

# Description and Analysis of the First Spectrum of Iodine

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An extensive survey of the spectra of iodine has led to a list of more than 900 lines emitted by neutral atoms in the region from 23070 Å in the infrared to 1195 Å in the extreme ultraviolet. Wavelengths between 12304 Å and 2061 Å were derived from measurements of spectrograms obtained with gratings of high dispersion. Wavelengths of lines outside these limits are the computed values for lines observed on photometric tracings of the infrared, inaccessible to photography, and in the ultraviolet with a vacuum-grating spectrograph. For many of the lines Zeeman patterns were obtained in a magnetic field of about 37,000 oersteds. With these data many of the lines have been classified as combinations between odd levels from the electron configurations  $5s^2 5p^4 np$  and  $5s^2 5p^4 nf$ , and even levels from the configurations  $5s^2 5p^4 ns$  and  $5s^2 5p^4 nd$ . Among these levels several sets have been recognized as forming Rydberg sequences that are in close agreement in placing the ground state  $5p^5 {}^2P_{1/2}$  of I I at 84,340 cm<sup>-1</sup> below the ground state  $5p^4 {}^3P_2$  of I II. This gives 10.45 electron-volts for the ionization potential of the neutral iodine atom. A strong infrared line at 13148.8 Å is explained as a magnetic dipole transition between the levels of the ground term  $5p^5 {}^2P^o$ .

## 1. Introduction

The spectra absorbed and emitted by iodine in its atomic and molecular states have been the object of many investigations. In volumes 5 and 8 of his *Handbuch der Spectroscopie*, Kayser lists 352 papers, which appeared up to 1933, dealing with various aspects of the spectral behavior of this heavy member of the halogen family. Since that date additional papers have appeared, of which some are cited below. But in spite of this abundant material, representing a vast amount of work, knowledge of the first spectrum of iodine, I I, emitted by neutral atoms, has remained scanty and fragmentary, largely owing to the fact that important parts of the spectrum lie in the not easily accessible ultraviolet and infrared regions. It is the purpose of this paper to present a new description of the first spectrum of iodine and an analysis of its term structure.

## 2. Experimental Procedure

The investigations of the first spectrum of iodine at the Bureau were made at two different times under different experimental conditions. The first series of observations was made more than 30 years ago when chlorine and bromine [1]<sup>1</sup> also were being investigated. The light source was a Geissler tube of Pyrex glass into which a small amount of dry iodine vapor could be admitted from time-to-time to replace that which was adsorbed on the walls of the tube or absorbed by its aluminum electrodes. The lamp was similar to that used in the experiments on chlorine and bromine, and was excited to luminescence in the same way, with an uncondensed discharge from the high-voltage side of a 40-kv transformer. The spectrograms were recorded on plates sensitized to the green, orange, red, and infrared regions of the spectrum by bathing ordinary photoplates in solutions of the photosensitizing dyes available at that time; namely, pinaverdol, pinacyanol, dicyanin, and the newly

discovered kryptocyanin. The spectrographs carried concave gratings of 21-ft radius of curvature, ruled with 7,500 and 20,000 lines per in. and set up in Wadsworth mountings. Each exposure to the light source was made with one-half the length of the spectrograph slit covered with a colored-glass filter so that both the first-order spectrum and the overlapping second order could be obtained at the same time. Each plate was exposed also to light from the iron arc, in both the first and second orders, to obtain the necessary standard wavelengths for use in deriving the iodine wavelengths. Because the capillary of the discharge became discolored after a run of a few hours, it was necessary to make exposures of nearly 24-hr duration in order to photograph the lines of longest wavelength recorded on the plates.

Measurement of these spectrograms yielded a list of approximately 400 wavelengths, with estimated intensities, extending from 9732 Å in the infrared to 3820 Å in the ultraviolet. This list was seemingly the most extensive description of the first spectrum of iodine then available and was being prepared for publication when the paper, "The Arc Spectrum of Iodine," by Evans [2] appeared. A comparison of his list and ours showed that they were essentially identical. This fact and also the fact that his paper contains the first real results for a classification of the iodine spectrum outside of the Schumann region induced us to defer publication of our results until a substantial addition could be made to the description and analysis of the spectrum.

The second series of observations was made at various times during the period from 1953 to 1957. Improved apparatus and new experimental procedures have made it possible to advance the description of the iodine spectrum beyond the limits reached in the earlier work, and also to obtain Zeeman-effect observations that have led to a revision and extension of its term structure. The new light source was an electrodeless discharge tube of the type described by Corliss, Westfall, and Bozman [3]. It was excited to luminescence in a field of 2,450 Mc

<sup>1</sup> Figures in brackets indicate the literature references at the end of this paper.

from a microwave oscillator. The plates used to record the spectra were EK types 103a-O-UV, 103 a-O, I-F, I-N, I-Q, and I-Z, according to the region investigated, and, where required, were hypersensitized in an ammonia bath by the method recommended by Burka [4]. Four concave-grating spectrographs and a Hilger E 1 quartz-prism spectrograph were used to obtain the spectrograms. The spectrographs carrying the gratings with 7,500 and 15,000 lines per in. were used for the infrared and red regions where many new strong lines were found. For the shorter regions the grating with 30,000 lines per in. was used as well as the one with 15,000. For the extreme ultraviolet both the Hilger E 1 instrument and a 2-m glass grating ruled with 30,000 lines per in. and mounted in a vacuum chamber were used. All the spectrograms bore exposures to the iron arc or other sources of standard lines to be used in the determination of the iodine wavelengths.

For the Zeeman-effect observations the Weiss water-cooled magnet of the Bureau was used. With a current of 160 amp through the coils and a gap of 5 mm between the pole pieces, a field of approximately 37,000 oersteds was produced. The source between the pole pieces was also an electrodeless lamp of the type mentioned above, but of diameter 4 mm. A Wollaston prism of quartz placed between the light source and the projection lens of the spectrograph separated the two polarizations on the slit, with space between them for a no-field exposure. On plates appropriate for the regions under investigation resolved magnetic patterns were recorded for nearly all the strong lines of I<sub>I</sub> from 2062 Å in the ultraviolet out to 11246 Å in the infrared. Zeeman patterns were recorded also for some I<sub>II</sub> lines of long wavelength.

### 3. Results

The observational data and the deductions from atomic theory that are essential for the description of the spectrum of iodine and its term structure are embodied in the tables appended to this paper. In tables 1, 2, and 3 are listed, in the first three columns, the wavelengths of the lines of I<sub>I</sub>, their estimated intensities and characteristics, and their wave numbers in vacuum. The letters after the intensity numbers have the following significance: *c*=partially resolved hyperfine structure (hfs); *d*=double; *w*=widened line owing to unresolved hfs; *h*=hazy, diffuse; *Z*=Zeeman pattern given in table 4. The letters *A* and *B* indicate the type of shading displayed by unresolved patterns; thus: *A*=; *B*=. The term combinations in the fourth column of the tables are based on *g*- and *J*-values derived from the Zeeman-effect patterns of table 4.

Although the intensities are estimates made according to the usual practice of spectroscopists, an attempt has been made to bring them into closer relationship with photometric standards than is possible in a compressed linear scale in which the numbers are roughly proportional to the logarithms of the actual intensities. In table 2 the intensities are the measured heights of the peaks of the lines above the noise level of the recorder tracings.

In tables 1 and 3, however, an attempt has been made to bring the estimated intensities into harmony with a photometric scale that reflects the enormous range in the strength of the lines. On this scale the faintest lines are arbitrarily assigned an intensity 1, and the strongest lines, that occur in the multiplet 6s<sup>4</sup>P—6p<sup>4</sup>P° are designated as 10<sup>5</sup> times as strong. This ratio is based on an accurate determination of the relative strengths of Rowland ghosts to their parent lines. Thus, for the strongest lines estimates were made of the intensities of their ghosts, and then multiplied by the corresponding factors to establish the intensities of the parent lines. Although the scale was thus established to represent true relative strengths of lines in short ranges of the spectrum, no attempt was made to standardize them over longer ranges of wavelengths. In particular the intensity scale used in the vacuum region is several orders of magnitude less than that of the visible and infrared. A few of the longest wavelengths listed in table 1 were measured also on the infrared recordings described above. The intensities derived for them from these observations are given in parentheses following the estimated photographic intensities, thereby giving a comparison of the relative sensitivity of the two modes of observation in a region near the limit of photographic detection.

The terms to be expected theoretically on the assumption that *LS*-coupling governs the orbital and spin angular momenta of the atom are given in table 5. Those actually found in this investigation are given in tables 6 and 7. These terms make it possible to calculate accurate wavelengths for lines in the infrared and extreme ultraviolet regions that lie beyond the reach of photographic recording with high-dispersion spectrographs. Such lines as have been observed by other investigators are given in tables 2 and 3 with wavelengths calculated from the terms of tables 6 and 7.

Lines listed in table 2 were observed by C. J. Humphreys [5] at the Naval Ordnance Laboratory, Corona, and by E. K. Plyler [6] at the Bureau. In this work they used electrodeless-discharge lamps, similar to those mentioned above in conjunction with their recording infrared spectrometers. These observations, which were made expressly for this investigation, verify all but two of the new infrared lines measured by Eshbach and Fisher [7] and add several lines not previously observed. All the ultraviolet lines in table 3 were measured on spectrograms obtained with the 2-m vacuum-grating spectrograph of the Bureau. These data not only confirm the descriptions of I<sub>I</sub> given by Turner [8], La Croute [9], McLeod [10], and Hellerman [11], but increase by a factor of more than 4 the number of lines reported by these earlier observers. The light source was an electrodeless discharge of the type described above but modified to incorporate a LiF window sealed to the tube with an O-ring. The wavelengths listed in the first column are not the values derived directly from the measurements, but are values calculated from the energy levels given in tables 6 and 7. They were derived, therefore, indirectly from international

secondary standard wavelengths, and are believed to be correct to less than 0.005 Å. They are recommended for use as standards in the vacuum region.

A problem of prime importance for the analysis of the first spectrum of iodine is the evaluation of the separation of the levels of the ground state  $5p^5 \ ^2P^o$ . The lines at 2061.6 and 1830.4 Å, which are due to the transitions  $5p^5 \ ^2P_{0\frac{1}{2}} - 6s \ ^2P_{1\frac{1}{2}}$  and  $5p^5 \ ^2P_{1\frac{1}{2}} - 6s \ ^4P_{2\frac{1}{2}}$ , may be used for this purpose. These lines have been measured several times by different observers, but the wavelengths reported for them are only approximately correct and are afflicted with the errors that are inherent in the reference lines against which they were measured. In the present work the mean wavelength of the longer line has been determined as 2061.633 Å, from seven observations, made with high dispersion in the higher orders of the gratings, relative to international secondary standards in the iron arc spectrum.

The wavelength of the shorter line was determined as 1830.380 Å from measurements relative to internal standards selected from the iodine spectrum itself. These lines are at 1876, 1844, 1799, 1702, and 1593 Å, and appear with 1830 Å on spectrograms made with the vacuum-grating spectrograph. We have determined accurate wavelengths for them, from measurements of lines of longer wavelength on high-dispersion spectrograms, by making use of the combination principle. Thus, we find the following level-separations, which are mean values of the wave-number differences between numerous pairs of well-measured lines:

$$6s \ ^4P_{0\frac{1}{2}} - 6s \ ^2P_{1\frac{1}{2}} = 4803.39 \text{ cm}^{-1}$$

$$6s \ ^4P_{1\frac{1}{2}} - 6s \ ^2P_{1\frac{1}{2}} = 5726.93 \text{ cm}^{-1}$$

$$6s \ ^2P_{0\frac{1}{2}} - 6s \ ^2P_{1\frac{1}{2}} = 7093.88 \text{ cm}^{-1}$$

$$6s' \ ^2D_{1\frac{1}{2}} - 6s \ ^2P_{1\frac{1}{2}} = 10262.33 \text{ cm}^{-1}$$

$$nd \ 5.1_{1\frac{1}{2}} - 6s \ ^2P_{1\frac{1}{2}} = 14262.05 \text{ cm}^{-1}$$

By adding each of these numbers to 48489.73 cm<sup>-1</sup>, the wavenumber of 2061 Å, we obtain the wavenumbers, and thence the exact wavelengths, of the selected standards. These are given in table 3.

With the accurate values thus established for the two transitions given above we have:

$$1830.380 \text{ Å is } 5p^5 \ ^2P_{1\frac{1}{2}} - 6s \ ^4P_{2\frac{1}{2}} = 54633.46 \text{ cm}^{-1}$$

$$6s \ ^4P_{2\frac{1}{2}} - 6s \ ^2P_{1\frac{1}{2}} = 1459.42 \text{ cm}^{-1}$$

$$1782.758 \text{ Å is } 5p^5 \ ^2P_{1\frac{1}{2}} - 6s \ ^2P_{1\frac{1}{2}} = 56092.88 \text{ cm}^{-1}$$

$$2061.633 \text{ Å is } 5p^5 \ ^2P_{0\frac{1}{2}} - 6s \ ^2P_{1\frac{1}{2}} = 48489.73 \text{ cm}^{-1}$$

$$5p^5 \ ^2P_{1\frac{1}{2}} - 5p^5 \ ^2P_{0\frac{1}{2}} = 7603.15 \text{ cm}^{-1}$$

#### 4. Term Structure of I<sub>I</sub>

The spectrum to be expected theoretically for neutral iodine atoms, if *LS*-coupling governs their behavior under excitation, is that based on the terms given in table 5. In the unexcited state of the atom

the electron configuration is  $5s^2 \ 5p^5$ , which yields an inverted  $^2P$  term of odd parity. The higher states of even and odd parity arise when excitation of the atom leads to the electron configurations listed in the first column of table 5. Ionization of the atom leaves it in one of the states represented by the  $^3P$ ,  $^1D$ , and  $^1S$  terms of the basic electron configuration  $5s^2 \ 5p^4$  of the ion I<sup>+</sup>. These three terms give rise, therefore, to the three families of terms that are likely to produce the strongest lines of the spectrum I<sub>I</sub>. A fourth family also is expected based on the addition of *s*, *p*, *d*, etc., electrons to the  $5p^6$  configuration. The terms from these configurations will be doublets, and the lines arising from their combinations probably will be among the weaker lines of the spectrum.

The first real regularity in the spectrum of iodine was announced by Turner [12] who found the wave-number interval of approximately 7,600 cm<sup>-1</sup> recurring among several pairs of the lines observed by him in the Schumann region. Subsequently he correctly suggested that this difference represents the separation of the levels in the ground term  $^2P$  of the  $5s^2 \ 5p^5$  electron configuration. From this starting point all further advances in the interpretation of the first spectrum of iodine have been made. Evans [2], Deb [13], and Murakawa [14], in their analyses, have recognized these lines as resulting from the  $6s \rightarrow 5p$  transition, but they are not in agreement on their designations of the individual levels of the  $5p^4 \ 6s$  configuration. Inasmuch as the interpretations of I<sub>I</sub> offered by these investigators all rest on the *6s*-levels it is, therefore, not surprising that there is disagreement among them.

The classifications of the lines of I<sub>I</sub> given in this paper likewise are based on the *6s*-levels. With *g*- and *J*-values derived from well-resolved Zeeman patterns it is now possible to designate these levels with certainty and also some of the higher *np*-levels with which they combine. Increasing excitation of the iodine atoms brings into play still higher levels from the *np*, *nd*, *ns*, and *nf* electrons, but it is difficult to designate them with certainty because the *g*-values indicate a breakdown of the orbital and spin momenta to a coupling scheme between pure *LS*-coupling and *jj*-coupling, probably *jl*-coupling described by Racah [15].

Most of the levels of I<sub>I</sub> that are given in tables 6 and 7 result from the addition of *ns*, *np*, *nd*, and *nf* electrons to the lowest energy states  $^3P$ ,  $^1D$ , and  $^1S$  of I<sub>II</sub>. These were reported first by LaCroix [9] whose analysis, giving the relative positions of these terms, shows that the levels of  $^3P$  are separated by large differences in wave number. This fact has been an important guide in the analysis of I<sub>I</sub>, for nearly all the terms derived from  $^3P$  show similar characteristics. The eight levels of the  $5s^2 \ 5p^4 \ 6s$  electron configuration all closely conform to the pattern of spacing exhibited by their parentage. This is confirmed by the *g*-values of these levels, which are nearly equal to the *g*'s for *LS*-coupling, their sum being  $\Sigma g_{obs} = 11.985$  as compared with  $\Sigma g_{LS} = 12.000$ .

Of the levels coming from configurations with

*np*, *nd*, and *nf* electrons, only those derived from  $^3P$  levels can be designated with certainty. Here again the classifications rest on level intervals and *g*-values; but the *g*-values now show marked deviations from *LS-g*'s, owing to configuration interaction, and there is a tendency for levels to form pairs. In the case of the 13 levels derived by adding a *6p*-electron to the parent term  $^3P$ , *g*-sharing is such as to remove all resemblance to *LS g*-values, yet their sums are nearly equal, namely  $\Sigma g_{obs} = 17.847$  and  $\Sigma g_{LS} = 18.000$ . Few if any levels from the parent terms,  $^1D$  and  $^1S$ , are definitely established.

## 5. Series and Ionization Potential.

Before the first attempts were made to unravel the spectrum of iodine, several investigators reported the results of their measurements of critical and ionizing potentials in iodine vapor. Thus, in the period from 1920 to 1924, values of the ionization potential from 10 to 10.5 ev were reported by Found [16], Mohler and Foote [17], Duffendack [18], and Mackay [19]. From these results it was evident that the separation  $^2P_{1\frac{1}{2}} - ^3P_2$  between the ground states of I I and I II is approximately  $85,000 \text{ cm}^{-1}$ . Although the series announced by Deb [9] and by Price [20] give limits in close agreement with this value; yet, in the light of the present analysis, their series must be considered fictitious and the ionization potentials derived from them fortuitous.

The first physically real series of I I was given by Evans [2] in his analysis of the spectrum. Two of his levels in combination with the *6s*-term accounted for two pairs of strong lines separated by the same difference in wave number. The limit and ionization potential derived from them are within the range of the experimental values cited above. This series has been confirmed and extended in the present work. Other series of three and more members have been found also, as set forth in table 8. Four of these series, with four and five members, representing the migration of the *s*, *p*, and *d* electrons, have been used to calculate the separation of the ground states of I I and I II and the corresponding ionization potential. A Ritz formula  $R/(m+\alpha+\beta/m^2)^2$  has been evaluated for the variable terms of these series, and with the values given in the table for  $\alpha$  and  $\beta$ , it is found to represent the series very closely. In fact the first solutions of this formula were found to fit series of three members closely enough to predict higher series members that were subsequently found. The individual determinations of the interval  $^2P_{1\frac{1}{2}} - ^3P_2$  between the ground states of I I and I II are in close agreement among themselves, and when an unweighted average of them is taken, yield a value of  $84,340 \text{ cm}^{-1}$ . From this an ionization potential of 10.45 ev is derived for the work required to remove a *p*-electron from the configuration  $5s^2 5p^5$ .

## 6. The Continuous Spectra

An outstanding feature of the spectra of iodine emitted by the electrodeless lamps used in this in-

vestigation is a succession of continua extending from 4800 Å in the blue to 2900 Å in the ultraviolet. These bands have been encountered by nearly all observers who have studied the spectra emitted by iodine molecules and atoms subjected to various modes of excitation. Perhaps the best description of them is that published by Curtis and Evans [21]. Our observations are in agreement with their view that these continua belong to two systems, the one consisting of a single band or of several broad overlapping bands between 4800 and 4035 Å, the other consisting of a very broad strong band followed by several groups of much fainter and narrower bands. In each of these groups the individual bands are about 25 Å in width, nearly equally spaced, and increase in intensity to a maximum near the center of the group and then decline. The mean separation of the bands is  $210 \text{ cm}^{-1}$ , which is exactly the separation of the vibrational levels of the ground state of the  $I_2$  molecule as found by Kimura and Miyanashi [22] in their study of the ultraviolet absorption bands of  $I_2$ . This fact casts doubt on the results of various investigators who have sought to ascribe the bands to atomic recombination processes. A description of these emission features as recorded on our spectrograms is given in table 9. Their presence is unwelcome in investigations of I I because they mask completely the fainter and widened lines and attenuate differences in intensity of measurable lines that appear on the continuous background.

## 7. Discussion

The analysis of the first spectrum of iodine, presented in the preceding pages, shows clearly that in its excited states the atom has departed from *LS*-coupling and has reached a stage in intermediate coupling. Since the *g*-value of  $5p^5 2P_{0\frac{1}{2}}$ , of the ground state, is very closely that given by *LS*-coupling, we may assume that the *g*-value of the lowest member of the ground term,  $5p^5 2P_{1\frac{1}{2}}$ , is also that for *LS*-coupling. In the first excited states, however, arising from the electron configuration  $5s^2 5p^4 6s$ , the departure from *LS*-coupling is marked. Although the *g*-values deviate only slightly from the *LS*-values, the levels of the  $^4P$  and  $^2P$  terms show the separations that are characteristic of their parent term,  $^3P$ , the ground state of I II. This is in accord with the scheme for the  $p^4$  configuration in intermediate coupling as illustrated by Condon and Shortley [23].

In the configurations containing *np*-electrons, in which  $n \geq 6$ , almost all resemblance to *LS*-coupling has disappeared from the term structure. The *g*-values for all the levels deviate strongly from the *LS*-values, except those for  $^4D_{3\frac{1}{2}}$  and  $^4D_{2\frac{1}{2}}$  for which the deviation is slight. The levels fall into a pair structure, of the kind prescribed by Racah [15] for *jl*-coupling, in which the level separations bear no resemblance whatsoever to those of *LS*-coupling. The *g*-values calculated for the *np*-levels with Racah's formula fit the observed *g*'s more closely than do the *g*'s of *LS*-coupling. A similar situation holds for levels from configurations with *nd*-electrons, and also

with  $nf$ -electrons. The  $nf$ -levels fall very close to each other so that lines originating in them are separated by intervals less than those imposed by a magnetic field, say, of 35,000 oersteds. All of these lines for which magnetic patterns appear on our spectrograms of the Zeeman effect show the unsymmetrical structures, due to Paschen-Back interaction, similar to those described by Kiess and Shortley [24] for lines of oxygen and nitrogen.

These matters raise the question as to the appropriateness of the designations used for the energy levels of a heavy atom such as iodine. It is obvious that in a complex spectrum manifesting in its various states a transition from one coupling scheme to another, no single scheme of notation will be adequate or satisfactory. The only notation scheme that has achieved a status of widespread usage and permanence is the one devised for spectra built on  $LS$ -coupling. It is here emphasized that the use of it, in this paper, for levels that do not result from this coupling scheme, is for convenience in designating them, and not for attaching to them the quantum significance usually conveyed by the  $LS$ -symbols.

An inspection of table 1 will show that some lines have resisted all attempts to classify them. There is no doubt, however, that they belong to I II; but the necessary links to connect them to known or, as yet, unknown levels have not been found. One of the lines left unclassified after the bulk of the analysis had been completed is the relatively strong infrared line at 13149.19 Å as measured by Eshback and Fisher [7]. This line, measured by us also on the recordings of the infrared spectrum of iodine by Plyler and by Humphreys, has a wave number practically identical with the separation of the levels in the  $^2P_0^o$  ground term of the iodine atom. Accordingly it is designated in table 2 as the forbidden transition  $^2P_{0\frac{1}{2}} \rightarrow ^2P_{1\frac{1}{2}}$  and a wavelength corresponding to the wave number 7603.15 cm<sup>-1</sup> is calculated for it. A similar transition has been reported by Edlén [25] for the isoelectronic spectrum Xe II. We have tried to photograph this line on EK-IZ plates so as to get a more accurate value of its wavelength, but the experiments were unsuccessful. Our belief that our designation of it is correct is substantiated by the discovery of similar transitions between the metastable levels of I II, as reported by Martin and Corliss in their forthcoming paper on I II.

TABLE 1. Wavelengths and term combinations of I II

Wavelength	Intensity	Wave number	Designation
12304.77	10(435)	8124.71	$6p\ 4S_{1\frac{1}{2}} - nd\ 6_{1\frac{1}{2}}$
12136.08	5w(180)	8237.65	$nd\ 5_{1\frac{1}{2}} - np\ 2_{1\frac{1}{2}}$
12097.44	2	8263.96	$nd\ 5_{2\frac{1}{2}} - 6p'\ 2P_{1\frac{1}{2}}$
12033.94	60	8307.57	$6p\ 4D_{3\frac{1}{2}} - nd\ 10_{3\frac{1}{2}}$
12023.74	8	8314.61	$6s\ 2P_{0\frac{1}{2}} - 6p\ 2P_{0\frac{1}{2}}$
11996.92	75(510)	8333.20	$6p\ 2D_{2\frac{1}{2}} - nd\ 10_{3\frac{1}{2}}$
11866.00	2	8425.14	$nd\ 5_{1\frac{1}{2}} - 7p\ 2P_{0\frac{1}{2}}$
11779.17	30	8487.24	$6p\ 4S_{1\frac{1}{2}} - nd\ 7_{1\frac{1}{2}}$
11778.01	45	8488.08	
11761.74	2	8499.82	$nd\ 5_{1\frac{1}{2}} - 8p\ 2D_{2\frac{1}{2}}$

In conclusion we acknowledge our indebtedness to several of our colleagues for data used in this investigation. Both C. J. Humphreys and E. K. Plyler made observations of the infrared spectrum of iodine beyond the reach of photography. Their data are presented in table 3. W. F. Meggers and R. Zalubas measured the magnetic patterns of numerous iodine lines during their investigations of spectra emitted by various electrodeless metal-halide lamps. Finally, W. C. Martin, Jr., made new measurements of iodine spectra in the extreme ultraviolet that have surpassed in extent and accuracy earlier descriptions of these spectra. It is a pleasure for us to thank each for his contribution to this paper.

## 8. References

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Wavelength	Intensity	Wave number	Designation
11650.40	1w	8581.05	$nd\ 5_{2\frac{1}{2}} - 8p\ 4P_{2\frac{1}{2}}$
11610.60	5	8610.47	$6s'\ 2D_{1\frac{1}{2}} - 7p\ 4P_{2\frac{1}{2}}$
11588.23	40(365)	8627.09	$6s\ 2P_{0\frac{1}{2}} - 6p\ 2S_{0\frac{1}{2}}$
11558.56	100(660)	8649.23	$6p\ 4S_{1\frac{1}{2}} - nd\ 8_{1\frac{1}{2}}$
11538.57	2	8664.22	$nd\ 5_{2\frac{1}{2}} - 8p\ 4S_{1\frac{1}{2}}$
11498.65	15	8694.30	$6s'\ 2D_{1\frac{1}{2}} - 7p\ 4S_{1\frac{1}{2}}$
11486.80	8	8703.27	$\{ 6p\ 4D_{1\frac{1}{2}} - 7d\ 4P_{2\frac{1}{2}}$
11465.76	25w	8719.25	$nd\ 5_{2\frac{1}{2}} - 8p\ 2D_{2\frac{1}{2}}$
11458.07	6	8725.09	$nd\ 3_{2\frac{1}{2}} - 4f\ 1_{2\frac{1}{2}}$
11457.08	8	8725.84	$nd\ 3_{2\frac{1}{2}} - 4f\ 2_{1\frac{1}{2}}$

TABLE 1. Wavelengths and term combinations of Ii—Continued

Wavelength	Intensity	Wave number	Designation	Wavelength	Intensity	Wave number	Designation
11451. 13	50	8730. 38	$6p\ ^4P_{0\frac{1}{2}}-nd\ 11_{1\frac{1}{2}}$	10788. 18	10	9266. 87	$6s'\ ^2D_{1\frac{1}{2}}-7p\ ^4P_{0\frac{1}{2}}$
11447. 72	100	8732. 98	$\left\{ \begin{array}{l} 6p\ ^4P_{0\frac{1}{2}}-nd\ 8_{1\frac{1}{2}} \\ nd\ 3_{2\frac{1}{2}}-4f\ 3_{0\frac{1}{2}} \end{array} \right.$	10777. 97	4	9275. 65	$6p\ ^2D_{1\frac{1}{2}}-11s\ ^4P_{2\frac{1}{2}}$
11429. 56	15	8746. 86		10771. 79	20	9280. 98	$6p\ ^4D_{1\frac{1}{2}}-10s\ ^4P_{2\frac{1}{2}}$
11428. 40	50	8747. 74	$nd\ 4_{1\frac{1}{2}}-4f\ 5_{2\frac{1}{2}}$	10722. 07	2	9324. 01	$6p\ ^2P_{0\frac{1}{2}}-nd\ 27_{1\frac{1}{2}}$
11420. 33	75	8753. 92	$nd\ 2_{1\frac{1}{2}}-4f\ 0.1_{0\frac{1}{2}}$	10706. 79	15	9337. 31	$7s\ ^4P_{2\frac{1}{2}}-6f\ 6_{1\frac{1}{2}}$
11415. 66	3	8757. 50	$nd\ 2_{1\frac{1}{2}}-4f\ 1_{2\frac{1}{2}}$	10696. 02	100	9346. 72	$nd\ 5.1_{1\frac{1}{2}}-8p\ ^2P_{0\frac{1}{2}}$
11410. 07	75	8761. 79	$6p\ ^4P_{1\frac{1}{2}}-nd\ 16_{1\frac{1}{2}}$	10685. 82	100Z	9355. 64	$6p\ ^4P_{1\frac{1}{2}}-nd\ 19.1_{0\frac{1}{2}}$
11401. 46	2	8768. 42	$nd\ 3_{2\frac{1}{2}}-4f\ 4_{1\frac{1}{2}}$	10603. 64	4	9428. 15	$6p\ ^2D_{1\frac{1}{2}}-nd\ 35_{0\frac{1}{2}}$
11397. 98	10	8771. 09		10588. 59	6	9441. 55	$7s\ ^2P_{1\frac{1}{2}}-7p\ ^4D_{1\frac{1}{2}}$
11397. 28	8	8771. 63	$nd\ 3_{2\frac{1}{2}}-4f\ 5_{2\frac{1}{2}}$	10578. 22	20	9450. 80	$6p\ ^2P_{1\frac{1}{2}}-nd\ 38_{0\frac{1}{2}}$
11396. 50	10c	8772. 23	$6p\ ^4D_{0\frac{1}{2}}-nd\ 34_{0\frac{1}{2}}$	10545. 62	15	9480. 02	$nd\ 5.1_{1\frac{1}{2}}-5f\ 1_{1\frac{1}{2}}$
11375. 25	75	8788. 61	$nd\ 4.1_{1\frac{1}{2}}-4f\ 5.1_{1\frac{1}{2}}$	10539. 72	50	9485. 33	$nd\ 5.1_{1\frac{1}{2}}-5f\ 2_{0\frac{1}{2}}$
11373. 78	1	8789. 75	$6s\ ^2P_{0\frac{1}{2}}-6p\ ^4D_{1\frac{1}{2}}$	10534. 95	10	9489. 62	$nd\ 5.1_{1\frac{1}{2}}-5f\ 3_{2\frac{1}{2}}$
11372. 10	125	8791. 05	$nd\ 4.1_{1\frac{1}{2}}-4f\ 6_{1\frac{1}{2}}$	10515. 40	100	9507. 27	$6p\ ^4D_{3\frac{1}{2}}-nd\ 14_{2\frac{1}{2}}$
11366. 90	1	8795. 07	$nd\ 4_{1\frac{1}{2}}-4f\ 6_{1\frac{1}{2}}$	10494. 08	$30w,+g?$	9526. 58	$nd\ 5.1_{1\frac{1}{2}}-5f\ 5_{1\frac{1}{2}}$
11356. 37	125	8803. 23		10487. 23	10	9532. 80	$6p\ ^2D_{2\frac{1}{2}}-nd\ 14_{2\frac{1}{2}}$
11353. 67	75	8805. 32	$\left\{ \begin{array}{l} 6p\ ^4S_{1\frac{1}{2}}-nd\ 9_{2\frac{1}{2}} \\ 6p\ ^4D_{1\frac{1}{2}}-nd\ 27_{1\frac{1}{2}} \end{array} \right.$	10469. 23	3	9549. 19	
11351. 81	18	8806. 76	$nd\ 2_{1\frac{1}{2}}-4f\ 4_{1\frac{1}{2}}$	10466. 54	5000Z	9551. 65	$6s\ ^2P_{1\frac{1}{2}}-6p\ ^2D_{2\frac{1}{2}}$
11347. 85	100d	8809. 84	$nd\ 2_{1\frac{1}{2}}-4f\ 5_{2\frac{1}{2}}$	10459. 55	1	9558. 03	
11343. 23	75	8813. 42	$6s\ ^2P_{1\frac{1}{2}}-6p\ ^4P_{2\frac{1}{2}}$	10455. 45	2	9561. 77	
11313. 27	100	8836. 76	$6p\ ^4P_{1\frac{1}{2}}-nd\ 17_{2\frac{1}{2}}$	10445. 35	5	9571. 03	$6p\ ^2P_{0\frac{1}{2}}-nd\ 28.2_{1\frac{1}{2}}$
11298. 66	4	8848. 19	$\left\{ \begin{array}{l} nd\ 5.1_{1\frac{1}{2}}-8p\ ^4P_{0\frac{1}{2}} \\ 6p\ ^4D_{0\frac{1}{2}}-nd\ 35_{0\frac{1}{2}} \end{array} \right.$	10438. 81	$3,+g?$	9577. 02	
11293. 40	65d?	8852. 32	$nd\ 5_{2\frac{1}{2}}-8p\ ^4D_{3\frac{1}{2}}$	10435. 34	100d?Z	9580. 10	
11290. 54	25w	8854. 56	$nd\ 2_{1\frac{1}{2}}-4f\ 5.1_{1\frac{1}{2}}$	10428. 39	6	9586. 60	$6p\ ^2P_{0\frac{1}{2}}-nd\ 39_{2\frac{1}{2}}$
11287. 49	75d	8856. 95	$nd\ 2_{1\frac{1}{2}}-4f\ 6_{1\frac{1}{2}}$	10416. 61	75Z	9597. 44	$6p\ ^4S_{1\frac{1}{2}}-nd\ 11_{1\frac{1}{2}}$
11271. 03	1	8869. 88	$nd\ 5.1_{1\frac{1}{2}}-8p\ ^4P_{1\frac{1}{2}}$	10412. 80	10	9600. 95	$6s'\ ^2D_{2\frac{1}{2}}-7p\ ^4P_{0\frac{1}{2}}$
11246. 77	125	8889. 02	$\left\{ \begin{array}{l} 6p\ ^4P_{2\frac{1}{2}}-nd\ 9_{2\frac{1}{2}} \\ 6p\ ^2D_{1\frac{1}{2}}-8d\ ^4P_{2\frac{1}{2}} \end{array} \right.$	10391. 74	400Z	9620. 40	$6s\ ^2P_{0\frac{1}{2}}-6p\ ^2D_{1\frac{1}{2}}$
11236. 56	400Z	8897. 09	$6s\ ^2P_{1\frac{1}{2}}-6p\ ^4S_{1\frac{1}{2}}$	10375. 20	400Z	9635. 74	
11187. 21	5	8936. 34		10354. 93	8	9654. 60	
11179. 11	10	8942. 82	$6p\ ^2D_{2\frac{1}{2}}-nd\ 11_{1\frac{1}{2}}$	10348. 02	7	9661. 04	$6p\ ^4D_{0\frac{1}{2}}-6s''\ ^2S_{0\frac{1}{2}}$
11176. 21	50	8945. 14	$6s'\ ^2D_{2\frac{1}{2}}-7p\ ^4P_{2\frac{1}{2}}$	10326. 53	3	9665. 54	$6p\ ^4D_{1\frac{1}{2}}-nd\ 31_{1\frac{1}{2}}$
11172. 79	8	8947. 87	$6s'\ ^2D_{1\frac{1}{2}}-7p\ ^4P_{0\frac{1}{2}}$	10325. 90	75	9681. 15	$6p\ ^4P_{0\frac{1}{2}}-nd\ 11_{1\frac{1}{2}}$
11169. 39	75	8950. 60		10322. 56	100	9684. 87	$6s\ ^2D_{1\frac{1}{2}}-nd\ 37_{2\frac{1}{2}}$
11147. 15	10	8968. 46	$6p\ ^2S_{0\frac{1}{2}}-nd\ 27_{1\frac{1}{2}}$	10318. 20	35	9688. 96	$nd\ 5.1_{1\frac{1}{2}}-np\ 3_{1\frac{1}{2}}$
11140. 20	6	8974. 05	$6p\ ^2P_{1\frac{1}{2}}-nd\ 32.2_{0\frac{1}{2}}$	10313. 72	4	9693. 17	$6p\ ^4P_{1\frac{1}{2}}-nd\ 20_{0\frac{1}{2}}$
11138. 10	2	8975. 74		10310. 20	50	9696. 48	$6s\ ^2P_{0\frac{1}{2}}-nd\ 1_{1\frac{1}{2}}$
11116. 62	7	8993. 09		10286. 07	8	9719. 23	$6p\ ^4D_{1\frac{1}{2}}-8d\ ^4P_{2\frac{1}{2}}$
11093. 70	4	9011. 66		10274. 34	$4w$	9730. 32	$nd\ 5_{2\frac{1}{2}}-5f\ 5_{1\frac{1}{2}}$
11084. 16	2	9019. 42	$6p\ ^2P_{1\frac{1}{2}}-11s\ ^4P_{2\frac{1}{2}}$	10266. 04	5	9738. 19	
11072. 33	60	9029. 06	$6s'\ ^2D_{2\frac{1}{2}}-7p\ ^4S_{1\frac{1}{2}}$	10242. 83	10	9760. 26	$6p\ ^4D_{2\frac{1}{2}}-9d\ ^4D_{3\frac{1}{2}}$
11059. 56	4	9039. 48	$\left\{ \begin{array}{l} 6p\ ^4D_{0\frac{1}{2}}-nd\ 36.1_{0\frac{1}{2}} \\ 6p\ ^2P_{1\frac{1}{2}}-nd\ 33_{0\frac{1}{2}} \end{array} \right.$	10241. 29	20	9761. 73	
11053. 79	15	9044. 20	$6p\ ^4P_{1\frac{1}{2}}-nd\ 18_{2\frac{1}{2}}$	10238. 82	1000Z	9764. 08	$6s\ ^2P_{1\frac{1}{2}}-6p\ ^4P_{0\frac{1}{2}}$
11050. 04	2	9047. 27		10232. 06	35	9770. 53	$nd\ 5.1_{1\frac{1}{2}}-np\ 4_{1\frac{1}{2}}$
11020. 60	250Z	9071. 44	$6p\ ^2P_{0\frac{1}{2}}-nd\ 10_{3\frac{1}{2}}$	10211. 60	5	9790. 11	$7s\ ^4P_{2\frac{1}{2}}-np\ 5_{3\frac{1}{2}}$
11017. 14	100Z	9074. 29	$6p\ ^4P_{1\frac{1}{2}}-nd\ 19_{1\frac{1}{2}}$	10201. 82	7	9799. 49	$nd\ 4.1_{1\frac{1}{2}}-6p'\ ^2P_{1\frac{1}{2}}$
10991. 19	4	9095. 71	$6p\ ^4D_{1\frac{1}{2}}-nd\ 28.2_{1\frac{1}{2}}$	10172. 91	300Z	9827. 34	$nd\ 3_{2\frac{1}{2}}-6p'\ ^2P_{1\frac{1}{2}}$
10979. 70	6	9105. 23	$6p\ ^2P_{1\frac{1}{2}}-nd\ 34_{0\frac{1}{2}}$	10166. 00	8	9834. 02	$6p\ ^2D_{1\frac{1}{2}}-nd\ 39_{2\frac{1}{2}}$
10970. 39	4	9112. 95	$6p\ ^4D_{2\frac{1}{2}}-nd\ 31_{1\frac{1}{2}}$				$6p\ ^4S_{1\frac{1}{2}}-nd\ 13_{2\frac{1}{2}}$
10914. 32	15	9159. 77	$6p\ ^4D_{2\frac{1}{2}}-8d\ ^4D_{3\frac{1}{2}}$	10158. 64	400Z	9841. 15	$6p\ ^4P_{1\frac{1}{2}}-nd\ 21_{2\frac{1}{2}}$
10897. 87	65	9173. 60	$6s'\ ^2D_{2\frac{1}{2}}-7p\ ^4D_{3\frac{1}{2}}$	10147. 70	1	9851. 77	$6p\ ^4D_{3\frac{1}{2}}-5d\ ^4D_{3\frac{1}{2}}$
10894. 66	70	9176. 30		10141. 83	100Z	9857. 47	$6p\ ^4P_{0\frac{1}{2}}-nd\ 15_{0\frac{1}{2}}$
10891. 47	75d?	9178. 99	$\left\{ \begin{array}{l} 6p\ ^2P_{0\frac{1}{2}}-7d\ ^4P_{2\frac{1}{2}} \\ 6p\ ^2D_{2\frac{1}{2}}-nd\ 13_{2\frac{1}{2}} \end{array} \right.$	10133. 56	40	9865. 51	$nd\ 2_{1\frac{1}{2}}-6p'\ ^2P_{1\frac{1}{2}}$
10889. 23	18	9180. 88	$6p\ ^2P_{1\frac{1}{2}}-nd\ 35_{0\frac{1}{2}}$	10132. 38	3	9866. 66	$6p\ ^2D_{2\frac{1}{2}}-5d\ ^4D_{3\frac{1}{2}}$
10856. 80	2	9208. 31	$6p\ ^4D_{2\frac{1}{2}}-nd\ 32_{1\frac{1}{2}}$	10131. 16	750Z	9867. 85	$6s\ ^2P_{0\frac{1}{2}}-6p\ ^2P_{1\frac{1}{2}}$
10841. 34	4	9221. 43	$6p\ ^2D_{1\frac{1}{2}}-nd\ 32.2_{0\frac{1}{2}}$	10126. 07	7	9872. 81	$6p\ ^2P_{0\frac{1}{2}}-nd\ 29_{1\frac{1}{2}}$

TABLE 1. Wavelengths and term combinations of I<sub>I</sub>—Continued

Wavelength	Intensity	Wave number	Designation	Wavelength	Intensity	Wave number	Designation
10109. 70	5	9888. 79	<i>nd</i> 5 <sub>2½</sub> — <i>np</i> 3 <sub>1½</sub>	9390. 14	1	10646. 55	<i>nd</i> 5 <sub>2½</sub> —9 <i>p</i> 2D <sub>2½</sub>
10074. 13	7	9923. 71	<i>nd</i> 4 <sub>1½</sub> — <i>np</i> 1.1 <sub>2½</sub>	9379. 77	2	10658. 32	6 <i>p</i> 2P <sub>0½</sub> — <i>nd</i> 34 <sub>0½</sub>
10066. 72	7	9931. 01	6 <i>p</i> 2S <sub>0½</sub> — <i>nd</i> 32.1 <sub>0½</sub>	9374. 34	3d	10664. 49	6 <i>p</i> 4D <sub>1½</sub> — <i>nd</i> 39 <sub>2½</sub>
10050. 11	2w	9947. 42	<i>nd</i> 3 <sub>2½</sub> — <i>np</i> 1.1 <sub>2½</sub>	9365. 16	25c	10674. 95	<i>nd</i> 2 <sub>1½</sub> —8 <i>p</i> 4P <sub>1½</sub>
10034. 64	2	9962. 76	6 <i>p</i> 4D <sub>3½</sub> — <i>nd</i> 37 <sub>2½</sub>	9358. 69	1	10682. 33	6 <i>p</i> 2S <sub>0½</sub> — <i>nd</i> 38 <sub>0½</sub>
10030. 35	2	9967. 03	6 <i>p</i> 4P <sub>0½</sub> — <i>nd</i> 16 <sub>1½</sub>	9335. 05	1000Z	10709. 38	6 <i>s</i> 4P <sub>1½</sub> —6 <i>p</i> 4D <sub>2½</sub>
10023. 10	22	9974. 24	<i>nd</i> 5 <sub>2½</sub> — <i>np</i> 4 <sub>2½</sub>	9321. 95	200Z	10724. 43	6 <i>p</i> 4S <sub>1½</sub> — <i>nd</i> 15 <sub>0½</sub>
10011. 68	20	9985. 61	<i>nd</i> 2 <sub>1½</sub> — <i>np</i> 1.1 <sub>2½</sub>	9313. 64	5	10734. 02	6 <i>p</i> 2P <sub>0½</sub> — <i>nd</i> 35 <sub>0½</sub>
10003. 05	500Z	9994. 22	{ 6 <i>s</i> 4P <sub>1½</sub> —6 <i>p</i> 2S <sub>0½</sub>	9227. 74	600Z	10833. 92	6 <i>p</i> 4S <sub>1½</sub> — <i>nd</i> 16 <sub>1½</sub>
9992. 54	85	10004. 72	6 <i>p</i> 2P <sub>1½</sub> —6 <i>s'</i> 2S <sub>0½</sub>	9213. 57	3	10850. 58	<i>nd</i> 5.1 <sub>1½</sub> —6 <i>f</i> 1 <sub>2½</sub>
9992. 24	4	10005. 02	<i>nd</i> 1 <sub>1½</sub> —4 <i>f</i> 0.1 <sub>0½</sub>	9201. 90	12	10864. 33	<i>nd</i> 5.1 <sub>1½</sub> —6 <i>f</i> 4 <sub>2½</sub>
9963. 30	400Z	10034. 09	6 <i>p</i> 4D <sub>3½</sub> —5 <i>d</i> 4F <sub>4½</sub>	9195. 07	3d	10872. 40	<i>nd</i> 5.1 <sub>1½</sub> —6 <i>f</i> 6 <sub>1½</sub>
9954. 63	200d	10042. 82	<i>nd</i> 2 <sub>1½</sub> — <i>np</i> 2 <sub>1½</sub>	9180. 20	70Z	10890. 02	6 <i>p</i> 4P <sub>0½</sub> — <i>nd</i> 20 <sub>0½</sub>
9939. 30	100	10058. 31	<i>nd</i> 1 <sub>1½</sub> —4 <i>f</i> 4 <sub>1½</sub>	9164. 38	20	10908. 82	6 <i>p</i> 4S <sub>1½</sub> — <i>nd</i> 17 <sub>2½</sub>
9933. 25	8	10064. 44	6 <i>p</i> 2P <sub>0½</sub> — <i>nd</i> 30 <sub>0½</sub>	9156. 91	500Z	10917. 72	{ 6 <i>s</i> 4P <sub>0½</sub> —6 <i>p</i> 2S <sub>0½</sub>
9892. 35	50	10106. 05	<i>nd</i> 1 <sub>1½</sub> —4 <i>f</i> 5.1 <sub>1½</sub>	9150. 63	3	10925. 21	6 <i>p</i> 2P <sub>0½</sub> — <i>nd</i> 36.1 <sub>0½</sub>
9889. 95	40c	10108. 46	<i>nd</i> 1 <sub>1½</sub> —4 <i>f</i> 6 <sub>1½</sub>	9128. 03	600Z	10952. 26	6 <i>s'</i> 2D <sub>1½</sub> —4 <i>f</i> 1 <sub>2½</sub>
9855. 04	1	10144. 31	<i>nd</i> 3 <sub>2½</sub> —8 <i>p</i> 4P <sub>2½</sub>	9113. 91	12000Z	10969. 23	6 <i>s</i> 2P <sub>1½</sub> —6 <i>p</i> 4P <sub>1½</sub>
9842. 75	150Z	10156. 98	6 <i>s</i> 4P <sub>1½</sub> —6 <i>p</i> 4D <sub>1½</sub>	9098. 86	1000Z	10987. 37	6 <i>s</i> 4P <sub>1½</sub> —6 <i>p</i> 2D <sub>1½</sub>
9835. 52	5	10164. 45	6 <i>p</i> 2P <sub>0½</sub> — <i>nd</i> 30 <sub>0½</sub>	9094. 50	150	10992. 64	6 <i>p</i> 4P <sub>2½</sub> — <i>nd</i> 17 <sub>2½</sub>
9832. 28	4	10167. 80	<i>nd</i> 4 <sub>1½</sub> —7 <i>p</i> 2P <sub>0½</sub>	9087. 16	300Z	11001. 52	6 <i>s'</i> 2D <sub>1½</sub> —4 <i>f</i> 4 <sub>1½</sub>
9827. 53	1	10172. 71	<i>np</i> 1 <sub>1½</sub> —6 <i>s'</i> 2S <sub>0½</sub>	9084. 70	6	11004. 50	6 <i>s'</i> 2D <sub>1½</sub> —4 <i>f</i> 5 <sub>1½</sub>
9818. 10	2	10182. 48	<i>nd</i> 2 <sub>1½</sub> —8 <i>p</i> 4P <sub>2½</sub>	9082. 55	2	11007. 10	6 <i>p</i> 2P <sub>0½</sub> — <i>nd</i> 38.2 <sub>1½</sub>
9817. 59	1	10183. 01	6 <i>p</i> 4D <sub>3½</sub> — <i>nd</i> 34 <sub>0½</sub>	9079. 34	50Z	11010. 99	6 <i>s</i> 4P <sub>2½</sub> —6 <i>p</i> 2D <sub>2½</sub>
9813. 53	200Z	10187. 22	6 <i>p</i> 4S <sub>1½</sub> — <i>nd</i> 14 <sub>2½</sub>	9058. 33	15000Z	11036. 53	6 <i>s</i> 4P <sub>2½</sub> —6 <i>p</i> 4D <sub>3½</sub>
9808. 26	1	10192. 70	6 <i>p</i> 2P <sub>0½</sub> — <i>nd</i> 31.1 <sub>1½</sub>	9047. 91	20c	11049. 24	6 <i>s'</i> 2D <sub>1½</sub> —4 <i>f</i> 5.1 <sub>1½</sub>
9800. 89	100Z	10200. 36	6 <i>s</i> 2P <sub>0½</sub> —6 <i>p</i> 4D <sub>0½</sub>	9046. 90	3	11050. 47	
9787. 21	2	10214. 62	6 <i>p</i> 2S <sub>0½</sub> — <i>nd</i> 32.2 <sub>0½</sub>	9046. 36	5	11051. 13	
9774. 97	10c	10227. 41	<i>nd</i> 3 <sub>2½</sub> —8 <i>p</i> 4S <sub>1½</sub>	9043. 85	1	11054. 20	{ <i>nd</i> 5 <sub>2½</sub> —6 <i>f</i> 1 <sub>2½</sub>
9773. 50	6	10228. 94	6 <i>p</i> 4D <sub>3½</sub> — <i>nd</i> 17 <sub>2½</sub>	9042. 30	15	11056. 10	{ 6 <i>s</i> 4P <sub>1½</sub> — <i>nd</i> 1 <sub>1½</sub>
9772. 17	7c	10230. 34	<i>nd</i> 2 <sub>1½</sub> —7 <i>p</i> 2P <sub>0½</sub>	9034. 43	2d?	11065. 73	<i>nd</i> 5 <sub>2½</sub> —6 <i>f</i> 3 <sub>1½</sub>
9749. 20	40cZ	10254. 44	6 <i>p</i> 2D <sub>2½</sub> — <i>nd</i> 17 <sub>2½</sub>	9032. 64	3	11067. 92	<i>nd</i> 5 <sub>2½</sub> —6 <i>f</i> 4 <sub>2½</sub>
9747. 23	3c	10256. 51	9022. 40	5000Z	11080. 48	6 <i>s</i> 4P <sub>0½</sub> —6 <i>p</i> 2D <sub>1½</sub>	
9733. 56	300	10270. 92	6 <i>p</i> 4P <sub>3½</sub> — <i>nd</i> 14 <sub>2½</sub>	9018. 05	10c	11085. 83	<i>nd</i> 4.1 <sub>1½</sub> —8 <i>p</i> 2P <sub>0½</sub>
9731. 73	5000Z	10272. 85	6 <i>s</i> 4P <sub>2½</sub> —6 <i>p</i> 4P <sub>2½</sub>	9007. 77	40	11098. 48	6 <i>p</i> 4P <sub>3½</sub> —5 <i>d</i> 2F <sub>3½</sub>
9725. 47	30Z	10279. 46	6 <i>p</i> 4P <sub>0½</sub> — <i>nd</i> 19 <sub>1½</sub>	9004. 39	6c	11102. 65	6 <i>p</i> 4S <sub>1½</sub> — <i>nd</i> 18 <sub>2½</sub>
9710. 58	8	10295. 22	8993. 13	400Z	11116. 55	6 <i>p</i> 4S <sub>1½</sub> — <i>nd</i> 19 <sub>1½</sub>	
9701. 28	5	10305. 09	8969. 04	300Z	11146. 41	6 <i>p</i> 4S <sub>1½</sub> — <i>nd</i> 19 <sub>1½</sub>	
9673. 43	1	10334. 76	<i>nd</i> 2 <sub>1½</sub> —8 <i>p</i> 2D <sub>2½</sub>	8964. 69	400Z	11151. 81	{ <i>nd</i> 5.1 <sub>1½</sub> —7 <i>p</i> 2S <sub>0½</sub>
9663. 06	1	10345. 86	6 <i>p</i> 2S <sub>0½</sub> — <i>nd</i> 34 <sub>0½</sub>	8925. 97	225cZ	11200. 20	{ <i>nd</i> 2 <sub>1½</sub> —8 <i>p</i> 2P <sub>0½</sub>
9653. 06	3000dZ	10356. 57	6 <i>s</i> 4P <sub>2½</sub> —6 <i>p</i> 4S <sub>1½</sub>	8902. 23	60c	11230. 06	6 <i>p</i> 4P <sub>2½</sub> — <i>nd</i> 19 <sub>1½</sub>
9649. 61	2000Z	10360. 27	6 <i>p</i> 2D <sub>2½</sub> —5 <i>d</i> 2F <sub>3½</sub>	8899. 72	3	11233. 23	6 <i>p</i> 4D <sub>3½</sub> — <i>nd</i> 21 <sub>2½</sub>
9623. 21	15	10388. 69	8898. 50	10000dZ	11234. 77	{ 6 <i>p</i> 2S <sub>0½</sub> —6 <i>s'</i> 2S <sub>0½</sub>	
9598. 22	2000Z	10415. 74	<i>nd</i> 3 <sub>2½</sub> —8 <i>p</i> 4D <sub>3½</sub>	8884. 70	30c	11252. 21	6 <i>s</i> 4P <sub>1½</sub> —6 <i>p</i> 2P <sub>1½</sub>
9593. 62	2	10420. 74	8879. 55	90d	11258. 74	6 <i>p</i> 2D <sub>2½</sub> — <i>nd</i> 21 <sub>2½</sub>	
9579. 02	20c	10436. 62	8878. 76	75c	11259. 74	<i>nd</i> 3 <sub>2½</sub> —5 <i>f</i> 4 <sub>3½</sub>	
9573. 22	1	10442. 94	8874. 18	25	11265. 56	<i>nd</i> 3 <sub>2½</sub> —5 <i>f</i> 4.1 <sub>3½</sub>	
9566. 81	1	10449. 94	8873. 86	30d	11265. 97	6 <i>p</i> 4D <sub>3½</sub> — <i>nd</i> 21.1 <sub>3½</sub>	
9555. 72	10	10462. 07	8870. 66	6	11270. 03	<i>nd</i> 4 <sub>1½</sub> —5 <i>f</i> 5 <sub>1½</sub>	
9528. 48	18	10491. 98	8868. 00	75	11273. 41		
9526. 90	30	10493. 72	8865. 53	1	11276. 54	6 <i>s'</i> 2D <sub>2½</sub> —4 <i>f</i> 0 <sub>1½</sub>	
9516. 92	1h	10504. 72	8864. 95	1c	11277. 28	<i>nd</i> 3 <sub>2½</sub> —5 <i>f</i> 4.2 <sub>3½</sub>	
9514. 07	1	10507. 86	8862. 33	1	11280. 61		
9496. 69	1	10527. 09	8858. 79	3c	11285. 12	<i>nd</i> 2 <sub>1½</sub> —5 <i>f</i> 1 <sub>1½</sub>	
9466. 34	50cZ	10560. 85	8857. 50	3000Z	11286. 77	6 <i>s'</i> 2D <sub>2½</sub> —4 <i>f</i> 1 <sub>1½</sub>	
9438. 25	1h	10592. 28	8854. 68	1c	11290. 36	<i>nd</i> 2 <sub>1½</sub> —5 <i>f</i> 2 <sub>0½</sub>	
9427. 15	3000Z	10604. 75	8853. 80	2000Z	11291. 49	6 <i>p</i> 2D <sub>2½</sub> — <i>nd</i> 21.1 <sub>3½</sub>	
9426. 71	4000Z	10605. 25	8853. 24	1000Z	11292. 20	6 <i>s'</i> 2D <sub>2½</sub> —4 <i>f</i> 2 <sub>3½</sub>	
9423. 42	1	10608. 94	8852. 65	50d	11292. 95	6 <i>s'</i> 2D <sub>2½</sub> —4 <i>f</i> 2.1 <sub>3½</sub>	
9398. 73	200c	10636. 82	8851. 30	100c	11294. 68	<i>nd</i> 2 <sub>1½</sub> —5 <i>f</i> 3 <sub>2½</sub>	

TABLE 1. Wavelengths and term combinations of I<sub>I</sub>—Continued

Wavelength	Intensity	Wave number	Designation	Wavelength	Intensity	Wave number	Designation
8848.33	80	11298.47		8023.01	300Z	12460.73	$6s' \ ^2D_{1\frac{1}{2}} - 8p \ ^4S_{1\frac{1}{2}}$
8847.14	250Z	11299.99	$6s' \ ^2D_{2\frac{1}{2}} - 4f \ 3_{3\frac{1}{2}}$			$6p \ ^4S_{1\frac{1}{2}} - 8s \ ^4P_{2\frac{1}{2}}$	
8822.08	50c	11332.09	$nd \ 2_{1\frac{1}{2}} - 5f \ 5_{1\frac{1}{2}}$	8009.61	2	12481.57	
8818.89	150c	11336.18	$6s' \ ^2D_{2\frac{1}{2}} - 4f \ 4_{1\frac{1}{2}}$	8003.63	1000Z	12490.90	$6s \ ^4P_{0\frac{1}{2}} - 6p \ ^4D_{0\frac{1}{2}}$
8816.65	275cZ	11339.07	$6s' \ ^2D_{2\frac{1}{2}} - 4f \ 5_{2\frac{1}{2}}$	7998.01	40	12499.68	$6s' \ ^2D_{1\frac{1}{2}} - 8p \ ^2D_{2\frac{1}{2}}$
				7995.63	3	12503.40	
8812.40	100d	11345.82					
8809.86	10	11347.80		7974.48	200Z	12536.56	$nd \ 1_{1\frac{1}{2}} - 5f \ 1_{1\frac{1}{2}}$
8805.41	15d	11353.54	$6p \ ^4P_{1\frac{1}{2}} - 7s \ ^4P_{0\frac{1}{2}}$	7969.48	500Z	12544.42	$6p \ ^4P_{2\frac{1}{2}} - 8s \ ^4P_{2\frac{1}{2}}$
8780.10	100c	11386.27	$6s' \ ^2D_{2\frac{1}{2}} - 4f \ 6_{1\frac{1}{2}}$	7968.61	5	12545.80	$nd \ 1_{1\frac{1}{2}} - 5f \ 3_{2\frac{1}{2}}$
8748.22	250dZ	11427.76	$6p \ ^4S_{1\frac{1}{2}} - nd \ 19_{1\frac{1}{2}}$	7960.43	50d	12558.68	$6p \ ^4P_{0\frac{1}{2}} - 7s \ ^4P_{0\frac{1}{2}}$
			$nd \ 4_{1\frac{1}{2}} - np \ 3_{1\frac{1}{2}}$	7955.90	40dZ	12565.84	$6p \ ^4S_{1\frac{1}{2}} - 8s \ ^2P_{1\frac{1}{2}}$
8729.70	200c	11452.01	$nd \ 3_{2\frac{1}{2}} - np \ 3_{1\frac{1}{2}}$	7951.99	200c	12572.01	$6s' \ ^2D_{2\frac{1}{2}} - np \ 2_{1\frac{1}{2}}$
8700.80	500cZ	11490.04	$nd \ 2_{1\frac{1}{2}} - np \ 3_{1\frac{1}{2}}$	7944.85	250cZ	12583.32	$nd \ 1_{1\frac{1}{2}} - 5f \ 5_{1\frac{1}{2}}$
8680.36	10d	11517.10	$nd \ 1_{1\frac{1}{2}} - 8p \ ^4S_{1\frac{1}{2}}$	7941.34	1	12588.87	
8664.95	1500cZ	11537.60	$nd \ 3_{2\frac{1}{2}} - np \ 4_{2\frac{1}{2}}$	7927.94	15c	12610.15	
8642.60	200d	11567.42	$6s \ ^4P_{1\frac{1}{2}} - 6p \ ^4D_{0\frac{1}{2}}$	7927.10	50c	12611.49	$nd \ 4_{1\frac{1}{2}} - 6f \ 6_{1\frac{1}{2}}$
8636.40	175dZ	11575.72	$nd \ 2_{1\frac{1}{2}} - np \ 4_{2\frac{1}{2}}$	7922.23	20c	12619.24	$nd \ 3_{2\frac{1}{2}} - 6f \ 2_{2\frac{1}{2}}$
8560.30	30d	11678.63	$nd \ 5_{1\frac{1}{2}} - 7p \ ^4D_{2\frac{1}{2}}$	7914.70	15c	12631.25	$nd \ 3_{2\frac{1}{2}} - 6f \ 4_{2\frac{1}{2}}$
8551.60	3cw	11690.51		7909.08	3	12640.22	
8545.52	300dZ	11698.82	$6p \ ^4P_{0\frac{1}{2}} - 8s \ ^2P_{1\frac{1}{2}}$	7907.86	25	12642.17	
8503.32	2cw	11756.89	$6p \ ^4S_{1\frac{1}{2}} - nd \ 20_{0\frac{1}{2}}$	7904.45	10d	12647.63	
8490.67	1cw	11774.40		7903.27	3	12649.51	$6p \ ^4P_{2\frac{1}{2}} - 8s \ ^2P_{1\frac{1}{2}}$
8486.11	1000Z	11780.73	$6p \ ^4D_{3\frac{1}{2}} - 8s \ ^4P_{2\frac{1}{2}}$	7899.41	3c	12655.70	$nd \ 2_{1\frac{1}{2}} - 6f \ 1_{2\frac{1}{2}}$
8467.80	5	11806.21	$6p \ ^2D_{2\frac{1}{2}} - 8s \ ^4P_{2\frac{1}{2}}$	7897.98	600	12657.99	
8451.46	60d	11829.03	$6p \ ^4P_{1\frac{1}{2}} - 7s \ ^4P_{0\frac{1}{2}}$	7892.43	15c	12666.89	$nd \ 2_{1\frac{1}{2}} - 6f \ 3_{1\frac{1}{2}}$
8443.19	8c	11840.61		7890.85	25c	12669.42	$nd \ 2_{1\frac{1}{2}} - 6f \ 4_{2\frac{1}{2}}$
8427.41	20c	11862.80	$6s \ ^2P_{0\frac{1}{2}} - 7p \ ^4S_{1\frac{1}{2}}$	7886.00	5c	12677.22	
8418.95	4	11874.71	$nd \ 5_{2\frac{1}{2}} - 7f \ 1_{2\frac{1}{2}}$	7885.72	5c	12677.66	
8416.54	50	11878.11		7864.56	1	12711.77	
8414.32	40	11881.24	$6p \ ^4P_{1\frac{1}{2}} - 6d \ ^4P_{2\frac{1}{2}}$	7861.20	15	12717.21	$6s' \ ^2D_{2\frac{1}{2}} - 8p \ ^4P_{2\frac{1}{2}}$
8413.59	5c	11882.26	$nd \ 5_{2\frac{1}{2}} - 7p \ ^4D_{2\frac{1}{2}}$	7846.21	12c	12741.51	
8393.30	10000cZ	11910.99	$6s \ ^4P_{0\frac{1}{2}} - 6p \ ^2D_{1\frac{1}{2}}$	7813.39	50	12795.03	$6s' \ ^2D_{2\frac{1}{2}} - 8p \ ^4S_{1\frac{1}{2}}$
8391.70	200Z	11913.27	$6p \ ^2D_{2\frac{1}{2}} - 8s \ ^2P_{1\frac{1}{2}}$	7792.51	4d	12829.31	
8382.49	2c	11926.36	$6p \ ^4S_{1\frac{1}{2}} - nd \ 21_{2\frac{1}{2}}$	7789.48	5	12834.30	$6s' \ ^2D_{2\frac{1}{2}} - 8p \ ^2D_{2\frac{1}{2}}$
8352.73	40	11968.85	$nd \ 1_{1\frac{1}{2}} - 8p \ ^4P_{1\frac{1}{2}}$	7778.39	15	12852.60	$6p \ ^4P_{1\frac{1}{2}} - 8s \ ^4P_{2\frac{1}{2}}$
8333.19	50	11996.91	$6p \ ^4P_{2\frac{1}{2}} - nd \ 21_{2\frac{1}{2}}$	7768.13	25d	12869.57	$6s' \ ^2D_{1\frac{1}{2}} - 8p \ ^4P_{1\frac{1}{2}}$
8322.16	3c	12012.38	$nd \ 4_{1\frac{1}{2}} - 9p \ ^4P_{2\frac{1}{2}}$	7758.80	40d	12885.05	$6p \ ^4P_{1\frac{1}{2}} - 9s \ ^2P_{1\frac{1}{2}}$
8305.80	40c	12036.47	$nd \ 3_{2\frac{1}{2}} - 9p \ ^4P_{2\frac{1}{2}}$	7728.92	25	12934.86	
8289.50	60d	12060.14	$6s' \ ^2D_{1\frac{1}{2}} - 6p \ ^2P_{1\frac{1}{2}}$	7715.72	15c	12956.99	$nd \ 2_{1\frac{1}{2}} - 7p \ ^2S_{0\frac{1}{2}}$
8285.18	8d	12066.42		7700.20	2000cZ	12983.11	$6s' \ ^2D_{2\frac{1}{2}} - 8p \ ^4D_{3\frac{1}{2}}$
8279.57	2d	12074.21	$nd \ 2_{1\frac{1}{2}} - 9p \ ^4P_{0\frac{1}{2}}$	7696.92	50c	12988.64	
8262.54	30	12099.50		7671.01	200dZ	13032.51	$6p \ ^4P_{0\frac{1}{2}} - nd \ 21_{2.0\frac{1}{2}}$
8260.04	200Z	12103.15		7670.02	10d	13034.19	$6p \ ^4P_{0\frac{1}{2}} - 7s \ ^4P_{1\frac{1}{2}}$
8258.84	1	12104.91	$nd \ 4.1_{1\frac{1}{2}} - 9p \ ^4S_{1\frac{1}{2}}$	7611.16	5	13134.99	$nd \ 3_{2\frac{1}{2}} - 7p \ ^4D_{1\frac{1}{2}}$
8257.74	15	12106.52		7604.88	200Z	13145.85	$6s \ ^4P_{1\frac{1}{2}} - 7p \ ^4P_{2\frac{1}{2}}$
8251.08	80Z	12116.30	$6s \ ^2P_{0\frac{1}{2}} - 7p \ ^4P_{0\frac{1}{2}}$	7588.60	400Z	13174.04	$6p \ ^4P_{0\frac{1}{2}} - nd \ 22_{1\frac{1}{2}}$
8247.74	15d	12121.21		7571.25	100	13204.24	$6s' \ ^2D_{2\frac{1}{2}} - 8p \ ^4P_{0\frac{1}{2}}$
8240.05	4000Z	12132.52	$nd \ 3_{2\frac{1}{2}} - 9p \ ^4S_{1\frac{1}{2}}$	7556.65	500dZ	13229.74	$6s \ ^4P_{1\frac{1}{2}} - 7p \ ^4S_{1\frac{1}{2}}$
8222.57	500dZ	12158.31	$6s \ ^4P_{0\frac{1}{2}} - 6p \ ^2P_{1\frac{1}{2}}$	7554.18	2000Z	13234.07	$6p \ ^4D_{3\frac{1}{2}} - 6d \ ^4D_{3\frac{1}{2}}$
8213.95	50d	12171.07	$nd \ 2_{1\frac{1}{2}} - 9p \ ^4S_{1\frac{1}{2}}$	7547.03	5h	13246.60	$6p \ ^2D_{2\frac{1}{2}} - 7s \ ^4P_{1\frac{1}{2}}$
8206.49	2	12182.28	$nd \ 4.1_{1\frac{1}{2}} - 9p \ ^2D_{2\frac{1}{2}}$	7539.66	200	13259.55	$6p \ ^2D_{2\frac{1}{2}} - 6d \ ^4D_{3\frac{1}{2}}$
8187.94	30c	12209.73	$nd \ 3_{2\frac{1}{2}} - 9p \ ^2D_{2\frac{1}{2}}$	7531.81	200Z	13273.37	$6s' \ ^2D_{2\frac{1}{2}} - 8p \ ^4P_{0\frac{1}{2}}$
8179.00	50d	12223.08	$6p \ ^4P_{1\frac{1}{2}} - 7s \ ^2P_{0\frac{1}{2}}$	7490.52	500cZ	13346.54	$6s' \ ^2D_{1\frac{1}{2}} - 8p \ ^2P_{0\frac{1}{2}}$
8169.38	800cZ	12237.47	$6s' \ ^2D_{1\frac{1}{2}} - np \ ^2P_{1\frac{1}{2}}$	7483.97	4	13358.26	
8162.22	1	12248.20	$nd \ 2_{1\frac{1}{2}} - 9p \ ^2D_{2\frac{1}{2}}$	7476.45	25	13371.65	$6s \ ^4P_{1\frac{1}{2}} - 7p \ ^2D_{2\frac{1}{2}}$
8105.60	50dZ	12333.76	$6p \ ^4P_{1\frac{1}{2}} - nd \ 24_{1\frac{1}{2}}$	7468.99	5000Z	13385.01	$6p \ ^4D_{3\frac{1}{2}} - 6d \ ^4F_{4\frac{1}{2}}$
8090.76	1000Z	12356.38	$6p \ ^4P_{1\frac{1}{2}} - nd \ 25_{2\frac{1}{2}}$	7468.45	10	13385.98	
8065.70	300dZ	12394.77	$6s' \ ^2D_{2\frac{1}{2}} - 6p' \ ^2P_{1\frac{1}{2}}$	7468.16	25d	13386.50	$6p \ ^2D_{2\frac{1}{2}} - nd \ 22_{1\frac{1}{2}}$
8046.13	4	12424.75	$6s' \ ^2D_{1\frac{1}{2}} - 7p \ ^2P_{0\frac{1}{2}}$	7448.11	2d	13422.54	$nd \ 1_{1\frac{1}{2}} - 9p \ ^4S_{1\frac{1}{2}}$
8043.74	100000Z	12428.61	$6s \ ^4P_{2\frac{1}{2}} - 6p \ ^4P_{1\frac{1}{2}}$	7446.37	300dZ	13425.67	$6p \ ^4S_{1\frac{1}{2}} - 7s \ ^4P_{0\frac{1}{2}}$
8039.85	100dZ	12434.62	$6s \ ^2P_{0\frac{1}{2}} - 7p \ ^4P_{1\frac{1}{2}}$	7444.91	100d	13428.29	$6p \ ^4P_{0\frac{1}{2}} - 7s \ ^2P_{0\frac{1}{2}}$

TABLE 1. Wavelengths and term combinations of II—Continued

Wavelength	Intensity	Wave number	Designation	Wavelength	Intensity	Wave number	Designation
6570.38	150Z	15215.62	$6p\ ^4P_{3/2}$ —nd 28.2 <sub>1½</sub>	6183.98	5c	16166.34	$6p\ ^4P_{3/2}$ —nd 28.2 <sub>1½</sub>
6566.49	1000Z	15224.64	$6p\ ^2D_{3/2}$ —nd 28.3 <sub>1½</sub>	6173.62	3h	16193.48	$6s'\ ^2D_{3/2}$ —np 6 <sub>2½</sub>
6564.80	200cZ	15228.56	$6s\ ^2P_{3/2}$ —6p' $^2P_{1/2}$	6168.71	3	16206.37	
6560.82	300Z	15237.79	$6p\ ^2D_{3/2}$ —nd 28.1 <sub>2½</sub>	6147.43	50	16262.47	$6p\ ^4S_{1/2}$ —10s $^4P_{2/2}$
6547.34	50	15269.17		6115.97	100Z	16346.12	$6p\ ^4P_{2/2}$ —10s $^4P_{2/2}$
6489.11	6	15406.18	$6s\ ^2P_{0/2}$ —np 2 <sub>1½</sub>	6101.71	20	16384.31	
6488.10	300Z	15408.58	$6s\ ^2P_{1/2}$ —6p $^2P_{0/2}$	6082.43	1000Z	16436.26	$6s\ ^2P_{1/2}$ —6p $^4D_{2/2}$
6479.89	60	15428.11	$6p\ ^2D_{3/2}$ —nd 28.2 <sub>1½</sub>	6073.46	50Z	16460.53	$6s\ ^4P_{0/2}$ —4f 4 <sub>1½</sub>
6479.24	50	15429.65	$6p\ ^4P_{1/2}$ —nd 37 <sub>2½</sub>	6055.96	80Z	16508.10	$6s\ ^4P_{0/2}$ —4f 5.1 <sub>1½</sub>
6477.39	20	15434.06	$6p\ ^4P_{1/2}$ —nd 38 <sub>1½</sub>	6055.03	30d	16510.63	$6s\ ^4P_{0/2}$ —4f 6 <sub>1½</sub>
6455.00	50Z	15487.60	$6s\ ^4P_{1/2}$ —4f 1 <sub>3½</sub>	6053.49	300dZ	16514.83	$6s\ ^2P_{0/2}$ —8p $^2P_{0/2}$
6442.58	30	15517.45	$6p\ ^4P_{0/2}$ —nd 29 <sub>1½</sub>	6044.41	60	16539.64	
6434.49	40Z	15536.96	$6s\ ^4P_{1/2}$ —4f 4 <sub>1½</sub>	6042.71	100	16544.30	
6433.28	30Z	15539.88	$6s\ ^4P_{1/2}$ —4f 5 <sub>2½</sub>	6041.24	10	16548.31	
6415.70	100	15582.46		6024.08	2000dZ	16595.46	{ $6s'\ ^2D_{2/2}$ —np 7 <sub>1½</sub>
							$6s\ ^4P_{1/2}$ —6p' $^2P_{1/2}$
6414.80	12d	15584.64	$6s\ ^4P_{1/2}$ —4f 5.1 <sub>1½</sub>	6015.37	40	16619.49	$6p\ ^4D_{3/2}$ —9d $^4D_{3/2}$
6413.84	12d	15586.98	$6s\ ^4P_{1/2}$ —4f 6 <sub>1½</sub>	6004.99	20	16648.22	$6s\ ^2P_{0/2}$ —5f 1 <sub>1½</sub>
6411.22	50cZ	15593.35	$6s\ ^2P_{0/2}$ —7p $^2P_{0/2}$	6003.54	3	16652.23	$6p\ ^4S_{1/2}$ —nd 31 <sub>1½</sub>
6378.70	200	15672.85	$6s'\ ^2D_{2/2}$ —np 5 <sub>3½</sub>	5984.86	300	16704.21	{ $6p\ ^4D_{3/2}$ —9d $^4F_{4/2}$
6376.33	80d	15678.68	$6s'\ ^2D_{1/2}$ —7p $^4D_{2/2}$	5976.53	12		$6p\ ^4S_{1/2}$ —nd 31.1 <sub>1½</sub>
6371.68	400Z	15690.12	$6p\ ^4S_{1/2}$ —7d $^4P_{2/2}$	5973.50	60	16735.98	
6367.28	400dZ	15700.96	$6p\ ^4S_{1/2}$ —nd 26 <sub>1½</sub>	5969.36	1	16747.58	
6366.67	5c	15702.47	$6s'\ ^2D_{2/2}$ —7p $^4D_{1/2}$	5968.26	150	16750.67	$6s\ ^2P_{1/2}$ —6p $^2D_{1/2}$
6366.05	10cw	15703.98		5967.81	4	16751.92	$6s\ ^4P_{1/2}$ —np 1.1 <sub>2½</sub>
6359.16	500Z	15721.31	$6s\ ^2P_{1/2}$ —6p $^2S_{0/2}$	5967.46	3	16752.92	$6p\ ^4D_{3/2}$ —nd 36 <sub>2½</sub>
6355.57	150	15729.89	$6p\ ^2D_{3/2}$ —nd 27 <sub>1½</sub>	5966.76	100	16754.88	$6p\ ^4S_{1/2}$ —nd 32.1 <sub>0½</sub>
6339.44	1000Z	15769.91	$6p\ ^4P_{3/2}$ —7d $^4D_{3/2}$	5966.40	300cZ	16772.76	$6s\ ^4P_{1/2}$ —np 2 <sub>0½</sub>
6337.85	2000Z	15773.87	$6p\ ^4P_{3/2}$ —7d $^4P_{2/2}$	5956.87	300	16782.70	{ $6p\ ^4P_{2/2}$ —8d $^4D_{3/2}$
6333.50	400	15784.70	$6p\ ^4P_{3/2}$ —nd 26 <sub>1½</sub>	5955.00			$6s\ ^2P_{1/2}$ —np 1 <sub>1½</sub>
6330.37	800Z	15792.51	$6p\ ^4S_{1/2}$ —nd 27 <sub>1½</sub>	5954.38	150	16787.96	$6p\ ^4P_{2/2}$ —8d $^4P_{2/2}$
6323.82	70	15808.86	$6p\ ^4P_{0/2}$ —nd 30 <sub>0½</sub>	5947.78	2	16789.72	$6p\ ^4P_{2/2}$ —8d $^4P_{2/2}$
6313.13	500Z	15835.63	$6p\ ^4S_{1/2}$ —nd 27.1 <sub>2½</sub>	5943.07	1	16808.34	$6s\ ^4P_{1/2}$ —np 2 <sub>0½</sub>
6312.50	100	15837.21	$6p\ ^4P_{0/2}$ —nd 31.1 <sub>1½</sub>	5934.03	25	16821.67	$6p\ ^4D_{3/2}$ —nd 37 <sub>2½</sub>
6303.93	1	15858.74	$6s'\ ^2D_{1/2}$ —np 6 <sub>3½</sub>	5932.02	8cw	16847.29	$6p\ ^2D_{3/2}$ —nd 37 <sub>2½</sub>
6297.00	100dZ	15876.20	$6p\ ^4P_{0/2}$ —nd 27 <sub>1½</sub>	5911.17	20	16853.00	$6s\ ^2P_{0/2}$ —np 3 <sub>1½</sub>
6295.24	30	15880.63	$6p\ ^4P_{0/2}$ —nd 32 <sub>1½</sub>	5894.03	2000Z	16912.45	$6s\ ^2P_{1/2}$ —8p $^4P_{3/2}$
6293.98	1000Z	15883.81	$6s\ ^2P_{1/2}$ —6p $^4D_{1/2}$	5882.24	70Z	16961.63	$6s\ ^2P_{1/2}$ —6p $^2P_{1/2}$
6292.36	10cw, I II?	15887.90	$6p\ ^4P_{0/2}$ —nd 32.1 <sub>0½</sub>	5886.68	3	16995.62	$6s\ ^4P_{1/2}$ —8p $^4S_{1/2}$
6290.61	100	15892.32	$6p\ ^4S_{1/2}$ —nd 28.1 <sub>2½</sub>	5885.04	6	17034.90	$6s\ ^4P_{1/2}$ —8p $^2D_{3/2}$
6280.03	2w	15919.10	$6p\ ^4P_{2/2}$ —nd 27.1 <sub>2½</sub>	5882.85	50	17038.98	$6p\ ^4D_{3/2}$ —nd 33 <sub>0½</sub>
6262.78	40	15962.94	$6p\ ^4P_{3/2}$ —nd 28 <sub>3½</sub>	5882.20	50	17148.36	$6p\ ^4D_{3/2}$ —10d $^4F_{4/2}$
6259.12	25	15972.27		5881.19	5h	17167.94	$6p\ ^4P_{2/2}$ —11s $^4P_{2/2}$
6249.14	40	15997.78	$6p\ ^2D_{3/2}$ —nd 31 <sub>1½</sub>	5880.65	70Z	17206.37	$6p\ ^2D_{3/2}$ —nd 40 <sub>2½</sub>
6246.14	200	16005.53	$6s'\ ^2D_{2/2}$ —7f 1 <sub>2½</sub>	5878.65	80Z	17294.31	$6s\ ^2P_{1/2}$ —6p $^4D_{0/2}$
6245.38	80	16007.42	$6p\ ^4P_{2/2}$ —nd 28 <sub>3½</sub>	5876.33	1000dZ	17343.27	$6s\ ^4P_{2/2}$ —6p $^4D_{0/2}$
6244.72	40	16009.10	$6s'\ ^2D_{2/2}$ —7f 2 <sub>3½</sub>	5875.06	100	17383.29	{ $6p\ ^4P_{2/2}$ —8p $^4P_{0/2}$
6244.48	800Z	16009.72	$6s\ ^2P_{0/2}$ —8p $^4P_{1/2}$	5874.98	150Z	17404.96	$6s\ ^4P_{1/2}$ —8p $^4P_{1/2}$
6244.00	100	16010.96	$6p\ ^2D_{3/2}$ —8d $^4D_{3/2}$	5874.98	30	17407.74	$6p\ ^4S_{1/2}$ —nd 36 <sub>2½</sub>
6243.17	50d	16013.08	$6s'\ ^2D_{2/2}$ —7p $^4D_{3/2}$	5873.63	6cw	17439.07	$6s\ ^2P_{1/2}$ —6p $^4D_{0/2}$
6242.70	10	16014.29	$6s'\ ^2D_{2/2}$ —10p $^4D_{3/2}$	5873.63	40	17491.29	$6p\ ^4P_{2/2}$ —nd 36 <sub>2½</sub>
6240.83	200	16019.09	$6p\ ^4D_{3/2}$ —8d $^4D_{3/2}$	5873.63	50	17520.40	$6p\ ^4P_{2/2}$ —nd 37 <sub>2½</sub>
6238.12	8	16026.04	$6p\ ^4D_{3/2}$ —8d $^4P_{2/2}$	5873.63	5	17585.52	$6p\ ^4P_{2/2}$ —nd 37 <sub>2½</sub>
6233.50	100Z	16037.92	$6s\ ^2P_{0/2}$ —8p $^4P_{1/2}$	5873.63	5	17589.98	$6s\ ^4P_{1/2}$ —nd 36 <sub>2½</sub>
6230.91	20	16044.58	$6p\ ^2D_{3/2}$ —8d $^4D_{3/2}$	5873.63	5	17618.93	$6p\ ^4P_{2/2}$ —nd 36 <sub>2½</sub>
6228.91	4h	16049.73	$6p\ ^2D_{3/2}$ —nd 31.1 <sub>1½</sub>	5873.63	5	17637.01	$6p\ ^4P_{2/2}$ —nd 39 <sub>2½</sub>
6228.20	40	16051.57	$6p\ ^2D_{3/2}$ —8d $^4P_{2/2}$	5873.63	2	17735.01	$6p\ ^4P_{2/2}$ —nd 40 <sub>2½</sub>
6216.17	1c	16082.62	$6p\ ^4S_{1/2}$ —nd 28.2 <sub>1½</sub>	5873.63	2	17860.87	$6s\ ^4P_{1/2}$ —8p $^2P_{0/2}$
6213.10	500Z	16090.58	$6p\ ^4D_{3/2}$ —8d $^4F_{4/2}$	5873.63	3	17882.05	$6s\ ^4S_{1/2}$ —nd 38.1 <sub>2½</sub>
6212.14	3c	16093.06	$6p\ ^2D_{3/2}$ —nd 32 <sub>1½</sub>	5873.63	2	17882.05	$6p\ ^4P_{2/2}$ —nd 38.2 <sub>1½</sub>
6191.88	800Z	16145.73		5873.63	30cw		

TABLE 1. Wavelengths and term combinations of Ir—Continued

Wave-length	Intensity	Wave number	Designation	Wave-length	Intensity	Wave number	Designation
7444. 08	8	13429. 80		7018. 42	30	14244. 30	$6p\ ^4P_{\frac{3}{2}} - nd\ 23_{\frac{3}{2}}$
7437. 41	8d	13441. 01		7018. 24	100	14244. 66	$6p\ ^4D_{\frac{5}{2}} - 9s\ ^4P_{\frac{1}{2}}$
7421. 98	6	13469. 78		7010. 20	20	14261. 00	
7420. 01	200	13473. 37		7006. 16	5	14269. 23	$6s'\ ^2D_{\frac{1}{2}} - 9p\ ^4P_{\frac{5}{2}}$
7418. 96	12	13475. 27		6993. 41	100d	14295. 24	$6p\ ^4S_{\frac{1}{2}} - 7s\ ^2P_{\frac{3}{2}}$
7418. 26	100	13476. 54	$nd\ 2_{\frac{1}{2}} - 7f\ 1_{\frac{5}{2}}$	6991. 89	30	14298. 34	
7416. 48	500Z	13479. 78	$6s'\ ^2D_{\frac{1}{2}} - 5f\ 1_{\frac{5}{2}}$	6989. 78	500	14302. 66	$6p\ ^2D_{\frac{3}{2}} - 9s\ ^2P_{\frac{1}{2}}$
7414. 50	50	13483. 38	$6s\ ^4P_{\frac{1}{2}} - 7p\ ^4P_{\frac{3}{2}}$	6986. 51	400	14309. 36	
7413. 60	200dZ	13485. 01	$6s'\ ^2D_{\frac{1}{2}} - 5f\ 2_{\frac{5}{2}}$	6985. 13	40	14312. 18	$6p\ ^4P_{\frac{1}{2}} - nd\ 29_{\frac{1}{2}}$
7411. 20	200Z	13489. 38	$6s'\ ^2D_{\frac{1}{2}} - 5f\ 3_{\frac{5}{2}}$	6959. 09	50	14365. 73	$6s'\ ^2D_{\frac{1}{2}} - 9p\ ^4S_{\frac{1}{2}}$
7410. 50	1000Z	13490. 66		6940. 98	50d	14403. 22	
7402. 06	5000Z	13506. 04	$6p\ ^2D_{\frac{3}{2}} - nd\ 23_{\frac{3}{2}}$	6939. 71	8w	14405. 86	$6p\ ^4S_{\frac{1}{2}} - nd\ 24_{\frac{1}{2}}$
7390. 78	60c	13526. 65	$6s'\ ^2D_{\frac{1}{2}} - 5f\ 5_{\frac{1}{2}}$	6939. 21	15c	14406. 90	$6s\ ^4P_{\frac{3}{2}} - 7p\ ^4P_{\frac{5}{2}}$
7385. 58	15d	13536. 18		6928. 82	100	14428. 50	$6p\ ^4S_{\frac{1}{2}} - nd\ 25_{\frac{3}{2}}$
7384. 08	150dZ	13538. 92	$6p\ ^4P_{\frac{3}{2}} - nd\ 24_{\frac{1}{2}}$	6922. 05	2	14442. 62	$6s'\ ^2D_{\frac{1}{2}} - 9p\ ^2D_{\frac{3}{2}}$
7341. 23	20	13617. 95	$6p\ ^4P_{\frac{1}{2}} - 7d\ ^4P_{\frac{3}{2}}$	6899. 61	20d	14489. 58	$6p\ ^4P_{\frac{3}{2}} - nd\ 24_{\frac{1}{2}}$
7336. 76	1cw	13626. 75	$nd\ 3_{\frac{1}{2}} - np\ 6_{\frac{3}{2}}$	6856. 75	50	14580. 15	$6p\ ^4P_{\frac{1}{2}} - nd\ 31_{\frac{1}{2}}$
7335. 40	3	13628. 77	$6p\ ^4P_{\frac{1}{2}} - nd\ 26_{\frac{1}{2}}$	6849. 46	10	14595. 66	
7333. 72	5cw	13631. 90	$6p\ ^4P_{\frac{1}{2}} - nd\ 26.1_{\frac{3}{2}}$	6845. 63	40	14603. 84	$6p\ ^4P_{\frac{1}{2}} - nd\ 30_{\frac{3}{2}}$
7306. 15	2	13683. 33		6843. 07	15	14609. 29	$6s'\ ^2D_{\frac{1}{2}} - 9p\ ^4P_{\frac{3}{2}}$
7305. 43	400cZ	13684. 69	$6s'\ ^2D_{\frac{1}{2}} - np\ 3_{\frac{1}{2}}$				
7286. 44	40	13720. 35	$6p\ ^4P_{\frac{1}{2}} - nd\ 27_{\frac{1}{2}}$	6832. 42	1	14632. 07	$6p\ ^4P_{\frac{1}{2}} - nd\ 31.1_{\frac{1}{2}}$
7269. 97	25d	13751. 43	$6p\ ^2D_{\frac{3}{2}} - nd\ 24_{\frac{1}{2}}$	6831. 56	3	14633. 90	$6p\ ^4P_{\frac{1}{2}} - 8d\ ^4P_{\frac{3}{2}}$
7263. 61	25	13763. 48	$6p\ ^4P_{\frac{1}{2}} - nd\ 27.1_{\frac{1}{2}}$	6812. 30	50	14675. 29	$6p\ ^4P_{\frac{1}{2}} - nd\ 32_{\frac{1}{2}}$
7259. 98	100cZ	13770. 35	$6s'\ ^2D_{\frac{1}{2}} - np\ 4_{\frac{3}{2}}$	6808. 85	10	14682. 71	$6p\ ^4P_{\frac{1}{2}} - nd\ 32.1_{\frac{1}{2}}$
				6789. 23	60Z	14725. 15	$6s\ ^4P_{\frac{3}{2}} - 7p\ ^4P_{\frac{1}{2}}$
7258. 06	200Z	13774. 00	$6p\ ^2D_{\frac{3}{2}} - nd\ 25_{\frac{3}{2}}$	6788. 04	20	14727. 72	$nd\ 1_{\frac{1}{2}} - 7f\ 1_{\frac{1}{2}}$
7243. 49	25	13801. 70	$6s\ ^4P_{\frac{1}{2}} - 7p\ ^4P_{\frac{3}{2}}$	6784. 58	50	14735. 25	$nd\ 1_{\frac{1}{2}} - 7p\ ^4D_{\frac{3}{2}}$
7237. 84	500Z	13812. 48	$6s'\ ^2D_{\frac{1}{2}} - 5f\ 1_{\frac{1}{2}}$	6765. 27	2	14777. 30	$6s'\ ^2D_{\frac{1}{2}} - 9p\ ^2D_{\frac{3}{2}}$
7236. 78	1000Z	13814. 50		6741. 52	300c	14829. 36	
7235. 01	200	13817. 88		6739. 44	100	14833. 94	$6p\ ^4P_{\frac{3}{2}} - nd\ 26_{\frac{1}{2}}$
7233. 81	100	13820. 17	$6p\ ^4P_{\frac{1}{2}} - nd\ 28.1_{\frac{1}{2}}$	6738. 05	100d	14837. 00	$6p\ ^4P_{\frac{3}{2}} - nd\ 26.1_{\frac{1}{2}}$
7231. 82	200Z	13823. 98	$6s'\ ^2D_{\frac{1}{2}} - 5f\ 3_{\frac{3}{2}}$	6736. 53	100	14840. 35	
7230. 13	80	13827. 21	$6s'\ ^2D_{\frac{1}{2}} - 5f\ 4_{\frac{3}{2}}$	6732. 03	400Z	14850. 27	$6s'\ ^2D_{\frac{1}{2}} - 6f\ 1_{\frac{3}{2}}$
7228. 94	30	13829. 48	$6s'\ ^2D_{\frac{1}{2}} - 5f\ 4.1_{\frac{3}{2}}$	6726. 92	200d	14861. 55	$6s'\ ^2D_{\frac{1}{2}} - 6f\ 3_{\frac{1}{2}}$
7227. 30	700Z	13832. 62		6722. 73	20cw	14870. 81	$6s'\ ^2D_{\frac{1}{2}} - 6f\ 5_{\frac{3}{2}}$
7221. 10	5	13844. 50	$6s'\ ^2D_{\frac{1}{2}} - 5f\ 4.2_{\frac{3}{2}}$	6722. 12	8cw	14872. 15	$6s'\ ^2D_{\frac{1}{2}} - 6f\ 6_{\frac{1}{2}}$
7212. 50	20c	13861. 00	$6s'\ ^2D_{\frac{1}{2}} - 5f\ 5_{\frac{1}{2}}$	6702. 35	5c	14916. 02	$nd\ 1_{\frac{1}{2}} - np\ 6_{\frac{3}{2}}$
7192. 52	300cZ	13899. 51	$6p\ ^4S_{\frac{1}{2}} - nd\ 21.2_{\frac{3}{2}}$	6698. 56	25Z	14924. 47	$6s'\ ^2D_{\frac{1}{2}} - 9p\ ^4D_{\frac{3}{2}}$
7191. 66	400dZ	13901. 17	$6p\ ^4S_{\frac{1}{2}} - 7s\ ^4P_{\frac{1}{2}}$	6698. 46	200Z	14924. 69	$6p\ ^4S_{\frac{1}{2}} - 9s\ ^4P_{\frac{2}{2}}$
7182. 79	300	13918. 34	$nd\ 1_{\frac{1}{2}} - 6f\ 3_{\frac{3}{2}}$	6698. 10	60	14925. 49	$6p\ ^4P_{\frac{3}{2}} - nd\ 27_{\frac{1}{2}}$
7178. 03	30	13927. 56	$nd\ 1_{\frac{1}{2}} - 6f\ 5_{\frac{1}{2}}$	6697. 29	500cZ	14927. 30	
7177. 35	30d	13928. 89	$nd\ 1_{\frac{1}{2}} - 6f\ 6_{\frac{1}{2}}$	6683. 92	50	14957. 16	$6p\ ^4S_{\frac{1}{2}} - 9s\ ^2P_{\frac{1}{2}}$
7164. 79	1000Z	13953. 30	$6p\ ^4S_{\frac{1}{2}} - 6d\ ^4P_{\frac{3}{2}}$	6662. 10	400Z	15006. 15	$6p\ ^4D_{\frac{3}{2}} - 7d\ ^4D_{\frac{3}{2}}$
7148. 63	400cZ	13984. 84	$6p\ ^4P_{\frac{3}{2}} - 7s\ ^4P_{\frac{1}{2}}$	6661. 11	500Z	15008. 38	$6p\ ^4P_{\frac{3}{2}} - 9s\ ^4P_{\frac{2}{2}}$
7142. 06	2000Z	13997. 71	$6p\ ^4P_{\frac{3}{2}} - 6d\ ^4D_{\frac{3}{2}}$	6660. 34	100	15010. 11	$6p\ ^4D_{\frac{3}{2}} - 7d\ ^4P_{\frac{3}{2}}$
7137. 12	10	14007. 39		6650. 79	50	15031. 66	$6p\ ^2D_{\frac{3}{2}} - 7d\ ^4D_{\frac{3}{2}}$
7135. 55	60	14010. 48	$6p\ ^4P_{\frac{1}{2}} - nd\ 28.2_{\frac{1}{2}}$	6644. 26	30	15046. 43	$6p\ ^4P_{\frac{1}{2}} - nd\ 33_{\frac{1}{2}}$
7131. 06	10cw	14019. 32	$6s'\ ^2D_{\frac{1}{2}} - np\ 3_{\frac{1}{2}}$	6619. 66	5000Z	15102. 35	$6p\ ^2D_{\frac{3}{2}} - nd\ 26_{\frac{1}{2}}$
7122. 05	1200Z	14037. 04	$6p\ ^4P_{\frac{1}{2}} - 6d\ ^4P_{\frac{3}{2}}$	6604. 07	18c	15138. 00	$6p\ ^4D_{\frac{3}{2}} - 7d\ ^4F_{\frac{4}{2}}$
7120. 05	500Z	14040. 98	$6p\ ^4S_{\frac{1}{2}} - nd\ 22_{\frac{1}{2}}$	6598. 08	12c	15151. 74	$6s'\ ^2D_{\frac{1}{2}} - 7p\ ^2S_{\frac{3}{2}}$
7107. 43	65c	14065. 91	$nd\ 2_{\frac{1}{2}} - np\ 7_{\frac{1}{2}}$	6588. 67	4	15173. 38	$6p\ ^4P_{\frac{1}{2}} - nd\ 35_{\frac{1}{2}}$
7095. 17	80	14090. 22	$6p\ ^4P_{\frac{1}{2}} - 9s\ ^2P_{\frac{1}{2}}$	6585. 27	1000Z	15181. 22	$6p\ ^2D_{\frac{3}{2}} - nd\ 27.1_{\frac{1}{2}}$
7087. 76	200c	14104. 95	$6s'\ ^2D_{\frac{1}{2}} - np\ 4_{\frac{1}{2}}$	6583. 75	2000Z	15184. 73	$6s'\ ^2D_{\frac{1}{2}} - 6f\ 1_{\frac{3}{2}}$
7085. 05	5c	14110. 34	$6s\ ^2P_{\frac{1}{2}} - 4f\ 0_{\frac{1}{2}}$	6582. 92	300	15186. 64	$6s'\ ^2D_{\frac{1}{2}} - 6f\ 2_{\frac{3}{2}}$
7077. 87	300	14124. 65	$6p\ ^4P_{\frac{3}{2}} - nd\ 22_{\frac{1}{2}}$	6581. 30	25	15190. 37	
7063. 59	300Z	14153. 21		6580. 53	200	15192. 16	
7055. 30	20	14169. 84	$6s\ ^2P_{\frac{1}{2}} - 4f\ 4_{\frac{1}{2}}$	6578. 78	50	15196. 20	$6s'\ ^2D_{\frac{1}{2}} - 6f\ 3_{\frac{1}{2}}$
7045. 11	5	14190. 34	$6p\ ^4P_{\frac{1}{2}} - 10s\ ^4P_{\frac{2}{2}}$	6577. 68	100d	15198. 74	$6s'\ ^2D_{\frac{1}{2}} - 6f\ 4_{\frac{1}{2}}$
7031. 62	8w	14217. 56	$6s\ ^2P_{\frac{1}{2}} - 4f\ 5.1_{\frac{1}{2}}$	6575. 35	80	15204. 13	$6p\ ^4D_{\frac{3}{2}} - nd\ 28_{\frac{3}{2}}$
7030. 40	5	14220. 02	$6s\ ^2P_{\frac{1}{2}} - 4f\ 6_{\frac{1}{2}}$	6574. 21	20cw	15206. 75	$6s'\ ^2D_{\frac{1}{2}} - 6f\ 6_{\frac{1}{2}}$

TABLE I. Wavelengths and term combinations of II—Continued

Wave-length	Intensity	Wave number	Designation	Wave-length	Intensity	Wave number	Designation
5590. 20	8cw	17883. 50	$6s\ ^4P_{0\frac{1}{2}} - 7p\ ^2P_{0\frac{1}{2}}$	4392. 09	40Z	22761. 83	$6s\ ^2P_{1\frac{1}{2}} - 8p\ ^2D_{0\frac{1}{2}}$
5586. 36	400cZ	17895. 79	$6s\ ^4P_{2\frac{1}{2}} - 6p\ ^4D_{2\frac{1}{2}}$	4389. 85	5c	22773. 44	$6s\ ^4P_{2\frac{1}{2}} - 4f\ ^6I_{1\frac{1}{2}}$
5579. 05	40	17919. 24	$6s\ ^4P_{0\frac{1}{2}} - 8p\ ^4S_{1\frac{1}{2}}$	4321. 84	500dZ	23131. 81	$6s\ ^2P_{1\frac{1}{2}} - 8p\ ^4P_{1\frac{1}{2}}$
5549. 32	25	18015. 24	$6s\ ^4P_{1\frac{1}{2}} - 5f\ ^1I_{1\frac{1}{2}}$	4318. 36	3	23150. 45	
5546. 41	70	18024. 69	$6s\ ^4P_{1\frac{1}{2}} - 5f\ ^3I_{2\frac{1}{2}}$	4317. 52	2	23154. 95	
5544. 80	1	18029. 92	$6s\ ^2P_{0\frac{1}{2}} - 6f\ ^3I_{1\frac{1}{2}}$	4292. 36	3d	23290. 68	
5536. 03	1	18058. 47	$6p\ ^4S_{1\frac{1}{2}} - 6s''\ ^2S_{0\frac{1}{2}}$	4282. 75	10d	23342. 94	
5534. 98	10h	18061. 91	$6s\ ^4P_{1\frac{1}{2}} - 5f\ ^5I_{1\frac{1}{2}}$	4273. 36	4	23394. 23	
5500. 95	150cZ	18173. 64	$6s\ ^4P_{2\frac{1}{2}} - 6p\ ^2D_{1\frac{1}{2}}$	4265. 33	2	23438. 27	
5486. 94	3d	18220. 03	$6s\ ^4P_{1\frac{1}{2}} - np\ ^3I_{0\frac{1}{2}}$	4234. 54	75dZ	23608. 69	$6s\ ^2P_{1\frac{1}{2}} - 8p\ ^2P_{0\frac{1}{2}}$
5461. 24	7hl	18305. 79	$6s\ ^4P_{1\frac{1}{2}} - np\ ^4S_{1\frac{1}{2}}$	4209. 82	30Z	23747. 32	$6s\ ^2P_{1\frac{1}{2}} - 5f\ ^2I_{0\frac{1}{2}}$
5427. 94	10	18418. 08		4209. 06	20	23751. 60	$6s\ ^2P_{1\frac{1}{2}} - 5f\ ^3I_{2\frac{1}{2}}$
5427. 06	600c	18421. 08	$6s\ ^4P_{2\frac{1}{2}} - 6p\ ^2P_{1\frac{1}{2}}$	4203. 72	35dZ	23781. 78	$6s\ ^4P_{2\frac{1}{2}} - 6p'\ ^2P_{0\frac{1}{2}}$
5316. 36	1	18804. 64	$6s\ ^4P_{1\frac{1}{2}} - 9p\ ^4P_{2\frac{1}{2}}$	4202. 51	8d	23788. 62	$6s\ ^2P_{1\frac{1}{2}} - 5f\ ^5I_{1\frac{1}{2}}$
5300. 99	25	18859. 15		4189. 16	4	23864. 43	
5297. 17	20Z	18872. 77	$6s\ ^2P_{1\frac{1}{2}} - 7p\ ^4P_{2\frac{1}{2}}$	4174. 70	1	23947. 09	$6s\ ^2P_{1\frac{1}{2}} - np\ ^3I_{1\frac{1}{2}}$
5278. 73	10	18938. 69	$6s\ ^4P_{0\frac{1}{2}} - 5f\ ^1I_{1\frac{1}{2}}$	4172. 61	5w	23959. 08	$6s\ ^4P_{2\frac{1}{2}} - np\ ^2I_{0\frac{1}{2}}$
5273. 72	30Z	18956. 69	$6s\ ^2P_{1\frac{1}{2}} - 7p\ ^4S_{1\frac{1}{2}}$	4159. 85	25c	24032. 58	$6s\ ^2P_{1\frac{1}{2}} - np\ ^4I_{1\frac{1}{2}}$
5265. 69	40Z	18985. 59	$6s\ ^4P_{0\frac{1}{2}} - 5f\ ^5I_{1\frac{1}{2}}$	4148. 41	75dZ	24098. 85	$6s\ ^4P_{2\frac{1}{2}} - 8p\ ^4P_{2\frac{1}{2}}$
5238. 26	20	19085. 01		4134. 15	100dZ	24181. 97	$6s\ ^4P_{2\frac{1}{2}} - 8p\ ^4S_{1\frac{1}{2}}$
5234. 57	1000Z	19098. 46	$6s\ ^2P_{1\frac{1}{2}} - 7p\ ^2D_{2\frac{1}{2}}$	4129. 21	200Z	24210. 90	
5204. 15	300Z	19210. 10	$6s\ ^2P_{1\frac{1}{2}} - 7p\ ^4P_{0\frac{1}{2}}$	4127. 43	15	24221. 34	$6s\ ^4P_{2\frac{1}{2}} - 8p\ ^2D_{0\frac{1}{2}}$
5196. 77	10	19237. 38	$6s\ ^2P_{0\frac{1}{2}} - 7p\ ^2D_{1\frac{1}{2}}$	4125. 08	3w	24235. 14	
5194. 86	1	19244. 44	$6s\ ^2P_{0\frac{1}{2}} - 7p\ ^2P_{1\frac{1}{2}}$	4102. 23	200	24370. 13	$6s\ ^4P_{2\frac{1}{2}} - 8p\ ^4D_{0\frac{1}{2}}$
5154. 03	3	19396. 90	$6s\ ^4P_{1\frac{1}{2}} - 6f\ ^3I_{0\frac{1}{2}}$	4069. 48	8	24566. 25	
5153. 37	3	19399. 39	$6s\ ^4P_{1\frac{1}{2}} - 6f\ ^4S_{1\frac{1}{2}}$	4065. 33	40d	24591. 33	$6s\ ^4P_{2\frac{1}{2}} - 8p\ ^4P_{1\frac{1}{2}}$
5145. 52	400Z	19428. 98	$6s\ ^2P_{0\frac{1}{2}} - np\ ^7I_{0\frac{1}{2}}$	4059. 27	15	24628. 04	$6s\ ^2P_{1\frac{1}{2}} - 9p\ ^4S_{1\frac{1}{2}}$
5119. 29	10000Z	19528. 53	$6s\ ^2P_{1\frac{1}{2}} - 7p\ ^4P_{1\frac{1}{2}}$	4046. 63	80	24704. 97	$6s\ ^2P_{1\frac{1}{2}} - 9p\ ^2D_{2\frac{1}{2}}$
5029. 34	4	19877. 58	$6s\ ^4P_{1\frac{1}{2}} - 10p\ ^4P_{2\frac{1}{2}}$	3990. 82	2w	25050. 45	
4947. 59	15	20206. 24	$6s\ ^4P_{1\frac{1}{2}} - 7f\ ^1I_{0\frac{1}{2}}$	3978. 76	10	25126. 38	$6s\ ^2P_{1\frac{1}{2}} - 6f\ ^4I_{0\frac{1}{2}}$
4945. 74	7	20213. 79	$6s\ ^4P_{1\frac{1}{2}} - 7p\ ^4D_{2\frac{1}{2}}$	3977. 52	5w	25134. 21	$6s\ ^2P_{1\frac{1}{2}} - 6f\ ^6I_{0\frac{1}{2}}$
4919. 80	3	20320. 37	$6s\ ^4P_{0\frac{1}{2}} - 6f\ ^3I_{0\frac{1}{2}}$	3964. 89	20c	25214. 27	$6s\ ^4P_{2\frac{1}{2}} - 5f\ ^4S_{1\frac{1}{2}}$
4916. 94	200Z	20332. 19	$6s\ ^4P_{2\frac{1}{2}} - 7p\ ^4P_{2\frac{1}{2}}$	3934. 91	20	25406. 37	$6s\ ^2P_{2\frac{1}{2}} - np\ ^3I_{1\frac{1}{2}}$
4902. 00	75Z	20394. 16	$6s\ ^4P_{1\frac{1}{2}} - np\ ^6I_{2\frac{1}{2}}$	3933. 73	2	25413. 93	$6s\ ^2P_{1\frac{1}{2}} - 7p\ ^2S_{0\frac{1}{2}}$
4896. 75	200	20416. 02	$6s\ ^4P_{2\frac{1}{2}} - 7p\ ^4S_{1\frac{1}{2}}$	3921. 68	55d	25492. 08	$6s\ ^4P_{2\frac{1}{2}} - np\ ^4I_{2\frac{1}{2}}$
4882. 68	12	20474. 86		3918. 60	6	25512. 12	
4874. 55	2	20509. 00		3902. 02	6	25620. 52	
4862. 96	60Z	20557. 88	$6s\ ^4P_{2\frac{1}{2}} - 7p\ ^2D_{0\frac{1}{2}}$	3900. 55	8	25630. 17	$6s\ ^2P_{1\frac{1}{2}} - 7p\ ^4D_{1\frac{1}{2}}$
4862. 32	1000Z	20560. 59	$6s\ ^4P_{2\frac{1}{2}} - 7p\ ^4D_{3\frac{1}{2}}$	3893. 84	20	25674. 34	
4850. 51	60Z	20610. 65	$6s\ ^4P_{0\frac{1}{2}} - 7p\ ^2S_{0\frac{1}{2}}$	3853. 86	4	25940. 68	$6s\ ^2P_{1\frac{1}{2}} - 7p\ ^4D_{2\frac{1}{2}}$
4850. 35	50Z	20611. 37	$6s\ ^4P_{1\frac{1}{2}} - 7p\ ^2P_{1\frac{1}{2}}$	3847. 63	15	25982. 68	
4827. 57	35Z	20708. 59		3846. 41	25d	25990. 92	$6s\ ^4P_{2\frac{1}{2}} - 9p\ ^4P_{0\frac{1}{2}}$
4802. 46	5	20816. 86		3840. 95	50	26027. 87	
4800. 20	50	20826. 66	$6s\ ^4P_{0\frac{1}{2}} - 7p\ ^4D_{1\frac{1}{2}}$	3827. 24	20	26121. 11	$6s\ ^2P_{1\frac{1}{2}} - np\ ^6I_{2\frac{1}{2}}$
4763. 31	250Z	20987. 95	$6s\ ^4P_{2\frac{1}{2}} - 7p\ ^4P_{1\frac{1}{2}}$	3820. 91	25d	26164. 38	$6s\ ^4P_{2\frac{1}{2}} - 9p\ ^2D_{0\frac{1}{2}}$
4701. 52	5	21263. 79	$6s\ ^2P_{1\frac{1}{2}} - 4f\ ^4I_{1\frac{1}{2}}$	3796. 69	15	26331. 29	$6s\ ^2P_{1\frac{1}{2}} - 7p\ ^2D_{1\frac{1}{2}}$
4700. 88	40Z	21266. 68	$6s\ ^2P_{1\frac{1}{2}} - 4f\ ^5I_{2\frac{1}{2}}$	3762. 04	6w	26573. 80	$6s\ ^4P_{2\frac{1}{2}} - 6f\ ^2I_{3\frac{1}{2}}$
4690. 90	20c	21311. 92	$6s\ ^2P_{1\frac{1}{2}} - 4f\ ^5I_{1\frac{1}{2}}$	3760. 34	3w	26585. 81	$6s\ ^4P_{2\frac{1}{2}} - 6f\ ^4S_{1\frac{1}{2}}$
4690. 49	35Z	21313. 79	$6s\ ^2P_{1\frac{1}{2}} - 4f\ ^6I_{0\frac{1}{2}}$	3698. 10	2w	27033. 25	$6s\ ^4P_{2\frac{1}{2}} - 6f\ ^4I_{2\frac{1}{2}}$
4643. 84	1	21527. 89	$6s\ ^4P_{0\frac{1}{2}} - 7p\ ^2D_{0\frac{1}{2}}$	3694. 42	1	27060. 17	$6s\ ^4P_{2\frac{1}{2}} - np\ ^5I_{3\frac{1}{2}}$
4642. 32	8	21534. 94	$6s\ ^4P_{0\frac{1}{2}} - 7p\ ^2P_{1\frac{1}{2}}$	3693. 90	20d	27063. 98	$6s\ ^4P_{2\frac{1}{2}} - 10p\ ^4P_{0\frac{1}{2}}$
4602. 86	12	21719. 56	$6s\ ^4P_{0\frac{1}{2}} - np\ ^7I_{0\frac{1}{2}}$	3690. 42	8d	27089. 50	$6s\ ^4P_{2\frac{1}{2}} - 7p\ ^4D_{1\frac{1}{2}}$
4478. 56	100Z	22322. 36	$6s\ ^2P_{1\frac{1}{2}} - 6p'\ ^2P_{1\frac{1}{2}}$	3684. 42	5	27133. 62	
4443. 26	15d	22499. 70	$6s\ ^2P_{1\frac{1}{2}} - np\ ^2P_{0\frac{1}{2}}$	3648. 59	1w	27400. 07	$6s\ ^4P_{2\frac{1}{2}} - 7p\ ^4D_{0\frac{1}{2}}$
4409. 12	15Z	22673. 92	$6s\ ^4P_{2\frac{1}{2}} - 4f\ ^1I_{2\frac{1}{2}}$	3624. 72	5c	27580. 50	$6s\ ^4P_{2\frac{1}{2}} - np\ ^6I_{2\frac{1}{2}}$
4408. 95	20	22674. 78		3607. 55	5w	27711. 77	
4407. 94	25Z	22679. 96	$6s\ ^4P_{2\frac{1}{2}} - 4f\ ^2I_{3\frac{1}{2}}$	3597. 29	3w	27790. 80	$6s\ ^4P_{2\frac{1}{2}} - 7p\ ^2D_{1\frac{1}{2}}$
4407. 86	20c, Z	22680. 38	$\begin{cases} 6s\ ^2P_{1\frac{1}{2}} - 6p'\ ^2P_{1\frac{1}{2}} \\ \{ 6s\ ^2P_{1\frac{1}{2}} - 7p\ ^2P_{0\frac{1}{2}} \end{cases}$	3596. 39	1w	27797. 76	$6s\ ^4P_{2\frac{1}{2}} - 7p\ ^2P_{0\frac{1}{2}}$
4406. 54	10Z	22687. 19	$\begin{cases} 6s\ ^4P_{2\frac{1}{2}} - 4f\ ^2I_{3\frac{1}{2}} \\ \{ 6s\ ^4P_{2\frac{1}{2}} - 4f\ ^3I_{2\frac{1}{2}} \end{cases}$	3552. 78	3w	28138. 96	
4398. 99	40cZ	22726. 13	$6s\ ^4P_{2\frac{1}{2}} - 4f\ ^5I_{2\frac{1}{2}}$	2061. 633	2000	48489. 73	$5p^5\ ^2P_{0\frac{1}{2}} - 6s\ ^2P_{1\frac{1}{2}}$

TABLE 2. Infrared lines of I I

Computed wave length	Intensity	Wave number	Designation	Computed wave length	Intensity	Wave number	Designation
23070. 01	95	4333. 45	$5d\ 4D_{3/2} - 5f\ 3_{2/3}$	15074. 52	125	6631. 90	$nd\ 11_{1/2} - 6f\ 4_{3/2}$
23001. 54	100	4346. 35	$7p\ 4S_{1/2} - nd\ 24_{1/2}$	15052. 50	25	6641. 60	$nd\ 2_{1/2} - 7p\ 2D_{3/2}$
22308. 77	150	4481. 32	$6p\ 4D_{3/2} - nd\ 5_{3/2}$	15032. 73	310	6650. 34	$6p\ 2D_{3/2} - 7s\ 2P_{1/2}$
22226. 19	250	4497. 97	$6p\ 4P_{3/2} - nd\ 5_{1/2}$				$7s\ 4P_{1/2} - 8p\ 4S_{1/2}$
22182. 49	375	4506. 83	$6p\ 2D_{3/2} - nd\ 5_{2/2}$	14460. 38	275	6913. 56	$6p\ 2P_{3/2} - 7s\ 4P_{3/2}$
21569. 51	110	4634. 91	$6p\ 2P_{3/2} - nd\ 19_{1/2}$	14287. 74	400	6997. 10	$6p\ 4D_{3/2} - 7s\ 4P_{3/2}$
20153. 00	30	4960. 69	$6s\ 4P_{3/2} - 6p\ 4P_{3/2}$				$nd\ 5.1_{1/2} - 4f\ 5_{3/2}$
19910. 52	45	5021. 10	$nd\ 13_{2/2} - 5f\ 3_{3/2}$	14272. 18	220	7004. 73	$nd\ 5.1_{1/2} - 4f\ 6_{1/2}$
19835. 64	15	5040. 05	$nd\ 5_{2/2} - 7p\ 2D_{3/2}$	14176. 65	50	7051. 93	$nd\ 5_{2/2} - 4f\ 1_{3/2}$
19824. 93	10	5042. 78	$nd\ 5_{2/2} - 7p\ 4D_{3/2}$	13970. 89	35	7156. 15	$nd\ 5_{2/2} - 4f\ 2_{3/2}$
19426. 10	15	5146. 31	$6s'\ 2D_{1/2} - 6p\ 2P_{3/2}$	13958. 54	275	7162. 12	$nd\ 5_{2/2} - 4f\ 2.1_{3/2}$
19370. 06	260	5161. 20	$\{ 7p\ 4P_{3/2} - nd\ 27_{1/2}$	13869. 12	75	7208. 30	$nd\ 5_{2/2} - 4f\ 5_{3/2}$
19105. 35	300	5232. 71	$6p\ 4S_{1/2} - nd\ 5_{2/2}$				$6p\ 4P_{3/2} - nd\ 6_{1/2}$
19072. 11	220	5241. 83	$6p\ 4P_{3/2} - 7s\ 2P_{1/2}$	13774. 59	65	7257. 76	$6p\ 4S_{1/2} - 7s\ 2P_{1/2}$
19060. 64	10	5244. 98	$\{ 5d\ 4F_{3/2} - 9p\ 4D_{3/2}$	13685. 85	200	7304. 82	$a\ 13387. 8$
			$6s\ 4P_{1/2} - 6p\ 4P_{1/2}$	13467. 4			$13148. 85$
			$6p\ 4P_{3/2} - nd\ 5_{2/2}$	13119. 27	300	7603. 15	$5p^5\ 2P_{3/2} - 5p^5\ 2P_{3/2}$
18982. 85	35	5266. 48	$nd\ 5.1_{1/2} - 7p\ 4P_{3/2}$	13116. 03	110	7620. 29	$6p\ 4P_{3/2} - nd\ 7_{1/2}$
18634. 52	10	5364. 92	$6p\ 4S_{1/2} - nd\ 5.1_{1/2}$				$7p\ 4D_{3/2} - 10d\ 4F_{3/2}$
18348. 37	240	5448. 59	$6p\ 4P_{3/2} - nd\ 5.1_{1/2}$	12846. 14	125	7782. 32	$6p\ 4P_{3/2} - nd\ 8_{1/2}$
18276. 25	110	5470. 09	$nd\ 5_{2/2} - 7p\ 4P_{3/2}$	12843. 83		7783. 74	$6p\ 2P_{3/2} - 7s\ 2P_{3/2}$
16213. 94	110	6165. 85	$6s\ 4P_{3/2} - 6p\ 4P_{3/2}$	12265. 16	50	8150. 95	$6p\ 4D_{3/2} - 7d\ 4P_{3/2}$
			$6p\ 4D_{3/2} - 7s\ 4P_{3/2}$	12135. 81	90	8237. 82	$nd\ 5.1_{1/2} - np\ 2P_{3/2}$
16192. 66	30	6173. 96	$6s'\ 2D_{1/2} - 6p\ 4D_{3/2}$				
16038. 15	400	6233. 44	$6p\ 4D_{3/2} - 7s\ 4P_{3/2}$				
15972. 67	115	6258. 95	$6p\ 2D_{3/2} - 7s\ 4P_{3/2}$				
15583. 89	250	6415. 13	$6p\ 4P_{3/2} - nd\ 7_{1/2}$				
			$nd\ 4_{1/2} - 7p\ 4S_{1/2}$				
15528. 30	280	6438. 10	$\{ 6p\ 4P_{3/2} - 7s\ 2P_{1/2}$				
			$6p\ 4D_{3/2} - 7s\ 4P_{3/2}$				

<sup>a</sup> Observed wavelength.

TABLE 3. Wavelengths of I I in the ultraviolet

Wavelength $\lambda_{vac}$	Intensity	Wave number	Designation	Wavelength $\lambda_{vac}$	Intensity	Wave number	Designation
1876. 415	2000	53293. 12	$2P_{3/2} - 6s\ 4P_{3/2}$	1457. 470			
1844. 451	15000	54216. 66	$2P_{3/2} - 6s\ 4P_{1/2}$	1457. 389	5000	68612. 02	$2P_{1/2} - nd\ 4_{1/2}$
1830. 380	75000	54633. 46	$2P_{3/2} - 6s\ 4P_{2/2}$	1453. 179	5000	68615. 84	$2P_{3/2} - nd\ 4.1_{1/2}$
1799. 091	5000	55583. 61	$2P_{3/2} - 6s\ 2P_{3/2}$	1446. 260	5000	68814. 63	$2P_{3/2} - nd\ 19.1_{0/2}$
1782. 758	12000	56092. 88	$2P_{3/2} - 6s\ 2P_{1/2}$	1429. 539	800	69143. 83	$2P_{3/2} - nd\ 20_{0/2}$
			$2P_{3/2} - 6s\ 2P_{1/2}$			69952. 62	$2P_{3/2} - 8s\ 2P_{1/2}$
1702. 068	15000	58752. 06	$2P_{3/2} - 6s'\ 2D_{1/2}$	1425. 490	8000	70151. 32	$2P_{1/2} - nd\ 5_{2/2}$
1675. 174	1500	59695. 30	$2P_{3/2} - nd\ 1_{1/2}$	1421. 364	2000	70354. 93	$2P_{3/2} - 7s\ 4P_{3/2}$
1642. 137	2000	60896. 27	$2P_{3/2} - 6s\ 4P_{3/2}$	1412. 180	200	70812. 51	$2P_{3/2} - 7s\ 4P_{3/2}$
1640. 780	2500	60946. 62	$2P_{3/2} - nd\ 2_{1/2}$	1402. 793		71286. 35	$2P_{3/2} - nd\ 21.2_{0/2}$
1639. 106	200	61008. 87	$2P_{3/2} - nd\ 4_{1/2}$	1402. 758	15	71288. 01	$2P_{3/2} - 7s\ 4P_{3/2}$
1617. 604	5000	61819. 81	$2P_{1/2} - 6s\ 4P_{3/2}$	1400. 014	2000	71427. 84	$2P_{3/2} - nd\ 22_{1/2}$
1593. 580	5000	62751. 78	$2P_{3/2} - nd\ 5.1_{1/2}$	1395. 049	30	71682. 11	$2P_{3/2} - 7s\ 2P_{3/2}$
1582. 610	1500	63186. 76	$2P_{3/2} - 6s\ 2P_{3/2}$	1392. 898	2000	71792. 77	$2P_{3/2} - nd\ 24_{1/2}$
1545. 794	80	64691. 68	$2P_{3/2} - 7s\ 2P_{1/2}$	1390. 750	3000	71903. 44	$2P_{3/2} - 7s\ 4P_{2/2}$
1526. 448	2500	65511. 57	$2P_{3/2} - nd\ 6_{1/2}$	1383. 225	4000	72294. 83	$2P_{3/2} - 7s\ 2P_{1/2}$
1518. 047	15000	65874. 10	$2P_{3/2} - nd\ 7_{1/2}$	1382. 284	1200	72344. 05	$2P_{3/2} - 9s\ 2P_{1/2}$
1514. 678	5000	66020. 64	$2P_{3/2} - 6s'\ 2D_{2/2}$	1368. 217	2500	73087. 83	$2P_{3/2} - nd\ 26_{1/2}$
1514. 323	2000	66036. 13	$2P_{3/2} - nd\ 8_{1/2}$	1367. 714	2500	73114. 72	$2P_{3/2} - nd\ 6_{1/2}$
1507. 041	5000	66355. 21	$2P_{3/2} - 6s'\ 2D_{1/2}$	1366. 506	800	73179. 34	$2P_{3/2} - nd\ 27_{1/2}$
1492. 888	5000	66984. 25	$2P_{3/2} - nd\ 11_{1/2}$	1361. 111	3000	73469. 39	$2P_{3/2} - nd\ 28.2_{1/2}$
1485. 918	1000	67298. 45	$2P_{3/2} - nd\ 1_{1/2}$	1360. 965	5000	73477. 25	$2P_{3/2} - nd\ 7_{1/2}$
1465. 828	2500	68220. 83	$2P_{3/2} - nd\ 16_{1/2}$	1357. 971	3000	73639. 28	$2P_{3/2} - nd\ 8_{1/2}$
1459. 145	4000	68533. 28	$2P_{3/2} - nd\ 19_{1/2}$	1355. 542	2000	73771. 21	$2P_{3/2} - nd\ 29_{1/2}$
1458. 794	2500	68549. 77	$2P_{3/2} - nd\ 2_{1/2}$	1355. 099	5000	73795. 35	$2P_{3/2} - nd\ 9_{2/2}$
1457. 981	10000	68588. 00	$2P_{3/2} - nd\ 3_{2/2}$	1350. 206	600	74062. 75	$2P_{3/2} - nd\ 30_{0/2}$

TABLE 3. Wavelengths of I<sub>I</sub> in the ultraviolet—Continued

Wavelength λ <sub>vac</sub>	Intensity	Wave number	Designation	Wavelength λ <sub>vac</sub>	Intensity	Wave number	Designation
1349. 691	10	74091. 05	$^2P_{0\frac{1}{2}}-nd\ 31.1_{1\frac{1}{2}}$	1236. 362	400	80882. 46	$^2P_{1\frac{1}{2}}-nd\ 28.1_{2\frac{1}{2}}$
1348. 903	800	74134. 29	$^2P_{0\frac{1}{2}}-nd\ 32.1_{1\frac{1}{2}}$	<sup>a</sup> 1233. 517	50	81069. 0	$^2P_{1\frac{1}{2}}-nd\ 28.2_{1\frac{1}{2}}$
1348. 768	40	74141. 74	$^2P_{0\frac{1}{2}}-nd\ 32.1_{0\frac{1}{2}}$	1233. 463	300	81072. 54	$^2P_{1\frac{1}{2}}-nd\ 28.2_{0\frac{1}{2}}$
1343. 626	1000	74425. 46	$^2P_{0\frac{1}{2}}-nd\ 32.2_{0\frac{1}{2}}$	<sup>a</sup> 1232. 914	10	81108. 7	$^2P_{1\frac{1}{2}}-10s\ ^4P_{2\frac{1}{2}}$
1342. 449	60	74490. 73	$^2P_{0\frac{1}{2}}-nd\ 33.0_{0\frac{1}{2}}$	1230. 732	400	81252. 48	
1341. 264	100	74556. 55	$^2P_{0\frac{1}{2}}-nd\ 34.0_{0\frac{1}{2}}$	1228. 888	500	81374. 36	$^2P_{1\frac{1}{2}}-nd\ 29.1_{0\frac{1}{2}}$
1340. 709	1500	74587. 40	$^2P_{1\frac{1}{2}}-nd\ 11_{1\frac{1}{2}}$	<sup>a</sup> 1228. 041	100	81430. 5	
1339. 903	800	74632. 28	$^2P_{0\frac{1}{2}}-nd\ 35.0_{0\frac{1}{2}}$	1224. 856	400	81642. 25	$^2P_{1\frac{1}{2}}-nd\ 31.1_{0\frac{1}{2}}$
<sup>a</sup> 1338. 210	20	74726. 7		1224. 501	300	81665. 90	$^2P_{1\frac{1}{2}}-nd\ 30.0_{0\frac{1}{2}}$
<sup>b</sup> 1336. 478	1000	74823. 52	{ $^2P_{1\frac{1}{2}}-nd\ 13.2_{2\frac{1}{2}}$ $^2P_{0\frac{1}{2}}-nd\ 36.1_{0\frac{1}{2}}$	{ 1224. 077 1224. 049	600	{ 81694. 20 81696. 06	$^2P_{1\frac{1}{2}}-8d\ ^4P_{2\frac{1}{2}}$
1335. 238	200	74893. 04	$^2P_{0\frac{1}{2}}-nd\ 38.0_{0\frac{1}{2}}$	1223. 430	100	81737. 44	$^2P_{1\frac{1}{2}}-nd\ 32.1_{0\frac{1}{2}}$
1333. 232	5	75005. 69	$^2P_{0\frac{1}{2}}-nd\ 38.2_{1\frac{1}{2}}$	<sup>a</sup> 1219. 327	10	82012. 4	
<sup>a</sup> 1330. 714	30	75147. 6		1219. 087	70	82028. 61	$^2P_{1\frac{1}{2}}-nd\ 32.2_{0\frac{1}{2}}$
1330. 189	2000	75177. 26	$^2P_{1\frac{1}{2}}-nd\ 14.2_{2\frac{1}{2}}$	<sup>a</sup> 1218. 909	70	82040. 6	
1325. 463	10	75445. 32	$^2P_{0\frac{1}{2}}-6s''\ ^2S_{0\frac{1}{2}}$				
1318. 844	30	75823. 98	$^2P_{1\frac{1}{2}}-nd\ 16.1_{0\frac{1}{2}}$	<sup>a</sup> 1218. 711	100	82053. 9	
1317. 542	3000	75898. 91	$^2P_{1\frac{1}{2}}-nd\ 17.2_{1\frac{1}{2}}$	1218. 411	200	82074. 10	$^2P_{1\frac{1}{2}}-11s\ ^4P_{2\frac{1}{2}}$
1313. 947	3000	76106. 57	$^2P_{1\frac{1}{2}}-nd\ 18.2_{2\frac{1}{2}}$	1218. 118	40	82093. 88	$^2P_{1\frac{1}{2}}-nd\ 33.0_{0\frac{1}{2}}$
1313. 432	1500	76136. 43	$^2P_{1\frac{1}{2}}-nd\ 19.1_{0\frac{1}{2}}$	1217. 142	150	82159. 70	$^2P_{1\frac{1}{2}}-nd\ 34.0_{0\frac{1}{2}}$
1302. 983	3000	76746. 98	$^2P_{1\frac{1}{2}}-nd\ 20.0_{0\frac{1}{2}}$	<sup>c</sup> 1216. 021	?	82235. 43	$^2P_{1\frac{1}{2}}-nd\ 35.0_{0\frac{1}{2}}$
1300. 335	10000	76903. 28	$^2P_{1\frac{1}{2}}-nd\ 21.2_{1\frac{1}{2}}$	<sup>a</sup> 1214. 631	50	82329. 5	
<sup>a</sup> 1299. 012	50	76981. 6		1213. 627	40	82397. 62	$^2P_{1\frac{1}{2}}-nd\ 36.2_{1\frac{1}{2}}$
1291. 143	300	77450. 76	$^2P_{1\frac{1}{2}}-8s\ ^4P_{2\frac{1}{2}}$	1213. 199	60	82426. 68	$^2P_{1\frac{1}{2}}-nd\ 36.1_{0\frac{1}{2}}$
1289. 395	3000	77555. 77	$^2P_{1\frac{1}{2}}-8s\ ^2P_{1\frac{1}{2}}$	1212. 242	60	82491. 80	$^2P_{1\frac{1}{2}}-nd\ 37.2_{1\frac{1}{2}}$
1275. 255	1500	78415. 66	$^2P_{1\frac{1}{2}}-7s\ ^4P_{0\frac{1}{2}}$	1212. 177	60	82496. 19	$^2P_{1\frac{1}{2}}-nd\ 38.0_{0\frac{1}{2}}$
1267. 596	600	78889. 50	$^2P_{0\frac{1}{2}}-nd\ 21.2_{0\frac{1}{2}}$	1210. 880	10	82584. 58	$^2P_{1\frac{1}{2}}-nd\ 38.1_{2\frac{1}{2}}$
1267. 569		78891. 16	$^2P_{1\frac{1}{2}}-7s\ ^4P_{1\frac{1}{2}}$	1210. 524	60	82608. 84	$^2P_{1\frac{1}{2}}-nd\ 38.2_{1\frac{1}{2}}$
1266. 731	150	78943. 37	$^2P_{1\frac{1}{2}}-6d\ ^4P_{2\frac{1}{2}}$	1210. 050	60	82641. 21	$^2P_{1\frac{1}{2}}-nd\ 39.2_{1\frac{1}{2}}$
1265. 326	40	79030. 99	$^2P_{1\frac{1}{2}}-nd\ 22.1_{1\frac{1}{2}}$	<sup>a</sup> 1208. 466	50	82749. 5	
1261. 269	800	79285. 26	$^2P_{1\frac{1}{2}}-7s\ ^2P_{0\frac{1}{2}}$	<sup>a</sup> 1207. 964	20	82783. 9	
1259. 510	3000	79395. 92	$^2P_{1\frac{1}{2}}-nd\ 24.1_{1\frac{1}{2}}$	1206. 988	10	82850. 87	$^2P_{1\frac{1}{2}}-nd\ 40.2_{1\frac{1}{2}}$
1259. 153	2500	79418. 49	$^2P_{1\frac{1}{2}}-nd\ 25.2_{2\frac{1}{2}}$	<sup>a</sup> 1205. 430	25	82957. 9	
1251. 335	600	79914. 68	$^2P_{1\frac{1}{2}}-9s\ ^4P_{2\frac{1}{2}}$	1204. 116	40	83048. 47	$^2P_{1\frac{1}{2}}-6s''\ ^2S_{0\frac{1}{2}}$
1250. 826	400	79947. 20	$^2P_{1\frac{1}{2}}-9s\ ^2P_{1\frac{1}{2}}$	<sup>a</sup> 1201. 348	30	83239. 8	
1249. 969	15	80001. 95	$^2P_{1\frac{1}{2}}-nd\ 25.1_{0\frac{1}{2}}$	<sup>a</sup> 1200. 946	30	83267. 7	
1239. 463	70	80680. 12	$^2P_{0\frac{1}{2}}-7d\ ^4P_{2\frac{1}{2}}$	<sup>a</sup> 1200. 711	25	83284. 0	
1239. 296	70	80690. 98	$^2P_{1\frac{1}{2}}-nd\ 26.1_{1\frac{1}{2}}$	<sup>a</sup> 1196. 786	10	83557. 1	
1239. 249	70	80694. 00	$^2P_{1\frac{1}{2}}-nd\ 26.1_{0\frac{1}{2}}$	<sup>a</sup> 1195. 288	15	83661. 8	
1237. 892	300	80782. 49	$^2P_{1\frac{1}{2}}-nd\ 27.1_{1\frac{1}{2}}$				
1237. 231	200	80825. 65	$^2P_{1\frac{1}{2}}-nd\ 27.1_{2\frac{1}{2}}$				

<sup>a</sup> Wavelengths measured by C. H. Corliss and W. C. Martin.<sup>b</sup> Also I<sub>II</sub>.<sup>c</sup> Masked by I<sub>II</sub> and Ly<sub>α</sub>.TABLE 4. Zeeman effect of I<sub>I</sub>

Wavelength	Magnetic patterns	Wavelength	Magnetic patterns
11236. 56	(0.117, <b>0.356</b> ) 1.240, <b>1.479</b> , 1.721	10325. 90	(0.194) 1.163, 1.441, <b>1.805</b>
11020. 60	( <b>0.123</b> , 0.370, 0.643) <b>0.629</b> , 0.844, 1.044	10238. 82	(0.086) <b>1.298</b> , 1.494
11017. 14	(0.524) ...	10172. 91	(0.000w) 0.907
10685. 82	(0.754) ...	10158. 64	(0.213) 1.348w
10466. 54	( <b>0.079</b> , 0.240) <b>0.976</b> , 1.147, 1.299, 1.488	10141. 83	(0.370) 0.482, <b>1.152</b>
10435. 34	(0.257) ...	10131. 16	(0.275) 1.095, <b>1.612</b>
10416. 61	(0.428, <b>1.230</b> ) 0.369, <b>1.187</b> , 1.985	10003. 05	(0.118) 1.500, <b>1.747</b>
10391. 74	(0.249) 1.042, <b>1.563</b>	9963. 30	(0.000w) 1.053w A
10375. 20	(0.000w) 0.943w A	9842. 75	(0.141, <b>0.449</b> ) 1.184, <b>1.472</b> , 1.785
10375. 20	( <b>0.059</b> , 0.252, 0.419, 0.603, 0.775) <b>1.034</b> , 1.211, 1.383, 1.586, 1.753, 1.897	9813. 53	( <b>0.179</b> , 0.553) <b>0.702</b> , 1.077, 1.445

TABLE 4. Zeeman effect of I I—Continued

Wave-length	Magnetic patterns	Wave-length	Magnetic patterns
9800. 89	(0.170) 0.952	7944. 85	(0.000) 0.929
9749. 20	(0.000) 1.178	7700. 20	( <b>0.074</b> , 0.265) 0.959
9744. 83	(0.141) . . .	7671. 01	(0.330) 1.223
9731. 73	(0.150) 1.502	7604. 88	( <b>0.068</b> , 0.212) 1.376
9725. 47	(0.245) <b>0.798</b> , 1.290	7588. 60	(0.180) 1.053
9653. 06	(0.000) 1.525	7556. 65	(0.037) 1.598
9649. 61	(0.000) 0.982 A	7554. 18	(0.182) 1.313, <b>1.418</b>
9598. 22	(0.000) 1.053	7531. 81	(0.000) 1.480
9466. 34	(0.282) 1.818	7490. 52	(0.000) 1.244
9427. 15	( <b>0.096</b> , 0.274, 0.451) <b>0.879</b> , 1.050, 1.223, 1.405	7468. 99	( <b>0.045</b> , 0.143, 0.237) <b>1.006</b> , 1.088, 1.187
9426. 71	(0.654) 1.886	7446. 37	(0.344) . . .
9335. 05	( <b>0.118</b> , 0.348) <b>1.005</b> , 1.250, 1.500, 1.738	7416. 48	(0.523) 0.606, <b>0.989</b> , 1.346
9321. 95	(0.395) 1.216, <b>1.991</b>	7413. 60	(0.000w) 0.933w
9227. 74	(0.114, <b>0.300</b> ) 1.294, <b>1.505</b> , 1.705	7411. 20	(0.000) 1.186
9180. 20	(0.157) 1.114	7410. 50	(0.000w) 1.120
9156. 91	(0.055, <b>0.585</b> ) 1.531, <b>1.968</b>	7402. 06	(0.000W) 0.993 A
9128. 03	( <b>0.514</b> , 1.401) <b>0.647</b> , 1.613	7384. 08	(0.000) 1.379
9113. 91	(0.000) 1.385	7305. 43	(0.430) 1.167
9098. 86	(0.170, <b>0.504</b> ) 1.118, <b>1.460</b> , 1.790	7259. 98	(0.000) 1.393
9087. 16	(0.476) 0.650, <b>1.009</b> , 1.317	7258. 06	(0.072) 1.196
9079. 34	(0.174, 0.478, <b>0.868</b> ) 0.682, 1.036, <b>1.372</b> , 1.738, 2.117	7237. 84	( <b>0.128</b> , 0.361, 0.599) 0.590
9058. 33	( <b>0.090</b> , 0.268, 0.455) <b>0.934</b> , 1.107, 1.294, 1.503, 1.698	7236. 78	( <b>0.060</b> , 0.198, 0.341, 0.454) <b>0.807</b> , 0.938, 1.050
9022. 40	(0.622) <b>0.717</b> , 1.939	7231. 82	(0.362) 1.159
8993. 13	( <b>0.196</b> , 0.552) <b>0.628</b> , 1.033, 1.419	7227. 30	(0.606) . . .
8969. 04	(0.289, <b>0.853</b> ) 1.311	7192. 52	(0.390) 0.932, <b>1.758</b>
8964. 69	(0.000) 1.098	7191. 66	(0.000) 1.589
8925. 97	(0.444, <b>0.743</b> ) 0.821, 1.102, <b>1.372</b> , 1.676, 1.971	7164. 79	( <b>0.130</b> , 0.418) <b>0.967</b> , 1.251
8898. 50	(0.157, <b>0.452</b> ) 1.185, <b>1.492</b> , 1.793	7148. 63	(. . .) 1.402
8857. 50		7142. 06	( <b>0.075</b> , 0.246, 0.417) <b>0.947</b> , 1.106, 1.277
8853. 80	P-B	7122. 05	(0.247, <b>0.386</b> ) . . . <b>1.293</b> , 1.435
8853. 24		7120. 05	(0.206, <b>0.616</b> ) 0.965, <b>1.437</b> , 1.838
8847. 14	(0.427) 0.840	7085. 05	(0.000) 1.171
8816. 65	(0.362) 1.083, <b>1.289</b> , 1.458	7063. 59	(0.501) 1.092
8748. 22	(0.266) 1.338	6789. 23	(0.553) 0.920
8700. 80	(0.269) 0.910, 1.091	6732. 03	(0.000 W) 0.939 w
8664. 95	(0.062) 1.198	6698. 46	(0.000) 1.492 A
8636. 40	(0.098) 1.477	6697. 29	( <b>0.112</b> , 0.368, 0.643) . . .
8545. 52	(0.098) 1.268 A	6662. 10	(0.243 w) 1.325 w
8486. 11	( <b>0.072</b> , 0.224, 0.375) <b>1.038</b>	6661. 11	(0.092) 1.530
8393. 30	(0.640) <b>0.659</b> , 1.951	6619. 66	(0.000 W) 1.553 A
8391. 70	( <b>0.067</b> , 0.184) <b>1.049</b> , 1.143, 1.303	6585. 27	(0.275) 0.670, <b>0.898</b>
8260. 04	(0.218w) 1.284w	6583. 75	(0.000 W) 1.028 A
8251. 08	(0.389) 1.183	6570. 38	(0.165) 1.086
8240. 05	(0.0W) 1.043 A	6566. 49	( <b>0.086</b> , 0.275) <b>0.824</b> , 0.983, 1.135
8222. 57	(0.622) <b>0.721</b> , 1.950	6564. 80	(0.285) . . .
8169. 38	( <b>0.117</b> , 0.325) 0.762	6560. 82	(0.143 w) 1.204
8105. 60	(0.000) 1.396	6488. 10	(0.078) 1.310, <b>1.450</b>
8090. 76	( <b>0.111</b> , 0.382) <b>0.809</b> , 1.029, 1.264	6455. 00	
8065. 70	(0.000) <b>1.343</b>	6434. 49	P-B
8043. 74	( <b>0.082</b> , 0.242) 1.855 B	6433. 28	
8039. 85	(0.339) . . .	6415. 70	(0.000 w, 0.303) 1.154 A
8023. 01	(0.000) 1.524	6411. 22	(0.339) 1.140
8003. 63	(0.714) 1.868	6371. 68	( <b>0.142</b> , 0.354) <b>1.007</b> , 1.246, 1.500
7974. 48	(0.128), 0.616, <b>0.920</b>	6367. 28	(0.217, <b>0.707</b> ) 0.604, <b>1.090</b> , 1.585
7969. 48	(0.106) 1.536	6359. 16	(0.000) 1.375
7955. 90	(0.234) . . .	6339. 44	(0.000 w, 0.186, 0.377) <b>0.930</b> , 1.113
		6337. 85	(0.217, <b>0.367</b> ) 1.598
		6330. 37	(0.228) 1.575
		6313. 13	(0.195, <b>0.371</b> ) 1.257
		6297. 00	(0.000 w) 1.540

TABLE 4. Zeeman effect of I I—Continued

Wave-length	Magnetic patterns	Wave-length	Magnetic patterns
6293. 98	(0.000) 1.350	5204. 15	(0.074) <b>1.298</b> , 1.434
6244. 48	(0.000 <i>W</i> ) . . .	5145. 52	(0.000 <i>w</i> ) 0.8854
6233. 50	(0.278) . . .	5119. 29	(0.116) 1.415
6213. 10	(0.000 <i>W</i> ) 1.097 <i>A</i>	4916. 94	(0.217) 1.538
6191. 88	(0.000 <i>W</i> ) 1.050	4902. 00	(0.000 <i>w</i> ) 1.129 <i>A</i>
6115. 97	(0.000 <i>w</i> , 0.205) 1.529 <i>B</i>	4862. 96	(0.216, 0.549, <b>0.913</b> ) 0.712, 1.078, <b>1.421</b> , 1.749, 2.096
6082. 43	(0.000) 1.388	4862. 32	(0.000) 1.073
6073. 46	} P-B	4850. 51	} P-B
6055. 96		4850. 35	
6055. 03		4827. 57	(0.495) 0.990
6053. 49	(0.000) 0.888	4763. 31	(0.000) 1.630
6024. 08	(0.172, <b>0.406</b> ) 1.213, <b>1.489</b> , 1.759	4700. 88	(0.000 <i>W</i> ) 0.9584
5981. 26	(0.124) 1.345	4690. 49	(0.518) 0.841, <b>1.433</b> , . . .
5960. 40	( <b>0.358</b> , 1.036) <b>0.000d</b> , 0.643, 1.273	4478. 56	(0.000) 1.377
5894. 03	(0.000) 1.346	4409. 12	(. . .) 0.626
5882. 24	(0.000 <i>w</i> ) 1.699	4408. 01	( <b>0.238</b> , 0.699, 1.228) <b>0.000</b> , 0.328, 0.877, 1.361, 1.914
5780. 65	(0.122) 1.261, <b>1.505</b>	4234. 54	(0.000 <i>d</i> ) 1.243, <b>1.595</b>
5764. 33	( <b>0.131</b> , 0.349) 1.243, 1.473, 1.749, <b>2.014</b>	4209. 82	(. . .) 1.198
5743. 90	(0.478) . . .	4203. 72	(0.000) 1.786
5586. 36	(0.365) 1.280, <b>1.483</b> , 1.707	4148. 41	(0.000) 1.515
5500. 95	( <b>0.121</b> , 0.416) 1.443, 1.795, <b>2.070</b>	4134. 15	(0.000) 1.442
5297. 17	(0.000) 1.595	4129. 21	(0.000) 1.122
5273. 72	(0.289) 1.272, <b>1.447</b> , 1.657	2061. 63	(0.335) 1.005, <b>1.669</b>

TABLE 5. Predicted terms of I I

Electron configuration	Terms		
5s <sup>2</sup> 5p <sup>5</sup>	<sup>2</sup> P <sup>o</sup>		
5s 5p <sup>6</sup>	<sup>2</sup> S		
5s <sup>2</sup> 5p <sup>4</sup> (I II)	<sup>3</sup> P	<sup>1</sup> D	<sup>1</sup> S
5s <sup>2</sup> 5p <sup>4</sup> ns	<sup>2</sup> S <sup>o</sup> , <sup>2</sup> P <sup>o</sup> , <sup>2</sup> D <sup>o</sup> , <sup>4</sup> S <sup>o</sup> , <sup>4</sup> P <sup>o</sup> , <sup>4</sup> D <sup>o</sup>	<sup>2</sup> D <sup>o</sup> , <sup>2</sup> F <sup>o</sup>	<sup>2</sup> S
5s <sup>2</sup> 5p <sup>4</sup> np	<sup>2</sup> P <sup>o</sup> , <sup>2</sup> D <sup>o</sup> , <sup>2</sup> F <sup>o</sup> , <sup>4</sup> P <sup>o</sup> , <sup>4</sup> D <sup>o</sup> , <sup>4</sup> F <sup>o</sup>	<sup>2</sup> P <sup>o</sup> , <sup>2</sup> D <sup>o</sup> , <sup>2</sup> F <sup>o</sup> , <sup>2</sup> G <sup>o</sup>	<sup>2</sup> P <sup>o</sup>
5s <sup>2</sup> 5p <sup>4</sup> nd	<sup>2</sup> D <sup>o</sup> , <sup>2</sup> F <sup>o</sup> , <sup>2</sup> G <sup>o</sup> , <sup>4</sup> D <sup>o</sup> , <sup>4</sup> F <sup>o</sup> , <sup>4</sup> G <sup>o</sup>	<sup>2</sup> S <sup>o</sup> , <sup>2</sup> P <sup>o</sup> , <sup>2</sup> D <sup>o</sup> , <sup>2</sup> F <sup>o</sup> , <sup>2</sup> G <sup>o</sup> , <sup>2</sup> H <sup>o</sup>	<sup>2</sup> D
5s <sup>2</sup> 5p <sup>4</sup> nf			<sup>2</sup> F <sup>o</sup>

TABLE 6. Odd terms of I I

Electron configuration	Term symbol	Level	$\Delta\nu$	Observed <i>g</i>	Electron configuration	Term symbol	Level	$\Delta\nu$	Observed <i>g</i>
5s <sup>2</sup> 5p <sup>5</sup>	5p <sup>5</sup> <sup>2</sup> P <sub>1/2</sub>	0. 00	-7603. 15	0. 673	5s <sup>2</sup> 5p <sup>4</sup> ( <sup>3</sup> P) 6p	6p <sup>2</sup> D <sub>5/2</sub>	65644. 49	-7162. 72	1. 217
	<sup>2</sup> P <sub>3/2</sub>	7603. 15				<sup>2</sup> D <sub>3/2</sub>	72807. 21		
5s <sup>2</sup> 5p <sup>4</sup> ( <sup>3</sup> P) 6p	6p <sup>4</sup> P <sub>3/2</sub>	64906. 34	-2155. 78	1. 524	5s <sup>2</sup> 5p <sup>4</sup> ( <sup>3</sup> P) 6p	6p <sup>4</sup> D <sub>3/2</sub>	65670. 00	-6859. 17	1. 420
	<sup>4</sup> P <sub>1/2</sub>	67062. 12		1. 415		<sup>4</sup> D <sub>5/2</sub>	72529. 17	552. 43	1. 370
5s <sup>2</sup> 5p <sup>4</sup> ( <sup>3</sup> P) 6p	<sup>4</sup> P <sub>5/2</sub>	65856. 96	1205. 16	1. 556	6p <sup>4</sup> S <sub>1/2</sub>	71976. 74		-1410. 44	1. 317
	6p <sup>4</sup> S <sub>1/2</sub>	64990. 01		1. 619		<sup>4</sup> D <sub>3/2</sub>	73387. 18		1. 137

TABLE 6. Odd terms of Ir—Continued

Electron configuration	Term symbol	Level	$\Delta\nu$	Observed $g$	Electron configuration	Term symbol	Level	$\Delta\nu$	Observed $g$
$5s^2 5p^4(^3P)6p$	$6p\ ^2P_{0\frac{1}{2}}$	71501.52	1553.04	1.239	$5s^2 5p^4(^3P)8p$	$8p\ ^4P_{2\frac{1}{2}}$	78732.31	−492.47	1.454 1.30
	$2P_{1\frac{1}{2}}$	73054.56		1.329		$4P_{1\frac{1}{2}}$	79224.78		
$5s^2 5p^4(^3P)6p$	$6p\ ^2S_{0\frac{1}{2}}$	71813.97	1.377	−655.64	$5s^2 5p^4(^3P)8p$	$4P_{0\frac{1}{2}}$	79203.11	21.67	1.71
	$np\ 1P_{1\frac{1}{2}}$	72875.75				$8p\ ^4S_{1\frac{1}{2}}$	78815.61		
$5s^2 5p^4(^3P)7p$	$7p\ ^4P_{2\frac{1}{2}}$	74965.77	318.28	1.472	$5s^2 5p^4(^3P)8p$	$8p\ ^2D_{2\frac{1}{2}}$	78854.86	1.11	1.37 1.02
	$4P_{1\frac{1}{2}}$	75621.41		1.483	$5s^2 5p^4(^3P)8p$	$8p\ ^4D_{3\frac{1}{2}}$	79003.70		
	$4P_{0\frac{1}{2}}$	75303.13		1.53	$5s^2 5p^4(^3P)8p$	$8p\ ^2P_{0\frac{1}{2}}$	79701.73		
$5s^2 5p^4(^3P)7p$	$7p\ ^4S_{1\frac{1}{2}}$	75049.57		1.506	$5s^2 5p^4(^3P)5f$	$5f\ 1P_{1\frac{1}{2}}$	79835.03		
$5s^2 5p^4(^3P)7p$	$7p\ ^2D_{2\frac{1}{2}}$	75191.37	−7232.81	1.24		$5f\ 2P_{0\frac{1}{2}}$	79840.23		
	$2D_{1\frac{1}{2}}$	82424.18		1.27		$5f\ 3P_{2\frac{1}{2}}$	79844.58		
$5s^2 5p^4(^3P)7p$	$7p\ ^4D_{3\frac{1}{2}}$	75194.10	310.82	1.42		$5f\ 4P_{3\frac{1}{2}}$	79847.88		
	$4D_{2\frac{1}{2}}$	82033.80		1.39		$5f\ 4.1P_{3\frac{1}{2}}$	79853.40		
	$4D_{1\frac{1}{2}}$	81722.98		1.39		$5f\ 4.2P_{3\frac{1}{2}}$	79865.20		
	$4D_{0\frac{1}{2}}$	...		1.39		$5f\ 5P_{1\frac{1}{2}}$	79881.74		
$5s^2 5p^4(^3P)7p$	$7p\ ^2P_{0\frac{1}{2}}$	78780.04	−3651.16	1.48	$5s^2 5p^4(^3P)9p$	$np\ 3P_{1\frac{1}{2}}$	80039.94	1.32	80125.57
	$2P_{1\frac{1}{2}}$	82431.20		1.48	$5s^2 5p^4(^3P)9p$	$np\ 4P_{2\frac{1}{2}}$	80624.45		
$5s^2 5p^4(^3P)7p$	$7p\ ^2S_{0\frac{1}{2}}$	81506.80	582.54	1.48	$5s^2 5p^4(^3P)6f$	$9p\ ^4P_{3\frac{1}{2}}$	80720.85	80945.44	81205.39 81207.32 81216.75 81219.30 81226.02 81227.38
	$4f\ 0P_{1\frac{1}{2}}$	77297.15				$6f\ 1P_{1\frac{1}{2}}$	81205.39		
	$4f\ 0.1P_{0\frac{1}{2}}$	77303.58				$6f\ 2P_{0\frac{1}{2}}$	81207.32		
	$4f\ 1P_{3\frac{1}{2}}$	77307.47				$6f\ 3P_{2\frac{1}{2}}$	81216.75		
	$4f\ 2P_{3\frac{1}{2}}$	77313.12				$6f\ 4P_{3\frac{1}{2}}$	81219.30		
	$4f\ 2.1P_{3\frac{1}{2}}$	77313.76				$6f\ 5P_{4\frac{1}{2}}$	81226.02		
	$4f\ 3P_{3\frac{1}{2}}$	77320.66				$6f\ 6P_{5\frac{1}{2}}$	81227.38		
	$4f\ 4P_{1\frac{1}{2}}$	77356.76							
	$4f\ 5P_{3\frac{1}{2}}$	77359.62							
$5s^2 5p^4(^3P)4J$	$4f\ 5.1P_{1\frac{1}{2}}$	77404.49	−3651.16	1.48	$5s^2 5p^4(^3P)10p$	$10p\ ^4P_{2\frac{1}{2}}$	81693.55	82026.20 82030.35	81697.54 82035.00
	$4f\ 6P_{1\frac{1}{2}}$	77406.86				$10p\ ^4D_{3\frac{1}{2}}$	82026.20		
$5s^2 5p^4(^1D)6p$	$6p'\ ^2P_{1\frac{1}{2}}$	78415.36			$5s^2 5p^4(^3P)7f$	$7f\ 1P_{1\frac{1}{2}}$	82030.35		
$5s^2 5p^4(^3P)6s$	$np\ 1.1P_{0\frac{1}{2}}$	78535.48			$5s^2 5p^4(^3P)7f$	$7f\ 2P_{0\frac{1}{2}}$	82214.04		
$5s^2 5p^4(^3P)6s$	$np\ 2P_{1\frac{1}{2}}$	78592.75		1.00	$5s^2 5p^4(^3P)7s$	$np\ 6P_{2\frac{1}{2}}$	82615.84		

TABLE 7. Even terms of Ir

Electron configuration	Term symbol	Level	$\Delta\nu$	Observed $g$	Electron configuration	Term symbol	Level	$\Delta\nu$	Observed $g$
$5s^2 5p^4(^3P)6s$	$6s\ ^4P_{2\frac{1}{2}}$	54633.46	−7186.35	1.576	$5s^2 5p^4(^1S)6s$	$6s''\ ^2S_{0\frac{1}{2}}$	83048.47	−6987.72	+475.50
	$4P_{1\frac{1}{2}}$	61819.81		1.618	$5s^2 5p^4(^3P)7s$	$7s\ ^4P_{2\frac{1}{2}}$	71903.44		
$5s^2 5p^4(^3P)6s$	$4P_{0\frac{1}{2}}$	60896.27	+923.54	2.561		$4P_{1\frac{1}{2}}$	78891.16	−6990.43	1.454 1.30
	$6s\ ^2P_{1\frac{1}{2}}$	56092.88		1.385		$4P_{0\frac{1}{2}}$	78415.66		
$5s^2 5p^4(^1D)6s$	$2P_{0\frac{1}{2}}$	63186.76	−7093.88	0.799	$5s^2 5p^4(^3P)7s$	$7s\ ^2P_{1\frac{1}{2}}$	72294.83	−492.47	21.67
	$6s'\ ^2D_{2\frac{1}{2}}$	66020.64		1.258		$2P_{0\frac{1}{2}}$	79285.26		
$5s^2 5p^4(^1D)6s$	$2D_{1\frac{1}{2}}$	66355.21	−334.57	0.828					

TABLE 7. Even terms of II—Continued

Electron configuration	Term symbol	Level	$\Delta\nu$	Observed $g$	Electron configuration	Term symbol	Level	$\Delta\nu$	Observed $g$
$5s^2\ 5p^4(^3P)8s$	$\begin{cases} 8s\ ^4P_{2\frac{1}{2}} \\ 2P_{1\frac{1}{2}} \end{cases}$	77450. 76 77555. 77			$5s^2\ 5p^4(^3P)nd$	$\begin{cases} nd\ 11_{1\frac{1}{2}} \\ nd\ 13_{2\frac{1}{2}} \\ nd\ 14_{2\frac{1}{2}} \\ nd\ 15_{0\frac{1}{2}} \\ nd\ 16_{1\frac{1}{2}} \end{cases}$	74587. 40 74823. 48 75177. 26 75714. 44 75823. 98	0. 78 1. 26 0. 80 1. 41	
$5s^2\ 5p^4(^3P)9s$	$\begin{cases} 9s\ ^4P_{2\frac{1}{2}} \\ 2P_{1\frac{1}{2}} \end{cases}$	79914. 68 79947. 20							
$5s^2\ 5p^4(^3P)10s$	$\begin{cases} 10s\ ^4P_{2\frac{1}{2}} \\ 2P_{1\frac{1}{2}} \end{cases}$	81252. 48 ...	1. 53		$5s^2\ 5p^4(^3P)nd$	$\begin{cases} nd\ 17_{2\frac{1}{2}} \\ nd\ 18_{2\frac{1}{2}} \\ nd\ 19_{1\frac{1}{2}} \\ nd\ 19_{1\frac{1}{2}} \\ nd\ 20_{0\frac{1}{2}} \end{cases}$	75898. 91 76106. 57 76136. 43 76417. 78 76746. 98	1. 14 1. 23 1. 04	
$5s^2\ 5p^4(^3P)11s$	$\begin{cases} 11s\ ^4P_{2\frac{1}{2}} \\ 2P_{1\frac{1}{2}} \end{cases}$	82074. 10 ...							
$5s^2\ 5p^4(^3P)5d$	$\begin{cases} 5d\ ^4D_{3\frac{1}{2}} \\ ^4P_{2\frac{1}{2}} \\ ^4F_{4\frac{1}{2}} \\ 2F_{3\frac{1}{2}} \end{cases}$	75511. 13 ... 75704. 09 76004. 78	1. 15		$5s^2\ 5p^4(^3P)nd$	$\begin{cases} nd\ 21_{2\frac{1}{2}} \\ nd\ 21_{1\frac{3}{2}} \\ nd\ 21_{2\frac{1}{2}} \\ nd\ 22_{0\frac{1}{2}} \\ nd\ 23_{3\frac{1}{2}} \end{cases}$	76903. 28 76935. 98 78889. 50 79030. 99 79150. 55		1. 23
$5s^2\ 5p^4(^3P)6d$	$\begin{cases} 6d\ ^4D_{3\frac{1}{2}} \\ ^4P_{2\frac{1}{2}} \\ ^4F_{4\frac{1}{2}} \end{cases}$	78904. 05 78943. 37 79055. 01							
$5s^2\ 5p^4(^3P)7d$	$\begin{cases} 7d\ ^4D_{3\frac{1}{2}} \\ ^4P_{2\frac{1}{2}} \\ ^4F_{4\frac{1}{2}} \end{cases}$	80676. 17 80680. 12 80772. 35							
$5s^2\ 5p^4(^3P)8d$	$\begin{cases} 8d\ ^4D_{3\frac{1}{2}} \\ ^4P_{2\frac{1}{2}} \\ ^4F_{4\frac{1}{2}} \end{cases}$	81689. 02 81696. 06 81760. 58							
$5s^2\ 5p^4(^3P)9d$	$\begin{cases} 9d\ ^4D_{3\frac{1}{2}} \\ ^4P_{2\frac{1}{2}} \\ ^4F_{4\frac{1}{2}} \end{cases}$	82289. 46 ... 82374. 21							
$5s^2\ 5p^4(^3P)10d$	$\begin{cases} 10d\ ^4D_{3\frac{1}{2}} \\ ^4P_{2\frac{1}{2}} \\ ^4F_{4\frac{1}{2}} \end{cases}$	... 82818. 36							
$5s^2\ 5p^4(^3P)nd$	$\begin{cases} nd\ 1_{1\frac{1}{2}} \\ nd\ 2_{1\frac{1}{2}} \\ nd\ 3_{2\frac{1}{2}} \\ nd\ 4_{1\frac{1}{2}} \\ nd\ 4.1_{1\frac{1}{2}} \\ nd\ 5_{2\frac{1}{2}} \end{cases}$	67298. 45 68549. 77 68588. 00 68612. 02 68615. 84 70151. 32							
$5s^2\ 5p^4(^3P)nd$	$\begin{cases} nd\ 5.1_{1\frac{1}{2}} \\ nd\ 6_{1\frac{1}{2}} \\ nd\ 7_{1\frac{1}{2}} \\ nd\ 8_{1\frac{1}{2}} \\ nd\ 9_{2\frac{1}{2}} \\ nd\ 10_{3\frac{1}{2}} \end{cases}$	70354. 93 73114. 72 73477. 25 73639. 28 73795. 35 73977. 68	1. 21						

TABLE 8. Series of II

$n$	$ns\ ^4P_{2\frac{1}{2}}$		$np\ ^4P_{2\frac{1}{2}}$		$nd\ ^4F_{4\frac{1}{2}}$		$nf\ X_{3\frac{1}{2}}$	
	Current term	Rydberg denominator	Current term	Rydberg denominator	Current term	Rydberg denominator	Current term	Rydberg denominator
4							7019	3. 95402
5							4492	4. 94261
6	29707	1. 92197	19434	2. 37627	8636	3. 56468	3133	5. 91829
7	12437	2. 97043	9374	3. 42148	5285	4. 55674	2310	6. 89240
8	6889	3. 99116	5608	4. 42357	3568	5. 54580		
9	4425	4. 97989	3716	5. 43424	2579	6. 52305		
10	3088	5. 96126	2643	6. 44359	1955	7. 49209		
11	2266	6. 95900						
	$\alpha = -4.00389$		$\alpha = -3.51112$		$\alpha = -1.48320$		$\alpha = -0.11660$	
	$\beta = -2.52065$		$\beta = -3.98279$		$\beta = +1.15995$		$\beta = +1.16647$	

TABLE 9. *Continua in the spectrum of iodine*

Wave-lengths of maxima	Wave num- bers of maxima	Inten- sity	$\Delta\nu$	Remarks	Wave- lengths of maxima	Wave num- bers of maxima	Inten- sity	$\Delta\nu$	Remarks
4625	21616	1000		Band extends from 4820 to 4400 Å Minimum at 4360 Å	3142	31818	5	193	Minimum at 3136 Å
4285	23331	100	1715	Band extends from 4325 to 4240 Å	3123	32011	20	217	Minimum at 3114 Å
3525	28361	25	325	Minimum at 3510 Å	3080	32458	10	230	Minimum at 3091 Å
3485	28686	50	503	Minimum at 3475 Å Band extends from 3470 to 3380 Å	3062	32649	20	211	Minimum at 3074 Å
3425	29189	500	1290	Minimum at 3340 Å	3042	32864	25	215	Minimum at 3053 Å
3280	30479	5	206	Minimum at 3270 Å	3025	33048	15	184	Minimum at 3032 Å
3258	30685	10	228	Minimum at 3245 Å	3008	33235	20	234	Minimum at 2997 Å
3234	30913	25	211	Minimum at 3221 Å	2987	33469	20	214	Minimum at 2976 Å
3212	31124	50	205	Minimum at 3199 Å	2968	33683	10	182	Minimum at 2959 Å
3191	31329	25	208	Minimum at 3177 Å	2952	33861	15	173	Minimum at 2943 Å
3170	31537	10	281	Minimum at 3155 Å	2937	34038	5		

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