

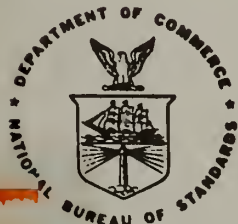
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# A Review of Compartment Fire Models

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U.S. DEPARTMENT OF COMMERCE  
National Bureau of Standards  
National Engineering Laboratory  
Center for Fire Research  
Washington, DC 20234

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## **A REVIEW OF COMPARTMENT FIRE MODELS**

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**U.S. DEPARTMENT OF COMMERCE, Malcolm Baldrige, Secretary**  
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- Figure 5. One possible configuration of the neutral plane height ( $N$ ) and the lower layer/upper layer discontinuity ( $Z_L$ ). Also indicated in the figure is the notation used for vent specification ( $B, H_L, H_U$ ).
- Figure 6. The nine possible flow conditions. The " $SA_{ij}$ " notation is taken from Tanaka [18]. Compartment two is assumed to be ambient. The selection rules determine which flow formula is to be used to calculate the vent flow.



## 1. INTRODUCTION

The purpose of this review is to bring together the various fire models which have been constructed over the years, in the hope of defining a general framework for further research in the field of fire modeling. In this context we will attempt to formulate each model (or numerical implementation) in terms of similar variables. No attempt will be made to justify or rederive each facet of each model. Rather we will elucidate the assumptions and enumerate the equations behind each model. References (1-16, 18-19) describe the models in detail, and Appendix B summarizes them.

The specific topic is the one-room model. All of the physics and chemistry which has been developed for a one-room model will be discussed. The model is assumed to be embedded in a world of uniform temperature  $T_a$  and reference pressure  $P_a$ . (The outside wall is at  $T_e$  which may not be the same as  $T_a$ .) A more general treatment is possible, but in discussing doorway phenomena in particular, the problem of multi-room connections would simply cloud the issue of what physics is actually present in each of the models.

The discussion is broken down into the basic conservation equations, the source and sink terms for these equations, and then the subsidiary equations which deal with interaction of various objects in a single compartment fire. The governing equations are divided in this manner to facilitate the development of a model (and numerical code) which will be modular in scope as well as general in application. With such a model we would hope that we can test various fire, fire spread, and entrainment models.

The notation used is detailed in Appendix A. In addition to being consistent within the monograph, the symbols chosen were those most commonly used by the various authors. In other cases, for example where several symbols were used, it is hoped that a judicious choice will make the text clearer. For symbols which are used only within one section,

or where the usage differs slightly, these symbols are defined in the text. Appendix B contains a bibliography of the models discussed in this report. Within the text they are referred to as (NBS-I, NBS-II, BRI, Harvard, Cal Tech and Dayton). This indicates the general geographic origin of the models. There are several references for each one, usually generated as one aspect or another is studied in detail. Rather than attempting to reference each model completely at each occurrence in the text, they will be referred to by the above generic names, and for the interested reader a more detailed explanation is provided in Appendix B.

## 2. THE GOVERNING EQUATIONS

The basic equations describe mass, momentum and energy transfer from point to point in a fire related environment. These equations will be formulated in a way which describes the flow from one volume to another. Thus they are couched in the integral formulation of "control volumes". These control volumes will be of sufficient size that we will require only a few to describe any system of interest. The choice is based on the premise that the details which occur within such a volume do not concern us (at present), but their mutual interaction does.

In considering dynamic systems, it is necessary to solve a problem self-consistently in terms of conservation of mass, energy and momentum. If such is not done, then some of the dynamics may be obscured or even lost. In particular, discussion of movement of the zone interfaces should be self-consistent.

By appropriate manipulation the conservation equations for mass and energy can be transformed into the form

$$\frac{dm}{dt} = \sum_i \dot{m}_i \quad (1)$$

$$c_p \dot{m} \frac{dT}{dt} - A_d Z \frac{dP}{dt} = \dot{Q} + \sum_i h_i \dot{m}_i \quad (2)$$



and

$$P = \rho RT \quad (3)$$

where  $\dot{Q}$  is the net energy input to the volume due to radiation (from all sources), convection and conduction, and  $h_i$  (the enthalpy of the  $i^{\text{th}}$  object in the control volume) is relative to the initial temperature of the volume from which the mass parcel  $\dot{m}_i \delta t$  came. The equation of state for an ideal gas is usually used for closure of the system. More correctly it should be written

$$P = P(\rho, T) \quad (4)$$

especially for applications to fire problems which are not ideal gas problems. However, for the case of an ideal gas, the derivations and discussions are simplified, and generalizations can be discussed later. The sign convention is that positive fluxes on the right hand side of an equation will increase the quantity being calculated on the left hand side, that is, transfer into a volume is indicated by a positive flux on the right-hand side.

The basic differences which arise among the models can be discussed using these equations as a basis. In an alternative form of the energy equation, written in terms of the internal energy (not enthalpy) an explicit work term appears

$$P\delta V$$

in place of the time derivative of pressure in equation (2). If the energy equation is to be solved in this form, then this work term should be carried explicitly.

Table I shows which of the zone models include, at least in principle, either the pressure derivative or the volume work term in their derivations but none of these models retain this term in the energy equation.

Table I

<u>Model</u>	<u>Included</u>
NBS-I	Yes
BRI	No
Harvard	No
Cal Tech	Yes
Dayton	No

The most important contribution to the pressure term  $\left\{ v \delta t \frac{dP}{dt} \Big|_{\text{ext}} \right\}$  is due to exothermic reactions and forced ventilation. This term arises, for example, when chemical energy release results in an increase in the volume occupied by the gas as well as the internal energy of the gas. If the gas were not allowed to expand, then the pressure in the compartment would increase. With the usual leaks present in rooms, the internal pressure change is essentially negligible and in general this term is small. However, it may be appropriate to retain this term since not all problems will necessarily allow such an approximation.

## 2.1 Form and Special Assumptions

The general form of the zone (control volume) model is to divide each compartment into two zones: an upper zone which contains a hot layer, and a lower layer which is often, but not always, considered to be at ambient conditions. There may exist one or more fires and plumes in a room and these can usually be considered to be part of the upper zone. Mass and energy transfer between the zones is provided by the plumes and mixing at vents as well as radiation between layers. This is true for all of the models and is possibly a deficiency since there is experimental evidence that shows mass flow can occur along walls and other boundaries. In general the plume, once created, simply transfers mass and energy from one zone to another. Another set of equations could be written for the plume, but as long as it is in quasi-steady state, considering it to be part of the upper zone is quite sufficient.

Another way of looking at the plume is to consider it so small in mass, energy content and volume that it can be ignored except as a transfer mechanism. For some problems, however, the plume must be considered a separate zone along with the concomitant conservation equations.

Mass and convective enthalpy transfer to and from a layer can exist at an arbitrary number of ports, although only the BRI model allows direct vertical transfer for multi-level buildings. Finally, radiation and convection are allowed to and from walls, floors and ceilings. The Harvard model includes detailed heat transfer to the fire source and other internal objects as well.

The specific formulation for a single compartment will be discussed. It will be in a form which allows general connectivity to other compartments. Then a discussion of the physical phenomena is in order. These are the pieces which then become part of the general flow equations.

## 2.2 Equations

For a given compartment there exist three equations and four unknowns for each zone. There are several variations of these equations which have been used.

The general form of eqns. (1-3) together with the subsidiary equations

$$H = Z_u + Z_l \quad (5)$$

and

$$P_u = P_l \text{ at the zone boundary} \quad (6)$$

provide closure for the model. An implicit assumption is that the cross section (A) is known, and is usually assumed to be constant,  $A \neq A(z)$ , where  $AZ = V$ . The subscripts "u" and "l" represent "upper" and "lower", respectively, whereas "i" refers to openings to other zones, other compartments or external volumes.

With the assumption of two zones per compartment, there exist two sets of equations (1-3), one for each layer. In some cases, of course, one zone may be infinitesimally small, such as the upper layer at  $t = 0$ . Specifically there are four differential (1-2) and two algebraic equations (3), together with the two subsidiary equations (5-6) for each compartment. Substitutions can be made using the subsidiary equations which still leaves unknown the manner in which the other variables will be found. For numerical purposes, as well as having common grounds for comparison of the models, we will write the equations in terms of

$$m_u = Z_u A \rho_u, \quad m_\ell = Z_\ell A \rho_\ell, \quad V_u, \quad V_\ell, \quad T_u, \quad T_\ell, \quad P_u \quad \text{and} \quad P_\ell$$

This yields six equations in eight unknowns,

$$\frac{dm_u}{dt} = \sum_i \dot{m}_{i,u} \tag{7}$$

$$\frac{dm_\ell}{dt} = \sum_i \dot{m}_{i,\ell} \tag{8}$$

$$c_p m_u \frac{dT_u}{dt} = V_u \frac{dP_u}{dt} + \dot{Q}_u + \sum_i h_{i,u} \dot{m}_{i,u} \tag{9}$$

$$c_p m_\ell \frac{dT_\ell}{dt} = V_\ell \frac{dP_\ell}{dt} + \dot{Q}_\ell + \sum_i h_{i,\ell} \dot{m}_{i,\ell} \tag{10}$$

$$P_u = \rho_u R T_u \tag{11}$$

$$P_\ell = \rho_\ell R T_\ell$$

together with the subsidiary assumptions  $P_u = P_\ell$  at the interface between the zones and  $V = V_\ell + V_u$ . The set of equations can, and usually is, simplified somewhat. The pressure consists of a reference pressure at the floor and a hydrostatic term



$$P(Z) = \bar{P}(\text{floor}) - \int_0^Z \rho g dZ.$$

The hydrostatic term is considerably smaller than the reference pressure,  $\bar{P}(\text{floor})$ . Thus, change in pressure as a function of height is important only for calculating the height of the neutral plane and the relative pressure between two vents, the latter being important for vent flow calculations. This simplifies the equations in the following manner

$$P_u = \rho_u RT_u \approx \rho_\ell RT_\ell = P_\ell \equiv P,$$

which reduces by one the number of independent variables. It should be noted that the reference pressure  $P = P_r = \bar{P}(\text{floor})$  is not independent of time, so that the term  $dP/dt \neq 0$ . This yields, finally, five equations in five independent unknowns. The specific set differs for each of the models, but all of them can be recast into this form.

The exact form in which these equations are solved varies from model to model. Nevertheless, they all have the same source terms; that is, the right-hand side of equations (7-11) will be similar for each model, with differences showing up in the number of terms and the exact formulation for each term.

If we ignore, for the moment, the problem of enthalpy of formation, then the enthalpy transfer terms can be written as

$$h_j \dot{m}_{j,i} = \dot{m}_j c_p (T_j - T_i)$$

where  $T_i$  is the temperature of the layer which is being changed and  $T_j$  is the temperature of the layer from which the gas is coming. With these assumptions, approximations and insertions, the four differential equations can be written as

$$\frac{d}{dt} m_u = \dot{m}_p + \sum_i \dot{m}_{i,u} \quad (12a)$$

$$\frac{d}{dt} m_\ell = - \dot{m}_e + \sum_i \dot{m}_{i,\ell} \quad (12b)$$

$$c_p m_u \frac{d}{dt} T_u = v_u \frac{dP}{dt} + \dot{Q}_u + \dot{m}_f c_p (T_R - T_u) + \dot{m}_e c_p (T_\ell - T_u) + \sum_i \dot{m}_i c_p (T_i - T_u) \quad (12c)$$

$$c_p m_\ell \frac{d}{dt} T_\ell = v_\ell \frac{dP}{dt} + \dot{Q}_\ell - \dot{m}_e c_p (T_u - T_\ell) + \sum_i \dot{m}_i c_p (T_i - T_\ell) \quad (12d)$$

where  $\dot{Q}_u$  and  $\dot{Q}_\ell$  are the heating, or cooling, terms for the upper and lower layers, respectively. In the present discussion

$$\dot{Q}_u = \dot{Q}_R + \dot{Q}_C + \dot{Q}_f + \dot{Q}_o$$

and

$$\dot{Q}_\ell = 0 \text{ (assumed in all the models under discussion).}$$

The term  $\dot{Q}_f$  includes the heat of pyrolysis.

In the following sections we will discuss the source term for eqn. (12) and discuss the differences which occur in each of the models.

### 3. RADIATIVE AND CONVECTIVE LOSS AND GAIN

The radiative and convective heat loss and gain terms are contained in eqns. (12c, 12d) and are represented by the term  $\dot{Q}$ . The formalism used here is somewhat more complex than is needed strictly for loss and gain to a layer; however, the same terms will show up again in discussion of the subsidiary equations that describe heating of the walls, floor, ceiling, and other objects. Sources of heat which increase internal energy are positive. Once again  $\dot{Q} = \dot{Q}_R + \dot{Q}_C + \dot{Q}_f + \dot{Q}_o$ .



### 3.1 Radiative Loss and Gain

The terms which contribute heat to an absorbing layer are the same (in form) for all layers so we will discuss these in terms of a general layer contribution. Radiation can leave a layer by going to another layer, to the walls, exiting through a vent, or heating up an object. Similarly, a layer can be heated by absorption of radiation from these surfaces and objects. The formalism which we will use for the geometry is that used by Siegel and Howell [17] and is shown in figure (1). The radiative transfer can be done with a great deal of generality; however, most models at least make the assumption that zones and surfaces either radiate and absorb like a black body, or a grey body with some constant emissivity ( $\epsilon < 1$ ).

A further assumption consonant with the stratified zone assumption is that emission and absorption are constant throughout a gas layer. In application to a two layer model, the assumption is made in all implementations that the lower layer is diathermous. This is not a necessary assumption but greatly simplifies the radiation transfer calculations. In future work, the case of a lower layer which can emit and absorb should probably be included, especially if interlayer mixing is included.

The radiation transfer problem is always broken into two parts: first is transfer to and from gases, walls and surfaces; second is transfer to and from objects and fires. There are two reasons for such a division. The first is that, although complicated, radiation to and from walls, gases and surfaces can be treated more or less completely. The primary problem which arises here is that not a sufficient amount of information is known about wall emissivity. For objects and fires, however, not only are the absorption and emission coefficients not well defined, the geometrical factors are not well defined and may change in time as well. Usually the models assume some average shape for the flame and plume and make reasonable estimates for the various view factors. For the most part, the discrepancy is not important. However,

when flashover is about to occur, this is probably not a valid assumption. Furthermore, with current fire sources, positive feedback may increase the pyrolysis rate and may lead to unstable numerical solutions. For these reasons, we will describe the two sources of radiation transfer separately.

The notation used for the wall, gas, surface interaction is shown in figure (1). All of the zone models can be cast into this form. The primary differences are the sophistication with which the geometrical factors are calculated, and the approximations which are used in finding the emissivity and transmission factors for the layers. With the assumption of blackbody emission, and using eq. (17.20) of Siegal and Howell [17] we obtain

$$\sum_{j=1}^N \left[ \frac{\delta_{kj}}{\epsilon_j} - F_{kj} \tau_{kj} \left( \frac{1-\epsilon_j}{\epsilon_j} \right) \right] \dot{Q}_j = \sum_{j=1}^N \left[ (\delta_{kj} - F_{kj} \tau_{kj}) \epsilon_j \sigma T_j^4 - F_{kj} \bar{\alpha}_{kj} \epsilon_g \sigma T_g^4 \right] \quad (13)$$

for radiation to surface (k) from surface (j).

Of interest here is the net influx of radiation to the upper gas layer, which can be written as

$$\dot{Q}_g = - \sum_k \dot{Q}_k A_k.$$

The notation used is:

- $F_{kj}$  = geometrical view factor of surface (k) by surface (j)
- $\tau_{kj}$  = transmission coefficient from surface (k) to surface (j)
- $\bar{\alpha}_{ki}$  = absorption coefficient
- $\dot{Q}_j$  = net radiation flux to surface (j) - watts/m<sup>2</sup>
- $\dot{Q}_g$  = net radiation to the upper gas layer - watts

$\dot{Q}_u, \dot{Q}_l$  = net radiation to the upper and lower layer (boundary) surfaces

$\sigma$  = Stephan-Boltzman constant =  $5.67 \times 10^{-8}$  watts/m<sup>2</sup>K<sup>4</sup>

L = mean beam length - (m)

$\alpha$  = absorption coefficient of the upper gas layer (m<sup>-1</sup>)

We assume that the grey, upper zone, can be treated as an equivalent radiating sphere with a mean beam length given by

$$L = 4V/A$$

where V is the volume of the gas and A is its surface area. This yields an effective emissivity for the upper layer of

$$\epsilon_g = 1 - \exp(-\alpha L).$$

For a two zone model, with the assumption that the lower layer is diathermous, we have the two terms, as shown in detail by Tanaka [10]

$$\dot{Q}_u = A_u \epsilon_u \Pi_u / D, \quad \dot{Q}_l = A_l \epsilon_l \Pi_l / D \quad \text{and} \quad \dot{Q}_g = -\dot{Q}_u + \dot{Q}_l$$

plus additional loss terms for a vent. The factors  $A_u$  and  $A_l$  are the upper and lower boundary surface areas minus the vents and other openings. The terms are given by

$$D = \{1 - (1-\epsilon_u)(1-\epsilon_g)F_{uu}\} \{1 - (1-\epsilon_l)F_{ll}\} - \{(1-\epsilon_u)(1-\epsilon_l)(1-\epsilon_g)^2 F_{ul}F_{lu}\} \quad (14)$$

$$\begin{aligned} \Pi_u = & \{[1 - (1-\epsilon_g)F_{uu}] \{1 - (1-\epsilon_l)F_{ll}\} \\ & - \{(1-\epsilon_l)(1-\epsilon_g)^2 F_{ul}F_{lu}\}] \sigma T_{uw}^4 \\ & - (1-\epsilon_g)F_{ul} \epsilon_l \sigma T_{lw}^4 - [1 + (1-\epsilon_l)\{(1-\epsilon_g)F_{ul}F_{lu} - F_{ll}\}] \epsilon_g \sigma T_g^4 \end{aligned} \quad (15)$$

$$\begin{aligned}
\Pi_{\ell} = & \{ [1 - (1-\epsilon_u)(1-\epsilon_g)F_{uu}] (1-F_{\ell\ell}) - (1-\epsilon_u)(1-\epsilon_g)^2 F_{ul} F_{\ell u} \} \sigma T_{\ell w}^4 \\
& - (1-\epsilon_g) F_{\ell u} \epsilon_u \sigma T_{uw}^4 \\
& - \{ [1 - (1-\epsilon_u)(1-\epsilon_g)F_{uu} F_{\ell u} + (1-\epsilon_u)(1-\epsilon_g)F_{\ell u}] \epsilon_g \sigma T_g^4
\end{aligned} \tag{16}$$

Equations (14-16) represent the most general case. These equations can be simplified somewhat if we do not consider the effect of grey walls. Then we have, as an example,

$$\epsilon_{\ell} = \epsilon_u = 1.$$

With this approximation, the factors become

$$\dot{Q}_u = A_u P_u / D,$$

$$\dot{Q}_{\ell} = A_{\ell} P_{\ell} / D,$$

$$D = 1,$$

$$\Pi_u = \{ 1 - (1-\epsilon_g) F_{uu} \} \sigma T_{uw}^4 - (1-\epsilon_g) F_{\ell\ell} \sigma T_{\ell w}^4 - \epsilon_g \sigma T_g^4$$

$$\Pi_{\ell} = \{ 1 - F_{\ell\ell} \} \sigma T_{\ell w}^4 - (1-\epsilon_g) F_{\ell\ell} \sigma T_{uw}^4 - F_{\ell u} \epsilon_g \sigma T_g^4$$

where  $F_{uu} = 1 - \frac{A_d}{A_u}$ ,  $F_{ul} = \frac{A_d}{A_u}$ ,  $F_{lu} = \frac{A_d}{A_{\ell}}$ , and  $F_{\ell\ell} = 1 - \frac{A_d}{A_{\ell}}$ .

Finally, the net radiation to the upper gas layer is given by

$$\dot{Q}_g = - \sum_k \epsilon_k A_k \Pi_k / D \tag{17}$$



### 3.2 Implementation of Radiation Formulae

The various models differ in the number of these terms which are included, and when included, the form taken for the view factors and calculation of emissivity. Table II shows the level of completeness of each model.

Table II

Model	Comments
NBS-I	Complete
NBS-II	No radiative transfer - effective radiative loss proportional to $H_c$
BRI	Complete - no radiation through vents, does not include radiation from fire source
Harvard	Complete - ambient lower wall, vent not treated self-consistently.
Cal Tech	No radiation transfer - effective radiative loss proportional to $H_c$
Dayton	Considers only total radiation ( $F_{ij} = 1$ ), $T_{uw} = T_{lw}$ , no radiation through vents

NOTE: "complete" refers to the formalism discussed above.

Two terms which have not yet been discussed are first, the radiation due to a fire source, and second, the radiation interchange through a vent. The former is a very difficult problem in that a fire source is generally very irregular in shape, and usually changing in time. The latter is done very much in an ad-hoc manner, that is an overlay on the usual radiation treatment, primarily due to the difficulty of calculating the actual configuration factors for vents and other arbitrary openings. For these reasons, the effects of these two sources of radiation will simply be presented for the models where appropriate. The source terms for radiation exchange with vents, walls and a zone are

shown in Table III for the three models which have a complete radiative transfer scheme. This example is presented for comparison purposes and is for  $\epsilon_u = \epsilon_l = 1$ . The notation is that used in figures (1) and (2), and given earlier. All three use  $\epsilon_g$  (lower) = 0.0; therefore we use the notation  $\epsilon_g$  refers to the upper gas layer only, whereas  $\epsilon_u$  and  $\epsilon_l$  refer to the upper and lower wall emissivities, respectively.

TABLE III

Model	$\dot{Q}_R$
NBS-I	$-\sigma[\epsilon_g T_u^4 A + (1-\epsilon_g) T_{uw}^4 (A_u + A_{uv}) - T_{uw}^4 A_u - \epsilon_g T_{lw}^4 A_d - T_a^4 A_{uv}] + 1/2 E_f$
Harvard	$-\sigma[\epsilon_g T_u^4 A - \epsilon_g T_{uw}^4 A_u - \epsilon_g T_{lw}^4 (A_d + A_{lv})] + 1/2 E_f f(Z_u, Z_l)$
BRI	$-\sigma[\epsilon_g T_u^4 A - \epsilon_g A_u T_{uw}^4 - \epsilon_g A_d T_{lw}^4]$

Note:  $A = A_u + A_d + A_{uv}$ ,  $A_v = A_{uv} + A_{lv}$ ;  $\epsilon_u = \epsilon_l = 1.0$

$E_f$  = radiation from a fire source and  $f(Z_u, Z_l)$  is its view factor to the upper layer

### 3.3 Convection

Convection is the mechanism by which the gas layers lose (or gain) energy to the walls. The other physical process, conduction, which is intimately connected with convection, will be discussed in the section on subsidiary equations - heat loss by the wall and heating of interior objects.

The basic formulation is the same for all models which incorporate convective heat loss. The primary differences lie in the method used to change the wall temperature, redistribute energy and the calculation of the heat loss coefficient -  $h_c$ . Table IV summarizes the results.



These formulae are applied in the models in a general way although very little data exists for detailed validation. The formula used in NBS-I is specific to crib fires in a single room and that used in NBS-II was an effort to make such a detailed comparison between a model and a particular experiment.

TABLE IV

Model	Form ( $q_c =$ )	$u/\ell$	Note	Constant
NBS-I	$h_c (T_g - T_w) A_w$	both	1	$h_c = \begin{cases} h_\ell = 10 \text{ W/m}^2\text{-K} \\ \text{or} \\ h_u = 0.88 Q^{1/3} / A^{3/8} \end{cases}$ $\lambda_c = \text{parameter}$ $t_c, L_c$ are characteristic scales for the room
NBS-II	$\frac{(1-\lambda_c) \dot{Q}_o \tau_c}{\rho_1 C_p T_a L_c} \dot{q}$	upper	$\dot{q} = \dot{Q}(t) / \dot{Q}_o$	
BRI	$A_w \left( \frac{k}{\ell} \right) \begin{cases} 0.21 (Gr \cdot Pr)^{1/3}, T_g > T_u \\ 0.012 (Gr \cdot Pr)^{1/3}, T_g < T_u \end{cases}$	upper	2	$Gr = \frac{g \ell^3  T_g - T_u }{\nu^2 T_g}$ $Pr = 0.7$
Harvard	$h(T_g - T_w) A_w + \delta \dot{q}$	upper	3	$h = \min(50, 5 + 45 \frac{(T_L - T_w)}{100}) \text{ W/m}^2\text{-K}$
Cal Tech	$C_1 \dot{Q}$	upper		
Dayton	$h_c (T_g - T_{wall}) A_w$	upper		$h_c = 11.21 \text{ W/m}^2\text{-K}$

1. Applies to both upper and lower walls; as a slice of the layer reaches a new section of the wall, the wall is assumed to equilibrate instantaneously.
2.  $\nu = 7.18 \times 10^{-10} T^{7/4}$ ,  $g = 9.8 \text{ m/sec}$ ,  $\ell = \text{length over which convective heat transfer takes place}$ . Applies to the upper layer only.
3. Applies to upper surface only; " $\delta q$ " is an approximation for the amount of energy required to bring the new slice of heated wall to the same conditions as the hot upper wall. As the layer retreats  $\delta q$  is set to zero, that is, no energy is put into the lower layer by cooling of this new slice of lower wall.

$$\delta \dot{q} = \int_0^t \dot{q}_c dt [\dot{\epsilon}_u / \epsilon_u] [A_w], \text{ where } A_w \text{ is the area of ceiling, and}$$

$$\dot{\epsilon}_u (\text{thermal energy in the upper layer}) > 0$$

#### 4. ENTHALPY AND MASS FLUX

The primary mechanisms which provide for mixing and transfer of mass and energy between the two layers are flow provided via the plume and mixing at a door and other vents. These phenomena are complicated if one considers the general multiroom case. In the following section we will consider the situation as it exists in current models with one

room vented to an ambient atmosphere and make no attempt to describe the general interaction. In the case of the plume, this implies that the plume starts at a fire source and no combustion occurs during the ascent of a smoke parcel. As the gas rises, it entrains air from the lower layer. This mixture is then deposited in the upper layer. For the problem of vent flow, the exterior will be a single "layer" whose pressure is calculated from hydrostatic equilibrium at temperature  $T_a$ . This is the external air temperature and is generally somewhat (usually) below  $T_e$ , the external wall temperature.

#### 4.1 Plumes

A fire generates a plume which transports mass and energy from the fire into the upper layer. In addition, the plume entrains mass from the lower layer and transports this mass and energy into the upper layer. The energy and mass ( $\dot{Q}_f$  and  $\dot{m}_f$ ) generated by the fire are common to all models. The form of entrainment used by each of the models is shown in Table V.

There are several possible problems in the implementation of plume theory in numerical applications. Specifically, the enthalpy which is generated by a heat source due to combustion is relative to some reference temperature. It is important, to be a self-consistent, to use this same temperature as a reference in calculating the energy necessary for pyrolysis and the net energy added to the fuel to bring it up to the pyrolysis temperature. In particular, in order to go from the fuel temperature to the pyrolysis temperature, a certain amount of energy must be subtracted from the lower layer; of course there is also some radiative heating from the fire and from both layers. Finally, in using Table V, one must be careful that the formula is applicable to the problem. For example, the form given for NBS-I is specific to a crib fire.

TABLE V

	$\dot{m}_e [\dot{m}_p = \dot{m}_e + \dot{m}_f]$	Comments
NBS-I	$0.055(\dot{m}_b \Delta H)^{1/2} (D-0.53)n[D-0.53]$	$n(x) = \text{Heaviside step function}$ $= \begin{cases} 0 & x \leq 0 \\ 1 & x > 0 \end{cases}$
NBS-II	$0.21 \rho_L (gZ)^{1/2} Z^2 (Q^*)^{1/3}$	
BRI	$0.522 \left( \frac{\rho_a^2 \dot{Q}}{C_p T_a} \right)^{1/3} Z^{5/3} \left( 1 + 0.44 \frac{\theta}{T} \right)^{-2/3}$	$\frac{\theta}{T_L} = 4.35 \left( \frac{\dot{Q}}{P_m C_p T_L} \right)^{2/3} Z^{-5/3}$
Harvard	$\pi \rho_L (b^2 u - R^2 u_F)$	$u_F = (C_o/X)^{1/3}; x = R/1.2a;$ $u = (C_o/H_p)^{1/3}; b = 1.2aH_p;$ $C_o = 25gE_f/48\pi a^2 C_p T_L P_L$ $H_p = h_R - h_f - h_L + X = \text{plume length}$
Cal Tech	$1.2 \times 0.21 \rho_\infty (gZ)^{1/2} Z^2 (Q^*)^{1/3}$	$Q^* = \dot{Q}/P_\infty C_p T_\infty Z^2 (gZ)^{1/2}$
Dayton	$\dot{m}_f \left\{ \left[ \omega \left( C_1 \left( \frac{Z}{y_o} \right) + 1 \right)^{5/2} - 1 \right] + \left[ \left( C_2 \left( \frac{Z-Z_B}{y_B} \right)^{1/5} + 1 \right)^{5/3} - 1 \right] \right\}$	$\omega = \gamma C_p T_L / (\gamma C_p T_L + \chi_L \Delta H_c)$ $C_1 = 0.80 \frac{E^{4/5}}{\omega^{3/5}} (1-\omega)^{1/5} (gy_o/u_o^2)^{1/5}$ $C_2 = 1.09 E_p^{4/5} (gy_B/u_B^2)^{1/5} (1-\rho_B/\rho_L)^{1/5}$ $Z_B = 1.49 E_c^{-4/5} [\omega/(1-\omega)]^{1/5} (\omega \rho_L \rho_o + \gamma \chi_{LO_2})^{2/5} (\rho_o u_o / \rho_a \sqrt{gy_o})^{2/5}$ $L = \text{layer}$

#### 4.2 Door Jets and Mixing

Flow at vents is governed by the pressure difference across a vent which occurs at each zone level. In the control volume approximation the general momentum equation for the zones is not solved (a possible source of error). Instead, the momentum transfer at the boundaries is included by using Bernoulli's solution for forced flow. This is augmented for restricted openings by using "flow coefficients." These latter modifications deal with the problem of constriction of velocity streamlines at an orifice. As shown in figure (3), the general case for multilayer flow can be quite complicated.

There are actually two cases which apply to this type of flow. The first, and usually thought of in fire problems, is the case where air, smoke or some other fluid is pulled from a compartment by buoyancy. The second type of flow is the piston effect, and is particularly important in the initial stages of a fire. Rather than depending on a difference in density between two gases, the flow is forced by a pressure difference across an orifice generated, for example, by combustion in a compartment.

Figure (4) shows the four configurations which will be considered. The situation is specific for one room venting to an unconfined, ambient atmosphere. Cases (b and c) shown in figure (4) can be considered as two manifestations of the same geometry. An important distinction must be made in the physics of the two cases, however. In case (b) both buoyancy and the piston effect are present whereas in case (c), the flow is driven only by buoyancy.

A further complication which arises is that of determining the position of the neutral plane relative to the layer discontinuity height. The somewhat complex interaction leads to a large set of flow conditions even in the case of a single compartment, at least for cases (c) and (d). In general, all models proceed on the basis that flow is governed by Bernoulli's equation at an opening and the driving force is a combination of the ground (reference) pressure and the hydrostatic change in pressure as a function of height. The differences in the implementation of the models lie in the formalism actually used to calculate the flow, which terms are included (i.e., mixing between the upper and lower layers), and which regimes are included.

Table VI shows which of the four regimes (filling, buoyant, flow and choked flow) are included in each model. The terms are explained in figure (4).



Table VI  
Flow regimes included in each model

<u>Model</u>	<u>Regimes</u>
NBS-I	c
NBS-II	a
BRI	a-d
Harvard	a-c
Cal Tech	a-c
Dayton	a-c

The most systematic approach is done by Tanaka [18]. Therefore we will use this notation, since all other models can be cast into this form. The one piece which is missing from the BRI model is the intra-room layer mixing which will be taken from NBS-I.

The notation is:

S = smoke

A = air

ij = flow from room i to room j

P = reference pressure (at the floor)

The order of the letters indicate which type of layer is flowing into which other types. For example,  $SA_{j1}$  indicates that fluid from a smoke layer in room (j) is flowing into an air layer in room (i). Of course, in the present case, only one room exists with the remaining space being an infinite, ambient atmosphere. Therefore, only the term  $AA_{12}$ ,  $SA_{12}$ ,  $AA_{21}$ , and  $AS_{21}$  can be non-zero. Table VII shows the various terms where the notation is shown in figure (5).

Mixing between the upper and lower zones can be important at doorways, and possibly at the walls. Little energy will be transported by this mechanism, but it does provide a means by which smoke can be injected into the lower layer, thereby providing a means to change the

optical properties of that layer. At present, the only work which incorporates this phenomena is that of Quintiere et al. [13], and is for vents only. The contribution to the mass input to the lower layer is given by

$$AA_{21}/SA_{12} = 0.5 (T_a/T_g) \frac{N-Z_u}{N} \quad (18)$$

TABLE VII

Neutral Plane	Flow	Term	Condition
$N \leq Z_\ell$		$SA_{12} = \begin{cases} 2/3CB(2g\rho_u \rho_u-\rho_a )^{1/2}\{(\min(H_h, Z_\ell)-N)^{3/2}-(\max(H_\ell, Z_\ell)-N)^{3/2}\} \\ 0 \end{cases}$	$H_h > Z_\ell, H_\ell < Z_a$ $H_h < Z_\ell$
		$AA_{12} = \begin{cases} C'B(2\rho_a P_a-P )^{1/2} \{ \min(H_h, Z_\ell) - H_\ell \} \\ 0 \end{cases}$	$H_\ell < Z_\ell$ $H_\ell > Z_\ell$
$Z_\ell < N < \infty$		$SA_{12} = \begin{cases} 2/3CB(2g\rho_u \rho_u-\rho_a )^{1/2} \{ (H_h-N)^{3/2} - (\max(H_\ell, N) - N)^{3/2} \} \\ 0 \end{cases}$	$H_h \geq X, H_\ell < \infty$ $H_h < X \text{ or } H_\ell > \infty$
		$AS_{21} = \begin{cases} 2/3CB(2g\rho_a \rho_a-\rho_u )^{1/2} \{ (N-\max(Z_\ell, H_\ell))^{3/2} - (N-\min(H_h, X))^{3/2} \} \\ 0 \end{cases}$	$H_h \geq Z_\ell, H_\ell < X$ $H_h < Z_\ell \text{ or } H_\ell > X$
		$AA_{21} = \begin{cases} C'B(2\rho_a P_a-P )^{1/2} \{ \min(H_h, Z_\ell) - H_\ell \} \\ 0 \end{cases}$	$H_\ell \leq Z_\ell$ $H_\ell > Z_\ell$

$c = C' = 0.68$

### 4.3 Special Cases and Selection Tables

Figure 6 shows the possible combinations of the neutral plane and vent configurations. The neutral plane is given from the hydrostatic equation by knowing the pressures at the floor and realizing that at the neutral plane the pressure differential is zero. For the single room case, there is only one possible choice for the neutral plane. If the pressures are equal at the floor and everywhere (within a vent) the densities are the same, then the neutral plane is well above the opening height ( $H_u$ ). If, however, the situation is as shown in figure (5), then we have



$$P_i(Z) = P - \rho_l g Z_\ell - \rho_u g (Z - Z_\ell) \quad (\text{internal})$$

and (19)

$$P_e(Z) = P_a - \rho_a g Z. \quad (\text{external})$$

At the neutral plane we have  $P_i(N) = P_e(N)$  and therefore

$$N = \left( \frac{P_a - P}{\rho_a - \rho_u} \right) + \left( \frac{\rho_l - \rho_u}{\rho_a - \rho_u} \right) Z_\ell \quad (20)$$

defines the position of the neutral plane. Normally the flow coefficients will be approximately equal

$$C \sim C' \sim 0.6 - 0.7$$

for both air and smoke.

## 5. SUBSIDIARY EQUATIONS

The term "subsidiary equations" covers all of the interactions which do not directly influence mass or energy balance of the upper or lower layer, or terms which are not yet on a firm mathematical basis. The former includes effects such as thermal conductivity through walls and an example of the latter would be fire sources and the effect on them of radiation for the upper gas layer.

### 5.1 Fire Source

The most difficult aspects of modeling a fire (source) are deciding where the fire actually burns and elucidating the effect of radiation on the fuel source. The latter problem includes both self-radiation, that is radiation from the flame back to the fuel source as well as radiation from walls, layers and other objects to the fuel. Many of the models avoid the problem by specifying the burning rate as a function of time. For some purposes this is adequate. However, for studying flashover, multiple ignition and similar problems, a fire model must be specified.

The heat release rate per unit mass being  $h_c$  (generally not a strong function of temperature), the enthalpy flux into the upper layer is

$$\dot{Q}_f = h_c \dot{m}_v - c_p (T_u - T_v) \dot{m}_v - \dot{Q}_p$$

Essentially, the energy introduced into the upper layer is a product of the heat release rate at the pyrolysis temperature with a reduction for the energy required to gasify the fuel and subsequent pyrolysis.

For a specified burning rate, this is sufficient. In order to treat a self-consistent fire, however, one must include the interaction of a fire source with its environment. A self-consistent fire source must include the effects of oxygen vitiation, radiation to the fuel surface, radiation by the fuel surface, and heating by conduction. The NBS-I, Harvard and Dayton models have fire source algorithms which include one or more of these effects, as shown in Table VIII.

Table VIII

NBS-I	Oxygen, radiation
Harvard	Radiation, re-radiation, conduction
Dayton	Oxygen, radiation

All three models attempt to include the effects of radiation from both the upper layer, walls and the flame itself. Clearly, prior to ignition, the radiation from "other" sources is most important. However, once ignition occurs, radiation from the flame to walls, the gas layer and other objects should generally dominate. In addition, the Harvard model includes re-radiation from the fuel source itself. This latter effect should not be important for gas burners or pool fires but may be effective in reducing the burning rate of charring materials.

An important difference in the models is the method of including the geometrical (view) factors. As before, the intent is to include only the radiation which an object "sees." A fire changes shape in time, and rather quickly at that. This leads to the use of approximate flame shapes, the most common being cones (Harvard) or cylinders (NBS-I, Dayton). When including incident radiation due to walls and the gas layers, a complete treatment of the problem allows for absorption by the plume and flame. Finally allowance has to be made for effects of oxygen vitiation. In principle, a more complete kinetics scheme should be included, especially as the detailed species which are released in combustion become accessible experimentally. Nevertheless, the effect of oxygen (or lack) must be taken into account. The (NBS-I) model does these in an ad hoc but reasonably effective way, namely

$$\dot{m}_f = \dot{m}_{free} + \dot{m}_R + \dot{m}_o \quad (19)$$

where  $\dot{m}_f$  = free burning rate of a fire

$\dot{m}_R$  = enhancement due to radiation effects

$\dot{m}_o$  = decrement due to vitiation

For NBS-I, which considered the burning of cribs, the  $\dot{m}_R$  term is given by

$$\dot{m}_R = \frac{\dot{q}_{top} \cdot A_{top} + \dot{q}_{side} \cdot A_{side}}{L_{vap}}$$

where  $L_{vap}$  is the gasification energy and is a measured quantity. For this particular experiment,  $L_{vap}$  ignored the contribution (decrement) due to heating of the fuel. Thus to do this problem more generally a loss term,  $\dot{m}_{target}$ , should be included in eqn. (19). The term "Top" and "Side" refer to the parts of the cribs which were being modeled. The Harvard model proceeds in a similar fashion except that the view factors which determine the  $\dot{q}$  term are calculated numerically based on the object position and geometry relative to the ceiling and walls. In the case of NBS-I, the view factors are calculated analytically, once again based on the geometry of the room and object (fire source). One aspect



of the Harvard model is the inclusion of absorption of the flame radiation by the fire source.

Oxygen deprivation has two parts. The first is the oxygen concentration near the fuel source. The second is the amount of burning which occurs in the plume (a flame) above the fuel source. This latter is important since, as the upper layer descends, the functional form of combustion changes. The NBS-I model does this by

$$\dot{m}_o = - \dot{m}_{\text{free}} \left\{ 1 - \frac{Y_{\text{ox}}}{0.23} \right\}$$

and the Harvard model accomplishes this same effect by changing the size of the cone of the flame. As the layer moves down, the cone is truncated, reducing the pyrolysis rate, thus reducing the mass flow. The Dayton model accomplishes this using a method similar to NBS-I. The Dayton model uses the Steward [20] and Fang [21] plume models (see Table V), however.

The last piece in this analysis concerns the pyrolysis itself. In addition to the problem of heating the fuel from its reference state  $T_R$ , there is also the problem of a phase change, for example, from solid to gas, where the combustion finally occurs. From the point of view of modeling a fire, the only reason to be concerned with this process is that a phase change is sensitive to temperature and pressure; thus the fire history affects the burning rate  $\dot{m}_i$ . An important part of this process is the production of toxic substances, such as carbon-monoxide and cyanide. Since this is in large part the motivation for studying the fire problem, a great deal more effort must be spent on this aspect of modeling a fire source.

As is obvious, our understanding of the fire source itself is minimal. Further, the source to be used is specific to the problem to be studied. At this point there is no agreed upon way to include a general, self-consistent fire source in a model.

## 5.2 Radiation Between Objects

The calculation of radiation effects between objects suffers from problems similar to those of the fire source itself. Interobject radiation is an important phenomenon only if multiple targets are present and one is interested in flashover. The only two models which address this issue are the Dayton [2] and Harvard models [19]. The primary difficulty arises in calculating view factors between the objects. This has an analogy in several other fields. One must calculate the projection onto a surface (the target) of the surface subtended by the solid angle (view factor) of the emitter. Except for simple objects such as spheres and planar surfaces, this calculation is difficult. Current research in modeling of solid objects, which deals with piecewise continuous approximations to a surface, will aid in automating and generalizing these calculations. At the present time, however, the methods used in the above mentioned models will have to suffice.

## 5.3 Conduction

Conduction of heat through solids occurs in two places: the compartment walls and interior objects. The techniques used are similar for both cases. Generally a slab is cut into  $N$  intermediate slices ( $N+1$  nodes). Then the heat conduction equation

$$\frac{dT}{dt} = \nabla (\alpha \nabla T) \approx \alpha \nabla^2 T = \alpha \frac{\partial^2}{\partial x^2} T$$

is solved for each element of this finite grid. On either side of a finite slab, boundary conditions are imposed for radiation (net) and convective heating and cooling. The number of nodes is chosen to reduce the error to some reasonable value, say 5%. Use of  $N > 20$  will improve precision without a concomitant increase in accuracy. For very thin walls only the surface nodes are necessary.

For the first time step, an initial temperature profile must be assumed. The usual assumption is uniform ambient temperature throughout

the solid. After the first step, the previous distribution can be assumed. As can be seen, this implementation differs markedly from the usual "control volume" approach.

The actual calculation usually employs a finite central difference, forward (explicit) time step scheme.

$$T_j(t + \delta t) = (1 - a\delta t) T_j(t) + \frac{a\delta t}{2} (T_{j+1}(t) + T_{j-1}(t) + 1)$$

with  $\delta t$  = time step interval

$$a = \text{differential coefficient} = 2 \alpha / (\delta x)^2$$

At the surface, the boundary condition

$$b \delta t \dot{q}$$

with  $b = (\rho c \delta x)^{-1}$ ,

$c$  = heat capacity per unit mass.

The flux  $\dot{q}$  may arise from convection or radiation and may be positive or negative.

When this formulation is applied, particularly to walls, ceilings and floors, a decision must be made as to the division of surfaces which are in the upper and lower layers, and what is done to the energy balance as the wall surface immersed in each layer changes. This method of heat conduction is done by NBS-I, BRI, Harvard and Dayton, although the Dayton implementation is not as complete as the others.

An alternative is to use an average conduction coefficient and assume that the energy loss is proportional to the temperature difference times this heat loss factor. Such a formulation is applicable to heat loss in early stages of a fire when only the ceiling is affected and principally by convective heating. At later stages, however, this model is clearly not appropriate. The models which use this method are NBS-II and Cal Tech.



## 6. CONCLUSIONS

We have examined the models which use the control volume concept in analyzing fire growth and spread. While none of the models is complete, all of them have features which can be well utilized in the formation of a general fire model. These features reflect the specialties of the respective authors.

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## Appendix A - Notation

- A area ( $m^2$ );  $A_u$ ,  $A_l$ ,  $A_d$  are the upper and lower compartment surface areas in contact with the upper and lower gas layer, Fig. (1), respectively.  $A_d$  is the interface area between the upper and lower layers. In section 4.1, "A" is used as a variable in the flow equations to indicate air.
- B width of a vent (m)
- C, C' flow coefficient  $\approx 0.6-0.7$  for both smoke and air
- c specific heat -  $c_p$ ,  $c_v$  (J/kg/k)
- $\dot{E}$  energy release rate (J/s)
- $F_{ij}$  view factor - relative area of "i" as seen by "j" (dimensionless)
- g acceleration of gravity ( $9.8 \text{ m/s}^2$ )
- H height (m),  $H_u$ ,  $H_l$ , are the upper and lower limits of a vent - Fig. (2)
- h enthalpy (J/kg/k)
- $h_c$  heat of combustion - theoretical (J/kg)
- i, j compartment indices
- L mean beam length (m) equivalent opaque sphere
- m mass (kg)
- $\dot{m}$  mass flow (kg/s):  
 $\dot{m}_v$  - rate of release of volatiles  
 $\dot{m}_e$  - entrained into a plume  
 $\dot{m}_f$  - fuel release  
 $\dot{m}_{ij}$  - mass entering room "i" from room "j"  
 $\dot{m}_p$  - flow rate in plume ( $\dot{m}_p = \dot{m}_f + \dot{m}_e$ )
- N height of the neutral plane (m)

P pressure (pa):  $\bar{P} \rightarrow P$  - floor reference pressure

$P_e$  - Eqn. (19)

$P_a$  - outside ambient pressure

$\dot{Q}$  rate heat is added or lost (J/s):

$\dot{Q}_u, \dot{Q}_l$  - upper, lower zones,  
respectively

$\dot{Q}_f$  - fire ( $h_c \dot{m}_v$ )

$\dot{Q}_o$  - objects

$\dot{Q}_R$  - radiation

$\dot{Q}_c$  - convection by walls

$\dot{Q}_g$  - radiation added to upper  
gas layer

$\dot{Q}_k$  - radiation from surface "k"

$\dot{Q}_p$  - combustion energy lost by  
formation of volatiles

R gas constant for specific mixture

S smoke - section 4.1

t time (s)

T temperature (k):  $T_a$  - ambient

$T_c$  - external wall

$T_u$  - upper wall

$T_l$  - lower wall

$T_R$  - reference temperature for enthalphy flow

$T_g$  - upper zone temperature

$T_v$  - volatile temperature

V volume ( $m^3$ )

$Z$  layer thickness (m)  
 $\alpha$  absorption coefficient of upper gas layer ( $m^{-1}$ ), thermal diffusivity ( $m^2/s$ )  
 $\gamma$  ratio of specific heat ( $c_p/c_v$ )  
 $\epsilon$  emissivity (dimensionless):  $\epsilon_i$  - surface "i"  
 $\epsilon_g$  - upper gas layer  
 $\epsilon_u$  - upper compartment surface  
 $\epsilon_l$  - lower compartment surface  
 $\rho$  mass density ( $kg/m^3$ )  
 $\kappa$  thermal conductivity (j/msk)  
 $\delta_{ij}$  Kronecker delta = 0  $i \neq j$   
= 1  $i = j$

Subscripts - In general "u" and "l" indicate upper and lower gas layer, respectively. For area and emissivity variables, reference is to the compartment itself.



## Appendix B - Summary of the Models

The references contain reasonably complete descriptions of the models. In addition, copies of the computer codes are available from the authors. The following list is a suggested starting point for each of the models, and is the primary summary of the contents and capabilities of each:

<u>Model</u>	<u>References</u>
NBS-I	13
NBS-II	5
BRI	18
Harvard	19
Cal Tech	6
Dayton	2

To summarize, the most complete transport phenomena are contained in the BRI model, and the most consistent radiation transport in NBS-I. Both the Dayton and Harvard models attempt to deal with the problem of a self-consistent fire source model. As indicated in section 5.1, this is a difficult problem given our understanding of flames shapes, turbulent mixing and the chemical kinetics of fire oriented combustion.

None of the models handles the numerics well. Part of the problem is mixing differential and algebraic equations, both of which are stiff. (Stiff in this context refers to two or more processes with widely (greater than a factor of ten) varying time constants.) Another part of the problem is in the discreteness of the source functions and their derivatives. Finally, one has to look carefully for possible instabilities in the numerics themselves when the approximation, discussed in section 1, are made.

# RADIATION PATH BETWEEN TWO SURFACES WITH AN ISOTHERMAL GAS BETWEEN THEM

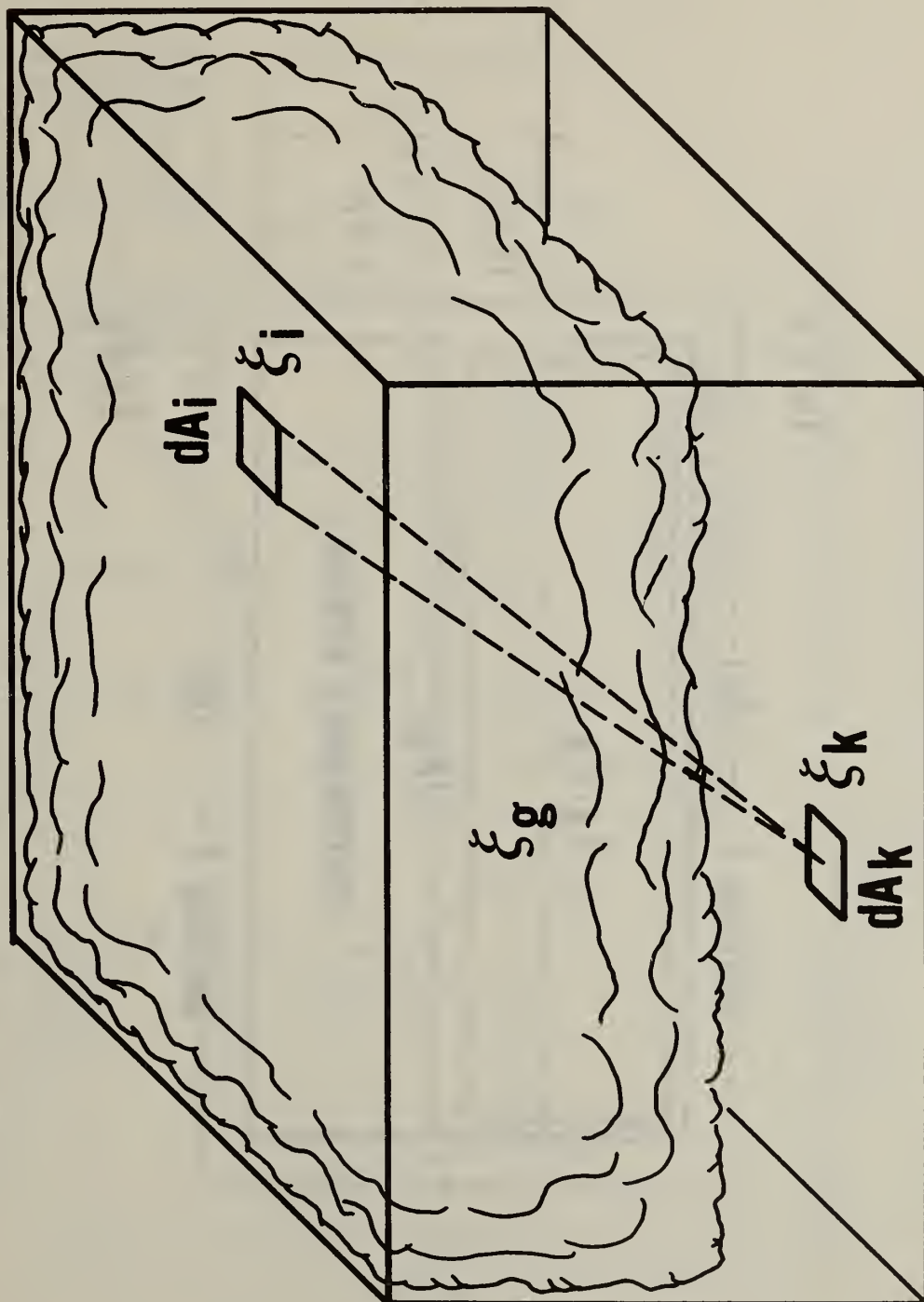


Figure 1. Geometry used in calculating the radiative transfer. The box is a special case of the general geometry used in ref. [18].



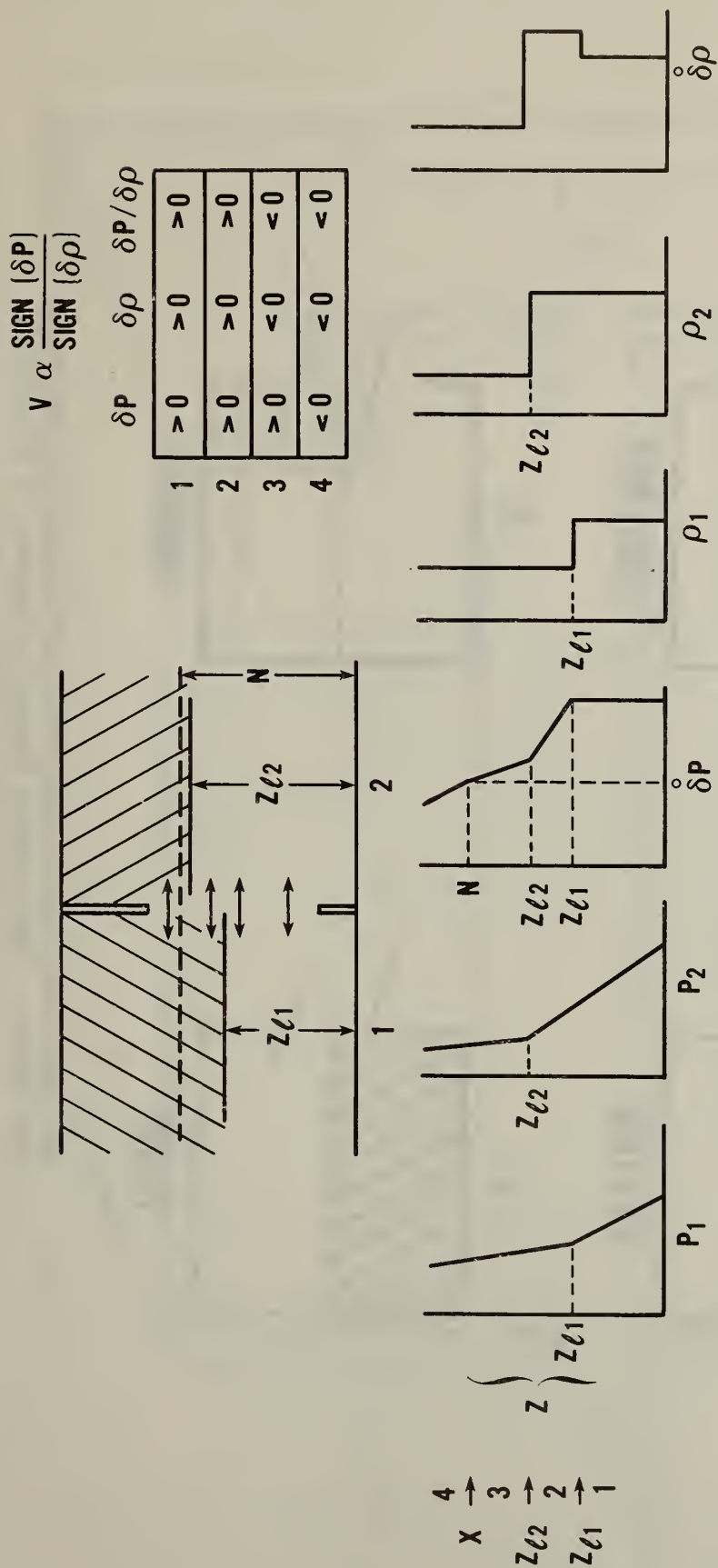


Figure 3. Selection rules for determining the direction of flow driven solely by buoyancy.

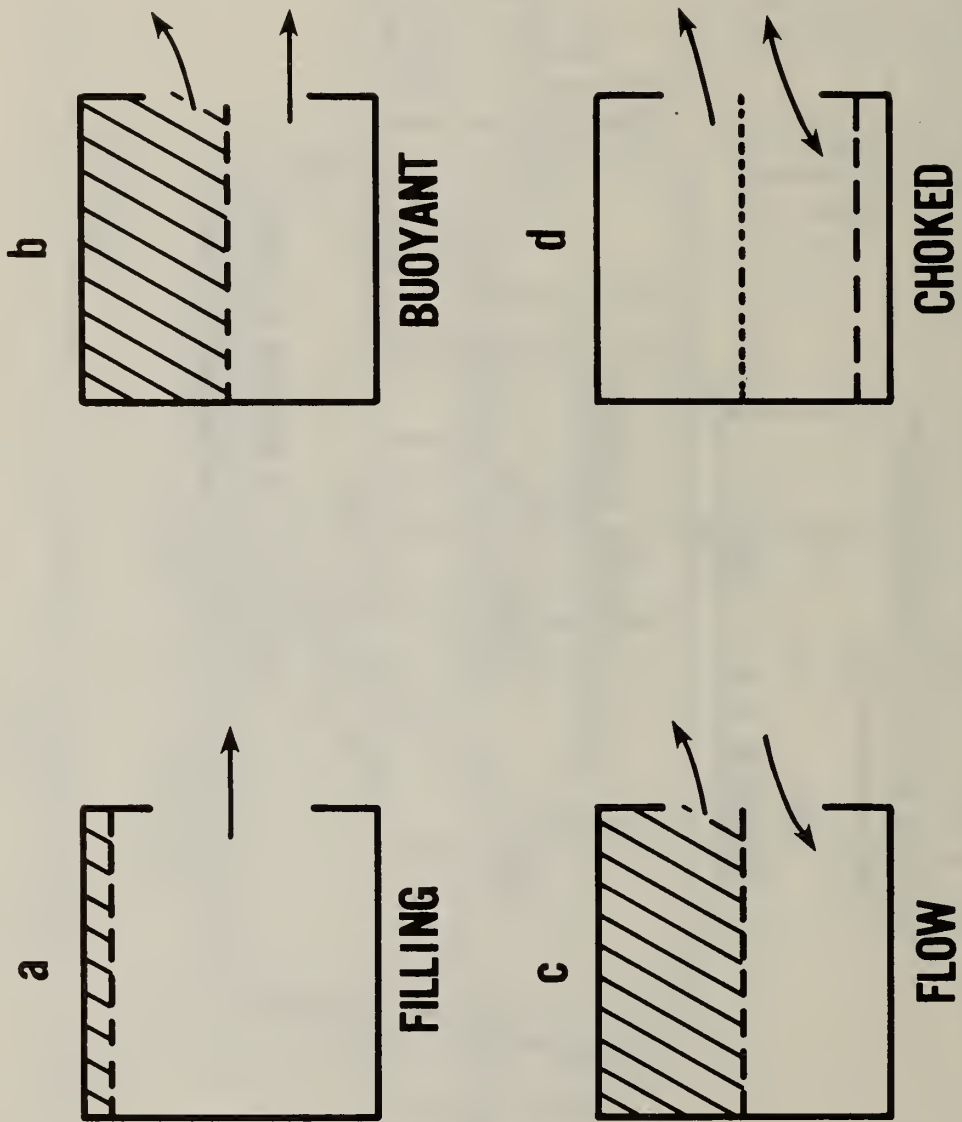


Figure 4. The four flow regimes considered in the single compartment flow models. The external region is assumed to be ambient.



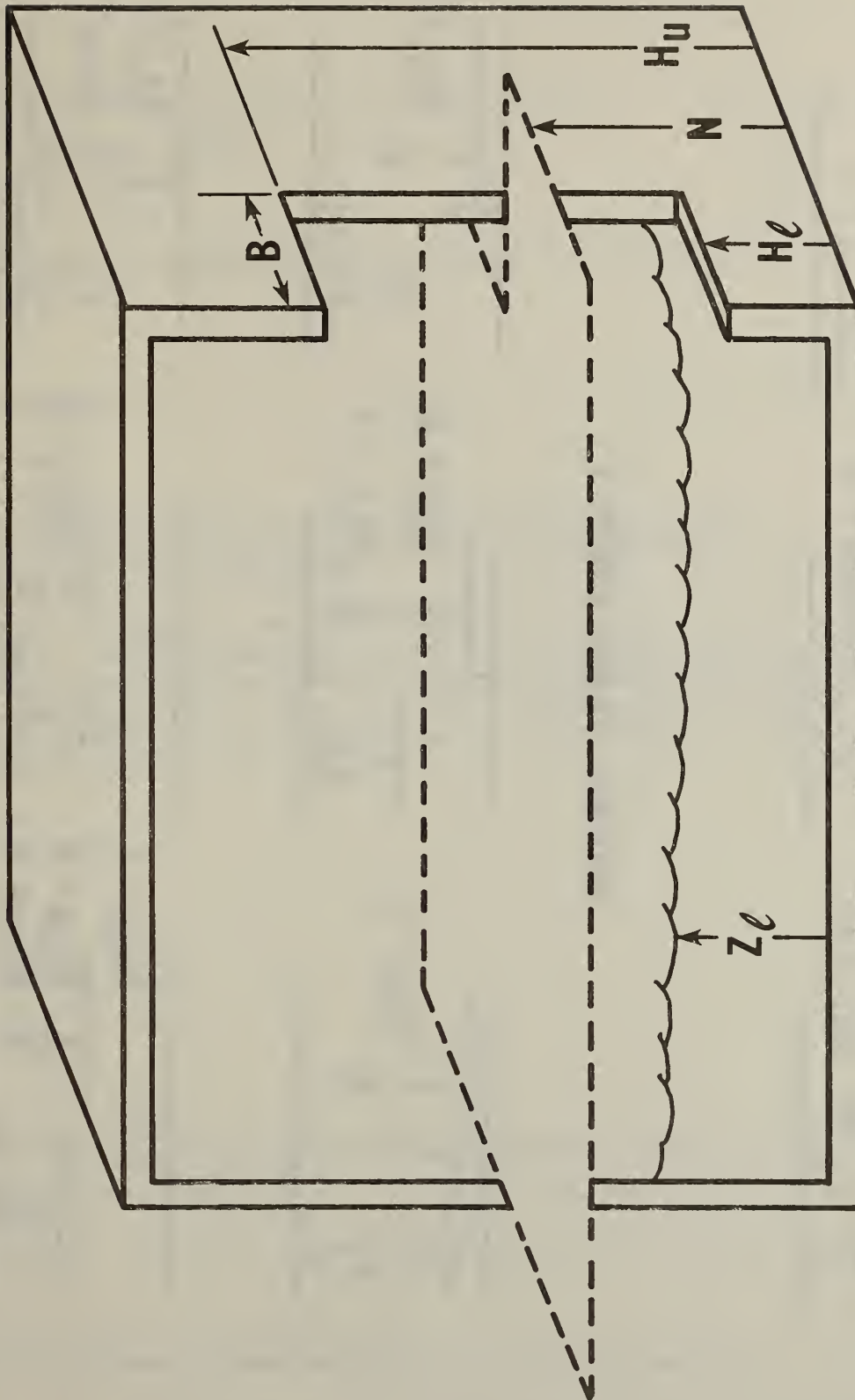
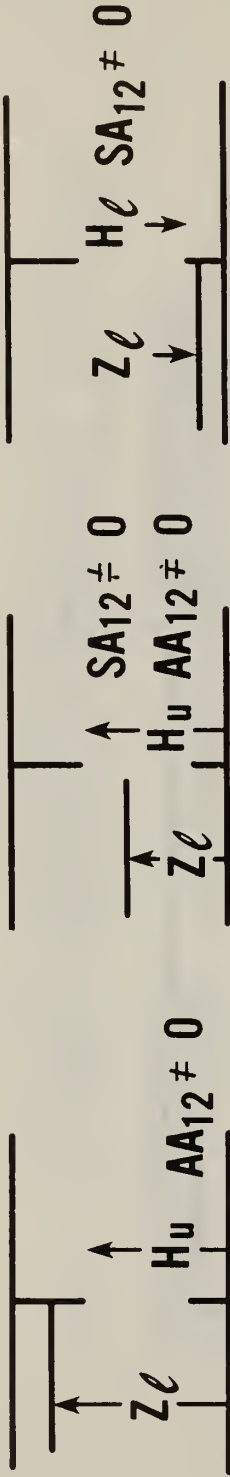


Figure 5. One possible configuration of the neutral plane height ( $N$ ) and the lower layer/upper layer discontinuity ( $Z_\ell$ ). Also indicated in the figure is the notation used for vent specification ( $B, H_\ell, H_u$ ).

## NEUTRAL PLANE BELOW $Z_\ell$



## NEUTRAL PLANE ABOVE $Z_\ell$

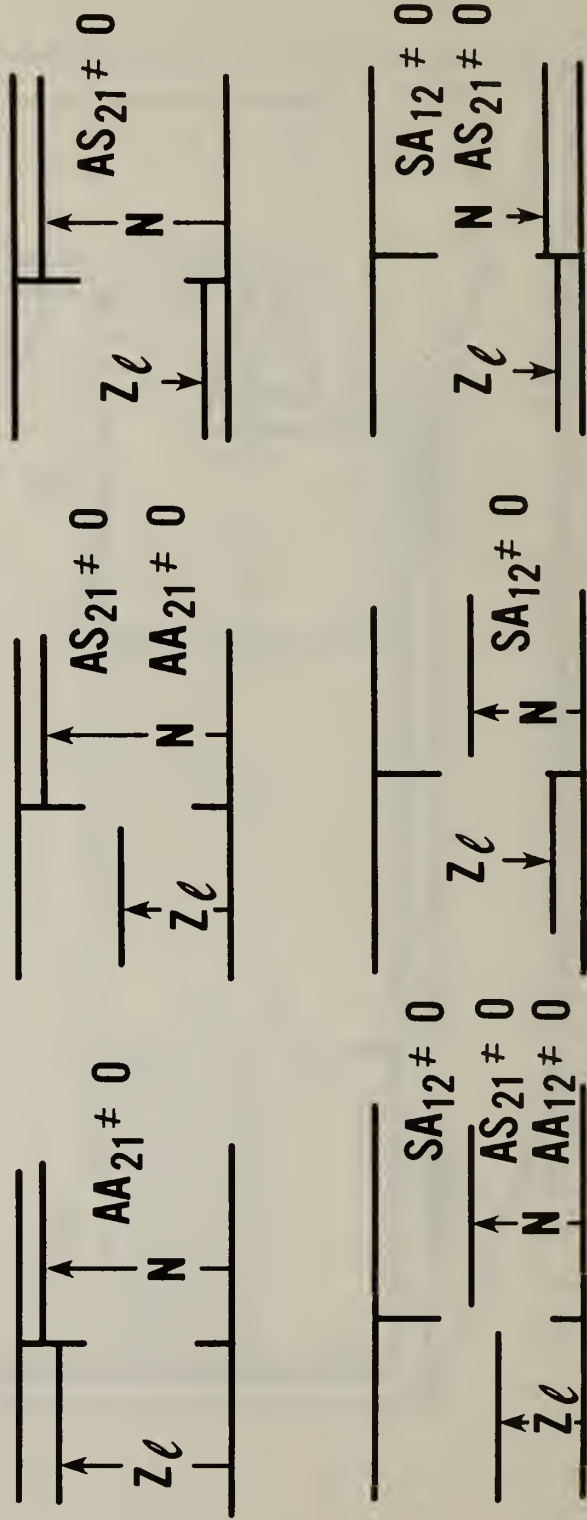


Figure 6. The nine possible flow conditions. The "SA<sub>ij</sub>" notation is taken from Tanaka [18]. Compartment two is assumed to be ambient. The selection rules determine which flow formula is to be used to calculate the vent flow.

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<b>10. SUPPLEMENTARY NOTES</b>  <input type="checkbox"/> Document describes a computer program; SF-185, FIPS Software Summary, is attached.			
<b>11. ABSTRACT</b> <i>(A 200-word or less factual summary of most significant information. If document includes a significant bibliography or literature survey, mention it here)</i> <p>We have examined the concept of zone modeling as it has been applied to fire problems. The existing models which embody the zone concept are compared in order to ascertain the state-of-the-art in understanding of fire growth and fire spread processes. This review is intended to be a starting point for future modeling efforts and thus the discussion centers around the use of the various submodels.</p>			
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