV}	² S		² P ^o			² D						
		nf (n ≥ 4)			ng (n ≥ 5)						
2s ² (¹ S)nx			² F ^o			² G					
2s 2p(³ P ^o)nx	{	⁴ D ² D	⁴ F ² F	⁴ G ² G	⁴ F ^o ² F ^o	⁴ G ^o ² G ^o	⁴ H ^o ² H ^o				
2s 2p(¹ P ^o)nx'		² D	² F	² G	² F ^o	² G ^o	² H ^o				
2p ² (³ P)nx''	{	⁴ D ^o ² D ^o	⁴ F ^o ² F ^o	⁴ G ^o ² G ^o	⁴ F ² F	⁴ G ² G	⁴ H ² H				
2p ² (¹ D)nx'''		² P ^o	² D ^o	² F ^o	² G ^o	² H ^o	² D	² F	² G	² H	² I
2p ² (¹ S)nx ^{IV}			² F ^o			² G					

TABLE 7.—PREDICTED TERMS OF THE CI ISOELECTRONIC SEQUENCE

Config. 1s ² +	Predicted Terms										
2s ² 2p ²	{ _{1S} ³ P _{1D}										
2s 2p ³	{ ⁵ S ^o _{3S^o} ³ P ^o _{1P^o} ³ D ^o _{1D^o}										
2p ⁴	{ _{1S} ³ P _{1D}										
	ns (n ≥ 3)		np (n ≥ 3)			nd (n ≥ 3)			nf (n ≥ 4)	
2s ² 2p(² P ^o)nx	{ _{1P^o} ³ P ^o		_{1S} _{1P} _{1D}	³ S ³ P ³ D	_{1P^o} _{1D^o} _{1F^o}	³ P ^o ³ D ^o ³ F ^o	_{1D} _{1F} _{1G}	³ D ³ F ³ G	_{1D^o} _{1F^o} _{1G^o}	
2s 2p ² (⁴ P)nx	{ _{3P} ⁵ P		_{3S^o} _{3P^o} _{3D^o}	⁵ S ^o ⁵ P ^o ⁵ D ^o	_{3P} _{3D} _{3F}	⁵ P ⁵ D ⁵ F	_{3D^o} _{3F^o} _{3G^o}	⁵ D ^o ⁵ F ^o ⁵ G ^o	_{3D^o} _{3F^o} _{3G^o}	
2s 2p ² (² D)nx'	{ _{1D} ³ D		_{1P^o} _{1D^o} _{1F^o}	³ P ^o ³ D ^o ³ F ^o	_{1S} _{1P} _{1D} _{1F} _{1G}	³ S ³ P ³ D ³ F ³ G	_{1P^o} _{1D^o} _{1F^o} _{1G^o} _{1H^o}	³ D ^o ³ F ^o ³ G ^o ³ H ^o	_{1D^o} _{1F^o} _{1G^o} _{1H^o}	
2s 2p ² (² S)nx''	{ _{1S} ³ S		_{1P^o} ³ P ^o	³ S ^o ³ P ^o	_{1D} ³ D	³ S ³ P ³ D ³ F ³ G	_{1F^o} ³ F ^o	³ S ^o ³ P ^o ³ D ^o ³ F ^o ³ G ^o ³ H ^o	_{1F^o} ³ F ^o	
2s 2p ² (² P)nx'''	{ _{1P} ³ P		_{1S^o} _{1P^o} _{1D^o}	³ S ^o ³ P ^o ³ D ^o	_{1P} _{1D} _{1F}	³ P ³ D ³ F	_{1D^o} _{1F^o} _{1G^o}	³ D ^o ³ F ^o ³ G ^o	_{1D^o} _{1F^o} _{1G^o}	
2p ³ (⁴ S ^o)nx ^{IV}	{ ⁵ S ^o _{3S^o}		_{3P} ⁵ P	⁵ S ^o ⁵ P ^o	_{3D^o} ⁵ D ^o	⁵ P ⁵ D ⁵ F	_{3F} ⁵ F	⁵ S ^o ⁵ P ^o ⁵ D ^o ⁵ F ^o ⁵ G ^o ⁵ H ^o	_{3F} ⁵ F	

TABLE 8. PREDICTED TERMS OF THE NI ISOELECTRONIC SEQUENCE

Config. 1s ² +	Predicted Terms												
2s ² 2p ³	{ ⁴ S ^o 2P ^o 2D ^o												
2s 2p ⁴	{ ₂ S ⁴ P ² P 2D												
2p ⁵	2P ^o												
	ns (n ≥ 3)			np (n ≥ 3)			nd (n ≥ 3)			nf (n ≥ 4)		
2s ² 2p ² (³ P)nx	{ ⁴ P ² P			⁴ S ^o ⁴ P ^o ⁴ D ^o ² S ^o ² P ^o ² D ^o			⁴ P ⁴ D ⁴ F ² P ² D ² F			⁴ D ^o ⁴ F ^o ⁴ G ^o ² D ^o ² F ^o ² G ^o		
2s ² 2p ² (¹ D)nx'	2D			2P ^o 2D ^o 2F ^o			2S 2P 2D 2F 2G	2P ^o 2D ^o 2F ^o 2G ^o 2H ^o				
2s ² 2p ² (¹ S)nx''	2S			2P ^o			2D			2F ^o		
2s 2p ³ (⁵ S ^o)nx'''	{ ⁶ S ^o ⁴ S ^o			⁶ P ⁴ P			⁶ D ^o ⁴ D ^o			⁶ F ⁴ F		
2s 2p ³ (³ D ^o)nx ^{iv}	{ ⁴ D ^o ² D ^o			⁴ P ⁴ D ⁴ F ² P ² D ² F			⁴ S ^o ⁴ P ^o ⁴ D ^o ⁴ F ^o ⁴ G ^o ² S ^o ² P ^o ² D ^o ² F ^o ² G ^o	⁴ P ⁴ D ⁴ F ⁴ G ⁴ H ² P ² D ² F ² G ² H				
2s 2p ³ (³ P ^o)nx ^v	{ ⁴ P ^o ² P ^o			⁴ S ⁴ P ⁴ D ² S ² P ² D			⁴ P ^o ⁴ D ^o ⁴ F ^o ² P ^o ² D ^o ² F ^o			⁴ D ⁴ F ⁴ G ² D ² F ² G		

TABLE 9. PREDICTED TERMS OF THE O I ISOELECTRONIC SEQUENCE

Config. 1s ² +	Predicted Terms																
2s ² 2p ⁴	{ ₁ S		³ P			₁ D											
2s 2p ⁵	{		³ P ^o ₁ P ^o														
	ns (n ≥ 3)					np (n ≥ 3)					nd (n ≥ 3)						
2s ² 2p ³ (⁴ S ^o) nx	{ ⁵ S ^o ³ S ^o							⁵ P ³ P			⁵ D ^o ³ D ^o						
2s ² 2p ³ (² D ^o) nx'	{		³ D ^o ₁ D ^o			³ P ₁ P			³ D ₁ D	³ F ₁ F	³ S ^o ₁ S ^o	³ P ^o ₁ P ^o	³ D ^o ₁ D ^o	³ F ^o ₁ F ^o	³ G ^o ₁ G ^o		
2s ² 2p ³ (² P ^o) nx''	{		³ P ^o ₁ P ^o			³ S ₁ S	³ P ₁ P	³ D ₁ D	³ P ^o ₁ P ^o					³ D ^o ₁ D ^o	³ F ^o ₁ F ^o		
2s 2p ⁴ (⁴ P) nx'''	{		⁵ P ³ P			⁵ S ^o ³ S ^o	⁵ P ^o ³ P ^o	⁵ D ^o ³ D ^o	⁵ P ³ P					⁵ D ³ D	⁵ F ³ F		
2s 2p ⁴ (² D) nx ^{IV}	{		³ D ₁ D			³ P ^o ₁ P ^o			³ D ^o ₁ D ^o	³ F ^o ₁ F ^o	³ S ₁ S	³ P ₁ P	³ D ₁ D	³ F ₁ F	³ G ₁ G		
2s 2p ⁴ (² S) nx ^V	{ ₁ S ³ S							³ P ^o ₁ P ^o			³ D ₁ D						
2s 2p ⁴ (² P) nx ^{VI}	{		³ P ₁ P			³ S ^o ₁ S ^o	³ P ^o ₁ P ^o	³ D ^o ₁ D ^o	³ P ₁ P					³ D ₁ D	³ F ₁ F		
	nf (n ≥ 4)															
2s ² 2p ³ (⁴ S ^o) nx	{		⁵ F ³ F													
2s ² 2p ³ (² D ^o) nx'	{ ₁ P ³ P		³ D ₁ D	³ F ₁ F	³ G ₁ G	³ H ₁ H										
2s ² 2p ³ (² P ^o) nx''	{		³ D ₁ D			³ F ₁ F	³ G ₁ G									
2s 2p ⁴ (⁴ P) nx'''	{		⁵ D ^o ³ D ^o			⁵ F ^o ³ F ^o	⁵ G ^o ³ G ^o									
2s 2p ⁴ (² D) nx ^{IV}	{ ₁ P ^o ³ P ^o		³ D ^o ₁ D ^o	³ F ^o ₁ F ^o	³ G ^o ₁ G ^o	³ H ^o ₁ H ^o										
2s 2p ⁴ (² S) nx ^V	{		³ F ^o ₁ F ^o													
2s 2p ⁴ (² P) nx ^{VI}	{		³ D ^o ₁ D ^o			³ F ^o ₁ F ^o	³ G ^o ₁ G ^o									

TABLE 10. PREDICTED TERMS OF THE FI ISOELECTRONIC SEQUENCE

Config. 1s ² +	Predicted Terms									
2s ² 2p ⁵	2P°									
2s 2p ⁶	2S									
	ns (n≥3)			np (n≥3)			nd (n≥3)			
2s ² 2p ⁴ (3P)nx	{	4P		4S°	4P°	4D°	4P	4D	4F	
		2P		2S°	2P°	2D°	2P	2D	2F	
2s ² 2p ⁴ (1D)nx'			2D		2P°	2D°	2F°	2S	2P	2D
2s ² 2p ⁴ (1S)nx''	2S				2P°				2D	
2s 2p ⁵ (3P°)nx'''	{	4P°		4S	4P	4D	4P°	4D°	4F°	
		2P°		2S	2P	2D	2P°	2D°	2F°	
	nf (n≥4)								
2s ² 2p ⁴ (3P)nx	{	4D°	4F°	4G°					
		2D°	2F°	2G°					
2s ² 2p ⁴ (1D)nx'	2P°	2D°	2F°	2G°	2H°				
2s ² 2p ⁴ (1S)nx''			2F°						
2s 2p ⁵ (3P°)nx'''	{	4D	4F	4G					
		2D	2F	2G					

TABLE 11. PREDICTED LEVELS OF THE Ne I ISOELECTRONIC SEQUENCE

Config. $1s^2+$	Predicted Terms									
$2s^2 2p^6$	$1S$									
	$ns (n \geq 3)$	$np (n \geq 3)$			$nd (n \geq 3)$			$nf (n \geq 4)$	
$2s^2 2p^5(^2P^\circ)nx$	{	$3P^\circ$	$3S$	$3P$	$3D$	$3P^\circ$	$3D^\circ$	$3F^\circ$	$3D$	$3F$
		$1P^\circ$	$1S$	$1P$	$1D$	$1P^\circ$	$1D^\circ$	$1F^\circ$	$1D$	$1F$
$2s 2p^6(^2S)nx$	{	$3S$		$3P^\circ$		$3D$		$3F^\circ$		
	$1S$			$1P^\circ$		$1D$		$1F^\circ$		
<i>jl</i> -Coupling Notation										
Config. $1s^2 2s^2+$	Predicted Pairs									
	$ns (n \geq 3)$	$np (n \geq 3)$			$nd (n \geq 3)$			$nf (n \geq 4)$	
$2p^5(^2P_{1/2})nx$	$[1\frac{1}{2}]^\circ$	$[\frac{1}{2}]$			$[\frac{1}{2}]^\circ$			$[1\frac{1}{2}]$	
		$[2\frac{1}{2}]$			$[3\frac{1}{2}]^\circ$			$[4\frac{1}{2}]$	
		$[1\frac{1}{2}]$			$[1\frac{1}{2}]^\circ$			$[2\frac{1}{2}]$	
		$[1\frac{1}{2}]$			$[2\frac{1}{2}]^\circ$			$[3\frac{1}{2}]$	
$2p^5(^2P_{3/2})nx'$	$[\frac{1}{2}]^\circ$	$[1\frac{1}{2}]$			$[2\frac{1}{2}]^\circ$			$[3\frac{1}{2}]$	
		$[\frac{1}{2}]$			$[1\frac{1}{2}]^\circ$			$[2\frac{1}{2}]$	

TABLE 12. PREDICTED TERMS OF THE Mg I ISOELECTRONIC SEQUENCE

Config. 1s ² 2s ² 2p ⁶ +	Predicted Terms									
3s ²	1S									
3s(2S)3p	{ 3P ^o 1P ^o									
3p ²	{ 1S 3P 1D									
	ns (n ≥ 4)			np (n ≥ 4)			nd (n ≥ 3)			
3s(2S)nx	{ 3S 1S			3P ^o 1P ^o			3D 1D			
3p(2P ^o)nx	{ 3P ^o 1P ^o			3S 1S	3P 1P	3D 1D	3P ^o 1P ^o	3D ^o 1D ^o	3F ^o 1F ^o	
	n _f (n ≥ 4)			ng (n ≥ 5)			nh (n ≥ 6)		
3s(2S)nx	{ 3F ^o 1F ^o			3G 1G			3H ^o 1H ^o		
3p(2P ^o)nx	3D 1D	3F 1F	3G 1G	3F ^o 1F ^o	3G ^o 1G ^o	3H ^o 1H ^o	3G 1G	3H 1H	3I 1I

TABLE 13. PREDICTED TERMS OF THE Al I ISOELECTRONIC SEQUENCE

Config. 1s ² 2s ² 2p ⁶ +	Predicted Terms													
3s ² (1S)3p	2P ^o													
3s 3p ²	{ 2S 4P 2P 2D													
3p ³	{ 4S ^o 2P ^o 2D ^o													
	ns (n ≥ 4)			np (n ≥ 4)			nd (n ≥ 3)			nf (n ≥ 4)			ng (n ≥ 5)	
3s ² (1S)nx	2S			2P ^o			2D			2F ^o			2G	
3s 3p(3P ^o)nx	{ 4P ^o 2P ^o			4S 2S	4P 2P	4D 2D	4P ^o 2P ^o	4D ^o 2D ^o	4F ^o 2F ^o	4D 2D	4F 2F ^o	4G 2G	4F ^o 2F ^o	4G ^o 2G ^o
3s 3p(1P ^o)nx'	2P ^o			2S	2P	2D	2P ^o	2D ^o	2F ^o	2D	2F	2G	2F ^o	2G ^o

TABLE 14. PREDICTED TERMS OF THE Si I ISOELECTRONIC SEQUENCE

Config. $1s^2\ 2s^2\ 2p^6+$	Predicted Terms											
$3s^2\ 3p^2$	$\left\{ \begin{array}{ccc} & {}^3P & \\ {}^1S & & {}^1D \end{array} \right.$											
$3s\ 3p^3$	$\left\{ \begin{array}{ccc} {}^5S^\circ & & \\ {}^3S^\circ & {}^3P^\circ & {}^3D^\circ \\ & {}^1P^\circ & {}^1D^\circ \end{array} \right.$											
$3p^4$	$\left\{ \begin{array}{ccc} & {}^3P & \\ {}^1S & & {}^1D \end{array} \right.$											
	$ns\ (n \geq 4)$		$np\ (n \geq 4)$			$nd\ (n \geq 3)$			$nf\ (n \geq 4)$		
$3s^2\ 3p(^2P^\circ)nx$	$\left\{ \begin{array}{c} {}^3P^\circ \\ {}^1P^\circ \end{array} \right.$		$\begin{array}{c} {}^3S \\ {}^1S \end{array}$	$\begin{array}{c} {}^3P \\ {}^1P \end{array}$	$\begin{array}{c} {}^3D \\ {}^1D \end{array}$	$\begin{array}{c} {}^3P^\circ \\ {}^1P^\circ \end{array}$	$\begin{array}{c} {}^3D^\circ \\ {}^1D^\circ \end{array}$	$\begin{array}{c} {}^3F^\circ \\ {}^1F^\circ \end{array}$	$\begin{array}{c} {}^3D \\ {}^1D \end{array}$	$\begin{array}{c} {}^3F \\ {}^1F \end{array}$	$\begin{array}{c} {}^3G \\ {}^1G \end{array}$
$3s\ 3p^2(^4P)nx$	$\left\{ \begin{array}{c} {}^5P \\ {}^3P \end{array} \right.$		$\begin{array}{c} {}^5S^\circ \\ {}^3S^\circ \end{array}$	$\begin{array}{c} {}^5P^\circ \\ {}^3P^\circ \end{array}$	$\begin{array}{c} {}^5D^\circ \\ {}^3D^\circ \end{array}$	$\begin{array}{c} {}^5P \\ {}^3P \end{array}$	$\begin{array}{c} {}^5D \\ {}^3D \end{array}$	$\begin{array}{c} {}^5F \\ {}^3F \end{array}$	$\begin{array}{c} {}^5D^\circ \\ {}^3D^\circ \end{array}$	$\begin{array}{c} {}^5F^\circ \\ {}^3F^\circ \end{array}$	$\begin{array}{c} {}^5G^\circ \\ {}^3G^\circ \end{array}$

TABLE 15. PREDICTED TERMS OF THE P I ISOELECTRONIC SEQUENCE

Config. $1s^2\ 2s^2\ 2p^6+$	Predicted Terms										
$3s^2\ 3p^3$	$\left\{ \begin{array}{ccc} {}^4S^\circ & {}^2P^\circ & {}^2D^\circ \\ {}^2S & {}^2P & {}^2D \end{array} \right.$										
$3s\ 3p^4$	$\left\{ \begin{array}{ccc} {}^4P & & {}^2D \\ {}^2S & & \end{array} \right.$										
$3p^5$	${}^2P^\circ$										
	$ns\ (n\geq 4)$	$np\ (n\geq 4)$			$nd\ (n\geq 3)$			$nf\ (n\geq 4)$		
$3s^2\ 3p^2({}^3P)nx$	$\left\{ \begin{array}{c} {}^4P \\ {}^2P \end{array} \right.$	$\begin{array}{c} {}^4S^\circ \\ {}^2S^\circ \end{array}$	$\begin{array}{c} {}^4P^\circ \\ {}^2P^\circ \end{array}$	$\begin{array}{c} {}^4D^\circ \\ {}^2D^\circ \end{array}$	$\begin{array}{c} {}^4P \\ {}^2P \end{array}$	$\begin{array}{c} {}^4D \\ {}^2D \end{array}$	$\begin{array}{c} {}^4F \\ {}^2F \end{array}$	$\begin{array}{c} {}^4D^\circ \\ {}^2D^\circ \end{array}$	$\begin{array}{c} {}^4F^\circ \\ {}^2F^\circ \end{array}$	$\begin{array}{c} {}^4G^\circ \\ {}^2G^\circ \end{array}$
$3s^2\ 3p^2({}^1D)nx'$	2D		${}^2P^\circ\ {}^2D^\circ$	${}^2F^\circ$	${}^2S\ {}^2P$	${}^2D\ {}^2F$	2G	${}^2P^\circ\ {}^2D^\circ$	${}^2F^\circ\ {}^2G^\circ$	${}^2H^\circ$
$3s^2\ 3p^2({}^1S)nx''$	2S		${}^2P^\circ$			2D			${}^2F^\circ$	

TABLE 16. PREDICTED TERMS OF THE Si ISOELECTRONIC SEQUENCE

Config. $1s^2 2s^2 2p^6 +$	Predicted Terms									
$3s^2 3p^4$	$\left\{ \begin{array}{l} {}^3P \\ {}^1D \end{array} \right.$									
$3s 3p^5$	$\left\{ \begin{array}{l} {}^3P^\circ \\ {}^1P^\circ \end{array} \right.$									
	$ns (n \geq 4)$	$np (n \geq 4)$			$nd (n \geq 3)$			$nf (n \geq 4)$		
$3s^2 3p^3 ({}^4S^\circ) nx$	$\left\{ \begin{array}{l} {}^5S^\circ \\ {}^3S^\circ \end{array} \right.$	$\begin{array}{l} {}^5P \\ {}^3P \end{array}$			$\begin{array}{l} {}^5D^\circ \\ {}^3D^\circ \end{array}$			$\begin{array}{l} {}^5F \\ {}^3F \end{array}$		
$3s^2 3p^3 ({}^2D^\circ) nx'$	$\left\{ \begin{array}{l} {}^3D^\circ \\ {}^1D^\circ \end{array} \right.$	$\begin{array}{l} {}^3P \\ {}^1P \end{array}$	$\begin{array}{l} {}^3D \\ {}^1D \end{array}$	$\begin{array}{l} {}^3F \\ {}^1F \end{array}$	$\begin{array}{l} {}^3S^\circ \\ {}^1S^\circ \end{array}$	$\begin{array}{l} {}^3P^\circ \\ {}^1P^\circ \end{array}$	$\begin{array}{l} {}^3D^\circ \\ {}^1D^\circ \end{array}$	$\begin{array}{l} {}^3F^\circ \\ {}^1F^\circ \end{array}$	$\begin{array}{l} {}^3G^\circ \\ {}^1G^\circ \end{array}$	$\begin{array}{l} {}^3P \\ {}^1P \end{array} \quad \begin{array}{l} {}^3D \\ {}^1D \end{array} \quad \begin{array}{l} {}^3F \\ {}^1F \end{array} \quad \begin{array}{l} {}^3G \\ {}^1G \end{array} \quad \begin{array}{l} {}^3H \\ {}^1H \end{array}$
$3s^2 3p^3 ({}^2P^\circ) nx''$	$\left\{ \begin{array}{l} {}^3P^\circ \\ {}^1P^\circ \end{array} \right.$	$\begin{array}{l} {}^3S \\ {}^1S \end{array}$	$\begin{array}{l} {}^3P \\ {}^1P \end{array}$	$\begin{array}{l} {}^3D \\ {}^1D \end{array}$	$\begin{array}{l} {}^3P^\circ \\ {}^1P^\circ \end{array}$	$\begin{array}{l} {}^3D^\circ \\ {}^1D^\circ \end{array}$	$\begin{array}{l} {}^3F^\circ \\ {}^1F^\circ \end{array}$	$\begin{array}{l} {}^3D \\ {}^1D \end{array}$	$\begin{array}{l} {}^3F \\ {}^1F \end{array}$	$\begin{array}{l} {}^3G \\ {}^1G \end{array}$
$3s 3p^4 ({}^4P) nx'''$	$\left\{ \begin{array}{l} {}^5P \\ {}^3P \end{array} \right.$	$\begin{array}{l} {}^5S^\circ \\ {}^3S^\circ \end{array}$	$\begin{array}{l} {}^5P^\circ \\ {}^3P^\circ \end{array}$	$\begin{array}{l} {}^5D^\circ \\ {}^3D^\circ \end{array}$	$\begin{array}{l} {}^5P \\ {}^3P \end{array}$	$\begin{array}{l} {}^5D \\ {}^3D \end{array}$	$\begin{array}{l} {}^5F \\ {}^3F \end{array}$	$\begin{array}{l} {}^5D^\circ \\ {}^3D^\circ \end{array}$	$\begin{array}{l} {}^5F^\circ \\ {}^3F^\circ \end{array}$	$\begin{array}{l} {}^5G^\circ \\ {}^3G^\circ \end{array}$
$3s 3p^4 ({}^2D) nx^{IV}$	$\left\{ \begin{array}{l} {}^3D \\ {}^1D \end{array} \right.$	$\begin{array}{l} {}^3P^\circ \\ {}^1P^\circ \end{array}$	$\begin{array}{l} {}^3D^\circ \\ {}^1D^\circ \end{array}$	$\begin{array}{l} {}^3F^\circ \\ {}^1F^\circ \end{array}$	$\begin{array}{l} {}^3S \\ {}^1S \end{array}$	$\begin{array}{l} {}^3P \\ {}^1P \end{array}$	$\begin{array}{l} {}^3D \\ {}^1D \end{array}$	$\begin{array}{l} {}^3F \\ {}^1F \end{array}$	$\begin{array}{l} {}^3G \\ {}^1G \end{array}$	$\begin{array}{l} {}^3P^\circ \\ {}^1P^\circ \end{array} \quad \begin{array}{l} {}^3D^\circ \\ {}^1D^\circ \end{array} \quad \begin{array}{l} {}^3F^\circ \\ {}^1F^\circ \end{array} \quad \begin{array}{l} {}^3G^\circ \\ {}^1G^\circ \end{array} \quad \begin{array}{l} {}^3H^\circ \\ {}^1H^\circ \end{array}$
$3s 3p^4 ({}^2S) nx^V$	$\left\{ \begin{array}{l} {}^3S \\ {}^1S \end{array} \right.$	$\begin{array}{l} {}^3P^\circ \\ {}^1P^\circ \end{array}$	$\begin{array}{l} {}^3D \\ {}^1D \end{array}$			$\begin{array}{l} {}^3F^\circ \\ {}^1F^\circ \end{array}$				
$3s 3p^4 ({}^2P) nx^{VI}$	$\left\{ \begin{array}{l} {}^3P \\ {}^1P \end{array} \right.$	$\begin{array}{l} {}^3S^\circ \\ {}^1S^\circ \end{array}$	$\begin{array}{l} {}^3P^\circ \\ {}^1P^\circ \end{array}$	$\begin{array}{l} {}^3D^\circ \\ {}^1D^\circ \end{array}$	$\begin{array}{l} {}^3P \\ {}^1P \end{array}$	$\begin{array}{l} {}^3D \\ {}^1D \end{array}$	$\begin{array}{l} {}^3F \\ {}^1F \end{array}$	$\begin{array}{l} {}^3D^\circ \\ {}^1D^\circ \end{array}$	$\begin{array}{l} {}^3F^\circ \\ {}^1F^\circ \end{array}$	$\begin{array}{l} {}^3G^\circ \\ {}^1G^\circ \end{array}$
$3p^5 ({}^2P^\circ) nx^{VII}$	$\left\{ \begin{array}{l} {}^3P^\circ \\ {}^1P^\circ \end{array} \right.$	$\begin{array}{l} {}^3S \\ {}^1S \end{array}$	$\begin{array}{l} {}^3P \\ {}^1P \end{array}$	$\begin{array}{l} {}^3D \\ {}^1D \end{array}$	$\begin{array}{l} {}^3P^\circ \\ {}^1P^\circ \end{array}$	$\begin{array}{l} {}^3D^\circ \\ {}^1D^\circ \end{array}$	$\begin{array}{l} {}^3F^\circ \\ {}^1F^\circ \end{array}$	$\begin{array}{l} {}^3D \\ {}^1D \end{array}$	$\begin{array}{l} {}^3F \\ {}^1F \end{array}$	$\begin{array}{l} {}^3G \\ {}^1G \end{array}$

TABLE 17. PREDICTED TERMS OF THE Cl I ISOELECTRONIC SEQUENCE

Config. $1s^2 2s^2 2p^6 +$	Predicted Terms									
$3s^2 3p^5$	${}^2P^\circ$									
$3s 3p^6$	2S									
	$ns (n \geq 4)$	$np (n \geq 4)$			$nd (n \geq 3)$			$nf (n \geq 4)$		
$3s^2 3p^4 ({}^3P) nx$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \end{array} \right.$	$\begin{array}{l} {}^4S^\circ \\ {}^2S^\circ \end{array}$	$\begin{array}{l} {}^4P^\circ \\ {}^2P^\circ \end{array}$	$\begin{array}{l} {}^4D^\circ \\ {}^2D^\circ \end{array}$	$\begin{array}{l} {}^4P \\ {}^2P \end{array}$	$\begin{array}{l} {}^4D \\ {}^2D \end{array}$	$\begin{array}{l} {}^4F \\ {}^2F \end{array}$	$\begin{array}{l} {}^4D^\circ \\ {}^2D^\circ \end{array}$	$\begin{array}{l} {}^4F^\circ \\ {}^2F^\circ \end{array}$	$\begin{array}{l} {}^4G^\circ \\ {}^2G^\circ \end{array}$
$3s^2 3p^4 ({}^1D) nx'$	2D			$\begin{array}{l} {}^2P^\circ \\ {}^2D^\circ \end{array}$	2S	2P	2D	2F	2G	$\begin{array}{l} {}^2P^\circ \\ {}^2D^\circ \end{array} \quad \begin{array}{l} {}^2F^\circ \\ {}^2G^\circ \end{array} \quad {}^2H^\circ$
$3s^2 3p^4 ({}^1S) nx''$	2S	${}^2P^\circ$			2D			${}^2F^\circ$		
$3s 3p^5 ({}^3P^\circ) nx'''$	$\left\{ \begin{array}{l} {}^4P^\circ \\ {}^2P^\circ \end{array} \right.$	$\begin{array}{l} {}^4S \\ {}^2S \end{array}$	$\begin{array}{l} {}^4P \\ {}^2P \end{array}$	$\begin{array}{l} {}^4D \\ {}^2D \end{array}$	$\begin{array}{l} {}^4P^\circ \\ {}^2P^\circ \end{array}$	$\begin{array}{l} {}^4D^\circ \\ {}^2D^\circ \end{array}$	$\begin{array}{l} {}^4F^\circ \\ {}^2F^\circ \end{array}$	$\begin{array}{l} {}^4D \\ {}^2D \end{array}$	$\begin{array}{l} {}^4F \\ {}^2F \end{array}$	$\begin{array}{l} {}^4G \\ {}^2G \end{array}$

TABLE 18. PREDICTED LEVELS OF THE Al ISOELECTRONIC SEQUENCE

Config. $1s^2 2s^2 2p^6 +$	Predicted Terms				
$3s^2 3p^6$	1S				
	$ns (n \geq 4)$	$np (n \geq 4)$	$nd (n \geq 3)$	$nf (n \geq 4)$
$3s^2 3p^5(^2P^\circ)nx$	$\left\{ \begin{array}{c} ^3P^\circ \\ ^1P^\circ \end{array} \right.$	$\begin{array}{ccc} ^3S & ^3P & ^3D \\ ^1S & ^1P & ^1D \end{array}$	$\begin{array}{ccc} ^3P^\circ & ^3D^\circ & ^3F^\circ \\ ^1P^\circ & ^1D^\circ & ^1F^\circ \end{array}$	$\begin{array}{ccc} ^3D & ^3F & ^3G \\ ^1D & ^1F & ^1G \end{array}$
$3s 3p^6(^2S)nx$	$\left\{ \begin{array}{c} ^3S \\ ^1S \end{array} \right.$	$\begin{array}{c} ^3P^\circ \\ ^1P^\circ \end{array}$	$\begin{array}{c} ^3D \\ ^1D \end{array}$	$\begin{array}{c} ^3F^\circ \\ ^1F^\circ \end{array}$
<i>jl</i> -Coupling Notation					
Config. $1s^2 2s^2 2p^6 3s^2 +$	Predicted Pairs				
	$ns (n \geq 4)$	$np (n \geq 4)$	$nd (n \geq 3)$	$nf (n \geq 4)$
$3p^5(^2P_{1/2})nx$	$[1\frac{1}{2}]^\circ$	$\left[\begin{array}{c} \frac{1}{2} \\ 2\frac{1}{2} \\ 1\frac{1}{2} \end{array} \right]$	$\left[\begin{array}{c} \frac{1}{2} \\ 3\frac{1}{2} \\ 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right]^\circ$	$\left[\begin{array}{c} 1\frac{1}{2} \\ 4\frac{1}{2} \\ 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right]$
$3p^5(^2P_{3/2})nx'$	$[\frac{1}{2}]^\circ$	$\left[\begin{array}{c} 1\frac{1}{2} \\ \frac{1}{2} \end{array} \right]$	$\left[\begin{array}{c} 2\frac{1}{2} \\ 1\frac{1}{2} \end{array} \right]^\circ$	$\left[\begin{array}{c} 3\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right]$

TABLE 19. PREDICTED TERMS OF THE Ca I ISOELECTRONIC SEQUENCE

Config. $1s^2\ 2s^2\ 2p^6\ 3s^2\ 3p^6+$	Predicted Terms												
$4s^2$	1S												
$3d^2$	$\left\{ \begin{array}{c} ^1S \\ ^3P \\ ^1D \\ ^3F \\ ^1G \end{array} \right.$												
$4p^2$	$\left\{ \begin{array}{c} ^1S \\ ^3P \\ ^1D \end{array} \right.$												
	$ns\ (n \geq 4)$				$np\ (n \geq 4)$				$nd\ (n \geq 3)$				
$4s(^2S)nx$	$\left\{ \begin{array}{c} ^3S \\ ^1S \end{array} \right.$					$\begin{array}{c} ^3P^\circ \\ ^1P^\circ \end{array}$		$\begin{array}{c} ^3D \\ ^1D \end{array}$					
$3d(^2D)nx'$	$\left\{ \begin{array}{c} ^3D \\ ^1D \end{array} \right.$					$\begin{array}{cc} ^3P^\circ & ^3D^\circ \\ ^1P^\circ & ^1D^\circ \end{array}$		$\begin{array}{c} ^3F^\circ \\ ^1F^\circ \end{array}$	$\begin{array}{cc} ^3S & ^3P \\ ^1S & ^1P \end{array}$	$\begin{array}{c} ^3D \\ ^1D \end{array}$	$\begin{array}{cc} ^3F & ^3G \\ ^1F & ^1G \end{array}$		
$4p(^2P^\circ)nx''$	$\left\{ \begin{array}{c} ^3P^\circ \\ ^1P^\circ \end{array} \right.$					$\begin{array}{c} ^3S \\ ^1S \end{array}$	$\begin{array}{c} ^3P \\ ^1P \end{array}$	$\begin{array}{c} ^3D \\ ^1D \end{array}$	$\begin{array}{ccc} ^3P^\circ & ^3D^\circ & ^3F^\circ \\ ^1P^\circ & ^1D^\circ & ^1F^\circ \end{array}$				
	$nf\ (n \geq 4)$				$ng\ (n \geq 5)$							
$4s(^2S)nx$	$\left\{ \begin{array}{c} ^3F^\circ \\ ^1F^\circ \end{array} \right.$					$\begin{array}{c} ^3G \\ ^1G \end{array}$						
$3d(^2D)nx'$	$\left\{ \begin{array}{ccccc} ^3P^\circ & ^3D^\circ & ^3F^\circ & ^3G^\circ & ^3H^\circ \\ ^1P^\circ & ^1D^\circ & ^1F^\circ & ^1G^\circ & ^1H^\circ \end{array} \right.$	$\begin{array}{c} ^3D \\ ^1D \end{array}$	$\begin{array}{c} ^3F \\ ^1F \end{array}$	$\begin{array}{c} ^3G \\ ^1G \end{array}$	$\begin{array}{c} ^3H \\ ^1H \end{array}$	$\begin{array}{c} ^3I \\ ^1I \end{array}$						
$4p(^2P^\circ)nx''$	$\left\{ \begin{array}{ccc} ^3D & ^3F & ^3G \\ ^1D & ^1F & ^1G \end{array} \right.$					$\begin{array}{ccc} ^3F^\circ & ^3G^\circ & ^3H^\circ \\ ^1F^\circ & ^1G^\circ & ^1H^\circ \end{array}$						

TABLE 20. PREDICTED TERMS OF THE Sc I ISOELECTRONIC SEQUENCE

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Predicted Terms				
$3d 4s^2$	2D				
$3d^3$	$\left\{ \begin{array}{ccccc} ^4P & & ^4F & & \\ ^2P & ^2D & ^2F & ^2G & ^2H \\ & ^2D & & & \end{array} \right.$				
	$ns (n \geq 4)$		$np (n \geq 4)$		
$3d 4s(^3D)nx$	$\left\{ \begin{array}{c} ^4D \\ ^2D \end{array} \right.$		$\begin{array}{ccc} ^4P^\circ & ^4D^\circ & ^4F^\circ \\ ^2P^\circ & ^2D^\circ & ^2F^\circ \end{array}$		
$3d 4s(^1D)nx$	2D		$\begin{array}{ccc} ^2P^\circ & ^2D^\circ & ^2F^\circ \end{array}$		
$3d^2(^3F)nx$	$\left\{ \begin{array}{c} ^4F \\ ^2F \end{array} \right.$		$\begin{array}{ccc} ^4D^\circ & ^4F^\circ & ^4G^\circ \\ ^2D^\circ & ^2F^\circ & ^2G^\circ \end{array}$		
$3d^2(^1D)nx$	2D		$\begin{array}{ccc} ^2P^\circ & ^2D^\circ & ^2F^\circ \end{array}$		
$3d^2(^1S)nx$	2S		$^2P^\circ$		
$3d^2(^3P)nx$	$\left\{ \begin{array}{c} ^4P \\ ^2P \end{array} \right.$		$\begin{array}{ccc} ^4S^\circ & ^4P^\circ & ^4D^\circ \\ ^2S^\circ & ^2P^\circ & ^2D^\circ \end{array}$		
$3d^2(^1G)nx$	2G		$\begin{array}{ccc} & & ^2F^\circ \\ & & ^2G^\circ \\ & & ^2H^\circ \end{array}$		
$4p^2(^3P)nx$	$\left\{ \begin{array}{c} ^4P \\ ^2P \end{array} \right.$		$\begin{array}{ccc} ^4S^\circ & ^4P^\circ & ^4D^\circ \\ ^2S^\circ & ^2P^\circ & ^2D^\circ \end{array}$		
	$nd (n \geq 3)$		$nf (n \geq 4)$	
$3d 4s(^3D)nx$	$\left\{ \begin{array}{ccccc} ^4S & ^4P & ^4D & ^4F & ^4G \\ ^2S & ^2P & ^2D & ^2F & ^2G \end{array} \right.$		$\begin{array}{ccccc} ^4P^\circ & ^4D^\circ & ^4F^\circ & ^4G^\circ & ^4H^\circ \\ ^2P^\circ & ^2D^\circ & ^2F^\circ & ^2G^\circ & ^2H^\circ \end{array}$	
$3d 4s(^1D)nx$	$\begin{array}{ccccc} ^2S & ^2P & ^2D & ^2F & ^2G \end{array}$		$\begin{array}{ccccc} ^2P^\circ & ^2D^\circ & ^2F^\circ & ^2G^\circ & ^2H^\circ \end{array}$	
$3d^2(^3F)nx$	$\left\{ \begin{array}{ccccc} ^4P & ^4D & ^4F & ^4G & ^4H \\ ^2P & ^2D & ^2F & ^2G & ^2H \end{array} \right.$		$\begin{array}{ccccc} ^4D^\circ & ^4F^\circ & ^4G^\circ & ^4H^\circ & ^4I^\circ \\ ^2D^\circ & ^2F^\circ & ^2G^\circ & ^2H^\circ & ^2I^\circ \end{array}$	
$3d^2(^1D)nx$	$\begin{array}{ccccc} ^2S & ^2P & ^2D & ^2F & ^2G \end{array}$		$\begin{array}{ccccc} ^2P^\circ & ^2D^\circ & ^2F^\circ & ^2G^\circ & ^2H^\circ \end{array}$	
$3d^2(^1S)nx$	2D		$^2F^\circ$	
$3d^2(^3P)nx$	$\left\{ \begin{array}{ccc} ^4P & ^4D & ^4F \\ ^2P & ^2D & ^2F \end{array} \right.$		$\begin{array}{ccc} ^4D^\circ & ^4F^\circ & ^4G^\circ \\ ^2D^\circ & ^2F^\circ & ^2G^\circ \end{array}$	
$3d^2(^1G)nx$	$^2D \quad ^2F \quad ^2G \quad ^2H \quad ^2I$		$\begin{array}{ccccc} & ^2F^\circ & ^2G^\circ & ^2H^\circ & ^2I^\circ & ^2K^\circ \end{array}$	
$4p^2(^3P)nx$	$\left\{ \begin{array}{ccc} ^4P & ^4D & ^4F \\ ^2P & ^2D & ^2F \end{array} \right.$		$\begin{array}{ccc} ^4D^\circ & ^4F^\circ & ^4G^\circ \\ ^2D^\circ & ^2F^\circ & ^2G^\circ \end{array}$	

TABLE 21. PREDICTED TERMS OF THE Tl I ISOELECTRONIC SEQUENCE

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Predicted Terms			
	ns ($n \geq 4$)	np ($n \geq 4$)	nd ($n \geq 4$)
$3d^2 4s^2$	$\left\{ \begin{array}{l} 3P \\ 1S \end{array} \right\}$	$\begin{array}{l} 3F \\ 1D \\ 1G \end{array}$		
$3d^4$	$\left\{ \begin{array}{l} 3P \\ 3P \end{array} \right\}$	$\begin{array}{l} 3D \\ 3D \end{array}$	$\begin{array}{l} 3F \\ 3F \end{array}$	$\begin{array}{l} 3H \\ 3H \end{array}$
	$\left\{ \begin{array}{l} 1S \\ 1S \end{array} \right\}$	$\begin{array}{l} 1D \\ 1D \end{array}$	$\begin{array}{l} 1F \\ 1F \end{array}$	$\begin{array}{l} 1G \\ 1G \end{array}$
$3d^2 4s(4F)nx$	$\left\{ \begin{array}{l} 3P \\ 1S \end{array} \right\}$	$\begin{array}{l} 3F \\ 1D \\ 1G \end{array}$	$\begin{array}{l} 3D \\ 1D \\ 1G \end{array}$	$\begin{array}{l} 3H \\ 1H \\ 1G \end{array}$
$3d^2(4F)nx$	$\left\{ \begin{array}{l} 3P \\ 3P \end{array} \right\}$	$\begin{array}{l} 3D \\ 3D \end{array}$	$\begin{array}{l} 3F \\ 3F \end{array}$	$\begin{array}{l} 3H \\ 3H \end{array}$
$3d^2 4s(2F)nx$	$\left\{ \begin{array}{l} 3P \\ 1S \end{array} \right\}$	$\begin{array}{l} 3D \\ 1D \end{array}$	$\begin{array}{l} 3F \\ 1F \end{array}$	$\begin{array}{l} 3H \\ 1H \end{array}$
$3d^2 4s(2D)nx$	$\left\{ \begin{array}{l} 3P \\ 1S \end{array} \right\}$	$\begin{array}{l} 3D \\ 1D \end{array}$	$\begin{array}{l} 3F \\ 1F \end{array}$	$\begin{array}{l} 3H \\ 1H \end{array}$
$3d^2(2G)nx$	$\left\{ \begin{array}{l} 3P \\ 1S \end{array} \right\}$	$\begin{array}{l} 3D \\ 1D \end{array}$	$\begin{array}{l} 3F \\ 1F \end{array}$	$\begin{array}{l} 3H \\ 1H \end{array}$
$3d^2(4P)nx$	$\left\{ \begin{array}{l} 3P \\ 3P \end{array} \right\}$	$\begin{array}{l} 3D \\ 3D \end{array}$	$\begin{array}{l} 3F \\ 3F \end{array}$	$\begin{array}{l} 3H \\ 3H \end{array}$
$3d^2(2P)nx$	$\left\{ \begin{array}{l} 3P \\ 1P \end{array} \right\}$	$\begin{array}{l} 3D \\ 1D \end{array}$	$\begin{array}{l} 3F \\ 1F \end{array}$	$\begin{array}{l} 3H \\ 1H \end{array}$
$3d^2 4s(4P)nx$	$\left\{ \begin{array}{l} 3P \\ 3P \end{array} \right\}$	$\begin{array}{l} 3D \\ 3D \end{array}$	$\begin{array}{l} 3F \\ 3F \end{array}$	$\begin{array}{l} 3H \\ 3H \end{array}$
$3d^2(2D)nx$	$\left\{ \begin{array}{l} 3P \\ 1D \end{array} \right\}$	$\begin{array}{l} 3D \\ 1D \end{array}$	$\begin{array}{l} 3F \\ 1F \end{array}$	$\begin{array}{l} 3H \\ 1H \end{array}$
$3d^2(2H)nx$	$\left\{ \begin{array}{l} 3P \\ 1H \end{array} \right\}$	$\begin{array}{l} 3D \\ 1H \end{array}$	$\begin{array}{l} 3F \\ 1H \end{array}$	$\begin{array}{l} 3H \\ 1H \end{array}$
$3d^2 4s(2G)nx$	$\left\{ \begin{array}{l} 3P \\ 1G \end{array} \right\}$	$\begin{array}{l} 3D \\ 1G \end{array}$	$\begin{array}{l} 3F \\ 1G \end{array}$	$\begin{array}{l} 3H \\ 1G \end{array}$
$3d^2 4s(2P)nx$	$\left\{ \begin{array}{l} 3P \\ 1P \end{array} \right\}$	$\begin{array}{l} 3D \\ 1P \end{array}$	$\begin{array}{l} 3F \\ 1P \end{array}$	$\begin{array}{l} 3H \\ 1P \end{array}$
$3d^2(2F)nx$	$\left\{ \begin{array}{l} 3P \\ 1F \end{array} \right\}$	$\begin{array}{l} 3D \\ 1F \end{array}$	$\begin{array}{l} 3F \\ 1F \end{array}$	$\begin{array}{l} 3H \\ 1F \end{array}$
$3d^2 4s(2S)nx$	$\left\{ \begin{array}{l} 3S \\ 1S \end{array} \right\}$	$\begin{array}{l} 3D \\ 1S \end{array}$	$\begin{array}{l} 3F \\ 1S \end{array}$	$\begin{array}{l} 3H \\ 1S \end{array}$
$3d 4s^2(2D)nx$	$\left\{ \begin{array}{l} 3D \\ 1D \end{array} \right\}$	$\begin{array}{l} 3D \\ 1D \end{array}$	$\begin{array}{l} 3F \\ 1D \end{array}$	$\begin{array}{l} 3H \\ 1D \end{array}$

TABLE 22. PREDICTED TERMS OF THE VI ISOELECTRONIC SEQUENCE

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Predicted Terms			
	$ns (n \geq 4)$	$np (n \geq 4)$	$nd (n \geq 4)$
$3d^3 4s^3$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \\ {}^2D \\ {}^2D \end{array} \right\}$		$\begin{array}{l} {}^4F \\ {}^2F \\ {}^2G \\ {}^2H \end{array}$	
$3d^3$	$\left\{ \begin{array}{l} {}^6S \\ {}^2S \end{array} \right\}$		$\begin{array}{l} {}^4P \\ {}^4D \\ {}^2D \\ {}^2F \\ {}^2G \\ {}^2H \end{array}$	$\begin{array}{l} {}^2I \\ {}^2H \end{array}$
$3d^3 4p^2 \dagger$	$\left\{ \begin{array}{l} {}^6P \\ {}^4P \\ {}^2P \end{array} \right\}$		$\begin{array}{l} {}^6D^* \\ {}^4D^* \\ {}^2D^* \end{array}$	$\begin{array}{l} {}^6F \\ {}^4F \\ {}^2F \\ {}^2G \\ {}^2G \end{array}$
$3d^4 ({}^6D) nx$	$\left\{ \begin{array}{l} {}^6D \\ {}^4D \end{array} \right\}$	$\begin{array}{l} {}^6P^o \\ {}^4P^o \end{array}$	$\begin{array}{l} {}^6D^o \\ {}^4D^o \end{array}$	$\begin{array}{l} {}^6F^o \\ {}^4F^o \end{array}$
$3d^3 4s ({}^6F) nx$		$\begin{array}{l} {}^6D^o \\ {}^4D^o \end{array}$	$\begin{array}{l} {}^6F^o \\ {}^4F^o \end{array}$	$\begin{array}{l} {}^6G^o \\ {}^4G^o \end{array}$
$3d^3 4s ({}^3F) nx$		$\begin{array}{l} {}^4D^o \\ {}^2D^o \end{array}$	$\begin{array}{l} {}^4F^o \\ {}^2F^o \end{array}$	$\begin{array}{l} {}^4G^o \\ {}^2G^o \end{array}$
$3d^4 ({}^3P) nx$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \end{array} \right\}$	$\begin{array}{l} {}^4S^o \\ {}^2S^o \end{array}$	$\begin{array}{l} {}^4P^o \\ {}^2P^o \end{array}$	$\begin{array}{l} {}^4D^o \\ {}^2D^o \end{array}$
$3d^4 ({}^3H) nx$			$\begin{array}{l} {}^4H \\ {}^2H \end{array}$	$\begin{array}{l} {}^4G^o \\ {}^2G^o \end{array}$
$3d^4 ({}^3F) nx$			$\begin{array}{l} {}^4F \\ {}^2F \end{array}$	$\begin{array}{l} {}^4G^o \\ {}^2G^o \end{array}$
				$\begin{array}{l} {}^4H^o \\ {}^2H^o \end{array}$
				$\begin{array}{l} {}^4I^o \\ {}^2I^o \end{array}$
				$\begin{array}{l} {}^4F \\ {}^2F \end{array}$
				$\begin{array}{l} {}^4D \\ {}^2D \end{array}$
				$\begin{array}{l} {}^4P \\ {}^2P \end{array}$
				$\begin{array}{l} {}^4F \\ {}^2F \end{array}$
				$\begin{array}{l} {}^4G \\ {}^2G \end{array}$
				$\begin{array}{l} {}^4H \\ {}^2H \end{array}$
				$\begin{array}{l} {}^4I \\ {}^2I \end{array}$
				$\begin{array}{l} {}^4K \\ {}^2K \end{array}$

$3d^3\ 4s(^6P)nx$	$\left\{ \begin{array}{l} {}^6P \\ {}^4P \end{array} \right\}$		${}^6S^\circ$ ${}^4S^\circ$	${}^6P^\circ$ ${}^4P^\circ$	${}^6D^\circ$ ${}^4D^\circ$	${}^6F^\circ$ ${}^4F^\circ$	6P 4P	6D 4D	6F 4F					
$3d^4(^3G)nx$	$\left\{ \begin{array}{l} {}^4G \\ {}^2G \end{array} \right\}$			${}^4F^\circ$ ${}^2F^\circ$	${}^4G^\circ$ ${}^2G^\circ$	${}^4H^\circ$ ${}^2H^\circ$		4D 2D	4F 2F	4H 2H	4I 2I			
$3d^3\ 4s(^3G)nx$	$\left\{ \begin{array}{l} {}^4G \\ {}^2G \end{array} \right\}$			${}^4F^\circ$ ${}^2F^\circ$	${}^4G^\circ$ ${}^2G^\circ$	${}^4H^\circ$ ${}^2H^\circ$		4D 2D	4F 2F	4H 2H	4I 2I			
$3d^4(^1G)nx$	$\left\{ \begin{array}{l} {}^4G \\ {}^2G \end{array} \right\}$			${}^2F^\circ$	${}^2G^\circ$	${}^2H^\circ$		2D	2F	2G	2H	2I		
$3d^4(^3D)nx$	$\left\{ \begin{array}{l} {}^4D \\ {}^2D \end{array} \right\}$			${}^4P^\circ$ ${}^2P^\circ$	${}^4D^\circ$ ${}^2D^\circ$			4S 2S	4P 2P	4F 2F	4G 2G			
$3d^3\ 4s(^1G)nx$		2G		${}^2F^\circ$	${}^2G^\circ$	${}^2H^\circ$			2D	2F	2G	2H	2I	
$3d^3\ 4s(^3P)nx$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \end{array} \right\}$		2I	${}^4S^\circ$ ${}^2S^\circ$	${}^4P^\circ$ ${}^2P^\circ$	${}^4D^\circ$ ${}^2D^\circ$			4P 2P	4D 2D	4F 2F			
$3d^4(^1I)nx$			2I				${}^2H^\circ$ ${}^2I^\circ$				2G	2H	2I	2K	2L
$3d^4(^1S)nx$	2S			${}^2P^\circ$						2D					
$3d^3\ 4s(^3H)nx$	$\left\{ \begin{array}{l} {}^4H \\ {}^2H \end{array} \right\}$				${}^4G^\circ$ ${}^2G^\circ$	${}^4H^\circ$ ${}^2H^\circ$	${}^4I^\circ$ ${}^2I^\circ$			4F 2F	4G 2G	4H 2H	4I 2I	4K 2K	
$3d^3\ 4s(^3D)nx$	$\left\{ \begin{array}{l} {}^4D \\ {}^2D \end{array} \right\}$			${}^4P^\circ$ ${}^2P^\circ$	${}^4D^\circ$ ${}^2D^\circ$	${}^4F^\circ$ ${}^2F^\circ$			4S 2S	4P 2P	4D 2D	4F 2F	4G 2G		
$3d^4(^1D)nx$	2D			${}^2P^\circ$	${}^2D^\circ$	${}^2F^\circ$			2S	2P	2D	2F	2G		
$3d^3\ 4s(^1P)nx$	2P			${}^2S^\circ$	${}^2P^\circ$	${}^2D^\circ$				2P	2D	2F			
$3d^3\ 4s(^1H)nx$		2H			${}^2G^\circ$	${}^2H^\circ$	${}^2I^\circ$				2F	2G	2H	2I	2K
$3d^3\ 4s(^1D)nx$	2D			${}^2P^\circ$	${}^2D^\circ$	${}^2F^\circ$			2S	2P	2D	2F	2G		
$3d^4(^1F)nx$	2F				${}^2D^\circ$	${}^2F^\circ$				2P	2D	2F	2G	2H	

†Incomplete—only limits of higher multiplicity considered.

*Two terms of this type predicted.

TABLE 23. THE CHEMICAL ELEMENTS—IONIZATION POTENTIALS*

Z	Element	Symbol	I. P.	Ground State	Z	Element	Symbol	I. P.	Ground State
1	Hydrogen	H	13. 595	$1s \quad 2S_{1/2}$	36	Krypton	Kr	13. 996	$(4s^2 \quad 4p^6) \quad 1S_0$
2	Helium	He	24. 580	$(1s^2) \quad 1S_0$	37	Rubidium	Rb	4. 176	$5s \quad 2S_{1/2}$
3	Lithium	Li	5. 390	$2s \quad 2S_{1/2}$	38	Strontium	Sr	5. 692	$5s^2 \quad 1S_0$
4	Beryllium	Be	9. 320	$2s^2 \quad 1S_0$	39	Yttrium	Y	6. 6	$4d \quad 5s^2 \quad 2D_{3/2}$
5	Boron	B	8. 296	$2s^2 \quad 2p \quad 2P_{1/2}^o$	40	Zirconium	Zr	6. 95	$4d^2 \quad 5s^2 \quad 3F_2$
6	Carbon	C	11. 264	$2s^2 \quad 2p^2 \quad 3P_0$	41	Columbium	Cb	6. 77	$4d^4 \quad 5s \quad 6D_{3/2}$
7	Nitrogen	N	14. 54	$2s^2 \quad 2p^3 \quad 4S_{1/2}$	42	Molybdenum	Mo	7. 18	$4d^5 \quad 5s \quad 7S_3$
8	Oxygen	O	13. 614	$2s^2 \quad 2p^4 \quad 3P_2$	43	Technetium	Tc		$4d^5 \quad 5s^2 \quad 6S_{21/2}$
9	Fluorine	F	17. 42	$2s^2 \quad 2p^5 \quad 2P_{1/2}$	44	Ruthenium	Ru	7. 5	$4d^7 \quad 5s \quad 5F_5$
10	Neon	Ne	21. 559	$(2s^2 \quad 2p^6) \quad 1S_0$	45	Rhodium	Rh	7. 7	$4d^8 \quad 5s \quad 4F_{41/2}$
11	Sodium	Na	5. 138	$3s \quad 2S_{1/2}$	46	Palladium	Pd	8. 33	$4d^{10} \quad 1S_0$
12	Magnesium	Mg	7. 644	$3s^2 \quad 1S_0$	47	Silver	Ag	7. 574	$5s \quad 2S_{1/2}$
13	Aluminum	Al	5. 984	$3s^2 \quad 3p \quad 2P_{1/2}^o$	48	Cadmium	Cd	8. 991	$5s^2 \quad 1S_0$
14	Silicon	Si	8. 149	$3s^2 \quad 3p^2 \quad 3P_0$	49	Indium	In	5. 785	$5s^2 \quad 5p \quad 2P_{1/2}^o$
15	Phosphorus	P	11. 0	$3s^2 \quad 3p^3 \quad 4S_{1/2}$	50	Tin	Sn	7. 332	$5s^2 \quad 5p^2 \quad 3P_0$
16	Sulfur	S	10. 357	$3s^2 \quad 3p^4 \quad 3P_2$	51	Antimony	Sb	8. 64	$5s^2 \quad 5p^3 \quad 4S_{1/2}$
17	Chlorine	Cl	13. 01	$3s^2 \quad 3p^5 \quad 2P_{1/2}$	52	Tellurium	Te	9. 01	$5s^2 \quad 5p^4 \quad 3P_2$
18	Argon	A	15. 755	$(3s^2 \quad 3p^6) \quad 1S_0$	53	Iodine	I	10. 44	$5s^2 \quad 5p^5 \quad 2P_{1/2}$
19	Potassium	K	4. 339	$4s \quad 2S_{1/2}$	54	Xenon	Xe	12. 127	$(5s^2 \quad 5p^6) \quad 1S_0$
20	Calcium	Ca	6. 111	$4s^2 \quad 1S_0$	55	Cesium	Cs	3. 893	$6s \quad 2S_{1/2}$
21	Scandium	Sc	6. 56	$3d \quad 4s^2 \quad 2D_{3/2}$	56	Barium	Ba	5. 210	$6s^2 \quad 1S_0$
22	Titanium	Ti	6. 83	$3d^2 \quad 4s^2 \quad 3F_2$	57	Lanthanum	La	5. 61	$5d \quad 6s^2 \quad 2D_{3/2}$
23	Vanadium	V	6. 74	$3d^3 \quad 4s^2 \quad 4F_{11/2}$	58	Cerium	Ce	(6. 91)	
24	Chromium	Cr	6. 76	$3d^5 \quad 4s \quad 7S_3$	59	Praseodymium	Pr	(5. 76)	
25	Manganese	Mn	7. 432	$3d^5 \quad 4s^2 \quad 6S_{21/2}$	60	Neodymium	Nd	(6. 31)	$4f^4 \quad 6s^2 \quad 5L_4$
26	Iron	Fe	7. 896	$3d^6 \quad 4s^2 \quad 5D_4$	61	Promethium	Pm		
27	Cobalt	Co	7. 86	$3d^7 \quad 4s^2 \quad 4F_{41/2}$	62	Samarium	Sm	5. 6	$4f^6 \quad 6s^2 \quad 7F_0$
28	Nickel	Ni	7. 633	$3d^8 \quad 4s^2 \quad 3F_4$	63	Europium	Eu	5. 67	$4f^7 \quad 6s^2 \quad 8S_{31/2}$
29	Copper	Cu	7. 723	$(3d^{10}) \quad 4s \quad 2S_{1/2}$	64	Gadolinium	Gd	6. 16	$4f^7 \quad 5d \quad 6s^2 \quad 9D_{11/2}$
30	Zinc	Zn	9. 391	$4s^2 \quad 1S_0$	65	Terbium	Tb	(6. 74)	
31	Gallium	Ga	6. 00	$4s^2 \quad 4p \quad 2P_{1/2}^o$	66	Dysprosium	Dy	(6. 82)	
32	Germanium	Ge	8. 13	$4s^2 \quad 4p^2 \quad 3P_0$	67	Holmium	Ho		
33	Arsenic	As	10. \pm	$4s^2 \quad 4p^3 \quad 4S_{1/2}$	68	Erbium	Er		
34	Selenium	Se	9. 750	$4s^2 \quad 4p^4 \quad 3P_2$	69	Thulium	Tm		$4f^{13} \quad 6s^2 \quad 2F_{31/2}$
35	Bromine	Br	11. 84	$4s^2 \quad 4p^5 \quad 2P_{1/2}$	70	Ytterbium	Yb	6. 2	$(4f^{14}) \quad 6s^2 \quad 1S_0$

TABLE 23. THE CHEMICAL ELEMENTS—IONIZATION POTENTIALS—Continued

Z	Element	Symbol	I. P.	Ground State	Z	Element	Symbol	I. P.	Ground State
71	Lutecium	Lu	5. 0	$5d \ 6s^2 \ ^2D_{3/2}$	88	Radium	Ra	5. 277	$7s^2 \ ^1S_0$
72	Hafnium	Hf	5. 5 \pm	$5d^2 \ 6s^2 \ ^3F_2$	89	Actinium	Ac		
73	Tantalum	Ta	6 \pm	$5d^3 \ 6s^2 \ ^4F_{1/2}$	90	Thorium	Th		$6d^2 \ 7s^2 \ ^3F_2$
74	Tungsten	W	7. 98	$5d^4 \ 6s^2 \ ^5D_0$	91	Protactinium	Pa		
75	Rhenium	Re	7. 87	$5d^5 \ 6s^2 \ ^6S_{5/2}$	92	Uranium	U	4 \pm	$5f^3 \ 6d \ 7s^2 \ ^5L_6^\circ$
76	Osmium	Os	8. 7	$5d^6 \ 6s^2 \ ^5D_4$	93	Neptunium	Np		
77	Iridium	Ir	9. 2	$5d^7 \ 6s^2 \ ^4F_{1/2}$	94	Plutonium	Pu		
78	Platinum	Pt	8. 96	$5d^9 \ 6s \ ^3D_3$	95	Americium	Am		
79	Gold	Au	9. 223	$(5d^{10}) \ 6s \ ^2S_{1/2}$	96	Curium	Cm		
80	Mercury	Hg	10. 434	$6s^2 \ ^1S_0$	97				
81	Thallium	Tl	6. 106	$6s^2 \ 6p \ ^2P_{1/2}^\circ$	98				
82	Lead	Pb	7. 415	$6s^2 \ 6p^2 \ ^3P_0$	99				
83	Bismuth	Bi	8 \pm	$6s^2 \ 6p^3 \ ^4S_{1/2}$	100				
84	Polonium	Po			101				
85	Astatine	At			102				
86	Radon	Rn	10. 745	$(6s^2 \ 6p^6) \ ^1S_0$	103				
87	Francium	Fa							

* Parentheses denote values that have been determined experimentally, but not yet confirmed by series.

TABLE 24. CHEMICAL SYMBOLS

Symbol	Element	Z	Symbol	Element	Z	Symbol	Element	Z	Symbol	Element	Z
A	Argon	18	Dy	Dysprosium	66	Mn	Manganese	25	S	Sulfur	16
Ac	Actinium	89	Er	Erbium	68	Mo	Molybdenum	42	Sb	Antimony	51
Ag	Silver	47	Eu	Europium	63	N	Nitrogen	7	Sc	Scandium	21
Al	Aluminum	13	F	Fluorine	9	Na	Sodium	11	Se	Selenium	34
Am	Americium	95	Fa	Francium	87	Nd	Neodymium	60	Si	Silicon	14
As	Arsenic	33	Fe	Iron	26	Ne	Neon	10	Sm	Samarium	62
At	Astatine	85	Ga	Gallium	31	Ni	Nickel	28	Sn	Tin	50
Au	Gold	79	Gd	Gadolinium	64	Np	Neptunium	93	Sr	Strontium	38
B	Boron	5	Ge	Germanium	32	O	Oxygen	8	Ta	Tantalum	73
Ba	Barium	56	H	Hydrogen	1	Os	Osmium	76	Tb	Terbium	65
Be	Beryllium	4	(D	Deuterium)	}	P	Phosphorus	15	Tc	Technetium	43
Bi	Bismuth	83	(T	Tritium)		Pa	Protactinium	91	Te	Tellurium	52
Br	Bromine	35	He	Helium	2	Pb	Lead	82	Th	Thorium	90
C	Carbon	6	Hf	Hafnium	72	Pd	Palladium	46	Ti	Titanium	22
Ca	Calcium	20	Hg	Mercury	80	Pm	Promethium	61	Tl	Thallium	81
Cb	Columbium	41	Ho	Holmium	67	Po	Polonium	84	Tm	Thulium	69
Cd	Cadmium	48	I	Iodine	53	Pr	Praseodymium	59	U	Uranium	92
Ce	Cerium	58	In	Indium	49	Pt	Platinum	78	V	Vanadium	23
Cl	Chlorine	17	Ir	Iridium	77	Pu	Plutonium	94	W	Tungsten	74
Cm	Curium	96	K	Potassium	19	Ra	Radium	88	Xe	Xenon	54
Co	Cobalt	27	Kr	Krypton	36	Rb	Rubidium	37	Y	Yttrium	39
Cr	Chromium	24	La	Lanthanum	57	Re	Rhenium	75	Yb	Ytterbium	70
Cs	Cesium	55	Li	Lithium	3	Rh	Rhodium	45	Zn	Zinc	30
Cu	Copper	29	Lu	Lutecium	71	Rn	Radon	86	Zr	Zirconium	40
			Mg	Magnesium	12	Ru	Ruthenium	44			

TABLE 25. THE PERIODIC SYSTEM*

1	1s	H 1	He 2
2	2s 2p	Li 3	Be 4 B 5 C 6 N 7 O 8 F 9 Ne 10
3	3s 3p	Na 11	Mg 12 Al 13 Si 14 P 15 S 16 Cl 17 A 18
4	4s 3d 4p	K 19	Ca 20 Sc 21 Ti 22 V 23 Cr 24 Mn 25 Fe 26 Co 27 Ni 28 Cu 29 Zn 30 Ga 31 Ge 32 As 33 Se 34 Br 35 Kr 36
5	5s 4d 5p	Rb 37	Sr 38 Y 39 Zr 40 Nb 41 Mo 42 Tc 43 Ru 44 Rh 45 Pd 46 Ag 47 Cd 48 In 49 Sn 50 Sb 51 Te 52 I 53 Xe 54
6	6s 4f 5d 6p	Cs 55	Ba 56 La 57 Ce 58 Pr 59 Nd 60 Pm 61 Sm 62 Eu 63 Gd 64 Tb 65 Dy 66 Ho 67 Er 68 Tm 69 Yb 70 Lu 71 Hf 72 Ta 73 W 74 Re 75 Os 76 Ir 77 Pt 78 Au 79 Hg 80 Tl 81 Pb 82 Bi 83 Po 84 At 85 Rn 86
7	7s 5f 6d	Fr 87	Ra 88 Ac 89 Th 90 Pa 91 U 92 Np 93 Pu 94 Am 95 Cm 96 97 98 99 100 101 102 103

*This arrangement is by Catalán. The electrons indicated in column two that are connected by braces have approximately the same binding energy. Consequently, for some elements one type of electron is preferred over another in the normal configuration, as for example, Cr, Nb, Pd, La, Ac, Th.

TABLE 26. INDEX—ISOELECTRONIC SEQUENCES
[The tabular entries are page numbers.]

Z	Element	Spectrum														
		I	II	III	IV	V	VI	VII	VIII	IX	X	XI	XII	XIII	XIV	XV
1	H, D, T	1, 3														
2	He	4	6													
3	Li	8	10	11												
4	Be	12	14	14	15											
5	B	16	17	19	19	20										
6	C	21	24	26	29	30	31									
7	N	32	35	38	40	42	43	44								
8	O	45	47	50	53	56	58	59	59							
9	F	60	62	64	66	69	71	74	75							
10	Ne	76	81	83	84	86	88									
11	Na	89	91	93	95	96	98	100	103	105						
12	Mg	106	108	109	111	113	114	117	119	121	122	123				
13	Al	124	126	129	130	131	133	135	136	138	140	142	143			
14	Si	144	147	148	150	151	152	154	156	157	159	160	162			
15	P	163	164	166	168	169	170	171	173	174	176	177	179	180		
16	S	181	183	185	187	188	189	190	191	193	194		194			
17	Cl	195	197	199	201	202	204	205	206	207	209	210				
18	A	211	216	218	220	222	223	224	224	225	226	226			226	
19	K	227	230	231	233	234	236	237	238	239	239	241				
20	Ca	242	245	247	248	249	251	252	253	254	255	255	257	258		258
21	Sc	259	262	263	264	265	266	267	268	269	270	271	272			
22	Ti	273	279	281	283	284	285	286	287	288	288	289	289	290		
23	V	291	298	301	303	304	304	305	306	306		307	307	308	309	

HYDROGEN

H

1 electron

$Z=1$

Ground state $1s\ ^2S_{1/2}$

$1s\ ^2S_{1/2}$ **109678.758** cm^{-1}

I. P. 13.595 volts

This table deals only with the light isotope of hydrogen, H^1 ; cf. page 3 for the other isotopes. The levels through $n=40$ have been calculated by J. E. Mack, "using $R_{\text{H}^1}=109677.581\ \text{cm}^{-1}$ and $\alpha^2=5.3256\times 10^{-5}$, and taking into account the Lamb-Retherford shift of the s -levels as well as the Sommerfeld-Dirac fine structure, according to the equation

$$\text{Level}_n - \text{Level}_\infty = R_A \left\{ -n^{-2}Z^2 + \alpha^2 n^{-3} Z^4 \left[-(J + \frac{1}{2})^{-1} + 3(4n)^{-1} + \Lambda_{nlZA} \right] + \dots \right\}.$$

Here A is the atomic weight, and α is the Sommerfeld fine-structure constant. The s -shift parameter Λ is appreciable only for $l=0$, and depends slowly upon n and Z and probably negligibly upon A ; it is found from the work of Lamb and Retherford to be 0.0485 ± 0.0002 for the $2s$ -level of hydrogen, and in the calculation of this table it is assumed to be independent of n .

The intervals are carried one place farther than the level values, insofar as they are accurately known.

The $1s\ ^2S_{1/2}$ level consists of two hyperfine structure components separated by $0.0473824 \pm 0.0000008\ \text{cm}^{-1}$, the lower of which has $F=0$ and the other $F=1$.

In any one-electron spectrum the correction arising from any modification ΔR of the value accepted for the Rydberg constant may be calculated to a close approximation from the equation

$$\Delta(\text{level}) = (1 - n^{-2})Z^2 \Delta R."$$

REFERENCES

- A. Fowler, *Report on Series in Line Spectra*, p. 89 (Fleetway Press, London, 1922). (T) (C L)
 F. Paschen und R. Götze, *Seriengesetze der Linienspektren*, p. 22 (Julius Springer, Berlin, 1922). (T) (C L)
 H. E. White, *Introduction to Atomic Spectra*, p. 33 (McGraw Hill Book Co., Inc., New York, N. Y., 1934). (G D)
 J. W. Drinkwater, O. Richardson, and W. E. Williams, *Proc. Roy. Soc. (London)* [A] **174**, 164 (1940). (Fine structure)
 C. E. Moore, *Princeton Obs. Contr. No. 20*, 1 (1945). (C L)
 H. A. Bethe, *Phys. Rev.* **72**, 339 (1947). (T)
 D. E. Nagle, R. S. Julian, and J. R. Zacharias, *Phys. Rev.* **72**, 971 (L) (1947). (hfs)
 J. E. Nafe and E. B. Nelson, *Phys. Rev.* **73**, 718 (1948). (hfs)
 H. Kuhn and G. W. Series, *Nature* **162**, 373 (1948). (Fine structure)
 W. E. Lamb, Jr., and R. E. Retherford, *Bul. Am. Phys. Soc.* **24**, No. 1, 59 (1949). (Fine structure)
 M. M. Kroll and W. E. Lamb, Jr., *Phys. Rev.* **75**, 388 (1949). (T)
 J. E. Mack, unpublished material (1949). (I P) (T) (C L)

H

H

Config.	Desig.	<i>J</i>	Level	Interval	Config	Desig.	<i>J</i>	Level	Interval
1s	1s ² S	½	0. 000		16s, etc.	16s ² S, etc.	½, etc.	109250. 33	
2p	2s ² S	½	82258. 907]]0.0354 0. 3651	17s, etc.	17s ² S, etc.	½, etc.	109299. 25	
2s		½	82258. 942		18s, etc.	18s ² S, etc.	½, etc.	109340. 25	
2p		1½	82259. 272		19s, etc.	19s ² S, etc.	½, etc.	109374. 94	
3p	3s ² S	½	97492. 198]]0. 010 0. 1082 0. 0361	20s, etc.	20s ² S, etc.	½, etc.	109404. 57	
3s		½	97492. 208		21s, etc.	21s ² S, etc.	½, etc.	109430. 06	
3p, 3d		1½	97492. 306		22s, etc.	22s ² S, etc.	½, etc.	109452. 15	
3d		2½	97492. 342		23s, etc.	23s ² S, etc.	½, etc.	109471. 428	
4p	4s ² S	½	102823. 835]]0. 004 0. 0456 0. 0152 0. 0076	24s, etc.	24s ² S, etc.	½, etc.	109488. 346	
4s		½	102823. 839		25s, etc.	25s ² S, etc.	½, etc.	109503. 274	
4p, 4d		1½	102823. 881		26s, etc.	26s ² S, etc.	½, etc.	109516. 513	
4d, 4f		2½	102823. 896		27s, etc.	27s ² S, etc.	½, etc.	109528. 309	
4f		3½	102823. 904		28s, etc.	28s ² S, etc.	½, etc.	109538. 863	
5p	5s ² S	½	105291. 615]]0. 002 0. 0233 0. 0078 0. 0039 0. 0024	29s, etc.	29s ² S, etc.	½, etc.	109548. 345	
5s		½	105291. 617		30s, etc.	30s ² S, etc.	½, etc.	109556. 894	
5p, 5d		1½	105291. 638		31s, etc.	31s ² S, etc.	½, etc.	109564. 629	
5d, 5f		2½	105291. 646		32s, etc.	32s ² S, etc.	½, etc.	109571. 651	
5f, 5g		3½	105291. 650		33s, etc.	33s ² S, etc.	½, etc.	109578. 044	
5g		4½	105291. 652		34s, etc.	34s ² S, etc.	½, etc.	109583. 881	
6p	6s ² S	½	106632. 135]]0. 001 0. 0136 0. 0045 0. 0022 0. 0014 0. 0009	35s, etc.	35s ² S, etc.	½, etc.	109589. 225	
6s		½	106632. 136		36s, etc.	36s ² S, etc.	½, etc.	109594. 130	
6p, 6d		1½	106632. 148		37s, etc.	37s ² S, etc.	½, etc.	109598. 643	
6d, 6f		2½	106632. 152		38s, etc.	38s ² S, etc.	½, etc.	109602. 804	
6f, 6g		3½	106632. 155		39s, etc.	39s ² S, etc.	½, etc.	109606. 649	
6g, 6h		4½	106632. 156		40s, etc.	40s ² S, etc.	½, etc.	109610. 210	
6h		5½	106632. 157						
7s, etc.	7s ² S, etc.	½, etc.	107440. 425 to . 439	0. 014					
8s, etc.	8s ² S, etc.	½, etc.	107965. 036 to . 045	0. 009					
9s, etc.	9s ² S, etc.	½, etc.	108324. 706 to . 714	0. 008					
10s, etc.	10s ² S, etc.	½, etc.	108581. 98						
11s, etc.	11s ² S, etc.	½, etc.	108772. 33						
12s, etc.	12s ² S, etc.	½, etc.	108917. 11						
13s, etc.	13s ² S, etc.	½, etc.	109029. 78						
14s, etc.	14s ² S, etc.	½, etc.	109119. 18						
15s, etc.	15s ² S, etc.	½, etc.	109191. 30						
						∞=Limit		109678. 758	

February 1949.

DEUTERIUM and TRITIUM

D and T

1 electron

 $Z=1$ Ground state $1s\ ^2S_{1/2}$ $1s\ ^2S_{1/2}\ D\ (H^2)\ 109708.596\ \text{cm}^{-1}$

I. P. D 13.598 volts

 $1s\ ^2S_{1/2}\ T\ (H^3)\ 109718.526\ \text{cm}^{-1}$

I. P. T 13.600 volts

The term values have been calculated by J. E. Mack, "using $R_D=109707.419$ and $R_T=109717.348\ \text{cm}^{-1}$, and taking into account the same fine structure as in hydrogen. Lamb and Retherford have found that the $2s$ -shift in deuterium is the same as in light hydrogen within about 0.5 percent. Levels not given here may be calculated from the hydrogen table with the aid of the correction equations

$$\text{Level}_D - \text{Level}_H = (1 - n^{-2})29.838\ \text{cm}^{-1} \text{ and } \text{Level}_T - \text{Level}_H = (1 - n^{-2})39.768\ \text{cm}^{-1}.$$

Nafe and Nelson have kindly communicated the results of their hyperfine structure measurements in tritium in advance of publication. In both isotopes the $1s$ -level has two hyperfine-structure components, the lower of which has the lower F -value. In deuterium the separation is $0.01092095 \pm 0.00000023\ \text{cm}^{-1}$, and the F -values are $1/2$ and $3/2$. In tritium the separation is $0.0505945 \pm 0.0000010\ \text{cm}^{-1}$, the F -values 0 and 1."

REFERENCES

- J. W. Drinkwater, O. Richardson, and W. E. Williams, Proc. Roy. Soc. (London) [A] **174**, 164 (1940). (Fine structure) (I S)
 D. E. Nagle, R. S. Julian, and J. R. Zacharias, Phys. Rev. **72**, 971 (L) (1947). (hfs)
 J. E. Nafe and E. B. Nelson, Phys. Rev. **73**, 718 (1948); **75**, in press (1949). (hfs)
 R. E. Retherford and W. E. Lamb, Jr., Bul. Am. Phys. Soc. **24**, No. 1, 59 (1949). (Fine structure)
 J. E. Mack, unpublished material (1949). (I P) (T) (C L)

D T

Config.	Desig.	J	Level	Level	Interval
$1s$	$1s\ ^2S$	$\frac{1}{2}$	0.000	0.000	
$2p$	$2s\ ^2S$	$\frac{1}{2}$	82281.285	82288.733]] 0.0354
$2s$		$\frac{1}{2}$	82281.320	82288.768	
$2p$		$1\frac{1}{2}$	82281.650	82289.098	
$3p$	$3s\ ^2S$	$\frac{1}{2}$	97518.721	97527.547]] 0.010
$3s$		$\frac{1}{2}$	97518.731	97527.558	
$3p, 3d$		$1\frac{1}{2}$	97518.829	97527.656	
$3d$		$2\frac{1}{2}$	97518.865	97527.692	
$4p$	$4s\ ^2S$	$\frac{1}{2}$	102851.808	102861.118]] 0.004
$4s$		$\frac{1}{2}$	102851.812	102861.122	
$4p, 4d$		$1\frac{1}{2}$	102851.854	102861.163	
$4d, 4f$		$2\frac{1}{2}$	102851.869	102861.178	
$4f$		$3\frac{1}{2}$	102851.877	102861.186	
$5p$	$5s\ ^2S$	$\frac{1}{2}$	105320.260	105329.792]] 0.002
$5s$		$\frac{1}{2}$	105320.262	105329.795	
$5p, 5d$		$1\frac{1}{2}$	105320.283	105329.816	
$5d, 5f$		$2\frac{1}{2}$	105320.291	105329.824	
$5f, 5g$		$3\frac{1}{2}$	105320.294	105329.827	
$5g$		$4\frac{1}{2}$	105320.297	105329.830	
$6p$	$6s\ ^2S$	$\frac{1}{2}$	106661.144	106670.798]] 0.001
$6s$		$\frac{1}{2}$	106661.145	106670.800	
$6p, 6d$		$1\frac{1}{2}$	106661.158	106670.812	
$6d, 6f$		$2\frac{1}{2}$	106661.162	106670.816	
$6f, 6g$		$3\frac{1}{2}$	106661.164	106670.818	
$6g, 6h$		$4\frac{1}{2}$	106661.166	106670.820	
$6h$		$5\frac{1}{2}$	106661.167	106670.821	
$7s, \text{etc.}$	$7s\ ^2S, \text{etc.}$	$\frac{1}{2}, \text{etc.}$	107469.654 to .669	107479.381 to .396	
	----- $\infty = \text{Limit}$	-----	109708.596	109718.526	

HELIUM

He I

2 electrons

 $Z=2$ Ground state $1s^2\ ^1S_0$ $1s^2\ ^1S_0\ 198305 \pm 15\ \text{cm}^{-1}$

I. P. 24.580 volts

Most of the terms are taken from Paschen-Götze with the term values subtracted from Paschen's limit as quoted by Robinson in 1937. Higher members of the $^1F^\circ$ and $^3F^\circ$ series are taken from Meggers and Dieke. The term $2p\ ^3P^\circ$ has been calculated from its combination with $2s\ ^3S_1$, using the resolved triplet as observed by Meggers, the intervals being $-0.078\ \text{cm}^{-1}$ and $-0.996\ \text{cm}^{-1}$. The components of $3p\ ^3P^\circ$ are based on Paschen's value of $3p\ ^3P_2^\circ$ and the intervals observed by Gibbs and Kruger; $-0.165\ \text{cm}^{-1}$ and $-0.192\ \text{cm}^{-1}$.

Some doubt exists regarding the correct classifications of lines attributed to doubly excited helium, such as those observed at 309.04 Å and 320.38 Å by Compton and Boyce, and at 320.392 Å and 357.507 Å by Kruger. Approximate theoretical computations of the energies of doubly excited levels have been made by a number of authors and are summarized by Wu. His classification of the line observed at 320.4 Å as $2p\ ^3P^\circ - 2p^2\ ^3P$ has been adopted and used for the calculation of $2p^2\ ^3P$.

Several references deal with intercombinations in He I, namely, those by Lyman, Hopfield, Paschen, Suga, and others. The term values based on the excellent long series have been adopted in the table, since it is believed that they are the most accurate.

REFERENCES

- F. Paschen und R. Götze, *Seriengesetze der Linienspektren* p. 22 (Julius Springer, Berlin, 1922). (T) (C L)
 T. Lyman, *Astroph. J.* **60**, 1 (1924). (T) (C L)
 K. T. Compton and J. C. Boyce, *J. Franklin Inst.* **205**, 497 (1928). (C L)
 F. Paschen, *Sitz. Berlin Akad. Wiss.* **30**, 662 (1929). (T) (C L)
 J. J. Hopfield, *Astroph. J.* **72**, 133 (1930). (T) (C L)
 P. G. Kruger, *Phys. Rev.* **36**, 855 (1930). (C L)
 R. C. Gibbs and P. G. Kruger, *Phys. Rev.* **37**, 1559 (1931). (T)
 W. F. Meggers and G. H. Dieke, *Bur. Std. J. Research* **9**, 121, RP462 (1932). (T) (C L)
 F. Paschen and R. Ritschl, *Ann. der Phys.* [5] **18**, 888 (1933). (T) (C L)
 H. E. White, *Introduction to Atomic Spectra* p. 209 (McGraw-Hill Book Co., Inc., New York, N. Y., 1934). (G D)
 W. F. Meggers, *J. Research Nat. Bur. Std.* **14**, 487, RP781 (1935). (C L)
 T. Suga, *Sci. Papers Inst. Phys. Chem. Research (Tokyo)* **34**, No. 740, 16 (1937). (C L)
 H. A. Robinson, *Phys. Rev.* **51**, 14 (1937). (I P)
 P. Jacquinet, *Compt. Rend.* **208**, 1896 (1939). (C L)
 T.-Y. Wu, *Phys. Rev.* **66**, 291 (1944). (C L)
 W. F. Meggers, *J. Opt. Soc. Am.* **36**, 431 (1946). (Summary hfs)

He I

He I

Config.	Desig.	<i>J</i>	Level	Config.	Desig.	<i>J</i>	Level
1s ²	1s ² 1S	0	0 ± 15	1s 7s	7s 1S	0	195973. 19
1s 2s	2s 3S	1	159850. 318	1s 7p	7p 3P°	2, 1, 0	196021. 72
1s 2s	2s 1S	0	166271. 70	1s 7d	7d 3D	3, 2, 1	196064. 00
1s 2p	2p 3P°	2 1 0	169081. 111 169081. 189 169082. 185	1s 7d	7d 1D	2	196064. 31
1s 2p	2p 1P°	1	171129. 148	1s 7f	7f 1F°	3	196065. 4
1s 3s	3s 3S	1	183231. 08	1s 7f	7f 3F°	4, 3, 2	196065. 51
1s 3s	3s 1S	0	184859. 06	1s 7p	7p 1P°	1	196073. 41
1s 3p	3p 3P°	2 1 0	185558. 92 185559. 085 185559. 277	1s 8s	8s 3S	1	196455. 79
1s 3d	3d 3D	3, 2, 1	186095. 90	1s 8s	8s 1S	0	196529. 03
1s 3d	3d 1D	2	186099. 22	1s 8p	8p 3P°	2, 1, 0	196561. 08
1s 3p	3p 1P°	1	186203. 62	1s 8d	8d 3D	3, 2, 1	196589. 42
1s 4s	4s 3S	1	190292. 46	1s 8d	8d 1D	2	196589. 73
1s 4s	4s 1S	0	190934. 50	1s 8f	8f 1F°	3	196590. 3
1s 4p	4p 3P°	2, 1, 0	191211. 42	1s 8f	8f 3F°	4, 3, 2	196590. 42
1s 4d	4d 3D	3, 2, 1	191438. 83	1s 8p	8p 1P°	1	196595. 56
1s 4d	4d 1D	2	191440. 71	1s 9s	9s 3S	1	196856. 37
1s 4f	4f 3F°	4, 3, 2	191446. 61	1s 9s	9s 1S	0	196907. 13
1s 4f	4f 1F°	3	191447. 24	1s 9p	9p 3P°	2, 1, 0	196929. 68
1s 4p	4p 1P°	1	191486. 95	1s 9d	9d 1D	2	196949. 49
1s 5s	5s 3S	1	193341. 33	1s 9d	9d 3D	3, 2, 1	196949. 63
1s 5s	5s 1S	0	193657. 78	1s 9f	9f 1F°	3	196950. 3
1s 5p	5p 3P°	2, 1, 0	193795. 07	1s 9f	9f 3F°	4, 3, 2	196950. 36
1s 5d	5d 3D	3, 2, 1	193911. 48	1s 9p	9p 1P°	1	196953. 95
1s 5d	5d 1D	2	193912. 54	1s 10s	10s 3S	1	197139. 76
1s 5f	5f 1F°	3	193914. 31	1s 10s	10s 1S	0	197176. 36
1s 5f	5f 3F°	4, 3, 2	193915. 79	1s 10p	10p 3P°	2, 1, 0	197192. 63
1s 5p	5p 1P°	1	193936. 75	1s 10d	10d 1D	2	197207. 08
1s 6s	6s 3S	1	194930. 46	1s 10d	10d 3D	3, 2, 1	197207. 30
1s 6s	6s 1S	0	195109. 17	1s 10f	10f 3F°	4, 3, 2	197208. 0
1s 6p	6p 3P°	2, 1, 0	195187. 21	1s 10p	10p 1P°	1	197210. 41
1s 6d	6d 3D	3, 2, 1	195254. 37	1s 11s	11s 3S	1	197347. 05
1s 6d	6d 1D	2	195255. 02	1s 11p	11p 3P°	2, 1, 0	197386. 98
1s 6f	6f 1F°	3	195256. 7	1s 11d	11d 1D	2	197397. 62
1s 6f	6f 3F°	4, 3, 2	195256. 82	1s 11d	11d 3D	3, 2, 1	197397. 75
1s 6p	6p 1P°	1	195269. 17	1s 11f	11f 3F°	4, 3, 2	197398. 6
1s 7s	7s 3S	1	195862. 63	1s 11p	11p 1P°	1	197400. 18
				1s 12s	12s 3S	1	197503. 69
				1s 12s	12s 1S	0	197524. 26

He I—Continued

He I—Continued

Config.	Desig.	<i>J</i>	Level	Config.	Desig.	<i>J</i>	Level
1s 12p	12p ³ P°	2, 1, 0	197534. 44	1s 16d	16d ³ D	3, 2, 1	197876. 41
1s 12d	12d ¹ D	2	197542. 54	1s 16p	16p ¹ P°	1	197877. 04
1s 12d	12d ³ D	3, 2, 1	197542. 67	1s 17p	17p ³ P°	2, 1, 0	197922. 51
1s 12p	12p ¹ P°	1	197544. 56	1s 17d	17d ³ D	3, 2, 1	197925. 33
1s 13s	13s ³ S	1	197624. 98	1s 17p	17p ¹ P°	1	197925. 87
1s 13p	13p ³ P°	2, 1, 0	197649. 07	1s 18p	18p ³ P°	2, 1, 0	197964. 02
1s 13s	13s ¹ S	0	197649. 78	1s 18d	18d ³ D	3, 2, 1	197966. 75
1s 13d	13d ¹ D	2	197655. 19	1s 18p	18p ¹ P°	1	197966. 80
1s 13d	13d ³ D	3, 2, 1	197655. 47	1s 19p	19p ³ P°	2, 1, 0	197999. 12
1s 13p	13p ¹ P°	1	197656. 95	1s 19d	19d ³ D	3, 2, 1	198001. 43
1s 14s	14s ³ S	1	197721. 13	1s 19p	19p ¹ P°	1	198001. 44
1s 14p	14p ³ P°	2, 1, 0	197739. 90	1s 20p	20p ³ P°	2, 1, 0	198029. 07
1s 14d	14d ¹ D	2	197744. 918	1s 20p	20p ¹ P°	1	198031. 02
1s 14d	14d ³ D	3, 2, 1	197744. 94	1s 20d	20d ³ D	3, 2, 1	198031. 41
1s 14p	14p ¹ P°	1	197746. 15	1s 21p	21p ³ P°	2, 1, 0	198054. 83
1s 15s	15s ³ S	1	197796. 63	1s 21d	21d ³ D	3, 2, 1	198056. 50
1s 15p	15p ³ P°	2, 1, 0	197813. 11	1s 22p	22p ³ P°	2, 1, 0	198077. 15
1s 15d	15d ³ D	3, 2, 1	197817. 05				
1s 15p	15p ¹ P°	1	197818. 12	He II (² S _{1/2})	<i>Limit</i>		198305
1s 16p	16p ³ P°	2, 1, 0	197872. 95	2p ²	2p ² ³ P	2, 1, 0	481198

August 1946.

He II

(H sequence; 1 electron)

Z=2Ground state 1s ²S_{1/2}1s ²S_{1/2} He³ 438889.040 cm⁻¹I. P. He³ 54.400 volts1s ²S_{1/2} He⁴ 438908.670 cm⁻¹I. P. He⁴ 54.403 volts

The levels have been calculated by J. E. Mack, "using $R_{\text{He}^4}=109722.264$ and taking into account the fine structure as in hydrogen, but with $\Delta=0.0402\pm0.009$, from the work of Skinner and Lamb on the 2s-level. The tentative experimental indication that Δ decreases with increasing n has been neglected. Assuming $R_{\text{He}^3}=109717.344$, the levels of He³ may be calculated to a close approximation from those of He⁴ by the equation

$$\text{Level}_{\text{He}^3, \text{II}} - \text{Level}_{\text{He}^4, \text{II}} = -(1-n^{-2})19.630 \text{ cm}^{-1}."$$

REFERENCES

- A. Fowler, *Report on Series in Line Spectra*, p. 95 (Fleetway Press, London, 1922). (T) (C L)
 F. Paschen und R. Götze, *Seriengesetze der Linienpektren*, p. 25 (Julius Springer, Berlin, 1922). (T) (C L)
 H. E. White, *Introduction to Atomic Spectra*, p. 33 (McGraw-Hill Book Co., New York, N. Y., 1934). (G D)
 C. E. Moore, Princeton Obs. Contr. No. 20, 1 (1945). (C L)
 H. A. Bethe, Phys. Rev. **72**, 339 (1947). (T)
 J. E. Mack and N. Austern, Phys. Rev. **72**, 972 (1947); **74**, 1262 (A) (1948). (Fine structure)
 G. R. Fowles, Phys. Rev. **73**, 639 (L) (1948); **74**, 219 (L) (1948). (Fine structure)
 H. Kopfermann and W. Paul, Nature **162**, 33 (L) (1948). (Fine structure)
 M. Skinner and W. E. Lamb, Jr., Bul. Am. Phys. Soc. **24**, No. 1, 59 (1949). (Fine structure)
 J. E. Mack, unpublished material (1949). (I P) (T) (C L)

He ³ II				He ⁴ II	
Config.	Desig.	<i>J</i>	Level	Level	Interval
1s	1s ² S	½	0.000	0.000	
2p	2s ² S	½	329164.390	329179.102]] 0.470
2s		½	329164.860	329179.572	
2p		1½	329170.135	329184.945	
3p	3s ² S	½	390123.179	390140.622]] 0.14
3s		½	390123.318	390140.761	
3p, 3d		1½	390124.910	390142.353	
3d		2½	390125.487	390142.930	
4p	4s ² S	½	411458.517	411476.917]] 0.06
4s		½	411458.576	411476.976	
4p, 4d		1½	411459.248	411477.648	
4d, 4f		2½	411459.491	411477.891	
4f		3½	411459.613	411478.013	
5p	5s ² S	½	421333.629	421352.472]] 0.03
5s		½	421333.659	421352.502	
5p, 5d		1½	421334.003	421352.846	
5d, 5f		2½	421334.128	421352.971	
5f, 5g		3½	421334.190	421353.033	
5g		4½	421334.228	421353.071	
6p	6s ² S	½	426697.845	426716.928]] 0.02
6s		½	426697.862	426716.945	
6p, 6d		1½	426698.062	426717.145	
6d, 6f		2½	426698.134	426717.217	
6f, 6g		3½	426698.170	426717.253	
6g, 6h		4½	426698.192	426717.275	
6h		5½	426698.206	426717.289	
7s, etc.	7s ² S, etc.	½, etc.	-----	429951.508 to .741	
8s, etc.	8s ² S, etc.	½, etc.	-----	432050.863 to 1.023	
9s, etc.	9s ² S, etc.	½, etc.	-----	433490.169 to .283	
10s, etc.	10s ² S, etc.	½, etc.	-----	434519.693 to .777	
11s, etc.	11s ² S, etc.	½, etc.	-----	435281.423 to .486	
12s, etc.	12s ² S, etc.	½, etc.	-----	435860.778 to .828	
13s, etc.	13s ² S, etc.	½, etc.	-----	436311.653 to .692	
14s, etc.	14s ² S, etc.	½, etc.	-----	436669.407 to .439	
15s, etc.	15s ² S, etc.	½, etc.	-----	436957.026 to 8.052	
	∞= <i>Limit</i>	-----	-----	438908. 670	

LITHIUM

Li I

3 electrons

 $Z=3$ Ground state $1s^2 2s^2 S_{\frac{1}{2}}$ $2s^2 S_{\frac{1}{2}}$ $43487.19 \pm 0.02 \text{ cm}^{-1}$

I. P. 5.390 volts

The analysis is from Fowler and Paschen-Götze. Meissner has generously furnished in advance of publication preliminary results of level splittings derived from observed fine structure of selected lines. These data are as follows:

Term	Interval (cm^{-1})	Line resolved (Å)	Term	Line resolved (Å)
$2p^2 P^{\circ}$	$0.3366 \pm 0.0005^*$	6707.912, .761	$3s^2 S$	8126.452, .231
$3d^2 D$	0.037 ± 0.001	6103.649, .538	$4s^2 S$	4971.745, .661
$4d^2 D$	0.015 ± 0.002	4602.894, .826	$5s^2 S$	4273.127, .066
$5d^2 D$	0.010 ± 0.003	4132.618, .562†	$6s^2 S$	3985.538, .485
$6d^2 D$	0.005 ± 0.003	3915.346, .295		

*Average of 6 determinations.

†Edlén and Lidén derive a mean value of $4132.60 \pm 0.02 \text{ Å}$ and the resulting corrected values quoted for $5d^2 D$ and the limit.

The values in the table for the above terms have been calculated from these wavelengths, except for $5d^2 D$. Jackson and Kuhn state that the multiplet splitting of $2p^2 P^{\circ} = 0.3372 \pm 0.0005 \text{ cm}^{-1}$.

The remaining terms given to two decimals have been calculated from the measures by France. The terms $ns^2 S$, $n=7$ to 11, and $nd^2 D$, $n=7$ to 12, are from Werner. All other term values are from Fowler's Report.

REFERENCES

- N. A. Kent, *Astroph. J.* **40**, 337 (1914). (T) (Z E)
A. S. King, *Astroph. J.* **44**, 169 (1916). (T)
A. Fowler, *Report on Series in Line Spectra*, p. 96 (Fleetway Press, London, 1922). (T) (C L)
F. Paschen und R. Götze, *Seriengesetze der Linienspektren*, p. 54, (Julius Springer, Berlin, 1922). (T) (C L)
S. Werner, *Studier over Spektroskopiske Lyskilder til Frembringelse af Gnistspektre med Resultater for Lithiums Gnistspektrum*, p. 67 (A. Aschehoug & Co., Dansk Forlag, Kobenhavn, 1927). (I P) (T) (C L)
R. W. France, *Proc. Roy. Soc. (London) [A]* **129**, 354 (1930). (I P) (T) (C L)
H. E. White, *Introduction to Atomic Spectra*, p. 77, 87, (McGraw-Hill Book Co., Inc., New York, N. Y. 1934). (G D)
D. A. Jackson and H. Kuhn, *Proc. Roy. Soc. (London) [A]* **173**, 278 (1939). (I S)
W. F. Meggers, *J. Opt. Soc. Am.* **36**, 431 (1946). (Summary hfs).
K. W. Meissner, L. G. Mundie, and P. Stelson, *Phys. Rev.* **74**, 932 (1948); **75**, 891 (L) (1949). (T) (C L)
B. Edlén and K. Lidén, *Phys. Rev.* **75**, 890 (L) (1949). (I P) (T)

Li I

Li I

Config.	Desig.	<i>J</i>	Level	Config.	Desig.	<i>J</i>	Level
2s	2s ² S	$\frac{1}{2}$	0. 00	12d	12d ² D	$1\frac{1}{2}, 2\frac{1}{2}$	42725
2p	2p ² P°	$\frac{1}{2}$ $1\frac{1}{2}$	14903. 66 14904. 00	13p	13p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	42832. 92
3s	3s ² S	$\frac{1}{2}$	27206. 12	14p	14p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	42923. 39
3p	3p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	30925. 38	15p	15p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	42995. 51
3d	3d ² D	$1\frac{1}{2}$ $2\frac{1}{2}$	31283. 08 31283. 12	16p	16p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43055. 34
4s	4s ² S	$\frac{1}{2}$	35012. 06	17p	17p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43105. 42
4p	4p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	36469. 55	18p	18p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43146. 96
4d	4d ² D	$1\frac{1}{2}$ $2\frac{1}{2}$	36623. 38 36623. 40	19p	19p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43181. 84
4f	4f ² F°	$2\frac{1}{2}, 3\frac{1}{2}$	36630. 2	20p	20p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43211. 39
5s	5s ² S	$\frac{1}{2}$	38299. 50	21p	21p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43237. 16
5p	5p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	39015. 56	22p	22p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43259. 14
5d	5d ² D	$1\frac{1}{2}$ $2\frac{1}{2}$	39094. 93 39094. 94	23p	23p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43278. 96
6f	5f ² F°	$2\frac{1}{2}, 3\frac{1}{2}$	39104. 5	24p	24p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43296. 03
6s	6s ² S	$\frac{1}{2}$	39987. 64	25p	25p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43311. 45
6p	6p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	40390. 84	26p	26p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43324. 81
6d	6d ² D	$1\frac{1}{2}$ $2\frac{1}{2}$	40437. 31 40437. 32	27p	27p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43336. 40
7s	7s ² S	$\frac{1}{2}$	40967. 9	28p	28p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43346. 39
7p	7p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	41217. 35	29p	29p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43354. 91
7d	7d ² D	$1\frac{1}{2}, 2\frac{1}{2}$	41246. 5	30p	30p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43363. 71
10d	10d ² D	$1\frac{1}{2}, 2\frac{1}{2}$	41489	31p	31p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43372. 06
8s	8s ² S	$\frac{1}{2}$	41587. 1	32p	32p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43378. 31
8p	8p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	41751. 63	33p	33p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43384. 9
8d	8d ² D	$1\frac{1}{2}, 2\frac{1}{2}$	41771. 3	34p	34p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43390. 3
9s	9s ² S	$\frac{1}{2}$	42003. 3	35p	35p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43395. 4
9p	9p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	42118. 27	36p	36p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43400. 5
9d	9d ² D	$1\frac{1}{2}, 2\frac{1}{2}$	42131. 3	37p	37p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43404. 7
10s	10s ² S	$\frac{1}{2}$	42298	38p	38p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43408. 6
10p	10p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	42379. 16	39p	39p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43412. 4
11s	11s ² S	$\frac{1}{2}$	42510	40p	40p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43416. 9
11p	11p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	42569. 1	41p	41p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43420. 9
11d	11d ² D	$1\frac{1}{2}, 2\frac{1}{2}$	42578	42p	42p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	43424. 3
12p	12p ² P°	$\frac{1}{2}, 1\frac{1}{2}$	42719. 14				
				Li II (¹ S ₀)	Limit		43487. 19

Li II

(He I sequence; 2 electrons)

 $Z=3$ Ground state $1s^2 {}^1S_0$ $1s^2 {}^1S_0$ 610079 \pm 25 cm^{-1} I. P. 75.6193 \pm 0.0031 volts

Singlet series have been published by both Schüler and Werner, the longer ones by Schüler. In the term list Schüler's rounded off values have been used for the terms $4s$ to $7s {}^1S$, $5d$ to $8d {}^1D$ and $8f {}^1F^\circ$. The limit is from Robinson and the $2p$ to $4p {}^1P^\circ$ terms are from Edlén. All the remaining terms are from Werner, who gives also an extrapolated value of $2s {}^1S_0$, entered in brackets in the table.

Intersystem combinations have not been observed, but the long series should give a reliable determination of the relative positions of the singlet and triplet terms.

REFERENCES

- H. Schüler. Zeit. Phys. **37**, 568 (1926). (T) (C L)
 S. Werner, Nature **116**, 574 (L) (1925); **118**, 154 (L) (1926). (T) (C L)
 S. Werner, *Studier over Spektroskopiske Lyskilder til Frembringelse af Gnistspektre med Resultater for Lithiums Gnistspektrum*, p. 59 (H. Aschehoug & Co., Dansk Forlag, Kobenhavn, 1927). (I P) (T) (C L).
 B. Edlén, Nova Acta Reg. Soc. Sci. Uppsala [IV] **9**, No. 6, 31 (1934). (T) (C L)
 H. E. White, *Introduction to Atomic Spectra* p. 209 (McGraw-Hill Book Co., Inc., New York, N. Y., 1934). (G D)
 H. A. Robinson, Phys. Rev. **51**, 14 (1937). (I P) (T) (C L)
 W. F. Meggers, J. Opt. Soc. Am. **36**, 431 (1946). (Summary hfs)

Li II

Li II

Author	Config.	Desig.	J	Level	Author	Config.	Desig.	J	Level
$1s^2 {}^1S$	$1s^2$	$1s^2 {}^1S$	0	0	4F	$1s 4f$	$4f {}^1F^\circ$	3	582645
$2s$	$1s 2s$	$2s {}^3S$	1	476046	$1s 4p {}^1P$	$1s 4p$	$4p {}^1P^\circ$	1	582832
$2s$	$1s 2s$	$2s {}^1S$	0	[490079]	$5s$	$1s 5s$	$5s {}^3S$	1	591184
$2p$	$1s 2p$	$2p {}^3P^\circ$	2, 1, 0	494273	5S	$1s 5s$	$5s {}^1S$	0	591984
$1s 2p {}^1P$	$1s 2p$	$2p {}^1P^\circ$	1	501816	$5p$	$1s 5p$	$5p {}^3P^\circ$	2, 1, 0	592141
$3s$	$1s 3s$	$3s {}^3S$	1	554761	$5d$	$1s 5d$	$5d {}^3D$	3, 2, 1	592505
3S	$1s 3s$	$3s {}^1S$	0	558779	5D	$1s 5d$	$5d {}^1D$	2	592508
$3p$	$1s 3p$	$3p {}^3P^\circ$	2, 1, 0	559501	5F	$1s 5f$	$5f {}^1F^\circ$	3	592523
$3d$	$1s 3d$	$3d {}^3D$	3, 2, 1	561245	$5f$	$1s 5f$	$5f {}^3F^\circ$	4, 3, 2	592527
3D	$1s 3d$	$3d {}^1D$	2	561276	5P	$1s 5p$	$5p {}^1P^\circ$	1	592639
$1s 3p {}^1P$	$1s 3p$	$3p {}^1P^\circ$	1	561749	$6s$	$1s 6s$	$6s {}^3S$	1	597122
$4s$	$1s 4s$	$4s {}^3S$	1	579982	6S	$1s 6s$	$6s {}^1S$	0	597574
4S	$1s 4s$	$4s {}^1S$	0	581590	$6p$	$1s 6p$	$6p {}^3P^\circ$	2, 1, 0	597666
$4p$	$1s 4p$	$4p {}^3P^\circ$	2, 1, 0	581897	$6d$	$1s 6d$	$6d {}^3D$	3, 2, 1	597876
$4d$	$1s 4d$	$4d {}^3D$	3, 2, 1	582612	6D	$1s 6d$	$6d {}^1D$	2	597877
4D	$1s 4d$	$4d {}^1D$	2	582631	$6f$	$1s 6f$	$6f {}^3F^\circ$	4, 3, 2	597886
$4f$	$1s 4f$	$4f {}^3F^\circ$	4, 3, 2	582644	6F	$1s 6f$	$6f {}^1F^\circ$	3	597886

Li II—Continued

Li II—Continued

Author	Config.	Desig.	<i>J</i>	Level	Author	Config.	Desig.	<i>J</i>	Level
7s	1s 7s	7s ³ S	1	600641	8D	1s 8d	8d ¹ D	2	603214
7S	1s 7s	7s ¹ S	0	600925	8f	1s 8f	8f ³ F°	4, 3, 2	603221
7d	1s 7d	7d ³ D	3, 2, 1	601115	8F	1s 8f	8f ¹ F°	3	603221
7D	1s 7d	7d ¹ D	2	601115					
7f	1s 7f	7f ³ F°	4, 3, 2	601121		Li III (² S _{1/2})	Limit		610079
7F	1s 7f	7f ¹ F°	3	601122					

May 1946.

Li III

(H sequence; 1 electron)

Z=3Ground state 1s ²S_{1/2}1s ²S_{1/2} Li⁶ III 987644.9 cm⁻¹I. P. Li⁶ III 122.419 volts1s ²S_{1/2} Li⁷ III 987657.8 cm⁻¹I. P. Li⁷ III 122.420 volts

Edlén and Ericson found two lines of the Lyman series, and Gale and Hoag found three more and the first Balmer line. Edlén points out that careful measurement of the Lyman line in orders up to the twelfth showed it definitely to the red of the value calculated from the Dirac theory, with an average discrepancy of about 20 cm⁻¹. This disagreement vanishes when the 1s-shift, calculated at 19 cm⁻¹, is taken into account, according to Mack.

J. E. Mack has calculated the terms listed here, "using $R_{Li^7}=109728.723$ and the same value of Δ as in He II, which probably makes the listed ionization energy too low by something between 0 and 2 cm⁻¹. Assuming $R_{Li^6}=109727.295$, the levels of Li⁶ may be found from the equation

$$\text{Level}_{Li^6} - \text{level}_{Li^7} = -(1 - n^{-2})12.9 \text{ cm}^{-1}."$$

REFERENCES

- H. G. Gale and J. B. Hoag, Phys. Rev. **37**, 1703 (A) (1931). (C L)
 B. Edlén and A. Ericson, Nature **125**, 233 (1930); **127**, 405 (1931); Zeit. Phys. **59**, 656 (1930). (CL)
 J. E. Mack, unpublished material (1949). (I P) (T) (C L)

Li III

Li III

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
1s	1s ² S	$\frac{1}{2}$	0. 0		5p	5p ² P°	$\frac{1}{2}$	948152. 2]] 0. 2 1. 89 0. 64 0. 31 0. 19
2p	2p ² P°	$\frac{1}{2}$	740731. 2]] 2. 4 29. 58	5s	5s ² S	$\frac{1}{2}$	948152. 4	
2s	2s ² S	$\frac{1}{2}$	740733. 6		5p, 5d	5d ² D, 5p ² P°	$1\frac{1}{2}$	948154. 1	
2p	2p ² P°	$1\frac{1}{2}$	740760. 8		5d, 5f	5d ² D, 5f ² F°	$2\frac{1}{2}$	948154. 8	
					5f, 5g	5g ² G, 5f ² F°	$3\frac{1}{2}$	948155. 1	
					5g	5g ² G	$4\frac{1}{2}$	948155. 3	
3p	3p ² P°	$\frac{1}{2}$	877915. 9]] 0. 7 8. 77 2. 92	6s, etc.	6s ² S, etc.	$\frac{1}{2}$, etc.	960223. 7 to 5. 5	
3s	3s ² S	$\frac{1}{2}$	877916. 6						
3p, 3d	3d ² D, 3p ² P°	$1\frac{1}{2}$	877924. 7						
3d	3d ² D	$2\frac{1}{2}$	877927. 6		7s, etc.	7s ² S, etc.	$\frac{1}{2}$, etc.	967502. 3 to 3. 5	
4p	4p ² P°	$\frac{1}{2}$	925929. 4]] 0. 3 3. 70 1. 23 0. 62					
4s	4s ² S	$\frac{1}{2}$	925929. 7						
4p, 4d	4d ² D, 4p ² P°	$1\frac{1}{2}$	925933. 1						
4d, 4f	4d ² D, 4f ² F°	$2\frac{1}{2}$	925934. 3						
4f	4f ² F°	$3\frac{1}{2}$	925934. 9						
						$\infty = \text{Limit}$		987657. 8	

February 1949.

BERYLLIUM

Be I

4 electrons

 $Z=4$ Ground state $1s^2 2s^2 {}^1S_0$ $2s^2 {}^1S_0$ 75192.29 cm^{-1}

I. P. 9.320 volts

All but four of the terms are from the work of Paschen or Paschen and Kruger. According to Paschen no intersystem combinations have been observed. The relative positions of the singlet and triplet terms are, however, excellently determined by long series with a relative uncertainty x not exceeding $\pm 2 \text{ cm}^{-1}$.

The predicted position of the resonance line, $2s^2 {}^1S_0 - 2p {}^3P_1^\circ$, is 4548.29 Å. Paton and Nusbaum have observed a line at 4553.07 Å to which they assign this classification, but their result has not been confirmed.

The term values of higher series members, calculated from the series formula but not substantiated by observation, are in brackets in the table.

Four terms are from Edlén's work: $2p^2 {}^1D$, $3p {}^3P^\circ$, $2p^2 {}^1S$, and $3p {}^3P$.

REFERENCES

- R. F. Paton and R. E. Nusbaum, Phys. Rev. **33**, 1093 (A) (1929). (C L)
 F. Paschen and P. G. Kruger, Ann. der Phys. [5] **8**, 1005 (1931). (T) (C L)
 F. Paschen, Ann. der Phys. [5] **12**, 514 (1932). (I P) (T) (C L)
 B. Edlén, Nova Acta Reg. Soc. Sci. Uppsala [IV] **9**, No. 6, 51 (1934). (T) (C L)
 H. E. White, *Introduction to Atomic Spectra*, p. 179 (McGraw-Hill Book Co., Inc., New York, N. Y., 1934). (G D)
 W. F. Meggers, J. Opt. Soc. Am. **36**, 431 (1946). (Summary hfs)

Be I

Be I

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$2s^2$	$2s^2 {}^1S$	0	0. 00		$2s({}^2S)3p$	$3p {}^1P^\circ$	1	[60187]	
$2s({}^2S)2p$	$2p {}^3P^\circ$	0	21979. 43+x	0. 68	$2s({}^2S)3d$	$3d {}^3D$	1, 2, 3	62054. 8 +x	
		1	21980. 11+x	2. 35	$2s({}^2S)3d$	$3d {}^1D$	2	64428. 15	
		2	21982. 46+x		$2s({}^2S)4s$	$4s {}^3S$	1	64507. 7 +x	
$2s({}^2S)2p$	$2p {}^1P^\circ$	1	42565. 3		$2s({}^2S)4s$	$4s {}^1S$	0	65245. 4	
$2s({}^2S)3s$	$3s {}^3S$	1	52082. 07+x		$2s({}^2S)4p$	$4p {}^3P^\circ$	0, 1, 2	[65949] +x	
$2s({}^2S)3s$	$3s {}^1S$	0	54677. 2		$2s({}^2S)4p$	$4p {}^1P^\circ$	1	[67228]	
$2p^2$	$2p^2 {}^1D$	2	56432. 5		$2s({}^2S)4d$	$4d {}^3D$	1, 2, 3	67943. 6 +x	
$2s({}^2S)3p$	$3p {}^3P^\circ$	0, 1, 2	58791. 6 +x		$2s({}^2S)4d$	$4d {}^1D$	2	68781. 2	
$2p^2$	$2p^2 {}^3P$	0	59694. 61+x	1. 40	$2s({}^2S)5s$	$5s {}^3S$	1	69009. 3 +x	
		1	59696. 01+x	2. 03					
		2	59698. 04+x						

Be I—Continued

Be I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
2s (2S) 5s	5s 1S	0	69322. 3		2s (2S) 9d	9d 3D	1, 2, 3	73803. 2 +x	
2s (2S) 5p	5p 3P°	0, 1, 2	[69634. 5] +x		2s (2S) 9d	9d 1D	2	73866. 9	
2s (2S) 5d	5d 3D	1, 2, 3	70606. 7 +x		2s (2S) 10s	10s 1S	0	73930. 4	
2s (2S) 5d	5d 1D	2	71002. 3		2s (2S) 10d	10d 3D	1, 2, 3	74070. 6 +x	
2s (2S) 6s	6s 3S	1	71161. 9 +x		2s (2S) 10d	10d 1D	2	74116. 7	
2s (2S) 6s	6s 1S	0	71320. 7		2s (2S) 11s	11s 1S	0	74163. 4	
2s (2S) 6p	6p 3P°	0, 1, 2	[71482. 9] +x		2s (2S) 11d	11d 3D	1, 2, 3	74268. 6 +x	
2p ²	2p ² 1S	0	71498. 9		2s (2S) 11d	11d 1D	2	74301. 4	
2s (2S) 6d	6d 3D	1, 2, 3	72030. 6 +x		2s (2S) 12d	12d 3D	1, 2, 3	74416. 3 +x	
2s (2S) 6d	6d 1D	2	72251. 1		2s (2S) 12d	12d 1D	2	74443. 2	
2s (2S) 7s	7s 3S	1	72355. 4 +x		Be II (2S _{1/2})	<i>Limit</i>	-----	75192. 29	
2s (2S) 7s	7s 1S	0	72448. 3		2p (2P°) 3s	3s 3P°	0 1 2	85554. 96+x 85557. 01+x 85560. 93+x	2. 05 3. 92
2s (2S) 7d	7d 3D	1, 2, 3	72881. 9 +x		2p (2P°) 3p	3p 3P	0 1 2	91901. 8 +x	
2s (2S) 7d	7d 1D	2	73017. 2		2p (2P°) 3d	3d 3D°	1 2 3	[94189.51]+x 94190. 11+x 94191. 26+x	0. 60 1. 15
2s (2S) 8s	8s 3S	1	73089. 1 +x		2p (2P°) 3d	3d 3P°	0 1 2	95162. 1 +x 95163. 1 +x 95165. 0 +x	1. 0 1. 9
2s (2S) 8s	8s 1S	0	73146. 7						
2s (2S) 8d	8d 3D	1, 2, 3	73429. 6 +x						
2s (2S) 8d	8d 1D	2	73519. 7						
2s (2S) 9s	9s 1S	0	73608. 5						

May 1946.

Be I OBSERVED TERMS*

Config. 1s ² +	Observed Terms		
2s ²	2s ² 1S		
2s(2S)2p	{ 2p 3P° 2p 1P°		
2p ²	{ 2p ² 3P 2p ² 1D		
	ns (n ≥ 3)	np (n ≥ 3)	nd (n ≥ 3)
2s(2S)nx	{ 3- 8s 3S 3-11s 1S	3p 3P°	3-12d 3D 3-12d 1D
2p(2P°)nx	3s 3P°	3p 3P	3d 3P° 3d 3D°

*For predicted terms in the spectra of the Be I isoelectronic sequence, see Introduction.

Be II

(Li I sequence; 3 electrons)

$Z=4$

Ground state $1s^2 2s \ ^2S_{1/2}$

$2s \ ^2S_{1/2} \ 146881.7 \text{ cm}^{-1}$

I. P. 18.206 volts

The analysis has been taken from the paper by Paschen and Kruger.

REFERENCES

F. Paschen and P. G. Kruger, *Ann. der Phys.* [5] **8**, 1014 (1931). (I P) (T) (C L)
H. E. White, *Introduction to Atomic Spectra* p. 98 (McGraw-Hill Book Co., Inc., New York, N. Y., 1934). (G D)
W. F. Meggers, *J. Opt. Soc. Am.* **36**, 431 (1946). (Summary hfs)

Be II					Be II				
Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
2s	2s 2S	$\frac{1}{2}$	0. 0	6. 6	5f	5f $^2F^\circ$	$2\frac{1}{2}, 3\frac{1}{2}$	129321. 9	
2p	2p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	31928. 8 31935. 4		6s	6s 2S	$\frac{1}{2}$	133559. 1	
3s	3s 2S	$\frac{1}{2}$	88231. 2		6p	6p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	134485. 6	
3p	3p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	96496. 4 96498. 2	1. 8	6d	6d 2D	$1\frac{1}{2}, 2\frac{1}{2}$	134682. 0	
3d	3d 2D	$1\frac{1}{2}, 2\frac{1}{2}$	98053. 2		6f	6f $^2F^\circ$	$2\frac{1}{2}, 3\frac{1}{2}$	134688. 1	
4s	4s 2S	$\frac{1}{2}$	115465. 2		7s	7s 2S	$\frac{1}{2}$	137226. 0	
4p	4p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	 118760		7p	7p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	 137796	
4d	4d 2D	$1\frac{1}{2}, 2\frac{1}{2}$	119422. 2		7d	7d 2D	$1\frac{1}{2}, 2\frac{1}{2}$	137920. 0	
4f	4f $^2F^\circ$	$2\frac{1}{2}, 3\frac{1}{2}$	119444. 6		7f	7f $^2F^\circ$	$2\frac{1}{2}, 3\frac{1}{2}$	137923. 1	
5s	5s 2S	$\frac{1}{2}$	127336. 1		8d	8d 2D	$1\frac{1}{2}, 2\frac{1}{2}$	140020. 4	
5p	5p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	 128970. 2		Be III (1S_0)	Limit	-----	146881. 7	
5d	5d 2D	$1\frac{1}{2}, 2\frac{1}{2}$	129311. 3						

April 1946.

Be III

(He I sequence; 2 electrons)

$Z=4$

Ground state $1s^2 \ ^1S_0$

$1s^2 \ ^1S_0 \ 1241225 \pm 100 \text{ cm}^{-1}$

I. P. 153.850 ± 0.012 volts

Both Robinson and Edlén report six lines of the singlet series observed, although the earlier members have also been measured by others. The range is between 81 A and 100 A. The singlet terms have been taken from Robinson's paper.

The relative absolute values of the triplet and singlet terms have been determined by extrapolation of $3d \ ^3D$ from He I and Li II, according to Edlén, who has generously furnished his unpublished term values of the triplets. Apparently no intersystem combinations have been observed in Be III, but the existence of the observed line $1s^2 \ ^1S_0-2p \ ^3P_1^\circ$ in the related spectra from B IV to Al XII, within the errors of measurement of the predicted positions, indicates that the uncertainty α is small.

REFERENCES

B. Edlén, *Nova Acta Reg. Soc. Sci. Uppsala* [IV] **9**, No. 6, 31 (1934). (T) (C L)
H. A. Robinson, *Phys. Rev.* **51**, 14 (1937). (I P) (T) (C L)
B. Edlén, unpublished material (Sept. 1947). (T)

Be III

Be III

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
1s ²	1s ² ¹ S	0	0		1s 4p	4p ¹ P°	1	1179830	
1s 2s	2s ³ S	1	956496 + <i>x</i>		1s 5p	5p ¹ P°	1	1201894	
1s 2p	2p ³ P°	0			1s 6p	6p ¹ P°	1	1213931	
		1	983348 + <i>x</i>	15	1s 7p	7p ¹ P°	1	1221135	
		2	983363 + <i>x</i>						
1s 2p	2p ¹ P°	1	997466						
1s 3p	3p ¹ P°	1	1132323		Be IV (² S _{1/2})	<i>Limit</i>		1241225	

September 1947.

Be IV

(H sequence; 1 electron)

Z=4Ground state 1s ²S_{1/2}1s ²S_{1/2} 1756004 cm⁻¹

I. P. 217.657 volts

Edlén and Ericson first observed this spectrum. Tyrén has observed three, and Robinson six, members of the principal series.

The terms in the table have been calculated by J. E. Mack, who has used $R_{\text{Be}}^{\circ}=109730.623$ and $\Lambda=0.040$.

REFERENCES

B. Edlén and A. Ericson, *Nature* **125**, 233 (1930); **127**, 405 (1931); *Zeit. Phys.* **59**, 656 (1930). (C L)

H. A. Robinson, *Phys. Rev.* **50**, 99 (1936). (C L)

F. Tyrén, *Zeit. Phys.* **98**, 771 (1936). (C L)

J. E. Mack, unpublished material (1949). (I P) (T) (C L)

Be IV

Be IV

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
1s	1s ² S	1/2	0		5p	5p ² P°	1/2	1685766	
2p	2p ² P°	1/2	1316965] 7 93. 5	5s	5s ² S	1/2	1685767]] 0. 5 6. 0 2. 0 1. 0 0. 6
2s	2s ² S	1/2	1316972		5p, 5d	5d ² D, 5p ² P°	1 1/2	1685772	
2p	2p ² P°	1 1/2	1317058		5d, 5f	5d ² D, 5f ² F°	2 1/2	1685774	
3p	3p ² P°	1/2	1560886] 2 27. 6 9. 2	5f, 5g	5g ² G, 5f ² F°	3 1/2	1695775	
3s	3s ² S	1/2	1560888		5g	5g ² G	4 1/2	1685776	
3p, 3d	3d ² D, 3p ² P°	1 1/2	1560913		6s, etc.	6s ² S, etc.	1/2	1707229 to 234	
3d	3d ² D	2 1/2	1560923		7s, etc.	7s ² S, etc.	1/2	1720170 to 173	
4p	4p ² P°	1/2	1646254] 1 11. 7 3. 9 1. 9					
4s	4s ² S	1/2	1646255						
4p, 4d	4d ² D, 4p ² P°	1 1/2	1646266						
4d, 4f	4d ² D	2 1/2	1646270						
4f	4f ² F°	3 1/2	1646272			∞ = <i>Limit</i>		1756004	

February 1949.

BORON

B I

5 electrons

 $Z=5$ Ground state $1s^2 2s^2 2p^2 P_{1/2}^{\circ}$ $2p^2 P_{1/2}^{\circ}$ 66930 cm^{-1}

I. P. 8.296 volts

The spectrum is incompletely observed, but 34 lines have been classified in the interval between 1378 Å and 2498 Å. The terms for which there is an entry in the column of the table headed "Authors", are from Edlén, but a correction of 90 cm^{-1} has been added to the limit as quoted from Selwyn (66840 cm^{-1}). Whitelaw and Mack have recalculated the limit and derived the value $B \text{ I } 2s^2 2p^2 P_{1/2}^{\circ} - B \text{ II } 2s^2 {}^1S_0 = 66930 \text{ cm}^{-1}$, using the 2D series alone because of extra-configurational perturbations in the 2S series. Selwyn averaged the limits from both the 2S and 2D series.

The remaining terms are from an unpublished manuscript kindly furnished by Clearman, who has extended the doublet series by further observations and confirmed the correction to the limit mentioned above. Clearman has also found two quartet terms. No intersystem combinations have been observed, as indicated by x in the table. Edlén estimates that $2p^2 P_{1/2}^{\circ} - 2p^2 P_{3/2}^{\circ} = 28800 \text{ cm}^{-1}$, by analogy with the observed intersystem combinations in C II and N III. The corresponding value of $2p^2 {}^4P_{1/2}$ is entered in brackets in the table and has been added to all of Clearman's values of quartet terms.

REFERENCES

- I. S. Bowen, *Phys. Rev.* **29**, 231 (1927). (T) (C L)
 E. W. H. Selwyn, *Proc. Phys. Soc. (London)* **41**, 401 (1929). (T) (C L)
 B. Edlén, *Nova Acta Reg. Soc. Sci. Uppsala* [IV] **9**, No. 6, 74 (1934). (T)
 H. E. White, *Introduction to Atomic Spectra* p. 115 (McGraw-Hill Book Co., Inc., New York, N. Y., 1934). (G D)
 N. G. Whitelaw and J. E. Mack, *Phys. Rev.* **47**, 677 (1935). (I P) (T)
 B. Edlén, *Zeit. Phys.* **98**, 564 (1936). (C L)
 W. Opeschowski and D. A. DeVries, *Physica* **6**, No. 9, 913 (1939). (I S)
 W. F. Meggers, *J. Opt. Soc. Am.* **36**, 431 (1946). (Summary hfs)
 H. E. Clearman Jr., unpublished material (Aug. 1947). (T) (C L)

B I

B I

Authors	Config.	Desig.	J	Level	Interval	Authors	Config.	Desig.	J	Level	Interval
$2p\ ^2P_1$ $\ ^2P_2$	$2s^2(^1S)2p$	$2p\ ^2P^o$	$\frac{1}{2}$ $1\frac{1}{2}$	0 16	16	$5d\ ^2D$	$2s^2(^1S)5d$	$5d\ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	62481	
	$2s\ 2p^2$	$2p^2\ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$[28805]+x$ $28810+x$ $28816+x$	5 6		$2s\ 2p^2$	$2p^2\ ^2S$	$\frac{1}{2}$	63561	
$2p'\ ^4P_3$						$6d\ ^2D$	$2s^2(^1S)6d$	$6d\ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	63847	
$3s\ ^2S_1$	$2s^2(^1S)3s$	$3s\ ^2S$	$\frac{1}{2}$	40040			$2s^2(^1S)7s$	$7s\ ^2S$	$\frac{1}{2}$	64156	
$2p'\ ^2D$	$2s\ 2p^2$	$2p^2\ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	47857			$2s^2(^1S)7d$	$7d\ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	64664	
$3d\ ^2D$	$2s^2(^1S)3d$	$3d\ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	54765			$2s^2(^1S)8d$	$8d\ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	65195	
$4s\ ^2S_1$	$2s^2(^1S)4s$	$4s\ ^2S$	$\frac{1}{2}$	55009			$2s^2(^1S)9s$	$9s\ ^2S$	$\frac{1}{2}$	65553	
$4d\ ^2D$	$2s^2(^1S)4d$	$4d\ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	59989			B II (1S_0)	Limit	-----	66930	
$5s\ ^2S_1$	$2s^2(^1S)5s$	$5s\ ^2S$	$\frac{1}{2}$	60146			$2s\ 2p^2$	$2p^2\ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	72535 72547	12
	$2s^2(^1S)6s$	$6s\ ^2S$	$\frac{1}{2}$	62098			$2p^3$	$2p^3\ ^4S^o$	$1\frac{1}{2}$	$97037+x$	

August 1947.

B I OBSERVED TERMS*

Config. $1s^2+$	Observed Terms	
$2s^2(^1S)2p$	$2p\ ^2P^o$	
$2s\ 2p^2$	$\left\{ \begin{array}{lll} 2p^2\ ^2S & 2p^2\ ^4P & 2p^2\ ^2D \\ & 2p^2\ ^2P & \end{array} \right.$	
$2p^3$	$2p^3\ ^4S^o$	
	$ns\ (n \geq 3)$	$nd\ (n \geq 3)$
$2s^2(^1S)nx$	$3-7s, 9s\ ^2S$	$3-8d\ ^2D$

*For predicted terms in the spectra of the B I isoelectronic sequence, see Introduction.

B II

(Be I sequence; 4 electrons)

 $Z=5$ Ground state $1s^2\ 2s^2\ ^1S_0$ $2s^2\ ^1S_0\ 202895\ \text{cm}^{-1}$

I. P. 25.149 volts

The terms are from Edlén, who remarks that the observed series, especially in the singlet system, are too short for the precise determination of the limits. By analogy with Be I, C III, and N IV, he interpolates the value of $2s^2\ ^1S_0 - 2p\ ^3P_1^o$ as $37340\ \text{cm}^{-1}$, which places the limit $2s^2\ ^1S_0$ at $202895.0\ \text{cm}^{-1}$. The absolute values of the singlet terms as published in Edlén's Monograph have therefore been increased by $249\ \text{cm}^{-1}$. The relative uncertainty x is probably less than this. No intersystem combinations have been observed.

An extrapolated value of $3s\ ^1S_0$ is given in brackets.

B II—Continued

REFERENCES

B. Edlén, Nova Acta Reg. Soc. Sci. Uppsala [IV] 9, No. 6, 51 (1934). (T) (C L)

B. Edlén, Zeit. Phys. 98, 561 (1936). (I P) (C L)

B II

B II

Edlén	Config.	Desig.	<i>J</i>	Level	Interval	Edlén	Config.	Desig.	<i>J</i>	Level	Interval
2s ¹ S ₀	2s ²	2s ² ¹ S	0	0.0		4p ³ P	2s(² S)4p	4p ³ P°	0, 1, 2	171544. 7+x	
2p ³ P ₀	2s(² S)2p	2p ³ P°	0	37333. 6+x	6. 4	4d ³ D	2s(² S)4d	4d ³ D	1, 2, 3	174072. 6+x	
³ P ₁			1	37340. 0+x	16. 4	4f ³ F	2s(² S)4f	4f ³ F°	2, 3, 4	174902. 5+x	
³ P ₂			2	37356. 4+x		4f ¹ F ₃	2s(² S)4f	4f ¹ F°	3	174921. 5	
2p ¹ P ₁	2s(² S)2p	2p ¹ P°	1	73396. 7		4d ¹ D ₂	2s(² S)4d	4d ¹ D	2	175546. 0	
2p' ³ P ₀	2p ²	2p ² ³ P	0	98910. 3+x	8. 4	5s ³ S ₁	2s(² S)5s	5s ³ S	1	180896. 5+x	
³ P ₁			1	98918. 7+x	14. 0	3s' ³ P ₀	2p(² P°)3s	3s ³ P°	0	181645. 2+x	9. 8
³ P ₂			2	98932. 7+x		³ P ₁			1	181655. 0+x	20. 9
						³ P ₂			2	181675. 9+x	
2p' ¹ D ₂	2p ²	2p ² ¹ D	2	102362. 1		5d ³ D	2s(² S)5d	5d ³ D	1, 2, 3	184633. 1+x	
2p' ¹ S ₀	2p ²	2p ² ¹ S	0	127662. 0		5f ³ F	2s(² S)5f	5f ³ F°	2, 3, 4	184908. 2+x	
3s ³ S ₁	2s(² S)3s	3s ³ S	1	129772. 9+x		3p' ¹ P ₁	2p(² P°)3p	3p ¹ P	1	189126. 6	
3s ¹ S ₀	2s(² S)3s	3s ¹ S	0	[135946]		3d' ³ F ₂₃	2p(² P°)3d	3d ³ F°	2, 3	194748? +x	12
3p ³ P ₀₁	2s(² S)3p	3p ³ P°	0, 1	143989. 7+x	3. 7	³ F ₄			4	194760? +x	
³ P ₂			2	143993. 4+x		3d' ¹ D ₂	2p(² P°)3d	3d ¹ D°	2	197721. 0	
3p ¹ P ₁	2s(² S)3p	3p ¹ P°	1	144102. 0		3d' ³ D	2p(² P°)3d	3d ³ D°	1, 2, 3	200484. 6+x	
3d ³ D	2s(² S)3d	3d ³ D	1, 2, 3	150649. 0+x							
3d ¹ D ₂	2s(² S)3d	3d ¹ D	2	154686. 9							
4s ³ S ₁	2s(² S)4s	4s ³ S	1	166344. 4+x							
4s ¹ S ₀	2s(² S)4s	4s ¹ S	0	167934. 2							
							B III (² S _{1/2})	Limit	-----	202895	

May 1946.

B II OBSERVED TERMS*

Config. 1s ² +	Observed Terms			
2s ²	2s ² ¹ S			
2s(² S)2p	{ 2p ³ P° 2p ¹ P°			
2p ²	{ 2p ² ³ P 2p ² ¹ D 2p ² ¹ S			
	ns (n ≥ 3)	np (n ≥ 3)	nd (n ≥ 3)	nf (n ≥ 4)
2s(² S)nx	{ 3-5s ³ S 4s ¹ S	3, 4p ³ P° 3p ¹ P°	3-5d ³ D 3, 4d ¹ D	4, 5f ³ F° 4f ¹ F°
2p(² P°)nx	{ 3s ³ P°	3p ¹ P	3d ³ D° 3d ³ F° 3d ¹ D°	

*For predicted terms in the spectra of the Be I isoelectronic sequence, see Introduction.

B III

(Li I sequence; 3 electrons)

 $Z=5$ Ground state $1s^2 2s^2 S_{1/2}$ $2s^2 S_{1/2}$ 305931.1 cm^{-1}

I. P. 37.920 volts

The terms are from Edlén. The absolute values are based on the assumption that n^* for $5g^2 G$ equals that of the corresponding term in C IV, where $5g^2 G - 6h^2 H^\circ$ has been observed. The precision of this term in B III is estimated to be within $\pm 1 \text{ cm}^{-1}$. The series are well represented by a Ritz formula.

Edlén gives four extrapolated term intervals, which are entered in brackets in the table.

REFERENCES

A. Ericson and B. Edlén, Zeit. Phys. **59**, 676 (1930). (T) (C L)B. Edlén, Nova Acta Reg. Soc. Sci. Uppsala [IV] **9**, No. 6, 37 (1934). (I P) (T) (C L) (G D)**B III****B III**

Edlén	Config.	Desig.	J	Level	Interval	Edlén	Config.	Desig.	J	Level	Interval
$2s^2 S_1$	$2s$	$2s^2 S$	$\frac{1}{2}$	0.0		$5p^2 P_2$	$5p$	$5p^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	265719.7	[2. 2]
$2p^2 P_1$ $2p^2 P_2$	$2p$	$2p^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	48358.5 48392.6	34.1	$5d^2 D_3$	$5d$	$5d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	266389.5	
$3s^2 S_1$	$3s$	$3s^2 S$	$\frac{1}{2}$	180201.8		$5f^2 F$	$5f$	$5f^2 F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	266416.5	
$3p^2 P_1$ $3p^2 P_2$	$3p$	$3p^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	192949.2 192959.4	10.2	$5g^2 G$	$5g$	$5g^2 G$	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ 4\frac{1}{2} \end{smallmatrix} \right\}$	266427.2	
$3d^2 D_3$	$3d$	$3d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	196071.2	[3. 4]	$6d^2 D_3$	$6d$	$6d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	278473.7	
$4s^2 S_1$	$4s$	$4s^2 S$	$\frac{1}{2}$	237695.5		$6f^2 F$	$6f$	$6f^2 F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	278491.7	
$4p^2 P_2$	$4p$	$4p^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	242832.4	[4. 3]	$6g^2 G$	$6g$	$6g^2 G$	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ 4\frac{1}{2} \end{smallmatrix} \right\}$	278497.5	
$4d^2 D_3$	$4d$	$4d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	244138.9	[1. 4]						
$4f^2 F$	$4f$	$4f^2 F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	244199.2							
$5s^2 S_1$	$5s$	$5s^2 S$	$\frac{1}{2}$	263156.2			B IV ($1S_0$)	Limit	-----	305931.1	

April 1946.

B IV

(He I sequence; 2 electrons)

 $Z=5$ Ground state $1s^2 1S_0$ $1s^2 1S_0$ 2091960 \pm 200 cm^{-1} I. P. 259.298 \pm 0.025 volts

The singlet terms are from Tyrén and the observed singlet combinations are in the range from 48 to 60 Å. The unit adopted by Tyrén, 10^3 cm^{-1} , has here been changed to cm^{-1} .

Relative absolute values of the triplet terms were derived by the extrapolation of $3d^3 D$ from He I and Li II, according to unpublished material generously furnished by Dr. Edlén. These calculations have confirmed the classification by Tyrén of a line at 61 Å as the inter-system combination $1s^2 1S_0 - 2p^3 P_1^2$. The triplet terms have been taken from Edlén's 1947 manuscript.

B IV—Continued

REFERENCES

- B. Edlén, Nova Acta Reg. Soc. Sci. Uppsala [IV] 9, No. 6, 31 (1934). (T) (C L)
 H. A. Robinson, Phys. Rev. 51, 14 (1937). (I P) (T) (C L)
 F. Tyrén, Nova Acta Reg. Soc. Sci. Uppsala [IV] 12, No. 1, 24 (1940). (I P) (T) (C L)
 B. Edlén, unpublished material (Sept. 1947). (T)

B IV

Config.	Desig.	<i>J</i>	Level	Interval
1s ²	1s ² 1S	0	0	
1s 2s	2s 3S	1	1601505	
1s 2p	2p 3P°	0	1636898	-16 52
		1	1636882	
		2	1636934	
1s 2p	2p 1P°	1	1658020	
1s 3p	3p 1P°	1	1898180	

B IV

Config.	Desig.	<i>J</i>	Level	Interval
1s 4p	4p 1P°	1	1982750	
1s 5p	5p 1P°	1	2022000	
1s 6p	6p 1P°	1	2043360	

B v (2S _{1/2})	Limit	-----	2091960	

September 1947.

B v

(H sequence; 1 electron)

Z=5Ground state 1s 2S_{1/2}1s 2S_{1/2} 2744063 cm⁻¹

I. P. 340.127 volts

Edlén first observed the Lyman line. Tyrén has observed three members of the series. The listed term values have been calculated by J. E. Mack for B¹¹v, "using $R_B^{11}=109731.835$ and $\Lambda=0.040$; a change of 1 percent in Λ would change the series limit by 1.46 cm⁻¹. For B¹⁰ the series limit is less by 13.6 cm⁻¹ than for B¹¹."

REFERENCES

- B. Edlén, Nova Acta Reg. Soc. Sci. Uppsala [IV] 9, No. 6, 28, 152 (1934). (T) (C L)
 F. Tyrén, Zeit. Phys. 98, 771 (1936). (C L)
 J. E. Mack, unpublished material (1949). (I P) (T) (C L)

B v

Config.	Desig.	<i>J</i>	Level	Interval
1s	1s 2S	½	0	
2p	2p 2P°	½	2057954	18 228. 3
2s	2s 2S	½	2057972	
2p	2p 2P°	1½	2058182	
3p	3p 2P°	½	2439151	5 67. 6 22. 6
3s	3s 2S	½	2439156	
3p, 3d	3d 2D, 3p 2P°	1½	2439218	
3d	3d 2D	2½	2439241	

B v

Config.	Desig.	<i>J</i>	Level	Interval
4p	4p 2P°	½	2572561	2 28. 5 9. 5 4. 8
4s	4s 2S	½	2572563	
4p, 4d	4p 2P°	1½	2572589	
4d, 4f	4d 2D, 4f 2F°	2½	2572599	
4f	4f 2F°	3½	2572603	
5s	5s 2S, etc.	½, etc,	2634306 to 330	

$\infty = \text{Limit}$				2744063

February 1949.

SEE REVISION IN NSRDS-NBS 3, Section 3, October 1970.

CARBON

C I

6 electrons

$Z=6$

Ground state $1s^2 2s^2 2p^2 {}^3P_0$

$2p^2 {}^3P_0$ 90878.3 cm^{-1}

I. P. 11.264 volts

The term assignments are taken from Edlén, who has revised and extended the earlier work on the analysis of this spectrum. Two extrapolated term values, derived from the irregular doublet law, are entered in brackets in the table.

The singlet and triplet terms are well connected by intersystem combinations. Only two quintet terms are known. They are connected with the rest by intersystem combinations based on the measures of the resonance lines by Shenstone.

One term, $5p {}^1S$, has been revised as suggested in the 1939 reference listed below.

Selected term values of C I have been improved from a study of the lines that have been clearly identified in the Infrared Solar Spectrum. Such precision cannot be expected from terms based on lines in the ultraviolet. As a starting point the value of $3s {}^3P_1^o = 60353.00 \text{ cm}^{-1}$ was adopted as correct, to agree with Shenstone's recent measures. Excellent agreement was found between the laboratory measures of Kiess (8335 Å to 11330 Å) and solar wave-numbers of lines identified as C I in the solar spectrum. Further to the red solar wavelengths surpass laboratory values in accuracy and give consistent internal separations within the multiplets.

In the course of this work all term values have been recalculated. Consequently, most of the listed values differ slightly from those published by Edlén. No changes have been made in his analysis, but the level $3d {}^3P_0^o$, calculated from solar wave-numbers, has been added to his list.

REFERENCES

- A. Fowler and E. W. H. Selwyn, Proc. Roy. Soc. (London) [A] **118**, 34 (1928). (T) (C L)
- S. B. Ingram, Phys. Rev. **34**, 421 (1929). (T) (C L)
- F. Paschen and G. Kruger, Ann. der Phys. [5] **7**, 1 (1930). (T) (C L)
- B. Edlén, Nova Acta Reg. Soc. Sci. Uppsala [IV] **9**, No. 6, 104 (1934). (I P) (T) (C L)
- H. E. White, *Introduction to Atomic Spectra* p. 266 (McGraw-Hill Book Co., Inc., New York, N. Y., 1934). (G D)
- C. C. Kiess, J. Research Nat. Bur. Std. **20**, 33, RP1062 (1938). (C L)
- W. F. Meggers, J. Opt. Soc. Am. **36**, 431 (1936). (Summary hfs)
- Y. Ishida, T. Tamura, and M. Fukushima, Sci. Papers Inst. Phys. Chem. Research (Tokyo) **36**, No. 936, 417 (1939). (T) (C L)
- H. D. Babcock and C. E. Moore, Carnegie Inst. Wash. Publ. 579 (1947). (Solar data)
- B. Edlén, Nature **159**, No. 4030, 129 (1947). (C L)
- A. G. Shenstone, Phys. Rev. **72**, 411 (1947). (T) (C L)

C I

Edlén	Config.	Desig.	<i>J</i>	Level	Interval	Edlén	Config.	Desig.	<i>J</i>	Level	Interval
$2p\ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s^2\ 2p^2$	$2p^2\ ^3P$	0 1 2	0. 0 16. 4 43. 5	16. 4 27. 1	$4d\ ^1D_2$	$2s^2\ 2p(^3P^\circ)4d$	$4d\ ^1D^\circ$	2	83500	
$2p\ ^1D_2$	$2s^2\ 2p^2$	$2p^2\ ^1D$	2	10193. 70		$4d\ ^3F_3$	$2s^2\ 2p(^3P^\circ)4d$	$4d\ ^3F^\circ$	2 3 4	83761	
$2p^1\ S_0$	$2s^2\ 2p^2$	$2p^2\ ^1S$	0	21648. 4		$4d\ ^3D_1$ $\ ^3D_2$ $\ ^3D_3$	$2s^2\ 2p(^3P^\circ)4d$	$4d\ ^3D^\circ$	1 2 3	83830 83837 83847	7 10
$3s\ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s^2\ 2p(^3P^\circ)3s$	$3s\ ^3P^\circ$	0 1 2	60333. 80 60353. 00 60393. 52	19. 20 40. 52	$5s\ ^1P_1$	$2s^2\ 2p(^3P^\circ)5s$	$5s\ ^1P^\circ$	1	83882. 5	
$3s\ ^1P_1$	$2s^2\ 2p(^3P^\circ)3s$	$3s\ ^1P^\circ$	1	61982. 20		$4d\ ^1F_3$	$2s^2\ 2p(^3P^\circ)4d$	$4d\ ^1F^\circ$	3	83949	
$2p'\ ^3D_3$ $\ ^3D_2$ $\ ^3D_1$	$2s\ 2p^3$	$2p^3\ ^3D^\circ$	3 2 1	64088. 56 64093. 19 64092. 01	-4. 63 1. 18	$4d\ ^3P_2$ $\ ^3P_1$ $\ ^3P_0$	$2s^2\ 2p(^3P^\circ)4d$	$4d\ ^3P^\circ$	2 1 0	84102. 6 84112	-9
$3p\ ^1P_1$	$2s^2\ 2p(^3P^\circ)3p$	$3p\ ^1P$	1	68858		$5p\ ^1P_1$	$2s^2\ 2p(^3P^\circ)5p$	$5p\ ^1P$	1	84852. 13	
$3p\ ^3D_1$ $\ ^3D_2$ $\ ^3D_3$	$2s^2\ 2p(^3P^\circ)3p$	$3p\ ^3D$	1 2 3	69689. 79 69710. 99 69744. 40	21. 20 33. 41	$5p\ ^3D_2$ $\ ^3D_3$	$2s^2\ 2p(^3P^\circ)5p$	$5p\ ^3D$	1 2 3	84952 84986. 2	34
$3p\ ^3S_1$	$2s^2\ 2p(^3P^\circ)3p$	$3p\ ^3S$	1	70744. 26		$5p\ ^1D_2$	$2s^2\ 2p(^3P^\circ)5p$	$5p\ ^1D$	2	85400. 38	
$3p\ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s^2\ 2p(^3P^\circ)3p$	$3p\ ^3P$	0 1 2	71352. 81 71365. 23 71385. 70	12. 42 20. 47	$5p\ ^1S_0$	$2s^2\ 2p(^3P^\circ)5p$	$5p\ ^1S$	0	85625. 84	
$3p\ ^1D_2$	$2s^2\ 2p(^3P^\circ)3p$	$3p\ ^1D$	2	72611. 06		$5d\ ^1D_2$	$2s^2\ 2p(^3P^\circ)5d$	$5d\ ^1D^\circ$	2	86187	
$3p\ ^1S_0$	$2s^2\ 2p(^3P^\circ)3p$	$3p\ ^1S$	0	73976. 23		$5d\ ^3F_2$ $\ ^3F_3$	$2s^2\ 2p(^3P^\circ)5d$	$5d\ ^3F^\circ$	2 3 4	86319 86326. 9	8
$2p'\ ^3P$	$2s\ 2p^3$	$2p^3\ ^3P^\circ$	2, 1, 0	75256. 3		$5d\ ^3D_2$ $\ ^3D_3$	$2s^2\ 2p(^3P^\circ)5d$	$5d\ ^3D^\circ$	1 2 3	86371. 3 86396	25
$3d\ ^1D_2$	$2s^2\ 2p(^3P^\circ)3d$	$3d\ ^1D^\circ$	2	77680. 5		$6s\ ^1P_1$	$2s^2\ 2p(^3P^\circ)6s$	$6s\ ^1P^\circ$	1	86413. 96	
$4s\ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s^2\ 2p(^3P^\circ)4s$	$4s\ ^3P^\circ$	0 1 2	78105. 23 78117. 06 78148. 36	11. 83 31. 30	$5d\ ^1F_3$	$2s^2\ 2p(^3P^\circ)5d$	$5d\ ^1F^\circ$	3	86450	
$3d\ ^3F_2$ $\ ^3F_3$ $\ ^3F_4$	$2s^2\ 2p(^3P^\circ)3d$	$3d\ ^3F^\circ$	2 3 4	78199. 34 78215. 82 78250. 22	16. 48 34. 40	$5d\ ^1P_1$	$2s^2\ 2p(^3P^\circ)5d$	$5d\ ^1P^\circ$	1	86491	
$3d\ ^3D_1$ $\ ^3D_2$ $\ ^3D_3$	$2s^2\ 2p(^3P^\circ)3d$	$3d\ ^3D^\circ$	1 2 3	78300. 8 78307 78316	6 9	$5d\ ^3P_2$ $\ ^3P_1$	$2s^2\ 2p(^3P^\circ)5d$	$5d\ ^3P^\circ$	2 1 0	86504 86517	-13
$4s\ ^1P_1$	$2s^2\ 2p(^3P^\circ)4s$	$4s\ ^1P^\circ$	1	78338		$6d\ ^1D_2$	$2s^2\ 2p(^3P^\circ)6d$	$6d\ ^1D^\circ$	2	87632	
$3d\ ^1F_3$	$2s^2\ 2p(^3P^\circ)3d$	$3d\ ^1F^\circ$	3	78531		$6d\ ^3F_2$ $\ ^3F_3$	$2s^2\ 2p(^3P^\circ)6d$	$6d\ ^3F^\circ$	2 3 4	87706 87713	7
$3d\ ^1P_1$	$2s^2\ 2p(^3P^\circ)3d$	$3d\ ^1P^\circ$	1	78727. 91		$6d\ ^3D_2$ $\ ^3D_3$	$2s^2\ 2p(^3P^\circ)6d$	$6d\ ^3D^\circ$	1 2 3	87752 87773	21
$3d\ ^3P_2$ $\ ^3P_1$	$2s^2\ 2p(^3P^\circ)3d$	$3d\ ^3P^\circ$	2 1 0	79311. 10 79319. 06 79323. 32	-7. 96 -4. 26	$7s\ ^1P_1$	$2s^2\ 2p(^3P^\circ)7s$	$7s\ ^1P^\circ$	1	87795. 3	
$4p\ ^3D_1$ $\ ^3D_2$ $\ ^3D_3$	$2s^2\ 2p(^3P^\circ)4p$	$4p\ ^3D$	1 2 3	80173. 29 80192. 49 80222. 74	19. 20 30. 25	$6d\ ^1F_3$	$2s^2\ 2p(^3P^\circ)6d$	$6d\ ^1F^\circ$	3	87807	
$4p\ ^1P_1$	$2s^2\ 2p(^3P^\circ)4p$	$4p\ ^1P$	1	80563. 57		$6d\ ^3P_2$ $\ ^3P_1$	$2s^2\ 2p(^3P^\circ)6d$	$6d\ ^3P^\circ$	2 1 0	87830 87839	-9
$4p\ ^3S_1$	$2s^2\ 2p(^3P^\circ)4p$	$4p\ ^3S$	1	81105. 70		$6d\ ^1P_1$	$2s^2\ 2p(^3P^\circ)6d$	$6d\ ^1P^\circ$	1	87831. 3	
$4p\ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s^2\ 2p(^3P^\circ)4p$	$4p\ ^3P$	0 1 2	81311. 52 81326. 33 81344. 48	14. 81 18. 15	$7d\ ^3F_2$ $\ ^3F_3$ $\ ^3F_4$	$2s^2\ 2p(^3P^\circ)7d$	$7d\ ^3F^\circ$	2 3 4	88541. 8 88547	5
$4p\ ^1D_2$	$2s^2\ 2p(^3P^\circ)4p$	$4p\ ^1D$	2	81770. 36			$2s^2\ 2p(^3P^\circ)7d$	$7d\ ^3D^\circ$	1 2 3	88607	
$4p\ ^1S_0$	$2s^2\ 2p(^3P^\circ)4p$	$4p\ ^1S$	0	82252. 31		$7d\ ^3D_3$					

C I—Continued

C I—Continued

Edlén	Config.	Desig.	<i>J</i>	Level	Interval	Edlén	Config.	Desig.	<i>J</i>	Level	Interval
7d ¹ F ₃	2s ² 2p(² P°)7d	7d ¹ F°	3	88624	-1		2s ² 2p(² P°)9d	9d ³ D°	1 2 3	89514	
7d ¹ P ₁	2s ² 2p(² P°)7d	7d ¹ P°	1	88632. 44		9d ³ D ₃					
7d ³ P ₂	2s ² 2p(² P°)7d	7d ³ P°	2 1 0	88639		9d ¹ F ₃	2s ² 2p(² P°)9d	9d ¹ F°	3		
8d ³ F ₃ ³ F ₂	2s ² 2p(² P°)8d	8d ³ F°	4 3 2	89081 89082		10d ³ D ₃	2s ² 2p(² P°)10d	10d ³ D°	1 2 3		
							2s ² 2p(² P°)11d	11d ³ D°	1 2 3		
8d ³ D ₃	2s ² 2p(² P°)8d	8d ³ D°	1 2 3	89146		11d ³ D ₃	C II (² P _{3/2})	Limit	-----	90878. 3	20. 7 24. 8
8d ¹ F ₃	2s ² 2p(² P°)8d	8d ¹ F°	3	89155		2p' ¹ D ₂	2s 2p ³	2p ³ ¹ D°	2	[97878]	
8d ³ P ₂	2s ² 2p(² P°)8d	8d ³ P°	2 1 0	89158			2s 2p ² (⁴ P)3s	3s ⁵ P	1 2 3	103541. 8 103562. 5 103587. 3	
9d ³ F ₂	2s ² 2p(² P°)9d	9d ³ F°	4 3 2	89450		2p' ³ S ₁	2s 2p ³	2p ³ ³ S°	1	105800. 5	
						2p' ¹ P ₁	2s 2p ³	2p ³ ¹ P°	1	[119878]	

September 1947.

C I OBSERVED TERMS*

Config. 1s ² +	Observed Terms					
2s ² 2p ²	{ 2p ² ¹ S 2p ² ³ P 2p ² ¹ D					
2s 2p ³	{ 2p ³ ⁵ S° 2p ³ ³ P° 2p ³ ³ D°					
	<i>ns</i> (<i>n</i> ≥ 3)		<i>np</i> (<i>n</i> ≥ 3)		<i>nd</i> (<i>n</i> ≥ 3)	
2s ² 2p(² P°) <i>nx</i>	{ 3, 4s ³ P° 3-7s ¹ P°		{ 3, 4p ³ S 3, 4p ³ P 3-5p ³ D 3-5p ¹ S 3-5p ¹ P 3-5p ¹ D		{ 3-8d ³ P° 3-11d ³ D° 3-9d ³ F° 3-7d ¹ P° 3-6d ¹ D° 3-9d ¹ F°	
2s 2p ² (⁴ P) <i>nx</i>	3s ⁵ P					

*For predicted terms in the spectra of the C I isoelectronic sequence, see Introduction.

C II

(B I sequence; 5 electrons)

 $Z=6$ Ground state $1s^2 2s^2 2p^2 P_{3/2}^0$ $2p^2 P_{3/2}^0$ 196659. 0 cm^{-1}

I. P. 24.376 volts

The term values for the doublets are taken from Edlén's Monograph. He has since rejected his $5p' {}^2D$ term. Intersystem combinations have been observed by Edlén (1936) and the resulting correction to the quartet terms as published in his Monograph, $+19.3 \text{ cm}^{-1}$, has been applied.

REFERENCES

- A. Fowler and E. W. H. Selwyn, Proc. Roy. Soc. (London) [A] **120**, 312 (1928). (T) (C L)
 B. Edlén, Nova Acta Reg. Soc. Sci. Uppsala [IV] **9**, No. 6, 74 (1934). (I P) (T) (C L) (G D)
 B. Edlén, Zeit. Phys. **98**, 561 (1936). (C L)
 B. Edlén, private communication (Dec. 1947). (T)

C II

C II

Edlén	Config.	Desig.	J	Level	Interval	Edlén	Config.	Desig.	J	Level	Interval
$2p^2 P_1$ $2p^2 P_2$	$2s^2 ({}^1S) 2p$	$2p^2 P^0$	$\frac{1}{2}$ $1\frac{1}{2}$	0. 0 64. 0	64. 0	$5s^2 S_1$	$2s^2 ({}^1S) 5s$	$5s^2 S$	$\frac{1}{2}$	173348. 18	
$2p' {}^4P_1$ $2p' {}^4P_2$ $2p' {}^4P_3$	$2s^2 2p^2$	$2p^2 {}^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	43000. 2 43021. 8 43050. 7	21. 6 28. 9	$5p^2 P_1$ $5p^2 P_2$	$2s^2 ({}^1S) 5p$	$5p^2 P^0$	$\frac{1}{2}$ $1\frac{1}{2}$	175287. 9 175295. 2	7. 3
$2p' {}^2D_3$ $2p' {}^2D_2$	$2s^2 2p^2$	$2p^2 {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	74930. 9 74933. 2	-2. 3	$3s' {}^2P_1$ $3s' {}^2P_2$	$2s^2 2p ({}^3P^0) 3s$	$3s^2 P^0$	$\frac{1}{2}$ $1\frac{1}{2}$	178194. 1 178220. 8	26. 7
$2p' {}^2S_1$	$2s^2 2p^2$	$2p^2 {}^2S$	$\frac{1}{2}$	96494. 1		$5d^2 D_3$	$2s^2 ({}^1S) 5d$	$5d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	178494. 8	
$2p' {}^2P_1$ $2p' {}^2P_2$	$2s^2 2p^2$	$2p^2 {}^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	110625. 1 110666. 3	41. 2	$5f^2 F$	$2s^2 ({}^1S) 5f$	$5f^2 F^0$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	178956. 46	
$3s^2 S_1$	$2s^2 ({}^1S) 3s$	$3s^2 S$	$\frac{1}{2}$	116537. 88		$6s^2 S_1$	$2s^2 ({}^1S) 6s$	$6s^2 S$	$\frac{1}{2}$	181258	
$3p^2 P_1$ $3p^2 P_2$	$2s^2 ({}^1S) 3p$	$3p^2 P^0$	$\frac{1}{2}$ $1\frac{1}{2}$	131724. 68 131735. 81	11. 13	$3p' {}^4D_1$ $3p' {}^4D_2$ $3p' {}^4D_3$ $3p' {}^4D_4$	$2s^2 2p ({}^3P^0) 3p$	$3p^2 D$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	181694. 50 181709. 20 181734. 21 181770. 48	14. 70 25. 01 36. 27
$2p'' {}^4S_2$	$2p^2$	$2p^2 {}^4S^0$	$1\frac{1}{2}$	142024. 4		$3p' {}^2P_1$ $3p' {}^2P_2$	$2s^2 2p ({}^3P^0) 3p$	$3p^2 P$	$\frac{1}{2}$ $1\frac{1}{2}$	182025. 0 182044. 5	19. 5
$3d^2 D_2$ $3d^2 D_3$	$2s^2 ({}^1S) 3d$	$3d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	145549. 99 145551. 44	1. 45	$6d^2 D_2$	$2s^2 ({}^1S) 6d$	$6d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	184064. 9	
$2p'' {}^2D_3$ $2p'' {}^2D_2$	$2p^2$	$2p^2 {}^2D^0$	$2\frac{1}{2}$ $1\frac{1}{2}$	150462. 8 150467. 9	-5. 1	$6f^2 F$	$2s^2 ({}^1S) 6f$	$6f^2 F^0$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	184376. 20	
$4s^2 S_1$	$2s^2 ({}^1S) 4s$	$4s^2 S$	$\frac{1}{2}$	157234. 43		$3p' {}^4S_2$	$2s^2 2p ({}^3P^0) 3p$	$3p^2 S$	$1\frac{1}{2}$	184688. 69	
$4p^2 P_1$ $4p^2 P_2$	$2s^2 ({}^1S) 4p$	$4p^2 P^0$	$\frac{1}{2}$ $1\frac{1}{2}$	162518. 70 162524. 62	5. 92	$3p' {}^4P_1$ $3p' {}^4P_2$ $3p' {}^4P_3$	$2s^2 2p ({}^3P^0) 3p$	$3p^2 P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	186425. 02 186441. 32 186463. 75	16. 30 22. 43
$3s' {}^4P_1$ $3s' {}^4P_2$ $3s' {}^4P_3$	$2s^2 2p ({}^3P^0) 3s$	$3s^2 P^0$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	166964. 70 166988. 46 167033. 43	23. 76 44. 97	$3p' {}^2D_2$ $3p' {}^2D_3$	$2s^2 2p ({}^3P^0) 3p$	$3p^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	188579. 3 188612. 7	33. 4
$4d^2 D_2$ $4d^2 D_3$	$2s^2 ({}^1S) 4d$	$4d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	168123. 92 168124. 33	0. 41	$3p' {}^2S_1$	$2s^2 2p ({}^3P^0) 3p$	$3p^2 S$	$\frac{1}{2}$	194571. 9	
$2p'' {}^2P_1$ $2p'' {}^2P_2$	$2p^2$	$2p^2 {}^2P^0$	$\frac{1}{2}$ $1\frac{1}{2}$	168731. 6 168750. 2	18. 6	$3d^2 {}^4F_2$ $3d^2 {}^4F_3$ $3d^2 {}^4F_4$ $3d^2 {}^4F_5$	$2s^2 2p ({}^3P^0) 3d$	$3d^2 F^0$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	195750. 8 195765. 1 195784. 7 195812. 3	14. 3 19. 6 27. 6
$4f^2 F$	$2s^2 ({}^1S) 4f$	$4f^2 F^0$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	168979. 05							

C II—Continued

C II—Continued

Edlén	Config.	Desig.	<i>J</i>	Level	Interval	Edlén	Config.	Desig.	<i>J</i>	Level	Interval
3d' ⁴ D ₁ ⁴ D ₂ ⁴ D ₃ ⁴ D ₄	2s 2p(³ P°)3d	3d ⁴ D°	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{2}{2}$ $\frac{3}{2}$	196556. 2 196561. 8 196570. 5 196580. 8	5. 6 8. 7 10. 3	4d' ² F ₄	2s 2p(³ P°)4d	4d ² F°	$\frac{2}{2}$ $\frac{3}{2}$	221502	
	C III (¹ S ₀)	Limit	-----	196659. 0		4f' ⁴ G ₃ ⁴ G ₄ ⁴ G ₅ ⁴ G ₆	2s 2p(³ P°)4f	4f ⁴ G	$\frac{2}{2}$ $\frac{3}{2}$ $\frac{4}{2}$ $\frac{5}{2}$	221543. 0 221553. 2 221574. 5 221603. 6	10. 2 21. 3 29. 1
3d' ² D ₂ ³ D ₃	2s 2p(³ P°)3d	3d ² D°	$\frac{1}{2}$ $\frac{2}{2}$	198426. 4 198437. 2	10. 8	4f' ² G ₄ ² G ₅	2s 2p(³ P°)4f	4f ² G	$\frac{3}{2}$ $\frac{4}{2}$	221585 221628	43
3d' ⁴ P ₃ ⁴ P ₂ ⁴ P ₁	2s 2p(³ P°)3d	3d ⁴ P°	$\frac{2}{2}$ $\frac{1}{2}$ $\frac{1}{2}$	198842. 0 198863. 5 198877. 7	-21. 5 -14. 2	4f' ⁴ D ₄ ⁴ D ₃ ⁴ D ₂	2s 2p(³ P°)4f	4f ⁴ D	$\frac{3}{2}$ $\frac{2}{2}$ $\frac{1}{2}$	221696. 5 221727. 4 221746. 3	-30. 9 -18. 9
3d' ² F ₃ ² F ₄	2s 2p(³ P°)3d	3d ² F°	$\frac{2}{2}$ $\frac{3}{2}$	199941. 4 199984. 2	42. 8	4f' ² D ₃ ² D ₂	2s 2p(³ P°)4f	4f ² D	$\frac{2}{2}$ $\frac{1}{2}$	221707. 9 221752. 9	-45. 0
3d' ² P ₂ ² P ₁	2s 2p(³ P°)3d	3d ² P°	$\frac{1}{2}$ $\frac{1}{2}$	202180. 3 202204. 4	-24. 1	4d' ² P ₂ ² P ₁	2s 2p(³ P°)4d	4d ² P°	$\frac{1}{2}$ $\frac{1}{2}$	222259. 1 222286. 0	-26. 9
4s' ⁴ P ₁ ⁴ P ₂ ⁴ P ₃	2s 2p(³ P°)4s	4s ⁴ P°	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{2}{2}$	209550. 26 209574. 28 209620. 36	24. 02 46. 08		2s 2p(³ P°)5s	5s ⁴ P°	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{2}{2}$		
4p' ² P ₁ ² P ₂	2s 2p(³ P°)4p	4p ² P	$\frac{1}{2}$ $\frac{1}{2}$	214406. 6 214429. 7	23. 1	5s' ⁴ P ₃				225813	
4p' ⁴ D ₁ ⁴ D ₂ ⁴ D ₃ ⁴ D ₄	2s 2p(³ P°)4p	4p ⁴ D	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{2}{2}$ $\frac{3}{2}$	214758. 3 214772. 6 214794. 6 214828. 0	14. 3 22. 0 33. 4	5p' ² P	2s 2p(³ P°)5p	5p ² P	{ $\frac{1}{2}$ $\frac{1}{2}$ }	227901	
4p' ⁴ S ₂	2s 2p(³ P°)4p	4p ⁴ S	$\frac{1}{2}$	215765. 6			2s 2p(³ P°)5d	5d ⁴ D°	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{2}{2}$ $\frac{3}{2}$		
4p' ⁴ P ₂ ⁴ P ₃	2s 2p(³ P°)4p	4p ⁴ P	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{2}{2}$	216378. 0 216397. 7	19. 7	5d' ⁴ P ₃	2s 2p(³ P°)5d	5d ⁴ P°	$\frac{2}{2}$ $\frac{1}{2}$ $\frac{1}{2}$	231050	
4p' ² D ₃	2s 2p(³ P°)4p	4p ² D	$\frac{1}{2}$ $\frac{2}{2}$	216927		5f' ² F	2s 2p(³ P°)5f	5f ² F	{ $\frac{2}{2}$ $\frac{3}{2}$ }	231221	
4d' ⁴ F ₂ ⁴ F ₃ ⁴ F ₄ ⁴ F ₅	2s 2p(³ P°)4d	4d ⁴ F°	$\frac{1}{2}$ $\frac{2}{2}$ $\frac{3}{2}$ $\frac{4}{2}$	219553. 8 219568. 5 219589. 2 219617. 0	14. 7 20. 7 27. 8		2s 2p(³ P°)5f	5f ⁴ F	$\frac{1}{2}$ $\frac{2}{2}$ $\frac{3}{2}$ $\frac{4}{2}$	231226. 8	
4d' ⁴ D ₂ ⁴ D ₃ ⁴ D ₄	2s 2p(³ P°)4d	4d ⁴ D°	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{2}{2}$ $\frac{3}{2}$	220127. 8 220137. 0 220147. 6	9. 2 10. 6	5f' ⁴ F ₅	2s 2p(³ P°)5f	5f ⁴ G	$\frac{2}{2}$ $\frac{3}{2}$ $\frac{4}{2}$ $\frac{5}{2}$	231499. 3	
4d' ² D ₂ ² D ₃	2s 2p(³ P°)4d	4d ² D°	$\frac{1}{2}$ $\frac{2}{2}$	220601. 1 220614. 2	13. 1	5f' ⁴ G ₆	2s 2p(³ P°)5f	5f ⁴ D	$\frac{3}{2}$ $\frac{2}{2}$ $\frac{1}{2}$ $\frac{1}{2}$	231520. 4	
4d' ⁴ P ₃ ⁴ P ₂ ⁴ P ₁	2s 2p(³ P°)4d	4d ⁴ P°	$\frac{2}{2}$ $\frac{1}{2}$ $\frac{1}{2}$	220808. 47 220828. 97 220840. 87	-20. 50 -11. 90		2s 2p(³ P°)6d	6d ⁴ D°	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{2}{2}$ $\frac{3}{2}$	236444	
4f' ² F ₃ ² F ₄	2s 2p(³ P°)4f	4f ² F	$\frac{2}{2}$ $\frac{3}{2}$	221089. 6 221098. 8	9. 2	6d' ⁴ D ₄	2s 2p(³ P°)6d	6d ⁴ P°	$\frac{2}{2}$ $\frac{1}{2}$ $\frac{1}{2}$	236605	
4f' ⁴ F ₄ ⁴ F ₅	2s 2p(³ P°)4f	4f ⁴ F	$\frac{1}{2}$ $\frac{2}{2}$ $\frac{3}{2}$ $\frac{4}{2}$	221106. 3 221107. 4	1. 1	6d' ⁴ P ₃	2s 2p(³ P°)6d				

Config. 1s ² +	Observed Terms			
2s ² (1S)2p	2p ² P°			
2s 2p ²	{ 2p ² ² S 2p ² ⁴ P 2p ² ² P 2p ² ² D			
2p ³	{ 2p ³ ⁴ S° 2p ³ ² P° 2p ³ ² D°			
	ns (n ≥ 3)	np (n ≥ 3)	nd (n ≥ 3)	nf (n ≥ 4)
2s ² (1S)nx	3-6s ² S	3-5p ² P°	3-6d ² D	4-6f ² F°
2s 2p(³ P°)nx	{ 3-5s ⁴ P° 3s ² P°	{ 3, 4p ⁴ S 3, 4p ⁴ P 3, 4p ⁴ D 3p ² S 3, 5p ² P 3, 4p ² D	{ 3-6d ⁴ P° 3-6d ⁴ D° 3, 4d ⁴ F° 3, 4d ² P° 3, 4d ² D° 3, 4d ² F°	{ 4, 5f ⁴ D 4, 5f ⁴ F 4, 5f ⁴ G 4f ² D 4, 5f ² F 4f ² G

*For predicted terms in the spectra of the B I isoelectronic sequence, see Introduction.

C III

SEE REVISION IN NSRDS-NBS 3, Section 3, October 1970.

(Be I sequence; 4 electrons)

Z=6

Ground state 1s² 2s² ¹S₀

2s² ¹S₀ 386159. 7 cm⁻¹

I. P. 47.864 volts

All but three terms are from Edlén's Monograph. For the terms 7d ³D, 8d ³D, and 9d ³D the revised values of Whitelaw and Mack have been used. Edlén has since rejected his 4d' ¹P term.

No intersystem combinations have been found with certainty. The long D-series determine the limits to about ±25 cm⁻¹. The uncertainty *x* in the relative positions of the singlets and triplets is, therefore, less than ±50 cm⁻¹ according to Edlén. No trace of the line predicted at 1910.7 ± 2 Å, 2s² ¹S₀ - 2p ³P₁^o, is visible on his plates. A line observed at 339 Å (294314.1 cm⁻¹) agrees within 4 cm⁻¹ with the calculated combination 2p ³P₁^o - 5d ¹D₂. This identification is uncertain, since it is not confirmed by other intersystem combinations.

REFERENCES

- B. Edlén, Nova Acta Reg. Soc. Sci. Uppsala [IV] **9**, No. 6, 51 (1934). (I P) (T) (C L) (G D)
 N. G. Whitelaw and J. E. Mack, Phys. Rev. **47**, 677 (1935). (T)
 B. Edlén, private communication (Dec. 1947). (T)

C III

C III

Edlén	Config.	Desig.	J	Level	Interval	Edlén	Config.	Desig.	J	Level	Interval
2s ¹ S ₀	2s ²	2s ² ¹ S	0	0. 0		2p' ¹ D ₂	2p ²	2p ² ¹ D	2	145875. 1	
2p ³ P ₀	2s(2S)2p	2p ³ P°	0	52315. 0+x	23. 0	2p' ¹ S ₀	2p ²	2p ² ¹ S	0	182520. 2	
³ P ₁			1	52338. 0+x	56. 8	3s ³ S ₁	2s(2S)3s	3s ³ S	1	238160. 7+x	
³ P ₂			2	52394. 8+x		3s ¹ S ₀	2s(2S)3s	3s ¹ S	0	247169. 5	
2p ¹ P ₁	2s(2S)2p	2p ¹ P°	1	102351. 4		3p ¹ P ₁	2s(2S)3p	3p ¹ P°	1	258931. 4	
2p' ³ P ₀	2p ³	2p ³ ³ P	0	137374. 0+x	29. 4						
³ P ₁			1	137403. 4+x	47. 1						
³ P ₂			2	137450. 5+x							

C III—Continued

C III—Continued

Edlén	Config.	Desig.	<i>J</i>	Level	Interval	Edlén	Config.	Desig.	<i>J</i>	Level	Interval
3p ³ P ₀ ³ P ₁ ³ P ₂	2s(²S)3p	3p ³ P°	0 1 2	259653. 8+x 259659. 3+x 259672. 1+x	5. 5 12. 8	5d ³ D ₃	2s(²S)5d	5d ³ D	1 2 3	345444 +x	
3d ³ D ₁ ³ D ₂ ³ D ₃	2s(²S)3d	3d ³ D	1 2 3	269957. 6+x 269959. 7+x 269962. 9+x	2. 1 3. 2	5g ³ G ₄ ³ G ₅	2s(²S)5g	5g ³ G	3 4 5	346525. 1+x 346526. 0+x	0. 9
3d ¹ D ₂	2s(²S)3d	3d ¹ D	2	276482. 7		5g ¹ G ₄	2s(²S)5g	5g ¹ G	4	346577. 5	
3s' ³ P ₀ ³ P ₁ ³ P ₂	2p(²P°)3s	3s ³ P°	0 1 2	308162. 9+x 308196. 2+x 308264. 8+x	33. 3 68. 6	5d ¹ D ₂	2s(²S)5d	5d ¹ D	2	346656. 0	
4s ³ S ₁	2s(²S)4s	4s ³ S	1	309404. 5+x		3d' ¹ P ₁	2p(²P°)3d	3d ¹ P°	1	346713. 1	
3s' ¹ P ₁	2p(²P°)3s	3s ¹ P°	1	310005. 2		5f ³ F ₂ ³ F ₃ ³ F ₄	2s(²S)5f	5f ³ F°	2 3 4	347099. 5+x 347101. 3+x 347103. 7+x	1. 8 2. 4
4s ¹ S ₀	2s(²S)4s	4s ¹ S	0	311720. 7		5f ¹ F ₃	2s(²S)5f	5f ¹ F°	3	348859. 5	
4p ³ P ₀₁ ³ P ₂	2s(²S)4p	4p ³ P°	0, 1 2	317743 +x 317748 +x	5	6s ³ S ₁	2s(²S)6s	6s ³ S	1	354796 +x	
3p' ¹ P ₁	2p(²P°)3p	3p ¹ P	1	319719. 4		6p ¹ P ₁	2s(²S)6p	6p ¹ P°	1	357088	
4d ³ D ₁ ³ D ₂ ³ D ₃	2s(²S)4d	4d ³ D	1 2 3	321358. 8+x 321375. 1+x 321398. 6+x	16. 3 23. 5		2s(²S)6d	6d ³ D	1 2 3	358046 +x	
4f ³ F ₂ ³ F ₃ ³ F ₄	2s(²S)4f	4f ³ F°	2 3 4	321949. 1+x 321955. 8+x 321964. 7+x	6. 7 8. 9	6g ³ G ₄ ³ G ₅	2s(²S)6g	6g ³ G	3 4 5	358638. 3+x 358639. 0+x	0. 7
4p ¹ P ₁	2s(²S)4p	4p ¹ P°	1	322403. 1		6g ¹ G ₄	2s(²S)6g	6g ¹ G	4	358688. 9	
4f ¹ F ₃	2s(²S)4f	4f ¹ F°	3	322701. 1		6d ¹ D ₂	2s(²S)6d	6d ¹ D	2	358725. 5	
3p' ³ D ₁ ³ D ₂ ³ D ₃	2p(²P°)3p	3p ³ D	1 2 3	323024. 0+x 323049. 4+x 323088. 2+x	25. 4 38. 8		2s(²S)6f	6f ³ F°	2 3 4	358800 +x	
4d ¹ D ₂	2s(²S)4d	4d ¹ D	2	324212. 0		6f ¹ F ₃	2s(²S)6f	6f ¹ F°	3	359122. 2	
3p' ³ S ₁	2p(²P°)3p	3p ³ S	1	327225. 7+x		7s ³ S ₁	2s(²S)7s	7s ³ S	1	363561 +x	
3p' ³ P ₀ ³ P ₁ ³ P ₂	2p(²P°)3p	3p ³ P	0 1 2	329633. 1+x 329654. 2+x 329690. 9+x	21. 1 36. 7	7p ¹ P ₁	2s(²S)7p	7p ¹ P°	1	364896	
3d' ¹ D ₂	2p(²P°)3d	3d ¹ D°	2	332690. 3		7d ³ D	2s(²S)7d	7d ³ D	1, 2, 3	365585 +x	
3p' ¹ D ₂	2p(²P°)3p	3p ¹ D	2	333116. 4		7d ¹ D ₂	2s(²S)7d	7d ¹ D	2	366027. 0	
3d' ³ F ₂ ³ F ₃ ³ F ₄	2p(²P°)3d	3d ³ F°	2 3 4	333333. 4+x 333358. 4+x 333395. 0+x	25. 0 36. 6	8p ¹ P ₁	2s(²S)8p	8p ¹ P°	1	369926	
3d' ³ D ₁ ³ D ₂ ³ D ₃	2p(²P°)3d	3d ³ D°	1 2 3	337602. 9+x 337616. 4+x 337636. 7+x	13. 5 20. 3	8d ³ D	2s(²S)8d	8d ³ D	1, 2, 3	370438 +x	
5s ³ S ₁	2s(²S)5s	5s ³ S	1	339881 +x		9d ³ D	2s(²S)9d	9d ³ D	1, 2, 3	373748 +x	
3d' ³ P ₂ ³ P ₁ ³ P ₀	2p(²P°)3d	3d ³ P°	2 1 0	340049. 5+x 340075. 8+x 340090. 3+x	-26. 3 -14. 5		2p(²P°)4s	4s ³ P°	0 1 2	376637 +x	
3d' ¹ F ₃	2p(²P°)3d	3d ¹ F°	3	341368. 5		4s' ³ P ₂					
5p ¹ P ₁	2s(²S)5p	5p ¹ P°	1	343255. 7		4p' ¹ P ₁	2p(²P°)4p	4p ¹ P	1	381104. 8	
	2s(²S)5p	5p ³ P°	0 1 2	344181 +x		4p' ³ D ₂ ³ D ₃	2p(²P°)4p	4p ³ D	1 2 3	381919 +x 381958 +x	39
5p ³ P ₂						4p' ³ P ₁ ³ P ₂	2p(²P°)4p	4p ³ P	0 1 2	384313 +x 384350 +x	37
3p' ¹ S ₀	2p(²P°)3p	3p ¹ S	0	345093. 9		4p' ¹ D ₂	2p(²P°)4p	4p ¹ D	2	385637. 5	
						4d' ¹ D ₂	2p(²P°)4d	4d ¹ D°	2	385816. 2	
							C IV (²S _{1/2})	Limit	-----	386159. 7	

C III—Continued

C III—Continued

Edlén	Config.	Desig.	<i>J</i>	Level	Interval	Edlén	Config.	Desig.	<i>J</i>	Level	Interval
4d' ³ D ₃	2p(² P°)4d	4d ³ D°	1	387646 +x		5d' ³ P ₂	2p(² P°)5d	5d ³ P°	2	410841 +x	
			2						1		
			3						0		
4d' ³ P ₂	2p(² P°)4d	4d ³ P°	2	388442 +x		6p' ³ D ₃	2p(² P°)6p	6p ³ D	1	421380 +x	
			1						2		
			0						3		
4d' ¹ F ₃	2p(² P°)4d	4d ¹ F°	3	388772. 2			2p(² P°)6p	6p ³ P	0		
5p' ¹ P ₁	2p(² P°)5p	5p ¹ P	1	407430. 4		6p' ³ P ₂			1		
									2	421967 +x	
5p' ³ D ₃	2p(² P°)5p	5p ³ D	1	407774 +x		6d' ³ D ₃	2p(² P°)6d	6d ³ D°	1	422881 +x	
			2						2		
			3						3		
5p' ³ P ₂	2p(² P°)5p	5p ³ P	0	408873 +x		6d' ³ P ₂	2p(² P°)6d	6d ³ P°	2	423058 +x	
			1						1		
			2						0		
5p' ¹ D ₂	2p(² P°)5p	5p ¹ D	2	409505. 0			2p(² P°)7p	7p ³ D	1		
5d' ¹ D ₂	2p(² P°)5d	5d ¹ D°	2	409682. 1		7p' ³ D ₃			2	429345 +x	
									3		
5d' ³ D ₃	2p(² P°)5d	5d ³ D°	1	410534 +x		7p' ³ P ₂	2p(² P°)7p	7p ³ P	0	429712 +x	
			2						1		
			3						2		

December 1947.

C III OBSERVED TERMS*

Config. 1s ² +	Observed Terms				
2s ²	2s ² ¹ S				
2s(² S)2p	{ 2p ³ P° 2p ¹ P°				
2p ²	{ 2p ² ³ P 2p ² ¹ D				
	ns (n ≥ 3)	np (n ≥ 3)	nd (n ≥ 3)	nf (n ≥ 4)	ng (n ≥ 5)
2s(² S)nx	{ 3-7s ³ S 3, 4s ¹ S	{ 3-5p ³ P° 3-8p ¹ P°	{ 3-9d ³ D 3-7d ¹ D	{ 4-6f ³ F° 4-6f ¹ F°	{ 5, 6g ³ G 5, 6g ¹ G
2p(² P°)nx	{ 3, 4s ³ P° 3s ¹ P°	{ 3p ³ S 3-7p ³ P 3-7p ³ D 3p ¹ S 3-5p ¹ P 3-5p ¹ D	{ 3-6d ³ P° 3-6d ³ D° 3d ³ F° 3d ¹ P° 3-5d ¹ D° 3, 4d ¹ F°		

*For predicted terms of the Be I isoelectronic sequence, see Introduction.

C IV

(Li I sequence; 3 electrons)

Z=6

Ground state $1s^2 2s^2 S_{\frac{1}{2}}$ $2s^2 S_{\frac{1}{2}}$ 520177.8 cm^{-1}

I. P. 64.476 volts

The terms are from Edlén. His extrapolated values of three intervals and the term values of the two high series members $8f^2 F^\circ$ and $8g^2 G$, etc., which were calculated from a well-determined series formula, are entered in brackets in the table.

REFERENCES

B. Edlén, Zeit. Astroph. **7**, 378 (1933). (T) (C L)B. Edlén, Nova Acta Reg. Soc. Sci. Uppsala [IV] **9**, No. 6, 40 (1934). (I P) (T) (C L) (G D)T.-Y. Wu, Phys. Rev. **58**, 1114 (1940). (C L)

C IV

C IV

Edlén	Config.	Desig.	J	Level	Interval	Edlén	Config.	Desig.	J	Level	Interval
$2s^2 S_1$	2s	$2s^2 S$	$\frac{1}{2}$	0.0		$6d^2 D$	6d	$6d^2 D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	471368	
$2p^2 P_1$ $2p^2 P_2$	2p	$2p^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	64484.2 64591.3	[107.1]	$6f^2 F$	6f	$6f^2 F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	471403.0	
$3s^2 S_1$	3s	$3s^2 S$	$\frac{1}{2}$	302847.9		$6g^2 G$	6g	$6g^2 G$	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ 4\frac{1}{2} \end{smallmatrix} \right\}$	471407.4	
$3p^2 P_1$ $3p^2 P_2$	3p	$3p^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	320048.5 320080.0	[31.5]	$6h^2 H$	6h	$6h^2 H^\circ$	$\left\{ \begin{smallmatrix} 4\frac{1}{2} \\ 5\frac{1}{2} \end{smallmatrix} \right\}$	471407.9	
$3d^2 D_2$ $3d^2 D_3$	3d	$3d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	324880.2 324890.9	[10.7]	$7s^2 S_1$	7s	$7s^2 S$	$\frac{1}{2}$	482659	
$4s^2 S_1$	4s	$4s^2 S$	$\frac{1}{2}$	401346.7		$7p^2 P$	7p	$7p^2 P^\circ$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	483931	
$4p^2 P_1$ $4p^2 P_2$	4p	$4p^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	408308.9 408322.2	13.3	$7d^2 D$	7d	$7d^2 D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	484309	
$4d^2 D_2$ $4d^2 D_3$	4d	$4d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	410333.8 410338.2	4.4	$7f^2 F$	7f	$7f^2 F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	484343.8	
$4f^2 F_4$	4f	$4f^2 F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	410434.1	[2.1]	$7g^2 G$	7g	$7g^2 G$	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ 4\frac{1}{2} \end{smallmatrix} \right\}$	484346.6	
$5s^2 S_1$	5s	$5s^2 S$	$\frac{1}{2}$	445366.1		$7h^2 H$	7h	$7h^2 H^\circ$	$\left\{ \begin{smallmatrix} 4\frac{1}{2} \\ 5\frac{1}{2} \end{smallmatrix} \right\}$	484346.9	
$5p^2 P_1$ $5p^2 P_2$	5p	$5p^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	448854 448861	[6.7]	$8p^2 P$	8p	$8p^2 P^\circ$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	492473	
$5d^2 D_3$	5d	$5d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	449887.4	[2.2]	8F	8f	$8f^2 F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	[492743]	
$5f^2 F$	5f	$5f^2 F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	449938.2		8GHK	8g, etc.	$8g^2 G$, etc.	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ \text{to} \\ 7\frac{1}{2} \end{smallmatrix} \right\}$	[492745]	
$5g^2 G$	5g	$5g^2 G$	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ 4\frac{1}{2} \end{smallmatrix} \right\}$	449948.4							
$6s^2 S_1$	6s	$6s^2 S$	$\frac{1}{2}$	468765							
$6p^2 P$	6p	$6p^2 P^\circ$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	470763							
							C V (1S_0)	Limit		520177.8	

April 1946.

C v

(He I sequence; 2 electrons)

Z=6

Ground state $1s^2 \ ^1S_0$ $1s^2 \ ^1S_0$ $3162450 \pm 300 \text{ cm}^{-1}$ I. P. $391.986 \pm 0.037 \text{ volts}$

The singlet terms are from Tyrén, who has reported (1940) nine lines visible on his spectrograms. His limit has been calculated from the series members $n=2$ to 6. The remaining singlet terms have been calculated from three classified lines at 32 Å given in his 1936 paper. He has also classified a line at 40.731 Å as the intersystem combination $1s^2 \ ^1S_0 - 2p \ ^3P_1^\circ$. His unit, 10^3 cm^{-1} has here been changed to cm^{-1} .

The triplet terms are from an unpublished manuscript kindly furnished by Edlén, who states that the absolute term values of the triplets are based on an extrapolation of $3d \ ^3D$ from He I and Li II. The relative positions of the singlet and triplet terms thus determined confirm the intersystem combination reported by Tyrén.

REFERENCES

- B. Edlén, Nova Acta Reg. Soc. Sci. Uppsala [IV] **9**, No. 6, 31 (1934). (C L)
 F. Tyrén, Zeit. Phys. **98**, 774 (1936). (C L)
 H. A. Robinson, Phys. Rev. **51**, 14 (1937). (I P) (T) (C L)
 F. Tyrén, Nova Acta Reg. Soc. Sci. Uppsala [IV] **12**, No. 1, 24 (1940). (I P) (T) (C L)
 B. Edlén, unpublished material (Sept. 1947). (T)

C v

C v

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$1s^2$	$1s^2 \ ^1S$	0	0		$1s \ 4p$	$4p \ ^1P^\circ$	1	2991680	
$1s \ 2s$	$2s \ ^3S$	1	2411266		$1s \ 5p$	$5p \ ^1P^\circ$	1	3053060	
$1s \ 2p$	$2p \ ^3P^\circ$	0	2455165	-13 136	$1s \ 6p$	$6p \ ^1P^\circ$	1	3086420	
		1	2455152		$1s \ 7p$	$7p \ ^1P^\circ$	1	3106750	
		2	2455288		$1s \ 8p$	$8p \ ^1P^\circ$	1	3118760	
$1s \ 2p$	$2p \ ^1P^\circ$	1	2483240						
$1s \ 3d$	$3d \ ^3D$	3, 2, 1	2857308						
$1s \ 3p$	$3p \ ^1P^\circ$	1	2859350		C VI ($^2S_{1/2}$)	Limit		3162450	

September 1947.

C VI

(H sequence; 1 electron)

 $Z=6$ Ground state $1s\ ^2S_{1/2}$ $1s\ ^2S_{1/2}$ 3951950 cm^{-1}

I. P. 489.84 volts

The first three members of the Lyman series have been observed by Tyrén. The terms listed below have been calculated by J. E. Mack, "using $R_{C^{12}}=109732.286$ and $\Lambda=0.040$. The series limit of C^{13} is higher by $14.0\ \text{cm}^{-1}$ than the one shown here."

REFERENCES

F. Tyrén, Zeit. Phys. **98**, 771 (1936). (C L)

J. E. Mack, unpublished material (1949). (I P) (T) (C L)

C VI

C VI

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$1s$	$1s\ ^2S$	$\frac{1}{2}$	0		$4p$	$4p\ ^2P^\circ$	$\frac{1}{2}$	3704957]] 5 59.2 19.7 9.9
$2p$	$2p\ ^2P^\circ$	$\frac{1}{2}$	2963768]] 38 473.3	$4s$	$4s\ ^2S$	$\frac{1}{2}$	3704961	
$2s$	$2s\ ^2S$	$\frac{1}{2}$	2963806		$4p, 4d$	$4d\ ^2D, 4p\ ^2P^\circ$	$1\frac{1}{2}$	3705016	
$2p$	$2p\ ^2P^\circ$	$1\frac{1}{2}$	2964241		$4d, 4f$	$4d\ ^2D, 4f\ ^2F^\circ$	$2\frac{1}{2}$	3705035	
					$4f$	$4f\ ^2F^\circ$	$3\frac{1}{2}$	3705045	
$3p$	$3p\ ^2P^\circ$	$\frac{1}{2}$	3512811]] 11 140.3 46.7	$5s, \text{etc.}$	$5s\ ^2S, \text{etc.}$	$\frac{1}{2}, \text{etc.}$	3793884 to 933	
$3s$	$3s\ ^2S$	$\frac{1}{2}$	3512822						
$3p, 3d$	$3d\ ^2D, 3p\ ^2P^\circ$	$1\frac{1}{2}$	3512951						
$3d$	$3d\ ^2D$	$2\frac{1}{2}$	3512998						
						$\infty = \text{Limit}$		3951950	

February 1949.

NITROGEN

N I

7 electrons

Z=7

Ground state $1s^2 2s^2 2p^3 {}^4S_{1\frac{1}{2}}^{\circ}$ $2p^3 {}^4S_{1\frac{1}{2}}^{\circ}$ 117345 cm^{-1}

I. P. 14.54 volts

The terms have been taken chiefly from the list prepared by Ekefors with extensions calculated from the classifications published in Tokyo. Unfortunately, no term list was included in the Tokyo papers. Consequently, considerable editing has been done in compiling terms from all the observational material. Revised values are suggested for a few levels and tentative values not in the literature are listed for $5d {}^4F_{2\frac{1}{2}}$, $5d {}^4F_{1\frac{1}{2}}$, $5d {}^4D_{3\frac{1}{2}}$, and $6d {}^4D_{3\frac{1}{2}}$. Further study is needed to verify the numerous blends resulting from practically coincident levels.

Intersystem combinations have been observed.

Kiess and Shortley have generously furnished g -values derived from the observed Zeeman effects of 18 infrared lines.

REFERENCES

- O. S. Duffendack and R. A. Wolfe, Phys. Rev. **34**, 409 (1929). (C L)
 S. B. Ingram, Phys. Rev. **34**, 421 (1929). (T) (C L)
 E. Ekefors, Zeit. Phys. **63**, 437 (1930). (I P) (T) (C L)
 H. E. White, *Introduction to Atomic Spectra* p. 260 (McGraw-Hill Book Co., Inc., New York, N. Y., 1934). (G D)
 B. Edlén, Bergstrand's Festskrift p. 135 (1938). (C L)
 M. Kamiyama, Sci. Papers Inst. Phys. Chem. Research (Tokyo) **36**, No. 933, 375 (1939). (C L)
 M. Kamiyama and T. Sugiura, Sci. Papers Inst. Phys. Chem. Research (Tokyo) **37**, Nos. 982 and 983, 479 (1940). (C L)
 M. Kamiyama and H. Noguchi, Sci. Papers Inst. Phys. Chem. Research (Tokyo) **39**, No. 1100, 475 (1942). (C L)
 J. R. Holmes, Phys. Rev. **63**, 41 (1943). (I S)
 W. F. Meggers, J. Opt. Soc. Am. **36**, 431 (1946). (Summary hfs)
 C. C. Kiess and G. Shortley, J. Research Nat. Bur. Std. **42**, 190, RP1961 (1949). (Z E)

N I

N I

Config.	Desig.	J	Level	Interval	Obs. g	Config.	Desig.	J	Level	Interval	Obs. g
$2s^2 2p^3$	$2p^3 {}^4S^{\circ}$	$1\frac{1}{2}$	0			$2s 2p^4$	$2p^4 {}^4P$	$2\frac{1}{2}$	88109.5		
$2s^2 2p^3$	$2p^3 {}^2D^{\circ}$	$2\frac{1}{2}$	19223	-8				$1\frac{1}{2}$	88153.4	-43.9	
		$1\frac{1}{2}$	19231					$\frac{1}{2}$	88173.0	-19.6	
$2s^2 2p^3$	$2p^3 {}^2P^{\circ}$	$\left\{ \begin{array}{c} 1\frac{1}{2} \\ \frac{1}{2} \end{array} \right\}$	28840			$2s^2 2p^2({}^3P)3p$	$3p {}^2S^{\circ}$	$\frac{1}{2}$	93582.3		
$2s^2 2p^2({}^3P)3s$	$3s {}^4P$	$\frac{1}{2}$	83285.5	33.8 46.7	2.670 1.735 1.603	$2s^2 2p^2({}^3P)3p$	$3p {}^4D^{\circ}$	$\frac{1}{2}$	94772.2	22.6	0.002
		$1\frac{1}{2}$	83319.3					$1\frac{1}{2}$	94794.8	37.3	1.19
		$2\frac{1}{2}$	83366.0					$2\frac{1}{2}$	94832.1	51.0	1.36
$2s^2 2p^2({}^3P)3s$	$3s {}^2P$	$\frac{1}{2}$	86131.4	91.8		$2s^2 2p^2({}^3P)3p$	$3p {}^4P^{\circ}$	$\frac{1}{2}$	94883.1		1.44
		$1\frac{1}{2}$	86223.2					$1\frac{1}{2}$	95476.5	18.4	2.671
								$2\frac{1}{2}$	95494.9	38.3	1.737
									95533.2		1.598

N I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$2s^2 2p^2(^3P)3p$	$3p\ ^4S^\circ$	$1\frac{1}{2}$	96751. 7		2. 004
$2s^2 2p^2(^3P)3p$	$3p\ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	96788. 2 96864. 2	76. 0	
$2s^2 2p^2(^3P)3p$	$3p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	97770. 1 97805. 8	35. 7	
$2s^2 2p^2(^1D)3s$	$3s'\ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	99665 99658	—7	
$2s^2 2p^2(^3P)4s$	$4s\ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	103618. 1 103668. 1 103736. 8	50. 0 68. 7	
$2s^2 2p^2(^3P)4s$	$4s\ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	104142. 2 104227. 4	85. 2	
$2s^2 2p^2(^3P)3d$	$3d\ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	104615. 4 104654. 9	—39. 5	
$2s^2 2p^2(^3P)3d$	$3d\ ^4F$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	104665 104684 104718 104767	19 34 49	
$2s^2 2p^2(^3P)3d$	$3d\ ^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	104810. 9 104882. 7	71. 8	
$2s^2 2p^2(^3P)3d$	$3d\ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	104864 104890 104957	26 67	
$2s^2 2p^2(^3P)3d$	$3d\ ^4D$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	104987 104998 105011 105020	11 13 9	
$2s^2 2p^2(^3P)3d$	$3d\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	105120. 8 105144. 3	23. 5	
$2s^2 2p^2(^3P)4p$	$4p\ ^2S^\circ$	$\frac{1}{2}$	106478. 6		
$2s^2 2p^2(^3P)4p$	$4p\ ^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	106760. 5 106780. 1 106816. 1 106870. 7	19. 6 36. 0 54. 6	
$2s^2 2p^2(^3P)4p$	$4p\ ^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	106982. 7 106998. 3 107039. 0	15. 6 40. 7	
$2s^2 2p^2(^3P)4p$	$4p\ ^4S^\circ$	$1\frac{1}{2}$	107447. 2		
$2s^2 2p^2(^3P)5s$	$5s\ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	109813. 5 109857. 8 109927. 9	44. 3 70. 1	
$2s^2 2p^2(^3P)5s$	$5s\ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	110029. 2 110108. 5	79. 3	
$2s^2 2p^2(^3P)4d$	$4d\ ^4F$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	110196 110214 110248 110304	18 34 56	
$2s^2 2p^2(^3P)4d$	$4d\ ^4D$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	110221 110275 110288 110339	54 13 51	
$2s^2 2p^2(^3P)4d$	$4d\ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	110221. 7 110244. 6	—22. 9	
$2s^2 2p^2(^3P)4d$	$4d\ ^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	110311 110373	62	

N I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$2s^2 2p^2(^3P)4d$	$4d\ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	110325 110351 110403	26 52	
$2s^2 2p^2(^3P)4d$	$4d\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	110448. 3 110470. 5	22. 2	
$2s^2 2p^2(^1D)3p$	$3p'\ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	110521. 9 110545. 8	23. 9	
$2s^2 2p^2(^1D)3p$	$3p'\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	112294. 8 112320. 8	26. 0	
$2s^2 2p^2(^3P)6s$	$6s\ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	112565. 9 112610. 6 112682. 6	44. 7 72. 0	
$2s^2 2p^2(^3P)6s$	$6s\ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	112735 112823	88	
$2s^2 2p^2(^3P)5d$	$5d\ ^4F$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	112751? 112763? 112799 112862	12 36 63	
$2s^2 2p^2(^3P)5d$	$5d\ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	112801 112816	—15	
$2s^2 2p^2(^3P)5d$	$5d\ ^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	112820 112890. 2	70	
$2s^2 2p^2(^3P)5d$	$5d\ ^4D$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	112825 112892?	67	
$2s^2 2p^2(^3P)5d$	$5d\ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	112855 112874 112912	19 38	
$2s^2 2p^2(^3P)5d$	$5d\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	112929. 2 112947. 5	18. 3	
$2s^2 2p^2(^3P)7s$	$7s\ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	114015? 114072? 114146	57 74	
$2s^2 2p^2(^3P)7s$	$7s\ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	114130 114163	33	
$2s^2 2p^2(^3P)6d$	$6d\ ^4F$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ \text{to} \\ 4\frac{1}{2} \end{array} \right\}$	114160		
$2s^2 2p^2(^3P)6d$	$6d\ ^4D$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	114182 114248?	66	
$2s^2 2p^2(^3P)6d$	$6d\ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	114193 114209	—16	
$2s^2 2p^2(^3P)6d$	$6d\ ^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	114196 114275	79	
$2s^2 2p^2(^3P)6d$	$6d\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	114232. 2 114290. 5	58. 3	
$2s^2 2p^2(^3P)6d$	$6d\ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	114259 114274	15	
$2s^2 2p^2(^3P)8s$	$8s\ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	114809 114890 114942	81 52	

N I—Continued

N I—Continued

Config.	Desig.	J	Level	Interval	Obs. g	Config.	Desig.	J	Level	Interval	Obs. g	
$2s^2\ 2p^2(^3P)8s$	$8s\ ^2P$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	114950	42. 6		$2s^2\ 2p^2(^3P)11s$	$11s\ ^2P$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	116107			
$2s^2\ 2p^2(^3P)7d$	$7d\ ^4D$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	114988			$2s^2\ 2p^2(^3P)11s$	$11s\ ^4P$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	116124			
$2s^2\ 2p^2(^3P)7d$	$7d\ ^2F$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	115004			$2s^2\ 2p^2(^3P)10d$	$10d\ ^2P$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ \frac{1}{2} \end{smallmatrix} \right\}$	116155			
$2s^2\ 2p^2(^3P)7d$	$7d\ ^2P$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ \frac{1}{2} \end{smallmatrix} \right\}$	115017			$2s^2\ 2p^2(^3P)10d$	$10d\ ^2F$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	116159			
$2s^2\ 2p^2(^3P)7d$	$7d\ ^2D$	$\begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 115057. 5 \\ 115100. 1 \end{smallmatrix}$			$2s^2\ 2p^2(^3P)10d$	$10d\ ^4D$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	116164			
$2s^2\ 2p^2(^3P)7d$	$7d\ ^4P$	$\begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix}$	115103			$2s^2\ 2p^2(^3P)10d$	$10d\ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	116240			
$2s^2\ 2p^2(^3P)9s$	$9s\ ^2P$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	115480	25		$2s^2\ 2p^2(^3P)10d$	$10d\ ^4P$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	116259			
$2s^2\ 2p^2(^3P)9s$	$9s\ ^4P$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	115483			$2s^2\ 2p^2(^3P)12s$	$12s\ ^2P$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	116305			
$2s^2\ 2p^2(^3P)8d$	$8d\ ^4D$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	115524			$2s^2\ 2p^2(^3P)12s$	$12s\ ^4P$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	116312			
$2s^2\ 2p^2(^3P)8d$	$8d\ ^2P$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ \frac{1}{2} \end{smallmatrix} \right\}$	115530			$2s^2\ 2p^2(^3P)11d$	$11d\ ^2P$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ \frac{1}{2} \end{smallmatrix} \right\}$	116351			
$2s^2\ 2p^2(^3P)8d$	$8d\ ^2F$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	115535			$2s^2\ 2p^2(^3P)11d$	$11d\ ^2F$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	116359			
$2s^2\ 2p^2(^3P)8d$	$8d\ ^2D$	$\begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 115597 \\ 115622 \end{smallmatrix}$			$2s^2\ 2p^2(^3P)11d$	$11d\ ^4D$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	116367			
$2s^2\ 2p^2(^3P)8d$	$8d\ ^4P$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	115618			$2s^2\ 2p^2(^3P)11d$	$11d\ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	116436			
$2s^2\ 2p^2(^3P)10s$	$10s\ ^2P$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	115842			$2s^2\ 2p^2(^3P)11d$	$11d\ ^4P$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	116441			
$2s^2\ 2p^2(^3P)10s$	$10s\ ^4P$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	115855			$2s^2\ 2p^2(^3P)13s$	$13s\ ^2P$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	116467			
$2s^2\ 2p^2(^3P)9d$	$9d\ ^4D$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	115887			$2s^2\ 2p^2(^3P)12d$	$12d\ ^2P$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ \frac{1}{2} \end{smallmatrix} \right\}$	116502			
$2s^2\ 2p^2(^3P)9d$	$9d\ ^2P$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ \frac{1}{2} \end{smallmatrix} \right\}$	115889	18		$2s^2\ 2p^2(^3P)12d$	$12d\ ^4P$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	116581			
$2s^2\ 2p^2(^3P)9d$	$9d\ ^2F$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	115902			$2s^2\ 2p^2(^3P)12d$	$12d\ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	116625			
$2s^2\ 2p^2(^3P)9d$	$9d\ ^2D$	$\begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 115973 \\ 115991 \end{smallmatrix}$			N II (3P_0)		<i>Limit</i>	117345			
$2s^2\ 2p^2(^3P)9d$	$9d\ ^4P$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	115990									

October 1947.

N I OBSERVED TERMS*

Config. $1s^2+$	Observed Terms					
$2s^2 2p^3$	{ $2p^3 {}^4S^\circ$ $2p^3 {}^2P^\circ$ $2p^3 {}^2D^\circ$					
$2s 2p^4$	$2p^4 {}^4P$					
	$ns (n \geq 3)$		$np (n \geq 3)$		$nd (n \geq 3)$	
$2s^2 2p^2({}^3P)nx$	{ $3-12s {}^4P$ $3-13s {}^2P$		$3, 4p {}^4S^\circ$ $3, 4p {}^2S^\circ$	$3, 4p {}^4P^\circ$ $3p {}^2P^\circ$	$3, 4p {}^4D^\circ$ $3p {}^2D^\circ$	$3-12d {}^4P$ $3-11d {}^4D$ $3-6d {}^4F$ $3-12d {}^2P$ $3-12d {}^2D$ $3-11d {}^2F$
$2s^2 2p^2({}^1D)nx'$	$3s' {}^2D$		$3p' {}^2P^\circ$ $3p' {}^2D^\circ$			

*For predicted terms in the spectra of the N I isoelectronic sequence, see Introduction.

N II

(C I sequence; 6 electrons)

$Z=7$

Ground state $1s^2 2s^2 2p^2 {}^3P_0$

$2p^2 {}^3P_0$ 238846.7 cm^{-1}

I. P. 29.605 volts

Edlén has revised and extended the earlier analysis of this spectrum. The terms are all taken from his Monograph, except those from the $4f$ configuration, which are from his 1936 paper, and his $3s' {}^3P$ and $5f$ -terms, which he has generously furnished in a private communication.

The singlet and triplet terms are well connected by intersystem combinations but the quintets are not so connected with the others. Edlén also suggests that by analogy with C I and O III the published absolute values of the quintet terms should be decreased by about 500 cm^{-1} . This correction has been applied in the table and should diminish the uncertainty x appreciably.

REFERENCES

- A. Fowler and L. J. Freeman, Proc. Roy. Soc. (London) [A] **114**, 662 (1927). (T) (C L)
 L. J. Freeman, Proc. Roy. Soc. (London) [A] **124**, 666 (1929). (T) (C L)
 B. Edlén, Nova Acta Reg. Soc. Sci. Uppsala [IV] **9**, No. 6, 109 (1934). (I P) (T) (C L) (G D of singlets)
 B. Edlén, Zeit. Phys. **98**, 564 (1936). (T) (C L)
 J. B. Green and H. N. Maxwell, Phys. Rev. **51**, 243 (1937). (Z E)
 B. Edlén, unpublished material (Dec. 1947). (T).

N II

N II

Edlén	Config.	Desig.	<i>J</i>	Level	Interval	Edlén	Config.	Desig.	<i>J</i>	Level	Interval
$2p\ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s^2\ 2p^2$	$2p^2\ ^3P$	0 1 2	0.0 49.1 131.3	49.1 82.2	$4p\ ^3S_1$	$2s^2\ 2p(^3P^\circ)4p$	$4p\ ^3S$	1	203532.8	
$2p\ ^1D_2$	$2s^2\ 2p^2$	$2p^2\ ^1D$	2	15315.7		$4p\ ^1D_2$	$2s^2\ 2p(^3P^\circ)4p$	$4p\ ^1D$	2	205350.7	
$2p\ ^1S_0$	$2s^2\ 2p^2$	$2p^2\ ^1S$	0	32687.1		$3s'\ ^5P_1$ $\ ^5P_2$ $\ ^5P_3$	$2s\ 2p^2(^4P)3s$	$3s\ ^5P$	1 2 3	205982.1+x 206038.1+x 206108.7+x	56.0 70.6
$2p'\ ^5S_2$	$2s\ 2p^3$	$2p^3\ ^5S^\circ$	2	47167.7+x		$4p\ ^1S_0$	$2s^2\ 2p(^2P^\circ)4p$	$4p\ ^1S$	0	206327.5	
$2p'\ ^3D_3$ $\ ^3D_2$ $\ ^3D_1$	$2s\ 2p^3$	$2p^3\ ^3D^\circ$	3 2 1	92237.9 92251.3 92252.9	-13.4 -1.6	$4d\ ^3F_2$ $\ ^3F_3$ $\ ^3F_4$	$2s^2\ 2p(^2P^\circ)4d$	$4d\ ^3F^\circ$	2 3 4	209675.3 209739.5 209825.3	64.2 85.8
$2p'\ ^1P_{12}$ $\ ^1P_0$	$2s\ 2p^3$	$2p^3\ ^1P^\circ$	2,1 0	109218.2 109224.8	-6.6	$4d\ ^1D_2$	$2s^2\ 2p(^2P^\circ)4d$	$4d\ ^1D^\circ$	2	209926.92	
$2p'\ ^1D_2$	$2s\ 2p^3$	$2p^3\ ^1D^\circ$	2	144189.1		$4d\ ^3D_1$ $\ ^3D_2$ $\ ^3D_3$	$2s^2\ 2p(^2P^\circ)4d$	$4d\ ^3D^\circ$	1 2 3	210239.8 210266.3 210301.9	26.5 35.6
$3s\ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s^2\ 2p(^2P^\circ)3s$	$3s\ ^3P^\circ$	0 1 2	148909.37 148940.97 149077.33	31.60 136.36	$4d\ ^3P_2$ $\ ^3P_1$ $\ ^3P_0$	$2s^2\ 2p(^2P^\circ)4d$	$4d\ ^3P^\circ$	2 1 0	210705.4 210751.5 210777.0	-46.1 -25.5
$3s\ ^1P_1$	$2s^2\ 2p(^2P^\circ)3s$	$3s\ ^1P^\circ$	1	149188.74		$4f\ ^1F_3$	$2s^2\ 2p(^2P^\circ)4f$	$4f\ ^1F$	3	211030.90	
$2p'\ ^3S_1$	$2s\ 2p^3$	$2p^3\ ^3S^\circ$	1	155129.9		$4f\ ^3F_2$ $\ ^3F_3$ $\ ^3F_4$	$2s^2\ 2p(^2P^\circ)4f$	$4f\ ^3F$	2 3 4	211033.71 211057.07 211061.03	23.36 3.96
$3p\ ^1P_1$	$2s^2\ 2p(^2P^\circ)3p$	$3p\ ^1P$	1	164611.60		$4d\ ^1F_3$	$2s^2\ 2p(^2P^\circ)4d$	$4d\ ^1F^\circ$	3	211104.8	
$3p\ ^3D_1$ $\ ^3D_2$ $\ ^3D_3$	$2s^2\ 2p(^2P^\circ)3p$	$3p\ ^3D$	1 2 3	166522.48 166583.26 166679.45	60.78 96.19	$4f\ ^3G_3$ $\ ^3G_4$ $\ ^3G_5$	$2s^2\ 2p(^2P^\circ)4f$	$4f\ ^3G$	3 4 5	211288.02 211295.65 211390.77	7.63 95.12
$2p'\ ^1P_1$	$2s\ 2p^3$	$2p^3\ ^1P^\circ$	1	166765.7		$4d\ ^1P_1$	$2s^2\ 2p(^2P^\circ)4d$	$4d\ ^1P^\circ$	1	211335.5	
$3p\ ^3S_1$	$2s^2\ 2p(^2P^\circ)3p$	$3p\ ^3S$	1	168893.04		$4f\ ^1G_4$	$2s^2\ 2p(^2P^\circ)4f$	$4f\ ^1G$	4	211402.89	
$3p\ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s^2\ 2p(^2P^\circ)3p$	$3p\ ^3P$	0 1 2	170573.38 170608.63 170667.00	35.25 58.37	$4f\ ^3D_3$ $\ ^3D_2$ $\ ^3D_1$	$2s^2\ 2p(^2P^\circ)4f$	$4f\ ^3D$	3 2 1	211411.25 211416.20 211487.28	-4.95 -71.08
$3p\ ^1D_2$	$2s^2\ 2p(^2P^\circ)3p$	$3p\ ^1D$	2	174212.93		$4f\ ^1D_2$	$2s^2\ 2p(^2P^\circ)4f$	$4f\ ^1D$	2	211491.16	
$3p\ ^1S_0$	$2s^2\ 2p(^2P^\circ)3p$	$3p\ ^1S$	0	178274.17		$3s'\ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s\ 2p^2(^4P)3s$	$3s\ ^3P$	0 1 2	211750.2 211780.6 211828.8	30.4 48.2
$3d\ ^3F_2$ $\ ^3F_3$ $\ ^3F_4$	$2s^2\ 2p(^2P^\circ)3d$	$3d\ ^3F^\circ$	2 3 4	186512.38 186571.80 186653.35	59.42 81.55	$5s\ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s^2\ 2p(^2P^\circ)5s$	$5s\ ^3P^\circ$	0 1 2	214212.4 214258.2 214385.3	45.8 127.1
$3d\ ^1D_2$	$2s^2\ 2p(^2P^\circ)3d$	$3d\ ^1D^\circ$	2	187092.20		$5s\ ^1P_1$	$2s^2\ 2p(^2P^\circ)5s$	$5s\ ^1P^\circ$	1	214828.0	
$3d\ ^3D_1$ $\ ^3D_2$ $\ ^3D_3$	$2s^2\ 2p(^2P^\circ)3d$	$3d\ ^3D^\circ$	1 2 3	187438.34 187462.38 187492.72	24.04 30.34		$2s^2\ 2p(^2P^\circ)5d$	$5d\ ^3D^\circ$	1 2 3	220717	
$3d\ ^3P_2$ $\ ^3P_1$ $\ ^3P_0$	$2s^2\ 2p(^2P^\circ)3d$	$3d\ ^3P^\circ$	2 1 0	188858.09 188909.89 188937.95	-51.80 -28.06	$5d\ ^3D_3$					
$3d\ ^1F_3$	$2s^2\ 2p(^2P^\circ)3d$	$3d\ ^1F^\circ$	3	189336.0		$5f\ ^3F_2$ $\ ^3F_3$ $\ ^3F_4$	$2s^2\ 2p(^2P^\circ)5f$	$5f\ ^3F$	2 3 4	221070.2 221074.3	4.1
$3d\ ^1P_1$	$2s^2\ 2p(^2P^\circ)3d$	$3d\ ^1P^\circ$	1	190121.15		$5d\ ^1F_3$	$2s^2\ 2p(^2P^\circ)5d$	$5d\ ^1F^\circ$	3	221137.6	
$4s\ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s^2\ 2p(^2P^\circ)4s$	$4s\ ^3P^\circ$	0 1 2	196541.09 196592.88 196712.17	51.79 119.29	$5f\ ^3G_3$ $\ ^3G_4$ $\ ^3G_5$	$2s^2\ 2p(^2P^\circ)5f$	$5f\ ^3G$	3 4 5	221227.7 221232.7 221302.2	5.0 69.5
$4s\ ^1P_1$	$2s^2\ 2p(^2P^\circ)4s$	$4s\ ^1P^\circ$	1	197859.28		$5f\ ^1G_4$	$2s^2\ 2p(^2P^\circ)5f$	$5f\ ^1G$	4	221312.1	
$4p\ ^1P_1$	$2s^2\ 2p(^2P^\circ)4p$	$4p\ ^1P$	1	202169.9		$3p'\ ^5D_0$ $\ ^5D_1$ $\ ^5D_2$ $\ ^5D_3$ $\ ^5D_4$	$2s\ 2p^2(^4P)3p$	$3p\ ^5D^\circ$	0 1 2 3 4	224027.1+x 224042.9+x 224072.3+x 224115.4+x 224169.3+x	15.8 29.4 43.1 53.9
$4p\ ^3D_1$ $\ ^3D_2$ $\ ^3D_3$	$2s^2\ 2p(^2P^\circ)4p$	$4p\ ^3D$	1 2 3	202714.94 202765.86 202862.06	50.92 96.20						
$4p\ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s^2\ 2p(^2P^\circ)4p$	$4p\ ^3P$	0 1 2	203164.7 203188.8 203259.7	24.1 70.9						

N II—Continued

N II—Continued

Edlén	Config.	Desig.	<i>J</i>	Level	Interval	Edlén	Config.	Desig.	<i>J</i>	Level	Interval
$3p' \ ^5P_1$ $\ ^5P_2$ $\ ^5P_3$	$2s \ 2p^2(^4P)3p$	$3p \ ^5P^\circ$	1 2 3	$225987.1+x$ $226011.2+x$ $226055.2+x$	24.1 44.0	$3d' \ ^5P_3$ $\ ^5P_2$ $\ ^5P_1$	$2s \ 2p^2(^4P)3d$	$3d \ ^5P$	3 2 1	$244737.4+x$ $244775.9+x$ $244802.0+x$	-38.5 -26.1
$3p' \ ^5S_2$	$2s \ 2p^2(^4P)3p$	$3p \ ^5S^\circ$	2	$230223.0+x$		$3d' \ ^5D_0$ $\ ^5D_1$ $\ ^5D_2$ $\ ^5D_3$ $\ ^5D_4$	$2s \ 2p^2(^4P)3d$	$3d \ ^5D$	0 1 2 3 4	$245319.8+x$ $245323.4+x$ $245331.3+x$ $245342.9+x$ $245356.9+x$	3.6 7.9 11.6 14.0
	N III ($^2P_{\frac{1}{2}}^{\circ}$)	Limit	-----	238846.7							
$3d' \ ^5F_1$ $\ ^5F_2$ $\ ^5F_3$ $\ ^5F_4$ $\ ^5F_5$	$2s \ 2p^2(^4P)3d$	$3d \ ^5F$	1 2 3 4 5	$243355.5+x$ $243371.2+x$ $243396.6+x$ $243430.2+x$ $243470.8+x$	15.7 25.4 33.6 40.6						

December 1947.

N II OBSERVED *g*-VALUES

Desig.	<i>J</i>	Obs. <i>g</i>	Desig.	<i>J</i>	Obs. <i>g</i>	Desig.	<i>J</i>	Obs. <i>g</i>
$3s \ ^3P^\circ$	1 2	1.455 1.502	$3p \ ^3S$	1	2.015	$3d \ ^1D^\circ$	2	0.986
$3s \ ^1P^\circ$	1	1.051	$3p \ ^3P$	1 2	1.530 1.497	$3d \ ^3D^\circ$	1 2 3	0.494 1.114 1.329
$3p \ ^1P$	1	1.005	$3p \ ^1D$	2	1.002	$3d \ ^3P^\circ$	2 1	1.504 1.487
$3p \ ^1D$	1 2 3	0.494 1.166 1.330	$3d \ ^3F^\circ$	3 4	1.079 1.250	$3d \ ^1P^\circ$	1	1.026

N II OBSERVED TERMS*

Config. $1s^2+$	Observed Terms			
$2s^2 \ 2p^2$	{ $2p^2 \ ^1S$ $2p^2 \ ^3P$ $2p^2 \ ^1D$ }			
$2s \ 2p^3$	{ $2p^3 \ ^5S^\circ$ $2p^3 \ ^3S^\circ$ $2p^3 \ ^3P^\circ$ $2p^3 \ ^3D^\circ$ $2p^3 \ ^1P^\circ$ $2p^3 \ ^1D^\circ$ }			
	$ns \ (n \geq 3)$	$np \ (n \geq 3)$	$nd \ (n \geq 3)$	$nf \ (n \geq 4)$
$2s^2 \ 2p(^2P^\circ)nx$	{ $3-5s \ ^3P^\circ$ $3-5s \ ^1P^\circ$ }	$3, 4p \ ^3S$ $3, 4p \ ^3P$ $3, 4p \ ^3D$ $3, 4p \ ^1S$ $3, 4p \ ^1P$ $3, 4p \ ^1D$	$3, 4d \ ^3P^\circ$ $3-5d \ ^3D^\circ$ $3, 4d \ ^3F^\circ$ $3, 4d \ ^1P^\circ$ $3, 4d \ ^1D^\circ$ $3-5d \ ^1F^\circ$	$4f \ ^3D$ $4, 5f \ ^3F$ $4, 5f \ ^3G$ $4f \ ^1D$ $4f \ ^1F$ $4, 5f \ ^1G$
$2s \ 2p^2(^4P)nx$	{ $3s \ ^5P$ $3s \ ^3P$ }	$3p \ ^5S^\circ$ $3p \ ^5P^\circ$ $3p \ ^5D^\circ$	$3d \ ^5P$ $3d \ ^5D$ $3d \ ^5F$	

*For predicted terms in the spectra of the C I isoelectronic sequence, see Introduction.

N III

(B I sequence; 5 electrons)

Z=7

Ground state $1s^2 2s^2 2p^2 P_{1/2}^o$ $2p^2 P_{1/2}^o$ 382625.5 cm^{-1}

I. P. 47.426 volts

All of the terms except those with a 4f-electron, have been taken from Edlén's Monograph. In 1936 Edlén published a revised and extended list of 4f-terms and the corresponding classified lines, including intersystem combinations. The observed correction to his previously published quartet terms -396.4 cm^{-1} , connecting them with the doublet terms has been incorporated into the present list.

REFERENCES

- L. J. Freeman, Proc. Roy. Soc. (London) [A] **121**, 318 (1928). (T) (C L)
 B. Edlén, Nova Acta Reg. Soc. Sci. Uppsala [IV] **9**, No. 6, 78 (1934). (I P) (T) (C L) (G D)
 B. Edlén, Zeit. Phys. **98**, 561 (1936). (T) (C L)

N III

N III

Edlén	Config.	Desig.	<i>J</i>	Level	Interval	Edlén	Config.	Desig.	<i>J</i>	Level	Interval
$2p^2 P_{1/2}$ $2p^2 P_{3/2}$	$2s^2(1S)2p$	$2p^2 P^o$	$\frac{1}{2}$ $\frac{3}{2}$	0. 0 174. 5	174. 5	$3s' \ ^4P_1$ $3s' \ ^4P_2$ $3s' \ ^4P_3$	$2s \ 2p(^3P^o)3s$	$3s \ ^4P^o$	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{3}{2}$	287535. 6 287598. 1 287713. 9	62. 5 115. 8
$2p' \ ^4P_1$ $2p' \ ^4P_2$ $2p' \ ^4P_3$	$2s \ 2p^2$	$2p^2 \ ^4P$	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{3}{2}$	57192. 1 57252. 0 57333. 2	59. 9 81. 2	$3s' \ ^2P_1$ $3s' \ ^2P_2$	$2s \ 2p(^3P^o)3s$	$3s \ ^2P^o$	$\frac{1}{2}$ $\frac{1}{2}$	297150. 2 297263. 1	112. 9
$2p' \ ^2D_3$ $2p' \ ^2D_2$	$2s \ 2p^2$	$2p^2 \ ^2D$	$\frac{3}{2}$ $\frac{1}{2}$	101023. 8 101031. 5	-7. 7	$4s \ ^2S_1$	$2s^2(1S)4s$	$4s \ ^2S$	$\frac{1}{2}$	301088. 2	
$2p' \ ^2S_1$	$2s \ 2p^2$	$2p^2 \ ^2S$	$\frac{1}{2}$	131003. 5		$3p' \ ^2P_1$ $3p' \ ^2P_2$	$2s \ 2p(^3P^o)3p$	$3p \ ^2P$	$\frac{1}{2}$ $\frac{1}{2}$	309132. 6 309185. 8	53. 2
$2p' \ ^2P_1$ $2p' \ ^2P_2$	$2s \ 2p^2$	$2p^2 \ ^2P$	$\frac{1}{2}$ $\frac{1}{2}$	145876. 1 145986. 5	110. 4	$3p' \ ^4D_1$ $3p' \ ^4D_2$ $3p' \ ^4D_3$ $3p' \ ^4D_4$	$2s \ 2p(^3P^o)3p$	$3p \ ^4D$	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{3}{2}$ $\frac{3}{2}$	309662. 8 309698. 3 309760. 5 309856. 7	35. 5 62. 2 96. 2
$2p'' \ ^4S_2$	$2p^3$	$2p^3 \ ^4S^o$	$\frac{1}{2}$	186802. 3		$4p \ ^2P_1$ $4p \ ^2P_2$	$2s^2(1S)4p$	$4p \ ^2P^o$	$\frac{1}{2}$ $\frac{1}{2}$	311691. 3 311716. 1	24. 8
$2p'' \ ^2D_3$ $2p'' \ ^2D_2$	$2p^3$	$2p^3 \ ^2D^o$	$\frac{3}{2}$ $\frac{1}{2}$	203072. 2 203088. 9	-16. 7	$3p' \ ^4S_2$	$2s \ 2p(^3P^o)3p$	$3p \ ^4S$	$\frac{1}{2}$	314224. 0	
$3s \ ^2S_1$	$2s^2(1S)3s$	$3s \ ^2S$	$\frac{1}{2}$	221302. 4		$3p' \ ^4P_1$ $3p' \ ^4P_2$ $3p' \ ^4P_3$	$2s \ 2p(^3P^o)3p$	$3p \ ^4P$	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{3}{2}$	317299. 9 317343. 4 317402. 3	43. 5 58. 9
$2p'' \ ^2P_1$ $2p'' \ ^2P_2$	$2p^3$	$2p^3 \ ^2P^o$	$\frac{1}{2}$ $\frac{1}{2}$	230404. 5 230408. 6	4. 1	$4d \ ^2D_2$ $4d \ ^2D_3$	$2s^2(1S)4d$	$4d \ ^2D$	$\frac{1}{2}$ $\frac{3}{2}$	317750. 8 317781. 8	31. 0
$3p \ ^2P_1$ $3p \ ^2P_2$	$2s^2(1S)3p$	$3p \ ^2P^o$	$\frac{1}{2}$ $\frac{1}{2}$	245665. 7 245701. 7	36. 0		$2s^2(1S)4f$	$4f \ ^2F^o$	$\frac{3}{2}$ $\frac{3}{2}$	320287. 5	
$3d \ ^2D_2$ $3d \ ^2D_3$	$2s^2(1S)3d$	$3d \ ^2D$	$\frac{1}{2}$ $\frac{3}{2}$	267238. 5 267244. 4	5. 9	$4f \ ^2F_4$					

N III—Continued

N III—Continued

Edlén	Config.	Desig.	<i>J</i>	Level	Interval	Edlén	Config.	Desig.	<i>J</i>	Level	Interval
$3p' \ ^2D_2$ $\ ^2D_3$	$2s \ 2p(^3P^\circ)3p$	$3p \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	320977. 4 321065. 8	88. 4	$4p' \ ^2D_2$ $\ ^2D_3$	$2s \ 2p(^3P^\circ)4p$	$4p \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	377883. 7 377970. 8	87. 1
$3p' \ ^2S_1$	$2s \ 2p(^3P^\circ)3p$	$3p \ ^2S$	$\frac{1}{2}$	327056. 8		$4p' \ ^4S_2$	$2s \ 2p(^3P^\circ)4p$	$4p \ ^4S$	$1\frac{1}{2}$	378440. 5	
$3d' \ ^4F_2$ $\ ^4F_3$ $\ ^4F_4$ $\ ^4F_5$	$2s \ 2p(^3P^\circ)3d$	$3d \ ^4F^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	330238. 4 330273. 5 330325. 3 330396. 7	35. 1 51. 8 71. 4	$4p' \ ^4P_1$ $\ ^4P_2$ $\ ^4P_3$	$2s \ 2p(^3P^\circ)4p$	$4p \ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	379307. 3 379352. 1 379405. 0	44. 8 52. 9
$3d' \ ^4D_1$ $\ ^4D_2$ $\ ^4D_3$ $\ ^4D_4$	$2s \ 2p(^3P^\circ)3d$	$3d \ ^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	332796. 6 332810. 0 332832. 0 332860. 3	13. 4 22. 0 28. 3	N IV (1S_0)		<i>Limit</i>	-----	382625. 5	
$5s \ ^2S_1$	$2s^2(^1S)5s$	$5s \ ^2S$	$\frac{1}{2}$	333713. 1		$4d' \ ^4F_3$ $\ ^4F_4$ $\ ^4F_5$	$2s \ 2p(^3P^\circ)4d$	$4d \ ^4F^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	384016 384065 384139	49 74
$3d' \ ^2D_2$ $\ ^2D_3$	$2s \ 2p(^3P^\circ)3d$	$3d \ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	334542. 2 334568. 9	26. 7	$4d' \ ^2D$	$2s \ 2p(^3P^\circ)4d$	$4d \ ^2D^\circ$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	385126	
$3d' \ ^4P_3$ $\ ^4P_2$ $\ ^4P_1$	$2s \ 2p(^3P^\circ)3d$	$3d \ ^4P^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	336213. 4 336268. 0 336303. 1	-54. 6 -35. 1	$4d' \ ^4D_2$ $\ ^4D_3$ $\ ^4D_4$	$2s \ 2p(^3P^\circ)4d$	$4d \ ^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	385296 385323 385352	27 29
$3d' \ ^2F_3$ $\ ^2F_4$	$2s \ 2p(^3P^\circ)3d$	$3d \ ^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	339744. 4 339855. 7	111. 3	$4d' \ ^4P_3$	$2s \ 2p(^3P^\circ)4d$	$4d \ ^4P^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	386246	
$5d \ ^2D_2$ $\ ^2D_3$	$2s^2(^1S)5d$	$5d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	341946. 2 341947. 9	1. 7	$4f' \ ^2F_3$ $\ ^2F_4$	$2s \ 2p(^3P^\circ)4f$	$4f \ ^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	386953. 4 386974	21
$3d' \ ^2P_2$ $\ ^2P_1$	$2s \ 2p(^3P^\circ)3d$	$3d \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	342693. 0 342763. 7	-70. 7	$4f' \ ^4F_3$ $\ ^4F_4$ $\ ^4F_5$	$2s \ 2p(^3P^\circ)4f$	$4f \ ^4F$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	387000. 8 387010. 3 387042. 3	9. 5 32. 0
$5f \ ^2F_4$	$2s^2(^1S)5f$	$5f \ ^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	342752. 0		$4d' \ ^2F_3$ $\ ^2F_4$	$2s \ 2p(^3P^\circ)4d$	$4d \ ^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	387728. 7 387811. 5	82. 8
$5g \ ^2G$	$2s^2(^1S)5g$	$5g \ ^2G$	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 4\frac{1}{2} \end{array} \right\}$	343116		$4f' \ ^4G_3$ $\ ^4G_4$ $\ ^4G_5$ $\ ^4G_6$	$2s \ 2p(^3P^\circ)4f$	$4f \ ^4G$	$2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$ $5\frac{1}{2}$	388039. 2 388082. 9 388134. 8 388198	43. 7 51. 9 63
$6d \ ^2D_3$	$2s^2(^1S)6d$	$6d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	354517		$4f' \ ^2G_4$ $\ ^2G_5$	$2s \ 2p(^3P^\circ)4f$	$4f \ ^2G$	$3\frac{1}{2}$ $4\frac{1}{2}$	388190. 3 388290. 0	99. 7
$6f \ ^2F_4$	$2s^2(^1S)6f$	$6f \ ^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	354955. 7		$4f' \ ^4D_4$ $\ ^4D_3$ $\ ^4D_2$ $\ ^4D_1$	$2s \ 2p(^3P^\circ)4f$	$4f \ ^4D$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	388273. 4 388310. 9 388359. 2 388386. 6	-37. 5 -48. 3 -27. 4
$6g \ ^2G$	$2s^2(^1S)6g$	$6g \ ^2G$	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 4\frac{1}{2} \end{array} \right\}$	355214		$4f' \ ^2D_3$ $\ ^2D_2$	$2s \ 2p(^3P^\circ)4f$	$4f \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	388376. 9 388442. 4	-65. 5
$4s' \ ^4P_1$ $\ ^4P_2$ $\ ^4P_3$	$2s \ 2p(^3P^\circ)4s$	$4s \ ^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	368525. 6 368538. 3 368704. 8	62. 7 116. 5	$\overline{3d'} \ ^2D_2$ $\ ^2D_3$	$2s \ 2p(^1P^\circ)3d$	$3d' \ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	396574. 9 396584. 8	9. 9
$\overline{3p'} \ ^2D_2$ $\ ^2D_3$	$2s \ 2p(^1P^\circ)3p$	$3p' \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	373342 373376	34	$5d' \ ^4D_4$	$2s \ 2p(^3P^\circ)5d$	$5d \ ^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	409017	
$4p' \ ^2P_1$ $\ ^2P_2$	$2s \ 2p(^3P^\circ)4p$	$4p \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	374747. 4 374805. 3	57. 9						
$4p' \ ^4D_1$ $\ ^4D_2$ $\ ^4D_3$ $\ ^4D_4$	$2s \ 2p(^3P^\circ)4p$	$4p \ ^4D$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	376756. 6 376803. 3 376863. 8 376953. 3	46. 7 60. 5 89. 5						
$\overline{3p'} \ ^2P_1$ $\ ^2P_2$	$2s \ 2p(^1P^\circ)3p$	$3p' \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	377591 377608	17						

June 1946.

N III OBSERVED TERMS*

Config. $1s^2 +$	Observed Terms		
$2s^2 ({}^1S)2p$	$2p {}^2P^\circ$		
$2s 2p^2$	{ $2p^2 {}^2S$ $2p^2 {}^4P$ $2p^2 {}^2P$ $2p^2 {}^2D$		
$2p^3$	{ $2p^3 {}^4S^\circ$ $2p^3 {}^2P^\circ$ $2p^3 {}^2D^\circ$		
	$ns (n \geq 3)$	$np (n \geq 3)$	$nd (n \geq 3)$
$2s^2 ({}^1S)nx$	$3-5s {}^2S$	$3, 4p {}^2P^\circ$	$3-6d {}^2D$
$2s 2p({}^3P^\circ)nx$	{ $3, 4s {}^4P^\circ$ $3s {}^2P^\circ$	$3, 4p {}^4S$ $3, 4p {}^4P$ $3, 4p {}^4D$	$3, 4d {}^4P^\circ$ $3-5d {}^4D^\circ$ $3, 4d {}^4F^\circ$ $3d {}^2P^\circ$ $3, 4d {}^2D^\circ$ $3, 4d {}^2F^\circ$
$2s 2p({}^1P^\circ)nx'$		$3p' {}^2P$ $3p' {}^2D$	
	$nf (n \geq 4)$	$ng (n \geq 5)$	
$2s^2 ({}^1S)nx$	$4-6f {}^2F^\circ$	$5, 6g {}^2G$	
$2s 2p({}^3P^\circ)nx$	{ $4f {}^4D$ $4f {}^4F$ $4f {}^4G$ $4f {}^2D$ $4f {}^2F$ $4f {}^2G$		

*For predicted terms in the spectra of the B I isoelectronic sequence, see Introduction.

N IV

(Be I sequence; 4 electrons)

$Z=7$

Ground state $1s^2 2s^2 {}^1S_0$

$2s^2 {}^1S_0$ 624851 cm^{-1}

I. P. 77.450 volts

The terms are from Edlén's papers. The absolute values of the singlet terms are uncertain, since only two members of the 1D -series have been observed. No intersystem combinations have been found. By analogy with N III, Edlén (1936) estimates that $2s^2 {}^1S_0 - 2p {}^3P_1 = 67200 \text{ cm}^{-1}$, which gives the absolute value of $2s^2 {}^1S_0$ as 624851 cm^{-1} instead of the earlier value 624499 cm^{-1} . The relative uncertainty x , therefore probably does not exceed $\pm 300 \text{ cm}^{-1}$.

The terms $4p {}^3P^\circ$, $4f {}^3F^\circ$, $5g {}^3G$, and $3d {}^3F^\circ$ are from the 1936 reference. Edlén obtains the $4f {}^3F^\circ$ term by assuming that $5g {}^3G$ is hydrogen-like (absolute value 70500 cm^{-1}) and adopting Freeman's identification of the $4f {}^3F^\circ - 5g {}^3G$ group of lines. The listed value of $5g {}^3G$ has been adjusted to fit Edlén's adopted value of $4f {}^3F^\circ$.

The estimated value of $3d {}^3F^\circ$ is included in the table in brackets.

REFERENCES

- L. J. Freeman, Proc. Roy. Soc. (London) [A] **127**, 330 (1930). (T) (C L)
 B. Edlén, Nova Acta Reg. Soc. Sci. Uppsala [IV] **9**, No. 6, 62 (1934). (T) (C L)
 B. Edlén, Zeit. Phys. **98**, 561 (1936). (I P) (C L)

N IV

N IV

Edlén	Config.	Desig.	<i>J</i>	Level	Interval	Edlén	Config.	Desig.	<i>J</i>	Level	Interval
2s ¹ S ₀	2s ²	2s ² ¹ S	0	0		3d' ³ F	2p(² P°)3d	3d ³ F°	2, 3, 4	[499851] +x	
2p ³ P ₀	2s(² S)2p	2p ³ P°	0	67136.4+x	63.2	4p ³ P	2s(² S)4p	4p ³ P°	0, 1, 2	503625 +x	
³ P ₁			1	67199.6+x	144.2	3d' ³ D ₁	2p(² P°)3d	3d ³ D°	1	505487 +x	31
³ P ₂			2	67343.8+x		³ D ₂			2	505518 +x	43
						³ D ₃			3	505561 +x	
2p ¹ P ₁	2s(² S)2p	2p ¹ P°	1	130695		3d' ¹ F ₃	2p(² P°)3d	3d ¹ F°	3	506292	
2p' ³ P ₀	2p ²	2p ² ³ P	0	175463.5+x	73.2	4p ¹ P ₁	2s(² S)4p	4p ¹ P°	1	507022	
³ P ₁			1	175536.7+x	124.8		2s(² S)4d	4d ³ D	1		
³ P ₂			2	175661.5+x					2		
2p' ¹ D ₂	2p ²	2p ² ¹ D	2	188885		4d ³ D ₃			3	511384 +x	
2p' ¹ S ₀	2p ²	2p ² ¹ S	0	235370		3d' ³ P ₂	2p(² P°)3d	3d ³ P°	2	511440 +x	-53
3s ³ S ₁	2s(² S)3s	3s ³ S	1	377206+x		³ P ₁			1	511493 +x	
3s ¹ S ₀	2s(² S)3s	3s ¹ S	0	388858					0		
3p ¹ P ₁	2s(² S)3p	3p ¹ P°	1	404521		4d ¹ D ₂	2s(² S)4d	4d ¹ D	2	514638	
3p ³ P ₀	2s(² S)3p	3p ³ P°	0	405893.2+x	15.8	4f ³ F ₂	2s(² S)4f	4f ³ F°	2	516631 +x	8
³ P ₁			1	405909.0+x	35.4	³ F ₃			3	516639 +x	11
³ P ₂			2	405944.4+x		³ F ₄			4	516650 +x	
3d ³ D ₁	2s(² S)3d	3d ³ D	1	419967.8+x	3.5	3d' ¹ P ₁	2p(² P°)3d	3d ¹ P°	1	519414	
³ D ₂			2	419971.3+x	8.1	4f ¹ F ₃	2s(² S)4f	4f ¹ F°	3	521868	
³ D ₃			3	419979.4+x		5p ¹ P ₁	2s(² S)5p	5p ¹ P°	1	550218	
3d ¹ D ₂	2s(² S)3d	3d ¹ D	2	429158			2s(² S)5d	5d ³ D	1		
3s' ³ P ₀	2p(² P°)3s	3s ³ P°	0	465223.0+x	77.6	5d ³ D ₃			2		
³ P ₁			1	465300.6+x	162.8				3	552731 +x	
³ P ₂			2	465463.4+x		5g ³ G	2s(² S)5g	5g ³ G	3, 4, 5	554419 +x	
3s' ¹ P ₁	2p(² P°)3s	3s ¹ P°	1	473032			2s(² S)6d	6d ³ D	1		
3p' ¹ P ₁	2p(² P°)3p	3p ¹ P	1	480880		6d ³ D ₃			2	574940 +x	
	2p(² P°)3p	3p ³ D	1						3	591043	
3p' ³ D ₂			2	484394 +x	131	4p' ¹ D ₂	2p(² P°)4p	4p ¹ D	2		
³ D ₃			3	484525 +x		4d' ³ D _{1,2}	2p(² P°)4d	4d ³ D°	1, 2	593665 +x	39
3p' ³ S ₁	2p(² P°)3p	3p ³ S	1	487542 +x		³ D ₃			3	593704 +x	
	2p(² P°)3p	3p ³ P	0				N v (² S _{1/2})	Limit	-----	624851	
3p' ³ P ₁			1	494240 +x	98		2p(² P°)5d	5d ³ D°	1		
³ P ₂			2	494338 +x					2		
3d' ¹ D ₂	2p(² P°)3d	3d ¹ D°	2	498315		5d' ³ D ₃			3	634198 +x	
3p' ¹ D ₂	2p(² P°)3p	3p ¹ D	2	499708							

May 1946.

N IV OBSERVED TERMS*

Config. 1s ² +	Observed Terms				
2s ²	2s ² 1S				
2s(2S)2p	{ 2p 3P° 2p 1P°				
2p ²	{ 2p ² 1S 2p ² 3P 2p ² 1D				
	ns (n≥3)	np (n≥3)		nd (n≥3)	nf (n≥4) ng (n≥5)
2s(2S)nx	{ 3s 3S 3s 1S	3, 4p 3P° 3-5p 1P°		3-6d 3D 3, 4d 1D	4f 3F° 4f 1F° 5g 3G
2p(2P°)nx	{ 3s 3P° 3s 1P°	3p 3S	3p 3P 3p 3D 3p 1P 3, 4p 1D	3d 3P° 3-5d 3D° 3d 1P° 3d 1D° 3d 1F°	

*For predicted terms in the spectra of the Be I isoelectronic sequence, see Introduction.

N V

(Li I sequence; 3 electrons)

$Z=7$

Ground state $1s^2\ 2s\ ^2S_{1/2}$

$2s\ ^2S_{1/2}\ 789532.9\ \text{cm}^{-1}$

I. P. 97.863 volts

Both Edlén and Cady have published analyses of this spectrum. Edlén has recently extended the earlier work and has generously furnished his revised term list in manuscript form. The observed term values in the table are from this unpublished list.

Edlén's extrapolated intervals and the term values for higher series members based on his calculations from the series formula are entered in brackets in the table. These have been taken from his 1933 and 1934 papers.

REFERENCES

- W. Cady, Phys. Rev. **44**, 821 (1933). (T) (C L)
 B. Edlén, Zeit. Astroph. **7**, 378 (1933). (T) (C L)
 B. Edlén, Nova Acta Reg. Soc. Sci. Uppsala [IV] **9**, No. 6, 41 (1934). (T) (C L)
 B. Edlén, unpublished material (Sept. 1947). (I P) (T)

N v

N v

Edlén	Config.	Desig.	<i>J</i>	Level	Interval	Edlén	Config.	Desig.	<i>J</i>	Level	Interval
2s ² S ₁	2s	2s ² S	$\frac{1}{2}$	0. 0		6GH	6g, 6h	6g ² G, etc.	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ \text{to} \\ 5\frac{1}{2} \end{smallmatrix} \right\}$	[713335]	
2p ² P ₁ ² P ₂	2p	2p ² P°	$\frac{1}{2}$ $1\frac{1}{2}$	80464. 9 80723. 3	258. 4	7S	7s	7s ² S	$\frac{1}{2}$	[731432]	
3s ² S ₁	3s	3s ² S	$\frac{1}{2}$	456134		7P	7p	7p ² P°	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	732993	
3p ² P ₁ ³ P ₂	3p	3p ² P°	$\frac{1}{2}$ $1\frac{1}{2}$	477777. 2 477851. 4	74. 2	7D	7d	7d ² D	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	[733516]	
3d ² D ₂ ² D ₃	3d	3d ² D	$1\frac{1}{2}$ $2\frac{1}{2}$	484403 484427	[24]	7F	7f	7f ² F°	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	[733547]	
4s ² S ₁	4s	4s ² S	$\frac{1}{2}$	606337		7GHI	7g, etc.	7g ² G, etc.	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ \text{to} \\ 6\frac{1}{2} \end{smallmatrix} \right\}$	[733552]	
4p ² P ₂	4p	4p ² P°	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	615150	[32]	8S	8s	8s ² S	$\frac{1}{2}$	[745260]	
4d ² D ₃	4d	4d ² D	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	617905	[10]	8P	8p	8p ² P°	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	[746311]	
	5s	5s ² S	$\frac{1}{2}$	673882		8D	8d	8d ² D	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	[746649]	
5p ² P ₂	5p	5p ² P°	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	678297	[16]	8F	8f	8f ² F°	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	[746670]	
5d ² D ₃	5d	5d ² D	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	679725	[5]	8GHIK	8g, etc.	8g ² G, etc.	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ \text{to} \\ 7\frac{1}{2} \end{smallmatrix} \right\}$	[746674]	
6S	6s	6s ² S	$\frac{1}{2}$	[709947]							
6p ² P	6p	6p ² P°	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	712464							
6d ² D	6d	6d ² D	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	713289							
6F	6f	6f ² F°	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	[713327]							
							N VI (¹ S ₀)	Limit	-----	789532. 9	

September 1947.

N VI

(He I sequence; 2 electrons)

Z = 7Ground state 1s² ¹S₀1s² ¹S₀ 4452800 ± 500cm⁻¹.

I. P. 551.925 ± 0.062 volts

Tyrén has observed the first three members of the singlet series. They are in the region from 23 Å to 28 Å. He lists also one intersystem combination—a line at 29.084 Å classified as 1s² ¹S₀—2p ³P₁. His unit, 10³ cm⁻¹, has here been changed to cm⁻¹.

Edlén has generously furnished his unpublished manuscript containing absolute values of the triplet terms extrapolated along the He I isoelectronic sequence. The relative positions of the singlet and triplet terms thus determined confirm the intersystem combination reported by Tyrén. The 2s ³S—2p ³P° combination has apparently not been observed, but Edlén regards the extrapolation from the irregular doublet law as very reliable. Brackets are used in the table to indicate extrapolated values not yet confirmed by observation.

REFERENCES

- F. Tyrén, Nova Acta Reg. Soc. Sci Uppsala [IV] 12, No. 1, 24 (1940). (I P) (T) (C L)
B. Edlén, unpublished material (Sept. 1947). (T)

N VI

N VI

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
1s ²	1s ² ¹ S	0	0		1s 3p	3p ¹ P ^o	1	4016390	
1s 2s	2s ² S	1	[3385890]		1s 4p	4p ¹ P ^o	1	4206810	
1s 2p	2p ³ P ^o	0	[3438270]	[10] [290]					
		1	3438280						
		2	[3438570]		N VII (² S _{1/2})	Limit		4452800	
1s 2p	2p ¹ P ^o	1	3473790						

September 1947.

N VII

(H sequence; 1 electron)

Z=7

Ground state 1s ²S_{1/2}

1s ²S_{1/2} 5379860 cm⁻¹

I. P. 666.83 volts

The first Lyman line has been observed by Tyrén. J. E. Mack has calculated the terms in the table, “using $R_N^{14}=109733.004$ and $\Delta=0.040$. The series limit of N¹⁵ is higher by 14.0 cm⁻¹ than the value given here.”

REFERENCES

F. Tyrén, Nova Acta Reg. Soc. Sci. Uppsala [IV] 12, No. 1, 24 (1940). (C L)
J. E. Mack, unpublished material (1949). (I P) (T) (C L)

N VII

Config.	Desig.	<i>J</i>	Level	Interval
1s	1s ² S	½	0	
2p	2p ² P ^o	½	4034535]] 70 876. 9
2s	2s ² S	½	4034605	
2p	2p ² P ^o	1½	4035412	
3s, etc.	3s ² S, etc.	½, etc.	4782035 to 381	
4s, etc.	4s ² S, etc.	½, etc.	5043625 to 789	
	∞=Limit	-----	5379860	

February 1949.

OXYGEN

O I

8 electrons

 $Z=8$ Ground state $1s^2 2s^2 2p^4 {}^3P_2$ $2p^4 {}^3P_2$ 109836.7 cm^{-1}

I. P. 13.614 volts

Edlén has published a detailed analysis of this spectrum in which he has revised and extended the earlier work by others. The terms have all been taken from his paper. For the higher series members not included in his main term table, $ns {}^5S^\circ$ and $ns {}^3S^\circ$ ($n=8$ to 11), and $nd {}^5D^\circ$ and $nd {}^3D^\circ$ ($n=8$ to 10) the observed values taken from his discussion of the series formulas (p. 15), in which he compares observed and calculated values, are listed below.

Two terms not derived from observed lines are entered in brackets: $11s {}^5S^\circ$, which is calculated from the series formula and $2s 2p^5 {}^1P^\circ$, which is extrapolated.

Intersystem combinations connect the terms of the singlet, triplet, and quintet systems.

Kiess and Shortley have observed g values for four levels as follows:

Desig.	Obs. g
$3s {}^5S_2^\circ$	1.999
$3p {}^5P_1$	2.506
5P_2	1.836
5P_3	1.666

REFERENCES

- A. Fowler, *Report on Series in Line Spectra* p. 166 (Fleetway Press, London, 1922). (T) (C L)
 R. Frerichs, *Phys. Rev.* **34**, 1239 (1929); **36**, 398 (1930). (T) (C L)
 H. E. White, *Introduction to Atomic Spectra* p. 266 (McGraw-Hill Book Co., Inc., New York, N. Y., 1934). (G D)
 K. R. More and C. A. Rieke, *Phys. Rev.* **50**, 1054 (1936). (Standard wavelengths)
 B. Edlén, *Kungl. Svenska Vetenskapsakad. Handl.* [3] **20**, No. 10, 31 pp. (1943). (I P) (T) (C L)
 W. F. Meggers, *J. Opt. Soc. Am.* **36**, 431 (1946). (Summary hfs)
 C. C. Kiess and G. Shortley, *J. Research Nat. Bur. Std.* **42**, 190, RP1961 (1949). (Z E)

O I

O I

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$2s^2\ 2p^4$	$2p^4\ ^3P$	2	0. 0	-158. 5 -68. 0	$2s^2\ 2p^3(^4S^\circ)4s$	$4s\ ^3S^\circ$	1	96225. 5	-0. 13 -0. 13
		1	158. 5		$2s^2\ 2p^3(^4S^\circ)3d$	$3d\ ^5D^\circ$	4	97420. 24	
		0	226. 5			3, 2 2, 1, 0	97420. 37 97420. 50		
$2s^2\ 2p^4$	$2p^4\ ^1D$	2	15867. 7						
$2s^2\ 2p^4$	$2p^4\ ^1S$	0	33792. 4		$2s^2\ 2p^3(^4S^\circ)3d$	$3d\ ^3D^\circ$	3, 2, 1	97488. 14	
$2s^2\ 2p^3(^4S^\circ)3s$	$3s\ ^5S^\circ$	2	73767. 81		$2s^2\ 2p^3(^4S^\circ)4p$	$4p\ ^5P$	1	99092. 64	0. 67 1. 21
$2s^2\ 2p^3(^4S^\circ)3s$	$3s\ ^3S^\circ$	1	76794. 69				2 3	99093. 31 99094. 52	
$2s^2\ 2p^3(^4S^\circ)3p$	$3p\ ^5P$	1 2 3	86625. 35 86627. 37 86631. 04	2. 02 3. 67	$2s^2\ 2p^3(^4S^\circ)4p$	$4p\ ^3P$	2, 1, 0	99680. 4	
$2s^2\ 2p^3(^4S^\circ)3p$	$3p\ ^3P$	2 1 0	88630. 84 88630. 30 88631. 00	0. 54 -0. 70	$2s^2\ 2p^3(^2D^\circ)3s$	$3s'\ ^3D^\circ$	3 2 1	101135. 04 101147. 21 101155. 10	-12. 17 -7. 89
$2s^2\ 2p^3(^4S^\circ)4s$	$4s\ ^5S^\circ$	2	95476. 43		$2s^2\ 2p^3(^4S^\circ)5s$	$5s\ ^5S^\circ$	2	102116. 21	
					$2s^2\ 2p^3(^4S^\circ)5s$	$5s\ ^3S^\circ$	1	102411. 65	

O I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$2s^2 2p^3(^2D^\circ)3s$	$3s' ^1D^\circ$	2	102661. 63		$2s^2 2p^3(^2D^\circ)3p$	$3p' ^1D$	2	116630. 51	
$2s^2 2p^3(^4S^\circ)4d$	$4d ^5D^\circ$	4 3 2 1 0	102865. 09		$2s^2 2p^3(^2D^\circ)4s$	$4s' ^1D^\circ$	2	122798. 7	
					$2s^2 2p^3(^2D^\circ)3d$	$3d' ^3P^\circ$	2 1 0	123296. 6 123355. 2 123386. 9	-58. 6 -31. 7
$2s^2 2p^3(^4S^\circ)4d$	$4d ^3D^\circ$	3, 2, 1	102908. 14		$2s^2 2p^3(^2D^\circ)3d$	$3d' ^3F^\circ$	4 3 2	124213. 18	
$2s^2 2p^3(^4S^\circ)5p$	$5p ^3P$	2, 1, 0	103869. 4						
$2s^2 2p^3(^4S^\circ)6s$	$6s ^5S^\circ$	2	105019. 0		$2s^2 2p^3(^2D^\circ)3d$	$3d' ^1G^\circ$	4	124238. 21	
$2s^2 2p^3(^4S^\circ)6s$	$6s ^3S^\circ$	1	105164. 90		$2s^2 2p^3(^2D^\circ)3d$	$3d' ^3G^\circ$	5 4 3	124239. 66 124258. 37 124252. 52	-18. 71 5. 85
$2s^2 2p^3(^4S^\circ)5d$	$5d ^5D^\circ$	4 to 0	105385. 3						
$2s^2 2p^3(^4S^\circ)5d$	$5d ^3D^\circ$	3, 2, 1	105408. 58		$2s^2 2p^3(^2D^\circ)3d$	$3d' ^1F^\circ$	3	124326. 32	
$2s^2 2p^3(^4S^\circ)6p$	$6p ^3P$	2, 1, 0	105911. 3		$2s^2 2p^3(^2D^\circ)4p$	$4p' ^3D$	3 2 1	125774. 51 125782. 09 125787. 14	-7. 58 -5. 05
$2s^2 2p^3(^4S^\circ)7s$	$7s ^5S^\circ$	2	106545. 1						
$2s^2 2p^3(^4S^\circ)7s$	$7s ^3S^\circ$	1	106627. 9		$2s 2p^5$	$2p^5 ^3P^\circ$	2 1 0	126266. 48 126339. 92 126383. 44	-73. 44 -43. 52
$2s^2 2p^3(^4S^\circ)6d$	$6d ^5D^\circ$	4 to 0	106751. 2						
$2s^2 2p^3(^4S^\circ)6d$	$6d ^3D^\circ$	3, 2, 1	106765. 8		$2s^2 2p^3(^2P^\circ)3p$	$3p'' ^3D$	3 2 1	127281. 85 127287. 62 127290. 93	-5. 77 -3. 31
$2s^2 2p^3(^4S^\circ)8s$	$8s ^5S^\circ$	2	107445. 4						
$2s^2 2p^3(^4S^\circ)8s$	$8s ^3S^\circ$	1	107497. 1		$2s^2 2p^3(^2P^\circ)3p$	$3p'' ^1P$	1	127667. 85	
$2s^2 2p^3(^4S^\circ)7d$	$7d ^5D^\circ$	4 to 0	107573. 1		$2s^2 2p^3(^2P^\circ)3p$	$3p'' ^1D$	2	128595. 02	
$2s^2 2p^3(^4S^\circ)7d$	$7d ^3D^\circ$	3, 2, 1	107582. 7		$2s^2 2p^3(^2D^\circ)5s$	$5s' ^1D^\circ$	2	129134. \pm	
$2s^2 2p^3(^4S^\circ)9s$	$9s ^5S^\circ$	2	108021. 4		$2s^2 2p^3(^2D^\circ)4d$	$4d' ^3F^\circ$	4 3 2	129666. 55	
$2s^2 2p^3(^4S^\circ)9s$	$9s ^3S^\circ$	1	108057. 6						
$2s^2 2p^3(^4S^\circ)8d$	$8d ^5D^\circ$	4 to 0	108105. 7		$2s^2 2p^3(^2D^\circ)4d$	$4d' ^1G^\circ$	4	129679. 49	
$2s^2 2p^3(^4S^\circ)8d$	$8d ^3D^\circ$	3, 2, 1	108116. 6		$2s^2 2p^3(^2D^\circ)4d$	$4d' ^3G^\circ$	5 4 3	129680. 14 129699. 16 129693. 08	-19. 02 6. 08
$2s^2 2p^3(^4S^\circ)10s$	$10s ^5S^\circ$	2	108412. 0						
$2s^2 2p^3(^4S^\circ)10s$	$10s ^3S^\circ$	1	108436. 1		$2s^2 2p^3(^2D^\circ)4d$	$4d' ^1F^\circ$	3	129736. 60	
$2s^2 2p^3(^4S^\circ)9d$	$9d ^5D^\circ$	4 to 0	108470. 2		$2s^2 2p^3(^2D^\circ)4d$	$4d' ^3P^\circ$	2 1 0	129969. 60 129979. 04 129984. 15	-9. 44 -5. 11
$2s^2 2p^3(^4S^\circ)9d$	$9d ^3D^\circ$	3, 2, 1	108477. 8						
$2s^2 2p^3(^4S^\circ)11s$	$11s ^5S^\circ$	2	[108688. 4]		$2s^2 2p^3(^2P^\circ)3p$	$3p'' ^1S$	0	130943. 21	
$2s^2 2p^3(^4S^\circ)11s$	$11s ^3S^\circ$	1	108707. 3		$2s^2 2p^3(^2D^\circ)6s$	$6s' ^1D^\circ$	2	131927. \pm	
$2s^2 2p^3(^4S^\circ)10d$	$10d ^5D^\circ$	4 to 0	108731. 5		$2s^2 2p^3(^2D^\circ)5d$	$5d' ^3F^\circ$	4 3 2	132190. 7 \pm	
$2s^2 2p^3(^4S^\circ)10d$	$10d ^3D^\circ$	3, 2, 1	108734. 4						
O II ($^4S_{1/2}$)	Limit	-----	109836. 7		$2s^2 2p^3(^2D^\circ)5d$	$5d' ^1G^\circ$	4	132197. 6 \pm	
$2s^2 2p^3(^2D^\circ)3p$	$3p' ^3D$	3 2 1	113294. 42 113294. 55 113298. 01	-0. 13 -3. 46	$2s^2 2p^3(^2D^\circ)5d$	$5d' ^3G^\circ$	5 4 3	132198. 1 132217. 8	-19. 7
$2s^2 2p^3(^2D^\circ)3p$	$3p' ^3F$	4 3 2	113714. 06 113721. 06 113726. 81	-7. 00 -5. 75					
$2s^2 2p^3(^2P^\circ)3s$	$3s'' ^3P^\circ$	2 1 0	113910. 20 113920. 63 113926. 80	-10. 43 -6. 17	$2s^2 2p^3(^2D^\circ)5d$	$5d' ^3P^\circ$	2, 1 0	132310. \pm	
$2s^2 2p^3(^2D^\circ)3p$	$3p' ^1F$	3	113995. 81		$2s^2 2p^3(^2D^\circ)7s$	$7s' ^1D^\circ$	2	133413. \pm	
$2s^2 2p^3(^2P^\circ)3s$	$3s'' ^1P^\circ$	1	115918. 30		$2s^2 2p^3(^2D^\circ)6d$	$6d' ^3P^\circ$	2, 1 0	133618. \pm	
					$2s 2p^5$	$2p^5 ^1P^\circ$	1	[189837]	

O I OBSERVED TERMS*

Config. $1s^2 +$	Observed Terms		
$2s^2 2p^4$	{ $2p^4 \ ^1S$ $2p^4 \ ^3P$ $2p^4 \ ^1D$		
$2s 2p^5$	$2p^5 \ ^3P^\circ$		
	$ns \ (n \geq 3)$	$np \ (n \geq 3)$	$nd \ (n \geq 3)$
$2s^2 2p^3(^4S^\circ)nx$	{ $3-10s \ ^5S^\circ$ $3-11s \ ^3S^\circ$	$3, 4p \ ^5P$ $3-6p \ ^3P$	$3-10d \ ^5D^\circ$ $3-10d \ ^3D^\circ$
$2s^2 2p^3(^2D^\circ)nx'$	{ $3s' \ ^3D^\circ$ $3-7s' \ ^1D^\circ$	$3, 4p' \ ^3D$ $3p' \ ^3F$ $3p' \ ^1D$ $3p' \ ^1F$	$3-6d' \ ^3P^\circ$ $3-5d' \ ^3F^\circ$ $3-5d' \ ^3G^\circ$ $3, 4d' \ ^1F^\circ$ $3-5d' \ ^1G^\circ$
$2s^2 2p^3(^2P^\circ)nx''$	{ $3s'' \ ^3P^\circ$ $3s'' \ ^1P^\circ$	$3p'' \ ^1S$ $3p'' \ ^1P$ $3p'' \ ^3D$ $3p'' \ ^1D$	

*For predicted terms in the spectra of the O I isoelectronic sequence, see Introduction.

O II

(N I sequence; 7 electrons)

$Z=8$

Ground state $1s^2 2s^2 2p^3 \ ^4S_{1\frac{1}{2}}$

$2p^3 \ ^4S_{1\frac{1}{2}} \ 283550.9 \text{ cm}^{-1}$

I. P. 35.146 volts

The terms are from Edlén's publications. He has summarized the earlier work on analysis by others and extended it by his observations in the far ultraviolet.

Edlén states that a number of the $5f$ -terms are very uncertain. These are followed by a "?" in the table. His estimated values of three terms from the (1S) limit in O III are given in brackets.

Mihul lists the observed Zeeman effects for 111 lines, which in general agree well with the theoretical patterns for the adopted classifications. From his data a number of g -values could be calculated, but many of the observed patterns are unresolved.

Although the analysis of O II is fairly complete, the measures by different observers are discordant. The term values could be greatly improved by a set of homogeneous observations. A monograph containing all classified lines of this spectrum is also needed.

The doublet and quartet terms are connected by intersystem combinations, but the sextet terms are not so connected with the rest. The relative uncertainty, x , may be a few hundred cm^{-1} .

REFERENCES

- I. S. Bowen, Phys. Rev. **29**, 231 (1927). (T) (C L)
 A. Fowler, Proc. Roy. Soc. (London) [A] **110**, 476 (1926). (T) (C L)
 C. Mihul, Ann. de Phys. [10] **9**, 294 (1928). (T) (C L) (Z E)
 H. N. Russell, Phys. Rev. **31**, 27 (1928). (T) (C L)
 B. Edlén, Nova Acta Reg. Soc. Sci. Uppsala [IV] **9**, No. 6, 136 (1934). (I P) (T) (C L) (G D)
 B. Edlén, Zeit. Phys. **93**, 728 (1935). (T) (C L)

O II

O II

Edlén	Config.	Desig.	<i>J</i>	Level	Interval	Edlén	Config.	Desig.	<i>J</i>	Level	Interval
$2p\ ^4S_2$	$2s^2\ 2p^3$	$2p^3\ ^4S^\circ$	$1\frac{1}{2}$	0. 0		$3d\ ^2P_2$	$2s^2\ 2p^2(^3P)3d$	$3d\ ^2P$	$1\frac{1}{2}$	233430. 10	-113.99
$2p\ ^2D_3$	$2s^2\ 2p^3$	$2p^3\ ^2D^\circ$	$2\frac{1}{2}$	26808. 4	-21. 0	2P_1			$\frac{1}{2}$	233544. 09	
2D_2			$1\frac{1}{2}$	26829. 4		$3d\ ^2D_2$	$2s^2\ 2p^2(^3P)3d$	$3d\ ^2D$	$1\frac{1}{2}$	234402. 48	51. 97
$2p\ ^2P_2$	$2s^2\ 2p^3$	$2p^3\ ^2P^\circ$	$1\frac{1}{2}$	40466. 9	-1. 5	2D_3			$2\frac{1}{2}$	234454. 45	
2P_1			$\frac{1}{2}$	40468. 4		$4s\ ^4P_1$	$2s^2\ 2p^2(^3P)4s$	$4s\ ^4P$	$\frac{1}{2}$	238626. 32	105. 22
$2p'\ ^4P_3$	$2s\ 2p^4$	$2p^4\ ^4P$	$2\frac{1}{2}$	119837. 7	-163. 4	4P_2			$1\frac{1}{2}$	238731. 54	
4P_2			$1\frac{1}{2}$	120001. 1		4P_3			$2\frac{1}{2}$	238892. 96	161. 42
4P_1			$\frac{1}{2}$	120083. 5	-82. 4	$4s\ ^2P_1$	$2s^2\ 2p^2(^3P)4s$	$4s\ ^2P$	$\frac{1}{2}$	240328. 75	
$2p'\ ^2D_3$	$2s\ 2p^4$	$2p^4\ ^2D$	$2\frac{1}{2}$	165987. 7	-8. 3	2P_2			$1\frac{1}{2}$	240516. 28	
2D_2			$1\frac{1}{2}$	165996. 0		$3s'\ ^6S_3$	$2s\ 2p^3(^6S^\circ)3s$	$3s'''\ ^6S^\circ$	$2\frac{1}{2}$	245395. 5 + <i>x</i>	
$3s\ ^4P_1$	$2s^2\ 2p^2(^3P)3s$	$3s\ ^4P$	$\frac{1}{2}$	185235. 36	105. 32	$4p\ ^4D_1$	$2s^2\ 2p^2(^3P)4p$	$4p\ ^4D^\circ$	$\frac{1}{2}$	245767. 80	48. 49
4P_2			$1\frac{1}{2}$	185340. 68		4D_2			$1\frac{1}{2}$	245816. 29	
4P_3			$2\frac{1}{2}$	185499. 20	158. 52	4D_3			$2\frac{1}{2}$	245902. 85	126. 10
$3s\ ^2P_1$	$2s^2\ 2p^2(^3P)3s$	$3s\ ^2P$	$\frac{1}{2}$	188888. 38		4D_4			$3\frac{1}{2}$	246028. 95	
2P_2			$1\frac{1}{2}$	189068. 37	179. 99	$4p\ ^2D_2$	$2s^2\ 2p^2(^3P)4p$	$4p\ ^2D^\circ$	$1\frac{1}{2}$	248009. 1	176. 2
$2p'\ ^2S_1$	$2s\ 2p^4$	$2p^4\ ^2S$	$\frac{1}{2}$	195710. 4		2D_3			$2\frac{1}{2}$	248185. 3	
$3p\ ^2S_1$	$2s^2\ 2p^2(^3P)3p$	$3p\ ^2S^\circ$	$\frac{1}{2}$	203942. 21	55. 54	$4p\ ^2P_1$	$2s^2\ 2p^2(^3P)4p$	$4p\ ^2P^\circ$	$\frac{1}{2}$	248425. 35	88. 88
$3p\ ^4D_1$	$2s^2\ 2p^2(^3P)3p$	$3p\ ^4D^\circ$	$\frac{1}{2}$	206730. 80		2P_2			$1\frac{1}{2}$	248514. 23	
4D_2			$1\frac{1}{2}$	206786. 34	91. 56		$2s^2\ 2p^2(^1S)3p$	$3p''\ ^2P^\circ$	$\frac{1}{2}$	[250251]	
4D_3			$2\frac{1}{2}$	206877. 90		$3\bar{d}\ ^2F_4$	$2s^2\ 2p^2(^1D)3d$	$3d'\ ^2F$	$3\frac{1}{2}$		
4D_4			$3\frac{1}{2}$	207002. 52	124. 62	2F_3			$2\frac{1}{2}$	251224. 1	
$3s\ ^2D_3$	$2s^2\ 2p^2(^1D)3s$	$3s'\ ^2D$	$2\frac{1}{2}$	206971. 3		-1. 0	$3\bar{d}\ ^2G_5$	$2s^2\ 2p^2(^1D)3d$	$3d'\ ^2G$	$4\frac{1}{2}$	252607. 7
2D_2			$1\frac{1}{2}$	206972. 3	2G_4				$3\frac{1}{2}$	252608. 9	
$3p\ ^4P_1$	$2s^2\ 2p^2(^3P)3p$	$3p\ ^4P^\circ$	$\frac{1}{2}$	208346. 17	46. 10	$3\bar{d}\ ^2D_2$	$2s^2\ 2p^2(^1D)3d$	$3d'\ ^2D$	$1\frac{1}{2}$	253046. 23	2. 12
4P_2			$1\frac{1}{2}$	208392. 27		2D_3			$2\frac{1}{2}$	253048. 35	
4P_3			$2\frac{1}{2}$	208484. 24	91. 97	$3\bar{d}\ ^2P_1$	$2s^2\ 2p^2(^1D)3d$	$3d'\ ^2P$	$\frac{1}{2}$	253789. 51	2. 36
$3p\ ^2D_2$	$2s^2\ 2p^2(^3P)3p$	$3p\ ^2D^\circ$	$1\frac{1}{2}$	211521. 98		2P_2			$1\frac{1}{2}$	253791. 87	
2D_3			$2\frac{1}{2}$	211712. 66	190. 68		$2s^2\ 2p^2(^3P)4d$	$4d\ ^4F$	$1\frac{1}{2}$	254481. 5	109. 2
$3p\ ^4S_2$	$2s^2\ 2p^2(^3P)3p$	$3p\ ^4S^\circ$	$1\frac{1}{2}$	212161. 94		$4d\ ^4F_4$			$2\frac{1}{2}$		
$2p'\ ^2P_2$	$2s\ 2p^4$	$2p^4\ ^2P$	$1\frac{1}{2}$	212593. 2	-169.2	4F_5		$3\frac{1}{2}$			
2P_1			$\frac{1}{2}$	212762. 4		59. 74		$2s^2\ 2p^2(^3P)4d$	$4d\ ^4D$	$\frac{1}{2}$	
$3p\ ^2P_1$	$2s^2\ 2p^2(^3P)3p$	$3p\ ^2P^\circ$	$\frac{1}{2}$	214169. 74	$4d\ ^4D_{2,3}$				$1\frac{1}{2}$		
2P_2			$1\frac{1}{2}$	214229. 48	23. 6	$3s'\ ^4S_2$	$2s\ 2p^3(^6S^\circ)3s$	$3s'''\ ^4S^\circ$	$1\frac{1}{2}$	254982. 2	
$3\bar{p}\ ^2F_3$	$2s^2\ 2p^2(^1D)3p$	$3p'\ ^2F^\circ$	$2\frac{1}{2}$	228723. 3		$4d\ ^4P_3$	$2s^2\ 2p^2(^3P)4d$	$4d\ ^4P$	$2\frac{1}{2}$	255104. 6	-21. 7
2F_4			$3\frac{1}{2}$	228746. 9	-21. 6	4P_2			$1\frac{1}{2}$	255140. 9	
$3\bar{p}\ ^2D_3$	$2s^2\ 2p^2(^1D)3p$	$3p'\ ^2D^\circ$	$2\frac{1}{2}$	229946. 6		4P_1			$\frac{1}{2}$	255162. 6	
2D_2			$1\frac{1}{2}$	229968. 2	54. 03	$4d\ ^2P_2$	$2s^2\ 2p^2(^3P)4d$	$4d\ ^2P$	$1\frac{1}{2}$	255172. 5	-108. 9
$3d\ ^4F_2$	$2s^2\ 2p^2(^3P)3d$	$3d\ ^4F$	$1\frac{1}{2}$	231296. 05		2P_1			$\frac{1}{2}$	255281. 4	
4F_3			$2\frac{1}{2}$	231350. 08	77. 91	$4d\ ^2F_3$	$2s^2\ 2p^2(^3P)4d$	$4d\ ^2F$	$2\frac{1}{2}$	255301. 3	163. 9
4F_4			$3\frac{1}{2}$	231427. 99		2F_4			$3\frac{1}{2}$	255465. 2	
4F_5			$4\frac{1}{2}$	231530. 26	102. 27	$3\bar{d}\ ^2S_1$	$2s^2\ 2p^2(^1D)3d$	$3d'\ ^2S$	$\frac{1}{2}$	255622. 4	-122. 6
$3d\ ^4P_3$	$2s^2\ 2p^2(^3P)3d$	$3d\ ^4P$	$2\frac{1}{2}$	232462. 83		$4f\ ^2D_3$	$2s^2\ 2p^2(^3P)4f$	$4f\ ^2D^\circ$	$2\frac{1}{2}$	255689. 6	
4P_2			$1\frac{1}{2}$	232536. 06	-73. 23	2D_2			$1\frac{1}{2}$	255812. 2	
4P_1			$\frac{1}{2}$	232602. 57		-66. 51	$4f\ ^4D_4$	$2s^2\ 2p^2(^3P)4f$	$4f\ ^4D^\circ$	$3\frac{1}{2}$	255691. 4
$3\bar{p}\ ^2P_1$	$2s^2\ 2p^2(^1D)3p$	$3p'\ ^2P^\circ$	$\frac{1}{2}$	232480. 1	4D_3				$2\frac{1}{2}$	255813. 1	-100
2P_2			$1\frac{1}{2}$	232526. 7	46. 6	4D_2			$1\frac{1}{2}$	255913. ±	
$3d\ ^4D_1$	$2s^2\ 2p^2(^3P)3d$	$3d\ ^4D$	$\frac{1}{2}$	232711. 70		34. 28	4D_1			$\frac{1}{2}$	255912. 0
4D_2			$1\frac{1}{2}$	232745. 98	1. 53						
4D_3			$2\frac{1}{2}$	232747. 51		6. 35					
4D_4			$3\frac{1}{2}$	232753. 86	162. 99						
$3d\ ^2F_3$	$2s^2\ 2p^2(^3P)3d$	$3d\ ^2F$	$2\frac{1}{2}$	232796. 27							
2F_4			$3\frac{1}{2}$	232959. 26							

O II—Continued

O II—Continued

Eldén	Config.	Desig.	<i>J</i>	Level	Interval	Eldén	Config.	Desig.	<i>J</i>	Level	Interval
4f ⁴ G ₃ ⁴ G ₄ ⁴ G ₅ ⁴ G ₆	2s ² 2p ² (³ P)4f	4f ⁴ G°	2½ 3½ 4½ 5½	255755.8 255759.4 255827.6 255977.5	3.6 68.2 149.9	5f ² G ₄ ² G ₅	2s ² 2p ² (³ P)5f	5f ² G°	3½ 4½	265763.0 265930.2	167.2
4f ² G ₄ ² G ₅	2s ² 2p ² (³ P)4f	4f ² G°	3½ 4½	255829.4 255983.6	154.2	5d ² D ₃	2s ² 2p ² (³ P)5d	5d ² D	1½ 2½	265856	
4d ² D ₂ ² D ₃	2s ² 2p ² (³ P)4d	4d ² D	1½ 2½	255843.1 255897.2	54.1	5f ⁴ F ₂ ⁴ F ₃ ⁴ F ₄ ⁴ F ₅	2s ² 2p ² (³ P)5f	5f ⁴ F°	1½ 2½ 3½ 4½	265928? 265961? 265985 265999	33 24 14
4f ⁴ F ₂ ⁴ F ₃ ⁴ F ₄ ⁴ F ₅	2s ² 2p ² (³ P)4f	4f ⁴ F°	1½ 2½ 3½ 4½	256083.5 256087.6 256123.1 256136.2	4.1 35.5 13.1	5f ² F ₃ ² F ₄	2s ² 2p ² (³ P)5f	5f ² F°	2½ 3½	265988? 265999?	11
4f ² F ₃ ² F ₄	2s ² 2p ² (³ P)4f	4f ² F°	2½ 3½	256125.8 256143.3	17.5	3p' ⁶ P ₂ ⁶ P ₃ ⁶ P ₄	2s 2p ³ (⁵ S°)3p	3p''' ⁶ P	1½ 2½ 3½	267763.39+x 267770.85+x 267783.40+x	7.46 12.55
5s ⁴ P ₁ ⁴ P ₂ ⁴ P ₃	2s ² 2p ² (³ P)5s	5s ⁴ P	½ 1½ 2½	257693.7 257797.9 257963.8	104.2 165.9	4d ² F ₃ ² F ₄	2s ² 2p ² (¹ D)4d	4d' ² F	2½ 3½	274739.2 274782.4	43.2
5s ² P ₁ ² P ₂	2s ² 2p ² (³ P)5s	5s ² P	½ 1½	258408.6 258601.7	193.1	4d ² D _{2,3}	2s ² 2p ² (¹ D)4d	4d' ² D	{ 1½ 2½ }	274920	
4s ² D ₃ ² D ₂	2s ² 2p ² (¹ D)4s	4s' ² D	2½ 1½	259286.2 259287.0	-0.8	4d ² P _{1,2}	2s ² 2p ² (¹ D)4d	4d' ² P	{ ½ 1½ }	275611?	
	2s ² 2p ² (³ P)5p	5p ⁴ D°	½ 1½ 2½ 3½	260959 261042 261180	83 138	4f ² G	2s ² 2p ² (¹ D)4f	4f' ² G°	{ 3½ 4½ }	275841.3	
5p ⁴ D ₂ ⁴ D ₃ ⁴ D ₄						4f ² F	2s ² 2p ² (¹ D)4f	4f' ² F°	{ 2½ 3½ }	275879.6	
	2s ² 2p ² (³ P)5p	5p ⁴ P°	½ 1½ 2½	261261.7 261354.3	92.6		2s ² 2p ² (¹ S)3d	3d'' ² D	{ 1½ 2½ }	[275951]	
5p ⁴ P ₂ ⁴ P ₃						4d ² S ₁	2s ² 2p ² (¹ D)4d	4d' ² S	½	275997?	
5p ² D ₂ ² D ₃	2s ² 2p ² (³ P)5p	5p ² D°	1½ 2½	261697.5 261869.4	171.9	4f ² D	2s ² 2p ² (¹ D)4f	4f' ² D°	{ 1½ 2½ }	276066.3	
	2s ² 2p ² (³ P)5d	5d ⁴ D	½ 1½ 2½ 3½	265220.3		4f ² H	2s ² 2p ² (¹ D)4f	4f' ² H°	{ 4½ 5½ }	276109.1	
5d ⁴ D _{2,3}						4f ² P	2s ² 2p ² (¹ D)4f	4f' ² P°	{ ½ 1½ }	276263.9?	
5d ⁴ P ₃ ⁴ P _{1,2}	2s ² 2p ² (³ P)5d	5d ⁴ P	2½ 1½ ½	265431.5 265468.2	-36.7	5s ² D _{2,3}	2s ² 2p ² (¹ D)5s	5s' ² D	{ 1½ 2½ }	278140	
	2s ² 2p ² (³ P)5d	5d ² F	2½ 3½	265578?		O III (³ P ₀) <i>Limit</i>			-----	283550.9	
5f ⁴ D ₄ ⁴ D ₃ ⁴ D ₂ ⁴ D ₁	2s ² 2p ² (³ P)5f	5f ⁴ D°	3½ 2½ 1½ ½	265639 265705? 265762? 265859?	-66 -57 -97	3d' ⁶ D ₅ ⁶ D ₄ ⁶ D ₃ ⁶ D ₂ ⁶ D ₁	2s 2p ³ (⁵ S°)3d	3d''' ⁶ D°	4½ 3½ 2½ 1½ ½	291895.90+x 291896.78+x 291898.01+x 291899.11+x 291899.81+x	-0.88 -1.23 -1.10 -0.70
5f ⁴ G ₃ ⁴ G ₄ ⁴ G ₅ ⁴ G ₆	2s ² 2p ² (³ P)5f	5f ⁴ G°	2½ 3½ 4½ 5½	265665? 265691 265761 265925	26 70 164	4s' ⁶ S ₃	2s 2p ³ (⁵ S°)4s	4s''' ⁶ S°	2½	298849.2 +x	

December 1947.

O II OBSERVED TERMS*

Config. $1s^2 +$	Observed Terms	
$2s^2 2p^3$	{ $2p^3 {}^4S^\circ$ $2p^3 {}^2P^\circ$ $2p^3 {}^2D^\circ$	
$2s 2p^4$	{ $2p^4 {}^2S$ $2p^4 {}^4P$ $2p^4 {}^2P$ $2p^4 {}^2D$	
	$ns (n \geq 3)$	$np (n \geq 3)$
$2s^2 2p^2({}^3P)nx$	{ $3-5s {}^4P$ $3-5s {}^2P$	$3p {}^4S^\circ$ $3, 5p {}^4P^\circ$ $3-5p {}^4D^\circ$ $3p {}^2S^\circ$ $3, 4p {}^2P^\circ$ $3-5p {}^2D^\circ$
$2s^2 2p^2({}^1D)nx'$	$3-5s' {}^2D$	$3p' {}^2P^\circ$ $3p' {}^2D^\circ$ $3p' {}^2F^\circ$
$2s 2p^3({}^6S^\circ)nx'''$	{ $3, 4s''' {}^6S^\circ$ $3s''' {}^4S^\circ$	$3p''' {}^6P$
	$nd (n \geq 3)$	$nf (n \geq 4)$
$2s^2 2p^2({}^3P)nx$	{ $3-5d {}^4P$ $3-5d {}^4D$ $3, 4d {}^4F$ $3, 4d {}^2P$ $3-5d {}^2D$ $3-5d {}^2F$	$4, 5f {}^4D^\circ$ $4, 5f {}^4F^\circ$ $4, 5f {}^4G^\circ$ $4f {}^2D^\circ$ $4, 5f {}^2F^\circ$ $4, 5f {}^2G^\circ$
$2s^2 2p^2({}^1D)nx'$	$3, 4d' {}^2S$ $3, 4d' {}^2P$ $3, 4d' {}^2D$ $3, 4d' {}^2F$ $3d' {}^2G$	$4f' {}^2P^\circ$ $4f' {}^2D^\circ$ $4f' {}^2F^\circ$ $4f' {}^2G^\circ$ $4f' {}^2H^\circ$
$2s 2p^3({}^6S^\circ)nx'''$	{ $3d''' {}^6D^\circ$	

*For predicted terms in the spectra of the N I isoelectronic sequence, see Introduction.

O III

(C I sequence; 6 electrons)

$Z=8$

Ground state $1s^2 2s^2 2p^2 {}^3P_0$

$2p^2 {}^3P_0$ 443193.5 cm^{-1}

I. P. 54.934 volts

The terms are from the papers by Edlén. The singlet, triplet and quintet terms are connected by intersystem combinations. Edlén has kindly furnished some unpublished results for inclusion here, namely, that intersystem combinations with quintet terms indicate that his published absolute values of these terms should be decreased by 418 cm^{-1} . This correction has been incorporated into the tabular values of the quintet terms.

REFERENCES

- C. Mihul, Ann. de Phys. [10] **9**, 326 (1928). (T) (C L) (Z E)
A. Fowler, Proc. Roy. Soc. (London) [A] **117**, 317 (1928). (T) (C L)
B. Edlén, Nova Acta Reg. Soc. Sci Uppsala [IV] **9**, No. 6, 115 (1934). (I P) (T) (C L) (G D)
B. Edlén, Zeit. Phys. **93**, 726 (1935). (T) (C L)
B. Edlén, Naturwiss. **30**, 279 (1942). (T) (C L)
B. Edlén, unpublished material (Dec. 1947). (T)

O III

O III

Edlén	Config.	Desig.	<i>J</i>	Level	Interval	Edlén	Config.	Desig.	<i>J</i>	Level	Interval
$2p\ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s^2\ 2p^2$	$2p^2\ ^3P$	0 1 2	0.0 113.4 306.8	113.4 193.4	$3s'\ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s\ 2p^2(^4P)3s$	$3s\ ^3P$	0 1 2	350026.1 350122.9 350302.3	96.8 179.4
$2p\ ^1D_2$	$2s^2\ 2p^2$	$2p^2\ ^1D$	2	20271.0		$4s\ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s^2\ 2p(^2P^\circ)4s$	$4s\ ^3P^\circ$	0 1 2	356732 356838 357111	106 273
$2p\ ^1S_0$	$2s^2\ 2p^2$	$2p^2\ ^1S$	0	43183.5		$4s\ ^1P_1$	$2s^2\ 2p(^2P^\circ)4s$	$4s\ ^1P^\circ$	1	358667.4	
$2p'\ ^5S_2$	$2s\ 2p^3$	$2p^3\ ^5S^\circ$	2	60312.1		$3p'\ ^3S_1$	$2s\ 2p^2(^4P)3p$	$3p\ ^3S^\circ$	1	363266.8	
$2p'\ ^3D_3$ $\ ^3D_2$ $\ ^3D_1$	$2s\ 2p^3$	$2p^3\ ^3D^\circ$	3 2 1	120025.4 120052.6 120058.5	-27.2 -5.9	$3p'\ ^5D_0$ $\ ^5D_1$ $\ ^5D_2$ $\ ^5D_3$ $\ ^5D_4$	$2s\ 2p^2(^4P)3p$	$3p\ ^5D^\circ$	0 1 2 3 4	365515.76 365550.60 365619.12 365719.16 365846.46	34.84 68.52 100.04 127.30
$2p'\ ^3P_2$ $\ ^3P_1$ $\ ^3P_0$	$2s\ 2p^3$	$2p^3\ ^3P^\circ$	2 1 0	142381.7 142382.8 142396.9	-1.1 -14.1	$4p\ ^1P_1$	$2s^2\ 2p(^2P^\circ)4p$	$4p\ ^1P$	1	365723.9	
$2p'\ ^1D_2$	$2s\ 2p^3$	$2p^3\ ^1D^\circ$	2	187049.4		$4p\ ^3D_1$ $\ ^3D_2$ $\ ^3D_3$	$2s^2\ 2p(^2P^\circ)4p$	$4p\ ^3D$	1 2 3	366486.91 366594.01 366801.04	107.10 207.03
$2p'\ ^3S_1$	$2s\ 2p^3$	$2p^3\ ^3S^\circ$	1	197086.7		$4p\ ^3S_1$	$2s^2\ 2p(^2P^\circ)4p$	$4p\ ^3S$	1	367952.20	
$2p'\ ^1P_1$	$2s\ 2p^3$	$2p^3\ ^1P^\circ$	1	210458.5		$3p'\ ^5P_1$ $\ ^5P_2$ $\ ^5P_3$	$2s\ 2p^2(^4P)3p$	$3p\ ^5P^\circ$	1 2 3	368526.37 368583.63 368684.75	57.26 101.12
$3s\ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s^2\ 2p(^2P^\circ)3s$	$3s\ ^3P^\circ$	0 1 2	267257.29 267375.65 267632.59	118.36 256.94	$4p\ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s^2\ 2p(^2P^\circ)4p$	$4p\ ^3P$	0 1 2	370326.7 370415.7 370524.2	89.0 108.5
$3s\ ^1P_1$	$2s^2\ 2p(^2P^\circ)3s$	$3s\ ^1P^\circ$	1	273080.07		$4p\ ^1D_2$	$2s^2\ 2p(^2P^\circ)4p$	$4p\ ^1D$	2	370900.6	
$2p''\ ^3P_2$ $\ ^3P_1$ $\ ^3P_0$	$2p^4$	$2p^4\ ^3P$	2 1 0	283758.9 283976.6 284073.3	-217.7 -96.7	$4p\ ^1S_0$	$2s^2\ 2p(^2P^\circ)4p$	$4p\ ^1S$	0	373046.2	
$3p\ ^1P_1$	$2s^2\ 2p(^2P^\circ)3p$	$3p\ ^1P$	1	290956.62		$3p'\ ^3D_1$ $\ ^3D_2$ $\ ^3D_3$	$2s\ 2p^2(^4P)3p$	$3p\ ^3D^\circ$	1 2 3	374575 374662.5 374798.6	88 136.1
$3p\ ^3D_1$ $\ ^3D_2$ $\ ^3D_3$	$2s^2\ 2p(^2P^\circ)3p$	$3p\ ^3D$	1 2 3	293865.26 294001.60 294221.65	136.34 220.05	$3p'\ ^5S_2$	$2s\ 2p^2(^4P)3p$	$3p\ ^5S^\circ$	2	376067.66	
$3p\ ^3S_1$	$2s^2\ 2p(^2P^\circ)3p$	$3p\ ^3S$	1	297557.50		$4d\ ^3F_2$	$2s^2\ 2p(^2P^\circ)4d$	$4d\ ^3F^\circ$	2 3 4	377375	
$2p''\ ^1D_2$	$2p^4$	$2p^4\ ^1D$	2	298289.4		$4d\ ^1D_2$	$2s^2\ 2p(^2P^\circ)4d$	$4d\ ^1D^\circ$	2	377687	
$3p\ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s^2\ 2p(^2P^\circ)3p$	$3p\ ^3P$	0 1 2	300228.21 300310.31 300440.85	82.10 130.54	$3p'\ ^3P_2$ $\ ^3P_1$ $\ ^3P_0$	$2s\ 2p^2(^4P)3p$	$3p\ ^3P^\circ$	2 1 0	378408.5 378420.9 378438.1	-12.4 -17.2
$3p\ ^1D_2$	$2s^2\ 2p(^2P^\circ)3p$	$3p\ ^1D$	2	306584.8		$4d\ ^3D_1$ $\ ^3D_2$ $\ ^3D_3$	$2s^2\ 2p(^2P^\circ)4d$	$4d\ ^3D^\circ$	1 2 3	379232 379293 379356	61 63
$3p\ ^1S_0$	$2s^2\ 2p(^2P^\circ)3p$	$3p\ ^1S$	0	313801.07		$4d\ ^3P_2$	$2s^2\ 2p(^2P^\circ)4d$	$4d\ ^3P^\circ$	2 1 0	380706	
$3d\ ^3F_2$ $\ ^3F_3$ $\ ^3F_4$	$2s^2\ 2p(^2P^\circ)3d$	$3d\ ^3F^\circ$	2 3 4	324462.46 324658.25 324836.41	195.79 178.16	$4d\ ^1F_3$	$2s^2\ 2p(^2P^\circ)4d$	$4d\ ^1F^\circ$	3	380782	
$3d\ ^1D_2$	$2s^2\ 2p(^2P^\circ)3d$	$3d\ ^1D^\circ$	2	324734.22		$4d\ ^1P_1$	$2s^2\ 2p(^2P^\circ)4d$	$4d\ ^1P^\circ$	1	381086	
$3d\ ^3D_1$ $\ ^3D_2$ $\ ^3D_3$	$2s^2\ 2p(^2P^\circ)3d$	$3d\ ^3D^\circ$	1 2 3	327227.94 327277.18 327350.90	49.24 73.72	$5s\ ^3P_2$	$2s^2\ 2p(^2P^\circ)5s$	$5s\ ^3P^\circ$	0 1 2	392221	
$3d\ ^3P_2$ $\ ^3P_1$ $\ ^3P_0$	$2s^2\ 2p(^2P^\circ)3d$	$3d\ ^3P^\circ$	2 1 0	329467.98 329531.98 329643.43	-114.00 -61.45	$5s\ ^1P_1$	$2s^2\ 2p(^2P^\circ)5s$	$5s\ ^1P^\circ$	1	392778	
$3d\ ^1F_3$	$2s^2\ 2p(^2P^\circ)3d$	$3d\ ^1F^\circ$	3	331820.2		$\overline{3s'}\ ^3D_1$ $\ ^3D_2$ $\ ^3D_3$	$2s\ 2p^2(^2D)3s$	$3s'\ ^3D$	1 2 3	394090 394126 394195	36 69
$3d\ ^1P_1$	$2s^2\ 2p(^2P^\circ)3d$	$3d\ ^1P^\circ$	1	332777.1							
$3s'\ ^5P_1$ $\ ^5P_2$ $\ ^5P_3$	$2s\ 2p^2(^4P)3s$	$3s\ ^5P$	1 2 3	338565.87 338690.34 338851.50	124.47 161.16						
$2p''\ ^1S_0$	$2p^4$	$2p^4\ ^1S$	0	343302.67							

O III—Continued

O III—Continued

Edlén	Config.	Desig.	<i>J</i>	Level	Interval	Edlén	Config.	Desig.	<i>J</i>	Level	Interval
$3d' \ ^5F_1$	$2s \ 2p^2(^4P)3d$	$3d \ ^5F$	1	394516.45	38.70	$7d \ ^1F_3$	$2s^2 \ 2p(^2P^\circ)7d$	$7d \ ^1F^\circ$	3	422977	
$\ ^5F_2$			2	394555.15	57.55	$\overline{3p'} \ ^1F_3$	$2s \ 2p^2(^2D)3p$	$3p' \ ^1F^\circ$	3	424998	
$\ ^5F_3$			3	394612.70	75.74	$\overline{3p'} \ ^1D_2$	$2s \ 2p^2(^2D)3p$	$3p' \ ^1D^\circ$	2	426338	
$\ ^5F_4$			4	394688.44	92.03	$4s' \ ^5P_1$	$2s \ 2p^2(^4P)4s$	$4s \ ^5P$	1	428487	119
$\ ^5F_5$			5	394780.47		$\ ^5P_2$			2	428606	163
						$\ ^5P_3$			3	428769	
$3d' \ ^5D_0$	$2s \ 2p^2(^4P)3d$	$3d \ ^5D$	0	398135.0	-3.6	$\overline{3p'} \ ^1P_1$	$2s \ 2p^2(^2D)3p$	$3p' \ ^1P^\circ$	1	430025	
$\ ^5D_1$			1	398131.4	-4.1						
$\ ^5D_2$			2	398127.3	10.1						
$\ ^5D_3$			3	398137.4	81.4						
$\ ^5D_4$			4	398218.8							
$3d' \ ^5P_3$	$2s \ 2p^2(^4P)3d$	$3d \ ^5P$	3	398474.3	-70.0	$4p' \ ^3S_1$	$2s \ 2p^2(^4P)4p$	$4p \ ^3S^\circ$	1	437015.0	
$\ ^5P_2$			2	398544.3	-38.5						
$\ ^5P_1$			1	398582.8		$4p' \ ^5D_0$	$2s \ 2p^2(^4P)4p$	$4p \ ^5D^\circ$	0		
$3d' \ ^3P_2$	$2s \ 2p^2(^4P)3d$	$3d \ ^3P$	2	400354.8	-109.9	$\ ^5D_1$			1	438241.0	62.2
$\ ^3P_1$			1	400464.7	-53.7	$\ ^5D_2$			2	438303.2	92.0
$\ ^3P_0$			0	400518.4		$\ ^5D_3$			3	438395.2	122.3
						$\ ^5D_4$			4	438517.5	
$3d' \ ^3F_2$	$2s \ 2p^2(^4P)3d$	$3d \ ^3F$	2	401379	96	$4p' \ ^5P_1$	$2s \ 2p^2(^4P)4p$	$4p \ ^5P^\circ$	1	439278.1	51.4
$\ ^3F_3$			3	401475.4	133.7	$\ ^5P_2$			2	439329.5	98.1
$\ ^3F_4$			4	401609.1		$\ ^5P_3$			3	439427.6	
$5d \ ^3F_2$	$2s^2 \ 2p(^2P^\circ)5d$	$5d \ ^3F^\circ$	2	401530			$2s \ 2p^2(^4P)4p$	$4p \ ^3D^\circ$	1		
			3						2		
			4						3		
$5d \ ^1D_2$	$2s^2 \ 2p(^2P^\circ)5d$	$5d \ ^1D^\circ$	2	401787		$4p' \ ^3D_3$				442710	
	$2s^2 \ 2p(^2P^\circ)5d$	$5d \ ^3D^\circ$	1				$O \ IV \ (^2P_{\frac{3}{2}})$	Limit	-----	443193.5	
			2								
$5d \ ^3D_3$			3	402530		$4d' \ ^5P_3$	$2s \ 2p^2(^4P)4d$	$4d \ ^5P$	3	450167	-70
$5d \ ^1F_3$	$2s^2 \ 2p(^2P^\circ)5d$	$5d \ ^1F^\circ$	3	403374		$\ ^5P_2$			2	450237	-54
$5d \ ^1P_1$	$2s^2 \ 2p(^2P^\circ)5d$	$5d \ ^1P^\circ$	1	403526		$\ ^5P_1$			1	450291	
$3d' \ ^3D_1$	$2s \ 2p^2(^4P)3d$	$3d \ ^3D$	1	405805.1	29.0	$\overline{3d'} \ ^3F$	$2s \ 2p^2(^2D)3d$	$3d' \ ^3F$	2, 3, 4	452855	
$\ ^3D_2$			2	405834.1	48.9	$\overline{3d'} \ ^3D$	$2s \ 2p^2(^2D)3d$	$3d' \ ^3D$	1, 2, 3	454174	
$\ ^3D_3$			3	405883.0		$\overline{3d'} \ ^3P$	$2s \ 2p^2(^2D)3d$	$3d' \ ^3P$	0, 1, 2	457634	
$6d \ ^1D_2$	$2s^2 \ 2p(^2P^\circ)6d$	$6d \ ^1D^\circ$	2	414675		$5d' \ ^5P_3$	$2s \ 2p^2(^4P)5d$	$5d \ ^5P$		473750	
	$2s^2 \ 2p(^2P^\circ)6d$	$6d \ ^3D^\circ$	1								
			2								
$6d \ ^3D_3$			3	415181					1		

December 1947.

O III OBSERVED TERMS*

Config. $1s^2+$	Observed Terms		
$2s^2 \ 2p^3$	$\{ 2p^2 \ ^1S \quad 2p^2 \ ^3P \quad 2p^2 \ ^1D$		
$2s \ 2p^3$	$\{ 2p^3 \ ^5S^\circ \quad 2p^3 \ ^3P^\circ \quad 2p^3 \ ^3D^\circ$ $\quad \quad \quad 2p^3 \ ^3S^\circ \quad 2p^3 \ ^1P^\circ \quad 2p^3 \ ^1D^\circ$		
$2p^4$	$\{ 2p^4 \ ^1S \quad 2p^4 \ ^3P \quad 2p^4 \ ^1D$		
	$ns \ (n \geq 3)$	$np \ (n \geq 3)$	$nd \ (n \geq 3)$
$2s^2 \ 2p(^2P^\circ)nx$	$\{ \quad \quad \quad 3-5s \ ^3P^\circ$ $\quad \quad \quad 3-5s \ ^1P^\circ$	$\quad \quad \quad 3, 4p \ ^3S \quad 3, 4p \ ^3P \quad 3, 4p \ ^3D$ $\quad \quad \quad 3, 4p \ ^1S \quad 3, 4p \ ^1P \quad 3, 4p \ ^1D$	$\quad \quad \quad 3, 4d \ ^3P^\circ \quad 3-6d \ ^3D^\circ \quad 3-5d \ ^3F^\circ$ $\quad \quad \quad 3-5d \ ^1P^\circ \quad 3-6d \ ^1D^\circ \quad 3-5, 7d \ ^1F^\circ$
$2s \ 2p^2(^4P)nx$	$\{ \quad \quad \quad 3, 4s \ ^5P$ $\quad \quad \quad 3s \ ^3P$	$\quad \quad \quad 3p \ ^5S^\circ \quad 3, 4p \ ^5P^\circ \quad 3, 4p \ ^5D^\circ$ $\quad \quad \quad 3, 4p \ ^3S^\circ \quad 3p \ ^3P^\circ \quad 3, 4p \ ^3D^\circ$	$\quad \quad \quad 3-5d \ ^5P \quad 3d \ ^5D \quad 3d \ ^5F$ $\quad \quad \quad 3d \ ^3P \quad 3d \ ^3D \quad 3d \ ^3F$
$2s \ 2p^2(^2D)nx'$	$\{ \quad \quad \quad 3s' \ ^3D$	$\quad \quad \quad 3p' \ ^1P^\circ \quad 3p' \ ^1D^\circ \quad 3p' \ ^1F^\circ$	$\quad \quad \quad 3d' \ ^3P \quad 3d' \ ^3D \quad 3d' \ ^3F$

* For predicted terms in the spectra of the C I isoelectronic sequence, see Introduction.

O IV

(B I sequence; 5 electrons)

 $Z=8$ Ground state $1s^2 2s^2 2p^3 P^\circ_{\frac{3}{2}}$ $2p^3 P^\circ_{\frac{3}{2}}$ 624396.5 cm^{-1}

I. P. 77.394 volts

Most of the terms are from Edlén's Monograph, corrected to agree with his 1935 paper, in which he adds several terms from $2p^2(^1D)$ and relabels his $2p^2(^3P)3s^2P$ term as $2p^2(^1D)3s^2D$. He also lists a combination in the visible, $3s'^2P^\circ - 3p'^2D$, from which a revised value of $3s'^2P^\circ$ has been calculated. A few other additions and corrections kindly communicated by Edlén have been incorporated into the table.

The term $6f^2F^\circ$ is from the paper by Whitelaw and Mack.

No intercombinations between the doublet and quartet terms have been observed, but the limits adopted by Edlén are based on well-established series, and the relative positions of the two groups of terms differ by probably only a small constant x .

REFERENCES

- L. J. Freeman, Proc. Roy. Soc. (London) [A] **127**, 330 (1930). (T) (C L)
 B. Edlén, Nova Acta Reg. Soc. Uppsala [IV] **9**, No. 6, 87 (1934). (I P) (T) (C L) (G D)
 P. G. Kruger and W. E. Shoupp, Phys. Rev. **44**, 105 (1933). (T) (C L)
 E. Edlén, Zeit. Phys. **93**, 726 (1935). (T) (C L)
 N. G. Whitelaw and J. E. Mack, Phys. Rev. **47**, 677 (1935). (T)
 B. Edlén, unpublished material (Dec. 1947). (T)

O IV

O IV

Edlén	Config.	Desig.	J	Level	Interval	Edlén	Config.	Desig.	J	Level	Interval
$2p^3 P^\circ_{\frac{3}{2}}$	$2s^2(^1S)2p$	$2p^3 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	0.0 386.5	386.5	$3s'^2 P^\circ_{\frac{3}{2}}$	$2s^2 2p(^3P^\circ)3s$	$3s^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	452808.0 453073.0	265.0
$2p'^4 P_1$ $4P_2$ $4P_3$	$2s^2 2p^2$	$2p^3 ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$71177.0+x$ $71308.4+x$ $71492.9+x$	131.4 184.5	$3p'^2 P_1$ $2P_2$	$2s^2 2p(^3P^\circ)3p$	$3p^2 P$	$\frac{1}{2}$ $1\frac{1}{2}$	467231.1 467346.5	115.4
$2p'^2 D_3$ $2D_2$	$2s^2 2p^2$	$2p^2 ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	126936.3 126950.3	-14.0	$3p'^4 D_1$ $4D_2$ $4D_3$ $4D_4$	$2s^2 2p(^3P^\circ)3p$	$3p^4 D$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	$468075.4+x$ $468154.2+x$ $468289.7+x$ $468499.4+x$	78.8 135.5 209.7
$2p'^2 S_1$	$2s^2 2p^2$	$2p^2 ^2S$	$\frac{1}{2}$	164366.9		$3p'^4 S_2$	$2s^2 2p(^3P^\circ)3p$	$3p^4 S$	$1\frac{1}{2}$	$474217.8+x$	
$2p'^2 P_1$ $2P_2$	$2s^2 2p^2$	$2p^2 ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	180481.3 180724.6	243.3	$3p'^4 P_1$ $4P_2$ $4P_3$	$2s^2 2p(^3P^\circ)3p$	$3p^4 P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$478587.7+x$ $478682.2+x$ $478811.3+x$	94.5 129.1
$2p''^4 S_2$	$2p^3$	$2p^3 ^4S^\circ$	$1\frac{1}{2}$	$231275.1+x$		$3p'^2 D_2$ $2D_3$	$2s^2 2p(^3P^\circ)3p$	$3p^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	482667.5 482923.1	255.6
$2p''^2 D_3$ $2D_2$	$2p^3$	$2p^3 ^2D^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$	255156.7 255186.0	-29.3	$4s^2 S_1$	$2s^2(^1S)4s$	$4s^2 S$	$\frac{1}{2}$	485823.1	
$2p''^2 P_1$ $2P_2$	$2p^3$	$2p^3 ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	289016.1 289024.0	7.9	$3p'^2 S_1$	$2s^2 2p(^3P^\circ)3p$	$3p^2 S$	$\frac{1}{2}$	492880	
$3s^2 S_1$	$2s^2(^1S)3s$	$3s^2 S$	$\frac{1}{2}$	357614.8		$3d'^4 F_2$ $4F_3$ $4F_4$ $4F_5$	$2s^2 2p(^3P^\circ)3d$	$3d^4 F^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	$494907.5+x$ $494986.3+x$ $495098.7+x$ $495252.8+x$	78.8 112.4 154.1
$3p^2 P_1$ $2P_2$	$2s^2(^1S)3p$	$3p^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	390161.1 390248.2	87.1	$3d'^4 D_1$ $4D_2$ $4D_3$ $4D_4$	$2s^2 2p(^3P^\circ)3d$	$3d^4 D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	$499506.4+x$ $499535.3+x$ $499582.0+x$ $499646.6+x$	28.9 46.7 64.6
$3d^2 D_2$ $2D_3$	$2s^2(^1S)3d$	$3d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	419533.5 419550.2	16.7						
$3s'^4 P_1$ $4P_2$ $4P_3$	$2s^2 2p(^3P^\circ)3s$	$3s^4 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$438588.5+x$ $438723.6+x$ $438970.5+x$	135.1 246.9						

O IV—Continued

O IV—Continued

Edlén	Config.	Desig.	<i>J</i>	Level	Interval	Edlén	Config.	Desig.	<i>J</i>	Level	Interval
$3d' \ ^2D_2$ $\ ^2D_3$	$2s \ 2p(^3P^\circ)3d$	$3d \ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	501511. 3 501566. 4	55. 1	$4d' \ ^2D_2$ $\ ^2D_3$	$2s \ 2p(^3P^\circ)4d$	$4d \ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	593627 593708	81
$3d' \ ^4P_3$ $\ ^4P_2$ $\ ^4P_1$	$2s \ 2p(^3P^\circ)3d$	$3d \ ^4P^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	503834. 5+x 503947. 9+x 504021. 7+x	-113. 4 -73. 8	$4f' \ ^2F_3$ $\ ^2F_4$	$2s \ 2p(^3P^\circ)4f$	$4f \ ^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	594007 594080	73
$4d \ ^2D_2$ $\ ^2D_3$	$2s^2(^1S)4d$	$4d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	510560 510567	7	$4f' \ ^2D_2$ $\ ^2D_3$	$2s \ 2p(^3P^\circ)4f$	$4f \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	594337 594542	205
$3d' \ ^2F_3$ $\ ^2F_4$	$2s \ 2p(^3P^\circ)3d$	$3d \ ^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	510746. 1 510978. 5	232. 4	$4d' \ ^2F_3$ $\ ^2F_4$	$2s \ 2p(^3P^\circ)4d$	$4d \ ^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	596299 596477	178
$3d' \ ^2P_2$ $\ ^2P_1$	$2s \ 2p(^3P^\circ)3d$	$3d \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	514217 514368	-151	$3p'' \ ^2S_1$	$2p^2(^3P)3p$	$3p'' \ ^2S^\circ$	$\frac{1}{2}$	597254	
$\overline{3s'} \ ^2P_1$ $\ ^2P_2$	$2s \ 2p(^1P^\circ)3s$	$3s' \ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	518684 518690	6	$8f \ ^2F$	$2s^2(^1S)8f$	$8f \ ^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	597352	
$5s \ ^2S_1$	$2s^2(^1S)5s$	$5s \ ^2S$	$\frac{1}{2}$	539368		$4d' \ ^2P_2$ $\ ^2P_1$	$2s \ 2p(^3P^\circ)4d$	$4d \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	597726 597863	-137
$\overline{3p'} \ ^2D_2$ $\ ^2D_3$	$2s \ 2p(^1P^\circ)3p$	$3p' \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	547311 547336	25	$\overline{3s''} \ ^2D_2$ $\ ^2D_3$	$2p^2(^1D)3s$	$3s''' \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	600092 600106	14
$\overline{3p'} \ ^2P_1$ $\ ^2P_2$	$2s \ 2p(^1P^\circ)3p$	$3p' \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	549792 549855	63		$2p^2(^3P)3p$	$3p'' \ ^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	602977	+x
$5d \ ^2D_3$	$2s^2(^1S)5d$	$5d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	552034		$3p'' \ ^4D_4$					
$5f \ ^2F$	$2s^2(^1S)5f$	$5f \ ^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	552490		$2p^2(^3P)3p$	$3p'' \ ^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	606434	+x	
$\overline{3p'} \ ^2S_1$	$2s \ 2p(^1P^\circ)3p$	$3p' \ ^2S$	$\frac{1}{2}$	554461		$3p'' \ ^2D_3$ $\ ^2D_2$	$2p^2(^3P)3p$	$3p'' \ ^2D^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$	616431 616460	-29
$4s' \ ^4P_1$ $\ ^4P_2$ $\ ^4P_3$	$2s \ 2p(^3P^\circ)4s$	$4s \ ^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	568638 +x 568773 +x 569020 +x	135 247	$3p'' \ ^4S_2$	$2p^2(^3P)3p$	$3p'' \ ^4S^\circ$	$1\frac{1}{2}$	616588	+x
	$2s \ 2p(^1P^\circ)3d$	$3d' \ ^1F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	570791			O v (1S_0)	Limit	-----	624396. 5	
$\overline{3d'} \ ^1F_4$	$2s \ 2p(^3P^\circ)4s$	$4s \ ^1P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	573696 573907	211	$\overline{3p''} \ ^1F$	$2p^2(^1D)3p$	$3p''' \ ^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	624882	
$4s' \ ^1P_1$ $\ ^1P_2$	$2s^2(^1S)6d$	$6d \ ^1D$	$1\frac{1}{2}$ $2\frac{1}{2}$	574373		$5p' \ ^1P_2$	$2s \ 2p(^3P^\circ)5p$	$5p \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	628496	
$6d \ ^1D_3$	$2s^2 \ 2p(^3P^\circ)4p$	$4p \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	575204 575373	169	$3d'' \ ^2F_4$	$2p^2(^3P)3d$	$3d'' \ ^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	630095	
$\overline{3d'} \ ^2D_2$ $\ ^2D_3$	$2s \ 2p(^1P^\circ)3d$	$3d' \ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	575819 575853	34	$5p' \ ^2D_2$ $\ ^2D_3$	$2s \ 2p(^3P^\circ)5p$	$5p \ ^1D$	$1\frac{1}{2}$ $2\frac{1}{2}$	630703 630879	176
$3s'' \ ^4P_1$ $\ ^4P_2$ $\ ^4P_3$	$2p^2(^3P)3s$	$3s'' \ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	576591 +x 576735 +x 576947 +x	144 212	$3d'' \ ^2D_3$ $\ ^2D_2$	$2p^2(^3P)3d$	$3d'' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	632426 632594	-168
$\overline{3d'} \ ^2P_1$ $\ ^2P_2$	$2s \ 2p(^1P^\circ)3d$	$3d' \ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	581721 581743	22	$5d' \ ^4D_4$	$2s \ 2p(^3P^\circ)5d$	$5d \ ^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	633896	+x
$4p' \ ^2D_2$ $\ ^2D_3$	$2s \ 2p(^3P^\circ)4p$	$4p \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	584552 584768	216	$5d' \ ^4P_3$	$2s \ 2p(^3P^\circ)5d$	$5d \ ^4P^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	634245. 5+x	
$7f \ ^1F$	$2s^2(^1S)7f$	$7f \ ^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	587850		$5d' \ ^2F_3$ $\ ^2F_4$	$2s \ 2p(^3P^\circ)5d$	$5d \ ^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	636024 636236	212
$4p' \ ^2S$	$2s \ 2p(^3P^\circ)4p$	$4p \ ^2S$	$\frac{1}{2}$	590071		$5d' \ ^2P_2$	$2s \ 2p(^3P^\circ)5d$	$5d \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	636492?	
	$2s \ 2p(^3P^\circ)4d$	$4d \ ^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$			$3d'' \ ^4P_3$ $\ ^4P_2$ $\ ^4P_1$	$2p^2(^3P)3d$	$3d'' \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	636851 +x 636950 +x 637012 +x	-99 -62
$4d' \ ^4D_4$				591767 +x		$\overline{3d''} \ ^2D$	$2p^2(^1D)3d$	$3d''' \ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	646859	
$4d' \ ^4P_3$	$2s \ 2p(^3P^\circ)4d$	$4d \ ^4P^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	592999 +x							

O IV—Continued

Edlén	Config.	Design.	<i>J</i>	Level	Interval
$\overline{3d''} \ ^2F_3$ $\ ^2F_4$	$2p^2(^1D)3d$	$3d''' \ ^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	651098 651117	19
$\overline{3d''} \ ^2P_2$ $\ ^2P_1$	$2p^2(^1D)3d$	$3d''' \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	653328 653411	-83
$6d' \ ^4D_4$	$2s \ 2p(^3P^o)6d$	$6d \ ^4D^o$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	656328 + <i>x</i>	
$\overline{4p'} \ ^2D_3$	$2s \ 2p(^1P^o)4p$	$4p' \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	656748	
$\overline{3d''} \ ^2S_1$	$2p^2(^1D)3d$	$3d''' \ ^2S$	$\frac{1}{2}$	659998	
$\overline{4d'} \ ^2D_3$	$2s \ 2p(^1P^o)4d$	$4d' \ ^2D^o$	$1\frac{1}{2}$ $2\frac{1}{2}$	668538	
$7d' \ ^4D_4$	$2s \ 2p(^3P^o)7d$	$7d \ ^4D^o$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	669705 + <i>x</i>	

December 1947.

O IV OBSERVED TERMS*

Config. $1s^2+$	Observed Terms			
	$ns \ (n \geq 3)$	$np \ (n \geq 3)$	$nd \ (n \geq 3)$	$nf \ (n \geq 4)$
$2s^2(^1S)2p$	$2p \ ^2P^o$			
$2s \ 2p^3$	$\{2p^3 \ ^2S$ $\ ^2P^o$ $\ ^2P^o$	$2p^3 \ ^2D$		
$2p^3$	$\{2p^3 \ ^4S^o$ $\ ^2P^o$ $\ ^2P^o$	$2p^3 \ ^2D^o$		
$2s^2(^1S)nx$	$3-5s \ ^2S$	$3p \ ^2P^o$	$3-6d$	$5, 7, 8f \ ^2F^o$
$2s \ 2p(^3P^o)nx$	$\{3, 4s \ ^4P^o$ $3, 4s \ ^2P^o$	$3p \ ^4P$ $3p \ ^2P$ $3-5p \ ^2P$	$3-7d \ ^4D^o$ $3, 4d \ ^2D^o$	$4f \ ^2F$
$2s \ 2p(^1P^o)nx'$	$3s' \ ^2P^o$	$3p' \ ^2S$ $3p' \ ^2P$ $3p' \ ^4P^o$	$3, 4d' \ ^2D^o$	$4f \ ^2D$
$2p^2(^3P)nx''$	$\{3s'' \ ^4P$	$3p'' \ ^4S^o$ $3p'' \ ^2S^o$	$3p'' \ ^4D^o$ $3p'' \ ^2D^o$	$4f \ ^2F$
$2p^2(^1D)nx'''$			$3p''' \ ^2F^o$	

*For predicted terms in the spectra of the Br isoelectronic sequence, see Introduction.

(Be I sequence; 4 electrons)

Z=8

Ground state $1s^2 2s^2 {}^1S_0$ $2s^2 {}^1S_0$ 918702 cm^{-1}

I. P. 113.873 volts

Edlén has revised and extended his published analysis and has generously furnished a manuscript copy of his complete term list in advance of publication, for inclusion here. He states that no intersystem combinations have been observed and that the relative uncertainty x in the position of the triplet terms with respect to the singlets may be $\pm 100 \text{ cm}^{-1}$.

In the published papers Edlén has used a prime to designate the terms from the ${}^2P^\circ$ limit in O VI.

REFERENCES

B. Edlén, Nova Acta Reg. Soc. Sci. Uppsala [IV] 9, No. 6, 62 (1934). (I P) (T) (C L)

B. Edlén, unpublished material (Dec. 1947). (I P) (T)

O V

O V

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$2s^2$	$2s^2 {}^1S$	0	0		$2p({}^2P^\circ)3p$	$3p {}^3S$	1	684124 + <i>x</i>	
$2s({}^2S)2p$	$2p {}^4P^\circ$	0	82121. 2+ <i>x</i>	136. 7 306. 2	$2p({}^2P^\circ)3p$	$3p {}^3P$	0	689585. 6+ <i>x</i>	114. 0 190. 7
		1	82257. 9+ <i>x</i>				1	689699. 6+ <i>x</i>	
		2	82564. 1+ <i>x</i>				2	689890. 3+ <i>x</i>	
$2s({}^2S)2p$	$2p {}^1P^\circ$	1	158798		$2p({}^2P^\circ)3d$	$3d {}^1D^\circ$	2	694646	
$2p^2$	$2p^2 {}^3P$	0	213641. 7+ <i>x</i>	155. 7 268. 8	$2p({}^2P^\circ)3p$	$3p {}^1D$	2	697170	
		1	213797. 4+ <i>x</i>		$2p({}^2P^\circ)3d$	$3d {}^3D^\circ$	1	704360 + <i>x</i>	64 103
		2	214066. 2+ <i>x</i>				2	704424 + <i>x</i>	
$2p^2$	$2p^2 {}^1D$	2	231722				3	704527 + <i>x</i>	
$2p^2$	$2p^2 {}^1S$	0	287909		$2p({}^2P^\circ)3p$	$3p {}^1S$	0	707630	
$2s({}^2S)3s$	$3s {}^3S$	1	547150. 0+ <i>x</i>		$2p({}^2P^\circ)3d$	$3d {}^3P^\circ$	2	708154 + <i>x</i>	-142 -83
$2s({}^2S)3s$	$3s {}^1S$	0	561278				1	708296 + <i>x</i>	
							0	708379 + <i>x</i>	
$2s({}^2S)3p$	$3p {}^1P^\circ$	1	580826		$2p({}^2P^\circ)3d$	$3d {}^1F^\circ$	3	712967	
$2s({}^2S)3p$	$3p {}^3P^\circ$	0	582983. 6+ <i>x</i>	36. 3 77. 3	$2p({}^2P^\circ)3d$	$3d {}^1P^\circ$	1	719277	
		1	583019. 9+ <i>x</i>		$2s({}^2S)4s$	$4s {}^3S$	1	722666 + <i>x</i>	
		2	583097. 2+ <i>x</i>						
$2s({}^2S)3d$	$3d {}^3D$	1	600925. 5+ <i>x</i>	10. 8 19. 8	$2s({}^2S)4s$	$4s {}^1S$	0	731667	
		2	600936. 3+ <i>x</i>		$2s({}^2S)4p$	$4p {}^3P^\circ$	0		18
		3	600956. 1+ <i>x</i>				1	736108 + <i>x</i>	
$2s({}^2S)3d$	$3d {}^1D$	2	612617				2	736126 + <i>x</i>	
$2p({}^2P^\circ)3s$	$3s {}^3P^\circ$	0	653099. 7+ <i>x</i>	162. 5 342. 8	$2s({}^2S)4p$	$4p {}^1P^\circ$	1	737883	
		1	653262. 2+ <i>x</i>		$2s({}^2S)4d$	$4d {}^3D$	1	742401 + <i>x</i>	6 14
		2	653605. 0+ <i>x</i>				2	742407 + <i>x</i>	
$2p({}^2P^\circ)3s$	$3s {}^1P^\circ$	1	664486				3	742421 + <i>x</i>	
$2p({}^2P^\circ)3p$	$3p {}^1P$	1	672695		$2s({}^2S)4d$	$4d {}^1D$	2	746280	
$2p({}^2P^\circ)3p$	$3p {}^3D$	1	677333 + <i>x</i>	199 315	$2s({}^2S)4f$	$4f {}^1F^\circ$	3	749857	
		2	677532 + <i>x</i>		$2s({}^2S)5s$	$5s {}^3S$	1	796263 + <i>x</i>	
		3	677847 + <i>x</i>						

O v—Continued

O v—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
2s(2S)5p	5p 1P°	1	802452		2s(2S)7p	7p 1P°	1	860874	
2s(2S)5d	5d 3D	1			2s(2S)7d	7d 3D	1		
		2					2		
		3	806625 +x				3	861975 +x	
2s(2S)5d	5d 1D	2	808351		2s(2S)7d	7d 1D	2	862419	
2p(2P°)4s	4s 1P°	1	824280		2s(2S)8p	8p 1P°	1	874447	
2p(2P°)4p	4p 1P	1	829588		2s(2S)8d	8d 3D	1		
2p(2P°)4p	4p 3D	1	831047 +x	166			2		
		2	831213 +x	291	2p(2P°)5p	5p 1P	1	898580	
		3	831504 +x		2p(2P°)5p	5p 3D	1		
2p(2P°)4p	4p 3S	1	832251 +x				2		
2p(2P°)4p	4p 3P	0					3	899671 +x	
		1	835151 +x	170	2p(2P°)5p	5p 3P	0		
		2	835321 +x				1		
2p(2P°)4d	4d 1D°	2	837834				2	901344 +x	
2p(2P°)4p	4p 1D	2	837864		2p(2P°)5p	5p 1D	2	902442	
2s(2S)6p	6p 1P°	1	839616		2p(2P°)5d	5d 1D°	2	902592	
2s(2S)6f	6f 1F°	3	840832		2p(2P°)5d	5d 3D°	1		
2s(2S)6d	6d 3D	1					2		
		2					3	904497 +x	
		3	841220 +x		2p(2P°)5d	5d 1F°	3	906404	
2p(2P°)4d	4d 3D°	1	841280 +x	94	O VI (2S½)	Limit	-----	918702	
		2	841374 +x	123	2p(2P°)6p	6p 1P	1	935093	
		3	841497 +x		2p(2P°)6p	6p 3D	1		
2s(2S)6d	6d 1D	2	842105				2		
2p(2P°)4d	4d 3P°	2	843290 +x	-107			3	935945 +x	
		1	843397 +x	-52	2p(2P°)6p	6p 3P	0		
		0	843449 +x				1		
2p(2P°)4d	4d 1F°	3	847129				2	936805 +x	
2p(2P°)4d	4d 1P°	1	847465		2p(2P°)6p	6p 1D	2	937341	

December 1947.

O v OBSERVED TERMS*

Config. 1s²+	Observed Terms			
2s²	2s² 1S			
2s(2S)2p	{	2p 3P°		
		2p 1P°		
2p²	{	2p² 3P		
		2p² 1D		
	ns (n ≥ 3)		np (n ≥ 3)	
2s(2S)nx	3-5s 3S 3, 4s 1S		3, 4p 3P° 3-8p 1P°	
2p(2P°)nx	3s 3P° 3, 4s 1P°		3, 4p 3S 3-6p 3P 3-6p 3D 3p 1S 3-6p 1P 3-6p 1D	3-8d 3D 3-7d 1D
			3, 4d 3P° 3-5d 3D° 3, 4d 1P° 3-5d 1D°	3-5d 1F°

*For predicted terms in the spectra of the Be I isoelectronic sequence, see Introduction.

O VI

(Li I sequence; 3 electrons)

 $Z=8$ Ground state $1s^2 2s^2 S_{\frac{1}{2}}$ $2s^2 S_{\frac{1}{2}}$ 1113999.5 cm^{-1}

I. P. 138.080 volts

This spectrum has been analyzed by Edlén. The observed term values have all been taken from a manuscript generously furnished by him in advance of publication. He remarks that the $np^2 P^\circ$ and $nd^2 D$ series have been observed in the vacuum spark further than given in the table. For series members beyond $n=6$ he states that the term values calculated from a Ritz formula are probably to be preferred.

In the table, extrapolated intervals and calculated term values are entered in brackets. They have been taken from the 1933 and 1934 references below, as have also the entries in column one.

REFERENCES

- B. Edlén, Zeit. Astroph. **7**, 378 (1933). (T) (C L)
 B. Edlén, Nova Acta Reg. Soc. Sci. Uppsala [IV] **9**, No. 6, 44 (1934). (T) (C L)
 F. Tyrén, Nova Acta Reg. Soc. Sci. Uppsala [IV] **12**, No. 1, 24 (1940). (C L)
 B. Edlén, unpublished material (Sept. 1947). (T)

O VI

O VI

Edlén	Config.	Desig.	J	Level	Interval	Edlén	Config.	Desig.	J	Level	Interval
$2s^2 S_1$	$2s$	$2s^2 S$	$\frac{1}{2}$	0. 0		6 F	$6f$	$6f^2 F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	[1004265]	
$2p^2 P_1$ $2p^2 P_2$	$2p$	$2p^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	96375. 0 96907. 5	532. 5	6 GH	$6g, 6h$	$6g^2 G, \text{etc.}$	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ \text{to} \\ 5\frac{1}{2} \end{smallmatrix} \right\}$	[1004276]	
$3s^2 S_1$	$3s$	$3s^2 S$	$\frac{1}{2}$	640039. 8		7 S	$7s$	$7s^2 S$	$\frac{1}{2}$	1030780	
$3p^2 P_1$ $3p^2 P_2$	$3p$	$3p^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	666113. 2 666269. 8	156. 6	7 P	$7p$	$7p^2 P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	1032630	
$3d^2 D_2$ $3d^2 D_3$	$3d$	$3d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	674625. 7 674676. 8	51. 1	7 D	$7d$	$7d^2 D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	1033324	
$4s^2 S_1$	$4s$	$4s^2 S$	$\frac{1}{2}$	852696		7 F	$7f$	$7f^2 F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	[1033382]	
$4p^2 P_1$ $4p^2 P_2$	$4p$	$4p^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	863333. 8 863397. 7	63. 9	7 GHI	$7g, \text{etc.}$	$7g^2 G, \text{etc.}$	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ \text{to} \\ 6\frac{1}{2} \end{smallmatrix} \right\}$	[1033389]	
$4d^2 D_2$ $4d^2 D_3$	$4d$	$4d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	866880. 1 866901. 5	21. 4	8 S	$8s$	$8s^2 S$	$\frac{1}{2}$	[1050543]	
$4f^2 F_3$ $4f^2 F_4$	$4f$	$4f^2 F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	867077. 7 867087. 5	9. 8	8 P	$8p$	$8p^2 P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	1051724	
	$5s$	$5s^2 S$	$\frac{1}{2}$	948690		8 F	$8f$	$8f^2 F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	[1052280]	
$5p^2 P_1$	$5p$	$5p^2 P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	954080	[33]	8 GHIK	$8g, \text{etc.}$	$8g^2 G, \text{etc.}$	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ \text{to} \\ 7\frac{1}{2} \end{smallmatrix} \right\}$	[1052285]	
$5d^2 D_3$	$5d$	$5d^2 D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	955856	[11]						
6 S	$6s$	$6s^2 S$	$\frac{1}{2}$	1000080		8 D	$8d$	$8d^2 D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	1052296	
6 P	$6p$	$6p^2 P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	1003130							
$6d^2 D_3$	$6d$	$6d^2 D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	1004178							
							O VII (1S_0)	Limit	-----	1113999. 5	

September 1947.

O VII

(He I sequence; 2 electrons)

 $Z=8$ Ground State $1s^2\ ^1S_0$ $1s^2\ ^1S_0$ 5963000 \pm 600 cm^{-1} I. P. 739.114 \pm 0.074 volts

Five singlet lines have been observed by Tyrén in the interval 17 Å to 21 Å. He has also observed one intersystem combination—a line at 21.804 Å classified as $1s^2\ ^1S_0-2p\ ^3P_1^\circ$. His unit $10^3\ \text{cm}^{-1}$ has here been changed to cm^{-1} .

The triplet terms are from Edlén, who has kindly furnished them in advance of publication. He remarks that the extrapolated absolute term values of the triplets relative to those of the singlets confirm the intersystem combination reported by Tyrén. The $2s^3S-2p\ ^3P^\circ$ combination has apparently not been observed, but Edlén regards the extrapolation from the irregular doublet law as very reliable. Brackets are used in the table to indicate extrapolated values not yet confirmed by observation.

REFERENCES

F. Tyrén, Nova Acta Reg. Soc. Sci. Uppsala [IV] 12, No. 1, 25 (1940). (I P) (T) (C L)

B. Edlén, unpublished material (Sept. 1947). (T)

O VII

O VII

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$1s^2$	$1s^2\ ^1S$	0	0		$1s\ 3p$	$3p\ ^1P^\circ$	1	5368550	
$1s\ 2s$	$2s\ ^3S$	1	4525340		$1s\ 4p$	$4p\ ^1P^\circ$	1	5628100	
$1s\ 2p$	$2p\ ^3P^\circ$	0	[4586170]	[60] [550]	$1s\ 5p$	$5p\ ^1P^\circ$	1	5748450	
		1	4586230		$1s\ 6p$	$6p\ ^1P^\circ$	1	5813950	
		2	[4586780]						
$1s\ 2p$	$2p\ ^1P^\circ$	1	4629200						
$1s\ 3p$	$3p\ ^3P^\circ$	0, 1, 2	5356380		O VIII ($^2S_{1/2}$)	Limit		5963000	
$1s\ 3d$	$3d\ ^3D$	3, 2, 1	5364990						

September 1947.

O VIII

(H sequence; 1 electron)

 $Z=8$ Ground state $1s\ ^2S_{1/2}$ $1s\ ^2S_{1/2}$ 7027970 cm^{-1}

I. P. 871.12 volts

Tyrén has observed the first Lyman line. J. E. Mack has calculated the terms in the table, "using $R_{O^{18}}=109733.539$, and $\Lambda=0.040$. The last digit is arbitrary, since the extrapolated $1s$ -shift is $957\ \text{cm}^{-1}$. The series limits of O^{17} and O^{18} are higher than that for O^{16} by 14.3 and $25.8\ \text{cm}^{-1}$, respectively."

REFERENCES

F. Tyrén, Nova Acta Reg. Soc. Sci. Uppsala [IV] 12, No. 1, 24 (1940). (C L)

J. E. Mack, unpublished material (1949). (I P) (T) (C L)

O VIII

O VIII

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$1s$	$1s\ ^2S$	$\frac{1}{2}$	0		$3s$, etc.	$3s\ ^2S$, etc.	$\frac{1}{2}$, etc.	6246978 to 7569	
$2p$	$2p\ ^2P^\circ$	$\frac{1}{2}$	5270363]] 120 1496					
$2s$	$2s\ ^2S$	$\frac{1}{2}$	5270483						
$2p$	$2p\ ^2P^\circ$	$1\frac{1}{2}$	5271859			$\infty = \text{Limit}$		7027970	

February 1949.

FLUORINE

F I

9 electrons

 $Z=9$ Ground state $1s^2 2s^2 2p^5 {}^2P_{1\frac{1}{2}}^{\circ}$ $2p^5 {}^2P_{1\frac{1}{2}}^{\circ}$ 140553.5 cm^{-1}

I. P. 17.42 volts

This spectrum is incompletely analyzed, but the terms from the 3P limit in F II are fairly well established. The terms listed have been taken from Edlén's later paper, supplemented by levels from further recent analysis by Lidén. The new levels have been generously furnished in manuscript form by Edlén, for inclusion here.

Intersystem combinations have been observed, connecting the doublet and quartet terms.

Edlén remarks that it is impossible to assign term designations to the levels labeled $3d$ X and $4d$ X, because of the departure from LS -coupling. He also states that the terms from 1D in F II need further confirmation. They are connected with the rest by only two ultraviolet lines, those observed by Bowen at 806.92 Å and 809.60 Å.

REFERENCES

- G. H. Carragan, *Astroph. J.* **63**, 145 (1926). (Z E)
 I. S. Bowen, *Phys. Rev.* **29**, 231 (1927). (T) (C L)
 B. Edlén, *Zeit. Phys.* **93**, 447 (1935). (C L)
 B. Edlén, *Zeit. Phys.* **98**, 445 (1936). (I P) (T) (C L)
 W. F. Meggers, *J. Opt. Soc. Am.* **36**, 431 (1946). (Summary hfs)
 B. Edlén, unpublished material (Dec. 1947). (T)
 K. Lidén, *Ark. Mat. Astr. Fys.* (Stockholm) **35A**, No. 24, p. 5 (1948). (T)

F I

F I

Edlén	Config.	Desig.	J	Level	Interval	Edlén	Config.	Desig.	J	Level	Interval
$2p {}^2P_2$ 2P_1	$2s^2 2p^5$	$2p^5 {}^2P^{\circ}$	$1\frac{1}{2}$ $\frac{1}{2}$	0. 0 404. 0	-404. 0	$3p {}^2D_3$ 2D_2	$2s 2p^4({}^3P)3p$	$3p {}^2D^{\circ}$	$2\frac{1}{2}$ $1\frac{1}{2}$	117623. 73 117873. 75	-250. 02
$3s {}^4P_3$ 4P_2 4P_1	$2s^2 2p^4({}^3P)3s$	$3s {}^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	102406. 50 102681. 24 102841. 20	-274. 74 -159. 96	$3p {}^2S_1$ $3p {}^4S_2$	$2s^2 2p^4({}^3P)3p$ $2s^2 2p^4({}^3P)3p$	$3p {}^2S^{\circ}$ $3p {}^4S^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$	118406. 09 118428. 62	
$3s {}^2P_2$ 2P_1	$2s^2 2p^4({}^3P)3s$	$3s {}^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	104731. 86 105057. 10	-325. 24	$3p {}^2P_2$ 2P_1	$2s^2 2p^4({}^3P)3p$	$3p {}^2P^{\circ}$	$1\frac{1}{2}$ $\frac{1}{2}$	118937. 61 119082. 63	-145. 02
$3p {}^4P_3$ 4P_2 4P_1	$2s^2 2p^4({}^3P)3p$	$3p {}^4P^{\circ}$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	115918. 70 116041. 69 116144. 39	-122. 99 -102. 70	$\overline{3s} {}^2D_3$ 2D_2	$2s^2 2p^4({}^1D)3s$	$3s' {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	123925. 50 123926. 56	-1. 06
$3p {}^4D_4$ 4D_3 4D_2 4D_1	$2s^2 2p^4({}^3P)3p$	$3p {}^4D^{\circ}$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	116988. 21 117164. 83 117309. 37 117392. 77	-176. 62 -144. 54 -83. 40	$3d {}^4D_4$ 4D_3 4D_2 4D_1	$2s^2 2p^4({}^3P)3d$	$3d {}^4D$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	128064. 90 128088. 63 128123. 51 128185. 80	-23. 73 -34. 88 -62. 29

F I—Continued

F I—Continued

Edlén	Config.	Desig.	<i>J</i>	Level	Interval	Edlén	Config.	Desig.	<i>J</i>	Level	Interval
3d X ₈	2s ² 2p ⁴ (³ P)3d	3d Z ₄		128141. 27			2s ² 2p ⁴ (³ P)4d	4d ⁴ F	4½ 3½ 2½ 1½	133606. 39 133923. 83 133932. 56 133972. 06	—317. 44 —8. 73 —39. 50
3d ⁴ F ₅ ⁴ F ₄ ⁴ F ₃ ⁴ F ₂	2s ² 2p ⁴ (³ P)3d	3d ⁴ F	4½ 3½ 2½ 1½	128219. 92 128515. 55 128526. 15 128612. 73	—295. 63 —10. 60 —86. 58		2s ² 2p ⁴ (³ P)4d	4d Z ₃		133607. 33	
3d X ₇	2s ² 2p ⁴ (³ P)3d	3d Z ₂		128220. 65			2s ² 2p ⁴ (³ P)4d	4d Z ₂		133624. 61	
3d X ₆	2s ² 2p ⁴ (³ P)3d	3d Z ₃		128221. 16			2s ² 2p ⁴ (³ P)4d	4d Z ₁		133644. 4	
3d X ₅	2s ² 2p ⁴ (³ P)3d	3d Y ₃		128339. 53			2s ² 2p ⁴ (³ P)4d	4d Y ₃		133911. 08	
3d X ₄	2s ² 2p ⁴ (³ P)3d	3d Y ₂	1½	128524. 09			2s ² 2p ⁴ (³ P)4d	4d Y ₂		133920. 20	
3d X ₃	2s ² 2p ⁴ (³ P)3d	3d Y ₁		128606. 88			2s ² 2p ⁴ (³ P)4d	4d Y ₁		133966. 47	
3d X ₂	2s ² 2p ⁴ (³ P)3d	3d X ₂		128698. 68			2s ² 2p ⁴ (³ P)4d	4d X ₂		134085. 53	
3d X ₁	2s ² 2p ⁴ (³ P)3d	3d X ₁		128713. 12			2s ² 2p ⁴ (³ P)4d	4d X ₁		134092. 03	
	2s ² 2p ⁴ (³ P)5s	5s ⁴ P	2½ 1½ ½	132596. 26 132745. 77 133009. 96	—149. 51 —264. 19	³ p ² F ₃ ² F ₄	2s ² 2p ⁴ (¹ D)3p	3p' ² F°	2½ 3½	137594. 63 137603. 44	8. 81
	2s ² 2p ⁴ (³ P)5s	5s ² P	1½ ½	132999. 16 133224. 10	—224. 94	³ p ² D ₂ ² D ₃	2s ² 2p ⁴ (¹ D)3p	3p' ² D°	1½ 2½	138700. 15 138708. 01	7. 86
	2s ² 2p ⁴ (³ P)4d	4d ⁴ D	3½ 2½ 1½ ½	133545. 27 133558. 14 133578. 15 133614. 10	—12. 87 —20. 01 —35. 95		F II (³ P ₂)	Limit	-----	140553. 5	
	2s ² 2p ⁴ (³ P)4d	4d Z ₁		133584. 35		2p' ² S ₁	2s 2p ⁶	2p ⁶ ² S	½	[168554]	

December 1947.

F I OBSERVED TERMS*

Config. 1s ² +	Observed Terms			
2s ² 2p ⁵	2p ⁵ ² P°			
	ns (n ≥ 3)	np (n ≥ 3)		nd (n ≥ 3)
2s ² 2p ⁴ (³ P)nx	{ 3, 5s ⁴ P 3, 5s ² P	3p ⁴ S° 3p ² S°	3p ⁴ P° 3p ² P°	3p ⁴ D° 3p ² D°
2s ² 2p ⁴ (¹ D)nx'	3s' ² D	3p' ² D°		3p' ² F°
				3, 4d ⁴ D 3, 4d ⁴ F

*For predicted terms in the spectra of the F I isoelectronic sequence, see Introduction.

(O I sequence; 8 electrons)

 $Z=9$ Ground state $1s^2 2s^2 2p^4 {}^3P_2$ $2p^4 {}^3P_2$ 282190.2 cm^{-1}

I. P. 34.98 volts

Bowen, Dingle, and Edlén have all contributed to the analysis of this spectrum. The singlet and triplet terms are taken from Edlén, who has revised and extended the earlier work. The quintet terms, except $5f {}^5F$, are from Dingle's paper. The term $5f {}^5F$ derived by Edlén agrees well with the $4f {}^5F$ term and Dingle's series limit.

The singlet and triplet terms are connected by intersystem combinations. The relative position of the quintets is determined by the series with the uncertainty x probably not exceeding 200 cm^{-1} .

Edlén lists a number of combinations that probably involve $2s^2 2p^3 ({}^2D^\circ) 4f$ terms at about $288600 \pm \text{cm}^{-1}$ above the ground state.

In a private communication Edlén has stated that his term published as $\overline{3d} {}^3D$ should have the designation $\overline{4s} {}^3P$. He has also revised his published value of $3d' {}^1S^\circ$.

REFERENCES

- H. Dingle, Proc. Roy. Soc. (London) [A] **128**, 600 (1930). (T) (C L)
 I. S. Bowen, Phys. Rev. **45**, 82 (1934). (T) (C L)
 B. Edlén, Zeit. Phys. **93**, 433 (1935). (I P) (T) (C L)
 B. Edlén, private communication (Dec. 1947). (T)

F II

F II

Edlén	Config.	Desig.	J	Level	Interval	Edlén	Config.	Desig.	J	Level	Interval
$2p {}^3P_2$ 3P_1 3P_0	$2s^2 2p^4$	$2p^4 {}^3P$	2 1 0	0.0 341.8 490.6	—341.8 —148.8						
$2p {}^1D_2$	$2s^2 2p^4$	$2p^4 {}^1D$	2	20873							
$2p {}^1S_0$	$2s^2 2p^4$	$2p^4 {}^1S$	0	44919							
$2p' {}^3P_2$ 3P_1 3P_0	$2s 2p^5$	$2p^5 {}^3P^\circ$	2 1 0	164797.7 165107.1 165281.0	—309.4 —173.9						
	$2s^2 2p^3 ({}^4S^\circ) 3s$	$3s {}^5S^\circ$	2	$176654.2 + x$							
$3s {}^3S_1$	$2s^2 2p^3 ({}^4S^\circ) 3s$	$3s {}^3S^\circ$	1	182865.2							
	$2s^2 2p^3 ({}^4S^\circ) 3p$	$3p {}^5P$	1 2 3	$202609.65 + x$ $202620.98 + x$ $202640.53 + x$	11.33 19.55						
$3p {}^3P_0$ 3P_1 3P_2	$2s^2 2p^3 ({}^4S^\circ) 3p$	$3p {}^3P$	0 1 2	207702.91 207699.91 207704.61	—3.00 4.70						
$\overline{3s} {}^3D_3$ 3D_2 3D_1	$2s^2 2p^3 ({}^2D^\circ) 3s$	$3s' {}^3D^\circ$	3 2 1	211866.62 211887.69 211900.72	—21.07 —13.03						
$\overline{3s} {}^1D_2$	$2s^2 2p^3 ({}^2D^\circ) 3s$	$3s' {}^1D^\circ$	2	215069.8							
$\overline{3s} {}^1P_1$	$2s^2 2p^3 ({}^2P^\circ) 3s$	$3s'' {}^1P^\circ$	1	227228.2							
$\overline{3s} {}^3P_2$ 3P_1 3P_0	$2s^2 2p^3 ({}^2P^\circ) 3s$	$3s'' {}^3P^\circ$	2 1 0	229550.83 229552.44 229555.10	—1.61 —2.66						
	$2s^2 2p^3 ({}^4S^\circ) 3d$	$3d {}^5D^\circ$	4 3 2 1 0								
	$2s^2 2p^3 ({}^4S^\circ) 3d$	$3d {}^3D^\circ$	1 2 3								
	$2s^2 2p^3 ({}^4S^\circ) 4s$	$4s {}^5S^\circ$	2	$235311.15 + x$							
	$2s^2 2p^3 ({}^2D^\circ) 3p$	$3p' {}^1P$	1	235643.1							
	$2s^2 2p^3 ({}^2D^\circ) 3p$	$3p' {}^3D$	1 2 3	236170.35 236173.07 236195.57	2.72 22.50						
	$2s^2 2p^3 ({}^4S^\circ) 4s$	$4s {}^3S^\circ$	1	236961.63							
	$2s^2 2p^3 ({}^2D^\circ) 3p$	$3p' {}^3F$	4 3 2	237507.91 237508.72 237509.37	—0.81 —0.65						
	$2s^2 2p^3 ({}^2D^\circ) 3p$	$3p' {}^1F$	3	238323.6							
	$2s 2p^5$	$2p^5 {}^1P^\circ$	1	239605.0							
	$2s^2 2p^3 ({}^2D^\circ) 3p$	$3p' {}^3P$	2 1 0	240093.10 240153.34 240179.91	—60.24 —26.57						
	$2s^2 2p^3 ({}^2D^\circ) 3p$	$3p' {}^1D$	2	246283.9							
	$2s^2 2p^3 ({}^4S^\circ) 4p$	$4p {}^3P$	0 1 2	246655.10 246662.55 246682.67	7.45 20.12						
	$2s^2 2p^3 ({}^2P^\circ) 3p$	$3p'' {}^3S$	1	253313.2							

F II—Continued

Edlén	Config.	Desig.	<i>J</i>	Level	Interval
$4d\ ^3D_3$	$2s^2\ 2p^3(^4S^\circ)4d$	$4d\ ^3D^\circ$	1 2 3	254016	
$4f\ ^3F$	$2s^2\ 2p^3(^4S^\circ)4f$	$4f\ ^3F$	4, 3, 2	254547. 3	
$\overline{3p}\ ^3D_3$	$2s^2\ 2p^3(^2P^\circ)3p$	$3p''\ ^3D$	3	254702. 30	-15. 06 -6. 60
$\overline{3p}\ ^3D_2$			2	254717. 36	
$\overline{3p}\ ^3D_1$			1	254723. 96	
	$2s^2\ 2p^3(^4S^\circ)4f$	$4f\ ^5F$	5 to 1	254703. 1+x	
$\overline{3p}\ ^1P_1$	$2s^2\ 2p^3(^2P^\circ)3p$	$3p''\ ^1P$	1	255606. 0	
$\overline{3p}\ ^3P_0$	$2s^2\ 2p^3(^2P^\circ)3p$	$3p''\ ^3P$	0	257253. 9	14. 9 23. 9
$\overline{3p}\ ^3P_1$			1	257268. 8	
$\overline{3p}\ ^3P_2$			2	257292. 7	
$\overline{3p}\ ^1D_2$	$2s^2\ 2p^3(^2P^\circ)3p$	$3p''\ ^1D$	2	258930. 0	
$5f\ ^5F$	$2s^2\ 2p^3(^4S^\circ)5f$	$5f\ ^5F$	5 to 1	264610 +x	
$\overline{3d}\ ^3F_2$	$2s^2\ 2p^3(^2D^\circ)3d$	$3d'\ ^3F^\circ$	2	264953. 12	5. 51 7. 28
$\overline{3d}\ ^3F_3$			3	264958. 63	
$\overline{3d}\ ^3F_4$			4	264965. 91	
$\overline{3d}\ ^1S_0$	$2s^2\ 2p^3(^2D^\circ)3d$	$3d'\ ^1S^\circ$	0	264994. 9	
$\overline{3d}\ ^3G_5$	$2s^2\ 2p^3(^2D^\circ)3d$	$3d'\ ^3G^\circ$	5	265255. 8	-12. 0 -21. 5
$\overline{3d}\ ^3G_4$			4	265267. 8	
$\overline{3d}\ ^3G_3$			3	265289. 3	
$\overline{3d}\ ^1G_4$	$2s^2\ 2p^3(^2D^\circ)3d$	$3d'\ ^1G^\circ$	4	265310. 1	
$\overline{3d}\ ^3D_3$	$2s^2\ 2p^3(^2D^\circ)3d$	$3d'\ ^3D^\circ$	3	265472. 70	-26. 04 -18. 40
$\overline{3d}\ ^3D_2$			2	265498. 74	
$\overline{3d}\ ^3D_1$			1	265517. 14	
$\overline{3d}\ ^1D_2$	$2s^2\ 2p^3(^2D^\circ)3d$	$3d'\ ^1D^\circ$	2	266270. 2	
$\overline{3p}\ ^1S_0$	$2s^2\ 2p^3(^2P^\circ)3p$	$3p''\ ^1S$	0	266338. 4	
$\overline{3d}\ ^3S_1$	$2s^2\ 2p^3(^2D^\circ)3d$	$3d'\ ^3S^\circ$	1	266360. 69	
$\overline{3d}\ ^3P_2$	$2s^2\ 2p^3(^2D^\circ)3d$	$3d'\ ^3P^\circ$	2	266454. 27	-44. 85 -17. 23
$\overline{3d}\ ^3P_1$			1	266499. 12	
$\overline{3d}\ ^3P_0$			0	266516. 35	
$\overline{3d}\ ^1F_3$	$2s^2\ 2p^3(^2D^\circ)3d$	$3d'\ ^1F^\circ$	3	266548. 7	
$\overline{3d}\ ^1P_1$	$2s^2\ 2p^3(^2D^\circ)3d$	$3d'\ ^1P^\circ$	1	267400. 3	
$\overline{4s}\ ^3D_3$	$2s^2\ 2p^3(^2D^\circ)4s$	$4s'\ ^3D^\circ$	3	269548. 7	-15. 5 -10. 3
$\overline{4s}\ ^3D_2$			2	269564. 2	
$\overline{4s}\ ^3D_1$			1	269574. 5	
$\overline{4s}\ ^1D_2$	$2s^2\ 2p^3(^2D^\circ)4s$	$4s'\ ^1D^\circ$	2	270508. 4	
	F III ($^4S_{1/2}$)	<i>Limit</i>	-----	282190. 2	
$\overline{3d}\ ^3F_4$	$2s^2\ 2p^3(^2P^\circ)3d$	$3d''\ ^3F^\circ$	4	282544. 7	-25. 0 -17. 2
$\overline{3d}\ ^3F_3$			3	282569. 7	
$\overline{3d}\ ^3F_2$			2	282586. 9	
$\overline{3d}\ ^1D_2$	$2s^2\ 2p^3(^2P^\circ)3d$	$3d''\ ^1D^\circ$	2	282774. 7	
$\overline{3d}\ ^3P_0$	$2s^2\ 2p^3(^2P^\circ)3d$	$3d''\ ^3P^\circ$	0	282897. 0	16. 4 34. 5
$\overline{3d}\ ^3P_1$			1	282913. 4	
$\overline{3d}\ ^3P_2$			2	282947. 9	
$\overline{3d}\ ^1F_3$	$2s^2\ 2p^3(^2P^\circ)3d$	$3d''\ ^1F^\circ$	3	283409. 4	
$\overline{3d}\ ^1P_1$	$2s^2\ 2p^3(^2P^\circ)3d$	$3d''\ ^1P^\circ$	1	284224. 8	
$\overline{3d}\ ^3D_3$	$2s^2\ 2p^3(^2P^\circ)4s$	$4s''\ ^3P^\circ$	2	286701. 9	-4. 7 -0. 7
$\overline{3d}\ ^3D_2$			1	286706. 6	
$\overline{3d}\ ^3D_1$			0	286707. 3	

F II OBSERVED TERMS*

Config. $1s^2+$	Observed Terms	F II OBSERVED TERMS*	
		<i>nf</i> ($n \geq 4$)	$4, 5f\ ^5F$ $4f\ ^3F$
$2s^2\ 2p^4$	$2p^4\ ^1D$	<i>nd</i> ($n \geq 3$)	$3d\ ^5D^\circ$ $3, 4d\ ^3D^\circ$ $3d'\ ^3D^\circ$ $3d'\ ^1D^\circ$ $3d''\ ^1D^\circ$
$2s\ 2p^5$	$2p^5\ ^3P$ $2p^5\ ^3P^\circ$ $2p^5\ ^1P^\circ$	<i>np</i> ($n \geq 3$)	$3p\ ^5P$ $3, 4p\ ^3P$ $3p'\ ^3P$ $3p'\ ^1P$ $3p''\ ^3P$ $3p''\ ^1P$ $3p''\ ^3S$ $3p''\ ^1S$
$2s^2\ 2p^3(^4S^\circ)nx$	$2p^4\ ^1S$	<i>ns</i> ($n \geq 3$)	$3, 4s\ ^5S^\circ$ $3, 4s\ ^3S^\circ$ $3, 4s'\ ^3D^\circ$ $3, 4s'\ ^1D^\circ$ $3, 4s''\ ^3P^\circ$ $3, 4s''\ ^1P^\circ$
$2s^2\ 2p^3(^2D^\circ)nx'$			
$2s^2\ 2p^3(^2P^\circ)nx''$			

* For predicted terms in the spectra of the O I isoelectronic sequence, see Introduction.

F III

(N I sequence; 7 electrons)

Z=9

Ground state $1s^2 2s^2 2p^3 {}^4S_{1/2}$ $2p^3 {}^4S_{1/2}$ 505410 cm^{-1}

I. P. 62.646 volts

The terms are from the paper by Edlén. With the aid of observations in the extreme ultra-violet he has extended the analysis by Bowen and Dingle and derived improved values of the series limits. He has found the sextet terms and estimated their position relative to the other terms. The value of x is somewhat uncertain. Bowen found 14 intersystem combinations connecting the doublet and quartet terms.

The term $3p'' {}^2P^\circ$ depends upon the combination with $3s'' {}^2S$, assigned to a pair of lines at 2920 Å. According to Edlén this classification is somewhat uncertain.

REFERENCES

H. Dingle, Proc. Roy. Soc. (London) [A] **122**, 144 (1929). (T) (C L)I. S. Bowen, Phys. Rev. **45**, 82 (1934). (T) (C L)B. Edlén, Zeit. Phys. **93**, 433 (1935). (I P) (T) (C L)

F III

F III

Edlén	Config.	Desig.	J	Level	Interval	Edlén	Config.	Desig.	J	Level	Interval
$2p {}^4S_2$	$2s^2 2p^3$	$2p^3 {}^4S^\circ$	$1\frac{1}{2}$	0		$3s {}^2P_1$	$2s^2 2p^2({}^3P)3s$	$3s {}^2P$	$\frac{1}{2}$	324489. 9	384. 5
$2p {}^2D_3$	$2s^2 2p^3$	$2p^3 {}^2D^\circ$	$2\frac{1}{2}$	34084	-36	2P_2			$1\frac{1}{2}$	324874. 4	
2D_2			$1\frac{1}{2}$	34120		$\overline{3s} {}^2D_3$	$2s^2 2p^2({}^1D)3s$	$3s' {}^2D$	$2\frac{1}{2}$	344016. 2	-3. 3
						2D_2			$1\frac{1}{2}$	344019. 5	
$2p {}^2P_{12}$	$2s^2 2p^3$	$2p^3 {}^2P^\circ$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ \frac{1}{2} \end{array} \right\}$	51558		$3p {}^2S_1$	$2s^2 2p^2({}^3P)3p$	$3p {}^2S^\circ$	$\frac{1}{2}$	344438. 4	
$2p' {}^4P_3$	$2s 2p^4$	$2p^4 {}^4P$	$2\frac{1}{2}$	151897. 9	-337. 4	$3p {}^4D_1$	$2s^2 2p^2({}^3P)3p$	$3p {}^4D^\circ$	$\frac{1}{2}$	348700. 5	114. 9
4P_2			$1\frac{1}{2}$	152235. 3		4D_2			$1\frac{1}{2}$	348815. 4	
4P_1			$\frac{1}{2}$	152410. 0		4D_3			$2\frac{1}{2}$	349005. 1	
						4D_4			$3\frac{1}{2}$	349264. 0	
$2p' {}^2D_3$	$2s 2p^4$	$2p^4 {}^2D$	$2\frac{1}{2}$	210240	-16	$3p {}^4P_1$	$2s^2 2p^2({}^3P)3p$	$3p {}^4P^\circ$	$\frac{1}{2}$	351234. 1	94. 3
2D_2			$1\frac{1}{2}$	210256		4P_2			$1\frac{1}{2}$	351328. 4	
$2p' {}^2S_1$	$2s 2p^4$	$2p^4 {}^2S$	$\frac{1}{2}$	248260		4P_3			$2\frac{1}{2}$	351517. 1	
$2p' {}^2P_2$	$2s 2p^4$	$2p^4 {}^2P$	$1\frac{1}{2}$	266559	-384	$3p {}^2D_2$	$2s^2 2p^2({}^3P)3p$	$3p {}^2D^\circ$	$1\frac{1}{2}$	355979. 6	390. 4
2P_1			$\frac{1}{2}$	266943		2D_3			$2\frac{1}{2}$	356370. 0	
$3s {}^4P_1$	$2s^2 2p^2({}^3P)3s$	$3s {}^4P$	$\frac{1}{2}$	316707. 3	211. 3	$3p {}^4S_2$	$2s^2 2p^2({}^3P)3p$	$3p {}^4S^\circ$	$1\frac{1}{2}$	357477. 0	
4P_2			$1\frac{1}{2}$	316918. 6							
4P_3			$2\frac{1}{2}$	317237. 5							

F III—Continued

F III—Continued

Edlén	Config.	Desig.	<i>J</i>	Level	Interval	Edlén	Config.	Desig.	<i>J</i>	Level	Interval
$3p\ ^2P_1$ 2P_2	$2s^2\ 2p^2(^3P)3p$	$3p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	360346. 2 360433. 1	86. 9	$4p\ ^4D_1$ 4D_2 4D_3 4D_4	$2s^2\ 2p^2(^3P)4p$	$4p\ ^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	426426. 0 426556. 4 426730. 7 426987. 5	130. 4 174. 3 256. 8
$\overline{3s}\ ^2S_1$	$2s^2\ 2p^2(^1S)3s$	$3s''\ ^2S$	$\frac{1}{2}$	372673. 0		$4p\ ^4P_1$ 4P_2 4P_3	$2s^2\ 2p^2(^3P)4p$	$4p\ ^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	427456. 7 427542. 4 427729. 3	85. 7 186. 9
$\overline{3p}\ ^2F_3$ 2F_4	$2s^2\ 2p^2(^1D)3p$	$3p'\ ^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	376806. 2 376871. 0	64. 8	$4p\ ^2D_2$ 2D_3	$2s^2\ 2p^2(^3P)4p$	$4p\ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	429105. 3 429500. 6	395. 3
$\overline{3p}\ ^2D_3$ 2D_2	$2s^2\ 2p^2(^1D)3p$	$3p'\ ^2D^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$	380242. 9 380299. 1	-56. 2	$4p\ ^2P_1$ 2P_2	$2s^2\ 2p^2(^3P)4p$	$4p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	431057. 1 431224. 2	167. 1
$\overline{3p}\ ^2P_1$ 2P_2	$2s^2\ 2p^2(^1D)3p$	$3p'\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	384350. 9 384485. 2	134. 3	$3p'\ ^4P_3$ 4P_2 4P_1	$2s\ 2p^3(^5S^\circ)3p$	$3p''' ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	434546. 3 434567. 0 434581. 6	-20. 7 -14. 6
$3d\ ^4F_2$ 4F_3 4F_4 4F_5	$2s^2\ 2p^2(^3P)3d$	$3d\ ^4F$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	387257. 3 387366. 2 387521. 8 387725. 5	108. 9 155. 6 203. 7	$\overline{4s}\ ^2D_{23}$	$2s^2\ 2p^2(^1D)4s$	$4s'\ ^2D$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	440830	
$3d\ ^2P_2$ 2P_1	$2s^2\ 2p^2(^3P)3d$	$3d\ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	389523. 5 389735. 7	-212. 2	$4d\ ^2P_2$ 2P_1	$2s^2\ 2p^2(^3P)4d$	$4d\ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	441159 441384	-225
$3d\ ^4D_1$ 4D_2 4D_3 4D_4	$2s^2\ 2p^2(^3P)3d$	$3d\ ^4D$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	390118. 4 390078. 3 390075. 7 390208. 4	-40. 1 -2. 6 132. 7	$4d\ ^4P_3$ 4P_2 4P_1	$2s^2\ 2p^2(^3P)4d$	$4d\ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	442153 442300 442378	-147 -78
$3d\ ^4P_3$ 4P_2 4P_1	$2s^2\ 2p^2(^3P)3d$	$3d\ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	390832. 3 390974. 0 391045. 2	-141. 7 -71. 2	$4d\ ^2F_3$ 2F_4	$2s^2\ 2p^2(^3P)4d$	$4d\ ^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	442280 442634	354
$3d\ ^2F_3$ 2F_4	$2s^2\ 2p^2(^3P)3d$	$3d\ ^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	391255. 6 391625. 5	369. 9	$\overline{3d}\ ^2D_{23}$	$2s^2\ 2p^2(^1S)3d$	$3d''\ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	442760	
$3s'\ ^6S_3$	$2s\ 2p^3(^5S^\circ)3s$	$3s''' ^6S^\circ$	$2\frac{1}{2}$	391910. 0 +x		$4d\ ^2D_2$ 2D_3	$2s^2\ 2p^2(^3P)4d$	$4d\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	444960 445008	48
$3d\ ^2D_2$ 2D_3	$2s^2\ 2p^2(^3P)3d$	$3d\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	395266. 1 395384. 1	118. 0	$3d'\ ^6D_5$ 6D_4 6D_3 6D_2 6D_1	$2s\ 2p^3(^5S^\circ)3d$	$3d''' ^6D^\circ$	$4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	462930. 1+x 462932. 7+x 462936. 5+x 462939. 9+x 462942. 4+x	-2. 6 -3. 8 -3. 4 -2. 5
$2p''\ ^2P_2$ 2P_1	$2p^5$	$2p^5\ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	401203 401721	-518	$5d\ ^4P_3$ $^4P_{12}$	$2s^2\ 2p^2(^3P)5d$	$5d\ ^4P$	$2\frac{1}{2}$ $\left\{ \begin{array}{l} 1\frac{1}{2} \\ \frac{1}{2} \end{array} \right\}$	465409 465541	-132
$3s'\ ^4S_2$	$2s\ 2p^3(^5S^\circ)3s$	$3s''' ^4S^\circ$	$1\frac{1}{2}$	404778		$5d\ ^2D_{23}$	$2s^2\ 2p^2(^3P)5d$	$5d\ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	466293	
$\overline{3p}\ ^2P_1$ 2P_2	$2s^2\ 2p^2(^1S)3p$	$3p''\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	406899. 2 406903. 3	4. 1	$4d\ ^2F_{34}$	$2s^2\ 2p^2(^1D)4d$	$4d'\ ^2F$	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	466810	
$\overline{3d}\ ^2F_4$ 2F_3	$2s^2\ 2p^2(^1D)3d$	$3d'\ ^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	413136. 1 413187. 1	-51. 0	$4d\ ^2D_{23}$	$2s^2\ 2p^2(^1D)4d$	$4d'\ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	466964	
$\overline{3d}\ ^2G_5$ 2G_4	$2s^2\ 2p^2(^1D)3d$	$3d'\ ^2G$	$4\frac{1}{2}$ $3\frac{1}{2}$	414887. 0 414890. 1	-3. 1	$4d\ ^2P_{12}$	$2s^2\ 2p^2(^1D)4d$	$4d'\ ^2P$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	467798	
$4s\ ^4P_1$ 4P_2 4P_3	$2s^2\ 2p^2(^3P)4s$	$4s\ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	415188 415714		$3d'\ ^4D_4$ 4D_3 $^4D_{12}$	$2s\ 2p^3(^5S^\circ)3d$	$3d''' ^4D^\circ$	$3\frac{1}{2}$ $2\frac{1}{2}$ $\left\{ \begin{array}{l} 1\frac{1}{2} \\ \frac{1}{2} \end{array} \right\}$	467868. 9 467869. 3 467870. 3	-0. 4 -1. 0
$\overline{3d}\ ^2D_2$ 2D_3	$2s^2\ 2p^2(^1D)3d$	$3d'\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	416160. 7 416178. 1	17. 4	$\overline{3s'}\ ^2D_3$ 2D_2	$2s\ 2p^3(^3D^\circ)3s$	$3s^{1\nu}\ ^2D^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$	474369 474413	-44
$4s\ ^2P_1$ 2P_2	$2s^2\ 2p^2(^3P)4s$	$4s\ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	417581 417968	387	$\overline{5d}\ ^2F_{34}$	$2s^2\ 2p^2(^1D)5d$	$5d'\ ^2F$	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	489494	
$\overline{3d}\ ^2P_1$ 2P_2	$2s^2\ 2p^2(^1D)3d$	$3d'\ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	418180. 6 418240. 9	60. 3	$\overline{5d}\ ^2D_{23}$	$2s^2\ 2p^2(^1D)5d$	$5d'\ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	490140	
$\overline{3d}\ ^2S_1$	$2s^2\ 2p^2(^1D)3d$	$3d'\ ^2S$	$\frac{1}{2}$	420997. 9							
$3p'\ ^6P_2$ 6P_3 6P_4	$2s\ 2p^3(^5S^\circ)3p$	$3p''' ^6P$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	425239. 6 +x 425261. 3 +x 425297. 4 +x	21. 7 36. 1						
$4p\ ^2S_1$	$2s^2\ 2p^2(^3P)4p$	$4p\ ^2S^\circ$	$\frac{1}{2}$	425388. 9							
							F IV(3P_0)	Limit	-----	505410	

F III Observed Terms*

Config. $1s^2+$	Observed Terms		
$2s^2 2p^3$	$\left\{ \begin{array}{lll} 2p^3 {}^4S^\circ & 2p^3 {}^2P^\circ & 2p^3 {}^2D^\circ \end{array} \right.$		
$2s 2p^4$	$\left\{ \begin{array}{lll} 2p^4 {}^2S & 2p^4 {}^4P & 2p^4 {}^2D \\ & 2p^4 {}^2P & \end{array} \right.$		
$2p^5$	$2p^5 {}^2P^\circ$		
	$ns (n \geq 3)$	$np (n \geq 3)$	$nd (n \geq 3)$
$2s^2 2p^2({}^3P)nx$	$\left\{ \begin{array}{l} 3, 4s {}^4P \\ 3, 4s {}^2P \end{array} \right.$	$\begin{array}{lll} 3p {}^4S^\circ & 3, 4p {}^4P^\circ & 3, 4p {}^4D^\circ \\ 3, 4p {}^2S^\circ & 3, 4p {}^2P^\circ & 3, 4p {}^2D^\circ \end{array}$	$\begin{array}{lll} 3-5d {}^4P & 3d {}^4D & 3d {}^4F \\ 3, 4d {}^2P & 3-5d {}^2D & 3, 4d {}^2F \end{array}$
$2s^2 2p^2({}^1D)nx'$	$3, 4s' {}^2D$	$3p' {}^2P^\circ \quad 3p' {}^2D^\circ \quad 3p' {}^2F^\circ$	$3d' {}^2S \quad 3, 4d' {}^2P \quad 3-5d' {}^2D \quad 3-5d' {}^2F \quad 3d' {}^2G$
$2s^2 2p^2({}^1S)nx''$	$3s'' {}^2S$	$3p'' {}^2P^\circ$	$3d'' {}^2D$
$2s 2p^3({}^6S^\circ)nx'''$	$\left\{ \begin{array}{l} 3s''' {}^6S^\circ \\ 3s''' {}^4S^\circ \end{array} \right.$	$\begin{array}{l} 3p''' {}^6P \\ 3p''' {}^4P \end{array}$	$\begin{array}{l} 3d''' {}^6D^\circ \\ 3d''' {}^4D^\circ \end{array}$
$2s 2p^3({}^3D^\circ)nx^{IV}$	$3s^{IV} {}^2D^\circ$		

*For predicted terms in the spectra of the N I isoelectronic sequence, see Introduction.

F IV

(C I sequence; 6 electrons)

$Z=9$

Ground state $1s^2 2s^2 2p^2 {}^3P_0$

$2p^2 {}^3P_0$ 703766.4 cm^{-1}

I. P. 87.23 volts

The first work on this spectrum was by Bowen. Edlén has greatly extended the earlier analysis. About 250 lines in the intervals 140 to 679 Å and 2171 to 3176 Å are now classified. The terms are from Edlén, who has rejected two terms in his published list, $4d' {}^3S$ and $\overline{3s'} {}^3S$. Extrapolated values are entered in brackets in the table.

The singlet and triplet terms are connected by intersystem combinations. No such combinations involving quintet terms have been observed. The uncertainty α may reach 50 to 100 cm^{-1} .

REFERENCES

- B. Edlén, Zeit. Phys. **92**, 19 (1934). (I P) (T) (C L)
 B. Edlén, private communication (Dec. 1947). (T)

F IV

F IV

Edlén	Config.	Desig.	<i>J</i>	Level	Interval	Edlén	Config.	Desig.	<i>J</i>	Level	Interval
$2p\ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s^2\ 2p^3$	$2p^2\ ^3P$	0 1 2	0.0 225.2 613.4	225.2 388.2	$3p'\ ^5P_1$ $\ ^5P_2$ $\ ^5P_3$	$2s\ 2p^2(^4P)3p$	$3p\ ^5P^\circ$	1 2 3	$542578.3+x$ $542693.2+x$ $542895.2+x$	114.9 202.0
$2p\ ^1D_2$	$2s^2\ 2p^3$	$2p^2\ ^1D$	2	25241		$3p'\ ^3D_1$ $\ ^3D_2$ $\ ^3D_3$	$2s\ 2p^2(^4P)3p$	$3p\ ^3D^\circ$	1 2 3	550918 551098 551866	180 268
$2p\ ^1S_0$	$2s^2\ 2p^3$	$2p^2\ ^1S$	0	53544			$2s\ 2p^2(^4P)3p$	$3p\ ^3P^\circ$	0		
$2p'\ ^5S_2$	$2s\ 2p^3$	$2p^3\ ^5S^\circ$	2	$74506+x$		$3p'\ ^3P_1$ $\ ^3P_2$			1 2	556051 556316	265
$2p'\ ^3D_3$ $\ ^3D_2$ $\ ^3D_1$	$2s\ 2p^3$	$2p^3\ ^3D^\circ$	3 2 1	147841.8 147888.9 147901.6	-47.1 -12.7	$4s\ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s^2\ 2p(^2P^\circ)4s$	$4s\ ^3P^\circ$	0 1 2	559747 559881 560304	134 423
$2p'\ ^3P_2$ $\ ^3P_1$ $\ ^3P_0$	$2s\ 2p^3$	$2p^3\ ^3P^\circ$	2 1 0	175237.0 175242.0 175264.1	-5.0 -22.1	$4s\ ^1P_1$	$2s^2\ 2p(^2P^\circ)4s$	$4s\ ^1P^\circ$	1	561267	
$2p'\ ^1D_2$	$2s\ 2p^3$	$2p^3\ ^1D^\circ$	2	228908		$\overline{3s'}\ ^3D_1$ $\ ^3D_2$ $\ ^3D_3$	$2s\ 2p^2(^2D)3s$	$3s'\ ^3D$	1 2 3	567900 568019 568175	119 156
$2p'\ ^3S_1$	$2s\ 2p^3$	$2p^3\ ^3S^\circ$	1	238297.2		$3d'\ ^5F_1$ $\ ^5F_2$ $\ ^5F_3$ $\ ^5F_4$ $\ ^5F_5$	$2s\ 2p^2(^4P)3d$	$3d\ ^5F$	1 2 3 4 5	$[576581]+x$ $576656.1+x$ $576768.2+x$ $576916.6+x$ $577100.1+x$	$[75]$ 112.1 148.4 183.5
$2p'\ ^1P_1$	$2s\ 2p^3$	$2p^3\ ^1P^\circ$	1	257390		$3d'\ ^5D_0$ $\ ^5D_1$ $\ ^5D_2$ $\ ^5D_3$ $\ ^5D_4$	$2s\ 2p^2(^4P)3d$	$3d\ ^5D$	0 1 2 3 4	$581806.1+x$ $581811.5+x$ $581828.6+x$ $581872.3+x$ $581977.6+x$	5.4 17.1 43.7 105.3
$2p''\ ^3P_2$ $\ ^3P_1$ $\ ^3P_0$	$2p^4$	$2p^4\ ^3P$	2 1 0	348327.0 348770.0 348963.0	-443.0 -193.0	$3d'\ ^5P_3$ $\ ^5P_2$ $\ ^5P_1$	$2s\ 2p^2(^4P)3d$	$3d\ ^5P$	3 2 1	$583547+x$ $583697+x$ $583798+x$	-150 -101
$3s\ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s^2\ 2p(^2P^\circ)3s$	$3s\ ^3P^\circ$	0 1 2	416417.3 416639.8 417143.4	222.5 503.6	$3d'\ ^3P_2$ $\ ^3P_1$ $\ ^3P_0$	$2s\ 2p^2(^4P)3d$	$3d\ ^3P$	2 1 0	585201 585425 585531	-224 -106
$3s\ ^1P_1$	$2s^2\ 2p(^2P^\circ)3s$	$3s\ ^1P^\circ$	1	423606.4		$\overline{3s'}\ ^1D_2$	$2s\ 2p^2(^2D)3s$	$3s'\ ^1D$	2	586263	
$3p\ ^3D_1$ $\ ^3D_2$ $\ ^3D_3$	$2s^2\ 2p(^2P^\circ)3p$	$3p\ ^3D$	1 2 3	451819.6 452081.1 452517.1	261.5 436.0	$4d\ ^3F_2$	$2s^2\ 2p(^2P^\circ)4d$	$4d\ ^3F^\circ$	2 3 4	586641	
$3p\ ^3S_1$	$2s^2\ 2p(^2P^\circ)3p$	$3p\ ^3S$	1	456884.3		$4d\ ^1D_2$	$2s^2\ 2p(^2P^\circ)4d$	$4d\ ^1D^\circ$	2	587130	
$3p\ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s^2\ 2p(^2P^\circ)3p$	$3p\ ^3P$	0 1 2	460215.2 460385.8 460640.6	170.6 254.8	$3d'\ ^3F_2$ $\ ^3F_3$ $\ ^3F_4$	$2s\ 2p^2(^4P)3d$	$3d\ ^3F$	2 3 4	588021 588223 588478	202 255
$3p\ ^1D_2$	$2s^2\ 2p(^2P^\circ)3p$	$3p\ ^1D$	2	469644.2		$4d\ ^3D_1$ $\ ^3D_2$ $\ ^3D_3$	$2s^2\ 2p(^2P^\circ)4d$	$4d\ ^3D^\circ$	1 2 3	589109 589188 589406	79 218
$3d\ ^3F_2$ $\ ^3F_3$ $\ ^3F_4$	$2s^2\ 2p(^2P^\circ)3d$	$3d\ ^3F^\circ$	2 3 4	492395.1 492358.8 493206.2	463.7 347.4	$4d\ ^3P_2$ $\ ^3P_1$ $\ ^3P_0$	$2s^2\ 2p(^2P^\circ)4d$	$4d\ ^3P^\circ$	2 1 0	590024 590201 590262	-177 -61
$3d\ ^1D_2$	$2s^2\ 2p(^2P^\circ)3d$	$3d\ ^1D^\circ$	2	492864		$4d\ ^1F_3$	$2s^2\ 2p(^2P^\circ)4d$	$4d\ ^1F^\circ$	3	592240	
$3d\ ^3D_1$ $\ ^3D_2$ $\ ^3D_3$	$2s^2\ 2p(^2P^\circ)3d$	$3d\ ^3D^\circ$	1 2 3	497481.4 497575.6 497729.1	94.2 153.5	$4d\ ^1P_1$	$2s^2\ 2p(^2P^\circ)4d$	$4d\ ^1P^\circ$	1	592674	
$3d\ ^3P_2$ $\ ^3P_1$ $\ ^3P_0$	$2s^2\ 2p(^2P^\circ)3d$	$3d\ ^3P^\circ$	2 1 0	500390.1 500602.1 500716.5	-212.0 -114.4	$3d'\ ^3D_1$ $\ ^3D_2$ $\ ^3D_3$	$2s\ 2p^2(^4P)3d$	$3d\ ^3D$	1 2 3	595331 595403 595481	72 78
$3s'\ ^5P_1$ $\ ^5P_2$ $\ ^5P_3$	$2s\ 2p^2(^4P)3s$	$3s\ ^5P$	1 2 3	$502723.0+x$ $502964.4+x$ $503282.4+x$	241.4 318.0	$\overline{3p'}\ ^1F_3$	$2s\ 2p^2(^2D)3p$	$3p'\ ^1F^\circ$	3	609811	
$3d\ ^1F_3$	$2s^2\ 2p(^2P^\circ)3d$	$3d\ ^1F^\circ$	3	505421.4		$\overline{3p'}\ ^1D_2$	$2s\ 2p^2(^2D)3p$	$3p'\ ^1D^\circ$	2	612830	
$3d\ ^1P_1$	$2s^2\ 2p(^2P^\circ)3d$	$3d\ ^1P^\circ$	1	506514		$\overline{3p'}\ ^1P_1$	$2s\ 2p^2(^2D)3p$	$3p'\ ^1P^\circ$	1	618889	
$3s'\ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s\ 2p^2(^4P)3s$	$3s\ ^3P$	0 1 2	519341 519539 519890	198 351	$5d\ ^3F_2$	$2s^2\ 2p(^2P^\circ)5d$	$5d\ ^3F^\circ$	2 3 4	629547	
$3p'\ ^3S_1$	$2s\ 2p^2(^4P)3p$	$3p\ ^3S^\circ$	1	534686							
$3p'\ ^5D_0$ $\ ^5D_1$ $\ ^5D_2$ $\ ^5D_3$ $\ ^5D_4$	$2s\ 2p^2(^4P)3p$	$3p\ ^5D^\circ$	0 1 2 3 4	$[538507]+x$ $538573.3+x$ $538709.2+x$ $538909.8+x$ $539166.1+x$	$[66]$ 135.9 200.6 256.3						

December 1947.

F IV OBSERVED TERMS*

*For predicted terms in the spectra of the C I isoelectronic sequence, see Introduction.

F V

(B I sequence; 5 electrons)

 $Z=9$ Ground state $1s^2 2s^2 2p \ ^2P_{\frac{1}{2}}^{\circ}$ $2p \ ^2P_{\frac{1}{2}}^{\circ} \ 921450 \text{ cm}^{-1}$

I. P. 114.214 volts

All of the terms are from an unpublished manuscript kindly furnished by Edlén. He has revised and extended his earlier analysis. The notation in the left column is from his published papers.

No intersystem combinations have been observed. The position of the quartet terms relative to the doublets may be in error by $\pm 100 \text{ cm}^{-1}$ according to Edlén. This uncertainty is indicated by x in the table.

REFERENCES

B. Edlén, Zeit. Phys. **89**, 597 (1934); **92**, 26 (1934); **94**, 56 (1935). (I P) (T) (C L).

B. Edlén, unpublished material (Dec. 1947). (I P) (T).

F V

F V

Edlén	Config.	Desig.	J	Level	Interval	Edlén	Config.	Desig.	J	Level	Interval
$2p \ ^2P_1$ 2P_2	$2s^2(^1S)2p$	$2p \ ^2P^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$	0 746	746	$3s' \ ^2P_1$ 2P_2	$2s \ 2p(^3P^{\circ})3s$	$3s \ ^2P^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$	638856 639365	509
$2p' \ ^4P_1$ 4P_2 4P_3	$2s \ 2p^3$	$2p^2 \ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$86035+x$ $86287+x$ $86651+x$	252 364	$3p' \ ^2P_1$ 2P_2	$2s \ 2p(^3P^{\circ})3p$	$3p \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	656208 656436	228
$2p' \ ^2D_3$ 2D_2	$2s \ 2p^2$	$2p^2 \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	152876 152898	-22	$3p' \ ^4D_1$ 4D_2 4D_3 4D_4	$2s \ 2p(^3P^{\circ})3p$	$3p \ ^4D$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	$657988+x$ $658134+x$ $658390+x$ $658791+x$	146 256 401
$2p' \ ^2S_1$	$2s \ 2p^2$	$2p^2 \ ^2S$	$\frac{1}{2}$	197565		$3p' \ ^4S_2$	$2s \ 2p(^3P^{\circ})3p$	$3p \ ^4S$	$1\frac{1}{2}$	$666240+x$	
$2p' \ ^2P_1$ 2P_2	$2s \ 2p^2$	$2p^2 \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	214881 215348	467	$3p' \ ^2D_2$ 2D_3	$2s \ 2p(^3P^{\circ})3p$	$3p \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	675932 676422	490
$2p' \ ^4S_2$	$2p^3$	$2p^3 \ ^4S^{\circ}$	$1\frac{1}{2}$	$276657+x$		$3p' \ ^2S_1$	$2s \ 2p(^3P^{\circ})3p$	$3p \ ^2S$	$\frac{1}{2}$	687806	
$2p'' \ ^2D_3$ 2D_2	$2p^3$	$2p^3 \ ^2D^{\circ}$	$2\frac{1}{2}$ $1\frac{1}{2}$	307226 307273	-47	$3d' \ ^4D_{12}$ 4D_3 4D_4	$2s \ 2p(^3P^{\circ})3d$	$3d \ ^4D^{\circ}$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \\ 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	$697817+x$ $697919+x$ $698055+x$	102 136
$2p'' \ ^2P_1$ 2P_2	$2p^3$	$2p^3 \ ^2P^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$	347418 347438	20	$3d' \ ^2D_2$ 2D_3	$2s \ 2p(^3P^{\circ})3d$	$3d \ ^2D^{\circ}$	$1\frac{1}{2}$ $2\frac{1}{2}$	699293 699389	96
$3s \ ^2S_1$	$2s^2(^1S)3s$	$3s \ ^2S$	$\frac{1}{2}$	524751		$3d' \ ^4P_3$ 4P_2 4P_1	$2s \ 2p(^3P^{\circ})3d$	$3d \ ^4P^{\circ}$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	$702908+x$ $703117+x$ $703259+x$	-209 -142
$3p \ ^2P_1$ 2P_2	$2s^2(^1S)3p$	$3p \ ^2P^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$	565367 565544	177	$\overline{3s'} \ ^2P_{12}$	$2s \ 2p(^1P^{\circ})3s$	$3s' \ ^2P^{\circ}$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	712755	
$3d \ ^2D_2$ 2D_3	$2s^2(^1S)3d$	$3d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	602476 602516	40						
$3s' \ ^4P_1$ 4P_2 4P_3	$2s \ 2p(^3P^{\circ})3s$	$3s \ ^4P^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$621138+x$ $621395+x$ $621863+x$	257 468						

F v—Continued

F v—Continued

Edlén	Config.	Desig.	<i>J</i>	Level	Interval	Eglén	Config.	Desig.	<i>J</i>	Level	Interval	
$3d' \ ^2F_3$ $\ ^2F_4$	$2s \ 2p(^3P^\circ)3d$	$3d \ ^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	712840 713306	466		$2s \ 2p(^3P^\circ)4d$	$4d \ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	841598 841695	97	
$4s \ ^2S_1$	$2s^2(^1S)4s$	$4s \ ^2S$	$\frac{1}{2}$	712936		$4d' \ ^4P_3$	$2s \ 2p(^3P^\circ)4d$	$4d \ ^4P^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	842452+x		
$3d' \ ^2P_2$ $\ ^2P_1$	$2s \ 2p(^3P^\circ)3d$	$3d \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	718472 718691	-219		$2s^2(^1S)6d$	$6d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	843497		
$4d \ ^2D_2$ $\ ^2D_3$	$2s^2(^1S)4d$	$4d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	744010 744036	26		$\overline{3p''} \ ^2F_3$ $\ ^2F_4$	$2p^2(^1D)3p$	$3p''' \ ^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	844112 844266	154
$\overline{3p'} \ ^2D_2$ $\ ^2D_3$	$2s \ 2p(^1P^\circ)3p$	$3p' \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	751406 751452	46		$4d' \ ^2F_3$ $\ ^2F_4$	$2s \ 2p(^3P^\circ)4d$	$4d \ ^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	847506 847817	311
$\overline{3p'} \ ^2P_1$ $\ ^2P_2$	$2s \ 2p(^1P^\circ)3p$	$3p' \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	752529 753656	127		$2p^2(^3P)3d$	$3d'' \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	853035 853442	-407	
$\overline{3p'} \ ^2S_1$	$2s \ 2p(^1P^\circ)3p$	$3p' \ ^2S$	$\frac{1}{2}$	760342			$2p^2(^1D)3p$	$3p''' \ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	854971		
$\overline{3d'} \ ^2F_{34}$	$2s \ 2p(^1P^\circ)3d$	$3d' \ ^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	783650		$3d'' \ ^4P_3$ $\ ^4P_2$ $\ ^4P_1$	$2p^2(^3P)3d$	$3d'' \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	860421+x 860619+x 860725+x	-198 -106	
$3s'' \ ^4P_1$ $\ ^4P_2$ $\ ^4P_3$	$2p^2(^3P)3s$	$3s'' \ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	784343+x 784604+x 785014+x	261 410	$\overline{3d''} \ ^2D$	$2p^2(^1D)3d$	$3d''' \ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	873904		
$\overline{3d'} \ ^2D_2$ $\ ^2D_3$	$2s \ 2p(^1P^\circ)3d$	$3d' \ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	787725 787764	39	$\overline{3d''} \ ^2F_{34}$	$2p^2(^1D)3d$	$3d''' \ ^2F$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	880312		
$\overline{3d'} \ ^2P_{12}$	$2s \ 2p(^1P^\circ)3d$	$3d' \ ^2P^\circ$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	793308		$\overline{3d''} \ ^2P_1$ $\ ^2P_2$	$2p^2(^1D)3d$	$3d''' \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	882930 883083	153	
$3s'' \ ^2P_1$ $\ ^2P_2$	$2p^2(^3P)3s$	$3s'' \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	797059 797519	460		$2s \ 2p(^3P^\circ)5s$	$5s \ ^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	892180+x		
$5d \ ^2D_3$	$2s^2(^1S)5d$	$5d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	808663 808677	14		$2s \ 2p(^3P^\circ)5p$	$5p \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	901487 902012	525	
$4s' \ ^4P_3$	$2s \ 2p(^3P^\circ)4s$	$4s \ ^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	810298+x			$2s \ 2p(^3P^\circ)5d$	$5d \ ^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	906074+x		
$\overline{3s''} \ ^2D$	$2p^2(^1D)3s$	$3s''' \ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	811075		$5d' \ ^4D$	$2s \ 2p(^3P^\circ)5d$	$5d \ ^4P^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	906565+x		
	$2p^2(^3P)3p$	$3p'' \ ^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	816518+x 816759+x 817101+x	241 342		FvI (1S_0)	<i>Limit</i>	-----	921450		
$3p'' \ ^4D_4$	$2p^2(^3P)3p$	$3p'' \ ^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	823375+x 823625+x	250		$2s \ 2p(^3P^\circ)6d$	$6d \ ^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	940921+x		
$3p'' \ ^4P_3$	$2s \ 2p(^3P^\circ)4p$	$4p \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	829436 829707	271		$2s \ 2p(^3P^\circ)6d$	$6d \ ^4P^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	941286+x		
$4p' \ ^2D_2$ $\ ^2D_3$	$2s \ 2p(^3P^\circ)4p$	$4p \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	833501 833920	419		$2p^2(^3P)4d$	$4d'' \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	998189+x		
$3p'' \ ^4S_2$	$2p^2(^3P)3p$	$3p'' \ ^4S^\circ$	$1\frac{1}{2}$	834790+x								
	$2s \ 2p(^3P^\circ)4p$	$4p \ ^2S$	$\frac{1}{2}$	838036								
	$2s \ 2p(^3P^\circ)4d$	$4d \ ^4D^\circ$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 1\frac{1}{2} \\ 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	841037+x 841095+x 841305+x	58 210							
$4d' \ ^4D_4$												

December 1947.

F V OBSERVED TERMS*

Config. $1s^2 +$	Observed Terms		
$2s^2(1S)2p$	$2p \ ^2P^\circ$		
$2s \ 2p^2$	$\left\{ \begin{array}{lll} 2p^2 \ ^2S & 2p^2 \ ^4P & 2p^2 \ ^2D \\ & 2p^2 \ ^2P & \end{array} \right.$		
$2p^3$	$\left\{ \begin{array}{lll} 2p^3 \ ^4S^\circ & 2p^3 \ ^2P^\circ & 2p^3 \ ^2D^\circ \end{array} \right.$		
	$ns \ (n \geq 3)$	$np \ (n \geq 3)$	$nd \ (n \geq 3)$
$2s^2(1S)nx$	$3, 4s \ ^2S$	$3p \ ^2P^\circ$	$3-6d \ ^2D$
$2s \ 2p(^3P^\circ)nx$	$\left\{ \begin{array}{l} 3-5s \ ^4P^\circ \\ 3s \ ^2P^\circ \end{array} \right.$	$\begin{array}{lll} 3p \ ^4S & & 3p \ ^4D \\ 3, 4p \ ^2S & 3, 4p \ ^2P & 3-5p \ ^2D \end{array}$	$\begin{array}{lll} 3-6d \ ^4P^\circ & 3-6d \ ^4D^\circ & \\ 3d \ ^2P^\circ & 3, 4d \ ^2D^\circ & 3, 4d \ ^2F^\circ \end{array}$
$2s \ 2p(^1P^\circ)nx'$	$3s' \ ^2P^\circ$	$\begin{array}{lll} 3p' \ ^2S & 3p' \ ^2P & 3p' \ ^2D \end{array}$	$\begin{array}{lll} 3d' \ ^2P^\circ & 3d' \ ^2D^\circ & 3d' \ ^2F^\circ \end{array}$
$2p^2(^3P)nx''$	$\left\{ \begin{array}{l} 3s'' \ ^4P \\ 3s'' \ ^2P \end{array} \right.$	$\begin{array}{lll} 3p'' \ ^4S^\circ & 3p'' \ ^4P^\circ & 3p'' \ ^4D^\circ \end{array}$	$\begin{array}{l} 3, 4d'' \ ^4P \\ 3d'' \ ^2P \end{array}$
$2p^2(^1D)nx'''$	$3s''' \ ^2D$	$3p''' \ ^2D^\circ \ 3p''' \ ^2F^\circ$	$3d''' \ ^2P \ 3d''' \ ^2D \ 3d''' \ ^2F$

*For predicted terms in the spectra of the B I isoelectronic sequence, see Introduction.

F VI

(Be I sequence; 4 electrons)

$Z=9$

Ground state $1s^2 2s^2 1S_0$

$2s^2 1S_0$ 1267581 cm^{-1}

I. P. 157.117 volts

Edlén has revised and extended his published analysis and has generously furnished a manuscript copy of his complete term list in advance of publication, for inclusion here.

In the published papers he has used a prime to designate the terms from the $^2P^\circ$ limit in F VII.

Intersystem combinations connecting the singlet and triplet systems of terms, have been observed.

REFERENCES

- B. Edlén, Zeit. Phys. **89**, 179 (1934). (I P) (T) (C L)
 B. Edlén, Zeit. Phys. **94**, 56 (1935). (T) (C L)
 B. Edlén, unpublished material (Dec. 1947). (I P) (T)

F VI

F VI

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$2s^2$	$2s^2\ ^1S$	0	0		$2p(^2P^o)3d$	$3d\ ^3P^o$	2 1 0	938524 938811 938958	-287 -147
$2s(^2S)2p$	$2p\ ^3P^o$	0 1 2	96601 96861 97437	260 576	$2p(^2P^o)3d$	$3d\ ^1F^o$	3	947305	
$2s(^2S)2p$	$2p\ ^1P^o$	1	186841		$2p(^2P^o)3d$	$3d\ ^1P^o$	1	953402	
$2p^2$	$2p^2\ ^3P$	0 1 2	251341 251635 252145	294 510	$2s(^2S)4s$	$4s\ ^3S$	1	989928	
$2p^2$	$2p^2\ ^1D$	2	274597		$2s(^2S)4s$	$4s\ ^1S$	0	997693	
$2p^2$	$2p^2\ ^1S$	0	340424		$2s(^2S)4p$	$4p\ ^1P^o$	1	1007852	
$2s(^2S)3s$	$3s\ ^3S$	1	747298		$2s(^2S)4d$	$4d\ ^3D$	1 2 3	1014439	
$2s(^2S)3s$	$3s\ ^1S$	0	764392		$2s(^2S)4d$	$4d\ ^1D$	2	1019363	
$2s(^2S)3p$	$3p\ ^1P^o$	1	787833		$2s(^2S)5s$	$5s\ ^3S$	1	1093463	
$2s(^2S)3p$	$3p\ ^3P^o$	0 1 2	790326 790474	148	$2s(^2S)5p$	$5p\ ^1P^o$	1	1099409	
$2s(^2S)3d$	$3d\ ^3D$	1, 2 3	812169 812208	39	$2s(^2S)5d$	$5d\ ^3D$	1 2 3	1106417	
$2s(^2S)3d$	$3d\ ^1D$	2	826853		$2s(^2S)5d$	$5d\ ^1D$	2	1108712	
$2p(^2P^o)3s$	$3s\ ^3P^o$	0 1 2	871160 871441 872078	281 637	$2p(^2P^o)4s$	$4s\ ^1P^o$	1	1112328	
$2p(^2P^o)3s$	$3s\ ^1P^o$	1	884290		$2p(^2P^o)4p$	$4p\ ^1P$	1	1115967	
$2p(^2P^o)3p$	$3p\ ^1P$	1	895287		$2p(^2P^o)4p$	$4p\ ^3D$	1 2 3	1117498 1117741 1118273	243 532
$2p(^2P^o)3p$	$3p\ ^3D$	1 2 3	900442 900785 901397	343 612	$2p(^2P^o)4p$	$4p\ ^3S$	1	1121377	
$2p(^2P^o)3p$	$3p\ ^3S$	1	909316		$2p(^2P^o)4p$	$4p\ ^3P$	0 1 2	1122468 1122662	194
$2p(^2P^o)3p$	$3p\ ^3P$	0 1 2	915196 915420 915770	224 350	$2p(^2P^o)4p$	$4p\ ^1D$	2	1126152	
$2p(^2P^o)3d$	$3d\ ^1D^o$	2	921821		$2p(^2P^o)4d$	$4d\ ^1D^o$	2	1126168	
$2p(^2P^o)3p$	$3p\ ^1D$	2	925393		$2p(^2P^o)4d$	$4d\ ^3D^o$	1 2 3	1130339	
$2p(^2P^o)3d$	$3d\ ^3D^o$	1 2 3	933586 933717 933920	131 203	$2p(^2P^o)4d$	$4d\ ^3P^o$	2 1 0	1131653 1131857	-204
$2p(^2P^o)3p$	$3p\ ^1S$	0	934633		$2p(^2P^o)4d$	$4d\ ^1F^o$	3	1135953	
					$2p(^2P^o)4d$	$4d\ ^1P^o$	1	1137535	

F VI—Continued

F VI—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$2s(^2S)6p$	$6p\ ^1P^\circ$	1	1154428		$2p(^2P^\circ)5d$	$5d\ ^3D^\circ$	1 2 3	1220940	
$2s(^2S)6d$	$6d\ ^3D$	1 2 3	1156097		$2p(^2P^\circ)5d$	$5d\ ^3P^\circ$	2 1 0	1221541	
$2s(^2S)6d$	$6d\ ^1D$	2	1157385		$2p(^2P^\circ)5d$	$5d\ ^1F^\circ$	3	1223598	
$2s(^2S)7p$	$7p\ ^1P^\circ$	1	1184469		$2p(^2P^\circ)5d$	$5d\ ^1P^\circ$	1	1224285	
$2s(^2S)7d$	$7d\ ^3D$	1 2 3	1185884		$2p(^2P^\circ)6p$	$6p\ ^3D$	1 2 3	1266672	
$3s(^2S)7d$	$7d\ ^1D$	2	1186611		F VII ($^2S_{1/2}$)	<i>Limit</i>	-----	1267581	
$2s(^2S)8d$	$8d\ ^3D$	1 2 3	1205139		$2p(^2P^\circ)6p$	$6p\ ^3P$	0 1 2	1267616	
$2p(^2P^\circ)5p$	$5p\ ^3D$	1 2 3	1215055		$2p(^2P^\circ)6p$	$6p\ ^1D$	2	1268554	
$2p(^3P^\circ)5p$	$5p\ ^3P$	0 1 2	1216995		$2p(^2P^\circ)6d$	$6d\ ^3D^\circ$	1 2 3	1269888	
$2p(^3P^\circ)5p$	$5p\ ^1D$	2	1218588		$2p(^2P^\circ)6d$	$6d\ ^1F^\circ$	3	1271437	
$2p(^2P^\circ)5d$	$5d\ ^1D^\circ$	2	1218786		$2p(^2P^\circ)7d$	$7d\ ^3D^\circ$	1 2 3	1299418	

December 1947.

F VI OBSERVED TERMS*

Config. $1s^2 +$	Observed Terms					
$2s^2$	$2s^2\ ^1S$					
$2s(^2S)2p$	{ $2p\ ^3P^\circ$ $2p\ ^1P^\circ$					
$2p^2$	{ $2p^2\ ^1S$ $2p^2\ ^3P$ $2p^2\ ^1D$					
	$ns\ (n \geq 3)$		$np\ (n \geq 3)$		$nd\ (n \geq 3)$	
$2s(^2S)nx$	{ $3-5s\ ^3S$ $3, 4s\ ^1S$		$3p\ ^3P^\circ$ $3-7p\ ^1P^\circ$		$3-8d\ ^3D$ $3-7d\ ^1D$	
$2p(^2P^\circ)nx$	{ $3s\ ^3P^\circ$ $3, 4s\ ^1P^\circ$		$3, 4p\ ^3S$ $3p\ ^1S$	$3-6p\ ^3P$ $3, 4p\ ^1P$	$3-6p\ ^3D$ $3-6p\ ^1D$	$3-5d\ ^3P^\circ$ $3-5d\ ^1P^\circ$ $3-7d\ ^3D^\circ$ $3-5d\ ^1D^\circ$ $3-6d\ ^1F^\circ$

*For predicted terms in the spectra of the Be I isoelectronic sequence, see Introduction.

F VII

(Li I sequence; 3 electrons)

$Z=9$

Ground state $1s^2 2s^2 S_{\frac{1}{2}}$

$2s^2 S_{\frac{1}{2}}$ 1493656 cm^{-1}

I. P. 185.139 volts

The analysis is by Edlén, who, in 1934, published a list of nine classified lines in the range between 86 Å and 134 Å. He has recently extended the analysis and has generously furnished his unpublished term list for use in the present compilation. All terms in the table have been taken from the later list, although the entries in column one are from the earlier paper.

Edlén remarks that the $np^2 P^\circ$ and $nd^2 D$ series have been observed in the vacuum spark further than indicated in the table, but beyond $n=6$ the term values calculated from a Ritz formula are probably to be preferred.

REFERENCES

- B. Edlén, Zeit. Phys. **89**, 179 (1934). (T) (C L)
 B. Edlén, unpublished material (Sept. 1947). (I P) (T)

F VII

F VII

Edlén	Config.	Desig.	J	Level	Interval	Edlén	Config.	Desig.	J	Level	Interval
$2s^2 S$	$2s$	$2s^2 S$	$\frac{1}{2}$	0			$6s$	$6s^2 S$	$\frac{1}{2}$	1339216	
$2p^2 P_1$ $2p^2 P_2$	$2p$	$2p^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	112258 113235	977		$6p$	$6p^2 P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	1342877	
$3s^2 S$	$3s$	$3s^2 S$	$\frac{1}{2}$	854625			$6d$	$6d^2 D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	1344141	
$3p^2 P_1$ $3p^2 P_2$	$3p$	$3p^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	885136 885418	282		$7s$	$7s^2 S$	$\frac{1}{2}$	1380775	
$3d^2 D_2$ $3d^2 D_3$	$3d$	$3d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	895632 895722	90		$7p$	$7p^2 P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	1382858	
$4s^2 S$	$4s$	$4s^2 S$	$\frac{1}{2}$	1140416			$7d$	$7d^2 D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	1383841	
$4p^2 P_2$	$4p$	$4p^2 P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	1152977			$8p$	$8p^2 P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	1408848	
$4d^2 D_3$	$4d$	$4d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	1157223 1157255	32		$8d$	$8d^2 D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	1409538	
	$5s$	$5s^2 S$	$\frac{1}{2}$	1269826							
	$5p$	$5p^2 P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	1276194							
$5d^2 D_3$	$5d$	$5d^2 D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	1278404							
							F VIII ($1S_0$)	Limit	-----	1493656	

September 1947.

F VIII

(He I sequence; 2 electrons)

 $Z=9$ Ground state $1s^2 1S_0$ $1s^2 1S_0$ 7693400 \pm 800 cm^{-1} I. P. 953.60 \pm 0.10 volts

Flemberg has classified three lines between 13 Å and 16 Å as the first three members of the singlet series. Tyrén has also observed the first two members of this series and classified a line at 16.951 Å as the intersystem combination $1s^2 1S_0 - 2p^3 P_1^o$. Tyrén's value of the limit is quoted here. The unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

Edlén has extended the analysis and has generously furnished his unpublished manuscript containing absolute values of the triplet terms extrapolated along the He I isoelectronic sequence. The relative positions of the singlet and triplet terms thus determined confirm the intersystem combination reported by Tyrén. The $2s^3 S - 2p^3 P^o$ combination has apparently not been observed, but Edlén regards the extrapolation from the irregular doublet law as very reliable. Brackets are used in the table to denote extrapolated values not yet confirmed by observation.

REFERENCES

- F. Tyrén, Nova Acta Reg. Soc. Sci. Uppsala [IV] **12**, No. 1, 25 (1940). (I P) (T) (C L)
 H. Flemberg, Ark. Mat. Astr. Fys. (Stockholm) **28A**, No. 18 p. 34 (1942). (T) (C L)
 B. Edlén, unpublished material (Sept. 1947). (T)

F VIII

F VIII

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$1s^3$	$1s^2 1S$	0	0		$1s 3d$	$3d^3 D$	3, 2, 1	[6912360]	
$1s 2s$	$2s^3 S$	1	[5829920]		$1s 3p$	$3p^1 P^o$	1	6916590	
$1s 2p$	$2p^3 P^o$	0	[5899150]	[160] [950]	$1s 4p$	$4p^1 P^o$	1	7256680	
		1	5899310						
		2	[5900260]						
$1s 2p$	$2p^1 P^o$	1	5949900		F IX ($^3S_{1/2}$)	Limit	-----	7693400	

September 1947.

NEON

Ne I

10 electrons

 $Z=10$ Ground state $1s^2 2s^2 2p^6 {}^1S_0$ $2p^6 {}^1S_0$ 173931.7 cm^{-1}

I. P. 21.559 volts

The present list has been compiled from an unpublished manuscript kindly furnished by Edlén, who has made a study of the terms of this spectrum and interpreted them with the aid of present atomic theory. His term array is based on that published by Meggers and Humphreys in 1933, although he has revised and extended their list. Three place values are from measures made with the interferometer. His predicted values of five f -levels are entered in brackets in the table.

Edlén has determined the new values of the series limits quoted here.

The classical work by Paschen on Ne I forms the basis of all subsequent investigations. His notation has, therefore, been retained in column one of the table, except for his fractional numerical prefixes for levels from an s -configuration, $m=1.5, 2.5$, etc., which are listed as 1, 2, etc., in accord with the 1933 term table mentioned above. The letters U, V, X, Y, Z adopted later when configurations involving f -electrons were found, are also entered in this column. Eleven levels in the latter group have J -values fixed by the observed combinations listed in the 1933 reference below. These J -values are entered in italics in the table.

Edlén suggested that a pair-coupling notation be adopted for Ne-like spectra to take into account the departure from LS -coupling. According to Shortley, LS -designations can be significantly assigned in only a few cases, in particular, for the following groups of levels:

Paschen	Desig.	Paschen	Desig.	Paschen	Desig.
$(n-2)s_5$	$ns {}^3P_2$	$2p_{10}$	$3p {}^3S_1$	$2p_5$	$3p {}^1P_1$
$(n-2)s_4$	$ns {}^3P_1$				
$(n-2)s_3$	$ns {}^3P_0$	$2p_9$	$3p {}^3D_3$	$2p_4$	$3p {}^3P_2$
		$2p_8$	$3p {}^3D_2$	$2p_3$	$3p {}^3P_0$
$(n-2)s_2$	$ns {}^1P_1$	$2p_7$	$3p {}^3D_1$	$2p_2$	$3p {}^3P_1$
		$2p_6$	$3p {}^1D_2$	$2p_1$	$3p {}^1S_0$

Consequently, the $j\bar{l}$ -coupling notation in the general form suggested by Racah is here introduced. The present arrangement has been suggested by Shortley, who has made a detailed investigation of the theoretical arrangement of the "pairs," to be used as a guide in preparing the present table. Pairs are separated only the case of $np [\frac{1}{2}]$, where the interval is large.

Twenty lines of Ne I in the range between 5852 Å and 7032 Å have been measured relative to the primary standard, and are regarded as accurate to eight figures. They have been adopted by the International Astronomical Union as secondary standards of wavelength.

Ne I—Continued

REFERENCES

- F. Paschen, Ann. der Phys. [4] **60**, 405 (1919). (T) (C L)
 W. Grotrian, Phys. Zeit. **21**, 638 (1920). (G D)
 F. Paschen, Ann. der Phys. [4] **63**, 201 (1920). (T) (C L)
 E. Back, Ann. der Phys. [4] **76**, 330 (1925). (Z E)
 N. Ryde, Zeit. Phys. **59**, 836 (1929). (T) (C L)
 K. Murakawa and T. Iwama, Sci. Papers Inst. Phys. Chem. Research (Tokyo) **13**, No. 254, 289 (1930). (Z E)
 W. F. Meggers and C. J. Humphreys, Bur. Std. J. Research **10**, 429, RP540 (1933). (T) (C L)
 J. C. Boyce, Phys. Rev. **46**, 378 (1934). (I P) (T) (C L)
 W. F. Meggers, J. Research Nat. Bur. Std. **14**, 490, RP781 (1935). (C L)
 Trans. Intern. Astr. Union **5**, 86 (1935).
 Y. Ishida and T. Tamura, Sci. Papers Inst. Phys. Chem. Research (Tokyo) **29**, 9 (1936). (T) (C L)
 P. Jacquinet, Compt. Rend. **202**, 1578 (1936). (Z E)
 R. Ritschl und H. Schober, Phys. Zeit. **38**, 6 (1937). (I S)
 C. J. Humphreys, J. Research Nat. Bur. Std. **20**, 24, RP1061 (1938). (T) (C L)
 J. B. Green and J. A. Peoples, Jr., Phys. Rev. **54**, 602 (1938). (Z E)
 G. Racah, Phys. Rev. **61**, 537 (L) (1942).
 B. Edlén, Ark. Mat. Astr. Fys. (Stockholm) **29A**, No. 32, p. 2 (1943). (C L)
 J. B. Green, Phys. Rev. **64**, 151 (1943). (Z E)
 B. Edlén, unpublished material (March 1948). (I P) (T)
 G. Shortley, unpublished material (Aug. 1947).

Ne I

Ne I

Paschen	Config.	Desig.	<i>J</i>	Level	Obs. <i>g</i>	Paschen	Config.	Desig.	<i>J</i>	Level	Obs. <i>g</i>
	$2p^6$	$2p^6\ ^1S$	0	0							
						$3s_1'''$	$2p^5(^2P_{3/2}^o)3d$	$3d' [2\frac{1}{2}]^o$	2	162410. 617	0. 781
						$3s_1''$			3	162412. 138	1. 125
$1s_5$	$2p^5(^2P_{1/2}^o)3s$	$3s [1\frac{1}{2}]^o$	2	134043. 790	1. 503	$3s_1''$	"	$3d' [1\frac{1}{2}]^o$	2	162421. 944	1. 242
$1s_4$			1	134461. 237	1. 464	$3s_1'$			1	162437. 642	0. 752
$1s_3$	$2p^5(^2P_{3/2}^o)3s$	$3s' [1\frac{1}{2}]^o$	0	134820. 591							
$1s_2$			1	135890. 670	1. 034	$3p_{10}$	$2p^5(^2P_{1/2}^o)4p$	$4p [1\frac{1}{2}]$	1	162519. 850	1. 929
	$2p^5(^2P_{1/2}^o)3p$	$3p [1\frac{1}{2}]$	1	148259. 746	1. 984	$3p_9$	"	$4p [2\frac{1}{2}]$	3	162832. 683	1. 328
$2p_{10}$			3	149659. 000	1. 329	$3p_8$			2	162901. 093	1. 112
$2p_9$	"	$3p [2\frac{1}{2}]$	2	149826. 181	1. 137	$3p_7$	"	$4p [1\frac{1}{2}]$	1	163014. 600	0. 974
$2p_8$			2	150123. 551	0. 669	$3p_6$			2	163040. 330	1. 360
$2p_7$	"	$3p [1\frac{1}{2}]$	1	150317. 821	1. 229	$3p_5$	"	$4p [1\frac{1}{2}]$	0	163403. 281	
$2p_6$			2	150919. 391		$3p_4$	$2p^5(^2P_{3/2}^o)4p$	$4p' [1\frac{1}{2}]$	1	163659. 248	0. 685
$2p_5$	"	$3p [1\frac{1}{2}]$	0	150919. 391		$3p_3$			2	163710. 581	1. 184
$2p_4$	$2p^5(^2P_{3/2}^o)3p$	$3p' [1\frac{1}{2}]$	1	150774. 072	0. 999	$3p_2$	"	$4p' [1\frac{1}{2}]$	1	163709. 699	1. 397
$2p_3$			2	150860. 468	1. 301	$3p_1$			0	164287. 864	
$2p_2$	"	$3p' [1\frac{1}{2}]$	1	151040. 413	1. 340						
$2p_1$			0	152972. 697		$3s_5$	$2p^5(^2P_{1/2}^o)5s$	$5s [1\frac{1}{2}]^o$	2	165830. 144	1. 492
						$3s_4$			1	165914. 756	1. 207
$2s_5$	$2p^5(^2P_{1/2}^o)4s$	$4s [1\frac{1}{2}]^o$	2	158603. 070		$3s_3$	$2p^5(^2P_{3/2}^o)5s'$	$5s' [1\frac{1}{2}]^o$	0	166608. 309	
$2s_4$			1	158797. 954		$3s_2$			1	166658. 484	1. 295
$2s_3$	$2p^5(^2P_{3/2}^o)4s$	$4s' [1\frac{1}{2}]^o$	0	159381. 94							
$2s_2$			1	159536. 57		$4d_6$	$2p^5(^2P_{1/2}^o)4d$	$4d [1\frac{1}{2}]^o$	0	166969. 639	
						$4d_5$			1	166977. 321	1. 391
$3d_6$	$2p^5(^2P_{1/2}^o)3d$	$3d [1\frac{1}{2}]^o$	0	161511. 590		$4d_4'$	"	$4d [3\frac{1}{2}]^o$	4	167002. 007	1. 251
$3d_5$			1	161526. 134	1. 397	$4d_4$			3	167003. 104	1. 040
$3d_4'$	"	$3d [3\frac{1}{2}]^o$	4	161592. 308	1. 249	$4d_3$	"	$4d [1\frac{1}{2}]^o$	2	167013. 535	1. 322
$3d_4$			3	161594. 081	1. 034	$4d_2$			1	167028. 957	0. 812
$3d_3$	"	$3d [1\frac{1}{2}]^o$	2	161609. 222	1. 356	$4d_1''$	"	$4d [2\frac{1}{2}]^o$	2	167049. 580	0. 990
$3d_2$			1	161638. 581	0. 860	$4d_1'$			3	167050. 639	1. 248
$3d_1''$	"	$3d [2\frac{1}{2}]^o$	2	161701. 623	0. 948	$4s_1'''$	$2p^5(^2P_{3/2}^o)4d$	$4d' [2\frac{1}{2}]^o$	2	167796. 939	0. 783
$3d_1'$			3	161703. 413	1. 249	$4s_1''$			3	167797. 865	1. 116

Ne I—Continued

Paschen	Config.	Desig.	<i>J</i>	Level	Obs. <i>g</i>
4s ₁ ^{''} 4s ₁	2p ⁵ (² P _{3/2} ^o)4d	4d' [1½] ^o	2 1	167798. 914 167809. 722	1. 230 0. 797
4X	2p ⁵ (² P _{3/2})4f	4f [1½]	1, 2	167054. 59	
4V	"	4f [4½]	4, 5	[167062. 5]	
4Y	"	4f [2½]	2, 3	167071. 08	
4Z	"	4f [3½]	3, 4	[167079. 1]	
4U	2p ⁵ (² P _{3/2} ^o)4f	4f' [2½]	2, 3	167848. 67	
4p ₁₀	2p ⁵ (² P _{1/2})5p	5p [½]	1	167451. 44	
4p ₉ 4p ₈	"	5p [2½]	3 2	167561. 03 167593. 18	
4p ₇ 4p ₆	"	5p [1½]	1 2	167641. 53 167650. 60	
4p ₃	"	5p [½]	0	167869. 17	
4p ₅ 4p ₄	2p ⁵ (² P _{3/2} ^o)5p	5p' [1½]	1 2	168357. 44 168380. 69	
4p ₂ 4p ₁	"	5p' [½]	1 0	168360. 57 168588. 83	
4s ₅ 4s ₄	2p ⁵ (² P _{1/2})6s	6s [1½] ^o	2 1	168926. 626 168969. 328	1. 500 1. 184
4s ₃ 4s ₂	2p ⁵ (² P _{3/2} ^o)6s	6s' [½] ^o	0 1	169707. 899 169729. 602	1. 313
5d ₆ 5d ₅	2p ⁵ (² P _{1/2})5d	5d [½] ^o	0 1	169484. 98 169490. 414	1. 383
5d ₄ ['] 5d ₄	"	5d [3½] ^o	4 3	169503. 612 169504. 258	1. 093
5d ₃ 5d ₂	"	5d [1½] ^o	2 1	169510. 540 169518. 977	1. 298 0. 791
5d ₁ ^{''} 5d ₁	"	5d [2½] ^o	2 3	169528. 241 169528. 862	
5s ₁ ^{'''} 5s ₁	2p ⁵ (² P _{3/2} ^o)5d	5d' [2½] ^o	2 3	170291. 291 170291. 650	
5s ₁ ^{''} 5s ₁	"	5d' [1½] ^o	2 1	170290. 934 170297. 98	1. 251 0. 809
5X	2p ⁵ (² P _{1/2})5f	5f [1½]	1, 2	169532. 22	
5V	"	5f [4½]	4, 5	[169536. 3]	
5Y	"	5f [2½]	2, 3	169540. 88	
5Z	"	5f [3½]	3, 4	[169545. 0]	
5U	2p ⁵ (² P _{3/2} ^o)5f	5f' [2½]	2, 3	170319. 71	
5p ₁₀	2p ⁵ (² P _{1/2})6p	6p [½]	1	169750. 11	
5p ₉ 5p ₈	"	6p [2½]	3 2	169799. 15 169816. 60	
5p ₇ 5p ₆	"	6p [1½]	1 2	169841. 45 169845. 79	

Ne I—Continued

Paschen	Config.	Desig.	<i>J</i>	Level	Obs. <i>g</i>
5p ₃	2p ⁵ (² P _{1/2})6p	6p [½]	0	169978. 70	
5p ₅ 5p ₄	2p ⁵ (² P _{3/2} ^o)6p	6p' [1½]	1 2	170586. 94 170599. 19	
5p ₂ 5p ₁	"	6p' [½]	1 0	170580. 35 170691. 32	
5s ₅ 5s ₄	2p ⁵ (² P _{1/2})7s	7s [1½] ^o	2 1	170534. 694 170559. 032	
5s ₃ 5s ₂	2p ⁵ (² P _{3/2} ^o)7s	7s' [½] ^o	0 1	171314. 84 171325. 997	1. 315
6d ₆ 6d ₅	2p ⁵ (² P _{1/2})6d	6d [½] ^o	0 1	170850. 252 170853. 315	1. 389
6d ₄ ['] 6d ₄	"	6d [3½] ^o	4 3	170860. 447 170860. 850	
6d ₃ 6d ₂	"	6d [1½] ^o	2 1	170864. 959 170869. 927	1. 331 0. 783
6d ₁ ^{''} 6d ₁	"	6d [2½] ^o	2 3	170874. 840 170875. 216	0. 971
6s ₁ ^{'''} 6s ₁	2p ⁵ (² P _{3/2} ^o)6d	6d' [2½] ^o	2 3	171644. 139 171644. 434	
6s ₁ ^{''} 6s ₁	"	6d' [1½] ^o	2 1	171641. 951 171646. 87	0. 857
6X	2p ⁵ (² P _{1/2})6f	6f [1½]	1, 2	170877. 72	
6V	"	6f [4½]	4, 5	170879. 95	
6Y	"	6f [2½]	2, 3	170882. 65	
6Z	"	6f [3½]	3, 4	170884. 95	
6U	2p ⁵ (² P _{3/2} ^o)6f	6f' [3½]	3, 4	171661. 87	
6p ₁₀	2p ⁵ (² P _{1/2})7p	7p [½]	1	171011. 31	
6p ₉ 6p ₈	"	7p [2½]	3 2	171034. 80 171045. 65	
6p ₇ 6p ₆	"	7p [1½]	1 2	171059. 96 171062. 18	
6p ₃	"	7p [½]	0	171150. 81	
6p ₅ 6p ₄	2p ⁵ (² P _{3/2} ^o)7p	7p' [1½]	1 2	171824. 2 171830. 0	
6p ₂ 6p ₁	"	7p' [½]	1 0	171832. 7 171915. 46	
6s ₅ 6s ₄	2p ⁵ (² P _{1/2})8s	8s [1½] ^o	2 1	171475. 295 171491. 464	
6s ₃ 6s ₂	2p ⁵ (² P _{3/2} ^o)8s	8s' [½] ^o	0 1	172256. 31 172263. 720	
7d ₆ 7d ₅	2p ⁵ (² P _{1/2})7d	7d [½] ^o	0 1	171671. 14 171673. 90	
7d ₄ ['] 7d ₄	"	7d [3½] ^o	4 3	171677. 455 171677. 714	

Ne I—Continued

Paschen	Config.	Desig.	<i>J</i>	Level	Obs. <i>g</i>
7d ₃ 7d ₂	2p ⁵ (² P _{1/2})7d	7d [1½] ^o	2 1	171683. 331 171684. 902	
7d' ₁ 7d' ₁	"	7d [2½] ^o	2 3	171687. 268 171687. 518	
7s ₁ ^{'''} 7s ₁ ^{'''}	2p ⁵ (² P _{3/2})7d	7d' [2½] ^o	2 3	172460. 407 172460. 602	
7s ₁ ^{''} 7s ₁ ^{''}	"	7d' [1½] ^o	2 1	172459. 85 172463. 02	
7X	2p ⁵ (² P _{1/2})7f	7f [1½]	1, 2	171688. 57	
7V	"	7f [4½]	4, 5	171689. 95	
7Y	"	7f [2½]	2, 3	171692. 07	
7Z	"	7f [3½]	3, 4	171693. 32	
7U	2p ⁵ (² P _{3/2})7f	{ 7f' [3½] 7f' [2½]	{ 3, 4 2, 3	172471. 45	
7p ₁₀	2p ⁵ (² P _{1/2})8p	8p [½]	1	171754. 2	
7p ₉ 7p ₈	"	8p [2½]	3 2	171789. 0 171793. 7	
7p ₇ 7p ₆	"	8p [1½]	1 2	171800. 3 171805. 1	
7p ₃	"	8p [½]	0	171833. 0	
7p ₄	2p ⁵ (² P _{3/2})8p	8p' [1½]	1 2	172575. 4	
7p ₂ 7p ₁	"	8p' [½]	1 0	172564. 8 172601. 7	
7s ₅ 7s ₄	2p ⁵ (² P _{1/2})9s	9s [1½] ^o	2 1	172073. 375 172082. 895	
7s ₃ 7s ₂	2p ⁵ (² P _{3/2})9s	9s' [½] ^o	0 1	172854. 12 172858. 96	
8d ₆ 8d ₅	2p ⁵ (² P _{1/2})8d	8d [½] ^o	0 1	172202. 33 172203. 86	
8d' ₄ 8d ₄	"	8d [3½] ^o	4 3	172207. 110 172207. 278	
8d ₃ 8d ₂	"	8d [1½] ^o	2 1	172208. 77 172211. 10	
8d' ₁ 8d' ₁	"	8d [2½] ^o	2 3	172213. 094 172213. 249	
8s ₁ ^{'''} 8s ₁ ^{'''}	2p ⁵ (² P _{3/2})8d	8d' [2½] ^o	2 3	172989. 185 172989. 263	
8s ₁ ^{''} 8s ₁ ^{''}	"	8d' [1½] ^o	2 1	172989. 06 172990. 96	
8X	2p ⁵ (² P _{1/2})8f	8f [1½]	1, 2	172214. 66	
8V	"	8f [4½]	4, 5	172215. 54?	
8Y	"	8f [2½]	2, 3	[172216. 7]	
8Z	"	8f [3½]	3, 4	172217. 64	

Ne I—Continued

8U	2p ⁵ (² P _{3/2})8f	{ 8f' [3½] 8f' [2½]	{ 3, 4 2, 3	172996. 63	
8p ₁₀	2p ⁵ (² P _{1/2})9p	9p [½]	1	172270. 4	
8p ₉ 8p ₈	"	9p [2½]	3 2	172284. 2 172288. 8	
8p _{7,6}	"	9p [1½]	1, 2	172293. 4	
8p ₃	"	9p [½]	0	172329. 3	
8p ₄	2p ⁵ (² P _{3/2})9p	9p' [1½]	1 2	173067. 4	
8p ₁	"	9p' [½]	1 0	173099. 3	
8s ₅ 8s ₄	2p ⁵ (² P _{1/2})10s	10s [1½] ^o	2 1	172477. 303 172483. 84	
8s ₃ 8s ₂	2p ⁵ (² P _{3/2})10s	10s' [½] ^o	0 1	173257. 24 173261. 41	
9d ₆ 9d ₅	2p ⁵ (² P _{1/2})9d	9d [½] ^o	0 1	172566. 85 172567. 88	
9d' ₄ 9d ₄	"	9d [3½] ^o	4 3	172569. 840 172570. 064	
9d ₃ 9d ₂	"	9d [1½] ^o	2 1	172571. 37 172572. 82	
9d' ₁ 9d' ₁	"	9d [2½] ^o	2 3	172574. 12 172574. 22	
9s ₁ ^{'''} 9s ₁ ^{'''}	2p ⁵ (² P _{3/2})9d	9d' [2½] ^o	2 3	173351. 45 173351. 50	
9s ₁ ^{''} 9s ₁ ^{''}	"	9d' [1½] ^o	2 1	173351. 49 173352. 75	
9V	2p ⁵ (² P _{1/2})9f	9f [4½]	4, 5	172575. 83	
9Y	"	9f [2½]	2, 3	172576. 8	
9Z	"	9f [3½]	3, 4	172577. 3	
9p ₁₀	2p ⁵ (² P _{1/2})10p	10p [½]	1	172621. 0	
9p ₉ 9p ₈	"	10p [2½]	3 2	172625. 2	
9p _{7,6}	"	10p [1½]	1, 2	172632. 2	
9p ₃	"	10p [½]	0	172667. 1	
9s ₅ 9s ₄	2p ⁵ (² P _{1/2})11s	11s [1½] ^o	2 1	172761. 79 172766. 55	
9s ₃ 9s ₂	2p ⁵ (² P _{3/2})11s	11s' [½] ^o	0 1	173542. 00 173545. 28	
10d ₆ 10d ₅	2p ⁵ (² P _{1/2})10d	10d [½] ^o	0 1	172826. 54 172827. 42	

Ne I—Continued

Ne I—Continued

Paschen	Config.	Desig.	<i>J</i>	Level	Obs. <i>g</i>	Paschen	Config.	Desig.	<i>J</i>	Level	Obs. <i>g</i>
10d ₄ ' 10d ₄	2p ⁵ (² P _{1/2} ^o)10d	10d [3½] ^o	4 3	172829. 11 172829. 20		11s ₁ ''' 11s ₁ ''	2p ⁵ (² P _{3/2} ^o)11d	11d' [2½] ^o	2 3	173802. 27 173802. 33	
10d ₃ 10d ₂	"	10d [1½] ^o	2 1	172829. 87 172831. 28		11s ₁ '	"	11d' [1½] ^o	2 1	173802. 75	
10d ₁ '' 10d ₁ '	"	10d [2½] ^o	2 3	172832. 20 172832. 24		11s ₈ 11s ₄	2p ⁵ (² P _{1/2})13s	13s [1½] ^o	2 1	173128. 02 173130. 76	
10s ₁ ''' 10s ₁ ''	2p ⁵ (² P _{3/2} ^o)10d	10d' [2½] ^o	2 3	173610. 45 173610. 52			2p ⁵ (² P _{3/2})12d	12d [½] ^o	0 1	173165. 56	
10s ₁ '' 10s ₁ '	"	10d' [1½] ^o	2 1	173610. 50 173611. 54		12d ₅	"	12d [3½] ^o	4 3	173166. 46 173166. 43	
10p _{7,6}	2p ⁵ (² P _{1/2})11p	11p [1½]	1, 2	172873. 9		12d ₄	"	12d [1½] ^o	2 1	173167. 03	
10s ₅ 10s ₄	2p ⁵ (² P _{1/2})12s	12s [1½] ^o	2 1	172970. 51 172974. 3'		12d ₃	"	12d [2½] ^o	2 3	173168. 14 173168. 43	
11d ₆ 11d ₅	2p ⁵ (² P _{1/2})11d	11d [½] ^o	0 1	173019. 37 173019. 86		12d ₁ '' 12d ₁ '	"	13d [½] ^o	0 1	173279. 46	
11d ₄ ' 11d ₄	"	11d [3½] ^o	4 3	173020. 86 173020. 82		13d ₅	"	13d [3½] ^o	4 3	173280. 05 173280. 12	
11d ₃	"	11d [1½] ^o	2 1	173022. 02		13d ₄ ' 13d ₄	"				
11d ₁ '' 11d ₁ '	"	11d [2½] ^o	2 3	173022. 95 173023. 27			Ne II (² P _{1/2})	Limit	-----	173931.7	
							Ne II (² P _{3/2})	Limit	-----	174712. 2	

March 1948.

Ne I OBSERVED LEVELS*

Config. 1s ² 2s ² +	Observed Terms			
2p ⁶	2p ⁶ 1S			
2p ⁵ (² P°)nx	ns (n≥3)	np (n≥3)		
	$\begin{Bmatrix} 3-13s & {}^3\text{P}^\circ \\ 3-11s & {}^1\text{P}^\circ \end{Bmatrix}$	$\begin{matrix} 3p & {}^3\text{S} \\ 3p & {}^1\text{S} \end{matrix}$	$\begin{matrix} 3p & {}^3\text{P} \\ 3p & {}^1\text{P} \end{matrix}$	$\begin{matrix} 3p & {}^3\text{D} \\ 3p & {}^1\text{D} \end{matrix}$
j <i>l</i> -Coupling Notation				
2p ⁵ (² P _{1/2})nx	Observed Pairs			
	ns (n≥3)	np (n≥3)	nd (n≥3)	nf (n≥4)
	3-13s [1½]°	$\begin{matrix} 3-10p [\frac{1}{2}] \\ 3-10p [2\frac{1}{2}] \\ 3-11p [1\frac{1}{2}] \end{matrix}$	$\begin{matrix} 3-13d [\frac{1}{2}]^\circ \\ 3-13d [3\frac{1}{2}]^\circ \\ 3-12d [1\frac{1}{2}]^\circ \\ 3-12d [2\frac{1}{2}]^\circ \end{matrix}$	$\begin{matrix} 4-8f [1\frac{1}{2}] \\ 6-9f [4\frac{1}{2}] \\ 4-7, 9f [2\frac{1}{2}] \\ 6-9f [3\frac{1}{2}] \end{matrix}$
2p ⁵ (² P _{3/2} °)nx'	3-11s' [½]°	$\begin{matrix} 3-9p' [1\frac{1}{2}] \\ 3-9p' [\frac{1}{2}] \end{matrix}$	$\begin{matrix} 3-11d' [2\frac{1}{2}]^\circ \\ 3-11d' [1\frac{1}{2}]^\circ \end{matrix}$	$\begin{matrix} 6-8f' [3\frac{1}{2}] \\ 4-8f' [2\frac{1}{2}] \end{matrix}$

*For predicted levels in the spectra of the Ne I isoelectronic sequence, see Introduction.

Ne II

(F I sequence; 9 electrons)

 $Z=10$ Ground state $1s^2 2s^2 2p^5 {}^2P_{1/2}^{\circ}$ $2p^5 {}^2P_{1/2}^{\circ}$ 331350 cm^{-1}

I. P. 41.07 volts

The terms are from Boyce, who has extended the analysis by further observations in the ultraviolet, and improved the earlier term values. The series limit is estimated from series of two members, the 3s and 4s terms.

Intersystem combinations connecting the doublet and quartet terms have been observed.

The values of the $3d' {}^2G$ and $3d' {}^2S$ terms have been corrected to agree with the observed combinations.

REFERENCES

- K. T. Compton and J. C. Boyce, J. Franklin Inst. **205**, 511 (1928). (T) (C L) (G D)
 T. L. de Bruin and C. J. Bakker, Zeit. Phys. **69**, 19 (1931). (T) (C L) (Z E)
 J. C. Boyce, Phys. Rev. **46**, 378 (1934). (I P) (T) (C L)

Ne II

Ne II

Config.	Desig.	J	Level	Interval	Obs. g	Config.	Desig.	J	Level	Interval	Obs. g
$2s^2 2p^5$	$2p^5 {}^2P^{\circ}$	$1\frac{1}{2}$ $\frac{1}{2}$	0 782	-782		$2s^2 2p^4({}^3P)3d$	$3d {}^4D$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	279139. 1 279220. 6 279326. 8 279425. 1	-81. 5 -106. 2 -98. 3	
$2s 2p^6$	$2p^6 {}^2S$	$\frac{1}{2}$	217050								
$2s^2 2p^4({}^3P)3s$	$3s {}^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	219133. 0 219650. 8 219949. 9	-517. 8 -299. 1	1. 60 1. 73 2. 67	$2s^2 2p^4({}^3P)3d$	$3d {}^4F$	$4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$	280174. 4 280702. 5 281028. 1 280949. 6	-528. 1 -325. 6 78. 5	
$2s^2 2p^4({}^3P)3s$	$3s {}^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	224089. 3 224701. 8	-612. 5	1. 33 0. 67	$2s^2 2p^4({}^3P)3d$	$3d {}^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	280264. 0 280799. 3	-535. 3	
$2s^2 2p^4({}^3P)3p$	$3p {}^4P^{\circ}$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	246194. 8 246417. 4 246599. 9	-222. 6 -182. 5	1. 60 1. 73 2. 67	$2s^2 2p^4({}^3P)3d$	$3d {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	280271. 0 280475. 6	-204. 6	
$2s^2 2p^4({}^1D)3s$	$3s' {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	246396. 5 246400. 0	-3. 5	1. 20 0. 80	$2s^2 2p^4({}^3P)3d$	$3d {}^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	280770. 2 280991. 7 281173. 5	221. 5 181. 8	
$2s^2 2p^4({}^3P)3p$	$3p {}^4D^{\circ}$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	249110. 8 249448. 0 249697. 7 249841. 8	-337. 2 -249. 7 -144. 1	1. 43 1. 37 1. 20 0. 00	$2s^2 2p^4({}^3P)3d$	$3d {}^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	281334. 5 281722. 3	387. 8	0. 70 1. 25
$2s^2 2p^4({}^3P)3p$	$3p {}^2D^{\circ}$	$2\frac{1}{2}$ $1\frac{1}{2}$	251013. 3 251524. 7	-511. 4	1. 20 0. 80	$2s^2 2p^4({}^3P)4s$	$4s {}^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	282000. 0 282376. 7 282682. 2	-376. 7 -305. 5	
$2s^2 2p^4({}^3P)3p$	$3p {}^2S^{\circ}$	$\frac{1}{2}$	252800. 8		1. 96	$2s^2 2p^4({}^3P)4s$	$4s {}^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	283323. 7 283896. 5	-572. 8	
$2s^2 2p^4({}^3P)3p$	$3p {}^4S^{\circ}$	$1\frac{1}{2}$	252956. 0			$2s^2 2p^4({}^3P)4d$	$4d {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	302321? 302452?	-131	
$2s^2 2p^4({}^3P)3p$	$3p {}^2P^{\circ}$	$1\frac{1}{2}$ $\frac{1}{2}$	254167. 0 254294. 0	-127. 0	1. 33 0. 71	$2s^2 2p^4({}^3P)4f$	$4f {}^4D^{\circ}$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	302830. 6 302845. 5 302905. 2 302991. 2	-14. 9 -59. 7 -86. 0	
$2s^2 2p^4({}^1D)3p$	$3p' {}^2F^{\circ}$	$2\frac{1}{2}$ $3\frac{1}{2}$	274366. 9 274411. 3	44. 4	0. 86 1. 14						
$2s^2 2p^4({}^1D)3p$	$3p' {}^2P^{\circ}$	$1\frac{1}{2}$ $\frac{1}{2}$	276278. 6 276514. 1	-235. 5	1. 33 0. 67	$2s^2 2p^4({}^3P)4d$	$4d {}^2P$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ \frac{1}{2} \end{array} \right\}$	302884?		
$2s^2 2p^4({}^1S)3s$	$3s'' {}^2S$	$\frac{1}{2}$	276678. 0		2. 00	$2s^2 2p^4({}^3P)4f$	$4f {}^4F^{\circ}$	$4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$	302905. 8 303530. 8 303826. 8 303511. 6	-625. 0 -295. 8 315. 0	
$2s^2 2p^4({}^1D)3p$	$3p' {}^2D^{\circ}$	$1\frac{1}{2}$ $2\frac{1}{2}$	277327. 6 277346. 1	18. 5	0. 80 1. 20						

Ne II—Continued

Ne II—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$2s^2 2p^4(^3P)4f$	$4f \ ^4G^\circ$	$5\frac{1}{2}$ $4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$	303475. 7 303465. 1 303701. 1 303602. 3	10. 6 —236. 0 98. 8		$2s^2 2p^4(^1D)4s$	$4s' \ ^2D$	$\left\{ \begin{matrix} 2\frac{1}{2} \\ 1\frac{1}{2} \end{matrix} \right\}$	306018?		
$2s^2 2p^4(^3P)4f$	$4f \ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	303465. 4 303882. 3	416. 9		$2s^2 2p^4(^1D)3d$	$3d' \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	306244. 8 306689. 8	445. 0	
$2s^2 2p^4(^1D)3d$	$3d' \ ^2G$	$4\frac{1}{2}$ $3\frac{1}{2}$	305366. 2 305367. 2	—1. 0		$2s^2 2p^4(^1D)3d$	$3d' \ ^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	307992. 2 308103. 3	—111. 1	
$2s^2 2p^4(^1S)3p$	$3p' \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	305399. 2 305409. 3	—10. 1	1. 33 0. 67	$2s^2 2p^4(^1D)3d$	$3d' \ ^2S$	$\frac{1}{2}$	309049. 7		
$2s^2 2p^4(^1D)3d$	$3d' \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	305568. 9 305584. 2	—15. 3		$2s^2 2p^4(^1S)3d$	$3d' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	327954. 7 327968. 2	—13. 5	
						Ne III (3P_2)	Limit	-----	331350		

March 1947.

Ne II OBSERVED TERMS*

Config. $1s^2 +$	Observed Terms	
$2s^2 2p^5$	$2p^5 \ ^2P^\circ$	
$2s \ 2p^6$	$2p^6 \ ^2S$	
	$ns \ (n \geq 3)$	$np \ (n \geq 3)$
$2s^2 2p^4(^3P)nx$	$\left\{ \begin{matrix} 3, 4s \ ^4P \\ 3, 4s \ ^2P \end{matrix} \right.$	$\begin{matrix} 3p \ ^4S^\circ & 3p \ ^4P^\circ & 3p \ ^4D^\circ \\ 3p \ ^2S^\circ & 3p \ ^2P^\circ & 3p \ ^2D^\circ \end{matrix}$
$2s^2 2p^4(^1D)nx'$	$3, 4s' \ ^2D$	$3p' \ ^2P^\circ \quad 3p' \ ^2D^\circ \quad 3p' \ ^2F^\circ$
$2s^2 2p^4(^1S)nx''$	$3s'' \ ^2S$	$3p'' \ ^2P^\circ$
	$nd \ (n \geq 3)$	$nf \ (n \geq 4)$
$2s^2 2p^4(^3P)nx$	$\left\{ \begin{matrix} 3d \ ^4P & 3d \ ^4D & 3d \ ^4F \\ 3, 4d \ ^2P & 3, 4d \ ^2D & 3d \ ^2F \end{matrix} \right.$	$\begin{matrix} 4f \ ^4D^\circ & 4f \ ^4F^\circ & 4f \ ^4G^\circ \\ 4f \ ^2D^\circ & & \end{matrix}$
$2s^2 2p^4(^1D)nx'$	$3d' \ ^2S \quad 3d' \ ^2P \quad 3d' \ ^2D \quad 3d' \ ^2F \quad 3d' \ ^2G$	
$2s^2 2p^4(^1S)nx''$	$3d'' \ ^2D$	

*For predicted terms in the spectra of the F I isoelectronic sequence, see Introduction.

Ne III

(O I sequence; 8 electrons)

 $Z=10$ Ground state $1s^2 2s^2 2p^4 {}^3P_2$ $2p^4 {}^3P_2$ 514148 cm^{-1} I. P. 64 ± 1 volts

This spectrum is incompletely analyzed. The terms have been taken from two references: triplet and quintet terms, de Bruin (1935); and singlet terms, Boyce (1934). The latter are located with respect to the ground state by means of the nebular lines at 3343 Å, 3868.74 Å, and 3967.51 Å. The relative positions of the quintet terms and the ionization potential are estimated, and the uncertainty, x , may be considerable.

REFERENCES

- T. L. de Bruin, *Zeit. Phys.* **77**, 505 (1932). (T) (C L)
 V. v. Keussler, *Zeit. Phys.* **85**, 1 (1933). (C L)
 J. C. Boyce, *Phys. Rev.* **46**, 378 (1934). (I P) (T) (C L)
 T. L. de Bruin, *Zeeman Verhandelingen* p. 413 (Martinus Nijhoff, The Hague, 1935). (I P) (T)
 J. C. Boyce, *Mon. Not. Roy. Astr. Soc.* **96**, 690 (1936). (C L)

Ne III

Ne III

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$2s^2 2p^4$	$2p^4 {}^3P$	2 1 0	0 650 927	—650 —277	$2s^2 2p^3({}^4S^\circ)3d$	$3d {}^3D^\circ$	1 2 3	398192.70 398196.83 398210.74	4.13 13.91
$2s^2 2p^4$	$2p^4 {}^1D$	2	25841		$2s^2 2p^3({}^4S^\circ)3d$	$3d {}^5D^\circ$	4 3 2 1 0	398946.98+x 398948.51+x 398952.34+x 398955.75+x	—1.53 —3.83 —3.41
$2s^2 2p^4$	$2p^4 {}^1S$	0	55747		$2s^2 2p^3({}^2D^\circ)3p$	$3p' {}^3P$	2 1 0	398986.64 399082.57 399125.12	—95.93 —42.55
$2s 2p^5$	$2p^5 {}^3P^\circ$	2 1 0	204292 204879 205204	—587 —325	$2s^2 2p^3({}^2D^\circ)3p$	$3p'' {}^3D$	3 2 1	409847.53 409845.08 409855.23	2.45 —10.15
$2s 2p^5$	$2p^5 {}^1P^\circ$	1	289479		$2s^2 2p^3({}^2P^\circ)3p$	$3p'' {}^3S$	1	410134.72	
$2s^2 2p^3({}^4S^\circ)3s$	$3s {}^5S^\circ$	2	314148 +x		$2s^2 2p^3({}^2P^\circ)3p$	$3p'' {}^3P$	0 1 2	412293.59 412313.11 412320.21	19.52 7.10
$2s^2 2p^3({}^4S^\circ)3s$	$3s {}^3S^\circ$	1	319444.90		$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^3F^\circ$	2 3 4	435527.90 435568.00 435620.80	40.10 52.80
$2s^2 2p^3({}^4S^\circ)3p$	$3p {}^5P$	1 2 3	352662.05+x 352692.93+x 352745.91+x	30.88 52.98	$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^3G^\circ$	5 4 3	436561.35 436588.34 436611.56	—26.99 —23.22
$2s^2 2p^3({}^2D^\circ)3s$	$3s' {}^3D^\circ$	3 2 1	353148.00 353177.16 353197.40	—29.16 —20.24	$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^3D^\circ$	3 2 1	436844.63 436914.39 436959.49	—69.76 —45.10
$2s^2 2p^3({}^4S^\circ)3p$	$3p {}^3P$	2 1 0	356776.52 356766.20 356776.52	10.32 —10.32	$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^3P^\circ$	2 1 0	439586.00 439707.81 439760.35	—121.81 —52.54
$2s^2 2p^3({}^2D^\circ)3s$	$3s' {}^1D^\circ$	2	357930		$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^3S^\circ$	1	440064.90	
$2s^2 2p^3({}^2P^\circ)3s$	$3s'' {}^3P^\circ$	2 1 0	374434.00 374460.75 374477.66	—26.75 —16.91					
$2s^2 2p^3({}^2P^\circ)3s$	$3s'' {}^1P^\circ$	1	379834						
$2s^2 2p^3({}^2D^\circ)3p$	$3p' {}^3D$	1 2 3	389058.24 389069.37 389139.05	11.13 69.68					
$2s^2 2p^3({}^2D^\circ)3p$	$3p' {}^3F$	2 3 4	391414.02 391429.94 391450.31	15.92 20.37					
					Ne IV (${}^4S_{3/2}$)	Limit	-----	514148	

Ne III OBSERVED TERMS*

Config. $1s^2 +$	Observed Terms	
$2s^2 2p^4$	$\left\{ \begin{array}{cc} 2p^4 \ ^1S & 2p^4 \ ^3P \\ & 2p^4 \ ^1D \end{array} \right.$	
$2s \ 2p^5$	$\left\{ \begin{array}{c} 2p^5 \ ^3P^\circ \\ 2p^5 \ ^1P^\circ \end{array} \right.$	
	$ns \ (n \geq 3)$	$np \ (n \geq 3)$
$2s^2 2p^3(^4S^\circ)nx$	$\left\{ \begin{array}{c} 3s \ ^5S^\circ \\ 3s \ ^3S^\circ \end{array} \right.$	$\begin{array}{c} 3p \ ^5P \\ 3p \ ^3P \end{array}$
$2s^2 2p^3(^2D^\circ)nx'$	$\left\{ \begin{array}{c} 3s' \ ^3D^\circ \\ 3s' \ ^1D^\circ \end{array} \right.$	$\begin{array}{ccc} 3p' \ ^3P & 3p' \ ^3D & 3p' \ ^3F \end{array}$
$2s^2 2p^3(^2P^\circ)nx''$	$\left\{ \begin{array}{c} 3s'' \ ^3P^\circ \\ 3s'' \ ^1P^\circ \end{array} \right.$	$\begin{array}{ccc} 3p'' \ ^3S & 3p'' \ ^3P & 3p'' \ ^3D \end{array}$
	$nd \ (n \geq 3)$	
$2s^2 2p^3(^4S^\circ)nx$	$\begin{array}{c} 3d \ ^5D^\circ \\ 3d \ ^3D^\circ \end{array}$	
$2s^2 2p^3(^2D^\circ)nx'$	$\begin{array}{ccccccc} 3d' \ ^3S^\circ & 3d' \ ^3P^\circ & 3d' \ ^3D^\circ & 3d' \ ^3F^\circ & 3d' \ ^3G^\circ \end{array}$	

*For predicted terms in the spectra of the O I isoelectronic sequence, see Introduction.

Ne IV

(N I sequence; 7 electrons)

$Z=10$

Ground state $1s^2 2s^2 2p^3 \ ^4S_{1/2}$

$2p^3 \ ^4S_{1/2} \ 783880 \text{ cm}^{-1}$

I. P. 97.16 volts

The analysis is by Paul and Polster, who have extended the earlier work by Boyce and published 111 classified lines in the interval from 140 Å to 786 Å. From series they derive the limit 781714 cm^{-1} and place the level $2p^3 \ ^2D_2^\circ$ at 38540 cm^{-1} above the ground state zero. No intersystem combinations have been observed.

On the basis of later analyses of the spectra in this sequence a slight adjustment in these values has been made by Robinson. The doublet terms have been increased by 2410 cm^{-1} and the limit by 2166 cm^{-1} to fit the isoelectronic sequence data. The later values have been adopted in the table. The uncertainty x , may be considerable.

REFERENCES

- F. W. Paul and H. D. Polster, Phys. Rev. **59**, 426 (1941). (I P) (T) (C L)
H. A. Robinson, unpublished material (March 1948). (I P) (T)

Ne IV

Ne IV

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$2s^2 2p^3$	$2p^3 \ ^4S^\circ$	$1\frac{1}{2}$	0		$2s^2 2p^2(^1D)3d$	$3d' \ ^2S$	$\frac{1}{2}$	$616482+x$	
$2s^2 2p^3$	$2p^3 \ ^2D^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$	$40950+x$ $40975+x$	-25	$2s^2 2p^2(^3P)4s$	$4s \ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	633465 633790 634413	325 623
$2s^2 2p^3$	$2p^3 \ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	$62157+x$ $62167+x$	10	$2s^2 2p^2(^3P)4s$	$4s \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	$635866+x$ $636475+x$	609
$2s 2p^4$	$2p^4 \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	183860 184477 184799	-617 -322	$2s^2 2p^2(^3P)4p$	$4p \ ^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	641908 642184 642472 642934	276 288 462
$2s 2p^4$	$2p^4 \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	$253807+x$ $253823+x$	-16	$2s^2 2p^2(^3P)4p$	$4p \ ^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	643239 643672 643975	433 303
$2s 2p^4$	$2p^4 \ ^2S$	$\frac{1}{2}$	$299351+x$		$2s^2 2p^2(^3P)4p$	$4p \ ^4S^\circ$	$1\frac{1}{2}$	648060	
$2s 2p^4$	$2p^4 \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	$319751+x$ $320452+x$	-701	$2s^2 2p^2(^1D)4s$	$4s' \ ^2D$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	$664124+x$	
$2s^2 2p^2(^3P)3s$	$3s \ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	478701 479079 479651	378 572	$2s^2 2p^2(^3P)4d$	$4d \ ^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	$670595+x$ $671252+x$	657
$2p^5$	$2p^5 \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	$484623+x$ $485585+x$	-962	$2s^2 2p^2(^3P)4d$	$4d \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	671402 672102 672676	-700 -574
$2s^2 2p^2(^3P)3s$	$3s \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	$488215+x$ $488917+x$	702	$2s 2p^3(^5S^\circ)3d$	$3d''' \ ^4D^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{array} \right\}$	672799	
$2s^2 2p^2(^1D)3s$	$3s' \ ^2D$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	$511411+x$		$2s^2 2p^2(^3P)4d$	$4d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	$673427+x$ $673587+x$	160
$2s^2 2p^2(^3P)3p$	$3p \ ^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	524391 524676 525017	285 341	$2s^2 2p^2(^3P)5s$	$5s \ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	693106 693717 694353	611 636
$2s^2 2p^2(^3P)3p$	$3p \ ^4S^\circ$	$1\frac{1}{2}$	532978		$2s^2 2p^2(^1D)4d$	$4d' \ ^2F$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	$697855+x$	
$2s^2 2p^2(^1S)3s$	$3s'' \ ^2S$	$\frac{1}{2}$	$551712+x$		$2s^2 2p^2(^1D)4d$	$4d' \ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	$699622+x$	
$2s^2 2p^2(^3P)3d$	$3d \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	$575968+x$ $576353+x$	-385	$2s^2 2p^2(^1D)4d$	$4d' \ ^2P$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	$701223+x$	
$2s^2 2p^2(^1S)3d$	$3d'' \ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	$576915+x$		$2s^2 2p^2(^1S)4d$	$4d'' \ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	$709460+x$	
$2s^2 2p^2(^3P)3d$	$3d \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	579307 579626 579737	-319 -111	$2s^2 2p^2(^1D)5s$	$5s' \ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	$724690+x$	
$2s^2 2p^2(^3P)3d$	$3d \ ^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	$579375+x$ $580095+x$	720	$2s^2 2p^2(^1D)5d$	$5d' \ ^2F$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	$740607+x$	
$2s^2 2p^2(^3P)3d$	$3d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	$586685+x$ $586918+x$	233	$2s^2 2p^2(^1D)6s$	$6s' \ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	$754597+x$	
$2s 2p^3(^5S^\circ)3s$	$3s''' \ ^4S^\circ$	$1\frac{1}{2}$	588021		Ne v (3P_0) <i>Limit</i>				783880
$2s^2 2p^2(^1D)3d$	$3d' \ ^2F$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	$605417+x$						
$2s^2 2p^2(^1D)3d$	$3d' \ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	$609118+x$						
$2s^2 2p^2(^1D)3d$	$3d' \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	$612668+x$ $612781+x$	113					

March 1948.

Ne IV OBSERVED TERMS*

Config. $1s^2+$	Observed Terms		
$2s^2 2p^3$	$\left\{ \begin{array}{l} 2p^3 \ ^4S^\circ \\ 2p^3 \ ^2P^\circ \\ 2p^3 \ ^2D^\circ \end{array} \right.$		
$2s \ 2p^4$	$\left\{ \begin{array}{l} 2p^4 \ ^4P \\ 2p^4 \ ^2P \\ 2p^4 \ ^2D \end{array} \right.$		
$2p^5$	$2p^5 \ ^2P^\circ$		
	$ns \ (n \geq 3)$	$np \ (n \geq 3)$	$nd \ (n \geq 3)$
$2s^2 2p^2(^3P)nx$	$\left\{ \begin{array}{l} 3-5s \ ^4P \\ 3, 4s \ ^2P \end{array} \right.$	$3, 4p \ ^4S^\circ \quad 3, 4p \ ^4P^\circ \quad 4p \ ^4D^\circ$	$\begin{array}{l} 3, 4d \ ^4P \\ 3d \ ^2P \end{array} \quad 3, 4d \ ^2D \quad 3, 4d \ ^2F$
$2s^2 2p^2(^1D)nx'$	$3-6s' \ ^2D$		$3d' \ ^2S \quad 3, 4d' \ ^2P \quad 3, 4d' \ ^2D \quad 3-5d' \ ^2F$
$2s^2 2p^2(^1S)nx''$	$3s'' \ ^2S$		$3, 4d'' \ ^2D$
$2s \ 2p^3(^5S^\circ)nx'''$	$3s''' \ ^4S^\circ$		$3d''' \ ^4D^\circ$

*For predicted terms in the spectra of the N I isoelectronic sequence, see Introduction.

Ne v

(C I sequence; 6 electrons)

$Z=10$

Ground state $1s^2 2s^2 2p^2 \ ^3P_0$

$2p^2 \ ^3P_0$ 1019950 cm^{-1}

I. P. 126.4 volts

Paul and Polster have classified a total of 56 lines of Ne v in the range 118 Å to 572 Å, as transitions among 47 energy levels. The absolute value of $2p^2 \ ^3P_0$ is calculated from the $nd \ ^3P^\circ$ and $nd \ ^3D^\circ$ series, in each of which two members have been observed.

The singlet and triplet terms are connected by the intersystem lines $2p^2 \ ^3P_{2,1} - 2p^2 \ ^1D_2$ observed in the spectra of gaseous nebula, as given by Bowen.

No intersystem combinations connecting the quintet terms with the rest have been observed, as indicated by the uncertainty x in the table. Paul and Polster estimate from isoelectronic sequence data that the term $2p^3 \ ^5S_2^\circ$ is $86700 \pm 300 \text{ cm}^{-1}$ above the ground state. From later data on this sequence Robinson places the value at 88842 cm^{-1} . The later value is entered in brackets and has been used in the present compilation for all quintet terms.

REFERENCES

- I. S. Bowen, Rev. Mod. Phys. **8**, 68 (1936). (C L)
 F. W. Paul and H. D. Polster, Phys. Rev. **59**, 428 (1941). (I P) (T) (C L)
 H. A. Robinson, unpublished material (March 1948). (T)

Ne V

Ne V

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$2s^2 2p^2$	$2p^2 \ ^3P$	0 1 2	0 414 1112	414 698	$2s^2 2p(^2P^o)3d$	$3d \ ^3D^o$	1 2 3	698231 698382 698735	151 353
$2s^2 2p^2$	$2p^2 \ ^1D$	2	30294		$2s^2 2p(^2P^o)3d$	$3d \ ^3P^o$	2 1 0	701765 702074 702459	-309 -385
$2s^2 2p^2$	$2p^2 \ ^1S$	0	63900		$2s^2 2p(^2P^o)3d$	$3d \ ^1P^o$	1	702412	
$2s 2p^3$	$2p^3 \ ^5S^o$	2	[88842]+ <i>x</i>		$2s^2 2p(^2P^o)3d$	$3d \ ^1F^o$	3	709956	
$2s 2p^3$	$2p^3 \ ^3D^o$	3 2 1	175834 175905 175927	-71 -22	$2s 2p^2(^4P)3s$	$3s \ ^3P$	0 1 2	719350 719527 720011	177 484
$2s 2p^3$	$2p^3 \ ^3P^o$	2, 1 0	208157 208193	-36	$2s^2 2p(^2P^o)4s$	$4s \ ^3P^o$	0, 1, 2	795279	
$2s 2p^3$	$2p^3 \ ^1D^o$	2	270564		$2s 2p^2(^4P)3d$	$3d \ ^5P$	3 2 1	799115+ <i>x</i> 799286+ <i>x</i> 799493+ <i>x</i>	-171 -207
$2s 2p^3$	$2p^3 \ ^3S^o$	1	279365		$2s^2 2p(^2P^o)4s$	$4s \ ^1P^o$	1	805284	
$2s 2p^3$	$2p^3 \ ^1P^o$	1	303812		$2s 2p^2(^4P)4s$	$4s \ ^5P$	1, 2, 3	822976+ <i>x</i>	
$2p^4$	$2p^4 \ ^3P$	2 1 0	412681 413466 413803	-785 -337	$2s^2 2p(^2P^o)4d$	$4d \ ^1D^o$	2	838623	
$2s^2 2p(^2P^o)3s$	$3s \ ^3P^o$	0 1 2	596230 596626 597492	396 866	$2s^2 2p(^2P^o)4d$	$4d \ ^3D^o$	1, 2, 3	842020	
$2s^2 2p(^2P^o)3s$	$3s \ ^1P^o$	1	605231		$2s^2 2p(^2P^o)4d$	$4d \ ^3P^o$	2, 1, 0	842914	
$2s^2 2p(^2P^o)3d$	$3d \ ^1D^o$	2	690691		$2s^2 2p(^2P^o)4d$	$4d \ ^1F^o$	3	847207?	
$2s 2p^2(^4P)3s$	$3s \ ^5P$	1 2 3	697507+ <i>x</i> 698059+ <i>x</i> 698512+ <i>x</i>	552 453	$2s 2p^2(^4P)4d$	$4d \ ^5P$	3, 2, 1	865282+ <i>x</i>	
					Ne VI ($^2P_{3/2}^o$)	Limit	-----	1019950	

March 1948.

Ne V OBSERVED TERMS*

Config. $1s^2+$	Observed Terms	
$2s^2 2p^2$	{ $2p^2 \ ^1S$ $2p^2 \ ^3P$ $2p^2 \ ^1D$	
$2s 2p^3$	{ $2p^3 \ ^5S^o$ $2p^3 \ ^3P^o$ $2p^3 \ ^3D^o$ $2p^3 \ ^3S^o$ $2p^3 \ ^1P^o$ $2p^3 \ ^1D^o$	
$2p^4$	$2p^4 \ ^3P$	
	$ns \ (n \geq 3)$	$nd \ (n \geq 3)$
$2s^2 2p(^2P^o)nx$	{ $3, 4s \ ^3P^o$ $3, 4s \ ^1P^o$	$3, 4d \ ^3P^o$ $3, 4d \ ^3D^o$ $3d \ ^1P^o$ $3, 4d \ ^1D^o$ $3, 4d \ ^1F^o$
$2s 2p^2(^4P)nx$	{ $3, 4s \ ^5P$ $3s \ ^3P$	$3, 4d \ ^5P$

*For predicted terms in the spectra of the Cr I isoelectronic sequence, see Introduction.

Ne VI

(B I sequence; 5 electrons)

 $Z=10$ Ground state $1s^2 2s^2 2p^2 P_{1/2}^\circ$ $2p^2 P_{1/2}^\circ$ 1274000 ± 1000 cm⁻¹I. P. 157.91 ± 0.12 volts

This spectrum is incompletely analyzed. Paul and Polster have classified 23 lines in the range from 110 Å to 562 Å. They have estimated the limit and ionization potential from isoelectronic data. No intersystem combinations have been observed but the uncertainty x is approximately known from their estimated value of $2p^2 {}^4P$ (entered in brackets in the table).

REFERENCE

F. W. Paul and H. D. Polster, Phys. Rev. **59**, 429 (1941). (I P) (T) (C L)

Ne VI

Ne VI

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$2s^2 ({}^1S) 2p$	$2p^2 {}^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	0 1316	1316	$2s 2p ({}^3P^\circ) 3s$	$3s {}^4P^\circ$	$\left\{ \begin{array}{c} \frac{1}{2} \\ \text{to} \\ 2\frac{1}{2} \end{array} \right\}$	$834113+x$	
$2s 2p^2$	$2p^2 {}^4P$	$\left\{ \begin{array}{c} \frac{1}{2} \\ \text{to} \\ 2\frac{1}{2} \end{array} \right\}$	$[99300]+x$		$2s 2p ({}^3P^\circ) 3p$	$3p {}^2P$	$\left\{ \begin{array}{c} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	878852	
$2s 2p^2$	$2p^2 {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	178998 179020	-22	$2s 2p ({}^3P^\circ) 3p$	$3p {}^2S$	$\frac{1}{2}$	900408	
$2s 2p^2$	$2p^2 {}^2S$	$\frac{1}{2}$	232587		$2s 2p ({}^3P^\circ) 3p$	$3p {}^2D$	$\left\{ \begin{array}{c} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	906373	
$2s 2p^2$	$2p^2 {}^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	249292 250112	820	$2s 2p ({}^3P^\circ) 3d$	$3d {}^4D^\circ$	$\left\{ \begin{array}{c} \frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{array} \right\}$	$924791+x$	
$2s^2 ({}^1S) 3s$	$3s {}^2S$	$\frac{1}{2}$	722610						
$2s^2 ({}^1S) 3p$	$3p {}^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	763096 763385	289	Ne VII (1S_0)	Limit		1274000	
$2s^2 ({}^1S) 3d$	$3d {}^2D$	$\left\{ \begin{array}{c} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	816405						

October 1946

SODIUM

Na I

11 electrons

 $Z=11$ Ground state $1s^2 2s^2 2p^6 3s^2 S_{1/2}$ $3s^2 S_{1/2}$ 41449.65 cm^{-1}

I. P. 5.138 volts

Thackeray has observed the $^2P^\circ$ series in absorption to $n=73$. His values are used for this series for $n=4$ to 59,* and for the 2D series for $n=8$ to 13.

Meissner and Luft have observed selected lines with an interferometer. Their results, including observed intervals of the $3-6d^2D$ terms (the four-place entries in the table) and improved absolute values of the $3-7s^2S$, $3p^2P^\circ$ and $3-7d^2D$ terms, have been used.

From infrared observations Rood and Sawyer have extended the nf^2F° series from $n=5$ to $n=11$, except for $n=8$. Their values have been used, a calculated value of $8f^2F^\circ$ being entered in brackets in the table.

The rest of the terms are from Fowler and Paschen-Götze, who published detailed analyses. By analogy with other spectra the designations $5g^2G$ and $6h^2H^\circ$ have been assigned to the terms calculated from Fowler's combinations labeled " $3\phi-4\phi$ " and " $4\phi-5\phi$ ", respectively.

REFERENCES

- A. Fowler, *Report on Series in Line Spectra* p. 99 (Fleetway Press, London, 1922). (I P) (T) (C L)
 F. Paschen und R. Götze, *Seriengesetze der Linienspektren* p. 56 (Springer, Berlin, 1922). (I P) (T) (C L)
 W. F. Meggers, *Bur. Std. J. Research* **10**, 673, RP558 (1933). (C L)
 H. E. White, *Introduction to Atomic Spectra* p. 77 (McGraw-Hill Book Co., Inc., New York, 1934). (G D)
 W. F. Meggers, *J. Research Nat. Bur. Std.* **14**, 487, RP781 (1935). (C L)
 K. W. Meissner und K. F. Luft, *Ann. der Phys.* [5] **29**, 698 (1937). (I P) (T) (C L)
 P. Rood and R. A. Sawyer, *Astroph. J.* **87**, 70 (1938). (T) (C L)
 W. F. Meggers, *J. Opt. Soc. Am.* **36**, 431 (1946). (Summary hfs)
 E. R. Thackeray, *Phys. Rev.* (1949). (In press). (I P) (T) (C L)

Na I

Na I

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
3s	3s 2S	$\frac{1}{2}$	0. 000		5p	5p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	35040. 27 35042. 79	2. 52
3p	3p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	16956. 183 16973. 379	17. 1963	6s	6s 2S	$\frac{1}{2}$	36372. 647	
4s	4s 2S	$\frac{1}{2}$	25739. 86		5d	5d 2D	$2\frac{1}{2}$ $1\frac{1}{2}$	37036. 781 37036. 805	-0. 0230
3d	3d 2D	$2\frac{1}{2}$ $1\frac{1}{2}$	29172. 855 29172. 904	-0. 0494	5f	5f $^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	37057. 6	
4p	4p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	30266. 88 30272. 51	5. 63	5g	5g 2G	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 4\frac{1}{2} \end{array} \right\}$	37060. 2	
5s	5s 2S	$\frac{1}{2}$	33200. 696		6p	6p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	37296. 51 37297. 76	1. 25
4d	4d 2D	$2\frac{1}{2}$ $1\frac{1}{2}$	34548. 754 34548. 789	-0. 0346	7s	7s 2S	$\frac{1}{2}$	38012. 074	
4f	4f $^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	34588. 6		6d	6d 2D	$2\frac{1}{2}$ $1\frac{1}{2}$	38387. 287 38387. 300	=0. 0124

*The last 14 members are not included because page proof had been prepared when the data were received.

Na I—Continued

Na I—Continued

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
6f	6f $^2F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	38400. 1	0. 74	14p	14p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	40814. 47	
6h	6h $^2H^\circ$	$\left\{ \begin{smallmatrix} 4\frac{1}{2} \\ 5\frac{1}{2} \end{smallmatrix} \right\}$	38403. 4		14d	14d 2D	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	40890. 0	
7p	7p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	38540. 40 38541. 14		15p	15p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	40901. 11	
8s	8s 2S	$\frac{1}{2}$	38968. 35		15d	15d 2D	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	40958	
7d	7d 2D	$2\frac{1}{2}$ $1\frac{1}{2}$	39200. 962 39200. 963	-0. 001	16p	16p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	40971. 16	0. 47
7f	7f $^2F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	39209. 2	0. 47	17p	17p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41028. 68	
8p	8p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	39298. 54 39299. 01		18p	18p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41076. 37	
9s	9s 2S	$\frac{1}{2}$	39574. 51		19p	19p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41116. 28	
8d	8d 2D	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	39729. 00	0. 47	20p	20p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41150. 39	0. 47
8f	8f $^2F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	[39734. 0]		21p	21p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41179. 22	
9p	9p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	39794. 53 39795. 00		22p	22p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41204. 28	
10s	10s 2S	$\frac{1}{2}$	39983. 0		23p	23p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41225. 88	
9d	9d 2D	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	40090. 57	0. 47	24p	24p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41244. 77	0. 47
9f	9f $^2F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	40093. 2		25p	25p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41261. 42	
10p	10p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	40137. 23		26p	26p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41276. 11	
11s	11s 2S	$\frac{1}{2}$	40273. 5		27p	27p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41289. 16	
10d	10d 2D	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	40349. 17	0. 47	28p	28p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41300. 74	0. 47
10f	10f $^2F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	40350. 9		29p	29p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41311. 09	
11p	11p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	40383. 16		30p	30p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41320. 34	
12s	12s 2S	$\frac{1}{2}$	40482. 9		31p	31p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41328. 87	
11f	11f $^2F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	40539	0. 47	32p	32p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41336. 50	0. 47
11d	11d 2D	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	40540. 35		33p	33p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41343. 49	
12p	12p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	40566. 03		34p	34p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41349. 70	
13s	13s 2S	$\frac{1}{2}$	40644. 6		35p	35p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41355. 50	
12d	12d 2D	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	40685. 8	0. 47	36p	36p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41360. 82	0. 47
13p	13p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	40705. 68						
14s	14s 2S	$\frac{1}{2}$	40769. 5						
13d	13d 2D	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	40798. 8						

Na I—Continued

Na I—Continued

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
37p	37p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41365.66		49p	49p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41402.25	
38p	38p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41370.11		50p	50p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41404.18	
39p	39p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41374.27		51p	51p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41406.03	
40p	40p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41378.04		52p	52p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41407.69	
41p	41p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41381.55		53p	53p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41409.80	
42p	42p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41384.84		54p	54p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41410.81	
43p	43p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41387.91		55p	55p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41412.20	
44p	44p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41390.73		56p	56p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41413.59	
45p	45p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41393.34		57p	57p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41414.89	
46p	46p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41395.77		58p	58p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41416.06	
47p	47p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41398.10		59p	59p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41417.18	
48p	48p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	41400.28		-----				
					Na II ($1S_0$)	Limit	-----	41449.65	

January 1949.

Na II

(Ne I sequence; 10 electrons)

 $Z=11$ Ground state $1s^2 2s^2 2p^6 {}^1S_0$ $2p^6 {}^1S_0$ 381528 cm^{-1}

I. P. 47.29 volts

The analysis has been taken from Söderqvist's Monograph except for the 5s- and 6s-levels, which are quoted from Vance's paper.

The term designations assigned by Söderqvist on the assumption of LS -coupling are listed under the heading "Author," with corresponding assignments added for the 5s- and 6s-levels.

As for Ne I, the $j\bar{l}$ -coupling notation in the general form suggested by Racah is adopted. Shortley has, however, pointed out that the configurations $2p^5 3s$, $2p^5 3p$, and $2p^5 3d$ are much closer to LS -coupling than they are to $j\bar{l}$ -coupling.

REFERENCES

- I. S. Bowen, Phys. Rev. **31**, 967 (1928). (T) (C L)
 S. Frisch, Zeit. Phys. **70**, 498 (1931). (T) (C L)
 B. B. Vance, Phys. Rev. **41**, 480 (1932). (T) (C L)
 J. Söderqvist, Nova Acta Reg. Soc. Sci. Uppsala [IV] **9**, No. 7, 22 (1934). (I P) (T) (C L) (G D)
 G. Racah, Phys. Rev. **61**, 537 (L) (1942).
 G. Shortley, unpublished material (1948).

Na II

Na II

Author	Config.	Desig.	<i>J</i>	Level	Author	Config.	Desig.	<i>J</i>	Level
2p ¹ S ₀	2p ⁶	2p ⁶ ¹ S	0	0. 00	3d ¹ P ₁	2p ⁵ (² P _{1/2} ^o)3d	3d [1½] ^o	1	331748. 77
3s ₅ ³ P ₂	2p ⁵ (² P _{1/2} ^o)3s	3s [1½] ^o	2	264928. 00	3d ¹ D ₂	2p ⁵ (² P _{3/2} ^o)3d	3d' [2½] ^o	2	332806. 06
3s ₄ ³ P ₁			1	265693. 29	³ D ₃			3	332845. 80
3s ₃ ³ P ₀	2p ⁵ (² P _{3/2} ^o)3s	3s' [½] ^o	0	266285. 36	3d ³ D ₂	"	3d' [1½] ^o	2	332966. 42
3s ₂ ¹ P ₁			1	268766. 67	³ D ₁			1	333166. 70
3p ₁₀ ³ S ₁	2p ⁵ (² P _{1/2} ^o)3p	3p [½]	1	293224. 12	4s ₅ ³ P ₂	2p ⁵ (² P _{1/2} ^o)4s	4s [1½] ^o	2	331500. 29
3p ₉ ³ D ₃	"	3p [2½]	3	297252. 52	4s ₄ ³ P ₁			1	331877. 67
3p ₈ ³ D ₂			2	297639. 34	4s ₃ ³ P ₀	2p ⁵ (² P _{3/2} ^o)4s	4s' [½] ^o	0	332713. 96
3p ₇ ³ D ₁	"	3p [1½]	1	298169. 14	4s ₂ ¹ P ₁			1	333111. 60
3p ₆ ¹ D ₂			2	299193. 75		2p ⁵ (² P _{1/2} ^o)5s	5s [1½] ^o	2	
3p ₃ ³ P ₀	"	3p [½]	0	300391. 59	5s ₄ ³ P ₁			1	353260
3p ₅ ¹ P ₁	2p ⁵ (² P _{3/2} ^o)3p	3p' [1½]	1	299889. 16	5s ₂ ¹ P ₁	2p ⁵ (² P _{3/2} ^o)5s	5s' [½] ^o	0	
3p ₄ ³ P ₂			2	300107. 71				1	354850
3p ₂ ³ P ₁	"	3p' [½]	1	300510. 92	4d ¹ P ₁	2p ⁵ (² P _{1/2} ^o)4d	4d [1½] ^o	1	353573
3p ₁ ¹ S ₀			0	308864. 54					
3d ³ P ₀	2p ⁵ (² P _{1/2} ^o)3d	3d [½] ^o	0	330553. 18	6s ₄ ³ P ₁	2p ⁵ (² P _{1/2} ^o)6s	6s [1½] ^o	2	
3d ³ P ₁			1	330640. 60				1	363500
3d ³ P ₂	"	3d [1½] ^o	2	330792. 85	6s ₂ ¹ P ₁	2p ⁵ (² P _{3/2} ^o)6s	6s' [½] ^o	0	
3d ³ F ₄	"	3d [3½] ^o	4	331126. 76				1	364960
3d ³ F ₃			3	331190. 49					
3d ³ F ₂	"	3d [2½] ^o	2	331669. 40		Na III (² P _{1/2} ^o)	Limit	-----	381528
3d ¹ F ₃			3	331711. 75		Na III (² P _{3/2} ^o)	Limit	-----	382892

August 1947.

Na II OBSERVED LEVELS*

Config. 1s ² 2s ² +	Observed Terms		
2p ⁶	2p ⁶ ¹ S		
	ns (n ≥ 3)	np (n ≥ 3)	nd (n ≥ 3)
2p ⁵ (² P ^o)nx	{ 3-6s ³ P ^o 3-6s ¹ P ^o	3p ³ S 3p ³ P 3p ³ D 3p ¹ S 3p ¹ P 3p ¹ D	3d ³ P ^o 3d ³ D ^o 3d ³ F ^o 3, 4d ¹ P ^o 3d ¹ D ^o 3d ¹ F ^o
<i>jl</i> -Coupling Notation			
	Observed Pairs		
	ns (n ≥ 3)	np (n ≥ 3)	nd (n ≥ 3)
2p ⁵ (² P _{1/2} ^o)nx	3-6s [1½] ^o	3p [½] 3p [2½] 3p [1½]	3d [½] ^o 3d [3½] ^o 3, 4d [1½] ^o 3d [2½] ^o
2p ⁵ (² P _{3/2} ^o)nx'	3-6s' [½] ^o	3p' [1½] 3p' [½]	3d' [2½] ^o 3d' [1½] ^o

*For predicted levels in the spectra of the Ne I isoelectronic sequence, see Introduction.

Na III

(F I sequence; 9 electrons)

 $Z=11$ Ground state is $1s^2 2s^2 2p^5 {}^2P_{1/2}^{\circ}$ $2p^5 {}^2P_{1/2}^{\circ}$ 578033 cm^{-1}

I. P. 71.65 volts

The terms are taken from the paper by Tombouliau, who has revised and extended the analysis by Söderqvist, but adopts the limit estimated by Söderqvist. The ${}^2P^{\circ}$ term from the 1S limit in Na IV has not been located to confirm Söderqvist's 2S and 2D terms from this limit.

Intersystem combinations have been observed, connecting the doublet and quartet terms.

REFERENCES

- J. Söderqvist, Nova Acta Reg. Soc. Sci. Uppsala [IV] **9**, No. 7, 39 (1934). (T) (C L) (G D)
D. H. Tombouliau, Phys. Rev. **54**, 347 (1938). (I P) (T) (C L)

Na III

Na III

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$2s^2 2p^5$	$2p^5 {}^2P^{\circ}$	$1\frac{1}{2}$ $\frac{1}{2}$	0 1364	-1364	$2s^2 2p^4({}^3P)3p$	$3p {}^4S^{\circ}$	$1\frac{1}{2}$	417415.5	
$2s 2p^6$	$2p^6 {}^2S$	$\frac{1}{2}$	264449		$2s^2 2p^4({}^3P)3p$	$3p {}^2P^{\circ}$	$1\frac{1}{2}$ $\frac{1}{2}$	418418.1 418556.9	-138.8
$2s^2 2p^4({}^3P)3s$	$3s {}^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	366165.3 367052.3 367561.9	-887.0 -509.6	$2s^2 2p^4({}^1S)3s$	$3s'' {}^2S$	$\frac{1}{2}$	435031	
$2s^2 2p^4({}^3P)3s$	$3s {}^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	373633.0 374681.4	-1048.4	$2s^2 2p^4({}^1D)3p$	$3p' {}^2F^{\circ}$	$2\frac{1}{2}$ $3\frac{1}{2}$	440472.0 440552.4	80.4
$2s^2 2p^4({}^1D)3s$	$3s' {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	399179.4 399182.7	-3.3	$2s^2 2p^4({}^1D)3p$	$3p' {}^2P^{\circ}$	$1\frac{1}{2}$ $\frac{1}{2}$	442710.5 443261.6	-551.1
$2s^2 2p^4({}^3P)3p$	$3p {}^4P^{\circ}$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	406200.9 406562.0 406876.0	-361.1 -314.0	$2s^2 2p^4({}^3P)3d$	$3d {}^4D$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	460267.8 460421.0 460605.6 460759.3	-153.2 -184.6 -153.7
$2s^2 2p^4({}^3P)3p$	$3p {}^4D^{\circ}$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	410987.9 411548.2 411963.9 412201.5	-560.3 -415.7 -237.6	$2s^2 2p^4({}^3P)3d$	$3d {}^4F$	$4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$	461877.4? 463112.8 463628.1 463462.2	-1235.4 -515.3 165.9
$2s^2 2p^4({}^3P)3p$	$3p {}^2D^{\circ}$	$2\frac{1}{2}$ $1\frac{1}{2}$	414281.0 415173.2	-892.2	$2s^2 2p^4({}^3P)3d$	$3d {}^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	462391.2 462963.6 463257.4	572.4 293.8
$2s^2 2p^4({}^3P)3p$	$3p {}^2S^{\circ}$	$\frac{1}{2}$	416910.2						

Na III—Continued

Na III—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$2s^2 2p^4(^3P)3d$	$3d \ ^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	463968. 8 465768. 8	—1800. 0	$2s^2 2p^4(^1D)3d$	$3d' \ ^2S$	$\frac{1}{2}$	497751. 2	
$2s^2 2p^4(^3P)3d$	$3d \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	464392. 1 465027. 9	—635. 8	$2s^2 2p^4(^1D)4s$	$4s' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	511410	
$2s^2 2p^4(^3P)3d$	$3d \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	465988. 0 466773. 0	785. 0	$2s^2 2p^4(^3P)4d$	$4d \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	514652	
$2s^2 2p^4(^3P)4s$	$4s \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	467773. 8 468528. 5 468949. 5	—754. 7 —421. 0	$2s^2 2p^4(^3P)4d$	$4d \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	515023 515379	356
$2s^2 2p^4(^3P)4s$	$4s \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	471446. 6 472250. 6	—804. 0	$2s^2 2p^4(^1S)3d$	$3d'' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	529465 529498	—33
$2s^2 2p^4(^1D)3d$	$3d' \ ^2G$	$4\frac{1}{2}$ $3\frac{1}{2}$	491928. 2		$2s^2 2p^4(^1D)4d$	$4d' \ ^2P$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	544227	
$2s^2 2p^4(^1D)3d$	$3d' \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	493191. 3 493289. 3	98. 0	$2s^2 2p^4(^1D)4d$	$4d' \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	544736	
$2s^2 2p^4(^1D)3d$	$3d' \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	493853. 2 494599. 0	745. 8	-----				
$2s^2 2p^4(^1D)3d$	$3d' \ ^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	495446. 8 495668. 6	—221. 8	Na IV (3P_2)	<i>Limit</i>	-----	578033	

March 1947.

Na III OBSERVED TERMS*

Config. $1s^2+$	Observed Terms		
$2s^2 2p^5$	$2p^5 \ ^2P^\circ$		
$2s \ 2p^6$	$2p^6 \ ^2S$		
	$ns \ (n \geq 3)$	$np \ (n \geq 3)$	$nd \ (n \geq 3)$
$2s^2 2p^4(^3P)nx$	$\left\{ \begin{array}{l} 3, 4s \ ^4P \\ 3, 4s \ ^2P \end{array} \right.$	$\begin{array}{l} 3p \ ^4S^\circ \ 3p \ ^4P^\circ \ 3p \ ^4D^\circ \\ 3p \ ^2S^\circ \ 3p \ ^2P^\circ \ 3p \ ^2D^\circ \end{array}$	$\begin{array}{l} 3d \ ^4P \ 3d \ ^4D \ 3d \ ^4F \\ 3, 4d \ ^2P \ 3, 4d \ ^2D \ 3d \ ^2F \end{array}$
$2s^2 2p^4(^1D)nx'$	$3, 4s' \ ^2D$	$3p' \ ^2P^\circ \ 3p' \ ^2D^\circ \ 3p' \ ^2F^\circ$	$3d' \ ^2S \ 3, 4d' \ ^2P \ 3, 4d' \ ^2D \ 3d' \ ^2F \ 3d' \ ^2G$
$2s^2 2p^4(^1S)nx''$	$3s'' \ ^2S$		$3d'' \ ^2D$

*For predicted terms in the spectra of the F I isoelectronic sequence, see Introduction.

Na IV

(O I sequence; 8 electrons)

 $Z=11$ Ground state $1s^2 2s^2 2p^4 {}^3P_2$ $2p^4 {}^3P_2$ 797741 cm^{-1}

I. P. 98.88 volts

The terms are from Söderqvist who has extended Vance's early work on this spectrum. In the 1946 reference Söderqvist states that the absolute values of the singlets as published in his Monograph should be decreased by 1000 cm^{-1} . This correction has been applied in the present list. The analysis is incomplete but 74 lines have been classified in the range 129 Å to 412 Å, and 40 terms found. No intersystem combinations have been observed and the uncertainty, x , may be considerable. The term $3d'' {}^3D$ has been calculated from its combination with $2p^5 {}^3P^o$ and added to the published list.

REFERENCES

- J. Söderqvist, Nova Acta Reg. Soc. Sci. Uppsala [IV] **9**, No. 7, 51 (1934). (I P) (T) (C L) (G D)
 J. Söderqvist, Ark. Mat. Astr. Fys. (Stockholm) **32A**, No. 19 p. 4 (1946). (C L)

Na IV

Na IV

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$2s^2 2p^4$	$2p^4 {}^3P$	2 1 0	0 1106 1576	-1106 -470	$2s^2 2p^3 ({}^2P^o) 3d$	$3d'' {}^3P^o$	2, 1, 0	663592	
$2s^2 2p^4$	$2p^4 {}^1D$	2	31118+x		$2s^2 2p^3 ({}^2P^o) 3d$	$3d'' {}^1D^o$	2	664904+x	
$2s^2 2p^4$	$2p^4 {}^1S$	0	66780+x		$2s^2 2p^3 ({}^2P^o) 3d$	$3d'' {}^3D^o$	3, 2, 1	665362	
$2s 2p^5$	$2p^5 {}^3P^o$	2 1 0	243682 244688 245238	-1006 -550	$2s^2 2p^3 ({}^2P^o) 3d$	$3d'' {}^1P^o$	1	665640+x	
$2s 2p^5$	$2p^5 {}^1P^o$	1	343972+x		$2s^2 2p^3 ({}^2P^o) 3d$	$3d'' {}^1F^o$	3	667696+x	
$2s^2 2p^3 ({}^4S^o) 3s$	$3s {}^3S^o$	1	486648		$2s^2 2p^3 ({}^4S^o) 4d$	$4d {}^3D^o$	3, 2, 1	684649	
$2s^2 2p^3 ({}^2D^o) 3s$	$3s' {}^3D^o$	3 2 1	525100 525119 525136	-19 -17	$2s^2 2p^3 ({}^2D^o) 4s$	$4s' {}^3D^o$	3, 2, 1	689755	
$2s^2 2p^3 ({}^2D^o) 3s$	$3s' {}^1D^o$	2	531696+x		$2s^2 2p^3 ({}^2D^o) 4s$	$4s' {}^1D^o$	2	692043+x	
$2s^2 2p^3 ({}^2P^o) 3s$	$3s'' {}^3P^o$	2, 1, 0	550176		$2s^2 2p^3 ({}^2P^o) 4s$	$4s'' {}^3P^o$	2, 1, 0	714937	
$2s^2 2p^3 ({}^2P^o) 3s$	$3s'' {}^1P^o$	1	557081+x		$2s^2 2p^3 ({}^2P^o) 4s$	$4s'' {}^1P^o$	1	716773+x	
$2s^2 2p^3 ({}^4S^o) 3d$	$3d {}^3D^o$	1 2 3	594893 594898 594941	5 43	$2s^2 2p^3 ({}^2D^o) 4d$	$4d' {}^3D^o$	3, 2, 1	730712	
$2s^2 2p^3 ({}^2D^o) 3d$	$3d' {}^3D^o$	3 2 1	638831 638942 638977	-111 -35	$2s^2 2p^3 ({}^2D^o) 4d$	$4d' {}^1P^o$	1	731948+x	
$2s^2 2p^3 ({}^2D^o) 3d$	$3d' {}^1P^o$	1	641468+x		$2s^2 2p^3 ({}^2D^o) 4d$	$4d' {}^3P^o$	2, 1, 0	732355	
$2s^2 2p^3 ({}^2D^o) 3d$	$3d' {}^3P^o$	2 1 0	643029 643304 (643396)	-275 (-92)	$2s^2 2p^3 ({}^2D^o) 4d$	$4d' {}^3S^o$	1	732940	
$2s^2 2p^3 ({}^2D^o) 3d$	$3d' {}^1D^o$	2	643912+x		$2s^2 2p^3 ({}^2D^o) 4d$	$4d' {}^1D^o$	2	733548+x	
$2s^2 2p^3 ({}^2D^o) 3d$	$3d' {}^3S^o$	1	644140		$2s^2 2p^3 ({}^2D^o) 4d$	$4d' {}^1F^o$	3	734195+x	
$2s^2 2p^3 ({}^4S^o) 4s$	$4s {}^3S^o$	1	644792		$2s^2 2p^3 ({}^2D^o) 5s$	$5s' {}^3D^o$	3, 2, 1	753352	
$2s^2 2p^3 ({}^2D^o) 3d$	$3d' {}^1F^o$	3	646711+x		$2s^2 2p^3 ({}^2P^o) 4d$	$4d'' {}^1D^o$	2	756045+x	
					$2s^2 2p^3 ({}^2P^o) 4d$	$4d'' {}^3D^o$	3, 2, 1	756367	
					$2s^2 2p^3 ({}^2P^o) 4d$	$4d'' {}^1F^o$	3	757261+x	
					$2s^2 2p^3 ({}^2D^o) 5d$	$5d' {}^3D^o$	3, 2, 1	772415	
					Na v (${}^4S_{3/2}$)	Limit	-----	797741	
					$2s 2p^4 ({}^4P) 3d$	$3d''' {}^3D$	3, 2, 1	813538	

Na IV OBSERVED TERMS*

Config. 1s ² +	Observed Terms	
2s ² 2p ⁴	{ 2p ⁴ ¹ S 2p ⁴ ³ P 2p ⁴ ¹ D	
2s 2p ⁵	{ 2p ⁵ ³ P ^o 2p ⁵ ¹ P ^o	
	ns (n ≥ 3)	nd (n ≥ 3)
2s ² 2p ³ (⁴ S ^o)nx	3, 4s ³ S ^o	3, 4d ³ D ^o
2s ² 2p ³ (² D ^o)nx'	{ 3-5s' ³ D ^o 3, 4s' ¹ D ^o	3, 4d' ³ S ^o 3, 4d' ³ P ^o 3-5d' ³ D ^o 3, 4d' ¹ D ^o 3, 4d' ¹ F ^o
2s ² 2p ³ (² P ^o)nx''	{ 3, 4s'' ³ P ^o 3, 4s'' ¹ P ^o	3d'' ³ P ^o 3, 4d'' ³ D ^o 3d'' ¹ P ^o 3, 4d'' ¹ D ^o 3, 4d'' ¹ F ^o
2s 2p ⁴ (⁴ P)nx'''		3d''' ³ D

*For predicted terms in the spectra of the O I isoelectronic sequence, see Introduction.

Na v

(N I sequence; 7 electrons)

Z=11

Ground state 1s² 2s² 2p³ ⁴S_{1/2}^o

2p³ ⁴S_{1/2}^o **1118170** cm⁻¹

I. P. 138.60 volts

Söderqvist has found 45 terms in this spectrum and classified 203 lines in the interval between 100 Å and 514 Å. No intersystem combinations have been observed. The series are short and the uncertainty, *x*, may be considerable.

REFERENCES

J. Söderqvist, Nova Acta Reg. Soc. Sci. Uppsala [IV] **9**, No. 7, 75 (1934). (T) (C L) (G D)
J. Söderqvist, Ark. Mat. Astr. Fys. (Stockholm) **32A**, No. 19 p. 4 (1946). (I P) (T) (C L)

Na v

Na v

Author	Config.	Desig.	<i>J</i>	Level	Interval	Author	Config.	Desig.	<i>J</i>	Level	Interval
2p ⁴ S ₂	2s ² 2p ³	2p ³ ⁴ S°	1½	0		$\overline{3s'}$ ⁴ D	2s 2p ³ (³ D°)3s	3s ^{IV} ⁴ D°	$\left\{ \begin{array}{c} \frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{array} \right\}$	878288	
2p ² D ₃ ² D ₂	2s ² 2p ³	2p ³ ² D°	$\frac{2\frac{1}{2}}{1\frac{1}{2}}$	$\frac{47570+x}{47595+x}$	-25	4s ⁴ P ₁ ⁴ P ₂ ⁴ P ₃	2s ² 2p ² (³ P)4s	4s ⁴ P	$\frac{1}{2}$ $\frac{1\frac{1}{2}}{2\frac{1}{2}}$	892244 892885 893822	641 937
2p ² P ₁ ² P ₂	2s ² 2p ³	2p ³ ² P°	$\frac{1}{2}$ $\frac{1\frac{1}{2}}$	$\frac{72454+x}{72493+x}$	39	$\overline{3s'}$ ² D	2s 2p ³ (³ D°)3s	3s ^{IV} ² D°	$\left\{ \begin{array}{c} \frac{1}{2} \\ \frac{1\frac{1}{2}}{2\frac{1}{2}} \end{array} \right\}$	$\frac{894095+x}{895944+x}$ $\frac{897147+x}{897147+x}$	
2p' ⁴ P ₃ ⁴ P ₂ ⁴ P ₁	2s 2p ⁴	2p ⁴ ⁴ P	$\frac{2\frac{1}{2}}{1\frac{1}{2}}$ $\frac{1\frac{1}{2}}{\frac{1}{2}}$	$\frac{215860}{216896}$ $\frac{217440}{217440}$	-1036 -544	4s ² P ₁ ² P ₂	2s ² 2p ² (³ P)4s	4s ² P	$\frac{1}{2}$ $\frac{1\frac{1}{2}}$	$\frac{895944+x}{897147+x}$	1203
2p' ² D ₃ ² D ₂	2s 2p ⁴	2p ⁴ ² D	$\frac{2\frac{1}{2}}{1\frac{1}{2}}$	$\frac{297116+x}{297150+x}$	-34	3d' ⁴ D	2s 2p ³ (⁵ S°)3d	3d''' ⁴ D°	$\left\{ \begin{array}{c} \frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{array} \right\}$	908717	
2p' ² S ₁	2s 2p ⁴	2p ⁴ ² S	$\frac{1}{2}$	349987+x		$\overline{3s'}$ ⁴ P	2s 2p ³ (³ P°)3s	3s ^V ⁴ P°	$\left\{ \begin{array}{c} \frac{1}{2} \\ \text{to} \\ 2\frac{1}{2} \end{array} \right\}$	919070	
2p' ² P ₂ ² P ₁	2s 2p ⁴	2p ⁴ ² P	$\frac{1\frac{1}{2}}{\frac{1}{2}}$	$\frac{371967+x}{373167+x}$	-1200	$\overline{4s}$ ² D	2s ² 2p ² (¹ D)4s	4s' ² D	$\left\{ \begin{array}{c} 2\frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	928053+x	
2p'' ² P ₂ ² P ₁	2p ⁵	2p ⁵ ² P°	$\frac{1\frac{1}{2}}{\frac{1}{2}}$	$\frac{567583+x}{569211+x}$	-1628	4d ² P ₂	2s ² 2p ² (³ P)4d	4d ² P	$\frac{1\frac{1}{2}}{\frac{1}{2}}$	937669+x	
3s ⁴ P ₁ ⁴ P ₂ ⁴ P ₃	2s ² 2p ² (³ P)3s	3s ⁴ P	$\frac{1}{2}$ $\frac{1\frac{1}{2}}$ $\frac{2\frac{1}{2}}$	671136 671790 672757	654 967	4d ⁴ D ₂₃ ⁴ D ₁	2s ² 2p ² (³ P)4d	4d ⁴ D	$\left\{ \begin{array}{c} 3\frac{1}{2} \\ 2\frac{1}{2} \\ 1\frac{1}{2} \\ \frac{1}{2} \end{array} \right\}$	939055 939858	-803
3s ² P ₁ ² P ₂	2s ² 2p ² (³ P)3s	3s ² P	$\frac{1}{2}$ $\frac{1\frac{1}{2}}$	$\frac{682470+x}{683673+x}$	1203	4d ² F ₃ ² F ₄	2s ² 2p ² (³ P)4d	4d ² F	$\frac{2\frac{1}{2}}{3\frac{1}{2}}$	940380+x 941392+x	1012
$\overline{3s}$ ² D	2s ² 2p ² (¹ D)3s	3s' ² D	$\left\{ \begin{array}{c} 2\frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	709277+x		4d ⁴ P ₃ ⁴ P ₂	2s ² 2p ² (³ P)4d	4d ⁴ P	$\frac{2\frac{1}{2}}{1\frac{1}{2}}$ $\frac{1\frac{1}{2}}{\frac{1}{2}}$	940716 940929	-213
$\overline{3s}$ ² S ₁	2s ² 2p ² (¹ S)3s	3s'' ² S	$\frac{1}{2}$	748640+x		4d ² D ₂ ² D ₃	2s ² 2p ² (³ P)4d	4d ² D	$\frac{1\frac{1}{2}}{2\frac{1}{2}}$	944022+x 944334+x	312
3d ² P ₂ ² P ₁	2s ² 2p ² (³ P)3d	3d ² P	$\frac{1\frac{1}{2}}{\frac{1}{2}}$	$\frac{792337+x}{792849+x}$	-512	$\overline{3p'}$ ² F ₄ ² F ₃	2s 2p ³ (³ D°)3p	3p ^{IV} ² F	$\frac{3\frac{1}{2}}{2\frac{1}{2}}$	949462+x 949984+x	-522
3d ⁴ D ₂₃ ⁴ D ₁	2s ² 2p ² (³ P)3d	3d ⁴ D	$\left\{ \begin{array}{c} 3\frac{1}{2} \\ 2\frac{1}{2} \\ 1\frac{1}{2} \\ \frac{1}{2} \end{array} \right\}$	$\frac{797060}{797270}$	-210	$\overline{4d}$ ² F	2s ² 2p ² (¹ D)4d	4d' ² F	$\left\{ \begin{array}{c} 3\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	973350+x	
3d ² F ₃ ² F ₄	2s ² 2p ² (³ P)3d	3d ² F	$\frac{2\frac{1}{2}}{3\frac{1}{2}}$	$\frac{797288+x}{798535+x}$	1247	$\overline{4d}$ ² D	2s ² 2p ² (¹ D)4d	4d' ² D	$\left\{ \begin{array}{c} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	974048+x	
3d ⁴ P ₃ ⁴ P ₂ ⁴ P ₁	2s ² 2p ² (³ P)3d	3d ⁴ P	$\frac{2\frac{1}{2}}{1\frac{1}{2}}$ $\frac{1\frac{1}{2}}{\frac{1}{2}}$	$\frac{798174}{798620}$ $\frac{798862}{798862}$	-446 -242	$\overline{3d'}$ ⁴ P ₃ ⁴ P ₂ ⁴ P ₁	2s 2p ³ (³ D°)3d	3d ^{IV} ⁴ P°	$\frac{2\frac{1}{2}}{1\frac{1}{2}}$ $\frac{1\frac{1}{2}}{\frac{1}{2}}$	$\frac{1004404}{1004626}$ $\frac{1004794}{1004794}$	-222 -168
3s' ⁴ S ₂	2s 2p ³ (⁵ S°)3s	3s''' ⁴ S°	$\frac{1}{2}$	801950		$\overline{3d'}$ ⁴ D	2s 2p ³ (³ D°)3d	3d ^{IV} ⁴ D°	$\left\{ \begin{array}{c} \frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{array} \right\}$	1008214	
3d ² D ₂ ² D ₃	2s ² 2p ² (³ P)3d	3d ² D	$\frac{1\frac{1}{2}}{2\frac{1}{2}}$	$\frac{808546+x}{808920+x}$	374	$\overline{3d'}$ ⁴ S ₂	2s 2p ³ (³ D°)3d	3d ^{IV} ⁴ S°	$\frac{1}{2}$	1008941	
$\overline{3d}$ ² F ₄ ² F ₃	2s ² 2p ² (¹ D)3d	3d' ² F	$\frac{3\frac{1}{2}}{2\frac{1}{2}}$	$\frac{828509+x}{828692+x}$	-183	$\overline{3d'}$ ² F ₄ ² F ₃	2s 2p ³ (³ D°)3d	3d ^{IV} ² F°	$\frac{3\frac{1}{2}}{2\frac{1}{2}}$	$\frac{1010088+x}{1010565+x}$	-477
$\overline{3d}$ ² D ₂ ² D ₃	2s ² 2p ² (¹ D)3d	3d' ² D	$\frac{1\frac{1}{2}}{2\frac{1}{2}}$	$\frac{832075+x}{832228+x}$	153	$\overline{5d}$ ² F	2s ² 2p ² (¹ D)5d	5d' ² F	$\left\{ \begin{array}{c} 3\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	1038208+x	
$\overline{3d}$ ² P ₁ ² P ₂	2s ² 2p ² (¹ D)3d	3d' ² P	$\frac{1}{2}$ $\frac{1\frac{1}{2}}$	$\frac{837431+x}{837723+x}$	292	$\overline{5d}$ ² D	2s ² 2p ² (¹ D)5d	5d' ² D	$\left\{ \begin{array}{c} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	1038845+x	
$\overline{3d}$ ² S ₁	2s ² 2p ² (¹ D)3d	3d' ² S	$\frac{1}{2}$	842067+x							
3p' ⁴ P	2s 2p ³ (⁵ S°)3p	3p''' ⁴ P	$\left\{ \begin{array}{c} \frac{1}{2} \\ \text{to} \\ 2\frac{1}{2} \end{array} \right\}$	847539							
$\overline{3d}$ ² D	2s ² 2p ² (¹ S)3d	3d'' ² D	$\left\{ \begin{array}{c} 1\frac{1}{2} \\ \text{to} \\ 2\frac{1}{2} \end{array} \right\}$	866780+x							
							Na vi (³ P ₀)	Limit		1118170	

Na V OBSERVED TERMS*

Config. $1s^2 +$	Observed Terms		
$2s^2 2p^3$	$\{ 2p^3 \ ^4S^\circ$	$2p^3 \ ^2P^\circ$	$2p^3 \ ^2D^\circ$
$2s \ 2p^4$	$\{ 2p^4 \ ^2S$	$2p^4 \ ^4P$ $2p^4 \ ^2P$	$2p^4 \ ^2D$
$2p^5$	$2p^5 \ ^2P^\circ$		
	$ns \ (n \geq 3)$	$np \ (n \geq 3)$	$nd \ (n \geq 3)$
$2s^2 2p^2(^3P)nx$	$\{ \begin{matrix} 3, 4s \ ^4P \\ 3, 4s \ ^2P \end{matrix}$		$\begin{matrix} 3, 4d \ ^4P & 3, 4d \ ^4D \\ 3, 4d \ ^2P & 3, 4d \ ^2D & 3, 4d \ ^2F \end{matrix}$
$2s^2 2p^2(^1D)nx'$		$3, 4s' \ ^2D$	$3d' \ ^2S \quad 3d' \ ^2P \quad 3-5d' \ ^2D \quad 3-5d' \ ^2F$
$2s^2 2p^2(^1S)nx''$	$3s'' \ ^2S$		$3d'' \ ^2D$
$2s \ 2p^3(^5S^\circ)nx'''$	$3s''' \ ^4S^\circ$	$3p''' \ ^4P$	$3d''' \ ^4D^\circ$
$2s \ 2p^3(^3D^\circ)nx^{IV}$	$\{ \begin{matrix} 3s^{IV} \ ^4D^\circ \\ 3s^{IV} \ ^2D^\circ \end{matrix}$	$3p^{IV} \ ^2F$	$3d^{IV} \ ^4S^\circ \quad 3d^{IV} \ ^4P^\circ \quad 3d^{IV} \ ^4D^\circ \quad 3d^{IV} \ ^2F^\circ$
$2s \ 2p^3(^3P^\circ)nx^V$	$3s^V \ ^4P^\circ$		

*For predicted terms in the spectra of the N I isoelectronic sequence, see Introduction.

Na VI

(C I sequence; 6 electrons)

$Z=11$

Ground state $1s^2 2s^2 2p^2 \ ^3P_0$

$2p^2 \ ^3P_0$ 1390558 cm^{-1}

I. P. 172.36 volts

The analysis is by Söderqvist, who has found 63 terms and classified 134 lines in the range between 80 Å and 638 Å. He determines the relative values of terms of different multiplicity from the series limits, although he lists a few observed singlet-triplet combinations. His term $2p^4 \ ^1D$ has been corrected to agree with the two observed combinations.

Söderqvist gives the quintet term $2p^3 \ ^5S_2^\circ$ at 103187 cm^{-1} above the ground state zero. From isoelectronic sequence data Robinson estimates this value as 103508 cm^{-1} . The later value has been used in the table and all quintet terms adjusted accordingly. The uncertainty, x , may be a few hundred cm^{-1} .

REFERENCES

- J. Söderqvist, Ark. Mat. Astr. Fys. (Stockholm) **32A**, No. 19 p. 4 (1946). (I P) (T) (C L)
H. A. Robinson, unpublished material (March 1948). (T)

Na VI

Na VI

Author	Config.	Desig.	<i>J</i>	Level	Interval	Author	Config.	Desig.	<i>J</i>	Level	Interval
2p 3P_0 3P_1 3P_2	2s ² 2p ²	2p ² 3P	0 1 2	0 698 1858	698 1160	3p' 3D	2s 2p ² (2D)3p	3p' $^3D^\circ$	1, 2, 3	1040223	
2p 1D_2	2s ² 2p ²	2p ² 1D	2	35358		3d' $^5D_{23}$	2s 2p ² (4P)3d	3d 5D	$\begin{Bmatrix} 0 \\ 1 \\ 2 \\ 3 \\ 4 \end{Bmatrix}$	1041771+x	
2p 1S_0	2s ² 2p ²	2p ² 1S	0	74274		3d' 5P_3 5P_2 5P_1	2s 2p ² (4P)3d	3d 5P	3 2 1	1045793+x 1046220+x 1046548+x	-427 -328
2p' 5S_2	2s 2p ³	2p ³ $^5S^\circ$	2	103508+x		3d' 3P_2 3P_1	2s 2p ² (4P)3d	3d 3P	2 1 0	1047408 1048104	-696
2p' 3D_3 3D_2 3D_1	2s 2p ³	2p ³ $^3D^\circ$	3 2 1	204131 204222 204260	-91 -38	3d' 3F_2 3F_3 3F_4	2s 2p ² (4P)3d	3d 3F	2 3 4	1053885 1054497 1055260	612 763
2p' 3P	2s 2p ³	2p ³ $^3P^\circ$	2, 1, 0	241341		3d' 3D_1 3D_2 3D_3	2s 2p ² (4P)3d	3d 3D	1 2 3	1067760 1067971 1068258	211 287
2p' 1D_2	2s 2p ³	2p ³ $^1D^\circ$	2	312175		$\overline{3p'}$ 1F_3	2s 2p ² (2D)3p	3p' $^1F^\circ$	3	1071896	
2p' 3S_1	2s 2p ³	2p ³ $^3S^\circ$	1	320589		$\overline{3p'}$ 1D_2	2s 2p ² (2D)3p	3p' $^1D^\circ$	2	1077752	
2p' 1P_1	2s 2p ³	2p ³ $^1P^\circ$	1	350179		4s 3P_2	2s ² 2p($^2P^\circ$)4s	4s $^3P^\circ$	0 1 2	1090756	
2p'' 3P_2 3P_1 3P_0	2p ⁴	2p ⁴ 3P	2 1 0	477277 478597 479156	-1320 -559	$\overline{3d'}$ 3F	2s 2p ² (2D)3d	3d' 3F	2, 3, 4	1125323	
2p'' 1D_2	2p ⁴	2p ⁴ 1D	2	539310		4d 3F_2	2s ² 2p($^2P^\circ$)4d	4d $^3F^\circ$	2 3 4	1128693	
3s 3P_1 3P_2	2s ² 2p($^2P^\circ$)3s	3s $^3P^\circ$	0 1 2	807324 808795	1471	$\overline{3d'}$ 3P	2s 2p ² (2D)3d	3d' 3P	0, 1, 2	1130631	
3s 1P_1	2s ² 2p($^2P^\circ$)3s	3s $^1P^\circ$	1	817598		4d 1D_2	2s ² 2p($^2P^\circ$)4d	4d $^1D^\circ$	2	1131032	
3p 3P_1 3P_2	2s ² 2p($^2P^\circ$)3p	3p 3P	0 1 2	872577 873287	710	4d 3D_1 3D_2 3D_3	2s ² 2p($^2P^\circ$)4d	4d $^3D^\circ$	1 2 3	1133491 1133871 1134746	380 875
3d 3F_2	2s ² 2p($^2P^\circ$)3d	3d $^3F^\circ$	2 3 4	919476		$\overline{3d'}$ 3D	2s 2p ² (2D)3d	3d' 3D	1, 2, 3	1134094	
3d 1D_2	2s ² 2p($^2P^\circ$)3d	3d $^1D^\circ$	2	920706		4d 3P_2	2s ² 2p($^2P^\circ$)4d	4d $^3P^\circ$	0 1 2	1136378	
3s' 5P_1 5P_2 5P_3	2s 2p ² (4P)3s	3s 5P	1 2 3	923059+x 923765+x 924708+x	706 943	4d 1F_3	2s ² 2p($^2P^\circ$)4d	4d $^1F^\circ$	3	1140721	
3d 3D_1 3D_2 3D_3	2s ² 2p($^2P^\circ$)3d	3d $^3D^\circ$	1 2 3	929774 929999 930510	225 511	$\overline{3d'}$ 3S_1	2s 2p ² (2D)3d	3d' 3S	1	1144276	
3d 3P_2 3P_1 3P_0	2s ² 2p($^2P^\circ$)3d	3d $^3P^\circ$	2 1 0	933915 934463 934745	-548 -282	$\overline{3d'}$ 1F_3	2s 2p ² (2D)3d	3d' 1F	3	1147708	
3d 1F_3	2s ² 2p($^2P^\circ$)3d	3d $^1F^\circ$	3	945309		$\overline{3d'}$ 1D_2	2s 2p ² (2D)3d	3d' 1D	2	1147735	
3d 1P_1	2s ² 2p($^2P^\circ$)3d	3d $^1P^\circ$	1	946392		$\overline{3d'}$ 1P_1	2s 2p ² (2D)3d	3d' 1P	1	1151140	
3s' 3P_0 3P_1 3P_2	2s 2p ² (4P)3s	3s 3P	0 1 2	949778 950367 951389	589 1022	4s' 5P_3	2s 2p ² (4P)4s	4s 5P	1 2 3	1205485+x	
3p' 3S_1	2s 2p ² (4P)3p	3p $^3S^\circ$	1	970835		4s' 3P_2	2s 2p ² (4P)4s	4s 3P	0 1 2	1214191	
3p' 3D_2 3D_3	2s 2p ² (4P)3p	3p $^3D^\circ$	1 2 3	996011 996734	723	5d 3D_3	2s ² 2p($^2P^\circ$)5d	5d $^3D^\circ$	1 2 3	1228205	
3p' 3P_1 3P_2	2s 2p ² (4P)3p	3p $^3P^\circ$	0 1 2	1005068 1005713	645	$\overline{3s'}$ 3D	2s 2p ² (2D)3s	3s' 3D	1, 2, 3	1016274	
$\overline{3s'}$ 3D	2s 2p ² (2D)3s	3s' 3D	1, 2, 3	1016274		$\overline{3s'}$ 1D_2	2s 2p ² (2D)3s	3s' 1D	2	1033221	

Na VI—Continued

Na VI—Continued

Author	Config.	Desig.	<i>J</i>	Level	Interval	Author	Config.	Desig.	<i>J</i>	Level	Interval
5d ¹ F ₃	2s ² 2p(² P°) 5d	5d ¹ F°	3	1230972		$\overline{4d}'$ ³ F	2s 2p ² (² D) 4d	4d' ³ F	2, 3, 4	1334585	
	2s 2p ² (⁴ P) 4d	4d ⁵ P	1			$\overline{4d}'$ ³ P	2s 2p ² (² D) 4d	4d' ³ P	0, 1, 2	1335519	
4d' ⁵ P ₃			2			$\overline{4d}'$ ³ D	2s 2p ² (² D) 4d	4d' ³ D	1, 2, 3	1337017	
			3	1250152+x			2s 2p ² (⁴ P) 5d	5d ⁵ P	1		
4d' ³ F ₂	2s 2p ² (⁴ P) 4d	4d ³ F	2	1253369	578				2		
³ F ₃			3	1253947	803	5d' ⁵ P ₃			3	1343510+x	
³ F ₄			4	1254750			Na VII (² P _{1/2} °)	Limit	-----	1390558	
4d' ³ D	2s 2p ² (⁴ P) 4d	4d ³ D	1, 2, 3	1258613		$\overline{5d}'$ ³ F	2s 2p ² (² D) 5d	5d' ³ F	2, 3, 4	1429862	
3p'' ³ P	1s ² 2p ³ (⁴ S°) 3p	3p ^{1v} ³ P	0, 1, 2	1265583							
6d ¹ F ₃	2s ² 2p(² P°) 6d	6d ¹ F°	3	1279991							

March 1948.

Na VI OBSERVED TERMS*

Config. 1s ² +	Observed Terms		
2s ² 2p ²	{ 2p ² ¹ S 2p ² ³ P 2p ² ¹ D		
2s 2p ³	{ 2p ³ ⁵ S° 2p ³ ³ P° 2p ³ ³ D° 2p ³ ³ S° 2p ³ ¹ P° 2p ³ ¹ D°		
2p ⁴	{ 2p ⁴ ³ P 2p ⁴ ¹ D		
	<i>ns</i> (<i>n</i> ≥ 3)	<i>np</i> (<i>n</i> ≥ 3)	<i>nd</i> (<i>n</i> ≥ 3)
2s ² 2p(² P°) <i>nx</i>	{ 3, 4s ³ P° 3s ¹ P°	3p ³ P	3-5d ³ P° 3-5d ³ D° 3, 4d ³ F° 3d ¹ P° 3, 4d ¹ D° 3-6d ¹ F°
2s 2p ² (⁴ P) <i>nx</i>	{ 3, 4s ⁵ P 3, 4s ³ P	3p ³ S° 3p ³ P° 3p ³ D°	3-5d ⁵ P 3d ⁵ D 3d ³ P 3, 4d ³ D 3, 4d ³ F
2s 2p ² (² D) <i>nx'</i>	{ 3s' ³ D 3s' ¹ D	3p' ³ D° 3p' ¹ D° 3p' ¹ F°	3d' ³ S 3, 4d' ³ P 3, 4d' ³ D 3-5d' ³ F 3d' ¹ P 3d' ¹ D 3d' ¹ F
2p ³ (⁴ S°) <i>nx</i> ^{IV}		3p ^{1v} ³ P	

*For predicted terms in the spectra of the C I isoelectronic sequence, see Introduction.

Na VII

(B I sequence; 5 electrons)

Z = 11Ground state 1s² 2s² 2p ²P_{1/2}°2p ²P_{1/2}° 1681679 cm⁻¹

I. P. 208.444 volts

All of the terms are taken from Söderqvist's later publication. The Grotrian diagram in the earlier paper should be extended to include the more complete analysis of 1944. He has classified 158 lines in the region between 62 Å and 491 Å.

The absolute values of the doublet terms are well determined. Those of the quartets are derived from the *nd* ⁴D° (*n* = 3, 4, 5) series; and the relative uncertainty *x*, may be a few hundred cm⁻¹. No intersystem combinations have been observed.

Na VII—Continued

REFERENCES

J. Söderqvist, Nova Acta Reg. Soc. Sci. Uppsala [IV] 9, No. 7, 93 (1934). (T) (C L) (G D)

J. Söderqvist, Ark. Mat. Astr. Fys. (Stockholm) 30A, No. 11 p. 9 (1944). (I P) (T) (C L)

Na VII

Na VII

Author	Config.	Desig.	J	Level	Interval	Author	Config.	Desig.	J	Level	Interval
$2p \ ^2P_1$ $\ ^2P_2$	$2s^2(^1S)2p$	$2p \ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	0 2139	2139	$\overline{3p'} \ ^2D_2$ $\ ^2D_3$	$2s \ 2p(^1P^\circ)3p$	$3p' \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	1251674 1252014	340
$2p' \ ^4P_1$ $\ ^4P_2$ $\ ^4P_3$	$2s \ 2p^2$	$2p^2 \ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$115187+x$ $115920+x$ $116987+x$	733 1067	$\overline{3p'} \ ^2P_1$ $\ ^2P_2$	$2s \ 2p(^1P^\circ)3p$	$3p' \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	1253353 1253779	426
$2p' \ ^2D_3$ $\ ^2D_2$	$2s \ 2p^2$	$2p^2 \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	205412 205448	-36	$\overline{3p'} \ ^2S_1$	$2s \ 2p(^1P^\circ)3p$	$3p' \ ^2S$	$\frac{1}{2}$	1258878	
$2p' \ ^2S_1$	$2s \ 2p^2$	$2p^2 \ ^2S$	$\frac{1}{2}$	264400		$3s'' \ ^4P_2$ $\ ^4P_3$	$2p^2(^3P)3s$	$3s'' \ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$1290221+x$ $1291755+x$	1534
$2p' \ ^2P_1$ $\ ^2P_2$	$2s \ 2p^2$	$2p^2 \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	283869 285189	1320	$\overline{3d'} \ ^2F$	$2s \ 2p(^1P^\circ)3d$	$3d' \ ^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	1292333	
$2p'' \ ^4S_2$	$2p^3$	$2p^3 \ ^4S^\circ$	$1\frac{1}{2}$	$367481+x$		$4s \ ^2S_1$	$2s^2(^1S)4s$	$4s \ ^2S$	$\frac{1}{2}$	1294914	
$2p'' \ ^2D_3$ $\ ^2D_2$	$2p^3$	$2p^3 \ ^2D^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$	412311 412395	-84	$\overline{3d'} \ ^2D_2$ $\ ^2D_3$	$2s \ 2p(^1P^\circ)3d$	$3d' \ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	1303445 1303643	198
$2p'' \ ^2P_1$ $\ ^2P_2$	$2p^3$	$2p^3 \ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	465017 465111	94	$\overline{3d'} \ ^2P$	$2s \ 2p(^1P^\circ)3d$	$3d' \ ^2P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	1306468	
$3s \ ^2S_1$	$2s^2(^1S)3s$	$3s \ ^2S$	$\frac{1}{2}$	951347		$\overline{3s''} \ ^2D_2$ $\ ^2D_3$	$2p^2(^1D)3s$	$3s''' \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	1331137 1331974	837
$3p \ ^2P_2$	$2s^2(^1S)3p$	$3p \ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	1008418		$4d \ ^2D_2$ $\ ^2D_3$	$2s^2(^1S)4d$	$4d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	1335809 1335889	80
$3d \ ^2D_2$ $\ ^2D_3$	$2s^2(^1S)3d$	$3d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	1060580 1060699	119		$2p^2(^3P)3p$	$3p'' \ ^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$		
$3s' \ ^4P_1$ $\ ^4P_2$ $\ ^4P_3$	$2s \ 2p(^3P^\circ)3s$	$3s \ ^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$1077458+x$ $1078190+x$ $1079520+x$	732 1330	$3p'' \ ^4D_4$				$1338659+x$	
$3s' \ ^2P_1$ $\ ^2P_2$	$2s \ 2p(^3P^\circ)3s$	$3s \ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	1103222 1104620	1398		$2p^2(^3P)3p$	$3p'' \ ^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$		$1345036+x$
$3p' \ ^2P_1$ $\ ^2P_2$	$2s \ 2p(^3P^\circ)3p$	$3p \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	1126810 1127431	621	$3p'' \ ^2D$	$2p^2(^3P)3p$	$3p'' \ ^2D^\circ$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	1348721	
$3p' \ ^2D_2$ $\ ^2D_3$	$2s \ 2p(^3P^\circ)3p$	$3p \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	1154779 1156180	1401	$3p'' \ ^4S_2$	$2p^2(^3P)3p$	$3p'' \ ^4S^\circ$	$1\frac{1}{2}$	$1363160+x$	
$3p' \ ^2S_1$	$2s \ 2p(^3P^\circ)3p$	$3p \ ^2S$	$\frac{1}{2}$	1172339		$\overline{3p''} \ ^2F_3$ $\ ^2F_4$	$2p^2(^1D)3p$	$3p''' \ ^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	1377822 1378295	473
	$2s \ 2p(^3P^\circ)3d$	$3d \ ^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	$1185931+x$ $1186190+x$ $1186666+x$	259 476	$3d'' \ ^2F_3$ $\ ^2F_4$	$2p^2(^3P)3d$	$3d'' \ ^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	1388500? 1388969?	469
$3d' \ ^4D_2$ $\ ^4D_3$ $\ ^4D_4$						$3d'' \ ^2D$	$2p^2(^3P)3d$	$3d'' \ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	1390448?	
$3d' \ ^2D_2$ $\ ^2D_3$	$2s \ 2p(^3P^\circ)3d$	$3d \ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	1186628 1187885	1257	$\overline{3p''} \ ^2D$	$2p^2(^1D)3p$	$3p''' \ ^2D^\circ$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	1392764	
$3d' \ ^4P_3$ $\ ^4P_2$ $\ ^4P_1$	$2s \ 2p(^3P^\circ)3d$	$3d \ ^4P^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	$1192538+x$ $1193059+x$ $1193402+x$	-521 -343	$3d'' \ ^4P_3$ $\ ^4P_2$ $\ ^4P_1$	$2p^2(^3P)3d$	$3d'' \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	$1399238+x$ $1399771+x$ $1400059+x$	-533 -288
$\overline{3s'} \ ^2P$	$2s \ 2p(^1P^\circ)3s$	$3s' \ ^2P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	1198287		$\overline{3d''} \ ^2D$	$2p^2(^1D)3d$	$3d''' \ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	1415636	
$3d' \ ^2F_3$ $\ ^2F_4$	$2s \ 2p(^3P^\circ)3d$	$3d \ ^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	1209908 1211236	1328		$2s \ 2p(^3P^\circ)4s$	$4s \ ^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$		$1423050+x$
$3d' \ ^2P_2$ $\ ^2P_1$	$2s \ 2p(^3P^\circ)3d$	$3d \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	1217189 1217955	-766	$4s' \ ^4P_3$					

Na VII—Continued

Na VII—Continued

Author	Config.	Desig.	J	Level	Interval	Author	Config.	Desig.	J	Level	Interval
$\overline{3d}''$ 2F_3 2F_4	$2p^2(1D)3d$	$3d'''$ 2F	$2\frac{1}{2}$ $3\frac{1}{2}$	1428717 1428798	81	$4p'$ 2D	$2s$ $2p(1P^\circ)4p$	$4p'$ 2D	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	1561885	
$\overline{3d}''$ 2P_1 2P_2	$2p^2(1D)3d$	$3d'''$ 2P	$\frac{1}{2}$ $1\frac{1}{2}$	1432135 1432606	471	$7d$ 2D	$2s^2(1S)7d$	$7d$ 2D	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	1570078	
$4s'$ 2P_2	$2s$ $2p(3P^\circ)4s$	$4s$ ${}^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	1432595		$\overline{4d}'$ 2F	$2s$ $2p(1P^\circ)4d$	$4d'$ ${}^2F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	1577813?	
$4p'$ 2P_1 2P_2	$2s$ $2p(3P^\circ)4p$	$4p$ 2P	$\frac{1}{2}$ $1\frac{1}{2}$	1442711 1443165	454	$5p'$ 2P	$2s$ $2p(3P^\circ)5p$	$5p$ 2P	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	1578354	
$4p'$ 2D_2 2D_3	$2s$ $2p(3P^\circ)4p$	$4p$ 2D	$1\frac{1}{2}$ $2\frac{1}{2}$	1452095 1453349	1254	$5p'$ 2D	$2s$ $2p(3P^\circ)5p$	$5p$ 2D	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	1583742	
$5d$ 2D_2 2D_3	$2s^2(1S)5d$	$5d$ 2D	$1\frac{1}{2}$ $2\frac{1}{2}$	1461518 1461588	70	$5d'$ 4D	$2s$ $2p(3P^\circ)5d$	$5d$ ${}^4D^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	1589481+x	
$4d'$ 4D_2 4D_3 4D_4	$2s$ $2p(3P^\circ)4d$	$4d$ ${}^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	1462587+x 1462631+x 1463462+x	44 831	$5d'$ 4P	$2s$ $2p(3P^\circ)5d$	$5d$ ${}^4P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	1590240+x	
$4d'$ 2D_3	$2s$ $2p(3P^\circ)4d$	$4d$ ${}^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	1464051		$5d'$ 2F_3 2F_4	$2s$ $2p(3P^\circ)5d$	$5d$ ${}^2F^\circ$	$\frac{2\frac{1}{2}}{3\frac{1}{2}}$	1592815 1593915	11
$4d'$ 4P_3	$2s$ $2p(3P^\circ)4d$	$4d$ ${}^4P^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	1465059+x		$8d$ 2D	$2s^2(1S)8d$	$8d$ 2D	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	1596400	
$4d'$ 2F_3 2F_4	$2s$ $2p(3P^\circ)4d$	$4d$ ${}^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	1471559 1472727	1168	$4p''$ 4D	$2p^2(3P)4p$	$4p''$ ${}^4D^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	1646320+x	
$4d'$ 2P_2 2P_1	$2s$ $2p(3P^\circ)4d$	$4d$ ${}^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	1473809 1474526	-717	$6d'$ 4P $6d'$ 4D	$2s$ $2p(3P^\circ)6d$	$\left\{ \begin{smallmatrix} 6d$ ${}^4P^\circ$ \\ $6d$ ${}^4D^\circ$ \end{smallmatrix} \right\}	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	1657724+x	
$6d$ 2D	$2s^2(1S)6d$	$6d$ 2D	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	1529463		$4d''$ 4P_3 4P_2	$2p^2(3P)4d$	$4d''$ 4P	$\frac{2\frac{1}{2}}{1\frac{1}{2}}$ $\frac{1\frac{1}{2}}{\frac{1}{2}}$	1668514+x 1668855+x	-341
$\overline{4s}'$ 2P	$2s$ $2p(1P^\circ)4s$	$4s'$ ${}^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	1538951		Na VIII ($1S^\circ$)		Limit	-----	1681679	

October 1946.

Na VII OBSERVED TERMS*

Config. $1s^2+$	Observed Terms		
$2s^2(1S)2p$	$2p$ ${}^2P^\circ$		
$2s$ $2p^2$	$\left\{ \begin{array}{l} 2p^2$ 2S $2p^2$ 4P $2p^2$ 2P $2p^2$ 2D \\ $2p^3$ ${}^4S^\circ$ $2p^3$ ${}^2P^\circ$ $2p^3$ ${}^2D^\circ$ \end{array} \right.		
$2p^3$			
	ns ($n \geq 3$)	np ($n \geq 3$)	nd ($n \geq 3$)
$2s^2(1S)nx$	$3, 4s$ 3S	$3p$ ${}^2P^\circ$	$3-8d$ 2D
$2s$ $2p(3P^\circ)nx$	$\left\{ \begin{array}{l} 3, 4s$ ${}^4P^\circ$ $3, 4s$ ${}^2P^\circ$ \end{array} \right.	$3p$ 2S $3-5p$ 2P $3-5p$ 2D	$3-6d$ ${}^4P^\circ$ $3-6d$ ${}^4D^\circ$ $3, 4d$ ${}^2P^\circ$ $3, 4d$ ${}^2D^\circ$ $3-5d$ ${}^4F^\circ$
$2s$ $2p(1P^\circ)nx'$	$3, 4s'$ ${}^2P^\circ$	$3p'$ 2S $3p'$ 2P $3, 4p'$ 2D	$3d'$ ${}^2P^\circ$ $3d'$ ${}^2D^\circ$ $3, 4d'$ ${}^2F^\circ$
$2p^2(3P)nx''$	$\left\{ \begin{array}{l} 3s''$ 4P \end{array} \right.	$3p''$ ${}^4S^\circ$ $3p''$ ${}^4P^\circ$ $3, 4p''$ ${}^4D^\circ$ $3p''$ ${}^2D^\circ$	$3, 4d''$ 4P $3d''$ 2D $3d''$ 2F
$2p^2(1D)nx'''$	$3s'''$ 2D	$3p'''$ ${}^2D^\circ$ $3p'''$ ${}^2F^\circ$	$3d'''$ 2P $3d'''$ 2D $3d'''$ 2F

*For predicted terms in the spectra of the Br isoelectronic sequence, see Introduction.

Na VIII

(Be I sequence; 4 electrons)

 $Z=11$ Ground state $1s^2 2s^2 {}^1S_0$ $2s^2 {}^1S_0$ 2131139 cm^{-1}

I. P. 264.155 volts

Eighty-six lines have been classified by Söderqvist, all but three of which are in the region between 51 Å and 117 Å. No intersystem combinations are known, but the absolute term values are well determined by the series, the relative uncertainty x being probably a few hundred cm^{-1} .

REFERENCE

J. Söderqvist, Ark. Mat. Astr. Fys. (Stockholm) **30A**, No. 11, p. 7 (1944). (I P) (T) (C L)

Na VIII

Na VIII

Author	Config.	Desig.	J	Level	Interval	Author	Config.	Desig.	J	Level	Interval
$2s {}^1S_0$	$2s^2$	$2s^2 {}^1S$	0	0		$3p' {}^1P_1$	$2p({}^2P^\circ)3p$	$3p {}^1P$	1	1432991	
$2p {}^3P_0$	$2s({}^2S)2p$	$2p {}^3P^\circ$	0	$126053+x$	730 1604	$3p' {}^3D_1$	$2p({}^2P^\circ)3p$	$3p {}^3D$	1	$1439584+x$	846 1620
3P_1			1	$126783+x$		3D_2			2	$1440430+x$	
3P_2			2	$128387+x$		3D_3			3	$1442050+x$	
$2p {}^1P_1$	$2s({}^2S)2p$	$2p {}^1P^\circ$	1	243223		$3p' {}^3S_1$	$2p({}^2P^\circ)3p$	$3p {}^3S$	1	$1452568+x$	
$2p' {}^3P_0$	$2p^2$	$2p^2 {}^3P$	0	$327667+x$	827 1405		$2p({}^2P^\circ)3p$	$3p {}^3P$	0		884
3P_1			1	$328494+x$		$3p' {}^3P_1$			1	$1460244+x$	
3P_2			2	$329899+x$		3P_2			2	$1461128+x$	
$2p' {}^1D_2$	$2p^2$	$2p^2 {}^1D$	2	361046		$3d' {}^1D_2$	$2p({}^2P^\circ)3d$	$3d {}^1D^\circ$	2	1469055	
$2p' {}^1S_0$	$2p^2$	$2p^2 {}^1S$	0	446099		$3p' {}^1D_2$	$2p({}^2P^\circ)3p$	$3p {}^1D$	2	1474598	
$3s {}^3S_1$	$2s({}^2S)3s$	$3s {}^3S$	1	$1240255+x$		$3p' {}^1S_0$	$2p({}^2P^\circ)3p$	$3p {}^1S$	0	1481521	
$3s {}^1S_0$	$2s({}^2S)3s$	$3s {}^1S$	0	1262799		$3d' {}^3D_1$	$2p({}^2P^\circ)3d$	$3d {}^3D^\circ$	1	$1485329+x$	292 628
$3p {}^1P_1$	$2s({}^2S)3p$	$3p {}^1P^\circ$	1	1294214		3D_2			2	$1485621+x$	
						3D_3			3	$1486249+x$	
$3d {}^3D_1$	$2s({}^2S)3d$	$3d {}^3D$	1	$1327399+x$	37 121	$3d' {}^3P_2$	$2p({}^2P^\circ)3d$	$3d {}^3P^\circ$	2	$1492167+x$	-642 -358
3D_2			2	$1327436+x$		3P_1			1	$1492809+x$	
3D_3			3	$1327557+x$		3P_0			0	$1493167+x$	
$3d {}^1D_2$	$2s({}^2S)3d$	$3d {}^1D$	2	1347756		$3d' {}^1F_3$	$2p({}^2P^\circ)3d$	$3d {}^1F^\circ$	3	1507690	
$3s' {}^3P_0$	$2p({}^2P^\circ)3s$	$3s {}^3P^\circ$	0	$1399858+x$	805 1714	$3d' {}^1P_1$	$2p({}^2P^\circ)3d$	$3d {}^1P^\circ$	1	1513677	
3P_1			1	$1400663+x$		$4s {}^3S_1$	$2s({}^2S)4s$	$4s {}^3S$	1	$1649682+x$	
3P_2			2	$1402377+x$		$4s {}^1S_0$	$2s({}^2S)4s$	$4s {}^1S$	0	1656830	
$3s' {}^1P_1$	$2p({}^2P^\circ)3s$	$3s {}^1P^\circ$	1	1426049		$4p {}^1P_1$	$2s({}^2S)4p$	$4p {}^1P^\circ$	1	1673388	

Na VIII—Continued

Na VIII—Continued

Author	Config.	Desig.	<i>J</i>	Level	Interval	Author	Config.	Desig.	<i>J</i>	Level	Interval
4 <i>d</i> ³ D	2 <i>s</i> (² S)4 <i>d</i>	4 <i>d</i> ³ D	1, 2, 3	1683549+ <i>x</i>	1283	5 <i>d</i> ¹ D ₂	2 <i>s</i> (² S)5 <i>d</i>	5 <i>d</i> ¹ D	2	1848978	
4 <i>d</i> ¹ D ₂	2 <i>s</i> (² S)4 <i>d</i>	4 <i>d</i> ¹ D	2	1689982		6 <i>p</i> ¹ P ₁	2 <i>s</i> (² S)6 <i>p</i>	6 <i>p</i> ¹ P ^o	1	1930912	
4 <i>p</i> ' ¹ P ₁	2 <i>p</i> (² P ^o)4 <i>p</i>	4 <i>p</i> ¹ P	1	1813205		6 <i>d</i> ³ D	2 <i>s</i> (² S)6 <i>d</i>	6 <i>d</i> ³ D	1, 2, 3	1933601+ <i>x</i>	
4 <i>p</i> ' ³ D ₂ ³ D ₃	2 <i>p</i> (² P ^o)4 <i>p</i>	4 <i>p</i> ³ D	1 2 3	1816179+ <i>x</i> 1817462+ <i>x</i>		6 <i>d</i> ¹ D ₂	2 <i>s</i> (² S)6 <i>d</i>	6 <i>d</i> ¹ D	2	1935242	
	2 <i>p</i> (² P ^o)4 <i>p</i>	4 <i>p</i> ³ P	0 1 2	1823044+ <i>x</i>		5 <i>p</i> ' ³ P	2 <i>p</i> (² P ^o)5 <i>p</i>	5 <i>p</i> ³ P	0, 1, 2	1988852+ <i>x</i>	
						5 <i>p</i> ' ¹ D ₂	2 <i>p</i> (² P ^o)5 <i>p</i>	5 <i>p</i> ¹ D	2	1990558	
4 <i>p</i> ' ³ P ₂						5 <i>d</i> ' ¹ D ₂	2 <i>p</i> (² P ^o)5 <i>d</i>	5 <i>d</i> ¹ D ^o	2	1991118	
4 <i>d</i> ' ¹ D ₂	2 <i>p</i> (² P ^o)4 <i>d</i>	4 <i>d</i> ¹ D ^o	2	1827472		5 <i>d</i> ' ³ D	2 <i>p</i> (² P ^o)5 <i>d</i>	5 <i>d</i> ³ D ^o	1, 2, 3	1994540+ <i>x</i>	
4 <i>p</i> ' ¹ D ₂	2 <i>p</i> (² P ^o)4 <i>p</i>	4 <i>p</i> ¹ D	2	1827658		5 <i>d</i> ' ³ P	2 <i>p</i> (² P ^o)5 <i>d</i>	5 <i>d</i> ³ P ^o	2, 1, 0	1995095+ <i>x</i>	
	2 <i>p</i> (² P ^o)4 <i>d</i>	4 <i>d</i> ³ D ^o	1 2 3	1833704+ <i>x</i>		5 <i>d</i> ' ¹ F ₃	2 <i>p</i> (² P ^o)5 <i>d</i>	5 <i>d</i> ¹ F ^o	3	1998029	
4 <i>d</i> ' ³ D ₃						6 <i>p</i> ' ³ D	2 <i>p</i> (² P ^o)6 <i>p</i>	6 <i>p</i> ³ D	1, 2, 3	2077097+ <i>x</i>	
4 <i>d</i> ' ³ P ₂	2 <i>p</i> (² P ^o)4 <i>d</i>	4 <i>d</i> ³ P ^o	2 1 0	1835175+ <i>x</i>		6 <i>d</i> ' ³ D	2 <i>p</i> (² P ^o)6 <i>d</i>	6 <i>d</i> ³ D ^o	1, 2, 3	2080630+ <i>x</i>	
						6 <i>d</i> ' ³ P	2 <i>p</i> (² P ^o)6 <i>d</i>	6 <i>d</i> ³ P ^o	2, 1, 0	2081335+ <i>x</i>	
4 <i>d</i> ' ¹ F ₃	2 <i>p</i> (² P ^o)4 <i>d</i>	4 <i>d</i> ¹ F ^o	3	1838762		6 <i>d</i> ' ¹ F ₃	2 <i>p</i> (² P ^o)6 <i>d</i>	6 <i>d</i> ¹ F ^o	3	2083106	
5 <i>p</i> ¹ P ₁	2 <i>s</i> (² S)5 <i>p</i>	5 <i>p</i> ¹ P ^o	1	1838911							
4 <i>d</i> ' ¹ P ₁	2 <i>p</i> (² P ^o)4 <i>d</i>	4 <i>d</i> ¹ P ^o	1	1843384			Na IX (² S _½)	<i>Limit</i>	-----	2131139	
5 <i>d</i> ³ D	2 <i>s</i> (² S)5 <i>d</i>	5 <i>d</i> ³ D	1, 2, 3	1848841+ <i>x</i>							

May 1946.

Na VIII OBSERVED TERMS*

Config. 1s ² +	Observed Terms		
2s ²	2s ² ¹ S		
2s(2S)2p	{ 2p ³ P ^o 2p ¹ P ^o		
2p ²	{ 2p ² ³ P 2p ² ¹ S 2p ² ¹ D		
	ns (n ≥ 3)	np (n ≥ 3)	nd (n ≥ 3)
2s(2S)nx	{ 3, 4s ³ S 3, 4s ¹ S	3-6p ¹ P ^o	3-6d ³ D 3-6d ¹ D
2p(2P ^o)nx	{ 3s ³ P ^o 3s ¹ P ^o	3p ³ S 3-5p ³ P 3, 4, 6p ³ D 3p ¹ S 3, 4p ¹ P 3-5p ¹ D	3-6d ³ P ^o 3-6d ³ D ^o 3, 4d ¹ P ^o 3-5d ¹ D ^o 3-6d ¹ F ^o

*For predicted terms in the spectra of the Be I isoelectronic sequence, see Introduction.

Na IX

(Li I sequence; 3 electrons)

 $Z=11$ Ground state $1s^2 2s^2 S_{1/2}$ $2s^2 S_{1/2}$ 2418520 cm^{-1}

I. P. 299.78 volts

The analysis is by Söderqvist, who has classified 22 lines in this spectrum. They occur in the region 81 Å to 44 Å, with the exception of one line at 681 Å.

Some of the relative levels have been connected by a study of the Rydberg denominators in the isoelectronic sequence rather than by the Ritz combination principle.

REFERENCE

J. Söderqvist, Ark. Mat. Astr. Fys. (Stockholm) **30A**, No. 11, p. 1 (1944). (I P) (T) (C L)

Na IX

Na IX

Author	Config.	Desig.	J	Level	Interval	Author	Config.	Desig.	J	Level	Interval
$2s^2 S_1$	$2s$	$2s^2 S$	$\frac{1}{2}$	0		$5p^2 P_{21}$	$5p$	$5p^2 P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	2059605	
$2p^2 P_1$ $2p^2 P_2$	$2p$	$2p^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	144038 146688	2650	$5d^2 D_2$ $2D_3$	$5d$	$5d^2 D$	$\frac{1}{2}$ $2\frac{1}{2}$	2062835 2062911	76
$3s^2 S_1$	$3s$	$3s^2 S$	$\frac{1}{2}$	1375944		$6p^2 P_{21}$	$6p$	$6p^2 P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	2169668	
$3p^2 P_1$ $2P_2$	$3p$	$3p^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	1415368 1416130	762	$6d^2 D_2$ $2D_3$	$6d$	$6d^2 D$	$\frac{1}{2}$ $2\frac{1}{2}$	2171366 2171553	187
$3d^2 D_2$ $2D_3$	$3d$	$3d^2 D$	$\frac{1}{2}$ $2\frac{1}{2}$	1429980 1430204	224	$7p^2 P_{21}$	$7p$	$7p^2 P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	2235886	
$4s^2 S_1$	$4s$	$4s^2 S$	$\frac{1}{2}$	1840336		$7d^2 D_2$ $2D_3$	$7d$	$7d^2 D$	$\frac{1}{2}$ $2\frac{1}{2}$	2237139 2237165	26
$4p^2 P_{21}$	$4p$	$4p^2 P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	1856665							
$4d^2 D_2$ $2D_3$	$4d$	$4d^2 D$	$\frac{1}{2}$ $2\frac{1}{2}$	1862222 1862572	350						
$5s^2 S_1$	$5s$	$5s^2 S$	$\frac{1}{2}$	2051922?			Na x ($1S_0$)	Limit	-----	2418520	

May 1946.

MAGNESIUM

Mg I

12 electrons

 $Z=12$ Ground state $1s^2 2s^2 2p^6 3s^2 {}^1S_0$ $3s^2 {}^1S_0$ 61669.14 cm^{-1}

I. P. 7.644 volts

The most complete term array is given in Paschen's 1931 paper, which has been extensively used in the present compilation.

Paschen lists the combinations $3d {}^3D - nf {}^3F^\circ$ ($n=4,5$) and $3d {}^1D - nf {}^1F^\circ$ ($n=4-9$), deriving from his infrared observations practically coincident values for the terms $nf {}^3F^\circ$ and $nf {}^1F^\circ$ for $n=4$ and $n=5$. Assuming that the two F-series were coincident throughout, Russell, Babcock, and the writer extended both series by the identification of Paschen's lines in the Infrared Solar Spectrum and by the discovery of the constant solar wave-number separation $3d {}^3D - 3d {}^1D$ for predicted successive series members. The constancy of this separation and the behavior of the solar lines in the disk and spot spectra leave no doubt as to the correctness of the identifications, although laboratory observations are lacking for confirmation of many of the lines. The term values in the table for the F-series ($nf {}^1F^\circ$ to $n=14$ and $nf {}^3F^\circ$ to $n=12$) have been calculated from solar data, with a slight adjustment to Paschen's absolute values of $3d {}^3D$ and $3d {}^1D$, as indicated in the 1945 reference below.

The three-decimal values listed for the terms $3p {}^3P^\circ$ and $3d {}^3D$ are from Meissner's paper.

Sawyer suggests that Paschen's $6d {}^1D$ term (58023.27 cm^{-1} in the table) may have the designation $3p^2 {}^1D$, in which case the n -values of the higher series members should be decreased by one unit. In accordance with the observations of Shenstone and Russell on related series, the $nd {}^1D$ series may well have absorbed the $3p^2 {}^1D$ term. The present analysis indicates that throughout the D-series the singlets are lower than the corresponding triplet terms.

The singlet and triplet terms are well connected by intersystem combinations.

REFERENCES

- F. Paschen, Sitz. Preuss. Akad. Wiss. **32**, 709 (1931). (I P) (T) (C L)
 F. Paschen, Ann. der Phys. [5] **12**, 511 (1932). (T) (C L)
 A. G. Shenstone and H. N. Russell, Phys. Rev. **39**, 431 (1932).
 H. E. White, *Introduction to Atomic Spectra* p. 179 (McGraw-Hill Book Co., Inc., New York, N. Y., 1934). (G D)
 K. W. Meissner, Ann. der Phys. [5] **31**, 518 (1938). (T) (C L)
 L. G. Mundie and K. W. Meissner, Phys. Rev. **65**, 265 (1944). (I S)
 H. D. Babcock and C. E. Moore, Astroph. J. **101**, 374 (1945). (T) (C L)

Mg I

Mg I

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
3s ²	3s ² 1S	0	0. 00		3s(2S)7d	7d 3D	3, 2, 1	59317. 4	
3s(2S)3p	3p 3P°	0 1 2	21850. 368 21870. 426 21911. 140	20. 058 40. 714	3s(2S)7f	7f 3F°	2, 3, 4	59400. 77	
3s(2S)3p	3p 1P°	1	35051. 36		3s(2S)7f	7f 1F°	3	59400. 77	
3s(2S)4s	4s 3S	1	41197. 37		3s(2S)9s	9s 3S	1	59648. 2	
3s(2S)4s	4s 1S	0	43503. 0		3s(2S)8d	8d 1D	2	59690. 02	
3s(2S)3d	3d 1D	2	46403. 14		3s(2S)8d	8d 3D	3, 2, 1	59880. 3	
3s(2S)4p	4p 3P°	0, 1 2	47847. 7 47851. 8	4. 1	3s(2S)8f	8f 3F°	2, 3, 4	59935. 38	
3s(2S)3d	3d 3D	3 2 1	47957. 035 47957. 018 47957. 047	0. 017 -0. 029	3s(2S)8f	8f 1F°	3	59935. 38	
3s(2S)4p	4p 1P°	1	49346. 6		3s(2S)10s	10s 3S	1	60103. 5	
3s(2S)5s	5s 3S	1	51872. 36		3s(2S)9d	9d 1D	2	60127. 31	
3s(2S)5s	5s 1S	0	52556. 37		3s(2S)9d	9d 3D	3, 2, 1	60263. 0	
3s(2S)4d	4d 1D	2	53134. 70		3s(2S)9f	9f 3F°	2, 3, 4	60301. 30	
3s(2S)4d	4d 3D	3, 2, 1	54192. 16		3s(2S)9f	9f 1F°	3	60301. 30	
3s(2S)5p	5p 3P°	0 1 2	 54252. 6		3s(2S)11s	11s 3S	1	60420. 2	
3s(2S)4f	4f 3F°	2, 3, 4	54676. 38		3s(2S)10d	10d 1D	2	60435. 15	
3s(2S)4f	4f 1F°	3	54676. 38		3s(2S)10d	10d 3D	3, 2, 1	60534. 5	
3s(2S)5p	5p 1P°	1	54699. 4		3s(2S)10f	10f 3F°	2, 3, 4	60562. 64	
3s(2S)6s	6s 3S	1	55891. 83		3s(2S)10f	10f 1F°	3	60562. 64	
3s(2S)6s	6s 1S	0	56187. 03		3s(2S)12s	12s 3S	1	60649. 2	
3s(2S)5d	5d 1D	2	56308. 43		3s(2S)11d	11d 1D	2	60658. 37	
3s(2S)5d	5d 3D	3, 2, 1	56968. 31		3s(2S)11d	11d 3D	3, 2, 1	60734. 0	
3s(2S)6p	6p 3P°	0, 1 2	57018. 8 57020. 1	1. 3	3s(2S)11f	11f 3F°	2, 3, 4	60755. 78	
3s(2S)5f	5f 3F°	2, 3, 4	57204. 22		3s(2S)11f	11f 1F°	3	60755. 78	
3s(2S)5f	5f 1F°	3	57204. 22		3s(2S)13s	13s 3S	1	60820. 9	
3p ²	3p ² 3P	0 1 2	57812. 72 57833. 28 57873. 89	20. 56 40. 61	3s(2S)13d	13d 3D	3, 2, 1	61002. 2	
3s(2S)7s	7s 3S	1	57853. 5		3s(2S)12d	12d 1D	2	60826. 6	
3s(2S)7s	7s 1S	0	58009. 46		3s(2S)12d	12d 3D	3, 2, 1	60884. 8	
3s(2S)6d	6d 1D	2	58023. 27		3s(2S)12f	12f 3F°	2, 3, 4	60902. 53	
3s(2S)6d	6d 3D	3, 2, 1	58442. 62		3s(2S)12f	12f 1F°	3	60902. 53	
3s(2S)7p	7p 3P°	0, 1, 2	58478. 4		3s(2S)14s	14s 3S	1	60952. 0	
3s(2S)6f	6f 3F°	2, 3, 4	58575. 54		3s(2S)13d	13d 1D	2	60955. 8	
3s(2S)6f	6f 1F°	3	58575. 54		3s(2S)13d	13d 3D	3, 2, 1	61002. 2	
3s(2S)8s	8s 3S	1	58962. 49		3s(2S)13f	13f 1F°	3	61016. 42	
3s(2S)7d	7d 1D	2	59041. 09		3s(2S)14d	14d 3D	3, 2, 1	61094. 6	
					3s(2S)14f	14f 1F°	3	61106. 98	
					Mg II (2S _{1/2})	Limit		61669. 14	
					3p(2P°)3d	3d 1F°	3	80693. 2	
					3p(2P°)3d	3d 3D°	1 2 3	83510. 73 83519. 98 83536. 22	9. 25 16. 24

July 1947.

MgI OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 +$	Observed Terms				
$3s^2$	$3s^2 \ ^1S$				
$3s(^2S)3p$	{	$3p \ ^3P^\circ$			
		$3p \ ^1P^\circ$			
$3p^2$		$3p^2 \ ^3P$			
		$ns \ (n \geq 4)$	$np \ (n \geq 4)$	$nd \ (n \geq 3)$	$nf \ (n \geq 4)$
$3s(^2S)nx$	{	$4-14s \ ^3S$ $4- \ 7s \ ^1S$	$4-7p \ ^3P^\circ$ $4, \ 5p \ ^1P^\circ$	$3-14d \ ^3D$ $3-13d \ ^1D$	$4-12f \ ^3F^\circ$ $4-14f \ ^1F^\circ$
$3p(^2P^\circ)nx$		{		$3d \ ^3D^\circ$ $3d \ ^1F^\circ$	

*For predicted terms in the spectra of the Mg I isoelectronic sequence, see Introduction.

Mg II

(Na I sequence; 11 electrons)

Z=12

Ground state 1s² 2s² 2p⁶ 3s 2S_{1/2}

3s 2S_{1/2} 121267.41 cm⁻¹

I. P. 15.03 volts

The analysis is from Fowler and Paschen-Götze. Mundie and Meissner calculate the separation of 3d 2D to be 1.000±0.002 cm⁻¹ (entered in brackets in the table). In 1913 A. S. King observed the line at 4481 Å (3d 2D—4f 2F°) as double, the violet component being about twice as strong as the red, thus indicating that the term 3d 2D is inverted.

REFERENCES

A. S. King, *Astroph. J.* **38**, 327 (1913).
F. Paschen und R. Götze, *Seriengesetze der Linienspektren*, p. 103 (Julius Springer, Berlin, 1922). (T) (C L)
A. Fowler, *Report on Series in Line Spectra*, p. 118 (Fleetway Press, London, 1922). (I P) (T) (C L)
R. F. Bacher and S. Goudsmit, *Atomic Energy States*, p. 273 (McGraw-Hill Book Co., Inc., New York and London, 1932). (T)
H. E. White, *Introduction to Atomic Spectra*, p. 98 (McGraw-Hill Book Co., Inc., New York, N. Y., 1934). (G D)
L. G. Mundie and K. W. Meissner, *Phys. Rev.* **65**, 272 (1944). (I S)

Mg II

Mg II

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
3s	3s 2S	$\frac{1}{2}$	0. 00	91. 55	7d	7d 2D	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	112198. 0	[−1. 000]
3p	3p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	35669. 42 35760. 97		7f	7f $^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	112301. 8	
4s	4s 2S	$\frac{1}{2}$	69805. 19		7g	7g 2G	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 4\frac{1}{2} \end{array} \right\}$	112310. 2	
3d	3d 2D	$2\frac{1}{2}$ $1\frac{1}{2}$	71490. 41 71491. 32		9s	9s 2S	$\frac{1}{2}$	114292. 2	
4p	4p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	80620. 8 80651. 3	30. 5	8d	8d 2D	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	114335. 7	
5s	5s 2S	$\frac{1}{2}$	92786. 2	14. 1	8f	8f $^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	114403. 6	
4d	4d 2D	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	93312. 1		8g	8g 2G	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 4\frac{1}{2} \end{array} \right\}$	114408. 6	
4f	4f $^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	93800. 0		9f	9f $^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	115845. 1	
5p	5p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	97454. 9 97469. 0		9g	9g 2G	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 4\frac{1}{2} \end{array} \right\}$	115848. 6	
6s	6s 2S	$\frac{1}{2}$	103198. 1	7. 6	10f	10f $^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	116875. 7	
5d	5d 2D	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	103421. 1		10g	10g 2G	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 4\frac{1}{2} \end{array} \right\}$	116878. 2	
5f	5f $^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	103690. 2		11f	11f $^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	117638. 3	
6p	6p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	105623. 1 105630. 7		11g	11g 2G	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 4\frac{1}{2} \end{array} \right\}$	117640. 6	
7s	7s 2S	$\frac{1}{2}$	108784. 7		12f	12f $^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	118218. 5	
6d	6d 2D	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	108900. 9		12g	12g 2G	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 4\frac{1}{2} \end{array} \right\}$	118220. 2	
6f	6f $^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	109062. 6						
6g	6g 2G	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 4\frac{1}{2} \end{array} \right\}$	109073. 2						
8s	8s 2S	$\frac{1}{2}$	112129. 8		Mg III (1S_0)	Limit	-----	121267. 41	

May 1947.

Mg III

(Ne I sequence; 10 electrons)

 $Z=12$ Ground state $1s^2 2s^2 2p^6 \ ^1S_0$ $2p^6 \ ^1S_0$ 646364 cm^{-1}

I. P. 80.12 volts

The analysis has been taken from Söderqvist's Monograph. The term designations he assigns on the assumption of LS -coupling are given with his notation under the heading "Author" in the table.

As for Ne I, the $j\bar{l}$ -coupling notation is introduced in the general form suggested by Racah. Shortley has, however, pointed out that the configurations $2p^5 3s$, $2p^5 3p$, and $2p^5 3d$ are much closer to LS -coupling than to $j\bar{l}$ -coupling.

REFERENCES

- J. Söderqvist, Nova Acta Reg. Soc. Sci. Uppsala [IV] 9, No. 7, 22 (1934). (I P) (T) (C L) (G D).
 G. Racah, Phys. Rev. 61, 537 (L) (1942).
 G. Shortley, unpublished material (1948).

Mg III

Mg III

Author	Config.	Desig.	<i>J</i>	Level	Author	Config.	Desig.	<i>J</i>	Level
$2p \ ^1S_0$	$2p^6$	$2p^6 \ ^1S$	0	0. 0	$4d \ ^3P_1$	$2p^5(^2P_{1/2})4d$	$4d [\frac{1}{2}]^\circ$	0 1	581747
$3s \ ^3P_2$ 3P_1	$2p^5(^2P_{1/2})3s$	$3s [1\frac{1}{2}]^\circ$	2 1	425649. 1 426877. 0	$4d \ ^1P_1$	"	$4d [1\frac{1}{2}]^\circ$	1	583448
$3s \ ^3P_0$ 1P_1	$2p^5(^2P_{3/2})3s$	$3s' [\frac{1}{2}]^\circ$	0 1	427861. 1 431539. 0	$4d \ ^3D_1$	$2p^5(^2P_{3/2})4d$	$4d' [1\frac{1}{2}]^\circ$	2 1	585473
$3p_{10} \ ^3S_1$	$2p^5(^2P_{1/2})3p$	$3p [\frac{1}{2}]$	1	467387. 3	$5s \ ^3P_1$	$2p^5(^2P_{1/2})5s$	$5s [1\frac{1}{2}]^\circ$	2 1	589116
$3p_9 \ ^3D_3$ $3p_8 \ ^3D_2$	"	$3p [2\frac{1}{2}]$	3 2	474062. 6 474663. 6	$5s \ ^1P_1$	$2p^5(^2P_{3/2})5s$	$5s' [\frac{1}{2}]^\circ$	0 1	591191
$3p_7 \ ^3D_1$ $3p_6 \ ^1D_2$	"	$3p [1\frac{1}{2}]$	1 2	475511. 4 477444. 9	$5d \ ^3P_1$	$2p^5(^2P_{1/2})5d$	$5d [\frac{1}{2}]^\circ$	0 1	605345
$3p_3 \ ^3P_0$	"	$3p [\frac{1}{2}]$	0	479275. 3	$5d \ ^1P_1$	"	$5d [1\frac{1}{2}]^\circ$	1	606230
$3p_5 \ ^1P_1$ $3p_4 \ ^3P_2$	$2p^5(^2P_{3/2})3p$	$3p' [1\frac{1}{2}]$	1 2	478383. 8 478855. 5	$5d \ ^3D_1$	$2p^5(^2P_{3/2})5d$	$5d' [1\frac{1}{2}]^\circ$	2 1	608332
$3p_2 \ ^3P_1$ $3p_1 \ ^1S_0$	"	$3p' [\frac{1}{2}]$	1 0	479465. 4 484439. 3	$6s \ ^3P_1$	$2p^5(^2P_{1/2})6s$	$6s [1\frac{1}{2}]^\circ$	2 1	609166
$3d \ ^3P_0$ 3P_1	$2p^5(^2P_{1/2})3d$	$3d [\frac{1}{2}]^\circ$	0 1	530186. 4 530429. 5	$6s \ ^1P_1$	$2p^5(^2P_{3/2})6s$	$6s' [\frac{1}{2}]^\circ$	0 1	611299
$3d \ ^3P_2$	"	$3d [1\frac{1}{2}]^\circ$	2	530972. 0	$6d \ ^1P_1$	$2p^5(^2P_{1/2})6d$	$6d [1\frac{1}{2}]^\circ$	1	618483
$3d \ ^3F_4$ 3F_3	"	$3d [3\frac{1}{2}]^\circ$	4 3	531569. 9 531838. 5	$6d \ ^3D_1$	$2p^5(^2P_{3/2})6d$	$6d' [1\frac{1}{2}]^\circ$	2 1	620598
$3d \ ^3F_2$ 1F_3	"	$3d [2\frac{1}{2}]^\circ$	2 3	532731. 8 532978. 0	$7d \ ^1P_1$	$2p^5(^2P_{1/2})7d$	$7d [1\frac{1}{2}]^\circ$	1	625958
$3d \ ^1P_1$	"	$3d [1\frac{1}{2}]^\circ$	1	534204. 1	$7d \ ^3D_1$	$2p^5(^2P_{3/2})7d$	$7d' [1\frac{1}{2}]^\circ$	2 1	628105
$3d \ ^1D_2$ 3D_3	$2p^5(^2P_{3/2})3d$	$3d' [2\frac{1}{2}]^\circ$	2 3	534782. 2 534931. 0	$8d \ ^1P_1$	$2p^5(^2P_{1/2})8d$	$8d [1\frac{1}{2}]^\circ$	1	630795
$3d \ ^3D_2$ 3D_1	"	$3d' [1\frac{1}{2}]^\circ$	2 1	535185. 9 536156. 7					
$4s \ ^3P_1$	$2p^5(^2P_{1/2})4s$	$4s [1\frac{1}{2}]^\circ$	2 1	546529		Mg IV ($^2P_{1/2}$)	Limit	-----	646364
$4s \ ^1P_1$	$2p^5(^2P_{3/2})4s$	$4s' [\frac{1}{2}]^\circ$	0 1	548727		Mg IV ($^2P_{3/2}$)	Limit	-----	648590

July 1947.

Mg IV

Mg IV

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$2s^2 2p^5$	$2p^5 \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	$\begin{matrix} 0 \\ 2226 \end{matrix}$	—2226	$2s^2 2p^4(^1D)3d$	$3d' \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	$\begin{matrix} 711622 \\ 711865 \end{matrix}$	243
$2s \ 2p^6$	$2p^6 \ ^2S$	$\frac{1}{2}$	311527		$2s^2 2p^4(^1D)3d$	$3d' \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	$\begin{matrix} 712120 \\ 713389 \end{matrix}$	1269
$2s^2 2p^4(^3P)3s$	$3s \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	$\begin{matrix} 543727.0 \\ 545143.5 \\ 545962.1 \end{matrix}$	$\begin{matrix} -1416.5 \\ -818.6 \end{matrix}$	$2s^2 2p^4(^1D)3d$	$3d' \ ^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	$\begin{matrix} 713660 \\ 714330 \end{matrix}$	
$2s^2 2p^4(^3P)3s$	$3s \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	$\begin{matrix} 553659 \\ 555338 \end{matrix}$	—1679	$2s^2 2p^4(^1D)3d$	$3d' \ ^2S$	$\frac{1}{2}$	714330	
$2s^2 2p^4(^1D)3s$	$3s' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	$\begin{matrix} 582571 \\ 582589 \end{matrix}$	—18	$2s^2 2p^4(^3P)4s$	$4s \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	$\begin{matrix} 723254 \\ 724809 \end{matrix}$	—1555
$2s^2 2p^4(^3P)3p$	$3p \ ^4P^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	$\begin{matrix} 596527.3 \\ 597071.9 \\ 597589.9 \end{matrix}$	$\begin{matrix} -544.6 \\ -518.0 \end{matrix}$	$2s^2 2p^4(^1S)3d$	$3d'' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	$\begin{matrix} 752927 \\ 752965 \end{matrix}$	—38
$2s^2 2p^4(^3P)3p$	$3p \ ^4D^\circ$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	$\begin{matrix} 603143.3 \\ 604007.4 \\ 604666.6 \end{matrix}$	$\begin{matrix} -864.1 \\ -659.2 \end{matrix}$	$2s^2 2p^4(^3P)4d$	$4d \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	$\begin{matrix} 767454 \\ 770799 \end{matrix}$	—3345
$2s^2 2p^4(^3P)3p$	$3p \ ^4S^\circ$	$1\frac{1}{2}$	612240.3		$2s^2 2p^4(^3P)4d$	$4d \ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$\begin{matrix} 767769 \\ 768728 \end{matrix}$	959
$2s^2 2p^4(^1S)3s$	$3s'' \ ^2S$	$\frac{1}{2}$	624102		$2s^2 2p^4(^3P)4d$	$4d \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	$\begin{matrix} 769397 \\ 770056 \end{matrix}$	659
$2s^2 2p^4(^3P)3d$	$3d \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	$\begin{matrix} 676837 \\ 677805 \end{matrix}$	—968	$2s^2 2p^4(^1S)4s$	$4s'' \ ^2S$	$\frac{1}{2}$	797062	
$2s^2 2p^4(^3P)3d$	$3d \ ^4F$	$4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$	677355		$2s^2 2p^4(^1D)4d$	$4d' \ ^2P$	$\left\{ \begin{matrix} \frac{1}{2} \\ 1\frac{1}{2} \end{matrix} \right\}$	802272	
$2s^2 2p^4(^3P)3d$	$3d \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	$\begin{matrix} 678403 \\ 680030 \end{matrix}$	—1627	$2s^2 2p^4(^1D)4d$	$4d' \ ^2D$	$\left\{ \begin{matrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{matrix} \right\}$	803023	
$2s^2 2p^4(^3P)3d$	$3d \ ^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	680510		$2s^2 2p^4(^1D)4d$	$4d' \ ^2S$	$\frac{1}{2}$	803769	
$2s^2 2p^4(^3P)3d$	$3d \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	$\begin{matrix} 681024 \\ 682471 \end{matrix}$	1447	$2s^2 2p^4(^3P)5d$	$5d \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	$\begin{matrix} 809677 \\ 811362 \end{matrix}$	—1685
					$2s^2 2p^4(^3P)5d$	$5d \ ^2P$	$\left\{ \begin{matrix} \frac{1}{2} \\ 1\frac{1}{2} \end{matrix} \right\}$	810543	
					Mg v (3P_2)	Limit	-----	881759	

March 1948.

Mg IV OBSERVED TERMS*

Config. $1s^2+$	Observed Terms			
$2s^2 2p^5$	$2p^5 \ ^2P^\circ$			
$2s \ 2p^6$	$2p^6 \ ^2S$			
	$ns \ (n \geq 3)$	$np \ (n \geq 3)$		$nd \ (n \geq 3)$
$2s^2 2p^4(^3P)nx$	$\left\{ \begin{matrix} 3s \ ^4P \\ 3, 4s \ ^2P \end{matrix} \right.$	$3p \ ^4S^\circ$	$3p \ ^4P^\circ$	$3p \ ^4D^\circ$
$2s^2 2p^4(^1D)nx'$				
$2s^2 2p^4(^1S)nx''$				
		$3s' \ ^2D$		
	$3, 4s'' \ ^2S$			
			$3, 4d' \ ^2S$	$3, 4d' \ ^2P$
				$3, 4d' \ ^2D$
				$3d' \ ^2F$
				$3d' \ ^2F$
				$3d'' \ ^2D$

*For predicted terms in the spectra of the FI isoelectronic sequence, see Introduction.

Mg v

(O I sequence; 8 electrons)

 $Z=12$ Ground state $1s^2 2s^2 2p^4 {}^3P_2$ $2p^4 {}^3P_2$ 1139421 cm^{-1}

I. P. 141.23 volts

Söderqvist has found 53 terms and classified 113 lines in this spectrum in the interval between 92 Å and 355 Å. No intersystem combinations have been observed and the uncertainty, x , may be considerable.

REFERENCE

J. Söderqvist, Ark. Mat. Astr. Fys. (Stockholm) **32A**, No. 19 p. 4 (1946). (I P) (T) (C L)

Mg v

Mg v

Author	Config.	Desig.	J	Level	Inter- val	Author	Config.	Desig.	J	Level	Inter- val
$2p$ 3P_2 3P_1 3P_0	$2s^2 2p^4$	$2p^4$ 3P	2 1 0	0 1780 2519	—1780 —739	$\overline{3d}$ 3D_3 3D_2 3D_1	$2s^2 2p^3({}^2P^\circ)3d$	$3d''$ ${}^3D^\circ$	3 2 1	902047 902441 902682	—394 —241
$2p$ 1D_2	$2s^2 2p^4$	$2p^4$ 1D	2	$36348+x$		$\overline{3d}$ 1P_1	$2s^2 2p^3({}^2P^\circ)3d$	$3d''$ ${}^1P^\circ$	1	$902907+x$	
$2p$ 1S_0	$2s^2 2p^4$	$2p^4$ 1S	0	$77712+x$		$\overline{3d}$ 1F_3	$2s^2 2p^3({}^2P^\circ)3d$	$3d''$ ${}^1F^\circ$	3	$905211+x$	
$2p'$ 3P_2 3P_1 3P_0	$2s 2p^5$	$2p^5$ ${}^3P^\circ$	2 1 0	283211 284827 285708	—1616 —881	$4s$ 3S_1	$2s^2 2p^3({}^4S^\circ)4s$	$4s$ ${}^3S^\circ$	1	910639	
$2p'$ 1P_1	$2s 2p^5$	$2p^5$ ${}^1P^\circ$	1	$397906+x$		$3s'$ 3P_2 3P_1	$2s 2p^4({}^4P)3s$	$3s'''$ 3P	2 1 0	940455 941048	—593
$3s$ 3S_1	$2s^2 2p^3({}^4S^\circ)3s$	$3s$ ${}^3S^\circ$	1	684544		$\overline{4s}$ 3D	$2s^2 2p^3({}^2D^\circ)4s$	$4s'$ ${}^3D^\circ$	3, 2, 1	962027	
$\overline{3s}$ 3D_3 3D_2 3D_1	$2s^2 2p^3({}^2D^\circ)3s$	$3s'$ ${}^3D^\circ$	3 2 1	727718 727763 727787	—45 —24	$4d$ 3D_1 3D_2 3D_3	$2s^2 2p^3({}^4S^\circ)4d$	$4d$ ${}^3D^\circ$	1 2 3	962378 962395 962427	17 32
$\overline{3s}$ 1D_2	$2s^2 2p^3({}^2D^\circ)3s$	$3s'$ ${}^1D^\circ$	2	$735976+x$		$\overline{4s}$ 1D_2	$2s^2 2p^3({}^2D^\circ)4s$	$4s'$ ${}^1D^\circ$	2	$965189+x$	
$\overline{3s}$ 3P_1 3P_2	$2s^2 2p^3({}^2P^\circ)3s$	$3s''$ ${}^3P^\circ$	0 1 2	756536 756589	53	$\overline{4s}$ 3P	$2s^2 2p^3({}^2P^\circ)4s$	$4s''$ ${}^3P^\circ$	0, 1, 2	990599^*	
$\overline{3s}$ 1P_1	$2s^2 2p^3({}^2P^\circ)3s$	$3s''$ ${}^1P^\circ$	1	$765049+x$		$\overline{4s}$ 1P_1	$2s^2 2p^3({}^2P^\circ)4s$	$4s''$ ${}^1P^\circ$	1	$993795+x$	
$3d$ 3D_1 3D_2 3D_3	$2s^2 2p^3({}^4S^\circ)3d$	$3d$ ${}^3D^\circ$	1 2 3	821963 821977 822071	14 94	$5s$ 3S_1	$2s^2 2p^3({}^4S^\circ)5s$	$5s$ ${}^3S^\circ$	1	1002125	
$\overline{3d}$ 3D	$2s^2 2p^3({}^2D^\circ)3d$	$3d'$ ${}^3D^\circ$	1, 2, 3	871221		$\overline{4d}$ 3D	$2s^2 2p^3({}^2D^\circ)4d$	$4d'$ ${}^3D^\circ$	1, 2, 3	1013878	
$\overline{3d}$ 1P_1	$2s^2 2p^3({}^2D^\circ)3d$	$3d'$ ${}^1P^\circ$	1	$873862+x$		$\overline{4d}$ 1P_1	$2s^2 2p^3({}^2D^\circ)4d$	$4d'$ ${}^1P^\circ$	1	$1015981+x$	
$\overline{3d}$ 3P_2 3P_1 3P_0	$2s^2 2p^3({}^2D^\circ)3d$	$3d'$ ${}^3P^\circ$	2 1 0	876762 877244 877444	—482 —200	$\overline{4d}$ 3P_2 3P_1	$2s^2 2p^3({}^2D^\circ)4d$	$4d'$ ${}^3P^\circ$	2 1 0	1017590 1017972	—382
$\overline{3d}$ 1D_2	$2s^2 2p^3({}^2D^\circ)3d$	$3d'$ ${}^1D^\circ$	2	$878028+x$		$\overline{4d}$ 1D_2	$2s^2 2p^3({}^2D^\circ)4d$	$4d'$ ${}^1D^\circ$	2	$1018840+x$	
$\overline{3d}$ 3S_1	$2s^2 2p^3({}^2D^\circ)3d$	$3d'$ ${}^3S^\circ$	1	879485		$\overline{4d}$ 1F_3	$2s^2 2p^3({}^2D^\circ)4d$	$4d'$ ${}^1F^\circ$	3	$1019913+x$	
$\overline{3d}$ 1F_3	$2s^2 2p^3({}^2D^\circ)3d$	$3d'$ ${}^1F^\circ$	3	$883210+x$		$\overline{3s'}$ 3D_1 3D_2 3D_3	$2s 2p^4({}^2D)3s$	$3s^{IV}$ 3D	1 2 3	1020311 1020375 1020468	64 93
$\overline{3d}$ 3P_0 3P_1 3P_2	$2s^2 2p^3({}^2P^\circ)3d$	$3d''$ ${}^3P^\circ$	0 1 2	898673 898904 899291	231 387	$3p'$ 3D	$2s 2p^4({}^4P)3p$	$3p'''$ ${}^3D^\circ$	1, 2, 3	1026283	
$\overline{3d}$ 1D_2	$2s^2 2p^3({}^2P^\circ)3d$	$3d''$ ${}^1D^\circ$	2	$901872+x$		$5d$ 3D	$2s^2 2p^3({}^4S^\circ)5d$	$5d$ ${}^3D^\circ$	1, 2, 3	1026774	
						$\overline{4d}$ 3P_1 3P_2	$2s^2 2p^3({}^2P^\circ)4d$	$4d''$ ${}^3P^\circ$	0 1 2	1042481 1042681	200

Mg v—Continued

Mg v—Continued

Author	Config.	Desig.	<i>J</i>	Level	Inter- val	Author	Config.	Desig.	<i>J</i>	Level	Inter- val
$\overline{4d}$ 3D	$2s^2 2p^3(^2P^\circ)4d$	$4d'' ^3D^\circ$	1, 2, 3	1043818		$\overline{5d}$ 1D_2	$2s^2 2p^2(^2D^\circ)5d$	$5d' ^1D^\circ$	2	1082461 + <i>x</i>	
$\overline{4d}$ 1D_2	$2s^2 2p^3(^2P^\circ)4d$	$4d'' ^1D^\circ$	2	1045766 + <i>x</i>		$\overline{5d}$ 1F_3	$2s^2 2p^3(^2D^\circ)5d$	$5d' ^1F^\circ$	3	1082855 + <i>x</i>	
$\overline{4d}$ 1P_1	$2s^2 2p^3(^2P^\circ)4d$	$4d'' ^1P^\circ$	1	1046201 + <i>x</i>		$\overline{5d}$ 1D_2	$2s^2 2p^3(^2P^\circ)5d$	$5d'' ^1D^\circ$	2	1110358 + <i>x</i>	
$\overline{4d}$ 1F_3	$2s^2 2p^3(^2P^\circ)4d$	$4d'' ^1F^\circ$	3	1046625 + <i>x</i>			Mg vI ($^4S_{1/2}$)	Limit		1139421	
$\overline{5s}$ 3D	$2s^2 2p^3(^2D^\circ)5s$	$5s' ^3D^\circ$	3, 2, 1	1054921		$4s'$ 3P_2	$2s 2p^4(^4P)4s$	$4s''' ^3P$	2 1 0	1161768	
$3d'$ 3D	$2s 2p^4(^4P)3d$	$3d''' ^3D$	1, 2, 3	1075102							
$\overline{5d}$ 3D	$2s^2 2p^3(^2D^\circ)5d$	$5d' ^3D^\circ$	1, 2, 3	1079431		$\overline{3d'}$ 3D_3	$2s 2p^4(^2D)3d$	$3d^{IV} ^3D$	3 2 1	1166471 1166552 1166626	-81 -74
$\overline{5d}$ 3P_2 3P_1	$2s^2 2p^3(^2D^\circ)5d$	$5d' ^3P^\circ$	2 1 0	1081883 1082146	-263	3D_1					
						$5s'$ 3P_2	$2s 2p^4(^4P)5s$	$5s''' ^3P$	2 1 0	1250956	

February 1947.

Mg v OBSERVED TERMS*

Config. $1s^2 +$	Observed Terms		
$2s^2 2p^4$	{ $2p^4 ^1S$ $2p^4 ^3P$ $2p^4 ^1D$		
$2s 2p^5$	{ $2p^5 ^3P^\circ$ $2p^5 ^1P^\circ$		
	$ns (n \geq 3)$	$np (n \geq 3)$	$nd (n \geq 3)$
$2s^2 2p^3(^4S^\circ)nx$	$3-5s ^3S^\circ$		$3-5d ^3D^\circ$
$2s^2 2p^3(^2D^\circ)nx'$	{ $3-5s' ^3D^\circ$ $3, 4s' ^1D^\circ$	$3d' ^3S^\circ$	$3-5d' ^3P^\circ$ $3-5d' ^3D^\circ$ $3-5d' ^1F^\circ$
$2s^2 2p^3(^2P^\circ)nx''$	{ $3, 4s'' ^3P^\circ$ $3, 4s'' ^1P^\circ$		$3, 4d'' ^3P^\circ$ $3, 4d'' ^3D^\circ$ $3, 4d'' ^1F^\circ$
$2s 2p^4(^4P)nx'''$	$3-5s''' ^3P$	$3p''' ^3D^\circ$	$3d''' ^3D$
$2s 2p^4(^2D)nx^{IV}$	$3s^{IV} ^3D$		$3d^{IV} ^3D$

*For predicted terms in the spectra of the O I isoelectronic sequence, see Introduction.

Mg VI

(N I sequence; 7 electrons)

 $Z=12$ Ground state $1s^2 2s^2 2p^3 ^4S_{1/2}$ $2p^3 ^4S_{1/2} 1507520 \text{ cm}^{-1}$

I. P. 186.86 volts

The analysis is by Söderqvist, who has found 56 terms and classified 124 lines in the range 72 Å to 403 Å. No intersystem combinations have been observed. The observations indicate an evident typographical error in the published absolute value of $2p^4 ^2P$, which has been corrected. The series are short and the uncertainty, *x*, may be considerable.

REFERENCE

J. Söderqvist, Ark. Mat. Astr. Fys. (Stockholm) **32A**, No. 19 p. 4 (1946). (I P) (T) (C L)

Mg VI

Mg VI

Author	Config.	Desig.	J	Level	Interval	Author	Config.	Desig.	J	Level	Interval
2p 4S_2	$2s^2 2p^3$	$2p^3 \ ^4S^\circ$	$1\frac{1}{2}$	0		$\overline{3d} \ ^2S_1$	$2s^2 2p^2(^1D)3d$	$3d' \ ^2S$	$\frac{1}{2}$	$1097978+x$	
2p 2D_3 2D_2	$2s^2 2p^3$	$2p^3 \ ^2D^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$	$54150+x$ $54171+x$	-21	$3p' \ ^4P$	$2s \ 2p^3(^3S^\circ)3p$	$3p''' \ ^4P$	$\left\{ \begin{array}{l} \frac{1}{2} \\ \text{to} \\ 2\frac{1}{2} \end{array} \right\}$	1100146	
2p 2P_1 2P_2	$2s^2 2p^3$	$2p^3 \ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	$82710+x$ $82832+x$	122	$\overline{3s'} \ ^4D$	$2s \ 2p^3(^3D^\circ)3s$	$3s^{IV} \ ^4D^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{array} \right\}$	1122023	
2p' 4P_3 4P_2 4P_1	$2s \ 2p^4$	$2p^4 \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	247945 249578 250445	-1633 -867	$\overline{3d} \ ^2D$	$2s^2 2p^2(^1S)3d$	$3d'' \ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	1123683+x	
2p' 2D_3 2D_2	$2s \ 2p^4$	$2p^4 \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	$340551+x$ $340584+x$	-33	$\overline{3s'} \ ^2D$	$2s \ 2p^3(^3D^\circ)3s$	$3s^{IV} \ ^2D^\circ$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	1149638+x	
2p' 2S_1	$2s \ 2p^4$	$2p^4 \ ^2S$	$\frac{1}{2}$	$400619+x$		$\overline{3s'} \ ^4P$	$2s \ 2p^3(^3P^\circ)3s$	$3s^V \ ^4P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ \text{to} \\ 2\frac{1}{2} \end{array} \right\}$	1172608	
2p' 2P_2 2P_1	$2s \ 2p^4$	$2p^4 \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	$423981+x$ $425938+x$	-1957	$3d' \ ^4D$	$2s \ 2p^3(^3S^\circ)3d$	$3d''' \ ^4D^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{array} \right\}$	1175396	
3s 4P_1 4P_2 4P_3	$2s^2 2p^2(^3P)3s$	$3s \ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	893943 894887 896443	944 1556	$\overline{3s'} \ ^2P_1$ 2P_2	$2s \ 2p^3(^3P^\circ)3s$	$3s^V \ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	$1191126+x$ $1191432+x$	306
3s 2P_1 2P_2	$2s^2 2p^2(^3P)3s$	$3s \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	$907202+x$ $909096+x$	1894		$2s^2 2p^2(^3P)4s$	$4s \ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	1196740	
$\overline{3s} \ ^2D$	$2s^2 2p^2(^1D)3s$	$3s' \ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	937628+x		$4s \ ^4P_3$					
$\overline{3s} \ ^2S_1$	$2s^2 2p^2(^1S)3s$	$3s'' \ ^2S$	$\frac{1}{2}$	$982218+x$		$4s \ ^2P_2$	$2s^2 2p^2(^3P)4s$	$4s \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	$1198265+x$	
3d 2P_2 2P_1	$2s^2 2p^2(^3P)3d$	$3d \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	$1038855+x$ $1039472+x$	-617	$3p' \ ^2F_4$ 2F_3	$2s \ 2p^3(^3D^\circ)3p$	$3p^{IV} \ ^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	$1222074+x$ $1222709+x$	-635
	$2s^2 2p^2(^3P)3d$	$3d \ ^4D$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	1045205 1045620	-415	$\overline{4s} \ ^2D$	$2s^2 2p^2(^1D)4s$	$4s' \ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	1234487+x	
3d $^4D_{23}$ 4D_1						$4d \ ^4D_{23}$	$2s^2 2p^2(^3P)4d$	$4d \ ^4D$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	1248829 1249500	-671
3d 2F_3 2F_4	$2s^2 2p^2(^3P)3d$	$3d \ ^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	$1045212+x$ $1047179+x$	1967	$4d \ ^2F_3$ 2F_4	$2s^2 2p^2(^3P)4d$	$4d \ ^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	$1251503+x$ $1253148+x$	1645
3s' 4S_2	$2s \ 2p^3(^3S^\circ)3s$	$3s''' \ ^4S^\circ$	$1\frac{1}{2}$	1046634		$4d \ ^4P_3$ 4P_2 4P_1	$2s^2 2p^2(^3P)4d$	$4d \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	1252238 1252662 1252866	-424 -204
3d 4P_3 4P_2 4P_1	$2s^2 2p^2(^3P)3d$	$3d \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	1047307 1047987 1048383	-680 -396	$4d \ ^2D_3$	$2s^2 2p^2(^3P)4d$	$4d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	$1257189+x$	
3d 2D_2 2D_3	$2s^2 2p^2(^3P)3d$	$3d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	$1060848+x$ $1061411+x$	563	$\overline{3d'} \ ^4P_3$ 4P_2 4P_1	$2s \ 2p^3(^3D^\circ)3d$	$3d^{IV} \ ^4P^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	1282028 1282398 1282668	-370 -270
$\overline{3d} \ ^2F_4$ 2F_3	$2s^2 2p^2(^1D)3d$	$3d' \ ^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	$1082132+x$ $1082438+x$	-306						
$\overline{3d} \ ^2D_2$ 2D_3	$2s^2 2p^2(^1D)3d$	$3d' \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	$1085361+x$ $1085718+x$	357						
$\overline{3d} \ ^2P_1$ 2P_2	$2s^2 2p^2(^1D)3d$	$3d' \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	$1092558+x$ $1093046+x$	488						

Mg VI—Continued

Mg VI—Continued

Author	Config.	Desig.	J	Level	Interval	Author	Config.	Desig.	J	Level	Interval
$\overline{3d}'$ 4D	$2s\ 2p^3(^3D^\circ)3d$	$3d^{IV}\ ^4D^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	1287044	-861	$5d$ $^4D_{23}$	$2s^2\ 2p^2(^3P)5d$	$5d$ 4D	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ 2\frac{1}{2} \\ 1\frac{1}{2} \\ \frac{1}{2} \end{smallmatrix} \right\}$	1342985	1746
$\overline{4d}$ 2F	$2s^2\ 2p^2(^1D)4d$	$4d' \ ^2F$	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	$1287104+x$		$5d$ 2F_3 2F_4	$2s^2\ 2p^2(^3P)5d$	$5d$ 2F	$2\frac{1}{2}$ $3\frac{1}{2}$	$1344310+x$ $1346056+x$	
$\overline{3d}'$ 4S_2	$2s\ 2p^3(^3D^\circ)3d$	$3d^{IV}\ ^4S^\circ$	$1\frac{1}{2}$	1287889		$5d$ 4P_3	$2s^2\ 2p^2(^3P)5d$	$5d$ 4P	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	1345550	
$\overline{3d}'$ 2F_4 2F_3	$2s\ 2p^3(^3D^\circ)3d$	$3d^{IV}\ ^2F^\circ$	$3\frac{1}{2}$ $2\frac{1}{2}$	$1288400+x$ $1289261+x$		$4d'$ 4D	$2s\ 2p^3(^5S^\circ)4d$	$4d''' \ ^4D^\circ$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	1373760	
$\overline{4d}$ 2D	$2s2p^2(^1D)4d$	$4d' \ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	$1289787+x$	973	$6s$ 4P_3	$2s^2\ 2p^2(^3P)6s$	$6s$ 4P	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	1380643	
$\overline{4d}$ 2P	$2s^2\ 2p^2(^1D)4d$	$4d' \ ^2P$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	$1292939+x$		$\overline{5d}$ 2F	$2s^2\ 2p^2(^1D)5d$	$5d' \ ^2F$	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	$1381572+x$	
$\overline{4d}$ 2S_1	$2s^2\ 2p^2(^1D)4d$	$4d' \ ^2S$	$\frac{1}{2}$	$1295321+x$		$\overline{5d}$ 2D	$2s^2\ 2p^2(^1D)5d$	$5d' \ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	$1383088+x$	
$5s$ 4P_2 4P_3	$2s^2\ 2p^2(^3P)5s$	$5s$ 4P	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	1317697 1318670		$5d'$ 4D	$2s\ 2p^3(^5S^\circ)5d$	$5d''' \ ^4D^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	1463928	
$4s'$ 4S_2	$2s\ 2p^3(^5S^\circ)4s$	$4s''' \ ^4S^\circ$	$1\frac{1}{2}$	1323609							
$\overline{4d}$ 2D	$2s^2\ 2p^2(^1S)4d$	$4d'' \ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	$1332285+x$							
$4p'$ 4P	$2s\ 2p^3(^5S^\circ)4p$	$4p''' \ ^4P$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	1340950							
							Mg VII (3P_0)	Limit	-----	1507520	

February 1947.

Mg VI OBSERVED TERMS*

Config. $1s^2+$	Observed Terms			
$2s^2\ 2p^3$	$\left\{ \begin{array}{lll} 2p^3\ ^4S^\circ & 2p^3\ ^2P^\circ & 2p^3\ ^2D^\circ \end{array} \right.$			
$2s\ 2p^4$	$\left\{ \begin{array}{lll} 2p^4\ ^2S & 2p^4\ ^4P & 2p^4\ ^2D \\ & 2p^4\ ^2P & \end{array} \right.$			
	$ns\ (n \geq 3)$	$np\ (n \geq 3)$	$nd\ (n \geq 3)$	
$2s^2\ 2p^2(^3P)nx$	$\left\{ \begin{array}{l} 3-6s\ ^4P \\ 3, 4s\ ^2P \end{array} \right.$		$3-5d\ ^4P$ $3d\ ^2P$	$3-5d\ ^4D$ $3, 4d\ ^2D$ $3-5d\ ^2F$
$2s^2\ 2p^2(^1D)nx'$		$3, 4s'\ ^2D$	$3, 4d'\ ^2S$	$3, 4d'\ ^2P$ $3-5d'\ ^2D$ $3-5d'\ ^2F$
$2s^2\ 2p^2(^1S)nx''$	$3s''\ ^2S$			$3, 4d''\ ^2D$
$2s\ 2p^3(^5S^\circ)nx'''$	$3, 4s''' \ ^4S^\circ$	$3, 4p''' \ ^4P$		$3-5d''' \ ^4D^\circ$
$2s\ 2p^3(^3D^\circ)nx^{IV}$	$\left\{ \begin{array}{l} 3s^{IV}\ ^4D^\circ \\ 3s^{IV}\ ^2D^\circ \end{array} \right.$	$3p^{IV}\ ^2F$	$3d^{IV}\ ^4S^\circ$	$3d^{IV}\ ^4P^\circ$ $3d^{IV}\ ^4D^\circ$ $3d^{IV}\ ^2F^\circ$
$2s\ 2p^3(^3P^\circ)nx^V$	$\left\{ \begin{array}{l} 3s^V\ ^4P^\circ \\ 3s^V\ ^2P^\circ \end{array} \right.$			

*For predicted terms in the spectra of the NI isoelectronic sequence, see Introduction.

Mg VII

(C I sequence; 6 electrons)

 $Z=12$ Ground state $1s^2 2s^2 2p^2 {}^3P_0$ $2p^2 {}^3P_0$ 1817734 cm^{-1}

I. P. 225.31 volts

Söderqvist has found 56 terms and classified 114 lines in this spectrum in the range 58 Å to 434 Å. He determines the relative values of the singlet, triplet, and quintet systems of terms from the series limits.

Söderqvist gives the quintet term $2p^3 {}^5S_2$ at 118134 cm^{-1} above the ground state zero. From isoelectronic sequence data Robinson estimates this value as 118620 cm^{-1} . The later value has been used in the table and all quintet terms adjusted accordingly.

The uncertainties x and y may be considerable.

REFERENCES

- J. Söderqvist, Ark. Mat. Astr. Fys. (Stockholm) **32A**, No. 19 p. 4 (1946). (I P) (T) (C L)
H. A. Robinson, unpublished material (March 1948). (T)

Mg VII

Mg VII

Author	Config.	Desig.	J	Level	Interval	Author	Config.	Desig.	J	Level	Interval
$2p$ 3P_0 3P_1 3P_2	$2s^2 2p^2$	$2p^2$ 3P	0 1 2	0 1127 2939	1127 1812	$3d$ 3D_1 3D_2 3D_3	$2s^2 2p({}^2P^\circ)3d$	$3d$ ${}^3D^\circ$	1 2 3	1191753 1192185 1193061	432 876
$2p$ 1D_2	$2s^2 2p^2$	$2p^2$ 1D	2	$41459+x$		$3d$ 3P_2 3P_1 3P_0	$2s^2 2p({}^2P^\circ)3d$	$3d$ ${}^3P^\circ$	2 1 0	1196770 1197469 1197872	-699 -403
$2p'$ 5S_2	$2s 2p^3$	$2p^3$ ${}^5S^\circ$	2	$118620+y$		$3s'$ 3P_0 3P_1 3P_2	$2s 2p^2({}^4P)3s$	$3s$ 3P	0 1 2	1211173 1212055 1213679	882 1624
$2p'$ 3D_3 3D_2 3D_1	$2s 2p^3$	$2p^3$ ${}^3D^\circ$	3 2 1	232865 232975 233027	-110 -52	$3d$ 1F_3	$2s^2 2p({}^2P^\circ)3d$	$3d$ ${}^1F^\circ$	3	$1212323+x$	
$2p'$ 3P	$2s 2p^3$	$2p^3$ ${}^3P^\circ$	2, 1, 0	274922		$3d$ 1P_1	$2s^2 2p({}^2P^\circ)3d$	$3d$ ${}^1P^\circ$	1	$1213297+x$	
$2p'$ 1D_2	$2s 2p^3$	$2p^3$ ${}^1D^\circ$	2	$354923+x$		$3p'$ 3S_1	$2s 2p^2({}^4P)3p$	$3p$ ${}^3S^\circ$	1	1235329	
$2p'$ 3S_1	$2s 2p^3$	$2p^3$ ${}^3S^\circ$	1	362128		$3p'$ 3D_2 3D_3	$2s 2p^2({}^4P)3p$	$3p$ ${}^3D^\circ$	1 2 3	1264827 1266076	1249
$2p'$ 1P_1	$2s 2p^3$	$2p^3$ ${}^1P^\circ$	1	$397655+x$		$3p'$ 3P	$2s 2p^2({}^4P)3p$	$3p$ ${}^3P^\circ$	0, 1, 2	1276520	
$3s$ 3P_0 3P_1 3P_2	$2s^2 2p({}^2P^\circ)3s$	$3s$ ${}^3P^\circ$	0 1 2	1047624 1048385 1050906	761 2521	$\overline{3s'}$ 3D	$2s 2p^2({}^2D)3s$	$3s'$ 3D	1, 2, 3	1285196	
$3s$ 1P_1	$2s^2 2p({}^2P^\circ)3s$	$3s$ ${}^1P^\circ$	1	$1061534+x$		$\overline{3p'}$ 3D	$2s 2p^2({}^2D)3p$	$3p'$ ${}^3D^\circ$	1, 2, 3	1299244	
$3p$ 3P_0 3P_1 3P_2	$2s^2 2p({}^2P^\circ)3p$	$3p$ 3P	0 1 2	1123745 1124937 1125850	1192 913	$\overline{3s'}$ 1D_2	$2s 2p^2({}^2D)3s$	$3s'$ 1D	2	$1305806+x$	
$3d$ 3F_2	$2s^2 2p({}^2P^\circ)3d$	$3d$ ${}^3F^\circ$	2 3 4	$1178758+x$		$2s 2p^2({}^4P)3d$	$3d$ 5D		0 1 2 3 4	} $1317618+y$	
$3s'$ 5P_1 5P_2 5P_3	$2s 2p^2({}^4P)3s$	$3s$ 5P	1 2 3	$1179696+y$ $1180484+y$ $1181963+y$	788 1479	$3d'$ ${}^5D_{23}$					
						$3d'$ 5P_3 5P_2 5P_1	$2s 2p^2({}^4P)3d$	$3d$ 5P	3 2 1	$1323222+y$ $1323889+y$ $1324311+y$	-667 -422
$3d$ 1D_2	$2s^2 2p({}^2P^\circ)3d$	$3d$ ${}^1D^\circ$	2	$1181424+x$							

Mg VII—Continued

Mg VII—Continued

Author	Config.	Desig.	<i>J</i>	Level	Interval	Author	Config.	Desig.	<i>J</i>	Level	Interval
$3d'$ 3P_2 3P_1 3P_0	$2s\ 2p^2(^4P)3d$	$3d\ ^3P$	2 1 0	1324975 1326033 1326568	-1058 -535	$4s'$ 5P_3	$2s\ 2p^2(^4P)4s$	$4s\ ^5P$	1 2 3	1549235+y	
$3d'$ 3F_2 3F_3 3F_4	$2s\ 2p^2(^4P)3d$	$3d\ ^3F$	2 3 4	1333173 1334115 1335328	942 1213	$4p'$ 3D_3	$2s\ 2p^2(^4P)4p$	$4p\ ^3D^\circ$	1 2 3	1579211	
$\overline{3p'}$ 1F_3	$2s\ 2p^2(^2D)3p$	$3p'\ ^1F^\circ$	3	1350497+x		$5d$ 3P	$2s^2\ 2p(^2P^\circ)5d$	$5d\ ^3P^\circ$	0, 1, 2	1597937	
$3d'$ 3D_1 3D_2 3D_3	$2s\ 2p^2(^4P)3d$	$3d\ ^3D$	1 2 3	1350626 1350948 1351359	322 411	$4d'$ 5P_3 5P_2 5P_1	$2s\ 2p^2(^4P)4d$	$4d\ ^5P$	3 2 1	1600167+y 1600760+y 1601134+y	-593 -374
$\overline{3p'}$ 1D_2	$2s\ 2p^2(^2D)3p$	$3p'\ ^1D^\circ$	2	1357681+x		$5d$ 1F_3	$2s^2\ 2p(^2P^\circ)5d$	$5d\ ^1F^\circ$	3	1600986+x	
$\overline{3d'}$ 3F	$2s\ 2p^2(^2D)3d$	$3d'\ ^3F$	2, 3, 4	1414307		$4d'$ 3F_2 3F_3 3F_4	$2s\ 2p^2(^4P)4d$	$4d\ ^3F$	2 3 4	1604844 1605621 1606747	777 1126
$\overline{3d'}$ 3P	$2s\ 2p^2(^2D)3d$	$3d'\ ^3P$	0, 1, 2	1420669		$6d$ 3P	$2s^2\ 2p(^2P^\circ)6d$	$6d\ ^3P^\circ$	0, 1, 2	1665781	
$\overline{3d'}$ 3D_2 3D_3	$2s\ 2p^2(^2D)3d$	$3d'\ ^3D$	1 2 3	1422040 1422614	574	$\overline{4d'}$ 3F	$2s\ 2p^2(^2D)4d$	$4d'\ ^3F$	2, 3, 4	1695880	
$\overline{3d'}$ 3S_1	$2s\ 2p^2(^2D)3d$	$3d'\ ^3S$	1	1435724		$2s\ 2p^2(^4P)5p$	$5p\ ^3D^\circ$	1 2 3		1717734	
$\overline{3d'}$ 1F_3	$2s\ 2p^2(^2D)3d$	$3d'\ ^1F$	3	1438863+x		$5p'$ 3D_3					
$\overline{3d'}$ 1D_2	$2s\ 2p^2(^2D)3d$	$3d'\ ^1D$	2	1439116+x		$2s\ 2p^2(^4P)5d$	$5d\ ^5P$	1 2 3		1727216+y	
$4d$ 1D_2	$2s^2\ 2p(^2P^\circ)4d$	$4d\ ^1D^\circ$	2	1466102+x		$5d'$ 5P_3					
$4d$ 3D_2 3D_3	$2s^2\ 2p(^2P^\circ)4d$	$4d\ ^3D^\circ$	1 2 3	1469556 1470420	864	$5d'$ 3F_4	$2s\ 2p^2(^4P)5d$	$5d\ ^3F$	2 3 4	1730140	
$4d$ 3P_2	$2s^2\ 2p(^2P^\circ)4d$	$4d\ ^3P^\circ$	0 1 2	1472144		$6d'$ 5P_3	$2s\ 2p^2(^4P)6d$	$6d\ ^5P$	1 2 3	1795347+y	
$4d$ 1F_3	$2s^2\ 2p(^2P^\circ)4d$	$4d\ ^1F^\circ$	3	1477931+x							
$4d$ 1P_1	$2s^2\ 2p(^2P^\circ)4d$	$4d\ ^1P^\circ$	1	1478676+x							
							Mg VIII ($^2P_{3/2}^\circ$)	Limit	-----	1817734	

March 1948.

Mg VII OBSERVED TERMS*

Config. $1s^2+$	Observed Terms		
$2s^2\ 2p^2$	$\{ 2p^2\ ^1S\quad 2p^2\ ^3P\quad 2p^2\ ^1D$		
$2s\ 2p^3$	$\{ 2p^3\ ^5S^\circ\quad 2p^3\ ^3S^\circ\quad 2p^3\ ^3P^\circ\quad 2p^3\ ^3D^\circ\quad 2p^3\ ^1P^\circ\quad 2p^3\ ^1D^\circ$		
	$ns\ (n \geq 3)$	$np\ (n \geq 3)$	$nd\ (n \geq 3)$
$2s^2\ 2p(^2P^\circ)nx$	$\{ 3s\ ^3P^\circ\quad 3s\ ^1P^\circ$	$3p\ ^3P$	$3-6d\ ^3P^\circ\quad 3, 4d\ ^3D^\circ\quad 3d\ ^3F^\circ\quad 3, 4d\ ^1P^\circ\quad 3, 4d\ ^1D^\circ\quad 3-5d\ ^1F^\circ$
$2s\ 2p^2(^4P)nx$	$\{ 3, 4s\ ^5P\quad 3s\ ^3P$	$3p\ ^3S^\circ\quad 3p\ ^3P^\circ\quad 3-5p\ ^3D^\circ$	$3-6d\ ^5P\quad 3d\ ^5D\quad 3d\ ^5F\quad 3d\ ^3P\quad 3d\ ^3D\quad 3-5d\ ^3F$
$2s\ 2p^2(^2D)nx'$	$\{ 3s'\ ^3D\quad 3s'\ ^1D$	$3p'\ ^3D^\circ\quad 3p'\ ^1D^\circ\quad 3p'\ ^1F^\circ$	$3d'\ ^3S\quad 3d'\ ^3P\quad 3d'\ ^3D\quad 3, 4d'\ ^3F\quad 3d'\ ^1D\quad 3d'\ ^1F$

*For predicted terms in the spectra of the CI isoelectronic sequence, see Introduction.

Mg VIII

(B I sequence; 5 electrons)

 $Z=12$ Ground state $1s^2 2s^2 2p^2 P^\circ_{1/2}$ $2p^2 P^\circ_{1/2}$ 2145679 cm^{-1}

I. P. 265.957 volts

The analysis is by Söderqvist, who has classified 118 lines, all but 9 of which lie between 52 Å and 97 Å. He remarks that the term values of $2p^3 {}^2P^\circ$ and $2p^3 {}^2D^\circ$ need further confirmation, since no combination of these terms with the doublets of the $2p^2$ configuration have been observed. These two terms and those calculated from combinations with them may require a slight adjustment but they are not seriously in error, as compared with the errors of measurement. Apparently the values extrapolated from the law of irregular doublets and those obtained from observed combinations confirm the terms fairly well.

The absolute values of the doublet terms are well determined from the $nd {}^2D$ series and $nd {}^2F^\circ$ series, both of which extend to $n=5$.

The absolute values of the quartet terms are obtained from the $nd {}^4D^\circ$ series ($n=3, 4, 5$). No intersystem combinations have been observed, and a small correction x may be needed to connect the doublet and quartet terms.

REFERENCE

J. Söderqvist, Ark. Mat. Astr. Fys. (Stockholm) **30A**, No. 11, p. 13 (1944). (I P) (T) (C L)

Mg VIII

Mg VIII

Author	Config.	Desig.	J	Level	Interval	Author	Config.	Desig.	J	Level	Interval
$2p^2 P_1$ $2p^2 P_2$	$2s^2 ({}^1S) 2p$	$2p^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	0 3304	3304	$3p'$ 2S_1	$2s 2p ({}^3P^\circ) 3p$	$3p^2 S$	$\frac{1}{2}$	1460911	
$2p'$ 4P_1 4P_2 4P_3	$2s 2p^2$	$2p^2 {}^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$130598+x$ $131763+x$ $133481+x$	1165 1718	$3d'$ 4D_2 4D_3 4D_4	$2s 2p ({}^3P^\circ) 3d$	$3d^4 D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	$1476964+x$ $1477341+x$ $1478182+x$	377 841
$2p'$ 2D_3 2D_2	$2s 2p^2$	$2p^2 {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	232281 232304	-23	$3d'$ 2D_2 2D_3	$2s 2p ({}^3P^\circ) 3d$	$3d^2 D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	1478358 1478706	348
$2p'$ 2S_1	$2s 2p^2$	$2p^2 {}^2S$	$\frac{1}{2}$	298283		$3d'$ 4P_3 4P_2 4P_1	$2s 2p ({}^3P^\circ) 3d$	$3d^4 P^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	$1484449+x$ $1485153+x$ $1485639+x$	-704 -486
$2p'$ 2P_1 2P_2	$2s 2p^2$	$2p^2 {}^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	318747 320742	1995	$\overline{3s'}$ 2P	$2s 2p ({}^1P^\circ) 3s$	$3s' {}^2P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	1486995	
$2p''$ 4S_2	$2p^3$	$2p^3 {}^4S^\circ$	$1\frac{1}{2}$	$414380+x$		$3d'$ 2F_3 2F_4	$2s 2p ({}^3P^\circ) 3d$	$3d^2 F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	1504992 1507043	2051
$2p''$ 2D_3 2D_2	$2p^3$	$2p^3 {}^2D^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$	465598 465738	-140	$3d'$ 2P_2 2P_1	$2s 2p ({}^3P^\circ) 3d$	$3d^2 P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	1513099 1514266	-1167
$2p''$ 2P_1 2P_2	$2p^3$	$2p^3 {}^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	524339 524486	147	$\overline{3p'}$ 2D_2 2D_3	$2s 2p ({}^1P^\circ) 3p$	$3p' {}^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	1548027 1548851	824
$3s^2 S_1$	$2s^2 ({}^1S) 3s$	$3s^2 S$	$\frac{1}{2}$	1210689		$\overline{3p'}$ 2P_1 2P_2	$2s 2p ({}^1P^\circ) 3p$	$3p' {}^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	1549955 1550564	609
$3d^2 D_2$ 2D_3	$2s^2 ({}^1S) 3d$	$3d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	1335863 1336033	170	$\overline{3p'}$ 2S_1	$2s 2p ({}^1P^\circ) 3p$	$3p' {}^2S$	$\frac{1}{2}$	1556517	
$3s'$ 4P_1 4P_2 4P_3	$2s 2p ({}^3P^\circ) 3s$	$3s^4 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$1352123+x$ $1353279+x$ $1355296+x$	1156 2017	$3s''$ 4P_1 4P_2 4P_3	$2p^2 ({}^3P) 3s$	$3s'' {}^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$1588737+x$ $1589965+x$ $1591973+x$	1228 2008
$3s'$ 2P_1 2P_2	$2s 2p ({}^3P^\circ) 3s$	$3s^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	1381466 1383731	2265	$\overline{3d'}$ 2F	$2s 2p ({}^1P^\circ) 3d$	$3d' {}^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	1597469	
$3p'$ 2P_1 2P_2	$2s 2p ({}^3P^\circ) 3p$	$3p^2 P$	$\frac{1}{2}$ $1\frac{1}{2}$	1408371 1409401	1030	$\overline{3d'}$ 2D_2 2D_3	$2s 2p ({}^1P^\circ) 3d$	$3d' {}^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	1607872 1608224	
$3p'$ 2D_2 2D_3	$2s 2p ({}^3P^\circ) 3p$	$3p^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	1440561 1442836	2275						

Mg VIII—Continued

Mg VIII—Continued

Author	Config.	Desig.	<i>J</i>	Level	Interval	Author	Config.	Desig.	<i>J</i>	Level	Interval
$\overline{3d'}$ 2P	$2s\ 2p(^1P^\circ)3d$	$3d' \ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	1610669			$2s\ 2p(^3P^\circ)4s$	$4s \ ^4P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	$1769549+x$	
$\overline{3s''}$ 2D	$2p^2(^1D)3s$	$3s''' \ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	1638646		$4s' \ ^4P_3$					
	$2p^2(^3P)3p$	$3p'' \ ^4D^\circ$	$\frac{1}{2}$			$4p' \ ^2P_2$	$2s\ 2p(^3P^\circ)4p$	$4p \ ^2P$	$\frac{1}{2}$	1814176	
$3p'' \ ^4D_4$			$\frac{1}{2}$			$4p' \ ^2D_3$	$2s\ 2p(^3P^\circ)4p$	$4p \ ^2D$	$\frac{1}{2}$	1825262	
$4s \ ^2S_1$	$2s^2(^1S)4s$	$4s \ ^2S$	$\frac{1}{2}$	1647879		$4d' \ ^2D$	$2s\ 2p(^3P^\circ)4d$	$4d \ ^2D^\circ$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	1837649	
	$2p^2(^3P)3p$	$3p'' \ ^4P^\circ$	$\frac{1}{2}$			$4d' \ ^4D$	$2s\ 2p(^3P^\circ)4d$	$4d \ ^4D^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	$1838017+x$	
$3p'' \ ^4P_3$			$\frac{1}{2}$	1653061+x		$4d' \ ^4P$	$2s\ 2p(^3P^\circ)4d$	$4d \ ^4P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	$1840084+x$	
$3p'' \ ^4S_2$	$2p^2(^3P)3p$	$3p'' \ ^4S^\circ$	$1\frac{1}{2}$	1674774+x		$4d' \ ^2F_3$	$2s\ 2p(^3P^\circ)4d$	$4d \ ^2F^\circ$	$\frac{2}{2}$	1846146	1879
$\overline{3p''}$ 2F	$2p^2(^1D)3p$	$3p''' \ ^2F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	1691070		$2F_4$			$\frac{3}{2}$	1848025	
$4d \ ^2D_2$	$2s^2(^1S)4d$	$4d \ ^2D$	$1\frac{1}{2}$	1693824	11	$5d \ ^2D_2$	$2s^2(^1S)5d$	$5d \ ^2D$	$1\frac{1}{2}$	1858322	97
$2D_3$			$2\frac{1}{2}$	1693835		$2D_3$			$2\frac{1}{2}$	1858419	
	$2p^2(^3P)3d$	$3d'' \ ^2F$	$2\frac{1}{2}$			$\overline{4d'}$ 2F	$2s\ 2p(^1P^\circ)4d$	$4d' \ ^2F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	$1964303?$	
$3d'' \ ^2F_4$			$3\frac{1}{2}$	1701860		$\overline{4d'}$ 2D	$2s\ 2p(^1P^\circ)4d$	$4d' \ ^2D^\circ$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	$1968694?$	
$3d'' \ ^2D$	$2p^2(^3P)3d$	$3d'' \ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	1703243?		$5d' \ ^4D$	$2s\ 2p(^3P^\circ)5d$	$5d \ ^4D^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	$2002221+x$	
$\overline{3p''}$ 2D	$2p^2(^1D)3p$	$3p''' \ ^2D^\circ$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	1708860		$5d' \ ^2F_3$	$2s\ 2p(^3P^\circ)5d$	$5d \ ^2F^\circ$	$\frac{2}{2}$	2005261	1391
$3d'' \ ^4P_3$	$2p^2(^3P)3d$	$3d'' \ ^4P$	$2\frac{1}{2}$	1716667+x	-814	$2F_4$			$\frac{3}{2}$	2006652	
$4P_2$			$1\frac{1}{2}$	1717481+x	-442		$2p^2(^3P)4p$	$4p'' \ ^4D^\circ$	$\frac{1}{2}$		
$4P_1$			$\frac{1}{2}$	1717923+x		$4p'' \ ^4D_4$			$1\frac{1}{2}$		
$\overline{3d''}$ 2D	$2p^2(^1D)3d$	$3d''' \ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	1733744					$2\frac{1}{2}$	$2042060+x$	
$\overline{3d''}$ 2F	$2p^2(^1D)3d$	$3d''' \ ^2F$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	1751987					$3\frac{1}{2}$		
$\overline{3d''}$ 2P_1	$2p^2(^1D)3d$	$3d''' \ ^2P$	$\frac{1}{2}$	1754593	965						
$2P_2$			$1\frac{1}{2}$	1755558							
							Mg IX (1S_0)	Limit		2145679	

October 1946.

Mg VIII OBSERVED TERMS*

Config. $1s^2+$	Observed Terms		
$2s^2(^1S)2p$	$2p \ ^2P^\circ$		
$2s \ 2p^2$	$\left\{ \begin{array}{l} 2p^2 \ ^2S \quad 2p^2 \ ^4P \quad 2p^2 \ ^2D \\ 2p^2 \ ^2P \end{array} \right.$		
$2p^3$	$\left\{ \begin{array}{l} 2p^3 \ ^4S^\circ \quad 2p^3 \ ^2P^\circ \quad 2p^3 \ ^2D^\circ \end{array} \right.$		
	$ns \ (n \geq 3)$	$np \ (n \geq 3)$	$nd \ (n \geq 3)$
$2s^2(^1S)nx$	$3, 4s \ ^2S$		$3-5d \ ^2D$
$2s \ 2p(^3P^\circ)nx$	$\left\{ \begin{array}{l} 3, 4s \ ^4P^\circ \\ 3s \ ^2P^\circ \end{array} \right.$	$3p \ ^2S \ 3, 4p \ ^2P \ 3, 4p \ ^2D$	$3, 4d \ ^4P^\circ \ 3-5d \ ^4D^\circ \ 3, 4d \ ^2P^\circ \ 3, 4d \ ^2D^\circ \ 3-5d \ ^2F^\circ$
$2s \ 2p(^1P^\circ)nx'$	$3s' \ ^2P^\circ$	$3p' \ ^2S \ 3p' \ ^2P \ 3p' \ ^2D$	$3d' \ ^2P^\circ \ 3, 4d' \ ^2D^\circ \ 3, 4d' \ ^2F^\circ$
$2p^2(^3P)nx''$	$3s'' \ ^4P$	$3p'' \ ^4S^\circ \ 3p'' \ ^4P^\circ \ 3, 4p'' \ ^4D^\circ$	$3d'' \ ^4P$
$2p^2(^1D)nx'''$			$3d'' \ ^2D \ 3d'' \ ^2F$
	$3s''' \ ^2D$	$3p''' \ ^2D^\circ \ 3p''' \ ^2F^\circ$	$3d''' \ ^2P \ 3d''' \ ^2D \ 3d''' \ ^2F$

*For predicted terms in the spectra of the B I isoelectronic sequence, see Introduction.

Mg IX

(Be I sequence; 4 electrons)

 $Z=12$ Ground state $1s^2 2s^2 {}^1S_0$ $2s^2 {}^1S_0$ 2645444 cm^{-1}

I. P. 327.90 volts

Sixty-five lines have been classified by Söderqvist. All but three lie in the range between 46 Å and 91 Å. No intersystem combinations are known, but the absolute term values are determined from series that are fairly well established. The relative uncertainty, x , is probably a few hundred cm^{-1} .

REFERENCE

J. Söderqvist, Ark. Mat. Astr. Fys. (Stockholm) **30A**, No. 11 p. 8 (1944). (I P) (T) (C L)

Mg IX

Mg IX

Author	Config.	Desig.	J	Level	Interval	Author	Config.	Desig.	J	Level	Interval
$2s$ 1S_0	$2s^2$	$2s^2$ 1S	0	0		$3d'$ 3P_2	$2p({}^2P^\circ)3d$	$3d$ ${}^3P^\circ$	2	$1815552+x$	-982 -528
$2p$ 3P_0	$2s({}^2S)2p$	$2p$ ${}^3P^\circ$	0	$140786+x$	1162 2472	3P_1			1	$1816534+x$	
3P_1			1	$141948+x$		3P_0			0	$1817062+x$	
3P_2			2	$144420+x$		$3d'$ 1F_3	$2p({}^2P^\circ)3d$	$3d$ ${}^1F^\circ$	3	1834337	
$2p$ 1P_1	$2s({}^2S)2p$	$2p$ ${}^1P^\circ$	1	271687		$3d'$ 1P_1	$2p({}^2P^\circ)3d$	$3d$ ${}^1P^\circ$	1	1841286	
$2p'$ 3P_0	$2p^2$	$2p^2$ 3P	0	$366194+x$	1299 2157	$4p$ 1P_1	$2s({}^2S)4p$	$4p$ ${}^1P^\circ$	1	2068680	
3P_1			1	$367493+x$		$4d$ 3D_1	$2s({}^2S)4d$	$4d$ 3D	1	$2080274+x$	54 50
3P_2			2	$369650+x$		3D_2			2	$2080328+x$	
$2p'$ 1D_2	$2p^2$	$2p^2$ 1D	2	404744		3D_3			3	$2080378+x$	
$2p'$ 1S_0	$2p^2$	$2p^2$ 1S	0	499444		$4d$ 1D_2	$2s({}^2S)4d$	$4d$ 1D	2	2087888	
$3s$ 3S_1	$2s({}^2S)3s$	$3s$ 3S	1	$1532749+x$			$2p({}^2P^\circ)4p$	$4p$ 3D	1		
$3s$ 1S_0	$2s({}^2S)3s$	$3s$ 1S	0	1558076		$4p'$ 3D_3			2		
$3p$ 1P_1	$2s({}^2S)3p$	$3p$ ${}^1P^\circ$	1	1593600			$2p({}^2P^\circ)4p$	$4p$ 3P	0		
$3d$ 3D_1	$2s({}^2S)3d$	$3d$ 3D	1	$1631321+x$	163 168	$4p'$ 3P_2			1		
3D_2			2	$1631484+x$					2	2235683	
3D_3			3	$1631652+x$		$4d'$ 1D_2	$2p({}^2P^\circ)4d$	$4d$ ${}^1D^\circ$	2	2240853	
$3d$ 1D_2	$2s({}^2S)3d$	$3d$ 1D	2	1654583		$4p'$ 1D_2	$2p({}^2P^\circ)4p$	$4p$ 1D	2	2241083	
$3s'$ 3P_0	$2p({}^2P^\circ)3s$	$3s$ ${}^3P^\circ$	0	$1710478+x$	1094 2533		$2p({}^2P^\circ)4d$	$4d$ ${}^3D^\circ$	1		
3P_1			1	$1711572+x$		$4d'$ 3D_3			2		
3P_2			2	$1714105+x$					3	$2248572+x$	
$3s'$ 1P_1	$2p({}^2P^\circ)3s$	$3s$ ${}^1P^\circ$	1	1742772		$4d'$ 3P_2	$2p({}^2P^\circ)4d$	$4d$ ${}^3P^\circ$	2	$2249773+x$	
$3p'$ 1P_1	$2p({}^2P^\circ)3p$	$3p$ 1P	1	1748116					1		
$3p'$ 3D_1	$2p({}^2P^\circ)3p$	$3p$ 3D	1	$1755785+x$	1018 2500	$4d'$ 1F_3	$2p({}^2P^\circ)4d$	$4d$ ${}^1F^\circ$	3	2256219	
3D_2			2	$1756803+x$		$4d'$ 1P_1	$2p({}^2P^\circ)4d$	$4d$ ${}^1P^\circ$	1	2258119	
3D_3			3	$1759303+x$		$5d$ 3D	$2s({}^2S)5d$	$5d$ 3D	1, 2, 3	$2285243+x$	
$3p'$ 3S_1	$2p({}^2P^\circ)3p$	$3p$ 3S	1	$1770688+x$		$5d$ 1D_2	$2s({}^2S)5d$	$5d$ 1D	2	2288385	
$3p'$ 3P_0	$2p({}^2P^\circ)3p$	$3p$ 3P	0	$1777886+x$	1117 1312	$5d'$ ${}^3D, {}^3P$	$2p({}^2P^\circ)5d$	$5d$ ${}^3P^\circ, {}^3D^\circ$	0 to 3	$2451942+x$	
3P_1			1	$1779003+x$		$5d'$ 1F_3	$2p({}^2P^\circ)5d$	$5d$ ${}^1F^\circ$	3	2454176	
3P_2			2	$1780315+x$							
$3d'$ 1D_2	$2p({}^2P^\circ)3d$	$3d$ ${}^1D^\circ$	2	1789287							
$3p'$ 1D_2	$2p({}^2P^\circ)3p$	$3p$ 1D	2	1795868							
$3d'$ 3D_1	$2p({}^2P^\circ)3d$	$3d$ ${}^3D^\circ$	1	$1807694+x$	493 995						
3D_2			2	$1808187+x$							
3D_3			3	$1809182+x$							
							Mg x (${}^2S_{1/2}$)	Limit		2645444	

Mg IX OBSERVED TERMS*

Config. 1s ² +	Observed Terms					
2s ²	2s ² 1S					
2s(2S)2p	{ 2p 3P° 2p 1P°					
2p ²	{ 2p ² 1S 2p ² 3P 2p ² 1D					
	ns (n ≥ 3)		np (n ≥ 3)		nd (n ≥ 3)	
2s(2S)nx	{ 3s 3S 3s 1S		3, 4p 1P°		3-5d 3D 3-5d 1D	
2p(2P°)nx	{ 3s 3P° 3s 1P°		3p 3S	3, 4p 3P 3p 1P	3, 4p 3D 3, 4p 1D	3-5d 3P° 3-5d 3D° 3, 4d 1P° 3, 4d 1D° 3-5d 1F°

*For predicted terms in the spectra of the Be I isoelectronic sequence, see Introduction.

Mg X

(Li I sequence; 3 electrons)

Z=12

Ground state 1s² 2s 2S_{1/2}

2s 2S_{1/2} 2963810 cm⁻¹ .

I. P. 367.36 volts

The present analysis results from the classification of nine lines in the region 65 Å to 44 Å. The transition 2s 2S—2p 2P° has not been reported. The predicted positions of these lines are at 625 Å and 609 Å.

Some of the relative levels have been connected by a study of the Rydberg denominators in the isoelectronic sequence rather than by the Ritz combination principle.

REFERENCE

J. Söderqvist, Ark. Mat. Astr. Fys. (Stockholm) **30A**, No. 11, p. 3 (1944). (I P) (T) (C L)

Mg x

Author	Config.	Desig.	<i>J</i>	Level	Inter- val
2s ² S ₁	2s	2s ² S	$\frac{1}{2}$	0	4047
2p ² P ₁ ² P ₂	2p	2p ² P°	$\frac{1}{2}$ $1\frac{1}{2}$	159929 163976	
3s ² S ₁	3s	3s ² S	$\frac{1}{2}$	1682648	
3p ² P ₁ ² P ₂	3p	3p ² P°	$\frac{1}{2}$ $1\frac{1}{2}$	1726519 1727832	1313
3d ² D ₂ ² D ₃	3d	3d ² D	$1\frac{1}{2}$ $2\frac{1}{2}$	1743410 1743880	470
4p ² P _{2,1}	4p	4p ² P°	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	2270148	512
4d ² D ₂ ² D ₃	4d	4d ² D	$1\frac{1}{2}$ $2\frac{1}{2}$	2277182 2277694	
Mg XI (¹ S ₀)		Limit	-----	2963810	

May 1946.

Mg XI

(He I sequence; 2 electrons)

Z=12Ground state 1s² ¹S₀1s² ¹S₀ 14209200 ± 2500 cm⁻¹

I. P. 1761.23 ± 0.31 volts

Flemberg has observed the four leading lines in this spectrum; they lie between 7 Å and 9 Å. He has calculated absolute term values on the assumption that the P-terms can be represented by a Ritz formula. The fourth line appeared on only one plate and was not used in the calculation of the limit.

The unit adopted by Flemberg, 10³ cm⁻¹, has here been changed to cm⁻¹.

REFERENCE

H. Flemberg, Ark. Mat. Astr. Fys. (Stockholm) **28A**, No. 18, p. 34 (1942). (I P) (T) (C L)

Mg XI

Config.	Desig.	<i>J</i>	Level
1s ²	1s ² ¹ S	0	0
1s 2p	2p ¹ P°	1	10907300
1s 3p	3p ¹ P°	1	12738400
1s 4p	4p ¹ P°	1	13381100
1s 5p	5p ¹ P°	1	13680600
-----	-----	---	-----
Mg XII (² S _½)	Limit	---	14209200

October 1946.

ALUMINUM

Al I

13 electrons

 $Z=13$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^2 P_{1/2}^\circ$ $3p^2 P_{1/2}^\circ$ 48279.16 cm^{-1}

I. P. 5.984 volts

The earlier analysis has been extended by Paschen and Ritschl, who have derived improved term values and extended the observations in the infrared and ultraviolet.

The terms $3p^2 {}^2P$ and $3p^2 {}^2S$ have been suggested by Bowen and Millikan and by Selwyn, respectively. The only combinations are with $3p^2 P^\circ$.

Paschen discusses the possibility that the term here called $3d {}^2D$ may be $3p^2 {}^2D$, in which case all subsequent members of the 2D series must have n decreased by one unit.

Intersystem combinations connecting the doublet and quartet terms have been observed.

REFERENCES

- A. Fowler, *Report on Series in Line Spectra*, p. 156 (Fleetway Press, London, 1922). (T) (C L)
 I. S. Bowen and R. A. Millikan, *Phys. Rev.* **26**, 160 (1925). (C L)
 E. W. H. Selwyn, *Proc. Phys. Soc. (London)* **41**, Part 4, No. 229, 402 (1929). (C L)
 F. Paschen, *Ann. der Phys.* [5] **12**, 516 (1932). (T) (C L)
 F. Paschen und R. Ritschl, *Ann. der Phys.* [5] **18**, 886 (1933). (I P) (T) (C L)
 W. F. Meggers, *J. Opt. Soc. Am.* **36**, 431 (1946). (Summary hfs)

Al I

Al I

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3s^2({}^1S)3p$	$3p^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	0.00 112.04	112.04	$3s^2({}^1S)4d$	$4d {}^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	38929.42 38933.96	4.54
$3s^2({}^1S)4s$	$4s {}^2S$	$\frac{1}{2}$	25347.69		$3s^2({}^1S)5p$	$5p^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	40271.98 40277.92	5.94
$3s^2 3p^2$	$3p^2 {}^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	29020.32 29066.90 29142.68	46.58 75.78	$3s^2({}^1S)4f$	$4f {}^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	41318.74	
$3s^2({}^1S)3d$	$3d {}^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	32435.45 32436.79	1.34	$3s^2({}^1S)6s$	$6s {}^2S$	$\frac{1}{2}$	42144.84	
$3s^2({}^1S)4p$	$4p^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	32949.84 32965.67	15.83	$3s^2({}^1S)5d$	$5d {}^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	42233.72 42237.71	3.99
$3s^2({}^1S)5s$	$5s {}^2S$	$\frac{1}{2}$	37689.32		$3s^2({}^1S)6p$	$6p^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	43334.95 43337.77	2.82

Al I—Continued

Al I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2(^1S)5f$	$5f \ ^2F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	43831. 08		$3s \ 3p^2$	$3p^2 \ ^2S$	$\frac{1}{2}$	51753. 0?	
$3s^2(^1S)6d$	$6d \ ^2D$	$\begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 44166. 48 \\ 44168. 88 \end{smallmatrix}$	2. 40	$3s \ 3p^2$	$3p^2 \ ^2P$	$\begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 56643. 0? \\ 56727. 3? \end{smallmatrix}$	84. 3
$3s^2(^1S)7s$	$7s \ ^2S$	$\frac{1}{2}$	44273. 16		$3s \ 3p(^3P^\circ)4s$	$4s \ ^4P^\circ$	$\begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 61691. 29 \\ 61747. 38 \\ 61843. 41 \end{smallmatrix}$	$\begin{smallmatrix} 56. 09 \\ 96. 03 \end{smallmatrix}$
$3s^2(^1S)7p$	$7p \ ^2P^\circ$	$\begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 44928. 4 \\ 44930. 4 \end{smallmatrix}$	2. 0	$3s \ 3p(^3P^\circ)3d$	$3d \ ^2D^\circ$	$\begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 67635. 3 \\ 67663. 2 \end{smallmatrix}$	27. 9
$3s^2(^1S)6f$	$6f \ ^2F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	45194. 65		$3s \ 3p(^3P^\circ)3d$	$3d \ ^2P^\circ$	$\begin{smallmatrix} 1\frac{1}{2} \\ \frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 71184. 7? \\ 71260. 7 \end{smallmatrix}$	-76. 0
$3s^2(^1S)7d$	$7d \ ^2D$	$\begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 45344. 16 \\ 45345. 60 \end{smallmatrix}$	1. 44	$3s \ 3p(^3P^\circ)3d$	$3d \ ^4D^\circ$	$\begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \\ 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 71235. 63 \\ 71244. 38 \\ 71260. 78 \\ 71286. 27 \end{smallmatrix}$	$\begin{smallmatrix} 8. 75 \\ 16. 40 \\ 25. 49 \end{smallmatrix}$
$3s^2(^1S)8s$	$8s \ ^2S$	$\frac{1}{2}$	45457. 27		$3s \ 3p(^3P^\circ)3d$	$3d \ ^4P^\circ$	$\begin{smallmatrix} 2\frac{1}{2} \\ 1\frac{1}{2} \\ \frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 72203. 77 \\ 72250. 29 \\ 72277. 68 \end{smallmatrix}$	$\begin{smallmatrix} -46. 52 \\ -27. 39 \end{smallmatrix}$
$3s^2(^1S)7f$	$7f \ ^2F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	46015. 73		$3s \ 3p(^3P^\circ)5s$	$5s \ ^2P^\circ$	$\begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 72979. 0 \\ 73077. 9 \end{smallmatrix}$	98. 9
$3s^2(^1S)8d$	$8d \ ^2D$	$\begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 46093. 9 \\ 46094. 27 \end{smallmatrix}$	0. 4	$3s \ 3p(^3P^\circ)4d$	$4d \ ^2P^\circ$	$\begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 76521. 8 \\ 76553. 7 \end{smallmatrix}$	31. 9
$3s^2(^1S)9s$	$9s \ ^2S$	$\frac{1}{2}$	46184. 5		$3s \ 3p(^3P^\circ)6s$	$6s \ ^2P^\circ$	$\begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 78612. 5 \\ 78710. 5 \end{smallmatrix}$	98. 0
$3s^2(^1S)9d$	$9d \ ^2D$	$\begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 46593. 28 \\ 46593. 83 \end{smallmatrix}$	0. 55	$3s \ 3p(^3P^\circ)5d$	$5d \ ^2F^\circ$	$\begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 80158. 0 \\ 80191. 9 \end{smallmatrix}$	33. 9
$3s^2(^1S)10s$	$10s \ ^2S$	$\frac{1}{2}$	46665. 7						
$3s^2(^1S)10d$	$10d \ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	46942. 3						
$3s^2(^1S)11d$	$11d \ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	47192. 0						
Al II (1S_0)	Limit	-----	48279. 16						

August 1947.

Al I OBSERVED TERMS*

Config. $1s^2 \ 2s^2 \ 2p^6 +$	Observed Terms			
$3s^2(^1S)3p$	$3p \ ^2P^\circ$			
$3s \ 3p^2$	$\left\{ \begin{smallmatrix} 3p^2 \ ^2S? \\ 3p^2 \ ^4P \\ 3p^2 \ ^2P? \end{smallmatrix} \right\}$			
	$ns \ (n \geq 4)$	$np \ (n \geq 4)$	$nd \ (n \geq 3)$	$nf \ (n \geq 4)$
$3s^2(^1S)nx$	$4-10s \ ^2S$	$4-7p \ ^2P^\circ$	$3-11d \ ^2D$	$4-7f \ ^2F^\circ$
$3s \ 3p(^3P^\circ)nx$	$\left\{ \begin{smallmatrix} 4s \ ^4P^\circ \\ 5, 6s \ ^2P^\circ \end{smallmatrix} \right\}$		$\begin{smallmatrix} 3d \ ^4P^\circ & 3d \ ^4D^\circ \\ 3, 4d \ ^2P^\circ & 3d \ ^2D^\circ \end{smallmatrix}$	$5d \ ^2F^\circ$

*For predicted terms in the spectra of the Al I isoelectronic sequence, see Introduction.

Al II

(Mg I sequence; 12 electrons)

 $Z=13$ Ground state $1s^2 2s^2 2p^6 3s^2 {}^1S_0$ $3s^2 {}^1S_0$ 151860.4 \pm 0.5 cm^{-1}

I. P. 18.823 volts

Sawyer and Paschen published a detailed analysis in 1927, from which most of the terms have been taken. Since then some revisions and extensions have been made, especially regarding the terms from the ${}^2P^\circ$ limit in Al III. The spectrum of Al II furnishes an excellent illustration of perturbed series and consequently is discussed in a number of theoretical papers on this subject. For example, Shenstone and Russell remark that one of the two lowest 1D terms should be $3p^2 {}^1D$. In accordance with their suggestions the terms labeled by Sawyer and Paschen $3 {}^1D$, $7 {}^3F$, and $12 {}^1P$ are here designated $3p^2 {}^1D$, $3d {}^3F^\circ$, and $4s {}^1P^\circ$?, respectively. These changes cause a decrease of one unit in the published values of n for all following series members in each of the three series.

In the 1927 paper the higher series members of the 3P and 3D series are assigned the J -values of the leading components (2 and 3, respectively). As the term intervals are known to be small, all three J -values for each term are entered in the table on the assumption that the terms are unresolved.

In 1933 Paschen and Ritschl published the detailed hyperfine structure separations they observed for a number of the components of triplet terms. From this paper the three new H-terms have been taken, and also slightly improved values of the terms $4s {}^1S$, $6s {}^3S$, $8p {}^3P^\circ$, $5f {}^1F^\circ$, and $5g {}^1, {}^3G$. It has been assumed that the singlet and triplet G-terms and also the singlet and triplet H-terms are coincident, since no multiplicities are assigned to them. Van Vleck and Whitelaw give the theoretical explanation of this for the G-terms.

Intersystem combinations connecting the singlet and triplet systems of terms have been observed.

REFERENCES

- R. A. Sawyer und F. Paschen, Ann. der Phys. [4] **84**, 1 (1927). (I P) (T) (C L)
 F. Paschen, Ann. der Phys. [5] **12**, 509 (1932). (T) (C L)
 A. G. Shenstone and H. N. Russell, Phys. Rev. **39**, 427 (1932). (T)
 F. Paschen und R. Ritschl, Ann. der Phys. [5] **18**, 872 (1933). (T) (C L) (hfs)
 J. H. Van Vleck and N. G. Whitelaw, Phys. Rev. **44**, 551 (1933).
 W. F. Meggers, J. Opt. Soc. Am. **36**, 431 (1946). (Summary hfs)

Al II

Al II

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3s^2$	$3s^2 {}^1S$	0	0.0		$3s(2S)4p$	$4p {}^3P^\circ$	0 1 2	105424.3 105438.4 105467.7	14.1 29.3
$3s(2S)3p$	$3p {}^3P^\circ$	0 1 2	37392.0 37453.8 37579.3	61.8 125.5	$3s(2S)4p$	$4p {}^1P^\circ$	1	106918.2	
$3s(2S)3p$	$3p {}^1P^\circ$	1	59849.7		$3s(2S)3d$	$3d {}^1D$	2	110087.5	
$3p^2$	$3p^2 {}^1D$	2	85479.0		$3s(2S)5s$	$5s {}^3S$	1	120089.8	
$3s(2S)4s$	$4s {}^3S$	1	91271.2		$3s(2S)5s$	$5s {}^1S$	0	121365.2	
$3p^2$	$3p^2 {}^3P$	0 1 2	94084.5 94146.8 94267.7	62.3 120.9	$3s(2S)4d$	$4d {}^3D$	3 2 1	121480.3 121480.9 121481.2	-0.6 -0.3
$3s(2S)4s$	$4s {}^1S$	0	95348.2		$3s(2S)4f$	$4f {}^3F^\circ$	2 3 4	123415.9 123418.0 123420.8	2.1 2.8
$3s(2S)3d$	$3d {}^3D$	3 2 1	95546.8 95547.9 95548.8	-1.1 -0.9	$3s(2S)4f$	$4f {}^1F^\circ$	3	123468.1	

Al II—Continued

Al II—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
3s(2S)4d	4d 1D	2	124792. 0		3s(2S)9s	9s 3S	1	144524. 3	
3s(2S)5p	5p 3P°	0 1 2	125700. 5 125706. 2 125719. 0	5. 7 12. 8	3s(2S)8d	8d 3D	3, 2, 1	144638. 9	
3s(2S)5p	5p 1P°	1	125866. 7		3s(2S)9s	9s 1S	0	144641. 9	
3s(2S)6s	6s 3S	1	132213. 2		3s(2S)8d	8d 1D	2	144780. 2	
3s(2S)6s	6s 1S	0	132776. 4		3s(2S)8f	8f 1F°	3	144781. 9	
3s(2S)5d	5d 3D	3 2, 1	132819. 7 132819. 9	-0. 2	3s(2S)9p	9p 1P°	1	144939. 1	
3s(2S)5f	5f 3F°	2 3 4	133435. 0 133440. 4 133447. 3	5. 4 6. 9	3s(2S)8g	8g 3G	3, 4, 5	144964. 7	
3s(2S)5f	5f 1F°	3	133679. 3		3s(2S)8g	8g 1G	4	144964. 7	
3s(2S)5d	5d 1D	2	133914. 1		3s(2S)8h	8h 3H°	4, 5, 6	144990. 0	
3s(2S)5g	5g 3G	3, 4, 5	134181. 2		3s(2S)8h	8h 1H°	5	144990. 0	
3s(2S)5g	5g 1G	4	134181. 2		3s(2S)8f	8f 3F°	2 3 4	145126. 5 145128. 9 145132. 1	2. 4 3. 2
3s(2S)6p	6p 1P°	1	134917. 3		3p(2P°)3d	3d 3D°	1, 2 3	145148 145152	4
3s(2S)6p	6p 3P°	0 1 2	135009. 0 135012. 1 135018. 9	3. 1 6. 8	3s(2S)9p	9p 3P°	0, 1, 2	145185?	
3s(2S)7s	7s 3S	1	138496. 7		3p(2P°)4s	4s 3P°	0 1 2	145773. 9 145832. 6 145959. 4	58. 7 126. 8
3s(2S)6f	6f 3F°	2 3 4	138518. 7 138536. 4 138559. 2	17. 7 22. 8	3s(2S)10s	10s 3S	1	146108. 8	
3s(2S)7s	7s 1S	0	138799. 3		3s(2S)9d	9d 3D	3, 2, 1	146185. 0	
3s(2S)6d	6d 3D	3, 2, 1	138811. 9		3s(2S)10s	10s 1S	0	146190. 1	
3s(2S)6f	6f 1F°	3	139242. 9		3s(2S)9d	9d 1D	2	146274. 4	
3s(2S)6d	6d 1D	2	139286. 8		3s(2S)9f	9f 1F°	3	146276. 5	
3s(2S)6g	6g 3G	3, 4, 5	139588. 7		3s(2S)10p	10p 1P°	1	146297. 5	
3s(2S)6g	6g 1G	4	139588. 7		3s(2S)9g	9g 3G	3, 4, 5	146414. 5	
3s(2S)7p	7p 1P°	1	139916. 7		3s(2S)9g	9g 1G	4	146414. 5	
3s(2S)7p	7p 3P°	0, 1, 2	140091. 2		3s(2S)9h	9h 3H°	4, 5, 6	146432. 8	
3p(2P°)3d	3d 3F°	2 3 4	141082. 4 141107. 5 141140. 5	25. 1 33. 0	3s(2S)9h	9h 1H°	5	146432. 8	
3s(2S)8s	8s 3S	1	142179. 8		3s(2S)9f	9f 3F°	2 3 4	146496. 7 146497. 8 146499. 2	1. 1 1. 4
3s(2S)8s	8s 1S	0	142360. 8		3s(2S)10p	10p 3P°	0, 1, 2	146577?	
3s(2S)7d	7d 3D	3, 2, 1	142362. 8		3p(2P°)3d	3d 3P°	0 1 2	146595. 0? 146596. 9 146599. 3	1. 9 2. 4
3s(2S)7f	7f 1F°	3	142601. 6		3s(2S)11s	11s 3S	1	147229. 0	
3s(2S)7d	7d 1D	2	142607. 0		3s(2S)11p	11p 1P°	1	147268. 8	
3s(2S)7g	7g 3G	3, 4, 5	142849. 2		3s(2S)10d	10d 3D	3, 2, 1	147282. 8	
3s(2S)7g	7g 1G	4	142849. 2		3s(2S)11s	11s 1S	0	147288. 8	
3s(2S)8p	8p 1P°	1	142958. 9		3s(2S)10d	10d 1D	2	147343. 2	
3s(2S)8p	8p 3P°	0, 1, 2	143170. 0		3s(2S)10f	10f 1F°	3	147344. 2	
3s(2S)7f	7f 3F°	2 3 4	143262. 7 143269. 8 143280. 6	7. 1 10. 8	3s(2S)10g	10g 3G	3, 4, 5	147451. 0	
					3s(2S)10g	10g 1G	4	147451. 0	
					3s(2S)10h	10h 3H°	4, 5, 6	147464. 7	

Al II—Continued

Al II—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
3s(2S)10 <i>h</i>	10 <i>h</i> 1H°	5	147464. 7	0. 4 0. 6	3s(2S)13 <i>f</i>	13 <i>f</i> 1F°	3	149199. 2	
3s(2S)10 <i>f</i>	10 <i>f</i> 3F°	2	147499. 8		3s(2S)13 <i>g</i>	13 <i>g</i> 3G	3, 4, 5	149252. 9	
		3	147500. 2		3s(2S)13 <i>g</i>	13 <i>g</i> 1G	4	149252. 9	
		4	147500. 8		3s(2S)13 <i>f</i>	13 <i>f</i> 3F°	2, 3, 4	149269. 5	
3s(2S)11 <i>p</i>	11 <i>p</i> 3P°	0, 1, 2	147572?	0. 4 0. 5	3s(2S)14 <i>p</i>	14 <i>p</i> 1P°	1	149434. 8	
3 <i>p</i> (2P°)4 <i>s</i> ?	4 <i>s</i> 1P°	1	148002. 0		3s(2S)15 <i>s</i>	15 <i>s</i> 1S	0	149554. 7	
3s(2S)12 <i>s</i>	12 <i>s</i> 3S	1	148052. 5		3s(2S)14 <i>f</i>	14 <i>f</i> 1F°	3	149568. 6	
3s(2S)11 <i>d</i>	11 <i>d</i> 3D	3, 2, 1	148090. 0		3s(2S)14 <i>f</i>	14 <i>f</i> 3F°	2, 3, 4	149625. 5	
3s(2S)12 <i>s</i>	12 <i>s</i> 1S	0	148097. 1		3s(2S)15 <i>p</i>	15 <i>p</i> 1P°	1	149748. 0	
3s(2S)11 <i>f</i>	11 <i>f</i> 1F°	3	148132. 6		3s(2S)16 <i>s</i>	16 <i>s</i> 1S	0	149856. 6	
3s(2S)11 <i>d</i>	11 <i>d</i> 1D	2	148132. 7		3s(2S)15 <i>f</i>	15 <i>f</i> 1F°	3	149866. 2	
3s(2S)11 <i>g</i>	11 <i>g</i> 3G	3, 4, 5	148217. 6		3s(2S)15 <i>f</i>	15 <i>f</i> 3F°	2, 3, 4	149913. 2	
3s(2S)11 <i>g</i>	11 <i>g</i> 1G	4	148217. 6		3s(2S)16 <i>p</i>	16 <i>p</i> 1P°	1	150007. 6	
3s(2S)11 <i>f</i>	11 <i>f</i> 3F°	2	148248. 7		3s(2S)16 <i>f</i>	16 <i>f</i> 1F°	3	150109. 7	
		3	148249. 1		3s(2S)16 <i>f</i>	16 <i>f</i> 3F°	2, 3, 4	150148. 4	
		4	148249. 6		3s(2S)17 <i>f</i>	17 <i>f</i> 1F°	3	150311. 1	
3s(2S)12 <i>p</i>	12 <i>p</i> 1P°	1	148579. 4		3s(2S)17 <i>f</i>	17 <i>f</i> 3F°	2, 3, 4	150343. 5	
3s(2S)13 <i>s</i>	13 <i>s</i> 3S	1	148673. 7		3s(2S)18 <i>f</i>	18 <i>f</i> 1F°	3	150479. 7	
3s(2S)13 <i>s</i>	13 <i>s</i> 1S	0	148706. 9		3s(2S)19 <i>f</i>	19 <i>f</i> 1F°	3	150622. 2	
3s(2S)12 <i>f</i>	12 <i>f</i> 1F°	3	148731. 6		3s(2S)20 <i>f</i>	20 <i>f</i> 1F°	3	150744. 1	
3s(2S)12 <i>g</i>	12 <i>g</i> 3G	3, 4, 5	148800. 4						
3s(2S)12 <i>g</i>	12 <i>g</i> 1G	4	148800. 4						
3s(2S)12 <i>f</i>	12 <i>f</i> 3F°	2, 3, 4	148822. 5						
3s(2S)13 <i>p</i>	13 <i>p</i> 1P°	1	149051. 9						
3s(2S)14 <i>s</i>	14 <i>s</i> 1S	0	149179. 8						
					Al III (2S _{1/2})	<i>Limit</i>		151860. 4	

July 1947.

Al II OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ +	Observed Terms				
3s ²	3s ² 1S				
3s(2S)3p	{	3p 3P° 3p 1P°			
3p ²		{	3p ² 3P 3p ² 1D		
	ns (n ≥ 4)		np (n ≥ 4)	nd (n ≥ 3)	
3s(2S)nx	{	4-13s 3S 4-16s 1S		4-11p 3P° 4-16p 1P°	3-11d 3D 3-11d 1D
3p(2P°)nx		{	4s 3P° 4s 1P°?		3d 3P°
	nf (n ≥ 4)		ng (n ≥ 5)	nh (n ≥ 6)	
3s(2S)nx	{	4-17f 3F° 4-20f 1F°		5-13g 3G 5-13g 1G	8-10h 3H° 8-10h 1H°

*For predicted terms in the spectra of the Mg I isoelectronic sequence, see Introduction.

Al III

(Na I sequence; 11 electrons)

 $Z=13$ Ground state $1s^2 2s^2 2p^6 3s^2 S_{1/2}$ $3s^2 S_{1/2}$ 229453.99 cm^{-1}

I. P. 28.44 volts

The analysis is by Paschen. Three terms, $6s^2 S$, $7s^2 S$ and $7p^2 P^\circ$ are from the paper by Ekefors, who extended the observations in the ultra-violet to 486 Å.

REFERENCES

F. Paschen, Ann. der Phys. [5] **71**, 148 (1923) and unpublished material. (I P) (T) (C L)E. Ekefors, Zeit. Phys. **51**, 471 (1928). (T) (C L)

Al III

Al III

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
3s	3s 2S	$\frac{1}{2}$	0. 00		6g	6g 2G	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ 4\frac{1}{2} \end{smallmatrix} \right\}$	202001. 32	
3p	3p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	53684. 1 53916. 6	232. 5	6h	6h $^2H^\circ$	$\left\{ \begin{smallmatrix} 4\frac{1}{2} \\ 5\frac{1}{2} \end{smallmatrix} \right\}$	202007. 32	
3d	3d 2D	$2\frac{1}{2}$ $1\frac{1}{2}$	115955. 03 115957. 31	-2. 28	7s	7s 2S	$\frac{1}{2}$	202904. 8	
4s	4s 2S	$\frac{1}{2}$	126162. 58		7p	7p $^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	205360	
4p	4p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	143632. 25 143712. 38	80. 13	7d	7d 2D	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	208880. 37	
4d	4d 2D	$2\frac{1}{2}$ $1\frac{1}{2}$	165785. 26 165786. 54	-1. 28	7f	7f $^2F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	209260. 98	
4f	4f $^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	167612. 05 167612. 43	0. 38	7g	7g 2G	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ 4\frac{1}{2} \end{smallmatrix} \right\}$	209282. 17	
5s	5s 2S	$\frac{1}{2}$	170636. 38		7h	7h $^2H^\circ$	$\left\{ \begin{smallmatrix} 4\frac{1}{2} \\ 5\frac{1}{2} \end{smallmatrix} \right\}$	209287. 52	
5p	5p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	178430. 49 178469. 64	39. 15	8d	8d 2D	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	213741. 42	
5d	5d 2D	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	188875. 52		8f	8f $^2F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	213992. 12	
5f	5f $^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	189875. 34 189875. 46	0. 12	8g	8g 2G	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ 4\frac{1}{2} \end{smallmatrix} \right\}$	214010. 67	
5g	5g 2G	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ 4\frac{1}{2} \end{smallmatrix} \right\}$	189927. 76		8h	8h $^2H^\circ$	$\left\{ \begin{smallmatrix} 4\frac{1}{2} \\ 5\frac{1}{2} \end{smallmatrix} \right\}$	214015. 8	
6s	6s 2S	$\frac{1}{2}$	191478. 5		9h	9h $^2H^\circ$	$\left\{ \begin{smallmatrix} 4\frac{1}{2} \\ 5\frac{1}{2} \end{smallmatrix} \right\}$	217255. 2	
6p	6p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	195620. 94 195641. 53	20. 59	-----				
6d	6d 2D	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	201374. 37		Al IV (1S_0)	Limit	-----	229453. 99	
6f	6f $^2F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	201969. 52						

May 1947.

Al IV

(Ne I sequence; 10 electrons)

 $Z=13$ Ground state $1s^2 2s^2 2p^6 {}^1S_0$ $2p^6 {}^1S_0$ 967783 cm^{-1}

I. P. 119.96 volts

The analysis has been taken from Söderqvist's Monograph. The term designations he assigns on the assumption of LS -coupling are given with his notation under the heading "Author" in the table.

As for Ne I, the jl -coupling notation in the general form suggested by Racah is introduced. Shortley has, however, pointed out that the configurations $2p^5 3s$, $2p^5 3p$, and $2p^5 3d$ are much closer to LS -coupling than to jl -coupling.

REFERENCES

- J. Söderqvist, Nova Acta Reg. Soc. Sci. Uppsala [IV] 9, No. 7, 34 (1934). (I P) (T) (C L).
 G. Racah, Phys. Rev. **61**, 537 (L) (1942).
 G. Shortley, unpublished material (1948).

Al IV

Al IV

Author	Config.	Desig.	J	Level	Author	Config.	Desig.	J	Level
$2p {}^1S_0$	$2p^6$	$2p^6 {}^1S$	0	0	$4s {}^3P_1$	$2p^5({}^2P_{1/2})4s$	$4s [1\frac{1}{2}]^o$	2 1	802936
$3s {}^3P_2$ 3P_1	$2p^5({}^2P_{1/2})3s$	$3s [1\frac{1}{2}]^o$	2 1	616646.7 618477.5	$4s {}^1P_1$	$2p^5({}^2P_{3/2}^o)4s$	$4s' [1\frac{1}{2}]^o$	0 1	806231
$3s {}^3P_0$ 1P_1	$2p^5({}^2P_{3/2}^o)3s$	$3s' [1\frac{1}{2}]^o$	0 1	619947.7 624720.5	$4d {}^3P_1$	$2p^5({}^2P_{1/2})4d$	$4d [1\frac{1}{2}]^o$	0 1	851956
$3p_{10} {}^3S_1$	$2p^5({}^2P_{1/2})3p$	$3p [1\frac{1}{2}]$	1	671635.5	$4d {}^1P_1$	"	$4d [1\frac{1}{2}]^o$	1	855286
$3p_9 {}^3D_3$ 3D_2	"	$3p [2\frac{1}{2}]$	3 2	680862.9 681686.7	$4d {}^3D_1$	$2p^5({}^2P_{3/2}^o)4d$	$4d' [1\frac{1}{2}]^o$	2 1	858671
$3p_7 {}^3D_1$ 1D_2	"	$3p [1\frac{1}{2}]$	1 2	682869.3 685732.8	$5s {}^3P_1$	$2p^5({}^2P_{1/2})5s$	$5s [1\frac{1}{2}]^o$	2 1	871391
$3p_3 {}^3P_0$	"	$3p [1\frac{1}{2}]$	0	688313.3	$5s {}^1P_1$	$2p^5({}^2P_{3/2}^o)5s$	$5s' [1\frac{1}{2}]^o$	0 1	874669
$3p_5 {}^1P_1$ 3P_2	$2p^5({}^2P_{3/2}^o)3p$	$3p' [1\frac{1}{2}]$	1 2	687456.8 687834.7	$5d {}^3P_1$	$2p^5({}^2P_{1/2})5d$	$5d [1\frac{1}{2}]^o$	0 1	894614
$3p_2 {}^3P_1$ 1S_0	"	$3p' [1\frac{1}{2}]$	1 0	688653.0 690244.9	$5d {}^1P_1$	"	$5d [1\frac{1}{2}]^o$	1	896138
$3d {}^3P_0$ 3P_1	$2p^5({}^2P_{1/2})3d$	$3d [1\frac{1}{2}]^o$	0 1	759197.4 759600.9	$5d {}^3D_1$	$2p^5({}^2P_{3/2}^o)5d$	$5d' [1\frac{1}{2}]^o$	2 1	899281
$3d {}^3P_2$	"	$3d [1\frac{1}{2}]^o$	2	761015.4	$6d {}^1P_1$	$2p^5({}^2P_{1/2})6d$	$6d [1\frac{1}{2}]^o$	1	918215
$3d {}^3F_4$ 3F_3	"	$3d [3\frac{1}{2}]^o$	4 3	761694.5 762277.1	$6d {}^3D_1$	$2p^5({}^2P_{3/2}^o)6d$	$6d' [1\frac{1}{2}]^o$	2 1	921362
$3d {}^3F_2$ 1F_3	"	$3d [2\frac{1}{2}]^o$	2 3	763502.8 764304.3					
$3d {}^1P_1$	"	$3d [1\frac{1}{2}]^o$	1	767040.6					
$3d {}^3D_3$ 1D_2	$2p^5({}^2P_{3/2}^o)3d$	$3d' [2\frac{1}{2}]^o$	3 2	767351.9 767536.2?		Al v (${}^2P_{1/2}$)	Limit	-----	967783
$3d {}^3D_2$ 3D_1	"	$3d' [1\frac{1}{2}]^o$	2 1	767756.1 770336.1		Al v (${}^2P_{3/2}^o$)	Limit	-----	971223

April 1947.

Al IV OBSERVED LEVELS*

Config. $1s^2 2s^2 +$	Observed Terms			
$2p^6$	$2p^6 \ ^1S$			
	$ns \ (n \geq 3)$	$np \ (n \geq 3)$		$nd \ (n \geq 3)$
$2p^5(^2P^\circ)nx$	$\begin{Bmatrix} 3-5s & ^3P^\circ \\ 3-5s & ^1P^\circ \end{Bmatrix}$	$\begin{Bmatrix} 3p & ^3S & 3p & ^3P & 3p & ^3D \\ 3p & ^1S & 3p & ^1P & 3p & ^1D \end{Bmatrix}$	$\begin{Bmatrix} 3-5d & ^3P^\circ & 3-6d & ^3D^\circ & 3d & ^3F^\circ \\ 3-6d & ^1P^\circ & 3d & ^1D^\circ & 3d & ^1F^\circ \end{Bmatrix}$	
<i>jl</i> -Coupling Notation				
	Observed Pairs			
	$ns \ (n \geq 3)$	$np \ (n \geq 3)$		$nd \ (n \geq 3)$
$2p^5(^2P_{1/2})nx$	$3-5s \ [1\frac{1}{2}]^\circ$	$\begin{Bmatrix} 3p \ [\frac{1}{2}] \\ 3p \ [2\frac{1}{2}] \\ 3p \ [1\frac{1}{2}] \end{Bmatrix}$		$\begin{Bmatrix} 3-5d \ [\frac{1}{2}]^\circ \\ 3d \ [3\frac{1}{2}]^\circ \\ 3-6d \ [1\frac{1}{2}]^\circ \\ 3d \ [2\frac{1}{2}]^\circ \end{Bmatrix}$
$2p^5(^2P_{3/2}^\circ)nx'$	$3-5s' \ [\frac{1}{2}]^\circ$	$\begin{Bmatrix} 3p' \ [1\frac{1}{2}] \\ 3p' \ [\frac{1}{2}] \end{Bmatrix}$		$\begin{Bmatrix} 3d' \ [2\frac{1}{2}]^\circ \\ 3-6d' \ [1\frac{1}{2}]^\circ \end{Bmatrix}$

*For predicted levels in the spectra of the Ne I isoelectronic sequence, see Introduction.

Al v

(F I sequence; 9 electrons)

$Z=13$

Ground state $1s^2 2s^2 2p^5 \ ^2P_{1/2}^\circ$

$2p^5 \ ^2P_{1/2}^\circ$ **1240600** cm^{-1}

I. P. 153.77 volts

The analysis published by Söderqvist in 1934 has been extended by Ferner to include 78 classified lines in the region between 85 Å and 281 Å. The present list has been compiled from unpublished material kindly furnished by Ferner.

Intersystem combinations connecting the doublet and quartet terms have been observed. All but one of the observed combinations are with the ground term.

Ferner's unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

By analogy with related spectra in the isoelectronic sequence Robinson has suggested the following changes in Ferner's term assignments:

Ferner	Robinson	Ferner	Robinson
$3d \ ^4P_{2/2}$	$3d \ ^2D_{2/2}$	$3d' \ ^2S_{1/2}$	$3d' \ ^2P_{1/2}$
$3d \ ^4D_{1/2}$ $\ ^4D_{2/2}$	$3d \ ^4F_{1/2}$ $3d \ ^4P_{2/2}$	$3d' \ ^2P_{1/2}$	$3d' \ ^2D_{1/2}$
$3d \ ^2D_{2/2}$	$3d \ ^2F_{2/2}$	$3d' \ ^2D_{1/2}$	$3d' \ ^2S_{1/2}$
$4d \ ^4D_{1/2}$ $\ ^4D_{2/2}$	$4d \ ^4P_{1/2}$ $4d \ ^2D_{2/2}$	$3d' \ ^2D_{2/2}$ $4d' \ ^2S_{1/2}$	$3d' \ ^2D_{2/2} \ ^2F_{2/2}$ $4d' \ ^2P_{1/2}$
$4d \ ^2D_{1/2}$ $\ ^2D_{2/2}$	$4d \ ^2P_{1/2}$ $4d \ ^2D_{1/2}$	$4d' \ ^2P_{1/2}$ $4d' \ ^2P_{1/2} \ ^2S_{1/2}^*$	$4d' \ ^2D$

*1100620.

He has also suggested a correction of $+1000 \text{ cm}^{-1}$ to Ferner's absolute term values. This correction has been made in the limit quoted here.

Al v—Continued

REFERENCES

- J. Söderqvist, Nova Acta Reg. Soc. Sci. Uppsala [IV] **9**, No. 7, 39 (1934). (T) (C L)
 E. Ferner, Ark. Mat. Astr. Fys. (Stockholm) **36A**, No. 1, p. 57 (1948). (I P) (T) (C L)
 H. A. Robinson, unpublished material (March 1948). (T) (C L)

Al v

Al v

Author	Config.	Desig.	J	Level	Interval	Author	Config.	Desig.	J	Level	Interval
$2p \ ^2P_2$ $\ ^2P_1$	$2s^2 2p^5$	$2p^5 \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	0 3440	-3440	$4d \ ^4D_3$ $\ ^4D_2$	$2s^2 2p^4(^3P)4d$	$4d \ ^4D$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	1062510 1062820	-310
$2p' \ ^2S_1$	$2s \ 2p^6$	$2p^6 \ ^2S$	$\frac{1}{2}$	358810							
$3s \ ^4P_3$ $\ ^4P_2$ $\ ^4P_1$	$2s^2 2p^4(^3P)3s$	$3s \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	751810 753960 755250	-2150 -1290	$4d \ ^4P_2$ $\ ^4P_3$	$2s^2 2p^4(^3P)4d$	$4d \ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	1063650 1064050	400
$3s \ ^2P_2$ $\ ^2P_1$	$2s^2 2p^4(^3P)3s$	$3s \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	764240 766790	-2550	$4d \ ^2P_1$ $\ ^2P_2$	$2s^2 2p^4(^3P)4d$	$4d \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	1065170 1067770	2600
$\overline{3s} \ ^2D_3$ $\ ^2D_2$	$2s^2 2p^4(^1D)3s$	$3s' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	796650 796680	-30	$4d \ ^2D_2$ $\ ^2D_3$	$2s^2 2p^4(^3P)4d$	$4d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	1065460 1066610	1150
$\overline{3s} \ ^2S_1$	$2s^2 2p^4(^1S)3s$	$3s'' \ ^2S$	$\frac{1}{2}$	843880		$\overline{4s} \ ^2S_1$	$2s^2 2p^4(^1S)4s$	$4s'' \ ^2S$	$\frac{1}{2}$	1089930	
$3d \ ^4D_3$ $\ ^4D_2$	$2s^2 2p^4(^3P)3d$	$3d \ ^4D$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	919900 920680	-780	$3s' \ ^2P_2$ $\ ^2P_1$	$2s \ 2p^5(^3P^\circ)3s$	$3s''' \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	1096180 1098350	-2170
						$\overline{4d} \ ^2P_1$ $\ ^2P_2$	$2s^2 2p^4(^1D)4d$	$4d' \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	1101400 1103380	1980
$3d \ ^4P_1$ $\ ^4P_2$ $\ ^4P_3$	$2s^2 2p^4(^3P)3d$	$3d \ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	921440 922120 922640	680 520	$\overline{4d} \ ^2S_1$	$2s^2 2p^4(^1D)4d$	$4d' \ ^2S$	$\frac{1}{2}$	1102540	
$3d \ ^2F_3$	$2s^2 2p^4(^3P)3d$	$3d \ ^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	923230		$\overline{4d} \ ^2D_3$	$2s^2 2p^4(^1D)4d$	$4d' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	1103190	
$3d \ ^2D_2$ $\ ^2D_3$	$2s^2 2p^4(^3P)3d$	$3d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	925430 926400	970	$5d \ ^4D_3$ $\ ^4D_2$	$2s^2 2p^4(^3P)5d$	$5d \ ^4D$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	1127550 1127730	-180
$3d \ ^2P_1$ $\ ^2P_2$	$2s^2 2p^4(^3P)3d$	$3d \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	925900 928410	2510	$5d \ ^2D_2$ $\ ^2D_3$	$2s^2 2p^4(^3P)5d$	$5d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	1129350 1130900	1550
$\overline{3d} \ ^2P_1$ $\ ^2P_2$	$2s^2 2p^4(^1D)3d$	$3d' \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	960420 961630	1210	$5d \ ^2P_1$ $\ ^2P_2$	$2s^2 2p^4(^3P)5d$	$5d \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	1129350 1131650	2300
$\overline{3d} \ ^2S_1$	$2s^2 2p^4(^1D)3d$	$3d' \ ^2S$	$\frac{1}{2}$	960860		$\overline{4d} \ ^2D_3$ $\ ^2D_2$	$2s^2 2p^4(^1S)4d$	$4d'' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	1149160 1149260	-100
$\overline{3d} \ ^2D_3$ $\ ^2D_2$	$2s^2 2p^4(^1D)3d$	$3d' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	962640 963330	-690	$6d \ ^2D_2$ $\ ^2D_3$	$2s^2 2p^4(^3P)6d$	$6d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	1163850 1165450	1600
$4s \ ^2P_2$ $\ ^2P_1$	$2s^2 2p^4(^3P)4s$	$4s \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	1005760 1008040	-2280	$\overline{5d} \ ^2S_1$	$2s^2 2p^4(^1D)5d$	$5d' \ ^2S$	$\frac{1}{2}$	1167380	
$\overline{3d} \ ^2D_3$ $\ ^2D_2$	$2s^2 2p^4(^1S)3d$	$3d'' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	1007150 1007290	-140	$\overline{5d} \ ^2P_2$	$2s^2 2p^4(^1D)5d$	$5d' \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	1168060	
$\overline{4s} \ ^2D_3$ $\ ^2D_2$	$2s^2 2p^4(^1D)4s$	$4s' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	1043430 1043480	-50						
							Al vi (3P_2)	Limit		1240600	

March 1948.

Al V OBSERVED TERMS*

Config. $1s^2+$	Observed Terms	
$2s^2 2p^5$	$2p^5 \ ^2P^\circ$	
$2s \ 2p^6$	$2p^6 \ ^2S$	
	$ns \ (n \geq 3)$	$nd \ (n \geq 3)$
$2s^2 2p^4(^3P)nx$	$\left\{ \begin{array}{l} 3s \ ^4P \\ 3, 4s \ ^2P \end{array} \right.$	$\begin{array}{lll} 3, 4d \ ^4P & 3-5d \ ^4D & \\ 3-5d \ ^2P & 3-6d \ ^2D & 3d \ ^2F \end{array}$
$2s^2 2p^4(^1D)nx'$	$3, 4s' \ ^2D$	$3-5d' \ ^2S \quad 3-5d' \ ^2P \quad 3, 4d' \ ^2D$
$2s^2 2p^4(^1S)nx''$	$3, 4s'' \ ^2S$	$3, 4d'' \ ^2D$
$2s \ 2p^5(^3P^\circ)nx'''$	$3s''' \ ^2P^\circ$	

*For predicted terms in the spectra of the F I isoelectronic sequence, see Introduction.

Al VI

(O I sequence; 8 electrons)

$Z=13$

Ground state $1s^2 2s^2 2p^4 \ ^3P_2$

$2p^4 \ ^3P_2 \ 1536300 \text{ cm}^{-1}$

I. P. 190.42 volts

The analysis is by Ferner, who has extended the earlier work by Söderqvist. He has listed 45 terms and 89 classified lines. The later observations are in the region between 68 Å and 113 Å. Two intersystem combinations have been observed.

Ferner expresses all level values in units of 10^3 cm^{-1} but for uniformity all values listed below are given in cm^{-1} .

REFERENCES

- J. Söderqvist, Nova Acta Reg. Soc. Sci. Uppsala [IV] **9**, No. 7, 51 (1934). (T) (C L)
 E. Ferner, Ark. Mat. Astr. Fys. (Stockholm) **36A**, No. 1, p. 48 (1948). (I P) (T) (C L)

Al VI

Al VI

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$2s^2 2p^4$	$2p^4 \ ^3P$	$\begin{array}{l} 2 \\ 1 \\ 0 \end{array}$	$\begin{array}{l} 0 \\ 2736 \\ 3831 \end{array}$	$\begin{array}{l} -2736 \\ -1095 \end{array}$	$2s^2 2p^3(^2P^\circ)3s$	$3s'' \ ^3P^\circ$	$\begin{array}{l} 0 \\ 1 \\ 2 \end{array}$	$\begin{array}{l} 993660 \\ 993880 \end{array}$	$\begin{array}{l} 220 \end{array}$
$2s^2 2p^4$	$2p^4 \ ^1D$	2	41600		$2s^2 2p^3(^2P^\circ)3s$	$3s'' \ ^1P^\circ$	1	1003700	
$2s^2 2p^4$	$2p^4 \ ^1S$	0	88670		$2s^2 2p^3(^4S^\circ)3d$	$3d \ ^3D^\circ$	$\begin{array}{l} 1 \\ 2 \\ 3 \end{array}$	$\begin{array}{l} 1079460 \\ 1079490 \\ 1079610 \end{array}$	$\begin{array}{l} 30 \\ 120 \end{array}$
$2s \ 2p^5$	$2p^5 \ ^3P^\circ$	$\begin{array}{l} 2 \\ 1 \\ 0 \end{array}$	$\begin{array}{l} 323002 \\ 325470 \\ 326822 \end{array}$	$\begin{array}{l} -2468 \\ -1352 \end{array}$	$2s^2 2p^3(^2D^\circ)3d$	$3d' \ ^3F^\circ$	$\begin{array}{l} 4 \\ 3 \\ 2 \end{array}$	$\begin{array}{l} 1132180 \end{array}$	
$2s \ 2p^5$	$2p^5 \ ^1P^\circ$	1	451840		$2s^2 2p^3(^2D^\circ)3d$	$3d' \ ^3D^\circ$	$3, 2, 1$	1134170	
$2s^2 2p^3(^4S^\circ)3s$	$3s \ ^3S^\circ$	1	913130		$2s^2 2p^3(^2D^\circ)3d$	$3d' \ ^1P^\circ$	1	1136500	
$2s^2 2p^3(^2D^\circ)3s$	$3s' \ ^3D^\circ$	$3, 2, 1$	961100						
$2s^2 2p^3(^2D^\circ)3s$	$3s' \ ^1D^\circ$	2	970790						

Al VI—Continued

Al VI—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$2s^2 2p^3(^2D^\circ)3d$	$3d' \ ^3P^\circ$	2 1 0	1140840 1141670 1141910	-830 -240	$2s^2 2p^3(^2P^\circ)4s$	$4s'' \ ^1P^\circ$	1	1312070	
$2s^2 2p^3(^2D^\circ)3d$	$3d' \ ^1D^\circ$	2	1142220		$2s^2 2p^3(^2D^\circ)4d$	$4d' \ ^3D^\circ$	3, 2, 1	1339480	
$2s^2 2p^3(^2D^\circ)3d$	$3d' \ ^3S^\circ$	1	1145020		$2s^2 2p^3(^2D^\circ)4d$	$4d' \ ^1P^\circ$	1	1341090	
$2s^2 2p^3(^2D^\circ)3d$	$3d' \ ^1F^\circ$	3	1150250		$2s^2 2p^3(^2D^\circ)4d$	$4d' \ ^3P^\circ$	2 1 0	1343320	
$2s^2 2p^3(^2P^\circ)3d$	$3d'' \ ^3P^\circ$	0 1 2	1164220 1164620 1165260	400 640	$2s^2 2p^3(^2D^\circ)4d$	$4d' \ ^3S^\circ$	1	1345030	
$2s^2 2p^3(^2P^\circ)3d$	$3d'' \ ^3F^\circ$	4 3 2	1166530 1168690	-2160	$2s^2 2p^3(^2D^\circ)4d$	$4d' \ ^1D^\circ$	2	1345430	
$2s^2 2p^3(^2P^\circ)3d$	$3d'' \ ^1D^\circ$	2	1169150		$2s^2 2p^3(^2D^\circ)4d$	$4d' \ ^1F^\circ$	3	1346780	
$2s^2 2p^3(^2P^\circ)3d$	$3d'' \ ^3D^\circ$	3 2 1	1169390 1170650	-1260	$2s \ 2p^4(^2S)3s$	$3s^\vee \ ^3S$	1	1359890	
$2s^2 2p^3(^2P^\circ)3d$	$3d'' \ ^1P^\circ$	1	1171050		$2s^2 2p^3(^2P^\circ)4d$	$4d'' \ ^3P^\circ$	0 1 2	1371220	
$2s^2 2p^3(^2P^\circ)3d$	$3d'' \ ^1F^\circ$	3	1174450		$2s^2 2p^3(^2P^\circ)4d$	$4d'' \ ^3D^\circ$	3 2 1	1373440 1375140	-1700
$2s \ 2p^4(^4P)3s$	$3s''' \ ^3P$	2 1 0	1204550 1205500	-950	$2s^2 2p^3(^4S^\circ)5d$	$5d \ ^3D^\circ$	1, 2, 3	1375250	
$2s^2 2p^3(^4S^\circ)4s$	$4s \ ^3S^\circ$	1	1218290		$2s^2 2p^3(^2P^\circ)4d$	$4d'' \ ^1F^\circ$	3	1376860	
$2s^2 2p^3(^2D^\circ)4s$	$4s' \ ^3D^\circ$	3, 2, 1	1274550		$2s^2 2p^3(^2D^\circ)5s$	$5s' \ ^1D^\circ$	2	1405220	
$2s^2 2p^3(^2D^\circ)4s$	$4s' \ ^1D^\circ$	2	1279680		$2s^2 2p^3(^2P^\circ)5d$	$5d'' \ ^3P^\circ$	0 1 2	1465780	
$2s^2 2p^3(^4S^\circ)4d$	$4d \ ^3D^\circ$	1, 2, 3	1282960		$2s^2 2p^3(^2P^\circ)5d$	$5d'' \ ^3D^\circ$	3 2 1	1466990	
$2s \ 2p^4(^2D)3s$	$3s^\text{IV} \ ^3D$	3, 2, 1	1293290		Al VII ($^4S_{1/2}$)	Limit	-----	1536300	

February 1947.

Al VI OBSERVED TERMS*

Config. $1s^2 +$	Observed Terms			
$2s^2 2p^4$	{ $2p^4 \ ^1S$ $2p^4 \ ^3P$ $2p^4 \ ^1D$			
$2s \ 2p^5$	{ $2p^5 \ ^3P^\circ$ $2p^5 \ ^1P^\circ$			
	$ns \ (n \geq 3)$		$nd \ (n \geq 3)$	
$2s^2 2p^3(^4S^\circ)nx$	$3, 4s \ ^3S^\circ$		$3-5d \ ^3D^\circ$	
$2s^2 2p^3(^2D^\circ)nx'$	{ $3, 4s' \ ^3D^\circ$ $3-5s' \ ^1D^\circ$		$3, 4d' \ ^3S^\circ$	$3, 4d' \ ^3P^\circ$ $3, 4d' \ ^3D^\circ$ $3d' \ ^3F^\circ$ $3, 4d' \ ^1P^\circ$ $3, 4d' \ ^1D^\circ$ $3, 4d' \ ^1F^\circ$
$2s^2 2p^3(^2P^\circ)nx''$	{ $3s'' \ ^3P^\circ$ $3, 4s'' \ ^1P^\circ$		$3-5d'' \ ^3P^\circ$	$3-5d'' \ ^3D^\circ$ $3d'' \ ^3F^\circ$ $3d'' \ ^1P^\circ$ $3d'' \ ^1D^\circ$ $3, 4d'' \ ^1F^\circ$
$2s \ 2p^4(^4P)nx'''$	$3s''' \ ^3P$			
$2s \ 2p^4(^2D)nx^\text{IV}$	$3s^\text{IV} \ ^3D$			
$2s \ 2p^4(^2S)nx^\vee$	$3s^\vee \ ^3S$			

*For predicted terms in the spectra of the O I isoelectronic sequence, see Introduction.

Al VII

(N I sequence; 7 electrons)

 $Z=13$ Ground state $1s^2 2s^2 2p^3 {}^4S_{3/2}^\circ$ $2p^3 {}^4S_{3/2}^\circ$ 1951830 cm^{-1}

I. P. 241.93 volts

The analysis is from Ferner who kindly furnished his manuscript in advance of publication. He has extended the earlier work by Söderqvist to include 76 classified lines between 58 Å and 96 Å. One intersystem combination has been observed, but the relative positions of the doublet and quartet terms are determined from the series.

The unit used by Ferner, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

- J. Söderqvist, Nova Acta Reg. Soc. Sci. Uppsala [IV] 9, No. 7, 64 (1934). (T) (C L)
E. Ferner, Ark. Mat. Astr. Fys. (Stockholm) 36A, No. 1, p. 42 (1948). (I P) (T) (C L)

Al VII

Al VII

Author	Config.	Desig.	J	Level	Interval	Author	Config.	Desig.	J	Level	Interval
$2p {}^4S_2$	$2s^2 2p^3$	$2p^3 {}^4S^\circ$	$1\frac{1}{2}$	0		$3d {}^2D_2$	$2s^2 2p^2({}^3P)3d$	$3d {}^2D$	$1\frac{1}{2}$	1343710	820
$2p {}^2D_2$	$2s^2 2p^3$	$2p^3 {}^2D^\circ$	$1\frac{1}{2}$	60700	60	$3d {}^2D_3$			$2\frac{1}{2}$	1344530	
$2p {}^2D_3$	$2s^2 2p^3$		$2\frac{1}{2}$	60760		$\overline{3d} {}^2F_4$	$2s^2 2p^2({}^1D)3d$	$3d' {}^2F$	$3\frac{1}{2}$	1366720	-440
$2p {}^2P_1$	$2s^2 2p^3$	$2p^3 {}^2P^\circ$	$\frac{1}{2}$	93000	270	$\overline{3d} {}^2F_3$			$2\frac{1}{2}$	1367160	
$2p {}^2P_2$	$2s^2 2p^3$		$1\frac{1}{2}$	93270		$\overline{3d} {}^2D_2$	$2s^2 2p^2({}^1D)3d$	$3d' {}^2D$	$1\frac{1}{2}$	1369270	690
$2p' {}^4P_3$	$2s 2p^4$	$2p^4 {}^4P$	$2\frac{1}{2}$	280200	-2460	$\overline{3d} {}^2D_3$			$2\frac{1}{2}$	1369960	
$2p' {}^4P_2$	$2s 2p^4$		$1\frac{1}{2}$	282660	-1300	$\overline{3d} {}^2P_1$	$2s^2 2p^2({}^1D)3d$	$3d' {}^2P$	$\frac{1}{2}$	1378290	840
$2p' {}^4P_1$	$2s 2p^4$		$\frac{1}{2}$	283960		$\overline{3d} {}^2P_2$			$1\frac{1}{2}$	1379130	
$2p' {}^2D_3$	$2s 2p^4$	$2p^4 {}^2D$	$2\frac{1}{2}$	384260	-50	$3p' {}^4P$	$2s 2p^3({}^3S^\circ)3p$	$3p''' {}^4P$	$\left\{ \begin{array}{c} \frac{1}{2} \\ \text{to} \\ 2\frac{1}{2} \end{array} \right\}$	1383700	
$2p' {}^2D_2$	$2s 2p^4$		$1\frac{1}{2}$	384310		$\overline{3d} {}^2S_1$	$2s^2 2p^2({}^1D)3d$	$3d' {}^2S$	$\frac{1}{2}$	1384370	
$2p' {}^2S_1$	$2s 2p^4$	$2p^4 {}^2S$	$\frac{1}{2}$	451360		$\overline{3d} {}^2D$	$2s^2 2p^2({}^1S)3d$	$3d'' {}^2D$	$\left\{ \begin{array}{c} 1\frac{1}{2} \\ \text{to} \\ 2\frac{1}{2} \end{array} \right\}$	1410380	
$2p' {}^2P_2$	$2s 2p^4$	$2p^4 {}^2P$	$1\frac{1}{2}$	476090	-2960	$\overline{3d} {}^2D$			$\left\{ \begin{array}{c} \frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{array} \right\}$	1473060	
$2p' {}^2P_1$	$2s 2p^4$		$\frac{1}{2}$	479050		$3d' {}^4D$	$2s 2p^3({}^3S^\circ)3d$	$3d''' {}^4D^\circ$	$\left\{ \begin{array}{c} \frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{array} \right\}$	1591560	-610
$3s {}^4P_1$	$2s^2 2p^2({}^3P)3s$	$3s {}^4P$	$\frac{1}{2}$	1147100	1530	$4s {}^4P_2$	$2s^2 2p^2({}^3P)4s$	$4s {}^4P$	$\frac{1}{2}$	1540740	2110
$3s {}^4P_2$			$1\frac{1}{2}$	1148630	2290	$4s {}^4P_3$			$1\frac{1}{2}$	1542850	
$3s {}^4P_3$			$2\frac{1}{2}$	1150920		$4s {}^2P_2$	$2s^2 2p^2({}^3P)4s$	$4s {}^2P$	$\frac{1}{2}$	1540820	
$3s {}^2P_1$	$2s^2 2p^2({}^3P)3s$	$3s {}^2P$	$\frac{1}{2}$	1162360	2770	$\overline{3d} {}^4P_3$	$2s 2p^3({}^3D^\circ)3d$	$3d^{IV} {}^4P^\circ$	$2\frac{1}{2}$	1591560	-380
$3s {}^2P_2$			$1\frac{1}{2}$	1165130		$\overline{3d} {}^4P_2$			$1\frac{1}{2}$	1592170	
$\overline{3s} {}^2D$	$2s^2 2p^2({}^1D)3s$	$3s' {}^2D$	$\left\{ \begin{array}{c} 1\frac{1}{2} \\ \text{to} \\ 2\frac{1}{2} \end{array} \right\}$	1196680		$\overline{3d} {}^4P_1$			$\frac{1}{2}$	1592550	
$\overline{3s} {}^2S_1$	$2s^2 2p^2({}^1S)3s$	$3s'' {}^2S$	$\frac{1}{2}$	1246840		$\overline{3d}' {}^4D$	$2s 2p^3({}^3D^\circ)3d$	$3d^{IV} {}^4D^\circ$	$\left\{ \begin{array}{c} \frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{array} \right\}$	1598270	
$3d {}^2P_2$	$2s^2 2p^2({}^3P)3d$	$3d {}^2P$	$1\frac{1}{2}$	1315640	-780	$4d {}^2P_2$	$2s^2 2p^2({}^3P)4d$	$4d {}^2P$	$1\frac{1}{2}$	1598890	
$3d {}^2P_1$			$\frac{1}{2}$	1316420		$\overline{3d}' {}^4S_2$	$2s 2p^3({}^3D^\circ)3d$	$3d^{IV} {}^4S^\circ$	$1\frac{1}{2}$	1599300	
$3s' {}^4S_2$	$2s 2p^3({}^3S^\circ)3s$	$3s''' {}^4S^\circ$	$1\frac{1}{2}$	1322180		$4d {}^4D_{32}$	$2s^2 2p^2({}^3P)4d$	$4d {}^4D$	$3\frac{1}{2}$	1600670	-1070
$3d {}^2F_3$	$2s^2 2p^2({}^3P)3d$	$3d {}^2F$	$2\frac{1}{2}$	1323370	3020	$\overline{4d} {}^4D_{32}$			$2\frac{1}{2}$	1601740	
$3d {}^2F_4$			$3\frac{1}{2}$	1326390		$\overline{4d} {}^4D_1$			$1\frac{1}{2}$		
$3d {}^4D_{32}$	$2s^2 2p^2({}^3P)3d$	$3d {}^4D$	$3\frac{1}{2}$	1323940	-770				$\frac{1}{2}$		
$3d {}^4D_1$			$1\frac{1}{2}$	1324710							
$3d {}^4P_3$	$2s^2 2p^2({}^3P)3d$	$3d {}^4P$	$2\frac{1}{2}$	1326960	-1030						
$3d {}^4P_2$			$1\frac{1}{2}$	1327990	-560						
$3d {}^4P_1$			$\frac{1}{2}$	1328550							

Al VII—Continued

Al VII—Continued

Author	Config.	Desig.	<i>J</i>	Level	Inter- val	Author	Config.	Desig.	<i>J</i>	Level	Inter- val
$4d \ ^2F_3$ $\ ^2F_4$	$2s^2 2p^2(^3P)4d$	$4d \ ^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	1603550 1606260	2710		$2s^2 2p^2(^3P)5s$	$5s \ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	1702070	
$4d \ ^4P_3$	$2s^2 2p^2(^3P)4d$	$4d \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	1605240		$5s \ ^4P_3$					
						$5d \ ^2F_3$ $\ ^2F_4$	$2s^2 2p^2(^3P)5d$	$5d \ ^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	1729840 1732410	2570
$4d \ ^2D_2$ $\ ^2D_3$	$2s^2 2p^2(^3P)4d$	$4d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	1610820 1611560	740	$4d' \ ^4D_4$ $\ ^4D_3$	$2s \ 2p^3(^5S^{\circ})4d$	$4d''' \ ^4D^{\circ}$	$3\frac{1}{2}$ $2\frac{1}{2}$ $\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ \frac{1}{2} \end{smallmatrix} \right\}$	1739390 1739600 1739970	-210 -370
$4d \ ^2D_2$ $\ ^2D_3$	$2s^2 2p^2(^1D)4d$	$4d' \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	1646820 1647880	1060	$\ ^4D_{21}$					
$4d \ ^2F_{3/4}$	$2s^2 2p^2(^1D)4d$	$4d' \ ^2F$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	1647430		$5d \ ^2F_{4/3}$	$2s^2 2p^2(^1D)5d$	$5d' \ ^2F$	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	1773560	
$4d \ ^2S_1$	$2s^2 2p^2(^1D)4d$	$4d' \ ^2S$	$\frac{1}{2}$	1654160							
						Al VIII (3P_0)		Limit		1951830	

March 1947.

Al VII OBSERVED TERMS*

Config. $1s^2+$	Observed Terms		
$2s^2 2p^3$	$\left\{ \begin{array}{l} 2p^3 \ ^4S^{\circ} \quad 2p^3 \ ^2P^{\circ} \quad 2p^3 \ ^2D^{\circ} \end{array} \right.$		
$2s \ 2p^4$	$\left\{ \begin{array}{l} 2p^4 \ ^2S \quad 2p^4 \ ^4P \quad 2p^4 \ ^2P \quad 2p^4 \ ^2D \end{array} \right.$		
	$ns \ (n \geq 3)$	$np \ (n \geq 3)$	$nd \ (n \geq 3)$
$2s^2 2p^2(^3P)nx$	$\left\{ \begin{array}{l} 3-5s \ ^4P \\ 3, 4s \ ^2P \end{array} \right.$		$\begin{array}{l} 3, 4d \ ^4P \quad 3, 4d \ ^4D \\ 3, 4d \ ^2P \quad 3, 4d \ ^2D \quad 3-5d \ ^2F \end{array}$
$2s^2 2p^2(^1D)nx'$		$3s' \ ^2D$	$\begin{array}{l} 3, 4d' \ ^2S \quad 3d' \ ^2P \quad 3, 4d' \ ^2D \quad 3-5d' \ ^2F \end{array}$
$2s^2 2p^2(^1S)nx''$	$3s'' \ ^2S$		$3d'' \ ^2D$
$2s \ 2p^3(^5S^{\circ})nx'''$	$3s''' \ ^4S^{\circ}$	$3p''' \ ^4P$	$3, 4d''' \ ^4D^{\circ}$
$2s \ 2p^3(^3D^{\circ})nx^{IV}$			$3d^{IV} \ ^4S^{\circ} \quad 3d^{IV} \ ^4P^{\circ} \quad 3d^{IV} \ ^4D^{\circ}$

*For predicted terms in the spectra of the N I isoelectronic sequence, see Introduction.

Al VIII

(C I sequence; 6 electrons)

 $Z=13$ Ground state $1s^2 2s^2 2p^2 \ ^3P_0$ $2p^2 \ ^3P_0 \ 2300390 \text{ cm}^{-1}$

I. P. 285.13 volts

The analysis is by Ferner, who has generously furnished his manuscript in advance of publication. He has extended the earlier work by Söderqvist to include 77 classified lines in the region between 53 Å and 91 Å. The relative values of the singlet, triplet, and quintet systems of terms are determined from the series limits.

Ferner's unit, 10^3 cm^{-1} , has here been converted to cm^{-1} .

REFERENCES

- J. Söderqvist, Nova Acta Reg. Soc. Sci. Uppsala [IV], **9**, No. 7, 77 (1934). (T) (C L)
 E. Ferner, Ark. Mat. Astr. Fys. (Stockholm) **36A**, No. 1, p. 37 (1948). (I P) (T) (C L)

Al VIII

Al VIII

Author	Config.	Desig.	<i>J</i>	Level	Inter- val	Author	Config.	Desig.	<i>J</i>	Level	Inter- val
$2p \begin{smallmatrix} ^3P_0 \\ ^3P_1 \\ ^3P_2 \end{smallmatrix}$	$2s^2 2p^2$	$2p^2 \ ^3P$	0 1 2	0 1740 4440	1740 2700	$3d' \begin{smallmatrix} ^5P_3 \\ ^5P_2 \\ ^5P_1 \end{smallmatrix}$	$2s \ 2p^2(^4P)3d$	$3d \ ^5P$	3 2 1	$1631170+y$ $1632060+y$ $1632670+y$	—890 —610
$2p \ ^1D_2$	$2s^2 2p^2$	$2p^2 \ ^1D$	2	$46690+x$		$3d' \begin{smallmatrix} ^3P_2 \\ ^3P_1 \end{smallmatrix}$	$2s \ 2p^2(^4P)3d$	$3d \ ^3P$	2 1 0	1633840 1635440	—1600
$2p \ ^1S_0$	$2s^2 2p^2$	$2p^2 \ ^1S$	0	$96170+x$							
$2p' \ ^5S_2$	$2s \ 2p^3$	$2p^3 \ ^5S^\circ$	2	$133510+y$		$3d' \begin{smallmatrix} ^3F_2 \\ ^3F_3 \\ ^3F_4 \end{smallmatrix}$	$2s \ 2p^2(^4P)3d$	$3d \ ^3F$	2 3 4	1643590 1644990 1646790	1400 1800
$2p' \begin{smallmatrix} ^3D_3 \\ ^3D_2 \\ ^3D_1 \end{smallmatrix}$	$2s \ 2p^3$	$2p^3 \ ^3D^\circ$	3 2 1	262190 262320 262390	—130 —70	$\overline{3p'} \ ^1F_3$	$2s \ 2p^2(^2D)3p$	$3p' \ ^1F^\circ$	3	$1659180+x$	
$2p' \ ^3P$	$2s \ 2p^3$	$2p^3 \ ^3P^\circ$	0, 1, 2	309130		$\overline{\overline{3s'}} \ ^3S_1$	$2s \ 2p^2(^2S)3s$	$3s'' \ ^3S$	1	1662740	
$2p' \ ^1D_2$	$2s \ 2p^3$	$2p^3 \ ^1D^\circ$	2	$396990+x$		$3d' \begin{smallmatrix} ^3D_1 \\ ^3D_2 \\ ^3D_3 \end{smallmatrix}$	$2s \ 2p^2(^4P)3d$	$3d \ ^3D$	1 2 3	1664880 1665380 1665930	500 550
$2p' \ ^3S_1$	$2s \ 2p^3$	$2p^3 \ ^3S^\circ$	1	404220		$\overline{3p'} \ ^1D_2$	$2s \ 2p^2(^2D)3p$	$3p' \ ^1D^\circ$	2	$1667490+x$	
$2p' \ ^1P_1$	$2s \ 2p^3$	$2p^3 \ ^1P^\circ$	1	$444550+x$			$2s \ 2p^2(^2P)3s$	$3s''' \ ^3P$	0 1 2	1682590	
$3s \begin{smallmatrix} ^3P_0 \\ ^3P_1 \\ ^3P_2 \end{smallmatrix}$	$2s^2 2p(^2P^\circ)3s$	$3s \ ^3P^\circ$	0 1 2	1319280 1320450 1324080	1170 3630	$\overline{\overline{3s'}} \ ^3P_2$					
$3s \ ^1P_1$	$2s^2 2p(^2P^\circ)3s$	$3s \ ^1P^\circ$	1	$1335270+x$		$\overline{3d'} \ ^3F$	$2s \ 2p^2(^2D)3d$	$3d' \ ^3F$	2, 3, 4	1733950	
$3p \ ^3S_1$	$2s^2 2p(^2P^\circ)3p$	$3p \ ^3S$	1	1402180		$\overline{3d'} \ ^3D$	$2s \ 2p^2(^2D)3d$	$3d' \ ^3D$	1, 2, 3	1742250	
$3s' \begin{smallmatrix} ^5P_1 \\ ^5P_2 \\ ^5P_3 \end{smallmatrix}$	$2s \ 2p^2(^4P)3s$	$3s \ ^5P$	1 2 3	$1465810+y$ $1467470+y$ $1469680+y$	1660 2210	$\overline{3d'} \begin{smallmatrix} ^3P_2 \\ ^3P_1 \\ ^3P_0 \end{smallmatrix}$	$2s \ 2p^2(^2D)3d$	$3d' \ ^3P$	2 1 0	1745690 1747940 1749640	—1250 —1700
$3d \ ^3F_2$	$2s^2 2p(^2P^\circ)3d$	$3d \ ^3F^\circ$	2 3 4	$1468700+x$		$\overline{3d'} \ ^3S_1$	$2s \ 2p^2(^2D)3d$	$3d' \ ^3S$	1	1762090	
							$2s^2 2p(^2P^\circ)4s$	$4s \ ^3P^\circ$	0 1 2	1785380	
$3d \ ^1D_2$	$2s^2 2p(^2P^\circ)3d$	$3d \ ^1D^\circ$	2	$1471980+x$		$4s \ ^3P_2$					
$3d \begin{smallmatrix} ^3D_1 \\ ^3D_2 \\ ^3D_3 \end{smallmatrix}$	$2s^2 2p(^2P^\circ)3d$	$3d \ ^3D^\circ$	1 2 3	1484560 1485240 1486710	680 1470	$\overline{\overline{3d'}} \begin{smallmatrix} ^3D_2 \\ ^3D_3 \end{smallmatrix}$	$2s \ 2p^2(^2S)3d$	$3d'' \ ^3D$	1 2 3	1815990 1816950	960
$3d \begin{smallmatrix} ^3P_2 \\ ^3P_1 \\ ^3P_0 \end{smallmatrix}$	$2s^2 2p(^2P^\circ)3d$	$3d \ ^3P^\circ$	2 1 0	1490590 1491570 1492140	—980 —570	$\overline{\overline{3d'}} \ ^3F$	$2s \ 2p^2(^2P)3d$	$3d''' \ ^3F$	2, 3, 4	1831700	
						$\overline{\overline{3d'}} \ ^3D$	$2s \ 2p^2(^2P)3d$	$3d''' \ ^3D$	1, 2, 3	1840570	
$3s' \begin{smallmatrix} ^3P_1 \\ ^3P_2 \end{smallmatrix}$	$2s \ 2p^2(^4P)3s$	$3s \ ^3P$	0 1 2	1504810 1507220	2410	$\overline{\overline{\overline{3d'}}} \ ^3P_2$	$2s \ 2p^2(^2P)3d$	$3d''' \ ^3P$	0 1 2	1844390	
$3d \ ^1F_3$	$2s^2 2p(^2P^\circ)3d$	$3d \ ^1F^\circ$	3	$1509210+x$			$2s^2 2p(^2P^\circ)4d$	$4d \ ^3D^\circ$	1 2 3	1846180 1847490	1310
$3d \ ^1P_1$	$2s^2 2p(^2P^\circ)3d$	$3d \ ^1P^\circ$	1	$1510060+x$		$4d \begin{smallmatrix} ^3D_2 \\ ^3D_3 \end{smallmatrix}$					
$3p' \ ^3S_1$	$2s \ 2p^2(^4P)3p$	$3p \ ^3S^\circ$	1	1531270		$4d \ ^1P_1$	$2s^2 2p(^2P^\circ)4d$	$4d \ ^1P^\circ$	1	$1853670+x$	
$3p' \begin{smallmatrix} ^3D_1 \\ ^3D_2 \\ ^3D_3 \end{smallmatrix}$	$2s \ 2p^2(^4P)3p$	$3p \ ^3D^\circ$	1 2 3	1564140 1564840 1566840	700 2000	$4d' \begin{smallmatrix} ^5P_3 \\ ^5P_2 \\ ^5P_1 \end{smallmatrix}$	$2s \ 2p^2(^4P)4d$	$4d \ ^5P$	3 2 1	$1991450+y$ $1992250+y$ $1992760+y$	—800 —510
	$2s \ 2p^2(^4P)3p$	$3p \ ^3P^\circ$	0 1 2	1577760		$4d' \begin{smallmatrix} ^3F_3 \\ ^3F_4 \end{smallmatrix}$	$2s \ 2p^2(^4P)4d$	$4d \ ^3F$	2 3 4	1997710 1999710	2000
$3p' \ ^3P_2$											
$\overline{3s'} \ ^3D$	$2s \ 2p^2(^2D)3s$	$3s' \ ^3D$	1, 2, 3	1585400							
$\overline{3s'} \ ^1D_2$	$2s \ 2p^2(^2D)3s$	$3s' \ ^1D$	2	$1608440+x$			Al IX ($^2P_{3/2}^\circ$)	Limit	-----	2300390	

March 1948.

Al VIII OBSERVED TERMS*

Config. 1s ² +	Observed Terms		
2s ² 2p ²	{ 2p ² 1S 2p ² 3P 2p ² 1D		
2s 2p ³	{ 2p ³ 5S° 2p ³ 3P° 2p ³ 3D° 2p ³ 3S° 2p ³ 1P° 2p ³ 1D°		
	ns (n ≥ 3)	np (n ≥ 3)	nd (n ≥ 3)
2s ² 2p(2P°)nx	{ 3, 4s 3P° 3s 1P°	3p 3S	3d 3P° 3, 4d 3D° 3d 3F° 3, 4d 1P° 3d 1D° 3d 1F°
2s 2p ² (4P)nx	{ 3s 5P 3s 3P	3p 3S° 3p 3P° 3p 3D°	3, 4d 5P 3d 3D 3, 4d 3F 3d 3P
2s 2p ² (2D)nx'	{ 3s' 3D 3s' 1D	3p' 1D° 3p' 1F°	3d' 3S 3d' 3P 3d' 3D 3d' 3F
2s 2p ² (2S)nx''	3s'' 3S		3d'' 3D
2s 2p ² (2P)nx'''	3s''' 3P		3d''' 3P 3d''' 3D 3d''' 3F

*For predicted terms in the spectra of the C I isoelectronic sequence, see Introduction.

Al IX

(B I sequence; 5 electrons)

Z=13

Ground state 1s² 2s² 2p 2P_{1/2}°

2p 2P_{1/2}° 2663340 cm⁻¹

I. P. 330.1 volts

Ferner has extended the preliminary analysis by Söderqvist and now has 74 classified lines in the range between 43 Å and 77 Å. He kindly furnished his manuscript in advance of publication.

No intersystem combinations have been observed, as indicated by *x* in the table, but the absolute values of the doublet and quartet terms are determined from series. The quartet terms are not all connected by observed combinations.

Ferner's unit, 10³ cm⁻¹, has here been changed to cm⁻¹.

REFERENCE

J. Söderqvist, Nova Acta Reg. Soc. Sci. Uppsala [IV] **9**, No. 7, 90 (1934). (C L)
E. Ferner, Ark. Mat. Astr. Fys. (Stockholm) **36A**, No. 1, p. 30 (1948). (I P) (T) (C L)

Al IX

Al IX

Author	Config.	Desig.	J	Level	Interval	Author	Config.	Desig.	J	Level	Interval
$2p$ 2P_1 2P_2	$2s^2(^1S)2p$	$2p$ $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	0 4890	4890	$\overline{3p'}$ 2D_2 2D_3	$2s\ 2p(^1P^\circ)3p$	$3p'$ 2D	$1\frac{1}{2}$ $2\frac{1}{2}$	1875340 1876710	1370
$2p'$ 4P_1 4P_2 4P_3	$2s\ 2p^2$	$2p^2$ 4P	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$146310+x$ $148000+x$ $150490+x$	1690 2490	$\overline{3p'}$ 2P	$2s\ 2p(^1P^\circ)3p$	$3p'$ 2P	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	1878390	
$2p'$ 2D	$2s\ 2p^2$	$2p^2$ 2D	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	259720		$3s''$ 4P_1 4P_2 4P_3	$2p^2(^3P)3s$	$3s''$ 4P	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$1917920+x$ $1918850+x$ $1921100+x$	930 2250
$2p'$ 2S_1	$2s\ 2p^2$	$2p^2$ 2S	$\frac{1}{2}$	332650		$\overline{3d'}$ 2F	$2s\ 2p(^1P^\circ)3d$	$3d'$ $^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	1933050	
$2p'$ 2P_1 2P_2	$2s\ 2p^2$	$2p^2$ 2P	$\frac{1}{2}$ $1\frac{1}{2}$	353960 356950	2990	$\overline{3d'}$ 2D_2 2D_3	$2s\ 2p(^1P^\circ)3d$	$3d'$ $^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	1943380 1943980	600
$2p''$ 4S_2	$2p^3$	$2p^3$ $^4S^\circ$	$1\frac{1}{2}$	$461910+x$		$\overline{3d'}$ 2P	$2s\ 2p(^1P^\circ)3d$	$3d'$ $^2P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	1954710	
$2p''$ 2D_3 2D_2	$2p^3$	$2p^3$ $^2D^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$	519560 519740	-180		$2p^2(^3P)3p$	$3p''$ $^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	$1986800+x$	
$2p''$ 2P_1 2P_2	$2p^3$	$2p^3$ $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	584150 584390	240	$3p''$ 4D_4					
$3s$ 2S_1	$2s^2(^1S)3s$	$3s$ 2S	$\frac{1}{2}$	1501020			$2p^2(^3P)3p$	$3p''$ $^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$1991700+x$	
$3d$ 2D_2 2D_3	$2s^2(^1S)3d$	$3d$ 2D	$1\frac{1}{2}$ $2\frac{1}{2}$	1642140 1642380	240	$3p''$ 4P_3					
$3s'$ 4P_1 4P_2 4P_3	$2s\ 2p(^3P^\circ)3s$	$3s$ $^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$1657690+x$ $1659350+x$ $1662340+x$	1660 2990	$3p''$ 4S_2	$2p^2(^3P)3p$	$3p''$ $^4S^\circ$	$1\frac{1}{2}$	$2017670+x$	
$3s'$ 2P_1 2P_2	$2s\ 2p(^3P^\circ)3s$	$3s$ $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	1690880 1694110	3230	$\overline{3p''}$ 2D	$2p^2(^1D)3p$	$3p'''$ $^2D^\circ$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	2056120	
$3p'$ 2P_1 2P_2	$2s\ 2p(^3P^\circ)3p$	$3p$ 2P	$\frac{1}{2}$ $1\frac{1}{2}$	1720900 1722400	1500	$3d''$ 4P	$2p^2(^3P)3d$	$3d''$ 4P	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	$2065270+x$ $2066350+x$ $2067100+x$	-1080 -750
$3p'$ 2D_2 2D_3	$2s\ 2p(^3P^\circ)3p$	$3p$ 2D	$1\frac{1}{2}$ $2\frac{1}{2}$	1757500 1760970	3470	$4d$ 2D_2 2D_3	$2s^2(^1S)4d$	$4d$ 2D	$1\frac{1}{2}$ $2\frac{1}{2}$	2094020 2094490	470
$3p'$ 2S_1	$2s\ 2p(^3P^\circ)3p$	$3p$ 2S	$\frac{1}{2}$	1780950			$2s\ 2p(^3P^\circ)4d$	$4d$ $^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	$2254250+x$	
$3d'$ $^4D_{12}$ 4D_3 4D_4	$2s\ 2p(^3P^\circ)3d$	$3d$ $^4D^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \\ 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	$1799090+x$ $1799490+x$ $1800980+x$	400 1490	$4d'$ 4D_4					
$3d'$ 2D_2 2D_3	$2s\ 2p(^3P^\circ)3d$	$3d$ $^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	1800460 1800910	450	$4d'$ 4P_3	$2s\ 2p(^3P^\circ)4d$	$4d$ $^4P^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	$2256240+x$	
$\overline{3s'}$ 2P	$2s\ 2p(^1P^\circ)3s$	$3s'$ $^2P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	1807020		$4d'$ 2F_4	$2s\ 2p(^3P^\circ)4d$	$4d$ $^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	2265580	
$3d'$ 4P_3 4P_2 4P_1	$2s\ 2p(^3P^\circ)3d$	$3d$ $^4P^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	$1807490+x$ $1808530+x$ $1809210+x$	-1040 -680	$5d$ 2D_3	$2s^2(^1S)5d$	$5d$ 2D	$1\frac{1}{2}$ $2\frac{1}{2}$	2301150	
$3d'$ 2F_3 2F_4	$2s\ 2p(^3P^\circ)3d$	$3d$ $^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	1831260 1834300	3040	$\overline{4d'}$ 2D_3	$2s\ 2p(^1P^\circ)4d$	$4d'$ $^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	2393860	
$3d'$ 2P_2 2P_1	$2s\ 2p(^3P^\circ)3d$	$3d$ $^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	1840470 1842220	-1750						
							Al x (1S_0)	Limit		2663340	

August 1947.

Al IX OBSERVED TERMS*

Config. $1s^2 +$	Observed Terms		
$2s^2(1S)2p$	$2p \ ^2P^\circ$		
$2s \ 2p^2$	$\left\{ \begin{array}{l} 2p^2 \ ^2S \\ 2p^2 \ ^4P \\ 2p^2 \ ^2P \end{array} \right. \ 2p^2 \ ^2D$		
$2p^3$	$\left\{ \begin{array}{l} 2p^3 \ ^4S^\circ \\ 2p^3 \ ^2P^\circ \\ 2p^3 \ ^2D^\circ \end{array} \right.$		
	$ns \ (n \geq 3)$	$np \ (n \geq 3)$	$nd \ (n \geq 3)$
$2s^2(1S)nx$	$3s \ ^2S$		$3-5d \ ^2D$
$2s \ 2p(^3P^\circ)nx$	$\left\{ \begin{array}{l} 3s \ ^4P^\circ \\ 3s \ ^2P^\circ \end{array} \right.$	$3p \ ^2S \quad 3p \ ^2P \quad 3p \ ^2D$	$\begin{array}{l} 3, 4d \ ^4P^\circ \quad 3, 4d \ ^4D^\circ \\ 3d \ ^2P^\circ \quad 3d \ ^2D^\circ \end{array} \quad 3, 4d \ ^2F^\circ$
$2s \ 2p(^1P^\circ)nx'$	$3s' \ ^2P^\circ$	$3p' \ ^2P \quad 3p' \ ^2D$	$3d' \ ^2P^\circ \quad 3, 4d' \ ^2D^\circ \quad 3d' \ ^2F^\circ$
$2p^2(^3P)nx''$	$3s'' \ ^4P$	$3p'' \ ^4S^\circ \quad 3p'' \ ^4P^\circ \quad 3p'' \ ^4D^\circ$	$3d'' \ ^4P$
$2p^2(^1D)nx'''$		$3p''' \ ^2D^\circ$	

*For predicted terms in the spectra of the Br isoelectronic sequence, see Introduction.

Al X

(Be I sequence; 4 electrons)

$Z=13$

Ground state $1s^2 2s^2 \ ^1S_0$

$2s^2 \ ^1S_0 \ 3215340 \text{ cm}^{-1}$

I. P. 398.5 volts

Ferner has extended the preliminary analysis by Söderqvist and has classified 30 lines in the region between 44 Å and 63 Å. He has kindly furnished his manuscript in advance of publication.

No intersystem combinations have been observed, as indicated by x in the table, but absolute values of the singlet and triplet terms are known from the series.

Ferner's unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

- J. Söderqvist, Nova Acta Reg. Soc. Sci Uppsala [IV] **9**, No. 7, 94 (1934). (T) (C L)
 E. Ferner, Ark. Mat. Astr. Fys. (Stockholm) **36A**, No. 1, p. 27 (1948). (I P) (T) (C L)

Al x

Al x

Author	Config.	Desig.	J	Level	Inter- val	Author	Config.	Desig.	J	Level	Inter- val
2s 1S_0	$2s^2$	$2s^2 \ ^1S$	0	0		3p' 1P_1	$2p(^2P^\circ)3p$	3p 1P	1	2094730	
2p 3P_0 3P_1 3P_2	$2s(^2S)2p$	$2p \ ^3P^\circ$	0 1 2	$154850+x$ $156540+x$ $160200+x$	1690 3660	3p' 3D_1 3D_2 3D_3	$2p(^2P^\circ)3p$	3p 3D	1 2 3	$2101950+x$ $2103560+x$ $2107290+x$	1610 3730
2p 1P_1	$2s(^2S)2p$	$2p \ ^1P^\circ$	1	300400		3p' 3S_1	$2p(^2P^\circ)3p$	3p 3S	1	$2119440+x$	
2p' 3P	$2p^2$	$2p^2 \ ^3P$	0 1 2	$404300+x$ $406270+x$ $409460+x$	1970 3190	3p' 3P_1 3P_2	$2p(^2P^\circ)3p$	3p 3P	0 1 2	$2128300+x$ $2130180+x$	1880
2p' 1D_2	$2p^2$	$2p^2 \ ^1D$	2	448840		3d' 1D_2	$2p(^2P^\circ)3d$	3d $^1D^\circ$	2	2140690	
2p' 1S_0	$2p^2$	$2p^2 \ ^1S$	0	553270		3p' 1D_2	$2p(^2P^\circ)3p$	3p 1D	2	2148320	
3s 3S_1	$2s(^2S)3s$	3s 3S	1	$1855510+x$		3d' 3D_2 3D_3	$2p(^2P^\circ)3d$	3d $^3D^\circ$	1 2 3	$2161630+x$ $2163110+x$	1480
3s 1S_0	$2s(^2S)3s$	3s 1S	0	1884330		3d' 3P_2 3P_1	$2p(^2P^\circ)3d$	3d $^3P^\circ$	2 1 0	$2169960+x$ $2171350+x$	-1390
3p 1P_1	$2s(^2S)3p$	$3p \ ^1P^\circ$	1	1923850		3d' 1F_3	$2p(^2P^\circ)3d$	3d $^1F^\circ$	3	2192060	
3d 3D_1 3D_2 3D_3	$2s(^2S)3d$	3d 3D	1 2 3	$1965560+x$ $1965770+x$ $1966050+x$	210 280	4d 1D_2	$2s(^2S)4d$	4d 1D	2	2527470	
3d 1D_2	$2s(^2S)3d$	3d 1D	2	1992250		4d' 1F_3	$2p(^2P^\circ)4d$	4d $^1F^\circ$	3	2714560	
	$2p(^2P^\circ)3s$	3s $^3P^\circ$	0 1 2	$2056910+x$							
3s' 3P_2											
3s' 1P_1	$2p(^2P^\circ)3s$	3s $^1P^\circ$	1	2090980			Al xI ($^2S_{1/2}$)	-----	Limit	3215340	

August 1947.

Al x OBSERVED TERMS*

Config. $1s^2+$	Observed Terms			
$2s^2$	$2s^2 \ ^1S$			
$2s(^2S)2p$	{ $2p \ ^3P^\circ$ $2p \ ^1P^\circ$			
$2p^2$	{ $2p^2 \ ^3P$ $2p^2 \ ^1S$ $2p^2 \ ^1D$			
	$ns \ (n \geq 3)$	$np \ (n \geq 3)$	$nd \ (n \geq 3)$	
$2s(^2S)nx$	{ $3s \ ^3S$ $3s \ ^1S$	$3p \ ^1P^\circ$	$3d \ ^3D$ $3, 4d \ ^1D$	
$2p(^2P^\circ)nx$	{ $3s \ ^3P^\circ$ $3s \ ^1P^\circ$	$3p \ ^3S$ $3p \ ^3P$ $3p \ ^3D$ $3p \ ^1P$ $3p \ ^1D$	$3d \ ^3P^\circ$ $3d \ ^3D^\circ$ $3d \ ^1D^\circ$	$3, 4d \ ^1F^\circ$

*For predicted terms in the spectra of the Be I isoelectronic sequence, see Introduction.

Al XI

(Li I sequence; 3 electrons)

$Z=13$

Ground state $1s^2 2s\ ^2S_{1/2}$

$2s\ ^2S_{1/2}$ 3564900 cm^{-1}

I. P. 441.9 volts

The analysis is by Ferner, who kindly furnished his manuscript in advance of publication. Seven lines have been classified between 39 Å and 54 Å. Observations of the resonance lines have not been reported. Some of the relative levels have been connected by a study of the behavior of the Rydberg denominators rather than by the Ritz combination principle.

Ferner's unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCE

E. Ferner, Ark. Mat. Astr. Fys. (Stockholm) **36A**, No. 1, p. 25 (1948). (I P) (T) (C L)

Al XI

Config.	Desig.	<i>J</i>	Level	Inter- val
2s	2s 2S	½	0	5920
2p	2p 2P°	½	175900	
		1½	181820	
3s	3s 2S	½	2020460	1750
3p	3p 2P°	½	2068770	
		1½	2070520	
3d	3d 2D	1½	2087980	560
		2½	2088540	
4d	4d 2D	1½	2734140	
		2½		
-----	-----	---	-----	
Al xII (1S ₀)	<i>Limit</i>	---	3564900	

August 1947.

Al XII

(He I sequence; 2 electrons)

 $Z=13$ Ground state $1s^2\ ^1S_0$ $1s^2\ ^1S_0\ 16825000 \pm 3000\ \text{cm}^{-1}$ I. P. 2085.46 ± 0.37 volts

Flemberg has observed the first three members of the singlet series; the lines are in the region between 6 Å and 7 Å. He has calculated absolute term values on the assumption that the P-terms can be represented by a Ritz formula.

The unit adopted by Flemberg, $10^3\ \text{cm}^{-1}$, has here been changed to cm^{-1} .

REFERENCE

H. Flemberg, Ark. Mat. Astr. Fys. (Stockholm) **28A**, No. 18 p. 34 (1942). (I P) (T) (C L)

Al XII

Config.	Desig.	J	Level
$1s^2$	$1s^2\ ^1S$	0	0
$1s\ 2p$	$2p\ ^1P^\circ$	1	12891900
$1s\ 3p$	$3p\ ^1P^\circ$	1	15072700
$1s\ 4p$	$4p\ ^1P^\circ$	1	15838600
-----	-----	---	-----
Al XIII ($^2S_{1/2}$)	<i>Limit</i>	---	16825000

October 1946.

SEE REVISION IN NSRDS-NBS 3, Section 2, November 1967.

SILICON

Si I

14 electrons

 $Z=14$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^2 {}^3P_0$ $3p^2 {}^3P_0$ 65743.00 cm^{-1}

I. P. 8.149 volts

The terms are from Kiess, who has revised and extended the earlier work on analysis. He has published a complete list of classified lines extending from 1565 Å to 12270 Å. His notation has been adopted throughout, except for the following entries, which have been changed for uniformity:

Kiess	Desig.	Kiess	Desig.
$3p {}^3P$	$3p^2 {}^3P$	$3p' {}^3D^\circ$	$3p^3 {}^3D^\circ$
$3p {}^1D$	$3p^2 {}^1D$	x'	1°
$3p {}^1S$	$3p^2 {}^1S$	x''	2°

The singlet and triplet terms are connected by numerous intersystem combinations. No quintet terms have been found.

The Si I sequence invites further study from the theoretical point of view. In Si I the $3d {}^3D^\circ$ term is lower than the $3p^3 {}^3D^\circ$ term. In later members of the sequence the corresponding terms appear in the reverse order.

The extension by Kiess of the laboratory analysis to cover the infrared region has been of special astrophysical importance. The leading lines of Si I are strong in the solar spectrum. Conversely, the solar wave-number separations within the multiplets afford a valuable check on the accuracy of infrared solar wavelengths, provided the Si lines are unblended in the sun. The satisfactory internal agreement within the "solar" Si multiplets has also justified the use of this method to identify solar lines by prediction as unquestionably due to Si, although they have not yet been observed in the laboratory.

REFERENCES

- H. D. Babcock, C. E. Moore and W. P. Hoge, Mt. Wilson Contr. No. 534; *Astroph. J.* **83**, 118 (1936).
 C. C. Kiess, *J. Research Nat. Bur. Std.* **21**, 85, RP1124 (1938). (I P) (T) (C L) (E D)

Si I

Si I

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2\ 3p^2$	$3p^2\ ^3P$	0	0. 00	77. 15 146. 16	$3s^2\ 3p(^2P^\circ)5p$	$5p\ ^3D$	1	56978. 00	39. 26 180. 68
		1	77. 15				2	57017. 26	
		2	223. 31				3	57197. 94	
$3s^2\ 3p^2$	$3p^2\ ^1D$	2	6298. 81		$3s^2\ 3p(^2P^\circ)5p$	$5p\ ^3P$	0	57295. 76	32. 88 139. 54
$3s^2\ 3p^2$	$3p^2\ ^1S$	0	15394. 24				1	57328. 64	
							2	57468. 18	
$3s^2\ 3p(^2P^\circ)4s$	$4s\ ^3P^\circ$	0	39683. 10	77. 10 194. 92	$3s^2\ 3p(^2P^\circ)4d$	$4d\ ^3F^\circ$	2	57372. 44	78. 26 133. 15
		1	39760. 20				3	57450. 70	
		2	39955. 12				4	57583. 85	
$3s^2\ 3p(^2P^\circ)4s$	$4s\ ^1P^\circ$	1	40991. 74		$3s^2\ 3p(^2P^\circ)5p$	$5p\ ^3S$	1	57541. 86	
$3s^2\ 3p(^2P^\circ)3d$	$3d\ ^3D^\circ$	1	45276. 20	17. 40 28. 26	$3s^2\ 3p(^2P^\circ)5p$	$5p\ ^1D$	2	57797. 82	
		2	45293. 60				0	58311. 19	
		3	45321. 86						
$3s^2\ 3p(^2P^\circ)4p$	$4p\ ^1P$	1	47284. 20		$3s^2\ 3p(^2P^\circ)4f$	$4f\ ^1F$	3	58774. 18	
$3s^2\ 3p(^2P^\circ)3d$	$3d\ ^1D^\circ$	2	47351. 50		$3s^2\ 3p(^2P^\circ)4f$	$4f\ ^3F$	2	58775. 44	11. 36 2. 20
$3s^2\ 3p(^2P^\circ)4p$	$4p\ ^3D$	1	48020. 00	82. 38 161. 97			3	58786. 80	
		2	48102. 38				4	58789. 00	
		3	48264. 35						
$3s\ 3p^3$	$3p^3\ ^3D^\circ$	1	48399. 15	178. 45 296. 36	$3s^2\ 3p(^2P^\circ)4d$	$4d\ ^1F^\circ$	3	58893. 28	
		2	48577. 60				3	59035. 15	1. 85
		3	48873. 96						
$3s^2\ 3p(^2P^\circ)4p$	$4p\ ^3P$	0	49028. 17	32. 38 128. 06			4	59037. 00	16. 84
		1	49060. 55				5	59053. 84	
		2	49188. 61						
$3s^2\ 3p(^2P^\circ)4p$	$4p\ ^3S$	1	49399. 66		$3s^2\ 3p(^2P^\circ)5d$	$5d\ ^3D^\circ$	1	59056. 70	-24. 28 86. 09
$3s^2\ 3p(^2P^\circ)4p$	$4p\ ^3S$	1	49399. 66				2	59032. 42	
							3	59118. 51	
$3s^2\ 3p(^2P^\circ)3d$	$3d\ ^3F^\circ$	2	49850. 93	83. 19 137. 76	$3s^2\ 3p(^2P^\circ)4f$	$4f\ ^3D$	3	59109. 75	-81. 09 0. 44
		3	49934. 12				2	59190. 84	
		4	50071. 88				1	59190. 40	
$3s^2\ 3p(^2P^\circ)4p$	$4p\ ^1D$	2	50189. 43			1°	?	59109. 9	
$3s^2\ 3p(^2P^\circ)3d$	$3d\ ^3P^\circ$	2	50499. 44	-66. 51 -36. 20	$3s^2\ 3p(^2P^\circ)4f$	$4f\ ^1D$	2	59110. 91	
		1	50565. 95				?	59132. 5	
		0	50602. 15						
$3s^2\ 3p(^2P^\circ)4p$	$4p\ ^1S$	0	51611. 77		$3s^2\ 3p(^2P^\circ)6s$	$6s\ ^3P^\circ$	0	59220. 76	52. 52 232. 89
$3s^2\ 3p(^2P^\circ)3d$	$3d\ ^1F^\circ$	3	53362. 41				1	59273. 28	
							2	59506. 17	
$3s^2\ 3p(^2P^\circ)3d$	$3d\ ^1P^\circ$	1	53387. 17		$3s^2\ 3p(^2P^\circ)6s$	$6s\ ^1P^\circ$	1	59636. 34	
$3s^2\ 3p(^2P^\circ)4d$	$4d\ ^3D^\circ$	1	54184. 97	20. 15 52. 28	$3s^2\ 3p(^2P^\circ)5d$	$5d\ ^3P^\circ$	2	59917. 35	-92. 75 -32. 38
		2	54205. 12				1	60010. 10	
		3	54257. 40				0	60042. 48	
$3s^2\ 3p(^2P^\circ)5s$	$5s\ ^3P^\circ$	0	54244. 58	69. 32 213. 98	$3s^2\ 3p(^2P^\circ)5d$	$5d\ ^1D^\circ$	2	60299. 92	
		1	54313. 90				2	60645. 49	60. 41
		2	54527. 88						
$3s^2\ 3p(^2P^\circ)5s$	$5s\ ^1P^\circ$	1	54870. 99		$3s^2\ 3p(^2P^\circ)5d$	$5d\ ^3F^\circ$	2	60645. 49	143. 23
							3	60705. 90	
							4	60849. 13	
$3s^2\ 3p(^2P^\circ)5p$	$5p\ ^1P$	1	56425. 1		$3s^2\ 3p(^2P^\circ)5f$	$5f\ ^1D$	2	61303. 28	
$3s^2\ 3p(^2P^\circ)4d$	$4d\ ^1D^\circ$	2	56503. 00		$3s^2\ 3p(^2P^\circ)5f$	$5f\ ^3F$	2	61304. 50	0. 36 1. 71
							3	61304. 86	
$3s^2\ 3p(^2P^\circ)4d$	$4d\ ^3P^\circ$	2	56690. 94	-9. 90 -32. 40			4	61306. 57	
		1	56700. 84						
		0	56733. 24						

Si I—Continued

Si I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2 3p(^2P^\circ)5d$	$5d \ ^1P^\circ$	1	61308. 32		$3s^2 3p(^2P^\circ)6f$	$6f \ ^3F$	2 3 4	62668. 50	
$3s^2 3p(^2P^\circ)6d$	$6d \ ^3D^\circ$	1 2 3	61510. 71 61423. 93 61575. 80	—86. 78 151. 87	$3s^2 3p(^2P^\circ)8s$	$8s \ ^3P^\circ$	0 1 2	62753. 05 62808. 95 62923. 75	55. 90 114. 80
$3s^2 3p(^2P^\circ)5d$	$5d \ ^1F^\circ$	3	61424. 00		$3s^2 3p(^2P^\circ)6d$	$6d \ ^1F^\circ$	3	62802. 00	
$3s^2 3p(^2P^\circ)7s$	$7s \ ^3P^\circ$	0 1 2	61540. 00 61594. 80 61823. 44	54. 80 228. 64	$3s^2 3p(^2P^\circ)7d$	$7d \ ^3D^\circ$	1 2 3	62873. 90 62875. 18 62936. 30	1. 28 61. 12
$3s^2 3p(^2P^\circ)5f$	$5f \ ^3G$	3 4 5	61562. 37 61563. 75	1. 38	$3s^2 3p(^2P^\circ)8s$	$8s \ ^1P^\circ$	1	63130. 60	
$3s^2 3p(^2P^\circ)5f$	$5f \ ^3D$	3 2 1	61597. 12 61597. 90 61598. 60	—0. 78 —0. 70	$3s^2 3p(^2P^\circ)7d$	$7d \ ^3F^\circ$	2 3 4	63257. 61 63353. 70 63580. 63	96. 09 26. 93
$3s^2 3p(^2P^\circ)6d$	$6d \ ^3P^\circ$	2 1 0	61845. 96 61936. 86 61970. 28	—90. 90 —33. 42	$3s^2 3p(^2P^\circ)7d$	$7d \ ^1F^\circ$	3	63642. 55	
$3s^2 3p(^2P^\circ)7s$	$7s \ ^1P^\circ$	1	61881. 50		$3s^2 3p(^2P^\circ)8d$	$8d \ ^3D^\circ$	1 2 3	63758. 35	
$3s^2 3p(^2P^\circ)6d$	$6d \ ^1D^\circ$	2	62155. 20		$3s^2 3p(^2P^\circ)9s$	$9s \ ^1P^\circ$	1	63884. 95	
$3s^2 3p(^2P^\circ)6d$	$6d \ ^3F^\circ$	2 3 4	62349. 27 62376. 68 62534. 46	27. 41 157. 78	-----	-----	-----	-----	-----
					Si II ($^2P_{3/2}^\circ$)	Limit	-----	65743.00	

October 1947.

Si I OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 +$	Observed Terms		
$3s^2 3p^2$	$\left\{ \begin{array}{l} 3p^2 \ ^1S \quad 3p^2 \ ^3P \quad 3p^2 \ ^1D \\ 3p^3 \ ^3D^\circ \end{array} \right.$		
$3s \ 3p^3$			
$3s^2 3p(^2P^\circ)nx$	$ns \ (n \geq 4)$		$np \ (n \geq 4)$
	$\left\{ \begin{array}{l} 4-8s \ ^3P^\circ \\ 4-9s \ ^1P^\circ \end{array} \right.$		$\begin{array}{lll} 4, 5p \ ^3S & 4, 5p \ ^3P & 4, 5p \ ^3D \\ 4, 5p \ ^1S & 4, 5p \ ^1P & 4, 5p \ ^1D \end{array}$
	$nd \ (n \geq 3)$		$nf \ (n \geq 4)$
	$\left\{ \begin{array}{l} 3-6d \ ^3P^\circ \quad 3-8d \ ^3D^\circ \quad 3-7d \ ^3F^\circ \\ 3-5d \ ^1P^\circ \quad 3-6d \ ^1D^\circ \quad 3-7d \ ^1F^\circ \end{array} \right.$		$\begin{array}{lll} 4, 5f \ ^3D & 4-6f \ ^3F & 4, 5g \ ^3G \\ 4, 5f \ ^1D & 4f \ ^1F & \end{array}$

*For predicted terms in the spectra of the Si I isoelectronic sequence, see Introduction.

Si II

(Al I sequence; 13 electrons)

 $Z=14$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^2 P^\circ_{1/2}$ $3p^2 P^\circ_{1/2}$ 131818 cm^{-1}

I. P. 16.34 volts

The doublet terms from the 1S limit in Si III are from Fowler. His values of nf^2F° , $n=7$ to 9, are from his series formula and are indicated by brackets in the table, although they appear to be confirmed by observed combinations with $3p^2^2D$.

The $3p^2^2P$ term has been calculated from the data given by Bowen and Millikan in 1925.

The remaining terms are from Bowen, who pointed out in his 1928 paper that Fowler's term called " x " is $3p^2^2D$; and listed the two lines classified as $3p^2P^\circ-3p^2^2S$. This combination has been used to calculate $3p^2^2S$.

The quartet terms are from Bowen's 1932 paper. No intersystem combinations have been observed and the uncertainty, x , may be considerable. Bowen remarks that the relative positions of the doublet and quartet terms are only approximately determined by assuming that the difference between the terms $4s^2S$ and $4s^4P^\circ$ is equal to that between the terms $3s^2^1S$ and $3p^3P^\circ$ in Si III.

REFERENCES

- A. Fowler, Phil. Trans. Roy. Soc. London [A] **225**, 20 (1925). (I P) (T) (C L)
 I. S. Bowen and R. A. Millikan, Phys. Rev. **26**, 160 (1925). (T) (C L)
 I. S. Bowen, Phys. Rev. **31**, 37 (1928). (C L)
 I. S. Bowen, Phys. Rev. **39**, 13 (1932). (T) (C L)
 C. C. Kiess, J. Research Nat. Bur. Std. **21**, 205, RP1124 (1938). (C L)

Si II

Si II

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3s^2(^1S)3p$	$3p^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	0 287	287	$3s^2(^1S)5f$	$5f^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	113756.60	
$3s3p^2$	$3p^2^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	44080.3+ x 44190.9+ x 44364.4+ x	110.6 173.5	$3s^2(^1S)6p$	$6p^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	114048.7 114057.8	9.1
$3s3p^2$	$3p^2^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	55303.93 55319.84	15.91	$3s^2(^1S)7s$	$7s^2S$	$\frac{1}{2}$	117908.93	
$3s^2(^1S)4s$	$4s^2S$	$\frac{1}{2}$	65495.08		$3s3p(^3P^\circ)4s$	$4s^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	118118.0+ x 118234.0+ x 118433.9+ x	116.0 199.9
$3s3p^2$	$3p^2^2S$	$\frac{1}{2}$	76663.9		$3s^2(^1S)6d$	$6d^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	118516.6	
$3s^2(^1S)3d$	$3d^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	79334.89 79351.49	16.60	$3s^2(^1S)6f$	$6f^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	119307.57	
$3s^2(^1S)4p$	$4p^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	81185.98 81245.98	60.00	$3s^2(^1S)7f$	$7f^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	[122649]	
$3s3p^2$	$3p^2^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	83800 84004	204	$3p^3$	$3p^3^4S^\circ$	$1\frac{1}{2}$	124291.2+ x	
$3s^2(^1S)5s$	$5s^2S$	$\frac{1}{2}$	97966.60		$3s^2(^1S)8f$	$8f^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	[124814]	
$3s^2(^1S)4d$	$4d^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	101017.58 101018.88	1.30	$3s^2(^1S)9f$	$9f^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	[126294]	
$3s^2(^1S)4f$	$4f^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	103552.58		Si III (1S_0)	Limit	-----	131818	
$3s^2(^1S)5p$	$5p^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	103855.29 103879.60	24.31	$3s3p(^3P^\circ)4p$	$4p^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	135272.4+ x 135334.6+ x 135469.4+ x	62.2 134.8
$3s^2(^1S)6s$	$6s^2S$	$\frac{1}{2}$	111178.95		$3s3p(^3P^\circ)4p$	$4p^4S$	$1\frac{1}{2}$	136161.1+ x	
$3s^2(^1S)5d$	$5d^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	112389.2						

Si II OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ +	Observed Terms				
3s ² (1S)3p	3p 2P°				
3s 3p ²	{	3p ² 2S	3p ² 4P 3p ² 2P	3p ² 2D	
3p ³		3p ³ 4S°			
	ns (n ≥ 4)		np (n ≥ 4)	nd (n ≥ 3)	nf (n ≥ 4)
3s ² (1S)nx	4-7s 2S		4-6p 2P°	3-6d 2D	4-6f 2F°
3s 3p(3P°)nx	4s 4P°		4p 4S 4p 4P		

*For predicted terms in the spectra of the Al I isoelectronic sequence, see Introduction.

Si III

SEE REVISION IN NSRDS-NBS 3, Section 1, June 1965.

(Mg I sequence; 12 electrons) Z=14

Ground state 1s² 2s² 2p⁶ 3s² ¹S₀

3s² ¹S₀ 269940.6 cm⁻¹ I. P. 33.46 volts

The analysis is from Bowen, who has extended the earlier work of Fowler, by observations in the ultraviolet. Ninety-six lines have been classified in the interval 566 Å to 5739 Å. One intersystem combination, 3s² ¹S—3p ³P₁^o, is given, but Bowen states that the identification of this line is dubious. He remarks further that “the term values of the singlets and triplets can be independently determined with an accuracy that precludes any large shift in the relative position of the two systems, regardless of this identification.” The irregular doublet law for the isoelectronic sequence through P IV confirms this classification, as has been pointed out by Robinson.

Van Vleck and Whitelaw, by analogy with Al II, using a rigorous series formula, have recalculated the absolute value of 5g ³G as equal to 39831 cm⁻¹ as compared with Fowler’s value 39741 cm⁻¹ and Bowen’s value 39734.0 cm⁻¹.

REFERENCES

R. A. Sawyer und F. Paschen, Ann. der Phys. [IV] **84**, 8 (1927). (T)
I. S. Bowen, Phys. Rev. **39**, 8 (1932). (I P) (T) (C L)
H. A. Robinson, Phys. Rev. **51**, 731 (1937).
J. H. Van Vleck and N. G. Whitelaw, Phys. Rev. **44**, 560 (1933). (T)

Si III

Si III

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
3s ²	3s ² ¹ S	0	0. 0		3s(2S)4d	4d ¹ D	2	204329. 6	
3s(2S)3p	3p ³ P ^o	0	52630	128	3s(2S)5s	5s ³ S	1	206079. 6	
		1	52758	261	3s(2S)5s	5s ¹ S	0	207872. 5	
		2	53019		3s(2S)4f	4f ³ F ^o	2	209486. 7	27. 6
3s(2S)3p	3p ¹ P ^o	1	82883. 0				3	209464. 3	39. 5
3p ²	3p ² ¹ D	2	121946				4	209503. 8	
3s(2S)3d	3d ¹ D	2	122213. 0		3p(2P ^o)3d	3d ³ P ^o	2	216095	—98
3p ²	3p ² ³ P	0	129615	132			1	216193	—62
		1	129747	259			0	216255	
		2	130006		3p(2P ^o)3d	3d ¹ D ^o	1	217290	54
3s(2S)3d	3d ³ D	3	142847. 6	—2. 1			2	217344	51
		2	142849. 7	—2. 0			3	217395	
		1	142851. 7		3p(2P ^o)4s	4s ³ P ^o	0	226305	127
3s(2S)4s	4s ³ S	1	153281. 0				1	226432	295
3p ²	3p ² ¹ S	0	153443. 0				2	226727	
3s(2S)4s	4s ¹ S	0	159068. 4		3s(2S)5g	5g ³ G	3, 4, 5	230206. 6	
3s(2S)4p	4p ³ P ^o	0	175134. 0	33. 0	3s(2S)6g	6g ³ G	3, 4, 5	242379. 0	
		1	175167. 0	73. 2	3p(2P ^o)4p	4p ³ P	0	247776	83
		2	175240. 2				1	247859	214
3s(2S)4p	4p ¹ P ^o	1	176485. 9				2	248073	
3s(2S)4d	4d ³ D	3, 2, 1	201502. 5		Si IV (2S _{1/2})	Limit	-----	269940. 6	

July 1947.

Si III OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶	Observed Terms				
3s ²	3s ² ¹ S				
3s(2S)3p	{ 3p ³ P ^o 3p ¹ P ^o				
3p ²	{ 3p ² ³ P 3p ² ¹ S 3p ² ¹ D				
	ns (n ≥ 4)	np (n ≥ 4)	nd (n ≥ 3)	nf (n ≥ 4)	ng (n ≥ 5)
3s(2S)nx	{ 4, 5s ³ S 4, 5s ¹ S	4p ³ P ^o 4p ¹ P ^o	3, 4d ³ D 3, 4d ¹ D	4f ³ F ^o	5, 6g ³ G
3p(2P ^o)np	4s ³ P ^o	4p ³ P	3d ³ P ^o 3d ³ D ^o		

*For predicted terms in the spectra of the Mg I isoelectronic sequence, see Introduction.

Si IV

(Na I sequence; 11 electrons)

 $Z=14$ Ground state $1s^2 2s^2 2p^6 3s^2 S_{1/2}$ $3s^2 S_{1/2}$ 364097.7 cm^{-1}

I. P. 45.13 volts

The first detailed analysis by Fowler was extended and improved by Edlén and Söderqvist, who observed the spectrum from 815 Å to 4328 Å. The terms have been taken from their paper, extrapolated values being entered in brackets. They estimate the accuracy of the limit as probably within 2 or 3 cm^{-1} . One additional term, $8f^2 F^\circ$, has been taken from Fowler's paper and corrected slightly to agree with the rest.

The observations by McLennan and Shaver extend to the violet limit 458 Å and those by Millikan and Bowen extend to 361 Å.

REFERENCES

- R. A. Millikan and I. S. Bowen, Phys. Rev. **23**, 1 (1924). (C L)
 J. C. McLennan and W. W. Shaver, Trans. Roy. Soc. Canada [3] **18**, Sec III p. 14 (1924). (C L)
 A. Fowler, Phil. Trans. Roy. Soc. (London) [A] **225**, 38 (1925). (T) (C L)
 B. Edlén and J. Söderqvist, Zeit. Phys. **87**, 217 (1933). (I P) (T) (C L)

Si IV

Si IV

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
3s	3s 2S	$\frac{1}{2}$	0.0		6d	6d 2D	$\frac{1}{2}$ $\frac{2}{2}$	313923.4	
3p	3p $^2P^\circ$	$\frac{1}{2}$ $\frac{1}{2}$	71289.6 71749.9	460.3	6f	6f $^2F^\circ$	$\frac{2}{2}$ $\frac{3}{2}$	315231.6	
3d	3d 2D	$\frac{1}{2}$ $\frac{2}{2}$	160376.8		6g	6g 2G	$\left\{ \begin{array}{l} \frac{3}{2} \\ \frac{4}{2} \end{array} \right\}$	315306.8	
4s	4s 2S	$\frac{1}{2}$	193981.5		6h	6h $^2H^\circ$	$\left\{ \begin{array}{l} \frac{4}{2} \\ \frac{5}{2} \end{array} \right\}$	315320.0	
4p	4p $^2P^\circ$	$\frac{1}{2}$ $\frac{1}{2}$	218269.5 218431.3	161.8	7s	7s 2S	$\frac{1}{2}$	318744.5	
4d	4d 2D	$\frac{1}{2}$ $\frac{2}{2}$	250010.6		7p	7p $^2P^\circ$	$\frac{1}{2}$ $\frac{1}{2}$	[322347]	
4f	4f $^2F^\circ$	$\frac{2}{2}$ $\frac{3}{2}$	254129.4 254130.7	1.3	7d	7d 2D	$\frac{1}{2}$ $\frac{2}{2}$	[327369]	
5s	5s 2S	$\frac{1}{2}$	265420.4		7f	7f $^2F^\circ$	$\frac{2}{2}$ $\frac{3}{2}$	328201.5	
5p	5p $^2P^\circ$	$\frac{1}{2}$ $\frac{1}{2}$	276506.5 276581.8	75.3	7g	7g 2G	$\left\{ \begin{array}{l} \frac{3}{2} \\ \frac{4}{2} \end{array} \right\}$	328251.7	
5d	5d 2D	$\frac{1}{2}$ $\frac{2}{2}$	291499.2		7h	7h $^2H^\circ$	$\left\{ \begin{array}{l} \frac{4}{2} \\ \frac{5}{2} \end{array} \right\}$	328262	
5f	5f $^2F^\circ$	$\frac{2}{2}$ $\frac{3}{2}$	293721.0		8f	8f $^2F^\circ$	$\left\{ \begin{array}{l} \frac{2}{2} \\ \frac{3}{2} \end{array} \right\}$	[336619]	
5g	5g 2G	$\left\{ \begin{array}{l} \frac{3}{2} \\ \frac{4}{2} \end{array} \right\}$	293839.7						
6s	6s 2S	$\frac{1}{2}$	299679.6						
6p	6p $^2P^\circ$	$\frac{1}{2}$ $\frac{1}{2}$	305645 305687.6	43	Si v (1S_0)	Limit	-----	364097.7	

Si v

(Ne I sequence; 10 electrons)

 $Z=14$ Ground state $1s^2 2s^2 2p^6 {}^1S_0$ $2p^6 {}^1S_0$ 1345100 cm^{-1}

I. P. 166.73 volts

The analysis is by Ferner, who has extended the early work by Söderqvist. Thirteen lines have been classified in the region 78 Å to 118 Å, as combinations with the ground term.

Ferner's term designations assigned on the assumption of LS -coupling are given under the heading "Author" in the table.

As for Ne I, the jl -coupling notation in the general form suggested by Racah is introduced.

The unit used by Ferner, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

E. Ferner, Ark. Mat. Astr. Fys. (Stockholm) **28A**, No. 4 p. 4 (1941). (I P) (T) (C L).

G. Racah, Phys. Rev. **61**, 537 (L) (1942).

Si v

Si v

Author	Config.	Desig.	J	Level	Author	Config.	Desig.	J	Level
$2p {}^1S$	$2p^6$	$2p^6 {}^1S$	0	0	$4d {}^1P_1$	$2p^5({}^2P_{1/2})4d$	$4d [1\frac{1}{2}]^\circ$	1	1168550
$3s {}^3P_1$	$2p^5({}^2P_{1/2})3s$	$3s [1\frac{1}{2}]^\circ$	2 1	840560	$4d {}^3D_1$	$2p^5({}^2P_{3/2}^\circ)4d$	$4d' [1\frac{1}{2}]^\circ$	1	1174050
$3s {}^1P_1$	$2p^5({}^2P_{3/2}^\circ)3s$	$3s' [\frac{1}{2}]^\circ$	0 1	848460	$5d {}^1P_1$	$2p^5({}^2P_{1/2})5d$	$5d [1\frac{1}{2}]^\circ$	1	1232850
$3d {}^3P_1$	$2p^5({}^2P_{1/2})3d$	$3d [\frac{1}{2}]^\circ$	0 1	1018240	$5d {}^3D_1$	$2p^5({}^2P_{3/2}^\circ)5d$	$5d' [1\frac{1}{2}]^\circ$	1	1237520
$3d {}^1P_1$	"	$3d [1\frac{1}{2}]^\circ$	1	1029410	$6d {}^1P_1$	$2p^5({}^2P_{1/2})6d$	$6d [1\frac{1}{2}]^\circ$	1	1267380
$3d {}^3D_1$	$2p^5({}^2P_{3/2}^\circ)3d$	$3d' [1\frac{1}{2}]^\circ$	1	1036930	$6d {}^3D_1$	$2p^5({}^2P_{3/2}^\circ)6d$	$6d' [1\frac{1}{2}]^\circ$	1	1272090
$4s {}^3P_1$	$2p^5({}^2P_{1/2})4s$	$4s [1\frac{1}{2}]^\circ$	2 1	1100690		Si VI (${}^2P_{1/2}$)	Limit	-----	1345100
$4s {}^1P_1$	$2p^5({}^2P_{3/2}^\circ)4s$	$4s' [\frac{1}{2}]^\circ$	0 1	1105550		Si VI (${}^2P_{3/2}^\circ$)	Limit	-----	1350200

April 1947.

Si v OBSERVED LEVELS*

Config. 1s ² 2s ² +	Observed Terms	
2p ⁶	2p ⁶ 1S	
	ns (n ≥ 3)	nd (n ≥ 3)
2p ⁵ (² P°)nx	{ 3, 4s ³ P° 3, 4s ¹ P°	 3d ³ P° 3-6d ³ D° 3-6d ¹ P°
j <i>l</i> -Coupling Notation		
	Observed Pairs	
	ns (n ≥ 3)	nd (n ≥ 3)
	3, 4s [1½]°	3d [½]° 3-6d [1½]°
2p ⁵ (² P _{1/2})nx		
2p ⁵ (² P _{3/2})nx'	3, 4s' [½]°	3-6d' [1½]°

*For predicted levels in the spectra of the Ne I isoelectronic sequence, see Introduction.

Si vi

(F I sequence; 9 electrons)

Z=14

Ground state 1s² 2s² 2p⁵ ²P_{1/2}°

2p⁵ ²P_{1/2}° **1654800** cm⁻¹

I. P. 205.11 volts

The terms are from Ferner's paper. He has extended the earlier analysis by Söderqvist to include 63 classified lines in the range between 65 Å and 249 Å. All but two of the observed combinations are with the ground term. According to Ferner some of the term assignments are somewhat uncertain. The unit adopted by Ferner, 10³ cm⁻¹, has here been changed to cm⁻¹.

By analogy with related spectra in the isoelectronic sequence Robinson has suggested the following changes in Ferner's term assignments:

Ferner	Robinson	Ferner	Robinson
3d ⁴ F _{2/4}	3d ⁴ P _{2/4}	3d' ² S _{3/4}	3d' ² P _{1/4}
3d ⁴ P _{2/4}	3d ² D _{2/4}	3d' ² P _{1/4}	3d' ² D _{1/4}
3d ² D _{2/4}	3d ² F _{2/4}	3d' ² D _{2/4}	3d' ² F _{2/4}
4d ⁴ F _{2/4}	4d ² D _{2/4}	² D _{1/4}	3d' ² S _{3/4}
	4d ⁴ P _{2/4} *	3d' ² F _{2/4}	3d' ² D _{2/4}
4d ² D _{2/4}	4d ² D _{1/4}		4d' ² S _{3/4} **
² D _{1/4}	4d ² P _{1/4}	4d' ² S _{3/4}	4d' ² P _{1/4}
		4d' ² D _{2/4}	4d' ² D
		² D _{1/4}	4d' ² P _{1/4} ***

*1401250. **1446330. ***1445500.

REFERENCES

E. Ferner, Ark. Mat. Astr. Fys. (Stockholm) **28A**, No. 4, p. 5 (1941). (I P) (T) (C L)
H. A. Robinson, unpublished material (March 1948). (T) (C L)

Si VI

Si VI

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$2s^2 2p^5$	$2p^5 \ ^2P^\circ$	$1\frac{1}{2}$	$\begin{smallmatrix} 0 \\ 5100 \end{smallmatrix}$	-5100	$2s^2 2p^4(^3P)4s$	$4s \ ^2P$	$1\frac{1}{2}$	1329900	
$2s \ 2p^6$	$2p^6 \ ^2S$	$\frac{1}{2}$	406500		$2s^2 2p^4(^1D)4s$	$4s' \ ^2D$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	1371820	
$2s^2 2p^4(^3P)3s$	$3s \ ^4P$	$2\frac{1}{2}$	990460	-3180	$2s \ 2p^5(^3P^\circ)3s$	$3s''' \ ^2P^\circ$	$1\frac{1}{2}$	$\begin{smallmatrix} 1375840 \\ 1378830 \end{smallmatrix}$	-2990
$2s^2 2p^4(^3P)3s$	$3s \ ^2P$	$1\frac{1}{2}$	$\begin{smallmatrix} 1005440 \\ 1009140 \end{smallmatrix}$	-3700	$2s^2 2p^4(^3P)4d$	$4d \ ^4F$	$\begin{smallmatrix} 4\frac{1}{2} \\ 3\frac{1}{2} \\ 2\frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 1399110 \\ 1399450 \end{smallmatrix}$	-340
$2s^2 2p^4(^1D)3s$	$3s' \ ^2D$	$2\frac{1}{2}$	$\begin{smallmatrix} 1041450 \\ 1041500 \end{smallmatrix}$	-50	$2s^2 2p^4(^3P)4d$	$4d \ ^4P$	$\begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 1400880 \\ 1401740 \end{smallmatrix}$	860
$2s^2 2p^4(^1S)3s$	$3s'' \ ^2S$	$\frac{1}{2}$	1094460		$2s^2 2p^4(^3P)4d$	$4d \ ^2P$	$\begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 1402510 \\ 1406330 \end{smallmatrix}$	3820
$2s^2 2p^4(^3P)3d$	$3d \ ^4F$	$\begin{smallmatrix} 4\frac{1}{2} \\ 3\frac{1}{2} \\ 2\frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 1193290 \\ 1194330 \end{smallmatrix}$	-1040	$2s^2 2p^4(^3P)4d$	$4d \ ^2D$	$\begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 1403050 \\ 1404870 \end{smallmatrix}$	1820
$2s^2 2p^4(^3P)3d$	$3d \ ^4P$	$\begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 1194970 \\ 1196040 \\ 1197230 \end{smallmatrix}$	$\begin{smallmatrix} 1070 \\ 1190 \end{smallmatrix}$	$2s^2 2p^4(^1D)4d$	$4d' \ ^2S$	$\frac{1}{2}$	1444340	
$2s^2 2p^4(^3P)3d$	$3d \ ^2P$	$\begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 1200720 \\ 1204740 \end{smallmatrix}$	4020	$2s^2 2p^4(^1D)4d$	$4d' \ ^2D$	$\begin{smallmatrix} 2\frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 1445000 \\ 1445590 \end{smallmatrix}$	-590
$2s^2 2p^4(^3P)3d$	$3d \ ^2D$	$\begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 1201100 \\ 1202960 \end{smallmatrix}$	1860	$2s^2 2p^4(^1D)4d$	$4d' \ ^2P$	$\begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 1445030 \\ 1497100 \end{smallmatrix}$	
$2s^2 2p^4(^1D)3d$	$3d' \ ^2P$	$\begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 1239200 \\ 1242390 \end{smallmatrix}$	3190	$2s^2 2p^4(^1S)4d$	$4d'' \ ^2D$	$\begin{smallmatrix} 2\frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix}$	1497100	
$2s^2 2p^4(^1D)3d$	$3d' \ ^2S$	$\frac{1}{2}$	1241060		$2s^2 2p^4(^3P)5d$	$5d \ ^2D$	$\begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix}$	1497630	
$2s^2 2p^4(^1D)3d$	$3d' \ ^2D$	$\begin{smallmatrix} 2\frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 1242220 \\ 1243860 \end{smallmatrix}$	-1640	$2s^2 2p^4(^1D)5d$	$5d' \ ^2S$	$\frac{1}{2}$	1538370	
$2s^2 2p^4(^1D)3d$	$3d' \ ^2F$	$\begin{smallmatrix} 3\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix}$	1243020		$2s^2 2p(^1D)5d$	$5d' \ ^2P$	$\begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix}$	1538580	
$2s^2 2p^4(^1S)3d$	$3d'' \ ^2D$	$\begin{smallmatrix} 2\frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix}$	$\begin{smallmatrix} 1291510 \\ 1291800 \end{smallmatrix}$	-290	-----			-----	
$2s^2 2p^4(^3P)4s$	$4s \ ^4P$	$\begin{smallmatrix} 2\frac{1}{2} \\ 1\frac{1}{2} \\ \frac{1}{2} \end{smallmatrix}$	1322980		Si VII (3P_2)	Limit	-----	1654800	

March 1948.

Si VI OBSERVED TERMS*

Config. 1s ² +	Observed Terms			
2s ² 2p ⁵	2p ⁵ ²P°			
2s 2p ⁶	2p ⁶ ²S			
	ns (n ≥ 3)		nd (n ≥ 3)	
2s ² 2p ⁴ (³P)nx	{ 3, 4s ⁴P 3, 4s ²P		3, 4d ⁴P 3, 4d ²P	3, 4d ⁴F 3-5d ²D
2s ² 2p ⁴ (¹D)nx'	3, 4s' ²D		3-5d' ²S	3-5d' ²P 3, 4d' ²D 3d' ²F
2s ² 2p ⁴ (¹S)nx''	3s'' ²S		3, 4d'' ²D	
2s 2p ⁵ (³P°)nx'''	3s''' ²P°			

*For predicted terms in the spectra of the F I isoelectronic sequence, see Introduction.

Si VII

(O I sequence; 8 electrons)

 $Z=14$ Ground state $1s^2 2s^2 2p^4 {}^3P_2$ $2p^4 {}^3P_2$ 1988000 cm^{-1}

I. P. 246.41 volts

In 1941 Ferner published an analysis of this spectrum including 71 classified lines—64 in the region between 54 Å and 85 Å and 7 between 217 Å and 278 Å. The present term list is, however, based on later work kindly furnished by him in manuscript form.

Two intersystem combinations have been observed, connecting the triplet and singlet terms.

Ferner's unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

E. Ferner, Ark. Mat. Astr. Fys. (Stockholm) **28A**, No. 4 p. 3 (1941). (T) (C L)

E. Ferner, Ark. Mat. Astr. Fys. (Stockholm) **36A**, No. 1, p. 48 (1948). (I P) (T) (C L)

Si VII

Si VII

Config.	Desig.	J .	Level	Interval	Config.	Desig.	J .	Level	Interval
$2s^2 2p^4$	$2p^4 {}^3P$	2 1 0	0 4030 5570	—4030 —1540	$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^3D^\circ$	3 2 1	1428020 1428090	—70
$2s^2 2p^4$	$2p^4 {}^1D$	2	47000		$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^1P^\circ$	1	1429680	
$2s^2 2p^4$	$2p^4 {}^1S$	0	99780		$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^3P^\circ$	2 1 0	1435460 1436750 1437090	—1290 —340
$2s 2p^5$	$2p^5 {}^3P^\circ$	2 1 0	36 170 366780 368760	—3610 —1980	$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^1D^\circ$	2	1436760	
$2s^2 2p^5$	$2p^5 {}^1P^\circ$	1	506080		$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^3S^\circ$	1	1441230	
$2s^2 2p^3({}^4S^\circ)3s$	$3s {}^3S^\circ$	1	1172470		$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^1F^\circ$	3	1447870	
$2s^2 2p^3({}^2D^\circ)3s$	$3s' {}^3D^\circ$	1, 2, 3	1225150		$2s^2 2p^3({}^2P^\circ)3d$	$3d'' {}^3P^\circ$	0 1 2	1460290 1460860 1461860	570 1000
$2s^2 2p^3({}^2D^\circ)3s$	$3s' {}^1D^\circ$	2	1236320		$2s^2 2p^3({}^2P^\circ)3d$	$3d'' {}^1F^\circ$	4 3 2	1463270 1466490	—3220
$2s^2 2p^3({}^2P^\circ)3s$	$3s'' {}^3P^\circ$	0 1 2	1261610 1262040	430	$2s^2 2p^3({}^2P^\circ)3d$	$3d'' {}^1D^\circ$	2	1466910	
$2s^2 2p^3({}^2P^\circ)3s$	$3s'' {}^1P^\circ$	1	1273170		$2s^2 2p^3({}^2P^\circ)3d$	$3d'' {}^3D^\circ$	3 2 1	1467390 1470050	—2660
$2s^2 2p^3({}^4S^\circ)3d$	$3d {}^3D^\circ$	1, 2 2, 3	1367360 1367560	200	$2s^2 2p^3({}^2P^\circ)3d$	$3d'' {}^1P^\circ$	1	1470490	
$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^3F^\circ$	4 3 2	1426050		$2s^2 2p^3({}^2P^\circ)3d$	$3d'' {}^1F^\circ$	3	1474100	

Si VII—Continued

Si VII—Continued

Config.	Desig.	<i>J.</i>	Level	Inter- val	Config.	Desig.	<i>J.</i>	Level	Inter- val
$2s\ 2p^4(^4P)3s$	$3s'''^3P$	2 1 0	1499430		$2s^2\ 2p^3(^2D^{\circ})4d$	$4d'\ ^1F^{\circ}$	3	1714610	
$2s\ 2p^4(^2D)3s$	$3s^{IV}\ ^3D$	3 2 1	1590930		$2s^2\ 2p^3(^2P^{\circ})4d$	$4d''^3P^{\circ}$	0 1 2	1741130	
$2s^2\ 2p^3(^2D^{\circ})4s$	$4s'\ ^3D^{\circ}$	1, 2, 3	1631160		$2s^2\ 2p^3(^2P^{\circ})4d$	$4d''^3D^{\circ}$	3 2 1	1744440	
$2s^2\ 2p^3(^2D^{\circ})4s$	$4s'\ ^1D^{\circ}$	2	1635820		$2s^2\ 2p^3(^2P^{\circ})4d$	$4d''^1F^{\circ}$	3	1748200	
$2s^2\ 2p^3(^4S^{\circ})4d$	$4d\ ^3D^{\circ}$	1 2, 3	1643740		$2s^2\ 2p^3(^4S^{\circ})5d$	$5d\ ^3D^{\circ}$	1 2, 3	1769040	
$2s^2\ 2p^3(^2P^{\circ})4s$	$4s''^3P^{\circ}$	0 1 2	1669900		$2s^2\ 2p^3(^2D^{\circ})5d$	$5d'\ ^3D^{\circ}$	3, 2 1	1834120	
$2s^2\ 2p^3(^2D^{\circ})4d$	$4d'\ ^3D^{\circ}$	3, 2 1	1707070		$2s^2\ 2p^3(^2D^{\circ})5d$	$5d'\ ^3P^{\circ}$	2 1 0	1836140	
$2s^2\ 2p^3(^2D^{\circ})4d$	$4d'\ ^1P^{\circ}$	1	1707550		$2s\ 2p^4(^4P)4s$	$4s'''^3P$	2 1 0	1887680	
$2s^2\ 2p^3(^2D^{\circ})4d$	$4d'\ ^3P^{\circ}$	2 1 0	1711010						
$2s^2\ 2p^3(^2D^{\circ})4d$	$4d'\ ^3S^{\circ}$	1	1712680		Si VIII ($^4S_{1/2}$)	<i>Limit</i>	-----	1988000	

February 1947.

Si VII OBSERVED TERMS*

Config. $1s^2+$	Observed Terms	
$2s^2\ 2p^4$	$\{2p^4\ ^1S\quad 2p^4\ ^3P\quad 2p^4\ ^1D$	
$2s\ 2p^5$	$\{2p^5\ ^3P^{\circ}\quad 2p^5\ ^1P^{\circ}$	
	$ns\ (n \geq 3)$	$nd\ (n \geq 3)$
$2s^2\ 2p^3(^4S^{\circ})nx$	$3s\ ^3S^{\circ}$	$3-5d\ ^3D^{\circ}$
$2s^2\ 2p^3(^2D^{\circ})nx'$	$\{3, 4s'\ ^3D^{\circ}\quad 3, 4s'\ ^1D^{\circ}$	$3, 4d'\ ^3S^{\circ}\quad 3-5d'\ ^3P^{\circ}\quad 3-5d'\ ^3D^{\circ}\quad 3d'\ ^3F^{\circ}\quad 3, 4d'\ ^1P^{\circ}\quad 3d'\ ^1D^{\circ}\quad 3, 4d'\ ^1F^{\circ}$
$2s^2\ 2p^3(^2P^{\circ})nx''$	$\{3, 4s''^3P^{\circ}\quad 3s''^1P^{\circ}$	$3, 4d''^3P^{\circ}\quad 3, 4d''^3D^{\circ}\quad 3d''^3F^{\circ}\quad 3d''^1P^{\circ}\quad 3d''^1D^{\circ}\quad 3, 4d''^1F^{\circ}$
$2s\ 2p^4(^4P)nx'''$	$3, 4s'''^3P$	
$2s\ 2p^4(^2D)nx^{IV}$	$3s^{IV}\ ^3D$	

*For predicted terms in the spectra of the O I isoelectronic sequence, see Introduction.

Si VIII OBSERVED TERMS*

Config. $1s^2+$	Observed Terms		
$2s^3 2p^3$	$\{ 2p^3 \ ^4S^\circ \quad 2p^3 \ ^2P^\circ \quad 2p^3 \ ^2D^\circ$		
$2s \ 2p^4$	$\{ 2p^4 \ ^2S \quad 2p^4 \ ^4P \quad 2p^4 \ ^2P \quad 2p^4 \ ^2D$		
	$ns \ (n \geq 3)$	$np \ (n \geq 3)$	$nd \ (n \geq 3)$
$2s^2 2p^2(^3P)nx$	$\{ \quad \quad \quad 3s \ ^4P$ $\quad \quad \quad 3, 4s \ ^2P$		$3, 4d \ ^4P \ 3d \ ^4D$ $3d \ ^2P \ 3, 4d \ ^2D \ 3, 4d \ ^2F$
$2s^2 2p^2(^1D)nx'$		$3s' \ ^2D$	$3d' \ ^2S \ 3d' \ ^2P \ 3, 4d'^2D \ 3d' \ ^2F$
$2s \ 2p^3(^3S^\circ)nx'''$	$3s''' \ ^4S^\circ$	$3p''' \ ^4P$	$3d''' \ ^4D^\circ$

*For predicted terms in the spectra of the Ni isoelectronic sequence, see Introduction.

Si IX

(CI sequence; 6 electrons)

$Z=14$

Ground state $1s^2 2s^2 2p^2 \ ^3P_0$

$2p^2 \ ^3P_0$ 2838460 cm^{-1}

I. P. 351.83 volts

The terms have been taken from a manuscript by Ferner who generously submitted his revised analysis in advance of publication. A total of 42 lines have been classified, all but two of which are in the region between 44 Å and 65 Å. No combinations involving the terms $2p^3 \ ^1D^\circ$ and $2p^3 \ ^1P^\circ$ are listed.

The systems of terms of different multiplicity are not connected by intersystem combinations. Their relative positions are estimated by extrapolation along the isoelectronic sequence. The uncertainties, x and y , may be considerable.

Ferner's unit, 10^3 cm^{-1} , has here been converted to cm^{-1} .

REFERENCES

- E. Ferner, Ark. Mat. Astr. Fys. (Stockholm) **28A**, No. 4 p. 6 (1941). (T) (C L)
 E. Ferner, Ark. Mat. Astr. Fys. (Stockholm) **36A**, No. 1, p. 37 (1948). (I P) (T) (C L)

Si IX

Si IX

Author	Config.	Desig.	<i>J</i>	Level	Interval	Author	Config.	Desig.	<i>J</i>	Level	Interval
$2p \ ^3P_0$ $\ ^3P_1$ $\ ^3P_2$	$2s^2 2p^2$	$2p^2 \ ^3P$	0 1 2	0 2590 6460	2590 3870	$3d \ ^1F_3$	$2s^2 2p(^2P^\circ)3d$	$3d \ ^1F^\circ$	3	$1837810+x$	
$2p \ ^1D_2$	$2s^2 2p^2$	$2p^2 \ ^1D$	2	$52960+x$		$3d \ ^1P_1$	$2s^2 2p(^2P^\circ)3d$	$3d \ ^1P^\circ$	1	$1838540+x$	
$2p \ ^1S_0$	$2s^2 2p^2$	$2p^2 \ ^1S$	0	$107780+x$		$3p' \ ^3S_1$	$2s \ 2p^2(^4P)3p$	$3p \ ^3S^\circ$	1	1858590	
$2p' \ ^5S_2$	$2s \ 2p^3$	$2p^3 \ ^5S^\circ$	2	$150010+y$		$3p' \ ^3D_2$ $\ ^3D_3$	$2s \ 2p^2(^4P)3p$	$3p \ ^3D^\circ$	1 2 3	1896170 1899040	2870
$2p' \ ^3D_3$ $\ ^3D_2$ $\ ^3D_1$	$2s \ 2p^3$	$2p^3 \ ^3D^\circ$	3 2 1	292210 292360 292440	-150 -80	$\overline{3s'} \ ^3D$	$2s \ 2p^2(^2D)3s$	$3s' \ ^3D$	1, 2, 3	1917080	
$2p' \ ^3P$	$2s \ 2p^3$	$2p^3 \ ^3P^\circ$	2, 1, 0	344080		$3d' \ ^5P_3$ $\ ^5P_2$ $\ ^5P_1$	$2s \ 2p^2(^4P)3d$	$3d \ ^5P$	3 2 1	$1971270+y$ $1972500+y$ $1973460+y$	-1230 -960
$2p' \ ^1D_2$	$2s \ 2p^3$	$2p^3 \ ^1D^\circ$	2	$440410+x$		$3d' \ ^3P_2$	$2s \ 2p^2(^4P)3d$	$3d \ ^3P$	2 1 0	1973940	
$2p' \ ^3S_1$	$2s \ 2p^3$	$2p^3 \ ^3S^\circ$	1	446980		$3d' \ ^3F_2$ $\ ^3F_3$ $\ ^3F_4$	$2s \ 2p^2(^4P)3d$	$3d \ ^3F$	2 3 4	1985150 1987160 1989830	2010 2670
$2p' \ ^1P_1$	$2s \ 2p^3$	$2p^3 \ ^1P^\circ$	1	$492820+x$		$\overline{3p'} \ ^1F_3$	$2s \ 2p^2(^2D)3p$	$3p' \ ^1F^\circ$	3	$1999930+x$	
$3s \ ^3P_1$ $\ ^3P_2$	$2s^2 2p(^2P^\circ)3s$	$3s \ ^3P^\circ$	0 1 2	1623380 1628550	5170	$\overline{3p'} \ ^1D_2$	$2s \ 2p^2(^2D)3p$	$3p' \ ^1D^\circ$	2	$2009410+x$	
$3s \ ^1P_1$	$2s^2 2p(^2P^\circ)3s$	$3s \ ^1P^\circ$	1	$1640920+x$		$3d' \ ^3D_{3/2}$	$2s \ 2p^2(^4P)3d$	$3d \ ^3D$	1 2, 3	2011690	
$3s' \ ^5P_1$ $\ ^5P_2$ $\ ^5P_3$	$2s \ 2p^2(^4P)3s$	$3s \ ^5P$	1 2 3	$1784260+y$ $1786430+y$ $1789650+y$	2170 3220	$\overline{3d'} \ ^3F$	$2s \ 2p^2(^2D)3d$	$3d' \ ^3F$	2, 3, 4	2084940	
$3d \ ^3F_2$	$2s^2 2p(^3P^\circ)3d$	$3d \ ^3F^\circ$	2 3 4	$1789400+x$		$\overline{3d'} \ ^3D$	$2s \ 2p^2(^2D)3d$	$3d' \ ^3D$	1, 2, 3	2093650	
$3d \ ^1D_2$	$2s^2 2p(^2P^\circ)3d$	$3d \ ^1D^\circ$	2	$1794090+x$		$\overline{3d'} \ ^3F$	$2s \ 2p^2(^2P)3d$	$3d''' \ ^3F$	2, 3, 4	2190790	
$3d \ ^3D_1$ $\ ^3D_2$ $\ ^3D_3$	$2s^2 2p(^2P^\circ)3d$	$3d \ ^3D^\circ$	1 2 3	1808160 1809080 1811480	920 2400	$4d \ ^3D_2$ $\ ^3D_3$	$2s^2 2p(^2P^\circ)4d$	$4d \ ^3D^\circ$	1 2 3	2264270 2266400	2130
$3d \ ^3P_2$ $\ ^3P_1$ $\ ^3P_0$	$2s^2 2p(^2P^\circ)3d$	$3d \ ^3P^\circ$	2 1 0	1815690 1816940 1817670	-1250 -730		Si x ($^2P^\circ_{3/2}$)	Limit	-----	2838460	

March 1948.

Si IX OBSERVED TERMS*

Config. $1s^2+$	Observed Terms		
$2s^2 2p^2$	$\left\{ \begin{array}{cc} 2p^2 \ ^1S & 2p^2 \ ^3P \\ & 2p^2 \ ^1D \end{array} \right.$		
$2s \ 2p^3$	$\left\{ \begin{array}{ccc} 2p^3 \ ^5S^\circ & & \\ 2p^3 \ ^3S^\circ & 2p^3 \ ^3P^\circ & 2p^3 \ ^3D^\circ \\ & 2p^3 \ ^1P^\circ & 2p^3 \ ^1D^\circ \end{array} \right.$		
	$ns \ (n \geq 3)$	$np \ (n \geq 3)$	$nd \ (n \geq 3)$
$2s^2 2p(^2P^\circ)nx$	$\left\{ \begin{array}{c} 3s \ ^3P^\circ \\ 3s \ ^1P^\circ \end{array} \right.$		$\begin{array}{ccc} 3d \ ^3P^\circ & 3, 4d \ ^3D^\circ & 3d \ ^3F^\circ \\ 3d \ ^1P^\circ & 3d \ ^1D^\circ & 3d \ ^1F^\circ \end{array}$
$2s \ 2p^2(^4P)nx$	$\left\{ \begin{array}{c} 3s \ ^5P \end{array} \right.$	$3p \ ^3S^\circ \quad 3p \ ^3D^\circ$	$\begin{array}{ccc} 3d \ ^5P & & \\ 3d \ ^3P & 3d \ ^3D & 3d \ ^3F \end{array}$
$2s \ 2p^2(^2D)nx'$	$\left\{ \begin{array}{c} 3s' \ ^3D \end{array} \right.$	$3p' \ ^1D^\circ \quad 3p' \ ^1F^\circ$	$\begin{array}{ccc} & 3d' \ ^3D & 3d' \ ^3F \end{array}$
$2s \ 2p^2(^2P)nx'''$			$3d''' \ ^3F$

*For predicted terms in the spectra of the C I isoelectronic sequence, see Introduction.

Si x

(B 1 sequence; 5 electrons)

 $Z=14$ Ground state $1s^2 2s^2 2p^2 P_{1/2}^0$ $2p^2 P_{1/2}^0$ 3237400 cm^{-1}

I. P. 401.3 volts

Ferner has classified 29 lines in the range between 47 Å and 57 Å. He has kindly furnished his unpublished manuscript extending the analysis he published in 1941.

No intersystem combinations have been observed, as indicated by x in the table, but the absolute values of the doublet and quartet terms are determined from series. Extrapolated values are in brackets in the table.

The quartet terms are not all connected by observed combinations.

Ferner's unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

E. Ferner, Ark. Mat. Astr. Fys. (Stockholm) **28A**, No. 4, p. 18 (1941). (T) (C L)

E. Ferner, Ark. Mat. Astr. Fys. (Stockholm) **36A**, No. 1, p. 30 (1948). (I P) (T) (C L)

Si x

Si x

Author	Config.	Desig.	J	Level	Interval	Author	Config.	Desig.	J	Level	Interval
$2p^2 P_{1/2}^0$ $2p^2 P_{3/2}^0$	$2s^2(^1S)2p$	$2p^2 P^0$	$\frac{1}{2}$ $\frac{3}{2}$	0 6990	6990	$3p' ^2D_3$	$2s^2 2p(^3P^0)3p$	$3p^2 D$	$\frac{1}{2}$ $\frac{3}{2}$	2110260	
$2p' ^4P_1$ $2p' ^4P_2$ $2p' ^4P_3$	$2s^2 2p^2$	$2p^2 ^4P$	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{3}{2}$	$162060+x$ $164500+x$ $168090+x$	2440 3590	$3d' ^4D_{12}$ $3d' ^4D_3$ $3d' ^4D_4$	$2s^2 2p(^3P^0)3d$	$3d^2 D^0$	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{3}{2}$ $\frac{3}{2}$	$2151950+x$ $2152370+x$ $2154860+x$	420 2490
$2p' ^2D$	$2s^2 2p^2$	$2p^2 ^2D$	$\left\{ \frac{1}{2} \right.$ $\left. \frac{3}{2} \right\}$	287830		$3d' ^2D_2$ $3d' ^2D_3$	$2s^2 2p(^3P^0)3d$	$3d^2 D^0$	$\frac{1}{2}$ $\frac{3}{2}$	2153680 2154440	760
$2p' ^2S_1$	$2s^2 2p^2$	$2p^2 ^2S$	$\frac{1}{2}$	367650		$\overline{3s'} ^2P$	$2s^2 2p(^1P^0)3s$	$3s' ^2P^0$	$\left\{ \frac{1}{2} \right.$ $\left. \frac{1}{2} \right\}$	2158290	
$2p' ^2P_1$ $2p' ^2P_2$	$2s^2 2p^2$	$2p^2 ^2P$	$\frac{1}{2}$ $\frac{3}{2}$	389740 394000	4260	$3d' ^4P_3$	$2s^2 2p(^3P^0)3d$	$3d^2 P^0$	$\frac{3}{2}$ $\frac{1}{2}$ $\frac{1}{2}$	$2161950+x$	
$2p'' ^4S_2$	$2p^2$	$2p^2 ^4S^0$	$\frac{1}{2}$	$510190+x$		$3d' ^2F_3$ $3d' ^2F_4$	$2s^2 2p(^3P^0)3d$	$3d^2 F^0$	$\frac{3}{2}$ $\frac{3}{2}$	2188570 2193140	4570
$2p'' ^2D_3$ $2p'' ^2D_2$	$2p^2$	$2p^2 ^2D^0$	$\frac{3}{2}$ $\frac{1}{2}$	574360 574600	-240	$3d' ^2P_2$ $3d' ^2P_1$	$2s^2 2p(^3P^0)3d$	$3d^2 P^0$	$\frac{1}{2}$ $\frac{1}{2}$	2199190 2201770	-2580
$2p'' ^2P_1$ $2p'' ^2P_2$	$2p^2$	$2p^2 ^2P^0$	$\frac{1}{2}$ $\frac{3}{2}$	$[644560]$ $[644940]$	380	$\overline{3d'} ^2F$	$2s^2 2p(^1P^0)3d$	$3d' ^2F^0$	$\left\{ \frac{2}{2} \right.$ $\left. \frac{3}{2} \right\}$	2299860	
$3d^2 D_2$ $3d^2 D_3$	$2s^2(^1S)3d$	$3d^2 D$	$\frac{1}{2}$ $\frac{3}{2}$	1979260 1979730	470	$\overline{3d'} ^2D_2$ $\overline{3d'} ^2D_3$	$2s^2 2p(^1P^0)3d$	$3d' ^2D^0$	$\frac{1}{2}$ $\frac{3}{2}$	2310230 2311360	1130
$3s' ^4P_1$ $3s' ^4P_2$ $3s' ^4P_3$	$2s^2 2p(^3P^0)3s$	$3s^2 P^0$	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{3}{2}$	$1993860+x$ $1996180+x$ $2000570+x$	2320 4390	$3d'' ^4P_3$ $3d'' ^4P_2$	$2p^2(^3P)3d$	$3d'' ^4P$	$\frac{3}{2}$ $\frac{1}{2}$ $\frac{1}{2}$	$2445320+x$ $2446860+x$	-1540
$3s' ^2P_2$	$2s^2 2p(^3P^0)3s$	$3s^2 P^0$	$\frac{1}{2}$ $\frac{3}{2}$	2035810							
$3p' ^2P_2$	$2s^2 2p(^3P^0)3p$	$3p^2 P$	$\frac{1}{2}$ $\frac{3}{2}$	2066600			Si XI (1S_0)	Limit	-----	3237400	

August 1947.

Si x OBSERVED TERMS*

Config. 1s ² +	Observed Terms		
2s ² (¹ S)2p	2p ² P°		
2s 2p ²	{ 2p ² ² S 2p ² ⁴ P 2p ² ² D 2p ² ² P		
2p ³	{ 2p ³ ⁴ S° 2p ³ ² P° 2p ³ ² D°		
	ns (n ≥ 3)	np (n ≥ 3)	nd (n ≥ 3)
2s ² (¹ S)nx			3d ² D
2s 2p(³ P°)nx	{ 3s ⁴ P° 3s ² P°	3p ² P 3p ² D	3d ⁴ P° 3d ⁴ D° 3d ² F° 3d ² P° 3d ² D°
2s 2p(¹ P°)nx'			3d' ² D° 3d' ² F°
2p ² (³ P)nx''			3d'' ⁴ P

*For predicted terms in the spectra of the B I isoelectronic sequence, see Introduction.

Si XI

(Be I sequence; 4 electrons)

Z=14

Ground state 1s² 2s² ¹S₀

2s² ¹S₀ 3840470 cm⁻¹

I. P. 476.0 volts

Ferner has published a preliminary analysis giving the classifications of 12 lines in the region between 43 Å and 49 Å. He has recently extended the earlier work and generously furnished his revised term list in advance of publication, to be used in compiling the list below. No intersystem combinations have been observed, as indicated by *x* in the table. The unit adopted by Ferner, 10³ cm⁻¹, has here been changed to cm⁻¹.

REFERENCES

E. Ferner, Ark. Mat. Astr. Fys. (Stockholm) **28A**, No. 4 p. 20 (1941). (T) (C L)
E. Ferner, Ark. Mat. Astr. Fys. (Stockholm) **36A**, No. 1, p. 27 (1948). (I P) (T)

Si XI

Si XI

Author	Config.	Desig.	<i>J</i>	Level	Inter- val	Author	Config.	Desig.	<i>J</i>	Level	Inter- val
2s ¹ S ₀	2s ²	2s ² ¹ S	0	0		3d ¹ D ₂	2s(2S)3d	3d ¹ D	2	2361010	
2p ³ P ₀	2s(2S)2p	2p ³ P ^o	0	169140+x	2420		2p(2P ^o)3p	3p ³ D	1		
³ P ₁			1	171560+x	5250	3p' ³ D ₃			2		
³ P ₂			2	176810+x					3	2486810+x	
2p ¹ P ₁	2s(2S)2p	2p ¹ P ^o	1	329400		3d' ¹ D ₂	2p(2P ^o)3d	3d ¹ D ^o	2	2523240	
2p' ³ P ₀	2p ²	2p ² ³ P	0	443020+x	2890	3p' ¹ D ₂	2p(2P ^o)3p	3p ¹ D	2	2532140	
³ P ₁			1	445910+x	4560		2p(2P ^o)3d	3d ³ D ^o	1		
³ P ₂			2	450470+x		3d' ³ D ₂			2	2546810+x	
2p' ¹ D ₂	2p ²	2p ² ¹ D	2	493400		³ D ₃			3	2548970+x	
2p' ¹ S ₀	2p ²	2p ² ¹ S	0	607630		3d' ³ P ₂	2p(2P ^o)3d	3d ³ P ^o	2	2556220+x	
3s ¹ S ₀	2s(2S)3s	3s ¹ S	0	2241480					1		
3p ¹ P ₁	2s(2S)3p	3p ¹ P ^o	1	2285040		3d' ¹ F ₃	2p(2P ^o)3d	3d ¹ F ^o	0		
	2s(2S)3d	3d ³ D	1						3	2581130	
3d ³ D ₂			2	2331390+x	550						
³ D ₃			3	2331940+x			Si XII (2S _{1/2})	Limit	-----	3840470	

August 1947.

Si XI OBSERVED TERMS*

Config. 1s ² +	Observed Terms		
2s ²	2s ² ¹ S		
2s(2S)2p	{ 2p ³ P ^o 2p ¹ P ^o		
2p ²	{ 2p ² ³ P 2p ² ¹ D 2p ² ¹ S		
	<i>ns</i> (<i>n</i> ≥ 3)	<i>np</i> (<i>n</i> ≥ 3)	<i>nd</i> (≥ 3)
2s(2S) <i>nx</i>	{ 3s ¹ S	3p ¹ P ^o	3d ³ D 3d ¹ D
2p(2P ^o) <i>nx</i>	{	3p ³ D 3p ¹ D	3d ³ P ^o 3d ³ D ^o 3d ¹ D ^o 3d ¹ F ^o

*For predicted terms in the spectra of the Be I isoelectronic sequence, see Introduction.

Si XII

(Li I sequence; 3 electrons)

 $Z=14$ Ground state $1s^2 2s^2 S_{1/2}$ $2s^2 S_{1/2}$ 4221460 cm^{-1}

I. P. 523.2 volts

The classifications of three lines in the region 44 Å to 45 Å were published by Ferner in 1941, but no terms were given. His absolute term values based on later work, and kindly furnished in advance of publication, have been used in compiling the present list. Observations of the resonance lines have not been reported.

Ferner's unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

E. Ferner, Ark. Mat. Astr. Fys. (Stockholm) **28A**, No. 4 p. 21 (1941). (C L)

E. Ferner, Ark. Mat. Astr. Fys. (Stockholm) **36A**, No. 1, p. 25 (1948). (I P) (T)

Si XII

Config.	Desig.	J	Level	Interval
$2s$	$2s^2 S$	$\frac{1}{2}$	0	8390
$2p$	$2p^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	191900 200290	
$3s$	$3s^2 S$	$\frac{1}{2}$	2390580	
$3d$	$3d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	2463540 2464530	990
-----	-----	---	-----	
Si XIII ($1S_0$)	<i>Limit</i>	---	4221460	

August 1947.

PHOSPHORUS

P I

15 electrons

 $Z=15$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^3 {}^4S_{1/2}^{\circ}$ $3p^3 {}^4S_{1/2}^{\circ}$ 88560 cm^{-1}

I. P. 11.0 volts

Eleven terms have been found by Kiess, who extended earlier work on this spectrum by making the important observations in the infrared to 10813 Å. Robinson observed the ultra-violet region as far as 1323 Å and was able to extend the analysis.

The present list is taken from Robinson's paper, except for the term $4p {}^2P^{\circ}$, which has been adjusted to fit the observations by Kiess.

Intersystem combinations connecting the doublet and quartet terms have been observed.

There is not complete agreement about the configuration assignments of $3d {}^2P$ and $3p^4 {}^2P$, and those entered in the table are tentative.

REFERENCES

C. C. Kiess, Bur. Std. J. Research **8**, 393, RP425 (1932). (I P) (T) (C L)

H. A. Robinson, Phys. Rev. **49**, 297 (1936). (I P) (T) (C L)

W. F. Meggers, J. Opt. Soc. Am. **36**, 431 (1946). (Summary hfs)

P I

P I

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3s^2 3p^3$	$3p^3 {}^4S^{\circ}$	$1\frac{1}{2}$	0. 0		$3s^2 3p^2({}^3P)4p$	$4p {}^2P^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$	67971. 1 68088. 3	117. 2
$3s^2 3p^3$	$3p^3 {}^2D^{\circ}$	$1\frac{1}{2}$ $2\frac{1}{2}$	11361. 7 11376. 5	14. 8	$3s^2 3p^2({}^3P)4p$	$4p {}^2S^{\circ}$	$\frac{1}{2}$	68473. 2	
$3s^2 3p^3$	$3p^3 {}^2P^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$	18722. 4 18748. 1	25. 7	$3s^2 3p^2({}^3P)3d$	$3d {}^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	70391. 3 70690. 0	298. 7
$3s^2 3p^2({}^3P)4s$	$4s {}^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	55939. 23 56090. 59 56339. 68	151. 36 249. 09	$3s^2 3p^2({}^3P)3d$	$3d {}^4D$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	70637. 5 70778. 6	141. 1
$3s^2 3p^2({}^3P)4s$	$4s {}^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	57876. 8 58174. 4	297. 6	$3s 3p^4$	$3p^4 {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	71168. 3 71202. 6	-34. 3
$3s 3p^4$	$3p^4 {}^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	59533. 4 59713. 6 59818. 6	-180. 2 -105. 0	$3s^2 3p^2({}^3P)3d$	$3d {}^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	72386. 6 72494. 6 72571. 4	-108. 0 -76. 8
$3s^2 3p^2({}^1D)4s$	$4s' {}^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	65156. 6		$3s^2 3p^2({}^3P)3d$	$3d {}^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	72741. 9 72883. 5	141. 6
$3s^2 3p^2({}^3P)4p$	$4p {}^4D^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	65373. 6 65450. 2 65585. 1 65787. 3	76. 6 134. 9 202. 2	$3s 3p^4$	$3p^4 {}^2S$	$\frac{1}{2}$	72943. 3	
$3s^2 3p^2({}^3P)4p$	$4p {}^4P^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	66343. 4 66360. 2 66544. 1	16. 8 183. 9	$3s^2 3p^2({}^3P)3d$	$3d {}^2D?$	$1\frac{1}{2}$ $2\frac{1}{2}$	73248. 1	
$3s^2 3p^2({}^3P)4p$	$4p {}^2D^{\circ}$	$1\frac{1}{2}$ $2\frac{1}{2}$	66813. 1 66870. 2?	57. 1	$3s^2 3p^2({}^3P)5s$	$5s {}^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	75064. 6? 75211. 3? 75533. 4?	146. 7 322. 1
$3s^2 3p^2({}^3P)4p$	$4p {}^4S^{\circ}$	$1\frac{1}{2}$	66834. 5		-----				
$3s 3p^4$	$3p^4 {}^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	67908. 6 68126. 2	-217. 6	P II (3P_0)	Limit	-----	88560	

PI OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 +$	Observed Terms								
$3s^2 3p^3$	$\left\{ \begin{array}{lll} 3p^3 \ ^4S^\circ & 3p^3 \ ^2P^\circ & 3p^3 \ ^2D^\circ \end{array} \right.$								
$3s \ 3p^4$	$\left\{ \begin{array}{lll} 3p^4 \ ^4S & 3p^4 \ ^4P & \\ 3p^4 \ ^2S & 3p^4 \ ^2P & 3p^4 \ ^2D \end{array} \right.$								
	$ns \ (n \geq 4)$			$np \ (n \geq 4)$			$nd \ (n \geq 3)$		
$3s^2 3p^2(^3P)nx$	$\left\{ \begin{array}{ll} 4, 5s \ ^4P & \\ 4s \ ^2P & \end{array} \right.$			$4p \ ^4S^\circ$ $4p \ ^2S^\circ$	$4p \ ^4P^\circ$ $4p \ ^2P^\circ$	$4p \ ^4D^\circ$ $4p \ ^2D^\circ$	$3d \ ^4P$ $3d \ ^2P$	$3d \ ^4D$ $3d \ ^2D?$	$3d \ ^2F$
$3s^2 3p^2(^1D)nx'$	$4s' \ ^2D$								

*For predicted terms in the spectra of the P I isoelectronic sequence, see Introduction.

P II

(Si I sequence; 14 electrons)

 $Z=15$

Ground state $1s^2 2s^2 2p^6 3s^2 3p^2 \text{ } ^3\text{P}_0$

$$3p^2\ ^3P_0\ 158550.0\ \text{cm}^{-1}$$

I. P. 19.65 volts

The terms are mostly from the 1936 paper by Robinson, who has revised and extended the earlier analysis by Bowen. The singlet and triplet terms are well connected by inter-system combinations.

In his later paper Robinson adds two quintet terms, and makes a few corrections to his earlier list which have been incorporated here. The quintet terms are not connected by observation with the rest, as indicated by the uncertainty x and brackets denoting that the relative position of $3p^3\ ^5S^\circ$ is estimated.

REFERENCES

- I. S. Bowen, Phys. Rev. **29**, 510 (1927). (T) (C L)
S. Tolansky, Zeit. Pkys. **74**, 336 (1932). (hfs)
H. A. Robinson, Phys. Rev. **49**, 297 (1936). (I P) (T) (C L)
H. A. Robinson, Phys. Rev. **51**, 726 (1937). (T)

P II

P II

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2 3p^2$	$3p^2 {}^3P$	0 1 2	0. 0 166. 6 470. 3	166. 6 303. 7	$3s^2 3p({}^2P^\circ)4d$	$4d {}^3D^\circ$	3 2 1	127333. 6 127890. 2 127935. 7	-556. 6 -45. 5
$3s^2 3p^2$	$3p^2 {}^1D$	2	8882. 6		$3s^2 3p({}^2P^\circ)4d$	$4d {}^3P^\circ$	0 1 2	127368. 7 127601. 2 127951. 1	232. 5 349. 9
$3s^2 3p^2$	$3p^2 {}^1S$	0	21576. 4		$3s^2 3p({}^2P^\circ)4d$	$4d {}^1D^\circ$	2	129612. 0	
$3s 3p^3$	$3p^3 {}^5S^\circ$	2	[52450. 0]+ <i>x</i>		$3s^2 3p({}^2P^\circ)5p$	1 ($5p {}^3S?$)	1	129625. 5?	
$3s 3p^3$	$3p^3 {}^3D^\circ$	1 2 3	65251. 8 65272. 9 65307. 7	21. 1 34. 8		2	2	130239. 6	
$3s 3p^3$	$3p^3 {}^3P^\circ$	2 1 0	76764. 9 76813. 2 76824. 4	-48. 3 -11. 2	$3s^2 3p({}^2P^\circ)5p$	3	1, 2	130826. 2	
$3s 3p^3$	$3p^3 {}^1D^\circ$	2	77710. 8			4 ($5p {}^1D?$)	2	130913. 9	
$3s^2 3p({}^2P^\circ)4s$	$4s {}^3P^\circ$	0 1 2	86599. 0 86745. 1 87126. 1	146. 1 381. 0		5	2	130949. 6	
$3s^2 3p({}^2P^\circ)4s$	$4s {}^1P^\circ$	1	88893. 5		$3s^2 3p({}^2P^\circ)5p$	6 ($5p {}^1P?$)	1	130970. 0	
$3s 3p^3$	$3p^3 {}^1P^\circ$	1	102798. 4			7	2	131320. 5	
$3s^2 3p({}^2P^\circ)4p$	$4p {}^3D$	1 2 3	103166. 7 103340. 2 103668. 9	173. 5 328. 7		8	1, 2	131601. 9	
$3s^2 3p({}^2P^\circ)3d$	$3d {}^3P^\circ$	2 1 0	103632. 3 103755. 4 104219?	-123. 1 -464	$3s^2 3p({}^2P^\circ)4d$	$4d {}^1P^\circ$	1	131729. 1	
$3s^2 3p({}^2P^\circ)3d$	$3d {}^3D^\circ$	1 2 3	103935. 8 104053. 2 104101. 4	117. 4 48. 2	$3s^2 3p({}^2P^\circ)4d$	$4d {}^1F^\circ$	3	131764. 4	
$3s^2 3p({}^2P^\circ)4p$	$4p {}^3P$	0 1 2	105225. 5 105303. 6 105550. 9	78. 1 247. 3	$3s^2 3p({}^2P^\circ)4f$	11 ($4f {}^1D?$)	2	132082. 4	
$3s^2 3p({}^2P^\circ)3d$	$3d {}^1D^\circ$	2	105963. 1			12	2, 3	132134. 1	
$3s^2 3p({}^2P^\circ)4p$	$4p {}^3S$	1	106002. 5			13	2	132163. 6	
$3s^2 3p({}^2P^\circ)4p$	$4p {}^1D$	2	107924. 2			14	2	132206. 9	
$3s^2 3p({}^2P^\circ)3d$	$3d {}^1P^\circ$	1	108371. 8		$3s^2 3p({}^2P^\circ)4f$	15 ($4f {}^1F?$)	3	132236. 0	
$3s^2 3p({}^2P^\circ)4p$	$4p {}^1P$	1	108417. 4			16	2, 3	132354. 7	
$3s 3p^3$	$3p^3 {}^3S^\circ$	1	110254. 9			17	1	132371. 2	
	2°	2, 3	110456. 9?		$3s^2 3p({}^2P^\circ)5p$	19 ($5p {}^1S?$)	0, 1	132641. 5?	
$3s^2 3p({}^2P^\circ)4p$	$4p {}^1S$	0	111114. 8			20	1	133418. 8?	
$3s^2 3p({}^2P^\circ)5s$	$5s {}^3P^\circ$	0 1 2	123345. 4 123456. 7 123892. 0	111. 3 435. 3	$3s^2 3p({}^2P^\circ)6s$	$6s {}^3P^\circ$	0 1 2	137433 137486 138000	53 514
$3s^2 3p({}^2P^\circ)5s$	$5s {}^1P^\circ$	1	124433. 8		$3s^2 3p({}^2P^\circ)6s$	$6s {}^1P^\circ$	1	138058. 4	
$3s^2 3p({}^2P^\circ)4d$	$4d {}^3F^\circ$	2 3 4	124955. 9 125130. 6 125392. 7	174. 7 262. 1	$3s^2 3p({}^2P^\circ)5d$	$5d {}^3P^\circ$	0 1 2	139091. 9	
					$3s^2 3p({}^2P^\circ)6d$	$6d {}^3P^\circ$	0 1 2	145519. 8	
					P III (${}^2P_{3/2}^\circ$)	Limit	-----	158550. 0	
					$3s 3p({}^4P)3d$	$3d {}^5P$	3 2 1	160018. 2+ <i>x</i> 160144. 7+ <i>x</i> 160235. 2+ <i>x</i>	-126. 5 -90. 5

P II OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 +$	Observed Terms		
$3s^2 3p^2$	$\left\{ \begin{array}{ccc} 3p^2 \ ^1S & 3p^2 \ ^3P & 3p^2 \ ^1D \end{array} \right.$		
$3s \ 3p^3$	$\left\{ \begin{array}{ccc} 3p^3 \ ^5S^\circ & & \\ 3p^3 \ ^3S^\circ & 3p^3 \ ^3P^\circ & 3p^3 \ ^3D^\circ \\ & 3p^3 \ ^1P^\circ & 3p^3 \ ^1D^\circ \end{array} \right.$		
	$ns \ (n \geq 4)$	$np \ (n \geq 4)$	$nd \ (n \geq 3)$
$3s^2 3p(^2P^\circ)nx$	$\left\{ \begin{array}{ccc} 4-6s \ ^3P^\circ & & \\ 4-6s \ ^1P^\circ & & \end{array} \right.$	$\begin{array}{ccc} 4p \ ^3S & 4p \ ^3P & 4p \ ^3D \\ 4p \ ^1S & 4p \ ^1P & 4p \ ^1D \end{array}$	$\begin{array}{ccc} 3-6d \ ^3P^\circ & 3, 4d \ ^3D^\circ & 4d \ ^3F^\circ \\ 3, 4d \ ^1P^\circ & 3, 4d \ ^1D^\circ & 4d \ ^1F^\circ \end{array}$
$3s \ 3p^2(^4P)nx$			$3d \ ^5P$

*For predicted terms in the spectra of the Si I isoelectronic sequence, see Introduction.

P III

(Al I sequence; 13 electrons)

$Z=15$

Ground state $1s^2 2s^2 2p^6 3s^2 3p \ ^2P_{1/2}^\circ$

$3p \ ^2P_{1/2}^\circ \ 243290.0 \text{ cm}^{-1}$

I. P. 30.156 ± 0.003 volts

The terms have been taken from Robinson, who has revised and extended the earlier work on analysis. An evident misprint has been corrected here, i. e., the absolute term values of $4f \ ^4D$ should have been printed as negative.

Robinson has classified two lines as the intersystem combination $3p \ ^2P^\circ - 3p^2 \ ^4P$. He remarks that these must be considered as tentative classifications, but that they are consistent with the analogous transition in Al I.

REFERENCES

- R. A. Millikan and I. S. Bowen, Phys. Rev. **25**, 600 (1925). (T) (C L)
 I. S. Bowen, Phys. Rev. **39**, 13 (1932). (T) (C L)
 H. A. Robinson, Phys. Rev. **51**, 726 (1937). (I P) (T) (C L)

P III

P III

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2(^1S)3p$	$3p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	0. 0 559. 6	559. 6	$3s\ 3p(^3P^\circ)4s$	$4s\ ^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	184453. 4 184639. 3 185045. 2	185. 9 405. 9
$3s\ 3p^2$	$3p^2\ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	56919. 3 57125. 8 57454. 5	206. 5 328. 7	$3s\ 3p(^3P^\circ)3d$	$^21^\circ$	$1\frac{1}{2}$	184854. 1	
$3s\ 3p^2$	$3p^2\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	74915. 1 74944. 6	29. 5	$3s\ 3p(^3P^\circ)4s$	$4s\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	186920. 7	
$3s\ 3p^2$	$3p^3\ ^2S$	$\frac{1}{2}$	100201. 2		$3s^2(^1S)5p$	$5p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	191639. 5	
$3s\ 3p^2$	$3p^2\ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	109035. 7 109409. 7	374. 0	$3s^2(^1S)5d$	$5d\ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	200442. 8	
$3s^2(^1S)3d$	$3d\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	116873. 6 116884. 9	11. 3	$3s^2(^1S)6s$	$6s\ ^2S$	$\frac{1}{2}$	201103. 4	
$3s^2(^1S)4s$	$4s\ ^2S$	$\frac{1}{2}$	117834. 5		$3s^2(^1S)5f$	$5f\ ^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	202906. 4	
$3s^2(^1S)4p$	$4p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	141375. 7 141512. 8	137. 1	$3s^2(^1S)5g$	$5g\ ^2G$	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 4\frac{1}{2} \end{array} \right\}$	203782. 7	
$3p^3$	$3p^3\ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	147322. 4 147384. 3	61. 9	$3s\ 3p(^3P^\circ)4p$	$4p\ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	209938. 9 210055. 8 210306. 1	116. 9 250. 3
$3p^3$	$3p^3\ ^4S^\circ$	$1\frac{1}{2}$	159714. 6		$3s\ 3p(^3P^\circ)4p$	$4p\ ^4S$	$1\frac{1}{2}$	211339. 4	
$3p^3$	$3p^3\ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	170107. 2 170167. 0	-59. 8	$3s^2(^1S)6d$	$6d\ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	213982. 8	
$3s^2(^1S)4d$	$4d\ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	172429. 2		$3s^2(^1S)6f$	$6f\ ^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	215402. 0	
$3s\ 3p(^3P^\circ)3d$	$3d\ ^4P^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	173813. 4 173988. 4 174106. 2	-175. 0 -117. 8	$3s^2(^1S)6g$	$6g\ ^2G$	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 4\frac{1}{2} \end{array} \right\}$	215863. 2	
$3s\ 3p(^3P^\circ)3d$	$3d\ ^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	175260. 8 175314. 1 175376. 6 175427. 2	53. 3 62. 5 50. 6	$3s^2(^1S)7g$	$7g\ ^2G$	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 4\frac{1}{2} \end{array} \right\}$	223131. 0	
$3s^2(^1S)5s$	$5s\ ^2S$	$\frac{1}{2}$	176041. 0		P IV (1S_0)	<i>Limit</i>	-----	243290. 0	
$3s^2(^1S)4f$	$4f\ ^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	178653. 2		$3s\ 3p(^3P^\circ)4f$	$4f\ ^4D$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	248168. 4 248199. 4 248228. 4 248265. 5	-31. 0 -29. 0 -37. 1

September 1947.

P III OBSERVED TERMS*

Config. $1s^2\ 2s^2\ 2p^6+$	Observed Terms				
$3s^2(^1S)3p$	$3p\ ^2P^\circ$				
$3s\ 3p^2$	$\left\{ \begin{array}{lll} 3p^2\ ^2S & 3p^2\ ^4P & 3p^2\ ^2D \\ & 3p^2\ ^2P & \end{array} \right.$				
$3p^3$	$\left\{ \begin{array}{lll} 3p^3\ ^4S^\circ & 3p^3\ ^2P^\circ & 3p^3\ ^2D^\circ \end{array} \right.$				
	$ns\ (n \geq 4)$	$np\ (n \geq 4)$	$nd\ (n \geq 3)$	$nf\ (n \geq 4)$	$ng\ (n \geq 5)$
$3s^2(^1S)nx$	$4-6s\ ^2S$	$4-5p\ ^2P^\circ$	$3-6d\ ^2D$	$4-6f\ ^2F^\circ$	$5-7g\ ^2G$
$3s\ 3p(^3P^\circ)nx$	$\left\{ \begin{array}{l} 4s\ ^4P^\circ \\ 4s\ ^2P^\circ \end{array} \right.$	$4p\ ^4S\quad 4p\ ^4P$	$3d\ ^4P^\circ\quad 3d\ ^4D^\circ$	$4f\ ^4D^\circ$	

*For predicted terms in the spectra of the Al I isoelectronic sequence, see Introduction.

P IV

(Mg I sequence; 12 electrons)

 $Z=16$ Ground state $1s^2 2s^2 2p^6 3s^2 {}^1S_0$ $3s^2 {}^1S_0$ 414312.4 cm^{-1} I. P. 51.354 ± 0.013 volts

The analysis published by Bowen in 1932 has been extended by Robinson to include a total of 105 classified lines in the range from 283 Å to 4291 Å.

Intersystem combinations connecting the singlet and triplet terms have been observed. Robinson remarks that the observed combination $3s^2 {}^1S_0 - 3p {}^3P_1^o$ obeys the irregular doublet law very well.

REFERENCES

I. S. Bowen, Phys. Rev. **39**, 10 (1932). (T) (C L)H. A. Robinson, Phys. Rev. **51**, 727 (1937). (I P) (T) (C L)

P IV

P IV

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3s^2$	$3s^2 {}^1S$	0	0. 0		$3s({}^2S)4d$	$4d {}^1D$	2	296757. 8	
$3s({}^2S)3p$	$3p {}^3P^o$	0 1 2	67911. 6 68139. 0 68607. 4	227. 4 468. 4	$3p({}^2P^o)3d$	$3d {}^1P^o$	1	298327	
$3s({}^2S)3p$	$3p {}^1P^o$	1	105189. 9		$3s({}^2S)4f$	$4f {}^3F^o$	2 3 4	303115 303350 303659	235 309
$3s({}^2S)3d$	$3d {}^1D$	2	158138. 2		$3s({}^2S)5s$	$5s {}^3S$	1	309102. 4	
$3p^2$	$3p^2 {}^3P$	0 1 2	164935 165178 165646	243 468	$3p({}^2P^o)4s$	$4s {}^1P^o$	1	313078	
$3p^2$	$3p^2 {}^1D$	2	166144		$3s({}^2S)5s$	$5s {}^1S$	0	316627. 0	
$3s({}^2S)3d$	$3d {}^3D$	3, 2, 1	189389. 0		$3p({}^2P^o)4s$	$4s {}^3P^o$	0 1 2	317662 317948 318353	286 405
$3p^2$	$3p^2 {}^1S$	0	194588. 5		$3s({}^2S)5p$	$5p {}^3P^o$	0 1 2	320053 320126	73
$3s({}^2S)4s$	$4s {}^3S$	1	226888. 6		$3s({}^2S)5p$	$5p {}^1P^o$	1	320063. 5	
$3s({}^2S)4s$	$4s {}^1S$	0	233995. 0		$3s({}^2S)5d$	$5d {}^3D$	3 2 1	339635. 5 339639. 3 339642. 1	-3. 8 -2. 8
$3s({}^2S)4p$	$4p {}^3P^o$	0 1 2	256544. 1 256602. 7 256751. 3	58. 6 148. 6	$3s({}^2S)5d$	$5d {}^1D$	2	341004. 8?	
$3s({}^2S)4p$	$4p {}^1P^o$	1	257520. 2		$3s({}^2S)5f$	$5f {}^3F^o$	2 3 4	343309 343590	281
$3p({}^2P^o)3d$	$3d {}^1F^o$	3	276270?		$3s({}^2S)5g$	$5g {}^3G$	3, 4, 5	343688	
$3p({}^2P^o)3d$	$3d {}^1D^o$	2	276325?		$3s({}^2S)6s$	$6s {}^3S$	1	346672	
$3p({}^2P^o)3d$	$3d {}^3P^o$	2 1 0	281011 281251 281391	-240 -140	$3s({}^2S)6p$	$6p {}^1P^o$	1	352125?	
$3p({}^2P^o)3d$	$3d {}^3D^o$	1 2 3	283142 283239 283321	97 82					
$3s({}^2S)4d$	$4d {}^3D$	1 2 3	293233. 5 293238. 9 293246. 6	5. 4 7. 7	P v (${}^2S_{1/2}$)	Limit	-----	414312. 4	

July 1947.

P IV OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 +$	Observed Terms				
$3s^2$	$3s^2 \ ^1S$				
$3s(^2S)3p$	{ $3p \ ^3P^\circ$ $3p \ ^1P^\circ$				
$3p^2$	{ $3p^2 \ ^1S$ $3p^2 \ ^3P$ $3p^2 \ ^1D$				
	$ns \ (n \geq 4)$	$np \ (n \geq 4)$	$nd \ (n \geq 3)$	$nf \ (n \geq 4)$	$ng \ (n \geq 5)$
$3s(^2S)nx$	{ $4-6s \ ^3S$ $4, 5s \ ^1S$	$4, 5p \ ^3P^\circ$ $4-6p \ ^1P^\circ$	$3-5d \ ^3D$ $3-5d \ ^1D$	$4, 5f \ ^3F^\circ$	$5g \ ^3G$
$3p(^2P^\circ)nx$	{ $4s \ ^3P^\circ$ $4s \ ^1P^\circ$		$3d \ ^3P^\circ$ $3d \ ^3D^\circ$ $3d \ ^1P^\circ$ $3d \ ^1D^\circ$ $3d \ ^1F^\circ$		

* For predicted terms in the spectra of the Mg I isoelectronic sequence, see Introduction.

P v

(Na I sequence; 11 electrons)

$Z=15$

Ground state $1s^2 2s^2 2p^6 3s \ ^2S_{1/2}$

$3s \ ^2S_{1/2} \ 524462.9 \text{ cm}^{-1}$

I. P. 65.007 ± 0.003 volts

The analysis is from Robinson who has extended the earlier work by Bowen and Millikan. The total number of classified lines is 38, of which 31 are in the range between 210 Å and 1610 Å. The absolute value of $6h \ ^2H^\circ$ was extrapolated along the Na I isoelectronic sequence.

REFERENCE

H. A. Robinson, Phys. Rev. **51**, 732 (1937). (I P) (T) (C L)

P v

P v

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3s$	$3s \ ^2S$	$\frac{1}{2}$	0.0		$6s$	$6s \ ^2S$	$\frac{1}{2}$	427157	
$3p$	$3p \ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	88651.7 89446.3	794.6	$6p$	$6p \ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	435100.4	
$3d$	$3d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	204197.1 204208.3	11.2	$6d$	$6d \ ^2D$	{ $1\frac{1}{2}$ $2\frac{1}{2}$ }	445814	
$4s$	$4s \ ^2S$	$\frac{1}{2}$	272961.1		$6f$	$6f \ ^2F^\circ$	{ $2\frac{1}{2}$ $3\frac{1}{2}$ }	448061.7	
$4p$	$4p \ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	304161.3 304445.3	284.0	$6g$	$6g \ ^2G$	{ $3\frac{1}{2}$ $4\frac{1}{2}$ }	448216.8	
$4d$	$4d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	345398.4 345403.3	4.9	$6h$	$6h \ ^2H^\circ$	{ $4\frac{1}{2}$ $5\frac{1}{2}$ }	448247.4	
$4f$	$4f \ ^2F^\circ$	{ $2\frac{1}{2}$ $3\frac{1}{2}$ }	352595.3		$7s$	$7s \ ^2S$	$\frac{1}{2}$	455573	
$5s$	$5s \ ^2S$	$\frac{1}{2}$	376639.2		$7p$	$7p \ ^2P^\circ$	{ $\frac{1}{2}$ $1\frac{1}{2}$ }	460363	
$5p$	$5p \ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	391101.7 391242.4	140.7	$7d$	$7d \ ^2D$	{ $1\frac{1}{2}$ $2\frac{1}{2}$ }	466893	
$5d$	$5d \ ^2D$	{ $1\frac{1}{2}$ $2\frac{1}{2}$ }	410631.1		$7f$	$7f \ ^2F^\circ$	{ $2\frac{1}{2}$ $3\frac{1}{2}$ }	468530	
$5f$	$5f \ ^2F^\circ$	{ $2\frac{1}{2}$ $3\frac{1}{2}$ }	414458.7		$8p$	$8p \ ^2P^\circ$	{ $\frac{1}{2}$ $1\frac{1}{2}$ }	476181	
$5g$	$5g \ ^2G$	{ $3\frac{1}{2}$ $4\frac{1}{2}$ }	414684.4						
					P VI ($1S_0$)	Limit		524462.9	

June 1947.

P VI

(Ne I sequence; 10 electrons)

 $Z=15$ Ground state $1s^2 2s^2 2p^6 {}^1S_0$ $2p^6 {}^1S_0$ 1778250 cm^{-1}

I. P. 220.414 volts

The analysis is by Robinson who has generously furnished his manuscript in advance of publication. He has classified 23 lines in the range 57 Å to 91 Å, as combinations with the ground term. The term designations he assigns on the assumption of LS -coupling are given in the table under the heading "Author".

As for Ne I, the $j\bar{l}$ -coupling notation in the general form suggested by Racah is introduced. A predicted value of $7d [1\frac{1}{2}]^\circ$, is entered in brackets in the table, since the observed combination is a blend.

REFERENCES

G. Racah, Phys. Rev. **61**, 537 (L) (1942).

H. A. Robinson, unpublished material (June 1947). (I P) (T) (C L)

P VI

P VI

Author	Config.	Desig.	J	Level	Author	Config.	Desig.	J	Level
$2p {}^1S_0$	$2p^6$	$2p^6 {}^1S$	0	0	$5s {}^1P_1$	$2p^5({}^2P_{3/2})5s$	$5s'[1\frac{1}{2}]^\circ$	0 1	1582860
$3s {}^3P_1$	$2p^5({}^2P_{1/2})3s$	$3s [1\frac{1}{2}]^\circ$	2 1	1093240	$5d {}^3P_1$	$2p^5({}^2P_{3/2})5d$	$5d [1\frac{1}{2}]^\circ$	0 1	1613680
$3s {}^1P_1$	$2p^5({}^2P_{3/2})3s$	$3s'[1\frac{1}{2}]^\circ$	0 1	1103180	$5d {}^1P_1$	"	$5d [1\frac{1}{2}]^\circ$	1	1616320
$3d {}^3P_1$	$2p^5({}^2P_{1/2})3d$	$3d [1\frac{1}{2}]^\circ$	0 1	1306610	$5d {}^3D_1$	$2p^5({}^2P_{3/2})5d$	$5d'[1\frac{1}{2}]^\circ$	1	1622800
$3d {}^1P_1$	"	$3d [1\frac{1}{2}]^\circ$	1	1321910	$6s {}^1P_1$	$2p^5({}^2P_{3/2})6s$	$6s'[1\frac{1}{2}]^\circ$	0 1	1650930
$3d {}^3D_1$	$2p^5({}^2P_{3/2})3d$	$3d'[1\frac{1}{2}]^\circ$	1	1334210	$6d {}^1P_1$	$2p^5({}^2P_{1/2})6d$	$6d [1\frac{1}{2}]^\circ$	1	1666220
$4s {}^3P_1$	$2p^5({}^2P_{1/2})4s$	$4s [1\frac{1}{2}]^\circ$	2 1	1439840	$6d {}^3D_1$	$2p^5({}^2P_{3/2})6d$	$6d'[1\frac{1}{2}]^\circ$	1	1672940
$4s {}^1P_1$	$2p^5({}^2P_{3/2})4s$	$4s'[1\frac{1}{2}]^\circ$	0 1	1446740	$7d {}^1P_1$	$2p^5({}^2P_{1/2})7d$	$7d [1\frac{1}{2}]^\circ$	1	[1696180]
$4d {}^3P_1$	$2p^5({}^2P_{1/2})4d$	$4d [1\frac{1}{2}]^\circ$	0 1	1516530	$7d {}^3D_1$	$2p^5({}^2P_{3/2})7d$	$7d'[1\frac{1}{2}]^\circ$	1	1702790
$4d {}^1P_1$	"	$4d [1\frac{1}{2}]^\circ$	1	1523460	$8d {}^1P_1$	$2p^5({}^2P_{1/2})8d$	$8d [1\frac{1}{2}]^\circ$	1	1715440
$4d {}^3D_1$	$2p^5({}^2P_{3/2})4d$	$4d'[1\frac{1}{2}]^\circ$	1	1531210	$9d {}^1P_1$	$2p^5({}^2P_{1/2})9d$	$9d [1\frac{1}{2}]^\circ$	1	1726160
$5s {}^3P_1$	$2p^5({}^2P_{1/2})5s$	$5s [1\frac{1}{2}]^\circ$	2 1	1576040		P VII (${}^2P_{1/2}$)	Limit	-----	1778250
						P VII (${}^2P_{3/2}$)	Limit	-----	1785518

June 1947.

P VI OBSERVED LEVELS*

Config. $1s^2 2s^2 +$	Observed Terms	
$2p^6$	$2p^6 \ ^1S$	
	$ns \ (n \geq 3)$	$nd \ (n \geq 3)$
$2p^5(^2P^\circ)nx$	{ $\begin{array}{ll} 3-5s & ^3P^\circ \\ 3-6s & ^1P^\circ \end{array}$	$\begin{array}{ll} 3-5d & ^3P^\circ \\ 3-9d & ^1P^\circ \end{array} \quad 3-7d \ ^3D^\circ$
<i>jl</i> -Coupling Notation		
	Observed Pairs	
	$ns \ (n \geq 3)$	$nd \ (n \geq 3)$
$2p^5(^2P_{1/2})nx$	$3-5s \ [1\frac{1}{2}]^\circ$	$\begin{array}{l} 3-5d \ [\frac{1}{2}]^\circ \\ 3-9d \ [1\frac{1}{2}]^\circ \end{array}$
$2p^5(^2P_{3/2}^\circ)nx'$	$3-6s' \ [\frac{1}{2}]^\circ$	$3-7d' \ [1\frac{1}{2}]^\circ$

*For predicted levels in the spectra of the Ne I isoelectronic sequence, see Introduction.

P VII

(F I sequence; 9 electrons)

$Z=15$

Ground state $1s^2 2s^2 2p^5 \ ^2P_{1/2}^\circ$

$2p^5 \ ^2P_{1/2}^\circ \ 2124300 \text{ cm}^{-1}$

I. P. 263.31 volts

The analysis is by Robinson, who has generously furnished his manuscript in advance of publication. He has classified more than 70 lines in the region between 49 Å and 223 Å.

Intersystem combinations connecting the doublet and quartet terms have been observed.

REFERENCE

H. A. Robinson, unpublished material (March 1948). (I P) (T) (C L)

P VII

P VII

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$2s^2 2p^5$	$2p^5 \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	0 7268	-7268	$2s^2 2p^4(^3P)4s$	$4s \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	1695720 1701380	-5660
$2s \ 2p^6$	$2p^6 \ ^2S$	$\frac{1}{2}$	454732		$2s^2 2p^4(^1D)4s$	$4s' \ ^2D$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	1741710	
$2s^2 2p^4(^3P)3s$	$3s \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	1259730 1264170 1266000?	-4440 -1830	$2s^2 2p^4(^3P)4d$	$4d \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	1775510 1784030	-8520
$2s^2 2p^4(^3P)3s$	$3s \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	1277380 1282550	-5170	$2s^2 2p^4(^3P)4d$	$4d \ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	1778690	
$2s^2 2p^4(^1D)3s$	$3s' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	1317110		$2s^2 2p^4(^3P)4d$	$4d \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	1780190 1782260	2070
$2s^2 2p^4(^1S)3s$	$3s'' \ ^2S$	$\frac{1}{2}$	1375810		$2s^2 2p^4(^1S)4s$	$4s'' \ ^2S$	$\frac{1}{2}$	1801570	
$2s^2 2p^4(^3P)3d$	$3d \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	1496890 1500040	-3150	$2s^2 2p^4(^1D)4d$	$4d' \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	1827890 1829190	-1300
$2s^2 2p^4(^3P)3d$	$3d \ ^4F$	$4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$	1498400		$2s^2 2p^4(^1D)4d$	$4d' \ ^2D$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	1828630	
$2s^2 2p^4(^3P)3d$	$3d \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	1502040 1506730	-4690	$2s^2 2p^4(^1D)4d$	$4d' \ ^2S$	$\frac{1}{2}$	1830190	
$2s^2 2p^4(^3P)3d$	$3d \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	1505300 1511310	6010	$2s^2 2p^4(^3P)5s$	$5s \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	1865680	
$2s^2 2p^4(^3P)3d$	$3d \ ^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	1510050		$2s^2 2p^4(^1S)4d$	$4d'' \ ^2D$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	1885000	
$2s^2 2p^4(^1D)3d$	$3d' \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	1548480 1552170	3690	$2s^2 2p^4(^1S)5s$	$5s'' \ ^2S$	$\frac{1}{2}$	1913620	
$2s^2 2p^4(^1D)3d$	$3d' \ ^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	1552120		$2s \ 2p^5(^3P^\circ)3d$	$3d''' \ ^1^\circ$		1919310?	
$2s^2 2p^4(^1D)3d$	$3d' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	1553740 1554420	-680	$2s \ 2p^5(^3P^\circ)3d$	$3d''' \ ^2^\circ$		1921010?	
$2s^2 2p^4(^1D)3d$	$3d' \ ^2S$	$\frac{1}{2}$	1555560		$2s \ 2p^5(^3P^\circ)3d$	$3d''' \ ^3^\circ$		1922150?	
$2s^2 2p^4(^1S)3d$	$3d'' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	1606550 1606880	-330	$2s \ 2p^5(^3P^\circ)3d$	$3d''' \ ^4^\circ$		1925560?	
$2s \ 2p^5(^3P^\circ)3s$	$3s''' \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	1692160 1696860	-4700	$2s \ 2p^5(^3P^\circ)3d$	$3d''' \ ^5^\circ$		1931070?	
					$2s^2 2p^4(^1S)5d$	$5d'' \ ^2D$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	2013690	
					P VIII (3P_2)	Limit	-----	2124300	

March 1948.

P VII OBSERVED TERMS*

Config. 1s ² +	Observed Terms						
2s ² 2p ⁵	2p ⁵ ² P°						
2s 2p ⁶	2p ⁶ ² S						
	ns (n ≥ 3)			nd (n ≥ 3)			
2s ² 2p ⁴ (³ P)nx	{	3s ⁴ P 3-5s ² P		3, 4d ⁴ P 3, 4d ² P	3, 4d ² D	3d ⁴ F 3d ² F	
2s ² 2p ⁴ (¹ D)nx'			3, 4s' ² D	3, 4d' ² S	3, 4d' ² P	3, 4d' ² D	3d' ² F
2s ² 2p ⁴ (¹ S)nx''		3-5s'' ² S				3-5d'' ² D	
2s 2p ⁵ (³ P°)nx'''		3s''' ² P°					

*For predicted terms in the spectra of the F I isoelectronic sequence, see Introduction.

P VIII

(O I sequence; 8 electrons)

 $Z=15$ Ground state $1s^2 2s^2 2p^4 {}^3P_2$ $2p^4 {}^3P_2$ 2495000 cm^{-1}

I. P. 309.26 volts

The terms are from an unpublished manuscript kindly furnished by Robinson. No inter-system combinations have been observed and the uncertainty, x , may be considerable.

The unit adopted by Robinson, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCE

H A. Robinson, unpublished material (March 1948). (I P) (T)

P VIII

P VIII

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$2s^2 2p^4$	$2p^4 {}^3P$	2 1 0	0 5757 7826	—5757 —2069	$2s^2 2p^3({}^2P^\circ)3d$	$3d'' {}^3F^\circ$	4 3 2	1790480 1795030	—4550
$2s^2 2p^4$	$2p^4 {}^1D$	2	52450+x		$2s^2 2p^3({}^2P^\circ)3d$	$3d'' {}^1D^\circ$	2	1795430+x	
$2s^2 2p^4$	$2p^4 {}^1S$	0	110970+x		$2s^2 2p^3({}^2P^\circ)3d$	$3d'' {}^3D^\circ$	3 2 1	1796240 1800770	—4530
$2s 2p^5$	$2p^5 {}^3P^\circ$	2 1 0	403806 408913 411736	—5107 —2823	$2s^2 2p^3({}^2P^\circ)3d$	$3d'' {}^1P^\circ$	1	1800760+x	
$2s 2p^5$	$2p^5 {}^1P^\circ$	1	560680+x		$2s^2 2p^3({}^2P^\circ)3d$	$3d'' {}^1F^\circ$	3	1804930+x	
$2s^2 2p^3({}^4S^\circ)3s$	$3s {}^3S^\circ$	1	1462340		$2s^2 2p^3({}^4S^\circ)4s$	$4s {}^3S^\circ$	1	1958370	
$2s^2 2p^3({}^2D^\circ)3s$	$3s' {}^3D^\circ$	1, 2 3	1519740 1520030	290	$2s^2 2p^3({}^2D^\circ)4s$	$4s' {}^3D^\circ$	1 2 3	2029470	
$2s^2 2p^3({}^2D^\circ)3s$	$3s' {}^1D^\circ$	2	1532020+x		$2s^2 2p^3({}^2D^\circ)4s$	$4s' {}^1D^\circ$	2	2033320+x	
$2s^2 2p^3({}^2P^\circ)3s$	$3s'' {}^3P^\circ$	0 1 2	1559500 1560070 1561260	570 1190	$2s^2 2p^3({}^4S^\circ)4d$	$4d {}^3D^\circ$	1 2 3	2046710	
$2s^2 2p^3({}^2P^\circ)3s$	$3s'' {}^1P^\circ$	1	1573270+x		$2s^2 2p^3({}^2P^\circ)4s$	$4s'' {}^1P^\circ$	1	2073760+x	
$2s^2 2p^3({}^4S^\circ)3d$	$3d {}^3D^\circ$	1, 2 3	1685980 1686280	300	$2s^2 2p^3({}^2D^\circ)4d$	$4d' {}^3D^\circ$	3, 2, 1	2115510	
$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^3F^\circ$	4, 3, 2	1749870		$2s^2 2p^3({}^2D^\circ)4d$	$4d' {}^3P^\circ$	2 1 0	2119360	
$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^3D^\circ$	3, 2, 1	1753090		$2s^2 2p^3({}^2D^\circ)4d$	$4d' {}^3S^\circ$	1	2122020	
$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^1P^\circ$	1	1753830+x		$2s^2 2p^3({}^2D^\circ)4d$	$4d' {}^1F^\circ$	3	2123570+x	
$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^3P^\circ$	2 1 0	1760530 1762400	—1870	$2s^2 2p^3({}^4S^\circ)5d$	$5d {}^3D^\circ$	1 2 3	2210630	
$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^1D^\circ$	2	1761680+x		$2s^2 2p^3({}^2D^\circ)5s$	$5s' {}^1D^\circ$	1	2240920+x	
$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^3S^\circ$	1	1767880						
$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^1F^\circ$	3	1776050+x						
$2s^2 2p^3({}^2P^\circ)3d$	$3d'' {}^3P^\circ$	0 1 2	1787090 1788090 1789690	1000 1600	P IX (${}^4S_{1/2}$)	Limit	-----	2495000	

March 1948.

P VIII OBSERVED TERMS*

Config. $1s^2+$	Observed Terms	
$2s^2 2p^4$	$\left\{ \begin{array}{cc} 2p^4 \ ^1S & 2p^4 \ ^3P \\ & 2p^4 \ ^1D \end{array} \right.$	
$2s \ 2p^5$	$\left\{ \begin{array}{cc} & 2p^5 \ ^3P^\circ \\ & 2p^5 \ ^1P^\circ \end{array} \right.$	
	$ns \ (n \geq 3)$	$nd \ (n \geq 3)$
$2s^2 2p^3(^4S^\circ)nx$	$3, 4s \ ^3S^\circ$	$3-5d \ ^3D^\circ$
$2s^2 2p^3(^2D^\circ)nx'$	$\left\{ \begin{array}{cc} & 3, 4s' \ ^3D^\circ \\ & 3-5s' \ ^1D^\circ \end{array} \right.$	$3, 4d' \ ^3S^\circ \quad 3, 4d' \ ^3P^\circ \quad 3, 4d' \ ^3D^\circ \quad 3d' \ ^3F^\circ$ $\quad \quad \quad 3d' \ ^1P^\circ \quad 3d' \ ^1D^\circ \quad 3, 4d' \ ^1F^\circ$
$2s^2 2p^3(^2P^\circ)nx''$	$\left\{ \begin{array}{cc} & 3s'' \ ^3P^\circ \\ & 3, 4s'' \ ^1P^\circ \end{array} \right.$	$3d'' \ ^3P^\circ \quad 3d'' \ ^3D^\circ \quad 3d'' \ ^3F^\circ$ $\quad \quad \quad 3d'' \ ^1P^\circ \quad 3d'' \ ^1D^\circ \quad 3d'' \ ^1F^\circ$

*For predicted terms in the spectra of the O I isoelectronic sequence, see Introduction.

P IX

(N I sequence; 7 electrons)

$Z=15$

Ground state $1s^2 2s^2 2p^3 \ ^4S_{1/2}^\circ$

$2p^3 \ ^4S_{1/2}^\circ \ 3006200 \text{ cm}^{-1}$

I. P. 372.62 volts

The analysis is by Robinson, who has kindly furnished a manuscript copy in advance of publication. He has found 35 terms, and classified more than 100 lines in the region between 40 Å and 314 Å. Intersystem combinations connecting the doublet and quartet systems of terms have been observed.

REFERENCE

H. A. Robinson, unpublished material (March 1948). (I P) (T) (C L)

P IX

P IX

Config.	Desig.	J	Level	Internal	Config.	Desig.	J	Level	Interval
$2s^2 2p^3$	$2p^3 \ ^4S^\circ$	$1\frac{1}{2}$	0		$2p^5$	$2p^5 \ ^2P^\circ$	$1\frac{1}{2}$	898220 904700	-6480
$2s^2 2p^3$	$2p^3 \ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	73167 73730	563	$2s^2 2p^2(^3P)3s$	$3s \ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	1744000 1746250 1751850	2250 5600
$2s^2 2p^3$	$2p^3 \ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	113457 114430	973	$2s^2 2p^2(^3P)3s$	$3s \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	1764370 1768970	4600
$2s \ 2p^4$	$2p^4 \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	345390 350440 353050	-5050 -2610	$2s^2 2p^2(^1D)3s$	$3s' \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	1805940 1807340	1400
$2s \ 2p^4$	$2p^4 \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	472580 473090	-510	$2s^2 2p^2(^3P)3d$	$3d \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	1962630 1963830	-1200
$2s \ 2p^4$	$2p^4 \ ^2S$	$\frac{1}{2}$	552540		$2s \ 2p^3(^5S^\circ)3s$	$3s''' \ ^4S^\circ$	$1\frac{1}{2}$	1965970	
$2s \ 2p^4$	$2p^4 \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	580710 587010	-6300	$2s^2 2p^2(^3P)3d$	$3d \ ^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	1970380 1976610	6230

P IX—Continued

P IX—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$2s^2 2p^2(^3P)3d$	$3d \ ^4D$	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{2}{2}$ $\frac{3}{2}$	1973870 1975970	2100	$2s \ 2p^3(^3D^{\circ})3p$	$3p^{IV} \ ^2F$	$\frac{2}{2}$ $\frac{3}{2}$	2224980	
$2s^2 2p^2(^3P)3d$	$3d \ ^4P$	$\frac{2}{2}$ $\frac{1}{2}$ $\frac{1}{2}$	1977830 1979750 1980870	—1920 —1120	$2s \ 2p^3(^3D^{\circ})3d$	$3d^{IV} \ ^2F^{\circ}$	$\frac{3}{2}$ $\frac{2}{2}$	2309530 2312530	—3000
$2s^2 2p^2(^3P)3d$	$3d \ ^2D$	$\frac{1}{2}$ $\frac{2}{2}$	2000360 2001960	1600	$2s^2 2p^2(^3P)4s$	$4s \ ^4P$	$\frac{1}{2}$ $\frac{1}{2}$ $\frac{2}{2}$	2354100	
$2s^2 2p^2(^1D)3d$	$3d' \ ^2F$	$\frac{2}{2}$ $\frac{3}{2}$	2028530		$2s^2 2p^2(^3P)4s$	$4s \ ^2P$	$\frac{1}{2}$ $\frac{1}{2}$	2354120 2359520	5400
$2s^2 2p^2(^1D)3d$	$3d' \ ^2D$	$\frac{1}{2}$ $\frac{2}{2}$	2031610		$2s^2 2p^2(^3P)4d$	$4d \ ^2F$	$\frac{2}{2}$ $\frac{3}{2}$	2430900 2436400	5500
$2s^2 2p^2(^1D)3d$	$3d' \ ^2P$	$\frac{1}{2}$ $\frac{1}{2}$	2038670 2042470	3800	$2s^2 2p^2(^3P)4d$	$4d \ ^4P$	$\left\{ \begin{array}{c} \frac{1}{2} \\ \text{to} \\ \frac{2}{2} \end{array} \right\}$	2435220	
$2s \ 2p^3(^5S^{\circ})3p$	$3p''' \ ^4P$	$\left\{ \begin{array}{c} \frac{1}{2} \\ \text{to} \\ \frac{2}{2} \end{array} \right\}$	2043950		$2s^2 2p^2(^3P)4d$	$4d \ ^2D$	$\left\{ \begin{array}{c} \frac{1}{2} \\ \frac{2}{2} \end{array} \right\}$	2441100	
$2s^2 2p^2(^1D)3d$	$3d' \ ^2S$	$\frac{1}{2}$	2049150		$2s^2 2p^2(^1D)4d$	$4d' \ ^2F$	$\frac{2}{2}$ $\frac{3}{2}$	2480120	
$2s^2 2p^2(^1S)3d$	$3d'' \ ^2D$	$\left\{ \begin{array}{c} \frac{1}{2} \\ \frac{2}{2} \end{array} \right\}$	2079720		$2s^2 2p^2(^1D)4d$	$4d' \ ^2D$	$\left\{ \begin{array}{c} \frac{1}{2} \\ \frac{2}{2} \end{array} \right\}$	2487270	
$2s \ 2p^3(^3D^{\circ})3s$	$3s^{IV} \ ^2D^{\circ}$	$\left\{ \begin{array}{c} \frac{1}{2} \\ \frac{2}{2} \end{array} \right\}$	2103110		$2s^2 2p^2(^1S)4d$	$4d'' \ ^2D$	$\left\{ \begin{array}{c} \frac{1}{2} \\ \frac{2}{2} \end{array} \right\}$	2547080	
$2s \ 2p^3(^5S^{\circ})3d$	$3d''' \ ^4D^{\circ}$	$\left\{ \begin{array}{c} \frac{1}{2} \\ \text{to} \\ \frac{3}{2} \end{array} \right\}$	2161390		P x (3P_0)	Limit	-----	3006200	

March 1948.

P IX OBSERVED TERMS*

Config. $1s^2+$	Observed Terms		
$2s^2 2p^3$	$\left\{ \begin{array}{l} 2p^3 \ ^4S^{\circ} \\ 2p^3 \ ^2P^{\circ} \quad 2p^3 \ ^2D^{\circ} \end{array} \right.$		
$2s \ 2p^4$	$\left\{ \begin{array}{l} 2p^4 \ ^2S \\ 2p^4 \ ^4P \quad 2p^4 \ ^2P \quad 2p^4 \ ^2D \end{array} \right.$		
$2p^5$	$2p^5 \ ^2P^{\circ}$		
	$ns \ (n \geq 3)$	$np \ (n \geq 3)$	$nd \ (n \geq 3)$
$2s^2 2p^2(^3P)nx$	$\left\{ \begin{array}{l} 3, 4s \ ^4P \\ 3, 4s \ ^2P \end{array} \right.$		$\begin{array}{l} 3, 4d \ ^4P \quad 3d \ ^4D \\ 3d \ ^2P \quad 3, 4d \ ^2D \quad 3, 4d \ ^2F \end{array}$
$2s^2 2p^2(^1D)nx'$		$3s' \ ^2D$	$\begin{array}{l} 3d' \ ^2S \quad 3d' \ ^2P \quad 3, 4d' \ ^2D \quad 3, 4d' \ ^2F \end{array}$
$2s^2 2p^2(^1S)nx''$			$3, 4d'' \ ^2D$
$2s \ 2p^3(^5S^{\circ})nx'''$	$3s''' \ ^4S^{\circ}$	$3p''' \ ^4P$	$3d''' \ ^4D^{\circ}$
$2s \ 2p^3(^3D^{\circ})nx^{IV}$		$3s^{IV} \ ^2D^{\circ}$	$\begin{array}{l} 3p^{IV} \ ^2F \\ 3d^{IV} \ ^2F^{\circ} \end{array}$

*For predicted terms in the spectra of the N I isoelectronic sequence, see Introduction.

P x

(C I sequence; 6 electrons)

Z=15

Ground state $1s^2 2s^2 2p^2 {}^3P_0$ $2p^2 {}^3P_0$ 3432500 cm^{-1}

I. P. 425.46 volts

The analysis is from unpublished material kindly furnished by Robinson. He has found 36 terms and classified more than 70 lines in the region between 43 Å and 318 Å.

The singlet and triplet terms are connected by intersystem combinations. The connection of the quintet terms with the rest is based on Robinson's extrapolation of isoelectronic sequence data, as indicated by the uncertainty, x , and brackets in the table. The position of the level $2p^3 {}^3D_2$ is also extrapolated and entered in brackets.

REFERENCE

H. A. Robinson, unpublished material (March 1948). (I P) (T) (C L)

P x

P x

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$2s^2 2p^2$	$2p^2 {}^3P$	0 1 2	0 3390 8580	3390 5190	$2s^2 2p({}^2P^\circ)3d$	$3d {}^3P^\circ$	2 1 0	2171630 2173040 2173990	-1410 -950
$2s^2 2p^2$	$2p^2 {}^1D$	2	59330		$2s 2p^2({}^4P)3s$	$3s {}^3P$	0 1 2	2178420 2182320 2188220	3900 5900
$2s^2 2p^2$	$2p^2 {}^1S$	0	119430		$2s^2 2p({}^2P^\circ)3d$	$3d {}^1P^\circ$	1	2197500	
$2s 2p^3$	$2p^3 {}^5S^\circ$	2	[166580] + x		$2s^2 2p({}^2P^\circ)3d$	$3d {}^1F^\circ$	3	2197500	
$2s 2p^3$	$2p^3 {}^3D^\circ$	3 2 1	322790 [323010] 323160	[-220] -150	$2s 2p^2({}^4P)3p$	$3p {}^3S^\circ$	1	2216880	
$2s 2p^3$	$2p^3 {}^3P^\circ$	2 1 0	379660		$2s 2p^2({}^4P)3p$	$3p {}^3D^\circ$	3 2 1	2262660 2267280 2269510	-4620 -2230
$2s 2p^3$	$2p^3 {}^1D^\circ$	2	484377		$2s 2p^2({}^4P)3p$	$3p {}^3P^\circ$	2 1 0	2275380 2281140 2286080?	-5760 -4940
$2s 2p^3$	$2p^3 {}^3S^\circ$	1	490100		$2s 2p^2({}^2D)3s$	$3s' {}^3D$	1, 2, 3	2281000	
$2s 2p^3$	$2p^3 {}^1P^\circ$	1	541090		$2s 2p^2({}^1D)3s$	$3s' {}^1D$	2	2307970	
$2s^2 2p({}^2P^\circ)3s$	$3s {}^3P^\circ$	0 1 2	1954140 1955980 1963430	1840 7450	$2s 2p^2({}^4P)3d$	$3d {}^5D$	0 1 2 3 4	2331040 + x	
$2s^2 2p({}^2P^\circ)3s$	$3s {}^1P^\circ$	1	1976578		$2s 2p^2({}^4P)3d$	$3d {}^5P$	3 2 1	2342240 + x 2343760 + x 2344970 + x	-1520 -1210
$2s 2p^2({}^4P)3s$	$3s {}^5P$	1 2 3	2132450 + x 2135050 + x 2139320 + x	2600 4270	$2s 2p^2({}^4P)3d$	$3d {}^3P$	2 1 0	2345800 2351740 2354640	-5940 -2900
$2s^2 2p({}^2P^\circ)3d$	$3d {}^3F^\circ$	2 3 4	2140410		$2s 2p^2({}^4P)3d$	$3d {}^3F$	2 3 4	2355750 2358400 2362900	2650 4500
$2s^2 2p({}^2P^\circ)3d$	$3d {}^1D^\circ$	2	2147190						
$2s^2 2p({}^2P^\circ)3d$	$3d {}^3D^\circ$	1 2 3	2162410 2163500 2166800	1090 3300					

P X—Continued

P X—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
2s 2p ² (² D)3p	3p' ¹ F°	3	2371790	2000 1550	2s 2p ² (² D)3d	3d' ¹ D	2	2499250?	
2s 2p ² (² D)3p	3p' ¹ D°	2	2382480		2s 2p ² (² D)3d	3d' ¹ F	3	2499250?	
2s 2p ² (⁴ P)3d	3d ³ D	1	2385080		2s 2p ² (² D)3d	3d' ³ S	1	2509590?	
		2	2387080						
		3	2388630						
2s 2p ² (² D)3d	3d' ³ F	2, 3, 4	2467290		P XI (² P _{1/2})	Limit	-----	3432500	
2s 2p ² (² D)3d	3d' ³ D	1, 2, 3	2476100						

March 1948.

P X OBSERVED TERMS*

Config. 1s ² +	Observed Terms		
2s ² 2p ²	{ 2p ² ¹ S 2p ² ³ P 2p ² ¹ D		
2s 2p ³	{ 2p ³ ⁵ S° 2p ³ ³ P° 2p ³ ³ D° 2p ³ ³ S° 2p ³ ¹ P° 2p ³ ¹ D°		
	ns (n ≥ 3)	np (n ≥ 3)	nd (n ≥ 3)
2s ² 2p(² P°)nx	{ 3s ³ P° 3s ¹ P°		3d ³ P° 3d ³ D° 3d ³ F° 3d ¹ P° 3d ¹ D° 3d ¹ F°
2s 2p ² (⁴ P)nx	{ 3s ⁵ P 3s ³ P	3p ³ S° 3p ³ P° 3p ³ D°	3d ⁵ P 3d ⁵ D 3d ⁵ F 3d ³ P 3d ³ D 3d ³ F
2s 2p ² (² D)nx'	{ 3s' ³ D 3s' ¹ D	3p' ¹ D° 3p' ¹ F°	3d' ³ S 3d' ³ D 3d' ³ F 3d' ¹ D 3d' ¹ F

*For predicted terms in the spectra of the C I isoelectronic sequence, see Introduction.

P XI

(B I sequence; 5 electrons)

Z = 15Ground state 1s² 2s² 2p ²P_{1/2}°2p ²P_{1/2}° 3867500 cm⁻¹

I. P. 479.4 volts

The analysis is by Robinson, who has generously furnished his manuscript in advance of publication. He has classified 31 lines in the range from 42 Å to 325 Å. Some of the relative levels have been connected by a study of the behavior of the Rydberg denominators, rather than by the Ritz combination principle.

No intersystem combinations, connecting the doublet and quartet terms, have been observed, as indicated by *x* in the table. Robinson's extrapolated value of 2p² ⁴P_{1/2} is entered in brackets.

REFERENCE

H. A. Robinson, unpublished material (Feb. 1948). (I P) (T) (C L)

P XI

P XI

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$2s^2(^1S)2p$	$2p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	0 9700	9700	$2s\ 2p(^3P^\circ)3d$	$3d\ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	2539140 2540050	910
$2s\ 2p^2$	$2p^2\ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$[177900]+x$ $181300+x$ $186400+x$	3400 5100	$2s\ 2p(^1P^\circ)3s$	$3s'\ ^2P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	2541040	
$2s\ 2p^2$	$2p^2\ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	317190		$2s\ 2p(^3P^\circ)3d$	$3d\ ^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$2547290+x$	
$2s\ 2p^2$	$2p^2\ ^2S$	$\frac{1}{2}$	403330		$2s\ 2p(^3P^\circ)3d$	$3d\ ^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	2578000 2584000	6000
$2s\ 2p^2$	$2p^2\ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	425820 431650	5830	$2s\ 2p(^3P^\circ)3d$	$3d\ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	2589460 2593090	-3630
$2p^3$	$2p^3\ ^4S^\circ$	$1\frac{1}{2}$	$559500+x$		$2s\ 2p(^1P^\circ)3d$	$3d'\ ^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	2697820	
$2s^2(^1S)3s$	$3s\ ^2S$	$\frac{1}{2}$	2174060		$2s\ 2p(^1P^\circ)3d$	$3d'\ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	2707510 2709400	1890
$2s^2(^1S)3d$	$3d\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	2347470 2348130	660	$2p^2(^3P)3d$	$3d''\ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	$2856820+x$ $2858970+x$	-2150
$2s\ 2p(^3P^\circ)3s$	$3s\ ^4P^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	$2369930+x$ $2376130+x$ $2379730+x$	-6200 -3600	-----			-----	
$2s\ 2p(^3P^\circ)3s$	$3s\ ^2P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	2410070		P XII (1S_0)	Limit	-----	3867500	
$2s\ 2p(^3P^\circ)3d$	$3d\ ^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	$2536000+x$ $2540500+x$	4500					

February 1948.

P XI OBSERVED TERMS*

Config. $1s^2+$	Observed Terms	
$2s^2(^1S)2p$	$2p\ ^2P^\circ$	
$2s\ 2p^2$	$\left\{ \begin{array}{l} 2p^2\ ^2S \\ 2p^2\ ^4P \\ 2p^2\ ^2P \end{array} \right. \quad 2p^2\ ^2D$	
$2p^3$	$2p^3\ ^4S^\circ$	
	$ns\ (n \geq 3)$	$nd\ (n \geq 3)$
$2s^2(^1S)nx$	$3s\ ^2S$	$3d\ ^2D$
$2s\ 2p(^3P^\circ)nx$	$\left\{ \begin{array}{l} 3s\ ^4P^\circ \\ 3s\ ^2P^\circ \end{array} \right.$	$\begin{array}{l} 3d\ ^4P^\circ \quad 3d\ ^4D^\circ \\ 3d\ ^2P^\circ \quad 3d\ ^2D^\circ \quad 3d\ ^2F^\circ \end{array}$
$2s\ 2p(^1P^\circ)nx'$	$3s'\ ^2P^\circ$	$3d'\ ^2D^\circ \quad 3d'\ ^2F^\circ$
$2p^2(^3P)nx''$		$3d''\ ^4P$

*For predicted terms in the spectra of the Bi isoelectronic sequence, see Introduction.

P XII

(Be I sequence; 4 electrons)

 $Z=15$ Ground state $1s^2 2s^2 {}^1S_0$ $2s^2 {}^1S_0$ 4520500 cm^{-1}

I. P. 560.3 volts

The analysis is by Robinson, who has kindly furnished his manuscript on this spectrum in advance of publication. He has found 18 terms and classified 15 lines between 36 Å and 44 Å. Some of the relative terms have been connected by a study of the Rydberg denominators rather than by the Ritz combination principle.

No intersystem combinations have been observed, as indicated by the uncertainty x in the table. Robinson's extrapolated value of $2p {}^3P_0^\circ$ is entered in brackets.

REFERENCE

H. A. Robinson, unpublished material (Feb. 1948). (I P) (T) (C L)

P XII

P XII

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$2s^2$	$2s^2 {}^1S$	0	0		$2s({}^2S)3d$	$3d {}^1D$	2	2760490	
$2s({}^2S)2p$	$2p {}^3P^\circ$	0 1 2	$[183190] + x$ $186390 + x$ $192990 + x$	3200 6600	$2p({}^2P^\circ)3s$	$3s {}^1P^\circ$	1	2876720	
$2s({}^2S)2p$	$2p {}^1P^\circ$	1	358840		$2p({}^2P^\circ)3p$	$3p {}^1P$	1	2888690?	
$2p^2$	$2p^2 {}^3P$	0 1 2			$2p({}^2P^\circ)3p$	$3p {}^3D$	1 2 3		
								2897300 + x	
$2p^2$	$2p^2 {}^1D$	2	490990 + x		$2p({}^2P^\circ)3d$	$3d {}^1D^\circ$	2	2936160	
$2s({}^2S)3s$	$3s {}^3S$	1	538190		$2p({}^2P^\circ)3p$	$3p {}^1D$	2	2947770	
$2s({}^2S)3s$	$3s {}^1S$	0	2594640 + x		$2p({}^2P^\circ)3d$	$3d {}^3D^\circ$	1, 2, 3	2964340 + x	
$2s({}^2S)3p$	$3p {}^1P^\circ$	1	2629250		$2p({}^2P^\circ)3d$	$3d {}^1F^\circ$	3	3000210	
$2s({}^2S)3p$	$3p {}^3P^\circ$	1	2677740		$2p({}^2P^\circ)3d$	$3d {}^1P^\circ$	1	3011540	
$2s({}^2S)3d$	$3d {}^3D$	1 2 3	2726690 + x 2727190 + x 2727840 + x	500 650	-----			-----	
					P XIII (${}^2S_{1/2}$)	Limit	-----	4520500	

February 1948.

P XII OBSERVED TERMS*

Config. $1s^2 +$	Observed Terms		
$2s^2$	$2s^2 \ ^1S$		
$2s(^2S)2p$	$\left\{ \begin{array}{l} 2p \ ^3P^\circ \\ 2p \ ^1P^\circ \end{array} \right.$		
$2p^2$	$\left\{ \begin{array}{l} 2p^2 \ ^3P \\ 2p^2 \ ^1D \end{array} \right.$		
	$ns \ (n \geq 3)$	$np \ (n \geq 3)$	$nd \ (n \geq 3)$
$2s(^2S)nx$	$\left\{ \begin{array}{l} 3s \ ^3S \\ 3s \ ^1S \end{array} \right.$	$3p \ ^1P^\circ$	$\begin{array}{l} 3d \ ^3D \\ 3d \ ^1D \end{array}$
$2p(^2P^\circ)nx$	$\left\{ \begin{array}{l} 3s \ ^1P^\circ \end{array} \right.$	$\begin{array}{ll} 3p \ ^1P & 3p \ ^3D \\ 3p \ ^1D & 3p \ ^1D \end{array}$	$\begin{array}{lll} 3d \ ^1P^\circ & 3d \ ^3D^\circ & 3d \ ^1F^\circ \\ & 3d \ ^1D^\circ & \end{array}$

*For predicted terms in the spectra of the Be I isoelectronic sequence, see Introduction.

P XIII

(Li I sequence; 3 electrons)

$Z=15$

Ground state $1s^2 2s \ ^2S_{1/2}$

$2s \ ^2S_{1/2} \ 4933060 \text{ cm}^{-1}$

I. P. 611.45 volts

This spectrum is incompletely analyzed. Robinson has kindly furnished his unpublished manuscript giving seven classified lines; one at 110 Å and six between 35 Å and 38 Å. The resonance lines have not been observed. The absolute value of the ground term has been extrapolated from isoelectronic sequence data. Similarly, other relative levels have been connected by a study of the Rydberg denominators in the isoelectronic sequence rather than by the Ritz combination principle.

REFERENCE

H. A. Robinson, unpublished material (Feb. 1948). (I P) (T) (C L)

P XIII

Config.	Desig.	J	Level	Interval
$2s$	$2s \ ^2S$	$\frac{1}{2}$	0	
$2p$	$2p \ ^3P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	207720 219250	11530
$3s$	$3s \ ^2S$	$\frac{1}{2}$	2794900	
$3p$	$3p \ ^3P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	2844390 2850150	5760
$3d$	$3d \ ^3D$	$1\frac{1}{2}$ $2\frac{1}{2}$	2870260 2871620	1360
$4f$	$4f \ ^3F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	3772770?	
-----	-----	-----	-----	
P XIV (1S_0)	Limit	-----	4933060	

February 1948.

SULFUR

S I

16 electrons

 $Z=16$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^4 {}^3P_2$ $3p^4 {}^3P_2$ 83559.3 cm^{-1}

I. P. 10.357 volts

Edlén has revised and extended the earlier analyses and has generously furnished his manuscript term list in advance of publication, for inclusion here. Brackets denote values calculated from the series. For two such terms, however, $4f$ and $8f {}^5F$, combinations with $3d {}^5D^\circ$ have been observed.

Intersystem combinations connecting terms of all three multiplicities, have been observed.

REFERENCES

- R. Frerichs, Zeit. Phys. **80**, 150 (1933). (I P) (T) (C L)
 K. W. Meissner, O. Bartelt und L. Eckstein, Zeit. Phys. **86**, 54 (1933). (I P) (T) (C L)
 J. E. Ruedy, Phys. Rev. **44**, 757 (1933). (I P) (T) (C L)
 B. Edlén, Phys. Rev. **62**, 434 (1942). (T)
 W. F. Meggers, J. Opt. Soc. Am. **36**, 431 (1946). (Summary hfs)
 B. Edlén, unpublished material (Nov. 1946). (I P) (T)

S I

S I

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3s^2 3p^4$	$3p^4 {}^3P$	2	0.0		$3s^2 3p^3 ({}^2D^\circ) 4s$	$4s' {}^1D^\circ$	2	69238.7	
		1	396.8	-396.8			1	70165.9	0.9
		0	573.6	-176.8			2	70166.8	3.9
$3s^2 3p^4$	$3p^4 {}^1D$	2	9239.0		$3s^2 3p^3 ({}^4S^\circ) 3d$	$3d {}^3D^\circ$	3	70170.7	
$3s^2 3p^4$	$3p^4 {}^1S$	0	22181.4		$3s^2 3p^3 ({}^4S^\circ) 5s$	$5s {}^5S^\circ$	2	[70706]	
$3s^2 3p^3 ({}^4S^\circ) 4s$	$4s {}^5S^\circ$	2	52623.88		$3s^2 3p^3 ({}^4S^\circ) 5s$	$5s {}^3S^\circ$	1	71352.5	
$3s^2 3p^3 ({}^4S^\circ) 4s$	$4s {}^3S^\circ$	1	55331.15		$3s 3p^5$	$3p^5 {}^3P^\circ$	2	72025.5	-357.0
$3s^2 3p^3 ({}^4S^\circ) 4p$	$4p {}^5P$	1	63446.36	10.97			1	72382.5	-189.9
		2	63457.33	17.93			0	72572.4	
		3	63475.26		$3s^2 3p^3 ({}^4S^\circ) 5p$	$5p {}^5P$	1	73911.53	3.63
$3s^2 3p^3 ({}^4S^\circ) 4p$	$4p {}^3P$	0	64891.71	-2.48			2	73915.16	5.98
		1	64889.23	3.66			3	73921.14	
		2	64892.89		$3s^2 3p^3 ({}^4S^\circ) 5p$	$5p {}^3P$	2	74269.20	-1.08
$3s^2 3p^3 ({}^2D^\circ) 4s$	$4s' {}^3D^\circ$	1	67816.87	8.85			1	74270.28	-2.04
		2	67825.72	17.66			0	74272.32	
		3	67843.33		$3s^2 3p^3 ({}^4S^\circ) 4d$	$4d {}^5D^\circ$	4	74973.35	-0.95
$3p^3 ({}^4S^\circ) 3d$	$3d {}^5D^\circ$	4	67878.03	-12.42			3	74974.30	-1.13
		3	67890.45	2.20			2	74975.43	-0.88
		2	67888.25	2.28			1	74976.31	-0.59
		1	67885.97	1.30			0	74976.90	
		0	67884.67						

S I—Continued

S I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2 3p^3(^4S^\circ)4d$	$4d \ ^3D^\circ$	1 2 3	75952. 16 75952. 67 75956. 80	0. 51 4. 13	$3s^2 3p^3(^4S^\circ)7p$	$7p \ ^3P$	2 1 0	80113. 23 80120. 51 80124. 16	—7. 28 —3. 65
$3s^2 3p^3(^4S^\circ)6s$	$6s \ ^5S^\circ$	2	76464. 26		$3s^2 3p^3(^4S^\circ)6d$	$6d \ ^3D^\circ$	3 2 1	80182. 54 80183. 93 80185. 78	—1. 39 —1. 85
$3s^2 3p^3(^4S^\circ)4f$	$4f \ ^5F$	5 to 1	[76653]		$2s^2 3p^3(^4S^\circ)8s$	$8s \ ^5S^\circ$	2	80449. 30	
$3s^2 3p^3(^4S^\circ)4f$	$4f \ ^3F$	4, 3, 2	[76655]		$3s^2 3p^3(^4S^\circ)6f$	$6f \ ^5F$	5 to 1	80494. 73	
$3s^2 3p^3(^4S^\circ)6s$	$6s \ ^3S^\circ$	1	76720. 90		$3s^2 3p^3(^4S^\circ)6f$	$6f \ ^3F$	4, 3, 2	80495. 76	
$3s^2 3p^3(^2P^\circ)4s$	$4s'' \ ^3P^\circ$	0 1 2	77136. 10 77150. 59 77181. 41	14. 49 30. 82	$3s^2 3p^3(^4S^\circ)8s$	$8s \ ^3S^\circ$	1	80521. 99	
$3s^2 3p^3(^4S^\circ)6p$	$6p \ ^5P$	1 2 3	77851. 21 77856. 49	5. 28	$3s^2 3p^3(^4S^\circ)7d$	$7d \ ^5D^\circ$	4 3 2 1 0	80995. 48	
$3s^2 3p^3(^4S^\circ)6p$	$6p \ ^3P$	2 1 0	77891. 10		$3s^2 3p^3(^4S^\circ)8p$	$8p \ ^3P$	0, 1 2	80995. 90 80996. 33	0. 43
$3s^2 3p^3(^2D^\circ)4p$	$4p' \ ^3D$	1 2 3	78152. 45 78152. 00 78203. 38	—0. 45 51. 38	$3s^2 3p^3(^4S^\circ)7d$	$7d \ ^3D^\circ$	3 2 1	81080. 52 81082. 83 81084. 83	—2. 31 —2. 00
$3s^2 3p^3(^4S^\circ)5d$	$5d \ ^5D^\circ$	4 3, 2 2, 1, 0	78270. 30 78270. 72 78271. 19	—0. 42 —0. 47	$3s^2 3p^3(^4S^\circ)9s$	$9s \ ^5S^\circ$	2	81281. 76	
$3s^2 3p^3(^2P^\circ)4s$	$4s'' \ ^1P^\circ$	1	78290. 4		$3s^2 3p^3(^4S^\circ)7f$	$7f \ ^5F$	5 to 1	81309. 23	
$3s^2 3p^3(^2D^\circ)4p$	$4p' \ ^3F$	2 3 4	78410. 37 78436. 30 78463. 55	25. 93 27. 25	$3s^2 3p^3(^4S^\circ)7f$	$7f \ ^3F$	4, 3, 2	81310. 08	
$3s^2 3p^3(^2D^\circ)4p$	$4p' \ ^1F$	3	78638. 2		$3s^2 3p^3(^4S^\circ)9s$	$9s \ ^3S^\circ$	1	[81327. 3]	
$3s^2 3p^3(^4S^\circ)5d$	$5d \ ^3D^\circ$	3 2 1	78692. 24 78691. 78 78692. 99	0. 46 —1. 21	$3s^2 3p^3(^4S^\circ)8d$	$8d \ ^5D^\circ$	4 3 2 1 0	81628. 90	
$3s^2 3p^3(^4S^\circ)7s$	$7s \ ^5S^\circ$	2	79058. 24		$3s^2 3p^3(^4S^\circ)8d$	$8d \ ^3D^\circ$	3 2 1	81663. 4 81666 81668	—3 —2
$3s^2 3p^3(^4S^\circ)5f$	$5f \ ^5F$	5 to 1	79143. 18		$3s^2 3p^3(^4S^\circ)10s$	$10s \ ^5S^\circ$	2	81819. 40	
$3s^2 3p^3(^4S^\circ)5f$	$5f \ ^3F$	4, 3, 2	79144. 45		$3s^2 3p^3(^4S^\circ)8f$	$8f \ ^5F$	5 to 1	[81837. 3]	
$3s^2 3p^3(^4S^\circ)7s$	$7s \ ^3S^\circ$	1	79185. 74		$3s^2 3p^3(^4S^\circ)8f$	$8f \ ^3F$	4, 3, 2	[81837. 9]	
$3s^2 3p^3(^2D^\circ)4p$	$4p' \ ^3P$	2 1 0	79376. 34 79405. 74 79418. 45	—29. 40 —12. 71	$3s^2 3p^3(^4S^\circ)9d$	$9d \ ^5D^\circ$	4 3 2 1 0	82053. 94	
$3s^2 3p^3(^4S^\circ)7p$	$7p \ ^5P$	1 2 3	79785. 72		$3s^2 3p^3(^4S^\circ)10d$	$10d \ ^5D^\circ$	4 3 2 1 0	82353. 3	
$3s^2 3p^3(^4S^\circ)6d$	$6d \ ^5D^\circ$	4 3 2 1 0	79992. 36						
					S II ($^4S_{1/2}$)	Limit	-----	83559. 3	

S I OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 +$	Observed Terms			
$3s^2 3p^4$ $3s 3p^5$	$\left\{ \begin{array}{ccc} 3p^4 \ ^1S & 3p^4 \ ^3P & 3p^4 \ ^1D \\ & 3p^5 \ ^3P^o & \end{array} \right.$			
	$ns \ (n \geq 4)$	$np \ (n \geq 4)$	$nd \ (n \geq 3)$	$nf \ (n \geq 4)$
$3s^2 3p^3(^4S^o)nx$	$\left\{ \begin{array}{l} 4, 6-10s \ ^5S^o \\ 4-8s \ ^3S^o \end{array} \right.$	$\begin{array}{l} 4-7p \ ^5P \\ 4-8p \ ^3P \end{array}$	$\begin{array}{l} 3-10d \ ^5D^o \\ 3-8d \ ^3D^o \end{array}$	$\begin{array}{l} 4-8f \ ^5F \\ 5-7f \ ^3F \end{array}$
$3s^2 3p^3(^2D^o)nx'$	$\left\{ \begin{array}{l} 4s' \ ^3D^o \\ 4s' \ ^1D^o \end{array} \right.$	$\begin{array}{l} 4p' \ ^3P \quad 4p' \ ^3D \quad 4p' \ ^3F \\ 4p' \ ^1F \end{array}$		
$3s^2 3p^3(^2P^o)nx''$	$\left\{ \begin{array}{l} 4s'' \ ^3P^o \\ 4s'' \ ^1P^o \end{array} \right.$			

*For predicted terms in the spectra of the S I isoelectronic sequence, see Introduction.

S II

(P I sequence; 15 electrons)

$Z=16$

Ground state $1s^2 2s^2 2p^6 3s^2 3p^3 \ ^4S_{1\frac{1}{2}}$

$3p^3 \ ^4S_{1\frac{1}{2}} \ 188824.5 \text{ cm}^{-1}$

I. P. 23.4 ± 0.1 volts

The terms are from the paper by Hunter. He has revised and extended the earlier analyses of this spectrum.

The level labeled "x" in his list is here designated "1". The configuration assignments for this level and for the term called "(²P)" in the table are unknown. The latter is attributed by Robinson to $3s^2 3p^2 \ (^3P) 3d$ instead of the term at $118146.50 \text{ cm}^{-1}$.

Intersystem combinations, connecting the doublet and quartet systems of terms, have been established by L. and E. Bloch and confirmed by Hunter. They indicate a correction of $+317.17 \text{ cm}^{-1}$ to the absolute values of the doublet terms published by Ingram.

REFERENCES

- S. B. Ingram, Phys. Rev. **32**, 172 (1928). (I P) (T) (C L)
 L. et E. Bloch, Ann. de Phys. [10] **12**, 5 (1929). (T) (C L)
 M. Gilles, Ann. de Phys. [10] **15**, 301 (1931). (I P) (T) (C L) (Z E)
 O. Bartelt und L. Eckstein, Zeit. Phys. **86**, 77 (1933). (T) (C L)
 A. Hunter, Phil. Trans. Roy. Soc. London [A] **233**, 303 (1934). (I P) (T) (C L)
 H. A. Robinson, Phys. Rev. **49**, 297 (1936).

S II

S II

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2 3p^3$	$3p^3 {}^4S^\circ$	$1\frac{1}{2}$	0. 0		$3s^2 3p^2({}^3P)4p$	$4p {}^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	133268. 53 133399. 82	131. 29
$3s^2 3p^3$	$3p^3 {}^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	14851. 9 14883. 4	31. 5		1	$\frac{1}{2}?$	133359. 4	
$3s^2 3p^3$	$3p^3 {}^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	24524. 2 24572. 8	48. 6		2 (2P) 3	$\frac{1}{2}$ $1\frac{1}{2}$	139845. 6 140015. 7	170. 1
$3s 3p^4$	$3p^4 {}^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	79394. 8 79757. 9 79968. 0	-363. 1 -210. 1	$3s^2 3p^2({}^1D)4p$	$4p' {}^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	140229. 78 140318. 80	89. 02
$3s 3p^4$	$3p^4 {}^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	105599. 02 106044. 16	-445. 14	$3s^2 3p^2({}^1D)4p$	$4p' {}^2D^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$	140708. 51 140750. 00	-41. 49
$3s^2 3p^2({}^3P)4s$	$4s {}^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	109560. 50 109831. 28 110268. 33	270. 78 437. 05	$3s^2 3p^2({}^1D)4p$	$4p' {}^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	143488. 61 143623. 03	134. 42
$3s^2 3p^2({}^3P)3d$	$3d {}^4F$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	110176. 83 110313. 13 110508. 48 110766. 31	136. 30 195. 35 257. 83	$3s^2 3p^2({}^3P)5s$	$5s {}^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	150258. 20 150531. 12 150996. 27	272. 92 465. 15
$3s^2 3p^2({}^3P)4s$	$4s {}^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	112937. 33 113461. 22	523. 89	$3s^2 3p^2({}^3P)5s$	$5s {}^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	151383. 83 151910. 67	526. 84
$3s^2 3p^2({}^3P)3a$	$3d {}^4D$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	114162. 20 114200. 45 114230. 75 114279. 11	38. 25 30. 30 48. 36	$3s^2 3p^2({}^3P)4d$	$4d {}^4F$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	151959. 41 152094. 34 152304. 71 152615. 25	134. 93 210. 37 310. 54
$3s^2 3p^2({}^3P)3d$	$3d {}^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	114804. 11 115285. 31	481. 20	$3s^2 3p^2({}^3P)4d$	$4d {}^4D$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	153153. 66 153201. 72 153282. 80 153413. 52	48. 06 81. 08 130. 72
$3s^2 3p^2({}^3P)3d$	$3d {}^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	115817. 0 115870. 4 115892. 3	-53. 4 -21. 9	$3s^2 3p^2({}^3P)4d$	$4d {}^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	155818. 37 156029. 28 156148. 19	-210. 91 -118. 91
$3s^2 3p^2({}^3P)3d$	$3d {}^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	118146. 50		$3s^2 3p^2({}^3P)4d$	$4d {}^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	156121. 33 156603. 67	482. 34
$3s^2 3p^2({}^3P)3d$	$3d {}^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	119242. 13 119294. 70	52. 57	$3s^2 3p^2({}^3P)4d$	$4d {}^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	158666. 45 158826. 87	160. 42
$3s^2 3p^2({}^1D)4s$	$4s' {}^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	121528. 20 121529. 49	1. 29	$3s^2 3p^2({}^3P)5p$	$5p {}^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	164118. 6 164252. 0 164447. 3 164772. 7	133. 4 195. 3 325. 4
$3s^2 3p^2({}^3P)4p$	$4p {}^2S^\circ$	$\frac{1}{2}$	125485. 32		$3s^2 3p^2({}^1D)4d$	$4d' {}^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	164180. 63 164231. 78	-51. 15
$3s^2 3p^2({}^3P)4p$	$4p {}^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	127824. 93 127976. 21 128233. 07 128599. 11	151. 28 256. 86 366. 04	$3s^2 3p^2({}^3P)5p$	$5p {}^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	164279. 3 164317. 4 164459. 5	38. 1 142. 1
$3s^2 3p^2({}^3P)4p$	$4p {}^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	129787. 71 129858. 07 130134. 08	70. 36 276. 01	$3s^2 3p^2({}^1D)4d$	$4d' {}^2G$	$4\frac{1}{2}$ $3\frac{1}{2}$	164334. 94 164336. 71	-1. 77
$3s^2 3p^2({}^3P)4p$	$4p {}^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	130641. 00 131186. 86	545. 86	$3s^2 3p^2({}^3P)5p$	$5p {}^4S^\circ$	$1\frac{1}{2}$	165002. 45	
$3s^2 3p^2({}^3P)4p$	$4p {}^4S^\circ$	$1\frac{1}{2}$	131028. 76		S III (3P_0)	Limit	-----	188824. 5	

S II OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 +$	Observed Terms		
$3s^2 3p^3$	{ $3p^3 \ ^4S^\circ$ $3p^3 \ ^2P^\circ$ $3p^3 \ ^2D^\circ$		
$3s \ 3p^4$	{ $3p^4 \ ^4P$ $3p^4 \ ^2P$		
	$ns \ (n \geq 4)$	$np \ (n \geq 4)$	$nd \ (n \geq 3)$
$3s^2 3p^2(^3P)nx$	{ $4, 5s \ ^4P$ $4, 5s \ ^2P$	$4, 5p \ ^4S^\circ$ $4, 5p \ ^4P^\circ$ $4, 5p \ ^4D^\circ$ $4p \ ^2S^\circ$ $4p \ ^2P^\circ$ $4p \ ^2D^\circ$	$3, 4d \ ^4P$ $3, 4d \ ^4D$ $3, 4d \ ^4F$ $3d \ ^2P$ $3, 4d \ ^2D$ $3, 4d \ ^2F$
$3s^2 3p^2(^1P)nx'$	$4s' \ ^2D$	$4p' \ ^2P^\circ$ $4p' \ ^2D^\circ$ $4p' \ ^2F^\circ$	$4d' \ ^2F$ $4d' \ ^2G$

*For predicted terms in the spectra of the Pr isoelectronic sequence, see Introduction.

S III

(Si I sequence; 14 electrons)

$Z=16$

Ground state $1s^2 2s^2 2p^6 3s^2 3p^2 \ ^3P_0$

$3p^2 \ ^3P_0$ 282752 cm^{-1}

I. P. 35.0 ± 0.4 volts

The present term list has been compiled from those published by Hunter and by Robinson, although Ingram, Gilles, and others have contributed to the analysis.

Intersystem combinations connecting the singlet and triplet terms have been observed. Robinson derives from his measures a correction of -6 cm^{-1} to be applied to all terms higher than 140000 cm^{-1} . This correction has been introduced here. An estimated value of the interval of $3p^2 \ ^3P_{1,0}^\circ$ is entered in brackets in the table.

The quintet terms suggested by Gilles have been omitted, awaiting further confirmation.

REFERENCES

- S. B. Ingram, Phys. Rev. **33**, 907 (1929). (I P) (T) (C L)
M. Gilles, Ann. de Phys. [10] **15**, 322 (1931). (I P) (T) (C L) (Z E)
A. Hunter, Phil. Trans. Roy. Soc. London [A] **233**, 309 (1934). (I P) (T) (C L)
L. et E. Bloch, J. Phys. Rad. [7] **6**, No. 11, 441 (1935). (C L)
H. A. Robinson, Phys. Rev. **52**, 724 (1937). (T) (C L)

S III

S III

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2 3p^2$	$3p^2\ ^3P$	0	0. 0	297. 2 535. 3	$3s^2 3p(^2P^\circ)3d$	$3d\ ^3D^\circ$	1	147550. 32	140. 67 53. 55
		1	297. 2				2	147690. 99	
		2	832. 5				3	147744. 54	
$3s^2 3p^2$	$3p^2\ ^1D$	2	11320	$3s^2 3p(^2P^\circ)4s$	$4s\ ^1P^\circ$	1	148397. 8		
$3s^2 3p^2$	$3p^2\ ^1S$	0	27163	$3s^2 3p(^2P^\circ)4p$	$4p\ ^3D$	1	169770. 04	297. 27 581. 63	
$3s\ 3p^3$	$3p^3\ ^3D^\circ$	1	84018. 9	27. 5 53. 1	$3s^2 3p(^2P^\circ)4p$	$4p\ ^3P$	2		170067. 31
		2	84046. 4				3		170648. 94
		3	84099. 5				0	172631. 27	
$3s\ 3p^\circ$	$3p^3\ ^3P^\circ$	1	98743. 0	-22. 6 [-6]	$3s^2 3p(^2P^\circ)4p$	$4p\ ^3S$	1	172785. 77	154. 50 405. 96
		0	98765. 6				2	173191. 73	
		2					1	174036. 19	
$3s\ 3p^3$	$3p^3\ ^1D^\circ$	2	104159?	$3s^2 3p(^2P^\circ)4d$	$4d\ ^3F^\circ$	2	204578. 89	491. 86 489. 92	
$3s\ 3p^3$	$3p^3\ ^1P^\circ$	1	136839	$3s^2 3p(^2P^\circ)4d$	$4d\ ^3F^\circ$	3	205070. 75		
						4	205560. 67		
$3s\ 3p^3$	$3p^3\ ^3S^\circ$	1	138061. 4	$3s^2 3p(^2P^\circ)4d$	$4d\ ^3D^\circ$	1	206538. 87	132. 74 239. 36	
$3s^2 3p(^2P^\circ)3d$	$3d\ ^3P^\circ$	0	143095. 91	20. 28 7. 74	$3s^2 3p(^2P^\circ)5s$	$5s\ ^3P^\circ$	2		206671. 61
		1	143116. 19				3		206910. 97
		2	143123. 93				0	209773. 4	
$3s^2 3p(^2P^\circ)4s$	$4s\ ^3P^\circ$	0	146696. 19	40. 35 409. 46	$3s^2 3p(^2P^\circ)5s$	$5s\ ^1P^\circ$	1	209926. 1	152. 7 771. 5
		1	146736. 54				2	210697. 6	
		2	147146. 00				1	211326. 8	
					S IV ($^2P_{3/2}^\circ$)	Limit	-----	282752	

October 1947.

S III OBSERVED TERMS*

Config. $1s^2\ 2s^2\ 2p^6+$	Observed Terms						
$3s^2\ 3p^2$	$\left\{ \begin{array}{lll} 3p^2\ ^1S & 3p^2\ ^3P & 3p^2\ ^1D \end{array} \right.$						
$3s\ 3p^3$	$\left\{ \begin{array}{lll} 3p^3\ ^3S^\circ & 3p^3\ ^3P^\circ & 3p^3\ ^3D^\circ \\ & 3p^3\ ^1P^\circ & 3p^3\ ^1D^\circ \end{array} \right.$						
$3s^2\ 3p(^3P^\circ)nx$	$ns\ (n \geq 4)$		$np\ (n \geq 4)$			$nd\ (n \geq 3)$	
	$\left\{ \begin{array}{l} 4, 5s\ ^3P^\circ \\ 4, 5s\ ^1P^\circ \end{array} \right.$		$4p\ ^3S$	$4p\ ^3P$	$4p\ ^3D$	$3d\ ^3P^\circ$	$3, 4d\ ^3D^\circ\quad 4d\ ^3F^\circ$

*For predicted terms in the spectra of the Si I isoelectronic sequence, see Introduction.

S IV

(Al I sequence; 13 electrons)

 $Z=16$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^2 P_{1/2}^{\circ}$ $3p^2 P_{1/2}^{\circ}$ 381541.4 cm^{-1}

I. P. 47.29 volts

This spectrum is incompletely analyzed but 53 lines have been classified in the range from 519 Å to 3118 Å. For the doublet terms the authors' notation is entered in the first column of the table. The configurations are as given in Bacher and Goudsmit.

The quartet terms are from Bowen's 1932 paper. No intersystem combinations have been observed, as indicated by the uncertainty x . Bowen remarks that the relative positions of the doublet and quartet terms are only approximately determined, by assuming that the difference between the terms $4s^2 S$ and $4s^2 P^{\circ}$ is equal to that between the terms $3s^2 S$ and $3p^2 P^{\circ}$ in S V.

REFERENCES

- R. A. Millikan and I. S. Bowen, Phys. Rev. **25**, 600 (1925). (I P) (T) (C L)
 I. S. Bowen, Phys. Rev. **31**, 37 (1928). (T) (C L)
 R. F. Bacher and S. Goudsmit, *Atomic Energy States* p. 404 (McGraw-Hill Book Co. Inc., New York, N. Y., and London, 1932). (T)
 I. S. Bowen, Phys. Rev. **39**, 13 (1932). (T) (C L)
 L. Bloch et E. Bloch, J. Phys. Rad. [7] **6**, No. 11, 441 (1935). (C L)

S IV

S IV

Authors	Config.	Desig.	J	Level	Interval	Authors	Config.	Desig.	J	Level	Interval
$3p_2$ $3p_1$	$3s^2(^1S)3p$	$3p^2 P^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$	0.0 950.2	950.2	$4p_2$ $4p_1$	$3s^2(^1S)4p$	$4p^2 P^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$	213507.4 213717.4	210.0
	$3s^2 3p^2$	$3p^2 ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$71840 +x$ $72184 +x$ $72731 +x$	344 547		$3s^2 3p(^3P^{\circ})3d$	$3d^2 ^4P^{\circ}$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	$222854 +x$ $223143 +x$	-289
bD_2 bD_3	$3s^2 3p^2$	$3p^2 ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	94101.9 94148.1	46.2		$3s^2 3p(^3P^{\circ})3d$	$3d^2 ^4D^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	$224991 +x$ $225094 +x$ $225194 +x$ $225274 +x$	103 100 80
bS	$3s^2 3p^2$	$3p^2 ^2S$	$\frac{1}{2}$	123503.9							
bP_1 bP_2	$3s^2 3p^2$	$3p^2 ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	133617.9 134243.9	626.0	$4d$	$3s^2(^1S)4d$	$4d^2 D$	$\left\{ \frac{1}{2} \right.$ $\left. \frac{2}{2} \right\}$	255389.8	
$3d_2$ $3d_1$	$3s^2(^1S)3d$	$3d^2 ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	152127.1 152141.4	14.3		$3s^2 3p(^3P^{\circ})4s$	$4s^2 ^4P^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$263759 +x$ $264105 +x$ $264741 +x$	346 636
$4s$	$3s^2(^1S)4s$	$4s^2 ^2S$	$\frac{1}{2}$	181432.2		$5s$	$3s^2(^1S)5s$	$5s^2 ^2S$	$\frac{1}{2}$	271010.4	
	$3p^3$	$3p^3 ^4S^{\circ}$	$1\frac{1}{2}$	$197110 +x$							
cP	$3p^3$	$3p^3 ^2P^{\circ}$	$\left\{ \frac{1}{2} \right.$ $\left. \frac{1}{2} \right\}$	211368			S V (1S_0)	Limit	-----	381541.4	

September 1947.

S IV OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ +	Observed Terms		
3s ² (¹ S)3p	3p ² P°		
3s 3p ²	{	3p ² ² S	3p ² ⁴ P 3p ² ² P
3p ³		3p ³ ⁴ S°	3p ² ² D 3p ³ ² P°
	ns (n ≥ 4)		np (n ≥ 4) nd (n ≥ 3)
3s ² (¹ S)nx	4, 5s ² S		4p ² P° 3, 4d ² D
3s 3p(³ P°)nx	4s ⁴ P°		3d ⁴ P° 3d ⁴ D°

*For predicted terms in the spectra of the Al I isoelectronic sequence, see Introduction.

S V

(Mg I sequence; 12 electrons)Z=16

Ground state 1s² 2s² 2p⁶ 3s² ¹S₀

3s² ¹S₀ 584700 cm⁻¹I. P. 72.5 ± volts

This spectrum is incompletely analyzed, but Bowen has classified 30 lines in the range between 437 Å and 905 Å. He gives absolute values for only the triplet terms, but lists the singlet combination 3s² ¹S₀—3p ¹P₁^o, which has been used to calculate 3p ¹P₁^o in the table.

By extrapolation along the isoelectronic sequence the writer has estimated the limit 3s² ¹S₀ as approximately 584700 cm⁻¹, which places 3p ³P₀^o at 83071 cm⁻¹ above the ground state zero. These estimated values are entered in brackets in the table. The uncertainty, *x*, may be several hundred cm⁻¹. Bowen has estimated the error of the limit as probably not greater than ±1000 cm⁻¹.

REFERENCES

I. S. Bowen, Phys. Rev. **39**, 8 (1932). (T) (C L)
I. S. Bowen, letter (Sept. 1947). (T)

S V

S V

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval		
3s ²	3s ² ¹ S	0	0		3s(2S)4s	4s ³ S	1	311670 + <i>x</i>			
3s(2S)3p	3p ³ P°	0	[83071] + <i>x</i>	362 767	3p(2P°)3d	3d ³ P°	2	345376 + <i>x</i>	-374 -237		
		1	83433 + <i>x</i>				1	345750 + <i>x</i>			
		2	84200 + <i>x</i>				0	345987 + <i>x</i>			
3s(2S)3p	3p ¹ P°	1	127149		3p(2P°)3d	3d ³ D°	1	347883 + <i>x</i>	168 117		
3p ²	3p ² ³ P	0	200000 + <i>x</i>	417 769						2	348051 + <i>x</i>
		1	200417 + <i>x</i>							3	348168 + <i>x</i>
		2	201186 + <i>x</i>		-----						
3s(2S)3d	3d ³ D	1, 2, 3	234987 + <i>x</i>		S VI (2S _{1/2})	<i>Limit</i>	-----	[584700]			

S v OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ +	Observed Terms	
3s ²	3s ² 1S	
3s(2S)3p	{ 3p 3P° 3p 1P°	
3p ²	3p ² 3P	
	ns (n ≥ 4)	nd (n ≥ 3)
3s(2S)nx	4s 3S	3d 3D
3p(2P°)nx		3d 3P° 3d 3D°

*For predicted terms in the spectra of the Mg I isoelectronic sequence, see Introduction.

S VI

(Na I sequence; 11 electrons)

Z=16

Ground state 1s² 2s² 2p⁶ 3s 2S_{1/2}

3s 2S_{1/2} 710194 cm⁻¹

I. P. 88.029 ± 0.003 volts

The terms are from Robinson, who has extended the earlier analysis by Bowen and Millikan. There are 29 classified lines, all but 2 of which are in the region between 171 Å and 1117 Å. The absolute value of the ground state was extrapolated along the isoelectronic sequence.

REFERENCES

I. S. Bowen and R. A. Millikan, Phys. Rev. **25**, 295 (1925). (T) (C L)
H. A. Robinson, Phys. Rev. **52**, 724 (1937). (I P) (T) (C L)

S VIS VI

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
3s	3s 2S	1/2	0		5f	5f 2F° { 2 1/2 3 1/2 }		551848	
3p	3p 2P°	1/2 1 1/2	105874 107137	1263	5g	5g 2G { 3 1/2 4 1/2 }		552106	
3d	3d 2D	1 1/2 2 1/2	247420 247452	32	6s	6s 2S	1/2	573823	
4s	4s 2S	1/2	362983		6p	6p 2P°	1/2 1 1/2	583679	
4p	4p 2P°	1/2 1 1/2	401164 401621	457	6d	6d 2D { 1 1/2 2 1/2 }		596877	
4d	4d 2D	1 1/2 2 1/2	451785 451808	23	6f	6f 2F° { 2 1/2 3 1/2 }		600170	
4f	4f 2F° { 2 1/2 3 1/2 }		462653		7d	7d 2D { 1 1/2 2 1/2 }		627231	
5s	5s 2S	1/2	504112		7f	7f 2F° { 2 1/2 3 1/2 }		629395	
5p	5p 2P°	1/2 1 1/2	522030 522248	218					
5d	5d 2D	1 1/2 2 1/2	546021 546032	11	S VII (1S ₀)	Limit		710194	

S VII

(Ne I sequence; 10 electrons)

 $Z=16$ Ground state $1s^2 2s^2 2p^6 {}^1S_0$ $2p^6 {}^1S_0$ 2266990 cm^{-1}

I. P. 280.99 volts

Ferner has classified 16 lines between 46 Å and 72 Å as combinations with the ground term, and generously furnished his analysis in advance of publication. The term designations he assigns on the assumption of LS -coupling are given in the table under the heading "Author."

As for Ne I, the $j\bar{l}$ -coupling notation in the general form suggested by Racah is introduced. Ferner's unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

G. Racah, Phys. Rev. **61**, 537 (L) (1942).E. Ferner, Ark. Mat. Astr. Fys. (Stockholm) **36A**, No. 1, p. 62 (1948). (I P) (T) (C L)

S VII

S VII

Author	Config.	Desig.	J	Level	Author	Config.	Desig.	J	Level
$2p {}^1S_0$	$2p^6$	$2p^6 {}^1S$	0	0					
	$2p^5({}^2P_{1/2})3s$	$3s [1\frac{1}{2}]^\circ$	2		$5s {}^3P_1$	$2p^5({}^2P_{1/2})5s$	$5s [1\frac{1}{2}]^\circ$	$\frac{2}{1}$	1998920
$3s {}^3P_1$			1	1376220					
	$2p^5({}^2P_{3/2}^\circ)3s$	$3s' [\frac{1}{2}]^\circ$	0		$3p' \left\{ \begin{smallmatrix} {}^3P_1 \\ {}^1P_1 \end{smallmatrix} \right\}$	$2s 2p^6({}^2S)3p$	$3p \left\{ \begin{smallmatrix} {}^3P^\circ \\ {}^1P^\circ \end{smallmatrix} \right\}$	1	2000400
$3s {}^1P_1$			1	1388330					
	$2p^5({}^2P_{1/2})3d$	$3d [\frac{1}{2}]^\circ$	0		$5d {}^1P_1$	$2p^5({}^2P_{1/2})5d$	$5d [1\frac{1}{2}]^\circ$	1	2046080
$3d {}^3P_1$			1	1624770	$5d {}^3D_1$	$2p^5({}^2P_{3/2}^\circ)5d$	$5d' [1\frac{1}{2}]^\circ$	1	2055630
$3d {}^1P_1$	"	$3d [1\frac{1}{2}]^\circ$	1	1644630					
$3d {}^3D_1$	$2p^5({}^2P_{3/2}^\circ)3d$	$3d' [1\frac{1}{2}]^\circ$	1	1662210	$6d {}^1P_1$	$2p^5({}^2P_{1/2})6d$	$6d [1\frac{1}{2}]^\circ$	1	2113850
					$6d {}^3D_1$	$2p^5({}^2P_{3/2}^\circ)6d$	$6d' [1\frac{1}{2}]^\circ$	1	2123230
	$2p^5({}^2P_{1/2})4s$	$4s [1\frac{1}{2}]^\circ$	2						
$4s {}^3P_1$			1	1820230	$7d {}^3D_1$	$2p^5({}^2P_{3/2}^\circ)7d$	$7d' [1\frac{1}{2}]^\circ$	1	2163940
	$2p^5({}^2P_{3/2}^\circ)4s$	$4s' [\frac{1}{2}]^\circ$	0						
$4s {}^1P_1$			1	1829760					
						S VIII (${}^2P_{1/2}$)	Limit	-----	2266990
$4d {}^1P_1$	$2p^5({}^2P_{1/2})4d$	$4d [1\frac{1}{2}]^\circ$	1	1919500		S VIII (${}^2P_{3/2}$)	Limit	-----	2277120
$4d {}^3D_1$	$2p^5({}^2P_{3/2}^\circ)4d$	$4d' [1\frac{1}{2}]^\circ$	1	1930240					

August 1947.

S VII OBSERVED LEVELS*

Config. $1s^2\ 2s^2+$	Observed Terms		
$2p^6$	$2p^6\ ^1S$		
	$ns\ (n\geq 3)$	$nd\ (n\geq 3)$	$np\ (n\geq 3)$
	$2p^5(^2P^\circ)nx$	$\left\{\begin{array}{l}3-5s\ ^3P^\circ\\3-4s\ ^1P^\circ\end{array}\right.$	$\begin{array}{l}3d\ ^3P^\circ\quad 3-7d\ ^3D^\circ\\3-6d\ ^1P^\circ\end{array}$
$2p^6(^2S)nx$			
jl -Coupling Notation			
	Observed Pairs		
	$ns\ (n\geq 3)$	$nd\ (n\geq 3)$	
	$2p^5(^2P_{1/2}^\circ)nx$	$3-5s\ [1\frac{1}{2}]^\circ$	$\begin{array}{l}3d\ [1\frac{1}{2}]^\circ\\3-6d\ [1\frac{1}{2}]^\circ\end{array}$
$2p^5(^2P_{3/2}^\circ)nx'$	$3-4s'\ [1\frac{1}{2}]^\circ$	$3-7d'\ [1\frac{1}{2}]^\circ$	

*For predicted levels in the spectra of the Ne I isoelectronic sequence, see Introduction.

S VIII

(F I sequence; 9 electrons)

$Z=16$

Ground state $1s^2 2s^2 2p^5 \ ^2P_{1/2}^\circ$

$2p^5 \ ^2P_{1/2}^\circ$ **2652720** cm^{-1}

I. P. 328.80 volts

The analysis was furnished by Ferner in advance of publication. He has classified 44 lines in the interval between 44 Å and 65 Å. All but one of the observed combinations are with the ground term. In addition, Robinson has classified a pair of lines at 202.605 Å and 198.550 Å as $2p^5 \ ^2P^\circ - 2p^6 \ ^2S$.

Ferner's unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

- H. A. Robinson, Phys. Rev. **52**, 724 (1937). (C L)
 E. Ferner, Ark. Mat. Astr. Fys. (Stockholm) **36A**, No. 1, p. 57 (1948). (I P) (T) (C L)

S VIII

S VIII

Author	Config.	Desig.	J	Level	Interval	Author	Config.	Desig.	J	Level	Interval
$2p \ ^2P_2$ $\ ^2P_1$	$2s^2 2p^5$	$2p^5 \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	0 10130	-10130	$\overline{3d} \ ^2S_1$	$2s^2 2p^4(^1D)3d$	$3d' \ ^2S$	$\frac{1}{2}$	1894330	
$2p' \ ^2S_1$	$2s \ 2p^6$	$2p^6 \ ^2S$	$\frac{1}{2}$	503590		$\overline{3d} \ ^2F_3$	$2s^2 2p^4(^1D)3d$	$3d' \ ^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	1895520	
$3s \ ^4P_3$ $\ ^4P_2$ $\ ^4P_1$	$2s^2 2p^4(^3P)3s$	$3s \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	1559580 1565250 1569290	-5670 -4040	$\overline{3d} \ ^2D_3$ $\ ^2D_2$	$2s^2 2p^4(^1S)3d$	$3d'' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	1952100 1953010	-910
$3s \ ^2P_2$ $\ ^2P_1$	$2s^2 2p^4(^3P)3s$	$3s \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	1579700 1586650	-6950	$3s' \ ^2P_2$ $\ ^2P_1$	$2s \ 2p^5(^3P^\circ)3s$	$3s''' \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	2038530 2045040	-6510
$\overline{3s} \ ^2D_3$ $\ ^2D_2$	$2s^2 2p^4(^1D)3s$	$3s' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	1623380 1623610	-230	$4s \ ^2P_2$ $\ ^2P_1$	$2s^2 2p^4(^3P)4s$	$4s \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	2102340 2111240	-8900
$\overline{3s} \ ^2S_1$	$2s^2 2p^4(^1S)3s$	$3s'' \ ^2S$	$\frac{1}{2}$	1688170			$2s^2 2p^4(^3P)4d$	$4d \ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$		
	$2s^2 2p^4(^3P)3d$	$3d \ ^4D$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$			$4d \ ^4P_3$				2199830	
$3d \ ^4D_3$				1831370 1822510		$4d \ ^2D_2$ $\ ^2D_3$	$2s^2 2p^4(^3P)4d$	$4d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	2204100 2208530	4430
	$2s^2 2p^4(^3P)3d$	$3d \ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$		3910	$4d \ ^2P_2$	$2s^2 2p^4(^3P)4d$	$4d \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	2207770	
$3d \ ^2P_1$ $\ ^2P_2$	$2s^2 2p^4(^3P)3d$	$3d \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	1839250 1847550	8300	$\overline{4d} \ ^2S_1$	$2s^2 2p^4(^1D)4d$	$4d' \ ^2S$	$\frac{1}{2}$	2253570	
$3d \ ^2D_2$ $\ ^2D_3$	$2s^2 2p^4(^3P)3d$	$3d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	1842770 1847810	5040	$\overline{4d} \ ^2F_3$	$2s^2 2p^4(^1D)4d$	$4d' \ ^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	2254790	
$\overline{3d} \ ^2P_1$ $\ ^2P_2$	$2s^2 2p^4(^1D)3d$	$3d' \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	1888460 1897460	9000						
$\overline{3d} \ ^2D_3$ $\ ^2D_2$	$2s^2 2p^4(^1D)3d$	$3d' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	1892000 1898220	-6220						
							SIX (3P_2)	Limit			2652720

August 1947.

S VIII OBSERVED TERMS*

Config. 1s ² +	Observed Terms			
2s ² 2p ⁵	2p ⁵ ²P°			
2s 2p ⁶	2p ⁶ ²S			
	ns (n≥3)	nd (n≥3)		
2s ² 2p ⁴ (³P)nx	{ 3s ⁴P 3, 4s ²P	3, 4d ⁴P 3, 4d ²P	3d ⁴D 3, 4d ²D	
2s ² 2p ⁴ (¹D)nx'	3s¹ ²D	3, 4d' ²S	3d' ²P	3d' ²D 3, 4d' ²F
2s ² 2p ⁴ (¹S)nx''	3s'' ²S		3d'' ²D	
2s 2p ⁵ (³P°)nx'''	3s''' ²P°			

*For predicted terms in the spectra of the FI isoelectronic sequence, see Introduction.

S IX

(O I sequence; 8 electrons)

 $Z=16$ Ground state $1s^2 2s^2 2p^4 {}^3P_2$ $2p^4 {}^3P_2$ 3057300 cm^{-1}

I. P. 378.95 volts

Ferner has found 17 terms and classified 21 lines in this spectrum in the range from 46 Å to 56 Å. No intersystem combinations have been observed and the uncertainty, \pm , may be large. The unit adopted by Ferner, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCE

E. Ferner, Ark. Mat. Astr. Fys. (Stockholm) **36A**, No. 1, p. 48 (1948). (I P) (T) (C L)

S IX

S IX

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$2s^2 2p^4$	$2p^4 {}^3P$	2 1 0	0 7970 10630	—7970 —2660	$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^1D^\circ$	2	$2117140+x$	1790
					$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^3S^\circ$	1	2125310	
$2s^2 2p^4$	$2p^4 {}^1D$	2	$58000+x$		$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^1F^\circ$	3	$2134410+x$	
$2s^2 2p^4$	$2p^4 {}^1S$	0	$122300+x$		$2s^2 2p^3({}^2P^\circ)3d$	$3d'' {}^3P^\circ$	0 1 2	2144820 2146610	
$2s^2 2p^3({}^4S^\circ)3s$	$3s {}^3S^\circ$	1	1783150						
$2s^2 2p^3({}^2D^\circ)3s$	$3s' {}^3D^\circ$	3 2 1	1845770 1846340	—570	$2s^2 2p^3({}^2P^\circ)3d$	$3d'' {}^3F^\circ$	4 3 2	2154570	
$2s^2 2p^3({}^2D^\circ)3s$	$3s' {}^1D^\circ$	2	$1858500+x$		$2s^2 2p^3({}^2P^\circ)3d$	$3d'' {}^3D^\circ$	3 2 1	2156430	
$2s^2 2p^3({}^2P^\circ)3s$	$3s'' {}^1P^\circ$	1	$1904040+x$						
$2s^2 2p^3({}^4S^\circ)3d$	$3d {}^3D^\circ$	1, 2 3	2035220 2035870	650	$2s^2 2p^3({}^2P^\circ)3d$	$3d'' {}^1P^\circ$	1	$2162470+x$	
$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^3D^\circ$	3, 2, 1	2108190		S x (${}^4S_{1/2}$)		Limit	3057300	
$2s^2 2p^3({}^2D^\circ)3d$	$3d' {}^3P^\circ$	2 1 0	2116450 2119180	—2730					

August 1947.

S IX OBSERVED TERMS*

Config. $1s^2 +$	Observed Terms	
$2s^2 2p^4$	$\{ 2p^4 {}^1S \quad 2p^4 {}^3P \quad 2p^4 {}^1D$	
	$ns (n \geq 3)$	$nd (n \geq 3)$
$2s^2 2p^3({}^4S^\circ)nx$	$3s {}^3S^\circ$	$3d {}^1D^\circ$
$2s^2 2p^3({}^2D^\circ)nx'$	$\{ \quad \quad \quad 3s' {}^3D^\circ$ $\quad \quad \quad 3s' {}^1D^\circ$	$3d' {}^3S^\circ \quad 3d' {}^3P^\circ \quad 3d' {}^3D^\circ \quad 3d' {}^1D^\circ \quad 3d' {}^1F^\circ$
$2s^2 2p^3({}^2P^\circ)nx''$	$\{ \quad \quad \quad 3s'' {}^1P^\circ$	$3d'' {}^3P^\circ \quad 3d'' {}^3D^\circ \quad 3d'' {}^3F^\circ$ $3d'' {}^1P^\circ$

*For predicted terms in the spectra of the O I isoelectronic sequence, see Introduction.

S x

(N I sequence; 7 electrons)

Z=16

Ground state $1s^2 2s^2 2p^3 \text{}^4\text{S}_{1/2}^{\circ}$

$2p^3 \text{}^4\text{S}_{1/2}^{\circ}$ 3615900 cm⁻¹

I. P. 448.2 volts

The spectrum is very incompletely analyzed. Ferner has classified 4 lines between 44 Å and 47 Å and has generously furnished these classifications in advance of publication. The terms in the table have been derived from Ferner's data, adjusted by Robinson to fit the isoelectronic sequence data. All entries in brackets have been extrapolated along the isoelectronic sequence by Robinson. No intersystem combinations have been observed and the uncertainty, x , probably exceeds $\pm 1000\text{ cm}^{-1}$.
Ferner's unit, 10^3 cm^{-1} , has been changed to cm^{-1} in deriving the term values.

REFERENCES

E. Ferner, Ark. Mat. Astr. Fys. (Stockholm) **36A**, No. 1, p. 42 (1948). (C L)
H. A. Robinson, unpublished material (March 1948). (I P) (T)

S x

Config.	Desig.	<i>J</i>	Level	Interval
$2s^2 2p^3$	$2p^3 \text{}^4\text{S}^{\circ}$	$1\frac{1}{2}$	0	
$2s^2 2p^3$	$2p^3 \text{}^2\text{P}^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$	$[122230]+x$ $[123730]+x$	[1500]
$2s^2 2p^2(^3\text{P})3s$	$3s \text{}^4\text{P}$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	2092360 2098460	6100
$2s^2 2p^2(^3\text{P})3d$	$3d \text{}^2\text{D}$	$1\frac{1}{2}$ $2\frac{1}{2}$	$2375140 + x$ $2377300 + x$	2160
-----	-----	---	-----	
S xI ($^3\text{P}_0$)	Limit	---	[3615900]	

March 1948.

S xII

(B I sequence; 5 electrons)

Z=16

Ground state $1s^2 2s^2 2p \text{}^2\text{P}_{1/2}^{\circ}$

$2p \text{}^2\text{P}_{1/2}^{\circ}$ cm⁻¹

I. P. volts

By extrapolation along the B I isoelectronic sequence, Edlén estimates that the separation of the lowest term, $2p \text{}^2\text{P}_{1/2}^{\circ} - 2p \text{}^2\text{P}_{1/2}$, is 13266 cm^{-1} (7536 Å).

REFERENCE

B. Edlén, Zeit. Astroph. **22**, 58 (1942). (T)
July 1948.

CHLORINE

Cl I

17 electrons

 $Z=17$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^5 {}^2P_{1/2}^{\circ}$ $3p^5 {}^2P_{1/2}^{\circ}$ 104991 cm^{-1}

I. P. 13.01 volts

Most of the terms are from the analysis by Kiess, who has revised and extended the earlier work on this spectrum. Green and Lynn have observed the Zeeman effect and, with the aid of g -values, added a few terms to the list by Kiess. They list 11 unclassified lines for which both g -values are known.

Their miscellaneous levels are labeled in the table with numbers assigned by the writer, followed by their tentative designations entered in parentheses.

Intersystem combinations, connecting the doublet and quartet terms, have been observed

REFERENCES

- L. A. Turner, Phys. Rev. **27**, 401 (1926). (C L)
 O. Laporte, Nature **121**, 1021 (1928). (C L)
 C. C. Kiess, Bur. Std. J. Research **10**, 827, RP570 (1933). (I P) (T) (C L)
 B. Edlén, Zeit. Phys. **104**, 413 (1937). (I P) (C L)
 W. F. Meggers, J. Opt. Soc. Am. **36**, 431 (1946). (Summary hfs)
 J. B. Green and J. T. Lynn, Phys. Rev. **69**, 165 (1946). (T) (C L) (Z E)
 L. Davis, Jr., B. T. Feld, C. W. Zabel, and J. R. Zacharias, Phys. Rev. **73**, 525 (L) (1948). (hfs)

Cl I

Cl I

Config.	Desig.	J	Level	Interval	Obs. g	Config.	Desig.	J	Level	Interval	Obs. g
$3s^2 3p^5$	$3p^5 {}^2P^{\circ}$	$1\frac{1}{2}$ $\frac{1}{2}$	0 881	—881		$3s^2 3p^4({}^3P)4p$	$4p {}^2S^{\circ}$	$\frac{1}{2}$	85239.98		1.280
$3s^2 3p^4({}^3P)4s$	$4s {}^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	71954.00 72484.20 72822.64	—530.20 —338.44	1.599 1.722 2.652	$3s^2 3p^4({}^3P)4p$	$4p {}^2P^{\circ}$	$1\frac{1}{2}$ $\frac{1}{2}$	85438.04 85913.44	—475.40	1.327 1.379
$3s^2 3p^4({}^3P)4s$	$4s {}^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	74221.44 74861.24	—639.80	1.340 0.663	$3s^2 3p^4({}^3P)4p$	$4p {}^4S^{\circ}$	$1\frac{1}{2}$	85730.68		1.877
$3s^2 3p^4({}^3P)4p$	$4p {}^4P^{\circ}$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	82914.54 83126.59 83360.55	—212.05 —233.96	1.591 1.723 2.617	$3s^2 3p^4({}^1D)4p$	$4p' {}^2P^{\circ}$	$1\frac{1}{2}$ $\frac{1}{2}$	94309.67 94464.50	—154.83	1.328 0.872
$3s^2 3p^4({}^3P)4p$	$4p {}^4D^{\circ}$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	83889.64 84127.90 84480.91 84684.27	—238.26 —353.01 —203.36	1.422 1.308 1.163 0.059	$3s^2 3p^4({}^3P)5p$	$5p {}^4P^{\circ}$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	94477.93 94659.28 94969.43	—181.35 —310.15	1.559 1.722 2.309
$3s^2 3p^4({}^3P)4p$	$4p {}^4D^{\circ}$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	83889.64 84127.90 84480.91 84684.27	—238.26 —353.01 —203.36	1.422 1.308 1.163 0.059	$3s^2 3p^4({}^3P)5p$	$5p {}^4D^{\circ}$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	94727.91 94822.75 95309.43 95530.51	—94.84 —486.68 —221.08	1.420 1.247 1.147 1.409
$3s^2 3p^4({}^1D)4s$	$4s' {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	84115.68 84117.38	—1.70		$3s^2 3p^4({}^1D)4p$	$4p' {}^2F^{\circ}$	$2\frac{1}{2}$ $3\frac{1}{2}$	95140.05 95176.00	35.95	
$3s^2 3p^4({}^3P)4p$	$4p {}^2D^{\circ}$	$2\frac{1}{2}$ $1\frac{1}{2}$	84643.69 84984.04	—340.35	1.269 0.986	$3s^2 3p^4({}^3P)5p$	$5p {}^2D^{\circ}$	$2\frac{1}{2}$ $1\frac{1}{2}$	95396.31 95702.01	—305.70	1.352 1.321

Cl I—Continued

Cl I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$3s^2 3p^4(^3P)5p$	$5p\ ^2S^\circ$	$\frac{1}{2}$	95593. 28		0. 699	$3s^2 3p^4(^3P)5d$	$5d\ ^4F$	$4\frac{1}{2}$	99513. 68	—150. 47	1. 310
$3s^2 3p^4(^3P)5p$	$5p\ ^4S^\circ$	$1\frac{1}{2}$	95608. 30		1. 531			$3\frac{1}{2}$	99664. 15	—97. 37	1. 181
$3s^2 3p^4(^3P)4d$	$4d\ ^4D$	$3\frac{1}{2}$	95696. 49	—85. 92	1. 367			$2\frac{1}{2}$	99761. 52	—183. 90	1. 149
		$2\frac{1}{2}$	95782. 41	—110. 75	1. 209			$1\frac{1}{2}$	99945. 42		1. 240
		$1\frac{1}{2}$	95893. 16	—98. 02	0. 00	$3s^2 3p^4(^3P)4d$	$4d\ ^2P$	$1\frac{1}{2}$	99530. 10	—176. 90	1. 306
		$\frac{1}{2}$	95991. 18					$\frac{1}{2}$	99707. 00		1. 289
$3s^2 3p^4(^3P)5p$	$5p\ ^2P^\circ$	$1\frac{1}{2}$	96308. 84	—280. 80	1. 286	$3s^2 3p^4(^3P)6p$	$1^\circ\ (^2D?)$	$1\frac{1}{2}$	99564. 7		1. 32
		$\frac{1}{2}$	96589. 64		0. 712	$3s^2 3p^4(^3P)6p$	$2^\circ\ (^4D^\circ?)$	$\frac{1}{2}$	99582. 7		0. 49
$3s^2 3p^4(^1D)4p$	$4p'\ ^2D^\circ$	$2\frac{1}{2}$	96478. 38	—3. 32	0. 867	$3s^2 3p^4(^3P)7s?$	1 ($^4P?$)	$1\frac{1}{2}$	99677. 1		1. 73
		$1\frac{1}{2}$	96481. 70			$3s^2 3p^4(^3P)6p$	$6p\ ^2P^\circ$	$1\frac{1}{2}$	99819. 8	—79. 4	1. 28
$3s^2 3p^4(^3P)4d$	$4d\ ^4F$	$4\frac{1}{2}$	96490. 40	—236. 41	1. 097			$\frac{1}{2}$	99899. 2		0. 81
		$3\frac{1}{2}$	96726. 81	—214. 49	0. 967	$3s^2 3p^4(^3P)7s?$	2 ($^2P?$)	$\frac{1}{2}$	99968. 1		1. 21
		$2\frac{1}{2}$	96941. 30	—314. 25		$3s^2 3p^4(^3P)5d$	$5d\ ^4P$	$2\frac{1}{2}$	99984. 30	—248. 70	1. 589
$3s^2 3p^4(^3P)4d$	$4d\ ^2F$	$3\frac{1}{2}$	96829. 85	—350. 09				$1\frac{1}{2}$	100233. 00	65. 88	1. 470
		$2\frac{1}{2}$	97179. 94					$\frac{1}{2}$	100167. 12		
$3s^2 3p^4(^3P)6s$	$6s\ ^4P$	$2\frac{1}{2}$	97233. 37	—242. 83	1. 500	$3s^2 3p^4(^3P)7s?$	3 ($^2P?$)	$1\frac{1}{2}$	100046. 5		1. 42
		$1\frac{1}{2}$	97476. 20	—619. 76	1. 393	$3s^2 3p^4(^3P)5d$	$5d\ ^2F$	$3\frac{1}{2}$	100142. 41	—442. 87	1. 210
		$\frac{1}{2}$	98095. 96		1. 962			$2\frac{1}{2}$	100585. 28		1. 069
$3s^2 3p^4(^3P)4d$	$4d\ ^4P$	$2\frac{1}{2}$	97334. 60	—706. 20	1. 241	$3s^2 3p^4(^3P)5d$	$5d\ ^2D$	$2\frac{1}{2}$	100245. 32	—97. 66	
		$1\frac{1}{2}$	98040. 80	—600. 42	1. 620			$1\frac{1}{2}$	100342. 98		
		$\frac{1}{2}$	98641. 22			$3s^2 3p^4(^3P)5d$	$5d\ ^2P$	$1\frac{1}{2}$	100700. 3	—33. 1	1. 65
$3s^2 3p^4(^3P)4d$	$4d\ ^2D$	$2\frac{1}{2}$	97529. 85	—273. 61	1. 355			$\frac{1}{2}$	100733. 4		1. 59
		$1\frac{1}{2}$	97803. 46			$3s^2 3p^4(^3P)6d$	$6d\ ^4D$	$\frac{1}{2}$	100941. 9	99. 7	1. 010
$3s^2 3p^4(^3P)6p$	$6p\ ^4P^\circ$	$2\frac{1}{2}$			1. 91			$1\frac{1}{2}$	101041. 6	6. 9	1. 168
		$1\frac{1}{2}$	98911. 6					$2\frac{1}{2}$	101048. 47	—62. 87	1. 364
		$\frac{1}{2}$						$3\frac{1}{2}$	100985. 60		1. 377
$3s^2 3p^4(^3P)6p$	$6p\ ^4D^\circ$	$3\frac{1}{2}$			1. 32	$3s^2 3p^4(^3P)6d$	4 ($^4F?$)	$1\frac{1}{2}?$	101219. 0		1. 20
		$2\frac{1}{2}$	99015. 1			$3s^2 3p^4(^3P)6d$	5 ($^4P?$)	$2\frac{1}{2}$	101422. 4		1. 60
		$1\frac{1}{2}$				$3s^2 3p^4(^3P)6d$	6	$\frac{1}{2}$	101587. 4		0. 69
$3s^2 3p^4(^3P)5d$	$5d\ ^4D$	$3\frac{1}{2}$	99196. 02	—68. 69	1. 392	$3s^2 3p^4(^3P)6d$	7 ($^2F?$)	$2\frac{1}{2}$	101855. 0		1. 45
		$2\frac{1}{2}$	99264. 71	—85. 51	1. 358						
		$1\frac{1}{2}$	99350. 22	—53. 39	0. 363						
		$\frac{1}{2}$	99403. 61								
						Cl II (3P_2)	Limit	-----	104991		

January 1948.

Cl I OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 +$	Observed Terms					
$3s^2 3p^5$	$3p^5\ ^2P^\circ$					
	$ns\ (n \geq 4)$	$np\ (n \geq 4)$			$nd\ (n \geq 3)$	
$3s^2 3p^4(^3P)nx$	$\left\{ \begin{array}{l} 4, 6s\ ^4P \\ 4s\ ^2P \end{array} \right.$	$4, 5p\ ^4S^\circ$	$4-6p\ ^4P^\circ$	$4-6p\ ^4D^\circ$	$4, 5d\ ^4P$	$4-6d\ ^4D$
		$4, 5p\ ^2S^\circ$	$4-6p\ ^2P^\circ$	$4, 5p\ ^2D^\circ$	$4, 5d\ ^2P$	$4, 5d\ ^2D$
$3s^2 3p^4(^1D)nx'$	$4s'\ ^2D$	$4p'\ ^2P^\circ$	$4p'\ ^2D^\circ$	$4p'\ ^2F^\circ$	$4, 5d\ ^4F$	$4, 5d\ ^4F$

*For predicted terms in the spectra of the Cl I isoelectronic sequence, see Introduction.

Cl II

(S I sequence; 16 electrons)

Z=17

Ground state $1s^2 2s^2 2p^6 3s^2 3p^4 {}^3P_2$ $3p^4 {}^3P_2$ 192000 cm^{-1}

I. P. 23.80 volts

The terms are from the paper by Kiess and de Bruin, who have summarized, revised, and extended the earlier analysis by Murakawa and others. They give a complete list of classified lines; it extends from 558 Å to 9483 Å. Intersystem combinations connecting all three systems of terms, have been observed.

The two unclassified levels designated by them as x' and x'' are here labeled 1 and 2, respectively. The term they list as $4s' {}^3P$ is entered as " 3P " since its configuration is not definitely known.

The estimated position of $3p^4 {}^1S$ given by Edlén, is entered in brackets in the table.

REFERENCES

- C. C. Kiess and T. L. de Bruin, J. Research Nat. Bur. Std. **23**, 443, RP1244 (1939). (I P) (T) (C L) (G D)
 B. Edlén, Phys. Rev. **62**, 434 (1942). (T)
 S. Tolansky, Zeit. Phys. **74**, 336 (1932). (hfs)
 S. Tolansky, Zeit. Phys. **73**, 470 (1931). (I S)

Cl II

Cl II

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2 3p^4$	$3p^4 {}^3P$	2 1 0	0 697 996	—697 —299	$3s^2 3p^3({}^2D^\circ)3d$	$3d' {}^1P^\circ$	1	127726.9	
					$3s^2 3p^3({}^4S^\circ)4p$	$4p {}^5P$	1 2 3	128621.9 128662.5 128729.8	40.6 67.3
$3s^2 3p^4$	$3p^4 {}^1D$	2	11652						
$3s^2 3p^4$	$3p^4 {}^1S$	0	[27900]		$3s^2 3p^3({}^2D^\circ)4s$	$4s' {}^1D^\circ$	2	129065.4	
$3s 3p^5$	$3p^5 {}^3P^\circ$	2 1 0	93366.6 93998.7 94332.8	—632.1 —334.1	$3s^2 3p^3({}^4S^\circ)4p$	$4p {}^3P$	2 1 0	131767.4 131754.8 131768.0	12.6 —13.2
$3s^2 3p^3({}^4S^\circ)4s$	$4s {}^5S^\circ$	2	107878.5		$3s^2 3p^3({}^2D^\circ)3d$	$3d' {}^3G^\circ$	3 4 5	132162.1 132173.4 132191.3	11.3 17.9
$3s^2 3p^3({}^4S^\circ)3d$	$3d {}^5D^\circ$	4 3 2 1 0	110295.8 110296.8 110299.5 110302.0 110303.5	—1.0 —2.7 —2.5 —1.5	$3s^2 3p^3({}^2P^\circ)4s$	$4s'' {}^3P^\circ$	0 1 2	137770.1 137804.4 137877.6	34.3 73.2
$3s^2 3p^3({}^4S^\circ)4s$	$4s {}^3S^\circ$	1	112608.0		$3s^2 3p^3({}^2P^\circ)4s$	$4s'' {}^1P^\circ$	1	138623.0	
$3s 3p^5$	$3p^5 {}^1P^\circ$	1	115656.4		$3s^2 3p^3({}^2P^\circ)3d$	$3d'' {}^1P^\circ$	1	139350.0	
$3s^2 3p^3({}^4S^\circ)3d$	$3d {}^3D^\circ$	3 2 1	119809.9 119799.0 119842.1	10.9 —43.1	$3s^2 3p^3({}^2P^\circ)3d$	$3d'' {}^1D^\circ$	2	140259.1	
$3s^2 3p^3({}^2D^\circ)3d$	$3d' {}^1D^\circ$	2	121498.6		$3s^2 3p^3({}^2P^\circ)3d$	$3d'' {}^3D^\circ$	1 2 3	140740.0 141010.0 141349.6	270.0 339.6
$3s^2 3p^3({}^2D^\circ)3d$	$3d' {}^1F^\circ$	3	121635.1		$3s^2 3p^3({}^2P^\circ)3d$	$3d'' {}^3F^\circ$	4 3 2	143996.3 144174.5 144343.6	—178.2 —169.1
$3s^2 3p^3({}^2D^\circ)3d$	$3d' {}^3F^\circ$	2 3 4	126031.8 126219.1 126456.6	187.3 237.5	$3s^2 3p^3({}^2D^\circ)4p$	$4p' {}^1P$	1	145468.5	
$3s^2 3p^3({}^2D^\circ)4s$	$4s' {}^3D^\circ$	1 2 3	126725.1 126743.3 126782.8	18.2 39.5	$3s^2 3p^3({}^2P^\circ)3d$	$3d'' {}^3P^\circ$	0 1 2	 146012.9	

Cl II—Continued

Cl II—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval	
$3s^2\ 3p^3(^2D^\circ)4p$	$4p'\ ^3D$	1	146330. 0	3. 8 135. 2	$3s^2\ 3p^3(^4S^\circ)5d$	$5d\ ^5D^\circ$	0	$169799. 1$	0. 5 0. 6	
		2	146333. 8				1			
		3	146469. 0				2			
$3s^2\ 3p^3(^2D^\circ)4p$	$4p'\ ^3F$	2	147053. 7	72. 0 72. 7	$3s^2\ 3p^3(^2D^\circ)5s$	$5s'\ ^3D^\circ$	3	$170514. 7$	20. 4 40. 4	
		3	147125. 7				2			
		4	147198. 4				3			$170535. 1$
$3s^2\ 3p^3(^2D^\circ)4p$	$4p'\ ^1F$	3	147605. 7				$170575. 5$			
$3s^2\ 3p^3(^2D^\circ)4p$	$4p'\ ^3P$	2	149798. 3	-154. 1 -66. 6	$3s^2\ 3p^3(^4S^\circ)5d$	$5d\ ^3D^\circ$	3	$170973. 6$	-32. 2 -45. 7	
		1	149952. 4				2			$171005. 8$
		0	150019. 0				1			$171051. 5$
$3s^2\ 3p^3(^2D^\circ)3d$	$3d'\ ^3P^\circ$	2	150681. 4	-131. 3	$3s^2\ 3p^3(^2D^\circ)5s$	$5s'\ ^1D^\circ$	2	$171209. 2$		
		1	150812. 7							
		0								
$3s^2\ 3p^3(^2D^\circ)3d$	$3d'\ ^3D^\circ$	3	151092. 7	74. 1 -115. 2	$3s^2\ 3p^3(^2D^\circ)4d$	$4d'\ ^3F^\circ$	2	$172572. 6$	77. 7 90. 6	
		2	151018. 6				3			$172650. 3$
		1	151133. 8				4			$172740. 9$
$3s^2\ 3p^3(^4S^\circ)5s$	$5s\ ^5S^\circ$	2	152233. 1		$3s^2\ 3p^3(^2D^\circ)4d$	$4d'\ ^3G^\circ$	3	$173222. 7$	21. 2 33. 6	
$3s^2\ 3p^3(^2D^\circ)4p$	$4p'\ ^1D$	2	153257. 0	4			$173243. 9$			
		1		5			$173277. 5$			
$3s^2\ 3p^3(^2D^\circ)3d$	$3d'\ ^3S^\circ$	1	153571. 2				2	$174256. 3$		
$3s^2\ 3p^3(^4S^\circ)5s$	$5s\ ^3S^\circ$	1	153633. 1		$3s^2\ 3p^3(^2D^\circ)4d$	$4d'\ ^3D^\circ$	1	$174785. 7$	34. 9 32. 0	
$3s^2\ 3p^3(^4S^\circ)4d$	$4d\ ^5D^\circ$	0	154616. 7	2			$174820. 6$			
		1	154617. 8	3			$174852. 6$			
		2	154619. 6	1	$177423. 1$					
		3	154622. 6	0	$177693. 6$					
		4	154623. 8	1. 1 1. 8 3. 0 1. 2	$3s^2\ 3p^3(^2D^\circ)4d$	$4d'\ ^3S^\circ$	1	$177423. 1$		
$3p^5(^2P^\circ)4s$	$4s^{VII}\ ^3P^\circ$	2	157076. 6	-590. 2 -290. 0	$3s^2\ 3p^3(^2D^\circ)4d$	$4d'\ ^3P^\circ$	0	$177693. 6$	60. 6 62. 7	
		1	157666. 8				1			$177754. 2$
		0	157956. 8				2			$177816. 9$
$3s^2\ 3p^3(^2P^\circ)4p$	$4p''\ ^3S$	1	158177. 1		$3s^2\ 3p^3(^2D^\circ)4d$	$4d'\ ^1D^\circ$	2	$178539. 1$		
$3s^2\ 3p^3(^2P^\circ)4p$	$4p''\ ^3D$	1	158723. 7	44. 9 17. 8	$3s^2\ 3p^3(^2D^\circ)4d$	$4d'\ ^1P^\circ$	1	$179867. 0$		
		2	158768. 6							
		3	158786. 4							
$3s^2\ 3p^3(^2P^\circ)4p$	$4p''\ ^1D$	2	159574. 2		$3s^2\ 3p^3(^2P^\circ)5s$	$5s''\ ^3P^\circ$	0	$182337. 9$	34. 4 76. 4	
$3s^2\ 3p^3(^2P^\circ)4p$	$4p''\ ^1P$	2	159840. 3	1			$182372. 3$			
		1	159999. 6	2			$182448. 7$			
		2	160143. 4							
$3s^2\ 3p^3(^2P^\circ)4p$	$4p''\ ^3P$	2	161348. 4		$3s^2\ 3p^3(^2P^\circ)4d$	$4d''\ ^3F^\circ$	4	$184628. 1$	-27. 1 -3. 2	
$3s\ 3p^4(?)4s$	3P	0	159840. 3	3			$184655. 2$			
		1	159999. 6	2			$184658. 4$			
		2	160143. 4							
$3s^2\ 3p^3(^2P^\circ)4p$	$4p''\ ^3P$	2	161348. 4		$3s^2\ 3p^3(^2P^\circ)4d$	$4d''\ ^3P^\circ$	2	$185765. 0$	-140. 4	
$3s^2\ 3p^3(^2P^\circ)4p$	$4p''\ ^3P$	1	161634. 9	1			$185905. 4$			
		1	161654. 8	0						
		0	161671. 0							
$3s^2\ 3p^3(^4S^\circ)4d$	$4d\ ^3D^\circ$	3	161796. 5	-19. 9 -16. 2	$3s^2\ 3p^3(^2P^\circ)4d$	$4d''\ ^3D^\circ$	1	$185865. 2$		
		2	161907. 7				2			
		1	161989. 8				3			
		1	164210. 7	-111. 2 -82. 1	$3s^2\ 3p^3(^2D^\circ)6s$	$6s'\ ^3D^\circ$	1	$186844. 3$	16. 7 37. 3	
$3s^2\ 3p^3(^2P^\circ)4p$	$4p''\ ^1S$	2	165362. 1	2			$186861. 0$			
		0		3			$186898. 3$			
$3s^2\ 3p^3(^2P^\circ)4p$	$4p''\ ^1S$	0	165362. 1		$3s^2\ 3p^3(^2D^\circ)6s$	$6s'\ ^1D^\circ$	2	$187141. 4$		
$3s^2\ 3p^3(^4S^\circ)6s$	$6s\ ^5S^\circ$	2	168673. 6							
		$3s^2\ 3p^3(^4S^\circ)6s$	$6s\ ^3S^\circ$	1			169246. 6			
					Cl III ($^4S_{3/2}$)	Limit	192000			

January 1948.

ClII OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 +$	Observed Terms	
$3s^2 3p^4$	{ $3p^4 \ ^3P$ $3p^4 \ ^1D$ }	
$3s \ 3p^5$	{ $3p^5 \ ^3P^\circ$ $3p^5 \ ^1P^\circ$ }	
	$ns \ (n \geq 4)$	$np \ (n \geq 4)$
$3s^2 3p^3(^4S^\circ)nx$	{ $4-6s \ ^5S^\circ$ $4-6s \ ^3S^\circ$ }	$4p \ ^5P$ $4p \ ^3P$
$3s^2 3p^3(^2D^\circ)nx'$	{ $4-6s' \ ^3D^\circ$ $4-6s' \ ^1D^\circ$ }	$4p' \ ^3P$ $4p' \ ^1P$ $4p' \ ^3D$ $4p' \ ^1D$ $4p' \ ^3F$ $4p' \ ^1F$
$3s^2 3p^3(^2P^\circ)nx''$	{ $4, 5s'' \ ^3P^\circ$ $4s'' \ ^1P^\circ$ }	$4p'' \ ^3S$ $4p'' \ ^1S$ $4p'' \ ^3P$ $4p'' \ ^1P$ $4p'' \ ^3D$ $4p'' \ ^1D$
$3p^5(^2P^\circ)nx^{VII}$	$4s^{VII} \ ^3P^\circ$	
	$nd \ (n \geq 3)$	
$3s^2 3p^3(^4S^\circ)nx$	{ $3-5d \ ^5D^\circ$ $3-5d \ ^3D^\circ$ }	
$3s^2 3p^3(^2D^\circ)nx'$	{ $3, 4d' \ ^3S^\circ$ $3, 4d' \ ^3P^\circ$ $3, 4d' \ ^1P^\circ$ $3, 4d' \ ^3D^\circ$ $3, 4d' \ ^1D^\circ$ $3, 4d' \ ^3F^\circ$ $3, 4d' \ ^1F^\circ$ $3, 4d' \ ^3G^\circ$ }	
$3s^2 3p^3(^2P^\circ)nx''$	{ $3, 4d'' \ ^3P^\circ$ $3, 4d'' \ ^1P^\circ$ $3, 4d'' \ ^3D^\circ$ $3, 4d'' \ ^1D^\circ$ $3, 4d'' \ ^3F^\circ$ }	

*For predicted terms in the spectra of the S I isoelectronic sequence, see Introduction.

Cl III

(P I sequence; 15 electrons)

$Z=17$

Ground state $1s^2 2s^2 2p^6 3s^2 3p^3 \ ^4S_{1\frac{1}{2}}$

$3p^3 \ ^4S_{1\frac{1}{2}}^\circ$ **321936** cm^{-1}

I. P. 39.90 volts

The terms are from Bowen, who has greatly extended the early work on this spectrum. About 300 lines have been classified, and the observations range from 406 Å to 4971 Å. Inter-system combinations connecting the doublet and quartet terms have been observed.

Bowen remarks that because of perturbations the designations of the doublet levels of the $3d$ configuration are somewhat uncertain.

REFERENCES

- I. S. Bowen, Phys. Rev. **31**, 35 (1928). (I P) (T) (C L)
 I. S. Bowen, Phys. Rev. **45**, 401 (1934). (I P) (T) (C L)

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2 3p^3$	$3p^3 {}^4S^\circ$	$1\frac{1}{2}$	0. 0		$3s^2 3p^2({}^3P)4p$	$4p {}^4P^\circ$	$\frac{1}{2}$	204021. 6	
$3s^2 3p^3$	$3p^3 {}^2D^\circ$	$1\frac{1}{2}$	18053	67			$1\frac{1}{2}$	204124. 0	102. 4
		$2\frac{1}{2}$	18120				$2\frac{1}{2}$	204541. 2	417. 2
$3s^2 3p^3$	$3p^3 {}^2P^\circ$	$\frac{1}{2}$	29812	95	$3s^2 3p^2({}^3P)4p$	$4p {}^2D^\circ$	$1\frac{1}{2}$	205037. 3	
		$1\frac{1}{2}$	29907				$2\frac{1}{2}$	205946. 9	909. 6
$3s 3p^4$	$3p^4 {}^4P$	$2\frac{1}{2}$	98520	—610	$3s^2 3p^2({}^3P)4p$	$4p {}^4S^\circ$	$1\frac{1}{2}$	205938. 5	
		$1\frac{1}{2}$	99130	—345	$3s^2 3p^2({}^3P)4p$	$4p {}^2P^\circ$	$\frac{1}{2}$	209042. 1	
		$\frac{1}{2}$	99475				$1\frac{1}{2}$	209182. 8	140. 7
$3s^2 3p^2({}^3P)3d$	$3d {}^4F$	$1\frac{1}{2}$	146525. 6	224. 3	$3s^2 3p^2({}^1D)4p$	$4p' {}^2F^\circ$	$2\frac{1}{2}$	216524. 6	
		$2\frac{1}{2}$	146749. 9	323. 1			$3\frac{1}{2}$	216710. 4	185. 8
		$3\frac{1}{2}$	147073. 0	424. 9	$3s^2 3p^2({}^1D)4p$	$4p' {}^2D^\circ$	$2\frac{1}{2}$	217850. 2	
		$4\frac{1}{2}$	147497. 9				$1\frac{1}{2}$	217913. 1	—62. 9
$3s^2 3p^2({}^3P)3d$	$3d {}^4D$	$\frac{1}{2}$	151946. 4	—66. 5	$3s^2 3p^2({}^1D)4p$	$4p' {}^2P^\circ$	$\frac{1}{2}$	221862. 9	
		$1\frac{1}{2}$	151879. 9	—31. 3			$1\frac{1}{2}$	222100. 7	237. 8
		$2\frac{1}{2}$	151848. 6	104. 9	$3s^2 3p^2({}^3P)4d$	$4d {}^4F$	$1\frac{1}{2}$	239506. 3	
		$3\frac{1}{2}$	151953. 5				$2\frac{1}{2}$	239729. 9	223. 6
$3s^2 3p^2({}^3P)4s$	$4s {}^4P$	$\frac{1}{2}$	173736. 0	357. 8			$3\frac{1}{2}$	240075. 2	345. 3
		$1\frac{1}{2}$	174093. 8	520. 1			$4\frac{1}{2}$	240568. 4	493. 2
$3s^2 3p^2({}^3P)4s$	$4s {}^2P$	$\frac{1}{2}$	178369. 7	706. 4	$3s^2 3p^2({}^3P)4d$	$4d {}^4D$	$\frac{1}{2}$	241559. 4	
		$1\frac{1}{2}$	179076. 1				$1\frac{1}{2}$	241572. 4	13. 0
$3s^2 3p^2({}^3P)3d$	$3d {}^4P$	$2\frac{1}{2}$	179495. 2	—168. 3			$2\frac{1}{2}$	241685. 1	112. 7
		$1\frac{1}{2}$	179663. 5	—117. 5	$3s^2 3p^2({}^3P)4d$	$4d {}^4P$	$2\frac{1}{2}$	242822. 8	
		$\frac{1}{2}$	179781. 0				$1\frac{1}{2}$	243080. 7	—257. 9
$3s^2 3p^2({}^3P)3d$	$3d {}^2D$	$1\frac{1}{2}$	182076. 3	966. 4			$\frac{1}{2}$	243207. 2	—126. 5
		$2\frac{1}{2}$	183042. 7		$3s^2 3p^2({}^3P)4d$	$4d {}^2F$	$2\frac{1}{2}$	243828. 4	
$3s^2 3p^2({}^3P)3d$	$3d {}^2P$	$1\frac{1}{2}$	185838. 3	—382. 1			$3\frac{1}{2}$	244684. 9	856. 5
		$\frac{1}{2}$	186220. 4		$3s^2 3p^2({}^3P)5s$	$5s {}^4P$	$\frac{1}{2}$	244951. 5	
$3s^2 3p^2({}^1D)4s$	$4s' {}^2D$	$2\frac{1}{2}$	188390. 1	—58. 0			$1\frac{1}{2}$	245392. 4	440. 9
		$1\frac{1}{2}$	188448. 1				$2\frac{1}{2}$	246137. 2	744. 8
$3s^2 3p^2({}^1D)3d$	$3d' {}^2D$	$2\frac{1}{2}$	194959. 5	—308. 7	$3s^2 3p^2({}^3P)4d$	$4d {}^2D$	$1\frac{1}{2}$	248528. 2	
		$1\frac{1}{2}$	195268. 2				$2\frac{1}{2}$	248657. 7	129. 5
$3s^2 3p^2({}^1D)3d$	$3d' {}^2F$	$2\frac{1}{2}$	196137. 9	17. 9	$3s^2 3p^2({}^1D)4d$	$4d' {}^2D$	$1\frac{1}{2}$	254612. 7?	
		$3\frac{1}{2}$	196155. 8				$2\frac{1}{2}$	254683. 4?	70. 7
$3s^2 3p^2({}^1D)3d$	$3d' {}^2P$	$\frac{1}{2}$	198835. 5	148. 4	$3s^2 3p^2({}^1D)4d$	$4d' {}^2F$	$3\frac{1}{2}$	255086. 3	
		$1\frac{1}{2}$	198983. 9				$2\frac{1}{2}$	255140. 4	—54. 1
$3s^2 3p^2({}^3P)4p$	$4p {}^4D^\circ$	$\frac{1}{2}$	201073. 4	258. 6	$3s^2 3p^2({}^1D)5s$	$5s' {}^2D$	$2\frac{1}{2}$	258885. 8	
		$1\frac{1}{2}$	201332. 0	433. 1			$1\frac{1}{2}$	258890. 8	—5. 0
		$2\frac{1}{2}$	201765. 1	602. 5					
		$3\frac{1}{2}$	202367. 6						
					Cl IV (3P_0)	Limit		321936	

November 1947.

Cl III OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 +$	Observed Terms					
$3s^2 3p^3$	{ $3p^3 {}^4S^\circ$ $3p^3 {}^2P^\circ$ $3p^3 {}^2D^\circ$					
$3s 3p^4$	$3p^4 {}^4P$					
	$ns (n \geq 4)$		$np (n \geq 4)$		$nd (n \geq 3)$	
$3s^2 3p^2({}^3P)nx$	{ $4, 5s {}^4P$ $4s {}^2P$		$4p {}^4S^\circ$ $4p {}^4P^\circ$ $4p {}^4D^\circ$ $4p {}^2P^\circ$ $4p {}^2D^\circ$		$3, 4d {}^4P$ $3, 4d {}^4D$ $3, 4d {}^4F$ $3d {}^2P$ $3, 4d {}^2D$ $4d {}^2F$	
$3s^2 3p^2({}^1D)nx'$	$4, 5s' {}^2D$		$4p' {}^2P^\circ$ $4p' {}^2D^\circ$ $4p' {}^2F^\circ$		$3d' {}^2P$ $3, 4d' {}^2D$ $3, 4d' {}^2F$	

*For predicted terms in the spectra of the P I isoelectronic sequence, see Introduction.

Cl IV

(Si I sequence; 14 electrons)

 $Z=17$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^2 {}^3P_0$ $3p^2 {}^3P_0$ 431226 cm^{-1}

I. P. 53.5 volts

The analysis is by Bowen, who has classified 84 lines in the range between 318 Å and 3167 Å. The singlet and triplet terms are connected by intersystem combinations. Bowen classifies three lines (437 Å–440 Å) as $3p^3 {}^5S^\circ - 4s {}^5P$, but lists no quintet terms.

REFERENCES

- I. S. Bowen, Phys. Rev. **31**, 36 (1928). (C L)
 S. C. Deb, Acad. Sci. Allahabad Bul. **2**, 49 (1932). (I P) (T) (C L)
 I. S. Bowen, Phys. Rev. **45**, 401 (1934). (I P) (T) (C L)
 I. S. Bowen, Phys. Rev. **46**, 377 (1934). (T) (C L)

Cl IV

Cl IV

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2\ 3p^2$	$3p^2\ ^3P$	0	0	491 850	$3s^2\ 3p(^2P^\circ)4s$	$4s\ ^3P^\circ$	0	215026. 0	363. 3 1078. 8
		1	491				1	215389. 3	
		2	1341				2	216468. 1	
$3s^2\ 3p^2$	$3p^2\ ^1D$	2	13766		$3s^2\ 3p(^2P^\circ)4s$	$4s\ ^1P^\circ$	1	219454	
$3s^2\ 3p^2$	$3p^2\ ^1S$	0	32550		$3s^2\ 3p(^2P^\circ)4p$	$4p\ ^3D$	1	247575. 1?	451. 0 935. 1
$3s\ 3p^3$	$3p^3\ ^3D^\circ$	1	102752	35 82			2	248026. 1	
		2	102787					3	
		3	102869						
$3s\ 3p^3$	$3p^3\ ^3P^\circ$	2	120256	-18 -26	$3s^2\ 3p(^2P^\circ)4p$	$4p\ ^3P$	0	251471. 4	254. 4 670. 9
		1	120274				1	251725. 8	
		0	120300				2	252396. 7	
$3s\ 3p^3$	$3p^3\ ^3S^\circ$	1	164721		$3s^2\ 3p(^2P^\circ)5s$	$5s\ ^3P^\circ$	0	312747	244 1234
				1			312991		
				2			314225		
$3s\ 3p^3$	$3p^3\ ^1P^\circ$	1	166742		$3s^2\ 3p(^2P^\circ)5s$	$5s\ ^1P^\circ$	1	315121	
$3s^2\ 3p(^2P^\circ)3d$	$3d\ ^3P^\circ$	2	181643	-430 -227	-----	$Cl\ v\ (^2P_{3/2}^\circ)$	Limit	-----	431226
		1	182073						
		0	182300						
$3s^2\ 3p(^2P^\circ)3d$	$3d\ ^3D^\circ$	1	187008	166 172					
		2	187174						
		3	187346						

October 1947.

Cl IV OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 +$	Observed Terms		
$3s^2 3p^2$	$\left\{ \begin{array}{ccc} 3p^2 \ ^1S & 3p^2 \ ^3P & 3p^2 \ ^1D \end{array} \right.$		
$3s 3p^3$	$\left\{ \begin{array}{ccc} 3p^3 \ ^3S^\circ & 3p^3 \ ^3P^\circ & 3p^3 \ ^3D^\circ \\ & 3p^3 \ ^1P^\circ & \end{array} \right.$		
	$ns \ (n \geq 4)$	$np \ (n \geq 4)$	$nd \ (n \geq 3)$
$3s^2 3p(^3P^\circ)nx$	$\left\{ \begin{array}{ccc} 4, 5s & ^3P^\circ & \\ 4, 5s & ^1P^\circ & \end{array} \right.$	$4p \ ^3P \quad 4p \ ^3D$	$3d \ ^3P^\circ \quad 3d \ ^3D^\circ$

*For predicted terms in the spectra of the Si I isoelectronic sequence, see Introduction.

Cl V

(Al I sequence; 13 electrons)

$Z=17$

Ground state $1s^2 2s^2 2p^6 3s^2 3p \ ^2P_{1/2}^\circ$

$3p \ ^2P_{1/2}^\circ \ 547000 \text{ cm}^{-1}$

I. P. 67.80 volts

The analysis is by Bowen except for the revision of $3d \ ^4P^\circ$ and the addition of $5d \ ^2D$ suggested by Phillips and Parker. Forty-two lines have been classified in the interval between 236 Å and 894 Å.

No intersystem combinations connecting the doublet and quartet systems of terms have been observed, as indicated by x in the table.

REFERENCES

- I. S. Bowen, Phys. Rev. **31**, 37 (1928). (C L)
 S. C. Deb, Acad. Sci. Allahabad Bull. **2**, 43 (1932). (T) (C L)
 I. S. Bowen, Phys. Rev. **45**, 401 (1934). (I P) (T) (C L)
 L. W. Phillips and W. L. Parker, Phys. Rev. **60**, 306 (1941). (T) (C L)

Cl v

Cl v

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2(^1S)3p$	$3p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	0 1492	1492	$3s\ 3p(^3P^\circ)3d$	$3d\ ^4P^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	$269986+x$ $270423+x$ $270745+x$	-437 -322
$3s\ 3p^2$	$3p^2\ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$86000+x$ $86538+x$ $87381+x$	538 843	$3s\ 3p(^3P^\circ)3d$	$3d\ ^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	$272596+x$ $272757+x$ $272919+x$ $273020+x$	161 162 101
$3s\ 3p^2$	$3p^2\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	113234 113306	72	$3s^2(^1S)4d$	$4d\ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	349511	
$3s\ 3p^2$	$3p^2\ ^2S$	$\frac{1}{2}$	146644		$3s\ 3p(^3P^\circ)4s$	$4s\ ^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$353445+x$ $353978+x$ $354925+x$	533 947
$3s\ 3p^2$	$3p^2\ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	157931 158892	961	$3s^2(^1S)5d$	$5d\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	422949 423022	73
$3s^2(^1S)3d$	$3d\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	185861 185893	32					
$3p^3$	$3p^3\ ^4S^\circ$	$1\frac{1}{2}$	$233757+x$						
$3s^2(^1S)4s$	$4s\ ^2S$	$\frac{1}{2}$	256313		Cl VI (1S_0)	Limit		547000	

September 1947.

Cl v OBSERVED TERMS*

Config. $1s^2\ 2s^2\ 2p^6+$	Observed Terms	
$3s^2(^1S)3p$	$3p\ ^2P^\circ$	
$3s\ 3p^2$	$\left\{ \begin{array}{l} 3p^2\ ^2S \\ 3p^2\ ^4P \\ 3p^2\ ^2P \end{array} \right. \quad 3p^2\ ^2D$	
$3p^3$	$3p^3\ ^4S^\circ$	
	$ns\ (n \geq 4)$	$nd\ (n \geq 3)$
$3s^2(^1S)nx$	$4s\ ^2S$	$3-5d\ ^2D$
$3s\ 3p(^3P^\circ)nx$	$4s\ ^4P^\circ$	$3d\ ^4P^\circ\quad 3d\ ^4D^\circ$

*For predicted terms in the spectra of the Al I isoelectronic sequence, see Introduction.

Cl VI

(Mg I sequence; 12 electrons)

Z=17

Ground state $1s^2 2s^2 2p^6 3s^2 1S_0$

$3s^2 1S_0$ 780000± cm⁻¹

I. P. 96.7± volts

The analysis is incomplete. One singlet combination has been given by Bowen and Millikan, a line at 671.37 Å classified as $3s^2 1S_0-3p 1P_1^o$. The triplet terms are from Phillips and Parker, who have classified 34 lines in the range 194 Å to 736 Å.

From isoelectronic sequence data the writer has estimated the approximate value of the limit, and of $3p 3P_1^o$ above the ground state zero. All triplet terms have, consequently, been increased by 98147 cm⁻¹. The estimated values are entered in brackets in the table. The uncertainty, *x*, may be several hundred cm⁻¹.

REFERENCES

I. S. Bowen and R. A. Millikan, Phys. Rev. **25**, 597 (1925). (C L)
W. L. Parker and L. W. Phillips, Phys. Rev. **57**, 140 (1940). (T) (C L)
L. W. Phillips and W. L. Parker, Phys. Rev. **60**, 306 (1941). (T) (C L)

Cl VI					Cl VI					
Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval	
3s ²	3s ² ¹ S	0	0		3p(² P°)3d	3d ³ D°	1	411802 + <i>x</i>	273	
3s(² S)3p	3p ³ P°	0	[98147]+ <i>x</i>	553			2	412075 + <i>x</i>	153	
		1	98700 + <i>x</i>				3	412228 + <i>x</i>		
		2	99865 + <i>x</i>	1165						
3s(² S)3p	3p ¹ P°	1	148949		3s(² S)4d	4d ³ D	1	509868 + <i>x</i>	28	
							2	509896 + <i>x</i>	51	
							3	509947 + <i>x</i>		
3p ²	3p ² ³ P	0	234960 + <i>x</i>	636	3s(² S)4f	4f ³ F°	2, 3, 4	529889 + <i>x</i>		
		1	235596 + <i>x</i>	1201						
		2	236797 + <i>x</i>							
3s(² S)3d	3d ³ D	1	279845 + <i>x</i>	15	3s(² S)5d	5d ³ D	1	612058 + <i>x</i>	31	
		2	279860 + <i>x</i>	28			2	612089 + <i>x</i>		
		3	279888 + <i>x</i>				3			
3s(² S)4s	4s ³ S	1	407404 + <i>x</i>		Cl VII (² S _{1/2})	<i>Limit</i>	-----	[780000]		
3p(² P°)3d	3d ³ P°	2	409079 + <i>x</i>	−896						
		1	409975 + <i>x</i>	−787						
		0	410762 + <i>x</i>							

July 1947.

Cl VI OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 +$	Observed Terms		
$3s^2$	$3s^2 \ ^1S$		
$3s(^2S)3p$	{ $3p \ ^3P^\circ$ $3p \ ^1P^\circ$		
$3p^2$	$3p^2 \ ^3P$		
	$ns \ (n \geq 4)$	$nd \ (n \geq 3)$	$nf \ (n \geq 4)$
$3s(^2S)nx$	$4s \ ^3S$	$3-5d \ ^3D$	$4f \ ^3F^\circ$
$3p(^2P^\circ)nx$		$3d \ ^3P^\circ \quad 3d \ ^3D^\circ$	

*For predicted terms in the spectra of the Mg I isoelectronic sequence, see Introduction.

Cl VII

(Na I sequence; 11 electrons)

$Z=17$

Ground state $1s^2 2s^2 2p^6 3s \ ^2S_{1/2}$

$3s \ ^2S_{1/2} \ 921902 \text{ cm}^{-1}$

I. P. 114.27 volts

The resonance lines were observed by Bowen and Millikan. The analysis was extended by Phillips to include 22 classified lines in the interval between 174 Å and 813 Å. Absolute term values were derived from the $3d$ - nf series.

REFERENCES

- I. S. Bowen and R. A. Millikan, Phys. Rev. **25**, 295 (1925). (C L)
L. W. Phillips, Phys. Rev. **53**, 248 (1938). (I P) (T) (C L)

Cl VII

Cl VII

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3s$	$3s \ ^2S$	$\frac{1}{2}$	0		$4f$	$4f \ ^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	584086 584099	13
$3p$	$3p \ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	123001 124891	1890	$5s$	$5s \ ^2S$	$\frac{1}{2}$	647677	
$3d$	$3d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	290166 290239	73	$5d$	$5d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	697598 697619	21
$4s$	$4s \ ^2S$	$\frac{1}{2}$	464003		$5f$	$5f \ ^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	705398 705409	11
$4p$	$4p \ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	509197 509885	688	$6f$	$6f \ ^2F^\circ$	{ $2\frac{1}{2}$ $3\frac{1}{2}$ }	771549	
$4d$	$4d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	569142 569182	40	Cl VIII (1S_0)			Limit	921902

June 1947.

Cl VIII

(Ne I sequence; 10 electrons)

 $Z=17$ Ground state $1s^2 2s^2 2p^6 {}^1S_0$ $2p^6 {}^1S_0$ $2810000 \pm 500 \text{ cm}^{-1}$ I. P. 348.3 ± 0.1 volts

Edlén has classified 13 lines in the region between 39Å and 59Å, as combinations with the ground term. The terms from the (2S) limit in Cl IX need further confirmation.

As for Ne I the jl -coupling notation in the general form suggested by Racah is introduced. The unit 10^3 cm^{-1} used by Edlén has here been converted to cm^{-1} .

REFERENCES

B. Edlén, Zeit. Phys. **100**, 726 (1936). (I P) (T) (C L)G. Racah, Phys. Rev. **61**, 537 (L) (1942).

Cl VIII

Cl VIII

Edlén	Config.	Desig.	J	Level	Edlén	Config.	Desig.	J	Level
$2p {}^1S_0$	$2s^2 2p^6$	$2p^6 {}^1S$	0	0	$4d {}^1P_1$	$2s^2 2p^5({}^2P_{1/2})4d$	$4d [1\frac{1}{2}]^\circ$	1	2356820
					$4d {}^3D_1$	$2s^2 2p^5({}^2P_{3/2})4d$	$4d'[1\frac{1}{2}]^\circ$	1	2368550
$3s {}^3P_1$	$2s^2 2p^5({}^2P_{1/2})3s$	$3s [1\frac{1}{2}]^\circ$	2 1	1689450		$2s {}^2p^6({}^2S)3p$	$3p {}^3P^\circ$	2 1 0	2371580?
$3s {}^1P_1$	$2s^2 2p^5({}^2P_{3/2})3s$	$3s'[\frac{1}{2}]^\circ$	0 1	1704360	$3p' {}^3P_1$				
					$3p' {}^1P_1$	$2s {}^2p^6({}^2S)3p$	$3p {}^1P^\circ$	1	2401770?
$3d {}^3P_1$	$2s^2 2p^5({}^2P_{1/2})3d$	$3d [\frac{1}{2}]^\circ$	0 1	1972390	$5d {}^1P_1$	$2s^2 2p^5({}^2P_{1/2})5d$	$5d [1\frac{1}{2}]^\circ$	1	2521750
$3d {}^1P_1$	"	$3d [1\frac{1}{2}]^\circ$	1	1997040	$5d {}^3D_1$	$2s^2 2p^5({}^2P_{3/2})5d$	$5d'[1\frac{1}{2}]^\circ$	1	2534080
$3d {}^3D_1$	$2s^2 2p^5({}^2P_{3/2})3d$	$3d'[1\frac{1}{2}]^\circ$	1	2020730					
$4s {}^3P_1$	$2s^2 2p^5({}^2P_{1/2})4s$	$4s [1\frac{1}{2}]^\circ$	2 1	2242000		Cl IX (${}^2P_{1/2}$)	Limit	-----	2810000
$4s {}^1P_1$	$2s^2 2p^5({}^2P_{3/2})4s$	$4s'[\frac{1}{2}]^\circ$	0 1	2254200		Cl IX (${}^2P_{3/2}$)	Limit	-----	2823600

April 1947.

Cl VIII OBSERVED LEVELS*

Config. $1s^2+$	Observed Terms		
$2s^2\ 2p^6$	$2p^6\ ^1S$		
	$ns\ (n\geq 3)$	$np\ (n\geq 3)$	$nd\ (n\geq 3)$
$2s^2\ 2p^5(^2P^\circ)nx$	{ $\begin{matrix} 3, 4s & ^3P^\circ \\ 3, 4s & ^1P^\circ \end{matrix}$	{ $\begin{matrix} 3p & ^3P^\circ \\ 3p & ^1P^\circ \end{matrix}$	$\begin{matrix} 3d\ ^3P^\circ & 3-5d\ ^3D^\circ \\ 3-5d\ ^1P^\circ & \end{matrix}$
$2s\ 2p^6(^2S)nx$			
<i>jl</i> -Coupling Notation			
	Observed Pairs		
	$ns\ (n\geq 3)$		$nd\ (n\geq 3)$
	$3, 4s\ [1\frac{1}{2}]^\circ$		$\begin{matrix} 3d\ [\frac{1}{2}]^\circ \\ 3-5d\ [1\frac{1}{2}]^\circ \end{matrix}$
	$3, 4s'\ [\frac{1}{2}]^\circ$		$3-5d'\ [1\frac{1}{2}]^\circ$

*For predicted levels in the spectra of the Ne I isoelectronic sequence, see Introduction.

Cl IX

(F I sequence; 9 electrons)

$Z=17$

Ground state $1s^2 2s^2 2p^5 \ ^2P_{1/2}^\circ$

$2p^5 \ ^2P_{1/2}^\circ \ 3233000 \text{ cm}^{-1}$

I. P. 400.7 volts

Edlén has classified 34 lines in this spectrum in the interval 42 Å to 53 Å. The absolute value of the ground state has been extrapolated. Since no combinations between the two lowest terms have been observed, relative values have been extrapolated from the irregular doublet law for the three terms entered in brackets in the table. The uncertainty in the relative values may be large.

Levels from the $3d$ configurations with limits 3P and 1D in Cl x are labeled X since Edlén has been unable to assign term designations to them.

The unit used by Edlén, 10^3 cm^{-1} , has here been converted to cm^{-1} .

REFERENCE

B. Edlén, Zeit. Phys. **100**, 726 (1936). (I P) (T) (C L)

Cl IX

Cl IX

Edlén	Config.	Desig.	<i>J</i>	Level	Interval	Edlén	Config.	Desig.	<i>J</i>	Level	Interval
$2p \ ^2P_2$ $\ ^2P_1$	$2s^2 2p^5$	$2p^5 \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	$\begin{matrix} 0 \\ 13600 \end{matrix}$	-13600	$3d \ X_2$	$2s^2 2p^4(^3P)3d$	$3d \ X_2$		2209470	
$2p' \ ^2S_1$	$2s \ 2p^6$	$2p^6 \ ^2S$	$\frac{1}{2}$	[553400]		$3d \ X_1$	$2s^2 2p^4(^3P)3d$	$3d \ X_1$		2216710	
$3s \ ^4P_3$ $\ ^4P_2$ $\ ^4P_1$	$2s^2 2p^4(^3P)3s$	$3s \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	1888970 1896600 1901850	-7630 -5250	$\overline{3d} \ X_5$	$2s^2 2p^4(^1D)3d$	$3d' \ X_5$		2259280	
$3s \ ^2P_2$ $\ ^2P_1$	$2s^2 2p^4(^3P)3s$	$3s \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	1911950 1921050	-9100	$\overline{3d} \ X_4$	$2s^2 2p^4(^1D)3d$	$3d' \ X_4$		2263310	
$\overline{3s} \ ^2D_3$ $\ ^2D_2$	$2s^2 2p^4(^1D)3s$	$3s' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	1959790 1959960	-170	$\overline{3d} \ X_{2,3}$	$2s^2 2p^4(^1D)3d$	$3d' \ X_{2,3}$		2268000	
$\overline{\overline{3s}} \ ^2S_1$	$2s^2 2p^4(^1S)3s$	$3s'' \ ^2S$	$\frac{1}{2}$	2031080		$\overline{3d} \ X_1$	$2s^2 2p^4(^1D)3d$	$3d' \ X_1$		2272570	
$3d \ X_6$	$2s^2 2p^4(^3P)3d$	$3d \ X_6$		2196890		$\overline{\overline{3d}} \ ^2D_3$ $\ ^2D_2$	$2s^2 2p^4(^1S)3d$	$3d'' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	2328830 2330130	-1300
$3d \ X_5$	$2s^2 2p^4(^3P)3d$	$3d \ X_5$		2199540		$3s' \ ^2P_2$ $\ ^2P_1$	$2s \ 2p^5(^3P^\circ)3s$	$3s''' \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	[2415740] [2424380]	-8640
$3d \ X_4$	$2s^2 2p^4(^3P)3d$	$3d \ X_4$		2203850		$3d' \ ^2P_1$ $\ ^2P_2$	$2s \ 2p^5(^3P^\circ)3d$	$3d''' \ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	[2715940] [2722690]	6750
$3d \ X_3$	$2s^2 2p^4(^3P)3d$	$3d \ X_3$		2205950			Cl x (3P_2)	Limit	-----	[3233000]	

March 1947.

Cl IX OBSERVED TERMS*

Config. $1s^2 +$	Observed Terms	
$2s^2 2p^5$	$2p^5 \ ^2P^\circ$	
$2s \ 2p^6$	$2p^6 \ ^2S$	
	$ns \ (n \geq 3)$	$nd \ (n \geq 3)$
$2s^2 2p^4(^3P)nx$	$\left\{ \begin{matrix} 3s \ ^4P \\ 3s \ ^2P \end{matrix} \right.$	$3s' \ ^2D$
$2s^2 2p^4(^1D)nx'$		
$2s^2 2p^4(^1S)nx''$		
$2s \ 2p^5(^3P^\circ)nx'''$	$3s''' \ ^2P^\circ$	$3d''' \ ^2P^\circ$

*For predicted terms in the spectra of the F I isoelectronic sequence, see Introduction.

Cl x

(O I sequence; 8 electrons)

 $Z=17$ Ground state $1s^2 2s^2 2p^4 {}^3P_2$ $2p^4 {}^3P_2$ 3673000 cm^{-1}

I. P. 455.3 volts

Edlén has classified 15 lines between 39 Å and 47 Å. The absolute value of the ground term has been extrapolated from the isoelectronic sequence. Similarly, the singlet and triplet terms are connected only through the extrapolated value of $2p^4 {}^3P_2 - 2p^4 {}^1D_2$, and the uncertainty, x , may be large. The estimated value of $2p^5 {}^3P_2^o$ is given in brackets.

Edlén's term values expressed in units of 10^3 cm^{-1} are here changed to cm^{-1} .

REFERENCE

B. Edlén, Zeit. Phys. **100**, 732 (1936). (I P) (T) (C L).

Cl x

Cl x

Edlén	Config.	Desig.	J	Level	Interval	Edlén	Config.	Desig.	J	Level	Interval
$2p {}^3P_2$ 3P_1	$2s^2 2p^4$	$2p^4 {}^3P$	2 1 0	0 10880	-10880	$\overline{3s} {}^1P_1$	$2s^2 2p^3 ({}^2P^o) 3s$	$3s'' {}^1P^o$	1	$2262140 + x$	
$2p {}^1D$	$2s^2 2p^4$	$2p^4 {}^1D$	2	$61000 + x$		$3d {}^3D_2$ 3D_3	$2s^2 2p^3 ({}^4S^o) 3d$	$3d {}^3D^o$	1 2 3	2415360 2416040	680
$2p {}^1S$	$2s^2 2p^4$	$2p^4 {}^1S$	0	$130310 + x$		$\overline{3d} {}^3D$	$2s^2 2p^3 ({}^2D^o) 3d$	$3d' {}^3D^o$	3, 2, 1	2494700	
$2p' {}^3P$	$2s 2p^5$	$2p^5 {}^3P^o$	2 1 0	[487000]		$\overline{3d} {}^1D_2$	$2s^2 2p^3 ({}^2D^o) 3d$	$3d' {}^1D^o$	2	$2500380 + x$	
						$\overline{3d} {}^3P$	$2s^2 2p^3 ({}^2D^o) 3d$	$3d' {}^3P^o$	2, 1, 0	2502750	
$3s {}^3S_1$	$2s^2 2p^3 ({}^4S^o) 3s$	$3s {}^3S^o$	1	2184700		$\overline{3d} {}^1F_3$	$2s^2 2p^3 ({}^2D^o) 3d$	$3d' {}^1F^o$	3	$2520420 + x$	
$\overline{3s} {}^3D$	$2s^2 2p^3 ({}^2D^o) 3s$	$3s' {}^3D^o$	3, 2, 1	2202610		$\overline{3d} {}^3D$	$2s^2 2p^3 ({}^2P^o) 3d$	$3d'' {}^3D^o$	3, 2, 1	2547580	
$\overline{3s} {}^1D$	$2s^2 2p^3 ({}^2D^o) 3s$	$3s' {}^1D^o$	2	$2212650 + x$							
							Cl XI (${}^4S_{3/2}$)	Limit	-----	3673000	

March 1947.

Cl x OBSERVED TERMS*

Config. $1s^2 +$	Observed Terms	
$2s^2 2p^4$	$\left\{ \begin{array}{l} 2p^4 {}^1S \quad 2p^4 {}^3P \quad 2p^4 {}^1D \end{array} \right.$	
	$ns (n \geq 3)$	$nd (n \geq 3)$
$2s^2 2p^3 ({}^4S^o) nx$	$3s {}^3S^o$	$3d {}^3D^o$
$2s^2 2p^3 ({}^2D^o) nx'$	$\left\{ \begin{array}{l} 3s' {}^3D^o \\ 3s' {}^1D^o \end{array} \right.$	$3d' {}^3P^o \quad 3d' {}^3D^o \quad 3d' {}^1D^o \quad 3d' {}^1F^o$
$2s^2 2p^3 ({}^2P^o) nx''$	$\left\{ \begin{array}{l} 3s'' {}^1P^o \end{array} \right.$	$3d'' {}^3D^o$

*For predicted terms in the spectra of the O I isoelectronic sequence, see Introduction.

Cl XI

(N I sequence; 7 electrons)

Z=17

Ground state $1s^2 2s^2 2p^3 \ ^4S_{1/2}^\circ$

$2p^3 \ ^4S_{1/2}^\circ$ cm^{-1}

I. P. volts

This spectrum has not been analyzed, but Edlén has classified two lines as due to ClXI:

A	Int.	Wave No.	Desig.
40. 787 40. 392	0 0	2451760 2475740	$2p^3 \ ^2D^\circ - 3s' \ ^2D$ $2p^3 \ ^4S_{1/2}^\circ - 3s \ ^4P_{3/2}$

By extrapolation along the isoelectronic sequence, he lists combinations giving the relative positions of two other levels (entered in brackets in the table). From these data preliminary term values have been calculated and entered below. The uncertainty x is probably large.

The unit used by Edlén, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCE

B. Edlén, Zeit. Phys. **100**, 728 (1936). (C L)

Cl XI

Edlén	Config.	Desig.	J	Level
$2p \ ^4S_2$	$2s^2 2p^3$	$2p^3 \ ^4S^\circ$	$1\frac{1}{2}$	0
$2p \ ^2D_3$	$2s^2 2p^3$	$2p^3 \ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	$[94000]+x$
$2p \ ^2P_2$	$2s^2 2p^3$	$2p^3 \ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	$[143000]+x$
$3s \ ^4P_3$	$2s^2 2p^2(^3P)3s$	$3s \ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	2475740
$\overline{3s} \ ^2D$	$2s^2 2p^2(^1D)3s$	$3s' \ ^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	$2545760?+x$

February 1947.

ARGON

18 electrons

 $Z=18$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 {}^1S_0$ $3p^6 {}^1S_0$ 127109.9 cm^{-1}

I. P. 15.755 volts

The present list has been compiled from an unpublished manuscript kindly furnished by Edlén, who has made a study of this spectrum and interpreted it with the aid of present atomic theory. His term array is based on those published by Humphreys (1938) and by Meggers and Humphreys (1933), although he has revised and extended their lists. Three place entries are from interferometer measurements. The values of $4f[4\frac{1}{2}]$, $4f[3\frac{1}{2}]$, and $4f'[3\frac{1}{2}]$ are from unpublished data by Humphreys based on observations by Sittner.

The terms $ns'[1\frac{1}{2}]^\circ$ ($n=11$ to 16) and $nd'[1\frac{1}{2}]^\circ$ ($n=9$ to 14) have been calculated by the writer from the absorption series observed by Beutler in the region between 871 and 876 Å, and added to Edlén's list. Beutler lists these terms as blended.

Edlén has determined the new values of the series limits quoted here.

The Paschen notation used by Meissner, Rasmussen, Meggers, Humphreys, and others is entered in column one of the table in the same form as for Ne I. The letters U, V, W, X, Y, Z, adopted when configurations involving f electrons were found, are also entered in this column. Twenty-seven of these levels have J -values fixed by the observed combinations. These J -values are given in italics in the table.

Edlén suggested that a pair-coupling notation be adopted for Ne-like spectra to take into account the departure from LS -coupling. According to Shortley, LS -designations can be significantly assigned in only a few cases, in particular, for the following groups of levels:

Paschen	Desig.	Paschen	Desig.	Paschen	Desig.	Paschen	Desig.	Paschen	Desig.
$(n-3)s_5$	$ns {}^3P_2$	$2p_{10}$	$4p {}^3S_1$	$2p_5$	$4p {}^3P_0$	$4d_6$	$4d {}^3P_0$	$4d''_1$	$4d {}^3F_2$
$(n-3)s_4$	$ns {}^3P_1$	$2p_9$	$4p {}^3D_3$	$2p_4$	$4p {}^1P_1$	$4d_5$	$4d {}^3P_1$	$4d'_1$	$4d {}^1F_3$
$(n-3)s_3$	$ns {}^3P_0$	$2p_8$	$4p {}^3D_2$	$2p_3$	$4p {}^3P_2$	$4d'_4$	$4d {}^3F_4$	$4s''''_1$	$4d {}^1D_2$
$(n-3)s_2$	$ns {}^1P_1$	$2p_7$	$4p {}^3D_1$	$2p_2$	$4p {}^3P_1$	$4d_4$	$4d {}^3F_3$	$4s'''_1$	$4d {}^3D_3$
		$2p_6$	$4p {}^1D_2$	$2p_1$	$4p {}^1S_0$	$4d_3$	$4d {}^3P_2$	$4s''_1$	$4d {}^3D_2$
						$4d_2$	$4d {}^1P_1$	$4s'_1$	$4d {}^3D_1$

Consequently, the jl -coupling notation in the general form suggested by Racah is here introduced. The present arrangement has been suggested by Shortley, who has made a detailed investigation of the theoretical arrangement of the "pairs", to be used as a guide in preparing the present table. The pairs $nd[3\frac{1}{2}]^\circ$ and $nd[1\frac{1}{2}]^\circ$ are partially inverted as compared with Ne I.

No Grottrian diagram appears to have been published for this spectrum.

A I—Continued

REFERENCES

- K. W. Meissner, Zeit. Phys. **39**, 172 (1926); **40**, 839 (1927). (I P) (T) (C L)
 E. Rasmussen, *Serier i de Aedle Luftarters Spektre med Særligt Henblik paa Radiumemanation* p. 22 (Danske Erhvervs Annoncebureau's Forlag, Kobenhavn, 1932) Dissertation, Copenhagen. (T) (C L)
 E. Rasmussen, Zeit. Phys. **75**, 695 (1932). (T) (C L)
 W. F. Meggers and C. J. Humphreys, Bur. Std. J. Research **10**, 437, RP540 (1933). (T) (C L)
 R. M. Woods and B. J. Spence, Phys. Rev. **45**, 669 (1934). (C L)
 J. C. Boyce, Phys. Rev. **48**, 396 (1935). (I P) (T) (C L)
 H. Beutler, Zeit. Phys. **93**, 177 (1935). (I P) (T) (C L)
 H. Kopfermann und H. Krüger, Zeit. Phys. **105**, 389 (1937). (I S)
 J. B. Green, Phys. Rev. **52**, 736 (1937). (Z E)
 J. B. Green and B. Fried, Phys. Rev. **54**, 876 (1938). (Z E)
 P. Jacquinet, Compt. Rend. **206**, 1635 (1938). (Z E)
 C. J. Humphreys, J. Research Nat. Bur. Std. **20**, 26, RP1061 (1938) and unpublished data. (T) (C L)
 G. Racah, Phys. Rev. **61**, 537 (L) (1942).
 B. Edlén, Ark. Mat. Astr. Fys. (Stockholm) **29A**, No. 32 (1943). (C L)
 J. B. Green, Phys. Rev. **64**, 151 (1943). (Z E)
 G. Shortley, unpublished material (Aug. 1947).
 B. Edlén, unpublished material (April 1948). (I P) (T) (C L)
 W. R. Sittner, unpublished material (1949).

A I

A I

Au- thors	Config.	Desig.	<i>J</i>	Level	Obs. <i>g</i>	Au- thors	Config.	Desig.	<i>J</i>	Level	Obs. <i>g</i>
1 p_0	3 p^0	3 p^0 1S	0	0.0		3 p_{10}	3 $p^5(2P_{1/2})5p$	5 p [1/2]	1	116660.054	1.90
1 s_5 1 s_4	3 $p^5(2P_{1/2})4s$	4 s [1 1/2] ^o	2 1	93143.800 93750.639	1.506 1.404	3 p_9 3 p_8	"	5 p [2 1/2]	3 2	116942.815 116999.389	1.09
1 s_3 1 s_2	3 $p^5(2P_{3/2})4s$	4 s' [1/2] ^o	0 1	94553.707 95399.870	1.102	3 p_7 3 p_6	"	5 p [1 1/2]	1 2	117151.387 117183.654	1.01 1.42
2 p_{10}	3 $p^5(2P_{1/2})4p$	4 p [1/2]	1	104102.144	1.985	3 p_5	"	5 p [1/2]	0	117563.020	
2 p_9 2 p_8	"	4 p [2 1/2]	3 2	105462.804 105617.315	1.338 1.112	3 p_4 3 p_3	3 $p^5(2P_{3/2})5p$	5 p' [1 1/2]	1 2	118407.494 118469.117	0.61 1.18
2 p_7 2 p_6	"	4 p [1 1/2]	1 2	106087.305 106237.597	0.838 1.305	3 p_2 3 p_1	"	5 p' [1/2]	1 0	118459.662 118870.981	1.45
2 p_5	"	4 p [1/2]	0	107054.319		4 d_6 4 d_5	3 $p^5(2P_{1/2})4d$	4 d [1/2] ^o	0 1	118512.17 118651.447	1.467
2 p_4 2 p_3	3 $p^5(2P_{3/2})4p$	4 p' [1 1/2]	1 2	107131.755 107289.747	0.819 1.260	4 d'_4 4 d_4	"	4 d [3 1/2] ^o	4 3	119023.699 119212.98	1.255 1.077
2 p_2 2 p_1	"	4 p' [1/2]	1 0	107496.463 108722.668	1.380	4 d_3 4 d_2	"	4 d [1 1/2] ^o	2 1	118906.665 119847.81	1.437 0.768
3 d_6 3 d_5	3 $p^5(2P_{1/2})3d$	3 d [1/2] ^o	0 1	111667.87 111818.09		4 d'_1 4 d_1	"	4 d [2 1/2] ^o	2 3	119444.88 119566.11	0.908
3 d'_4 3 d_4	"	3 d [3 1/2] ^o	4 3	112750.22 113020.39		4 s_1'''' 4 s_1'''	3 $p^5(2P_{3/2})4d$	4 d' [2 1/2] ^o	2 3	120619.076 120753.52	0.987 1.133
3 d_3 3 d_2	"	3 d [1 1/2] ^o	2 1	112138.98 114147.75		4 s_1'' 4 s_1'	"	4 d' [1 1/2] ^o	2 1	120600.944 121011.979	1.057 0.877
3 d'_1 3 d_1	"	3 d [2 1/2] ^o	2 3	113426.05 113716.61		3 s_5 3 s_4	3 $p^5(2P_{1/2})6s$	6 s [1 1/2] ^o	2 1	119683.113 119760.22	1.500 1.184
3 s_1'''' 3 s_1'''	3 $p^5(2P_{3/2})3d$	3 d' [2 1/2] ^o	2 3	114641.04 114821.99		3 s_3 3 s_2	3 $p^5(2P_{3/2})6s$	6 s' [1/2] ^o	0 1	121096.67 121161.356	1.271
3 s_1'' 3 s_1'	"	3 d' [1 1/2] ^o	2 1	114805.18 115366.90		4X 4X	3 $p^5(2P_{1/2})4f$	4 f [1 1/2]	1 2	120188.34 120188.66	
2 s_5 2 s_4	3 $p^5(2P_{1/2})5s$	5 s [1 1/2] ^o	2 1	113468.55 113643.26		4V	"	4 f [4 1/2]	5 4	120207.32 120207.77	
2 s_3 2 s_2	3 $p^5(2P_{3/2})5s$	5 s' [1/2] ^o	0 1	114861.67 114975.07		4Y 4Y	"	4 f [2 1/2]	3 2	120229.81 120230.07	

A I—Continued

A I—Continued

Authors	Config.	Desig.	<i>J</i>	Level	Obs. <i>g</i>	Authors	Config.	Desig.	<i>J</i>	Level	Obs. <i>g</i>
4U	$3p^5(^2P_{1/2})4f$	4 <i>f</i> [3½]	3, 4	120250. 15		5 <i>p</i> ₂	$3p^5(^2P_{3/2})7p$	7 <i>p</i> ' [½]	1	124651. 05	
4W	$3p^5(^2P_{3/2})4f$	4 <i>f</i> ' [3½]	3, 4	121653. 40		5 <i>p</i> ₁			0	124749. 89	
4Z	"	4 <i>f</i> ' [2½]	3	121654. 32		6 <i>d</i> ₆	$3p^5(^2P_{1/2})6d$	6 <i>d</i> [½]°	0	123508. 96	
4Z			2	121654. 58		6 <i>d</i> ₅			1	123468. 034	1. 233
4 <i>p</i> ₁₀	$3p^5(^2P_{1/2})6p$	6 <i>p</i> [½]	1	121068. 804		6 <i>d</i> ₄ '	"	6 <i>d</i> [3½]°	4	123653. 238	1. 256
4 <i>p</i> ₉	"	6 <i>p</i> [2½]	3	121165. 431		6 <i>d</i> ₄			3	123773. 920	1. 052
4 <i>p</i> ₈			2	121191. 92		6 <i>d</i> ₃	"	6 <i>d</i> [1½]°	2	123808. 60	1. 206
4 <i>p</i> ₇	"	6 <i>p</i> [1½]	1	121257. 227		6 <i>d</i> ₁ '	"	6 <i>d</i> [2½]°	2	123826. 85	1. 107
4 <i>p</i> ₆			2	121270. 682		6 <i>d</i> ₁ '			3	123832. 50	1. 245
4 <i>p</i> ₅	"	6 <i>p</i> [½]	0	121470. 304		6 <i>s</i> ₁ '	$3p^5(^2P_{3/2})6d$	6 <i>d</i> ' [2½]°	2	125113. 48	0. 777
4 <i>p</i> ₄	$3p^5(^2P_{3/2})6p$	6 <i>p</i> ' [1½]	1	122609. 76		6 <i>s</i> ₁ '			3	125150. 00	1. 098
4 <i>p</i> ₃			2	122635. 128		6 <i>s</i> ₁ '	"	6 <i>d</i> ' [1½]°	2	125066. 501	1. 264
4 <i>p</i> ₂	"	6 <i>p</i> ' [½]	1	122601. 290		6 <i>s</i> ₁			1	125236. 23	
4 <i>p</i> ₁			0	122790. 612		5 <i>s</i> ₅	$3p^5(^2P_{1/2})8s$	8 <i>s</i> [1½]°	2	123903. 295	1. 50
5 <i>d</i> ₆	$3p^5(^2P_{1/2})5d$	5 <i>d</i> [½]°	0	121794. 158		5 <i>s</i> ₄			1	123935. 97	
5 <i>d</i> ₅			1	121932. 908	1. 400	5 <i>s</i> ₃	$3p^5(^2P_{3/2})8s$	8 <i>s</i> ' [½]°	0	125334. 75	
5 <i>d</i> ₄ '	"	5 <i>d</i> [3½]°	4	122036. 134	1. 253	5 <i>s</i> ₂			1	125353. 31	1. 26
5 <i>d</i> ₄			3	122160. 22	1. 076	6X	$3p^5(^2P_{1/2})6f$	6 <i>f</i> [1½]	1	124041. 20	
5 <i>d</i> ₃	"	5 <i>d</i> [1½]°	2	122086. 974	1. 387	6X			2	124041. 38	
5 <i>d</i> ₂			1	122514. 29	0. 813	6V	"	6 <i>f</i> [4½]	4, 5	124046. 64	
5 <i>d</i> ₁ '	"	5 <i>d</i> [2½]°	2	122282. 134	0. 941	6Y	"	6 <i>f</i> [2½]	3	124051. 44	
5 <i>d</i> ₁ '			3	122329. 72	1. 199	6Y			2	124051. 65	
5 <i>s</i> ₁ '	$3p^5(^2P_{3/2})5d$	5 <i>d</i> ' [2½]°	2	123505. 536	0. 802	6U	"	6 <i>f</i> [3½]	3, 4	124058. 36	
5 <i>s</i> ₁ '			3	123557. 459	1. 127	6W	$3p^5(^2P_{3/2})6f$	6 <i>f</i> ' [3½]	3, 4	125482. 70	
5 <i>s</i> ₁ '	"	5 <i>d</i> ' [1½]°	2	123372. 987	1. 265	6Z	"	6 <i>f</i> ' [2½]	3	125483. 16	
5 <i>s</i> ₁			1	123815. 53	0. 846	6Z			2	125483. 34	
4 <i>s</i> ₅	$3p^5(^2P_{1/2})7s$	7 <i>s</i> [1½]°	2	122440. 109	1. 506		$3p^5(^2P_{1/2})8p$	8 <i>p</i> [½]	1	124311. 72	
4 <i>s</i> ₄			1	122479. 459	1. 164	6 <i>p</i> ₁₀					
4 <i>s</i> ₃	$3p^5(^2P_{3/2})7s$	7 <i>s</i> ' [½]°	0	123873. 07		6 <i>p</i> ₉	"	8 <i>p</i> [2½]	3	124349. 04	
4 <i>s</i> ₂			1	123882. 30	1. 296	6 <i>p</i> ₈			2	124356. 73	
5X	$3p^5(^2P_{1/2})5f$	5 <i>f</i> [1½]	1	122686. 20		6 <i>p</i> ₇	"	8 <i>p</i> [1½]	1	124376. 38	
5X			2	122686. 40		6 <i>p</i> ₆			2	124381. 01	
5V	"	5 <i>f</i> [4½]	4, 5	122695. 70		6 <i>p</i> ₅	"	8 <i>p</i> [½]	0	124439. 41	
5Y	"	5 <i>f</i> [2½]	3	122707. 94		6 <i>p</i> ₄	$3p^5(^2P_{3/2})8p$	8 <i>p</i> ' [1½]	1	125783. 8	
5Y			2	122708. 18		6 <i>p</i> ₃			2	125791. 94	
5U	"	5 <i>f</i> [3½]	3, 4	122717. 90		6 <i>p</i> ₂	"	8 <i>p</i> ' [½]	1	125777. 3	
5W	$3p^5(^2P_{3/2})5f$	5 <i>f</i> ' [3½]	3, 4	124135. 74		6 <i>p</i> ₁			0	125831. 45	
5Z	"	5 <i>f</i> ' [2½]	3	124137. 29		7 <i>d</i> ₆	$3p^5(^2P_{1/2})7d$	7 <i>d</i> [½]°	0	124526. 75	
5Z			2	124137. 45		7 <i>d</i> ₅			1	124554. 939	
5 <i>p</i> ₁₀	$3p^5(^2P_{1/2})7p$	7 <i>p</i> [½]	1	123172. 09		7 <i>d</i> ₄ '	"	7 <i>d</i> [3½]°	4	124609. 917	
5 <i>p</i> ₉	"	7 <i>p</i> [2½]	3	123205. 83		7 <i>d</i> ₄			3	124649. 549	
5 <i>p</i> ₈			2	123220. 73		7 <i>d</i> ₃	"	7 <i>d</i> [1½]°	2	124603. 957	
5 <i>p</i> ₇	"	7 <i>p</i> [1½]	1	123254. 99		7 <i>d</i> ₂			1	124788. 39	
5 <i>p</i> ₆			2	123261. 593		7 <i>d</i> ₁ '	"	7 <i>d</i> [2½]°	2	124692. 02	
5 <i>p</i> ₅	"	7 <i>p</i> [½]	0	123385. 13		7 <i>d</i> ₁ '			3	124715. 16	
5 <i>p</i> ₄	$3p^5(^2P_{3/2})7p$	7 <i>p</i> ' [1½]	1	124643. 54		7 <i>s</i> ₁ '	$3p^5(^2P_{3/2})7d$	7 <i>d</i> ' [2½]°	2	126064. 50	
5 <i>p</i> ₃			2	124658. 52		7 <i>s</i> ₁ '			3	126089. 56	

Au- thors	Config.	Desig.	<i>J</i>	Level	Obs. <i>g</i>	Au- thors	Config.	Desig.	<i>J</i>	Level	Obs. <i>g</i>
7s'' ₁	3p ⁵ (² P _{3/2})7d	7d' [1½]°	2 1	126053. 21		9d' ₄ 9d ₄	3p ⁵ (² P _{1/2})9d	9d [3½]°	4 3	125631. 69 125652. 04	
6s ₅ 6s ₄	3p ⁵ (² P _{1/2})9s	9s [1½]°	2 1	124771. 67 124782. 77		9d ₃ 9d ₂	"	9d [1½]°	2 1	125637. 93 125718. 12	
6s ₃ 6s ₂	3p ⁵ (² P _{3/2})9s	9s' [½]°	0 1	126202. 82 126211. 57		9d' ₁ 9d ₁	"	9d [2½]°	2 3	125671. 53 125680. 52	
7X 7X	3p ⁵ (² P _{1/2})7f	7f [1½]	1 2	124857. 27 124857. 42		9s' ₁	3p ⁵ (² P _{3/2})9d'	9d' [1½]°	2 1	127130	
7V	"	7f [4½]	4, 5	124860. 64		8s ₅ 8s ₄	3p ⁵ (² P _{1/2})11s	11s [1½]°	2 1	125709. 45 125715. 50	
7Y 7Y	"	7f [2½]	3 2	124865. 04 124865. 19		8s ₂	3p ⁵ (² P _{3/2})11s	11s' [½]°	0 1	127130	
7U	"	7f [3½]	3, 4	124868. 77		9X	3p ⁵ (² P _{1/2})9f	9f [1½]	1, 2	125748. 9	
7W	3p ⁵ (² P _{3/2})7f	7f' [3½]	3, 4	126294. 90		9V	"	9f [4½]	4, 5	125750. 39	
7Z	"	7f' [2½]	3 2	126295. 02		9Y	"	9f [2½]	3 2	125752. 8	
7p ₁₀	3p ⁵ (² P _{1/2})9p	9p [½]	1	125039. 60		9U	"	9f [3½]	3, 4	125754. 21	
7p ₉ 7p ₈	"	9p [2½]	3 2	125054. 1 125059. 8		9p ₁₀	3p ⁵ (² P _{1/2})11p	11p [½]	1	125844. 3	
7p ₇ 7p ₆	"	9p [1½]	1 2	125072. 6 125074. 9		9p ₇ 9p ₆	"	11p [1½]	1 2	125853. 3 125853. 8	
7p ₅	"	9p [½]	0	125122. 54		9p ₅	"	11p [½]	0	125888. 9	
7p ₁	3p ⁵ (² P _{3/2})9p	9p' [½]	1 0	126524. 2		10d ₆ 10d ₅	3p ⁵ (² P _{1/2})10d	10d [½]°	0 1	125895. 72 125898. 64	
8d ₆ 8d ₅	3p ⁵ (² P _{1/2})8d	8d [½]°	0 1	125163. 00 125135. 898		10d' ₄ 10d ₄	"	10d [3½]°	4 3	125922. 53 125932. 59	
8d' ₄ 8d ₄	"	8d [3½]°	4 3	125219. 88 125269. 52		10d ₃	"	10d [1½]°	2 1	125906. 61	
8d ₃	"	8d [1½]°	2 1	125282. 97		10d' ₁ 10d ₁	"	10d [2½]°	2 3	125945. 72 125957. 40	
8d' ₁ 8d ₁	"	8d [2½]°	2 3	125291. 45 125293. 65		10s' ₁	3p ⁵ (² P _{3/2})10d	10d' [1½]°	2 1	127410	
7s ₅ 7s ₄	3p ⁵ (² P _{1/2})10s	10s [1½]°	2 1	125329. 99 125331. 93		9s ₅ 9s ₄	3p ⁵ (² P _{1/2})12s	12s [1½]°	2 1	125979. 41 125984. 35	
8X	3p ⁵ (² P _{1/2})8f	8f [1½]	1, 2	125386. 41		9s ₂	3p ⁵ (² P _{3/2})12s	12s' [½]°	0 1	127410	
8V	"	8f [4½]	4, 5	125388. 65		10p ₁₀ 10p ₅	3p ⁵ (² P _{1/2})12p	12p [½]	1 0	126072. 6 126101. 7	
8Y 8Y	"	8f [2½]	3 2	125391. 04 125391. 17		11d ₆ 11d ₅	3p ⁵ (² P _{1/2})11d	11d [½]°	0 1	126114. 66 126099. 49	
8U	"	8f [3½]	3, 4	125393. 79		11d' ₄ 11d ₄	"	11d [3½]°	4 3	126135. 42 126154. 55	
8p ₁₀	3p ⁵ (² P _{1/2})10p	10p [½]	1	125505. 5		11d ₃	"	11d [1½]°	2 1	126159. 9	
8p ₉	"	10p [2½]	3 2	125519. 9		11d' ₁ 11d ₁	"	11d [2½]°	2 3	126162. 5 126163. 24	
8p ₇ 8p ₆	"	10p [1½]	1 2	125531. 5 125533. 8							
8p ₅	"	10p [½]	0	125561. 9							
9d ₆ 9d ₅	3p ⁵ (² P _{1/2})9d	9d [½]°	0 1	125595. 11 125613. 12							

A I—Continued

A I—Continued

Au- thors	Config.	Desig.	<i>J</i>	Level	Obs. <i>g</i>	Au- thors	Config.	Desig.	<i>J</i>	Level	Obs. <i>g</i>
11s ₁ '	3p ⁵ (² P _{3/2} ^o)11d	11d' [1½] ^o	2 1	127610		13d ₃	3p ⁵ (² P _{3/2} ^o)13d	13d [1½] ^o	2 1	126420. 8	
10s ₈ 10s ₄	3p ⁵ (² P _{1/2} ^o)13s	13s [1½] ^o	2 1	126178. 27 126181. 30		13d ₁ ' 13d ₁ '	"	13d [2½] ^o	2 3	126432. 1 126435. 5	
10s ₂	3p ⁵ (² P _{3/2} ^o)13s	13s' [½] ^o	0 1	127610		13s ₁ '	3p ⁵ (² P _{3/2} ^o)13d	13d' [1½] ^o	2 1	127880	
11p ₅	3p ⁵ (² P _{1/2} ^o)13p	13p [½]	1 0	126270. 0		14d ₆ 14d ₅	3p ⁵ (² P _{1/2} ^o)14d	14d [½] ^o	0 1	126508. 1 126510. 06	
12d ₆ 12d ₅	3p ⁵ (² P _{1/2} ^o)12d	12d [½] ^o	0 1	126281. 3 126292. 71		14d ₄ ' 14d ₄	"	14d [3½] ^o	4 3	126517. 41 126521. 71	
12d ₄ ' 12d ₄	"	12d [3½] ^o	4 3	126295. 79 126305. 28		14d ₃	"	14d [1½] ^o	2 1	126514. 8	
12d ₃	3p ⁵ (² P _{1/2} ^o)13d	12d [1½] ^o	2 1	126302. 6		14d ₁ '	"	14d [2½] ^o	2 3	126530. 1	
12d ₁ ' 12d ₁	"	12d [2½] ^o	2 3	126313. 1 126316. 1		14s ₁ '	3p ⁵ (² P _{3/2} ^o)14d	14d' [1½] ^o	2 1	127970	
12s ₁ '	3p ⁵ (² P _{3/2} ^o)12d	12d' [1½] ^o	2 1	127760			A II (² P _{1/2} ^o)	<i>Limit</i>	-----	127109. 9	
11s ₅ 11s ₄	3p ⁵ (² P _{1/2} ^o)14s	14s [1½] ^o	2 1	126328. 80 126332. 0		12s ₂	3p ⁵ (² P _{3/2} ^o)15s	15s' [½] ^o	0 1	127880	
11s ₂	3p ⁵ (² P _{3/2} ^o)14s	14s' [½] ^o	0 1	127760		13s ₂	3p ⁵ (² P _{3/2} ^o)16s	16s' [½] ^o	0 1	127970	
13d ₅	3p ⁵ (² P _{1/2} ^o)13d	13d [½] ^o	0 1	126412. 99			A II (² P _{3/2} ^o)	<i>Limit</i>	-----	128541. 3	
13d ₄ ' 13d ₄	"	13d [3½] ^o	4 3	126419. 65 126426. 07							

April 1948.

A I OBSERVED LEVELS*

Config. 1s ² 2s ² 2p ⁶ 3s ² +	Observed Terms			
3p ⁶	3p ⁶ 1S			
	ns (n ≥ 4)	np (n ≥ 4)		nd (n ≥ 3)
3p ⁵ (² P ^o)nx	{ 4-16s ³ P ^o 4-9, 11-16s ¹ P ^o	4p ³ S 4p ¹ S	4p ³ P 4p ¹ P	4p ³ D 4p ¹ D
		4d ³ P ^o 4d ¹ P ^o	4d ³ D ^o 4d ¹ D ^o	4d ³ F ^o 4d ¹ F ^o
<i>jl</i> -Coupling Notation				
	Observed Pairs			
	ns (n ≥ 4)	np (n ≥ 4)		nd (n ≥ 3)
3p ⁵ (² P _{1/2} ^o)nx	4-14s [1½] ^o	4-13p [½] 4-10p [2½] 4-11p [1½]		3-14d [½] ^o 3-14d [3½] ^o 3-14d [1½] ^o 3-14d [2½] ^o
3p ⁵ (² P _{3/2} ^o)nx'	4-9, 11-16s' [½] ^o	4-8p' [1½] 4-9p' [½]		3-7d' [2½] ^o 3-7, 9-14d' [1½] ^o
				4-9f [1½] 5-9f [4½] 4-9f [2½] 5-9f [3½]
				4-7f' [3½] 4-7f' [2½]

*For predicted levels in the spectra of the A I isoelectronic sequence, see Introduction.

(Cl I sequence; 17 electrons)

Z=18

Ground state $1s^2 2s^2 2p^6 3s^2 3p^5 {}^2P_{1/2}^{\circ}$ $3p^5 {}^2P_{1/2}^{\circ} 222820 \pm 300 \text{ cm}^{-1}$

I. P. 27.62 volts

A monograph containing the complete and detailed analysis of this spectrum is needed. Most of the analysis is by de Bruin, but his work has been revised and extended by a number of investigators who are not in complete agreement on all details of interpretation.

The term list published by Boyce forms the basis of the present compilation, but the later additions and revisions by Minnhagen, Edlén, and de Bruin have been incorporated into the present list. The writer has prepared a complete multiplet array for this spectrum and in dubious cases she has attempted to adopt the term assignments that appear to be best confirmed from the multiplet evidence.

One term labeled " 2P " in the table, (" $a {}^2P$ " in the published papers), has as yet no configuration assignment. Three miscellaneous levels assigned by de Bruin (1937) to the $4f$ configuration have been omitted pending further confirmation.

The doublet and quartet terms are well connected by observed intersystem combinations. Edlén has derived the series limit quoted here from the $({}^3P)ns {}^4P {}^2P$ series ($n=4, 5, 6$).

REFERENCES

- T. L. de Bruin, Zeit. Phys. **51**, 108 (1928); Proc. Roy. Acad. Amsterdam **31**, No. 7, 771 (1928). (I P) (T) (C L)
 T. L. de Bruin, Zeit. Phys. **61**, 307 (1930); Proc. Roy. Acad. Amsterdam **33**, No. 2, 198 (1930). (I P) (T) (C L)
 A. H. Rosenthal, Ann. der Phys. [5] **4**, 49 (1930). (T) (C L)
 J. C. Boyce, Phys. Rev. **48**, 397 (1935). (I P) (T) (C L)
 T. L. de Bruin, Proc. Roy. Acad. Amsterdam **40**, No. 4, 340 (1937). (T) (C L)
 B. Edlén, Zeit. Phys. **104**, 413 (1937). (I P) (T) (C L)
 R. Bezler, Zeit. Phys. **116**, 480 (1940). (Z E)
 L. Minnhagen, Ark. Mat. Astr. Fys. (Stockholm) **34A**, No. 22 p. 4 (1947). (T) (C L)
 L. Minnhagen, Ark. Mat. Astr. Fys. (Stockholm) **35A**, No. 16 p. 3 (1948). (E D)

A II

A II

Config.	Desig.	J	Level	Interval	Obs. g	Config.	Desig.	J	Level	Interval	Obs. g
$3s^2 3p^5$	$3p^5 {}^2P^{\circ}$	$1\frac{1}{2}$ $\frac{1}{2}$	0. 0 1432. 0	-1432. 0		$3s^2 3p^4({}^1D)4s$	$4s' {}^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	148620. 98 148843. 29	222. 31	0. 803 1. 202
$3s 3p^6$	$3p^6 {}^2S$	$\frac{1}{2}$	108722. 5			$3s^2 3p^4({}^3P)3d$	$3d {}^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	149180. 18 150148. 54	-968. 36	
$3s^2 3p^4({}^3P)3d$	$3d {}^4D$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	132328. 22 132482. 12 132631. 64 132738. 60	-153. 90 -149. 52 -106. 96		$3s^2 3p^4({}^3P)3d$	$3d {}^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	150475. 82 151088. 18	612. 36	
$3s^2 3p^4({}^3P)4s$	$4s {}^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	134242. 62 135086. 88 135602. 62	-844. 26 -515. 74	1. 598 1. 722 2. 650	$3s^2 3p^4({}^3P)4p$	$4p {}^4P^{\circ}$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	155044. 07 155352. 04 155709. 02	-307. 97 -356. 98	1. 599 1. 720 2. 638
$3s^2 3p^4({}^3P)4s$	$4s {}^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	138244. 51 139259. 22	-1014. 71	1. 334 0. 676	$3s^2 3p^4({}^3P)4p$	$4p {}^4D^{\circ}$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	157234. 93 157674. 30 158168. 71 158429. 05	-439. 37 -494. 41 -260. 34	1. 427 1. 334 1. 199 0. 000
$3s^2 3p^4({}^3P)3d$	$3d {}^4F$	$4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$	142187. 42 142718. 01 143108. 63 143372. 48	-530. 59 -390. 62 -263. 85		$3s^2 3p^4({}^3P)4p$	$4p {}^2D^{\circ}$	$2\frac{1}{2}$ $1\frac{1}{2}$	158731. 20 159394. 32	-663. 12	1. 241 0. 918
$3s^2 3p^4({}^3P)3d$	$3d {}^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	144710. 90 145669. 84	958. 94		$3s^2 3p^4({}^3P)4p$	$4p {}^2P^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$	159707. 46 160240. 35	532. 89	0. 983 1. 244
$3s^2 3p^4({}^3P)3d$	$3d {}^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	147229. 17 147504. 12 147876. 98	274. 95 372. 86		$3s^2 3p^4({}^3P)4p$	$4p {}^4S^{\circ}$	$1\frac{1}{2}$	161049. 65		1. 987
						$3s^2 3p^4({}^3P)4p$	$4p {}^2S^{\circ}$	$\frac{1}{2}$	161090. 31		1. 695

A II—Continued

A II—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$3s^2 3p^4(^1S)4s$	$4s'' \ ^2S$	$\frac{1}{2}$	167308. 66		1. 993	$3s^2 3p^4(^3P)4d$	$4d \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	192557. 77 192712. 93	—155. 16	1. 198 0. 833
$3s^2 3p^4(^1D)4p$	$4p' \ ^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	170401. 88 170531. 29	129. 41	0. 857 1. 140	$3s^2 3p^4(^3P)4f$	$4f \ ^4F^\circ$	$4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$	194800. 97 194822. 95 194862. 31 194997. 65	—21. 98 —39. 36 —135. 34	
$3s^2 3p^4(^1D)4p$	$4p' \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	172214. 74 172817. 14	—602. 40	1. 332 0. 677	$3s^2 3p^4(^3P)4f$	$4f \ ^4D^\circ$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	194883. 96 195032. 13 195298. 62 195282. 50	—148. 17 —266. 49 16. 12	
$3s^2 3p^4(^1D)3d$	$3d' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	172336. 47 172830. 63	—494. 16		$3s^2 3p^4(^1D)5s$	$5s' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	195865. 61 195867. 73	—2. 12	
$3s^2 3p^4(^1D)4p$	$4p' \ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	173348. 78 173394. 33	45. 55	0. 804 1. 202	$3s^2 3p^4(^3P)4f$	$4f \ ^1D^\circ$	$1\frac{1}{2}$	196077. 40		
$3s^2 3p^4(^1D)3d$	$3d' \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	174410. 74 174821. 94?	—411. 20		$3s^2 3p^4(^3P)4f$	$4f \ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	196091. 04 196622. 78 196633. 93	11. 15	
$3s^2 3p^4(^3P)5s$	$5s \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	181595. 04 182223. 06 182952. 14	—628. 02 —729. 08	1. 603 1. 609 2. 550	$3s^2 3p^4(^1D)4d$	$4d' \ ^2G$	$3\frac{1}{2}$ $4\frac{1}{2}$	198595. 91 198604. 78	8. 87	
$3s^2 3p^4(^3P)5s$	$5s \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	183091. 83 183915. 58	—823. 75	1. 445 0. 816	$3s^2 3p^4(^3P)6s$	$6s \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	198813. 17 199138. 92 200111. 16	—325. 75 —972. 24	
$3s^2 3p^4(^3P)4d$	$4d \ ^4D$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	183676. 42 183798. 22 183986. 83 184193. 12	—121. 80 —188. 61 —206. 29	1. 427 1. 370 1. 198 0. 380	$3s^2 3p^4(^1D)4d$	$4d' \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	199447. 56 199982. 96	535. 40	0. 670
$3s^2 3p^4(^1D)3d$	$3d' \ ^2S$	$\frac{1}{2}$	184094. 10			$3s^2 3p^4(^1D)4d$	$4d' \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	199525. 96 199680. 58	154. 62	1. 196
$3s^2 3p^4(^3P)4d$	$4d \ ^4F$	$4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$	185093. 92 185625. 47 186075. 06 186341. 39	—531. 55 —449. 59 —266. 33	1. 330 1. 217 1. 045 0. 612	$3s^2 3p^4(^3P)6s$	$6s \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	200032. 65 200624. 00	—591. 35	
$3s^2 3p^4(^3P)4d$	$4d \ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	186172. 32 186471. 32 186891. 92	299. 00 420. 60	2. 600 1. 494 1. 588	$3s^2 3p^4(^1D)4d$	$4d' \ ^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	200139. 84 200235. 70	—95. 86	0. 862
$3s^2 3p^4(^1S)3d$	$3d'' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	186728. 28 186750. 78	—22. 50		$3s^2 3p^4(^3P)5d$	$5d \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	204418. 50 204515. 81	—97. 31	
$3s^2 3p^4(^3P)4d$	$4d \ ^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	186817. 12 187589. 62	—772. 50	1. 167 0. 861	$3s^2 3p^4(^3P)5d$	$5d \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	204586. 40		
$3s^2 3p^4(^3P)4d$	$4d \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	189935. 62 190593. 62	658. 00	0. 667 1. 322	$3s^2 3p^4(^1D)4d$	$4d' \ ^2S$	$\frac{1}{2}$	205243. 96		2. 004
$3s^2 3p^4(^3P)5p$	$5p \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	190106. 84 190196. 80	—89. 96		$3s^2 3p^4(^1D)4f$	$4f' \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	208592. 90		
$3s^2 3p^4(^3P)5p$	$5p \ ^2D^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$	190508. 00			$3s^2 3p^4(^1D)6s$	$6s' \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	212932. 88 212934. 30	1. 42	
$3s^2 3p^4(^3P)5p$	$5p \ ^2S^\circ$	$\frac{1}{2}$	191708. 46			A III (3P_2)	Limit		222820		
$3s^2 3p^4(^1S)4p$	$4p'' \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	191975. 16 192334. 09	—358. 93	1. 332 0. 760						

A II OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6$	Observed Terms		
$3s^2 3p^5$	$3p^5 \ ^2P^\circ$		
$3s \ 3p^6$	$3p^6 \ ^2S$		
	$ns \ (n \geq 4)$	$np \ (n \geq 4)$	
$3s^2 3p^4(^3P)nx$	$\left\{ \begin{array}{l} 4-6s \ ^4P \\ 4-6s \ ^2P \end{array} \right.$	$4p \ ^4S^\circ$ $4, 5p \ ^2S^\circ$	$4p \ ^4P^\circ$ $4, 5p \ ^2P^\circ$
$3s^2 3p^4(^1D)nx'$		$4p \ ^4D^\circ$ $4, 5p \ ^2D^\circ$	$4p' \ ^2P^\circ$ $4p' \ ^2D^\circ$ $4p'' \ ^2P^\circ$
$3s^2 3p^4(^1S)nx''$			$4p' \ ^2F^\circ$
	$nd \ (n \geq 3)$	$nf \ (n \geq 4)$	
$3s^2 3p^4(^3P)nx$	$\left\{ \begin{array}{lll} 3, 4d \ ^4P & 3, 4d \ ^4D & 3, 4d \ ^4F \\ 3-5d \ ^2P & 3-5d \ ^2D & 3, 4d \ ^2F \end{array} \right.$	$4f \ ^4D^\circ$ $4f \ ^2D^\circ$	
$3s^2 3p^4(^1D)nx'$		$4f' \ ^2P^\circ$	
$3s^2 3p^4(^1S)nx''$			

*For predicted terms in the spectra of the Cl I isoelectronic sequence, see Introduction.

A III

(S I sequence; 16 electrons)

$Z=18$

Ground state $1s^2 2s^2 2p^6 3s^2 3p^4 \ ^3P_2$

$3p^4 \ ^3P_2$ 329965.80 cm^{-1}

I. P. 40.90 volts

The terms are from de Bruin's 1937 paper except for singlets which are from Boyce and Edlén. The $3p^4 \ ^1S$ term, according to Edlén, is derived from the nebular line at 5191.4 Å, identified as the forbidden transition $3p^4 \ ^1D-3p^4 \ ^1S$.

Intersystem combinations connecting the three systems of terms have been observed.

Unfortunately, no complete or homogeneous list of classified lines exists. Such a list is needed to improve the present term values and to explain the numerical discrepancies in the various published papers. De Bruin's terms here designated $3d' \ ^3P^\circ$, $4d'' \ ^3P^\circ$, D° , F° , and $5s'' \ ^3P^\circ$ are apparently based on unpublished observational material.

REFERENCES

- V. v. Keussler, *Zeit. Phys.* **84**, 42 (1933). (I P) (T) (C L)
T. L. de Bruin, *Proc. Roy. Acad. Amsterdam* **36**, No. 7, 724 (1933). (T) (C L)
T. L. de Bruin, *Zeeman Verhandelingen* p. 414 (Martinus Nijhoff, The Hague, 1935). (T) (C L)
J. C. Boyce, *Phys. Rev.* **48**, 397 (1935). (I P) (T) (C L)
J. C. Boyce, *Phys. Rev.* **49**, 351 (1936). (T) (C L)
T. L. de Bruin, *Proc. Roy. Acad. Amsterdam* **40**, No. 4, 343 (1937). (I P) (T) (C L)
B. Edlén, *Phys. Rev.* **62**, 434 (1942). (T) (C L)

A III

A III

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2 3p^4$	$3p^4 \ ^3P$	2 1 0	0. 00 1112. 40 1570. 20	—1112. 40 —457. 80	$3s^2 3p^3(^2D^\circ)4p$	$4p' \ ^3P$	2 1 0	231341. 80 231627. 30 231754. 80	—285. 50 —127. 50
$3s^2 3p^4$	$3p^4 \ ^1D$	2	14010		$3s^2 3p^3(^2P^\circ)4p$	$4p'' \ ^3S$	1	239193. 48	
$3s^2 3p^4$	$3p^4 \ ^1S$	0	33267		$3s^2 3p^3(^2P^\circ)4p$	$4p'' \ ^3D$	1 2 3	240150. 66 240257. 59 240291. 66	106. 93 34. 07
$3s 3p^5$	$3p^5 \ ^3P^\circ$	2 1 0	113800. 70 114797. 60 115328. 40	—996. 90 —530. 80	$3s^2 3p^3(^2P^\circ)4p$	$4p'' \ ^3P$	0 1 2	242923. 96 243145. 76 243424. 97	221. 80 279. 21
$3s 3p^5$	$3p^5 \ ^1P^\circ$	1	144023		$3s^2 3p^3(^4S^\circ)4d$	$4d \ ^5D^\circ$	0 1 2 3 4	 246029. 76 246033. 79 246036. 64 246046. 57	 4. 03 2. 85 9. 93
$3s^2 3p^3(^4S^\circ)3d$	$3d \ ^5D^\circ$	0 1 2 3 4	 144882. 93 144885. 97 144892. 95 144907. 00	 3. 04 6. 98 14. 05	$3s^2 3p^3(^4S^\circ)5s$	$5s \ ^5S^\circ$	2	250712. 27	
$3s^2 3p^3(^4S^\circ)3d$	$3d \ ^3D^\circ$	3 2 1	156917. 62 156924. 68 157031. 40	—7. 06 —106. 72	$3s^2 3p^3(^4S^\circ)4d$	$4d \ ^3D^\circ$	1 2 3	252272. 92 252253. 69 252289. 02	—19. 23 35. 33
$3s^2 3p^3(^4S^\circ)4s$	$4s \ ^5S^\circ$	2	174375. 00		$3s^2 3p^3(^4S^\circ)5s$	$5s \ ^3S^\circ$	1	252575. 88	
$3s^2 3p^3(^4S^\circ)4s$	$4s \ ^3S^\circ$	1	180679. 00		$3s^2 3p^3(^2D^\circ)4d$	$4d' \ ^3F^\circ$	2 3 4	266722. 80 266877. 50 267071. 22	154. 70 193. 72
$3s^2 3p^3(^2D^\circ)3d$	$3d' \ ^3F^\circ$	4 3 2	186402. 15 186657. 20 186903. 05	—255. 05 —245. 85	$3s^2 3p^3(^2D^\circ)4d$	$4d' \ ^3G^\circ$	3 4 5	267782. 10 267833. 20 267895. 82	51. 10 62. 62
$3s^2 3p^3(^2D^\circ)3d$	$3d' \ ^3D^\circ$	1 2 3	187171. 12 187823. 05 188714. 05	651. 93 891. 00	$3s^2 3p^3(^2D^\circ)4d$	$4d' \ ^3D^\circ$	1 2 3	268978. 80 269012. 80 269000. 80	34. 00 —12. 00
$3s^2 3p^3(^2D^\circ)3d$	$3d' \ ^3P^\circ$	0 1 2	 188517. 32		$3s^2 3p^3(^2D^\circ)4d$	$4d' \ ^3P^\circ$	2 1 0	271507. 88 271672. 08 271696. 22	—164. 20 —24. 14
$3s^2 3p^3(^2D^\circ)4s$	$4s' \ ^3D^\circ$	1 2 3	196589. 20 196613. 91 196679. 80	24. 71 65. 89	$3s^2 3p^3(^2D^\circ)4d$	$4d' \ ^3S^\circ$	1	272068. 45	
$3s^2 3p^3(^4S^\circ)4p$	$4p \ ^5P$	1 2 3	204563. 53 204649. 24 204797. 37	85. 71 148. 13	$3s^2 3p^3(^2D^\circ)5s$	$5s' \ ^3D^\circ$	1 2 3	272127. 82 272188. 16 272250. 90	60. 34 62. 74
$3s^2 3p^3(^2D^\circ)3d$	$3d' \ ^3S^\circ$	1	204727. 47		$3s^2 3p^3(^2P^\circ)4d$	$4d'' \ ^3F^\circ$	2 3 4	281461. 97 281473. 82	11. 85
$3s^2 3p^3(^2P^\circ)4s$	$4s'' \ ^3P^\circ$	2 1 0	207233. 09 207532. 15 207673. 16	—299. 06 —141. 01	$3s^2 3p^3(^2P^\circ)4d$	$4d'' \ ^3P^\circ$	0 1 2	281947. 88 282000. 26 282099. 14	52. 38 98. 88
$3s^2 3p^3(^4S^\circ)4p$	$4p \ ^3P$	2 1 0	209151. 82 209127. 04 209166. 35	24. 78 —39. 31	$3s^2 3p^3(^2P^\circ)4d$	$4d'' \ ^3D^\circ$	3 2 1	283919. 78 284096. 26 284118. 51	—176. 48 —22. 25
$3s^2 3p^3(^2P^\circ)3d$	$3d'' \ ^3D^\circ$	3 2 1	210212. 26 211004. 85 211563. 83	—792. 59 —558. 98	$3s^2 3p^3(^2P^\circ)5s$	$5s'' \ ^3P^\circ$	0 1 2	285831. 20 285882. 00 286009. 21	50. 80 127. 21
$3s^2 3p^3(^2P^\circ)3d$	$3d'' \ ^3P^\circ$	2 1 0	213950. 87 214346. 70 214568. 49	—395. 83 —221. 79					
$3s^2 3p^3(^2D^\circ)4p$	$4p' \ ^3D$	1 2 3	225155. 18 225147. 93 225402. 59	—7. 25 254. 66	A IV ($^4S_{3/2}$)	Limit	-----	329965. 80	
$3s^2 3p^3(^2D^\circ)4p$	$4p' \ ^3F$	2 3 4	226355. 96 226503. 22 226646. 06	147. 26 142. 84					

A III OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 +$	Observed Terms			
$3s^2 3p^4$	{ $3p^4 \ ^1S$ $3p^4 \ ^3P$ $3p^4 \ ^1D$			
$3s 3p^5$	{ $3p^5 \ ^3P^o$ $3p^5 \ ^1P^o$			
	$ns \ (n \geq 4)$		$np \ (n \geq 4)$	
$3s^2 3p^3(^4S^o)nx$	{ $4, 5s \ ^5S^o$ $4, 5s \ ^3S^o$		$4p \ ^5P$ $4p \ ^3P$	
$3s^2 3p^3(^2D^o)nx'$	$4, 5s' \ ^3D^o$		$4p' \ ^3P$	$4p' \ ^3D$ $4p' \ ^3F$
$3s^2 3p^3(^2P^o)nx''$	$4, 5s'' \ ^3P^o$		$4p'' \ ^3S$ $4p'' \ ^3P$	$4p'' \ ^3D$
	$nd \ (n \geq 3)$			
$3s^2 3p^3(^4S^o)nx$	{ $3, 4d \ ^5D^o$ $3, 4d \ ^3D^o$			
$3s^2 3p^3(^2D^o)nx'$	$3, 4d' \ ^3S^o$	$3, 4d' \ ^3P^o$	$3, 4d' \ ^3D^o$	$3, 4d' \ ^3F^o$ $4d' \ ^3G^o$
$3s^2 3p^3(^2P^o)nx''$		$3, 4d'' \ ^3P^o$	$3, 4d'' \ ^3D^o$	$4d'' \ ^3F^o$

*For predicted terms in the spectra of the S I isoelectronic sequence, see Introduction.

A IV

(P I sequence; 15 electrons)

Z=18

Ground state $1s^2 2s^2 2p^6 3s^2 3p^3 \ ^4S_{1/2}$

$3p^3 \ ^4S_{1/2} \ 482400 \text{ cm}^{-1}$

I. P. 59.79 volts

The analysis is incomplete. Boyce has classified 26 lines in the range between 396 Å and 1197 Å and listed 8 terms.

De Bruin has extended the analysis and published the term list which is quoted here. Intersystem combinations connecting the doublet and quartet terms have been observed.

The ionization potential estimated by Edlén from isoelectronic sequence data has been used to calculate the limit (entered in brackets in the table).

REFERENCES

- J. C. Boyce, Phys. Rev. **48**, 401 (1935). (I P) (T) (C L)
 B. Edlén, *Zeeman Verhandelingen* p. 91 (Martinus Nijhoff, The Hague, 1935). (I P)
 T. L. de Bruin, Physica **3**, No. 8, 809 (1936). (T) (C L)
 A. B. Rao, Ind. J. Phys. **12**, 399 (1938). (T) (C L)

A IV

A IV

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2 3p^3$	$3p^3 \ ^4S^\circ$	$1\frac{1}{2}$	0. 00		$3s^2 3p^2(^3P)4p$	$4p \ ^4D^\circ$	$\frac{1}{2}$	285960. 17	
$3s^2 3p^3$	$3p^3 \ ^2D^\circ$	$1\frac{1}{2}$	21090	129			$1\frac{1}{2}$	286228. 80	268. 63
		$2\frac{1}{2}$	21219				$2\frac{1}{2}$	286751. 68	522. 88
							$3\frac{1}{2}$	287555. 83	804. 15
$3s^2 3p^3$	$3p^3 \ ^2P^\circ$	$\frac{1}{2}$	34854	181	$3s^2 3p^2(^3P)4p$	$4p \ ^4P^\circ$	$\frac{1}{2}$	289125. 88	
		$1\frac{1}{2}$	35035				$1\frac{1}{2}$	289237. 82	111. 94
							$2\frac{1}{2}$	289834. 68	596. 86
$3s 3p^4$	$3p^4 \ ^4P$	$2\frac{1}{2}$	117564	—951	$3s^2 3p^2(^3P)4p$	$4p \ ^2D^\circ$	$1\frac{1}{2}$	290256. 45	1411. 28
		$1\frac{1}{2}$	118515	—529			$2\frac{1}{2}$	291667. 73	
		$\frac{1}{2}$	119044		$3s^2 3p^2(^3P)4p$	$4p \ ^4S^\circ$	$1\frac{1}{2}$	291748. 70	
$3s 3p^4$	$3p^4 \ ^2D$	$1\frac{1}{2}$	145921	79	$3s^2 3p^2(^3P)4p$	$4p \ ^2P^\circ$	$\frac{1}{2}$	295674. 54	132. 23
		$2\frac{1}{2}$	146000				$1\frac{1}{2}$	295806. 77	
$3s 3p^4$	$3p^4 \ ^2P$	$1\frac{1}{2}$	166356	—1088	$3s^2 3p^2(^3P)4p$	$4p \ ^2S^\circ$	$\frac{1}{2}$	299563. 20	
		$\frac{1}{2}$	167444		$3s^2 3p^2(^1D)4p$	$4p' \ ^2F^\circ$	$2\frac{1}{2}$	304074. 29	325. 61
$3p^4$	$3p^4 \ ^2S$	$\frac{1}{2}$	177833				$3\frac{1}{2}$	304399. 90	
$3s^2 3p^2(^3P)4s$	$4s \ ^4P$	$\frac{1}{2}$	250219. 45	687. 15	$3s^2 3p^2(^1D)4p$	$4p' \ ^2D^\circ$	$2\frac{1}{2}$	306236. 28	—71. 97
		$1\frac{1}{2}$	250906. 60	1065. 40			$1\frac{1}{2}$	306308. 25	
		$2\frac{1}{2}$	251972. 00						
$3s^2 3p^2(^3P)4s$	$4s \ ^2P$	$\frac{1}{2}$	256093. 29	1255. 60					
		$1\frac{1}{2}$	257348. 89						
$3s^2 3p^2(^1D)4s$	$4s' \ ^2D$	$2\frac{1}{2}$	268151. 38	—20. 00					
		$1\frac{1}{2}$	268171. 38		A v (3P_0)	Limit	-----	[482400]	

November 1947.

A IV OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 +$	Observed Terms					
$3s^2 3p^3$	{ $3p^3 \ ^4S^\circ$ $3p^3 \ ^2P^\circ$ $3p^3 \ ^2D^\circ$					
$3s 3p^4$	{ $3p^4 \ ^2S$ $3p^4 \ ^4P$ $3p^4 \ ^2P$ $3p^4 \ ^2D$					
	<i>ns</i> ($n \geq 4$)			<i>np</i> ($n \geq 4$)		
$3s^2 3p^2(^3P)nx$	{ $4s \ ^4P$ $4s \ ^2P$			$4p \ ^4S^\circ$	$4p \ ^4P^\circ$	$4p \ ^4D^\circ$
				$4p \ ^2S^\circ$	$4p \ ^2P^\circ$	$4p \ ^2D^\circ$
$3s^2 3p^2(^1D)nx'$				$4p' \ ^2D^\circ$ $4p' \ ^2F^\circ$		
	$4s' \ ^2D$					

*For predicted terms in the spectra of the P I isoelectronic sequence, see Introduction.

(Si I sequence; 14 electrons)

Z=18

Ground state $1s^2 2s^2 2p^6 3s^2 3p^2 \ ^3P_0$

$3p^2 \ ^3P_0$ 605100 cm^{-1}

I. P. 75.0 volts

The terms have been taken from the paper by Phillips and Parker. This includes the earlier work by Boyce. Thirty-six lines have been classified in the region between 336 Å and 836 Å. Intersystem combinations connecting the singlet and triplet terms have been observed. No quintet terms have been found.

Using the method suggested by Edlén for extrapolation along the isoelectronic sequence, the writer has estimated the value of the limit quoted above and entered in brackets in the table.

REFERENCES

J. C. Boyce, Phys. Rev. **48**, 401 (1935). (I P) (T) (C L)
 B. Edlén, *Zeeman Verhandelingen* p. 91 (Martinus Nijhoff, The Hague, 1935). (I P)
 L. W. Phillips and W. L. Parker, Phys. Rev. **60**, 301 (1941). (T) (C L)

A v

A v

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2\ 3p^2$	$3p^2\ ^3P$	0	0	765 1267	$3s^2\ 3p(^2P^\circ)3d$	$3d\ ^3P^\circ$	2	217578	—708 —356
		1	765				1	218286	
		2	2032				0	218642	
$3s^2\ 3p^2$	$3p^2\ ^1D$	2	16301	46 132	$3s^2\ 3p(^2P^\circ)3d$	$3d\ ^3D^\circ$	1	224216	289 212
$3s\ 3p^3$	$3p^3\ ^3D^\circ$	1	121632				2	224505	
		2	121678				3	224717	
		3	121810		$3s^2\ 3p(^2P^\circ)4s$	$4s\ ^3P^\circ$	0	295742	507 1644
$3s\ 3p^3$	$3p^3\ ^3P^\circ$	2	141764	—9		1	296249		
		1, 0	141773			2	297893		
$3s\ 3p^3$	$3p^3\ ^3S^\circ$	1	191537		$3s^2\ 3p(^2P^\circ)4s$	$4s\ ^1P^\circ$	1	301300	
$3s\ 3p^3$	$3p^3\ ^1P^\circ$	1	195356		-----	-----	-----	[605100]	
					A VI ($^2P^\circ_{3/2}$)	Limit	-----		

October 1947.

A VI

(Al I sequence; 13 electrons)

 $Z=18$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^2 P_{1/2}^{\circ}$ $3p^2 P_{1/2}^{\circ} 736600 \text{ cm}^{-1}$

I. P. 91.3 volts

The analysis is by Phillips and Parker, who have classified 37 lines in the region between 180 Å and 596 Å. No intersystem combinations have been observed. They estimate that $3p^2 P_{1/2}^{\circ}$ is $100,000 \text{ cm}^{-1}$ above the ground state, with an uncertainty x equal to $\pm 1000 \text{ cm}^{-1}$. This value is entered in brackets in the table, and it has been added to the published values of all quartet terms.

Their limit, derived from the three members of the $3p^2 P^{\circ} - nd^2 D$ series is $721300 \pm 300 \text{ cm}^{-1}$ (I. P. 89.41 ± 0.04). Using the method suggested by Edlén, the writer has extrapolated the value of the limit quoted above and entered in brackets in the table. The uncertainty in this estimate is large because of the incompleteness of the isoelectronic sequence data.

REFERENCE

L. W. Phillips and W. L. Parker, Phys. Rev. **60**, 301 (1941). (I P) (T) (C L)

A VI

A VI

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3s^2(^1S)3p$	$3p^2 P^{\circ}$	$\frac{1}{2}$ $\frac{3}{2}$	0 2210	2210	$3s^2(^1S)3d$	$3d^2 D^{\circ}$	$\frac{1}{2}$ $\frac{3}{2}$	$319121+x$ $319393+x$	272 222
$3s^2(^1S)3p^2$	$3p^2^4 P$	$\frac{1}{2}$ $\frac{3}{2}$ $\frac{5}{2}$	$[100000]+x$ $100802+x$ $102034+x$	802 1232	$3s^2(^1S)4s$	$4s^2 S$	$\frac{1}{2}$	$319615+x$ $319747+x$	132
$3s^2(^1S)3p^2$	$3p^2^2 S$	$\frac{1}{2}$	169801		$3s^2(^1S)4s$	$4s^2 S$	$\frac{1}{2}$	342286	
$3s^2(^1S)3p^2$	$3p^2^2 P$	$\frac{1}{2}$ $\frac{3}{2}$	182182 183577	1395	$3s^2(^1S)4s$	$4s^2 P^{\circ}$	$\frac{1}{2}$ $\frac{3}{2}$	$453954+x$ $454716+x$ $456115+x$	762 1399
$3s^2(^1S)3d$	$3d^2 D$	$\frac{1}{2}$ $\frac{3}{2}$	218592 218657	65	$3s^2(^1S)4d$	$4d^2 D$	$\frac{1}{2}$ $\frac{3}{2}$	454760 454810	50
$3p^3$	$3p^3^4 S^{\circ}$	$\frac{1}{2}$	$270356+x$		$3s^2(^1S)5d$	$5d^2 D$	$\frac{1}{2}$ $\frac{3}{2}$	555330 555555	225
$3s^2(^1S)3p(^3P^{\circ})3d$	$3d^2 P^{\circ}$	$\frac{1}{2}$ $\frac{3}{2}$ $\frac{5}{2}$	$316199+x$ $316815+x$ $317298+x$	-616 -483					
					A VII ($1S_0$)	Limit		[736600]	

September 1947.

A VI OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 +$	Observed Terms	
$3s^2(^1S)3p$	$3p^2 P^{\circ}$	
$3s^2(^1S)3p^2$	$\begin{cases} 3p^2^2 S \\ 3p^2^4 P \\ 3p^2^2 P \end{cases}$	
$3p^3$	$3p^3^4 S^{\circ}$	
	$ns (n \geq 4)$	$nd (n \geq 3)$
$3s^2(^1S)nx$	$4s^2 S$	$3-5d^2 D$
$3s^2(^1S)3p(^3P^{\circ})nx$	$4s^2 P^{\circ}$	$3d^2 P^{\circ} \quad 3d^2 D^{\circ}$

*For predicted terms in the spectra of the Al I isoelectronic sequence, see Introduction.

A VII

(Mg I sequence; 12 electrons)

Z=18

Ground state $1s^2 2s^2 2p^6 3s^2 {}^1S_0$ $3s^2 {}^1S_0$ 1000400 cm^{-1}

I. P. 124.0 volts

Phillips and Parker have classified 25 lines in the interval between 151 Å and 644 Å. No intersystem combinations have been observed.

From the D-series they derive an absolute value of $3p {}^3P_0^\circ$ equal to $891000 \pm 200 \text{ cm}^{-1}$, and by extrapolation along the isoelectronic sequence estimate the absolute value of $3s^2 {}^1S_0$ as $1005000 \pm 1000 \text{ cm}^{-1}$.

From later data on this sequence the writer has extrapolated these values by the method suggested by Edlén, and adopted the revised entries given in the table in brackets.

REFERENCE

L. W. Phillips and W. L. Parker, Phys. Rev. **60**, 305 (1941). (I P) (T) (C L)

A VII

A VII

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2$	$3s^2 {}^1S$	0	0		$3s({}^2S)4p$	$4p {}^1P^\circ$	1	566362	
$3s({}^2S)3p$	$3p {}^3P^\circ$	0 1 2	[113095]+ <i>x</i> 113900 + <i>x</i> 115581 + <i>x</i>	805 1681	$3s({}^2S)4d$	$4d {}^3D$	1 2 3	634584 + <i>x</i> 634622 + <i>x</i> 634697 + <i>x</i>	38 75
$3s({}^2S)3p$	$3p {}^1P^\circ$	1	170720		$3s({}^2S)4f$	$4f {}^3F^\circ$	2, 3, 4	660092	
$3p^2$	$3p^2 {}^3P$	0 1 2	269829 + <i>x</i> 270770 + <i>x</i> 272554 + <i>x</i>	941 1784	$3s({}^2S)5d$	$5d {}^3D$	1 2 3	772300 + <i>x</i> 772325 + <i>x</i> 772355 + <i>x</i>	25 30
$3s({}^2S)3d$	$3d {}^3D$	1 2 3	324097 + <i>x</i> 324136 + <i>x</i> 324184 + <i>x</i>	39 48	A VIII (${}^2S_{1/2}$)				
$3s({}^2S)4s$	$4s {}^3S$	1	514083 + <i>x</i>		Limit		-----	[1000400]	

August 1947.

A VIII

(Na I sequence; 11 electrons)

Z=18

Ground state $1s^2 2s^2 2p^6 3s {}^2S_{1/2}$ $3s {}^2S_{1/2}$ 1157400 cm^{-1} I. P. 143.46 ± 0.05 volts

Phillips and Parker classified 23 lines in the interval 120 Å to 526 Å. The resonance lines calculated at 700.398 Å and 713.990 Å, have not been observed. Absolute term values were derived from four members of the 2D -series.

REFERENCE

L. W. Phillips and W. L. Parker, Phys. Rev. **60**, 305 (1941). (I P) (T) (C L)

A VIII

A VIII

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
3s	3s ² S	½	0		5s	5s ² S	½	812422	
3p	3p ² P°	½ 1½	140058 142776	2718	5p	5p ² P°	½ 1½	832245 832691	446
3d	3d ² D	1½ 2½	332576 332727	151	5d	5d ² D	1½ 2½	865084 865111	27
4s	4s ² S	½	575910		5f	5f ² F°	2½ 3½	875248 875277	29
4p	4p ² P°	½ 1½	628240 629237	997	6d	6d ² D	{ 1½ 2½ }	955560	
4d	4d ² D	1½ 2½	697471 697548	77					
4f	4f ² F°	2½ 3½	716818 716852	34	A IX (¹ S ₀)	Limit	-----	1157400	

June 1947.

A IX

(Ne I sequence; 10 electrons)

Z=18

Ground state 1s² 2s² 2p⁶ ¹S₀

2p⁶ ¹S₀ cm⁻¹

I. P. 421 volts

Two lines observed at 49.180 Å and 48.730 Å have been classified by Phillips and Parker as combinations with the ground term. The measurements may be in error by ±0.002 Å or ±100 cm⁻¹.

As for Ne I, the *jl*-coupling notation in the general form suggested by Racah is here introduced.

REFERENCES

- L. W. Phillips and W. L. Parker, Phys. Rev. **60**, 306 (1941). (T) (C L)
 G. Racah, Phys. Rev. **61**, 537 (L) (1942).
 B. Edlén, Zeit. Astroph. **22**, 62 (1942). (I P)

A IX

Authors	Config.	Desig.	<i>J</i>	Level
¹ S ₀	2p ⁶	2p ⁶ ¹ S	0	0
³ P ₁	2p ⁵ (² P _{1/2})3s	3s [1½]°	2 1	2033350
¹ P ₁	2p ⁵ (² P _{3/2})3s	3s' [½]°	0 1	2052120

April 1947.

A X

(F I sequence; 9 electrons)

Z=18

Ground state $1s^2 2s^2 2p^5 {}^2P_{1/2}^{\circ}$ $2p^5 {}^2P_{1/2}^{\circ}$ cm⁻¹

I. P. volts

This spectrum has not been analyzed. By interpolation along the F I isoelectronic sequence from F I through Ca XII, Edlén derives a reliable estimated value of the interval of the ground term, $2p^5 {}^2P_{1/2}^{\circ} - 2p^5 {}^2P_{3/2}^{\circ}$, equal to 18063 cm⁻¹. The faint coronal line observed at 5536 Å, wave number 18059 cm⁻¹, may thus be tentatively identified as this forbidden line of A X, according to Edlén.

REFERENCE

B. Edlén, Zeit. Astroph. **22**, 59 (1942). (T)

March 1947.

A XI

(O I sequence; 8 electrons)

Z=18

Ground state $1s^2 2s^2 2p^4 {}^3P_2$ $2p^4 {}^3P_2$ cm⁻¹

I. P. volts

This spectrum has not been analyzed. By extrapolation along the O I isoelectronic sequence Edlén estimates the separation $2p^4 {}^3P_2 - 2p^4 {}^3P_1$ to be approximately 14449 cm⁻¹, or 6919 Å. This line has not been identified in the solar corona.

REFERENCE

B. Edlén, Zeit. Astroph. **22**, 59 (1942). (T)

March 1947.

A XIV

(B I sequence; 5 electrons)

Z=18

Ground state $1s^2 2s^2 2p {}^2P_{1/2}^{\circ}$ $2p {}^2P_{1/2}^{\circ}$ cm⁻¹

I. P. volts

By extrapolation of the B I isoelectronic sequence, Edlén estimates that the separation of the lowest term $2p {}^2P_{1/2}^{\circ} - 2p {}^2P_{3/2}^{\circ}$, falls near enough to warrant tentative identification of the coronal line observed at 4359 Å (wave number 22935 cm⁻¹) as [A XIV].

REFERENCE

B. Edlén, Zeit. Astroph. **22**, 59 (1942). (T)

March 1947.

POTASSIUM

K I

19 electrons

Z=19

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 4s \ ^2S_{1/2}$ $4s \ ^2S_{1/2}$ 35009.78 cm^{-1}

I. P. 4.339 volts

H. R. Kratz has observed in absorption the $np \ ^2P^\circ$ series to $n=79$. He has generously furnished a list of his final term values in advance of publication, for inclusion here. His value of the limit is quoted. The series $ns \ ^2S$ ($n=4$ to 8), $nd \ ^2D$ ($n=3$ to 6), and $nf \ ^2F^\circ$ ($n=4$ to 9) are from Edlén, who revised the older values. Edlén remarks that the $ns \ ^2S$ and $nd \ ^2D$ series can best be continued by an extrapolation of the appropriate series formula, since the observed wavelengths are uncertain. This comment applies to the listed values of $ns \ ^2S$ ($n=9$ to 13), which are from Fowler's Report. Mack has furnished revised values of $nd \ ^2D$ ($n=8$ to 13), derived from observations of the forbidden transitions $6s-nd$ on the plates of Kratz. The last two members of this series are, respectively, 34213.1 and 34332.6.

From Paschen's classifications of far infrared lines Edlén concludes that the $5g \ ^2G$ and $6h \ ^2H^\circ$ terms are H-like. The terms derived from these calculations are entered in brackets in the table. Compared with all others, the terms $4f \ ^2F^\circ$, $5f \ ^2F^\circ$, and $5s \ ^2S$, derived from far infrared observations, are somewhat uncertain, according to Edlén.

No attempt has been made to give a complete bibliography of papers dealing with hyperfine structure of K I. From interferometric measures of the combinations $4p \ ^2P^\circ - nd \ ^2D$ ($n=5$ to 8) Masaki and Kobayakawa observe the following term intervals:

	$n=5$	6	7	8
$nd \ ^2D$	-0.503	-0.262	-0.158	-0.096
$4p \ ^2P_{3/2}^\circ - 4p \ ^2P_{1/2}^\circ$	57.600	57.600	57.599	57.600

The papers on Zeeman effect deal only with forbidden transitions of K I. From observations in a magnetic field of the lines at 4642 Å and 4641 Å ($4s \ ^2S - 3d \ ^2D$) Segrè and Bakker observe the interval of $3d \ ^2D$ to be $2.325 \pm 0.015 \text{ cm}^{-1}$.

The Kr^b resonance lines have been observed in absorption by Beutler and Guggenheimer at 662.38 Å and 653.31 Å. The $4s^2 \ ^2P^\circ$ term in the table has been calculated from these lines.

REFERENCES

- S. Datta, Proc. Roy. Soc. London [A] **101**, 539 (1922). (I P) (T) (C L)
 A. Fowler, *Report on Series in Line Spectra* p. 101 (Fleetway Press, London, 1922). (I P) (T) (C L)
 F. Paschen und R. Götze, *Seriengesetze der Linienspektren* p. 59 (Julius Springer, Berlin, 1922). (I P) (T) (C L)
 W. Grotrian, *Graphische Darstellung der Spektren von Atomen und Ionen mit ein, zwei and drei Valenzelektronen*, Part II, p. 29 (Julius Springer, Berlin, 1928). (G D)
 E. Segrè und C. J. Bakker, Zeit. Phys. **72**, 724 (1931). (Z E)
 H. Beutler und K. Guggenheimer, Zeit. Phys. **87**, 188 (1933). (T) (C L)
 W. F. Meggers, Bur. Std. J. Research **10**, 673, RP558 (1933). (C L)
 W. F. Meggers, J. Research Nat. Bur. Std. **14**, 497, RP781 (1935). (C L)
 B. Edlén, Zeit. Phys. **98**, 453 (1936). (I P) (T) (C L)
 O. Masaki and K. Kobayakawa, J. Sci. Hiroshima Univ. [A] **6**, 217 (1936). (C L)
 F. A. Jenkins and E. Segrè, Phys. Rev. **55**, 545 (1939). (Z E)
 W. F. Meggers, J. Opt. Soc. Am. **36**, 431 (1946). (Summary hfs)
 H. R. Kratz, unpublished material (Dec. 1947). (I P) (T)

K I

K I

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3p^6(^1S)4s$	$4s\ ^2S$	$\frac{1}{2}$	0. 00		$3p^6(^1S)11s$	$11s\ ^2S$	$\frac{1}{2}$	33598. 17	
$3p^6(^1S)4p$	$4p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	12985. 17 13042. 89	57. 72	$3p^6(^1S)9f$	$9f\ ^2F^\circ$	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	33652. 0	
$3p^6(^1S)5s$	$5s\ ^2S$	$\frac{1}{2}$	21026. 8		$3p^6(^1S)11p$	$11p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	33736. 60 33737. 44	0. 84
$3p^6(^1S)3d$	$3d\ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	21534. 42 21536. 75	-2. 33	$3p^6(^1S)10d$	$10d\ ^2D$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	33851. 76	
$3p^6(^1S)5p$	$5p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	24701. 44 24720. 20	18. 76	$3p^6(^1S)12s$	$12s\ ^2S$	$\frac{1}{2}$	33869. 7	
$3p^6(^1S)4d$	$4d\ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	27397. 01 27398. 11	-1. 10	$3p^6(^1S)12p$	$12p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	33972. 34 33972. 94	0. 60
$3p^6(^1S)6s$	$6s\ ^2S$	$\frac{1}{2}$	27450. 65		$3p^6(^1S)11d$	$11d\ ^2D$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	34056. 9	
$3p^6(^1S)4f$	$4f\ ^2F^\circ$	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	28127. 7		$3p^6(^1S)13s$	$13s\ ^2S$	$\frac{1}{2}$	34069. 3	
$3p^6(^1S)6p$	$6p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	28999. 29 29007. 70	8. 41	$3p^6(^1S)13p$	$13p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	34148. 15 34148. 63	0. 48
$3p^6(^1S)5d$	$5d\ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	30185. 18 30185. 69	-0. 51	$3p^6(^1S)14p$	$14p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	34282. 77 34283. 15	0. 38
$3p^6(^1S)7s$	$7s\ ^2S$	$\frac{1}{2}$	30274. 26		$3p^6(^1S)15p$	$15p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	34388. 16 34388. 46	0. 30
$3p^6(^1S)5f$	$5f\ ^2F^\circ$	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	30605. 6		$3p^6(^1S)16p$	$16p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	34472. 18 34472. 43	0. 25
$3p^6(^1S)5g$	$5g\ ^2G$	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 4\frac{1}{2} \end{array} \right\}$	[30619. 8]		$3p^6(^1S)17p$	$17p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	34540. 23 34540. 44	0. 21
$3p^6(^1S)7p$	$7p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	31069. 98 31074. 46	4. 48	$3p^6(^1S)18p$	$18p\ ^2P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	34596. 27	
$3p^6(^1S)6d$	$6d\ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	31695. 51 31695. 75	-0. 24	$3p^6(^1S)19p$	$19p\ ^2P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	34642. 78	
$3p^6(^1S)8s$	$8s\ ^2S$	$\frac{1}{2}$	31764. 95		$3p^6(^1S)20p$	$20p\ ^2P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	34681. 84	
$3p^6(^1S)6f$	$6f\ ^2F^\circ$	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	31953. 0		$3p^6(^1S)21p$	$21p\ ^2P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	34714. 98	
$3p^6(^1S)6h$	$6h\ ^2H^\circ$	$\left\{ \begin{array}{l} 4\frac{1}{2} \\ 5\frac{1}{2} \end{array} \right\}$	[31960. 6]		$3p^6(^1S)22p$	$22p\ ^2P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	34743. 37	
$3p^6(^1S)6g$	$6g\ ^2G$	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 4\frac{1}{2} \end{array} \right\}$	[31960. 8]		$3p^6(^1S)23p$	$23p\ ^2P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	34767. 78	
$3p^6(^1S)8p$	$8p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	32227. 42 32230. 12	2. 70	$3p^6(^1S)24p$	$24p\ ^2P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	34789. 03	
$3p^6(^1S)7a$	$7d\ ^2D$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	32598. 46		$3p^6(^1S)25p$	$25p\ ^2P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	34807. 62	
$3p^6(^1S)9s$	$9s\ ^2S$	$\frac{1}{2}$	32648. 17		$3p^6(^1S)26p$	$26p\ ^2P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	34823. 83	
$3p^6(^1S)7f$	$7f\ ^2F^\circ$	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	32764. 52		$3p^6(^1S)27p$	$27p\ ^2P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	34838. 30	
$3p^6(^1S)9p$	$9p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	32940. 34 32942. 08	1. 74	$3p^6(^1S)28p$	$28p\ ^2P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	34851. 11	
$3p^6(^1S)8d$	$8d\ ^2D$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	33178. 36		$3p^6(^1S)29p$	$29p\ ^2P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	34862. 52	
$3p^6(^1S)10s$	$10s\ ^2S$	$\frac{1}{2}$	33214. 39		$3p^6(^1S)30p$	$30p\ ^2P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	34872. 70	
$3p^6(^1S)8f$	$8f\ ^2F^\circ$	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	33291. 04		$3p^6(^1S)31p$	$31p\ ^2P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	34881. 94	
$3p^6(^1S)10p$	$10p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	33410. 34 33411. 54	1. 20	$3p^6(^1S)32p$	$32p\ ^2P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	34890. 20	
$3p^6(^1S)9d$	$9d\ ^2D$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	33572. 11						

K I—Continued

K I—Continued

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3p^6(^1S)33p$	$33p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34897. 75		$3p^6(^1S)58p$	$58p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34975. 15	
$3p^6(^1S)34p$	$34p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34904. 57		$3p^6(^1S)59p$	$59p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34976. 36	
$3p^6(^1S)35p$	$35p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34910. 79		$3p^6(^1S)60p$	$60p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34977. 50	
$3p^6(^1S)36p$	$36p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34916. 51		$3p^6(^1S)61p$	$61p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34978. 62	
$3p^6(^1S)37p$	$37p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34921. 69		$3p^6(^1S)62p$	$62p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34979. 60	
$3p^6(^1S)38p$	$38p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34926. 47		$3p^6(^1S)63p$	$63p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34980. 65	
$3p^6(^1S)39p$	$39p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34930. 91		$3p^6(^1S)64p$	$64p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34981. 58	
$3p^6(^1S)40p$	$40p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34934. 97		$3p^6(^1S)65p$	$65p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34982. 47	
$3p^6(^1S)41p$	$41p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34938. 72		$3p^6(^1S)66p$	$66p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34983. 27	
$3p^6(^1S)42p$	$42p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34942. 20		$3p^6(^1S)67p$	$67p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34984. 10	
$3p^6(^1S)43p$	$43p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34945. 49		$3p^6(^1S)68p$	$68p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34984. 83	
$3p^6(^1S)44p$	$44p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34948. 48		$3p^6(^1S)69p$	$69p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34985. 57	
$3p^6(^1S)45p$	$45p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34951. 26		$3p^6(^1S)70p$	$70p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34986. 25	
$3p^6(^1S)46p$	$46p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34953. 85		$3p^6(^1S)71p$	$71p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34986. 96	
$3p^6(^1S)47p$	$47p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34956. 32		$3p^6(^1S)72p$	$72p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34987. 53	
$3p^6(^1S)48p$	$48p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34958. 61		$3p^6(^1S)73p$	$73p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34988. 19	
$3p^6(^1S)49p$	$49p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34960. 73		$3p^6(^1S)74p$	$74p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34988. 85	
$3p^6(^1S)50p$	$50p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34962. 83		$3p^6(^1S)75p$	$75p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34989. 4	
$3p^6(^1S)51p$	$51p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34964. 67		$3p^6(^1S)76p$	$76p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34989. 9	
$3p^6(^1S)52p$	$52p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34966. 45		$3p^6(^1S)77p$	$77p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34990. 5	
$3p^6(^1S)53p$	$53p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34968. 09		$3p^6(^1S)78p$	$78p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34990. 8	
$3p^6(^1S)54p$	$54p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34969. 69		$3p^6(^1S)79p$	$79p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34991. 2	
$3p^6(^1S)55p$	$55p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34971. 17		-----				
$3p^6(^1S)56p$	$56p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34972. 57		K II (1S_0)	<i>Limit</i>		35009. 78	
$3p^6(^1S)57p$	$57p\ ^2P^\circ$	$\left\{ \begin{smallmatrix} \frac{1}{2} \\ 1\frac{1}{2} \end{smallmatrix} \right\}$	34973. 88		$3p^5(^3P_2)4s^2$	$4s^2\ ^2P^\circ$	$\frac{1}{2}$	150970	-2096
					$3p^5(^1P_1)4s^2$		$\frac{1}{2}$	153066	

(A I sequence; 18 electrons)

Z=19

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 {}^1S_0$ $3p^6 {}^1S_0$ 256637 cm^{-1}

I. P. 31.81 volts

Most of the levels were found by de Bruin, whose analysis is repeated in the three references listed under his name. The present list is taken from the paper by Bowen, who extended the earlier work by observations in the ultraviolet near 600 Å, which served to connect de Bruin's levels with the ground term. Bowen also determined the limit from the 4s- and 5s-series and extended the assignments of the Paschen notation to all but 2 of the 20 levels thus far identified in this spectrum. This notation is entered in column one of the table under the heading "A I".

As for A I, the jl -coupling notation in the general form suggested by Racah is adopted. The writer has suggested tentatively the tabular designation of the level labeled Y_{11} by de Bruin. The pairs $nd[3\frac{1}{2}]^\circ$ and $nd[1\frac{1}{2}]^\circ$ are partially inverted as compared with Ne I.

The LS -designations $ns {}^3P_{210}^\circ$, ${}^1P_1^\circ$ can probably be safely assigned to the levels ns_5 , ns_4 , ns_3 , ns_2 , respectively.

REFERENCES

- T. L. de Bruin, Zeit. Phys. **38**, 94, 1926; Proc. Royal Acad. Amsterdam **29**, No. 5, 713 (1926); Arch. Néerl. Sci. exactes et naturelles, [IIIA] **11**, 75 (1928). (T) (C L) (E D) (Z E)
 I. S. Bowen, Phys. Rev. **31**, 499 (1928). (I P) (T) (C L)
 G. Racah, Phys. Rev. **61**, 537 (L) (1942).

K II

K II

A I	de Bruin	Config.	Desig.	J	Level	A I	de Bruin	Config.	Desig.	J	Level
$1p_0$		$3p^6$	$3p^6 {}^1S$	0	0	$2p_2$ $2p_1$	P_9 P_{10}	$3p^5({}^2P_{3/2}^\circ)4p$	$4p' [\frac{1}{2}]$	1 0	190134. 8 194776. 1
$1s_5$ $1s_4$	X_2 X_3	$3p^5({}^2P_{1/2}^\circ)4s$	$4s [1\frac{1}{2}]^\circ$	2 1	162507. 0 163237. 0	$2s_5$ $2s_4$	Y_2 Y_3	$3p^5({}^2P_{1/2}^\circ)5s$	$5s [1\frac{1}{2}]^\circ$	2 1	212575. 5 212992. 9
$1s_3$ $1s_2$	X_7 X_8	$3p^5({}^2P_{3/2}^\circ)4s$	$4s' [\frac{1}{2}]^\circ$	0 1	165149. 5 166461. 5	$2s_3$ $2s_2$	Y_4 Y_5	$3p^5({}^2P_{3/2}^\circ)5s$	$5s' [\frac{1}{2}]^\circ$	0 1	214727. 0 215018. 8
$3d_6$ $3d_5$	X_4 X_5	$3p^5({}^2P_{1/2}^\circ)3d$	$3d [\frac{1}{2}]^\circ$	0 1	163436. 3 164496. 1			$3p^5({}^2P_{1/2}^\circ)4d$	$4d [\frac{1}{2}]^\circ$	0 1	
		"	$3d [3\frac{1}{2}]^\circ$	4		$4d_5$	Y_6	"	$4d [3\frac{1}{2}]^\circ$	4	215404. 9
$3d_4$	X_9	"	$3d [1\frac{1}{2}]^\circ$	3	170835. 4	$4d_4$	Y_9	"	$4d [1\frac{1}{2}]^\circ$	3	217726. 4
$3d_3$	X_6	"	$3d [1\frac{1}{2}]^\circ$	2 1	164932. 3	$4d_3$	Y_7	"	$4d [1\frac{1}{2}]^\circ$	2 1	215855. 8
$3d_1''$	X_{10}	"	$3d [2\frac{1}{2}]^\circ$	2 3	171526. 8	$4d_1''$	Y_{10}	"	$4d [2\frac{1}{2}]^\circ$	2 3	219196. 2
$2p_{10}$	P_1	$3p^5({}^2P_{1/2}^\circ)4p$	$4p [\frac{1}{2}]$	1	183208. 4		Y_8	$3p^5({}^2P_{3/2}^\circ)4d$	$4d' [?]^\circ$	2	217066. 3
$2p_9$ $2p_8$	P_2 P_3	"	$4p [2\frac{1}{2}]$	3 2	186388. 5 186685. 6		Y_{11}	"	$4d' [1\frac{1}{2}]^\circ$	2 1	223124. 1
$2p_7$ $2p_6$	P_4 P_5	"	$4p [1\frac{1}{2}]$	1 2	187531. 1 188154. 4						
$2p_5$	P_8	"	$4p [\frac{1}{2}]$	0	189772. 0			K III (${}^2P_{1/2}^\circ$)	Limit	-----	256637
$2p_4$ $2p_3$	P_6 P_7	$3p^5({}^2P_{3/2}^\circ)4p$	$4p' [1\frac{1}{2}]$	1 2	189243. 7 189661. 7			K III (${}^2P_{3/2}^\circ$)	Limit	-----	258803

May 1948.

K II OBSERVED LEVELS *

Config. $1s^2 2s^2 2p^6 3s^2 +$	Observed Terms		
$3p^6$	$3p^6 \ ^1S$		
	$ns \ (n \geq 4)$		
$3p^5(^2P^\circ)nx$	$\left\{ \begin{array}{l} 4, 5s \ ^3P^\circ \\ 4, 5s \ ^1P^\circ \end{array} \right.$		
<i>jl</i> -Coupling Notation			
	Observed Levels		
	$ns \ (n \geq 4)$	$np \ (n \geq 4)$	$nd \ (n \geq 3)$
	$4, 5s \ [1\frac{1}{2}]^\circ$	$4p \ [\frac{1}{2}]$ $4p \ [2\frac{1}{2}]$ $4p \ [1\frac{1}{2}]$	$3, 4d \ [\frac{1}{2}]^\circ$ $3, 4d \ [3\frac{1}{2}]^\circ$ $3, 4d \ [1\frac{1}{2}]^\circ$ $3, 4d \ [2\frac{1}{2}]^\circ$
$3p^5(^2P^\circ_{1/2})nx'$	$4, 5s' [\frac{1}{2}]^\circ$	$4p' [1\frac{1}{2}]$ $4p' [\frac{1}{2}]$	$4d' [1\frac{1}{2}]^\circ$

*For predicted levels in the spectra of the A I isoelectronic sequence, see Introduction.

K III

(Cl I sequence; 17 electrons)

$Z=19$

Ground state $1s^2 2s^2 2p^6 3s^2 3p^5 \ ^2P^\circ_{1/2}$

$3p^5 \ ^2P^\circ_{1/2} \ 369000 \text{ cm}^{-1}$

I. P. 46 volts

The analyses by various investigators are discordant, but nearly 80 lines have been classified in the range between 325 Å and 3885 Å.

From observed intersystem combinations Edlén has derived a correction of $+667.7 \text{ cm}^{-1}$ to the absolute values of the doublet terms given by de Bruin, to connect them with the quartet terms. Edlén also states that the limit derived by extrapolation along the isoelectronic sequence is 369000 cm^{-1} . This limit (entered in brackets in the table), indicates a correction of about -8000 cm^{-1} to the limit listed by de Bruin, 377000 cm^{-1} .

The doublet terms as given by Edlén and the quartet terms from de Bruin have been used in compiling the present list. The additional terms are from Tsien.

Kruger and Phillips designate as $4s'' \ ^2S_{1/2}$ the level at 246012 cm^{-1} , given by Tsien as $3d' \ ^2D_{1/2}$. Further study is needed to confirm the terms from the higher limits.

REFERENCES

- T. L. de Bruin, *Zeit. Phys.* **53**, 658 (1929). (IP) (T) (C L)
 B. Edlén, *Zeit. Phys.* **104**, 410 (1937). (I P) (T) (C L)
 P. G. Kruger and L. W. Phillips, *Phys. Rev.* **51**, 1087 (1937). (T) (C L)
 W.-Z. Tsien, *Chinese J. Phys.* **3**, No. 2, 118 (1939). (T) (C L)

K III

K III

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2 3p^5$	$3p^5 \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	$\begin{matrix} 0 \\ 2162 \end{matrix}$	-2162	$3s^2 3p^4(^1S)4s$	$4s'' \ ^2S$	$\frac{1}{2}$	241667	
$3p^6$	$3p^6 \ ^2S$	$\frac{1}{2}$	130609		$3s^2 3p^4(^3P)4p$	$4p \ ^2D^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$	$\begin{matrix} 243120.6 \\ 243448.2 \end{matrix}$	-327.6
$3s^2 3p^4(^3P)3d$	$3d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	$\begin{matrix} 190916 \\ 192082 \end{matrix}$	1166	$3s^2 3p^4(^3P)4p$	$4p \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	$\begin{matrix} 243947.4 \\ 245382.3 \end{matrix}$	-1434.9
$3s^2 3p^4(^3P)3d$	$3d \ ^2F$	$\left\{ \begin{matrix} 3\frac{1}{2} \\ 2\frac{1}{2} \end{matrix} \right\}$	201165		$3s^2 3p^4(^1D)3d$	$3d' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	$\begin{matrix} 244523 \\ 246012 \end{matrix}$	-1489
$3s^2 3p^4(^3P)4s$	$4s \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	$\begin{matrix} 207421.9 \\ 208687.8 \\ 209461.3 \end{matrix}$	$\begin{matrix} -1265.9 \\ -773.5 \end{matrix}$	$3s^2 3p^4(^3P)4p$	$4p \ ^4S^\circ$	$1\frac{1}{2}$	246625.6	
$3s^2 3p^4(^3P)4s$	$4s \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	$\begin{matrix} 212725.4 \\ 214232.3 \end{matrix}$	-1506.9	$3s^2 3p^4(^1D)3d$	$3d' \ ^2S$	$\frac{1}{2}$	250857	
$3s^2 3p^4(^1D)4s$	$4s' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	$\begin{matrix} 225051 \\ 225082 \end{matrix}$	-31		1		252040	
$3s^2 3p^4(^3P)4p$	$4p \ ^4P^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	$\begin{matrix} 237512.0 \\ 237912.2 \\ 238455.1 \end{matrix}$	$\begin{matrix} -400.2 \\ -542.9 \end{matrix}$	$3s^2 3p^4(^3P)5s$	$5s \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	$\begin{matrix} 262828 \\ 263770 \end{matrix}$	-942
$3s^2 3p^4(^3P)4p$	$4p \ ^4D^\circ$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	$\begin{matrix} 240829.9 \\ 241443.5 \\ 242165.3 \\ 242526.7 \end{matrix}$	$\begin{matrix} -613.6 \\ -721.3 \\ -361.4 \end{matrix}$	$3s^2 3p^4(^1D)5s$	$5s' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	$\begin{matrix} 289400 \\ 289515 \end{matrix}$	-115
$3s^2 3p^4(^1D)3d$	$3d' \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	$\begin{matrix} 241039 \\ 242548 \end{matrix}$	-1509	$3s^2 3p^4(^1S)3d$	$3d'' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	$\begin{matrix} 302404 \\ 303902 \end{matrix}$	-1498
						2		307429	
					K IV (3P_2)	Limit	-----	[369000]	

January 1948.

K III OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6$	Observed Terms						
$3s^2 3p^5$	$3p^5 \ ^2P^\circ$						
$3s \ 3p^6$	$3p^6 \ ^2S$						
	$ns \ (n \geq 4)$		$np \ (n \geq 4)$		$nd \ (n \geq 3)$		
$3s^2 3p^4(^3P)nx$	{ $4s \ ^4P$ $4, 5s \ ^2P$	$4p \ ^4S^\circ$	$1p \ ^4P^\circ$	$4p \ ^4D^\circ$			
$3s^2 3p^4(^1D)nx'$			$4p \ ^2P^\circ$	$4p \ ^2D^\circ$			
$3s^2 3p^4(^1S)nx''$	$4s'' \ ^2S$		$4, 5s' \ ^2D$		$3d' \ ^2S$	$3d' \ ^2P$	
					$3d' \ ^2D$	$3d'' \ ^2D$	

*For predicted terms in the spectra of the Cl I isoelectronic sequence, see Introduction.

K IV

(S r sequence; 16 electrons)

Z=19

Ground state $1s^2 2s^2 2p^6 3s^2 3p^4 {}^3P_2$ $3p^4 {}^3P_2$ 491300 cm^{-1}

I. P. 60.90 volts

The terms are from the papers by Bowen and by Tsien, with the revised values of $3p^4 {}^1S$ and $3p^5 {}^1P^\circ$ suggested by Edlén, and of $4s {}^3S^\circ$ by Mrs. Beckman. Colons have been added by the writer to some levels that appear to need further confirmation.

Nearly 60 lines have been classified in the region between 271 Å and 754 Å. Intersystem combinations connecting the singlet and triplet terms have been observed.

The limit is from Edlén's 1937 paper. He has derived it by extrapolation of isoelectronic sequence data.

REFERENCES

- M. Ram, Indian J. Phys. **8**, 155 (1933). (T) (C L)
 I. S. Bowen, Phys. Rev. **46**, 791 (1934). (T) (C L)
 B. Edlén, Zeit. Phys. **104**, 192 (1937). (I P)
 A. Beckman, *Bidrag till Kännedomen om Skandiums Spektrum i Yttersta Ultraviolett*, Akademisk Avhandling p. 79 (Almqvist and Wiksells Boktryckeri-A.-B., Uppsala, 1937). (C L)
 W.-Z. Tsien, Chinese J. Phys. **3**, No. 2, 131 (1939). (T) (C L)
 B. Edlén, Phys. Rev. **62**, 434 (1942). (T) (C L)

K IV

K IV

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2 3p^4$	$3p^4 {}^3P$	2 1 0	0 1673 2324	1673 651	$3s^2 3p^3({}^4S^\circ)4s$	$4s {}^3S^\circ$	1	260352	
					$3s^2 3p^3({}^2P^\circ)3d$	$3d'' {}^1P^\circ$	1	261445	
$3s^2 3p^4$	$3p^4 {}^1D$	2	16386		$3s^2 3p^3({}^2P^\circ)3d$	$3d'' {}^3D^\circ$	3 2 1	262831 263659	-828
$3s^2 3p^4$	$3p^4 {}^1S$	0	38548						
$3s 3p^5$	$3p^5 {}^3P^\circ$	2 1 0	134181 135659 136453	-1478 -794	$3s^2 3p^3({}^2P^\circ)3d$	$3d'' {}^1D^\circ$	2	273409	
					$3s^2 3p^3({}^2D^\circ)4s$	$4s' {}^3D^\circ$	1 2 3	277795 277851 277986	56 135
$3s 3p^5$	$3p^5 {}^1P^\circ$	1	171140						
$3s^2 3p^3({}^4S^\circ)3d$	$3d {}^3D^\circ$	3 2 1	189952 191204 191403	-252 -199	$3s^2 3p^3({}^2D^\circ)4s$	$4s' {}^1D^\circ$	2	282373	
					$3s^2 3p^3({}^2P^\circ)4s$	$4s'' {}^3P^\circ$	0 1 2	293384 293473 293720	89 247
$3s^2 3p^3({}^2D^\circ)3d$	$3d' {}^1F^\circ$	3	222420						
$3s^2 3p^3({}^2D^\circ)3d$	$3d' {}^3P^\circ$	2 1 0	225445 226090 227652	-645 -1562	$3s^2 3p^3({}^2P^\circ)4s$	$4s'' {}^1P^\circ$	1	298134	
					$3s^2 3p^3({}^4S^\circ)5s$	$5s {}^3S^\circ$	1	367890	
$3s^2 3p^3({}^2D^\circ)3d$	$3d' {}^1P^\circ$	1	235527:						
$3s^2 3p^3({}^3P^\circ)3d$	$3d'' {}^3P^\circ$	2 1 0	256034 257124 257811:	-1090 -687	K V (${}^4S_{3/2}$)	Limit	-----	491300	

December 1947.

K IV OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ +	Observed Terms	
3s ² 3p ⁴	{ 3p ⁴ ¹ S 3p ⁴ ³ P 3p ⁴ ¹ D	
3s 3p ⁵	{ 3p ⁵ ³ P ^o 3p ⁵ ¹ P ^o	
	ns (n ≥ 4)	nd (n ≥ 3)
3s ² 3p ³ (⁴ S ^o)nx	4, 5s ³ S ^o	3d ³ D ^o
3s ² 3p ³ (² D ^o)nx'	{ 4s' ³ D ^o 4s' ¹ D ^o	3d' ³ P ^o 3d' ¹ P ^o 3d' ¹ F ^o
3s ² 3p ³ (² P ^o)nx''	{ 4s'' ³ P ^o 4s'' ¹ P ^o	3d'' ³ P ^o 3d'' ³ D ^o 3d'' ¹ P ^o 3d'' ¹ D ^o

*For predicted terms in the spectra of the Si isoelectronic sequence, see Introduction.

K v

(P I sequence; 15 electrons) Z=19

Ground state 1s² 2s² 2p⁶ 3s² 3p³ ⁴S_{1/2}^o

3p³ ⁴S_{1/2}^o cm⁻¹ I. P. volts

The analysis is incomplete. The terms are from the paper by Tsien, who includes those given earlier by Bowen. Seventy-two lines have been classified in the interval between 294 Å and 825 Å.

The relative position of the doublet terms with respect to the quartet terms was estimated from the irregular doublet law. Tsien lists combinations of 3p³ ⁴S^o and 3p³ ²P^o with the level labeled "3", which are not in disagreement with this estimate.

REFERENCES

I. S. Bowen, Phys. Rev. **46**, 791 (1934). (T) (C L)
W.—Z. Tsien, Chinese J. Phys. **3**, No. 2, 136 (1939). (T) (C L)

K v

K v

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2 3p^3$	$3p^3 \ ^4S^\circ$	$1\frac{1}{2}$	0		$3s^2 3p^2(^3P)3d$	$3d \ ^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	262487 262874	387
$3s^2 3p^3$	$3p^3 \ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	24000 24237	237	$3s^2 3p^2(^3P)3d$	$3d \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	264741 264932	-191
$3s^2 3p^3$	$3p^3 \ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	39745 40064	319		3		268043	
$3s 3p^4$	$3p^4 \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	136639 138042 138806	-1403 -764		4		274375	
$3s 3p^4$	$3p^4 \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	161199 161564	365	$3s^2 3p^2(^1D)3d$	$3d' \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	281024	
	1		169703		$3s^2 3p^2(^1D)3d$	$3d' \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	290772	
	2		169886		$3s^2 3p^2(^1D)3d$	$3d' \ ^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	292710	
$3s 3p^4$	$3p^4 \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	194792 196319	-1527	$3s^2 3p^2(^1D)3d$	$3d' \ ^2S$	$1\frac{1}{2}$	292968	
$3s 3p^4$	$3p^4 \ ^2S$	$\frac{1}{2}$	205784		$3s^2 3p^2(^1S)3d$	$3d'' \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	304461 305978	1517
$3s^2 3p^2(^3P)3d$	$3d \ ^4F$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	206720 207165	445		5		307717	
						6		310120	
$3s^2 3p^2(^3P)3d$	$3d \ ^4D$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	222367 222711	-344	$3s^2 3p^2(^3P)4s$	$4s \ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	336628 337645 339172	1017 1527
$3s^2 3p^2(^3P)3d$	$3d \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	257865 259276 259726	-1411 -450	$3s^2 3p^2(^3P)4s$	$4s \ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	343726 345526	1800
$3s^2 3p^2(^3P)3d$	$3d \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	259205 260868	-1663	$3s^2 3p^2(^1D)4s$	$4s' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	356993 357033	-40

November 1947.

K v OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 +$	Observed Terms			
$3s^2 3p^3$	{ $3p^3 \ ^4S^\circ$ $3p^3 \ ^2P^\circ$ $3p^3 \ ^2D^\circ$			
$3s 3p^4$	{ $3p^4 \ ^2S$ $3p^4 \ ^4P$ $3p^4 \ ^2P$ $3p^4 \ ^2D$			
	<i>ns</i> ($n \geq 4$)		<i>nd</i> ($n \geq 3$)	
$3d^2 3p^2(^3P)nx$	{ $4s \ ^4P$ $4s \ ^2P$		$3d \ ^4P$ $3d \ ^4D$ $3d \ ^4F$ $3d \ ^2P$ $3d \ ^2D$ $3d \ ^2F$	
$3s^2 3p^2(^1D)nx'$	$4s' \ ^2D$		$3d' \ ^2S$ $3d' \ ^2P$ $3d' \ ^2D$ $3d' \ ^2F$	
$3s^2 3p^2(^1S)nx''$			$3d'' \ ^2D$	

*For predicted terms in the P I isoelectronic sequence, see Introduction.

K VI

(Si I sequence; 14 electrons)

Z=19

Ground state $1s^2 2s^2 2p^6 3s^2 3p^2 \ ^3P_0$

$3p^2 \ ^3P_0$ 804513 cm^{-1}

I. P. 99.7 volts

The analysis is chiefly by Whitford, with singlet terms added from Robinson's paper. Twenty-seven lines have been classified in the interval between 256 Å and 725 Å. Inter-system combinations connecting the singlet and triplet terms have been observed.

Using the method suggested by Edlén for extrapolation along the isoelectronic sequence, the writer has estimated the value of the limit quoted above and entered in brackets in the table.

REFERENCES

A. E. Whitford, Phys. Rev. **46**, 793 (1934). (T) (C L)
H. A. Robinson, Phys. Rev. **52**, 725 (1937). (T) (C L)

K VI					K VI				
Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2\ 3p^2$	$3p^2\ ^3P$	0	0	1131 1793	$3s\ 3p^3$	$3p^3\ ^1P^\circ$	1	223840	—1172 —539
		1	1131		$3s^2\ 3p(^2P^\circ)3d$	$3d\ ^3P^\circ$	2	252332	
		2	2924				1	253504	
$3s^2\ 3p^2$	$3p^2\ ^1D$	2	18973				0	254043	
$3s\ 3p^3$	$3p^3\ ^3D^\circ$	1	140743	53 170	$3s^2\ 3p(^2P^\circ)4s$	$4s\ ^3P^\circ$	0	387421	
		2	140796				1	388114	
		3	140966				2	390493	
$3s\ 3p^3$	$3p^3\ ^3P^\circ$	2, 1, 0	163434	-----					
$3s\ 3p^3$	$3p^3\ ^3S^\circ$	1	218316	K VII ($^2P^\circ_{3/2}$)		Limit	-----	[804513]	

October 1947.

K VII

(Al I sequence; 13 electrons)

Z=19

Ground state $1s^2 2s^2 2p^6 3s^2 3p^2 P_{3/2}^{\circ}$ $3p^2 P_{3/2}^{\circ}$ 950200 cm^{-1}

I. P. 118 volts

Both Whitford and Phillips have worked on the analysis of this spectrum. Thirty lines have been classified in the interval between 175 Å and 671 Å. No intersystem combinations have been observed, but Phillips estimates that $3p^2 {}^4P_{3/2}$ is approximately 114000cm^{-1} above the ground state. This value is entered in brackets in the table. The uncertainty x may exceed $\pm 1000\text{cm}^{-1}$.

Using the method suggested by Edlén, the writer has extrapolated the value of the limit quoted above and entered in brackets in the table. The uncertainty in this estimate is large owing to the incompleteness of the isoelectronic sequence data.

REFERENCES

A. E. Whitford, Phys. Rev. **46**, 793 (1934). (T) (C L)L. W. Phillips, Phys. Rev. **55**, 708 (1939). (T) (C L)

K VII

K VII

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2(1S)3p$	$3p^2 P^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$	0 3129	3129	$3s 3p({}^3P^{\circ})3d$	$3d {}^4D^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	$365092+x$ $365463+x$ $365778+x$ $365916+x$	371 315 138
$3s 3p^2$	$3p^2 {}^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$[114000]+x$ $115145+x$ $116871+x$	1145 1726	$3s^2(1S)4s$	$4s {}^2S$	$\frac{1}{2}$	439297	
$3s 3p^2$	$3p^2 {}^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	151882 152049	167	$3s 3p({}^3P^{\circ})4s$	$4s {}^4P^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$565314+x$ $566443+x$ $568375+x$	1129 1932
$3s 3p^2$	$3p^2 {}^2S$	$\frac{1}{2}$	193079		$3s^2(1S)4d$	$4d {}^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	570812 570969	157
$3s 3p^2$	$3p^2 {}^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	206507 208434	1927					
$3s^2(1S)3d$	$3d {}^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	250668 250787	119					
$3p^3$	$3p^3 {}^4S^{\circ}$	$1\frac{1}{2}$	$307479+x$		K VIII ($1S_0$)	Limit	-----	[950200]	

September 1947.

K VII OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 +$	Observed Terms	
$3s^2(1S)3p$	$3p^2 P^{\circ}$	
$3s 3p^2$	$\left\{ \begin{array}{l} 3p^2 {}^2S \\ 3p^2 {}^4P \\ 3p^2 {}^2P \end{array} \right. \quad 3p^2 {}^2D$	
$3p^3$	$3p^3 {}^4S^{\circ}$	
	$ns (n \geq 4)$	$nd (n \geq 3)$
$3s^2(1S)nx$	$4s {}^2S$	$3, 4d {}^2D$
$3s 3p({}^3P^{\circ})nx$	$4s {}^4P^{\circ}$	$3d {}^4D^{\circ}$

*For predicted terms in the spectra of the Al I isoelectronic sequence, see Introduction.

K VIII

(Mg I sequence; 12 electrons)

Z=19

Ground state $1s^2 2s^2 2p^6 3s^2 \ ^1S_0$

$3s^2 \ ^1S_0$ 1247000± cm⁻¹

I. P. 155± volts

Twenty-six lines have been classified in the range between 155 Å and 938 Å. The triplet terms are from Parker and Phillips; the singlets from Tsien. By extrapolation along the sequence Mrs. Beckman has classified a line at 774.738 Å as the intersystem combination $3s^2 \ ^1S_0-3p \ ^3P_1^\circ$. The listed values of the triplet terms have been adjusted to fit this assignment.

From isoelectronic sequence data the writer has extrapolated the value of the limit, using the method suggested by Edlén. This value is entered in brackets in the table. Although this estimate may be in error by more than ±1000 cm⁻¹, it gives an approximate value of the ionization potential.

REFERENCES

A. Beckman, *Bidrag till K nnedomen om Skandiums Spektrum i Yttersta Ultraviolet*, Akademisk Avhandling p. 55 (Almqvist and Wiksells Boktryckeri-A.-B., Uppsala, 1937). (C L)
W.-Z. Tsien, *Chinese J. Phys.* **3**, No. 2, 142 (1939). (T) (C L)
W. L. Parker and L. W. Phillips, *Phys. Rev.* **57**, 140 (1940). (T) (C L)

K VIII

K VIII

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3s^2$	$3s^2 \ ^1S$	0	0		$3s(^2S)3d$	$3d \ ^3D$	1 2 3	368004 368060 368132	56 72
$3s(^2S)3p$	$3p \ ^3P^\circ$	0 1 2	127968 129080 131452	1112 2372	$3s(^2S)4s$	$4s \ ^3S$	1	631654	
$3s(^2S)3p$	$3p \ ^1P^\circ$	1	192540. 2		$3s(^2S)4d$	$4d \ ^3D$	1 2 3	770165 770260 770401	95 141
$3s(^2S)3d$	$3d \ ^1D$	2	299117. 4		$3s(^2S)4f$	$4f \ ^3F^\circ$	2, 3, 4	801511	
$3p^2$	$3p^2 \ ^3P$	0 1 2	304669 306035 308608	1366 2573					
					K IX ($^2S_{1/2}$)	Limit	-----	[1247000±]	

March 1948.

K IX

(Na I sequence; 11 electrons)

 $Z=19$ Ground state $1s^2 2s^2 2p^6 3s^2 S_{1/2}$ $3s^2 S_{1/2}$ 1419425 cm^{-1}

I. P. 175.94 volts

All but two of the terms are from the paper by Kruger and Phillips, who extended the earlier work by Edlén and Whitford. Absolute term values are based on three members of the 2D -series.

The two terms $5s^2 S$ and $5g^2 G$ have been added from the paper by Tsien, but adjusted to agree with the term array by Kruger and Phillips.

Twenty-five lines have been classified, in the range from 112 Å to 636 Å.

REFERENCES

W.-Z. Tsien, Chinese J. Phys. **3**, No. 2, 145 (1939). (T) (C L)

P. G. Kruger and L. W. Phillips, Phys. Rev. **55**, 352 (1939). (I P) (T) (C L)

K IX

K IX

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
3s	3s 2S	$\frac{1}{2}$	0		5s	5s 2S	$\frac{1}{2}$	979901	
3p	3p $^2P^\circ$	$\frac{1}{2}$ $\frac{3}{2}$	157159 160925	3766	5g	5g 2G	$\frac{4}{2}$ $\frac{3}{2}$	1044250 1044298	-48
3d	3d 2D	$\frac{1}{2}$ $\frac{3}{2}$	374788 375080	292	5d	5d 2D	$\frac{1}{2}$ $\frac{3}{2}$	1049114 1049174	60
4s	4s 2S	$\frac{1}{2}$	698902		5f	5f $^2F^\circ$	$\frac{2}{2}$ $\frac{3}{2}$	1061120 1061172	52
4p	4p $^2P^\circ$	$\frac{1}{2}$ $\frac{3}{2}$	758174 759615	1441					
4d	4d 2D	$\frac{1}{2}$ $\frac{3}{2}$	836703 836861	158	K x (1S_0)	Limit		1419425	
4f	4f $^2F^\circ$	$\frac{2}{2}$ $\frac{3}{2}$	860763 860842	79					

June 1947.

K X

(Ne I sequence; 10 electrons)

 $Z=19$ Ground state $1s^2 2s^2 2p^6 ^1S_0$ $2p^6 ^1S_0$ 4064300 cm^{-1}

I. P. 503.8 volts

Eleven lines between 29 Å and 41 Å have been classified by Edlén and Tyrén as combinations with the ground term. Their absolute term values have been extrapolated along the Ne I isoelectronic sequence.

By analogy with Ne I, $j\bar{l}$ -coupling notation in the general form suggested by Racah is introduced.

The unit adopted by Edlén and Tyrén, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

B. Edlén and F. Tyrén, Zeit. Phys. **101**, 206 (1936). (I P) (T) (C L)

G. Racah, Phys. Rev. **61**, 537 (L) (1942).

K x					K x				
Authors	Config.	Desig.	<i>J</i>	Level	Authors	Config.	Desig.	<i>J</i>	Level
2 <i>p</i> ¹ S ₀	2 <i>s</i> ² 2 <i>p</i> ⁶	2 <i>p</i> ⁶ ¹ S	0	0	3 <i>p</i> ' ³ P ₁	2 <i>s</i> 2 <i>p</i> ⁶ (² S)3 <i>p</i>	3 <i>p</i> ³ P ^o	$\frac{2}{1}$ 0	3219400
3 <i>s</i> ³ P ₁	2 <i>s</i> ² 2 <i>p</i> ⁵ (² P _{1/2})3 <i>s</i>	3 <i>s</i> [1½] ^o	$\frac{2}{1}$	2407300	3 <i>p</i> ' ¹ P ₁	2 <i>s</i> 2 <i>p</i> ⁶ (² S)3 <i>p</i>	3 <i>p</i> ¹ P ^o	1	3237600
3 <i>s</i> ¹ P ₁	2 <i>s</i> ² 2 <i>p</i> ⁵ (² P _{3/2} ^o)3 <i>s</i>	3 <i>s</i> ' [½] ^o	0 1	2430300	4 <i>d</i> ¹ P ₁	2 <i>s</i> ² 2 <i>p</i> ⁵ (² P _{1/2})4 <i>d</i>	4 <i>d</i> [1½] ^o	1	3356400
3 <i>d</i> ³ P ₁	2 <i>s</i> ² 2 <i>p</i> ⁵ (² P _{1/2})3 <i>d</i>	3 <i>d</i> [½] ^o	0 1	2760200	4 <i>d</i> ³ D ₁	2 <i>s</i> ² 2 <i>p</i> ⁵ (² P _{3/2} ^o)4 <i>d</i>	4 <i>d</i> ' [1½] ^o	1	3379700
3 <i>d</i> ¹ P ₁	"	3 <i>d</i> [1½] ^o	1	2794900		K xI (² P _{1/2})	Limit	-----	4064300
3 <i>d</i> ³ D ₁	2 <i>s</i> ² 2 <i>p</i> ⁵ (² P _{3/2} ^o)3 <i>d</i>	3 <i>d</i> ' [1½] ^o	1	2832300		K xI (² P _{3/2} ^o)	Limit	-----	4087775
4 <i>s</i> ³ P ₁	2 <i>s</i> ² 2 <i>p</i> ⁵ (² P _{1/2})4 <i>s</i>	4 <i>s</i> [1½] ^o	2 1	3205100					
4 <i>s</i> ¹ P ₁	2 <i>s</i> ² 2 <i>p</i> ⁵ (² P _{3/2} ^o)4 <i>s</i>	4 <i>s</i> ' [½] ^o	0 1	3232400					

April 1947.

K x OBSERVED LEVELS*

Config. 1 <i>s</i> ² +	Observed Terms		
2 <i>s</i> ² 2 <i>p</i> ⁶	2 <i>p</i> ⁶ ¹ S		
	<i>ns</i> (<i>n</i> ≥ 3)	<i>np</i> (<i>n</i> ≥ 3)	<i>nd</i> (<i>n</i> ≥ 3)
2 <i>s</i> ² 2 <i>p</i> ⁵ (² P ^o) <i>nx</i>	{ 3, 4 <i>s</i> ³ P ^o 3, 4 <i>s</i> ¹ P ^o		3 <i>d</i> ³ P ^o 3, 4 <i>d</i> ³ D ^o 3, 4 <i>d</i> ¹ P ^o
2 <i>s</i> 2 <i>p</i> ⁶ (² S) <i>nx</i>	{	3 <i>p</i> ³ P ^o 3 <i>p</i> ¹ P ^o	
<i>jl</i> -Coupling Notation			
	Observed Pairs		
	<i>ns</i> (<i>n</i> ≥ 3)		<i>nd</i> (<i>n</i> ≥ 3)
2 <i>s</i> ² 2 <i>p</i> ⁵ (² P _{1/2}) <i>nx</i>	3, 4 <i>s</i> [1½] ^o		3 <i>d</i> [½] ^o 3, 4 <i>d</i> [1½] ^o
2 <i>s</i> ² 2 <i>p</i> ⁵ (² P _{3/2} ^o) <i>nx</i> '	3, 4 <i>s</i> ' [½] ^o		3, 4 <i>d</i> ' [1½] ^o

*For predicted levels in the spectra of the Ne I isoelectronic sequence, see Introduction.

K XI

(F I sequence; 9 electrons)

Z=19

Ground state $1s^2 2s^2 2p^5 \ ^2P^\circ_{1/2}$

$2p^5 \ ^2P^\circ_{1/2}$ cm⁻¹ I. P. volts

Edlén and Tyrén have classified 8 lines, which lie between 32 Å and 37 Å. They give no term array because the analysis is so incomplete. In the 1942 reference Edlén states that the interval of the ground term is known from his unpublished material to be 23475 cm⁻¹. From these data, preliminary term values have been calculated and listed below.

REFERENCES

B. Edlén and F. Tyrén, Zeit. Phys. **101**, 206 (1936). (C L)
B. Edlén, Zeit. Astroph. **22**, 59 (1942). (T)

K XI

Edlén	Config.	Desig.	J	Level	Inter- val
$2p \ ^2P_2$ $\ ^2P_1$	$2s^2 2p^5$	$2p^5 \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	$\overset{0}{23475}$	-23475
$3s \ ^4P_3$ $\ ^4P_2$	$2s^2 2p^4(^3P)3s$	$3s \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	2640600? 2652800?	-12200
$3s \ ^2P_2$	$2s^2 2p^4(^3P)3s$	$3s \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	2671300?	
$3s' \ ^2D_3$ $\ ^2D_2$	$2s^2 2p^4(^1D)3s$	$3s' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	2727600? 2728300?	-700
$3d$	$2s^2 2p^4(^3P)3d$	$3d \ X$		3047900?	
$\overline{3d}$	$2s^2 2p^4(^1D)3d$	$3d' \ X$		3107500?	

March 1947.

CALCIUM

Ca I

20 electrons

 $Z=20$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 {}^1S_0$ $4s^2 {}^1S_0$ 49304.80 cm^{-1}

I. P. 6.111 volts

The arc spectrum of calcium occupies an important place in the development of spectroscopic theory. In addition to the "regular" series, the terms involving two excited electrons were first discussed in the classical paper by Russell and Saunders in 1925.

Although the spectrum is well known, further observations in the infrared are urgently needed; and a monograph containing a homogeneous list of lines and term values should be prepared as soon as the analysis can be extended with the aid of these data.

The regular series terms, i. e., those from the 2S limit in Ca II, are from Fowler and Paschen-Götze. The rest are from Russell and Saunders and from unpublished analysis by Russell, who has generously furnished all of his data on this spectrum. The $6f {}^3F^o$ term has been resolved by Grafenberger. Three-place entries in the table are quoted from Wagman, who derived them from observations made with the interferometer. The writer has prepared a complete multiplet array and calculated all other values from the best available wavelength material. Colons indicate that the term values should be confirmed by further observations.

The singlet and triplet terms are connected by observed intersystem combinations.

REFERENCES

- F. Paschen und R. Götze, *Seriengesetze der Linienspektren*, p. 72 (Julius Springer, Berlin, 1922). (I P) (T) (C L)
 A. Fowler, *Report on Series in Line Spectra*, p. 121 (Fleetway Press, London, 1922). (I P) (T) (C L)
 H. N. Russell and F. A. Saunders, *Astroph. J.* **61**, 38 (1925). (I P) (T) (C L)
 E. Back, *Zeit. Phys.* **33**, 579 (1925). (Z E)
 H. N. Russell, unpublished material (1927?). (T) (C L)
 A. G. Shenstone and H. N. Russell, *Phys. Rev.* **39**, 417 (1932). (T)
 W. F. Meggers, *Bur. Std. J. Research* **10**, 676, RP558 (1933). (I P) (T) (C L)
 N. E. Wagman, *Univ. Pittsburgh Bul.* **34**, 1 (1937). (T) (C L)
 H. Grafenberger, *Ann. der Phys.* [5] **30**, 267 (1937). (C L)

Ca I

Ca I

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
4s ²	4s ² 1S	0	0. 000			4s(2S)4f	4f 1F°	3	42343. 554		
4s(2S)4p	4p 3P°	0	15157. 910			4s(2S)6p	6p 3P°	0	42514. 79:		
		1	15210. 067	52. 157				1	42518. 72	3. 93	
		2	15315. 948	105. 881				2	42526. 528	7. 81	
4s(2S)3d	3d 3D	1	20335. 344		0. 501	4s(2S)5d	5d 3D	1	42743. 058		
		2	20349. 247	13. 903	1. 162			2	42744. 776	1. 718	
		3	20370. 987	21. 740	1. 329			3	42747. 443	2. 667	
4s(2S)3d	3d 1D	2	21849. 610		1. 007	4s(2S)5d	5d 1D	2	42919. 074		
4s(2S)4p	4p 1P°	1	23652. 324			4s(2S)6p	6p 1P°	1	43933. 341		
4s(2S)5s	5s 3S	1	31539. 510			4s(2S)7s	7s 3S	1	43980. 798		
4s(2S)5s	5s 1S	0	33317. 25			4s(2S)7s	7s 1S	0	44276. 638		
3d(2D)4p	4p' 3F°	2	35730. 450		0. 754	4s(2S)5f	5f 3F°	2	44762. 620		
		3	35818. 712	88. 262	1. 076			3	44762. 822	0. 202	
		4	35896. 890	78. 178	1. 245			4	44763. 101	0. 279	
3d(2D)4p	4p' 1D°	2	35835. 400		0. 893	4s(2S)5f	5f 1F°	3	44804. 786		
4s(2S)5p	5p 3P°	0	36547. 671			4s(2S)7p	7p 3P°	0			
		1	36554. 722	7. 051				1	44957. 8		
		2	36575. 132	20. 410				2	44961. 6	3. 8	
3d(2D)4p	4p' 1P°	1	36731. 622			4s(2S)6d	6d 1D	2	44989. 882		
4s(2S)4d	4d 1D	2	37298. 312			4s(2S)6d	6d 3D	1	45049. 066		
4s(2S)4d	4d 3D	1	37748. 192					2	45050. 406	1. 340	
		2	37751. 884	3. 682				3	45052. 359	1. 953	
		3	37757. 462	5. 578		4s(2S)7p	7p 1P°	1	45425. 283		
3d(2D)4p	4p' 3D°	1	38192. 373			4s(2S)8s	8s 3S	1	45738. 732		
		2	38219. 094	26. 721		4s(2S)8s	8s 1S	0	45887. 31		
		3	38259. 102	40. 008		4s(2S)6f	6f 3F°	2	46164. 66		
4p ²	4p ² 3P	0	38417. 585					3	46164. 80	0. 14	
		1	38464. 844	47. 259				4	46164. 99	0. 19	
		2	38551. 588	86. 744		4s(2S)6f	6f 1F°	3	46182. 23		
3d(2D)4p	4p' 3P°	0	39333. 371			4s(2S)7d	7d 3D	1	46302. 18		
		1	39335. 316	1. 945				2	46303. 92	1. 74	
		2	39340. 078	4. 762				3	46306. 170	2. 25	
4s(2S)6s	6s 3S	1	40474. 275			4s(2S)7d	7d 1D	2	46309. 9		
3d(2D)4p	4p' 1F°	3	40537. 860			4s(2S)8p	8p 1P°	1	46479. 95		
4p ²	4p ² 1S	0	40690. 436			4s(2S)9s	9s 3S	1	46748. 21		
4p ²	4p ² 1D	2	40719. 867			4s(2S)9s	9s 1S	0	46835. 2		
4s(2S)5p	5p 1P°	1	41678. 997			4s(2S)7f	7f 3F°	2, 3, 4	47006. 11		
4s(2S)6s	6s 1S	0	41786. 312			4s(2S)7f	7f 1F°	3	47015. 137		
4s(2S)4f	4f 3F°	2	42170. 183								
		3	42170. 531	0. 348							
		4	42171. 006	0. 475							

Ca I—Continued

Ca I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
4s(2S)8d	8d 3D	1 2 3	47036. 32 47040. 00 47045. 384	3. 68 5. 38		4s(2S)12f	12f 3F°	2, 3, 4	48531. 4		
4s(2S)9p	9p 1P°	1	47184. 26			4s(2S)13d	13d 3D	1, 2, 3	48570. 7		
4s(2S)10s	10s 3S	1	47382. 10			4s(2S)13f	13f 3F°	2, 3, 4	48647. 1		
4s(2S)10s	10s 1S	0	47436. 9			4s(2S)14d	14d 3D	1, 2, 3	48676. 6		
3d(2D)5s	5s' 3D	1 2 3	47456. 1 47465. 9 47475. 7	9. 8 9. 8		4s(2S)15d	15d 3D	1, 2, 3	48762. 4		
4s(2S)8f	8f 3F°	2, 3, 4	47550. 11			4s(2S)16d	16d 3D	1, 2, 3	48830. 7		
4s(2S)8f	8f 1F°	3	47554. 97			Ca II (2S _{1/2})	Limit	-----	49304. 80		
4s(2S)10p	10p 1P°	1	47660. 8			3d(2D)5p	5p' 3F°	2 3 4	51235. 2: 51259. 5: 51318. 7:	24. 3 59. 2	
4s(2S)9d	9d 3D	1 2 3	47753. 3 47757. 5 47765. 5	4. 2 8. 0		3d(2D)4d	4d' 3D	1 2 3	51351. 1 51369. 6 51395. 5	18. 5 25. 9	
4s(2S)11s	11s 3S	1	47805. 85			3d(2D)4d	4d' 3G	3 4 5	51553. 6: 51579. 0: 51611. 5:	25. 4 31. 5	
4s(2S)11s	11s 1S	0	47843. 1			3d(2D)4d	4d' 3S	1	51571. 4		
4s(2S)9f	9f 3F°	2, 3, 4	47922. 2			3d(2D)5p	5p' 3D°	1 2 3	51710. 9 51734. 0 51766. 5	23. 1 32. 5	
4s(2S)9f	9f 1F°	3	47924. 9			3d(2D)4d	4d' 3F	2 3 4	53214. 6 53247. 9 53260. 4	33. 3 12. 5	
4s(2S)11p	11p 1P°	1	47998. 6			3d(2D)4d	4d' 3P	0 1 2	54282. 2 54288. 0 54304. 2	5. 8 16. 2	
4s(2S)10d	10d 3D	1 2 3	48032. 0 48033. 5 48036. 2	1. 5 2. 7		3d(2D)5d	5d' 3D	1 2 3	56444. 8 56469. 1 56494. 7	24. 3 25. 6	
4s(2S)12s	12s 3S	1	48103. 89			3d(2D)5d	5d' 3G	3 4 5	56526. 3: 56546. 6: 56578. 2:	20. 3 31. 6	
4s(2S)12s	12s 1S	0	48128. 2			3d(2D)5d	5d' 3S	1	56558. 8		
4s(2S)10f	10f 3F°	2, 3, 4	48186. 61			3d(2D)5d	5d' 3F	2 3 4	56900. 7: 56924. 1: 56979. 5:	23. 4 55. 4	
4s(2S)10f	10f 1F°	3	48188. 3			3d(2D)5d	5d' 3P	0 1 2	57601. 0 57617. 8 57638. 2	16. 8 20. 4	
4s(2S)12p	12p 1P°	1	48222. 9			3d(2D)6d	6d' 3P	0 1 2	59366. 8: 59392. 0:	25. 2	
4s(2S)11d	11d 3D	1, 2, 3	48259. 2								
4s(2S)13s	13s 3S	1	48320. 4								
4s(2S)11f	11f 3F°	2, 3, 4	48382. 90								
4s(2S)11f	11f 1F°	3	48385. 5								
4s(2S)13p	13p 1P°	1	48416. 0								
4s(2S)12d	12d 3D	1, 2, 3	48434. 8								
4s(2S)14s	14s 3S	1	48484. 7								
3d ²	3d ² 3P	0 1 2	48524. 130 48537. 673 48563. 630	13. 543 25. 957							

May 1948.

Ca I OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms	
$4s^2$	$4s^2 \ ^1S$	
$3d^2$	$3d^2 \ ^3P$	
$4p^2$	{ $4p^2 \ ^1S$ $4p^2 \ ^3P$ $4p^2 \ ^1D$ }	
	$ns \ (n \geq 5)$	$np \ (n \geq 4)$
$4s(^2S)nx$	{ $5-14s \ ^3S$ $5-12s \ ^1S$ }	$4-7p \ ^3P^\circ$ $4-13p \ ^1P^\circ$
$3d(^2D)nx'$	{ $5s' \ ^3D$ }	$4p' \ ^3P^\circ$ $4, 5p' \ ^3D^\circ$ $4, 5p' \ ^3F^\circ$ $4p' \ ^1P^\circ$ $4p' \ ^1D^\circ$ $4p' \ ^1F^\circ$
	$nd \ (n \geq 3)$	$nf \ (n \geq 4)$
$4s(^2S)nx$	{ $3-16d \ ^3D$ $3-7d \ ^1D$ }	$4-13f \ ^3F^\circ$ $4-11f \ ^1F^\circ$
$3d(^2D)nx'$	$4, 5d' \ ^3S$ $4-6d' \ ^3P$ $4, 5d' \ ^3D$ $4, 5d' \ ^3F$ $4, 5d' \ ^3G$	

*For predicted terms in the spectra of the Ca I isoelectronic sequence, see Introduction.

Ca II

(K I sequence; 19 electrons)

$Z=20$

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 4s \ ^2S_{1/2}$

$4s \ ^2S_{1/2} \ 95748.0 \text{ cm}^{-1}$

I. P. 11.87 volts

The analysis is chiefly from the paper by Saunders and Russell, who extended the earlier work on this spectrum. Their estimated value of $5g \ ^2G$ is entered in brackets. The terms $nd \ ^2D$ ($n=11$ to 16) and $nf \ ^2F^\circ$ ($n=8$ to 10) have been added from an unpublished manuscript by Shenstone who made additional observations in the region between 2897 Å and 3758 Å. Shenstone has also generously furnished his recent unpublished observations of the pair of lines at 8927.34 Å and 8912.10 Å, having intensities 20 and 15, respectively, and classified as $4d \ ^2D-4f \ ^2F^\circ$. These lines have been used to calculate the value of $4f \ ^2F^\circ$ listed in the table.

The three-place entries are quoted from Wagman's paper. They are derived from his observations made with the interferometer. The writer has made slight adjustments in the rest of the term values in order to fit the various sets of observations together.

A monograph on this spectrum is needed.

REFERENCES

- F. A. Saunders and H. N. Russell, *Astroph. J.* **62**, 1 (1925). (I P) (T) (C L)
H. E. White, *Introduction to Atomic Spectra*, p. 97 (McGraw-Hill Book Co., Inc., New York, N. Y., 1934). (E D)
N. E. Wagman, *Univ. Pittsburgh Bul.* **34**, 1 (1937). (T) (C L)
A. G. Shenstone, unpublished material (1930, 1946). (T) (C L)

Ca II

Ca II

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3p^6(^1S)4s$	$4s\ ^2S$	$\frac{1}{2}$	0. 00		$3p^6(^1S)7g$	$7g\ ^2G$	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ 4\frac{1}{2} \end{smallmatrix} \right\}$	86780. 9	
$3p^6(^1S)3d$	$3d\ ^2D$	$\frac{1}{2}$ $2\frac{1}{2}$	13650. 212 13710. 901	60. 689	$3p^6(^1S)8d$	$8d\ ^2D$	$\frac{1}{2}$ $2\frac{1}{2}$	87674. 0 87675. 7	1. 7
$3p^6(^1S)4p$	$4p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	25191. 541 25414. 427	222. 886	$3p^6(^1S)8f$	$8f\ ^2F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	88847. 6	
$3p^6(^1S)5s$	$5s\ ^2S$	$\frac{1}{2}$	52166. 982		$3p^6(^1S)8g$	$8g\ ^2G$	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ 4\frac{1}{2} \end{smallmatrix} \right\}$	88883. 8	
$3p^6(^1S)4d$	$4d\ ^2D$	$\frac{1}{2}$ $2\frac{1}{2}$	56839. 309 56858. 511	19. 202	$3p^6(^1S)9d$	$9d\ ^2D$	$\frac{1}{2}$ $2\frac{1}{2}$	89489. 8 89490. 8	1. 0
$3p^6(^1S)5p$	$5p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	60535. 0 60613. 2	78. 2	$3p^6(^1S)9f$	$9f\ ^2F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	90300. 0	
$3p^6(^1S)4f$	$4f\ ^2F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	68056. 96		$3p^6(^1S)9g$	$9g\ ^2G$	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ 4\frac{1}{2} \end{smallmatrix} \right\}$	90326. 4	
$3p^6(^1S)6s$	$6s\ ^2S$	$\frac{1}{2}$	70677. 61		$3p^6(^1S)10d$	$10d\ ^2D$	$\frac{1}{2}$ $2\frac{1}{2}$	90755. 3 90756. 1	0. 8
$3p^6(^1S)5d$	$5d\ ^2D$	$\frac{1}{2}$ $2\frac{1}{2}$	72722. 11 72730. 77	8. 66	$3p^6(^1S)10f$	$10f\ ^2F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	91338. 0	
$3p^6(^1S)6p$	$6p\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	74485. 8 74521. 7	35. 9	$3p^6(^1S)11d$	$11d\ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	91674. 0	
$3p^6(^1S)5f$	$5f\ ^2F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	78027. 8		$3p^6(^1S)12d$	$12d\ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	92360. 9	
$3p^6(^1S)5g$	$5g\ ^2G$	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ 4\frac{1}{2} \end{smallmatrix} \right\}$	[78163]		$3p^6(^1S)13d$	$13d\ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	92885. 0	
$3p^6(^1S)7s$	$7s\ ^2S$	$\frac{1}{2}$	79449. 9		$3p^6(^1S)14d$	$14d\ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	93299. 6	
$3p^6(^1S)6d$	$6d\ ^2D$	$\frac{1}{2}$ $2\frac{1}{2}$	80523. 47 80528. 06	4. 59	$3p^6(^1S)15d$	$15d\ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	93628. 8	
$3p^6(^1S)6f$	$6f\ ^2F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	83458. 4		$3p^6(^1S)16d$	$16d\ ^2D$	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	93896. 4	
$3p^6(^1S)6g$	$6g\ ^2G$	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ 4\frac{1}{2} \end{smallmatrix} \right\}$	83540. 0						
$3p^6(^1S)8s$	$8s\ ^2S$	$\frac{1}{2}$	84302. 6						
$3p^6(^1S)7d$	$7d\ ^2D$	$\frac{1}{2}$ $2\frac{1}{2}$	84935. 4 84938. 3	2. 9					
$3p^6(^1S)7f$	$7f\ ^2F^\circ$	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	86727. 5						
					Ca III (1S_0)	Limit		95748.0	

May 1948.

Ca III

(A I sequence; 18 electrons)

 $Z=20$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 {}^1S_0$ $3p^6 {}^1S_0$ 413127 cm^{-1}

I. P. 51.21 volts

This spectrum is incompletely analyzed. The present list has been compiled from the paper by Bowen, who has classified 137 lines in the region between 403 Å and 4081 Å.

The Paschen notation as given by Bowen is entered in column one of the table, under the heading "A I". Bowen remarks, however, that these assignments are in many cases doubtful for levels having the $3d$ configuration. The writer has, nevertheless, adopted them tentatively in order to introduce the jl -coupling notation in the general form suggested by Racah, as in the case of all spectra like A I. The pairs $nd[3\frac{1}{2}]^\circ$ and $nd[1\frac{1}{2}]^\circ$ are partially inverted as compared with Ne I.

The LS -designations $ns {}^3P_{210}^\circ {}^1P_1^\circ$ can probably be safely assigned to the levels ns_5, ns_4, ns_3, ns_2 , respectively.

REFERENCE

I. S. Bowen, Phys. Rev. **31**, 499 (1928). (I P) (T) (C L)

Ca III

Ca III

A I	Bowen	Config.	Desig.	J	Level	A I	Bowen	Config.	Desig.	J	Level
$1p_0$	$3p$	$3p^6$	$3p^6 {}^1S$	0	0. 0	$2p_5$	$4p_5$	$3p^5({}^2P_{1/2})4p$	$4p [\frac{1}{2}]$	0	282072
		$3p^5({}^2P_{1/2})3d$	$3d [\frac{1}{2}]^\circ$	0		$2p_4$	$4p_4$	$3p^5({}^2P_{3/2})4p$	$4p' [1\frac{1}{2}]$	1	281136. 3
$3d_5$	$3D_1$			1	203845. 1	$2p_3$	$4p_3$			2	281878. 8
		"	$3d [3\frac{1}{2}]^\circ$	4		$2p_2$	$4p_2$	"	$4p' [\frac{1}{2}]$	1	282568. 4
$3d_4$	$3D_3$			3	213378. 3					0	
$3d_3$	$3D_2$	"	$3d [1\frac{1}{2}]^\circ$	2	204835. 4			$3p^5({}^2P_{1/2})4d$	$4d [\frac{1}{2}]^\circ$	0	
$3d_2$	$3D_5$			1	224552. 4	$4d_5$	$4D_1$			1	322998. 9
$3d_1''$	$3D_4$	"	$3d [2\frac{1}{2}]^\circ$	2	214332. 3			"	$4d [3\frac{1}{2}]^\circ$	4	
				3		$4d_4$	$4D_3$			3	326182
$3s_1'''$	$3D_6$	$3p^5({}^2P_{3/2})3d$	$3d' [2\frac{1}{2}]^\circ$	2	225823. 2	$4d_3$	$4D_2$	"	$4d [1\frac{1}{2}]^\circ$	2	323650. 6
$3s_1''$	$3D_8$			3	228411. 6					1	
$3s_1''$	$3D_7$	"	$3d' [1\frac{1}{2}]^\circ$	2	227387. 8	$4d_1''$	$4D_4$	"	$4d [2\frac{1}{2}]^\circ$	2	323086. 5
$3s_1'$	$3D_9$			1	232831. 4					3	
								$3p^5({}^2P_{3/2})4d$	$4d' [2\frac{1}{2}]^\circ$	2	
$1s_5$	$4s_5$	$3p^5({}^2P_{1/2})4s$	$4s [1\frac{1}{2}]^\circ$	2	242543. 5	$4s_1'''$	$4D_5$			3	335285. 9
$1s_4$	$4s_4$			1	243927. 0						
$1s_3$	$4s_3$	$3p^5({}^2P_{3/2})4s$	$4s' [\frac{1}{2}]^\circ$	0	245608. 4	$2s_5$	$5s_5$	$3p^5({}^2P_{1/2})5s$	$5s [1\frac{1}{2}]^\circ$	2	327917
$1s_2$	$4s_2$			1	247693. 4	$2s_4$	$5s_4$			1	328580. 4
						$2s_3$	$5s_3$	$3p^5({}^2P_{3/2})5s$	$5s' [\frac{1}{2}]^\circ$	0	331042. 7
$2p_{10}$	$4p_{10}$	$3p^5({}^2P_{1/2})4p$	$4p [\frac{1}{2}]$	1	272185. 4	$2s_2$	$5s_2$			1	331398. 6
$2p_9$	$4p_9$	"	$4p [2\frac{1}{2}]$	3	277018. 8						
$2p_8$	$4p_8$			2	277377. 5						
$2p_7$	$4p_7$	"	$4p [1\frac{1}{2}]$	1	278616. 7			Ca IV (${}^2P_{1/2}$)	Limit	-----	413127
$2p_6$	$4p_6$			2	279738. 2			Ca IV (${}^2P_{3/2}$)	Limit	-----	416261

May 1948.

Ca III OBSERVED LEVELS*

Config. 1s ² 2s ² 2p ⁶ 3s ² +	Observed Terms		
3p ⁶	3p ⁶ ¹ S		
	ns (n ≥ 4)		
3p ⁵ (² P°)nx	{	4, 5s ³ P° 4, 5s ¹ P°	
j <i>l</i> -Coupling Notation			
	Observed Levels		
	ns (n ≥ 4)	np (n ≥ 4)	nd (n ≥ 3)
	4, 5s [1½]°	4p [½] 4p [2½] 4p [1½]	3, 4d [½]° 3, 4d [3½]° 3, 4d [1½]° 3, 4d [2½]°
3p ⁵ (² P° _{1/2})nx'	4, 5s' [½]°	4p' [1½] 4p' [½]	3, 4d' [2½]° 3d' [1½]°

*For predicted levels in the spectra of the A I isoelectronic sequence, see Introduction.

Ca IV

(Cl I sequence; 17 electrons)

Z=20

Ground state 1s² 2s² 2p⁶ 3s² 3p⁵ ²P°_{1/2}

3p⁵ ²P°_{1/2} 542000 cm⁻¹

I. P. 67 volts

Various investigators disagree about the interpretation of this spectrum. Tsien has published 34 classified lines in the region between 249 Å and 669 Å, all but one of which are due to combinations from the ground term. His terms are listed except for 4s⁴P, 4s²P, and 4s' ²D, which are from the paper by Kruger and Phillips. Further study of this spectrum is desirable to confirm the present analysis.

The limit (entered in brackets in the table) is from Edlén, who has estimated it by extrapolation along the isoelectronic sequence.

REFERENCES

I. S. Bowen, Phys. Rev. **31**, 498 (1928). (C L)
B. Edlén, Zeit. Phys. **104**, 410 (1947). (I P)
P. G. Kruger and L. W. Phillips, Phys. Rev. **51**, 1087 (1937). (T) (C L)
W.-Z. Tsien, Chinese J. Phys. **3**, No. 2, 118 (1939). (T) (C L)

Ca IV

Ca IV

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2 3p^5$	$3p^5 \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	0 3115	-3115	$3s^2 3p^4(^3P)4s$	$4s \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	298175 300249	-2074
$3s \ 3p^6$	$3p^6 \ ^2S$	$\frac{1}{2}$	152430		$3s^2 3p^4(^1D)3d$	$3d' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	303591	
$3s^2 3p^4(^3P)3d$	$3d \ ^4F$	$4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$	221944		$3s^2 3p^4(^1D)3d$	$3d' \ ^2S$	$\frac{1}{2}$	303844	
$3s^2 3p^4(^3P)3d$	$3d \ ^4D$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	227427 227827 228691	-400 -864	$3s^2 3p^4(^1D)4s$	$4s' \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	314079 314373	294
$3s^2 3p^4(^3P)3d$	$3d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	228429 230113	1684	$3s^2 3p^4(^3P)4p$	$4p \ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	329277	
$3s^2 3p^4(^3P)3d$	$3d \ ^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	266840		$3s^2 3p^4(^1S)4s$	$4s'' \ ^2S$	$\frac{1}{2}$	337207	
$3s^2 3p^4(^3P)4s$	$4s \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	291373 293011 294291	-1638 -1280	$3s^2 3p^4(^1D)5s$	$5s' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	399755 400949	-1194
					Ca V ($3p^4 \ ^3P_2$)	Limit	-----	[542000]	

March 1948.

Ca IV OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6$	Observed Terms		
$3s^2 3p^5$	$3p^5 \ ^2P^\circ$		
$3s \ 3p^6$	$3p^6 \ ^2S$		
	$ns \ (n \geq 4)$	$np \ (n \geq 4)$	$nd \ (n \geq 3)$
$3s^2 3p^4(^3P)nx$	$\left\{ \begin{array}{l} 4s \ ^4P \\ 4s \ ^2P \end{array} \right.$	$4p \ ^2P^\circ$	$\begin{array}{ll} 3d \ ^4D & 3d \ ^4F \\ 3d \ ^2D & 3d \ ^2F \end{array}$
$3s^2 3p^4(^1D)nx'$	$4, 5s' \ ^2D$		$3d' \ ^2S \quad 3d' \ ^2D$
$3s^2 3p^4(^1S)nx''$	$4s'' \ ^2S$		

*For predicted terms in the spectra of the Cl I isoelectronic sequence, see Introduction.

Ca V

(S I sequence; 16 electrons)

 $Z=20$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^4 \ ^3P_2$ $3p^4 \ ^3P_2 \ 680800 \text{ cm}^{-1}$

I. P. 84.39 volts

The terms are from the papers by Bowen and by Tsien with the revised value of $3p^5 \ ^1P^\circ$ suggested by Edlén.

More than 70 lines have been classified in the interval 184 Å to 656 Å. Intersystem combinations connecting the singlet and triplet terms have been observed.

REFERENCES

- M. Ram, Indian J. Phys. **8**, 167 (1933). (T) (C L)
 I. S. Bowen, Phys. Rev. **46**, 791 (1934). (T) (C L)
 B. Edlén, Zeit. Phys. **104**, 192 (1937). (I P)
 W.-Z. Tsien, Chinese J. Phys. **3**, No. 2, 131 (1939). (T) (C L)
 B. Edlén, Phys. Rev. **62**, 434 (1942). (T) (C L)

Ca v

Ca v

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2 3p^4$	$3p^4 \ ^3P$	2 1 0	0 2404 3276	-2404 -872	$3s^2 3p^3(^2D^\circ)4s$	$4s' \ ^3D^\circ$	1 2 3	369590 369696 369959	106 263
$3s^2 3p^4$	$3p^4 \ ^1D$	2	18831		$3s^2 3p^3(^2D^\circ)4s$	$4s' \ ^1D^\circ$	2	374728	
$3s^2 3p^4$	$3p^4 \ ^1S$	0	43847		$3s^2 3p^3(^2P^\circ)4s$	$4s'' \ ^3P^\circ$	0 1 2	387039 387226 387652	187 426
$3s 3p^5$	$3p^5 \ ^3P^\circ$	2 1 0	154664 156756 157897	-2092 -1141	$3s^2 3p^3(^2P^\circ)4s$	$4s'' \ ^1P^\circ$	1	392283	
$3s 3p^5$	$3p^5 \ ^1P^\circ$	1	197849		$3s^2 3p^3(^4S^\circ)5s$	$5s \ ^3S^\circ$	1	501127	
$3s^2 3p^3(^2D^\circ)3d$	$3d' \ ^1F^\circ$	3	254125		$3s^2 3p^3(^2D^\circ)5s$	$5s' \ ^3D^\circ$	1 2 3	524651 524770 525053	119 283
$3s^2 3p^3(^2P^\circ)3d$	$3d'' \ ^3P^\circ$	2 1 0	298204 299535	-1331	$3s^2 3p^3(^2D^\circ)5s$	$5s' \ ^1D^\circ$	2	526523	
$3s^2 3p^3(^2P^\circ)3d$	$3d'' \ ^1P^\circ$	1	302184		$3s^2 3p^3(^2P^\circ)5s$	$5s'' \ ^3P^\circ$	0 1 2	542249 542650	401
$3s^2 3p^3(^2P^\circ)3d$	$3d'' \ ^3D^\circ$	3 2 1	309834 310945	-1111	$3s^2 3p^3(^2P^\circ)5s$	$5s'' \ ^1P^\circ$	1	544143	
$3s^2 3p^3(^2P^\circ)3d$	$3d'' \ ^1D^\circ$	2	329230						
$3s^2 3p^3(^4S^\circ)4s$	$4s \ ^3S^\circ$	1	350914		Ca vi ($^4S_{1/2}$)	Limit		680800	

December 1947.

Ca v OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 +$	Observed Terms	
$3s^2 3p^4$	{ $3p^4 \ ^1S$ $3p^4 \ ^3P$ $3p^4 \ ^1D$	
$3s 3p^5$	{ $3p^5 \ ^3P^\circ$ $3p^5 \ ^1P^\circ$	
	$ns \ (n \geq 4)$	$nd \ (n \geq 3)$
$3s^2 3p^3(^4S^\circ)nx$	4, 5s $^3S^\circ$	
$3s^2 3p^3(^2D^\circ)nx'$	{	4, 5s' $^3D^\circ$ 4, 5s' $^1D^\circ$ $3d' \ ^1F^\circ$
$3s^2 3p^3(^2P^\circ)nx''$	{	4, 5s'' $^3P^\circ$ $3d'' \ ^3P^\circ$ $3d'' \ ^3D^\circ$ 4, 5s'' $^1P^\circ$ $3d'' \ ^1P^\circ$ $3d'' \ ^1D^\circ$

*For predicted terms in the spectra of the S I isoelectronic sequence, see Introduction.

Ca VI

(P I sequence; 15 electrons)

 $Z=20$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^3 {}^4S_{1\frac{1}{2}}$ $3p^3 {}^4S_{1\frac{1}{2}}$ cm⁻¹

I. P. volts

The terms are from the paper by Tsien, who includes those given earlier by Bowen. Fifty-three lines have been classified in the interval between 228 Å and 766 Å. For the term $3p^4 {}^2P$ the value given by Mrs. Beckman is quoted in place of that by Tsien.

The relative positions of the doublet and quartet systems of terms were estimated from the irregular doublet law. No intersystem combinations have been observed, as indicated by the uncertainty x in the table and the brackets around $3p^3 {}^2D_{1\frac{1}{2}}$.

REFERENCES

I. S. Bowen, Phys. Rev. **46**, 791 (1934). (T) (C L)A. Beckman, *Bidrag till Kännedomen om Skandiums Spektrum i Yttersta Ultraviolett*, Akademisk Avhandling p. 74 (Almqvist and Wiksells Boktryckeri -A.-B., Uppsala, 1937). (C L)W.-Z. Tsien, Chinese J. Phys. **3**, No. 2, 136 (1939). (T) (C L)

Ca VI

Ca VI

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3s^2 3p^3$	$3p^3 {}^4S^{\circ}$	$1\frac{1}{2}$	0			3		$303651+x$	
$3s^2 3p^3$	$3p^3 {}^2D^{\circ}$	$1\frac{1}{2}$ $2\frac{1}{2}$	$[27000]+x$ $27417+x$	417	$3s^2 3p^2({}^1D)3d$	$3d' {}^2S$	$\frac{1}{2}$	$320397+x$	
$3s^2 3p^3$	$3p^3 {}^2P^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$	$44754+x$ $45310+x$	556	$3s^2 3p^2({}^1D)3d$	$3d' {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	$321084+x$ $321584+x$	-500
$3s 3p^4$	$3p^4 {}^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	155792 157775 158833	-1983 -1058	$3s^2 3p^2({}^1D)3d$	$3d' {}^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	$332138+x$ $333492+x$	1354
	1 (2D) 2	$2\frac{1}{2}$ $1\frac{1}{2}$	$175758+x$ $176157+x$	-399	$3s^2 3p^2({}^1S)3d$	$3d'' {}^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	$360821+x$	
$3s 3p^4$	$3p^4 {}^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	$193412+x$ $193613+x$	201	$3s^2 3p^2({}^3P)3d$	$3d {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	$383743+x$	
$3s 3p^4$	$3p^4 {}^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	$223170+x$		$3s^2 3p^2({}^3P)4s$	$4s {}^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	433849 435286 437392	1437 2106
$3s 3p^4$	$3p^4 {}^2S$	$\frac{1}{2}$	$231318+x$		$3s^2 3p^2({}^3P)4s$	$4s {}^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	$442423+x$ $444890+x$	2467
$3s^2 3p^2({}^3P)3d$	$3d {}^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	$291165+x$		$3s^2 3p^2({}^1D)4s$	$4s' {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	$457458+x$ $457525+x$	-67
$3s^2 3p^2({}^3P)3d$	$3d {}^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	$294798+x$ $297250+x$	-2452					

November 1947.

Ca VI OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 +$	Observed Terms	
$3s^2 3p^3$	$\{ 3p^3 \ ^4S^\circ \quad 3p^3 \ ^2P^\circ \quad 3p^3 \ ^2D^\circ$	
$3s 3p^4$	$\{ 3p^4 \ ^2S \quad 3p^4 \ ^4P \quad 3p^4 \ ^2P \quad 3p^4 \ ^2D$	
	$ns \ (n \geq 4)$	$nd \ (n \geq 3)$
$3s^2 3p^2(^3P)nx$	$\{ \begin{matrix} 4s \ ^4P \\ 4s \ ^2P \end{matrix}$	$3d' \ ^2P \quad 3d \ ^2D \quad 3d \ ^2F$
$3s^2 3p^2(^1D)nx'$	$4s' \ ^2D$	$3d' \ ^2S \quad 3d' \ ^2P \quad 3d' \ ^2D$
$3s^2 3p^2(^1S)nx''$		$3d'' \ ^2D$

*For predicted terms in the spectra of the P I isoelectronic sequence, see Introduction.

Ca VII

(Si I sequence; 14 electrons)

$Z=20$

Ground state $1s^2 2s^2 2p^6 3s^2 3p^2 \ ^3P_0$

$3p^2 \ ^3P_0$ 1030000 cm^{-1}

I. P. 128 volts

The terms are from the paper by Phillips, who includes those found by Whitford and by Robinson. In the interval between 202 Å and 640 Å, 33 lines have been classified in all. Intersystem combinations connecting the singlet and triplet terms have been observed.

The limit entered in brackets in the table has been estimated by Phillips.

REFERENCES

- A. E. Whitford, Phys. Rev. **46**, 793 (1934). (T) (C L)
 H. A. Robinson, Phys. Rev. **52**, 725 (1937). (T) (C L)
 L. W. Phillips, Phys. Rev. **55**, 708 (1939). (I P) (T) (C L)

Ca VII

Ca VII

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3s^2 3p^2$	$3p^2 \ ^3P$	0 1 2	0 1627 4070	1627 2443	$3s^2 3p(^2P^\circ)3d$	$3d \ ^3P^\circ$	2 1 0	286232 288169 289011	—1937 —842
$3s^2 3p^2$	$3p^2 \ ^1D$	2	21870		$3s^2 3p(^2P^\circ)3d$	$3d \ ^3D^\circ$	1 2 3	302663 303151 303349	488 198
$3s 3p^3$	$3p^3 \ ^3D^\circ$	1 2 3	160160 160228 160527	68 299	$3s^2 3p(^2P^\circ)4s$	$4s \ ^3P^\circ$	0 1 2	490012 490918 494264	906 3346
$3s 3p^3$	$3p^3 \ ^3P^\circ$	2, 1, 0	185405						
$3s 3p^3$	$3p^3 \ ^3S^\circ$	1	245232						
$3s 3p^3$	$3p^3 \ ^1P^\circ$	1	252493		Ca VIII ($^2P_{3/2}^\circ$)	Limit	-----	[1030000]	

October 1947.

Ca VIII

(Al I sequence; 13 electrons)

 $Z=20$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^2 P_{1/2}^\circ$ $3p^2 P_{1/2}^\circ$ 1189000 cm^{-1}

I. P. 147 volts

The analysis is by Whitford and by Phillips. Thirty-five lines have been classified in the interval between 114 Å and 596 Å. No intersystem combinations have been observed, but Phillips estimates that $3p^2 {}^4P_{1/2}$ is approximately 128000 cm^{-1} above the ground state. This value is entered in brackets in the table. The uncertainty x may exceed $\pm 1000 \text{ cm}^{-1}$.

Using the method suggested by Edlén, the writer has extrapolated the value of the limit quoted above and entered in brackets in the table. The uncertainty in this estimate is large owing to the incompleteness of the isoelectronic sequence data.

REFERENCES

A. E. Whitford, Phys. Rev. **46**, 793 (1934). (T) (C L)L. W. Phillips, Phys. Rev. **55**, 708 (1939). (T) (C L)

Ca VIII

Ca VIII

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3s^2({}^1S)3p$	$3p^2 P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	0 4305	4305	$3s 3p({}^3P^\circ)3d$	$3d {}^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	$410725 + x$ $411283 + x$ $411664 + x$ $411782 + x$	558 381 118
$3s 3p^2$	$3p^2 {}^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$[128000] + x$ $129581 + x$ $131942 + x$	1581 2361	$3s^2({}^1S)4s$	$4s {}^2S$	$\frac{1}{2}$	547308	
$3s 3p^2$	$3p^2 {}^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	171573 171828	255	$3s 3p({}^3P^\circ)4s$	$4s {}^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$687650 + x$ $689017 + x$ $691726 + x$	1367 2709
$3s 3p^2$	$3p^2 {}^2S$	$\frac{1}{2}$	216590		$3s^2({}^1S)4d$	$4d {}^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	697981 698172	191
$3s 3p^2$	$3p^2 {}^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	231012 233584	2572	$3s^2({}^1S)5d$	$5d {}^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	872860 873070	210
$3s^2({}^1S)3d$	$3d {}^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	282362 282574	212					
$3p^3$	$3p^3 {}^4S^\circ$	$1\frac{1}{2}$	$344176 + x$		Ca IV (1S_0)	Limit	-----	[1189000]	

September 1947.

Ca VIII OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 +$	Observed Terms	
$3s^2({}^1S)3p$	$3p^2 P^\circ$	
$3s 3p^2$	$\begin{cases} 3p^2 {}^2S & 3p^2 {}^4P & 3p^2 {}^2D \\ & 3p^2 {}^2P \end{cases}$	
$3p^3$	$3p^3 {}^4S^\circ$	
	$ns (n \geq 4)$	$nd (n \geq 3)$
$3s^2({}^1S)nx$	$4s {}^2S$	$3-5d {}^2D$
$3s 3p({}^3P^\circ)nx$	$4s {}^4P^\circ$	$3d {}^4D^\circ$

*For predicted terms in the spectra of the Al I isoelectronic sequence, see Introduction.

Ca IX

(Mg I sequence; 12 electrons)

 $Z=20$ Ground state $1s^2 2s^2 2p^6 3s^2 {}^1S_0$ $3s^2 {}^1S_0$ 1519000 \pm cm^{-1} I. P. $188 \pm$ volts

Twenty-eight lines have been classified in the range between 100 Å and 828 Å. The triplet terms are from Parker and Phillips; the singlets from Tsien. By extrapolation along the sequence, Mrs. Beckman has classified a line at 693.824 Å as the intersystem combination $3s^2 {}^1S_0 - 3p {}^3P_1^o$. The listed values of the triplet terms have been adjusted to fit this assignment.

From isoelectronic sequence data, the writer has extrapolated the value of the limit, using the method suggested by Edlén. This value is entered in brackets in the table. Although this estimate may be in error by more than $\pm 1000 \text{ cm}^{-1}$, it gives an approximate value of the ionization potential.

REFERENCES

- A. Beckman, *Bidrag till Kännedomen om Skandiums Spektrum i Yttersta Ultraviolett*, Akademisk Avhandling p. 55 (Almqvist and Wiksells Boktryckeri -A.-B., Uppsala, 1937). (C L)
 W.-Z. Tsien, Chinese J. Phys. **3**, No. 2, 142 (1939). (T) (C L)
 W. L. Parker and L. W. Phillips, Phys. Rev. **57**, 140 (1940). (T) (C L)

Ca IX

Ca IX

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3s^2$	$3s^2 {}^1S$	0	0		$3s({}^2S)4s$	$4s {}^3S$	1	760002	
$3s({}^2S)3p$	$3p {}^3P^o$	0 1 2	142635 144130 147370	1495 3240	$3s({}^2S)4d$	$4d {}^3D$	1 2 3	916652 916780 916990	128 210
$3s({}^2S)3p$	$3p {}^1P^o$	1	214487.8		$3s({}^2S)4f$	$4f {}^3F^o$	2 3 4	954003 954023 954055	20 32
$3s({}^2S)3d$	$3d {}^1D$	2	335195.0		$3s({}^2S)5d$	$5d {}^3D$	1 2 3	1137720 1137880	160
$3p^2$	$3p^2 {}^3P$	0 1 2	339420 341333 344935	1913 3602					
$3s({}^2S)3d$	$3d {}^3D$	1 2 3	411525 411652 411858	127 206					
					Ca x (${}^2S_{1/2}$)	Limit	-----	[1519000]	

March 1948.

Ca x

(Na I sequence; 11 electrons)

$Z=20$

Ground state $1s^2 2s^2 2p^6 3s^2 S_{1/2}$

$3s^2 S_{1/2}$ 1704660 cm^{-1}

I. P. 211.29 volts

Kruger and Phillips extended the earlier analysis by Edlén. Their absolute term values are derived from three members of the 2D -series. One term, $5s^2 S$ has been added from the work of Tsien but adjusted to agree with those by Kruger and Phillips.

Twenty-two lines have been classified in the range from 93 Å to 574 Å.

REFERENCES

W.-Z. Tsien, Chinese J. Phys **3**, No. 2, 145 (1939). (T) (C L)

P. G. Kruger and L. W. Phillips, Phys. Rev. **55**, 352 (1939). (I P) (T) (C L)

Ca x

Ca x

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
3s	3s 2S	$\frac{1}{2}$	0		4f	4f $^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	1016113 1016208	95
3p	3p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	174214 179295	5081	5s	5s 2S	$\frac{1}{2}$	1170098	
3d	3d 2D	$1\frac{1}{2}$ $2\frac{1}{2}$	417113 417527	414	5d	5d 2D	$1\frac{1}{2}$ $2\frac{1}{2}$	1248686 1248791	105
4s	4s 2S	$\frac{1}{2}$	832838		5f	5f $^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	1263323 1263383	60
4p	4p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	899305 901210	1905	6f	6f $^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	1398140	
4d	4d 2D	$1\frac{1}{2}$ $2\frac{1}{2}$	987259 987484	225					
					Ca XI (1S_0)	Limit	-----	1704660	

June 1947.

Ca XI

(Ne I sequence; 10 electrons)

$Z=20$

Ground state $1s^2 2s^2 2p^6 ^1S_0$

$2p^6 ^1S_0$ 4774300 cm^{-1}

I. P. 591.8 volts

Eleven lines between 25 Å and 35 Å have been classified by Edlén and Tyrén as combinations with the ground term. Their absolute term values have been extrapolated along the Ne I isoelectronic sequence.

By analogy with Ne I, the $j\ell$ -coupling notation in the general form suggested by Racah is introduced.

The unit adopted by Edlén and Tyrén, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

B. Edlén and F. Tyrén, Zeit. Phys. **101**, 206 (1936). (I P) (T) (C L)

G. Racah, Phys. Rev **61**, 537 (L) (1942).

Ca XI

Ca XI

Authors	Config.	Desig.	<i>J</i>	Level	Authors	Config.	Desig.	<i>J</i>	Level
$2p\ ^1S_0$	$2s^2\ 2p^6$	$2p^6\ ^1S$	0	0	$3p'\ ^1P_1$	$2s\ 2p^6(^2S)3p$	$3p\ ^1P^{\circ}$	1	3708900
$3s\ ^3P_1$	$2s^2\ 2p^5(^2P^{\circ}_{1/2})3s$	$3s\ [1\frac{1}{2}]^{\circ}$	2 1	2810900	$4s\ ^3P_1$	$2s^2\ 2p^5(^2P^{\circ}_{1/2})4s$	$4s\ [1\frac{1}{2}]^{\circ}$	2 1	3753900
$3s\ ^1P_1$	$2s^2\ 2p^5(^2P^{\circ}_{3/2})3s$	$3s'\ [\frac{1}{2}]^{\circ}$	0 1	2839900	$4s\ ^1P_1$	$2s^2\ 2p^5(^2P^{\circ}_{3/2})4s$	$4s'\ [\frac{1}{2}]^{\circ}$	0 1	3781900
$3d\ ^3P_1$	$2s^2\ 2p^5(^2P^{\circ}_{1/2})3d$	$3d\ [\frac{1}{2}]^{\circ}$	0 1	3199300	$4d\ ^1P_1$	$2s^2\ 2p^5(^2P^{\circ}_{1/2})4d$	$4d\ [1\frac{1}{2}]^{\circ}$	1	3919000
$3d\ ^1P_1$	"	$3d\ [1\frac{1}{2}]^{\circ}$	1	3239700	$4d\ ^3D_1$	$2s^2\ 2p^5(^2P^{\circ}_{3/2})4d$	$4d'\ [1\frac{1}{2}]^{\circ}$	1	3948400
$3d\ ^3D_1$	$2s^2\ 2p^5(^2P^{\circ}_{3/2})3d$	$3d'\ [1\frac{1}{2}]^{\circ}$	1	3284300					
$3p'\ ^3P_1$	$2s\ 2p^6(^2S)3p$	$3p\ ^3P^{\circ}$	2 1 0	3692900		Ca XII ($^2P^{\circ}_{1/2}$)	-----	Limit	4774300
						Ca XII ($^2P^{\circ}_{3/2}$)	-----	Limit	4804328

April 1947.

Ca XI OBSERVED LEVELS*

Config. 1s ² +	Observed Terms		
2s ² 2p ⁶	2p ⁶ 1S		
	ns (n ≥ 3)	np (n ≥ 3)	nd (n ≥ 3)
	2s ² 2p ⁵ (² P°)nx	{ 3, 4s ³ P° 3, 4s ¹ P°	
2s 2p ⁶ (² S)nx	{	3p ³ P° 3p ¹ P°	
j <i>l</i> -Coupling Notation			
	Observed Pairs		
	ns (n ≥ 3)		nd (n ≥ 3)
	2s ² 2p ⁵ (² P _{1/2} ^o)nx	3, 4s [1½] ^o	3d [½] ^o 3, 4d [1½] ^o
2s ² 2p ⁵ (² P _{3/2} ^o)nx'	3, 4s'[½] ^o	3, 4d'[1½] ^o	

*For predicted levels in the spectra of the Ne I isoelectronic sequence, see Introduction.

Ca XII

(F I sequence; 9 electrons)

 $Z=20$ Ground state $1s^2 2s^2 2p^5 {}^2P_{1/2}^\circ$ $2p^5 {}^2P_{1/2}^\circ$ cm^{-1}

I. P. 655 volts

Edlén and Tyrén have classified 9 lines in the range 27 Å to 32 Å. They have published no term array because the analysis is so incomplete. In the 1942 paper Edlén lists the interval of the ground term as 30028 cm^{-1} , a value based on unpublished material. From these data preliminary term values have been calculated and entered in the table.

REFERENCES

B. Edlén and F. Tyrén, Zeit. Phys. **101**, 206 (1936). (C L)B. Edlén, Zeit. Astroph. **22**, 59 (1942). (I P) (T)

Ca XII

Edlén	Config.	Desig.	J	Level	Interval
$2p {}^2P_2$ $2p {}^2P_1$	$2s^2 2p^5$	$2p^5 {}^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	0 30028	-30028
$3s {}^4P_3$ $3s {}^4P_2$	$2s^2 2p^4({}^3P)3s$	$3s {}^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	3062300 3077100	-14800
$3s {}^2P_2$ $3s {}^2P_1$	$2s^2 2p^4({}^3P)3s$	$3s {}^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	3097900	
$\overline{3s} {}^2D_3$ $\overline{3s} {}^2D_2$	$2s^2 2p^4({}^1D)3s$	$3s' {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	3158600 3158900	-300
$\overline{3d}$	$2s^2 2p^4({}^1D)3d$	$3d' X$		3574200	
$\overline{\overline{3d}} {}^2D_3$ $\overline{\overline{3d}} {}^2D_2$	$2s^2 2p^4({}^1S)3d$	$3d'' {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	3648000 3652400	-4400

March 1947.

Ca XIII

(O I sequence; 8 electrons)

 $Z=20$ Ground state $1s^2 2s^2 2p^4 {}^3P_2$ $2p^4 {}^3P_2$ cm^{-1}

I. P. volts

This spectrum has not been analyzed. Edlén suggests the possibility that the line observed in the coronal spectrum at 4086.3 Å (24465 cm^{-1}) may be due to the forbidden transition $2p^4 {}^3P_2 - 2p^4 {}^3P_1$ of Ca XIII. This separation for the leading components of the ground term is not inconsistent with that extrapolated along the O I isoelectronic sequence.

REFERENCE

B. Edlén, *Zeit. Astroph.* **22**, 62 (1942). (T)

March 1947.

Ca XV

(C I sequence; 6 electrons)

 $Z=20$ Ground state $1s^2 2s^2 2p^2 {}^3P_0$ $2p^2 {}^3P_0$ cm^{-1} .

I. P. volts

An extrapolation of the ground term interval along the C I isoelectronic sequence indicates that the separations of the components of the ground term, $2s^2 2p^2 {}^3P$, should be approximately 17700 cm^{-1} , according to Edlén. He suggests that the line observed in the solar corona at 5694.42 Å, wave number 17556 cm^{-1} , may tentatively be identified as [Ca xv]?, $2s^2 2p^2 {}^3P_0 - 2s^2 2p^2 {}^3P_1$.

REFERENCE

B. Edlén, *Zeit. Astroph.* **22**, 59 (1942). (T)

March 1947

SCANDIUM

Sc I

21 electrons

 $Z=21$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d 4s^2 {}^2D_{1/2}$ $a {}^2D_{1/2}$ 52920 cm^{-1}

I. P. 6.56 volts

The analysis is chiefly from the paper by Russell and Meggers with some additions from unpublished manuscript generously furnished by Russell. In the published analysis the terms $a {}^4P$, $y {}^4P^\circ$, and $z {}^4S^\circ$ were unconnected with the rest and $a {}^4P_{3/2}$ was assigned the value x . The connection is now established from observed combinations.

Similarly, the group $a {}^2P$, $v {}^2D^\circ$, $z {}^2S^\circ$ and $u {}^2D^\circ$ were connected with the rest only by the relation $a {}^2P_{3/2}=y$. Ufford has predicted the relative position of $a {}^2P$. His estimated value, $a {}^2P_{3/2}=21400$, is entered in brackets in the table and has been added to all levels in this group of terms. The uncertainty is indicated by y since the group is not connected with the rest by observed combinations.

The two terms, $f {}^4P$ and $x {}^4D^\circ$ have been added from the unpublished material mentioned above. The limit is also from a recalculation of the series recently made by Russell for inclusion here.

Russell and Meggers have noted that the assignment of the limit terms to the two triads $z {}^2P^\circ$ $z {}^2D^\circ$ $z {}^2F^\circ$, $y {}^2P^\circ$ $y {}^2D^\circ$ $y {}^2F^\circ$ is uncertain. One triad has the limit $a {}^3D$ in Sc II and the other, $a {}^1D$. Russell, in discussing the behavior of the d electrons in related spectra, concludes that the higher triad has as its limit the term of higher multiplicity. (See 1927 reference below.)

The doublet and quartet terms are connected by observed intersystem combinations.

In the 1925 paper mentioned below some observed Zeeman patterns are given. Catalán has calculated from these patterns the g -values listed in the table.

REFERENCES

- S. Goudsmit, J. van der Mark, and P. Zeeman, Proc. Roy. Acad. Amsterdam **28**, No. 2, 127 (1925). (Z E)
 H. N. Russell and W. F. Meggers, Sci. Papers Bur. Std. **22**, No. 558, 340 (1927). (I P) (T) (C L) (G D)
 H. N. Russell, Astroph. J. **66**, 201 (1927); Mt. Wilson Contr. No. 341 (1927).
 C. W. Ufford, unpublished material (July 1941). (T)
 H. N. Russell, unpublished material (Jan. 1934, May 1948). (I P) (T) (C L)
 M. A. Catalán, unpublished material (June 1948). (Z E)
 W. F. Meggers, J. Opt. Soc. Am. **36**, 431 (1946). (Summary hfs.)

Sc I

Sc I

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$3d\ 4s^2$	$a\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	0. 00 168. 34	168. 34	0. 79 1. 20	$3d^2(a\ ^3F)4p$	$y\ ^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	32637. 40 32659. 21 32696. 84 32751. 54	21. 81 37. 63 54. 70	
$3d^2(a\ ^3F)4s$	$a\ ^4F$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	11520. 15 11557. 64 11610. 24 11677. 31	37. 49 52. 60 67. 07		$3d^2(a\ ^3F)4p$	$z\ ^2G^\circ$	$3\frac{1}{2}$ $4\frac{1}{2}$	33056. 19 33151. 40	95. 21	
$3d^2(a\ ^3F)4s$	$a\ ^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	14926. 24 15041. 98	115. 74		$3d^2(a\ ^3F)4p$	$x\ ^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	33154. 01 33278. 64	124. 63	
$3d\ 4s(a\ ^3D)4p$	$z\ ^4F^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	15672. 55 15756. 51 15881. 76 16026. 52	83. 96 125. 25 144. 76		$3d^2(a\ ^3F)4p$	$x\ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	33615. 06 33707. 25	92. 19	
$3d\ 4s(a\ ^3D)4p$	$z\ ^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	16009. 71 16021. 78 16141. 04 16210. 80	12. 07 119. 26 69. 76		$3d^3$	$e\ ^4F$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	33763. 57 33798. 68 33846. 62 33906. 40	35. 11 47. 94 59. 78	
$3d\ 4s(a\ ^1D)4p$	$z\ ^2D^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$	16022. 72 16096. 86	-74. 14		$3d\ 4s(a\ ^3D)5s$	$e\ ^4D$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	34390. 25 34422. 85 34480. 05 34567. 10	32. 60 57. 20 87. 05	
$3d^2(b\ ^1D)4s$	$b\ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	17012. 98 17025. 36	-12. 38		$3d\ 4s(a\ ^3D)5s$	$e\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	35671. 00 35745. 57	74. 57	
$3d^2(a\ ^3P)4s$	$a\ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	17918. 85 17947. 98 18000. 25	29. 13 52. 27		$3d^3$	$f\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	36276. 76 36330. 49	53. 73	
$3d\ 4s(a\ ^3D)4p$	$z\ ^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	18504. 05 18515. 77 18571. 40	11. 72 55. 63		$3d^3$	$e\ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	36492. 82 36515. 76 36572. 80	22. 94 57. 04	
$3d\ 4s(a\ ^1D)4p$	$z\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	18711. 08 18855. 76	144. 73		$3d^2(b\ ^1D)4p$	$w\ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	36934. 15 37039. 77	105. 62	
$3d^2(a\ ^1G)4s$	$a\ ^2G$	$4\frac{1}{2}$ $3\frac{1}{2}$	20237. 10 20239. 92	-2. 82		$3d\ 4s(a\ ^3D)4d$	$e\ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	37085. 72 37148. 25	62. 53	
$3d\ 4s(a\ ^1D)4p$	$z\ ^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	21032. 78 21085. 84	53. 06		$3d^2(b\ ^1D)4p$	$w\ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	37086. 31 37125. 72	-39. 41	
$3d^2(a\ ^3P)4s$	$a\ ^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	[21400]+ <i>y</i> 21480. 40 + <i>y</i>	80. 40		$3d^2(a\ ^3P)4p$	$x\ ^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	37486. 48 37553. 34 37717. 11	66. 86 163. 77	
$3d\ 4s(a\ ^3D)4p$	$y\ ^2P^\circ$	$\left\{ \begin{array}{l} \frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	24656. 80			$3d\ 4s(a\ ^3D)4d$	$g\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	37780. 83 37855. 50	74. 67	
$3d\ 4s(a\ ^3D)4p$	$y\ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	24866. 18 25014. 15	147. 97	0. 82 1. 17	$3d^2(a\ ^3P)4p$	$z\ ^4S^\circ$	$1\frac{1}{2}$	38179. 92		
$3d\ 4s(a\ ^3D)4p$	$y\ ^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	25584. 64 25724. 72	140. 08	0. 90 1. 14	$3d^2(a\ ^3P)4p$	$y\ ^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	38570. 64 38601. 50 38657. 93	30. 86 56. 43	
$3d^2(a\ ^3F)4p$	$z\ ^4G^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$ $5\frac{1}{2}$	29022. 87 29096. 20 29189. 83 29303. 52	73. 33 93. 63 113. 69		$3d\ 4s(a\ ^3D)4d$	$e\ ^2G$	$3\frac{1}{2}$ $4\frac{1}{2}$	38571. 70 38658. 23	86. 53	
$3d^2(a\ ^1S)4p$	$x\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	30573. 10 30706. 61	133. 51	0. 68	$3d\ 4s(a\ ^3D)4d$	$e\ ^2F$	$2\frac{1}{2}$ $3\frac{1}{2}$	38871. 60 38959. 16	87. 56	
$3d^2(a\ ^3F)4p$	$y\ ^4F^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	31172. 62 31215. 76 31275. 32 31350. 81	43. 14 59. 56 75. 49		$3d^2(a\ ^1G)4p$	$z\ ^2H^\circ$	$4\frac{1}{2}$ $5\frac{1}{2}$	39153. 42 39249. 27	95. 85	
						$3d^2(a\ ^1G)4p$	$y\ ^2G^\circ$	$3\frac{1}{2}$ $4\frac{1}{2}$	39392. 95 39423. 73	30. 78	

Sc I—Continued

Sc I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	
<i>3d 4s(a ³D)4d</i>	<i>f ⁴D</i>	½	39701. 30	20. 41 33. 22 44. 92		<i>3d²(a ³P)4p:</i>	<i>v ²D°</i>	1½	<i>43166. 52+y</i>	54. 22		
		2½	<i>43220. 74+y</i>									
		2½	39754. 93			<i>3d²(a ³P)4p:</i>	<i>z ²S°</i>	½	<i>43337. 03+y</i>			
		3½	39799. 85									
<i>3d 4s(a ³D)4d</i>	<i>e ⁴G</i>	2½	39861. 25	41. 40 55. 06 70. 52			<i>g ⁴D</i>	½	44598. 80			
		3½	39902. 65									
		4½	39957. 71									
		5½	40028. 23									
<i>3d²(a ¹G)4p</i>	<i>w ²F°</i>	2½	<i>39881. 25</i>	7. 86		<i>4p²(f ³P)3d</i>	<i>h ⁴F</i>	1½	44823. 06	86. 44		
3½	<i>39889. 11</i>	2½	44909. 50					106. 87				
<i>3d 4s(a ³D)4d</i>	<i>f ⁴F</i>	1½	40521. 21	33. 77 49. 04 66. 85				3½	45016. 37	109. 20		
		2½	40554. 98					4½	45125. 57			
		3½	40604. 02					<i>i ⁴F</i>	1½	47898. 95	47. 30	
		4½	40670. 87						2½	47946. 25		
	<i>h ²D</i>	1½	40802. 72	22. 93				3½	48071. 77	251. 81?		
		2½	40825. 65					4½	48323. 58?			
<i>3d 4s(a ³D)4d</i>	<i>f ⁴P</i>	½	41447. 02	27. 86 30. 77				<i>u ²D°</i>	1½	<i>51231. 50+y</i>	98. 04	
		1½	41474. 88						2½	<i>51329. 54+y</i>		
		2½	41505. 65									
<i>3d²(a ³F)5s</i>	<i>g ⁴F</i>	1½	41921. 94	38. 92 54. 71 69. 44		<i>Sc II (a ³D₁)</i>	<i>Limit</i>	-----	52920			
		2½	41960. 86									
		3½	42015. 57									
		4½	42085. 01									

June 1948.

Sc I OBSERVED TERMS*

Config. $1s^2\ 2s^2\ 2p^6\ 3s^2\ 3p^6+$	Observed Terms			
$3d\ 4s^2$	$a\ ^2D$			
$3d^3$	$\left\{ \begin{array}{l} e\ ^4P \\ f\ ^2D: \end{array} \right. \quad e\ ^4F$			
	$ns\ (n \geq 4)$	$np\ (n \geq 4)$		$nd\ (n \geq 3)$
$3d\ 4s(a\ ^3D)nx$	$\left\{ \begin{array}{l} e\ ^4D \\ e\ ^2D \end{array} \right.$	$\begin{array}{lll} z\ ^4P^\circ & z\ ^4D^\circ & z\ ^4F^\circ \\ y\ ^2P^\circ & y\ ^2D^\circ & y\ ^2F^\circ \end{array}$		$\begin{array}{llll} f\ ^4P & f\ ^4D & f\ ^4F & e\ ^4G \\ e\ ^2P & g\ ^2D & e\ ^2F & e\ ^2G \end{array}$
$3d\ 4s(a\ ^1D)nx$		$z\ ^2P^\circ \quad z\ ^2D^\circ \quad z\ ^2F^\circ$		
$3d^2(a\ ^3F)nx$	$\left\{ \begin{array}{l} a, g\ ^4F \\ a\ ^2F \end{array} \right.$	$\begin{array}{lll} & y\ ^4D^\circ & y\ ^4F^\circ \\ & x\ ^2D^\circ & x\ ^2F^\circ \end{array}$		$z\ ^4G^\circ \quad z\ ^2G^\circ$
$3d^2(b\ ^1D)nx$	$b\ ^2D$	$w\ ^2P^\circ \quad w\ ^2D^\circ$		
$3d^2(a\ ^1S)nx$		$x\ ^2P^\circ$		
$3d^2(a\ ^3P)nx$	$\left\{ \begin{array}{l} a\ ^4P \\ a\ ^2P \end{array} \right.$	$\begin{array}{ll} z\ ^4S^\circ & y\ ^4P^\circ \\ z\ ^2S^\circ: & x\ ^4D^\circ \\ & v\ ^2D^\circ: \end{array}$		
$3d^2(a\ ^1G)nx$		$a\ ^2G$		$w\ ^2F^\circ \quad y\ ^2G^\circ \quad z\ ^2H^\circ$
$4p^2(f\ ^3P)nx$				$h\ ^4F$

*For predicted terms in the spectra of the Sc I isoelectronic sequence, see Introduction.

(Ca I sequence; 20 electrons)

 $Z=21$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d 4s {}^3D_1$ $a {}^3D_1$ 104000 cm^{-1}

I. P. 12.89 volts

The analysis is from Russell and Meggers. All the terms are from the 1927 paper, except $y {}^1P^\circ$, which has been taken from the later reference. By analogy with Y II they assign $a {}^1S$ to the configuration $4s^2$ in place of the earlier assignment to $3d^2$.

The singlet and triplet terms are connected by observed intersystem combinations.

The g -values have been generously furnished by Catalán, who has calculated them from the observed Zeeman patterns given in the 1925 reference below.

REFERENCES

- S. Goudsmit, J. van der Mark, and P. Zeeman, Proc. Roy. Acad. Amsterdam **28**, No. 2, 130 (1925). (Z E)
 H. N. Russell and W. F. Meggers, Sci. Papers Bur. Std. **22**, No. 558, 331 (1927). (I P) (T) (C L) (G D)
 W. F. Meggers and H. N. Russell, Bur. Std. J. Research **2**, 761, RP 55 (1929). (T) (C L)
 M. A. Catalán, unpublished material (June 1948). (Z E)

Sc II

Sc II

Config.	Desig.	J	Level	Interval	Obs. g .	Config.	Desig.	J	Level	Interval	Obs. g
$3d({}^2D)4s$	$a {}^3D$	1 2 3	0.00 67.68 177.63	67.68 109.95	0.50 1.17 1.33	$3d({}^2D)5s$	$e {}^3D$	1 2 3	57551.46 57613.94 57743.37	62.48 129.43	
$3d({}^2D)4s$	$a {}^1D$	2	2540.97		1.00	$3d({}^2D)5s$	$e {}^1D$	2	58251.92		
$3d^2$	$a {}^3F_4$	2 3 4	4802.75 4883.42 4987.64	80.67 104.22	0.67 1.07 1.24	$3d({}^2D)4d$	$e {}^1F$	3	59528.22		
$3d^2$	$b {}^1D$	2	10944.51			$3d({}^2D)4d$	$f {}^3D$	1 2 3	59874.79 59929.18 60001.60	54.39 72.42	
$4s^2$	$a {}^1S$	0	11736.35			$3d({}^2D)4d$	$e {}^3G$	3 4 5	60266.95 60348.20 60456.97	81.25 108.77	
$3d^2$	$a {}^3P$	0 1 2	12074.00 12101.45 12154.34	27.45 52.89		$3d({}^2D)4d$	$e {}^1P$	1	60400.02		
$3d^2$	$a {}^1G$	4	14261.40			$3d({}^2D)4d$	$e {}^3S$	1	61071.10		
$3d({}^2D)4p$	$z {}^1D^\circ$	2	26081.32		1.00	$3d({}^2D)4d$	$e {}^3F$	2 3 4	63373.91 63444.43 63527.73	70.52 83.30	
$3d({}^2D)4p$	$z {}^3F^\circ$	2 3 4	27443.65 27602.32 27841.17	158.67 238.85	0.65 1.10 1.25	$3d({}^2D)4d$	$f {}^1D$	2	64366.15		
$3d({}^2D)4p$	$z {}^3D^\circ$	1 2 3	27917.69 28021.21 28161.03	103.52 139.82	0.51 1.16 1.33	$3d({}^2D)4d$	$e {}^3P$	0 1 2	64615.28 64646.08 64705.16	30.80 59.08	
$3d({}^2D)4p$	$z {}^3P^\circ$	0 1 2	29736.22 29742.12 29823.92	5.90 81.80	1.50	$3d({}^2D)4d$	$e {}^1S$	0	64942.79		
$3d({}^2D)4p$	$z {}^1P^\circ$	1	30815.65		1.00	$3d({}^2D)4d$	$e {}^1G$	4	65235.83		
$3d({}^2D)4p$	$z {}^1F^\circ$	3	32349.98		1.00	$4p^2$	$f {}^3P$	0 1 2	76242.40 76359.81 76588.48	117.41 228.67	
$4s({}^2S)4p$	$y {}^3P^\circ$	0 1 2	39001.59 39114.44 39344.90	112.85 230.46		-----					
$4s({}^2S)4p$	$y {}^1P^\circ$	1	55715.52			Sc III (${}^2D_{1/2}$)	Limit	-----	104000		

June 1948.

Sc II OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms		
$3d^3$	{ a^3P b^1D a^3F a^1G		
$4s^3$	a^1S		
$4p^2$	f^3P		
	ns ($n \geq 4$)	np ($n \geq 4$)	nd ($n \geq 4$)
$3d(^2D)nx$	{ a, e^3D a, e^1D	z^3P° z^3D° z^3F° z^1P° z^1D° z^1F°	e^3S e^3P f^3D e^3F e^3G e^1S e^1P f^1D e^1F e^1G
$4s(^2S)nx$	{	y^3P° y^1P°	

*A chart of predicted terms in the spectra of the Ca I isoelectronic sequence is given in the Introduction. Owing to the change in binding energies of the $3d$ and $4s$ electrons along this sequence, the arrangement of the charts of observed and predicted terms is not identical. In Sc II no primes are used to indicate higher limits, and the prefixes $a, b, \dots e, z, y$, replace those indicating the running electron.

Sc III

(K I sequence; 19 electrons)

 $Z=21$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^2 D_{1\frac{1}{2}}$ $3d^2 D_{1\frac{1}{2}}$ 199693.0 cm^{-1}

I. P. 24.75 volts

The early analysis by Gibbs and White was revised and extended by Smith. By analogy with Ti IV, Russell and Lang confirmed Smith's interpretation, added the $5s^2 S$ term, and predicted a number of series members. Their term array has been used for the present compilation, predicted values being entered in brackets. Fourteen lines in the range from 730 Å to 4069 Å have been classified.

REFERENCES

- R. C. Gibbs and H. E. White, Proc. Nat. Acad. Sci. **12**, 598 (1926). (T) (C L)
 S. Smith, Proc. Nat. Acad. Sci. **13**, 65 (1927). (I P) (T) (C L)
 H. N. Russell and R. J. Lang, Astroph. J. **66**, 19; Mt. Wilson Contr. No. 337 (1927). (I P) (T) (C L)

Sc III

Sc III

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3p^6(^1S)3d$	$3d^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	0.0 197.5	197.5	$3p^6(^1S)5d$	$5d^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	[148263] [148283]	20
$3p^6(^1S)4s$	$4s^2S$	$\frac{1}{2}$	25536.7		$3p^6(^1S)6s$	$6s^2S$	$\frac{1}{2}$	[149253]	
$3p^6(^1S)4p$	$4p^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	62102.2 62575.9	473.7	$3p^6(^1S)5f$	$5f^2F^\circ$	{ $2\frac{1}{2}$ $3\frac{1}{2}$ }	[159553]	
$3p^6(^1S)4d$	$4d^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	112254.2 112299.2	45.0	$3p^6(^1S)5g$	$5g^2G$	{ $3\frac{1}{2}$ $4\frac{1}{2}$ }	[160133]	
$3p^6(^1S)5s$	$5s^2S$	$\frac{1}{2}$	114863.8						
$3p^6(^1S)5p$	$5p^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	[128183] [128363]	180	Sc IV (1S_0)	Limit	-----	199693.0	
$3p^6(^1S)4f$	$4f^2F^\circ$	{ $2\frac{1}{2}$ $3\frac{1}{2}$ }	136871.0						

May 1948.

Sc IV

(A I sequence; 18 electrons)

 $Z=21$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 {}^1S_0$ $3p^6 {}^1S_0$ 596300 cm^{-1}

I. P. 73.9 volts

The analysis is seriously incomplete, but four lines between 215 Å and 298 Å have been independently classified, in the first two references quoted below, as combinations with the ground term. The two sets of wavelengths are not completely accordant, but the interpretation is the same in both papers.

The levels given in the table are from Mrs. Beckman's observations, and the limit is from the other paper. Mrs. Beckman's unit, 10^3 cm^{-1} , has here been changed to cm^{-1} , and all values have been rounded off in the last places. The limit may be in error by several hundred cm^{-1} .

For convenience, the Paschen notation has been added by the writer in column one of the table, under the heading "A I". As for A I, the jl -coupling notation in the general form suggested by Racah is here introduced, although LS -designations as indicated in column two under the heading "Authors" are perhaps preferable for the terms thus far identified.

REFERENCES

- A. Beckman, *Bidrag till Kännedomen om Skandiums Spektrum i Yttersta Ultraviolet*, Akademisk Avhandling p. 90 (Almqvist and Wiksells Boktryckeri -A.-B., Uppsala, 1937). (T) (C L)
 P. G. Kruger, S. G. Weissberg and L. W. Phillips, Phys. Rev. **51**, 1090 (1937). (I P) (T) (C L)
 G. Racah, Phys. Rev. **61**, 537 (L) (1942).

Sc IV

A I	Authors	Config.	Desig.	J	Level
$1p_0$	$3p^6 {}^1S$	$3p^6$	$3p^6 {}^1S$	0	0
$1s_4$	$3p^5 4s {}^3P^o$	$3p^5 ({}^2P_{1/2}) 4s$	$4s [1\frac{1}{2}]^o$	$\begin{smallmatrix} 2 \\ 1 \end{smallmatrix}$	335090
$1s_2$	$3p^5 4s {}^1P^o$	$3p^5 ({}^2P_{3/2}) 4s$	$4s' [\frac{1}{2}]^o$	$\begin{smallmatrix} 0 \\ 1 \end{smallmatrix}$	341010
$2s_4$	$3p^5 5s {}^3P^o$	$3p^5 [({}^2P_{1/2}) 5s$	$5s [1\frac{1}{2}]^o$	$\begin{smallmatrix} 2 \\ 1 \end{smallmatrix}$	460430
$2s_2$	$3p^5 5s {}^1P^o$	$3p^5 ({}^2P_{3/2}) 5s$	$5s' [\frac{1}{2}]^o$	$\begin{smallmatrix} 0 \\ 1 \end{smallmatrix}$	463990
		Sc V (${}^2P_{1/2}$)	Limit	-----	596300
		Sc V (${}^2P_{3/2}$)	Limit	-----	600630

May 1948.

Sc v

(ClI sequence; 17 electrons)

 $Z=21$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^5 {}^2P_{1/2}^o$ $3p^5 {}^2P_{1/2}^o$ 741000 cm^{-1}

I. P. 92 volts

Fifteen lines have been classified in the region from 228 Å to 587 Å, as combinations from the ground term. Two independent sets of term values have been published, that are in agreement except for the level $4s {}^4P_{2/2}$, for which Kruger and Phillips give 387508 cm^{-1} ; and the level $4s {}^4P_{1/2}$, which was not found by Mrs. Beckman. All other entries in the table are from the latter list. The unit adopted by Mrs. Beckman, 10^3 cm^{-1} , has here been changed to cm^{-1} .

From isoelectronic sequence data Edlén has estimated the limit given above and entered in brackets in the table.

REFERENCES

- A. Beckman, *Bidrag till Kännedomen om Skandiums Spektrum i Yttersta Ultraviolett*, Akademisk Avhandling p. 86 (Almqvist and Wiksells Boktryckeri -A.-B., Uppsala, 1937). (T) (C L)
 P. G. Kruger and L. W. Phillips, *Phys. Rev.* **51**, 1087 (1937). (T) (C L)
 B. Edlén, *Zeit. Phys.* **104**, 413 (1937). (I P)

Sc v

Config.	Desig.	J	Level	Interval
$3s^2 3p^5$	$3p^5 {}^2P^o$	$1\frac{1}{2}$ $\frac{1}{2}$	0 4328	—4328
$3s 3p^6$	$3p^6 {}^2S$	$\frac{1}{2}$	174412	
$3s^2 3p^4({}^3P)4s$	$4s {}^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	386387 388868 391575?	—2481 —2707
$3s^2 3p^4({}^3P)4s$	$4s {}^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	395503 398447	—2944
$3s^2 3p^4({}^1D)4s$	$4s' {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	410050 410133	—83
$3s^2 3p^4({}^1S)4s$	$4s'' {}^2S$	$\frac{1}{2}$	437512	
-----	-----	-----	-----	-----
Sc VI (3P_2)	Limit	-----	[741000]	

January 1948.

Sc VI

(S I sequence; 16 electrons)

 $Z=21$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^4 {}^3P_2$ $3p^4 {}^3P_2$ 896000 cm^{-1}

I. P. 111.1 volts

The analysis has been done independently by Mrs. Beckman and by Kruger and Pattin with results that are substantially in agreement. The triplet terms are quoted from the former and the singlets from the latter paper. Twenty-nine lines have been classified in the interval between 200 Å and 581 Å. The unit adopted by Mrs. Beckman, 10^3 cm^{-1} , has here been changed to cm^{-1} .

Intersystem combinations connecting the singlet and triplet terms have been observed. The limit is from Edlén, who has extrapolated it from isoelectronic sequence data.

REFERENCES

- A. Beckman, *Bidrag till Kännedomen om Skandiums Spektrum i Yttersta Ultraviolett*, Akademisk Avhandling p. 76 (Almqvist and Wiksells Boktryckeri -A.-B., Uppsala 1937). (T) (C L)
 P. G. Kruger and H. S. Pattin, *Phys. Rev.* **52**, 621 (1937). (T) (C L)
 B. Edlén, *Zeit. Phys.* **104**, 192 (1937). (I P)

Sc VI

Sc VI

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2\ 3p^4$	$3p^4\ ^3P$	2 1 0	0 3352 4453	—3352 —1101	$3s^2\ 3p^3(^2D^\circ)4s$	$4s'\ ^1D^\circ$	2	478354	261 715
					$3s^2\ 3p^3(^2P^\circ)4s$	$4s''\ ^3P^\circ$	0 1 2	491826 492087 492802	
$3s^2\ 3p^4$	$3p^4\ ^1D$	2	21397						
$3s^2\ 3p^4$	$3p^4\ ^1S$	0	49238		$3s^2\ 3p^3(^2P^\circ)4s$	$4s''\ ^1P^\circ$	1	497984	
$3s\ 3p^5$	$3p^5\ ^3P^\circ$	2 1 0	175344 178197 179784	—2853 —1587	-----		-----	-----	
					Sc VII ($^4S_{3/2}$)	<i>Limit</i>	-----	896000	
$3s^2\ 3p^3(^4S^\circ)4s$	$4s\ ^3S^\circ$	1	452070						
$3s^2\ 3p^3(^2D^\circ)4s$	$4s'\ ^3D^\circ$	1 2 3	472400 472563 473001	163 438					

January 1948.

Sc VII

(P I sequence; 15 electrons)

 $Z=21$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^3 {}^4S_{1\frac{1}{2}}^{\circ}$ $3p^3 {}^4S_{1\frac{1}{2}}^{\circ}$ cm^{-1}

I. P.

volts

The analysis is incomplete. Six multiplets have been published by Kruger and Pattin, who derive term intervals but give no term values. Mrs. Beckman has extended their analysis slightly and estimated the relative positions of the doublet and quartet systems of terms from isoelectronic sequence data. Her terms are, in general, quoted, except for the term $3p^4 {}^4P$, which is based on the wavelengths by Kruger and Pattin.

Twenty lines have been classified in the interval between 182 Å and 571 Å. No inter-system combinations have been observed, as indicated by the uncertainty x in the table and brackets around $3p^3 {}^2D_{1\frac{1}{2}}^{\circ}$.

The unit adopted by Mrs. Beckman, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

P. G. Kruger and H. S. Pattin, Phys. Rev. **52**, 624 (1937). (C L)

A. Beckman, *Bidrag till Kännedomen om Skandiums Spektrum i Yttersta Ultraviolet*, Akademisk Avhandling p. 71 (Almqvist and Wiksells Boktryckeri -A.-B., Uppsala, 1937). (T) (C L)

Sc VII

Sc VII

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3s^2 3p^3$	$3p^3 {}^4S^{\circ}$	$1\frac{1}{2}$	0		$3s^2 3p^2({}^3P)3d$	$3d {}^2P$	$1\frac{1}{2}$	$329950 + x$	-3410
$3s^2 3p^3$	$3p^3 {}^2D^{\circ}$	$1\frac{1}{2}$ $2\frac{1}{2}$	$[30000] + x$ $30670 + x$	670	$3s^2 3p^2({}^3P)4s$	$4s {}^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$333360 + x$ 541670 $543600?$ 546490	1930 2890
$3s^2 3p^3$	$3p^3 {}^2P^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$	$49840 + x$ $50740 + x$	900	$3s^2 3p^2({}^3P)4s$	$4s {}^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	$551940 + x$ $555200 + x$	3260
$3s 3p^4$	$3p^4 {}^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	175050 177760 179200	-2710 -1440	$3s^2 3p^2({}^1D)4s$	$4s' {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	$568860 + x$ $568990 + x$	-130

December 1947.

Sc VII OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 +$	Observed Terms	
$3s^2 3p^3$	{ $3p^3 {}^4S^{\circ}$ $3p^3 {}^2P^{\circ}$ $3p^3 {}^2D^{\circ}$ $3p^4 {}^4P$	
$3s 3p^4$		
	$ns (n \geq 4)$	$nd (n \geq 3)$
$3s^2 3p^2({}^3P)nx$	{ $4s {}^4P$ $4s {}^2P$	$3d {}^2P$
$3s^2 3p^2({}^1D)nx'$		
	$4s' {}^2D$	

*For predicted terms in the spectra of the P I isoelectronic sequence, see Introduction.

Sc VIII

(Si I sequence; 14 electrons)

Z=21

Ground state $1s^2 2s^2 2p^6 3s^2 3p^2 {}^3P_0$ $3p^2 {}^3P_0$ 1280000 cm^{-1}

I. P. 159 volts

The analysis is incomplete. The results by Kruger and Phillips are not entirely in agreement with those by Mrs. Beckman. The present list has been compiled from the three references below. One term, $4s {}^1P_1^o$, has been calculated from its combination with $3p^2 {}^1D_2$ as given by Mrs. Beckman. Twenty-five lines are classified in the region between 164 Å and 494 Å. Intersystem combinations connecting the singlet and triplet terms have been observed. The limit, entered in brackets in the table, has been estimated by Phillips.

REFERENCES

- A. Beckman, *Bidrag till Kännedomen om Skandiums Spektrum i Yttersta Ultraviolett*, Akademisk Avhandling p. 65 (Almqvist and Wiksells Boktryckeri -A.-B., Uppsala, 1937). (T) (C L)
 P. G. Kruger and L. W. Phillips, Phys. Rev. **52**, 97 (1937). (T) (C L)
 L. W. Phillips, Phys. Rev. **55**, 708 (1939). (I P) (T) (C L)

Sc VIII

Sc VIII

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2 3p^2$	$3p^2 {}^3P$	0 1 2	0 2280 5510	2280 3230	$3s^2 3p({}^2P^o)3d$	$3d {}^3P^o$	2 1 0	319570 322540 323670	-2970 -1130
$3s^2 3p^2$	$3p^2 {}^1D$	2	25030		$3s^2 3p({}^2P^o)4s$	$4s {}^3P^o$	0 1 2	603540 604610 609180	1070 4570
$3s 3p^3$	$3p^3 {}^3P^o$	$\left\{ \begin{array}{c} 2 \\ 1 \\ 0 \end{array} \right\}$	207760		$3s^2 3p({}^2P^o)4s$	$4s {}^1P^o$	1	614100	
$3s 3p^3$	$3p^3 {}^3S^o$	1	271680						
$3s 3p^3$	$3p^3 {}^1P^o$	1	281520		Sc IX (${}^2P_{3/2}^o$)	Limit	-----	[1280000]	

October 1947.

Sc IX

(Al I sequence; 13 electrons)

 $Z=21$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^2 P_{1/2}^{\circ}$ $3p^2 P_{1/2}^{\circ} 1456000 \text{ cm}^{-1}$

I. P. 180 volts

The analysis is incomplete, but 17 lines have been classified in the region between 119 Å and 537 Å. The listed term values have been calculated by the writer from the combinations given in the references below.

No intersystem combinations have been observed. Using the method of extrapolation suggested by Edlén, the writer has estimated that $3p^2 {}^4P_{1/2}$ is about 141000 cm^{-1} above the ground state. This value is entered in brackets in the table and has been added to all quartet terms. The uncertainty x may well exceed $\pm 1000 \text{ cm}^{-1}$. Similarly, she has extrapolated the value of the limit quoted above and entered in brackets in the table. The uncertainty in this estimate is large owing to the incompleteness of the isoelectronic sequence data.

REFERENCES

- A. Beckman, *Bidrag till Kännedomen om Skandiums Spektrum i Yttersta Ultraviolett*, Akademisk Avhandling p. 59 (Almqvist and Wiksells Boktryckeri -A.-B., Uppsala 1937). (T) (C L)
P. G. Kruger and L. W. Phillips, Phys. Rev. **52**, 97 (1937). (T) (C L)

Sc IX

Sc IX

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3s^2(1S)3p$	$3p^2 P^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$	0 5760	5760	$3s^2(1S)4s$	$4s^2 S$	$\frac{1}{2}$	666260	
$3s 3p^2$	$3p^2 {}^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$[141000] + x$ $143120 + x$ $146280 + x$	2120 3160	$3s 3p(3P^{\circ})4s$	$4s^2 P^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	$819550 + x$ $821490 + x$ $825120 + x$	1940 3630
$3s 3p^2$	$3p^2 {}^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	191760		$3s^2(1S)4d$	$4d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	837210 837450	240
$3s 3p^2$	$3p^2 {}^2S$	$\frac{1}{2}$	240410		$Sc\ x\ (1S_0)$		<i>Limit</i>	$[1456000]$	
$3s 3p^2$	$3p^2 {}^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	255830 259150	3320					
$3s^2(1S)3d$	$3d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	313860 314210	350					

October 1947.

Sc IX OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 +$	Observed Terms	
$3s^2(1S)3p$	$3p^2 P^{\circ}$	
$3s 3p^2$	$\left\{ \begin{array}{l} 3p^2 {}^2S \quad 3p^2 {}^4P \quad 3p^2 {}^2D \\ \quad \quad 3p^2 {}^2P \end{array} \right.$	
	$ns\ (n \geq 4)$	$nd\ (n \geq 3)$
$3s^2(1S)nx$	$4s^2 S$	$3, 4d^2 D$
$3s 3p(3P^{\circ})nx$	$4s^2 P^{\circ}$	

*For predicted terms in the spectra of the Al I isoelectronic sequence, see Introduction.

(Mg I sequence; 12 electrons)

Z=21

Ground state $1s^2 2s^2 2p^6 3s^2 \ ^1S_0$

$3s^2 \ ^1S_0$ 1819530 cm^{-1}

I. P. 225.5 volts

The terms are from the paper by Mrs. Beckman, who has classified 26 lines in the region between 76 Å and 628 Å. She lists one intersystem combination, $3s^2 \ ^1S_0-3p \ ^3P_1^\circ$, and derives absolute term values from the $3d \ ^3D-nf \ ^3F^\circ$ series ($n=4, 5, 6$).

Parker and Phillips have independently found four triplet terms $3p \ ^3P^\circ$, $3d \ ^3D$, $4s \ ^3S$, and $4f \ ^3F^\circ$. Their arrangement of the $3p \ ^3P^\circ-4s \ ^3S$ and $3d \ ^3D-4f \ ^3F^\circ$ multiplets is identical with Mrs. Beckman's but they differ from her in the interpretation of the group of lines ascribed to $3p \ ^3P^\circ-3d \ ^3D$.

Their resulting terms that differ from those listed below (adjusted to the same zero point) are as follows:

Desig.	Level	Desig.	Level
$3d \ ^3D_3$	455510	$4f \ ^3F_4^\circ$	1117757
$\ ^3D_2$	455199	$\ ^3F_3^\circ$	1117710
$\ ^3D_1$	455007	$\ ^3F_2^\circ$	1117689

By extrapolation along the isoelectronic sequence, using the method suggested by Edlén, the writer calculates the limit to be approximately 1818600 cm^{-1} (I. P. 225.4), or about 1000 cm^{-1} lower than that derived by Mrs. Beckman from the $^3F^\circ$ series.

The unit adopted by Mrs. Beckman, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

A. Beckman, *Bidrag till K nnedomen om Skandiums Spektrum i Yttersta Ultraviolet*, Akademisk Avhandling p. 53 (Almqvist and Wiksells Boktryckeri -A.-B., Uppsala, 1937). (I P) (T) (C L) (G D)
W. L. Parker and L. W. Phillips, *Phys. Rev.* **57**, 140 (1940). (T) (C L)

Sc x

Sc x

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3s^2$	$3s^2 \ ^1S$	0	0		$3s(^2S)5p$	$5p \ ^1P^\circ$	1	1309880	
$3s(^2S)3p$	$3p \ ^3P^\circ$	0 1 2	157230 159210 163530	1980 4320	$3s(^2S)5d$	$5d \ ^3D$	1 2 3	1351120	
$3s(^2S)3p$	$3p \ ^1P^\circ$	1	236490		$3s(^2S)5f$	$5f \ ^3F^\circ$	2 3 4	1374440 1374550	110
$3s(^2S)3d$	$3d \ ^3D$	1 2 3	458710 459030 459470	320 440	$3s(^2S)6f$	$6f \ ^3F^\circ$	2 3 4	1511130	
$3s(^2S)4s$	$4s \ ^3S$	1	899250						
$3s(^2S)4p$	$4p \ ^1P^\circ$	1	980600						
$3s(^2S)4d$	$4d \ ^3D$	1 2 3	1074060 1074250 1074530	190 280					
$3s(^2S)4f$	$4f \ ^3F^\circ$	2 3 4	1121400 1121550 1121740	150 190					
					Sc XI ($^2S_{1/2}$)	Limit	-----	1819530	

Sc XI

(Na I sequence; 11 electrons)

 $Z=21$ Ground state $1s^2 2s^2 2p^6 3s^2 S_{1/2}$ $3s^2 S_{1/2}$ 2015030 cm^{-1}

I. P. 249.76 volts

The analysis is by Mrs. Beckman who has extended the work of Edlén and of Kruger and Phillips. She has published 30 classified lines in the interval from 62 Å to 168 Å.

The absolute value of the ground state is extrapolated from isoelectronic sequence data.

The unit adopted by Mrs. Beckman, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

B. Edlén, Zeit. Phys. **100**, 621 (1936). (T) (C L)

A. Beckman, *Bidrag till Kännedom om Skandiums Spektrum i Yttersta Ultraviolet*, Akademisk Avhandling, p. 45 (Almqvist and Wiksells Boktryckeri -A.-B., Uppsala, 1937). (I P) (T) (C L) (G D)

Sc XI

Sc XI

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
3s	3s 2S	$\frac{1}{2}$	0		5f	5f $^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	1482160 1482210	50
3p	3p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	191030 197720	6690	6s	6s 2S	$\frac{1}{2}$	1588790	
3d	3d 2D	$1\frac{1}{2}$ $2\frac{1}{2}$	459410 460030	620	6p	6p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	1609480	
4s	4s 2S	$\frac{1}{2}$	977470		6d	6d 2D	$1\frac{1}{2}$ $2\frac{1}{2}$	1635020	
4p	4p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	1051340 1053870	2530	6f	6f $^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	1645030	
4d	4d 2D	$1\frac{1}{2}$ $2\frac{1}{2}$	1148560 1148830	270	7d	7d 2D	$1\frac{1}{2}$ $2\frac{1}{2}$	1736700	
4f	4f $^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	1182570 1182680	110	7f	7f $^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	1743430	
5s	5s 2S	$\frac{1}{2}$	1382110		-----				
5p	5p $^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	1418280 1419550	1270	Sc XII (1S_0)	Limit	-----	2015030	
5d	5d 2D	$1\frac{1}{2}$ $2\frac{1}{2}$	1464770 1464870	100					

June 1947.

Sc XII

(Ne I sequence; 10 electrons)

Z=21

Ground state $1s^2 2s^2 2p^6\ ^1S_0$

$2p^6\ ^1S_0$ 5539700 cm^{-1}

I. P. 686.6 volts

Edlén and Tyrén have classified five lines in the range 26 Å to 30 Å, as combinations with the ^o ground term. Their absolute term values are based on extrapolation along the Ne I isoelectronic sequence. Their unit, 10³ cm⁻¹, has here been changed to cm⁻¹.

As for Ne I, the *jl*-coupling notation in the general form suggested by Racah is introduced.

REFERENCES

B. Edlén and F. Tyrén, Zeit. Phys. **101**, 210 (1936). (I P) (T) (C L)
G. Racah, Phys. Rev. **61**, 537 (L) (1942).

Sc XII

Authors	Config.	Desig.	<i>J</i>	Level
$2p\ ^1S_0$	$2p^6$	$2p^6\ ^1S$	0	0
$3s\ ^3P_1$	$2p^5(^2P^{\circ}_{1/2})3s$	$3s\ [1\frac{1}{2}]^{\circ}$	$\begin{smallmatrix} 2 \\ 1 \end{smallmatrix}$	3245100
$3s\ ^1P_1$	$2p^5(^2P^{\circ}_{3/2})3s$	$3s'\ [\frac{1}{2}]^{\circ}$	$\begin{smallmatrix} 0 \\ 1 \end{smallmatrix}$	3280800
$3d\ ^3P_1$	$2p^5(^2P^{\circ}_{1/2})3d$	$3d\ [\frac{1}{2}]^{\circ}$	$\begin{smallmatrix} 0 \\ 1 \end{smallmatrix}$	3668400
$3d\ ^1P_1$	"	$3d\ [1\frac{1}{2}]^{\circ}$	1	3714700
$3d\ ^3D_1$	$2p^5(^2P^{\circ}_{3/2})3d$	$3d'\ [1\frac{1}{2}]^{\circ}$	1	3767300
	-----	-----	---	-----
	Sc XIII ($^2P^{\circ}_{1/2}$)	<i>Limit</i>	---	5539700
	Sc XIII ($^2P^{\circ}_{3/2}$)	<i>Limit</i>	---	5577400

April 1947.

TITANIUM

Ti I

22 electrons

 $Z=22$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^2 4s^2 {}^3F_2$ $a {}^3F_2$ 55138 cm^{-1}

I. P. 6.83 volts

The arc spectrum of titanium was one of the first highly complex spectra to be analyzed fairly completely. The detailed analysis published by Russell in 1927 contains 142 terms based on 422 multiplets, and lists 1394 classified lines. Singlet, triplet, and quintet terms are connected by intersystem combinations. This paper, which represents the work of many early contributions as well, by King, Meggers, Kiess, Babcock, and many others, is concluded with the noteworthy statement "The present theories of atomic and spectral structure suffice to give a most satisfactory account, in full and complete detail, of all the features of the very complex spectrum of titanium."

From infrared observations Kiess and Meggers have added the terms $d {}^3P$ and $a {}^5D$. In 1940 Russell added $e {}^3H$ and in 1947 he revised the configuration assignments for inclusion here, as given in column one of the table.

The term values given to three places in the table are from the 1928 paper by Kiess, who calculated them from lines he observed with the interferometer.

Approximate g -values have been calculated by the writer from the Zeeman patterns observed by King and Babcock and quoted by Russell (1927). Most of the observed patterns are unresolved, and consequently the observed g -values differ from the theoretical ones, by a few percent in some cases. They verify the analysis, however, with remarkable consistency. Colons indicate that the observational data are insufficient to give an independent g -value. It is highly desirable to extend this work with the aid of Harrison's unpublished Zeeman observations of titanium.

Both Many and Rohrllich have made theoretical investigations of this spectrum. In the former paper the reality of the term $a {}^1S_0$ at 15166.59 is questioned and this term has been rejected by Russell. Rohrllich has suggested that the ${}^1P^\circ$ term at 39265.80 may be a ${}^1D^\circ$ term. This change has been adopted in the table and the labels of higher ${}^1P^\circ$ and ${}^1D^\circ$ terms changed accordingly, since it has been noted by Russell that this term may equally well be a ${}^1D^\circ$ term. In cases where Rohrllich's configuration assignments differ from those of Russell a colon is entered in column one after the configuration.

REFERENCES

- H. N. Russell, *Astroph. J.* **66**, 347 (1927); *Mt. Wilson Contr. No.* 345 (1927). (I P) (T) (C L) (G D) (Z E)
 C. C. Kiess, *Bur. Std. J. Research* **1**, 77, RP4 (1928). (T) (C L)
 W. F. Meggers and C. C. Kiess, *Bur. Std. J. Research* **9**, 310, RP473 (1932). (T) (C L)
 C. C. Kiess, *J. Research Nat. Bur. Std.* **20**, 35 (RP1062) (1938). (T) (C L)
 H. N. Russell, unpublished material (May 1940, April 1947). (T) (C L)
 A. Many, *Phys. Rev.* **70**, 511 (1946).
 F. Rohrllich, *Phys. Rev.* **74**, 1381 (1948).
 C. E. Moore, unpublished material (June 1948). (Z E)

Ti I

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$3d^2 4s^2$	a^3F	2	0. 000	170. 132	0. 66	$3d^2 4s(a^2F)4p$	z^3F°	2	19323. 003		0. 67
		3	170. 132	216. 741	1. 08			3	19421. 580	98. 577	1. 07
		4	386. 873		1. 25			4	19573. 980	152. 400	1. 26
$3d^3(b^4F)4s$	a^5F	1	6556. 86	41. 97	0. 00	$3d^2 4s(a^2F)4p$	z^3D°	1	19937. 878	68. 171	1. 16
		2	6598. 83	62. 17	0. 99			2	20006. 049	120. 023	1. 34
		3	6661. 00	81. 79	1. 25			3	20126. 072		
		4	6742. 79	100. 21	1. 35	$3d^3(a^2P)4s$	a^1P	1	20062. 98		1. 03
		5	6843. 00		1. 41	$3d^3(b^2D)4s$	b^1D	2	20209. 64		1. 01:
$3d^2 4s^2$	a^1D	2	7255. 29		1. 02	$3d^3(a^2H)4s$	a^1H	5	20795. 65		1. 01
$3d^2 4s^2$	a^3P	0	8436. 630	55. 807		$3d^2 4s(a^2F)4p:$	z^3G°	3	21469. 534	118. 986	0. 75
		1	8492. 437	109. 916	1. 50			4	21588. 520	151. 223	1. 05
		2	8602. 353		1. 49			5	21739. 743		1. 21
$3d^3(b^4F)4s$	b^3F	2	11531. 812	108. 008	0. 67	$3d^2 4s(a^2F)4p$	z^1D°	2	22081. 15		1. 00
		3	11639. 820	137. 000	1. 08	$3d^2 4s(a^2F)4p$	z^1F°	3	22404. 69		1. 00
		4	11776. 820		1. 26	$3d^2 4s(a^2F)4p$	z^1G°	4	24694. 81		0. 97
$3d^2 4s^2$	a^1G	4	12118. 46		0. 98	$3d^2 4s(b^4P)4p$	z^3S°	1	24921. 19		1. 99
$3d^3(a^4P)4s$	a^5P	1	13981. 75	46. 72	2. 50	$3d^2 4s(b^4P)4p$	z^5S°	2	25102. 88		1. 93
		2	14028. 47	77. 21	1. 82	$3d^2 4s(a^4F)4p:$	y^3F°	2	25107. 453	119. 783	1. 06
		3	14105. 68		1. 66			3	25227. 236	161. 109	1. 21?
$3d^3(a^2G)4s$	a^3G	3	15108. 153	48. 650	0. 74			4	25388. 345		
		4	15156. 803	63. 597	1. 06	$3d^2 4s(a^4F)4p:$	y^3D°	1	25317. 842	121. 088	0. 50
		5	15220. 400		1. 21			2	25438. 930	204. 794	1. 17
$3d^2 4s(a^4F)4p$	z^5G°	2	15877. 18	98. 41	0. 39			3	25643. 724		1. 33
		3	15975. 59	130. 49	0. 93	$3d^2 4s(a^4F)4p:$	y^3D°	1	25317. 842	121. 088	0. 50
		4	16106. 08	161. 43	1. 15			2	25438. 930	204. 794	1. 17
		5	16267. 51	191. 20	1. 25			3	25643. 724		1. 33
		6	16458. 71		1. 33	$3d^2 4s(b^4P)4p$	z^3P°	2	25493. 78	—43. 61	1. 47
$3d^2 4s(a^4F)4p$	z^5F°	1	16817. 19	58. 00	0. 00			1	25537. 39		1. 50
		2	16875. 19	86. 23		$3d^2 4s(b^4P)4p:$	y^5D°	0	25605. 03	30. 71	
		3	16961. 42	113. 89	1. 26:			1	25635. 74	64. 21	
		4	17075. 31	140. 13	1. 34			2	25699. 95	97. 65	
		5	17215. 44		1. 42			3	25797. 60	129. 22	1. 52
$3d^3(b^2D)4s$	a^3D	1	17369. 59	54. 52	0. 49			4	25926. 82		
		2	17424. 11	116. 22	1. 17	$3d^3(b^4F)4p$	y^5G°	2	26494. 37	70. 06	0. 34
		3	17540. 33		1. 34			3	26564. 43	92. 98	0. 91
$3d^3(a^2P)4s$	b^3P	0	17995. 75	65. 79				4	26657. 41	115. 57	1. 15
		1	18061. 54	83. 86				5	26772. 98	137. 71	1. 25
		2	18145. 40					6	26910. 69		1. 34
$3d^3(a^2H)4s$	a^3H	4	18037. 28	103. 97	0. 80	$3d^3(b^4F)4p:$	x^3F°	2	26803. 462	89. 484	0. 66
		5	18141. 252	51. 342	1. 02			3	26892. 946	132. 721	1. 06
		6	18192. 594		1. 17			4	27025. 667		1. 23
$3d^3(a^2G)4s$	b^1G	4	18287. 62		1. 02	$3d^3(b^4F)4p$	x^3D°	1	27355. 065	62. 972	0. 51
$3d^2 4s(a^4F)4p$	z^5D°	0	18462. 83	20. 03				2	27418. 037	62. 040	1. 17
		1	18482. 86	42. 21	1. 65?	$3d^3(b^4F)4p:$	y^3G°	3	27480. 077		1. 36
		2	18525. 07	68. 92	1. 50			3	27480. 077		
		3	18593. 99	101. 24	1. 49			3	27499. 033	115. 660	0. 75
		4	18695. 23		1. 51			4	27614. 693	135. 463	1. 05
$3d^3(a^4P)4s$	c^3P	0	18818. 23	7. 66		$3d^2 4s(b^4P)4p$	z^5P°	1	27665. 57	74. 62	
		1	18825. 89	85. 66	1. 54?			2	27740. 19	147. 55	
		2	18911. 55		1. 54:			3	27887. 74		

Ti I—Continued

Ti I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$3d^2 4s(a^2D)4p:$	y^1D°	2	27906.91		0.98	$3d^2 4s(b^2P)4p:$	y^1P°	1	34947.02		
$3d^3(b^4F)4p$	y^5F°	1	28596.45		0.00	$3d^2 4s(b^2P)4p:$	x^1D°	2	35035.11		
		2	28638.82	42.37	1.01	$3d^2 4s(b^2P)4p$	y^3S°	1	35439.43		2.18
		3	28702.70	63.88	1.24	$3d^3(a^2G)4p$	y^3H°	4	35454.099		0.79
		4	28788.39	85.69	1.34			5	35559.662	105.563	1.04
		5	28896.08	107.69	1.40			6	35685.188	125.526	1.17
$3d^4$	a^5D	0	28772.86			$3d^3(a^4P)4p$	w^5D°	0	35503.40		
		1	28791.62	18.76				1	35527.76	24.36	1.51
		2	28828.51	36.89				2	35577.14	49.38	1.53
		3	28882.44	53.93				3	35652.95	75.81	1.46
		4	28952.10	69.66				4	35757.51	104.56	1.46
$3d^2 4s(b^4P)4p:$	w^3D°	1	29661.272		0.51	$3d^2 4s(a^4F)5s$	e^5F	1	35959.07		0.00
		2	29768.686	107.414	1.16			2	36013.57	54.50	1.03?
		3	29912.292	143.606	1.34			3	36096.47	82.90	1.24
$3d^3(b^2F)4s$	a^1F	3	29818.31					4	36208.92	112.45	1.34
$3d^3(b^4F)4p$	x^5D°	0	29829.16					5	36351.43	142.51	1.42
		1	29855.26	26.10	1.46	$3d^2 4s(b^2G)4p:$	y^1G°	4	36000.25		1.00
		2	29907.29	52.03	1.50	$3d^4$	b^3G	3	36065.75		
		3	29986.24	78.95	1.49			4	36132.21	66.46	
		4	30060.34	74.10	1.49			5	36200.94	68.73	
$3d^2 4s(a^4F)4p:$	x^3G°	3	29914.773			$3d^3(a^4P)4p$	y^5P°	1	36298.43		2.47
		4	29971.106	56.333				2	36340.67	42.24	1.81
		5	30039.246	68.140	1.19	$3d^3(b^2D)4p:$	w^3P°	0	36414.58	73.91	1.66
$3d^2 4s(a^2D)4p:$	v^3D°	1	31184.089		0.51			1	37090.65		
		2	31190.663	6.574	1.17			2	37173.03	82.38	1.53
		3	31206.014	15.351	1.34	$3d^3(a^4P)4p$	y^5S°	2	37325.47	152.44	1.48
$3d^2 4s(b^2G)4p:$	w^3G°	3	31373.862		0.75	$3d^2 4s(a^4F)5s$	e^3F	2	37359.13		1.99
		4	31489.486	115.624	1.05			3	37538.71		
		5	31628.698	139.212	1.19			4	37659.97	121.26	1.11
$3d^2 4s(a^2D)4p:$	y^3P°	0	31685.90			$3d^3(a^2G)4p$	v^3G°	3	37824.69	164.72	1.27
		1	31725.75	39.85				4	37851.91		
		2	31805.94	80.19	1.47	$3d^2 4s(b^2P)4p:$	u^3D°	1	37976.78	124.87	0.53
$3d^2 4s(b^2G)4p$	z^3H°	4	31830.016		0.80			2	38159.71	182.93	1.14:
		5	31914.304	84.288	1.04	$3d^3(a^2P)4p$	z^1S°	0	38200.94		
		6	32013.555	99.251	1.17	$3d^3(a^2G)4p$	t^3F°	2	38451.29		0.66
$3d^2 4s(a^2D)4p$	y^1F°	3	32857.76		0.99?	$3d^2 4s(b^2G)4p:$	x^1F°	3	37622.63		0.94
$3d^2 4s(b^2P)4p:$	x^3P°	0	33085.14			$3d^3(b^2D)4p:$	u^3F°	2	37654.77		0.65
		1	33090.55	5.41	1.46			3	37743.96	89.19	1.08
		2	33114.49	23.94	1.46	$3d^2 4s(a^2D)4p:$	w^3F°	2	33655.898		
$3d^2 4s(a^2D)4p:$	w^3F°	2	33655.898		0.66			3	33680.162		
		3	33680.162	24.264	1.09	$3d^2 4s(b^2P)4p:$	u^3D°	1	33700.897	20.735	1.26
		4	33700.897	20.735	1.26	$3d^3(a^2P)4p$	z^1S°	0	33660.73		0.94?
$3d^2 4s(a^2D)4p:$	z^1P°	1	33660.73			$3d^3(a^2G)4p$	t^3F°	2	33980.685		0.63
$3d^2 4s(b^2G)4p$	v^3F°	2	33980.685		0.63			3	34078.612	97.927	1.10
		3	34078.612	97.927	1.10			4	34205.001	126.389	1.23
		4	34205.001	126.389	1.23	$3d^4$	d^3P	0	34170.95		
$3d^4$	d^3P	1	34327.96	157.01				1	34327.96	207.08	
		2	34535.04	207.08		$3d^3(a^2H)4p$	z^3I°	5	38572.75		0.81
$3d^2 4s(b^2G)4p:$	z^1H°	5	34700.31		1.02			6	38669.03	96.28	1.02
						$3d^3(b^2D)4p:$	t^3D°	1	38779.97	110.94	1.15
								2	38654.23		0.54:
								3	38699.95	45.72	
									38764.96	65.01	1.32

Ti I—Continued

Ti I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$3d^3(a^2G)4p:$	x^1G°	4	38959.53		1.02		w^3H°	4	41780.95		
$3d^3(b^2D)4p$	x^1P°	1	39078.00					5	41895.15	114.20	
$3d^3(b^4F)5s$	f^5F	1	39107.25	42.01		$3d^24s(a^4F)5p$	v^5D°	0	41822.99	31.02	
		2	39149.26	65.12				1	41854.01	52.60	
		3	39214.38	87.98				2	41906.61	79.32	
		4	39302.36	110.42				3	41985.93	106.59	
		5	39412.78					4	42092.52		
$3d^3(a^2H)4p$	x^3H°	4	39115.99	36.15	0.88?	$3d^24s(a^4F)4d$	e^5H	3	41823.19	93.86	
		5	39152.14	46.25	1.02			4	41917.05	100.96	1.15
		6	39198.39		1.18			5	42018.01	105.76	1.22
$3d^3(a^2P)4p$	w^1D°	2	39265.80		1.06:			6	42123.77	81.82	1.28
$3d^3(b^4F)5s$	f^3F	2	39526.89	114.09		$3d^24s(a^4F)4d$	e^5D	0	41871.56	29.80	
		3	39640.98	144.96				1	41901.36	57.15	
		4	39785.94					2	41958.51	94.21	
$3d^3(a^4P)4p$	s^3D°	1	39662.15	23.95	0.52			3	42052.72	131.94	
		2	39686.10	29.41				4	42184.66		
		3	39715.51		1.31:	$3d^24s(a^4F)4d$	g^3F	2	41871.87	116.52	
$3d^3(b^2D)4p$	w^1F°	3	40303.04		1.05:			3	41988.39	118.67	
$3d^3(a^2H)4p$	z^1I°	6	40319.80		1.03	$3d^3(a^2P)4p:$	u^3P°	2	41928.59	-15.36	
$3d^3(a^4P)4p:$	v^3P°	0	40369.76	14.82				1	41943.95	-15.51	
		1	40384.58	82.46				0	41959.46		
		2	40467.04				q^3D°	1	42146.39	60.49	
$3d^3(a^2P)4p$	r^3D°	1	40556.07	114.53	0.49			2	42206.88	104.43	1.32
		2	40670.60	173.59				3	42311.31		
		3	40844.19				p^3D°	1	42193.94	75.79	
$3d^3(a^2P)4p$	x^3S°	1	40844.19					2	42269.73	106.98	
	w^1G°	4	40883.30		0.95:	$3d^24s(a^4F)4d$	e^5P	1	42611.58	112.53	
$3d^3(a^2G)4p:$	y^1H°	5	41039.93		1.03			2	42724.11	134.79	1.64
$3d^24s(a^2F)5s$	e^1F	3	41087.31		1.01	$3d^24s(a^2S)4p:$	w^1P°	1	42927.55		1.00:
$3d^3(a^2H)4p$	u^3G°	3	41169.82	85.62	0.73	$3d^24s(a^4F)4d$	g^5F	1	43034.08	46.84	
		4	41255.44	86.18	1.03			2	43080.92	67.23	
		5	41341.62		1.19			3	43148.15	83.84	
$3d^24s(a^4F)4d$	e^3G	3	41194.42	174.44				4	43231.99	98.08	
		4	41368.86	112.27				5	43330.07		
		5	41481.13				r^3F°	2	43467.55	115.59	
	s^3F°	2	41337.43	120.19	0.66			3	43583.14	161.41	
		3	41457.62	166.51	1.09	$3d^3(a^2H)4p:$	v^1G°	4	43674.31		0.95
		4	41624.13		1.24						
$3d^24s(a^4F)4d$	e^3H	4	41515.09	41.24		$3d^3(b^2D)4p$	v^1D°	2	43710.28		0.98:
		5	41556.33	58.69							
		6	41615.02			$3d^3(b^4F)4d$	f^5H	3	43843.82	57.92	0.91
$3d^3(a^2G)4p:$	v^1F°	3	41585.24					4	43901.74	69.81	1.11
$3d^24s(a^4F)4d$	e^5G	2	41714.35	43.12				5	43971.55	79.82	1.21
		3	41757.47	61.23	1.12			6	44051.37	83.28	1.29
		4	41818.70	84.78	1.24			7	44134.65		
		5	41903.48	115.74	1.34		o^3D°	1	43975.62	103.77	1.18?
		6	42019.22					2	44079.39	153.76	
								3	44233.15		

Ti I—Continued

Ti I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
	<i>t</i> ³ G°	3				3d ² 4s(a ² F)4d	<i>i</i> ³ F	2			
		4	44162.44					3	47038.16		
		5	44375.57	213.13				4	47194.68	156.52	
3d ³ (a ² H)4p	<i>x</i> ¹ H°	5	44163.24		1.03	3d ³ (b ⁴ F)6s	<i>i</i> ⁵ F	1			
3d ³ (b ⁴ F)4d	<i>f</i> ⁵ D	0						2			
		1						3			
		2						4			
		3	44254.39					5	47777.32		
		4	44381.17	126.78		3d ² 4s(a ⁴ F)5d	<i>g</i> ⁵ H	3	47840.62		
3d ² 4s(a ² D)5s	<i>e</i> ¹ D	2	44581.16					4	47913.61	72.99	
								5	47994.32	80.71	
								6	48106.83	112.51	
								7	48262.83	156.00	
	<i>q</i> ³ F°	2	44825.26			3d ² 4s(a ⁴ F)5d	<i>h</i> ⁵ G	2	47870.61		
		3	44923.00	97.74				3	47936.79	66.18	
		4	45041.02	118.02				4	48018.08	81.29	
3d ³ (a ⁴ P)4p	<i>w</i> ³ S°	1	44857.89					5	48119.47	101.39	
								6	48233.47	114.00	
	<i>n</i> ³ D°	1	44966.36			3d ² 4p ²	<i>j</i> ⁵ F	1	48058.85		
		2	45063.94	97.58				2	48107.42	48.57	
		3	45206.34	142.40				3	48208.87	101.45	
3d ² 4s(a ² S)4p.	<i>t</i> ³ P°	0	45040.70					4	48328.81	119.94	
		1	45090.73	50.03				5	48462.11	133.30	
		2	45178.06	87.33		3d ² 4s(a ⁴ F)5d	<i>g</i> ⁵ D	0			
3d ² 4s(a ² F)4d	<i>e</i> ¹ H	5	45485.35					1			
3d ³ (b ⁴ F)4d?	<i>f</i> ⁵ G	2						2			
		3	45689.89					3	48059.82		
		4	45711.28	21.39				4	48186.11	126.29	
		5	45756.45?	45.17		3d ³ (b ² F)4p.	<i>u</i> ¹ F°	3	48365.09		
		6	45904.73	148.28		3d ² 4s(a ⁴ F)5d	<i>k</i> ⁵ F	1			
3d ² 4s(a ² F)4d	<i>f</i> ³ H	4	45721.89		0.80			2	48519.21		
		5	45832.50	110.61	1.03			3	48588.28	69.07	
		6	45960.39	127.89	1.17			4	48672.66	84.38	
3d ² 4s(a ⁴ F)6s	<i>h</i> ⁵ F	1	45764.71					5	48771.73	99.07	
		2	45813.01	48.30		3d ² 4p ²	<i>e</i> ³ D	1	48724.83		
		3	45893.26	80.25				2	48724.34	-0.49	
		4	46007.62	114.36				3	48839.74	115.40	
		5	46157.76	150.14		3d ² 4p ²	<i>h</i> ⁵ D	0	48802.32		
3d ² 4s(a ² F)4d	<i>e</i> ¹ G	4	46068.04					1	48859.51	57.19	
3d ² 4s(b ⁴ P)5s	<i>e</i> ³ P	0						2	48915.07	55.56	
		1						3	49024.43	109.36	
		2	46244.60					4	49036.46	12.03	
3d ³ (b ² F)4p.	<i>u</i> ¹ G°	4	46257.67		0.95		<i>f</i> ³ D	1			
3d ² 4s(a ⁴ F)6s	<i>h</i> ³ F	2						2	49571.69		
		3						3	49619.72	48.03	
		4	46530.45				<i>f</i> ¹ D	2	50128.08		
3d ² 4s(a ² F)4d	<i>f</i> ¹ F	3	46650.26				<i>f</i> ¹ G	4	52125.98		
3d ² 4p ²	<i>g</i> ⁵ G	2	46943.91				<i>e</i> ¹ P	1	53663.32		
		3	47030.28	86.37							
		4	47139.86	109.58							
		5	47280.69	140.83							
		6	47446.84	166.15							
						Ti II (a ⁴ F _{1/2})	<i>Limit</i>	-----	55138		

Ti 1 OBSERVED TERMS

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms									
$3d^2 4s^2$	{	a^1P		a^3F		a^1G				
$3d^4$	{	d^3P		a^5D				b^3G		
$3d^2 4p^2$	{		h^5D e^3D		i^5F		g^5G			
	$ns (n \geq 4)$					$np (n \geq 4)$				
$3d^2 4s(a^4F)nx$	{		e, h^5F e, h^3F				z, v^5D° y^3D°	z^5F° y^3F°	z^5G° x^3G°	
$3d^3(b^4F)nx$	{		a, f, i^5F b, f^3F				x^5D° x^3D°	y^5F° x^3F°	y^5G° y^3G°	
$3d^2 4s(a^2F)nx$	{			e^1F			z^3D° z^1D°	z^3F° z^1F°	z^3G° z^1G°	
$3d^2 4s(a^2D)nx$	{		e^1D				y^3P° z^1P°	v^3D° y^1D°	w^3F° y^1F°	
$3d^3(a^2G)nx$	{				a^3G b^1G				t^3F° v^1F°	v^3G° x^1G°
$3d^3(a^4P)nx$	{	a^5P c^3P					y^5S° w^3S°	y^5P° v^3P°	w^5D° s^3D°	
$3d^3(a^2P)nx$	{	b^3P a^1P					x^3S° z^1S°	u^3P°	r^3D° w^1D°	
$3d^2 4s(b^4P)nx$	{	e^3P					z^5S° z^3S°	z^5P° z^3P°	y^5D° w^3D°	
$3d^3(b^2D)nx$	{		a^3D b^1D					w^3P° x^1P°	t^3D° u^1D°	u^3F° w^1F°
$3d^3(a^2H)nx$	{				a^3H a^1H					u^3G° v^1G°
$3d^2 4s(b^2G)nx$	{									x^3H° x^1H°
$3d^2 4s(b^2P)nx$	{									z^3I° z^1I°
$3d^2 4s(b^2P)nx$	{									u^3G° v^1G°
$3d^2 4s(b^2P)nx$	{									x^3H° x^1H°
$3d^3(b^2F)nx$				a^1F						z^3H° z^1H°
$3d^2 4s(a^2S)nx$	{									u^3F° y^1F°
	$nd (n \geq 4)$									
$3d^2 4s(a^4F)nx$	{	e^5P	e, g^5D	g, k^5F g^3F	e, h^5G e^3G	e, g^5H e^3H				
$3d^3(b^4F)nx$			f^5D		$f^5G?$	f^5H				
$3d^2 4s(a^2F)nx$	{			i^3F f^1F	e^1G	f^3H e^1H				

*For predicted terms in the spectra of the Ti I isoelectronic sequence, see Introduction.

Ti II

(Sc I sequence; 21 electrons)

 $Z=22$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^2 4s \ ^4F_{1/2}$ $a \ ^4F_{1/2} \ 110000 \text{ cm}^{-1}$

I. P. 13.63 volts

This spectrum has been analyzed by Russell. His detailed analysis published in 1927 contains 50 terms derived from 164 multiplets, and includes 529 classified lines. The doublet and quartet terms are connected by observed intersystem combinations.

The configuration assignments are of considerable theoretical interest, as indicated, for example, in the references to the papers by Ufford, Racah, and Many listed below. Many has interchanged the configurations given by Russell to the two low 4F terms. From a detailed study of the series relations Russell has recently shown conclusively that his original assignments were correct, namely that the lower term ($a \ ^4F$) has the configuration $3d^2 (a \ ^3F)4s$ and that the higher one ($b \ ^4F$) should be ascribed to $3d^3$.

Approximate g -values have been determined by Catalán from the Zeeman patterns observed by King and Babcock and quoted by Russell (1927). Very few patterns have been resolved and consequently the observed g -values differ from the theoretical ones by a few percent in some cases. Colons indicate that LS -coupling has been assumed and a theoretical g -value introduced in order to utilize the observed data. It is highly desirable to extend this work with the aid of Harrison's unpublished Zeeman observations of titanium.

REFERENCES

- H. N. Russell, *Astroph. J.* **66**, 283 (1927); Mt. Wilson Contr. No. 344 (1927). (I P) (T) (C L) (G D) (Z E)
 C. W. Ufford, *Phys. Rev.* **44**, 732 (1933).
 G. Racah, *Phys. Rev.* **62**, 438 (1942).
 A. Many, *Phys. Rev.* **70**, 511 (1946).
 H. N. Russell, *Phys. Rev.* **74**, 689 (1948).
 M. A. Catalán, unpublished material (June 1948). (Z E)

Ti II

Ti II

Config.	Desig.	J	Level	Interval	Obs. g	Config.	Desig.	J	Level	Interval	Obs. g
$3d^2(a \ ^3F)4s$	$a \ ^4F$	$1\frac{1}{2}$	0. 00			$3d^3$	$a \ ^4P$	$\frac{1}{2}$	9363. 71		2. 63
		$2\frac{1}{2}$	93. 94	93. 94				$1\frac{1}{2}$	9395. 76	32. 05	1. 74
		$3\frac{1}{2}$	225. 47	131. 53				$2\frac{1}{2}$	9518. 05	122. 29	1. 60:
		$4\frac{1}{2}$	393. 22	167. 75							
$3d^3$	$b \ ^4F$	$1\frac{1}{2}$	907. 96			$3d^3$	$a \ ^2P$	$\frac{1}{2}$	9850. 90		0. 66
		$2\frac{1}{2}$	983. 80	75. 84				$1\frac{1}{2}$	9975. 92	125. 02	1. 33
		$3\frac{1}{2}$	1087. 21	103. 41				$\frac{3}{2}$	9872. 87		2. 60
		$4\frac{1}{2}$	1215. 58	128. 37				$1\frac{1}{2}$	9930. 74	57. 87	1. 72:
$3d^2(a \ ^3F)4s$	$a \ ^2F$	$2\frac{1}{2}$	4628. 61		0. 86:	$3d^2(a \ ^3P)4s$	$b \ ^4P$	$2\frac{1}{2}$	10024. 74	94. 00	1. 60:
		$3\frac{1}{2}$	4897. 60	268. 99	1. 14:						
$3d^2(a \ ^1D)4s$	$a \ ^2D$	$1\frac{1}{2}$	8710. 47		0. 80	$3d^3$	$b \ ^2D$	$1\frac{1}{2}$	12628. 77		0. 80:
		$2\frac{1}{2}$	8744. 27	33. 80	1. 20:			$2\frac{1}{2}$	12758. 15	129. 38	1. 20:
$3d^3$	$a \ ^2G$	$3\frac{1}{2}$	8997. 69		0. 89:	$3d^3$	$a \ ^2H$	$4\frac{1}{2}$	12676. 99		0. 91:
		$4\frac{1}{2}$	9118. 15	120. 46	1. 11:			$5\frac{1}{2}$	12774. 81	97. 82	1. 09:
$3d^2(a \ ^1G)4s$	$b \ ^2G$	$4\frac{1}{2}$	15257. 53			$3d^2(a \ ^1G)4s$	$b \ ^2G$	$4\frac{1}{2}$	15257. 53		1. 11:
		$3\frac{1}{2}$	15265. 60					$3\frac{1}{2}$	15265. 60	-8. 07	0. 89:

Ti II—Continued

Ti II—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$3d^2(a^3P)4s$	b^2P	$\frac{1}{2}$ $1\frac{1}{2}$	16515. 79 16625. 25	109. 46	0. 66 1. 33	$3d^2(a^1G)4p$	x^2F°	$3\frac{1}{2}$ $2\frac{1}{2}$	47466. 80 47625. 17	-158. 37	1. 14: 0. 86:
$3d^3$	b^2F	$3\frac{1}{2}$ $2\frac{1}{2}$	20891. 88 20951. 77	-59. 89	1. 14: 0. 86:	$3d\ 4s(a^3D)4p$	x^4D°	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	52329. 78 52458. 98 52471. 48 52631. 07	129. 20 12. 50 159. 59	
$3d^2(a^1S)4s$	a^2S	$\frac{1}{2}$	21338. 00:			$3d\ 4s(a^3D)4p$	x^2P°	$\frac{1}{2}$ $1\frac{1}{2}$	53121. 48 53128. 17	6. 69	
$3d\ 4s^2$	c^2D	$1\frac{1}{2}$ $2\frac{1}{2}$	24961. 34 25193. 04	231. 70	0. 80: 1. 20:	$3d\ 4s(a^3D)4p$	w^2D°	$2\frac{1}{2}$ $1\frac{1}{2}$	53554. 90 53596. 70	-41. 80	
$3d^2(a^3F)4p$	z^4G°	$2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$ $5\frac{1}{2}$	29544. 37 29734. 45 29968. 08 30240. 68	190. 08 233. 63 272. 60	0. 57: 0. 98:	$3d\ 4s(a^3D)4p$	y^4P°	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	56223. 13 56249. 11 56325. 94	25. 98 76. 83	
$3d^2(a^3F)4p$	z^4F°	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	30836. 52 30958. 70 31113. 61 31300. 92	122. 18 154. 91 187. 31	0. 40: 1. 03: 1. 24:	$3d\ 4s(a^3D)4p$	w^2F°	$2\frac{1}{2}$ $3\frac{1}{2}$	59321. 79 59467. 81	146. 02	
$3d^2(a^3F)4p$	z^2F°	$2\frac{1}{2}$ $3\frac{1}{2}$	31207. 44 31490. 82	283. 38	0. 86: 1. 14:	$3d^2(a^3F)5s$	e^4F	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	62180. 02 62271. 25 62409. 58 62594. 27	91. 23 138. 33 184. 69	
$3d^2(a^3F)4p$	z^2D°	$1\frac{1}{2}$ $2\frac{1}{2}$	31756. 50 32025. 50	269. 00	0. 92 1. 20	$3d^2(a^3F)5s$	e^2F	$2\frac{1}{2}$ $3\frac{1}{2}$	63168. 23 63444. 76	276. 53	
$3d^2(a^3F)4p$	z^4D°	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	32532. 38 32602. 51 32697. 94 32767. 02	70. 13 95. 43 69. 08	0. 00 1. 20 1. 37 1. 43:	$3d^2(a^3F)4d$	e^4G	$2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$ $5\frac{1}{2}$	64884. 65 64977. 57 65094. 29 65241. 60	92. 92 116. 72 147. 31	
$3d^2(a^3F)4p$	z^2G°	$3\frac{1}{2}$ $4\frac{1}{2}$	34543. 36 34748. 50	205. 14	0. 89: 1. 11:	$3d^2(a^3F)4d$	e^4H	$3\frac{1}{2}$ $4\frac{1}{2}$ $5\frac{1}{2}$ $6\frac{1}{2}$	65184. 72 65307. 45 65445. 85 65589. 10	122. 73 138. 40 143. 25	
$3d^2(a^3P)4p$	z^2S°	$\frac{1}{2}$	37430. 55		2. 09	$3d^2(a^3F)4d$	f^2F	$2\frac{1}{2}$ $3\frac{1}{2}$	65312. 71 65458. 65	145. 94	
$3d^2(a^1D)4p$	y^2D°	$1\frac{1}{2}$ $2\frac{1}{2}$	39233. 44 39476. 87	243. 43	0. 80: 1. 20:	$3d^2(a^3F)4d$	e^4D	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	66767. 43? 66816. 49 66937. 70 66996. 67	49. 06 121. 21 58. 97	
$3d^3(a^1D)4p$	z^2P°	$1\frac{1}{2}$ $\frac{1}{2}$	39602. 90 39674. 64	-71. 74	1. 21 0. 67:	$3d^2(a^3F)4d$	e^2G	$3\frac{1}{2}$ $4\frac{1}{2}$	67604. 20 67820. 87	216. 67	
$3d^2(a^1D)4p$	y^2F°	$2\frac{1}{2}$ $3\frac{1}{2}$	39926. 83 40074. 71	147. 88	0. 86: 1. 14:	$3d^2(a^3F)4d$	e^2H	$4\frac{1}{2}$ $5\frac{1}{2}$	68328. 95 68582. 34	253. 39	
$3d^2(a^3P)4p$	z^4S°	$1\frac{1}{2}$	40027. 28			$3d^2(a^3F)4d$	f^4F	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	68767. 66 68845. 14 68950. 39 69081. 35	77. 48 105. 25 130. 96	
$3d^2(a^3P)4p$	y^4D°	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	40330. 25 40425. 80 40581. 80 40798. 37	95. 55 156. 00 216. 57		$3d\ 4s(b^1D)4p$	v^2D°	$1\frac{1}{2}$ $2\frac{1}{2}$	69327. 32 69622. 15	294. 83	
$3d^2(a^3P)4p$	z^4P°	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	41996. 74 42068. 85 42208. 84	72. 11 139. 99		$3d\ 4s(b^1D)4p$	v^2F°	$2\frac{1}{2}$ $3\frac{1}{2}$	70606. 35 70893. 00	286. 65	
$3d^2(a^1G)4p$	y^2G°	$3\frac{1}{2}$ $4\frac{1}{2}$	43740. 77 43780. 99	40. 22	0. 89: 1. 11:						
$3d^2(a^3P)4p$	x^2D°	$2\frac{1}{2}$ $1\frac{1}{2}$	44902. 42 44914. 80	-12. 38	1. 20: 0. 80:						
$3d^2(a^3P)4p$	y^2P°	$\frac{1}{2}$ $1\frac{1}{2}$	45472. 89 45548. 90	76. 01	0. 66: 1. 33:						
$3d^2(a^1G)4p$	z^2H°	$4\frac{1}{2}$ $5\frac{1}{2}$	45673. 75 45908. 56	234. 81							
						Ti III (a^3F_2)	Limit	-----	110000		

Ti II OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms		
$3d^3$	{ a^4P a^2P b^2D b^4F b^2F a^2G a^2H		
$3d 4s^2$	c^2D		
	$ns (n \geq 4)$	$np (n \geq 4)$	$nd (n \geq 4)$
$3d^2(a^3F)nx$	{ a, e^4F a, e^2F	z^4D° z^4F° z^4G° z^2D° z^2F° z^2G°	e^4D f^4F e^4G e^4H f^2F e^2G e^2H
$3d^2(a^1D)nx$	a^2D	z^2P° y^2D° y^2F°	
$3d^2(a^3P)nx$	{ b^4P b^2P	z^4S° z^4P° y^4D° z^2S° y^2P° x^2D°	
$3d^2(a^1S)nx$	a^2S		
$3d^2(a^1G)nx$	b^2G	x^2F° y^2G° z^2H°	
$3d 4s(a^3D)nx$	{	y^4P° x^4D° x^2P° w^2D° w^2F°	
$3d 4s(b^1D)nx$		v^2D° v^2F°	

*A chart of predicted terms in the spectra of the ScI isoelectronic sequence is given in the Introduction. Owing to the difference in binding energies of the $3d$ and $4s$ electrons along this sequence, the charts of observed and predicted terms are not similarly arranged for Ti II.

Ti III

(Ca I sequence; 20 electrons)

 $Z=22$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^2 {}^3F_2$ a^3F_2 227000 cm^{-1}

I. P. 28.14 volts

The analysis is by Russell and Lang who have classified 84 lines in the interval between 1002 Å and 2984 Å.

The singlet and triplet terms are connected by observed intersystem combinations.

REFERENCE

H. N. Russell and R. J. Lang, *Astroph. J.* **66**, 25 ; Mt. Wilson Contr. No. 337 (1927). (I P) (T) (C L)

Ti III

Ti III

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3d^2$	a^3F	2	0.0	183.7 238.2	$3d(^2D)4p$	z^1F°	3	83116.58	
		3	183.7		$3d(^2D)4p$	z^1P°	1	83795.70	
		4	421.9		$3d(^2D)4d$	e^3G	3	129096.3	
$3d^2$	a^1D	2	8472.6				4	129256.0	159.7
$3d^2$	a^3P	0	10536.4	67.1 117.6			5	129472.6	216.6
		1	10603.5		$3d(^2D)4d$	e^3D	1		
		2	10721.1				2	129873.9	
$3d^2$	a^1S	0	14052.7?				3	130019.5	145.6
$3d^2$	a^1G	4	14398.5		$3d(^2D)4d$	e^3S	2	132854.6	
					$3d(^2D)4d$	e^3F	2	133067.2	
							3	133209.7	142.5
$3d(^2D)4s$	a^3D	1	38063.50	134.48 227.21			4	133373.7	164.0
		2	38197.98		$3d(^2D)4d$	e^3P	0	135543.8	
		3	38425.19				1	135602.4	58.6
$3d(^2D)4s$	b^1D	2	41703.65				2	135724.1	121.7
$3d(^2D)4p$	z^1D°	2	75197.43		$4s(^2S)4p$	y^3P°	0	137262	
							1	137490	228
							2	137971	481
$3d(^2D)4p$	z^3D°	1	76999.70	166.95 257.55					
		2	77166.65						
		3	77424.20						
$3d(^2D)4p$	z^3F°	2	77421.48	324.70 412.53					
		3	77746.18						
		4	78158.71						
$3d(^2D)4p$	z^3P°	0	80943.95	-5.93 85.58					
		1	80938.02						
		2	81023.60						
					Ti IV (2D) _{13/2}	Limit		227000	

June 1948.

Ti III OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms		
$3d^2$	{ a^1S a^3P a^1D a^3F a^1G		
	ns ($n \geq 4$)	np ($n \geq 4$)	nd ($n \geq 4$)
	{ a^3D b^1D	z^3P° z^3D° z^3F° z^1P° z^1D° z^1F°	e^3S e^3P e^3D e^3F e^3G
$3d(^2D)nx$			
$4s(^2S)nx$		y^3P°	

*A chart of predicted terms in the spectra of the Ca I isoelectronic sequence is given in the Introduction. Owing to the change in binding energies of the $3d$ and $4s$ electrons along this sequence, the arrangement of the charts of observed and predicted terms is not identical. In Ti III no primes are used to indicate higher limits, and the prefixes $a, b \dots e, z, y$ replace those indicating the running electron.

Ti IV

(K I sequence; 19 electrons)

Z=22

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^2 D_{1/2}$

$3d^2 D_{1/2}$ 348817.8 cm⁻¹

I. P. 43.24 volts

The analysis is from Russell and Lang, who have revised and extended the early work of Gibbs and White. Thirty-one lines have been classified in the range between 423 Å and 5492 Å.

REFERENCES

R. C. Gibbs and H. E. White, Proc. Nat. Acad. Sci. **12**, 598 (1926). (T) (C L)
H. N. Russell and R. J. Lang, Astroph. J. **66**, 15 (1927); Mt. Wilson Contr. No. 337 (1927). (I P) (T) (C L)

Ti IV

Ti IV

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3p^6(^1S)3d$	$3d^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	0. 0 384. 3	384. 3	$3p^6(^1S)5d$	$5d^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	258827. 2 258866. 7	39. 5
$3p^6(^1S)4s$	$4s^2S$	$\frac{1}{2}$	80378. 6		$3p^6(^1S)6s$	$6s^2S$	$\frac{1}{2}$	265835. 8	
$3p^6(^1S)4p$	$4p^2P^o$	$\frac{1}{2}$ $1\frac{1}{2}$	127912. 5 128730. 9	818. 4	$3p^6(^1S)5g$	$5g^2G$	$\left\{ \begin{matrix} 3\frac{1}{2} \\ 4\frac{1}{2} \end{matrix} \right\}$	278501. 1	
$3p^6(^1S)4d$	$4d^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	196794. 8 196880. 5	85. 7	$3p^6(^1S)6h$	$6h^2H^o$	$\left\{ \begin{matrix} 4\frac{1}{2} \\ 5\frac{1}{2} \end{matrix} \right\}$	300012. 5	
$3p^6(^1S)5s$	$5s^2S$	$\frac{1}{2}$	212395. 8		$3p^6(^1S)7h$	$7h^2H^o$	$\left\{ \begin{matrix} 4\frac{1}{2} \\ 5\frac{1}{2} \end{matrix} \right\}$	312973. 5	
$3p^6(^1S)5p$	$5p^2P^o$	$\frac{1}{2}$ $1\frac{1}{2}$	230597. 6 230913. 4	315. 8	-----	-----	-----	-----	-----
$3p^6(^1S)4f$	$4f^2F^o$	$2\frac{1}{2}$ $3\frac{1}{2}$	236125. 3 236132. 5	7. 2	Ti V (1S_0)	Limit	-----	348817. 8	

May 1948.

Ti v

(A I sequence; 18 electrons)

Z=22

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 \ ^1S_0$

$3p^6 \ ^1S_0$ 805500 cm^{-1}

I. P. 99.8 volts

Four lines are classified in the region between 163 Å and 228 Å, as combinations with the ground term. The levels in the table are from the 1937 reference, and all values have been rounded off in the last places.

For convenience, the Paschen notation has been added by the writer in column one of the table, under the heading "A I". As for A I, the jl -coupling notation in the general form suggested by Racah is here introduced, although LS -designations, as indicated in column two under the heading "Authors", are perhaps preferable for the terms thus far identified.

REFERENCES

P. G. Kruger and S. G. Weissberg, Phys. Rev. **48**, 659 (1935). (C L)
P. G. Kruger, S. G. Weissberg and L. W. Phillips, Phys. Rev. **51**, 1090 (1937). (I P) (T) (C L)
G. Racah, Phys. Rev. **61**, 537 (L) (1942).

Ti v

A I	Authors	Config.	Desig.	J	Level
$1p_0$	$3p^6 \ ^1S$	$3p^6$	$3p^6 \ ^1S$	0	0
$1s_4$	$3p^6 4s \ ^3P^o$	$3p^6(^3P_{1/2})4s$	$4s [1\frac{1}{2}]^o$	$\begin{smallmatrix} 2 \\ 1 \end{smallmatrix}$	$\begin{smallmatrix} 436880 \\ 1 \end{smallmatrix}$
$1s_2$	$3p^6 4s \ ^1P^o$	$3p^6(^3P_{3/2}^o)4s$	$4s' [\frac{1}{2}]^o$	$\begin{smallmatrix} 0 \\ 1 \end{smallmatrix}$	$\begin{smallmatrix} 443780 \\ 1 \end{smallmatrix}$
$2s_4$	$3p^6 5s \ ^3P^o$	$3p^6(^3P_{1/2})5s$	$5s [\frac{1}{2}]^o$	$\begin{smallmatrix} 2 \\ 1 \end{smallmatrix}$	$\begin{smallmatrix} 608090 \\ 1 \end{smallmatrix}$
$2s_2$	$3p^6 5s \ ^1P^o$	$3p^6(^3P_{3/2}^o)5s$	$5s' [\frac{1}{2}]^o$	$\begin{smallmatrix} 0 \\ 1 \end{smallmatrix}$	$\begin{smallmatrix} 612970 \\ 1 \end{smallmatrix}$
		Ti VI ($^3P_{1/2}$)	Limit	-----	805500
		Ti VI ($^3P_{3/2}^o$)	Limit	-----	811330

May 1948.

Ti VI

(Cl I sequence; 17 electrons)

 $Z=22$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^5 {}^2P_{1\frac{1}{2}}^{\circ}$ $3p^5 {}^2P_{1\frac{1}{2}}^{\circ}$ 966000 cm^{-1}

I. P. 120 volts

All of the terms except $3p^6 {}^2S$ are from the paper by Edlén. Twelve lines in the region between 182 Å and 524 Å have been classified as combinations from the ground term. Edlén has estimated the value of the limit by extrapolation along the isoelectronic sequence, as indicated by brackets in the table. His unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

- S. G. Weissberg and P. G. Kruger, Phys. Rev. **49**, 872 (A) (1936). (C L)
 B. Edlén, Zeit. Phys. **104**, 407 (1937). (I P) (T) (C L)

Ti VI

Config.	Desig.	J	Level	Interval
$3s^2 3p^5$	$3p^5 {}^2P^{\circ}$	$1\frac{1}{2}$ $\frac{1}{2}$	0 5840	—5840
$3s 3p^6$	$3p^6 {}^2S$	$\frac{1}{2}$	196620	
$3s^2 3p^4({}^3P)4s$	$4s {}^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	495390	
$3s^2 3p^4({}^3P)4s$	$4s {}^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	502580 506440	—3860
$3s^2 3p^4({}^1D)4s$	$4s' {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	518820 518930	—110
$3s^2 3p^4({}^1S)4s$	$4s'' {}^2S$	$\frac{1}{2}$	549000	
-----	-----	---	-----	
Ti VII (3P_2)	Limit	---	[966000]	

January 1948.

Ti VII

(S I sequence; 16 electrons)

 $Z=22$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^4 {}^3P_2$ $3p^4 {}^3P_2$ 1136000 cm^{-1}

I. P. 140.8 volts

All the terms are from Edlén's paper except $3p^5 {}^3P^\circ$, which is from Kruger and Pattin, who have estimated the value entered in brackets in the table. Twenty-four lines have been classified in the region between 164 Å and 200 Å. The limit is from Edlén, who has extrapolated it from isoelectronic sequence data.

The singlet and triplet terms are connected by two observed intersystem combinations.

The unit adopted by Edlén, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

B. Edlén, Zeit. Phys. **104**, 188 (1937). (I P) (T) (C L)

P. G. Kruger and H. S. Pattin, Phys. Rev. **52**, 622 (1937). (T) (C L)

Ti VII

Ti VII

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3s^2\ 3p^4$	$3p^4\ ^3P$	2	0	—4540 —1360	$3s^2\ 3p^3(^2D^\circ)4s$	$4s'\ ^1D^\circ$	2	592930	440 1130
		1	4540		$3s^2\ 3p^3(^2P^\circ)4s$	$4s''\ ^3P^\circ$	0	607550	
		0	5900			1	607990		
$3s^2\ 3p^4$	$3p^4\ ^1D$	2	24120		2	609120			
$3s^2\ 3p^4$	$3p^4\ ^1S$	0	54770	$3s^2\ 3p^3(^2P^\circ)4s$	$4s''\ ^1P^\circ$	1	614790		
$3s\ 3p^5$	$3p^5\ ^3P^\circ$	2	196260	—3800 —[2140]	Ti VIII ($^4S_{3/2}$)	Limit	-----	1136000	
		1	200060						
		0	[202200]						
$3s^2\ 3p^3(^4S^\circ)4s$	$4s\ ^3S^\circ$	1	564240						
$3s^2\ 3p^3(^2D^\circ)4s$	$4s'\ ^3D^\circ$	1	586100	220 680					
		2	586320						
		3	587000						

January 1948.

Ti VIII

(Pr sequence; 15 electrons)

 $Z=22$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^3 {}^4S_{1\frac{1}{2}}^{\circ}$ $3p^3 {}^4S_{1\frac{1}{2}}^{\circ}$ cm^{-1}

I. P. volts

The analysis is incomplete. Kruger and Pattin have observed 15 lines between 150 Å and 162 Å and arranged them in five multiplets that give intervals consistent with those found in related isoelectronic spectra.

By a rough extrapolation of $3p^3 {}^4S_{1\frac{1}{2}}^{\circ} - 3p^3 {}^2D_{1\frac{1}{2}}^{\circ}$ along the isoelectronic sequence the writer has estimated the value of $3p^3 {}^2D_{1\frac{1}{2}}^{\circ}$ entered in brackets in the table. She has calculated the terms listed below from the observed multiplets. The uncertainty x in the estimated position of the doublet terms relative to the quartets may well exceed $\pm 500 \text{ cm}^{-1}$.

REFERENCE

P. G. Kruger and H. S. Pattin, Phys. Rev. **52**, 624 (1937). (C L)

Ti VIII

Config.	Desig.	J	Level	Interval
$3s^2 3p^3$	$3p^3 {}^4S^{\circ}$	$1\frac{1}{2}$	0	
$3s^2 3p^3$	$3p^3 {}^2D^{\circ}$	$1\frac{1}{2}$ $2\frac{1}{2}$	$[33000] + x$ $34080 + x$	1080
$3s^2 3p^3$	$3p^3 {}^2P^{\circ}$	$\frac{1}{2}$ $1\frac{1}{2}$	55000 56460	1460
$3s^2 3p^2({}^3P)4s$	$4s {}^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	660130 662850 666500	2720 3650
$3s^2 3p^2({}^3P)4s$	$4s {}^2P$	$\frac{1}{2}$ $1\frac{1}{2}$	$672220 + x$ $676450 + x$	4230
$3s^2 3p^2({}^1D)4s$	$4s' {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	$691260 + x$ $691490 + x$	-230

December 1947.

Ti IX

(Si I sequence; 14 electrons)

Z=22

Ground state $1s^2 2s^2 2p^6 3s^2 3p^2 \ ^3P_0$

$3p^2 \ ^3P_0$ 1560000 cm^{-1}

I. P. 193 volts

The analysis is very incomplete, but seven lines have been classified by Phillips in the interval 281 Å to 341 Å as combinations among three triplet terms. He states that the interval $3p^2 \ ^3P_0-3p^2 \ ^3P_1$ of the ground term has been extrapolated along the sequence, since no combinations from the ground state $3p^2 \ ^3P_0$ are known. The first interval is, therefore, entered in brackets in the table, as well as his estimated value of the limit.

REFERENCE

L. W. Phillips, Phys. Rev. **55**, 709 (1939). (I P) (T) (C L)

Ti IX

Config.	Desig.	<i>J</i>	Level	Interval
$3s^2 3p^2$	$3p^2 \ ^3P$	0	0	[3100] 4210
		1	3100	
		2	7310	
$3s 3p^3$	$3p^3 \ ^3S^\circ$	1	299920	
$3s^2 3p(^3P^\circ)3d$	$3d \ ^3P^\circ$	2	352460	-4340 -1580
		1	356800	
		0	358380	
-----	-----	---	-----	
Ti x ($^3P_{3/2}^\circ$)	Limit	---	[1560000]	

October 1947.

Ti X

(Al I sequence; 13 electrons)

Z=22

Ground state $1s^2 2s^2 2p^6 3s^2 3p \ ^2P_{3/2}^\circ$

$3p \ ^2P_{3/2}^\circ$ cm^{-1}

I. P. volts

This spectrum has not been analyzed, but Edlén has classified two lines as follows:

I. A.	Int.	Wave No.	Desig.
101. 355	[2]	986630	$3p \ ^2P^\circ-4d \ ^2D$
102. 107	2	979360	

His unit, 10^8 cm^{-1} , is here changed to cm^{-1} .

REFERENCE

B. Edlén, Zeit. Phys. **103**, 540 (1936). (C L)
December 1947.

Ti XI

(Mg I sequence; 12 electrons)

 $Z=22$ Ground state $1s^2 2s^2 2p^6 3s^2 {}^1S_0$ $3s^2 {}^1S_0$ 2142000 cm^{-1}

I. P. 266 volts

Edlén has classified 14 lines in the region between 71 Å and 126 Å. No intersystem combinations have been observed and the triplet terms are not all connected by observed combinations. He has determined the relative positions of the various groups of terms and also the ionization potential by extrapolation along the isoelectronic sequence. His estimated value of the limit is entered in brackets in the table.

His unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCE

B. Edlén, Zeit. Phys. **103**, 536 (1936). (I P) (T) (C L)

Ti XI

Ti XI

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3s^2$	$3s^2 {}^1S$	0	0		$3s({}^2S)4f$	$4f {}^3F^o$	2 3 4		
$3s({}^2S)3p$	$3p {}^3P^o$	0 1 2	$172370+x$ $174920+x$ $180550+x$	2550 5630	$3s({}^2S)5d$	$5d {}^3D$	1 2 3	$1297420+x$	
$3s({}^2S)3d$	$3d {}^3D$	1 2 3	$504150+x$		$3s({}^2S)5f$	$5f {}^3F^o$	2 3 4	$1577370+x$ $1603570+x$	
$3s({}^2S)4s$	$4s {}^3S$	1	$1050030+x$						
$3s({}^2S)4p$	$4p {}^1P^o$	1	1139970						
$3s({}^2S)4d$	$4d {}^3D$	1 2 3	$1243080+x$ $1243350+x$ $1243770+x$	270 420					
					Ti XII (${}^2S_{1/2}$)	Limit	-----	[2142000]	

August 1947.

Ti XII

(Na I sequence; 11 electrons)

 $Z=22$ Ground state $1s^2 2s^2 2p^6 3s {}^2S_{1/2}$ $3s {}^2S_{1/2}$ 2351530 cm^{-1}

I. P. 291.47 volts

Edlén has classified 16 lines in the interval 60 Å to 116 Å, and extrapolated the absolute value of the ground term from isoelectronic sequence data.

The unit adopted by Edlén, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCE

B. Edlén, Zeit. Phys. **100**, 621 (1936). (I P) (T) (C L)

Ti XII

Ti XII

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
3s	3s ² S	½	0						
3p	3p ² P°	½ 1½	208300 216960	8660	5p	5p ² P°	½ 1½	1645820 1647310	1490
3d	3d ² D	1½ 2½	502370 503260	890	5d	5d ² D	1½ 2½	1697530 1697740	210
4s	4s ² S	½	1133370		5f	5f ² F°	2½ 3½	1717270 1717410	140
4p	4p ² P°	½ 1½	1214330 1217670	3340	6f	6f ² F°	2½ 3½	1911470	
4d	4d ² D	1½ 2½	1321380 1321840	460					
4f	4f ² F°	2½ 3½	1360770 1360930	160	Ti XIII (¹ S ₀)	Limit		2351530	

June 1947.

Ti XIII

(Ne I sequence; 10 electrons)

Z=22

Ground state 1s² 2s² 2p⁶ ¹S₀

2p⁶ ¹S₀ 6360600 cm⁻¹

I. P. 788.4 volts

Edlén and Tyrén have classified five lines in the interval between 23 Å and 26 Å, as combinations with the ground term. Their absolute term values are based on extrapolation along the Ne I isoelectronic sequence. Their unit, 10³ cm⁻¹, has here been changed to cm⁻¹.

As for Ne I, the *jl*-coupling notation in the general form suggested by Racah is introduced.

REFERENCES

B. Edlén and F. Tyrén, Zeit. Phys. **101**, 210 (1936). (I P) (T) (C L)
G. Racah, Phys. Rev. **61**, 537 (L) (1942).

Ti XIII

Authors	Config.	Desig.	<i>J</i>	Level
2p ¹ S ₀	2p ⁶	2p ⁶ ¹ S	0	0
3s ³ P ₁	2p ⁵ (² P _{1/2})3s	3s [1½]°	2 1	3709200
3s ¹ P ₁	2p ⁵ (² P _{3/2})3s	3s' [½]°	0 1	3753600
3d ³ P ₁	2p ⁵ (² P _{1/2})3d	3d [½]°	0 1	4168200
3d ³ P ₁	"	3d [1½]°	1	4219800
3d ³ D ₁	2p ⁵ (² P _{3/2})3d	3d' [1½]°	1	4281600
	Ti XIV (² P _{1/2})	Limit	-----	6360600
	Ti XIV (² P _{3/2})	Limit	-----	6407500

April 1947.

VANADIUM

V I

23 electrons

 $Z=23$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^3 4s^2 {}^4F_{1\frac{1}{2}}$ $a {}^4F_{1\frac{1}{2}} 54361 \text{ cm}^{-1}$

I. P. 6.74 volts

The arc spectrum of vanadium has been studied since 1923. The early contributions of Meggers, Laporte, Landé, Bechert, Sommer, and many others culminated in the extensive analysis of this highly complex spectrum published by Meggers and Russell in 1936. They list 60 doublet terms, 60 quartet terms, and 28 sextet terms from 634 multiplets, and give 2186 classified lines extending from 2082 Å to 11911 Å. The terms of all three multiplicities are connected by observed intersystem combinations.

The configuration assignments of many of the odd doublet and quartet terms are extremely uncertain and a number of terms are unassigned. No limit assignment has been attempted for the sextet triad $x {}^6P^\circ$, $w {}^6D^\circ$, and $x {}^6F^\circ$, which comes from $3d^4 5p$, and for two quartet triads which may arise from $3d^3 4s 5p$. Rohrlich has suggested that some of the configurations of odd terms from $d^3 sp$ and $d^4 p$ should be interchanged.

Zeeman observations by Babcock of more than 900 lines provided the large array of g -values which greatly facilitated the analysis. Much of this material was generously furnished in manuscript form for inclusion in the 1936 paper. A discussion of the g -sums by Russell and Babcock appears in the 1935 reference below.

Six terms, and miscellaneous odd levels were added by the writer in 1939 from additional observations of the spectrum between 1848 Å and 2173 Å.

REFERENCES

- H. N. Russell and H. D. Babcock, *Zeeman Verhandelingen* p. 286 (Martinus Nijhoff, The Hague 1935). (Z E)
 W. F. Meggers and H. N. Russell, *J. Research Nat. Bur. Std.* **17**, 125, RP906 (1936). (I P) (T) (C L) (Z E)
 C. E. Moore, *Phys. Rev.* **55**, 710 (1939). (T) (C L)
 W. F. Meggers, *J. Opt. Soc. Am.* **36**, 431 (1946). (Summary hfs.)
 F. Rohrlich, *Phys. Rev.* **74**, 1393 (1948).

V I

V I

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$3d^3 4s^2$	a^4F	$1\frac{1}{2}$	0. 00	137. 38	0. 40	$3d^4(a^3H)4s$	b^2H	$4\frac{1}{2}$	19023. 47	121. 66	0. 91
		$2\frac{1}{2}$	137. 38	186. 04	1. 01			$5\frac{1}{2}$	19145. 13		1. 08
		$3\frac{1}{2}$	323. 42	229. 60	1. 20	$3d^4(b^3F)4s$	a^2F	$2\frac{1}{2}$	19026. 34	51. 81	0. 86
		$4\frac{1}{2}$	553. 02		1. 28			$3\frac{1}{2}$	19078. 15		1. 14
$3d^4(a^5D)4s$	a^6D	$\frac{1}{2}$	2112. 32	40. 88	3. 29	$3d^5$	a^6S	$2\frac{1}{2}$	20202. 49		
		$1\frac{1}{2}$	2153. 20	66. 93	1. 82	$3d^3 4s(a^3F)4p$	z^4D°	$\frac{1}{2}$	20606. 43	81. 32	-0. 04
		$2\frac{1}{2}$	2220. 13	91. 24	1. 61			$1\frac{1}{2}$	20687. 75	140. 73	1. 21
		$3\frac{1}{2}$	2311. 37	113. 52	1. 53			$2\frac{1}{2}$	20828. 48	204. 04	1. 35
		$4\frac{1}{2}$	2424. 89		1. 52			$3\frac{1}{2}$	21032. 52		1. 45
$3d^4(a^5D)4s$	a^4D	$\frac{1}{2}$	8412. 94	63. 26	0. 00	$3d^4(a^3D)4s$	b^4D	$3\frac{1}{2}$	20767. 57	-21. 56	1. 45
		$1\frac{1}{2}$	8476. 20	102. 32	1. 19			$2\frac{1}{2}$	20789. 13	-23. 86	1. 25
		$2\frac{1}{2}$	8578. 52	137. 20	1. 35			$1\frac{1}{2}$	20812. 99	-17. 21	1. 20
		$3\frac{1}{2}$	8715. 72		1. 39			$\frac{1}{2}$	20830. 20		0. 10
$3d^3 4s^2$	a^4P	$\frac{1}{2}$	9544. 54	92. 42	2. 59	$3d^4(a^3G)4s$	b^2G	$4\frac{1}{2}$	21603. 17	-43. 22	1. 11
		$1\frac{1}{2}$	9636. 96	187. 62	1. 70			$3\frac{1}{2}$	21646. 39		0. 86
		$2\frac{1}{2}$	9824. 58		1. 55	$3d^3 4s(a^3F)4p$	z^4G°	$2\frac{1}{2}$	21841. 45	122. 05	0. 55
$3d^3 4s^2$	a^2G	$3\frac{1}{2}$	10892. 50	208. 15	0. 88			$3\frac{1}{2}$	21963. 50	157. 67	0. 96
		$4\frac{1}{2}$	11100. 65		1. 13			$4\frac{1}{2}$	22121. 17	192. 82	1. 16
$3d^3 4s^2$	a^2P	$1\frac{1}{2}$	13801. 53	-9. 37	1. 20			$5\frac{1}{2}$	22313. 99		1. 24
		$\frac{1}{2}$	13810. 90		0. 64	$3d^3 4s(a^3F)4p$	z^4F°	$1\frac{1}{2}$	23088. 06	122. 50	0. 39?
$3d^3 4s^2$	a^2D	$1\frac{1}{2}$	14514. 75	34. 08	0. 97			$2\frac{1}{2}$	23210. 56	142. 53	0. 98?
		$2\frac{1}{2}$	14548. 83		1. 17			$3\frac{1}{2}$	23353. 09	166. 75	1. 23
$3d^4(a^3H)4s$	a^4H	$3\frac{1}{2}$	14910. 04	39. 26	0. 65	$3d^3 4s(a^3F)4p$	z^2D°	$1\frac{1}{2}$	23608. 80	326. 35	0. 76
		$4\frac{1}{2}$	14949. 30	51. 54	0. 94			$2\frac{1}{2}$	23935. 15		1. 32?
		$5\frac{1}{2}$	15000. 84	62. 10	1. 10	$3d^4(a^5D)4p$	z^6P°	$1\frac{1}{2}$	24648. 10	79. 75	2. 34
		$6\frac{1}{2}$	15062. 94		1. 18			$2\frac{1}{2}$	24727. 85	110. 71	1. 85
$3d^4(a^3P)4s$	b^4P	$\frac{1}{2}$	15078. 25	192. 17	2. 60			$3\frac{1}{2}$	24838. 56		1. 67
		$1\frac{1}{2}$	15270. 42	301. 48	1. 68	$3d^4(a^5D)4p$	z^4P°	$\frac{1}{2}$	24770. 62	144. 54	2. 54
		$2\frac{1}{2}$	15571. 90		1. 54			$1\frac{1}{2}$	24915. 16	215. 80	1. 71
$3d^3 4s^2$	a^2H	$4\frac{1}{2}$	15103. 77	161. 06	0. 90			$2\frac{1}{2}$	25130. 96		1. 59
		$5\frac{1}{2}$	15264. 83		1. 07	$3d^4(a^5D)4p$	y^6F°	$\frac{1}{2}$	24789. 36	40. 82	-0. 58
$3d^4(b^3F)4s$	b^4F	$1\frac{1}{2}$	15664. 75	24. 05	0. 39			$1\frac{1}{2}$	24830. 18	68. 55	1. 02
		$2\frac{1}{2}$	15688. 80	35. 42	1. 05			$2\frac{1}{2}$	24898. 73	94. 15	1. 23
		$3\frac{1}{2}$	15724. 22	46. 50	1. 22			$3\frac{1}{2}$	24992. 88	118. 62	1. 37
		$4\frac{1}{2}$	15770. 72		1. 31			$4\frac{1}{2}$	25111. 50	142. 03	1. 41
$3d^3 4s(a^5F)4p$	z^6G°	$1\frac{1}{2}$	16361. 45	88. 40	0. 00	$3d^4(a^5D)4p$	y^4F°	$1\frac{1}{2}$	25930. 51	73. 71	0. 42
		$2\frac{1}{2}$	16449. 85	122. 69	0. 78			$2\frac{1}{2}$	26004. 22	117. 82	0. 98
		$3\frac{1}{2}$	16572. 54	156. 21	1. 10			$3\frac{1}{2}$	26122. 04	49. 92	1. 15
		$4\frac{1}{2}$	16728. 75	188. 40	1. 22			$4\frac{1}{2}$	26171. 96		1. 23
		$5\frac{1}{2}$	16917. 15	219. 29	1. 26	$3d^3 4s(a^3F)4p$	z^2G°	$3\frac{1}{2}$	26021. 89	323. 05	0. 92
		$6\frac{1}{2}$	17136. 44		1. 43			$4\frac{1}{2}$	26344. 94		1. 13
$3d^4(a^3G)4s$	a^4G	$2\frac{1}{2}$	17054. 87	62. 05	0. 59	$3d^4(a^5D)4p$	y^4D°	$\frac{1}{2}$	26182. 60	66. 88	-0. 06
		$3\frac{1}{2}$	17116. 92	65. 06	0. 96			$1\frac{1}{2}$	26249. 48	103. 11	1. 17
		$4\frac{1}{2}$	17181. 98	60. 07	1. 14			$2\frac{1}{2}$	26352. 59	127. 69	1. 34
		$5\frac{1}{2}$	17242. 05		1. 27			$3\frac{1}{2}$	26480. 28		1. 39
$3d^3 4s(a^5F)4p$	z^6D°	$\frac{1}{2}$	18085. 82	40. 45	3. 20	$3d^4(a^5D)4p$	y^6D°	$\frac{1}{2}$	26397. 36	40. 32	3. 25
		$1\frac{1}{2}$	18126. 27	71. 81	1. 76			$1\frac{1}{2}$	26437. 68	68. 20	1. 86
		$2\frac{1}{2}$	18198. 08	104. 19	1. 58			$2\frac{1}{2}$	26505. 88	98. 89	1. 59
		$3\frac{1}{2}$	18302. 27	135. 80	1. 56			$3\frac{1}{2}$	26604. 77	133. 54	1. 58
		$4\frac{1}{2}$	18438. 07		1. 55			$4\frac{1}{2}$	26738. 31		1. 50
$3d^3 4s(a^5F)4p$	z^6F°	$\frac{1}{2}$	18120. 12	53. 94	-0. 44	$3d^3 4s(a^3F)4p$	z^2F°	$2\frac{1}{2}$	27187. 77	283. 11	1. 01?
		$1\frac{1}{2}$	18174. 06	84. 83	1. 14			$3\frac{1}{2}$	27470. 88		1. 01
		$2\frac{1}{2}$	18258. 89	113. 57	1. 28	$3d^3 4s(a^5P)4p$	x^6D°	$\frac{1}{2}$	28313. 68	55. 08	3. 23
		$3\frac{1}{2}$	18372. 46	141. 00	1. 28			$1\frac{1}{2}$	28368. 76	93. 39	1. 82
		$4\frac{1}{2}$	18513. 46	166. 66	1. 38			$2\frac{1}{2}$	28462. 15	133. 49	1. 58
		$5\frac{1}{2}$	18680. 12		1. 42			$3\frac{1}{2}$	28595. 64	172. 49	1. 52
$3d^4(a^3P)4s$	b^2P	$\frac{1}{2}$	18805. 05	384. 23	0. 67			$4\frac{1}{2}$	28768. 13		1. 47
		$1\frac{1}{2}$	19189. 28		1. 37						

V I—Continued

V I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$3d^3 4s(a^5P)4p$	z^4S°	$1\frac{1}{2}$	28621. 27			$3d^3 4s(a^5F)4p$	v^4D°	$\frac{1}{2}$	34477. 40		0. 00
	y^6P°	$1\frac{1}{2}$	29202. 80		2. 32			$1\frac{1}{2}$	34537. 21	59. 81	1. 05
		$2\frac{1}{2}$	29296. 43	93. 63	1. 76			$2\frac{1}{2}$	34619. 52	82. 31	1. 28
		$3\frac{1}{2}$	29418. 17	121. 74	1. 62			$3\frac{1}{2}$	34747. 06	127. 54	1. 35
$3d^3 4s(c^3P)4p$	y^4P°	$\frac{1}{2}$	30021. 57		2. 67		u^4D°	$\frac{1}{2}$	35012. 91		
		$1\frac{1}{2}$	30094. 52	72. 95	1. 74			$1\frac{1}{2}$	35092. 36	79. 45	1. 12
		$2\frac{1}{2}$	30120. 78	26. 26	1. 67			$2\frac{1}{2}$	35225. 04	132. 68	1. 32
								$3\frac{1}{2}$	35379. 11	154. 07	1. 33
$3d^3 4s(b^3G)4p?$	y^4G°	$2\frac{1}{2}$	30635. 60		0. 53	$3d^4(a^3P)4p?$	y^4S°	$1\frac{1}{2}$	36408. 23		1. 85
		$3\frac{1}{2}$	30694. 34	58. 74	0. 93		x^2D°	$1\frac{1}{2}$	36416. 49		0. 89
		$4\frac{1}{2}$	30771. 72	77. 38	1. 13			$2\frac{1}{2}$	36700. 78	284. 29	1. 13
		$5\frac{1}{2}$	30864. 34	92. 62	1. 21		x^2G°	$3\frac{1}{2}$	36461. 26		0. 85
$3d^3 4s(a^5P)4p$	z^6S°	$2\frac{1}{2}$	30832. 58			$3d^4(b^3F)4p$		$4\frac{1}{2}$	36538. 58	77. 32	1. 05
$3d^3 4s(b^3G)4p$	x^4F°	$1\frac{1}{2}$	31200. 12		0. 38	$3d^3 4s(b^1D)4p$	y^2P°	$\frac{1}{2}$	36477. 75		0. 74
		$2\frac{1}{2}$	31228. 98	28. 86	1. 01			$1\frac{1}{2}$	36580. 46	102. 71	1. 17
		$3\frac{1}{2}$	31268. 15	39. 17	1. 21		x^4P°	$2\frac{1}{2}$	36611. 81		1. 54
		$4\frac{1}{2}$	31317. 50	49. 35	1. 32			$1\frac{1}{2}$	36814. 80	-202. 99	1. 77
$3d^3 4s(a^5F)4p$	x^4G°	$2\frac{1}{2}$	31398. 09		0. 53	$3d^4(a^3P)4p?$		$\frac{1}{2}$	36695. 49	119. 31	2. 51
		$3\frac{1}{2}$	31541. 18	143. 09	0. 95		w^2G°	$3\frac{1}{2}$	36628. 82		
		$4\frac{1}{2}$	31721. 73	180. 55	1. 12			$4\frac{1}{2}$	36828. 33	199. 51	0. 65?
		$5\frac{1}{2}$	31937. 18	215. 45	1. 20		w^4G°	$2\frac{1}{2}$	36763. 41		
	z^2S°	$\frac{1}{2}$	31786. 19		2. 30	$3d^3 4s(b^3H)4p?$		$3\frac{1}{2}$	36822. 86	59. 45	1. 06
	y^2S°	$\frac{1}{2}$	31962. 30		2. 21			$4\frac{1}{2}$	36897. 88	75. 02	1. 17
	x^4D°	$\frac{1}{2}$	32348. 89		0. 08		x^2F°	$2\frac{1}{2}$	36766. 00	40. 54	1. 26
		$1\frac{1}{2}$	32456. 45	107. 56	1. 17			$3\frac{1}{2}$	36925. 88	159. 88	0. 89
$3d^3 4s(b^3G)4p$		$2\frac{1}{2}$	32660. 26	203. 81	1. 29	$3d^5$	e^4F	$1\frac{1}{2}$	36983. 63		
		$3\frac{1}{2}$	32891. 06	230. 80	1. 35			$2\frac{1}{2}$	36989. 20	5. 57	
	z^4H°	$3\frac{1}{2}$	32692. 09		0. 68			$3\frac{1}{2}$	37025. 60	36. 40	
		$4\frac{1}{2}$	32788. 22	96. 13	0. 98			$4\frac{1}{2}$	37075. 64	50. 04	
		$5\frac{1}{2}$	32897. 81	109. 59	1. 11	$3d^4(a^5D)5s$	e^6D	$\frac{1}{2}$	37116. 68		3. 08
		$6\frac{1}{2}$	32963. 90	66. 09	1. 21			$1\frac{1}{2}$	37158. 36	41. 68	1. 87
	z^2P°	$\frac{1}{2}$	32724. 86		0. 73?			$2\frac{1}{2}$	37227. 44	69. 08	1. 61
		$1\frac{1}{2}$	32767. 88	43. 02	1. 22			$3\frac{1}{2}$	37322. 09	94. 65	1. 64
$3d^3 4s(a^5F)4p$	w^4F°	$1\frac{1}{2}$	32738. 14		0. 52	$3d^4(a^3H)4p$	v^2G°	$3\frac{1}{2}$	37174. 68		0. 99
		$2\frac{1}{2}$	32846. 74	108. 60	1. 01			$4\frac{1}{2}$	37361. 95	187. 27	1. 05
		$3\frac{1}{2}$	32988. 82	142. 08	1. 18		y^2H°	$4\frac{1}{2}$	37180. 90		0. 73
		$4\frac{1}{2}$	33155. 30	166. 48	1. 30			$5\frac{1}{2}$	37210. 85	29. 95	1. 08
	y^2G°	$4\frac{1}{2}$	33306. 96		1. 03	$3d^4(a^3H)4p$	z^4I°	$4\frac{1}{2}$	37285. 03		0. 87
		$3\frac{1}{2}$	33360. 31	-53. 35	0. 91			$5\frac{1}{2}$	37315. 83	30. 80	0. 96
	y^2F°	$3\frac{1}{2}$	33481. 45		1. 11			$6\frac{1}{2}$	37404. 25	88. 42	1. 08
		$2\frac{1}{2}$	33527. 64	-46. 19	0. 85			$7\frac{1}{2}$	37518. 36	114. 11	1. 15
	z^2H°	$4\frac{1}{2}$	33640. 18		0. 92	$3d^4(b^3F)4p?$	w^2F°	$2\frac{1}{2}$	37342. 66		0. 84
		$5\frac{1}{2}$	33695. 32	55. 14	1. 09			$3\frac{1}{2}$	37475. 08	132. 42	1. 08
	w^4D°	$\frac{1}{2}$	33966. 72		0. 09	$3d^3 4s(a^5F)5s$	e^6F	$\frac{1}{2}$	37374. 98		-0. 72
		$1\frac{1}{2}$	33976. 02	9. 30	0. 80			$1\frac{1}{2}$	37423. 17	48. 19	1. 05
$3d^3 4s(c^3P)4p$		$2\frac{1}{2}$	34065. 61	89. 59	1. 30			$2\frac{1}{2}$	37503. 14	79. 97	1. 30
		$3\frac{1}{2}$	34128. 04	62. 43	1. 35			$3\frac{1}{2}$	37614. 97	111. 83	1. 33
	1°		34019. 12					$4\frac{1}{2}$	37758. 07	143. 10	1. 43
	v^4F°	$1\frac{1}{2}$	34030. 04		0. 86	$3d^4(b^3F)4p$	w^2D°	$1\frac{1}{2}$	37457. 50		0. 80
		$2\frac{1}{2}$	34167. 84		1. 32?			$2\frac{1}{2}$	37762. 54	295. 04	1. 18
		$3\frac{1}{2}$	34374. 81	137. 80	1. 21		y^4H°	$3\frac{1}{2}$	37481. 86		0. 76
		$4\frac{1}{2}$	34529. 81	206. 97	1. 41			$4\frac{1}{2}$	37516. 95	35. 59	1. 05
	y^2D°	$1\frac{1}{2}$	34428. 76		0. 73	$3d^3 4s(b^3H)4p?$		$5\frac{1}{2}$	37565. 88	48. 93	1. 09
		$2\frac{1}{2}$	34486. 80	58. 04	1. 18			$6\frac{1}{2}$	37626. 44	60. 56	1. 24

V I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$3d^4(b\ ^3F)4p$	$v\ ^4G^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$ $5\frac{1}{2}$	37498.76 37556.00 37644.41 37764.89	57.24 88.41 120.48	0.60 1.02 1.15 1.22
$3d^4(a\ ^3H)4p?$	$z\ ^2I^\circ$	$5\frac{1}{2}$ $6\frac{1}{2}$	37530.29 37606.32	76.03	0.94 1.06
$3d^4(b\ ^3F)4p$	$t\ ^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	37757.24 37834.98 37959.66 38115.65	77.74 124.68 155.99	0.01 1.18 1.33 1.35
$3d^4(a\ ^5D)5s$	$e\ ^4D$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	37940.08 38003.93 38106.32 38242.46	63.85 102.39 136.14	
$3d^4(a\ ^3H)4p$	$x\ ^2H^\circ$	$4\frac{1}{2}$ $5\frac{1}{2}$	38123.76 38220.63	96.87	0.88 1.10
$3d^4(a\ ^3H)4p$	$x\ ^4H^\circ$	$3\frac{1}{2}$ $4\frac{1}{2}$ $5\frac{1}{2}$ $6\frac{1}{2}$	38245.75 38323.87 38404.96 38482.96	78.12 81.09 78.00	0.67 0.93 1.11 1.22
	$u\ ^2G^\circ$	$4\frac{1}{2}$ $3\frac{1}{2}$	38529.78 38610.94	-81.16	0.99 0.88?
$3d^3\ 4s(a\ ^1H)4p?$	$y\ ^2I^\circ$	$5\frac{1}{2}$ $6\frac{1}{2}$	39008.60 39081.10	72.50	0.92 1.06
$3d^3\ 4s(a\ ^5F)5s$	$f\ ^4F$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	39127.23 39241.34 39398.82 39597.01	114.11 157.48 198.19	0.46? 1.03 1.22? 1.33?
$3d^3\ 4s(a\ ^5P)4p$	$w\ ^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	39237.10 39248.90 39422.66	11.80 173.76	2.57 1.60 1.52
$3d^4(b\ ^3F)4p$	$u\ ^4F^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	39266.60 39300.48 39341.76 39391.02	33.88 41.28 49.26	0.54 1.00 1.21 1.30
$3d^3\ 4s(c\ ^3P)4p$	$x\ ^4S^\circ$	$1\frac{1}{2}$	39847.24		2.00
$3d^4(a\ ^3P)4p$	$s\ ^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	39877.62 39935.07 39999.89 40125.79	57.45 64.82 125.90	0.01 1.10 1.33 1.38
$3d^4(a\ ^3P)4p$	$v\ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	39884.43 40119.26	234.83	0.92 1.14
$3d^4(a\ ^3H)4p$	$u\ ^4G^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$ $5\frac{1}{2}$	39962.17 40001.18 40038.95 40063.78	39.01 37.77 24.83	0.53 0.99 1.19 1.23
	$v\ ^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	40153.51 40587.35	433.84	1.01
$3d^3\ 4s(a\ ^1P)4p?$	$u\ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	40225.38 40325.77	100.39	0.70 1.12
$3d^3\ 4s(a\ ^1P)4p?$	$x\ ^2S^\circ$	$\frac{1}{2}$	40299.81		
$3d^4(a\ ^3G)4p$	$w\ ^4H^\circ$	$3\frac{1}{2}$ $4\frac{1}{2}$ $5\frac{1}{2}$ $6\frac{1}{2}$	40314.83 40378.70 40452.38 40535.62	63.87 73.68 83.24	0.65 0.92 1.08 1.22

V I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
	$x\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	40328.62 40437.42	108.80	1.52
	$w\ ^2P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$	40693.76		
	$w\ ^2H^\circ$	$5\frac{1}{2}$ $4\frac{1}{2}$	40919.68 40980.54	-60.86	0.96? 0.99
$3d^4(a\ ^3G)4p$	$t\ ^4F^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	41389.49 41428.93 41492.29 41599.36	39.44 63.36 107.07	0.42 0.89? 1.15 1.23
$3d^3\ 4s(b\ ^1G)4p?$	$t\ ^2G^\circ$	$3\frac{1}{2}$ $4\frac{1}{2}$	41436.58 41539.14	102.56	0.90 1.04
$3d^3\ 4s(a\ ^1H)4p?$	$v\ ^2H^\circ$	$4\frac{1}{2}$ $5\frac{1}{2}$	41501.41 41659.71	158.30	0.87 1.05
$3d^4(a\ ^3G)4p$	$t\ ^4G^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$ $5\frac{1}{2}$	41654.70 41758.41 41860.54 41918.24	103.71 102.13 57.70	0.58 1.03 1.20 1.20
	$v\ ^4P^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	41751.78 41848.47 42009.93	96.69 161.46	2.56 1.62 1.48
$3d^3\ 4s(a\ ^5P)4p$	$r\ ^4D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	41928.47 41999.10 42138.00 42245.61	70.63 138.90 107.61	0.04 1.20 1.33 1.36
$3d^3\ 4s(b\ ^1D)4p?$	$u\ ^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	41950.35 42020.93	70.58	0.84 1.11
$3d^4(a\ ^5D)4d$	$e\ ^6G$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$ $5\frac{1}{2}$ $6\frac{1}{2}$	42033.84 42070.05 42114.17 42177.31 42257.32 42353.42	36.21 44.12 63.14 80.01 96.10	1.08 1.23 1.32 1.35
$3d^3\ 4s(b\ ^1G)4p$	$u\ ^2H^\circ$	$4\frac{1}{2}$ $5\frac{1}{2}$	42079.14 42220.69	141.55	0.85 1.06
$3d^4(a\ ^5D)4d$	$e\ ^6P$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	42164.74		1.44?
	2°	$3\frac{1}{2}$	42236.66		
$3d^3\ 4s(a\ ^1P)4p?$	$v\ ^2P^\circ$	$1\frac{1}{2}$ $\frac{1}{2}$	42318.42 42480.62	-162.20	1.34 1.14
	$w\ ^2S^\circ$	$\frac{1}{2}$	42362.04		1.50?
$3d^4(a\ ^5D)4d$	$f\ ^6F$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$ $5\frac{1}{2}$	42363.62 42506.32 42577.98	142.70 71.66	1.39
$3d^4(a\ ^5D)4d$	6D	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	42404.89 42553.62	148.73	1.61
	$w\ ^6D^\circ$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	42480.31 42587.41 42725.33	107.10 137.92	

V I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$3d^3 4s(a^5P)4p$	w^4S°	$1\frac{1}{2}$	42969. 49		1. 94
	s^4F°	$1\frac{1}{2}$	42981. 34		
		$2\frac{1}{2}$	43051. 31	69. 97	
		$3\frac{1}{2}$	43147. 09	95. 78	
		$4\frac{1}{2}$	43266. 15	119. 06	
	q^4D°	$\frac{1}{2}$	43249. 44		
		$1\frac{1}{2}$	43308. 83	59. 39	
		$2\frac{1}{2}$	43410. 82	101. 99	
		$3\frac{1}{2}$	43555. 12	144. 30	
	u^4P°	$\frac{1}{2}$	43443. 33		1. 46
		$1\frac{1}{2}$	43503. 99	60. 66	
		$2\frac{1}{2}$	43585. 59	81. 60	
$3d^3 4s(a^5F)4d$	e^6H	$2\frac{1}{2}$	43649. 40		0. 38
		$3\frac{1}{2}$	43706. 82	57. 42	0. 88
		$4\frac{1}{2}$	43787. 60	90. 78	1. 11
		$5\frac{1}{2}$	43894. 15	106. 55	1. 18
		$6\frac{1}{2}$	44028. 33	134. 18	1. 30
		$7\frac{1}{2}$	44189. 95	161. 62	1. 38
	x^6F°	$\frac{1}{2}$	43707. 97?		
		$1\frac{1}{2}$	43845. 80?	137. 83	
		$2\frac{1}{2}$	43959. 24?	113. 44	
		$3\frac{1}{2}$	44026. 29?	67. 05	
		$4\frac{1}{2}$	44202. 51?	176. 22	
		$5\frac{1}{2}$			
$3d^3 4s(a^5F)4d$	f^6G	$1\frac{1}{2}$	43818. 02		0. 38?
		$2\frac{1}{2}$	43847. 16	29. 14	0. 78
		$3\frac{1}{2}$	43911. 93	64. 77	1. 12
		$4\frac{1}{2}$	44005. 14	93. 21	1. 26
		$5\frac{1}{2}$	44139. 69	134. 55	1. 34
		$6\frac{1}{2}$	44327. 04	187. 35	1. 35
$3d^3 4s(b^1G)4p$	t^2F°	$3\frac{1}{2}$	43873. 79		1. 04?
		$2\frac{1}{2}$	43875. 25	—1. 46	0. 86
$3d^3 4s(a^3F)5s$	e^2F	$2\frac{1}{2}$	43918. 58		0. 89
		$3\frac{1}{2}$	44066. 05	147. 47	1. 18
$3d^3 4s(a^3F)5s$	x^6P°	$1\frac{1}{2}$			
		$2\frac{1}{2}$			
		$3\frac{1}{2}$	43988. 00?		
	s^4G°	$2\frac{1}{2}$	43999. 68		
		$3\frac{1}{2}$	44043. 36	43. 68	0. 98
$3d^4(a^3G)4p$	t^2H°	$4\frac{1}{2}$	44145. 77		0. 90
		$5\frac{1}{2}$	44184. 02	38. 25	1. 06?
	f^6P	$1\frac{1}{2}$	44443. 67		
		$2\frac{1}{2}$	44532. 60	88. 93	
$3d^4(a^3G)4p$	s^2G°	$3\frac{1}{2}$	44690. 47	157. 87	
		$4\frac{1}{2}$	44463. 28	—32. 15	1. 09
		$3\frac{1}{2}$	44495. 43		0. 91
		$3\frac{1}{2}$			
$3d^4(a^3G)4p$	p^4D°	$\frac{1}{2}$	44514. 34		
		$1\frac{1}{2}$	44554. 25	39. 91	1. 22
		$2\frac{1}{2}$	44616. 68	62. 43	1. 37?
		$3\frac{1}{2}$	44700. 88	84. 20	1. 32?
$3d^3 4s(a^5F)4d$	g^6D	$\frac{1}{2}$			
		$1\frac{1}{2}$	44844. 83		
		$2\frac{1}{2}$	44921. 08	76. 25	
		$3\frac{1}{2}$	45056. 61	135. 53	1. 55?
		$4\frac{1}{2}$	45157. 74	101. 13	

V I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$3d^3 4s(a^1H)4p?$	r^4F°	$1\frac{1}{2}$	44973. 60		0. 58?
		$2\frac{1}{2}$	45049. 17	75. 57	
		$3\frac{1}{2}$	45058. 62	9. 45	0. 97
		$4\frac{1}{2}$	45145. 16	86. 54	1. 26
	q^4F°	$1\frac{1}{2}$	45066. 56		0. 59
		$2\frac{1}{2}$	45107. 21	40. 65	0. 93
		$3\frac{1}{2}$	45157. 72	50. 51	1. 05
		$4\frac{1}{2}$	45237. 16	79. 44	1. 22
	u^2P°	$\frac{1}{2}$			
		$1\frac{1}{2}$	45159. 15		1. 66?
	r^2G°	$3\frac{1}{2}$	45175. 92		0. 98
		$4\frac{1}{2}$	45361. 42	185. 50	1. 14
$3d^3 4s(a^5F)4d$	$2a^\circ$	$5\frac{1}{2}$	45353. 69		
	g^6F	$\frac{1}{2}$			
		$1\frac{1}{2}$	45638. 54		
		$2\frac{1}{2}$	45700. 25	61. 71	
		$3\frac{1}{2}$	45743. 62	43. 37	1. 26
		$4\frac{1}{2}$	45813. 25	69. 63	
		$5\frac{1}{2}$	46034. 58	221. 33	
	p^4F°	$1\frac{1}{2}$	45648. 86		0. 60
		$2\frac{1}{2}$	45688. 41	39. 55	
		$3\frac{1}{2}$	45760. 03	71. 62	1. 02
		$4\frac{1}{2}$	45891. 55	131. 52	1. 32
$3d^4(a^3D)4p$	t^2P°	$1\frac{1}{2}$	45654. 50		1. 24?
		$\frac{1}{2}$	45946. 66	—292. 16	
	o^4D°	$\frac{1}{2}$	45702. 14		
		$1\frac{1}{2}$	45762. 24	60. 10	0. 96?
$3d^3 4s(b^3G)4p$	r^4G°	$2\frac{1}{2}$	46052. 79		0. 56
		$3\frac{1}{2}$	46139. 06	86. 27	0. 96
		$4\frac{1}{2}$	46243. 64	104. 58	1. 15
		$5\frac{1}{2}$	46363. 42	119. 78	1. 19
	4°	$1\frac{1}{2}$	46322. 39:		
		$2\frac{1}{2}$			
	5°	$2\frac{1}{2}$	46500. 64		
	6°	$1\frac{1}{2}$	46707. 18		
	t^4P°	$\frac{1}{2}$	46851. 10		
		$1\frac{1}{2}$	46862. 73	11. 63	
		$2\frac{1}{2}$	46868. 10	5. 37	
$3d^3 4s(b^3G)4p$	s^2F°	$2\frac{1}{2}$	46996. 84		
		$3\frac{1}{2}$	47143. 24	146. 40	1. 02
	7°	$3\frac{1}{2}$	47348. 14		
$3d^3 4s(b^3G)4p$	3°	$1\frac{1}{2}$	47423. 18		
	s^2H°	$4\frac{1}{2}$	47611. 77		1. 01?
		$5\frac{1}{2}$	47701. 55	89. 78	0. 94
	8°	$3\frac{1}{2}$	47615. 56		
	9°	$\frac{1}{2}$	47682. 68		
		$2\frac{1}{2}$			
	q^4G°	$2\frac{1}{2}$	47690. 5		
		$3\frac{1}{2}$	47823. 24	132. 7	
		$4\frac{1}{2}$	48014. 18	190. 94	
		$5\frac{1}{2}$	48191. 04	176. 86	

VI—Continued

VI—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>		
$3d^3\ 4s(b\ ^3G)4p$	<i>o</i> $^4F^\circ$	$1\frac{1}{2}$	47801. 6	114. 3 223. 5 189. 4	0. 89 1. 08	$3d^3\ 4p^2$	24°	$2\frac{1}{2}$	50130. 6	135. 13			
		$2\frac{1}{2}$	47915. 9				25°	$3\frac{1}{2}?$	50154. 35				
		$3\frac{1}{2}$	48139. 4				26°	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	50333. 59				
		$4\frac{1}{2}$	48328. 8				27°	$1\frac{1}{2}$	50355. 89				
	10°	$2\frac{1}{2}$	47809. 20				<i>r</i> $^2F^\circ$	$2\frac{1}{2}$	50404. 14				
	11°	$2\frac{1}{2}$	47925. 49:				$3\frac{1}{2}$	50539. 27					
	<i>q</i> $^2G^\circ$	$3\frac{1}{2}$	47959. 82	197. 75			28°	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	50438. 35				
		$4\frac{1}{2}$	48157. 57				<i>p</i> $^4G^\circ$	$2\frac{1}{2}$	50452. 6:				
	12°	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	48001. 8:				$3\frac{1}{2}$	50579. 6					
	13°	$3\frac{1}{2}$	48023. 68				$4\frac{1}{2}$	50742. 4					
	14°	$3\frac{1}{2}$	48047. 63				$5\frac{1}{2}$	50933. 58:					
	15°	$2\frac{1}{2}$	48070. 91				29°	$3\frac{1}{2}$	50529. 67				
	16°	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	48201. 79				<i>h</i> 6G	$1\frac{1}{2}$	50584. 27				
	17°	$3\frac{1}{2}?$	48289. 8				$2\frac{1}{2}$	50654. 72					
$3d^4(a\ ^3P)4p?$	<i>v</i> $^2S^\circ$	$\frac{1}{2}$	48844. 67	94. 03 156. 54 143. 78	2. 03		$3\frac{1}{2}$	50751. 83					
		$1\frac{1}{2}$	48844. 67			$4\frac{1}{2}$	50876. 00						
	18°	$2\frac{1}{2}$	48881. 48			$5\frac{1}{2}$	51026. 30						
	19°	$2\frac{1}{2}$	48964. 99			$6\frac{1}{2}$	51201. 12						
	20°	$1\frac{1}{2}$	49000. 82			30°	$3\frac{1}{2}$	50595. 73					
	<i>n</i> $^4D^\circ$	$\frac{1}{2}$	49189. 74			<i>n</i> $^4F^\circ$	$1\frac{1}{2}$	50909. 7					
		$1\frac{1}{2}$	49283. 77			$2\frac{1}{2}$	51021. 2						
		$2\frac{1}{2}$	49440. 31			$3\frac{1}{2}$	51174. 50						
		$3\frac{1}{2}$	49584. 09			$4\frac{1}{2}$	51366. 6						
	21°	$2\frac{1}{2}$	49302. 61			<i>m</i> $^4D^\circ$	$\frac{1}{2}$	50976. 5:					
	22°	$\left\{ \begin{smallmatrix} 3\frac{1}{2} \\ 4\frac{1}{2} \end{smallmatrix} \right\}$	49341. 90:			$1\frac{1}{2}$	51067. 7						
		$4\frac{1}{2}$	49341. 90:			$2\frac{1}{2}$	51212. 2:						
	$3d^3\ 4s(b\ ^3D)4s$	<i>t</i> $^2D^\circ$	$2\frac{1}{2}$			49689. 01	-33. 87	1. 25		$3\frac{1}{2}$	51398. 1:		
		$1\frac{1}{2}$	49722. 88			31°			$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	51194. 2			
$3d^3\ 4s(a\ ^5F)5d$	<i>f</i> 6H	$2\frac{1}{2}$	49717. 57	79. 61 77. 94 108. 04 181. 10 137. 37			32°	$1\frac{1}{2}$	51830. 69				
		$3\frac{1}{2}$	49797. 18			33°	$\left\{ \begin{smallmatrix} 2\frac{1}{2} \\ 3\frac{1}{2} \end{smallmatrix} \right\}$	52008. 09					
		$4\frac{1}{2}$	49875. 12			$3d^3\ 4s(b\ ^3H)4p?$	<i>p</i> $^2G^\circ$	$3\frac{1}{2}$	52774. 08				
		$5\frac{1}{2}$	49983. 16			$4\frac{1}{2}$	52947. 98						
		$6\frac{1}{2}$	50164. 26			$3d^3\ 4s(b\ ^3H)4p?$	<i>r</i> $^2H^\circ$	$4\frac{1}{2}$	54081. 51				
		$7\frac{1}{2}$	50301. 63			$5\frac{1}{2}$	54251. 26						
$3d^3\ 4s(a\ ^5F)5d$	<i>g</i> 6G	$1\frac{1}{2}$		143. 20 182. 22 94. 46									
		$2\frac{1}{2}$											
		$3\frac{1}{2}$	49789. 17										
		$4\frac{1}{2}$	49932. 37										
		$5\frac{1}{2}$	50114. 59										
$3d^3\ 4s(b\ ^3H)4p$		$6\frac{1}{2}$	50209. 05										
	<i>x</i> $^2I^\circ$	$5\frac{1}{2}$	49977. 90	142. 79	0. 91 1. 06								
		$6\frac{1}{2}$	50120. 69										
	23°	$3\frac{1}{2}?$	50090. 28										
						V II ($a\ ^5D_0$)	Limit	-----	54361				
							34°	$2\frac{1}{2}$	55202. 44				
							35°	$\left\{ \begin{smallmatrix} 1\frac{1}{2} \\ 2\frac{1}{2} \end{smallmatrix} \right\}$	55877. 82:				
							<i>s</i> $^2P^\circ$	$1\frac{1}{2}$	57561. 36?	-182. 76			
								$\frac{1}{2}$	57744. 12?				

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms	
$3d^3 4s^2$	$\left\{ \begin{array}{l} a^4P \\ a^2P \end{array} \right. \quad a^2D \quad a^4F \quad a^2G \quad a^2H$	
$3d^5$	$\left\{ a^6S \right. \quad e^4F$	
$3d^3 4p^2$	h^6G	
$ns \ (n \geq 4)$		
$3d^4(a^5D)nx$	$\left\{ \begin{array}{l} a, e^6D \\ a, e^4D \end{array} \right.$	
$3d^3 4s(a^5F)nx$	$\left\{ \begin{array}{l} e^6F \\ f^4F \end{array} \right.$	
$3d^3 4s(a^3F)nx$	e^2F	
$3d^4(a^3P)nx$	$\left\{ \begin{array}{l} b^4P \\ b^2P \end{array} \right.$	
$3d^4(a^3H)nx$	$\left\{ \begin{array}{l} a^4H \\ b^2H \end{array} \right.$	
$3d^4(b^3F)nx$	$\left\{ \begin{array}{l} b^4F \\ a^2F \end{array} \right.$	
$3d^4(a^3G)nx$	$\left\{ \begin{array}{l} a^4G \\ b^2G \end{array} \right.$	
$3d^4(a^3D)nx$	b^4D	
$np \ (n \geq 4)$		$nd \ (n \geq 4)$
$3d^4(a^5D)nx$	$\left\{ \begin{array}{l} z^6P^\circ \\ z^4P^\circ \end{array} \right. \quad y^6D^\circ \quad y^4D^\circ \quad y^6F^\circ \quad y^4F^\circ$	$e^6P \quad f^6D \quad f^6F \quad e^6G$
$3d^3 4s(a^5F)nx$	$\left\{ \begin{array}{l} z^6D^\circ \\ v^4D^\circ \end{array} \right. \quad z^6F^\circ \quad z^4F^\circ \quad z^6G^\circ \quad x^4G^\circ$	$f^6P \quad g^6D \quad g^6F \quad f, g^6G \quad e, f^6H$
$3d^3 4s(a^3F)nx$	$\left\{ \begin{array}{l} z^4D^\circ \\ z^2D^\circ \end{array} \right. \quad z^4F^\circ \quad z^2F^\circ \quad z^4G^\circ \quad z^2G^\circ$	
$3d^4(a^3P)nx$	$\left\{ \begin{array}{l} y^4S^\circ? \\ v^2S^\circ? \end{array} \right. \quad x^4P^\circ? \quad s^4D^\circ \quad v^2D^\circ$	
$3d^4(a^3H)nx$	$\left\{ \begin{array}{l} u^4G^\circ \\ v^2G^\circ \end{array} \right. \quad x^4H^\circ \quad x^2H^\circ \quad z^4I^\circ \quad z^2I^\circ?$	
$3d^4(b^3F)nx$	$\left\{ \begin{array}{l} t^4D^\circ \\ w^2D^\circ \end{array} \right. \quad u^4F^\circ \quad w^2F^\circ? \quad v^4G^\circ \quad x^2G^\circ$	
$3d^3 4s(a^3P)nx$	$\left\{ \begin{array}{l} z^6S^\circ \\ w^4S^\circ \end{array} \right. \quad y^6P^\circ \quad w^4P^\circ \quad x^6D^\circ \quad r^4D^\circ$	
$3d^4(a^3G)nx$	$\left\{ \begin{array}{l} t^4F^\circ \\ s^2F^\circ \end{array} \right. \quad t^4G^\circ \quad s^2G^\circ \quad w^4H^\circ \quad t^2H^\circ$	
$3d^3 4s(b^3G)nx$	$\left\{ \begin{array}{l} x^4F^\circ \\ s^2F^\circ \end{array} \right. \quad y^4G^\circ? \quad q^2G^\circ \quad z^4H^\circ \quad s^2H^\circ$	
$3d^4(a^3D)nx$	$o^4D^\circ \quad p^4F^\circ$	
$3d^3 4s(b^3G)nx$	$t^2F^\circ \quad t^2G^\circ? \quad u^2H^\circ$	
$3d^3 4s(c^3P)nx$	$x^4S^\circ \quad y^4P^\circ \quad w^4D^\circ$	
$3d^3 4s(b^3H)nx$	$\left\{ \begin{array}{l} w^4G^\circ? \\ p^2G^\circ? \end{array} \right. \quad y^4H^\circ? \quad r^2H^\circ? \quad x^2I^\circ$	
$3d^3 4s(b^3D)nx$	t^2D°	
$3d^3 4s(a^3P)nx$	$x^2S^\circ? \quad v^2P^\circ? \quad u^2D^\circ?$	
$3d^3 4s(a^3H)nx$	$r^2G^\circ? \quad v^2H^\circ? \quad y^2I^\circ?$	
$3d^3 4s(b^3D)nx$	$y^2P^\circ \quad u^2F^\circ?$	

*For predicted terms in the spectra of the VI isoelectronic sequence, see Introduction.

V II

(Ti I sequence; 22 electrons)

 $Z=23$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^4 {}^5D_0$ $a {}^5D_0$ 114600 cm^{-1}

I. P. 14.2 volts

The analysis is from the paper by Meggers and the writer, who published 89 terms and 1456 classified lines in the region from 1313 Å to 7015 Å. The terms of the three multiplicities are connected by observed intersystem combinations.

The g -values were calculated from unpublished data kindly furnished by Babcock and given in the 1940 reference below.

This is the first spectrum in which all theoretical terms (except the highest singlet, 1S), arising from the electron configuration d^4 have been established.

Many has discussed the configuration assignments and suggests from theoretical calculations that the term $c {}^1D$ at 44658 cm^{-1} be assigned to $3d^3 4s$. The two other terms which he criticizes, $b {}^3P$ and $c {}^3P$, were published in 1940 with precisely the limits he suggests.

Although intensively sought, series have not been found, probably because this spectrum has been observed only with condensed sparks at atmospheric pressure. The limit, entered in brackets in the table, was estimated by Russell from isoelectronic sequence data.

When the analysis of V III has been extended, the prefixes b , c , assigned by the writer to the limits may be changed. The limits here called $a {}^2F$, $b {}^2G$, and $c {}^2D$ have not yet been observed in V III.

REFERENCES

- H. N. Russell, *Astroph. J.* **66**, 233 (1927); *Mt. Wilson Contr. No.* 342 (1927). (I P)
 W. F. Meggers and C. E. Moore, *J. Research Nat. Bur. Std.* **25**, 83 RP1317 (1940). (I P) (T) (C L) (E D) (Z E)
 A. Many, *Phys. Rev.* **70**, 511 (1946).

V II

V II

Config.	Desig.	J	Level	Interval	Obs. g	Config.	Desig.	J	Level	Interval	Obs. g
$3d^4$	$a {}^5D$	0	0.00			$3d^4$	$a {}^3H$	4	12545.15		0.83:
		1	36.05	36.05				5	12621.57	76.42	1.02
		2	106.63	70.58				6	12706.15	84.58	1.27:
		3	208.89	102.26		$3d^4$	$b {}^3F$	2	13490.84		0.59
		4	339.21	130.32				3	13542.68	51.84	1.06
$3d^3(a {}^4F)4s$	$a {}^5F$							4	13609.00	66.32	1.19
		1	2604.82			$3d^3(a {}^4P)4s$	$a {}^5P$	1	13511.71		2.39
		2	2687.01	82.19	0.97			2	13594.73	83.02	1.78
		3	2808.76	121.75	1.20						
		4	2968.22	159.46	1.30:			3	13741.61	146.88	1.62
$3d^3(a {}^4F)4s$	$a {}^3F$	5	3162.80	194.58	1.28:	$3d^4$	$a {}^3G$				
		2	8640.21		0.65			3	14461.73		0.74
		3	8841.97	201.76	1.04			4	14556.09	94.36	1.00
$3d^4$	$a {}^3P$	4	9097.81	255.84	1.22	$3d^4$	$a {}^3G$	5	14655.63	99.54	1.17
		0	11295.51			$3d^3(a {}^2G)4s$	$b {}^3G$	3	16340.97		0.76
		1	11514.76	219.25	1.48			4	16421.51	80.54	1.03
		2	11908.27	393.51	1.49			5	16533.00	111.49	1.16

V II—Continued

V II—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$3d^4$	$a\ ^1G$	4	17910. 98		0. 95	$3d^3(a\ ^4F)4p$	$z\ ^3D^\circ$	1	36954. 58		0. 24
$3d^4$	$a\ ^3D$	1	18269. 49		0. 49			2	37041. 11	86. 53	1. 08
		2	18293. 87	24. 38	1. 13			3	37205. 01	163. 90	1. 32
		3	18353. 89	60. 02	1. 30	$3d^3(a\ ^4F)4p$	$z\ ^5D^\circ$	0	37201. 41		
$3d^3(a\ ^2G)4s$	$b\ ^1G$	4	19112. 93		0. 98			1	37259. 42	58. 01	1. 39
$3d^3(a\ ^2P)4s$	$b\ ^3P$	2	19132. 69		1. 38			2	37369. 01	109. 59	1. 39
		1	19166. 19	—33. 50	1. 40			3	37520. 61	151. 60	1. 47
		0	19161. 27	4. 92				4	37531. 09	10. 48	1. 44
$3d^4$	$a\ ^1I$	6	19191. 50		0. 96:	$3d^3(a\ ^4F)4p$	$z\ ^3G^\circ$	3	39234. 05		0. 84
$3d^4$	$a\ ^1S$	0	19902. 60					4	39403. 77	169. 72	1. 03
								5	39612. 97	209. 20	1. 19
$3d^3(a\ ^4P)4s$	$c\ ^3P$	0	20156. 64			$3d^3(a\ ^4F)4p$	$z\ ^3F^\circ$	2	40001. 66		0. 65
		1	20089. 56	—67. 08	1. 35			3	40195. 52	193. 86	1. 02
		2	20343. 00	253. 44	1. 36			4	40430. 10	234. 58	1. 22
$3d^3(a\ ^2H)4s$	$b\ ^3H$	4	20242. 32		0. 82	$3d^3(c\ ^2D)4s$	$c\ ^3D$	3	44098. 46		1. 27:
		5	20280. 19	37. 87	1. 01			2	44159. 43	—60. 97	1. 14:
		6	20363. 22	83. 03	1. 14			1	44200. 97: ?	—41. 54	0. 50:
$3d^3(a\ ^2D)4s$	$b\ ^3D$	1	20522. 14		0. 58	$3d^4$	$c\ ^1D$	2	44657. 99		
		2	20617. 05	94. 91	1. 25	$3d^3(a\ ^4P)4p$	$z\ ^3P^\circ$	0	46586. 43		
		3	20622. 99	5. 94	1. 26			1	46690. 43	104. 00	1. 44
$3d^4$	$a\ ^1D$	2	20980. 92		1. 02			2	46739. 98	49. 55	1. 48
$3d^3(a\ ^2P)4s$	$a\ ^1P$	1	22273. 54		0. 97	$3d^3(a\ ^4P)4p$	$z\ ^5P^\circ$	1	46754. 59		2. 28
$3d^3(a\ ^2H)4s$	$a\ ^1H$	5	23391. 09		1. 04			2	46879. 94	125. 35	1. 65
$3d^3(a\ ^2D)4s$	$b\ ^1D$	2	25191. 08		0. 99			3	47051. 89	171. 95	1. 58
$3d^4$	$a\ ^1F$	3	26839. 82		0. 97	$3d^3(a\ ^4P)4p$	$y\ ^5D^\circ$	0	47027. 88		
$3d^4$	$c\ ^3F$	2	30267. 46		0. 67			1	47107. 98	80. 10	1. 43
		3	30306. 40	38. 94	1. 06	$3d^3(a\ ^2G)4p$	$z\ ^3H^\circ$	4	47056. 32		
		4	30318. 63	12. 23	1. 25			5	47297. 08	240. 76	1. 01
$3d^3(a\ ^2F)4s$	$d\ ^3F$	4	30613. 97		1. 23			6	47607. 79	310. 71	1. 13
		3	30641. 71	—27. 74	1. 05	$3d^3(a\ ^2P)4p$	$z\ ^1S^\circ$	0	48258. 28		
		2	30673. 14	—31. 43	0. 67	$3d^3(a\ ^2G)4p$	$y\ ^3G^\circ$	3	48579. 96		0. 67
$3d^4$	$d\ ^3P$	2	32040. 76		1. 38			4	48730. 76	150. 80	1. 02
		1	32299. 24	—258. 48	1. 48			5	48853. 04	122. 28	1. 22
		0	32420. 04	—120. 80		$3d^3(a\ ^2G)4p$	$y\ ^3F^\circ$	2	49201. 66		0. 63
$3d^3(a\ ^2F)4s$	$b\ ^1F$	3	34228. 79		1. 00			3	49210. 78	9. 12	0. 99
$3d^3(a\ ^4F)4p$	$z\ ^5G^\circ$	2	34592. 72		0. 31			4	49268. 61	57. 83	1. 18
		3	34745. 72	153. 00	0. 93	$3d^3(a\ ^2G)4p$	$z\ ^1F^\circ$	3	49568. 45		0. 97
		4	34946. 55	200. 83	1. 14	$3d^3(a\ ^2G)4p$	$z\ ^1H^\circ$	5	49593. 41		0. 95
		5	35193. 13	246. 58	1. 16	$3d^3(a\ ^2G)4p$	$z\ ^1G^\circ$	4	49723. 68		0. 96
		6	35483. 39	290. 26		$3d^3(a\ ^4P)4p$	$z\ ^5S^\circ$	2	49731. 32		
$3d^4$	$c\ ^1G$	4	36425. 07		0. 96	$3d^3(a\ ^2P)4p$	$z\ ^1D^\circ$	2	49898. 22		0. 93
$3d^3(a\ ^4F)4p$	$z\ ^5F^\circ$	1	36489. 34		0. 35	$3d^3(a\ ^4P)4p$	$y\ ^3D^\circ$	1	50473. 76		0. 49
		2	36673. 51	184. 17	1. 08			2	50775. 47	301. 71	1. 11
		3	36919. 23	245. 72	1. 24			3	51085. 77	310. 30	1. 27
		4	37150. 57	231. 34		$3d^3(a\ ^2P)4p$	$y\ ^3P^\circ$	0	50662. 36		
		5	37352. 39	201. 82	1. 40:			1	50738. 82	76. 46	1. 39
								2	51123. 31	384. 49	1. 51

V II—Continued

V II—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$3d^3(a^2H)4p$	y^3H°	4 5 6	52082.88 52153.55 52252.70	70.67 99.15	0.70 0.98 1.04:	$3d^3(a^2F)4p$	x^1G°	4	65790.28		0.94
$3d^3(a^2P)4p$	z^3S°	1	52181.18		1.85	$3d^24s(b^2F)4p$	y^5G°	2 3 4 5 6	66228.4: 66667.3: 66962.7: 67356.0: 67795.7:?	438.9 295.4 393.3 439.7	
$3d^3(a^2D)4p$	x^3F°	2 3 4	52245.68 52391.94 52657.51	146.26 265.57	0.68 1.07 1.18:	$3d^3(a^2F)4p$	x^1F°	3	66303.88		0.95
$3d^3(a^2P)4p$	x^3D°	1 2 3	52604.11 52700.03 52767.86	95.92 67.33	0.63 1.10 1.26	$3d^24s(b^4F)4p$	v^3F°	2 3 4	67737.8 67905.1 68147.2	167.3 242.1	
$3d^3(a^2P)4p$	z^1P°	1	52803.75		0.92	$3d^24s(b^4F)4p$	u^3D°	1 2 3	68759.4 68797.7 68945.0	38.3 147.3	
$3d^3(a^2H)4p$	z^3I°	5 6 7	52877.99 53076.82 53319.52	198.83 242.70	0.84: 0.98 1.11:	$3d^24s(b^4F)4p$	v^3G°	3 4 5	69644.2 69912.1 70227.8	267.9 315.7	
$3d^3(a^2D)4p$	w^3D°	1 2 3	53751.46 53868.63 53927.19	117.17 58.56	0.49: 1.10 1.37	$3d^24s(b^2G)4p$	x^1H°	5	70936.4		
$3d^3(a^2H)4p$	y^1G°	4	54144.20		1.00	$3d^24s(b^2G)4p$	w^1G°	4	72292.2:		
$3d^3(a^2D)4p$	x^3P°	2 1 0	54715.63 54717.85 54813.45	—2.22 —95.60		$3d^3(a^4F)4d$	e^5H	3 4 5 6 7	72447.96: 72550.71 72680.20: 72837.00: 73020.35:	102.75 129.49 156.80 183.35	
$3d^3(a^2D)4p$	y^1F°	3	55142.01		0.94	$3d^3(a^4F)4d$	e^5P	1 2 3	72517.84: 72674.28 72908.17	156.44 233.89	
$3d^3(a^2H)4p$	x^3G°	5 4 3	55206.87 55304.34 55349.63	—97.47 —45.29	1.15 1.02 0.82	$3d^3(a^4F)4d$	e^5D	0 1 2 3 4	72682.06:? 72789.23:? 72951.00:	107.17 161.77	
$3d^3(a^2H)4p$	z^1I°	6	55403.38		1.01:	$3d^3(a^4F)4d$	e^5G	2 3 4 5 6	73026.76 73145.68 73278.92 73416.63 73498.93:	118.92 133.24 137.71 82.30	
$3d^3(a^2H)4p$	y^1H°	5	55499.38		1.03:	$3d^3(a^4F)4d$	e^5F	1 2 3 4 5	73222.72: 73293.82:?	71.10	
$3d^3(a^4P)4p$	y^3S°	1	55663.27		1.92	$3d^24s(b^2G)4p$	w^1F°	3	74664.5		
$3d^3(a^2D)4p$	y^1P°	1	56171.49		1.05:	$3d^3(c^2D)4p?$	t^3D°	1 2 3	75715.45:? 75758.29 75848.13	42.84 89.84	0.50: 1.14: 1.27:
$3d^3(a^2D)4p$	y^1D°	2	57342.59		0.98		u^3F°	2 3 4	76220.4 76385.8 76643.5	165.4 257.7	
$3d^3(a^2F)4p$	w^3F°	2 3 4	62085.02 62133.39 62176.24	48.37 42.85	0.58: 1.00 1.36:		2°	3	76405.4		
	1°	4	62761.9				w^1D°	2	78791.3:		
$3d^24s(b^4F)4p$	y^5F°	1 2 3 4 5	63548.5: 63657.2 63816.9 64026.6 64287.1	108.7 159.7 209.7 260.5			3°	3	79040.4		
$3d^3(a^2F)4p$	w^3G°	3 4 5	64057.39 64130.84 64229.10	73.45 98.26	0.72: 1.02						
$3d^2(a^2F)4p$	x^1D°	2	64586.23		1.03:						
$3d^3(a^2F)4p$	v^3D°	3 2 1	64603.53 64804.13 64930.76	—200.60 —126.63	1.22: 1.02: 0.46:						
$3d^24s(b^4F)4p$	x^5D°	0 1 2 3 4	65783.4 65816.2 65885.3 65996.7 66158.6	32.8 69.1 111.4 161.9							
						V III ($a^4F_{1/2}$)	Limit	----	[114600]		

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms	
$3d^4$	$\left\{ \begin{array}{cccccc} a^5D & b^3F & a^3G & a^3H & & \\ a^3P & a^3D & & & & \\ d^3P & c^3F & & & & \\ a^1S & a^1D & a^1F & a^1G & a^1I & \\ & c^1D & & c^1G & & \end{array} \right.$	
	$ns (n \geq 4)$	$np (n \geq 4)$
$3d^3(a^4F)nx$	$\left\{ \begin{array}{c} a^5F \\ a^3F \end{array} \right.$	$\begin{array}{ccc} z^5D^\circ & z^5F^\circ & z^5G^\circ \\ z^3D^\circ & z^3F^\circ & z^3G^\circ \end{array}$
$3d^3(a^2P)nx$	$\left\{ \begin{array}{c} b^3P \\ a^1P \end{array} \right.$	$\begin{array}{ccc} z^3S^\circ & y^3P^\circ & x^3D^\circ \\ z^1S^\circ & z^1P^\circ & z^1D^\circ \end{array}$
$3d^3(a^4P)nx$	$\left\{ \begin{array}{c} a^5P \\ c^3P \end{array} \right.$	$\begin{array}{ccc} z^5S^\circ & z^5P^\circ & y^5D^\circ \\ y^3S^\circ & z^3P^\circ & y^3D^\circ \end{array}$
$3d^3(a^2G)nx$	$\left\{ \begin{array}{c} b^3G \\ b^1G \end{array} \right.$	$\begin{array}{ccc} y^3F^\circ & y^3G^\circ & z^3H^\circ \\ z^1F^\circ & z^1G^\circ & z^1H^\circ \end{array}$
$3d^3(a^2D)nx$	$\left\{ \begin{array}{c} b^3D \\ b^1D \end{array} \right.$	$\begin{array}{ccc} x^3P^\circ & w^3D^\circ & x^3F^\circ \\ y^1P^\circ & y^1D^\circ & y^1F^\circ \end{array}$
$3d^3(a^2H)nx$	$\left\{ \begin{array}{c} b^3H \\ a^1H \end{array} \right.$	$\begin{array}{ccc} & & x^3G^\circ \\ & & y^1G^\circ \end{array}$ $\begin{array}{ccc} y^3H^\circ & z^3I^\circ & \\ y^1H^\circ & z^1I^\circ & \end{array}$
$3d^3(a^2F)nx$	$\left\{ \begin{array}{c} d^3F \\ b^1F \end{array} \right.$	$\begin{array}{ccc} v^3D^\circ & w^3F^\circ & w^3G^\circ \\ x^1D^\circ & x^1F^\circ & x^1G^\circ \end{array}$
$3d^2 4s(b^4F)nx$	$\left\{ \right.$	$\begin{array}{ccc} x^5D^\circ & y^5F^\circ & y^5G^\circ \\ u^3D^\circ & v^3F^\circ & v^3G^\circ \end{array}$
$3d^2 4s(b^2G)nx$		$\begin{array}{ccc} w^1F^\circ & w^1G^\circ & x^1H^\circ \end{array}$
$3d^3(c^2D)nx$	c^3D	t^3D°
	$nd (n \geq 4)$	
$3d^3(a^4F)nx$	$e^5P \quad e^5D \quad e^5F \quad e^5G \quad e^5H$	

*A chart of predicted terms in the spectra of the Ti I isoelectronic sequence is given in the Introduction. Owing to the differences in binding energy of the $3d$ and $4s$ electrons the arrangement of the charts of predicted and observed terms is different for V II.

V III

(Sc I sequence; 21 electrons)

 $Z=23$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^3 {}^4F_{1/2}$ $a {}^4F_{1/2} 246000 \text{ cm}^{-1}$

I.P. 29.7 volts

The analysis is by White, who has classified 120 lines in the interval between 1117 Å and 2595 Å. The limit (entered in brackets in the table) is derived from his extrapolation of isoelectronic sequence data.

The doublet and quartet terms are connected by observed intersystem combinations.

The reality of the term $a {}^2P$ is questioned in the paper by Many.

REFERENCES

- H. E. White, Phys. Rev. **33**, 672 (1929). (I P) (T) (C L)
A. Many, Phys. Rev. **70**, 513 (1946).

V III

V III

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3d^3$	$a\ ^4F$	$1\frac{1}{2}$	0	145	$3d^2(a\ ^3F)4p$	$z\ ^4F^\circ$	$1\frac{1}{2}$	86716	221
		$2\frac{1}{2}$	145	194			$2\frac{1}{2}$	86937	281
		$3\frac{1}{2}$	339	244			$3\frac{1}{2}$	87218	326
		$4\frac{1}{2}$	583				$4\frac{1}{2}$	87544	
$3d^3$	$a\ ^2P$	$\frac{1}{2}$	11207	180	$3d^2(a\ ^3F)4p$	$z\ ^2F^\circ$	$2\frac{1}{2}$	87881	448
		$1\frac{1}{2}$	11387				$3\frac{1}{2}$	88329	
$3d^3$	$a\ ^4P$	$\frac{1}{2}$	11513	77	$3d^2(a\ ^3F)4p$	$z\ ^2D^\circ$	$1\frac{1}{2}$	88560	386
		$1\frac{1}{2}$	11590	181			$2\frac{1}{2}$	88946	
		$2\frac{1}{2}$	11771						
$3d^3$	$a\ ^2G$	$3\frac{1}{2}$	11966	221	$3d^2(a\ ^3F)4p$	$z\ ^4D^\circ$	$\frac{1}{2}$	89004	187
		$4\frac{1}{2}$	12187				$1\frac{1}{2}$	89191	267
$3d^3$	$a\ ^2D$	$1\frac{1}{2}$	16229	147			$2\frac{1}{2}$	89458	—40
		$2\frac{1}{2}$	16376				$3\frac{1}{2}$	89418	
$3d^3$	$a\ ^2H$	$4\frac{1}{2}$	16822	155	$3d^2(a\ ^3F)4d$	$e\ ^4H$	$3\frac{1}{2}$	141269	217
		$5\frac{1}{2}$	16977				$4\frac{1}{2}$	141486	247
$3d^2(a\ ^3F)4s$	$b\ ^4F$	$1\frac{1}{2}$	43941	167			$5\frac{1}{2}$	141733	258
		$2\frac{1}{2}$	44108	236			$6\frac{1}{2}$	141991	
		$3\frac{1}{2}$	44344	301					
		$4\frac{1}{2}$	44645						
$3d^2(a\ ^3F)4s$	$b\ ^2F$	$2\frac{1}{2}$	49329	478	-----	-----	---	-----	
		$3\frac{1}{2}$	49807		V IV ($a\ ^3F_2$)	Limit	---	[240000]	
$3d^2(a\ ^3F)4p$	$z\ ^4G^\circ$	$2\frac{1}{2}$	85523	351					
		$3\frac{1}{2}$	85874	431					
		$4\frac{1}{2}$	86305	503					
		$5\frac{1}{2}$	86808						

June 1948.

V III OBSERVED TERMS*

Config. $1s^2\ 2s^2\ 2p^6\ 3s^2\ 3p^6+$	Observed Terms		
$3d^3$	{ $a\ ^4P$ $a\ ^4F$ $a\ ^2P$ $a\ ^2D$ $a\ ^2G$ $a\ ^2H$		
	$ns\ (n\geq 4)$ $np\ (n\geq 4)$ $nd\ (n\geq 4)$		
	{ $b\ ^4F$ $z\ ^4D^\circ$ $z\ ^4F^\circ$ $z\ ^4G^\circ$ $b\ ^2F$ $z\ ^2D^\circ$ $z\ ^2F^\circ$ $z\ ^2G^\circ$ $e\ ^4H$		
$3d^2(a\ ^3F)nx$	{ $b\ ^4F$ $z\ ^4D^\circ$ $z\ ^4F^\circ$ $z\ ^4G^\circ$ $b\ ^2F$ $z\ ^2D^\circ$ $z\ ^2F^\circ$ $z\ ^2G^\circ$ $e\ ^4H$		

*For predicted terms in the spectra of the Sc I isoelectronic sequence, see Introduction.

V IV

(Ca I sequence; 20 electrons)

 $Z=23$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^2 {}^3F_2$ $a {}^3F_2$ 391000 cm^{-1}

I. P. 48 volts

White has classified 64 lines in the region between 675 Å and 2269 Å, and extrapolated the limit from isoelectronic sequence data. The limit derived from his ionization potential is entered in brackets in the table.

From a study of related spectra, Edlén has rejected White's $3d {}^1S_0$ term, and his four intersystem combinations. Edlén suggests that the line observed at 734.36 Å (136173 cm^{-1}) may be designated a $a {}^1D_2 - z {}^3F_2$, which decreases White's singlet terms by 698 cm^{-1} . This change has been adopted here.

REFERENCE

H. E. White, Phys. Rev. **33**, 538 (1929). (I P) (T) (C L)

B. Edlén- unpublished material (Feb. 1949). (T) (C L)

V IV

V IV

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3d^2$	$a {}^3F$	2 3 4	0 318 730	318 412	$3d({}^2D)4p$	$z {}^3F^\circ$	2 3 4	147133 147653 148365	520 712
$3d^2$	$a {}^1D$	2	10960		$3d({}^2D)4p$	$z {}^3P^\circ$	0 1 2	151446 151424 151564	-22 140
$3d^2$	$a {}^3P$	0 1 2	13121 13238 13453	117 215	$3d({}^2D)4p$	$z {}^1F^\circ$	3	153920	
$3d^2$	$a {}^1G$	4	18389		$3d({}^2D)4p$	$z {}^1P^\circ$	1	155567	
$3d({}^2D)4s$	$a {}^3D$	1 2 3	96195 96410 96795	215 385	$3d({}^2D)4d$	$e {}^3G$	3 4 5	217835 218097 218461	262 364
$3d({}^2D)4s$	$b {}^1D$	2	100204		$3d({}^2D)4d$	$e {}^3F$	2 3 4	223510 223833 224263	323 430
$3d({}^2D)4p$	$z {}^1D^\circ$	0	144276						
$3d({}^2D)4p$	$z {}^3D^\circ$	1 2 3	146116 146426 146851	310 425					
					V v (${}^2D_{1\frac{1}{2}}$)	Limit	---	[391000]	

Feb. 1949.

V IV OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms		
$3d^2$	{ $a {}^3P$ $a {}^1D$ $a {}^3F$ $a {}^1G$ }		
	$ns (n \geq 4)$	$np (n \geq 4)$	$nd (n \geq 4)$
$3d({}^2D)nx$	{ $a {}^3D$ $b {}^1D$ }	$z {}^3P^\circ$ $z {}^1D^\circ$ $z {}^3F^\circ$ $z {}^1P^\circ$ $z {}^1D^\circ$ $z {}^1F^\circ$	$e {}^3F$ $e {}^3G$

*A chart of predicted terms in the spectra of the Ca I isoelectronic sequence is given in the Introduction. Owing to the change in binding energies of the $3d$ and $4s$ electrons along this sequence, the arrangement of the charts of observed and predicted terms is not identical. In V IV the prefixes a , b , . . . e , z replace those indicating the running electron.

(K I sequence; 19 electrons)

 $Z=23$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d \ ^2D_{1\frac{1}{2}}$ $3d \ ^2D_{1\frac{1}{2}} \ 526000 \text{ cm}^{-1}$

I. P. 65.2 volts

The terms have been calculated from the data published by Gibbs and White, who classified 11 lines in the region between 286 Å and 1716 Å. From these data Kruger and Weissberg have calculated the limit by fitting a Ritz-Rydberg formula to the 2S terms. Their limit in round numbers is quoted here.

REFERENCES

- R. C. Gibbs and H. E. White, Phys. Rev. **33**, 162 (1929). (C L)
 P. G. Kruger and S. G. Weissberg, Phys. Rev. **52**, 317 (1937). (I P)

V v

Config.	Desig.	J	Level	Interval
$3p^6(^1S)3d$	$3d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	0 620	620
$3p^6(^1S)4s$	$4s \ ^2S$	$\frac{1}{2}$	148100	
$3p^6(^1S)4p$	$4p \ ^2P^o$	$\frac{1}{2}$ $1\frac{1}{2}$	206347 207617	1270
$3p^6(^1S)5s$	$5s \ ^2S$	$\frac{1}{2}$	328167	
$3p^6(^1S)4f$	$4f \ ^2F^o$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	349204	
$3p^6(^1S)6s$	$6s \ ^2S$	$\frac{1}{2}$	403933	
-----	-----	---	-----	
V v I (1S_0)	<i>Limit</i>	---	526000	

May 1948.

V VI

(A I sequence; 18 electrons)

 $Z=23$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 \ ^1S_0$ $3p^6 \ ^1S_0 \ 1040100 \text{ cm}^{-1}$

I. P. 128.9 volts

Four lines are classified in the region between 128 Å and 182 Å, as combinations with the ground term. The values listed in the table have been rounded off in the last places.

For convenience, the Paschen notation has been added by the writer in column one under the heading "A I". As for A I, the jl -coupling notation in the general form suggested by Racah is here introduced, although LS -designations, as indicated in column two under the heading "Authors", are perhaps preferable for the terms thus far identified.

REFERENCES

- P. G. Kruger and S. G. Weissberg, Phys. Rev. **48**, 659 (1935). (I P) (T) (C L)
 P. G. Kruger, S. G. Weissberg and L. W. Phillips, Phys. Rev. **51**, 1090 (1937). (I P) (T)
 G. Racah, Phys. Rev. **61**, 537 (L) (1942).

V VI

A I	Authors	Config.	Desig.	J	Level
$1p_0$	$3p^6 \ ^1S$	$3p^6$	$3p^6 \ ^1S$	0	0
$1s_4$	$3p^5 \ 4s \ ^3P^o$	$3p^5(^2P_{1/2}^o)4s$	$4s \ [1\frac{1}{2}]^o$	2 1	549300
$1s_2$	$3p^5 \ 4s \ ^1P^o$	$3p^5(^2P_{3/2}^o)4s$	$4s' \ [\frac{1}{2}]^o$	0 1	557650
$2s_4$	$3p^5 \ 5s \ ^3P^o$	$3p^5(^2P_{1/2}^o)5s$	$5s \ [1\frac{1}{2}]^o$	2 1	771760
$2s_2$	$3p^5 \ 5s \ ^1P^o$	$3p^5(^2P_{3/2}^o)5s$	$5s' \ [\frac{1}{2}]^o$	0 1	778920
		-----	-----	---	-----
		V VII ($^2P_{1/2}^o$)	<i>Limit</i>	---	1040100
		V VII ($^2P_{3/2}^o$)	<i>Limit</i>	---	1047760

May 1948.

V VII

(Cl I sequence; 17 electrons)

 $Z=23$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^5 \ ^2P_{1/2}^o$ $3p^5 \ ^2P_{1/2}^o$ 1216000 cm^{-1}

I. P. 151 volts

All of the terms except $3p^6 \ ^2S$ are from the paper by Edlén. Thirteen lines in the region between 148 Å and 472 Å have been classified as combinations from the ground state. Edlén has estimated the value of the limit by extrapolation along the isoelectronic sequence, as indicated by brackets in the table. His unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

S. G. Weissberg and P. G. Kruger, Phys. Rev. **49**, 872 (A) (1936). (C L)
 B. Edlén, Zeit. Phys. **104**, 407 (1937). (I P) (T) (C L)

V VII

Config.	Desig.	J	Level	Interval
$3s^2 3p^5$	$3p^5 \ ^2P^o$	$1\frac{1}{2}$ $\frac{1}{2}$	0 7660	—7660
$3s \ 3p^6$	$3p^6 \ ^2S$	$\frac{1}{2}$	219160	
$3s^2 3p^4(^3P)4s$	$4s \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $\frac{1}{2}$	608640 612810 615480	—4170 —2670
$3s^2 3p^4(^3P)4s$	$4s \ ^2P$	$1\frac{1}{2}$ $\frac{1}{2}$	620650 625570	—4920
$3s^2 3p^4(^1D)4s$	$4s' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	638540 638710	—170
$3s^2 3p^4(^1S)4s$	$4s'' \ ^2S$	$\frac{1}{2}$	671580	
-----	-----	---	-----	
V VIII (3P_2)	<i>Limit</i>	---	[1216000]	

January 1948.

V VIII

(S I sequence; 16 electrons)

 $Z=23$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^4 {}^3P_2$ $3p^4 {}^3P_2$ 1401000 cm^{-1}

I. P. 173.7 volts

The analysis is by Edlén, who has classified 19 lines in the range between 135 Å and 147 Å. He has extrapolated the limit from isoelectronic sequence data. The singlet and triplet terms are connected by two observed intersystem combinations.

Edlén's unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCE

B. Edlén, Zeit. Phys. **104**, 188 (1937). (I P) (T) (C L)

V VIII

V VIII

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3s^2 3p^4$	$3p^4 {}^3P$	2	0	-6000 -1580	$3s^2 3p^3 ({}^2D^\circ) 4s$	$4s' {}^1D^\circ$	2	718450	630 1770
		1	6000						
		0	7580						
$3s^2 3p^4$	$3p^4 {}^1D$	2	27120		$3s^2 3p^3 ({}^2P^\circ) 4s$	$4s'' {}^3P^\circ$	0	734246	
							1	734870	
							2	736640	
$3s^2 3p^4$	$3p^4 {}^1S$	0	60720		$3s^2 3p^3 ({}^2P^\circ) 4s$	$4s'' {}^1P^\circ$	1	742790	
$3s^2 3p^3 ({}^4S^\circ) 4s$	$4s {}^3S^\circ$	1	687250		-----	-----	---	-----	
$3s^2 3p^3 ({}^2D^\circ) 4s$	$4s' {}^3D^\circ$	1	710600	310 1080	V IX (${}^4S_{3/2}$)	Limit	---	1401000	
		2	710910						
		3	711990						

January 1948.

V IX

(P I sequence; 15 electrons)

 $Z=23$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^3 {}^4S_{1/2}^\circ$ $3p^3 {}^4S_{1/2}^\circ$ cm^{-1}

I. P. volts

Kruger and Pattin have observed 6 lines near 126 Å, and arranged them in two multiplets that give intervals consistent with those found in related isoelectronic spectra.

By a rough extrapolation of $3p^3 {}^4S_{1/2}^\circ - 3p^3 {}^2D_{1/2}^\circ$ along the isoelectronic sequence, the writer has estimated the value of $3p^3 {}^2D_{1/2}^\circ$ (entered in brackets in the table), and calculated the terms listed below from the multiplets given by Kruger and Pattin. The uncertainty x in the estimated position of the doublet terms relative to the quartets may exceed $\pm 500 \text{ cm}^{-1}$.

REFERENCE

P. G. Kruger and H. S. Pattin, Phys. Rev. **52**, 624 (1937). (C L)

V IX

Config.	Desig.	J	Level	Interval
$3s^2 3p^3$	$3p^3 \ ^4S^\circ$	$1\frac{1}{2}$	0	
$3s^2 3p^3$	$3p^3 \ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	$[36000]+x$ $37520+x$	1520
$3s^2 3p^2(^3P)4s$	$4s \ ^4P$	$\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	789070 792700 797320	3630 4620
$3s^2 3p^2(^1D)4s$	$4s' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	$824500+x$ $824860+x$	-360

December 1947.

V XI

(Al I sequence; 13 electrons)

 $Z=23$ Ground state $1s^2 2s^2 2p^6 3s^2 3p \ ^2P_{\frac{1}{2}}$ $3p \ ^2P_{\frac{1}{2}}^o$ cm^{-1}

I. P. volts

This spectrum has not been analyzed, but Edlén has classified two lines as follows:

I. A.	Int.	Wave No.	Desig.
87. 166 87. 868	3 4	1147240 1138070	$\left. \vphantom{\begin{matrix} 1147240 \\ 1138070 \end{matrix}} \right\} 3p \ ^2P^\circ - 4d \ ^2D$

His unit, 10^3 cm^{-1} , is here changed to cm^{-1} .

REFERENCE

B. Edlén, Zeit. Phys. **103**, 540 (1936). (C L)

December 1947.

V XII

(Mg I sequence; 12 electrons)

 $Z=23$ Ground state $1s^2 2s^2 2p^6 3s^2 \ ^1S_0$ $3s^2 \ ^1S_0$ 2490000 cm^{-1}

I. P. 309 volts

Edlén has classified 15 lines in the region between 61 Å and 106 Å. No intersystem combinations have been observed, and the triplet terms are not all connected by observed combinations. He has determined the relative positions of the various groups of terms and also the ionization potential by extrapolation along the isoelectronic sequence. His estimated value of the limit is entered in brackets in the table.

His unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCE

B. Edlén, Zeit. Phys. **103**, 536 (1936). (I P) (T) (C L)

V XII

V XII

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
3s ²	3s ² 1S	0	0		3s(2S)4f	4f 3F°	2 3 4	1485160+x	
3s(2S)3p	3p 3P°	0 1 2	188350+x 191450+x 198610+x	3100 7160	3s(2S)5d	5d 3D	1 2 3	1818660+x 1818910+x	250
3s(2S)3d	3d 3D	1 2 3	549580+x		3s(2S)5f	5f 3F°	2 3 4	1848960+x	
3s(2S)4s	4s 3S	1	1212500+x		-----	-----	---	-----	
3s(2S)4p	4p 1P°	1	1310500		V XIII (2S _½)	Limit	---	[2490000]	
3s(2S)4d	4d 3D	1 2 3	1424530+x 1424850+x 1425410+x	320 560					

August 1947.

V XIII

(Na I sequence; 11 electrons)

Z=23

Ground state 1s² 2s² 2p⁶ 3s 2S_½

3s 2S_½ 2713130 cm⁻¹

I. P. 336.29 volts

Edlén has classified 15 lines in the interval 52 Å to 99 Å, and extrapolated the absolute value of the ground term from isoelectronic sequence data.

The unit adopted by Edlén, 10³ cm⁻¹, has here been changed to cm⁻¹.

REFERENCE

B. Edlén, Zeit. Phys. **100**, 621 (1936). (I P) (T) (C L)

V XIII

V XIII

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
3s	3s 2S	½	0		4f	4f 2F°	2½ 3½	1550290 1550510	220
3p	3p 2P°	½ 1½	225350 236430	11080	5p	5p 2P°	½ 1½	1889360 1891430	2070
3d	3d 2D	1½ 2½	545500 546730	1230	5d	5d 2D	1½ 2½	1946050 1946360	310
4s	4s 2S	½	1300330		5f	5f 2F°	2½ 3½	1968740	
4p	4p 2P°	½ 1½	1388410 1392780	4370	-----	-----	---	-----	
4d	4d 2D	1½ 2½	1505740 1506340	600	V XIV (1S ₀)	Limit	---	2713130	

June 1947.

V XIV

(Ne I sequence; 10 electrons)

 $Z=23$ Ground state $1s^2 2s^2 2p^6 {}^1S_0$ $2p^6 {}^1S_0$ 7237600 cm^{-1}

I. P. 897.1 volts

Edlén and Tyrén have classified four lines in the region between 20 Å and 23 Å, as combinations with the ground term. They have derived absolute term values by extrapolation along the Ne I isoelectronic sequence. Their unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

As for Ne I, the jl -coupling notation in the general form suggested by Racah is introduced.

REFERENCES

- B. Edlén and F. Tyrén, Zeit. Phys. **101**, 210 (1936). (I P) (T) (C L).
 G. Racah, Phys. Rev. **61**, 537 (L) (1942).

V XIV

Authors	Config.	Desig.	J	Level
$2p {}^1S_0$	$2p^6$	$2p^6 {}^1S$	0	0
$3s {}^3P_1$	$2p^5({}^2P_{1/2})3s$	$3s [1\frac{1}{2}]^\circ$	$\begin{smallmatrix} 2 \\ 1 \end{smallmatrix}$	4202700
$3s {}^1P_1$	$2p^5({}^2P_{3/2}^\circ)3s$	$3s' [\frac{1}{2}]^\circ$	$\begin{smallmatrix} 0 \\ 1 \end{smallmatrix}$	4257100
$3d {}^1P_1$	$2p^5({}^2P_{1/2})3d$	$3d [1\frac{1}{2}]^\circ$	1	4757800
$3d {}^3D_1$	$2p^5({}^2P_{3/2})3d$	$3d' [1\frac{1}{2}]^\circ$	1	4827200
	-----	-----	---	-----
	V xv (${}^2P_{1/2}$)	Limit	---	7237600
	V xv (${}^2P_{3/2}^\circ$)	Limit	---	7295300

April 1947.

U.S. DEPT. OF COMM. BIBLIOGRAPHIC DATA SHEET		1. PUBLICATION OR REPORT NO. NBS-NSRDS 35, Vol. I	2. Gov't Accession No.	3. Recipient's Accession No.
4. TITLE AND SUBTITLE Atomic Energy Levels as Derived from the Analyses of Optical Spectra. Vol. I, ^1H to ^{23}V			5. Publication Date December	
			6. Performing Organization Code	
7. AUTHOR(S) Charlotte E. Moore			8. Performing Organization	
9. PERFORMING ORGANIZATION NAME AND ADDRESS NATIONAL BUREAU OF STANDARDS DEPARTMENT OF COMMERCE WASHINGTON, D.C. 20234			10. Project/Task/Work Unit No.	
			11. Contract/Grant No.	
12. Sponsoring Organization Name and Address Same			13. Type of Report & Period Covered Final	
			14. Sponsoring Agency Code	
15. SUPPLEMENTARY NOTES				
16. ABSTRACT (A 200-word or less factual summary of most significant information. If document includes a significant bibliography or literature survey, mention it here.) This series of three volumes is a critical compilation of atomic energy levels prepared at the National Bureau of Standards from the analyses of optical spectra. Volume I contains data on the spectra of hydrogen, deuterium, tritium, helium, lithium, beryllium, boron, carbon, nitrogen, oxygen, fluorine, neon, sodium, magnesium, aluminum, silicon, phosphorus, sulfur, chlorine, argon, potassium, calcium, scandium, titanium, and vanadium (^1H to ^{23}V). Volume II covers the spectra of chromium, manganese, iron, cobalt, nickel, copper, zinc, gallium, germanium, arsenic, selenium, bromine, krypton, rubidium, strontium, yttrium, zirconium, and niobium (^{24}Cr to ^{41}Nb). Volume III includes the spectra of molybdenum, technetium, ruthenium, rhodium, palladium, silver, cadmium, indium, tin, antimony, tellurium, iodine, xenon, cesium, barium, lanthanum; hafnium, tantalum, tungsten, rhenium, osmium, iridium, platinum, gold, mercury, thallium, lead, bismuth, polonium, radon, radium, and actinium (^{42}Mo to ^{57}La ; ^{72}Hf to ^{89}Ac).				
17. KEY WORDS (Alphabetical order, separated by semicolons) Energy levels; H-V.				
18. AVAILABILITY STATEMENT <input checked="" type="checkbox"/> UNLIMITED. <input type="checkbox"/> FOR OFFICIAL DISTRIBUTION. DO NOT RELEASE TO NTIS.		19. SECURITY CLASS (THIS REPORT) UNCLASSIFIED	21. NO. OF PAGES 359	
		20. SECURITY CLASS (THIS PAGE) UNCLASSIFIED	22. Price \$5.00	

Publications in the National Standard Reference Data Series National Bureau of Standards

You may use this listing as your order form by checking the proper box of the publication(s) you desire or by providing the full identification of the publication you wish to purchase. The full letter symbols with each publication number and full title of the publication and author must be given in your order, e.g. NSRDS-NBS-21, Kinetic Data on Gas Phase Unimolecular Reactions, by S. W. Benson and H. E. O'Neal.

Pay for publications by check, money order, or Superintendent of Documents coupons or deposit account. Make checks and money orders payable to Superintendent of Documents. Foreign remittances should be made either

by international money order or draft on an American bank. Postage stamps are not acceptable.

No charge is made for postage to destinations in the United States and possessions, Canada, Mexico, and certain Central and South American countries. To other countries, payments for documents must cover postage. Therefore, one-fourth of the price of the publication should be added for postage.

Send your order together with remittance to Superintendent of Documents, Government Printing Office, Washington, D.C. 20402.

- ☐ NSRDS-NBS 1, **National Standard Reference Data System—Plan of Operation**, by E. L. Brady and M. B. Wallenstein, 1964 (15 cents), SD Catalog No. C13.48:1.
- ☐ NSRDS-NBS 2, **Thermal Properties of Aqueous Uni-univalent Electrolytes**, by V. B. Parker, 1965 (45 cents), SD Catalog No. C13.48:2.
- ☐ NSRDS-NBS 3, Sec. 1, **Selected Tables of Atomic Spectra, Atomic Energy Levels and Multiplet Tables, Si II, Si III, Si IV**, by C. E. Moore, 1965 (35 cents), SD Catalog No. C13.48:3/Sec.1.
- ☐ NSRDS-NBS 3, Sec. 2, **Selected Tables of Atomic Spectra, Atomic Energy Levels and Multiplet Tables, Si I**, by C. E. Moore, 1967 (20 cents), SD Catalog No. C13.48:3/Sec.2.
- ☐ NSRDS-NBS 3, Sec. 3, **Selected Tables of Atomic Spectra, Atomic Energy Levels and Multiplet Tables, C I, C II, C III, C IV, C V, C VI**, by C. E. Moore, 1970 (\$1), SD Catalog No. C13.48:3/Sec.3.
- ☐ NSRDS-NBS 3, Sec. 4, **Selected Tables of Atomic Spectra, Atomic Energy Levels and Multiplet Tables, N IV, N V, N VI, N VII**, by C. E. Moore, 1971 (55 cents), SD Catalog No. C13.48:3/Sec.4.
- ☐ NSRDS-NBS 4, **Atomic Transition Probabilities, Vol. I, Hydrogen Through Neon**, by W. L. Wiese, M. W. Smith, and B. M. Glennon, 1966 (\$2.50), SD Catalog No. C13.48:4/Vol.1.
- ☐ NSRDS-NBS 5, **The Band Spectrum of Carbon Monoxide**, by P. H. Krupenie, 1966 (70 cents), SD Catalog No. C13.48:5.
- ☐ NSRDS-NBS 6, **Tables of Molecular Vibrational Frequencies, Part 1**, by T. Shimanouchi, 1967 (40 cents), SD Catalog No. C13.48:6/Pt.1.
- ☐ NSRDS-NBS 7, **High Temperature Properties and Decomposition of Inorganic Salts, Part 1. Sulfates**, by K. H. Stern and E. L. Weise, 1966 (35 cents), SD Catalog No. C13.48:7/Pt.1.
- ☐ NSRDS-NBS 8, **Thermal Conductivity of Selected Materials**, by R. W. Powell, C. Y. Ho, and P. E. Liley, 1966 (\$3). PB189698*
- ☐ NSRDS-NBS 9, **Tables of Bimolecular Gas Reactions**, by A. F. Trotman-Dickenson and G. S. Milne, 1967 (\$2), SD Catalog No. C13.48:9.
- ☐ NSRDS-NBS 10, **Selected Values of Electric Dipole Moments for Molecules in the Gas Phase**, by R. D. Nelson, Jr., D. R. Lide, Jr., and A. A. Maryott, 1967 (40 cents), SD Catalog No. C13.48:10.
- ☐ NSRDS-NBS 11, **Tables of Molecular Vibrational Frequencies, Part 2**, by T. Shimanouchi, 1967 (30 cents), SD Catalog No. C13.48:11/Pt.2.
- ☐ NSRDS-NBS 12, **Tables for the Rigid Asymmetric Rotor: Transformation Coefficients from Symmetric to Asymmetric Bases and Expectation Values of P_2^2 , P_2^1 , and P_2^0** , by R. H. Schwendeman, 1968 (60 cents), SD Catalog No. C13.48:12.
- ☐ NSRDS-NBS 13, **Hydrogenation of Ethylene on Metallic Catalysts**, by J. Horiuti and K. Miyahara, 1968 (\$1), SD Catalog No. C13.48:13.
- ☐ NSRDS-NBS 14, **X-Ray Wavelengths and X-Ray Atomic Energy Levels**, by J. A. Bearden, 1967 (40 cents), SD Catalog No. C13.48:14.
- ☐ NSRDS-NBS 15, **Molten Salts: Vol. 1, Electrical Conductance, Density, and Viscosity Data**, by G. J. Janz, F. W. Dampier, G. R. Lakshminarayanan, P. K. Lorenz, and R. P. T. Tomkins, 1968 (\$3), SD Catalog No. C13.48:15/Vol.1.
- ☐ NSRDS-NBS 16, **Thermal Conductivity of Selected Materials, Part 2**, by C. Y. Ho, R. W. Powell, and P. E. Liley, 1968 (\$2), SD Catalog No. C13.48:16/Pt.2.
- ☐ NSRDS-NBS 17, **Tables of Molecular Vibrational Frequencies, Part 3**, by T. Shimanouchi, 1968 (30 cents), SD Catalog No. C13.48:17/Pt.3.
- ☐ NSRDS-NBS 18, **Critical Analysis of the Heat-Capacity Data of the Literature and Evaluation of Thermodynamic Properties of Copper, Silver, and Gold From 0 to 300°K**, by G. T. Furukawa, W. G. Saba, and M. L. Reilly, 1968 (40 cents), SD Catalog No. C13.48:18.
- ☐ NSRDS-NBS 19, **Thermodynamic Properties of Ammonia as an Ideal Gas**, by L. Haar, 1968 (20 cents), SD Catalog No. C13.48:19.
- ☐ NSRDS-NBS 20, **Gas Phase Reaction Kinetics of Neutral Oxygen Species**, by H. S. Johnston, 1968 (45 cents), SD Catalog No. C13.48:20.
- ☐ NSRDS-NBS 21, **Kinetic Data on Gas Phase Unimolecular Reactions**, by S. W. Benson and H. E. O'Neal, 1970 (\$7), SD Catalog No. C13.48:21.
- ☐ NSRDS-NBS 22, **Atomic Transition Probabilities, Vol. II, Sodium Through Calcium, A Critical Data Compilation**, by W. L. Wiese, M. W. Smith, and B. M. Miles, 1969 (\$4.50), SD Catalog No. C13.48:22/Vol.II.
- ☐ NSRDS-NBS 23, **Partial Grotrian Diagrams of Astrophysical Interest**, by C. E. Moore and P. W. Merrill, 1968 (55 cents), SD Catalog No. C13.48:23.
- ☐ NSRDS-NBS 24, **Theoretical Mean Activity Coefficients of Strong Electrolytes in Aqueous Solutions from 0 to 100° C**, by Walter J. Hamer, 1968 (\$4.25), SD Catalog No. C13.48:24.
- ☐ NSRDS-NBS 25, **Electron Impact Excitation of Atoms**, by B. L. Moiseiwitsch and S. J. Smith, 1968 (\$2), SD Catalog No. C13.48:25.
- ☐ NSRDS-NBS 26, **Ionization Potentials, Appearance Potentials, and Heats of Formation of Gaseous Positive Ions**, by J. L. Franklin, J. G. Dillard, H. M. Rosenstock, J. T. Herron, K. Draxl, and F. H. Field, 1969 (\$4), SD Catalog No. C13.48:26.
- ☐ NSRDS-NBS 27, **Thermodynamic Properties of Argon from the Triple Point to 300 K at Pressures to 1000 Atmospheres**, by A. L. Gosman, R. D. McCarty, and J. G. Hust, 1969 (\$1.25), SD Catalog No. C13.48:27.
- ☐ NSRDS-NBS 28, **Molten Salts: Vol. 2, Section 1, Electrochemistry of Molten Salts: Gibbs Free Energies and Excess Free Energies From Equilibrium-Type Cells**, by G. J. Janz and C. G. M. Dijkhuis. Section 2, **Surface Tension Data**, by G. J. Janz, G. R. Lakshminarayanan, R. P. T. Tomkins, and J. Wong, 1969 (\$2.75), SD Catalog No. C13.48:28/Vol.2.
- ☐ NSRDS-NBS 29, **Photon Cross Sections, Attenuation Coefficients, and Energy Absorption Coefficients From 10 keV to 100 GeV**, by J. H. Hubbell, 1969 (75 cents), SD Catalog No. C13.48:29.

* Available from National Technical Information Service, Springfield, Virginia 22151.

- ☐ NSRDS-NBS 30, **High Temperature Properties and Decomposition of Inorganic Salts, Part 2. Carbonates**, by K. H. Stern and E. L. Weise, 1969 (45 cents), SD Catalog No. C13.48:30/Pt. 2.
- ☐ NSRDS-NBS 31, **Bond Dissociation Energies in Simple Molecules**, by B. deB. Darwent, 1970 (55 cents), SD Catalog No. C13.48:31.
- ☐ NSRDS-NBS 32, **Phase Behavior in Binary and Multicomponent Systems at Elevated Pressures: *n*-Pentane and Methane-*n*-Pentane**, by V. M. Berry and B. H. Sage, 1970 (70 cents), SD Catalog No. C13.48:32.
- ☐ NSRDS-NBS 33, **Electrolytic Conductance and the Conductances of the Halogen Acids in Water**, by W. J. Hamer and H. J. DeWane, 1970 (50 cents), SD Catalog No. C13.48:33.
- ☐ NSRDS-NBS 34, **Ionization Potentials and Ionization Limits Derived from the Analyses of Optical Spectra**, by C. E. Moore, 1970 (75 cents), SD Catalog No. C13.48:34.
- ☐ NSRDS-NBS 35, **Atomic Energy Levels as Derived From the Analyses of Optical Spectra. Vol. I, ¹H to ²³V; Vol. II, ²⁴Cr to ⁴¹Nb; Vol. III, ⁴²Mo to ⁵⁷La, ⁷²Hf to ⁸⁹Ac**, by C. E. Moore, 1971 (In preparation), SD Catalog No. C13.48:35/Vols. I, II, and III.
- ☐ NSRDS-NBS 36, **Critical Micelle Concentrations of Aqueous Surfactant Systems**, by P. Mukerjee and K. J. Mysels, 1971 (\$3.75), SD Catalog No. C13.48:36.
- ☐ NSRDS-NBS 37, **JANAF Thermochemical Tables, 2d Edition**, by D. R. Stull, H. Prophet, et al., 1971 (\$9.75), SD Catalog No. C13.48:37.
- ☐ NSRDS-NBS 38, **Critical Review of Ultraviolet Photoabsorption Cross Sections for Molecules of Astrophysical and Aeronomic Interest**, by R. D. Hudson, 1970 (\$1), SD Catalog No. C13.48:38.

Announcement of New Publications on Standard Reference Data

Superintendent of Documents,
Government Printing Office,
Washington, D.C. 20402

Dear Sir:

Please add my name to the announcement list of new publications to be issued in the series:
National Standard Reference Data Series—National Bureau of Standards.

Name_____

Company_____

Address_____

City_____ State_____ Zip Code_____

(Notification Key N337)

