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THERMAL EXPANSION OF LEAD

By Peter Hidnert and W. T. Sweeney

ABSTRACT

Measurements have been made on the linear thermal expansion of three samples of cast lead between room temperature and 300° C. and the results have been correlated with data obtained by other investigators between 1740 and 1931.

correlated with data obtained by other investigators between 1740 and 1931. A curve has been derived which shows the linear thermal expansion of lead between -253° and $+300^{\circ}$ C. The summary gives average coefficients of expansion for various temperature ranges between -250° and $+300^{\circ}$ C.

A comparison of the indirect results by Kopp and Matthiessen with the direct data by other observers, indicates that lead expands the same in all directions.

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I. INTRODUCTION

The thermal expansion of lead has been of considerable interest for nearly 200 years. Its measurement has been the object of more than 25 investigations. A summary of available data obtained by various investigators is given in Table 1.

Between 1740 and 1831, a number of determinations of the coefficient of linear expansion were made for the range from 0° to 100° C. In 1831, Daniell reported the changes in length of a bar of lead heated from 17° to 100° C. and to the point of fusion, respectively. From that time up to 1930, no measurements on the linear thermal expansion of lead above 110° C. have been located in the literature. The present authors appear to have been the first observers after Daniell to report data on the linear thermal expansion of lead above 110° C. Their abstract giving coefficients of linear expansion on heating for various temperature ranges between 20° and 300° C. was published ¹ in February, 1930. In October of the same year, Uffelmann ² published coefficients of linear expansion for various temperatures between 80° and 280° C.

The present investigation was undertaken in order to obtain reliable data on the linear thermal expansion of lead above 100° C. The data obtained between 20° and 300° C. have been correlated with results by other observers.

The authors wish to express appreciation to H. W. Bearce, W. Souder, and H. S. Rawdon for valuable suggestions, and to H. S. Krider for assistance during the preparation of the manuscript.

¹ Hidnert and Sweeney. (See Table 1.)

² See Table 1.

	Remarks	Computed in 1932 from data by Daniell. Coefficients of linear expansion were computed by Kopp from coefficients of cubical expansion obtained by density method	Computed in 1932 from from his data on cubical expansion obtained by weighing 2 samples of lead in water at different temperatures. Computed in 1932 from coefficient of cubical expansion reported by Buff. Computed in 1932 from coefficient of cubical expansion reported by With and Computed in 1932 from coefficient of cubical expansion reported by Vicentini and Omodei.
servers	Coeffi- cient of linear expan- sion pen	210 22 22 22 22 22 22 22 22 22 22 22 22 22	29. 2 29. 5 29. 1 29. 5 29. 4 29. 5 29. 5 20. 5
TADA I. Summary of expansion and on lead of various obs	Temperature or temperature range	$\left\{\begin{array}{c} \circ C \\ 0 \ 100 \$	$\begin{cases} 0 \text{ to 100} \\ 100 \text{ temperature to 100.} \\ 325 \\ 325 \\ (2) 325 \\ -180 \text{ to } +20 \\ -180 \text{ to } +20 \end{cases}$
	Material	Lead do do do do do Lead, cast Lead, cast in a	we l1 - sm ok ed mold and var- nished. Lead, cast Lead wire Lead wire Lead rod made from drawn wire. Lead, cast
	Reference	 Hist. Acad. Roy. Sci., p. 230, 1745 Phil. Trans., vol. 48, pt. 2, p. 598, 1754 Phil. Trans., vol. 48, pt. 2, p. 598, 1754 Mém. Sci. Math. Phys. Inst. Imp. France, 2d pt., p. 89, 1811. Phil. Trans., vol. 121 (pt. 1), p. 443, 1831, or Phil. Mag., series 3, vol. 1, pp. 197, 261, 1832. Ann. Chem. Pharm., vol. 81, p. 1, 1852 Rep. Brit. Assoc., vol. 28, p. 46, 1859; Proc. Roy. Soc. (London), vol. 10, p. 315, 1859-60 and Chem. News. vol. 3, pp. 315, 357, 371, 1861. 	Physik ú. Chemie, vol. 130, p. 50, 1867. Compt. Rend., vol. 68, p. 1125, 1869 or Ann. Physik u. Chemie, vol. 138, p. 26, 1869. Ann. Physik u. Chemie, vol. 145, p. 626, 1872. (Erlangen. Besold. 1872. 8° 1–20), Fortsch. Physik, vol. 29, p. 579, 1873. Ann. Physik u. Chemie, vol. 160, p. 497, 1877. Proc. Roy. Soc. (London), vol. 25, p. 280, 1876–77. Atti Accad. Sci. Torino, vol. 23, p. 38, 1887–88. Phys. Rev., vol. 27, p. 1, 1908.
	Date	1740 (?) 1745 Before 1754 Before 1754 Before 1779 Before 1808 1811 1831 1852 1852 1859 1856	1869 1872 1872 1877 1877 1877 1888 1888 1908
	Observers	Muschembroeck ¹ Bouguer Buliout Bulliout Smeaton Herbert 1. Berthoud ¹ Lavoisier and Laplace ² Daniell and Laplace ² Calvert, Johnson, and Lowe.	Fizeau

TABLE 1.-Summary of expansion data on lead by various ob

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Sweet	ney			Thern	nal Expansion of	Lead	
	(Determined expansion of lead by comparison with expansion of fused quartz. Linde- mann states that he neglected the expan- sion of the latter for it is extremely small	(and wrenth the experimental error.	Computed in 1932 from the data reported by Rhert.	Average values obtained on 3 samples from observations on heating.		Determined relative expansion of lead and fused quarts; the curve of the relative change of length is nearly horizontal in the region of the superconductivity point of lead (-266° C) . The authors state that as there was no discontinuity observed in the thermal expansion curve for lead, it seems fair to conclude that when with the lowering temperatures the superconduc- tivity point was passed, no sudden change in the crystal lattice of the metal accom- panied the appearance of superconduc- tivity.	
27.0 29.3 28.8	19.8 20.3 26.6	29.3 29.5	24. 7 26. 1 20. 8	28.9 29.0 31.4 28.9 28.9 28.9 28.9 28.9 28.9 28.9 28.9	299 1 299 1 299 2 299 2 200 200		
$\left\{ \begin{array}{c} -190 \text{ to } +17 \\ +17 \text{ to } 100 \\ 18 \end{array} \right $	$\begin{cases} -253 \text{ to } -192 \\ -190 \text{ to } -183 \\ -190 \text{ to } +20 \end{cases}$	+15 to 110 10 to 100	$\left[\begin{array}{c} -253 \text{ to } 0 \\ -190 \text{ to } 0 \\ -953 \text{ to } -190 \end{array}\right]$	$\begin{array}{c} 0 \\ 0 \\ 2 \\ 2 \\ 2 \\ 2 \\ 2 \\ 2 \\ 2 \\ 2 \\$, 2560 2560 2560 2560 2560 2560 2560 2560	-268 to -246	
Lead rod	Lead, chemically pure.	Lead, cast		Lead, cast	Lead, annealed	Lead	tre, p. 1, 1808).
Ann. Physik, vol. 33, p. 33, 1910	Physik. Z., vol. 12, p. 1197, 1911	Öfversigt Finska Vetenskaps-Soc. Fördh., vol. 54, 1911–12, Afd. A., N: 0 24. J. Inst. Metals, vol. 31, p. 75, 1924	Z. Physik, vol. 47, p. 712, 1928	Phys. Rev., vol. 35, p. 296, 1930	Phil. Mag., vol. 10, p. 633, 1930	Trans. Roy. Soc. Can., III, series 3, vol. 25, p. 1, 1931.	m. Sci. Math. Phys. Inst. France, second semest
1910	1911	1912 1924	1928	1930	1930	1931	ryeau (Mé
Grüneisen	Lindemann	Rauramo and Saarialho Friend and Vallance	Ebert	Hidnert and Sweeney	Uffelmann	McLennan, Allen, and Wilhelm.	1 Cited by Guyton de Mo

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² Cited by Daniell (Puil. Mag., vol. 10, pp. 191, 268, 330, 1831). ³ Computed in 1932 from Matthiessen's equation.

II. MATERIALS INVESTIGATED

Three samples of cast lead were investigated. The purity and the method of casting of the samples are indicated in Table 2. The length of each sample was about 300 mm and the cross section 10 mm square.

III. APPARATUS

The furnace shown in Figure 1 of Scientific Paper of the Bureau of Standards No. 488, was used for the measurements of the linear thermal expansion of sample 1001 and the white furnace shown at the extreme left of Figure 1 of Scientific Paper of the Bureau of Standards No. 524 was used for samples 1144 and 1215. Figure 4 of the latter paper indicates the method used in mounting samples 1144 and 1215 in the furnace.

Expansion measurements were made by means of micrometer microscopes, which were sighted on fine wires suspended from or in contact with the ends of the specimen. For a detailed description of the apparatus and the methods used the reader should refer to the publications mentioned.

IV. RESULTS

Observations were made on the linear thermal expansion of three cast samples of lead at various temperatures between room temperature and 300° C. and the results obtained are shown in Figure 1. The expansion curves are plotted from different origins to display the individual characteristics of each curve.

The average coefficients of expansion given in Table 2 were derived from the observations on heating and on cooling. This table also gives the differences in length before and after the expansion tests. The plus (+) sign indicates an increase in length and the minus (-)sign a decrease in length.

	Lead con- tent ¹	Method of casting	Test No.	Heating or cooling	Aver	Change in length			
Sam- ple					20° to 60° C.	20° to 100° C.	20° to 200° C.	20° to 300° C.	after heating to 300° C. and cooling to room tem- pera- ture
	Per cent				X10-8				
² 1001		Preheated steel mold	1	Heating	29.2	29.6	31.2	$\times 10^{-6}$ 32.5	Per cent $\left.\right\} +0.02$
³ 1144	99.9	Sand mold	1	Heating	28.3	29.5 28.6 29.5	30. 4 29. 5	31.0 31.2 31.2	.00
³ 1215	99.8	Sand mold, from same ingot as sample 1001.	$ \left\{\begin{array}{c} 1\\ 2 \end{array}\right. $	{Heating \Cooling {Heating \Cooling	29.0	28.9 28.8 28.8	29.7 29.4 29.8	30. 9 31. 1 31. 0 31. 0	$\left. \left. \right\}01 \\ .00 \\ \right\}$
		Average			28.8	29.1	30.0	31.3	

TABLE 2.—Average coefficients of linear expansion of cast lead

Analysis by H. A. Buchheit, of this bureau.
 See sample 1215.
 Density for samples 1144 and 1215, 11.310 and 11.329 g/cm³ at 25° C., respectively, determined by Miss E. E. Hill, of this bureau.

The differences obtained in the coefficients of expansion of the three samples of cast lead are probably due to variations in the methods of casting.



FIGURE 1.-Linear thermal expansion of three cast samples of lead

Sample 1001 which was cast in a preheated steel mold, has larger coefficients of expansion than the samples cast in sand molds. From the observations obtained on cooling, it appears that the coefficients of expansion of sample 1001 on the next heating would be less than those obtained on the first heating.

The curve in Figure 2 represents the linear thermal expansion of lead between -253° and $+300^{\circ}$ C. The portion of the curve between -253° and $+20^{\circ}$ C. represents the data obtained by previous



FIGURE 2.—Comparison of the average expansion curve derived in the present investigation on lead, with data from the observers

The lower portion of the curve from -253° to $+20^{\circ}$ C. was derived from data by Lindemann, Ebert, Dorsey, and Grüneisen, and the upper portion of the curve from 20° to 300° C. represents the average of all results obtained in the present investigation on three samples of lead.

investigators, and the portion between 20° and 300° C., the average of all results obtained by the present authors on the three samples of lead. The values obtained by other observers are included in this figure for comparison. Most of these values show good agreement. Bouguer and Muschembroeck obtained the lowest values and Berthoud the highest value at 100° C.

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Kopp and Matthiessen's data for linear expansion obtained indirectly from density measurements agree closely with direct measurements made by other observers, and therefore it appears that the expansion of lead is the same in all directions.

The coefficient of linear expansion derived from Vicentini and Omodei's value for the coefficient of cubical expansion of lead near the melting point, appears to be too low.

McLennan, Allen, and Wilhelm published a curve which shows the relative change of length measured against the change of length of fused quartz near absolute zero. Their conclusion relating to the lattice structure of lead, which is based on the assumption that the expansion curve of lead is almost horizontal in the region of the superconductivity point $(-266^{\circ} \text{ C.})$, does not appear to be justified. Since the expansion of fused quartz is not known near absolute zero, it is not possible definitely to determine from their curve the actual expansion of lead in this region. It is possible that when these observers cooled lead and fused quartz in the region of the superconductivity point of lead, the contraction of the lead nearly balanced the expansion of the fused quartz³ and thus they obtained a nearly horizontal curve for the relative expansion.

V. SUMMARY

Data have been obtained on the linear thermal expansion of three samples of cast lead. Observations were taken at various temperatures between room temperature and 300° C., and the data have been correlated with available data by other investigators.

The average coefficients of linear expansion for various temperature ranges between -250° and $+300^{\circ}$ C., as derived from the expansion curve in Figure 2 are as follows:

•		V10 -
-250° to +	· 20° C	25.1
-200° to	20° C	26.5
-100° to	20° C	28.3
$+20^{\circ}$ to \cdot	60° C	28.8
20° to	100° C	29.1
20° to	200° C	30.0
20° to	300° C	31.3

WASHINGTON, September 12, 1932.

³ Souder and Hidnert, B. S. Sci. Paper No. 524 (fig. 11). 141809-32----9