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A11101888558

/National Bureau of Standards circular
QC100 .U555 C467;V.2;1952 C.1 NBS-PUB-R

ATOMIC ENERGY LEVELS

As Derived From the Analyses of Optical Spectra

Volume II
24Cr—41Nb



CIRCULAR 467

UNITED STATES DEPARTMENT OF COMMERCE
NATIONAL BUREAU OF STANDARDS

UNITED STATES DEPARTMENT OF COMMERCE, Charles Sawyer, Secretary
NATIONAL BUREAU OF STANDARDS, A. V. Astin, Director

ATOMIC ENERGY LEVELS

As Derived From the Analyses of Optical Spectra

Volume II

The Spectra of Chromium, Manganese, Iron, Cobalt,
Nickel, Copper, Zinc, Gallium, Germanium, Arsenic,
Selenium, Bromine, Krypton, Rubidium, Strontium,
Yttrium, Zirconium, and Niobium

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Circular of the National Bureau of Standards 467

Issued August 15, 1952

National Bureau of Standards

SEP 2 1952

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Preface

The present volume is the second of a series being prepared at the National Bureau of Standards as part of a general program on the compilation of atomic energy levels derived from analyses of optical spectra. Since the publication in 1932 of "*Atomic Energy States as Derived from the Analyses of Optical Spectra*" by Robert F. Bacher and Samuel Goudsmit, the data on atomic spectra have accumulated in sufficient quantity to fill several volumes.

The first volume appeared in print June 15, 1949. That work contains the energy levels of 206 spectra of the first 23 elements, Hydrogen through Vanadium. The present volume includes similar data for 152 spectra of the elements having atomic numbers 24 through 41, Chromium through Niobium. The date of completion of the manuscript for each spectrum is entered at the end of the separate tables of energy levels. At present, structure is recognized in more than 500 spectra of 85 elements, so the program is still far from completion.

The general arrangement of Volumes I and II is identical. The adopted form of presentation is that recommended by the majority of interested workers who received a questionnaire proposed by the National Research Council Committee on Line Spectra of the Elements.

The manuscript has been prepared by Charlotte E. Moore under the direction of William F. Meggers, Chief of the Spectroscopy Section of the Atomic and Radiation Physics Division. A number of important gaps among the spectra of elements 24 to 41 have been filled, owing to the efforts of many spectroscopists to furnish data in advance of publication. The cordial cooperation of these individuals as well as that of National Research Council Committee as a whole, is greatly appreciated. This program was initiated while E. U. Condon was Director of the Bureau, and the present Volume was practically completed during his tenure.

A. V. ASTIN, *Director*.

WASHINGTON, D. C., June 30, 1952.

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1. Introduction

The present work is a continuation of a project started in 1946 at the National Bureau of Standards. Because Bacher and Goudsmit did not contemplate revising their extremely useful book on "*Atomic Energy States as Derived from the Analyses of Optical Spectra*" published in 1932, it was decided to prepare a critical compendium of Atomic

Energy Levels, in the Spectroscopy Section of this Bureau. The National Research Council Committee on Line Spectra of the Elements met at that time to discuss details, and this Committee has continued its enthusiastic support of the program.

2. Scope of the Present Tables

Requests to extend the scope of the tables are still being received. A critical compilation of data on X-ray spectra is needed.

No attempt has been made here to include data on hyperfine structure ascribed to atomic nuclei, except for H, D, and T. In Volume II reference is made under the individual spectra to the bibliography on this subject by Mack.¹ The earlier bibliography by Meggers² was mentioned in the references to relevant spectra in Volume I. In addition, selected papers are listed for a few spectra, particularly if they are not included in these two references summarizing hyperfine structure data. The present Volumes are, however, definitely not adequate for serious workers in this field.

Since the present tables include only the atomic energy levels derived from the analyses of optical spectra, the users must consult the separate references for lists of the lines observed in the various spectra, for line intensities, and for line classifications. To simplify this situation

the writer is preparing an "Ultraviolet Multiplet Table"³ of selected spectra, along with the tabulation of energy levels. For each Volume of Atomic Energy Levels (called AEL for brevity) there is a corresponding Section of the Ultraviolet Multiplet Table (abbreviated as UVMT in the text that follows). For example, Volume I (AEL) contains the energy levels of spectra of elements with atomic numbers 1 to 23, i. e., Hydrogen through Vanadium. Section 1 of the UVMT contains selected multiplets from spectra of these same elements. Similarly, just as the present Volume (AEL) covers the range of elements of atomic numbers 24 through 41, so will Section 2 of the UVMT contain multiplets of spectra of these same elements.

The UVMT, together with the "Revised Multiplet Table"⁴ (which is limited to lines of wavelength longer than 3000 Å), will to some extent fulfill the requests for tables of leading lines of selected spectra to accompany the energy levels.

3. Arrangement

The Introduction to Volume I (AEL) describes in detail the arrangement of the data, the notation, and the columns of the tables. Since Volume II is arranged in exactly the same style, the description need not be repeated here.† For convenience, the meaning of the letters in parentheses following the literature references is, however, repeated. These letters denote the following:

I P Ionization potential.
T Terms.

C L Classified lines.
G D Grotrian diagram.
Z E Zeeman effect.
I S Isotope shift.
hfs Hyperfine structure.

Briefly stated, these letters describe the scope and content of the paper in question. If no letters follow a reference, the paper is referred to in the remarks on the individual spectrum. For example, this is generally true of theoretical papers.

¹ J. E. Mack, *Rev. Mod. Phys.* **22**, No. 1, 64-76 (1950).

² W. F. Meggers, *J. Opt. Soc. Am.* **36**, 431 (1946).

† One minor change has been introduced for technical reasons. In Volume I the inner quantum number $J=\frac{1}{2}$ is entered as " $\frac{1}{2}$." In successive volumes this quantity is entered as " $0\frac{1}{2}$ " throughout.

³ C. E. Moore, *Circ. Nat. Bur. Std.* **488**; Section 1, H to V (1950); Section 2, Cr to Nb, in press (1952).

⁴ C. E. Moore, *Princeton Univ. Obs. Contr.* No. 20 (1945).

4. Tables of Predicted and Observed Arrays of Terms

As in Volume I, arrays of observed terms follow the individual tables of energy levels for the more complex spectra in cases where the analysis is not seriously in-

complete. Similar arrays of terms predicted from theory for spectra of the different isoelectronic sequences are included in Tables 1 to 17 (pages XII to XXVII) as follows:

Table	Sequence	Table	Sequence	Table	Sequence
1	Cr I	7	Zn I	13	Kr I
2	Mn I	8	Ga I	14	Sr I
3	Fe I	9	Ge I	15	Y I
4	Co I	10	As I	16	Zr I
5	Ni I	11	Se I	17	Nb I
6	Cu I	12	Br I		

These tables follow exactly the same pattern as was used for the sequences in Volume I.

For the Kr I sequence, as for Ne I and Ar I, the table gives both predicted terms (*LS*-coupling) and predicted pairs of levels (*jl*-coupling). The spectra of the inert-gas type have an interesting history. Meissner in 1919⁵ pointed out that well known series in Ne I converge to different limits. In the present notation, the separate limits, representing the two components of the limit term

⁵ K. W. Meissner, *Ann. der Phys.* [4] 58, 356 (1919).

of the ion are clearly indicated. The pair-coupling notation in the general form suggested by Racah has been adopted⁶ on the recommendation of Edlén, to take into account the departure from *LS*-coupling. Shortley has prepared an array of the theoretical arrangement of the pairs, for the writer to use as a guide in listing the energy levels of spectra of this type. This is discussed in Volume I. The Paschen notation is also retained in the table of levels of Kr I.

⁶ G. Racah, *Phys. Rev.* 61, 537 (L) (1942).

5. The Periodic Table

5.1 The Chemical Elements by Atomic Number; Ionization Potentials (Table 18)

In Volume I, Table 23, the ionization potentials and ground states were recorded for the first spectra throughout the periodic table. Some revisions of these data are given at the end of Volume II. Further revisions have resulted during the course of the work on Volume II. All ionization potentials in both Volumes as revised to date are entered in Table 18. Column one gives the atomic number *Z*; column two the chemical symbol of the element. Successive columns contain the ionization potentials of the atoms or ions in successive stages of ionization, I denoting first spectra (neutral atoms); II, second spectra (singly ionized atoms); etc.

Throughout these Volumes the ionization potentials

are derived by multiplying the limit in cm^{-1} by the factor 0.00012395, to express it in electron volts. This factor was recommended by Birge in 1941 (See Volume I, page IX.)

5.2 The Chemical Elements by Chemical Symbol (Table 19)

In Table 24 of Volume I the chemical elements were arranged in alphabetical order of their chemical symbols to facilitate cross reference to the book by Bacher and Goudsmit, "*Atomic Energy States*." This table is repeated in Volume II as Table 19, with elements 97 (Bk)⁷ and 98 (Cf)⁸ added, together with the changes recently adopted by the International Union of Chemistry.⁹ The additions and corrections are as follows:

Older			Corrected		
Symbol	Element	<i>Z</i>	Symbol	Element	<i>Z</i>
-----	-----		Bk	Berkelium	97
-----	-----		Cf	Californium	98
Fa	Francium	87	Fr	Francium	87
Lu	Lutecium	71	Lu	Lutetium	71
Cb	Columbium	41	Nb	Niobium	41
Pm	Prometheum	61	Pm	Promethium	61
W	Tungsten	74	W	Wolfram	74

⁷ S. G. Thompson, A. Ghiorso, and G. T. Seaborg, *Phys. Rev.* 77, 838 (L) (1950).

⁸ S. G. Thompson, K. Street, Jr., A. Ghiorso, and G. T. Seaborg, *Phys. Rev.* 80, 790 (1950).

⁹ See Report by E. Wichers, *J. Am. Chem. Soc.* 72, No. 4, 1431 (1950).

5.3 The Periodic System (Table 20)

For convenience, the Periodic System as given on page XLII of Volume I is repeated here as Table 20. Three changes have been made: The chemical symbol of element 41, formerly Cb is now Nb; elements 97 (Bk), and 98 (Cf) are added. The general arrangement is similar to that given on page 333 of "*The Theory of Atomic Spectra*" by Condon and Shortley.¹⁰

5.4 Index—Isoelectronic Sequences (Table 21)

This table contains the index to the data in Volume II, i. e., the spectra from Cr through Nb. The arrangement is identical with that of the corresponding Index, Table 26, in Volume I.

Column 1 contains the atomic number, followed by the chemical symbol. Across the top the successive stages of ionization are indicated, I denoting first spectra (neutral atoms), II second spectra (singly ionized atoms), III third spectra, etc. The numbers indicate the page on

¹⁰ E. U. Condon and G. H. Shortley, *The Theory of Atomic Spectra*, (The Macmillan Co., New York, N. Y., The University Press, Cambridge, Eng., Corrected edition, 1951).

which the data for the individual spectra may be found.

In this table, isoelectronic spectra appear on the diagonals. Alternate diagonals are printed in bold face type to emphasize the spectra of each sequence. Blanks occur for spectra in which structure has not yet been recognized.

No sequences are carried beyond Nb in this Volume, but they will be continued in later Volumes for spectra of elements with $Z > 41$. The sequences started but not completed in Volume II are as follows:

Sequence	Spectrum
Co I	Mo XVI, Pd XX—Sn XXIV
Cu I	Pd XVII—Sb XXXIII
Br I	Mo VIII
Kr I	Mo VII
Rb I	Mo VI
Sr I	Mo V
Y I	Mo IV
Zr I	Mo III
Nb I	Mo II

6. Acknowledgments

It was stated in the Introduction to Volume I that "Many scientific workers and many institutions at home and abroad are represented in this work." The same comment applies emphatically to Volume II. The inspiration to carry on such a project may be attributed directly to the large number of interested workers who have been supporting this program. At the National Bureau of Standards, E. U. Condon, during his tenure as Director, took a keen personal interest in the entire project. W. F. Meggers, Chief of the Spectroscopy Section, and Chairman of the Committee on Line Spectra, has spared no effort in giving expert advice on all difficult problems. He has carefully supervised the work and provided much spectroscopic material for inclusion here. Similarly, C. C. Kiess has arranged his entire schedule of work to meet the demands of this program. His splendid analyses of the very complex spectra Cr I and Cr II are lasting tributes illustrative of his interest in the work. The infrared observations of atomic spectra by C. J. Humphreys, Chief of the Infrared Spectroscopy Section, have already added important data, and promise to yield much more in the near future.

The writer has also received most valuable assistance from Princeton. H. N. Russell has been an active and interested consultant. Reference has been made to his work on binding energies, and in addition, he has read most of the manuscript. A. G. Shenstone has not only placed on his program the spectra most urgently needed to fill serious gaps in this Volume, but has also stimulated similar programs in other laboratories. For example, the third spectra of the elements Mn to Cu would have been seriously incomplete without his collaboration.

Three other members of the Committee on Line Spectra have also responded generously. J. E. Mack has provided unpublished data on Rb I. G. R. Harrison has supplied a number of workers with plates and films of Zeeman patterns observed at the Massachusetts Institute of Technology. W. E. Albertson has discussed some of the future problems, particularly in regard to rare earth spectra.

The international cooperation has also been gratifying. Edlén has read a large part of the manuscript, contributed valuable suggestions on a number of spectra in the iron group, and furnished unpublished data on high ionization spectra of the elements from Se through Zr, as well as on Kr I. M. A. Catalán has made two visits

to the United States in the interest of this program. His extensive work on Mn I and his study of the Mn III spectrum represent only a part of his contributions. Through the cordial collaboration of Harrison, Meggers, and Shenstone, he has also secured enough material to enable him and his staff in Madrid to continue this work.

Other analyses have been carried out especially for inclusion here. At Lehigh University C. W. Curtis has made an extensive study of Mn II, and N. E. Hager Jr., is extending the analysis of Co II. At Dartmouth and Cornell, F. L. Moore, Jr., has continued work on Cr III and Cr IV and expects to start Mn IV soon—all done in collaboration with Shenstone. Miss Dorothy W. Weeks of Wilson College, has furnished g -values for a number of spectra, with the aid of films from Harrison's laboratory. Unpublished material has also been received from R. A. Fisher (Fe I), S. Glad (Fe III), C. W. Gartlein (Ge II, As II), K. W. Meissner, K. L. Andrew (Ge I, Ge II), and F. W. Paul (Kr IX). Miss Olga García-Riquelme in Madrid (Mn); and L. Wilets (Cu III) and L. E. Gibson (Zn II), in Princeton, have also assisted.

The American Philosophical Society and the American Academy of Arts and Sciences have granted funds for work in connection with this program.

The users of this book owe a debt of gratitude to every one of these individuals and institutions for the material recorded in advance of publication.

The writer could never have prepared these Volumes without the services of experts in many fields of activity. Miss Sarah A. Jones, the Librarian at the Bureau, and her staff have continued their competent assistance with the many literature references. J. L. Mathusa and his staff in the Publications Section of the Bureau, have handled the difficult problems entailed in publishing this material with skill that has attracted the attention of many users. The personnel in the Government Printing Office have been equally cooperative. Mrs. Isabel D. Murray has provided much technical assistance in the preparation of the manuscript. Mrs. Marion W. Maddox has carefully typed the text accompanying the separate spectra.

The writer enjoys a lasting pleasure in recalling and recording the generous assistance she has received from all who have so willingly contributed to this compilation of Atomic Energy Levels.

Tables

Predicted Terms Cr I to Nb I

Ionization Potentials

The Periodic System

Index—Isoelectronic Sequence

TABLE 1. PREDICTED TERMS OF THE CI I ISOELECTRONIC SEQUENCE

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Predicted Terms			
	$ns (n \geq 4)$	$np (n \geq 4)$	$nd (n \geq 4)$
$3d^4 4s^2$	$\left\{ \begin{array}{l} {}^5D \\ {}^3P \quad {}^3D \\ {}^3P \end{array} \right.$ $\left\{ \begin{array}{l} {}^3G \quad {}^3H \\ {}^1G \\ {}^1G \end{array} \right.$ $\left\{ \begin{array}{l} {}^1I \\ {}^1I \end{array} \right.$			
$3d^6$	$\left\{ \begin{array}{l} {}^5D \\ {}^3P \quad {}^3D \\ {}^3P \end{array} \right.$ $\left\{ \begin{array}{l} {}^3G \quad {}^3H \\ {}^1G \\ {}^1G \end{array} \right.$ $\left\{ \begin{array}{l} {}^1I \\ {}^1I \end{array} \right.$			
$3d^5(6S)nx$	$\left\{ \begin{array}{l} {}^7S \\ {}^5S \end{array} \right.$	$\left\{ \begin{array}{l} {}^7P^o \\ {}^5P^o \end{array} \right.$	$\left\{ \begin{array}{l} {}^7D \\ {}^5D \end{array} \right.$	$\left\{ \begin{array}{l} {}^7D \\ {}^5D \end{array} \right.$
$3d^4 4s(6D)nx$		$\left\{ \begin{array}{l} {}^7P^o \\ {}^5P^o \end{array} \right.$	$\left\{ \begin{array}{l} {}^7D^o \\ {}^5D^o \end{array} \right.$	$\left\{ \begin{array}{l} {}^7F \\ {}^5F \end{array} \right.$
$3d^4 4s(4D)nx$		$\left\{ \begin{array}{l} {}^5P^o \\ {}^3P^o \end{array} \right.$	$\left\{ \begin{array}{l} {}^5D^o \\ {}^3D^o \end{array} \right.$	$\left\{ \begin{array}{l} {}^5F \\ {}^3F \end{array} \right.$
$3d^5(4G)nx$			$\left\{ \begin{array}{l} {}^5G \\ {}^3G \end{array} \right.$	$\left\{ \begin{array}{l} {}^5F \\ {}^3F \end{array} \right.$
$3d^5(4P)nx$	$\left\{ \begin{array}{l} {}^5P \\ {}^3P \end{array} \right.$	$\left\{ \begin{array}{l} {}^5P^o \\ {}^3P^o \end{array} \right.$	$\left\{ \begin{array}{l} {}^5D^o \\ {}^3D^o \end{array} \right.$	$\left\{ \begin{array}{l} {}^5P \\ {}^3P \end{array} \right.$
$3d^5(4D)nx$		$\left\{ \begin{array}{l} {}^5P^o \\ {}^3P^o \end{array} \right.$	$\left\{ \begin{array}{l} {}^5D^o \\ {}^3D^o \end{array} \right.$	$\left\{ \begin{array}{l} {}^5F \\ {}^3F \end{array} \right.$
$3d^4 4s(4P)nx$	$\left\{ \begin{array}{l} {}^5P \\ {}^3P \end{array} \right.$	$\left\{ \begin{array}{l} {}^5P^o \\ {}^3P^o \end{array} \right.$	$\left\{ \begin{array}{l} {}^5D^o \\ {}^3D^o \end{array} \right.$	$\left\{ \begin{array}{l} {}^5P \\ {}^3P \end{array} \right.$
$3d^5(2I)nx$		$\left\{ \begin{array}{l} {}^3H^o \\ {}^1H^o \end{array} \right.$	$\left\{ \begin{array}{l} {}^3K^o \\ {}^1K^o \end{array} \right.$	$\left\{ \begin{array}{l} {}^3G \\ {}^1G \end{array} \right.$
$3d^4 4s(4H)nx$		$\left\{ \begin{array}{l} {}^5H^o \\ {}^3H^o \end{array} \right.$	$\left\{ \begin{array}{l} {}^5F \\ {}^3F \end{array} \right.$	$\left\{ \begin{array}{l} {}^5K \\ {}^3K \end{array} \right.$
$3d^4 4s(4F)nx$		$\left\{ \begin{array}{l} {}^5D^o \\ {}^3D^o \end{array} \right.$	$\left\{ \begin{array}{l} {}^5G^o \\ {}^3G^o \end{array} \right.$	$\left\{ \begin{array}{l} {}^5F \\ {}^3F \end{array} \right.$
$3d^5(2D)nx$	$\left\{ \begin{array}{l} {}^3D \\ {}^1D \end{array} \right.$	$\left\{ \begin{array}{l} {}^3P^o \\ {}^1P^o \end{array} \right.$	$\left\{ \begin{array}{l} {}^3F^o \\ {}^1F^o \end{array} \right.$	$\left\{ \begin{array}{l} {}^3P \\ {}^1P \end{array} \right.$
$3d^5(2F)nx$		$\left\{ \begin{array}{l} {}^3D^o \\ {}^1D^o \end{array} \right.$	$\left\{ \begin{array}{l} {}^3G^o \\ {}^1G^o \end{array} \right.$	$\left\{ \begin{array}{l} {}^3P \\ {}^1P \end{array} \right.$

$3d^5(4F)nx$	{	$5F$ $3F$	$5D^\circ$ $3D^\circ$	$5F^\circ$ $3F^\circ$	$5G^\circ$ $3G^\circ$	$5P$ $3P$	$5D$ $3D$	$5F$ $3F$	$5G$ $3G$	$5H$ $3H$
$3d^4 4s(4G)nx$	{			$5F^\circ$ $3F^\circ$	$5G^\circ$ $3G^\circ$		$5D$ $3D$	$5F$ $3F$	$5G$ $3G$	$5H$ $3H$
$3d^4 4s(2H)nx$	{				$3G^\circ$ $1G^\circ$			$3F$ $1F$	$3G$ $1G$	$3H$ $1H$
$3d^4 4s(2F)nx$	{	$3P$ $1P$	$3D^\circ$ $1D^\circ$				$3P$ $1P$	$3D$ $1D$	$3F$ $1F$	$3G$ $1G$
$3d^5(2F)nx$	{		$3D^\circ$ $1D^\circ$	$3F^\circ$ $1F^\circ$	$3G^\circ$ $1G^\circ$		$3P$ $1P$	$3D$ $1D$	$3F$ $1F$	$3G$ $1G$
$3d^4 4s(2G)nx$	{				$3G^\circ$ $1G^\circ$			$3D$ $1D$	$3G$ $1G$	$3H$ $1H$
$3d^4 4s(4D)nx$	{	$5D$ $3D$	$5P^\circ$ $3P^\circ$	$5F^\circ$ $3F^\circ$			$5S$ $3S$	$5D$ $3D$	$5F$ $3F$	$5G$ $3G$
$3d^5(2G)nx$	{			$3F^\circ$ $1F^\circ$	$3G^\circ$ $1G^\circ$		$3D$ $1D$	$3F$ $1F$	$3G$ $1G$	$3H$ $1H$
$3d^4 4s(2G)nx$	{			$3F^\circ$ $1F^\circ$	$3G^\circ$ $1G^\circ$		$3D$ $1D$	$3F$ $1F$	$3G$ $1G$	$3H$ $1H$
$3d^4 4s(2F)nx$	{		$3D^\circ$ $1D^\circ$	$3F^\circ$ $1F^\circ$	$3G^\circ$ $1G^\circ$		$3P$ $1P$	$3D$ $1D$	$3F$ $1F$	$3G$ $1G$
$3d^4 4s(2I)nx$	{				$3H^\circ$ $1H^\circ$				$3G$ $1G$	$3H$ $1H$
$3d^5(2S)nx$	{ ^{3S} _{1S} }		$3P^\circ$ $1P^\circ$					$3D$ $1D$		
$3d^4 4s(2D)nx$	{		$3P^\circ$ $1P^\circ$	$3D^\circ$ $1D^\circ$	$3F^\circ$ $1F^\circ$		$3S$ $1S$	$3P$ $1P$	$3D$ $1D$	$3F$ $1F$
$3d^4 4s(2S)nx$	{ ^{3S} _{1S} }		$3P^\circ$ $1P^\circ$					$3D$ $1D$		
$3d^5(2D)n$	{		$3P^\circ$ $1P^\circ$	$3D^\circ$ $1D^\circ$	$3F^\circ$ $1F^\circ$		$3S$ $1S$	$3P$ $1P$	$3D$ $1D$	$3F$ $1F$
$3d^4 4s(2D)nx$	{		$3P^\circ$ $1P^\circ$	$3D^\circ$ $1D^\circ$	$3F^\circ$ $1F^\circ$		$3S$ $1S$	$3P$ $1P$	$3D$ $1D$	$3F$ $1F$
$3d^5(2G)nx$	{			$3F^\circ$ $1F^\circ$	$3G^\circ$ $1G^\circ$				$3G$ $1G$	$3H$ $1H$

TABLE 2. PREDICTED TERMS OF THE Mn I ISOELECTRONIC SEQUENCE*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Predicted Terms			
	$ns (n \geq 4)$	$np (n \geq 4)$	$nd (n \geq 4)$	$nd (n \geq 4)$
$3d^5 4s^2$	$\left\{ \begin{array}{l} {}^6S \\ {}^2S \end{array} \right\}$ $\left\{ \begin{array}{l} {}^4P \\ {}^2P \end{array} \right\}$ $\left\{ \begin{array}{l} {}^4D \\ {}^2D \\ {}^2D \end{array} \right\}$ $\left\{ \begin{array}{l} {}^4F \\ {}^2F \\ {}^2F \end{array} \right\}$ $\left\{ \begin{array}{l} {}^4G \\ {}^2G \\ {}^2G \end{array} \right\}$ $\left\{ \begin{array}{l} {}^2H \\ {}^2H \end{array} \right\}$ $\left\{ \begin{array}{l} {}^2I \end{array} \right\}$			
$3d^7$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \end{array} \right\}$ $\left\{ \begin{array}{l} {}^4D \\ {}^2D \\ {}^2D \end{array} \right\}$ $\left\{ \begin{array}{l} {}^4F \\ {}^2F \\ {}^2F \end{array} \right\}$ $\left\{ \begin{array}{l} {}^4G \\ {}^2G \\ {}^2G \end{array} \right\}$ $\left\{ \begin{array}{l} {}^2H \\ {}^2H \end{array} \right\}$			
$3d^5 4p^2 \dagger$	$\left\{ \begin{array}{l} {}^8S \\ {}^6S \\ {}^4S \end{array} \right\}$ $\left\{ \begin{array}{l} {}^6P \\ {}^4P \end{array} \right\}$ $\left\{ \begin{array}{l} {}^6D \\ {}^4D \end{array} \right\}$ $\left\{ \begin{array}{l} {}^6F \\ {}^4F \end{array} \right\}$ $\left\{ \begin{array}{l} {}^6G \\ {}^4G \end{array} \right\}$			
$3d^5 4s(7S)nc$	$\left\{ \begin{array}{l} {}^6S \\ {}^4S \end{array} \right\}$	$\left\{ \begin{array}{l} {}^8P^o \\ {}^6P^o \end{array} \right\}$ $\left\{ \begin{array}{l} {}^6P^o \\ {}^4P^o \end{array} \right\}$ $\left\{ \begin{array}{l} {}^6D^o \\ {}^4D^o \end{array} \right\}$ $\left\{ \begin{array}{l} {}^6F^o \\ {}^4F^o \end{array} \right\}$ $\left\{ \begin{array}{l} {}^6G^o \\ {}^4G^o \end{array} \right\}$		$\left\{ \begin{array}{l} {}^8D \\ {}^6D \end{array} \right\}$ $\left\{ \begin{array}{l} {}^6D \\ {}^4D \end{array} \right\}$ $\left\{ \begin{array}{l} {}^6P \\ {}^4P \end{array} \right\}$ $\left\{ \begin{array}{l} {}^6D \\ {}^4D \end{array} \right\}$ $\left\{ \begin{array}{l} {}^6F \\ {}^4F \end{array} \right\}$ $\left\{ \begin{array}{l} {}^6G \\ {}^4G \end{array} \right\}$
$3d^5 4s(6S)nc$	$\left\{ \begin{array}{l} {}^6S \\ {}^4S \end{array} \right\}$	$\left\{ \begin{array}{l} {}^6P^o \\ {}^4P^o \end{array} \right\}$ $\left\{ \begin{array}{l} {}^6D^o \\ {}^4D^o \end{array} \right\}$ $\left\{ \begin{array}{l} {}^6F^o \\ {}^4F^o \end{array} \right\}$ $\left\{ \begin{array}{l} {}^6G^o \\ {}^4G^o \end{array} \right\}$		$\left\{ \begin{array}{l} {}^6D \\ {}^4D \end{array} \right\}$ $\left\{ \begin{array}{l} {}^6P \\ {}^4P \end{array} \right\}$ $\left\{ \begin{array}{l} {}^6D \\ {}^4D \end{array} \right\}$ $\left\{ \begin{array}{l} {}^6F \\ {}^4F \end{array} \right\}$ $\left\{ \begin{array}{l} {}^6G \\ {}^4G \end{array} \right\}$
$3d^5 (5D)nc$				
$3d^5 4s(6G)nc$				
$3d^5 4s(6P)nc$				
$3d^5 4s(6D)nc$				
$3d^5 4s(6G)nc$				
$3d^5 4s(6P)nc$				
$3d^5 4s(6D)nc$				
$3d^5 4s(6F)nc$				
$3d^5 4s(6F)nc$				

$3d^5 4s(3I)nx$		$\begin{Bmatrix} 4I \\ 2I \end{Bmatrix}$	$\begin{matrix} 4H^\circ & 4I^\circ & 4K^\circ \\ 2H^\circ & 2I^\circ & 2K^\circ \end{matrix}$	$\begin{matrix} 4G \\ 2G \end{matrix}$	$\begin{matrix} 4H \\ 2H \end{matrix}$	$\begin{matrix} 4I \\ 2I \end{matrix}$	$\begin{matrix} 4K \\ 2K \end{matrix}$	$\begin{matrix} 4L \\ 2L \end{matrix}$
$3d^6(3P)nx$	$\begin{Bmatrix} 4P \\ 2P \end{Bmatrix}$		$\begin{matrix} 4S^\circ & 4P^\circ & 4D^\circ \\ 2S^\circ & 2P^\circ & 2D^\circ \end{matrix}$	$\begin{matrix} 4H^\circ & 4I^\circ & 4K^\circ \\ 2H^\circ & 2I^\circ & 2K^\circ \end{matrix}$	$\begin{matrix} 4G \\ 2G \end{matrix}$	$\begin{matrix} 4H \\ 2H \end{matrix}$	$\begin{matrix} 4I \\ 2I \end{matrix}$	$\begin{matrix} 4K \\ 2K \end{matrix}$
$3d^6(3H)nx$		$\begin{matrix} 4H \\ 2H \end{matrix}$		$\begin{matrix} 4G^\circ & 4H^\circ & 4I^\circ \\ 2G^\circ & 2H^\circ & 2I^\circ \end{matrix}$	$\begin{matrix} 4G \\ 2G \end{matrix}$	$\begin{matrix} 4H \\ 2H \end{matrix}$	$\begin{matrix} 4I \\ 2I \end{matrix}$	$\begin{matrix} 4K \\ 2K \end{matrix}$
$3d^6(3F)nx$		$\begin{matrix} 4F \\ 2F \end{matrix}$		$\begin{matrix} 4G^\circ & 4F^\circ & 4D^\circ \\ 2G^\circ & 2F^\circ & 2D^\circ \end{matrix}$	$\begin{matrix} 4G \\ 2G \end{matrix}$	$\begin{matrix} 4H \\ 2H \end{matrix}$	$\begin{matrix} 4I \\ 2I \end{matrix}$	$\begin{matrix} 4K \\ 2K \end{matrix}$
$3d^6(3G)nx$		$\begin{matrix} 4G \\ 2G \end{matrix}$		$\begin{matrix} 4F^\circ & 4G^\circ & 4H^\circ \\ 2F^\circ & 2G^\circ & 2H^\circ \end{matrix}$	$\begin{matrix} 4G \\ 2G \end{matrix}$	$\begin{matrix} 4H \\ 2H \end{matrix}$	$\begin{matrix} 4I \\ 2I \end{matrix}$	$\begin{matrix} 4K \\ 2K \end{matrix}$
$3d^6(1I)nx$		$\begin{matrix} 2I \end{matrix}$		$\begin{matrix} 2H^\circ & 2I^\circ & 2K^\circ \end{matrix}$	$\begin{matrix} 2G \\ 2G \end{matrix}$	$\begin{matrix} 2H \\ 2H \end{matrix}$	$\begin{matrix} 2K \\ 2K \end{matrix}$	$\begin{matrix} 2L \\ 2L \end{matrix}$
$3d^6(3D)nx$	$\begin{Bmatrix} 4D \\ 2D \end{Bmatrix}$		$\begin{matrix} 4P^\circ & 4D^\circ & 4F^\circ \\ 2P^\circ & 2D^\circ & 2F^\circ \end{matrix}$	$\begin{matrix} 4S^\circ & 4P^\circ & 4D^\circ \\ 2S^\circ & 2P^\circ & 2D^\circ \end{matrix}$	$\begin{matrix} 4G \\ 2G \end{matrix}$	$\begin{matrix} 4H \\ 2H \end{matrix}$	$\begin{matrix} 4I \\ 2I \end{matrix}$	$\begin{matrix} 4K \\ 2K \end{matrix}$
$3d^6(1G)nx$	$\begin{matrix} 2S \end{matrix}$	$\begin{matrix} 2G \end{matrix}$		$\begin{matrix} 2F^\circ & 2G^\circ & 2H^\circ \end{matrix}$	$\begin{matrix} 2G \\ 2G \end{matrix}$	$\begin{matrix} 2H \\ 2H \end{matrix}$	$\begin{matrix} 2I \\ 2I \end{matrix}$	$\begin{matrix} 2K \\ 2K \end{matrix}$
$3d^6(1S)nx$								
$3d^6(1D)nx$		$\begin{matrix} 2D \end{matrix}$						
$3d^6(1F)nx$		$\begin{matrix} 2F \end{matrix}$						
$3d^6(3P)nx$	$\begin{Bmatrix} 4P \\ 2P \end{Bmatrix}$		$\begin{matrix} 4S^\circ & 4P^\circ & 4D^\circ \\ 2S^\circ & 2P^\circ & 2D^\circ \end{matrix}$	$\begin{matrix} 4G^\circ & 4F^\circ & 4D^\circ \\ 2G^\circ & 2F^\circ & 2D^\circ \end{matrix}$	$\begin{matrix} 4G \\ 2G \end{matrix}$	$\begin{matrix} 4H \\ 2H \end{matrix}$	$\begin{matrix} 4I \\ 2I \end{matrix}$	$\begin{matrix} 4K \\ 2K \end{matrix}$
$3d^6(3F)nx$		$\begin{matrix} 4F \\ 2F \end{matrix}$		$\begin{matrix} 4G^\circ & 4F^\circ & 4D^\circ \\ 2G^\circ & 2F^\circ & 2D^\circ \end{matrix}$	$\begin{matrix} 4G \\ 2G \end{matrix}$	$\begin{matrix} 4H \\ 2H \end{matrix}$	$\begin{matrix} 4I \\ 2I \end{matrix}$	$\begin{matrix} 4K \\ 2K \end{matrix}$
$3d^4 4s^2(5D)nx$	$\begin{Bmatrix} 6D \\ 4D \end{Bmatrix}$		$\begin{matrix} 6P^\circ & 6D^\circ & 6F^\circ \\ 4P^\circ & 4D^\circ & 4F^\circ \end{matrix}$	$\begin{matrix} 6S^\circ & 6P^\circ & 6D^\circ \\ 4S^\circ & 4P^\circ & 4D^\circ \end{matrix}$	$\begin{matrix} 6G \\ 4G \end{matrix}$	$\begin{matrix} 6H \\ 4H \end{matrix}$	$\begin{matrix} 6I \\ 4I \end{matrix}$	$\begin{matrix} 6K \\ 4K \end{matrix}$
$3d^5 4s(7S)nx$	$\begin{Bmatrix} 8F^\circ \\ 6F^\circ \end{Bmatrix}$	$ng (n \geq 4)$	$ng (n \geq 5)$	$\begin{matrix} 8G \\ 6G \end{matrix}$	$\begin{matrix} 8G \\ 6G \end{matrix}$	$\begin{matrix} 8I \\ 6I \end{matrix}$	$\begin{matrix} 8K \\ 6K \end{matrix}$	$\begin{matrix} 8L \\ 6L \end{matrix}$
$3d^5 4s(5S)nx$	$\begin{Bmatrix} 6F^\circ \\ 4F^\circ \end{Bmatrix}$	$ng (n \geq 4)$	$ng (n \geq 5)$	$\begin{matrix} 6G \\ 4G \end{matrix}$	$\begin{matrix} 6G \\ 4G \end{matrix}$	$\begin{matrix} 6I \\ 4I \end{matrix}$	$\begin{matrix} 6K \\ 4K \end{matrix}$	$\begin{matrix} 6L \\ 4L \end{matrix}$
$3d^6(5D)nx$	$\begin{Bmatrix} 6P^\circ & 6D^\circ \\ 4P^\circ & 4D^\circ \end{Bmatrix}$	$ng (n \geq 4)$	$ng (n \geq 5)$	$\begin{matrix} 6D^\circ & 6F^\circ & 6H^\circ \\ 4D^\circ & 4F^\circ & 4H^\circ \end{matrix}$	$\begin{matrix} 6G \\ 4G \end{matrix}$	$\begin{matrix} 6H \\ 4H \end{matrix}$	$\begin{matrix} 6I \\ 4I \end{matrix}$	$\begin{matrix} 6K \\ 4K \end{matrix}$

*The general arrangement of the limit terms in this table is in order of increasing value of the terms of Mn II, where the terms from the configuration $3d^5 4s$ are low, followed by the terms from $3d^6$, which are low in Fe III, listed in order of increasing value in this spectrum.

†Incomplete, only the limit $6S$ considered.

TABLE 3. PREDICTED TERMS OF THE Fe I ISOELECTRONIC SEQUENCE

Config. $1s^2 2s^2 2p^6$ $3s^2 3p^6 +$	Predicted Terms				
$3d^6 4s^2$	$\begin{cases} {}^5D \\ {}^3P \\ {}^3D \\ {}^3F \\ {}^3F \\ {}^3G \\ {}^3H \\ {}^1S \\ {}^1S \\ {}^1D \\ {}^1F \\ {}^1G \\ {}^1G \\ {}^1I \end{cases}$				
$3d^8$	$\begin{cases} {}^3P \\ {}^1S \\ {}^1D \\ {}^3F \\ {}^1G \end{cases}$				
	$ns (n \geq 4)$	$np (n \geq 4)$	$nd (n \geq 4)$	
$3d^6 4s({}^6D)nx$	$\begin{cases} {}^7D \\ {}^5D \end{cases}$	$\begin{cases} {}^7P^\circ & {}^7D^\circ & {}^7F^\circ \\ {}^5P^\circ & {}^5D^\circ & {}^5F^\circ \end{cases}$	$\begin{cases} {}^7S & {}^7P & {}^7D & {}^7F & {}^7G \\ {}^5S & {}^5P & {}^5D & {}^5F & {}^5G \end{cases}$	
$3d^7({}^4F)nx$	$\begin{cases} {}^5F \\ {}^3F \end{cases}$	$\begin{cases} {}^5D^\circ & {}^5F^\circ & {}^5G^\circ \\ {}^3D^\circ & {}^3F^\circ & {}^3G^\circ \end{cases}$	$\begin{cases} {}^5P & {}^5D & {}^5F & {}^5G & {}^5H \\ {}^3P & {}^3D & {}^3F & {}^3G & {}^3H \end{cases}$	
$3d^6 4s({}^4D)nx$	$\begin{cases} {}^5D \\ {}^3D \end{cases}$	$\begin{cases} {}^5P^\circ & {}^5D^\circ & {}^5F^\circ \\ {}^3P^\circ & {}^3D^\circ & {}^3F^\circ \end{cases}$	$\begin{cases} {}^5S & {}^5P & {}^5D & {}^5F & {}^5G \\ {}^3S & {}^3P & {}^3D & {}^3F & {}^3G \end{cases}$	
$3d^7({}^4P)nx$	$\begin{cases} {}^5P \\ {}^3P \end{cases}$	$\begin{cases} {}^5S^\circ & {}^5P^\circ & {}^5D^\circ \\ {}^3S^\circ & {}^3P^\circ & {}^3D^\circ \end{cases}$	$\begin{cases} {}^5P & {}^5D & {}^5F \\ {}^3P & {}^3D & {}^3F \end{cases}$	
$3d^7({}^2G)nx$	$\begin{cases} {}^3G \\ {}^1G \end{cases}$	$\begin{cases} {}^3F^\circ & {}^3G^\circ & {}^3H^\circ \\ {}^1F^\circ & {}^1G^\circ & {}^1H^\circ \end{cases}$	$\begin{cases} {}^3D & {}^3F & {}^3G & {}^3H & {}^3I \\ {}^1D & {}^1F & {}^1G & {}^1H & {}^1I \end{cases}$	
$3d^7({}^2P)nx$	$\begin{cases} {}^3P \\ {}^1P \end{cases}$	$\begin{cases} {}^3S^\circ & {}^3P^\circ & {}^3D^\circ \\ {}^1S^\circ & {}^1P^\circ & {}^1D^\circ \end{cases}$	$\begin{cases} {}^3P & {}^3D & {}^3F \\ {}^1P & {}^1D & {}^1F \end{cases}$	
$3d^7({}^2H)nx$	$\begin{cases} {}^3H \\ {}^1H \end{cases}$	$\begin{cases} {}^3G^\circ & {}^3H^\circ & {}^3I^\circ \\ {}^1G^\circ & {}^1H^\circ & {}^1I^\circ \end{cases}$	$\begin{cases} {}^3F & {}^3G & {}^3H & {}^3I & {}^3K \\ {}^1F & {}^1G & {}^1H & {}^1I & {}^1K \end{cases}$	
$3d^7({}^2D)nx$	$\begin{cases} {}^3D \\ {}^1D \end{cases}$	$\begin{cases} {}^3P^\circ & {}^3D^\circ & {}^3F^\circ \\ {}^1P^\circ & {}^1D^\circ & {}^1F^\circ \end{cases}$	$\begin{cases} {}^3S & {}^3P & {}^3D & {}^3F & {}^3G \\ {}^1S & {}^1P & {}^1D & {}^1F & {}^1G \end{cases}$	
$3d^6 4s({}^4P)nx$	$\begin{cases} {}^5P \\ {}^3P \end{cases}$	$\begin{cases} {}^5S^\circ & {}^5P^\circ & {}^5D^\circ \\ {}^3S^\circ & {}^3P^\circ & {}^3D^\circ \end{cases}$	$\begin{cases} {}^5P & {}^5D & {}^5F \\ {}^3P & {}^3D & {}^3F \end{cases}$	
$3d^6 4s({}^4H)nx$	$\begin{cases} {}^5H \\ {}^3H \end{cases}$	$\begin{cases} {}^5G^\circ & {}^5H^\circ & {}^5I^\circ \\ {}^3G^\circ & {}^3H^\circ & {}^3I^\circ \end{cases}$	$\begin{cases} {}^5F & {}^5G & {}^5H & {}^5I & {}^5K \\ {}^3F & {}^3G & {}^3H & {}^3I & {}^3K \end{cases}$	
$3d^6 4s({}^4F)nx$	$\begin{cases} {}^5F \\ {}^3F \end{cases}$	$\begin{cases} {}^5D^\circ & {}^5F^\circ & {}^5G^\circ \\ {}^3D^\circ & {}^3F^\circ & {}^3G^\circ \end{cases}$	$\begin{cases} {}^5P & {}^5D & {}^5F & {}^5G & {}^5H \\ {}^3P & {}^3D & {}^3F & {}^3G & {}^3H \end{cases}$	
$3d^5 4s^2({}^4S)nx$	$\begin{cases} {}^7S \\ {}^5S \end{cases}$	$\begin{cases} {}^7P^\circ \\ {}^5P^\circ \end{cases}$	$\begin{cases} {}^7D \\ {}^5D \end{cases}$	
$3d^6 4s({}^4G)nx$	$\begin{cases} {}^5G \\ {}^3G \end{cases}$	$\begin{cases} {}^5F^\circ & {}^5G^\circ & {}^5H^\circ \\ {}^3F^\circ & {}^3G^\circ & {}^3H^\circ \end{cases}$	$\begin{cases} {}^5D & {}^5F & {}^5G & {}^5H & {}^5I \\ {}^3D & {}^3F & {}^3G & {}^3H & {}^3I \end{cases}$	
$3d^6 4s({}^2P)nx$	$\begin{cases} {}^3P \\ {}^1P \end{cases}$	$\begin{cases} {}^3S^\circ & {}^3P^\circ & {}^3D^\circ \\ {}^1S^\circ & {}^1P^\circ & {}^1D^\circ \end{cases}$	$\begin{cases} {}^3P & {}^3D & {}^3F \\ {}^1P & {}^1D & {}^1F \end{cases}$	
$3d^6 4s({}^2H)nx$	$\begin{cases} {}^3H \\ {}^1H \end{cases}$	$\begin{cases} {}^3G^\circ & {}^3H^\circ & {}^3I^\circ \\ {}^1G^\circ & {}^1H^\circ & {}^1I^\circ \end{cases}$	$\begin{cases} {}^3F & {}^3G & {}^3H & {}^3I & {}^3K \\ {}^1F & {}^1G & {}^1H & {}^1I & {}^1K \end{cases}$	
$3d^6 4s({}^2F)nx$	$\begin{cases} {}^3F \\ {}^1F \end{cases}$	$\begin{cases} {}^3D^\circ & {}^3F^\circ & {}^3G^\circ \\ {}^1D^\circ & {}^1F^\circ & {}^1G^\circ \end{cases}$	$\begin{cases} {}^3P & {}^3D & {}^3F & {}^3G & {}^3H \\ {}^1P & {}^1D & {}^1F & {}^1G & {}^1H \end{cases}$	
$3d^6 4s({}^2G)nx$	$\begin{cases} {}^3G \\ {}^1G \end{cases}$	$\begin{cases} {}^3F^\circ & {}^3G^\circ & {}^3H^\circ \\ {}^1F^\circ & {}^1G^\circ & {}^1H^\circ \end{cases}$	$\begin{cases} {}^3D & {}^3F & {}^3G & {}^3H & {}^3I \\ {}^1D & {}^1F & {}^1G & {}^1H & {}^1I \end{cases}$	
$3d^6 4s({}^4D)nx$	$\begin{cases} {}^5D \\ {}^3D \end{cases}$	$\begin{cases} {}^5P^\circ & {}^5D^\circ & {}^5F^\circ \\ {}^3P^\circ & {}^3D^\circ & {}^3F^\circ \end{cases}$	$\begin{cases} {}^5S & {}^5P & {}^5D & {}^5F & {}^5G \\ {}^3S & {}^3P & {}^3D & {}^3F & {}^3G \end{cases}$	
$3d^7({}^2F)nx$	$\begin{cases} {}^3F \\ {}^1F \end{cases}$	$\begin{cases} {}^3D^\circ & {}^3F^\circ & {}^3G^\circ \\ {}^1D^\circ & {}^1F^\circ & {}^1G^\circ \end{cases}$	$\begin{cases} {}^3P & {}^3D & {}^3F & {}^3G & {}^3H \\ {}^1P & {}^1D & {}^1F & {}^1G & {}^1H \end{cases}$	

TABLE 4. PREDICTED TERMS OF THE Co I ISOELECTRONIC SEQUENCE*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Predicted Terms			
	$ns (n \geq 4)$	$np (n \geq 4)$	$nd (n \geq 4)$
$3d^7 4s^2$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \\ {}^2D \\ {}^2D \\ {}^2D \end{array} \right.$	$\begin{array}{l} {}^4F \\ {}^2F \\ {}^2G \\ {}^2H \end{array}$	$\begin{array}{l} {}^4F \\ {}^2F \\ {}^2G \\ {}^2H \end{array}$
$3d^6$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \\ {}^2D \\ {}^2D \end{array} \right.$	$\begin{array}{l} {}^4F \\ {}^2F \\ {}^2G \\ {}^2H \end{array}$	$\begin{array}{l} {}^4F \\ {}^2F \\ {}^2G \\ {}^2H \end{array}$
$3d^8 ({}^3F) nx$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \end{array} \right.$	$\begin{array}{l} {}^4S^{\circ} \\ {}^2S^{\circ} \end{array}$	$\begin{array}{l} {}^4D^{\circ} \\ {}^2D^{\circ} \end{array}$	$\begin{array}{l} {}^4P \\ {}^2P \\ {}^4D \\ {}^2D \\ {}^4F \\ {}^2F \\ {}^4G \\ {}^2G \end{array}$
$3d^8 ({}^3P) nx$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \end{array} \right.$	$\begin{array}{l} {}^4S^{\circ} \\ {}^2S^{\circ} \end{array}$	$\begin{array}{l} {}^4D^{\circ} \\ {}^2D^{\circ} \end{array}$	$\begin{array}{l} {}^4P \\ {}^2P \\ {}^4D \\ {}^2D \\ {}^4F \\ {}^2F \\ {}^4G \\ {}^2G \end{array}$
$3d^8 ({}^1G) nx$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \end{array} \right.$	$\begin{array}{l} {}^4S^{\circ} \\ {}^2S^{\circ} \end{array}$	$\begin{array}{l} {}^4D^{\circ} \\ {}^2D^{\circ} \end{array}$	$\begin{array}{l} {}^4P \\ {}^2P \\ {}^4D \\ {}^2D \\ {}^4F \\ {}^2F \\ {}^4G \\ {}^2G \end{array}$
$3d^8 ({}^1D) nx$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \end{array} \right.$	$\begin{array}{l} {}^4S^{\circ} \\ {}^2S^{\circ} \end{array}$	$\begin{array}{l} {}^4D^{\circ} \\ {}^2D^{\circ} \end{array}$	$\begin{array}{l} {}^4P \\ {}^2P \\ {}^4D \\ {}^2D \\ {}^4F \\ {}^2F \\ {}^4G \\ {}^2G \end{array}$
$3d^7 4s ({}^6F) nx$	$\left\{ \begin{array}{l} {}^6P \\ {}^4P \end{array} \right.$	$\begin{array}{l} {}^6S^{\circ} \\ {}^4S^{\circ} \end{array}$	$\begin{array}{l} {}^6D^{\circ} \\ {}^4D^{\circ} \end{array}$	$\begin{array}{l} {}^6P \\ {}^4P \\ {}^6D \\ {}^4D \\ {}^6F \\ {}^4F \\ {}^6G \\ {}^4G \end{array}$
$3d^7 4s ({}^6P) nx$	$\left\{ \begin{array}{l} {}^6P \\ {}^4P \end{array} \right.$	$\begin{array}{l} {}^6S^{\circ} \\ {}^4S^{\circ} \end{array}$	$\begin{array}{l} {}^6D^{\circ} \\ {}^4D^{\circ} \end{array}$	$\begin{array}{l} {}^6P \\ {}^4P \\ {}^6D \\ {}^4D \\ {}^6F \\ {}^4F \\ {}^6G \\ {}^4G \end{array}$
$3d^7 4s ({}^3F) nx$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \end{array} \right.$	$\begin{array}{l} {}^4S^{\circ} \\ {}^2S^{\circ} \end{array}$	$\begin{array}{l} {}^4D^{\circ} \\ {}^2D^{\circ} \end{array}$	$\begin{array}{l} {}^4P \\ {}^2P \\ {}^4D \\ {}^2D \\ {}^4F \\ {}^2F \\ {}^4G \\ {}^2G \end{array}$
$3d^7 4s ({}^3P) nx$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \end{array} \right.$	$\begin{array}{l} {}^4S^{\circ} \\ {}^2S^{\circ} \end{array}$	$\begin{array}{l} {}^4D^{\circ} \\ {}^2D^{\circ} \end{array}$	$\begin{array}{l} {}^4P \\ {}^2P \\ {}^4D \\ {}^2D \\ {}^4F \\ {}^2F \\ {}^4G \\ {}^2G \end{array}$
$3d^7 4s ({}^3H) nx$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \end{array} \right.$	$\begin{array}{l} {}^4S^{\circ} \\ {}^2S^{\circ} \end{array}$	$\begin{array}{l} {}^4D^{\circ} \\ {}^2D^{\circ} \end{array}$	$\begin{array}{l} {}^4P \\ {}^2P \\ {}^4D \\ {}^2D \\ {}^4F \\ {}^2F \\ {}^4G \\ {}^2G \end{array}$
$3d^7 4s ({}^3G) nx$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \end{array} \right.$	$\begin{array}{l} {}^4S^{\circ} \\ {}^2S^{\circ} \end{array}$	$\begin{array}{l} {}^4D^{\circ} \\ {}^2D^{\circ} \end{array}$	$\begin{array}{l} {}^4P \\ {}^2P \\ {}^4D \\ {}^2D \\ {}^4F \\ {}^2F \\ {}^4G \\ {}^2G \end{array}$
$3d^7 4s ({}^3F) nx$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \end{array} \right.$	$\begin{array}{l} {}^4S^{\circ} \\ {}^2S^{\circ} \end{array}$	$\begin{array}{l} {}^4D^{\circ} \\ {}^2D^{\circ} \end{array}$	$\begin{array}{l} {}^4P \\ {}^2P \\ {}^4D \\ {}^2D \\ {}^4F \\ {}^2F \\ {}^4G \\ {}^2G \end{array}$
$3d^7 4s ({}^3D) nx$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \end{array} \right.$	$\begin{array}{l} {}^4S^{\circ} \\ {}^2S^{\circ} \end{array}$	$\begin{array}{l} {}^4D^{\circ} \\ {}^2D^{\circ} \end{array}$	$\begin{array}{l} {}^4P \\ {}^2P \\ {}^4D \\ {}^2D \\ {}^4F \\ {}^2F \\ {}^4G \\ {}^2G \end{array}$
$3d^7 4s ({}^3P) nx$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \end{array} \right.$	$\begin{array}{l} {}^4S^{\circ} \\ {}^2S^{\circ} \end{array}$	$\begin{array}{l} {}^4D^{\circ} \\ {}^2D^{\circ} \end{array}$	$\begin{array}{l} {}^4P \\ {}^2P \\ {}^4D \\ {}^2D \\ {}^4F \\ {}^2F \\ {}^4G \\ {}^2G \end{array}$

*Within a given configuration the limit terms are listed in the general order predicted by the Hund theory.

TABLE 5. PREDICTED TERMS OF THE Ni I ISOELECTRONIC SEQUENCE

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ +	Predicted Terms			
	ns ($n \geq 4$)	np ($n \geq 4$)	nd ($n \geq 4$)	
$3d^8 4s^2$	{ 3P 1D 3F 1G			
$3d^{10}$	1S			
$3d^8(2D)nx$	{ 3D 1D	$^3P^\circ$ $^3D^\circ$ $^3F^\circ$ $^1P^\circ$ $^1D^\circ$ $^1F^\circ$	3S 3D 3F 3G 1S 1D 1F 1G	
$3d^8 4s(4F)nx$	{ 5F 3F	$^5D^\circ$ $^5F^\circ$ $^5G^\circ$ $^3D^\circ$ $^3F^\circ$ $^3G^\circ$	5P 5D 5F 5G 5H 3P 3D 3F 3G 3H	
$3d^8 4s(2F)nx$	{ 3F 1F	$^3D^\circ$ $^3F^\circ$ $^3G^\circ$ $^1D^\circ$ $^1F^\circ$ $^1G^\circ$	3P 3D 3F 3G 3H 1P 1D 1F 1G 1H	
$3d^8 4s(2D)nx$	{ 3D 1D	$^3P^\circ$ $^3D^\circ$ $^3F^\circ$ $^1P^\circ$ $^1D^\circ$ $^1F^\circ$	3S 3D 3F 3G 1S 1D 1F 1G	
$3d^8 4s(4P)nx$	{ 5P 3P	$^5S^\circ$ $^5P^\circ$ $^5D^\circ$ $^3S^\circ$ $^3P^\circ$ $^3D^\circ$	5P 5D 5F 3P 3D 3F	
$3d^8 4s(2P)nx$	{ 3P 1P	$^3S^\circ$ $^3P^\circ$ $^3D^\circ$ $^1S^\circ$ $^1P^\circ$ $^1D^\circ$	3P 3D 3F 1P 1D 1F	
$3d^8 4s(2G)nx$	{ 3G 1G	$^3F^\circ$ $^3G^\circ$ $^3H^\circ$ $^1F^\circ$ $^1G^\circ$ $^1H^\circ$	3D 3F 3G 3H 3I 1D 1F 1G 1H 1I	
$3d^8 4s(2S)nx$	{ 3S 1S	$^3P^\circ$ $^1P^\circ$	3D 3F 3G 3H 3I 1D 1F 1G 1H 1I	
$3d^8(2D)nx$	{ $^3P^\circ$ $^3D^\circ$ $^3F^\circ$ $^3G^\circ$ $^3H^\circ$ $^1P^\circ$ $^1D^\circ$ $^1F^\circ$ $^1G^\circ$ $^1H^\circ$	3D 3F 3G 3H 3I 1D 1F 1G 1H 1I	
$3d^8 4s(4F)nx$	{ $^5D^\circ$ $^5F^\circ$ $^5G^\circ$ $^5H^\circ$ $^5I^\circ$ $^3D^\circ$ $^3F^\circ$ $^3G^\circ$ $^3H^\circ$ $^3I^\circ$	5F 5G 5H 5I 5K 3F 3G 3H 3I 3K	
$3d^8 4s(2F)nx$	{ $^3D^\circ$ $^3F^\circ$ $^3G^\circ$ $^3H^\circ$ $^3I^\circ$ $^1D^\circ$ $^1F^\circ$ $^1G^\circ$ $^1H^\circ$ $^1I^\circ$	3F 3G 3H 3I 3K 1F 1G 1H 1I 1K	
$3d^8 4s(2D)nx$	{ $^3P^\circ$ $^3D^\circ$ $^3F^\circ$ $^3G^\circ$ $^3H^\circ$ $^1P^\circ$ $^1D^\circ$ $^1F^\circ$ $^1G^\circ$ $^1H^\circ$	3D 3F 3G 3H 3I 1D 1F 1G 1H 1I	
$3d^8 4s(4P)nx$	{ $^5D^\circ$ $^5F^\circ$ $^5G^\circ$ $^3D^\circ$ $^3F^\circ$ $^3G^\circ$	5F 5G 5H 3F 3G 3H	
$3d^8 4s(2P)nx$	{ $^3D^\circ$ $^3F^\circ$ $^3G^\circ$ $^1D^\circ$ $^1F^\circ$ $^1G^\circ$	3F 3G 3H 3I 1F 1G 1H 1I	
$3d^8 4s(2G)nx$	{ $^3F^\circ$ $^3G^\circ$ $^3H^\circ$ $^3I^\circ$ $^1F^\circ$ $^1G^\circ$ $^1H^\circ$ $^1I^\circ$	3G 3H 3I 3L 1G 1H 1I 1K 1L	
$3d^8 4s(2S)nx$	{ $^3F^\circ$ $^1F^\circ$	3G 3I 1G 1I	

TABLE 6. PREDICTED TERMS OF THE Cu I ISOELECTRONIC SEQUENCE

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Predicted Terms					
	$ns (n \geq 4)$	$np (n \geq 4)$	$nd (n \geq 4)$	$nf (n \geq 4)$	$ng (n \geq 5)$
$3d^0 4s^2$	$2D$					
$3d^{10}(1S)nx$	$2S$	$2P^o$	$2D$	$2F^o$	$2G$
$3d^9 4s(3D)nx'$	$4D$	$4P^o$	$4S$	$4D^o$	$4F$	$4H$
	$2D$	$2P^o$	$2S$	$2D^o$	$2F$	$2H$
$3d^9 4s(1D)nx''$	$2D$	$2P^o$	$2S$	$2D^o$	$2F$	$2H$

TABLE 7. PREDICTED TERMS OF THE Zn I ISOELECTRONIC SEQUENCE

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Predicted Terms					
	$ns (n \geq 4)$	$np (n \geq 4)$	$nd (n \geq 4)$	$nf (n \geq 4)$	$ng (n \geq 5)$
$3d^{10} 4s^2$	$1S$					
$3d^{10} 4p^2$	$3P$	$3P^o$	$3D$	$3F^o$	$3G$	$3H$
	$1D$	$1P^o$	$1D$	$1F^o$	$1G$	$1H$
$3d^{10} 4d^2$	$3P$	$3P^o$	$3D$	$3F^o$	$3G$	$3H$
	$1S$	$1P^o$	$1D$	$1F^o$	$1G$	$1H$
$3d^{10} 4s(2S)nx$	$3S$	$3P^o$	$3D$	$3F^o$	$3G$	$3H$
	$1P^o$	$1P^o$	$1D$	$1F^o$	$1G$	$1H$
$3d^{10} 4p(2P^o)nx$	$3S$	$3P^o$	$3D$	$3F^o$	$3G$	$3H$
	$1P^o$	$1P^o$	$1D$	$1F^o$	$1G$	$1H$
$3d^9 4s^2(2D)nx'$	$3D$	$3P^o$	$3S$	$3D^o$	$3F$	$3H$
	$1D$	$1P^o$	$1S$	$1D^o$	$1F$	$1H$

TABLE 8. PREDICTED TERMS OF THE Ga I ISOELECTRONIC SEQUENCE

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} +$	Predicted Terms					
$4s^2(1S)4f$	$2P^{\circ}$					
$4s 4p^2$	$\left\{ \begin{array}{l} 4P \\ 2S \quad 2P \quad 2D \end{array} \right.$					
$4p^3$	$\left\{ \begin{array}{l} 4S^{\circ} \\ 2P^{\circ} \quad 2D^{\circ} \end{array} \right.$					
	$ns (n \geq 5)$	$np (n \geq 5)$	$nd (n \geq 4)$	$nf (n \geq 4)$	$ng (n \geq 5)$
$4s^2(1S)nx$	$2S$	$2P^{\circ}$	$2D$	$2F^{\circ}$	$2G$
$4s 4p(3P^{\circ})nx$	$\left\{ \begin{array}{l} 4P^{\circ} \\ 2P^{\circ} \end{array} \right.$	$\begin{array}{l} 4S \quad 4P \quad 4D \\ 2S \quad 2P \quad 2D \end{array}$	$\begin{array}{l} 4P^{\circ} \quad 4D^{\circ} \quad 4F^{\circ} \\ 2P^{\circ} \quad 2D^{\circ} \quad 2F^{\circ} \end{array}$	$\begin{array}{l} 4D \quad 4F \quad 4G \\ 2D \quad 2F \quad 2G \end{array}$	$\begin{array}{l} 4F^{\circ} \quad 4G^{\circ} \quad 4H^{\circ} \\ 2F^{\circ} \quad 2G^{\circ} \quad 2H^{\circ} \end{array}$
$4s 4p(1P^{\circ})nx'$	$2P^{\circ}$	$2S \quad 2P \quad 2D$	$2P^{\circ} \quad 2D^{\circ} \quad 2F^{\circ}$	$2D \quad 2F \quad 2G$	$2F^{\circ} \quad 2G^{\circ} \quad 2H^{\circ}$

TABLE 9. PREDICTED TERMS OF THE Ge I ISOELECTRONIC SEQUENCE

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} +$	Predicted Terms					
$4s^2 4p^2$	$\left\{ \begin{array}{l} 3P \\ 1S \quad 1D \end{array} \right.$					
$4s 4p^3$	$\left\{ \begin{array}{l} 5S^{\circ} \\ 3S^{\circ} \quad 3P^{\circ} \quad 3D^{\circ} \\ 1P^{\circ} \quad 1D^{\circ} \end{array} \right.$					
	$ns (n \geq 5)$	$np (n \geq 5)$	$nd (n \geq 4)$	$nf (n \geq 4)$	
$4s^2 4p(2P^{\circ})nx$	$\left\{ \begin{array}{l} 3P^{\circ} \\ 1P^{\circ} \end{array} \right.$	$\begin{array}{l} 3S \quad 3P \quad 3D \\ 1S \quad 1P \quad 1D \end{array}$	$\begin{array}{l} 3P^{\circ} \quad 3D^{\circ} \quad 3F^{\circ} \\ 1P^{\circ} \quad 1D^{\circ} \quad 1F^{\circ} \end{array}$	$\begin{array}{l} 3D \quad 3F \quad 3G \\ 1D \quad 1F \quad 1G \end{array}$	
$4s 4p^2(4P)nx$	$\left\{ \begin{array}{l} 5P \\ 3P \end{array} \right.$	$\begin{array}{l} 5S^{\circ} \quad 5P^{\circ} \quad 5D^{\circ} \\ 3S^{\circ} \quad 3P^{\circ} \quad 3D^{\circ} \end{array}$	$\begin{array}{l} 5P \quad 5D \quad 5F \\ 3P \quad 3D \quad 3F \end{array}$	$\begin{array}{l} 5D^{\circ} \quad 5F^{\circ} \quad 5G^{\circ} \\ 3D^{\circ} \quad 3F^{\circ} \quad 3G^{\circ} \end{array}$	

TABLE 10. PREDICTED TERMS OF THE As I ISOELECTRONIC SEQUENCE

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} +$	Predicted Terms					
$4s^2 4p^3$	$\left\{ \begin{array}{l} 4S^{\circ} \\ 2P^{\circ} \quad 2D^{\circ} \end{array} \right.$					
$4s 4p^4$	$\left\{ \begin{array}{l} 4P \\ 2S \quad 2P \quad 2D \end{array} \right.$					
$4p^5$	$2P^{\circ}$					
	$ns (n \geq 5)$	$np (n \geq 5)$	$nd (n \geq 4)$	$nf (n \geq 4)$	
$4s^2 4p^2(3P)nx$	$\left\{ \begin{array}{l} 4P \\ 2P \end{array} \right.$	$\begin{array}{l} 4S^{\circ} \quad 4P^{\circ} \quad 4D^{\circ} \\ 2S^{\circ} \quad 2P^{\circ} \quad 2D^{\circ} \end{array}$	$\begin{array}{l} 4P \quad 4D \quad 4F \\ 2P \quad 2D \quad 2F \end{array}$	$\begin{array}{l} 4D^{\circ} \quad 4F^{\circ} \quad 4G^{\circ} \\ 2D^{\circ} \quad 2F^{\circ} \quad 2G^{\circ} \end{array}$	
$4s^2 4p^2(1D)nx'$		$2D$	$2P^{\circ} \quad 2D^{\circ} \quad 2F^{\circ}$	$2S \quad 2P \quad 2D \quad 2F \quad 2G$	$2P^{\circ} \quad 2D^{\circ} \quad 2F^{\circ} \quad 2G^{\circ} \quad 2H^{\circ}$
$4s^2 4p^2(1S)nx''$	$2S$	$2P^{\circ}$	$2D$		$2F^{\circ}$

TABLE 11. PREDICTED TERMS OF THE Se I ISOELECTRONIC SEQUENCE

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} +$	Predicted Terms			
	$ns (n \geq 5)$	$np (n \geq 5)$	$nd (n \geq 4)$	$nf (n \geq 4)$
$4s^2 4p^4$	$\left\{ \begin{array}{l} {}^1S \\ {}^3P \end{array} \right.$	1D		
$4s 4p^5$	$\left\{ \begin{array}{l} {}^3P^o \\ {}^1P^o \end{array} \right.$			
$4s^2 4p^3({}^1S^o)nx$	$\left\{ \begin{array}{l} {}^6S^o \\ {}^8S^o \end{array} \right.$	$\begin{array}{l} {}^5P \\ {}^3P \end{array}$	$\begin{array}{l} {}^5D^o \\ {}^3D^o \end{array}$	$\begin{array}{l} {}^5F \\ {}^3F \end{array}$
$4s^2 4p^3({}^2D^o)nx'$	$\left\{ \begin{array}{l} {}^3D^o \\ {}^1D^o \end{array} \right.$	$\begin{array}{l} {}^3P \\ {}^1P \\ {}^1D \\ {}^1F \end{array}$	$\begin{array}{l} {}^3S^o \\ {}^1S^o \\ {}^3D^o \\ {}^1D^o \end{array}$	$\begin{array}{l} {}^3P \\ {}^1P \\ {}^3D \\ {}^1D \\ {}^3F \\ {}^1F \\ {}^3G \\ {}^1G \\ {}^3H \\ {}^1H \end{array}$
$4s^2 4p^3({}^2P^o)nx''$	$\left\{ \begin{array}{l} {}^3P^o \\ {}^1P^o \end{array} \right.$	$\begin{array}{l} {}^3S \\ {}^1S \\ {}^3P \\ {}^1P \\ {}^3D \\ {}^1D \end{array}$	$\begin{array}{l} {}^3P^o \\ {}^1P^o \\ {}^3D^o \\ {}^1D^o \end{array}$	$\begin{array}{l} {}^3D \\ {}^1D \\ {}^3F \\ {}^1F \\ {}^3G \\ {}^1G \end{array}$

TABLE 12. PREDICTED TERMS OF THE Br I ISOELECTRONIC SEQUENCE

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} +$	Predicted Terms				
	$ns (n \geq 5)$	$np (n \geq 5)$	$nd (n \geq 4)$	$nf (n \geq 4)$	$ng (n \geq 5)$
$4s^2 4p^5$					
$4s 4p^6$					
$4s^2 4p^4({}^3P)nx$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \end{array} \right.$	$\begin{array}{l} {}^4S^o \\ {}^2S^o \\ {}^4P^o \\ {}^2P^o \end{array}$	$\begin{array}{l} {}^4P \\ {}^2P \\ {}^4D \\ {}^2D \\ {}^4F \\ {}^2F \end{array}$	$\begin{array}{l} {}^4D^o \\ {}^2D^o \\ {}^4F^o \\ {}^2F^o \end{array}$	$\begin{array}{l} {}^4F \\ {}^2F \\ {}^4G \\ {}^2G \end{array}$
$4s^2 4p^4({}^1D)nx'$	2D	$\begin{array}{l} {}^3P^o \\ {}^2P^o \\ {}^3D^o \\ {}^2F^o \end{array}$	$\begin{array}{l} {}^2S \\ {}^2P \\ {}^2D \\ {}^2F \\ {}^2G \end{array}$	$\begin{array}{l} {}^2S \\ {}^2P^o \\ {}^2D^o \\ {}^2F^o \\ {}^2G^o \\ {}^2H^o \end{array}$	$\begin{array}{l} {}^2D \\ {}^2F \\ {}^2G \\ {}^2H \\ {}^2I \end{array}$
$4s^2 4p^4({}^1S)nx''$		2S	2D	${}^2F^o$	2G
$4s 4p^5({}^3P^o)nx'''$	$\left\{ \begin{array}{l} {}^4P^o \\ {}^2P^o \end{array} \right.$	$\begin{array}{l} {}^4S \\ {}^2S \\ {}^4P \\ {}^2P \\ {}^4D \\ {}^2D \end{array}$	$\begin{array}{l} {}^4P^o \\ {}^2P^o \\ {}^4D^o \\ {}^2D^o \\ {}^4F^o \\ {}^2F^o \end{array}$	$\begin{array}{l} {}^4D \\ {}^2D \\ {}^4F \\ {}^2F \end{array}$	$\begin{array}{l} {}^4P^o \\ {}^2P^o \\ {}^4G^o \\ {}^2G^o \\ {}^4H^o \\ {}^2H^o \end{array}$

TABLE 13. PREDICTED LEVELS OF THE KI I ISOELECTRONIC SEQUENCE

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} +$	Predicted Terms				
$4s^2 4p^6$	1S				
	$ns (n \geq 5)$	$np (n \geq 5)$	$nd (n \geq 4)$	$nf (n \geq 4)$	-----
$4s^2 4p^5(^2P^o)nx$	$\begin{cases} ^3P^o \\ ^1P^o \end{cases}$	$\begin{matrix} ^3S & ^3P & ^3D \\ ^1S & ^1P & ^1D \end{matrix}$	$\begin{matrix} ^3P^o & ^3D^o & ^3F^o \\ ^1P^o & ^1D^o & ^1F^o \end{matrix}$	$\begin{matrix} ^3D & ^3F & ^3G \\ ^1D & ^1F & ^1G \end{matrix}$	----- ----- -----
$4s 4p^6(^2S)nx'$	$\begin{cases} ^3S \\ ^1S \end{cases}$	$\begin{matrix} ^3P^o \\ ^1P^o \end{matrix}$	$\begin{matrix} ^3D \\ ^1D \end{matrix}$	$\begin{matrix} ^3F^o \\ ^1F^o \end{matrix}$	----- -----
<i>jl</i> -Coupling Notation					
Config. $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 +$	Predicted Pairs				
	$ns (n \geq 5)$	$np (n \geq 5)$	$nd (n \geq 4)$	$nf (n \geq 4)$	-----
$4p^5(^2P_{1/2})nx$	$[1\frac{1}{2}]^o$	$\begin{matrix} [0\frac{1}{2}] \\ [2\frac{1}{2}] \\ [1\frac{1}{2}] \end{matrix}$	$\begin{matrix} [0\frac{1}{2}]^o \\ [3\frac{1}{2}]^o \\ [1\frac{1}{2}]^o \\ [2\frac{1}{2}]^o \end{matrix}$	$\begin{matrix} [1\frac{1}{2}] \\ [4\frac{1}{2}] \\ [2\frac{1}{2}] \\ [3\frac{1}{2}] \end{matrix}$	----- ----- ----- -----
$4p^5(^2P_{3/2})nx'$	$[0\frac{1}{2}]^o$	$\begin{matrix} [1\frac{1}{2}] \\ [0\frac{1}{2}] \end{matrix}$	$\begin{matrix} [2\frac{1}{2}]^o \\ [1\frac{1}{2}]^o \end{matrix}$	$\begin{matrix} [3\frac{1}{2}] \\ [2\frac{1}{2}] \end{matrix}$	----- -----

TABLE 14. PREDICTED TERMS OF THE S_{f1} ISOELECTRONIC SEQUENCE

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{l_0} 4s^2 4p^6 +$	Predicted Terms			
$5s^2$	1S			
$4d^2$	$\left\{ \begin{array}{cccc} ^3P & & ^3F & \\ ^1S & ^1D & & ^1G \end{array} \right.$			
$5p^2$	$\left\{ \begin{array}{cc} ^3P & \\ ^1S & ^1D \end{array} \right.$			
	$ns (n \geq 5)$	$np (n \geq 5)$	$nd (n \geq 4)$	
$5s(^2S)nx$	$\left\{ \begin{array}{c} ^3S \\ ^1S \end{array} \right.$	$\begin{array}{c} ^3P^\circ \\ ^1P^\circ \end{array}$	$\begin{array}{c} ^3D \\ ^1D \end{array}$	
$4d(^2D)nx'$	$\left\{ \begin{array}{c} ^3D \\ ^1D \end{array} \right.$	$\begin{array}{ccc} ^3P^\circ & ^3D^\circ & ^3F^\circ \\ ^1P^\circ & ^1D^\circ & ^1F^\circ \end{array}$	$\begin{array}{ccccc} ^3S & ^3P & ^3D & ^3F & ^3G \\ ^1S & ^1P & ^1D & ^1F & ^1G \end{array}$	
$5p(^2P^\circ)nx''$	$\left\{ \begin{array}{c} ^3P^\circ \\ ^1P^\circ \end{array} \right.$	$\begin{array}{ccc} ^3S & ^3P & ^3D \\ ^1S & ^1P & ^1D \end{array}$	$\begin{array}{ccc} ^3P^\circ & ^3D^\circ & ^3F^\circ \\ ^1P^\circ & ^1D^\circ & ^1F^\circ \end{array}$	
	$nf (n \geq 4)$	$ng (n \geq 5)$		-----
$5s(^2S)nx$	$\left\{ \begin{array}{c} ^3F^\circ \\ ^1F^\circ \end{array} \right.$	$\begin{array}{c} ^3G \\ ^1G \end{array}$		----- -----
$4d(^2D)nx'$	$\left\{ \begin{array}{ccccc} ^3P^\circ & ^3D^\circ & ^3F^\circ & ^3G^\circ & ^3H^\circ \\ ^1P^\circ & ^1D^\circ & ^1F^\circ & ^1G^\circ & ^1H^\circ \end{array} \right.$	$\begin{array}{ccccc} ^3D & ^3F & ^3G & ^3H & ^3I \\ ^1D & ^1F & ^1G & ^1H & ^1I \end{array}$		----- ----- -----
$5p(^2P^\circ)nx''$	$\left\{ \begin{array}{ccc} ^3D & ^3F & ^3G \\ ^1D & ^1F & ^1G \end{array} \right.$	$\begin{array}{ccc} ^3F^\circ & ^3G^\circ & ^3H^\circ \\ ^1F^\circ & ^1G^\circ & ^1H^\circ \end{array}$		----- -----

TABLE 15. PREDICTED TERMS OF THE Y I ISOELECTRONIC SEQUENCE

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 +$	Predicted Terms		
	$ns (n \geq 5)$	$np (n \geq 5)$	$nd (n \geq 4)$
$4d 5s^2$	2D		
$4d^3$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \\ {}^2D \end{array} \right.$	$\left\{ \begin{array}{l} {}^4F \\ {}^2F \\ {}^2G \\ {}^2H \end{array} \right.$	
$5s^2(1S)nx$	2S		2D
$4d 5s(3D)nx$	$\left\{ \begin{array}{l} {}^4D \\ {}^2D \end{array} \right.$	$\left\{ \begin{array}{l} {}^4D^\circ \\ {}^2D^\circ \\ {}^4F^\circ \\ {}^2F^\circ \end{array} \right.$	$\left\{ \begin{array}{l} {}^4S \\ {}^2S \\ {}^4P \\ {}^2P \\ {}^4D \\ {}^2D \\ {}^4F \\ {}^2F \\ {}^4G \\ {}^2G \end{array} \right.$
$4d 5s(1D)nx$	2D	${}^2D^\circ$	2D
$4d^2(3F)nx$	$\left\{ \begin{array}{l} {}^4F \\ {}^2F \end{array} \right.$	$\left\{ \begin{array}{l} {}^4D^\circ \\ {}^2D^\circ \\ {}^4F^\circ \\ {}^2F^\circ \end{array} \right.$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \\ {}^4D \\ {}^2D \\ {}^4F \\ {}^2F \\ {}^4G \\ {}^2G \end{array} \right.$
$4d^2(3P)nx$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \end{array} \right.$	$\left\{ \begin{array}{l} {}^4D^\circ \\ {}^2D^\circ \\ {}^4S^\circ \\ {}^2S^\circ \end{array} \right.$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \\ {}^4D \\ {}^2D \\ {}^4F \\ {}^2F \\ {}^4G \\ {}^2G \end{array} \right.$
$4d^2(1G)nx$		2G	
$4d^2(1D)nx$	2D	${}^2D^\circ$	2D
$4d^2(1S)nx$	2S	${}^2P^\circ$	2S
$5p^2(3P)nx$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \end{array} \right.$	$\left\{ \begin{array}{l} {}^4D^\circ \\ {}^2D^\circ \end{array} \right.$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \\ {}^4D \\ {}^2D \\ {}^4F \\ {}^2F \end{array} \right.$
$5p^2(1D)nx$	2D	${}^2D^\circ$	2D
$5p^2(1S)nx$	2S	${}^2P^\circ$	2S

TABLE 17. PREDICTED TERMS OF THE Nb I ISOELECTRONIC SEQUENCE

Config.	Predicted Terms		
	$ns (n \geq 5)$	$np (n \geq 5)$	$nd (n \geq 5)$
$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 +$			
$4d^5$	$\left\{ \begin{array}{l} {}^6S \\ {}^2S \end{array} \right.$ $\left\{ \begin{array}{l} {}^4P \quad {}^4D \quad {}^4F \quad {}^4G \\ {}^2D \quad {}^2F \quad {}^2G \quad {}^2H \end{array} \right.$ $\left\{ \begin{array}{l} {}^2I \\ {}^2P \quad {}^2D \quad {}^2D \end{array} \right.$ $\left\{ \begin{array}{l} {}^4P \quad {}^4F \\ {}^2P \quad {}^2D \quad {}^2F \quad {}^2G \quad {}^2H \\ {}^2D \end{array} \right.$		
$4d^3 5s^2$	$\left\{ \begin{array}{l} {}^4P \quad {}^4F \\ {}^2P \quad {}^2D \quad {}^2F \quad {}^2G \quad {}^2H \\ {}^2D \end{array} \right.$		
$4d^4 ({}^5D) nx$	$\left\{ \begin{array}{l} {}^6D \\ {}^4D \end{array} \right.$	$\left\{ \begin{array}{l} {}^6P^\circ \quad {}^6D^\circ \quad {}^6F^\circ \\ {}^4P^\circ \quad {}^4D^\circ \quad {}^4F^\circ \end{array} \right.$	$\left\{ \begin{array}{l} {}^6S \\ {}^4S \end{array} \right.$ $\left\{ \begin{array}{l} {}^6P \quad {}^6D \quad {}^6F \quad {}^6G \\ {}^4P \quad {}^4D \quad {}^4F \quad {}^4G \end{array} \right.$ $\left\{ \begin{array}{l} {}^6S \\ {}^4S \end{array} \right.$
$4d^3 5s ({}^3F) nx$	$\left\{ \begin{array}{l} {}^6F \\ {}^4F \end{array} \right.$	$\left\{ \begin{array}{l} {}^6D^\circ \quad {}^6F^\circ \quad {}^6G^\circ \\ {}^4D^\circ \quad {}^4F^\circ \quad {}^4G^\circ \end{array} \right.$	$\left\{ \begin{array}{l} {}^6P \quad {}^6D \quad {}^6F \quad {}^6G \quad {}^6H \\ {}^4P \quad {}^4D \quad {}^4F \quad {}^4G \quad {}^4H \end{array} \right.$
$4d^4 ({}^3P) nx$	$\left\{ \begin{array}{l} {}^4P \\ {}^2P \end{array} \right.$	$\left\{ \begin{array}{l} {}^4S^\circ \quad {}^4D^\circ \\ {}^2S^\circ \quad {}^2D^\circ \end{array} \right.$	$\left\{ \begin{array}{l} {}^4P \quad {}^4D \quad {}^4F \\ {}^2P \quad {}^2D \quad {}^2F \end{array} \right.$
$4d^3 5s ({}^3F) nx$	$\left\{ \begin{array}{l} {}^4F \\ {}^2F \end{array} \right.$	$\left\{ \begin{array}{l} {}^4D^\circ \quad {}^4F^\circ \quad {}^4G^\circ \\ {}^2D^\circ \quad {}^2F^\circ \quad {}^2G^\circ \end{array} \right.$	$\left\{ \begin{array}{l} {}^4P \quad {}^4D \quad {}^4F \quad {}^4G \quad {}^4H \\ {}^2P \quad {}^2D \quad {}^2F \quad {}^2G \quad {}^2H \end{array} \right.$
$4d^4 ({}^3H) nx$	$\left\{ \begin{array}{l} {}^4H \\ {}^2H \end{array} \right.$	$\left\{ \begin{array}{l} {}^4G^\circ \quad {}^4H^\circ \quad {}^4I^\circ \\ {}^2G^\circ \quad {}^2H^\circ \quad {}^2I^\circ \end{array} \right.$	$\left\{ \begin{array}{l} {}^4F \quad {}^4G \quad {}^4H \quad {}^4I \\ {}^2F \quad {}^2G \quad {}^2H \quad {}^2I \end{array} \right.$ $\left\{ \begin{array}{l} {}^4K \\ {}^2K \end{array} \right.$
$4d^4 ({}^3G) nx$	$\left\{ \begin{array}{l} {}^4G \\ {}^2G \end{array} \right.$	$\left\{ \begin{array}{l} {}^4F^\circ \quad {}^4G^\circ \quad {}^4H^\circ \\ {}^2F^\circ \quad {}^2G^\circ \quad {}^2H^\circ \end{array} \right.$	$\left\{ \begin{array}{l} {}^4D \quad {}^4F \quad {}^4G \quad {}^4H \quad {}^4I \\ {}^2D \quad {}^2F \quad {}^2G \quad {}^2H \quad {}^2I \end{array} \right.$

TABLE 18. THE CHEMICAL ELEMENTS—IONIZATION POTENTIALS

Z	Element	Spectrum																			
		I	II	III	IV	V	VI	VII	VIII	IX	X	XI	XII	XIII	XIV	XV	XVI	XVII	XVIII	XIX	
1	H	13.595																			
2	He	24.580	54.403																		
3	Li	5.390	75.619	122.419																	
4	Be	9.320	18.206	153.850	217.657																
5	B	8.296	25.149	37.920	259.298	340.127															
6	C	11.264	24.376	47.864	64.476	391.986	489.84														
7	N	14.54	29.605	47.426	77.450	97.863	551.925	666.83													
8	O	13.614	35.146	54.934	77.394	113.873	138.080	739.114	871.12												
9	F	17.418	34.98	62.646	87.23	114.214	157.117	185.139	953.60												
10	Ne	21.559	41.07	64	97.16	126.4	157.91														
11	Na	5.138	47.29	71.65	98.88	138.60	172.36	208.444	264.155	299.78											
12	Mg	7.644	15.03	80.12	109.29	141.23	186.86	225.31	265.957	327.90	367.36	1761.23									
13	Al	5.984	18.823	28.44	119.96	153.77	190.42	241.93	285.13	330.1	398.5	441.9	2085.46								
14	Si	8.149	16.34	33.46	45.13	166.73	205.11	246.41	303.87	351.83	401.3	476.0	523.2								
15	P	10.55	19.65	30.156	51.354	65.007	220.414	263.31	309.26	372.62	425.46	479.4	560.3	611.4							
16	S	10.357	23.4	35.0	47.29	72.5	88.029	280.99	328.80	378.95	448.2										
17	Cl	13.01	23.80	39.90	53.5	67.80	96.7	114.27	348.3	400.7	455.3										
18	A	15.755	27.62	40.90	59.79	75.0	91.3	124.0	143.46	421											
19	K	4.339	31.81	46	60.90	99.7	118	155	175.94	503.8											
20	Ca	6.111	11.87	51.21	67	84.39	128	147	188	211.29	591.8	655									
21	Sc	6.56	12.80	24.75	73.9	92	111.1	159	180	226	250	687									
22	Ti	6.83	13.57	27.47	43.24	99.8	120	140.8	193	266	291.5	788.4									
23	V	6.74	14.65	29.31	48	65.2	128.9	151	173.7	309	336.3	897.1									
24	Cr	6.763	16.49	30.95	49.6	73	90.6	161.1	185	209.6	355	384.2	1013.0								
25	Mn	7.432	15.64	33.69	76	119.24	196	222	248	435	1135.9										
26	Fe	7.90	16.18	30.64				151	235	262	290	355	457	489							
27	Co	7.86	17.05	33.49							305	350	512	547	1403						
28	Ni	7.633	18.15	35.16																	
29	Cu	7.724	20.29	36.83																	671

30	Zn	9.391	17.96	39.70															
31	Ga	6.00	20.51	30.70	64.2														
32	Ge	7.88	15.93	34.21	45.7	93.4													
33	As	9.81	20.2	28.3	50.1	62.6	127.5												
34	Se	9.75	21.5	32.0	42.9	73.1	81.7	155											
35	Br	11.84	21.6	35.9				193											
36	Kr	13.996	24.56	36.9															
37	Rb	4.176	27.5	40						277									
38	Sr	5.692	11.027		57									324					
39	Y	6.5	12.4	20.5		77													
40	Zr	6.95	14.03	24.8	33.97		99												
41	Nb	6.77	14	28.1	38.3	50	103	125											
																			374

TABLE 19. CHEMICAL SYMBOLS

Symbol	Element	Z	Symbol	Element	Z	Symbol	Element	Z	Symbol	Element	Z
A	Argon	18	Dy	Dysprosium	66	Mn	Manganese	25	Ru	Ruthenium	44
Ac	Actinium	89	Er	Erbium	68	Mo	Molybdenum	42	S	Sulfur	16
Ag	Silver	47	Eu	Europium	63	N	Nitrogen	7	Sb	Antimony	51
Al	Aluminum	13	F	Fluorine	9	Na	Sodium	11	Sc	Scandium	21
Am	Americium	95	Fe	Iron	26	Nb	Niobium	41	Se	Selenium	34
As	Arsenic	33	Fr	Francium	87	Nd	Neodymium	60	Si	Silicon	14
At	Astatine	85	Ga	Gallium	31	Ne	Neon	10	Sb	Samarium	62
Au	Gold	79	Gd	Gadolinium	64	Ni	Nickel	28	Sn	Tin	50
B	Boron	5	Ge	Germanium	32	Np	Neptunium	93	Sr	Strontium	38
Ba	Barium	56	H	Hydrogen	1	O	Oxygen	8	Ta	Tantalum	73
Bc	Beryllium	4	(D	Deuterium)		Os	Osmium	76	Tb	Terbium	65
Bi	Bismuth	83	(T	Tritium)		P	Phosphorus	15	Tc	Technetium	43
Bk	Berkelium	97	He	Helium	2	Pa	Protactinium	91	Te	Tellurium	52
Br	Bromine	35	Hf	Hafnium	72	Pb	Lead	82	Th	Thorium	90
Br	Bromine	35	Hg	Mercury	80	Pd	Palladium	46	Ti	Titanium	22
C	Carbon	6	I	Iodine	67	Pm	Promethium	61	Tl	Thallium	81
Ca	Calcium	20	Hl	Holmium	53	Po	Polonium	84	Tl	Thulium	69
Cd	Cadmium	48	I	Iodine	53	Pr	Praseodymium	59	U	Uranium	92
Ce	Cerium	58	In	Indium	49	Pt	Platinum	78	V	Vanadium	23
Cf	Californium	98	Ir	Iridium	77	Pu	Plutonium	94	W	Wolfram	74
Cl	Chlorine	17	K	Potassium	19	Ra	Radium	88	Xe	Xenon	54
Cl	Chlorine	17	Kr	Krypton	36	Rb	Rubidium	37	Y	Yttrium	39
Cm	Curium	96	La	Lanthanum	57	Rc	Rhenium	75	Yb	Ytterbium	70
Co	Cobalt	27	Li	Lithium	3	Rh	Rhodium	45	Zn	Zinc	30
Co	Chromium	24	L	Lutetium	71	Ru	Ru	12	Zr	Zirconium	40
Cr	Chromium	24	Lu	Lutetium	71						
Cs	Cesium	55	Mg	Magnesium	12						
Cu	Copper	29									

TABLE 20. THE PERIODIC SYSTEM*

1	1s	H 1	He 2
2	2s	Li 3	Be 4
	2p	B 5	C N O F Ne 6 7 8 9 10
3	3s	Na 11	Mg 12
	3p	Al 13	Si P S Cl A 14 15 16 17 18
4	{4s	K 19	Ca 20
	{3d	Sc Ti V Cr Mn Fe Co Ni Cu Zn 21 22 23 24 25 26 27 28 29 30	
5	4p	Ga 31	Ge As Se Br Kr 32 33 34 35 36
	{5s	Rb 37	Sr 38
6	{4d	Y Zr Nb Mo Tc Ru Rh Pd Ag Cd 39 40 41 42 43 44 45 46 47 48	
	5p	In 49	Sn Sb Te I Xe 50 51 52 53 54
7	{6s	Cs 55	Ba 56
	{4f	La Ce Pr Nd Pm Sm Eu Gd Tb Dy Ho Er Tm Yb 57 58 59 60 61 62 63 64 65 66 67 68 69 70	
8	{5d	Lu 71	Hf Ta W Re Os Ir Pt Au Hg 72 73 74 75 76 77 78 79 80
	6p	Tl 81	Pb Bi Po At Rn 82 83 84 85 86
9	{7s	Fr 87	Ra 88
	{5f	Ac Th Pa U Np Pu Am Cm Bk Cf 89 90 91 92 93 94 95 96 97 98 99 100 101 102	
10	{6d	103 104 105 106 107 108 109 110 111 112	

*This arrangement is by Catalán. See J. Cabrera, Física general Zaragoza, Spain (1950). The electrons indicated in column two that are connected by braces have approximately the same binding energy. Consequently, for some elements one type of electron is preferred over another in the normal configuration, as for example, Cr, Nb, Pd, La, Ac, Th.

TABLE 21. INDEX—ISOELECTRONIC SEQUENCES
 [The tabular entries are page numbers]

Z	Element	Spectrum																		
		I	II	III	IV	V	VI	VII	VIII	IX	X	XI	XII	XIII	XIV	XV	XVI	XVII	XVIII	XIX
24	Cr	1	10	14	16	18	20	21	22	22	22	23	24	25	25					
25	Mn	27	32	37	38	40	41	42	43	44	45		46	46	47	47				
26	Fe	49	55	60	65	66	67	69	70	70	71		73	74	74	75	76			
27	Co	78	83	85	88	88	89	90	92					93	93	94	95	96		
28	Ni	97	101	103		104	104	105	106	108		108	108		109	110	110			
29	Cu	111	115	121				123												123
30	Zn	124	126	128	129															
31	Ga	130	131	133	134															
32	Ge	135	137	139	140	141														
33	As	142	144	146	147	148	149													
34	Se	150	152	154	156	156	157	158												
35	Br	159	161	163	164	165	166	167	167	168										
36	Kr	169	173	176	178					179										
37	Rb	180	184	185						186	186	187	188	188	188					
38	Sr	189	191		193							194	194	195	195					
39	Y	196	199	201		202						203	204	204	204	204				
40	Zr	205	209	212	213		214					215	215	215	215					
41	Nb	216	221	223	224	225	226	227												

CHROMIUM

Cr I

24 electrons

Z=24

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^5 4s^1 {}^7S_3$

$a {}^7S_3$ $54565 \pm 1 \text{ cm}^{-1}$

I. P. 6.763 volts

The arc spectrum of chromium has an interesting history. As early as 1907 Miller found pairs of lines having similar Zeeman patterns, which were shown by Richter in 1914 to be members of the two outstanding triplets in the blue and violet. This early work inspired a systematic search, started in 1918 at the National Bureau of Standards, for all groups of lines exhibiting recurring constant wave number differences, and additional triplets in the near infrared were reported by Kiess and Meggers in 1920.

This spectrum is unique in that the discovery of further regularities in Cr I was made independently and almost simultaneously at three places. In 1922 when Catalán announced his discovery of multiplets in the spectra of manganese, he also reported a few in Cr I. At nearly the same time Fr. Gieseler, from her study of the Zeeman effect, found similar groups of related lines in Cr I. Shortly afterwards C. C. Kiess and H. K. Kiess reported the arrangement of the strong triplets in two systems of series including both sharp and diffuse members.

The present term list has been furnished by C. C. Kiess in advance of publication, especially for inclusion here. He has revised and extended the earlier work by Catalán and others. Approximately 2800 lines are classified in the range between 1988 Å and 11610 Å, resulting from combinations among 155 terms. Four multiplicities, singlets, triplets, quintets, and septets, are known, the terms of each being connected by observed intersystem combinations. The limit is excellently determined from three series of 7S , 5S , and ${}^7P^\circ$ terms, two of three members each, and one of four members, which are accurately represented by a Ritz formula.

In 1945 Kiess generously furnished the writer with his analysis for inclusion in her "*Revised Multiplet Table*." His later work has introduced changes in both the notation and the analysis she used. In order to avoid confusion these changes are recorded below, the first entry (R M T), denoting the data used in the "*Revised Multiplet Table*", the second, (A E L), the present revision of that material:

Changes in Notation			
R M T	A E L	R M T	A E L
$f {}^5P$	$e {}^5P$	$s {}^3G^\circ$	$r {}^3G^\circ$
$z {}^1H^\circ$	$y {}^1H^\circ$	$q {}^3H^\circ$	$p {}^3H^\circ$
$z {}^1I^\circ$	$y {}^1I^\circ$	$w {}^3I_5^\circ$	$x {}^1H^\circ$
$y {}^1I^\circ$	$x {}^1I^\circ$	$u' {}^5F^\circ$	$t {}^5F^\circ$
$w {}^3F^\circ$	$v {}^3F^\circ$	$t {}^5F^\circ$	$s {}^5F^\circ$
$v {}^3F^\circ$	$u {}^3F^\circ$	$s {}^5F^\circ$	$r {}^5F^\circ$
$u {}^3F^\circ$	$s {}^3F^\circ$	$r {}^5F^\circ$	$q {}^5F^\circ$
$t {}^3F^\circ$	$r {}^3F^\circ$	1_4°	2_4°

Cr I—Continued

Changes in Analysis		
Desig.	R M T	A E L
$g \ ^1D_1$	47699. 18	47700. 18
$v \ ^3P_2^{\circ}$	Missing	56802. 50
$t \ ^3P_{2,1}^{\circ}$	-----	Rejected
$x \ ^3D_1^{\circ}$	48681. 53:	48839. 90
$u \ ^3H_4^{\circ}$	Missing	55915. 50
$w \ ^3I_5^{\circ}$	60005. 4	59806. 27
$w \ ^5D_0^{\circ}$	Missing	46081. 27
$t \ ^5D_0^{\circ}$	52000. 59	51999. 62
$s \ ^5D_2^{\circ}$	53540. 8	53558. 05

All three-place g -values in the table and those for the levels $a \ ^5P_3$, $z \ ^5S_2^{\circ}$, and $u \ ^5P_{2,1}^{\circ}$ have been calculated by Kiess from Zeeman patterns observed at the Massachusetts Institute of Technology. The rest are from the paper by Catalán and Sancho.

This detailed analysis will be published by the National Bureau of Standards in the near future. With this splendid piece of work Kiess has provided another highly complex spectrum which affords an excellent confirmation of Hund's general theory of atomic spectra.

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Cr I

Cr I

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$3d^5(a^6S)4s$	<i>a</i> ⁷ S	3	0.00		2.007	$3d^4 4s^2$	<i>b</i> ³ G	3	27597.22		
$3d^5(a^6S)4s$	<i>a</i> ⁵ S	2	7593.16		2.006			4	27703.84	106.62	
								5	27816.88	113.04	
$3d^4 4s^2$	<i>a</i> ⁵ D	0	7750.78		0.000	$3d^4 4s(a^6D)4p$	<i>y</i> ⁷ P ^o	2	27728.87		2.341
		1	7810.82	60.04	1.501			3	27820.23	91.36	1.929
		2	7927.47	116.65	1.496			4	27935.26	115.03	1.761
		3	8095.21	167.74	1.501						
		4	8307.57	212.36	1.497	$3d^5(b^4D)4s$	<i>a</i> ³ D	3	28637.00	-45.18	
$3d^5(a^4G)4s$	<i>a</i> ⁶ G	2	20517.40		0.37			2	28682.18	2.75	
		3	20520.92	3.52	0.93			1	28679.43		
		4	20523.69	2.77	1.13	$3d^4 4s(a^6D)4p$	<i>y</i> ⁶ P ^o	1	29420.90		2.513
		5	20523.94	0.25	1.25			2	29584.62	163.72	1.836
		6	20519.60	-4.34	1.33			3	29824.75	240.13	1.669
$3d^5(a^4P)4s$	<i>a</i> ⁵ P	3	21840.84		1.60	$3d^4 4s(a^4D)4p$	<i>z</i> ⁵ F ^o	1	30787.30		0.002
		2	21847.88	-7.04	1.847			2	30858.82	71.52	0.997
		1	21856.94	-9.06	2.500			3	30965.46	106.64	1.245
$3d^4 4s^2$	<i>a</i> ³ P	0	23163.27					4	31106.37	140.91	1.345
		1	23512.00	348.73				5	31280.35	173.98	1.396
		2	24093.16	581.16		$3d^5(a^2D)4s$	<i>b</i> ³ D	3	31009.00	-19.33	
$3d^5(a^6S)4p$	<i>z</i> ⁷ P ^o	2	23305.01		2.334			2	31028.33	-20.52	
		3	23386.35	81.34	1.92	$3d^5(a^2I)4s$	<i>a</i> ³ I	7	31048.00		-1.33
		4	23498.84	112.49	1.752			6	31049.33		-6.02
$3d^4 4s^2$	<i>a</i> ³ H	4	23933.90					5	31055.35		
		5	24056.11	122.21		$3d^5(b^4F)4s$	<i>a</i> ⁶ F	1	31352.42		2.79
		6	24200.23	144.12				2	31355.21		9.12
$3d^5(b^4D)4s$	<i>b</i> ⁵ D	0	24277.06		0.00			3	31364.33		13.63
		1	24286.54	9.48	1.48			4	31377.96		15.44
		2	24299.89	13.35	1.51			5	31393.40		
		3	24303.94	4.05	1.55	$3d^4 4s^2$	<i>a</i> ¹ G	4	31987.06		
		4	24282.34	-21.60	1.51						
$3d^5(a^4G)4s$	<i>a</i> ³ G	3	24833.86			$3d^4 4s^2$	<i>a</i> ¹ I	6	32097.36		
		4	24897.55	63.69							
		5	25038.61	141.06		$3d^5(a^2F)4s$	<i>b</i> ³ F	2	33040.10		20.64
$3d^4 4s^2$	<i>a</i> ³ F	2	24940.61					3	33060.74		52.53
		3	25106.34	165.73				4	33113.27		
		4	25177.39	71.05		$3d^4 4s(a^6D)4p$	<i>z</i> ⁶ D ^o	0	33338.20		-0.003
$3d^4 4s(a^6D)4p$	<i>z</i> ⁷ F ^o	0	24971.21					1	33423.79	85.59	1.499
		1	25010.64	39.43	1.52			2	33542.11	118.32	1.497
		2	25089.20	78.56	1.50			3	33671.55	129.44	1.497
		3	25206.02	116.82	1.49	$3d^4 4s(a^4D)4p$	<i>z</i> ³ P ^o	0	33762.56		0.00
		4	25359.62	153.60	1.51			1	33897.26	134.70	1.49
		5	25548.64	189.02	1.51			2	34190.49	293.23	1.55
		6	25771.40	222.76	1.53	$3d^5(a^2I)4s$	<i>b</i> ¹ I	6	33762.74		
$3d^5(a^6S)4p$	<i>z</i> ⁶ P ^o	3	26787.50		1.670	$3d^4 4s^2$	<i>c</i> ³ D	1	33906.65		29.00
		2	26796.28	-8.78	1.830			2	33935.65		-0.77
		1	26801.93	-5.65	2.512			3	33934.88		
$3d^5(a^4P)4s$	<i>b</i> ³ P	0	27163.20			$3d^6$	<i>c</i> ⁵ D	4	35398.02		-103.24
		1	27176.22	13.02				3	35501.26		-71.68
		2	27223.05	46.83				2	35572.94		-45.57
$3d^4 4s(a^6D)4p$	<i>z</i> ⁷ D ^o	1	27300.19		3.01			1	35618.51		-22.18
		2	27382.18	81.99	1.99			0	35640.69		
		3	27500.37	118.19	1.76	$3d^5(b^4F)4s$	<i>c</i> ³ F	2	35807.90		5.83
		4	27649.71	149.34	1.66			3	35813.73		49.09
		5	27825.45	175.74	1.61			4	35862.82		

Cr I—Continued

Cr I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>		
$3d^5(b^2H)4s$	b^3H	4	35870. 53			$3d^5(a^4G)4p$	z^5G°	2	42515. 35		0. 35		
		5	35884. 40	13. 87	3			42538. 81	23. 46				
		6	35934. 02	49. 62	4			42564. 85	26. 04				
$3d^4 4s(a^4D)4p$	z^3F°	2	35897. 87			$3d^4 4s(b^4P)4p$	z^5S°	2	43124. 88		1. 96		
		3	36034. 22	136. 35	5			42589. 25	24. 40				
		4	36212. 15	177. 93	6			42605. 81	16. 56				
$3d^5(b^2F)4s$	d^3F	2	36558. 55			$3d^5(a^6S)4d$	e^5D	4	44050. 87	-17. 85			
		3	36552. 13	-6. 42	3			44068. 72	-12. 18				
		4	36577. 73	25. 60	2			44080. 90	-8. 02				
$3d^5(a^6S)5s$	e^7S	3	36895. 73			0	44092. 80	-3. 88					
$3d^5(b^2G)4s$	c^3G	3	37205. 88			$3d^5(a^4P)4p$	w^5P°	1	44125. 90		2. 74		
		4	37244. 17	38. 29	2			44186. 92	61. 02				
		5	37233. 50	-10. 67	3			44259. 36	72. 44				
$3d^5(a^6S)5s$	e^5S	2	37883. 34			$3d^4 4s(a^4H)4p$	z^5I°	4	44246. 70		1. 68		
$3d^5(b^2H)4s$	a^1H	5	38537. 68		5			44307. 96	61. 26				
					6			44393. 10	85. 14				
					7			44514. 44	121. 34				
$3d^4 4s(a^4D)4p$	z^3D°	1	38597. 06			8	44666. 55	152. 11					
		2	38730. 67	133. 61	$3d^4 4s(a^4H)4p$	y^5G°	2	44299. 98		0. 35			
		3	38911. 33	180. 66			3	44373. 34	73. 36				
			4	44534. 46			161. 12						
$3d^5(b^2G)4s$	b^1G	4	39158. 63			5	44591. 46	57. 00					
$3d^4 4s(a^4D)4p$	y^5F°	1	40906. 46			$3d^4 4s(b^4P)4p$	v^5P°	1	44666. 74		1. 25		
		2	40971. 29	64. 83	2			44875. 19	208. 45				
		3	41086. 26	114. 97	3			45113. 22	238. 03				
		4	41224. 78	138. 52									
		5	41393. 47	168. 69									
$3d^4 4s(a^4D)4p$	x^5P°	1	40930. 31			$3d^4 4s(a^4F)4p$	x^5F°	1	45201. 84		2. 47		
		2	40982. 97	52. 66	2			45225. 20	23. 36				
		3	41043. 35	60. 38	3			45255. 51	30. 31				
$3d^4 4s(a^4D)4p$	y^5D°	0	41224. 80			$3d^5(a^4G)4p$	z^3H°	4	45286. 08		1. 41		
		1	41289. 17	64. 37	5			45306. 45	20. 37				
		2	41409. 03	119. 86	6			45348. 73	-5. 45				
		3	41575. 10	166. 07	5			45354. 18	-4. 45				
$3d^4 4s(a^4H)4p$	z^5H°	4	41782. 19	207. 09	1. 500	$3d^5(a^4G)4p$	y^5H°	4	45358. 63		0. 52		
		3	42025. 60	54. 21				3	45566. 02	48. 86			
		4	42079. 81	73. 93				4	45614. 88	48. 40			
		5	42153. 74	98. 43				5	45663. 28	44. 08			
		6	42252. 17	135. 15				6	45707. 36	34. 13			
$3d^4 4s(b^4P)4p$	x^5D°	7	42387. 32			$3d^5(a^6S)6s$	f^7S	3	45643. 38		1. 29		
		0	42218. 37	74. 59	-0. 003			$3d^5(a^4P)4p$	y^3P°	0		45722. 59	
		1	42292. 96	145. 86	1. 501					1		45719. 20	-3. 39
		2	42438. 82	209. 44	1. 494					2		45734. 32	15. 12
3	42648. 26	260. 31	1. 498										
$3d^5(a^6S)5p$	x^7P°	4	42908. 57			$3d^5(a^4G)4p$	y^3F°	2	45966. 45		2. 05		
		2	42238. 04	16. 07				3	46000. 36	33. 91			
		3	42254. 11	21. 09				4	46058. 20	57. 84			
$3d^5(a^6S)4d$	e^7D	4	42275. 20			$3d^5(a^6S)6s$	f^5S	2	45967. 81		1. 24		
		1	42253. 42	1. 10				$3d^5(a^4P)4p$	y^3D°	1		46077. 09	
		2	42254. 52	1. 74	2					46109. 26		32. 17	
		3	42256. 26	2. 11	3					46174. 40		65. 14	
		4	42258. 37	2. 69	1. 55								
5	42261. 06												

Cr I—Continued

Cr I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>		
$3d^5(a^4P)4p$	w^5D°	0	46081.27			$3d^4 4s(b^4G)4p$	u^5F°	1	47877.55				
		1	46298.32	217.05				2	47917.93	40.38	0.00		
		2	46349.50	51.18				3	47974.53	56.60	1.04		
		3	46368.35	18.85				4	48014.40	39.87	1.36		
		4	46422.46	54.11				5	47985.76	-28.64	1.38		
$3d^4 4s(a^6D)5s$	f^7D	1	46448.60	76.24	2.99	$3d^4 4s(c^4D)4p$	t^5F°	1	48210.04				
		2	46524.84	112.37	1.99			2	48217.83	7.79			
		3	46637.21	145.85	1.77			3	48251.91	34.08			
		4	46783.06	175.92	1.63			4	48042.83	-209.08			
		5	46958.98		1.61			5					
$3d^5(a^4G)4p$	w^5F°	1	46678.35	-1.29		$3d^4 4s(b^4P)4p$	x^3P°	0	48226.36				
		2	46677.06	11.18				1	48331.30	104.94			
		3	46688.24	32.30	1.25			2	48458.67	127.37			
		4	46720.54	-15.56									
		5	46704.98		1.37			$3d^4 4s(a^4H)4p$	y^3H°	4	48288.37	22.02	
$3d^4 4s(a^4H)4p$	z^3G°	3	46846.77	58.26	5	48310.39	134.96						
		4	46905.03	80.84	6	48445.35							
$3d^4 4s(c^4D)4p$	u^5P°	3	46878.61	-89.09	1.68	$3d^4 4s(a^6D)5s$	f^5D	0	48488.23	19.33			
		2	46967.70	-54.05	1.83			1	48507.56	51.01			
$3d^4 4s(b^4G)4p$	x^5G°	1	47021.75		2.43			2	48558.57	103.02			
		2	47047.47	78.23	0.45			3	48661.59	162.91	1.46		
$3d^5(a^4G)4p$	y^3G°	3	47125.70	64.17	0.96	$3d^4 4s(a^4F)4p$	x^3G°	3	48515.08	47.08			
		4	47189.87	38.93				4	48562.16	224.23			
		5	47228.80	-6.53	1.27			5	48786.39				
		6	47222.27		1.44								
		$3d^5(a^4G)4p$	y^3G°	3	47048.48			6.43		$3d^4 4s(b^4P)4p$	x^3D°	3	48636.14
				4	47054.91	0.40		1	48839.90			187.68	
$3d^4 4s(b^4P)4p$	z^3S°	1	47088.40			2	49027.58	283.28					
		3	47586.06	44.37		3	49310.86						
$3d^4 4s(a^4H)4p$	z^3I°	6	47630.43	62.20		$3d^5(a^6S)7s$	g^7S	3	49177.83				
		7	47692.63					$3d^4 4s(b^4G)4p$	w^3G°	3	49370.70	83.24	
		$3d^4 4s(b^4G)4p$	x^5H°	3	47621.31					67.20		4	49453.94
4	47688.51			105.31		5	49538.06						
$3d^4 4s(b^4G)4p$	x^5H°	5	47793.82	148.47		$3d^4 4s(a^4F)4p$	w^5G°			2	49466.77	52.95	
		6	47942.29	197.89						3	49519.72	53.31	1.04
		7	48140.18					4	49573.03	44.58			
		$3d^5(b^4D)4p$	v^5F°	1	47629.66			1.85		5	49617.61	17.55	
				2	47631.51			4.74		6	49635.16		1.35
$3d^5(a^6S)6p$	w^7P°	3	47636.25	3.59	1.34			$3d^5(a^4P)4p$	y^3S°	1	49477.04		
		4	47639.84	4.92		2	49586.38			131.50			
		5	47644.76			3	49717.88			145.62			
		4	47697.44	11.15		4	49863.50						
		$3d^5(a^6S)5d$	g^7D	2	47708.59	10.49		$3d^5(b^4D)4p$	t^5P°	1	49588.97	9.11	2.48
3	47719.08					2	49598.08			214.38	1.88		
4	47702.30			1.35		3	49812.46				1.77		
$3d^5(a^6S)5d$	g^7D	3	47702.30	2.36		$3d^5(b^4D)4p$	x^3F°	4	49620.69	-29.53			
		4	47704.66	5.14				3	49650.22	-2.54			
		5	47709.80					2	49652.76				
		$3d^5(b^4D)4p$	v^5D°	0	47788.08	-15.78	0.00	$3d^5(a^4P)4p$	y^5S°	2	49822.59		2.00
				1	47772.30	13.80	1.37			$3d^4 4s(a^6D)5p$	s^5F°	1	50018.80
2	47786.10			28.30	1.39	2	50057.61	44.43					
3	47814.40			52.08	1.53	3	50102.04	108.83	1.27				
4	47866.48				1.50	4	50210.87	42.40	1.25				
				5	50253.27		1.39						

Cr I—Continued

Cr I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>				
$3d^5(b^4D)4p$	w^3D°	1	50105.54			$3d^4 4s(a^4D)5p$	q^5F°	1	54198.23						
		2	50184.10	78.56				2	54252.19	53.96					
		3	50264.48	80.38				3	54328.95	76.76					
$3d^4 4s(c^4D)4p$	u^5D°	4	50557.56		1.54		e^5F	1	54296.76		86.60				
		3	50628.11	-70.55	1.54			2	54383.36	92.93					
		2	50654.76	-26.65	1.51			3	54476.29	96.55					
		1	50662.77	-8.01	1.46			4	54572.84	87.47					
		0	50661.20	1.57	0.00			5	54660.31						
$3d^4 4s(a^4F)4p$	w^3F°	2	50890.15			$3d^5(a^2I)4p$	z^3K°	6	54316.83		88.11				
		3	50950.42	60.27				7	54404.94	93.33					
		4	51059.79	109.37				8	54498.27						
$3d^5(b^4D)4p$	w^3P°	0	51176.88			Cr II($^6S_{2/2}$)	Limit	54565							
		1	51246.87	69.99											
		2	51286.52	39.65											
$3d^5(a^2I)4p$	z^1H°	5	51401.24			$3d^4 4s(a^4D)5s$	g^5D	0							
$3d^4 4s(a^4F)4p$	t^5D°	0	51999.62					1	54671.90		146.65				
		1	52003.06	3.44				2	54818.55	168.27					
		2	52012.44	9.38				3	54986.82	222.19					
		3	52031.72	19.28		4	55209.01								
$3d^5(a^2I)4p$	y^3I°	5	52591.94			$3d^4 4s(a^2H)4p$	w^3H°	4	54786.55		62.63				
		6	52660.61	68.67				5	54799.18	87.64					
		7	52677.88	17.27				6	54886.82						
		2°	52720.07					$3d^4 4s(a^2H)4p$	z^1I°	6	54800.26				
$3d^4 4s(b^4G)4p$	x^3H°	6	52914.94		$3d^4 4s(a^4D)5s$	e^3D	1			54804.69		169.95			
		5	52885.39	29.55						2	54974.64	230.15			
		4	52963.44	-78.05				3	55204.79						
$3d^5(b^4F)4p$	r^5F°	1	53011.65			$3d^5(a^2I)4p$	v^3H°	4	54810.94		118.78				
		2	53037.52	25.87				5	54929.72	-63.15					
		3	53073.90	36.38				6	54866.57						
		4	53117.54	43.64		$3d^5(a^2D)4p$	v^3D°	1	54956.59		196.04				
		5	53172.33	54.79	1.42			2	55152.63	299.01					
3				3	55451.64										
$3d^4 4s(a^6D)4d$	e^7G	1	53148.35			$3d^5(a^2I)4p$	z^1K°	7	54970.23						
		2	53177.87	29.52				$3d^4 4s(c^2F)4p$	v^3F°	2	54992.93		108.94		
		3	53228.49	50.62						3	55101.87	105.53			
		4	53298.90	70.41		4	55207.40								
		$3d^4 4s(a^6D)4d$	e^7F	5	53393.50	94.60		$3d^5(a^2F)4p$	u^3F°	4	55120.77		-231.95		
				6	53517.85	124.35				3	55352.72	-120.95			
				7	53662.64	144.79				2	55473.67				
$3d^4 4s(a^6D)4d$	e^7F			0	53376.7			$3d^5(a^2I)4p$	y^1I°	6	55516.69				
				1	53414.1	37.4				$3d^4 4s(a^2H)4p$	x^3I°	5	55686.46		54.65
				2	53467.4	53.3						6	55741.11	57.99	
				3	53563.0	95.6		7	55799.10						
		$3d^4 4s(a^6D)4d$	e^7F	4	53703.9	140.9		$3d^5(b^2H)4p$	u^3H°	6	55908.12		33.14		
				5	53848.1	144.2				5	55874.98	-40.52			
6	54040.2			192.1		4	55915.50								
$3d^4 4s(a^6D)5p$	s^5D°			0				$3d^4 4s(a^2H)4p$	y^1H°	5	55945.08				
				1						$3d^5(b^4F)4p$	v^5G°	2	56155.12		54.69
				2	53558.05	82.69						3	56209.81	69.75	
		3	53640.74	142.03		4	56279.56					82.30			
4	53782.77			5	56361.86	87.24									
$3d^4 4s(a^2G)4p$	v^3G°	3	53804.84	122.75		6	56449.10								
		4	53927.59	150.54											
		5	54078.13												

Cr I—Continued

Cr I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	
$3d^5(a^2D)4p$	v^3P°	2	56591.88	-130.72 -79.90		$3d^5(b^4F)4p$	q^3F°	2	60253.00	73.04 41.34		
		1	56722.60					3	60326.04			
		0	56802.50					4	60367.38			
$3d^5(b^2H)4p$	u^3G°	3	56985.67	47.93 54.65		$3d^4 4s(b^2I)4p$	v^3I°	5	60427.63	99.92 129.42		
		4	57033.60					6	60527.55			
		5	57083.25					7	60656.97			
$3d^4 4s(a^2P)4p$	u^3P°	2	57087.70	-44.89 -22.00		$3d^5(b^2H)4p$	x^1I°	6	60441.42	-36.09 -14.22		
		1	57132.59					5	60467.85			
		0	57154.59						4			60503.94
$3d^4 4s(a^4D)6p$	p^5F°	1	57096.62	4.04 85.94 50.90 90.16		$3d^5(b^2F)4p$	s^3D°	3	60615.84	-14.03 -48.25		
		2	57100.66					2	60629.87			
		3	57186.60					1	60678.12			
		4	57237.50					2	60819.50			
		5	57327.66						4			60960.58
$3d^5(b^2F)4p$	t^3F°	2	57220.67	55.75 59.05		$3d^4 4s(c^2G)4p$	p^3F°	2	60819.50 60960.58	141.08		
		3	57276.42									
		4	57335.47									
$3d^5(a^4G)5s$	e^5G	2	57350.65	10.59 11.54 10.15 6.39		$3d^4 4s(c^2G)4p$	s^3H°	4	60870.63	137.44 183.57		
		3	57361.24					5	61008.07			
		4	57372.78					6	61191.64			
		5	57382.93					3	61078.28			
		6	57389.32						4			61123.20
									5			61161.35
$3d^4 4s(c^2G)4p$	t^3G°	3	57557.03	30.33 115.00		$3d^5(b^2G)4p$	r^3G°	3	61078.28	44.92 38.15		
		4	57587.36					4	61123.20			
		5	57702.36					5	61161.35			
$3d^4 4s(a^4D)5p$	r^5D°	0	57958.42	36.62 68.76 83.96 144.86		$3d^4 4s(b^4P)5s$	e^5P	1	61558.17	129.39 162.61		
		1	57995.04					2	61687.56			
		2	58063.80					3	61850.17			
		3	58147.76					2	61675.72			
		4	58292.62						3			61930.05
$3d^5(a^2D)4p$	e^3G	3	57984.94	7.21 -1.92		$3d^5(a^4G)4d$	f^5G	2		62646.60	15.36 9.04 19.96 -17.04	
		4	57992.15					3		62661.96		
		5	57990.23					4	62671.00			
								5	62690.96			
								6	62673.92			
$3d^5(a^2D)4p$	s^3F°	2	58162.84	39.81 -34.76		$3d^4 4s(b^2I)4p$	r^3H°	4	62762.06	68.20 72.77		
		3	58202.65					5	62830.26			
		4	58167.89					6	62903.03			
$3d^4 4s(a^2P)4p$	u^3D°	1	58725.28	134.95 261.92		$3d^5(b^2G)4p$	q^3H°	4	63116.80	27.56 38.58		
		2	58860.23					5	63144.36			
		3	59122.15					6	63182.94			
$3d^4 4s(a^2G)4p$	t^3H°	4	58728.29	26.29 20.78		$3d^5(d^2G)4p$	p^3H°	4	63841.81	85.46 70.59		
		5	58754.58					5	63927.27			
		6	58775.36					6	63997.86			
$3d^5(a^2F)4p$	t^3D°	1	58870.20	-98.17 152.09		$3d^4 4s(a^4H)5s$	e^5H	3	64712.04	39.38 50.66 34.22 103.98		
		2	58772.03					4	64751.42			
		3	58924.12					5	64802.08			
$3d^4 4s(a^2G)4p$	r^3F°	2	59357.90	59.11 70.70		$3d^5(d^2G)4p?$	p^3G°	3	66008.95	85.11 86.28		
		3	59417.01					4	66094.06			
		4	59487.71					5	66180.34			
$3d^5(b^2H)4p$	w^3I°	5	59806.27	78.00 73.19								
		6	59884.27									
		7	59957.46									
$3d^5(b^2H)4p$	x^1H°	5	60005.60									
$3d^4 4s(a^2P)4p$	x^3S°	1	60084.09									

Cr I OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms			
$3d^4 4s^3$	a^3P	\bar{a}^5D c^3D	a^3F	b^3G a^1G a^3H a^1I
$3d^6$		c^5D		
		$ns (n \geq 4)$		$np (n \geq 4)$
$3d^5(a^6S)nx$	$\{ \begin{matrix} a, e, f, g^1S \\ a, e, f, g^3S \end{matrix} \}$			z, x, w^7P^o z^5P^o
$3d^4 4s(a^6D)nx$	$\{ \begin{matrix} f^7D \\ f^5D \end{matrix} \}$			y^7P^o z^7D^o z^5D^o z, s, p^5F^o
$3d^4 4s(a^4D)nx$	$\{ \begin{matrix} g^5D \\ e^3D \end{matrix} \}$			x^5P^o z^3P^o y, r^5D^o z^3D^o y, q^5F^o z^3F^o
$3d^5(a^4G)nx$	$\{ \begin{matrix} a^5P \\ b^3P \end{matrix} \}$		a, e^5G a^3G	w^5F^o y^5F^o z^5G^o y^3G^o y^5H^o z^3H^o
$3d^5(a^4P)nx$	$\{ \begin{matrix} a^5P \\ b^3P \end{matrix} \}$			w^5P^o y^3P^o w^5D^o y^3D^o
$3d^5(b^4D)nx$	$\{ \begin{matrix} b^5D \\ a^3D \end{matrix} \}$			t^5P^o w^3P^o v^5F^o w^3D^o x^5F^o x^3F^o
$3d^4 4s(b^4P)nx$	$\{ e^5P \}$			v^5P^o x^3P^o x^5D^o x^3D^o
$3d^5(a^2I)nx$	$\{ \begin{matrix} a^3I \\ b^1I \end{matrix} \}$			y^5S^o y^3S^o
$3d^4 4s(a^4H)nx$	$\{ e^5H \}$			v^3H^o z^1H^o y^3I^o y^1I^o z^3K^o z^1K^o
$3d^4 4s(a^4F)nx$				y^5G^o z^3G^o z^5H^o y^3H^o z^5I^o z^3I^o
$3d^5(a^2D)nx$				x^5P^o w^3P^o w^5G^o x^3G^o
		b^3D		x^5F^o w^3F^o s^3F^o

$3d^5(a^3F)nx$	{	b^3F	t^3D^o	u^3F^o	u^3F^o	
$3d^5(b^3F)nx$		a^5F c^3F		r^5F^o q^3F^o	v^5G^o	
$3d^4 4s(b^4G)nx$				u^5F^o	x^5G^o w^3G^o	x^5H^o x^3H^o
$3d^4 4s(a^2H)nx$	{		x^3S^o	u^3P^o	u^3D^o	x^3I^o z^1I^o
$3d^4 4s(a^2P)nx$						
$3d^5(b^3F)nx$	{	d^3F		t^3F^o	s^3D^o	
$3d^5(b^3H)nx$			b^3H a^1H			
$3d^4 4s(a^2G)nx$	{			r^3F^o	v^3G^o	t^3H^o
$3d^4 4s(c^4D)nx$				u^5P^o	t^5F^o	u^5D^o
$3d^5(b^2G)nx$	{				c^3G b^1G	r^3G^o q^3H^o
$3d^4 4s(c^2G)nx$					p^3F^o	t^3G^o
$3d^4 4s(c^2F)nx$	{					v^3F^o
$3d^4 4s(b^2I)nx$						
$3d^5(d^2G)nx$						v^3I^o
$3d^5(a^6S)nx$	{	e, g^7D e^5D	$nd (n \geq 4)$			
$3d^4 4s(a^6D)nx$		e^7F				
$3d^5(a^4G)nx$		e^7G f^5G				

*For predicted terms in the spectra of the Cr I isoelectronic sequence, see Vol. II, Introduction.

Cr II—Continued

Cr II—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$3d^6$	b^2H	$4\frac{1}{2}$ $5\frac{1}{2}$	35610. 50 35707. 66	97. 16		$3d^3 4s^2$	e^2G	$3\frac{1}{2}$ $4\frac{1}{2}$	54444. 19 54678. 95	234. 76	
$3d^5$	a^2G	$3\frac{1}{2}$ $4\frac{1}{2}$	36101. 82 36272. 66	170. 84		$3d^3 4s^2$	c^4P	$2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	55023. 30 55081. 77	—58. 4	
$3d^4(a^3D)4s$	c^4D	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	38269. 67 38315. 00 38362. 56 38396. 36	—45. 33 —47. 56 —33. 80		$3d^3 4s^2$	b^2P	$0\frac{1}{2}$ $1\frac{1}{2}$	59130. 51		
$3d^4(a^1G)4s$	b^2G	$3\frac{1}{2}$ $4\frac{1}{2}$	38509. 07 38563. 15	54. 08	0. 910 1. 100	$3d^3 4s^2$	e^2D	$1\frac{1}{2}$ $2\frac{1}{2}$	59527. 05 59570. 23	43. 18	
$3d^4(a^3G)4s$	c^2G	$3\frac{1}{2}$ $4\frac{1}{2}$	39684. 00 39824. 52	140. 52		$3d^4(a^3H)4p$	z^4H°	$3\frac{1}{2}$ $4\frac{1}{2}$ $5\frac{1}{2}$ $6\frac{1}{2}$	63601. 20 63706. 62 63849. 11 64030. 85	105. 42 142. 49 181. 74	0. 680 1. 030 1. 138 1. 234
$3d^5$	c^2F	$2\frac{1}{2}$ $3\frac{1}{2}$	39742. 36 39877. 28	134. 92		$3d^4(a^3P)4p$	y^4D°	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	63802. 41 64061. 82 64448. 84 64924. 30	259. 41 387. 02 475. 46	0. 000 1. 199 1. 380 1. 411
$3d^4(a^1I)4s$	b^2I	$6\frac{1}{2}$ $5\frac{1}{2}$	40202. 14 40228. 44	—26. 30		$3d^4(a^3P)4p$	z^2S°	$0\frac{1}{2}$	65029. 67		
$3d^4(a^1S)4s$	a^2S	$0\frac{1}{2}$	40415. 34			$3d^4(a^3H)4p$	z^4G°	$2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$ $5\frac{1}{2}$	65156. 84 65257. 03 65384. 04 65709. 53	100. 19 127. 01 325. 49	0. 593 0. 920 1. 120 1. 265
$3d^4(a^3D)4s$	b^2D	$2\frac{1}{2}$ $1\frac{1}{2}$	42898. 12 42986. 73	—88. 61		$3d^4(a^3H)4p$	z^4I°	$4\frac{1}{2}$ $5\frac{1}{2}$ $6\frac{1}{2}$ $7\frac{1}{2}$	65217. 61 65419. 95 65618. 41 65812. 63	202. 34 198. 46 194. 22	
$3d^5$	b^2S	$0\frac{1}{2}$	44307. 44			$3d^4(a^3H)4p$	z^2G°	$3\frac{1}{2}$ $4\frac{1}{2}$	65543. 06 65680. 15	137. 09	
$3d^4(a^1D)4s$	c^2D	$1\frac{1}{2}$ $2\frac{1}{2}$	45669. 54 45730. 74	61. 20		$3d^4(a^3P)4p$	y^4P°	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	66256. 77 66355. 13 66727. 16	98. 36 372. 03	2. 545 1. 671 1. 502
$3d^4(a^5D)4p$	z^6F°	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$ $5\frac{1}{2}$	46823. 64 46905. 52 47040. 54 47227. 50 47464. 94 47751. 98	81. 88 135. 02 186. 96 237. 44 287. 04	—0. 689 1. 124 1. 314 1. 378 1. 416	$3d^4(a^3H)4p$	z^2G°	$3\frac{1}{2}$ $4\frac{1}{2}$	65543. 06 65680. 15	137. 09	
$3d^5$	d^2D	$2\frac{1}{2}$ $1\frac{1}{2}$	47354. 63 47372. 75	—18. 12		$3d^4(a^3P)4p$	y^4P°	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	66256. 77 66355. 13 66727. 16	98. 36 372. 03	2. 545 1. 671 1. 502
$3d^4(a^5D)4p$	z^6P°	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	48399. 19 48491. 39 48632. 36	92. 20 140. 97	2. 382 1. 875 1. 710	$3d^4(a^3P)4p$	z^2D°	$1\frac{1}{2}$ $2\frac{1}{2}$	66649. 71 67012. 28	362. 57	
$3d^4(a^5D)4p$	z^4P°	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	48749. 57 49006. 15 49351. 96	256. 58 345. 81	2. 844 1. 802 1. 628	$3d^4(a^3P)4p$	z^2P°	$0\frac{1}{2}$ $1\frac{1}{2}$	66872. 12 67070. 48	198. 36	
$3d^4(a^5D)4p$	z^6D°	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	49493. 00 49564. 80 49706. 47 49646. 25 49838. 43	71. 80 141. 67 —60. 22 192. 18	3. 155 1. 824 1. 624 1. 577 1. 570	$3d^4(a^3F)4p$	y^4G°	$2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$ $5\frac{1}{2}$	67344. 42 67334. 20 67353. 60 67369. 33	—10. 22 19. 40 15. 73	
$3d^4(a^5D)4p$	z^6D°	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	49493. 00 49564. 80 49706. 47 49646. 25 49838. 43	71. 80 141. 67 —60. 22 192. 18	3. 155 1. 824 1. 624 1. 577 1. 570	$3d^4(a^3F)4p$	y^4F°	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	67379. 92 67387. 30 67393. 80 67448. 82	7. 38 6. 50 55. 02	
$3d^4(a^1F)4s$	d^2F	$3\frac{1}{2}$ $2\frac{1}{2}$	50667. 33 50687. 63	—20. 30		$3d^4(a^3H)4p$	z^2I°	$5\frac{1}{2}$ $6\frac{1}{2}$	67506. 34 67589. 06	82. 72	
$3d^4(a^5D)4p$	z^4F°	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	51584. 44 51669. 75 51789. 21 51943. 04	85. 31 119. 46 153. 83	0. 406 1. 025 1. 248 1. 338	$3d^4(a^3F)4p$	x^4D°	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	67859. 91 67870. 50 67868. 05 67875. 68	10. 59 —2. 45 7. 63	
$3d^5$	d^2G	$3\frac{1}{2}$ $4\frac{1}{2}$	52298. 12 52321. 30	23. 18		$3d^4(a^3P)4p$	z^4S°	$1\frac{1}{2}$	68305. 73		1. 978
$3d^3 4s^2$	c^4F	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	53051. 55 53271. 07 53566. 22 53923. 57	219. 52 295. 15 357. 35		$3d^4(a^3H)4p$	z^2H°	$4\frac{1}{2}$ $5\frac{1}{2}$	68477. 11 68737. 99	260. 88	
$3d^4(a^5D)4p$	z^4D°	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	54418. 08 54499. 70 54625. 76 54784. 67	81. 62 126. 06 158. 91	0. 007 1. 178 1. 376 1. 430	$3d^4(a^3F)4p$	z^2F°	$2\frac{1}{2}$ $3\frac{1}{2}$	68583. 44 68760. 00	176. 56	
						$3d^4(a^3G)4p$	y^4H°	$3\frac{1}{2}$ $4\frac{1}{2}$ $5\frac{1}{2}$ $6\frac{1}{2}$	68843. 51 68992. 55 69170. 60 69388. 40	149. 04 178. 05 217. 80	

Cr II—Continued

Cr II—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$3d^4(a^3G)4p$	x^4F°	$1\frac{1}{2}$	69348.36			$3d^4(a^1D)4p$	w^2P°	$0\frac{1}{2}$	82854.00		
		$2\frac{1}{2}$	69478.06	129.70	$1\frac{1}{2}$			82920.03	66.03		
		$3\frac{1}{2}$	69506.16	28.10	$3d^4(a^5D)5s$		e^4D	$0\frac{1}{2}$	84208.28	110.26	
		$4\frac{1}{2}$	69498.27	-7.89				$1\frac{1}{2}$	84318.54	175.66	
$3d^4(a^3F)4p$	y^2D°	$1\frac{1}{2}$	69638.77			$2\frac{1}{2}$	84494.20	231.76			
		$2\frac{1}{2}$	69954.20	315.43	$3\frac{1}{2}$	84725.96					
$3d^4(a^3F)4p$	y^2G°	$3\frac{1}{2}$	69903.46			$3d^4(a^1F)4p$	u^2F°	$2\frac{1}{2}$	84604.99	72.40	
		$4\frac{1}{2}$	70107.83	204.37	$3\frac{1}{2}$			84677.39			
$3d^4(a^3G)4p$	x^4G°	$2\frac{1}{2}$	70317.04			$3d^4(a^1F)4p$	v^2G°	$3\frac{1}{2}$	85573.42	366.08	
		$3\frac{1}{2}$	70427.22	110.18	$4\frac{1}{2}$			85939.50			
		$4\frac{1}{2}$	70679.22	252.00	$3d^4(a^1F)4p$	v^2D°	$2\frac{1}{2}$	86507.38	-3.70		
		$5\frac{1}{2}$	70879.95	200.73			$1\frac{1}{2}$	86511.08			
$3d^4(a^3G)4p$	y^2H°	$4\frac{1}{2}$	70394.46			$3d^4(a^5D)4d$	e^6G	$1\frac{1}{2}$	86594.82	59.61	
		$5\frac{1}{2}$	70399.04	4.58	$2\frac{1}{2}$			86654.43	83.93		
$3d^4(a^3G)4p$	y^2F°	$2\frac{1}{2}$	70584.64			$3\frac{1}{2}$	86738.36	108.94			
		$3\frac{1}{2}$	70852.24	267.60	$4\frac{1}{2}$	86847.30	133.12				
$3d^4(a^3G)4p$	x^2G°	$3\frac{1}{2}$	72648.79			$5\frac{1}{2}$	86980.42	156.92			
		$4\frac{1}{2}$	72716.91	68.12	$6\frac{1}{2}$	87137.34					
$3d^4(a^3D)4p$	w^4D°	$0\frac{1}{2}$	73406.68			$3d^4(a^5D)4d$	e^6P	$1\frac{1}{2}$	86667.95	23.82	
		$1\frac{1}{2}$	73411.94	5.26	$2\frac{1}{2}$			86691.77	90.30		
		$2\frac{1}{2}$	73436.27	24.33	$3d^4(a^5D)4d$	f^6D	$0\frac{1}{2}$	87314.0	40.1		
		$3\frac{1}{2}$	73485.60	49.33			$1\frac{1}{2}$	87354.62	58.65		
$3d^4(a^3D)4p$	x^2F°	$3\frac{1}{2}$	74114.48			$2\frac{1}{2}$	87413.27	101.83			
		$2\frac{1}{2}$	74436.14	-321.66	$3\frac{1}{2}$	87515.10	172.56				
$3d^4(a^3D)4p$	w^4F°	$1\frac{1}{2}$	74273.48			$3d^4(a^5D)4d$	e^6F	$0\frac{1}{2}$	87542.12	52.48	
		$2\frac{1}{2}$	74318.86	45.38	$1\frac{1}{2}$			87594.60	71.40		
		$3\frac{1}{2}$	74423.84	104.98	$2\frac{1}{2}$	87666.00	92.88				
		$4\frac{1}{2}$	74504.51	80.67	$3\frac{1}{2}$	87758.88	189.82				
$3d^4(a^1I)4p$	y^2I°	$5\frac{1}{2}$	74421.76			$4\frac{1}{2}$	87948.70	52.62			
		$6\frac{1}{2}$	74424.35	2.59	$5\frac{1}{2}$	88001.32					
$3d^4(a^1I)4p$	x^2H°	$4\frac{1}{2}$	74455.90			$3d^4(a^5D)4d$	e^4G	$2\frac{1}{2}$	89056.10	118.00	
		$5\frac{1}{2}$	74707.42	251.52	$3\frac{1}{2}$			89174.10	151.50		
$3d^4(a^3D)4p$	x^4P°	$2\frac{1}{2}$	74484.25			$4\frac{1}{2}$	89325.60	183.03			
		$1\frac{1}{2}$	74718.05	-233.80	$3d^4(a^5D)4d$	e^4P	$0\frac{1}{2}$	88426.2	210.5		
		$0\frac{1}{2}$	74920.80	-202.75			$1\frac{1}{2}$	88636.7	286.5		
$3d^4(a^1I)4p$	z^2K°	$6\frac{1}{2}$	74743.33			$2\frac{1}{2}$	88923.2				
		$7\frac{1}{2}$	74958.80	215.47	$3d^4(a^5D)4d$	f^4D	$0\frac{1}{2}$	89269.88	67.82		
$3d^4(a^3D)4p$	y^2P°	$0\frac{1}{2}$	74854.08					$1\frac{1}{2}$	89337.70	137.38	
		$1\frac{1}{2}$	74984.93	130.85	$2\frac{1}{2}$	89475.08	146.17				
$3d^4(a^1G)4p$	w^2G°	$3\frac{1}{2}$	75716.74			$3\frac{1}{2}$	89621.25				
		$4\frac{1}{2}$	75810.10	93.36	$3d^4(a^5D)4d$	e^4F	$1\frac{1}{2}$	90512.50	78.60		
$3d^4(a^1G)4p$	w^2F°	$3\frac{1}{2}$	76879.03					$2\frac{1}{2}$	90591.10	134.40	
		$2\frac{1}{2}$	76987.78	-108.75	$3\frac{1}{2}$	90725.50	125.50				
$3d^4(a^1G)4p$	w^2H°	$5\frac{1}{2}$	77078.96			$3d^4(b^3F)4p$	u^2G°	$3\frac{1}{2}$	90986.31	117.05	
		$4\frac{1}{2}$	77270.40	-191.44	$4\frac{1}{2}$			91103.36			
$3d^4(a^1S)4p$	x^2P°	$1\frac{1}{2}$	77713.66			$3d^4(b^3F)4p$	u^2D°	$2\frac{1}{2}$	91426.31	-130.23	
		$0\frac{1}{2}$	77777.58	-63.92	$1\frac{1}{2}$			91556.54			
$3d^4(a^3D)4p$	x^2D°	$2\frac{1}{2}$	77935.24			$3d^4(a^5D)4d$	e^6S	$2\frac{1}{2}$	91954.78		
		$1\frac{1}{2}$	78109.64	-174.40	$3d^4(a^3G)5s$			f^4G	$2\frac{1}{2}$		
$3d^4(a^1D)4p$	w^2D°	$1\frac{1}{2}$	80288.25				$3\frac{1}{2}$				
		$2\frac{1}{2}$	80420.43	132.18	$4\frac{1}{2}$	105365.2	56.7				
$3d^4(a^1D)4p$	v^2F°	$2\frac{1}{2}$	81232.91			$5\frac{1}{2}$	105421.9				
		$3\frac{1}{2}$	81432.36	199.45							
$3d^4(a^5D)5s$	e^6D	$0\frac{1}{2}$	82692.26			-----	-----	-----	-----	-----	-----
		$1\frac{1}{2}$	82763.45	71.19	$Cr III (a^5D_0)$	Limit	-----	133060			
		$2\frac{1}{2}$	82881.30	117.85							
		$3\frac{1}{2}$	83041.20	159.90							
		$4\frac{1}{2}$	83240.20	199.00							

Cr II OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms	
$3d^5$	$\left\{ \begin{array}{l} b^6S \\ b^2S \end{array} \right.$	$\begin{array}{l} a^4P \quad b^4D \quad b^4F \quad a^4G \\ a^2D \quad a^2F \quad a^2G \\ c^2D \quad c^2F \quad d^2G \quad b^2H \quad a^2I \end{array}$
$3d^3 4s$	$\left\{ \begin{array}{l} c^4P \\ b^2P \end{array} \right.$	$\begin{array}{l} e^2D \quad c^4F \\ e^2G \end{array}$
	$ns (n \geq 4)$	$np (n \geq 4)$
$3d^4(a^5D)nx$	$\left\{ \begin{array}{l} a,e^6D \\ a,e^4D \end{array} \right.$	$\begin{array}{l} z^6P^\circ \quad z^6D^\circ \quad z^6F^\circ \\ z^4P^\circ \quad z^4D^\circ \quad z^4F^\circ \end{array}$
$3d^4(a^3P)nx$	$\left\{ \begin{array}{l} b^4P \\ a^2P \end{array} \right.$	$\begin{array}{l} z^4S^\circ \quad y^4P^\circ \quad y^4D^\circ \\ z^2S^\circ \quad z^2P^\circ \quad z^2D^\circ \end{array}$
$3d^4(a^3H)nx$	$\left\{ \begin{array}{l} a^4H \\ a^2H \end{array} \right.$	$\begin{array}{l} z^4G^\circ \quad z^4H^\circ \quad z^4I^\circ \\ z^2G^\circ \quad z^2H^\circ \quad z^2I^\circ \end{array}$
$3d^4(a^3F)nx$	$\left\{ \begin{array}{l} a^4F \\ b^2F \end{array} \right.$	$\begin{array}{l} x^4D^\circ \quad y^4F^\circ \quad y^4G^\circ \\ y^2D^\circ \quad z^2F^\circ \quad y^2G^\circ \end{array}$
$3d^4(a^3G)nx$	$\left\{ \begin{array}{l} b,f^4G \\ c^2G \end{array} \right.$	$\begin{array}{l} x^4F^\circ \quad x^4G^\circ \quad y^4H^\circ \\ y^2F^\circ \quad x^2G^\circ \quad y^2H^\circ \end{array}$
$3d^4(a^3D)nx$	$\left\{ \begin{array}{l} c^4D \\ b^2D \end{array} \right.$	$\begin{array}{l} x^4P^\circ \quad w^4D^\circ \quad w^4F^\circ \\ y^2P^\circ \quad x^2D^\circ \quad x^2F^\circ \end{array}$
$3d^4(a^1G)nx$	b^2G	$\begin{array}{l} w^2F^\circ \quad w^2G^\circ \quad w^2H^\circ \end{array}$
$3d^4(a^1I)nx$	b^2I	$\begin{array}{l} x^2H^\circ \quad y^2I^\circ \quad z^2K^\circ \end{array}$
$3d^4(a^1S)nx$	a^2S	x^2P°
$3d^4(a^1D)nx$	c^2D	$\begin{array}{l} w^2P^\circ \quad w^2D^\circ \quad v^2F^\circ \end{array}$
$3d^4(a^1F)nx$	d^2F	$\begin{array}{l} v^2D^\circ \quad u^2F^\circ \quad v^2G^\circ \end{array}$
$3d^4(b^3F)nx$		$\begin{array}{l} u^2D^\circ \quad u^2G^\circ \end{array}$
	$nd (n \geq 4)$	
$3d^4(a^5D)nx$	$\left\{ \begin{array}{l} e^6S \quad e^6P \quad f^6D \quad e^6F \quad e^6G \\ e^4P \quad f^4D \quad e^4F \quad e^4G \end{array} \right.$	

*For predicted terms of the VI isoelectronic sequence, see Volume I, pp. xxxviii and xxxix.

Cr III

(Ti I sequence; 22 electrons)

Z=24

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^4 {}^5D_0$ $a {}^5D_0$ 249700 cm^{-1}

I. P. 30.95 volts

The early work on Cr III by White and by Bowen resulted in the classification of 112 lines from 16 triplet and quintet terms. Bowen found strong intersystem combinations connecting the two systems of terms.

Recently, F. L. Moore, Jr., has extended the observations and analysis, utilizing also unpublished wavelengths by Kiess and Lang. He has observed more than 2300 lines between 712 Å and 3924 Å, of which approximately 750 Cr III lines are classified. About 60 terms are known, including eight singlets, also connected with the rest by observed intersystem combinations. The limit quoted here has been estimated by Catalán from a study of isoelectronic data. From the $ns^{5,3}F$ series ($n=4, 5$) Moore derives the value 253064. White estimated the value of the limit as about 222000 cm^{-1} .

This spectrum is of considerable astrophysical interest. For this reason Moore has made the present investigation and furnished his results especially for inclusion here. All of the terms in the table have been taken from his manuscript.

REFERENCES

- H. E. White, Phys. Rev. **33**, 914 (1929). (T) (C L)
 I. S. Bowen, Phys. Rev. **52**, 1153 (1937). (T) (C L)
 F. L. Moore, Jr., unpublished material (May 1951). (I P) (T) (C L)
 M. A. Catalán, unpublished material (April 1951). (I P)

Cr III

Cr III

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval	
$3d^4$	$a {}^5D$	0	0.0		$3d^3(a {}^4F)4s$	$c {}^3F$	2	56650.5		
		1	59.9	59.9			3	56992.2	341.7	
		2	181.9	122.0			4	57422.1	429.9	
		3	355.8	173.9		$a {}^5P$	1	63038.6	133.6	
		4	575.0	219.2			2	63172.2	248.1	
$3d^4$	$a {}^3P$	0	16770.9		$3d^3(a {}^2G)4s$	$b {}^3G$	3	65890.9	136.9	
		1	17167.4	396.5			4	66027.8	194.9	
		2	17850.0	682.6			5	66222.7		
$3d^4$	$a {}^3H$	4	17272.8	122.7	$3d^3(a {}^4P)4s$	$c {}^3P$	0			
		5	17395.5	133.3			1	66602.2	579.5	
		6	17528.8				2	67181.7		
$3d^4$	$a {}^3F$	2	18451.0	59.0	$3d^3(a {}^2G)4s$	$c {}^1G$	4	68575.4		
		3	18510.0	71.6			$d {}^3P$	0	69510.6	270.3
		4	18581.6					1	69780.9	510.9
$3d^4$	$a {}^3G$	3	20702.0	149.3	2	70291.8				
		4	20851.3	143.3	$3d^3(a {}^2D)4s$	$b {}^3D$	1	70980.2	185.9	
		5	20994.6				2	71166.1	155.7	
$3d^4$	$a {}^3D$	3	25725.8	-55.2	$3d^3(a {}^2H)4s$	$a {}^1H$	5	71408.2		
		2	25781.0	-67.2			$b {}^3H$	4	71676.2	59.8
		1	25848.2					5	71736.0	132.5
$3d^4$	$a {}^1F$	3	41052.8		$3d^3(a {}^2F)4s$	$d {}^3F$	2			
		$b {}^3F$	4	43286.4			-35.3	3	81664.5	591.4
			3	43321.7			17.6	4	82255.9	
$3d^4$	$b {}^3P$	2	43304.1		$3d^3(a {}^4F)4p$	$z {}^5G^\circ$	2	93765.6	263.6	
		0	48759.7	163.6			3	94029.2	345.5	
		1	48923.3	544.2			4	94374.7	424.9	
$a {}^5F$	2	49467.5		5			94799.6	504.5		
	1	49491.0	135.8	6			95304.1			
$3d^3(a {}^4F)4s$	$a {}^5F$	2	49626.8	200.9						
		3	49827.7	261.5						
		4	50089.2	319.0						
		5	50408.2							
		4	50408.2							
$3d^4$	$b {}^1G$	4	51093.1							

Cr III—Continued

Cr III—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3d^3(a^4F)4p$	z^5D°	0	95778.6		$3d^3(a^2H)4p$	x^3G°	5	120699.7	
		1	96148.5	369.9			4	120747.8	-48.1
		2	96385.9	237.4			3	120765.3	-17.5
		3	96713.1	327.2					
		4	97097.1	384.0					
$3d^3(a^4F)4p$	z^5F°	1	96773.5		$3d^3(a^2F)4p$	w^3F°	2	128753.9	
		2	96921.4	147.9			3	128782.2	28.3
		3	97120.3	198.9			4	128849.4	67.2
		4	97358.7	238.4					
		5	97618.3	259.6					
$3d^3(a^4F)4p$	z^3D°	1	97076.9		$3d^3(a^2F)4p$	w^3G°	3	131118.1	
		2	97306.1	229.2			4	131267.3	149.2
		3	97683.0	376.9			5	131448.8	181.5
$3d^3(a^4F)4p$	z^3G°	3	99840.7		$3d^3(a^2F)4p$	v^3D°	1		
		4	100099.5	258.8			2	138362.4?	
		5	100421.4	321.9			3	138975.5	613.1
$3d^3(a^4F)4p$	z^3F°	2	101443.7		$3d^3(b^2D)4p$	v^3F°	2		
		3	101745.3	301.6			3	150972.5	
		4	102099.6	354.3			4		
$3d^3(a^4P)4p$	z^5P°	1	108248.3		$3d^3(a^4F)4d$	e^5H	3	152927.3	
		2	108459.2	210.9			4	153099.1	171.8
		3	108793.3	334.1			5	153314.6	215.5
$3d^3(a^4P)4p$	y^5D°	0	109145.9		$3d^3(a^4F)4d$	e^5F	1	153729.8	
		1	109237.4	91.5			2	153893.5	163.7
		2	109434.2	196.8			3	154085.7	192.2
		3	109721.4	287.2			4	154457.1	371.4
		4	110154.0	432.6			5	154849.6	392.5
$3d^3(a^2G)4p$	z^3H°	4	109533.5		$3d^3(a^4F)4d$	e^5D	0		
		5	109943.9	410.4			1		
		6	110505.0	561.1			2		
$3d^3(a^4P?)4p$	z^3P°	0			$3d^3(a^4F)5s$	f^5F	1	157303.2	
		1					2	157435.1	131.9
		2	109569.7	268.3			3	157637.3	202.2
$3d^3(a^2G)4p$	y^3G°	3	111375.0		$3d^3(a^4F)4d$	e^3H	4	157908.0	
		4	111643.3	210.7			5	158241.8	333.8
		5	111854.0						
$3d^3(a^2G)4p$	y^3F°	4	112371.2		$3d^3(a^4F)4d$	e^3G	3	157668.9	
		3	112466.2	-95.0			4	158066.7	376.2
		2	112398.5	67.7			5	158442.9	
$3d^3(a^4P)4p$	z^5S°	2	113355.7		$3d^3(a^4F)5s$	f^3F	2	159031.8	
							3	159375.6	343.8
$3d^3(a^2G)4p$	z^1G°	4	114354.9		$3d^3(a^4F)4d$	e^3F	2	158328.3	
		3	114715.9	465.9			3	158463.6	135.3
		2	115181.8	370.9			4	158623.7	160.1
$3d^3(a^4P)4p$	y^3D°	1	114715.9		$3d^3(a^2G)4d$	f^3H	4	168582.3	
		2	115181.8	465.9			5		
		3	115552.7	370.9			6		
$3d^3(a^2H)4p$	y^3H°	4	115570.2		$3d^3(a^2H)4d$	g^3H	4	173565.9	
		5	115668.9	98.7			5	173926.6	360.7
		6	115843.5	174.6			6		
$3d^3(a^2D)4p$	x^3F°	2	116391.6		$3d^3(a^2H)4d$	f^3I	5	173200.5	
		3	116532.1	140.5			6	173662.8	462.3
		4	116967.0	434.9			7	173876.3	213.5
$3d^3(a^2H)4p$	y^1G°	4	117099.3		$Cr\ IV\ (a^4F_{1/2})$	<i>Limit</i>	-----	249700	
		5	117145.2						
		6	117487.5	342.3					
		7	117922.3	434.8					
$3d^3(a^2G)4p$	z^1H°	5	117186.9						
$3d^3(a^2D)4p$	w^3D°	1	118055.2						
		2	118423.3	368.1					
		3	118598.2	174.9					
$3d^3(a^2H)4p$	y^1H°	5	119039.9						

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ +	Observed Terms		
3d ⁴	$\left\{ \begin{array}{llll} a^5D & a^3F & a^3G & a^3H \\ a^3P & a^3D & b^3F & \\ b^3P & & a^1F & b^1G \end{array} \right.$		
	<i>ns</i> (<i>n</i> ≥ 4)	<i>np</i> (<i>n</i> ≥ 4)	<i>nd</i> (<i>n</i> ≥ 4)
3d ³ (a ⁴ F) <i>nx</i>	$\left\{ \begin{array}{l} a, f^5F \\ c, f^3F \end{array} \right.$	$\begin{array}{lll} z^5D^\circ & z^5F^\circ & z^5G^\circ \\ z^3D^\circ & z^3F^\circ & z^3G^\circ \end{array}$	$\begin{array}{llll} e^5D & e^5F & e^5G & e^5H \\ & e^3F & e^3G & e^3H \end{array}$
3d ³ (a ⁴ P) <i>nx</i>	$\left\{ \begin{array}{l} a^5P \\ c^3P \end{array} \right.$	$\begin{array}{lll} z^5S^\circ & z^5P^\circ & y^5D^\circ \\ & z^3P^\circ & y^3D^\circ \end{array}$	
3d ³ (a ² P) <i>nx</i>	<i>d</i> ³ P		
3d ³ (a ² G) <i>nx</i>	$\left\{ \begin{array}{l} b^3G \\ c^1G \end{array} \right.$	$\begin{array}{lll} y^3F^\circ & y^3G^\circ & z^3H^\circ \\ & z^1G^\circ & z^1H^\circ \end{array}$	<i>f</i> ³ H
3d ³ (a ² D) <i>nx</i>	<i>b</i> ³ D	$\begin{array}{ll} w^3D^\circ & x^3F^\circ \end{array}$	
3d ³ (a ² H) <i>nx</i>	$\left\{ \begin{array}{l} b^3H \\ a^1H \end{array} \right.$	$\begin{array}{lll} & x^3G^\circ & y^3H^\circ & z^3I^\circ \\ & y^1G^\circ & y^1H^\circ & \end{array}$	<i>g</i> ³ H <i>f</i> ³ I
3d ³ (a ² F) <i>nx</i>	<i>d</i> ³ F	$\begin{array}{lll} v^3D^\circ & w^3F^\circ & w^3G^\circ \end{array}$	
3d ³ (b ² D) <i>nx</i>		<i>v</i> ³ F ^o	

*For predicted terms in the spectra of the Ti I isoelectronic sequence, see Volume I, p. xxxvii.

Cr IV

(Sc I sequence; 21 electrons)

Z=24

Ground state 1s² 2s² 2p⁶ 3s² 3p⁶ 3d³ ⁴F_{1/2}

*a*⁴F_{1/2} 400000? cm⁻¹

I. P. 49.6 volts

The analysis was begun by White in 1929 and carried further by Bowen in 1937. It has recently been revised and extended by F. L. Moore, Jr., especially for inclusion here. The new observations extend from 500 Å to 2050 Å. At present Moore has more than 200 classified lines. The doublet and quartet terms are connected by observed intersystem combinations.

Moore's provisional value of the limit, derived from the *ns* ^{4,2}F series (*n*=4,5), is quoted above.

The prefix "a" has here been ascribed by the writer to the limit terms ¹G and ¹S, although these terms have not yet been observed in Cr v.

REFERENCES

- H. E. White, Phys. Rev. **33**, 672 (1929). (I P) (T) (C L)
 I. S. Bowen, Phys. Rev. **52**, 1153 (1937). (T) (C L)
 F. L. Moore, Jr., unpublished material (May 1951). (I P) (T) (C L)

Cr IV

Cr IV

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3d^3$	a^4F	$1\frac{1}{2}$	0		$3d^2(a^3P)4p$	y^4D°	$0\frac{1}{2}$		
		$2\frac{1}{2}$	244	244			$1\frac{1}{2}$	173674	429
		$3\frac{1}{2}$	561	317			$2\frac{1}{2}$	174103	751
		$4\frac{1}{2}$	956	395			$3\frac{1}{2}$	174854	
$3d^3$	a^4P	$0\frac{1}{2}$	14072		$3d^2(a^1D)4p$	y^2F°	$2\frac{1}{2}$		
		$1\frac{1}{2}$	14215	143			$3\frac{1}{2}$	175166	
		$2\frac{1}{2}$	14481	266					
$3d^3$	a^2P	$1\frac{1}{2}$	14185		$3d^2(a^1G)4p$	y^2G°	$3\frac{1}{2}$	177926	
		$0\frac{1}{2}$	14317	-132			$4\frac{1}{2}$	178036	110
$3d^3$	a^2G	$3\frac{1}{2}$	15064		$3d^2(a^1S)4p$	x^2P°	$0\frac{1}{2}$	182029	
		$4\frac{1}{2}$	15414	350			$1\frac{1}{2}$	182114	85
$3d^3$	a^2D	$1\frac{1}{2}$	20218		$3d^2(a^1G)4p$	z^2H°	$4\frac{1}{2}$	182391	
		$2\frac{1}{2}$	20667	449			$5\frac{1}{2}$	183453	62
$3d^3$	a^2H	$4\frac{1}{2}$	21078		$3d^2(a^1G)4p$	x^2F°	$2\frac{1}{2}$		
		$5\frac{1}{2}$	21328	250			$3\frac{1}{2}$	184868	
$3d^3$	a^2F	$3\frac{1}{2}$	36490		$3d^2(a^3F)4d$	e^4G	$2\frac{1}{2}$	232573	
		$2\frac{1}{2}$	37062	-572			$3\frac{1}{2}$	232905	332
$3d^2(a^3F)4s$	b^4F	$1\frac{1}{2}$	104003		$3d^2(a^3F)4d$	e^2G	$3\frac{1}{2}$	232889	
		$2\frac{1}{2}$	104265	262			$4\frac{1}{2}$	233521	632
		$3\frac{1}{2}$	104637	372					
		$4\frac{1}{2}$	105113	476					
$3d^2(a^3F)4s$	b^2F	$2\frac{1}{2}$	109950		$3d^2(a^3F)4d$	e^4H	$3\frac{1}{2}$	233365	
		$3\frac{1}{2}$	110700	750			$4\frac{1}{2}$	233718	353
$3d^2(a^1G)4s$	b^2G	$4\frac{1}{2}$	127202		$3d^2(a^3F)4d$	e^2F	$5\frac{1}{2}$	234108	
		$3\frac{1}{2}$	127218	-16			$6\frac{1}{2}$	234508	400
$3d^2(a^3F)4p$	z^4G°	$2\frac{1}{2}$	157369		$3d^2(a^3F)4d$	e^4F	$2\frac{1}{2}$	233446	
		$3\frac{1}{2}$	157942	573			$3\frac{1}{2}$	233647	201
		$4\frac{1}{2}$	158638	696					
		$5\frac{1}{2}$	159458	820					
$3d^2(a^3F)4p$	z^4F°	$1\frac{1}{2}$	158536		$3d^2(a^3F)5s$	$f^4F?$	$1\frac{1}{2}$	234094	
		$2\frac{1}{2}$	158901	365			$2\frac{1}{2}$	252073	
		$3\frac{1}{2}$	159360	459			$3\frac{1}{2}$	252312	239
		$4\frac{1}{2}$	159872	512			$4\frac{1}{2}$	252683	371
$3d^2(a^3F)4p$	z^2F°	$2\frac{1}{2}$	160313		$3d^2(a^3F)5s$	$f^2F?$	$2\frac{1}{2}$	253426	
		$3\frac{1}{2}$	160946	633			$3\frac{1}{2}$	253979	553
$3d^2(a^3F)4p$	z^4D°	$0\frac{1}{2}$	160512		$3d^2(a^1G)4d$	e^2H	$4\frac{1}{2}$	258397	
		$1\frac{1}{2}$	160999	487			$5\frac{1}{2}$	259806	1409
		$2\frac{1}{2}$	161506	507					
		$3\frac{1}{2}$	162072	566					
$3d^2(a^3F)4p$	z^2D°	$1\frac{1}{2}$	161765		$3d^2(a^1G)4d$	f^2G	$3\frac{1}{2}$	259504	
		$2\frac{1}{2}$	162310	545			$4\frac{1}{2}$		
$3d^2(a^3F)4p$	z^2G°	$3\frac{1}{2}$	164920		$3d^2(a^3F)4f$	y^4G°	$5\frac{1}{2}$	300340	
		$4\frac{1}{2}$	165440	520			$3\frac{1}{2}$	300366	
$3d^2(a^1D)4p$	y^2D°	$2\frac{1}{2}$	169260				$2\frac{1}{2}$		
		$1\frac{1}{2}$	169849	-589					
$3d^2(a^3P)4p$	x^2D°	$1\frac{1}{2}$			Cr v (3F_2)	<i>Limit</i>		400000?	
		$2\frac{1}{2}$	174978						

May 1951.

Cr IV OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ +	Observed Terms							
3d ³	$\left\{ \begin{array}{cccc} a^4P & a^4F & & \\ a^2P & a^2D & a^2F & a^2G & a^2H \end{array} \right.$							
	<i>ns</i> (<i>n</i> ≥ 4)	<i>np</i> (<i>n</i> ≥ 4)		<i>nd</i> (<i>n</i> ≥ 4)	<i>nf</i> (<i>n</i> ≥ 4)			
3d ² (a ³ F) <i>nx</i>	$\left\{ \begin{array}{l} b, f^4F \\ b, f^2F \end{array} \right.$	<i>z</i> ⁴ D ^o	<i>z</i> ⁴ F ^o	<i>z</i> ⁴ G ^o	<i>e</i> ⁴ F	<i>e</i> ⁴ G	<i>e</i> ⁴ H	<i>y</i> ⁴ G ^o ?
3d ² (a ¹ D) <i>nx</i>		<i>z</i> ² D ^o	<i>z</i> ² F ^o	<i>z</i> ² G ^o	<i>e</i> ² F	<i>e</i> ² G		
3d ² (a ³ P) <i>nx</i>	$\left\{ \begin{array}{l} \\ \\ \\ \end{array} \right.$	<i>y</i> ² D ^o	<i>y</i> ² F ^o					
3d ² (a ¹ G) <i>nx</i>			<i>x</i> ² D ^o					
3d ² (a ¹ S) <i>nx</i>		<i>b</i> ² G	<i>x</i> ² P ^o	<i>x</i> ² F ^o	<i>y</i> ² G ^o	<i>z</i> ² H ^o	<i>f</i> ² G	<i>e</i> ² H

*For predicted terms in the spectra of the Sc I isoelectronic sequence, see Volume I, p. xxxvi.

Cr v

(Ca I sequence; 20 electrons)

Z=24

Ground state 1s² 2s² 2p⁶ 3s² 3p⁶ 3d² ³F₂

*a*³F₂ 589700 cm⁻¹

I. P. 73 volts

This spectrum is incompletely analyzed. In 1929 White published 55 classified lines in the region between 433 Å and 1820 Å. Cady has also made some observations and revised White's intersystem combinations and singlet terms. Edlén, who obtained Cady's unpublished material from Bowen, has furnished this material, together with additional revisions. One term (¹S₀) has been rejected, and three (*a*¹G, *b*¹D, and *z*¹F^o) are not connected with the rest, as indicated by *x* in the table. Edlén's estimated value of *a*¹G is entered in brackets.

The observed tabular values are from Cady's list for terms having configurations 3d² and 3d(²D) 4p. The remaining term values have been adjusted to these by means of the combinations observed by White. The small numerical differences in the two lists of triplet terms are due only to discrepancies in the measured wavelengths used for each, and not to differences in interpretation.

White's estimated ionization potential, based on extrapolation of isoelectronic sequence data, has been used to derive the limit entered in brackets in the table.

REFERENCES

- H. E. White, Phys. Rev. **33**, 538 (1929). (I P) (T) (C L)
 W. M. Cady, unpublished material (1939?). (T) (C L)
 B. Edlén, unpublished material (1949). (T)

Cr v

Cr v

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3d^2$	a^3F	2	0		$3d(^2D)4p$	z^3F°	2	229556	
		3	513	513			3	230328	772
		4	1146	633			4	231408	1080
$3d^2$	a^1D	2	13200		$3d(^2D)4p$	z^3P°	0	234680	-55
$3d^2$	a^3P	0	15500	184	$3d(^2D)4p$	z^1F°	1	234625	232
		1	15684	368			2	234857	
		2	16052				3	237568+x	
$3d^2$	a^1G	4	[22060]+x		$3d(^2D)4p$	z^1P°	1	239939	
$3d(^2D)4s$	a^3D	1	167184	313	$3d(^2D)4d$	e^3G	3	319451	360
		2	167497	602			4	319811	595
		3	168099				5	320406	
$3d(^2D)4s$	b^1D	2	171736+x		-----	-----	-----	-----	-----
$3d(^2D)4p$	z^1D°	2	226130		Cr vi ($^2D_{13}$)	Limit	-----	[589700]	
$3d(^2D)4p$	z^3D°	1	228006	491					
		2	228497	630					
		3	229127						

March 1949.

Cr v OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms			
$3d^2$	{ a^3P a^1D a^3F a^1G			
	ns ($n \geq 4$)		np ($n \geq 4$)	
$3d(^2D)nx$	{ a^3D b^1D		z^3P° z^1P°	z^3D° z^1D°
			z^3F° z^1F°	e^3G

*An array of predicted terms in the spectra of the Ca I isoelectronic sequence is given in Vol. 1, p. xxxv. Owing to the change in binding energies of the $3d$ and $4s$ electrons along this sequence the arrangement of the arrays of observed and predicted terms is not identical. In Cr v the prefixes a, b, \dots, e, z replace those indicating the running electron.

Cr VI

(K I sequence; 19 electrons)

 $Z=24$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^2 D_{1/2}$ $3d^2 D_{1/2}$ 730900 cm^{-1}

I. P. 90.6 volts

The analysis is very incomplete. Six lines have been classified, in the range between 209A and 1455 A. The terms have been calculated chiefly from the data by Gibbs and White. The limit is from an estimate by Kruger and Weissberg, based on isoelectronic sequence data.

REFERENCES

- R. C. Gibbs and H. E. White, Proc. Nat. Acad. Sci. **12**, 676 (1926). (C L)
 R. C. Gibbs and H. E. White, Phys. Rev. **33**, 162 (1929). (C L)
 P. G. Kruger and S. G. Weissberg, Phys. Rev. **52**, 316 (1937). (I P) (C L)

Cr VI

Config.	Desig.	J	Level	Interval
$3p^6(^1S)3d$	$3d^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	0 957	957
$3p^6(^1S)4s$	$4s^2S$	$0\frac{1}{2}$	227775	
$3p^6(^1S)4p$	$4p^2P^\circ$	$0\frac{1}{2}$ $1\frac{1}{2}$	296489 298311	1822
$3p^6(^1S)4f$	$4f^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	476255	
-----	-----	-----	-----	
Cr VII (1S_0)	<i>Limit</i>	-----	[730900]	

May 1948.

Cr VII

(A I sequence; 18 electrons)

 $Z=24$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 {}^1S_0$ $3p^6 {}^1S_0$ 1299700 cm^{-1}

I. P. 161.1 volts

Four lines are classified in the region between 104 Å and 148 Å, as combinations with the ground term. The values listed in the table have been rounded off in the last places.

For convenience, the Paschen notation has been added by the writer in column one under the heading "A I". As for A I, the jl -coupling notation in the general form suggested by Racah is here introduced, although LS -designations as indicated in column two under the heading "Authors" are perhaps preferable for the terms thus far identified.

REFERENCES

- P. G. Kruger and S. G. Weissberg, Phys. Rev. **48**, 659 (1935). (I P) (T) (C L)
 P. G. Kruger, S. G. Weissberg and L. W. Phillips, Phys. Rev. **51**, 1090, (1937). (I P) (T)
 G. Racah, Phys. Rev. **61**, 537 (L) (1942).

Cr VII

A I	Authors	Config.	Desig.	J	Level
$1p_0$	$3p^6 {}^1S$	$3p^6$	$3p^6 {}^1S$	0	0
$1s_4$	$3p^5 4s {}^3P^\circ$	$3p^5 ({}^2P_{1/2}) 4s$	$4s [1\frac{1}{2}]^\circ$	2 1	672330
$1s_2$	$3p^5 4s {}^1P^\circ$	$3p^5 ({}^2P_{3/2}) 4s$	$4s' [0\frac{1}{2}]^\circ$	0 1	682440
$2s_4$	$3p^5 5s {}^3P^\circ$	$3p^5 ({}^2P_{1/2}) 5s$	$5s [1\frac{1}{2}]^\circ$	2 1	951110
$2s_2$	$3p^5 5s {}^1P^\circ$	$3p^5 ({}^2P_{3/2}) 5s$	$5s' [0\frac{1}{2}]^\circ$	0 1	960340
		-----	-----	---	-----
		Cr VIII (${}^2P_{1/2}$)	<i>Limit</i>	---	1299700
		Cr VIII (${}^2P_{3/2}$)	<i>Limit</i>	---	1309640

May 1948.

Cr VIII

(Cl I sequence; 17 electrons)

Z=24

Ground state $1s^2 2s^2 2p^6 3s^2 3p^5 {}^2P_{1/2}$ $3p^5 {}^2P_{1/2}$ 1491000 cm^{-1}

I. P. 185 volts

All of the terms except $3p^6 {}^2S$ are from the paper by Edlén. Twelve lines in the region between 124 Å and 430 Å have been classified as combinations from the ground term. Edlén has estimated the value of the limit by extrapolation along the isoelectronic sequence, as indicated by brackets in the table.

His unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

- S. G. Weissberg and P. G. Kruger, Phys. Rev. **49**, 872 (A) (1936). (C L)
 B. Edlén, Zeit Phys. **104**, 407 (1937). (I P) (T) (C L)

Cr VIII

Config.	Desig.	J	Level	Interval
$3s^2 3p^5$	$3p^5 {}^2P^{\circ}$	$1\frac{1}{2}$	0	-9900
		$0\frac{1}{2}$	9900	
$3s 3p^6$	$3p^6 {}^2S$	$0\frac{1}{2}$	242131	
$3s^2 3p^4({}^3P)4s$	$4s {}^4P$	$2\frac{1}{2}$	735880	-5180
		$1\frac{1}{2}$	741060	
		$0\frac{1}{2}$		
$3s^2 3p^4({}^3P)4s$	$4s {}^2P$	$1\frac{1}{2}$	749650	-6090
		$0\frac{1}{2}$	755740	
$3s^2 3p^4({}^1D)4s$	$4s' {}^2D$	$2\frac{1}{2}$	769240	-320
		$1\frac{1}{2}$	769560	
$3s^2 3p^4({}^1S)4s$	$4s'' {}^2S$	$0\frac{1}{2}$	805260	
-----	-----	-----	-----	
Cr IX (3P_2)	<i>Limit</i>	-----	[1491000]	

January 1948.

Cr IX

(S I sequence; 16 electrons)

Z=24

Ground state $1s^2 2s^2 2p^6 3s^2 3p^4 {}^3P_2$ $3p^4 {}^3P_2$ 1691000 cm^{-1}

I. P. 209.6 volts

Edlén has classified 11 lines in the interval between 117 Å and 123 Å, and extrapolated the value of the limit from isoelectronic sequence data.

The singlet and triplet terms are not connected by observed intersystem combinations. By analogy with Fe XI, Edlén has revised the interpolated values of the singlet terms for inclusion here, and the uncertainty, x , is probably not large.

Edlén's unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCE

- B. Edlén, Zeit. Phys. **104**, 188 (1937). (I P) (T) (C L)
 B. Edlén, letter (January 1950). (T)

Cr IX

Config.	Desig.	<i>J</i>	Level	Interval
$3s^2 3p^4$	$3p^4 \ ^3P$	2	0	
		1	7860	-7860
		0	9600	-1740
$3s^2 3p^4$	$3p^4 \ ^1D$	2	$30270+x$	
$3s^2 3p^4$	$3p^4 \ ^1S$	0	$66940+x$	
$3s^2 3p^3 (^4S^\circ)4s$	$4s \ ^3S^\circ$	1	821150	
$3s^2 3p^3 ({}^2D^\circ)4s$	$4s' \ ^3D^\circ$	1	845930	350
		2	846280	
		3	847870	
$3s^2 3p^3 ({}^2D^\circ)4s$	$4s' \ ^1D^\circ$	2	$854720+x$	
$3s^2 3p^3 ({}^2P^\circ)4s$	$4s'' \ ^1P^\circ$	1	$881800+x$	
-----	-----	-----	-----	
Cr x (${}^4S_{1/2}$)	<i>Limit</i>	-----	[1691000]	

January 1950.

Cr XII

(Al I sequence; 13 electrons)

 $Z=24$ Ground state $1s^2 2s^2 2p^6 3s^2 3p \ ^2P_{0\frac{1}{2}}$ $3p \ ^2P_{0\frac{1}{2}}$ cm⁻¹

I. P. volts

This spectrum has not been analyzed, but Edlén has classified 2 lines as follows:

I. A.	Int.	Wave No.	Desig.
75. 815	2	1319000	} $3p \ ^2P^\circ - 4d \ ^2D$
76. 488	3	1307400	

His unit, 10^3 cm^{-1} , is here changed to cm^{-1} .

REFERENCE

B. Edlén, Zeit. Phys. **103**, 540 (1936). (C L)

December 1947.

(Mg I sequence; 12 electrons)

 $Z=24$ Ground state $1s^2 2s^2 2p^6 3s^2 1S_0$ $3s^2 1S_0$ 2862000 cm^{-1}

I. P. 355 volts

Edlén has classified 14 lines in the region between 53 Å and 91 Å. No intersystem combinations have been observed and the triplet terms are not all connected by observed combinations. He has determined the relative positions of the various groups of terms and also the ionization potential by extrapolation along the isoelectronic sequence. His estimated value of the limit is entered in brackets in the table. His unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCE

B. Edlén, Zeit. Phys. **103**, 536 (1936). (I P) (T) (C L)

Cr XIII

Cr XIII

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3s^2$	$3s^2 1S$	0	0		$3s(2S)4f$	$4f 3F^\circ$	2 3 4		
$3s(2S)3p$	$3p 3P^\circ$	0 1 2	$203030+x$ $206960+x$ $216150+x$	3930 9190				$1682940+x$	
$3s(2S)3d$	$3d 3D$	1 2 3	$594270+x$		$3s(2S)5d$	$5d 3D$	1 2 3	$2076100+x$	
$3s(2S)4s$	$4s 3S$	1	$1384840+x$		$3s(2S)5f$	$5f 3F^\circ$	2 3 4	$2110160+x$	
$3s(2S)4p$	$4p 1P^\circ$	1	1492920		-----	-----	-----	-----	
$3s(2S)4d$	$4d 3D$	1 2 3	$1615620+x$ $1616020+x$ $1616750+x$	400 730	Cr XIV ($2S_{1/2}$)	<i>Limit</i>	-----	[2862000]	

August 1947.

Cr XIV

(Na I sequence; 11 electrons)

 $Z=24$ Ground state $1s^2 2s^2 2p^6 3s^2 S_{1/2}$ $3s^2 S_{1/2}$ 3099630 cm^{-1}

I. P. 384.20 volts

Edlén classified 16 lines in the interval 46 Å to 86 Å and extrapolated the absolute value of the ground term from isoelectronic sequence data. His unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCE

B. Edlén, Zeit. Phys. **100**, 621 (1936). (I P) (T) (C L)

Cr XIV

Cr XIV

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
3s	$3s^2 S$	$0\frac{1}{2}$	0		5p	$5p^2 P^\circ$	$0\frac{1}{2}$ $1\frac{1}{2}$	2149290 2152020	2730
3p	$3p^2 P^\circ$	$0\frac{1}{2}$ $1\frac{1}{2}$	242400 256320	13920	5d	$5d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	2210440 2210900	460
3d	$3d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	588910 590620	1710	5f	$5f^2 F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	2236380 2236550	170
4s	$4s^2 S$	$0\frac{1}{2}$	1478250		6f	$6f^2 F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	2500370	
4p	$4p^2 P^\circ$	$0\frac{1}{2}$ $1\frac{1}{2}$	1574180 1579550	5370					
4d	$4d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	1700870 1701760	890	----- Cr xv (1S_0)	----- <i>Limit</i>	-----	----- [3099630]	
4f	$4f^2 F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	1750930 1751220	290					

June 1947.

Cr XV

(Ne I sequence; 10 electrons)

 $Z=24$ Ground state $1s^2 2s^2 2p^6 ^1S_0$ $2p^6 ^1S_0$ 8172300 cm^{-1}

I. P. 1013.0 volts

Nine lines have been classified by Tyrén in the interval between 15 Å and 21 Å, as combinations with the ground term. His absolute term values have been derived by extrapolation along the Ne I isoelectronic sequence.

By analogy with Ne I the jl -coupling notation in the general form suggested by Racah is introduced.

The unit adopted by Tyrén, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

- F. Tyrén, Zeit. Phys. **111**, 314 (1938). (I P) (T) (C L)
G. Racah, Phys. Rev. **61**, 537 (L) (1942).

Cr XV

Cr XV

Author	Config.	Desig.	<i>J</i>	Level	Author	Config.	Desig.	<i>J</i>	Level
2p ¹ S	2p ⁶	2p ⁶ ¹ S	0	0	3p' ³ P ₁	2p ⁶ (² S)3p	3p ³ P ^o	2 1 0	5894500
3s ³ P ₁	2p ⁵ (² P _{1/2})3s	3s [1½] ^o	2 1	4727500	3p' ¹ P ₁	2p ⁶ (² S)3p	3p ¹ P ^o	1	5921000
3s ¹ P ₁	2p ⁵ (² P _{3/2})3s	3s' [0½] ^o	0 1	4793200	4d ¹ P ₁	2p ⁶ (² P _{1/2})4d	4d [1½] ^o	1	6576000
3d ³ P ₁	2p ⁵ (² P _{1/2})3d	3d [0½] ^o	0 1	5259000	4d ³ D ₁	2p ⁶ (² P _{3/2})4d	4d' [1½] ^o	1	6641000
3d ¹ P ₁	"	3d [1½] ^o	1	5324200		-----	-----	---	-----
3d ³ D ₁	2p ⁵ (² P _{3/2})3d	3d' [1½] ^o	1	5406300		Cr XVI (² P _{1/2})	Limit	---	[8172300]
						Cr XVI (² P _{3/2})	<i>Limit</i>	---	8242400

April 1947.

Cr XV OBSERVED LEVELS*

Config. 1s ² +	Observed Terms		
2s ² 2p ⁶	2p ⁶ ¹ S		
	<i>ns</i> (<i>n</i> ≥ 3)	<i>np</i> (<i>n</i> ≥ 3)	<i>nd</i> (<i>n</i> ≥ 3)
2s ² 2p ⁵ (² P ^o) <i>nx</i>	{ 3s ³ P ^o 3s ¹ P ^o		3d ³ P ^o 3, 4d ³ D ^o 3, 4d ¹ P ^o
2s 2p ⁶ (² S) <i>nx</i>	{	3p ³ P ^o 3p ¹ P ^o	
<i>jl</i> —Coupling Notation			
Config. 1s ² +	Observed Pairs		
	<i>ns</i> (<i>n</i> ≥ 3)		<i>nd</i> (<i>n</i> ≥ 3)
2s ² 2p ⁵ (² P _{1/2}) <i>nx</i>	3s [1½] ^o		3d [0½] ^o 3, 4d [1½] ^o
2s ² 2p ⁵ (² P _{3/2}) <i>nx</i> '	3s' [0½] ^o		3, 4d' [1½] ^o

*For predicted levels in the spectra of the Ne I isoelectronic sequence, see Vol. I, p. xxxi.

MANGANESE

Mn I

25 electrons

Z=25

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^5 4s^2 {}^6S_{2\frac{1}{2}}$ $a {}^6S_{2\frac{1}{2}}$ 59960 cm^{-1}

I. P. 7.432 volts

Manganese occupies one of the most important places in the development of spectrum analysis. It was in the spectra of this element that M. A. Catalán first discovered multiplets. This led directly to further analysis of complex spectra, and culminated in the development of the Hund theory, which is beautifully confirmed in detail by observation.

During his recent visit to the United States, Catalán extended his early work on Mn I as well as that of Back, Meggers, Russell, Dunham, and others, to include more than 1500 classified lines and over 400 multiplets. The observations of the spectrum now cover the region from 600 Å to 17607 Å. The present term list has been furnished by him in advance of publication especially for inclusion here. Some terms and configurations have been contributed by Olga García-Riquelme in connection with her doctoral thesis prepared under the direction of Catalán.

The terms of all four multiplicities are connected by observed intersystem combinations.

A detailed discussion of asymmetrical Zeeman patterns of Mn I is given in the references. For example, Catalán lists two g values, 1.068 and 1.003, for the level $z {}^6F_{1\frac{1}{2}}^o$, and two (1.068 and 0.906) for $y {}^6F_{1\frac{1}{2}}^o$, one for each of the magnetic levels. Only one entry for each of these levels is included in the table.

REFERENCES

- M. A. Catalán, Phil. Trans. Roy. Soc. (London) [A] **223**, 127 (1922). (I P) (T) (C L)
 J. E. Mack, Rev. Mod. Phys. **22**, No. 1 64, (1950). (Summary hfs)
 M. A. Catalán and R. Velasco, J. Opt. Soc. Am. **40**, 653 (1950). (Z E)
 M. A. Catalán, J. Research Nat. Bur. Std. **47**, 502, RP2278 (1951). (Z E)
 M. A. Catalán and Olga García-Riquelme, unpublished material (November 1951). (I P) (T) (C L) (Z E)

Mn I

Mn I

Config.	Desig.	J	Level	Interval	Obs. g	Config.	Desig.	J	Level	Interval	Obs. g		
$3d^5 4s^2$	$a {}^6S$	$2\frac{1}{2}$	0. 00		1. 999	$3d^5 4s^2$	$a {}^4P$	$2\frac{1}{2}$	27201. 54		1. 597		
$3d^6(a {}^5D)4s$	$a {}^6D$	$4\frac{1}{2}$	17052. 29	—229. 71	1. 559	$3d^5 4s^2$	$a {}^4P$	$1\frac{1}{2}$	27248. 00	—46. 46	1. 730		
		$3\frac{1}{2}$	17282. 00	—169. 52	1. 584			$0\frac{1}{2}$	27281. 85	—33. 85	2. 666		
		$2\frac{1}{2}$	17451. 52	—116. 96	1. 657			$3d^5 4s^2$	$b {}^4D$	$3\frac{1}{2}$	30354. 21	—65. 40	1. 423
		$1\frac{1}{2}$	17568. 48	—68. 67	1. 866					$2\frac{1}{2}$	30419. 61	—6. 10	
		$0\frac{1}{2}$	17637. 15		3. 327					$1\frac{1}{2}$	30425. 71	13. 97	
$3d^5 4s(a {}^7S)4p$	$z {}^8P^o$	$2\frac{1}{2}$	18402. 46	129. 18	2. 284	$3d^5 4s(a {}^5S)4p$	$z {}^4P^o$	$2\frac{1}{2}$	31001. 15	—75. 27	1. 600		
		$3\frac{1}{2}$	18531. 64	173. 73	1. 938			$1\frac{1}{2}$	31076. 42	—48. 53	1. 732		
		$4\frac{1}{2}$	18705. 37		1. 779			$0\frac{1}{2}$	31124. 95				
$3d^6(a {}^5D)4s$	$a {}^4D$	$3\frac{1}{2}$	23296. 67	—252. 53	1. 427	$3d^6(a {}^3P)4s$	$b {}^4P$	$2\frac{1}{2}$	33825. 49	—637. 88	1. 604		
		$2\frac{1}{2}$	23549. 20	—170. 32	1. 368			$1\frac{1}{2}$	34463. 37	—381. 89			
		$1\frac{1}{2}$	23719. 52	—99. 35	1. 198			$0\frac{1}{2}$	34845. 26				
		$0\frac{1}{2}$	23818. 87		0. 000								
$3d^5 4s(a {}^7S)4p$	$z {}^6P^o$	$1\frac{1}{2}$	24779. 32	8. 73	2. 364	$3d^6(a {}^3H)4s$	$a {}^4H$	$6\frac{1}{2}$	34138. 88	—111. 64	1. 231		
		$2\frac{1}{2}$	24788. 05	14. 20	1. 714			$5\frac{1}{2}$	34250. 52	—93. 38	1. 135		
		$3\frac{1}{2}$	24802. 25					$4\frac{1}{2}$	34343. 90	—79. 37	0. 971		
				$3\frac{1}{2}$	34423. 27				0. 665				
$3d^5 4s^2$	$a {}^4G$	$5\frac{1}{2}$	25265. 74	—19. 69	1. 270	$3d^6(a {}^3F)4s$	$a {}^4F$	$4\frac{1}{2}$	34938. 70	—102. 67	1. 328		
		$4\frac{1}{2}$	25285. 43	—2. 31				$3\frac{1}{2}$	35041. 37	—73. 61	1. 238		
		$3\frac{1}{2}$	25287. 74	6. 70				$2\frac{1}{2}$	35114. 98	—50. 07			
		$2\frac{1}{2}$	25281. 04					$1\frac{1}{2}$	35165. 05				

Mn I—Continued

Mn I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>		
$3d^5 4s(a^5S)4p$	y^6P°	$1\frac{1}{2}$	35689.98		2.400	$3d^5 4s(a^5G)4p$	y^6F°	$5\frac{1}{2}$	48021.43		1.460		
		$2\frac{1}{2}$	35725.85	35.87	1.886			$4\frac{1}{2}$	48168.01	-146.58	1.432		
		$3\frac{1}{2}$	35769.97	44.12	1.712			$3\frac{1}{2}$	48225.99	-57.98	1.403		
$3d^6(a^3G)4s$	b^4G	$5\frac{1}{2}$	37420.24		1.263	$3d^5 4s(a^5S)5s$	f^6S	$2\frac{1}{2}$	49415.35				
		$4\frac{1}{2}$	37630.82	-210.58	1.163			$3d^5 4s(a^5S)5s$	e^4S	$1\frac{1}{2}$	49591.51		
		$3\frac{1}{2}$	37737.22	-106.40	0.989					$3\frac{1}{2}$	49888.08		1.711
		$2\frac{1}{2}$	37789.93	-52.71						$2\frac{1}{2}$	50012.53	-124.45	1.888
				$1\frac{1}{2}$	50099.16	-86.63	2.398						
$3d^6(a^3H)4s$	a^2H	$5\frac{1}{2}$	38008.70										
		$4\frac{1}{2}$	38120.18	-111.48									
$3d^6(a^3F)4s$	a^2F	$3\frac{1}{2}$	38669.60			$3d^5 4s(a^5P)4p$	v^6P°	$3\frac{1}{2}$	49888.08		1.711		
		$2\frac{1}{2}$						$2\frac{1}{2}$	50012.53	-124.45	1.888		
								$1\frac{1}{2}$	50099.16	-86.63	2.398		
$3d^5 4s(a^7S)5s$	e^6S	$3\frac{1}{2}$	39431.31		2.000	$3d^5 4s(a^5G)4p$	z^4H°	$3\frac{1}{2}$	50065.46		7.13		
$3d^5 4s(a^7S)5s$	e^6S	$2\frac{1}{2}$	41403.93					$4\frac{1}{2}$	50072.59		8.72		
								$5\frac{1}{2}$	50081.31		13.29		
								$6\frac{1}{2}$	50094.60				
$3d^6(a^5D)4p$	z^6D°	$4\frac{1}{2}$	41789.48		1.556	$3d^5 4s(a^7S)6s$	f^6S	$3\frac{1}{2}$	50157.63		1.995		
		$3\frac{1}{2}$	41932.64	-143.16	1.587			$3\frac{1}{2}$					
		$2\frac{1}{2}$	42053.73	-121.09	1.653	$3d^5 4s(a^5G)4p$	y^4F°	$4\frac{1}{2}$	50341.30		-17.98		
		$1\frac{1}{2}$	42143.57	-89.84	1.867			$3\frac{1}{2}$	50359.28	-17.98	-13.95		
		$0\frac{1}{2}$	42198.56	-54.99	3.317			$2\frac{1}{2}$	50373.23	-13.95	-10.04		
								$1\frac{1}{2}$	50383.27	-10.04			
$3d^6(a^5D)4p$	z^6F°	$5\frac{1}{2}$	43314.23		1.464	$3d^5 4s(b^5D)4p$	x^6F°	$0\frac{1}{2}$	50818.64		44.41		
		$4\frac{1}{2}$	43428.58	-114.35	1.431			$1\frac{1}{2}$	50863.05		68.37		
		$3\frac{1}{2}$	43524.08	-95.50	1.395			$2\frac{1}{2}$	50931.42		83.53		
		$2\frac{1}{2}$	43595.50	-71.42	1.310			$3\frac{1}{2}$	51014.95		85.54		
		$1\frac{1}{2}$	43644.45	-48.95	1.068			$4\frac{1}{2}$	51100.49		68.69		
		$0\frac{1}{2}$	43672.96	-28.51	-0.602			$5\frac{1}{2}$	51169.18				
$3d^6(a^5D)4p$	z^4F°	$4\frac{1}{2}$	44288.76		1.317	$3d^5 4s(a^7S)6s$	g^6S	$2\frac{1}{2}$	50904.68				
		$3\frac{1}{2}$	44523.45	-234.69	1.240			$2\frac{1}{2}$					
		$2\frac{1}{2}$	44696.29	-172.84	1.030	$3d^5 4s(a^5P)4p$	x^4P°	$1\frac{1}{2}$	51305.41		-140.24		
		$1\frac{1}{2}$	44814.73	-89.84	0.400			$0\frac{1}{2}$	51445.65	-107.13	1.728		
									51552.78		2.664		
$3d^7$	c^4F	$1\frac{1}{2}$	44978.98		34.57	$3d^5 4s(a^5G)4p$	z^4G°	$2\frac{1}{2}$	51515.63		14.98		
		$2\frac{1}{2}$	45013.55	34.57	1.717			$3\frac{1}{2}$	51530.61		15.66		
		$3\frac{1}{2}$	45060.81	47.26	1.885			$4\frac{1}{2}$	51546.27		14.66		
		$4\frac{1}{2}$	45130.16	69.35	2.399			$5\frac{1}{2}$	51560.93		1.273		
$3d^6(a^5D)4p$	x^6P°	$3\frac{1}{2}$	44993.92		1.717	$3d^7$	c^4P	$2\frac{1}{2}$	51638.17		-80.05		
		$2\frac{1}{2}$	45156.11	-162.19	1.885			$1\frac{1}{2}$	51718.22		-69.70		
		$1\frac{1}{2}$	45259.17	-103.06	2.399			$0\frac{1}{2}$	51787.92				
$3d^6(a^6D)4p$	z^4D°	$3\frac{1}{2}$	45754.27		1.427	$3d^5 4s(b^5D)4p$	u^6P°	$1\frac{1}{2}$	52015.00		113.65		
		$2\frac{1}{2}$	45940.93	-186.66	1.372			$2\frac{1}{2}$	52128.65		124.59		
		$1\frac{1}{2}$	46083.89	-142.96	1.200			$3\frac{1}{2}$	52253.24				
		$0\frac{1}{2}$	46169.93	-86.04	0.000	$3d^5 4s(a^7S)5d$	f^6D	$1\frac{1}{2}$					
$3d^5 4s(a^7S)5p$	y^6P°	$2\frac{1}{2}$	45981.40		20.52			$2\frac{1}{2}$	52702				
		$3\frac{1}{2}$	46001.92	20.52	26.95			$3\frac{1}{2}$					
		$4\frac{1}{2}$	46028.87	26.95				$4\frac{1}{2}$	52703		1		
								$5\frac{1}{2}$	52705		2		
$3d^5 4s(a^7S)4d$	e^6D	$1\frac{1}{2}$	46706.09		0.94	$3d^5 4s(a^7S)5d$	f^6D	$4\frac{1}{2}$	52726.39		-4.02		
		$2\frac{1}{2}$	46707.03	0.94	1.30			$3\frac{1}{2}$	52730.41		-2.81		
		$3\frac{1}{2}$	46708.33	1.30	1.82			$2\frac{1}{2}$	52733.22		-1.79		
		$4\frac{1}{2}$	46710.15	1.82	2.43			$1\frac{1}{2}$	52735.01		-0.82		
		$5\frac{1}{2}$	46712.58	2.43				$0\frac{1}{2}$	52735.83				
$3d^6(a^5D)4p$	y^4P°	$2\frac{1}{2}$	46901.13		1.595	$3d^5 4s(a^7S)5d$	f^6D	$4\frac{1}{2}$	52726.39		-4.02		
		$1\frac{1}{2}$	47154.51	-253.38	1.732			$3\frac{1}{2}$	52730.41		-2.81		
		$0\frac{1}{2}$	47299.29	-144.78	2.666			$2\frac{1}{2}$	52733.22		-1.79		
								$1\frac{1}{2}$	52735.01		-0.82		
								$0\frac{1}{2}$	52735.83				
$3d^5 4s(a^7S)4d$	e^6D	$4\frac{1}{2}$	47207.28		-4.78	$3d^5 4s(b^5D)4p$	x^6D°	$4\frac{1}{2}$	52758.11		-111.99		
		$3\frac{1}{2}$	47212.06	-4.78	-3.55			$3\frac{1}{2}$	52870.10		-13.77		
		$2\frac{1}{2}$	47215.61	-3.55	-2.54			$2\frac{1}{2}$	52883.87		0.00		
		$1\frac{1}{2}$	47218.15	-2.54	-1.49			$1\frac{1}{2}$	52883.87		0.77		
		$0\frac{1}{2}$	47219.64	-1.49				$0\frac{1}{2}$	52883.10				
$3d^5 4s(a^7S)5p$	w^6P°	$3\frac{1}{2}$	47387.62		1.713	$3d^5 4s(a^7S)4f$	z^6F°	$0\frac{1}{2}$					
		$2\frac{1}{2}$	47659.52	-271.90	1.952			to					
		$1\frac{1}{2}$	47782.43	-122.91	2.336			$6\frac{1}{2}$	52977.96				
$3d^5 4s(a^5P)4p$	y^6D°	$0\frac{1}{2}$	47452.16		3.174	$3d^5 4s(a^5P)4p$	y^4D°	$0\frac{1}{2}$	53101.32		1.87		
		$1\frac{1}{2}$	47466.66	14.50	287.33			$1\frac{1}{2}$	53103.19		6.02		
		$2\frac{1}{2}$	47753.99	287.33	1.820			$2\frac{1}{2}$	53109.21		14.88		
		$3\frac{1}{2}$	47774.52	20.53	1.594			$3\frac{1}{2}$	53124.09		1.423		
		$4\frac{1}{2}$	47903.80	129.28	1.540								

Mn I—Continued

Mn I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	
$3d^5 4s(a^7S)6p$	t^6P°	$3\frac{1}{2}$	53261.42	-30.16 -19.79		$3d^5 4s(b^3P)4p$	y^4S°	$1\frac{1}{2}$	57512.16		2.000	
		$2\frac{1}{2}$	53291.58									
		$1\frac{1}{2}$	53311.37									
$3d^5 4s(a^7S)7s$	g^8S	$3\frac{1}{2}$	54180			$3d^5 4s(a^7S)8d$	i^8D	$1\frac{1}{2}$ to $5\frac{1}{2}$	57651.93			
$3d^5 4s(a^5P)4p$	z^4S°	$1\frac{1}{2}$	54218.71			$3d^5 4s(a^7S)7f$	w^6F°	$0\frac{1}{2}$ to $6\frac{1}{2}$		57695.3		
$3d^5 4s(a^7S)7s$	h^6S	$2\frac{1}{2}$	54460.30			$3d^6(a^3H)4p$	y^4G°	$5\frac{1}{2}$ $4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$	58075.06 58110.24 58136.69 58159.73	-35.18 -26.45 -23.04	1.269 1.168 0.980 0.578	
$3d^5 4s(a^5S)4d$	g^6D	$4\frac{1}{2}$	54938.71	-6.57 -5.38 -2.36 3.12		$3d^5 4s(a^7S)9d$	j^8D	$1\frac{1}{2}$ to $5\frac{1}{2}$	58199.7			
		$3\frac{1}{2}$	54945.28									
		$2\frac{1}{2}$	54950.66									
		$1\frac{1}{2}$	54953.02									
		$0\frac{1}{2}$	54949.90									
	x^4D°	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	55107.52 55186.17 55279.91	-78.65 -93.74	1.407 1.365 0.826							
$3d^5 4s(b^5D)4p$	w^4P°	$2\frac{1}{2}$	55405.14	36.48 -88.54		$3d^6(a^3H)4p$	y^4H°	$6\frac{1}{2}$ $5\frac{1}{2}$ $4\frac{1}{2}$ $3\frac{1}{2}$	58338.67 58427.30 58485.52 58519.90	-88.63 -58.22 -34.38	1.228 1.133 0.968 0.665	
		$1\frac{1}{2}$	55368.66									
		$0\frac{1}{2}$	55457.20									
$3d^5 4s(a^7S)6d$	g^8D	$1\frac{1}{2}$	55374 55375 55376	1 1		$3d^6(a^3H)4p$	z^4I°	$7\frac{1}{2}$ $6\frac{1}{2}$ $5\frac{1}{2}$ $4\frac{1}{2}$	58852.60 58843.39 58851.49 58866.66	9.21 -8.10 -15.17		
		$2\frac{1}{2}$										
		$3\frac{1}{2}$										
		$4\frac{1}{2}$										
		$5\frac{1}{2}$										
$3d^5 4s(a^7S)5f$	y^6F°	$0\frac{1}{2}$ to $6\frac{1}{2}$	55499.08			$3d^6(a^3P)4p$	t^4P°	$2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	59116.60 59384.45 59568.29	-267.85 -183.84		
$3d^5 4s(a^7S)6d$	h^6D	$4\frac{1}{2}$	55681.9	-6.2 -2.7 -1.1 -0.5		$3d^6(a^3P)4p$	w^4F°	$4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$	59257.44 59290.23 59360.72 59416.15	-32.79 -70.49 -55.43	1.327	
		$3\frac{1}{2}$	55688.1									
		$2\frac{1}{2}$	55690.8									
		$1\frac{1}{2}$	55691.9									
		$0\frac{1}{2}$	55692.4									
	1°	$2\frac{1}{2}$	55923.81			$3d^6(a^3P)4p$	v^4D°	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	59339.49 59600.35 59989.77 60141.98	-260.86 -389.42 -152.21		
$3d^5 4s(a^5S)5p$	s^6P°	$1\frac{1}{2}$	55996.62	11.29 4.51		$3d^6(a^3F)4p$	u^4D°	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	59470.14 59480.80 59527.89 59527.36	-10.66 -47.09 0.53		
		$2\frac{1}{2}$	56007.91									
		$3\frac{1}{2}$	56012.42									
$3d^5 4s(a^7S)8s$	h^8S	$3\frac{1}{2}$	56144.16			$3d^6(a^3H)4p$	z^2I°	$6\frac{1}{2}$ $5\frac{1}{2}$	59617.12 59827.88	-210.76		
$3d^6(a^5D)5s$	i^6D	$4\frac{1}{2}$	56189.45	-166.76 -134.58 -77.14 -98.13		$3d^6(a^3F)4p$	x^4G°	$5\frac{1}{2}$ $4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$	59652.90 59731.94 59784.31 59817.70	-79.04 -52.37 -33.39		
		$3\frac{1}{2}$	56356.21									
		$2\frac{1}{2}$	56490.79									
		$1\frac{1}{2}$	56567.93									
		$0\frac{1}{2}$	56666.06									
	e^4D	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	56462.08 56561.95 56601.63 56646.9?	-99.87 -39.68 -45.3		Mn II (7S_3)	<i>Limit</i>	-----	59960			
$3d^5 4s(a^7S)7d$	h^8D	$1\frac{1}{2}$ to $5\frac{1}{2}$	56801.4			$3d^6(a^3P)4p$	x^4S°	$1\frac{1}{2}$	60126.81			
$3d^5 4s(a^7S)6f$	x^6F°	$0\frac{1}{2}$ to $6\frac{1}{2}$	56863			$3d^6(a^3F)4p$	z^2G°	$4\frac{1}{2}$ $3\frac{1}{2}$	60668.49 60739.42	-70.93		
$3d^5 4p^2$	e^6P	$2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	57086.33 57218.15 57388.90	131.82 170.75		$3d^5 4s(^3I)4p$	x^4H°	$6\frac{1}{2}$ $5\frac{1}{2}$ $4\frac{1}{2}$ $3\frac{1}{2}$	60891.48 60933.73 60955.88 60957.21	-42.25 -22.15 -1.33	1.228 1.134	
$3d^5 4s(b^3P)4p$	u^4P°	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	57228.30 57360.78 57487.08	132.48 126.30	1.736	$3d^6(a^3F)4p$	y^2G°	$3\frac{1}{2}$ $4\frac{1}{2}$	60902.80 60938.97	36.17		
$3d^6(a^5D)5s$	f^4D	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	57305.62 57485.97 57621.90 57705.83	-180.35 -135.93 -83.93		$3d^6(a^3P)4p$	y^4I°	$7\frac{1}{2}$ $6\frac{1}{2}$ $5\frac{1}{2}$ $4\frac{1}{2}$	61204.54 61225.55 61225.77 61211.43	-21.01 -0.22 14.34		
						$3d^6(a^3F)4p$	w^4G°	$5\frac{1}{2}$ $4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$	61469.21 61485.34 61480.60 61471.23	-16.13 4.74 9.37		

Mn I—Continued

Mn I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>				
$3d^6(a^5D)4d$	<i>e</i> 6F	$5\frac{1}{2}$	61713. 62	—316. 56 —264. 48			15°	$2\frac{1}{2}$	64823. 21	—49. 70					
		$4\frac{1}{2}$	62030. 18				16°	$3\frac{1}{2}$	64988. 22						
		$3\frac{1}{2}$	62294. 66				<i>w</i> $^2G^\circ$	$4\frac{1}{2}$	65262. 28						
		$2\frac{1}{2}$						$3\frac{1}{2}$							
		$1\frac{1}{2}$													
		$0\frac{1}{2}$													
		2°	$3\frac{1}{2}$	61710. 98			17°	$3\frac{1}{2}$	65617. 37						
	3°	$4\frac{1}{2}$	61714. 52			18°	$4\frac{1}{2}$	65768. 81							
$3d^6(a^5D)4d$	<i>e</i> 6G	$6\frac{1}{2}$	62001. 09	—133. 36 —166. 18 —125. 85 —88. 11 —58. 52			<i>v</i> $^4G^\circ$	$2\frac{1}{2}$	65873. 40	2. 94 32. 58 —21. 61	1. 160 1. 259				
		$5\frac{1}{2}$	62134. 45					$3\frac{1}{2}$	65876. 34						
		$4\frac{1}{2}$	62300. 63					$4\frac{1}{2}$	65908. 92						
		$3\frac{1}{2}$	62426. 48					$5\frac{1}{2}$	65887. 31						
		$2\frac{1}{2}$	62514. 59												
		$1\frac{1}{2}$	62573. 11												
		<i>z</i> $^2F^\circ$	$3\frac{1}{2}$	62034. 04	—40. 98		<i>x</i> $^2D^\circ$	$2\frac{1}{2}$	65946. 87	—15. 03					
	$2\frac{1}{2}$	62075. 02	$1\frac{1}{2}$	65961. 90											
	6°	$4\frac{1}{2}$	62345. 73			<i>y</i> $^2F^\circ$	$2\frac{1}{2}$	66020. 63	128. 47						
		$2\frac{1}{2}$	62075. 02			$3\frac{1}{2}$	66149. 10								
$3d^6(a^3G)4p$	<i>v</i> $^4F^\circ$	$1\frac{1}{2}$	62390. 20	97. 16 17. 93 —112. 47	1. 329		<i>u</i> $^4H^\circ$	$3\frac{1}{2}$	66334. 47	21. 93 62. 15 150. 03					
		$2\frac{1}{2}$	62487. 36					$4\frac{1}{2}$	66356. 40						
		$3\frac{1}{2}$	62505. 29					$5\frac{1}{2}$	66418. 55						
		$4\frac{1}{2}$	62392. 82					$6\frac{1}{2}$	66568. 58						
$3d^5 4s(a^5F)4p$	<i>w</i> $^6D^\circ$	$4\frac{1}{2}$	62670. 81	—180. 66 90. 14 —26. 30 19. 47				19°	$1\frac{1}{2}$	66504. 21					
		$3\frac{1}{2}$	62851. 47					20°	$3\frac{1}{2}$	66600. 17					
		$2\frac{1}{2}$	62761. 33					21°	$4\frac{1}{2}$	66630. 92					
		$1\frac{1}{2}$	62787. 63					22°	$2\frac{1}{2}$	66654. 65					
		$0\frac{1}{2}$	62768. 16					23°	$2\frac{1}{2}, 3\frac{1}{2}$	66737. 82					
	7°	$2\frac{1}{2}$	63139. 70			24°	$2\frac{1}{2}, 3\frac{1}{2}$	66783. 05							
		<i>z</i> $^2H^\circ$	$5\frac{1}{2}$	63288. 78	85. 75										
		$4\frac{1}{2}$	63374. 53												
		8°	$3\frac{1}{2}$	63288. 78			1. 231		25°	$2\frac{1}{2}$	66837. 64				
		9°	$4\frac{1}{2}$	63347. 91					26°	$1\frac{1}{2}$	66843. 79				
		<i>w</i> $^4H^\circ$	$6\frac{1}{2}$	63363. 54	—94. 31 13. 24 49. 16				27°	$4\frac{1}{2}$	66855. 00				
			$5\frac{1}{2}$	63457. 85					28°	$2\frac{1}{2}$	66910. 02				
			$4\frac{1}{2}$	63444. 61					29°	$3\frac{1}{2}$	66981. 30				
			$3\frac{1}{2}$	63395. 45					30°	$4\frac{1}{2}$	67024. 41				
		<i>y</i> $^2H^\circ$	$5\frac{1}{2}$	63449. 13	—99. 36										
			$4\frac{1}{2}$	63548. 49											
		10°	$2\frac{1}{2}$	63523. 82					31°	$5\frac{1}{2}, 6\frac{1}{2}$	67206. 00			—71. 94	
		11°	$3\frac{1}{2}$	63546. 30					<i>w</i> $^2H^\circ$	$5\frac{1}{2}$	67504. 90				
		12°	$1\frac{1}{2}, 2\frac{1}{2}$	63583. 84						$4\frac{1}{2}$	67576. 84				
		<i>z</i> $^2D^\circ$	$2\frac{1}{2}$	63764. 90	—80. 42										
			$1\frac{1}{2}$	63845. 32											
		<i>x</i> $^2H^\circ$	$5\frac{1}{2}$	64051. 91	—3. 46										
			$4\frac{1}{2}$	64055. 37											
	13°	$3\frac{1}{2}$	64409. 69			<i>t</i> $^4G^\circ$	$5\frac{1}{2}$	67752. 84	—66. 33 —72. 19 —73. 68	1. 266					
	<i>x</i> $^2G^\circ$	$4\frac{1}{2}$	64585. 44	—63. 76			$4\frac{1}{2}$	67819. 17							
		$3\frac{1}{2}$	64649. 20			$3\frac{1}{2}$	67891. 36								
	14°	$0\frac{1}{2}$	64638. 68			$2\frac{1}{2}$	67965. 04								
	<i>y</i> $^2D^\circ$	$1\frac{1}{2}$	64683. 95	28. 99											
		$2\frac{1}{2}$	64712. 94												
	<i>v</i> $^4H^\circ$	$6\frac{1}{2}$	64731. 88	—87. 65 —68. 47 —32. 33	1. 236 1. 137 0. 974		$3d^5 4s(a^5G)5s$	<i>e</i> 4G	$5\frac{1}{2}$	68692. 02	—24. 20 —30. 45 —16. 1				
		$5\frac{1}{2}$	64819. 53						$4\frac{1}{2}$	68716. 22					
		$4\frac{1}{2}$	64888. 00						$3\frac{1}{2}$	68746. 67					
		$3\frac{1}{2}$	64920. 33						$2\frac{1}{2}$	68762. 8					
								33°	$5\frac{1}{2}$	69722. 96					

Mn I OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms			
	$ns (n \geq 4)$	$np (n \geq 4)$	$nd (n \geq 4)$	$nf (n \geq 4)$
$3d^5 4s^2$	a^6S	a^4P b^4D a^4G		
$3d^7$		c^4P c^4F		
$3d^5 4p^2$		e^8P		
$3d^5 4s(a^7S)nx$	$\{ \begin{matrix} e, f, g, h, sS \\ e, g, h, sS \end{matrix} \}$		z, y, sP^o z, w, t, oP^o	z, y, x, w, sF^o
$3d^5 4s(a^5S)nx$	$\{ \begin{matrix} f^6S \\ e^4S \end{matrix} \}$		y, s^6P^o z^4P^o	e^6D
$3d^6(a^5D)nx$	$\{ \begin{matrix} a, i, oD \\ a, f^4D \end{matrix} \}$		x^6P^o z^6D^o y^4P^o z^4D^o	e^6F e^6G
$3d^5 4s(a^5G)nx$	$\{ \begin{matrix} \end{matrix} \}$	i^4G	y^6F^o z^4G^o y^4F^o z^4H^o	
$3d^6(a^3P)nx$	$\{ \begin{matrix} b^4P \end{matrix} \}$		x^4S^o t^4P^o v^4D^o	
$3d^5 4s(a^5P)nx$	$\{ \begin{matrix} \end{matrix} \}$		v^6P^o y^6D^o x^4P^o y^4D^o	
$3d^6(a^3H)nx$	$\{ \begin{matrix} a^4H \\ a^2H \end{matrix} \}$		y^4G^o y^4H^o z^4I^o z^2I^o	
$3d^6(a^3F)nx$	$\{ \begin{matrix} a^4F \\ a^2F \end{matrix} \}$		u^4D^o w^4F^o x^4G^o	
$3d^5 4s(b^5D)nx$	$\{ \begin{matrix} \end{matrix} \}$		u^6P^o x^6D^o x^6F^o w^4P^o	
$3d^6(a^3G)nx$	$\{ \begin{matrix} b^4G \end{matrix} \}$		v^4F^o	
$3d^5 4s(b^3P)nx$			y^4S^o u^4P^o	
$3d^5 4s(a^5F)nx$			w^6D^o	
$3d^5 4s(^3I)nx$			x^4H^o y^4I^o	

*For predicted terms in the spectra of the Mn I isoelectronic sequence, see Vol. II, Introduction.

Mn II

(Cr I sequence; 24 electrons)

 $Z=25$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^5 4s^1 {}^7S_3$ $4s^1 {}^7S_3$ 126147 cm^{-1}

I. P. 15.64 volts

When Catalán announced his important discovery of multiplets in the spectrum of Mn I he listed, also, 24 lines of Mn II as members of four multiplets. Subsequently, Russell and, independently, Duffendack and Black, confirmed the regularities suggested by Catalán, and extended the list to include 33 lines classified as combinations among seven terms.

The leading investigation of this spectrum has, however, been carried on by Curtis, who has extended his earlier work and furnished the present analysis in advance of publication, for inclusion here. His line list now includes nearly 1100 classified lines in the interval from 900 Å to 6500 Å. The terms of all three multiplicities are connected by observed intersystem combinations. Curtis has calculated the limit from four members of the $3d^5({}^6S)nf^1F^o$ series, well represented by a Ritz formula.

The writer has changed the author's notation to that adopted for complex spectra and described in detail in the text of Volume I, p. XII, namely that with the prefixes $a, b, c, \dots z, y, x$, etc., throughout. For convenience in cross reference, Curtis' notation is entered in column 1.

This spectrum is important astrophysically. The present investigation will be extremely useful in the interpretation of stellar spectra.

The observed g -values have been derived by Catalán from observations of the Zeeman Effect made with the Bitter magnet and large grating spectrographs at the Massachusetts Institute of Technology, with the generous cooperation of Harrison.

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 C. W. Curtis, *Phys. Rev.* **53**, 474 (1938). (I P) (T) (C L)
 M. A. Catalán, unpublished material (May 1950). (Z E)
 C. W. Curtis, *J. Opt. Soc. Am.* **42**, 300, 1952. (I P) (T) (C L)

Mn II

Mn II

Curtis	Config.	Desig.	J	Level	Interval	Obs. g	Curtis	Config.	Desig.	J	Level	Interval	Obs. g
4s ⁷ S ₃	3d ⁵ (a ⁶ S)4s	a ⁷ S	3	0. 00		2. 00	4p ⁵ P ₃	3d ⁵ (a ⁶ S)4p	z ⁵ P ^o	3	43370. 37		1. 67
4s ⁵ S ₂	3d ⁵ (a ⁶ S)4s	a ⁵ S	2	9472. 86		2. 00	4p ⁵ P ₂		2	43484. 50	-114. 13	1. 83	
							4p ⁵ P ₁		1	43557. 03	-72. 53	2. 49	
3d ⁵ D ₄	3d ⁶	a ⁵ D	4	14325. 64	-267. 98	1. 49	b ⁵ D ₀	3d ⁴ 4s ²	c ⁵ D	0	54846. 0		
3d ⁵ D ₃			3	14593. 62	-187. 41	1. 49	b ⁵ D ₁		1	54938. 1	92. 1		
3d ⁵ D ₂			2	14781. 03	-120. 03	1. 49	b ⁵ D ₂		2	55115. 8	177. 7		
3d ⁵ D ₁			1	14901. 06	-58. 62	1. 49	b ⁵ D ₃		3	55371. 3	255. 5		
3d ⁵ D ₀			0	14959. 68		0. 00	b ⁵ D ₄		4	55696. 5	325. 2		
a ⁵ G ₆	3d ⁵ (a ⁴ G)4s	a ⁵ G	6	27546. 90	-24. 05	1. 33	z ⁵ G ₂	3d ⁵ (a ⁴ G)4p	z ⁵ G ^o	2	64456. 33	16. 80	1. 31
a ⁵ G ₅			5	27570. 95	-12. 35	1. 26	z ⁵ G ₃		3	64473. 13	20. 70	1. 26	
a ⁵ G ₄			4	27583. 30	-4. 93		z ⁵ G ₄		4	64493. 83	24. 74		
a ⁵ G ₃			3	27588. 23	-0. 80		z ⁵ G ₅		5	64518. 57	31. 15		
a ⁵ G ₂			2	27589. 03			z ⁵ G ₆		6	64549. 72			
3d ³ P ₂	3d ⁶	a ³ P	2	29869. 11	-815. 70		z ⁵ H ₃	3d ⁵ (a ⁴ G)4p	z ⁵ H ^o	3	65482. 66	83. 02	1. 30
3d ³ P ₁			1	30684. 81			z ⁵ H ₄		4	65565. 68	92. 62	1. 21	
3d ³ P ₀			0				z ⁵ H ₅		5	65658. 30	96. 20		
a ⁵ P ₃	3d ⁵ (a ⁴ P)4s	a ⁵ P	3	29889. 31	-29. 91	1. 66	z ⁵ H ₆		6	65754. 50	92. 11		
a ⁵ P ₂			2	29919. 22	-31. 90	1. 84	z ⁵ H ₇		7	65846. 61			
a ⁵ P ₁			1	29951. 12		2. 427							
3d ³ H ₆	3d ⁶	a ³ H	6	30523. 45	-155. 79	1. 15	z ⁵ F ₅	3d ⁵ (a ⁴ G)4p	z ⁵ F ^o	5	66542. 26	-100. 75	1. 40
3d ³ H ₅			5	30679. 24	-116. 51	1. 04	z ⁵ F ₄		4	66643. 01	-43. 44		
3d ³ H ₄			4	30795. 75		0. 81	z ⁵ F ₃		3	66686. 45	9. 89		
							z ⁵ F ₂		2	66676. 56	31. 78		
							z ⁵ F ₁		1	66644. 78			
3d ³ F ₄	3d ⁶	a ³ F	4	31514. 35	-147. 37	1. 25	z ⁵ D ₁	3d ⁵ (a ⁴ P)4p	z ⁵ D ^o	0			
3d ³ F ₃			3	31661. 72	-99. 21	1. 07	z ⁵ D ₂		1	66893. 79	7. 35		
3d ³ F ₂			2	31760. 93		0. 66	z ⁵ D ₃		2	66901. 14	107. 79		
a ⁵ D ₄	3d ⁵ (a ⁴ D)4s	b ⁵ D	4	32787. 60	-69. 35		z ⁵ D ₄		3	67008. 93	286. 23	1. 45	
a ⁵ D ₃			3	32856. 95	-1. 89				4	67295. 16		1. 49	
a ⁵ D ₂			2	32858. 84	22. 44		z ⁵ S ₂ ?	3d ⁵ (a ⁴ P)4p	z ⁵ S ^o	2	66929. 22		
a ⁵ D ₁			1	32836. 40	18. 30								
a ⁵ D ₀			0	32818. 10			z ³ H ₆	3d ⁵ (a ⁴ G)4p	z ³ H ^o	6	67744. 08	-101. 85	1. 16
a ³ G ₅	3d ⁶	a ³ G	5	33147. 39	-100. 93	1. 20	z ³ H ₅		5	67845. 93	-64. 33	1. 03	
a ³ G ₄			4	33248. 32	-30. 07	1. 04	z ³ H ₄		4	67910. 26		0. 78	
a ³ G ₃			3	33278. 39		0. 75							
3d ³ G ₅	3d ⁵ (a ⁴ G)4s	b ³ G	5	34761. 81	-148. 62	1. 21	z ³ F ₂	3d ⁵ (a ⁴ G)4p	z ³ F ^o	2	67766. 51	45. 24	0. 66
3d ³ G ₄			4	34910. 43	-94. 00	1. 05	z ³ F ₃		3	67811. 75	53. 88	1. 08	
3d ³ G ₃			3	35004. 43		0. 75	z ³ F ₄		4	67865. 63		1. 25	
a ³ P ₂	3d ⁵ (a ⁴ P)4s	b ³ P	2	36274. 29	-90. 01	1. 48	z ⁵ P ₃	3d ⁵ (a ⁴ P)4p	y ⁵ P ^o	3	68284. 38	-132. 96	1. 65
a ³ P ₁			1	36364. 30	-63. 59	1. 48	z ⁵ P ₂		2	68417. 34	-79. 03		
a ³ P ₀			0	36427. 89		0. 00	z ⁵ P ₁		1	68496. 37			
3d ³ D ₁	3d ⁶	a ³ D	1	37811. 57	36. 33	0. 52	z ³ P ₂	3d ⁵ (a ⁴ P)4p	z ³ P ^o	2	69044. 61	-171. 12	1. 49
3d ³ D ₂			2	37847. 90	3. 25	1. 17	z ³ P ₁		1	69215. 73	-103. 28	1. 49	
3d ³ D ₃			3	37851. 15		1. 35	z ³ P ₀		0	69319. 01		0. 00	
4p ⁷ P ₂	3d ⁵ (a ⁶ S)4p	z ⁷ P ^o	2	38366. 07	176. 89	2. 32	y ⁵ F ₁	3d ⁵ (a ⁴ D)4p	y ⁵ F ^o	1	70150. 39	80. 68	
4p ⁷ P ₃			3	38542. 96	263. 57	1. 947	y ⁵ F ₂		2	70231. 07	111. 51		
4p ⁷ P ₄			4	38806. 53		1. 76	y ⁵ F ₃		3	70342. 58	154. 86	1. 39?	
							y ⁵ F ₄		4	70497. 44	159. 74	1. 40	
							y ⁵ F ₅		5	70657. 18			
a ³ D ₃	3d ⁵ (a ⁴ D)4s	b ³ D	3	39808. 14	-5. 24	1. 34	z ³ G ₃	3d ⁵ (a ⁴ G)4p	z ³ G ^o	3	70517. 72	28. 30	0. 75
a ³ D ₂			2	39813. 38	-13. 19	1. 15	z ³ G ₄		4	70546. 02	-18. 70	1. 15?	
a ³ D ₁			1	39826. 57		0. 50	z ³ G ₅		5	70527. 32		1. 20	
a ⁵ F ₁	3d ⁵ (a ⁴ F)4s	a ⁵ F	1	43311. 0	28. 1		z ³ D ₃	3d ⁵ (a ⁴ P)4p	z ³ D ^o	3	70745. 08	-195. 16	1. 33
a ⁵ F ₂			2	43339. 1	55. 9		z ³ D ₂		2	70940. 24	-137. 94	1. 16	
a ⁵ F ₃			3	43395. 0	141. 9		z ³ D ₁		1	71078. 18		0. 50	
a ⁵ F ₄			4	43536. 9	-8. 6		y ⁵ P ₁	3d ⁵ (a ⁴ D)4p	x ⁵ P ^o	1	71263. 92	59. 23	
a ⁵ F ₅			5	43528. 3			y ⁵ P ₂		2	71323. 15	66. 99		
							y ⁵ P ₃		3	71390. 14			

Mn II—Continued

Mn II—Continued

Curtis	Config.	Desig.	J	Level	Interval	Obs. g	Curtis	Config.	Desig.	J	Level	Interval	Obs. g	
y^5D_4	$3d^5(a^4D)4p$	y^5D°	4	72010.75	-236.63	1.48		$3d^5(a^6S)5d$	f^7D	1				
y^5D_3			3	72247.38	-59.43		$5d^7D_2$		2	99892.5		2.3		
y^5D_2			2	72306.81	-13.81		$5d^7D_3$		3	99894.8		3.8		
y^5D_1			1	72320.62	-1.45		$5d^7D_4$		4	99898.6		4.5		
y^5D_0			0	72322.07			$5d^7D_5$		5	99903.1				
y^3D_1	$3d^5(a^4D)4p$	y^3D°	1	73385.15	10.82		$5d^5D_4$	$3d^5(a^6S)5d$	f^5D	4	100682.3		-5.8	
y^3D_2			2	73395.97	-0.85		$5d^5D_3$		3	100688.1		-4.5		
y^3D_3			3	73395.12			$5d^5D_2$		2	100692.6		-2.7		
y^3F_4	$3d^5(a^4D)4p$	y^3F°	4	73683.11	-97.70	1.24	$5d^5D_1$		1	100695.3				
y^3F_3			3	73780.81	-4.44	1.09			0					
y^3F_2			2	73785.25		0.75	e^5G_6	$3d^5(a^4G)5s$	e^5G	6	101467.58		-21.73	
z^3S_1	$3d^5(a^4P)4p$	z^3S°	1	73911.31			e^5G_5		5	101489.31		-9.72		
$5s^7S_3$	$3d^5(a^6S)5s$	e^7S	3	74559.97		2.00	e^5G_4		4	101499.03		-2.27		
							e^5G_3		3	101501.30		1.46		
							e^5G_2		2	101499.84				
	$3d^5(a^4D)4p$	y^3P°	0				e^3G_5	$3d^5(a^4G)5s$	e^3G	5	102680.09		-23.25	
y^3P_1			1	75719.59	199.14		e^3G_4		4	102703.34		-2.03		
y^3P_2			2	75918.73			e^3G_3		3	102705.37				
$5s^5S_2$	$3d^5(a^6S)5s$	e^5S	2	76374.56		2.00	e^5P_3	$3d^5(a^4P)5s$	e^5P	3	103803.44		-32.73	
$4d^7D_1$	$3d^5(a^6S)4d$	e^7D	1	79540.76	3.75		e^5P_2		2	103836.17		-31.97		
$4d^7D_2$			2	79544.51	5.77		e^5P_1		1	103868.14				
$4d^7D_3$			3	79550.28	8.10		e^3P_0	$3d^5(a^4P)5s$	e^3P	0	104914.18?		106.49	
$4d^7D_4$			4	79558.38	10.72		e^3P_1		1	105020.67		34.76		
$4d^7D_5$			5	79569.10			e^3P_2		2	105055.43				
$4d^5D_4$	$3d^5(a^6S)4d$	e^5D	4	82136.20	-8.14		e^5H_3	$3d^5(a^4G)4d$	e^5H	3	106157.4		6.8	
$4d^5D_3$			3	82144.34	-6.73		e^5H_4		4	106164.2		4.7		
$4d^5D_2$			2	82151.07	-4.65		e^5H_5		5	106168.9		1.0		
$4d^5D_1$			1	82155.72	-2.44		e^5H_6		6	106169.9		-2.2		
$4d^5D_0$			0	82158.16			e^5H_7		7	106167.7				
z^7P_2	$3d^4 4s(a^6D)4p?$	y^7P°	2	83255.8	119.8		w^5F_1	$3d^4 4s(b^4D)4p$	w^5F°	1	106265.3		108.4	
z^7P_3			3	83375.6	153.8		w^5F_2		2	106373.7		152.1		
z^7P_4			4	83529.4			w^5F_3		3	106525.8		181.5		
							w^5F_4		4	106707.3		186.5		
							w^5F_5		5	106893.8				
$5p^7P_2$	$3d^5(a^6S)5p?$	x^7P°	2	85895.27	65.14									
$5p^7P_3$			3	85960.41	96.99		w^5P_1	$3d^4 4s(b^4D)4p$	u^5P°	1	106479.2		270.8	
$5p^7P_4$			4	86057.40			w^5P_2		2	106750.0		422.8		
							w^5P_3		3	107172.8				
$5p^5P_3$	$3d^5(a^6S)5p$	w^5P°	3	86897.7	-39.1									
$5p^5P_2$			2	86936.8	-24.2		e^5I_4	$3d^5(a^4G)4d$	e^5I	4	106512.1		7.0	
$5p^5P_1$			1	86961.0			e^5I_5		5	106519.1		3.4		
							e^5I_6		6	106522.5		-2.7		
x^5P_1	$3d^4 4s(a^6D)4p?$	v^5P°	1	88839.6	239.3		e^5I_7		7	106519.8		-11.7		
x^5P_2			2	89078.9	349.9		e^5I_8		8	106508.1				
x^5P_3			3	89428.8			e^5D_4	$3d^5(a^4D)5s$	g^5D	4	106885.90		-64.34	
$6s^7S_3$	$3d^5(a^6S)6s$	f^7S	3	97728.0			e^5D_3		3	106950.24		-16.70		
							e^5D_2		2	106966.94		4.38		
$6s^5S_2$	$3d^5(a^6S)6s$	f^5S	2	98410.1			e^5D_1		1	106962.56		6.14		
							e^5D_0		0	106956.42				
$4f^7F_6$	$3d^5(a^6S)4f$	z^7F°	6	98423.5	-0.2									
$4f^7F_5$			5	98423.7	-0.1		e^3D_3	$3d^5(a^4D)5s$	e^3D	3	108102.25		-68.20	
$4f^7F_4$			4	98423.8	-0.2		e^3D_2		2	108170.45		-2.09		
$4f^7F_3$			3	98424.0	-0.1		e^3D_1		1	108172.54?				
$4f^7F_2$			2	98424.1			$7s^7S_3$	$3d^5(a^6S)7s$	g^7S	3	108126.2			
			1											
			0				$5f^7F_{6,5}$	$3d^5(a^6S)5f$	y^7F°	{6}	108409.8		-0.3	
$4f^5F_1$	$3d^5(a^6S)4f$	x^5F°	1	98461.76	0.58				{5}	108410.1		-0.1		
$4f^5F_2$			2	98462.34	0.82		$5f^7F_4$		4	108410.2		-0.1		
$4f^5F_3$			3	98463.16	1.00		$5f^7F_3$		3	108410.3				
$4f^5F_4$			4	98464.16	0.99		$5f^7F_2$		2					
$4f^5F_5$			5	98465.15					1					
									0					

Mn II—Continued

Mn II—Continued

Curtis	Config.	Desig.	J	Level	Interval	Obs. g	Curtis	Config.	Desig.	J	Level	Interval	Obs. g			
5f ⁵ F ₁	3d ⁵ (a ⁶ S)5f	v ⁵ F ^o	1	108435.6			6f ⁷ F ₆	3d ⁵ (a ⁶ S)6f	x ⁷ F ^o	6	113840.0					
5f ⁵ F ₂			2	108437.2	1.6		6f ⁷ F ₅			5	113840.1	-0.1				
5f ⁵ F ₃			3	108439.0	1.8		6f ⁷ F ₄			4	113840.2	-0.1				
5f ⁵ F ₄			4	108441.4	2.4					3	113840.3	-0.1				
5f ⁵ F ₅			5	108443.0	1.6					2						
7s ⁵ S ₂	3d ⁵ (a ⁶ S)7s	g ⁵ S	2	108447.6						1						
v ⁵ G ₂			y ⁵ G ^o	2	108485.4			8s ⁵ S ₂	3d ⁵ (a ⁶ S)8s	h ⁵ S	2	113895.2				
v ⁵ G ₃				3	108503.0	17.6		6g ⁵ G ₆			3d ⁵ (a ⁶ S)6g	f ⁵ G	(6)			
v ⁵ G ₄				4	108524.7	21.7		6g ⁵ G ₅					(5)			
v ⁵ G ₅				5	108550.7	26.0		6g ⁵ G ₄					(4)	113932.0		
v ⁵ G ₆				6	108587.9	37.2		6g ⁵ G ₃					(3)			
v ⁵ P ₁	t ⁵ P ^o	1	108726.4			6g ⁵ G ₂	(2)									
v ⁵ P ₂		2	108974.7	148.3			7G	3d ⁵ (a ⁶ S)6g	e ⁷ G	(7)						
v ⁵ P ₃		3	109378.9	404.2						(1)	113932					
v ⁵ F ₁	u ⁵ F ^o	1	108994.0			6f ⁵ F ₁	3d ⁵ (a ⁶ S)6f	r ⁵ F ^o	1	114023.5						
v ⁵ F ₂		2	109045.7	51.7		6f ⁵ F ₂			2	114024.8	1.3					
v ⁵ F ₃		3	109122.4	76.7		6f ⁵ F ₃			3	114026.3	1.5					
v ⁵ F ₄		4	109221.1	98.7		6f ⁵ F ₄			4	114027.8	1.5					
v ⁵ F ₅		5	109327.1	106.0		6f ⁵ F ₅			5	114028.8	-1.0					
w ⁵ D ₀	x ⁵ D ^o	0	109167.7			7d ⁷ D ₁	3d ⁵ (a ⁶ S)7d	h ⁷ D	1	114344.18						
w ⁵ D ₁		1	109235.3	67.6		7d ⁷ D ₂			2	114344.84	0.66					
w ⁵ D ₂		2	109343.7	108.4		7d ⁷ D ₃			3	114345.87	1.03					
w ⁵ D ₃		3	109476.3	132.6		7d ⁷ D ₄			4	114347.25	1.38					
w ⁵ D ₄	4	109607.8	131.5		7d ⁷ D ₅	5	114349.06	1.81								
6d ⁷ D ₁	3d ⁵ (a ⁶ S)6d	g ⁷ D	1	109241.31			7d ⁵ D ₄	3d ⁵ (a ⁶ S)7d	h ⁵ D	4	114932.2					
6d ⁷ D ₂			2	109242.36	1.05		7d ⁵ D ₃			3	114943.9	-11.7				
6d ⁷ D ₃			3	109243.95	1.59		7d ⁵ D ₂			2	114951.9	-8.0				
6d ⁷ D ₄			4	109246.15	2.20		7d ⁵ D ₁			1	114958.5	-4.6				
6d ⁷ D ₅			5	109249.00	2.85		7d ⁵ D ₀			0	114958.1	-1.6				
v ⁵ D ₀	3d ⁴ 4s(b ⁴ D)4p	w ⁵ D ^o	0	109958.0			9s ⁷ S ₃	3d ⁵ (a ⁶ S)9s	i ⁷ S	3	117031.1					
v ⁵ D ₁			1	109994.3	36.3		7f ⁷ F ₆			3d ⁵ (a ⁶ S)7f	w ⁷ F ^o	6	117112.9			
v ⁵ D ₂			2	110068.5	74.2		7f ⁷ F ₅					5	117113.0	-0.1		
v ⁵ D ₃			3	110204.9	136.4		7f ⁷ F _{4,3}					(4)	117113.1	-0.1		
v ⁵ D ₄	4	110428.7	223.8			(3)										
v ⁵ H ₃	y ⁵ H ^o	3	110547.5						2							
v ⁵ H ₄		4	110602.0	54.5					1							
v ⁵ H ₅		5	110692.2	90.2					0							
v ⁵ H ₆		6	110795.0	102.8												
v ⁵ H ₇		7	110926.0	131.0												
u ⁵ F ₁	t ⁵ F ^o	1	111017.5				3d ⁵ (a ⁶ S)7f	q ⁵ F ^o	1							
u ⁵ F ₂		2	111060.5	43.0							2					
u ⁵ F ₃		3	111115.4	54.9							3					
u ⁵ F ₄		4	111159.2	43.8		7f ⁵ F ₄			4	117137.8						
u ⁵ F ₅		5	111160.5	1.3		7f ⁵ F ₅			5	117148.3	10.5					
u ⁵ P ₁	s ⁵ P ^o	1	111162.3			s ⁵ F ₁	3d ⁵ (a ⁶ S)7f	p ⁵ F ^o	1	117164.7						
u ⁵ P ₂		2	111178.8	16.5		s ⁵ F ₂			2	117231.7	67.0					
u ⁵ P ₃		3	111212.8	34.0		s ⁵ F ₃			3	117314.6	82.9					
u ⁵ G ₂	x ⁵ G ^o	2	113093.0			s ⁵ F ₄			4	117399.3	84.7					
u ⁵ G ₃		3	113110.3	17.3		s ⁵ F ₅			5	117483.2	83.9					
u ⁵ G ₄		4	113181.7	71.4												
u ⁵ G ₅		5	113250.8	69.1		10s ⁷ S ₃	3d ⁵ (a ⁶ S)10s	j ⁷ S	3	119185.6						
u ⁵ G ₆	6	113323.0	72.2													
t ⁵ F ₁	s ⁵ F ^o	1	113645.5			8f ⁵ F	3d ⁵ (a ⁶ S)8f	o ⁵ F ^o	(1)							
t ⁵ F ₂		2	113645.0	-0.5							(5)	119253				
t ⁵ F ₃		3	113641.4	-3.6												
t ⁵ F ₄		4	113646.7	5.3												
t ⁵ F ₅		5	113658.0	11.3												
8s ⁷ S ₃	3d ⁵ (a ⁶ S)8s	h ⁷ S	3	113697.6			Mn III (a ⁶ S _{2,4})	Limit			126147					

Mn II OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms			
	$ns (n \geq 4)$	$np (n \geq 4)$	$nd (n \geq 4)$	
$3d^6$	a^5D a^3P	a^3F	a^3G	a^3H
$3d^4 4s^2$	c^5D			
$3d^5(a^5S)nx$	$\{ a, e, f, g, h, i, j \}^7S$ $\{ a, e, f, g, h \}^5S$		z, x^7P^o z, w^5P^o	e, f, g, h^7D e, f, h^5D
$3d^5(a^4G)nx$		a, e^5G b, e^3G	z^5F^o z^3F^o	z^5G^o z^3G^o
$3d^5(a^4F)nx$	$\{ a, e^5P$ b, e^3P		y^5P^o z^3P^o	z^5D^o z^3D^o
$3d^5(a^4D)nx$	$\{ b, g^5D$ b, e^3D		x^5P^o y^3P^o	y^5D^o y^3D^o
$3d^5(a^4F)nx$	a^5F		y^7P^o v^5P^o	y^5F^o y^3F^o
$3d^4 4s(a^6D)nx$			u^5P^o	w^5D^o w^5F^o
$3d^4 4s(b^4D)nx$				
$3d^5(a^6S)nx$	$\{ x, y, r, q, o \}^7F^o$ x, v, r, q, o^5F^o			e^7G f^5G

*For predicted terms in the spectra of the Cr I isoelectronic sequence, see Vol. II, Introduction.

Mn III

(V I sequence; 23 electrons)

Z=25

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^5 {}^6S_{2\frac{1}{2}}$ $a {}^6S_{2\frac{1}{2}}$ 271800 cm^{-1}

I. P. 33.69 volts

The early analysis by Gibbs and White has been extended by Catalán and Miss García-Riquelme especially for inclusion here. The present observations extend from 594 Å to 3997 Å. The resonance lines reported by Kruger and Miss Gilroy have been confirmed. Fifteen new terms are listed in the current manuscript.

Catalán has derived the limit from the $ns {}^6D$ series ($n=4,5$) by means of a Ritz formula, with the assumed value of α equal to 3.3×10^{-6} . The total number of classified lines is 173. Observed intersystem combinations connect the systems of terms of different multiplicity.

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R. C. Gibbs and H. E. White, Proc. Nat. Acad. Sci. **13**, 525 (1927). (T) (C L)P. G. Kruger and H. T. Gilroy, Phys. Rev. **48**, 720 (1935). (T) (C L)

M. A. Catalán and O. García-Riquelme, unpublished material (1951). (I P) (T) (C L)

Mn III

Mn III

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3d^5$	$a {}^6S$	$2\frac{1}{2}$	0. 0		$3d^4({}^6D)4p$	$z {}^4F^\circ$	$1\frac{1}{2}$	116581. 6	111. 9
$3d^5$	$a {}^4G$	$5\frac{1}{2}$	26824. 5	-27. 9			$2\frac{1}{2}$	116693. 5	159. 5
		$4\frac{1}{2}$	26852. 4	-7. 9			$3\frac{1}{2}$	116853. 0	211. 4
		$3\frac{1}{2}$	26860. 3	3. 4			$4\frac{1}{2}$	117064. 4	
		$2\frac{1}{2}$	26856. 9		$3d^4({}^6D)4p$	$z {}^4D^\circ$	$0\frac{1}{2}$	120977. 2	117. 0
$3d^5$	$a {}^4P$	$2\frac{1}{2}$	29168. 9	-74. 1			$1\frac{1}{2}$	121094. 2	176. 3
		$1\frac{1}{2}$	29243. 0				$2\frac{1}{2}$	121270. 5	213. 2
		$0\frac{1}{2}$					$3\frac{1}{2}$	121483. 7	
$3d^5$	$a {}^4D$	$3\frac{1}{2}$	32308. 9	-76. 8	$3d^4({}^6D)4d$	$e {}^6G$	$1\frac{1}{2}$	172444. 6	103. 4
		$2\frac{1}{2}$	32385. 7				$2\frac{1}{2}$	172548. 0	135. 6
		$1\frac{1}{2}$					$3\frac{1}{2}$	172683. 6	172. 6
		$0\frac{1}{2}$					$4\frac{1}{2}$	172856. 2	208. 2
							$5\frac{1}{2}$	173064. 4	243. 2
							$6\frac{1}{2}$	173307. 6	
$3d^5$	$a {}^4F$	$4\frac{1}{2}$	43574. 2	-30. 0	$3d^4({}^6D)4d$	$e {}^6P$	$1\frac{1}{2}$	172571. 7	44. 5
		$3\frac{1}{2}$	43604. 2	-66. 3			$2\frac{1}{2}$	172616. 2	142. 3
		$2\frac{1}{2}$	43670. 5	-5. 1			$3\frac{1}{2}$	172758. 5	
		$1\frac{1}{2}$	43675. 6		$3d^4({}^6D)4d$	$e {}^6D$	$0\frac{1}{2}$		
$3d^4({}^6D)4s$	$a {}^6D$	$0\frac{1}{2}$	62456. 7	111. 2			$1\frac{1}{2}$		
		$1\frac{1}{2}$	62567. 9	179. 4			$2\frac{1}{2}$		
		$2\frac{1}{2}$	62747. 3	241. 6			$3\frac{1}{2}$	174175. 7	243. 4
		$3\frac{1}{2}$	62988. 9	296. 3			$4\frac{1}{2}$	174419. 1	
		$4\frac{1}{2}$	63285. 2		$3d^4({}^6D)4d$	$e {}^6F$	$0\frac{1}{2}$		
$3d^4({}^6D)4s$	$b {}^4D$	$0\frac{1}{2}$	71396. 5	169. 0			$1\frac{1}{2}$		
		$1\frac{1}{2}$	71565. 5	267. 9			$2\frac{1}{2}$		
		$2\frac{1}{2}$	71833. 4	351. 4			$3\frac{1}{2}$	174576. 8	298. 8
		$3\frac{1}{2}$	72184. 8				$4\frac{1}{2}$	174875. 6	100. 7
							$5\frac{1}{2}$	174976. 3	
$3d^4({}^6D)4p$	$z {}^6F^\circ$	$0\frac{1}{2}$	110036. 8	137. 1	$3d^4({}^6D)5s$	$f {}^6D$	$0\frac{1}{2}$	176100. 0	106. 0
		$1\frac{1}{2}$	110173. 9	225. 7			$1\frac{1}{2}$	176206. 0	186. 0
		$2\frac{1}{2}$	110399. 6	313. 2			$2\frac{1}{2}$	176392. 0	250. 0
		$3\frac{1}{2}$	110712. 8	400. 0			$3\frac{1}{2}$	176642. 0	304. 4
		$4\frac{1}{2}$	111112. 8	490. 2			$4\frac{1}{2}$	176946. 4	
		$5\frac{1}{2}$	111603. 0						
$3d^4({}^6D)4p$	$z {}^6P^\circ$	$1\frac{1}{2}$	111778. 0	107. 4	$3d^4({}^6D)4d$	$e {}^6S$	$2\frac{1}{2}$	181951. 3	
		$2\frac{1}{2}$	111885. 4	174. 3					
		$3\frac{1}{2}$	112059. 7						
$3d^4({}^6D)4p$	$z {}^4P^\circ$	$0\frac{1}{2}$	112815. 8	264. 4					
		$1\frac{1}{2}$	113080. 2	598. 2	Mn IV (6D_0)	Limit		271800	
		$2\frac{1}{2}$	113678. 4						
$3d^4({}^6D)4p$	$z {}^6D^\circ$	$0\frac{1}{2}$	113993. 3	103. 4					
		$1\frac{1}{2}$	114096. 7	193. 1					
		$2\frac{1}{2}$	114289. 8	-78. 9					
		$3\frac{1}{2}$	114210. 9	292. 1					
		$4\frac{1}{2}$	114503. 0						

Mn III OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms		
$3d^5$	$\left\{ \begin{array}{l} a^6S \\ a^4P \quad a^4D \quad a^4F \quad a^4G \end{array} \right.$		
	$ns \ (n \geq 4)$	$np \ (n \geq 4)$	
$3d^4(^5D)nx$	$\left\{ \begin{array}{l} a, f^6D \\ b^4D \end{array} \right.$	$\begin{array}{l} z^6P^\circ \\ z^4P^\circ \end{array}$	$\begin{array}{l} z^6D^\circ \\ z^4D^\circ \end{array} \quad \begin{array}{l} z^6F^\circ \\ z^4F^\circ \end{array}$
	$nd \ (n \geq 4)$		
$3d^4(^5D)nx$	$e^6S \quad e^6P \quad e^6D \quad e^6F \quad e^6G$		

*For predicted terms in the spectra of the V I isoelectronic sequence, see Vol. I, p. xxxvii.

Mn IV

(Ti I sequence; 22 electrons)

Z=25

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^4 ^5D_0$

a^5D_0 cm^{-1}

I. P. volts

The analysis is from Bowen, who has extended White's early work and published 156 classified lines, 100 in the range 540 Å to 675 Å, and 56 in the range 1442 Å to 1973 Å. The triplet and quintet systems of terms are connected by observed intersystem combinations.

F. L. Moore, Jr., has recently observed this spectrum in more detail and with higher accuracy. He confirms the multiplet by White ($3d^3 4s ^5F - 3d^3 4p ^5G^\circ$) without doubt, and expects to carry the analysis much further in the near future.

No series have been found.

REFERENCES

- I. S. Bowen, Phys. Rev. **52**, 1153 (1937). (T) (C L)
 F. L. Moore, Jr., private communication (May 1951).

Mn IV

Mn IV

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3d^4$	a^5D	0	0	92	$3d^3(a^4F)4p$	z^5D°	0		
		1	92	188			1		
		2	280	267			2	170861	411
		3	547	333			3	171272	488
		4	880				4	171760	
$3d^4$	a^3P	0	20646	628	$3d^3(a^4F)4p$	z^5F°	1	171381	313
		1	21274	1047			2	171694	382
		2	22321				3	172076	395
$3d^4$	a^3H	4	21273	196	$3d^3(a^4F)4p$	z^3D°	4	172471	392
		5	21469	207			5	172863	
		6	21676						
$3d^4$	a^3F	2	22786	73	$3d^3(a^4F)4p$	z^3G°	1	172081	310
		3	22859	98			2	172391	557
		4	22957				3	172948	
$3d^4$	a^3G	3	25434	232	$3d^3(a^4F)4p$	z^3F°	3	175422	382
		4	25666	209			4	175804	479
		5	25875				5	176283	
$3d^3(a^4F)4s$	a^5F	1	111502	204	$3d^3(a^4P)4p$	z^5P°	2	177624	446
		2	111706	300			3	178070	503
		3	112006	398			4	178573	
		4	112402	475			1	184560	336
		5	112877				2	184896	534
$3d^3(a^4F)4s$	b^3F	2	119431	524	$3d^3(a^2G)4p$	y^3G°	3	188753	454
		3	119555	644			4	189207	346
		4	120599				5	189553	
$3d^3(a^4F)4p$	z^5G°	2	167885	410	$3d^3(a^2H)4p$	x^3G°	5	200186	-98
		3	168295	535			4	200284	-40
		4	168830	662			3	200324	
		5	169492	786					
		6	170278						

May 1951.

Mn IV OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms		
$3d^4$	$\{ a^3P \quad a^5D \quad a^3F \quad a^3G \quad a^3H$		
	$ns \ (n \geq 4)$	$np \ (n \geq 4)$	
$3d^3(a^4F)nx$	$\{$	a^5F	$z^5D^\circ \quad z^5F^\circ \quad z^5G^\circ$
$3d^3(a^4P)nx$		b^3F	$z^3D^\circ \quad z^3F^\circ \quad z^3G^\circ$
$3d^3(a^2G)nx$			z^5P°
$3d^3(a^2H)nx$			y^3G°
			x^3G°

*For predicted terms in the spectra of the Ti I isoelectronic sequence, see Vol. I, p. xxxvii.

Mn v

(Sc I sequence; 21 electrons)

Z=25

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^3 \ ^4F_{1\frac{1}{2}}$ $a \ ^4F_{1\frac{1}{2}} 613200 \pm \text{cm}^{-1}$

I. P. 76 volts

The analysis is incomplete. The first work was by White who found twelve multiplets giving three of the leading low terms from the $3d^3$ configuration, two terms from $3d^2 4s$ and five from $3d^2 4p$. Later Bowen extended the analysis and observed intersystem combinations connecting the doublet and quartet terms. About one hundred lines are now classified, in the region from 382 Å to 1620 Å.

The terms are mostly from Bowen's 1935 paper. Two of his terms have been added from the 1940 reference below. The two terms $b \ ^4F$ and $b \ ^2F$ are from White with values slightly adjusted by the writer to fit Bowen's term array, since the two sets of measures of the extreme ultraviolet lines are not completely accordant.

White's 2F term is here called $b \ ^2F$ on the assumption that the 2F term having the configuration $3d^3$, which is not yet known, will probably be lower and should, therefore, be designated $a \ ^2F$.

The limit (entered in brackets in the table) has been calculated from White's estimated ionization potential, which in turn is based on his extrapolation of isoelectronic sequence data.

REFERENCES

H. E. White, Phys. Rev. **33**, 672 (1929). (I P) (T) (C L)I. S. Bowen, Phys. Rev. **47**, 924 (1935). (T) (C L)S. Pasternack, Astroph. J. **92**, 140 (1940). (T)

Mn v

Mn v

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval		
$3d^3$	$a \ ^4F$	$1\frac{1}{2}$	0	349	$3d^2(a \ ^3F)4p$	$z \ ^2F^\circ$	$2\frac{1}{2}$	245486	821		
		$2\frac{1}{2}$	349	$3\frac{1}{2}$			246307				
		$3\frac{1}{2}$	827								
		$4\frac{1}{2}$	1406								
$3d^3$	$a \ ^4P$	$0\frac{1}{2}$	16420	160	$3d^2(a \ ^3F)4p$	$z \ ^4D^\circ$	$0\frac{1}{2}$	246530	-387		
		$1\frac{1}{2}$	16580	456			$1\frac{1}{2}$	246143			
		$2\frac{1}{2}$	17036				$2\frac{1}{2}$	246880			
$3d^3$	$a \ ^2G$	$3\frac{1}{2}$	17878	504	$3d^2(a \ ^3F)4p$	$z \ ^2D^\circ$	$1\frac{1}{2}$	248074			
		$4\frac{1}{2}$	18382				$2\frac{1}{2}$				
$3d^3$	$a \ ^2P$	$1\frac{1}{2}$	22902	-176	$3d^2(a \ ^3F)4p$	$z \ ^2G^\circ$	$3\frac{1}{2}$	250949	751		
		$0\frac{1}{2}$	23078				$4\frac{1}{2}$	251700			
$3d^3$	$a \ ^2D$	$2\frac{1}{2}$	24610	-41	$3d^2(a \ ^3P)4p$	$z \ ^4S^\circ$	$1\frac{1}{2}$	257424			
		$1\frac{1}{2}$	24651								
$3d^3$	$a \ ^2H$	$4\frac{1}{2}$	24953	362	$3d^2(a \ ^3P)4p$	$y \ ^4D^\circ$	$0\frac{1}{2}$	260808	652		
		$5\frac{1}{2}$	25315				$1\frac{1}{2}$			261460	
$3d^2(a \ ^3F)4s$	$b \ ^4F$	$1\frac{1}{2}$	176937	377	$3d^2(a \ ^3P)4p$	$z \ ^4P^\circ$	$0\frac{1}{2}$	264382	330		
		$2\frac{1}{2}$	177314				$1\frac{1}{2}$	264712			
		$3\frac{1}{2}$	177862				$2\frac{1}{2}$	265468			
		$4\frac{1}{2}$	178560								
$3d^2(a \ ^3F)4s$	$b \ ^2F$	$2\frac{1}{2}$	183528	1105	$3d^2(a \ ^1G)4p$	$y \ ^2G^\circ$	$3\frac{1}{2}$	265555	160		
		$3\frac{1}{2}$	184633				$4\frac{1}{2}$	265715			
$3d^2(a \ ^3F)4p$	$z \ ^4G^\circ$	$2\frac{1}{2}$	241906	847	$3d^2(a \ ^1G)4p$	$z \ ^2H^\circ$	$4\frac{1}{2}$	271475	1149		
		$3\frac{1}{2}$	242753				$5\frac{1}{2}$	272624			
		$4\frac{1}{2}$	243790								
		$5\frac{1}{2}$	245036								
$3d^2(a \ ^3F)4p$	$z \ ^4F^\circ$	$1\frac{1}{2}$	243115	552	Mn VI (3F_2)	<i>Limit</i>	-----	[613200]			
		$2\frac{1}{2}$	243667								
		$3\frac{1}{2}$	244370							703	
		$4\frac{1}{2}$	245125							755	

Mn V OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ +	Observed Terms	
3d	{ a ⁴ P a ² D a ⁴ F a ² G a ² H a ² P	
	ns (n ≥ 4)	np (n ≥ 4)
3d ² (a ³ F)nx	{ b ⁴ F b ² F	z ⁴ D° z ⁴ F° z ⁴ G° z ² D° z ² F° z ² G°
3d ² (a ³ P)nx		z ⁴ S° z ⁴ P° y ⁴ D°
3d ² (a ¹ G)nx		y ² G° z ² H°

*For predicted terms in the spectra of the Sc I isoelectronic sequence see Vol. I, p. XXXVI.

Mn VI

(Ca I sequence; 20 electrons)

Z=25

Ground state 1s² 2s² 2p⁶ 3s² 3p⁶ 3d² ³F₂

a ³F₂ cm⁻¹

I. P. volts

The analysis is incomplete. Thirty lines were classified by Cady in the range between 307 Å and 329 Å. The present term list is from an unpublished manuscript by Edlén, who has revised Cady's analysis, by using unpublished wavelengths by Bowen.

REFERENCES

- W. M. Cady, Phys. Rev. **43**, 324 (1933). (T) (C L)
B. Edlén, unpublished material (Feb. 1949). (T)

Mn VI

Mn VI

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
3d ²	a ³ F	2	0	746	3d(² D)4p	z ³ F°	2	323795	1053
		3	746	923			3	324848	
		4	1669				4	326369	
3d ²	a ¹ D	2	15336		3d(² D)4p	z ³ P°	0	329732	-97
3d ²	a ³ P	0	17782	275			1	329635	
		1	18057				2	329995	
		2	18628	571					360
3d ²	a ¹ G	4	25511		3d(² D)4p	z ¹ F°	3	333063	
3d(² D)4p	z ¹ D°	2	319811		3d(² D)4p	z ¹ P°	1	336126	
3d(²D)4p	z ³D°	1	321695	714					
		2	322409		874				
		3	323283						

March 1949.

Mn VI OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ +	Observed Terms
3d ²	{ a ³ P a ¹ D a ³ F a ¹ G
	np (n ≥ 4)
3d(² D)nx	{ z ³ P ^o z ³ D ^o z ³ F ^o z ¹ P ^o z ¹ D ^o z ¹ F ^o

*For predicted terms in the spectra of the Ca I isoelectronic sequence, see Vol. I, p. xxxv.

Mn VII

(K I sequence; 19 electrons)

Z=25

Ground state 1s² 2s² 2p⁶ 3s² 3p⁶ 3d ²D_{1/2}

3d ²D_{1/2} 962001 cm⁻¹

I. P. 119.24 volts

Twenty-two lines have been classified in the range from 111 Å to 1267 Å. The terms are from the paper by Kruger and Weissberg.

REFERENCES

R. C. Gibbs and H. E. White, Proc. Nat. Acad. Sci. **12**, 676 (1926). (C L)
P. G. Kruger and S. G. Weissberg, Phys. Rev. **52**, 316 (1937). (I P) (T) (C L)

Mn VII

Mn VII

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
3p ⁶ (¹ S)3d	3d ² D	1½ 2½	0 1355	1355	3p ⁶ (¹ S)6f	6f ² F ^o	{ 2½ 3½ }	807760	
3p ⁶ (¹ S)4s	4s ² S	0½	318734		3p ⁶ (¹ S)7f	7f ² F ^o	{ 2½ 3½ }	848838	
3p ⁶ (¹ S)4p	4p ² P ^o	0½ 1½	397647 400112	2465	3p ⁶ (¹ S)8f	8f ² F ^o	{ 2½ 3½ }	875534	
3p ⁶ (¹ S)5s	5s ² S	0½	613934		3p ⁶ (¹ S)9f	9f ² F ^o	{ 2½ 3½ }	893742	
3p ⁶ (¹ S)4f	4f ² F ^o	2½ 3½	615957 616108	151	-----	-----	-----	-----	
3p ⁶ (¹ S)5f	5f ² F ^o	2½ 3½	739771 739945	174	Mn VIII (¹ S ₀)	Limit	-----	962001	
3p ⁶ (¹ S)6s	6s ² S	0½	752144						

May 1948.

Mn VIII

(A I sequence; 18 electrons)

Z=25

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 {}^1S_0$ $3p^6 {}^1S_0$ 1585000 cm^{-1}

I. P. 196 volts

Two lines, at 122 Å and 124 Å, respectively, are classified as combinations with the ground term. The limit and the 5s-levels are apparently extrapolated from isoelectronic sequence data, as indicated by brackets in the table. The listed values have been rounded off in the last places.

For convenience, the Paschen notation has been added by the writer in column one under the heading "A I". As for A I, the jl -coupling notation in the general form suggested by Racah is here introduced, although LS -designations as indicated in column two under the heading "Authors" are perhaps preferable for the terms thus far identified.

REFERENCES

P. G. Kruger, S. G. Weissberg and L. W. Phillips, Phys. Rev. **51**, 1090 (1937). (I P) (T) (C L)
G. Racah, Phys. Rev. **61**, 537 (L) (1942).

Mn VIII

A I	Authors	Config.	Desig.	J	Level
$1p_0$	$3p^6 {}^1S$	$3p^6$	$3p^6 {}^1S$	0	0
$1s_4$	$3p^5 4s {}^3P^{\circ}$	$3p^5({}^2P_{1/2})4s$	$4s [1\frac{1}{2}]^{\circ}$	2 1	806100
$1s_2$	$3p^5 4s {}^1P^{\circ}$	$3p^5({}^2P_{3/2})4s$	$4s' [0\frac{1}{2}]^{\circ}$	0 1	818500
$2s_4$	$3p^5 5s {}^3P^{\circ}$	$3p^5({}^2P_{1/2})5s$	$5s [1\frac{1}{2}]^{\circ}$	2 1	[1146000]
$2s_2$	$3p^5 5s {}^1P^{\circ}$	$3p^5({}^2P_{3/2})5s$	$5s' [0\frac{1}{2}]^{\circ}$	0 1	[1159000]
		----- Mn IX (${}^2P_{1/2}$)	----- <i>Limit</i>	--- ---	----- [1585000]

May 1948.

Mn IX

(Cl I sequence; 17 electrons)

 $Z=25$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^5 {}^2P_{1/2}$ $3p^5 {}^2P_{1/2}$ 1791000 cm^{-1}

I. P. 222 volts

All of the terms except $3p^6 {}^2S$ are from the paper by Edlén. Eleven lines in the region between 105 Å and 395 Å have been classified as combinations from the ground term. Edlén has estimated the value of the limit by extrapolation along the isoelectronic sequence, as indicated by brackets in the table. His unit, 10^3cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

- S. G. Weissberg and P. G. Kruger, Phys. Rev. **49**, 872 (A) (1936). (C L)
 B. Edlén, Zeit. Phys. **104**, 407 (1937). (I P) (T) (C L)

Mn IX

Config.	Desig.	J	Level	Interval
$3s^2 3p^5$	$3p^5 {}^2P^{\circ}$	$1\frac{1}{2}$ $0\frac{1}{2}$	0 12530	-12530
$3p^5$	$3p^5 {}^2S$	$0\frac{1}{2}$	265291	
$3s^2 3p^4({}^3P)4s$	$4s {}^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	873580 880070	-6490
$3s^2 3p^4({}^3P)4s$	$4s {}^2P$	$1\frac{1}{2}$ $0\frac{1}{2}$	889560 896860	-7300
$3s^2 3p^4({}^1D)4s$	$4s' {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	910890 911310	-420
$3s^2 3p^4({}^1S)4s$	$4s'' {}^2S$	$0\frac{1}{2}$	950060	
-----	-----	-----	-----	
Mn x (3P_2)	<i>Limit</i>	-----	[1791000]	

January 1948.

Mn x

(S I sequence; 16 electrons)

Z=25

Ground state $1s^2 2s^2 2p^6 3s^2 3p^4 {}^3P_2$ $3p^4 {}^3P_2$ 2004000 cm^{-1}

I. P. 248 volts

Edlén has classified 11 lines in the interval between 100 Å and 104 Å, and extrapolated a value of the limit from isoelectronic sequence data.

The singlet and triplet terms are not connected by observed intersystem combinations. By analogy with Fe XI, Edlén has revised the interpolated values of the singlet terms for inclusion here, and the uncertainty, x , is probably not large.

Edlén's unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

- B. Edlén, Zeit. Phys. **104**, 188 (1937). (I P) (T) (C L)
 B. Edlén, letter (January 1950). (T)

Mn x

Config.	Desig.	J	Level	Interval
$3s^2 3p^4$	$3p^4 {}^3P$	2	0	-10000 -1700
		1	10000	
		0	11700	
$3s^2 3p^4$	$3p^4 {}^1D$	2	$33810+x$	
$3s^2 3p^4$	$3p^4 {}^1S$	0	$73400+x$	
$3s^2 3p^3({}^4S^{\circ})4s$	$4s {}^3S^{\circ}$	1	965990	
$3s^2 3p^3({}^2D^{\circ})4s$	$4s' {}^3D^{\circ}$	1	991770	430 1980
		2	992200	
		3	994180	
$3s^2 3p^3({}^2D^{\circ})4s$	$4s' {}^1D^{\circ}$	2	$1002150+x$	
$3s^2 3p^3({}^2P^{\circ})4s$	$4s'' {}^1P^{\circ}$	1	$1032080+x$	
-----	-----	-----	-----	
Mn XI (${}^4S_{3/2}$)	<i>Limit</i>	-----	[2004000]	

January 1950.

Mn XIII

(Al I sequence; 13 electrons)

Z=25

Ground state $1s^2 2s^2 2p^6 3s^2 3p^2 P_{1/2}$ $3p^2 P_{1/2}$ cm⁻¹

I. P. volts

This spectrum has not been analyzed, but Edlén has classified 2 lines as follows:

I. A.	Int.	Wave No.	Desig.
66. 574	1	1502090	} $3p^2 P^{\circ} - 4d^2 D$
67. 215	2	1487760	

His unit, 10³ cm⁻¹, is here changed to cm⁻¹.

REFERENCE

B. Edlén, Zeit. Phys. **103**, 540 (1936). (C L)

December 1947.

Mn XIV

(Mg I sequence; 12 electrons)

Z=25

Ground state $1s^2 2s^2 2p^6 3s^2 1S_0$ $3s^2 1S_0$ 3260000 cm⁻¹

I. P. 404 volts

Edlén has classified 13 lines in the region between 57 Å and 79 Å. No intersystem combinations have been observed and the triplet terms are not all connected by observed combinations. He has determined the relative positions of the various groups of terms and also the ionization potential by extrapolation along the isoelectronic sequence. His estimated value of the limit is entered in brackets in the table.

His unit, 10³ cm⁻¹, has here been changed to cm⁻¹.

REFERENCE

B. Edlén, Zeit. Phys. **103**, 536 (1936). (I P) (T) (C L)

Mn XIV

Mn XIV

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3s^2$	$3s^2 1S$	0	0		$3s(2S)4d$	$4d^2 D$	1	1818250+x	500 890
$3s(2S)3p$	$3p^3 P^{\circ}$	0	218910+x	4790 11380			2	1818750+x	
		1	223700+x				3	1819640+x	
		2	235080+x			$3s(2S)4f$	$4f^3 F^{\circ}$	2	
$3s(2S)3d$	$3d^3 D$	1					3		
		2					4	1892970+x	
		3	640240+x		$3s(2S)5f$	$5f^3 F^{\circ}$	2		
$3s(2S)4s$	$4s^3 S$	1	1569110+x				3		
							4	2387760+x	
$3s(2S)4p$	$4p^1 P^{\circ}$	1	1685630		-----	-----	-----	-----	
					Mn xv ($2S_{1/2}$)	Limit	-----	[3260000]	

August 1947.

Mn xv

(Na I sequence; 11 electrons)

 $Z=25$ Ground state $1s^2 2s^2 2p^6 3s^2 S_{0\frac{1}{2}}$ $3s^2 S_{0\frac{1}{2}}$ 3511210 cm^{-1}

I. P. 435 volts

Edlén has classified 12 lines in the interval 45 Å to 75 Å and extrapolated the absolute value of the ground term from isoelectronic sequence data. His unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCE

B. Edlén, *Zeit. Phys.* **100**, 621 (1936). (I P) (T) (C L)

Mn xv

Mn xv

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
3s	3s 2S	$0\frac{1}{2}$	0		4f	4f $^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	1962870 1963250	380
3p	3p $^2P^\circ$	$0\frac{1}{2}$ $1\frac{1}{2}$	259450 276650	17200	5d	5d 2D	$1\frac{1}{2}$ $2\frac{1}{2}$	2492130	
3d	3d 2D	$1\frac{1}{2}$ $2\frac{1}{2}$	632760 634950	2190	5f	5f $^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	2520370 2520600	230
4s	4s 2S	$0\frac{1}{2}$	1666950		-----	-----	-----	-----	
4p	4p $^2P^\circ$	$0\frac{1}{2}$ $1\frac{1}{2}$	1770410 1777150	6740	Mn xvi (1S_0)	Limit	-----	[3511210]	
4d	4d 2D	$1\frac{1}{2}$ $2\frac{1}{2}$	1906350 1907470	1120					

June 1947.

Mn xvi

(Ne I sequence; 10 electrons)

 $Z=25$ Ground state $1s^2 2s^2 2p^6 ^1S_0$ $2p^6 ^1S_0$ 9164300 cm^{-1}

I. P. 1135.9 volts

Tyrén has classified nine lines, between 13 Å and 18 Å, as combinations with the ground term. His absolute term values have been derived by extrapolation along the Ne I isoelectronic sequence.

By analogy with Ne I the jl -coupling notation in the general form suggested by Racah is introduced.

The unit adopted by Tyrén, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

F. Tyrén, *Zeit. Phys.* **111**, 314 (1938). (I P) (T) (C L)
G. Racah, *Phys. Rev.* **61**, 537 (L) (1942).

Mn XVI

Mn XVI

Author	Config.	Desig.	<i>J</i>	Level	Author	Config.	Desig.	<i>J</i>	Level
2 <i>p</i> ¹ S ₀	2 <i>s</i> ² 2 <i>p</i> ⁶	2 <i>p</i> ⁶ ¹ S	0	0	3 <i>p</i> ' ³ P ₁	2 <i>s</i> 2 <i>p</i> ⁶ (² S)3 <i>p</i>	3 <i>p</i> ³ P ^o	2 1 0	6530800
3 <i>s</i> ³ P ₁	2 <i>s</i> ² 2 <i>p</i> ⁵ (² P _{1/2})3 <i>s</i>	3 <i>s</i> [1½] ^o	2 1	5281200	3 <i>p</i> ¹ P ₁	2 <i>s</i> 2 <i>p</i> ⁶ (² S)3 <i>p</i>	3 <i>p</i> ¹ P ^o	1	6562500
3 <i>s</i> ¹ P ₁	2 <i>s</i> ² 2 <i>p</i> ⁵ (² P _{3/2})3 <i>s</i>	3 <i>s</i> '[0½] ^o	0 1	5360800	4 <i>d</i> ¹ P ₁	2 <i>s</i> ² 2 <i>p</i> ⁵ (² P _{1/2})4 <i>d</i>	4 <i>d</i> [1½] ^o	1	7348000
3 <i>d</i> ³ P ₁	2 <i>s</i> ² 2 <i>p</i> ⁵ (² P _{1/2})3 <i>d</i>	3 <i>d</i> [0½] ^o	0 1	5849700	4 <i>d</i> ³ D ₁	2 <i>s</i> ² 2 <i>p</i> ⁵ (² P _{3/2})4 <i>d</i> '	4 <i>d</i> '[1½] ^o	1	7429000
3 <i>d</i> ¹ P ₁		3 <i>d</i> [1½] ^o	1	5923500					
3 <i>d</i> ³ D ₁	2 <i>s</i> ² 2 <i>p</i> ⁵ (² P _{3/2})3 <i>d</i>	3 <i>d</i> '[1½] ^o	1	6018300		Mn XVII (² P _{1/2})	<i>Limit</i>	---	[9164300]
						Mn XVII (² P _{3/2})	<i>Limit</i>	---	9249600

April 1947.

Mn XVI OBSERVED LEVELS*

Config. 1 <i>s</i> ² +	Observed Terms		
2 <i>s</i> ² 2 <i>p</i> ⁶	2 <i>p</i> ⁶ ¹ S		
	<i>ns</i> (<i>n</i> ≥ 3)	<i>np</i> (<i>n</i> ≥ 3)	<i>nd</i> (<i>n</i> ≥ 3)
2 <i>s</i> ² 2 <i>p</i> ⁵ (² P ^o) <i>nx</i>	{ 3 <i>s</i> ³ P ^o 3 <i>s</i> ¹ P ^o	{ 3 <i>p</i> ³ P ^o 3 <i>p</i> ¹ P ^o	3 <i>d</i> ³ P ^o 3, 4 <i>d</i> ³ D ^o 3, 4 <i>d</i> ¹ P ^o
2 <i>s</i> 2 <i>p</i> ⁶ (² S) <i>nx</i>			
<i>jl</i> -Coupling Notation			
Config. 1 <i>s</i> ² +	Observed Pairs		
	<i>ns</i> (<i>n</i> ≥ 3)		<i>nd</i> (<i>n</i> ≥ 3)
2 <i>s</i> ² 2 <i>p</i> ⁵ (² P _{1/2}) <i>nx</i>	3 <i>s</i> [1½] ^o		3 <i>d</i> [0½] ^o 3, 4 <i>d</i> [1½] ^o
2 <i>s</i> ² 2 <i>p</i> ⁵ (² P _{3/2}) <i>nx</i> '	3 <i>s</i> '[0½] ^o		3, 4 <i>d</i> '[1½] ^o

*For predicted levels in the spectra of the Ne I isoelectronic sequence see Vol. I, p. xxxi.

IRON

Fe I

26 electrons

 $Z=26$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^6 4s^2 {}^5D_4$ $a {}^5D_4$ 63700 cm^{-1} I. P. 7.90 ± 0.01 volts

The terms are from the paper by Russell, Moore, and Weeks, which is "based on the work of many investigators, including unpublished studies by Miguel A. Catalán," which were not published because of the Spanish war. He contributed 89 new levels including the first singlet terms to be found in Fe I. Their table of all classified lines that have been observed in the laboratory extends from 1855 Å to 11973 Å and includes 3,606 lines. In addition they give a table of 1,254 predicted lines that have been accepted as present in the solar spectrum and graded "Good" or "Fair". It is reasonably certain that more faint solar lines may be correctly attributed to Fe I than are included in the present conservative list.

Intersystem combinations connecting the terms of all four multiplicities have been observed. One small change has been incorporated here. The miscellaneous level labeled 8° is here designated $x {}^3S^\circ$, as was tentatively suggested by the authors.

The ionization potential is well determined from three series.

The analysis could be carried further if laboratory observations of faint lines were extended. The present analysis is, however, remarkably complete for such a complex spectrum. The present lists contain 4,860 classified lines arising from combinations among 464 energy levels. With the aid of extensive Zeeman data all but 18 of these have been grouped into 147 terms, which combine to give 1,348 multiplets. "The general result of this analysis is that the iron arc spectrum, despite its complexity, is highly regular." The detailed confirmation of theory is conspicuous.

REFERENCES

- H. N. Russell, C. E. Moore, and D. W. Weeks, *Trans. Am. Phil. Soc.* **34**, Part 2, 111-207 (1944). (I P) (T)
(C L) (Z E)
J. E. Mack, *Rev. Mod. Phys.* **22**, No. 1, 64 (1950). (Summary hfs)
R. A. Fisher, unpublished infrared data (June 1950). ($x {}^7P^\circ$ term added in proof.)

Fe I

Fe I

Config.	Desig.	J	Level	Interval	Obs. g	Config.	Desig.	J	Level	Interval	Obs. g						
$3d^6 4s^2$	a^5D	4	0.000		1.496	$3d^6 4s(a^6D)4p$	z^5F^o	5	26874.562	-292.275	1.399						
		3	415.933	-415.933	1.497			4	27166.837	-227.866	1.355						
		2	704.003	-288.070	1.494			3	27394.703	-164.895	1.250						
		1	888.132	-184.129	1.498			2	27559.598	-106.764	1.004						
		0	978.074	-89.942				1	27666.362		-0.012						
$3d^7(a^4F)4s$	a^5F	5	6928.280		1.404	$3d^7(a^2P)4s$	a^1P	1	27543.00		0.817						
		4	7376.775	-448.495	1.349			$3d^7(a^2D)4s$	a^1D	2	28604.61		1.028				
		3	7728.071	-351.296	1.248					$3d^7(a^2H)4s$	a^1H	5	28819.98		1.000		
		2	7985.795	-257.724	0.995							$3d^6 4s(a^6D)4p$	z^5P^o	3	29056.341	-412.692	1.657
		1	8154.725	-168.930	-0.014									2	29469.033	-263.716	1.835
$3d^7(a^4F)4s$	a^3F	4	11976.260	-584.693	1.254	1	29732.749								2.487		
		3	12560.953	-407.620	1.086	$3d^6 4s^2$	a^1I	6	29313.04						1.014		
		2	12968.573		0.670			$3d^6 4s^2$	b^3D	1	29320.05			36.73	1.326		
		$3d^7(a^4P)4s$	a^5P	3	17550.210					-176.807	1.666	2	29356.78	15.08			
				2	17727.017					-200.394	1.820	3	29371.86				
1	17927.411				2.499					$3d^6 4s^2$	b^1G	4	29798.96			0.979	
$3d^6 4s^2$	a^3P			2	18378.215	-1174.278	1.506					$3d^6 4s(a^4D)4p$	z^3F^o	4		31307.272	-497.825
				1	19552.493	-485.37	1.500	3	31805.097					-328.917	1.086		
		0	20037.86			2	32134.014		0.682								
		$3d^6 4s(a^6D)4p$	z^7D^o	5	19350.894	-211.563	1.597	$3d^6 4s(a^4D)4p$	z^3D^o					3	31322.639	-363.738	1.321
				4	19562.457	-194.583	1.642			2	31686.377			-250.973	1.168		
3	19757.040			-155.471	1.746	1	31937.350				0.513						
2	19912.511			-107.137	2.008	$3d^6$	c^3F			4	32873.68	-539.10	1.264				
1	20019.648				2.999					3	33412.78	-352.55	1.066				
$3d^6 4s^2$	a^3H	6	19390.197	-230.839	1.163			2	33765.33		0.677						
		5	19621.036	-167.244	1.038			$3d^7(a^4F)4p$	y^5D^o	4	33095.962	-411.182	1.496				
		4	19788.230		0.811					3	33507.144	-294.451	1.492				
		$3d^6 4s^2$	b^3F	4	20641.144	-233.377	1.235			2	33801.595	-215.532	1.495				
				3	20874.521	-164.500	1.073			1	34017.127	-104.496	1.492				
2	21039.021				0.663	0	34121.623										
$3d^7(a^2G)4s$	a^3G			5	21715.770	-283.397	1.197	$3d^7(a^4F)4p$	y^5F^o	5	33695.418	-344.122	1.417				
				4	21999.167	-250.294	1.051			4	34039.540	-289.235	1.344				
		3	22249.461		0.756	3	34323.775			-218.460	1.244						
		$3d^6 4s(a^6D)4p$	z^7F^o	6	22650.427	-195.453	1.498			2	34547.235	-144.937	0.998				
				5	22845.880	-150.806	1.498			1	34692.172		-0.016				
4	22996.636			-114.262	1.493	$3d^6 4s(a^4D)4p$	z^3P^o	2	33946.965	-415.925	1.493						
3	23110.948			-81.560	1.513			1	34362.890	-192.75	1.496						
2	23192.508			-52.339	1.504			0	34555.64								
1	23244.847	-25.545	1.549	$3d^6 4s^2$	b^1D			2	34636.82								
0	23270.392							$3d^7(a^4F)4p$	z^5G^o	6	34843.980	61.532	1.332				
$3d^7(a^4P)4s$	b^3P	2	22838.360			-108.500	1.498			5	34782.448	-474.897	1.218				
		1	22946.860			-104.930	1.489			4	35257.345	-354.304	1.103				
		0	23051.790							3	35611.649	-244.775	0.887				
		$3d^6 4s(a^6D)4p$	z^7P^o	4	23711.467	-469.409	1.747			2	35856.424		0.335				
				3	24180.876	-326.052	1.908	$3d^7(a^4F)4p$	z^3G^o	5	35379.237	-388.354	1.248				
2	24506.928				2.333	4	35767.591			-311.804	1.100						
$3d^6 4s^2$	b^3G			5	23783.654	-335.200	1.200			3	36079.395		0.791				
				4	24118.854	-219.951	1.048			$3d^7(a^4F)4p$	y^3F^o	4	36686.204	-476.566	1.246		
		3	24338.805		0.761	3	37162.770					-358.416	1.086				
		$3d^7(a^2P)4s$	c^3P	2	24335.804	-436.256	1.484	2	37521.186				0.688				
				1	24772.060	-319.56	1.466	$3d^6 4s(a^4D)4p$	y^5P^o			3	36766.998	-390.596	1.661		
0	25091.62					2	37157.594					-251.981	1.836				
$3d^7(a^2G)4s$	a^1G			4	24574.690		1.001			1	37409.575		2.502				
				$3d^6 4s(a^6D)4p$	z^5D^o	4	25900.002			-240.191	1.502	$3d^7(b^2F)4s$	d^3F	2	36940.60	35.04	1.248
		3	26140.193			-199.515	1.500			3	36975.64			70.36			
		2	26339.708			-139.685	1.503	4	37046.00								
		1	26479.393			-71.102	1.495	$3d^7(a^4F)4p$	y^3D^o	3	38175.382			-502.685	1.324		
0	26550.495					2	38678.067			-317.697	1.151						
$3d^7(a^2H)4s$	b^3H	6	26105.95	-245.14	1.165	1	38995.764				0.493						
		5	26351.09	-276.55	1.032												
		4	26627.64		0.811												
		$3d^7(a^2D)4s$	a^3D	3	26225.03	-398.70	1.335										
				2	26623.73	217.24	1.178										
1	26406.49				0.731												

Fe I—Continued

Fe I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>				
$3d^6 4s(a^4D)4p$	x^5D°	4	39625. 829		1. 489	$3d^6 4s(b^4F)4p$	x^5G°	6	45608. 35:		1. 336				
		3	39969. 880	-344. 051	1. 504			5	45726. 18	-117. 83	1. 269				
		2	40231. 365	-261. 485	1. 501			4	45833. 24	-107. 06	1. 158				
		1	40404. 544	-173. 179	1. 498			3	45913. 53	-80. 29	0. 928				
		0	40491. 312	-86. 768				2	45964. 98	-51. 45	0. 323				
$3d^6 4s^2(a^6S)4p$	y^7P°	2	40052. 08		2. 340	$3d^6 4s(a^4H)4p$	z^2I°	7	45978. 04:		1. 149				
		3	40207. 12	155. 04	1. 908			6	46026. 98	-48. 94	1. 040				
		4	40421. 89	214. 77	1. 75:			5	46135. 92	-108. 94	0. 833				
$3d^6 4s(a^4D)4p$	x^5F°	5	40257. 367		1. 390	$3d^6 4s^2(a^6S)4p$	w^5P°	3	46137. 14		1. 658				
		4	40594. 47	-337. 10	1. 328			2	46313. 61	-176. 47	1. 822				
		3	40842. 13	-247. 66	1. 254			1	46410. 44	-96. 83	2. 436				
		2	41018. 06	-165. 93	0. 998										
		1	41130. 62	-112. 56	-0. 006										
$3d^7(b^2F)4s$	a^1F	3	40534. 18:			$3d^6 4s(b^4P)4p$	z^3S°	1	46600. 884		1. 888				
		X	3	40871. 46											
		2	41178. 36	-306. 90											
$3d^6 4s(b^4P)4p$	z^5S°	2	40895. 022		1. 985	$3d^7(a^4P)4p$	y^3P°	0	46672. 57						
								1	46901. 892	229. 32	1. 600				
$3d^6 4s(b^4P)4p$	x^5P°	3	42532. 76		1. 650	$3d^7(a^4P)4p$	u^5D°	4	46720. 85		1. 341				
		2	42859. 829	-327. 07	1. 822			3	46745. 03	-24. 18	1. 397				
		1	43079. 05	-219. 22	2. 464			2	46888. 582	-143. 55	1. 260				
$3d^6 4s(a^4H)4p$	y^5G°	6	42784. 387		1. 342	$3d^7(a^2G)4p$	x^3F°	4	46889. 207		1. 344				
		5	42911. 918	-127. 531	1. 203			3	47092. 776	-203. 569	1. 159				
		4	43022. 998	-111. 080	1. 024			2	47197. 074	-104. 298	0. 743				
		3	43137. 511	-114. 513	0. 905										
		2	43210. 044	-72. 533	0. 331			$3d^6 4s(a^4H)4p$	z^3H°	6	46982. 383		1. 200		
$3d^6 4s(a^6D)5s$	e^7D	5	42815. 855		1. 585	5	47008. 428			-26. 045	1. 060				
		4	43163. 327	-347. 472	1. 655	4	47106. 544			-98. 116	0. 880				
		3	43434. 633	-271. 306	1. 755										
		2	43633. 535	-198. 902	2. 009	$3d^7(a^4F)5s$	e^5F			5	47005. 510		1. 421		
		1	43763. 982	-130. 447	3. 002			4	47377. 967	-372. 457	1. 331				
				3	47755. 538			-377. 571	1. 236						
				2	48036. 667			-281. 129	0. 991						
				1	48221. 323			-184. 656	0. 007						
$3d^6 4s(a^4H)4p$	z^5H°	7				$3d^7(a^4P)4p$	w^3D°	3	47017. 239		1. 346				
		6	43321. 12:		1. 054			2	47136. 142	-118. 903	1. 216				
		5	42991. 66	329. 46	0. 871			1	47272. 095	-135. 953	0. 767				
		4	43108. 944	-117. 28	0. 871										
		3	43325. 93	-217. 04	0. 509										
$3d^6 4s(b^4P)4p$	w^5D°	4	43499. 54		1. 492	$3d^6 4s(a^4G)4p$	w^5G°	6	47363. 39		1. 306				
		3	43922. 70	-423. 16	1. 481			5	47420. 23	-56. 84	1. 305				
		2	44183. 64	-260. 94	1. 533			4	47590. 07	-169. 84	1. 145				
		1	44411. 18	-227. 54	1. 315			3	47693. 289	-103. 22	0. 931				
		0	44458. 96	-47. 78				2	47831. 20	-137. 91	0. 472				
$3d^6 4s(b^4F)4p$	w^5F°	5	44243. 67		1. 382	$3d^7(a^2G)4p$	z^1G°	4	47452. 770		1. 025				
		4	44022. 55	221. 12	1. 444			$3d^7(a^4P)4p$	y^3S°	1	47555. 63		1. 884		
		3	44166. 24	-143. 69	1. 351					$3d^6 4s(a^4G)4p$	v^5F°	5	47606. 10		1. 317
		2	44285. 48	-119. 24	1. 117							4	47930. 04	-323. 94	1. 264
		1	44378. 42:	-92. 94	0. 283							3	48122. 97	-192. 93	1. 236
				2	48238. 903	-115. 93	1. 267								
				1	48350. 62	-111. 72	0. 230								
$3d^6 4s(b^4F)4p$	v^5D°	4	44415. 13		1. 401	$3d^6 4s(b^4F)4p$	x^3G°	3	47834. 26		0. 668				
		3	44551. 44	-136. 31	1. 386			4	47812. 18	-22. 08	1. 061				
		2	44664. 13	-112. 69	1. 378			5	47834. 622	22. 44	1. 203				
		1	44760. 79	-96. 66	1. 389										
		0	44826. 92	-66. 13											
$3d^7(a^4P)4p$	y^5S°	2	44511. 86		1. 888	$3d^7(a^4F)5s$	e^3F	4	47960. 973		1. 288				
		$3d^6 4s(a^6D)5s$	e^5D	4	44677. 010				1. 502	3	48531. 896	-570. 923	1. 107		
				3	45061. 334			-384. 324	1. 508	2	48928. 423	-396. 527	0. 622		
				2	45333. 880			-272. 546	1. 503						
				1	45509. 155			-175. 275	1. 518						
0	45595. 084			-85. 929											
$3d^6 4s(b^4P)4p$	x^3D°	3	45220. 738		1. 352	$3d^7(a^4P)4p$	v^5P°	3	47966. 63		1. 646				
		2	45281. 889	-61. 151	1. 200			2	48163. 49	-196. 86	1. 740				
		1	45551. 833	-269. 944	0. 556			1	48289. 89	-126. 40	2. 213				
$3d^6 4s(a^4H)4p$	y^3G°	5	45294. 86		1. 207		w^3G°	5	48231. 33		1. 27:				
		4	45423. 456	-133. 60	1. 053			4	48361. 92	-130. 59	0. 934				
		3	45563. 026	-134. 570	0. 765			3	48475. 74	-113. 82	0. 584				

Fe I—Continued

Fe I—Continued

Config.	Desig	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>			
$3d^6 4s(b^4P)4p$	x^3P°	2	48304.707	-211.44 56.03	1.263 1.547	$3d^6 4s(a^6D)5p$	u^5F°	5	51016.72:	-364.76 -237.66 -208.45 -118.27				
		1	48516.15					4	51381.48					
		0	48460.12					3	51619.14:					
$3d^7(a^2G)4p$	z^1H°	5	48382.63		1.018			2	51827.59:					
$3d^7(a^2H)4p:$	y^1G°	4	48702.57		1.063	$3d^6 4s(a^4G)4p:$	x^3H°	6	51023.19	-45.58	1.161 1.038			
		2	49052.93	5	51068.77									
$3d^6 4s(b^4F)4p$	w^3F°	4	49108.94	-134.01 -190.23	1.181 1.165 0.677	$3d^6 4s(a^6D)5p$	t^5D°	4	51076.68	-284.78 -268.61 -206.80 -104.89	1.486			
		3	49242.950					3	51361.46					
		2	49433.18					2	51630.07:					
$3d^6 4s(b^4F)4p$	v^3D°	3	49135.08	-107.60 -54.98	1.211 0.954 0.562	$3d^6 4s(a^6D)4d$	f^5F	5	51103.237	-358.470 -142.439 -100.906 -49.482	1.384 1.355: 0.967			
		2	49242.68					4	51461.707					
		1	49297.66					3	51604.146					
$3d^7(a^2G)4p$	y^3H°	6	49434.20	-170.25 -122.61	1.17: 1.075 0.929	$3d^6 4s(a^6D)4d$	e^5S	2	51148.892		1.952			
		5	49604.45					$3d^6 4s(a^4G)4p:$	v^3F°	2	51201.33	163.97 -60.65	0.803 1.096 1.122	
		4	49727.058							3	51365.30			
										4	51304.65			
		1	51754.534											
$3d^7(a^2G)4p$	v^3G°	5	49460.92	-167.00 -222.69	1.163 0.914 0.763	$3d^6 4s(a^4D)5s$	e^3D	3	51294.262	-445.702 -299.975	1.345 1.125 0.801			
		4	49627.92					2	51739.964					
		3	49850.61					1	52039.939					
$3d^6 4s(a^6D)5p$	x^7P°	4	49558.5	-246.4 -241.0	0.92: 1.389 1.469	$3d^6 4s(a^4D)5s$	g^5D	4	51350.505	-420.072 -279.24 -164.51 -43.00	1.487 1.492 1.57: 0.801			
		3	49804.9					3	51770.577					
		2	50045.9					2	52049.82					
$3d^7(a^2D)4p:$	w^3P°	0	49951.36	91.89 143.62	1.389 1.469	$3d^7(a^2H)4p$	u^3G°	5	51373.96	-294.26 -157.58	1.140 1.067 0.801			
		1	50043.25					4	51668.22					
		2	50186.87					3	51825.80					
$3d^6 4s(a^6D)4d$	e^7F	6	50342.180	-491.305 -358.835 43.45 -182.22 123.05	1.490 1.505 1.617 1.499 2.490	$3d^6 4s(a^6D)4d$	e^7S	3	51570.16		1.92:			
		5	50833.485					4	51668.22		1.061			
		4	51192.320					3	51825.80					
		3	51148.87					4	51409.18		0.953			
		2	51331.090					5	51435.90:					
		1	51208.04					0						
$3d^6 4s(a^6D)4d$	f^7D	5	50377.92	-430.13 -53.80 -136.84 -49.41	1.510 1.574 1.844	$3d^6 4s(a^6D)5p$	u^5P°	3	51691.98:	-253.33 -165.0	2.633			
		4	50808.053					2	51945.31:					
		3	50861.85					1	52110.3:					
		2	50998.686											
		1	51048.10											
$3d^6 4s(a^6D)4d$	f^5D	4	50423.185	-111.250 -164.231 -181.486 -100.87	1.514 1.615 1.614 1.662	$3d^7(a^2P)4p$	y^1D°	2	51708.33		1.025			
		3	50534.435					$3d^6 4s(a^6D)4d$	e^5P	3	51837.279	-230.17 47.74	1.664 2.432	
		2	50698.666							2	52067.45			
		1	50880.152							1	52019.706			
		0	50981.02											
$3d^6 4s(a^6D)4d$	e^5G	6	50522.94	-180.97 -275.715 -239.432 -151.125	1.351 1.360 1.238 1.294 0.953	$3d^7(a^2P)4p$	u^3D°	3	51969.14	-327.82 -215.50	1.306 1.156 0.700			
		5	50703.912					$3d^7(a^2D)4p:$	t^3D°			2	52296.96	
		4	50979.627									1?	52180.82	502.11 -469.64
		3	51219.059									2	52682.93	
		2	51370.184									3	52213.29	
$3d^7(a^2G)4p:$	z^1F°	3	50586.89		1.018	$3d^7(a^2H)4p$	w^3H°	6	52431.47	-181.61 -155.70	1.177 1.033 0.810			
								5	52613.08					
$3d^6 4s(a^6D)4d$	e^7G	7	50651.76:	-316.11 -260.722 -106.34 -125.59 -79.24 -27.09	1.415 1.379 1.338 1.244	$3d^7(a^2H)4p$	y^3I°	7	52655.04:	141.45 -385.47	1.147 1.019 0.830			
		6	50967.873					$3d^7(a^2P)4p$	x^3S°			1	52857.84	
		5	51228.595											
		4	51334.94											
		3	51460.53											
2	51539.77													
	1	51566.86		-0.374										

Fe I—Continued

Fe I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>			
$3d^7(a^2P)4p:$	v^3P°	2	52916. 33	-892. 04	1. 495	$3d^7(a^4F)4d$	f^3F	4	54683. 39	-441. 58	1. 141			
		1	53808. 37					3	55124. 974			1. 071		
		0						2	55378. 842	-253. 868	0. 676			
	s^3D°	3	52953. 68:	-321. 59	1. 231	$3d^7(a^4F)4d$	w^1G°	4	54810. 82		1. 001			
		2	53275. 27					1						
$3d^7(a^4F)4d$	g^5F	5	53061. 28	-332. 44				2	54879. 720	-496. 397	1. 459			
		4	53393. 715						1			55376. 117	-350. 42	1. 459
		3	53830. 96						0			55726. 54:		
		2	54257. 52											
		1	54386. 16											
$3d^7(a^2H)4p$	z^1I°	6	53093. 60		1. 010									
$3d^7(a^4F)4d$	h^5D	4	53155. 13	-390. 75	1. 435	$3d^7(a^2D)4p:$	w^1D°	2	55754. 29		0. 990			
		3	53545. 882											
		2	53966. 720											
		1	54132. 48											
		0												
$3d^7(a^4F)4d$	f^5P	3	53160. 53	-408. 19				5	55907. 22	1. 66	1. 145			
		2	53568. 72						4			55905. 56	-192. 29	0. 857
		1	53925. 26						3			56097. 85		
$3d^7(a^4F)4d$	f^5G	6	53169. 21	-112. 53	1. 323	$3d^6\ 4s(b^2H)4p:$	u^3H°	6	56334. 01	-48. 68	1. 166			
		5	53281. 735					5	56382. 69			-40. 64	1. 029	
		4	53769. 020					4	56423. 33				0. 859	
		3	54161. 182											
		2	54375. 719											
$3d^7(a^2P)4p$	z^1P°	1	53229. 94		1. 266	$3d^7(a^2D)4p:$	u^3F°	4	56592. 76	-190. 57	1. 148			
				3	56783. 33			-75. 32	1. 077					
$3d^7(a^4F)4d$	e^5H	7	53275. 20:	-77. 82	1. 30:			3	56842. 70	0. 687				
		6	53353. 02:						4			56858. 65		
		5	53874. 30:											
		4	54237. 20											
		3	54491. 08											
	9°	4	53328. 87			$3d^6\ 4s(a^6D)5d$	1	5	56428. 06					
								$3d^6\ 4s(a^6D)5d$	2	4, 5	56452. 04			
$3d^6\ 4s(b^4D)4p$	t^5P°	3	53388. 68:	-723. 62	1. 70:	$3d^7(a^2D)4p:$	u^3F°			4	56592. 76	-190. 57	1. 148	
		2	54112. 30					3	56783. 33	-75. 32	1. 077			
		1	54271. 11					2	56858. 65		0. 687			
		3	53661. 13		1. 21:	$3d^6\ 4s(a^6D)5d$	3	4	56842. 70					
								v^1G°	4	56951. 27		1. 053		
$3d^7(a^2H)4p$	y^1H°	5	53722. 44		1. 03:	$3d^6\ 4s(b^2H)4p$	x^3I°			7	57027. 56:	-42. 69	1. 145	
				6	57070. 25			-34. 01	1. 028					
				5	57104. 26				0. 832					
$3d^7(a^4F)4d$	e^3G	5	53739. 488	-327. 08	1. 248			4	57550. 09	-90. 97	1. 235			
		4	54066. 57						3			57641. 06	-67. 70	0. 698
		3	54379. 44						2			57708. 76		
$3d^7(a^4F)4d$	f^3D	3	53747. 547	-319. 274	1. 258	$3d^6\ 4s(a^4D)4d$	i^5D	4	57697. 59	-116. 38	1. 384			
		2	54066. 821					3	57813. 97			-160. 19	1. 415	
		1	54449. 33					2	57974. 16					
								1						
								0						
$3d^6\ 4s(a^6D)6s$	g^7D	5	53800. 90	-323. 72	1. 586	$3d^6\ 4s(a^6D)7s$	h^7D	5	57897. 17					
		4	54124. 62					4						
		3	54413. 74:					3						
		2	54611. 72					2						
		1	54747. 74:					1						
$3d^7(a^4F)4d$	e^3H	6	53840. 68:	-426. 08	1. 225	$3d^6\ 4s(a^4D)4d$	g^5G	6	58001. 88	-269. 62	1. 40:			
		5	54266. 76:					5	58271. 50:			-248. 68		
		4	54555. 45:					4	58520. 18:			-189. 91		
								3	58710. 09:			-114. 72	0. 343	
		2	54611. 72			$3d^6\ 4s(a^4D)4d$	4	2	58213. 17					
		1	54747. 74:					r^3G°	5	59926. 62:	-245. 44	1. 190		
$3d^6\ 4s(b^4D)4d:$	$10^\circ(^5D^\circ?)$	3	53891. 54		1. 476	4	60172. 06			-192. 70			1. 030	
		$3d^6\ 4s(a^4G)4p:$	t^3G°	5	53983. 30		-254. 16			1. 234			3	60364. 76:
				4	54237. 46									
		3	54600. 35		0. 922	t^3H°	6	60365. 70:	-183. 48	1. 163				
								5			60549. 18	-208. 50	1. 040	
		3	54600. 35				4	60757. 68		0. 805				
	11°	3	54004. 82			q^3G°	5	60677. 23:	-77. 48					
$3d^6\ 4s(b^4D)4d:$	$12^\circ(^5F^\circ?)$	5	54013. 78		1. 356			4			60754. 71:	-52. 01		
		4	54301. 36					3			60806. 72			
	13°	4	54301. 36			-----	-----	-----	-----	-----	-----			
						Fe II ($a^6D_{15/2}$)	Limit		63700					

Fe I OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ +	Observed Terms							
3d ⁶ 4s ³	{ a ³ P	a ⁵ D b ³ D b ¹ D	b ³ F b ³ G b ¹ G	a ³ H a ¹ I				
3d ⁸			c ³ F					
	ns (n ≥ 4)		np (n ≥ 4)					
3d ⁶ 4s(a ⁶ D)nx	{ e, g, h ⁷ D e ⁵ D		z, x ⁷ P ^o z, u ⁵ P ^o	z ⁷ D ^o z, t ⁵ D ^o	z ⁷ F ^o z, u ⁵ F ^o			
3d ⁷ (a ⁴ F)nx	{ a, e ⁵ F a, e ³ F			y ⁵ D ^o y ³ D ^o	y ⁵ F ^o y ³ F ^o	z ⁵ G ^o z ³ G ^o		
3d ⁶ 4s(a ⁴ D)nx	{ g ⁵ D e ³ D		y ⁵ P ^o z ³ P ^o	x ⁵ D ^o z ³ D ^o	x ⁵ F ^o z ³ F ^o			
3d ⁷ (a ⁴ P)nx	{ a ⁵ P b ³ P		y ⁵ S ^o y ³ S ^o	v ⁵ P ^o y ³ P ^o	u ⁵ D ^o w ³ D ^o			
3d ⁷ (a ² G)nx		a ³ G a ¹ G			x ³ F ^o z ¹ F ^o	v ³ G ^o z ¹ G ^o	y ³ H ^o z ¹ H ^o	
3d ⁷ (a ² P)nx	{ c ³ P a ¹ P		x ³ S ^o	v ³ P ^o z ¹ P ^o	u ³ D ^o y ¹ D ^o			
3d ⁷ (a ² H)nx		b ³ H a ¹ H				u ³ G ^o y ¹ G ^o	w ³ H ^o y ¹ H ^o	y ³ I ^o z ¹ I ^o
3d ⁷ (a ² D)nx	{ a ³ D a ¹ D			w ³ P ^o	t ³ D ^o w ¹ D ^o	u ³ F ^o		
3d ⁶ 4s(b ⁴ P)nx			z ⁵ S ^o z ³ S ^o	x ⁵ P ^o x ³ P ^o	w ⁵ D ^o x ³ D ^o			
3d ⁶ 4s(a ⁴ H)nx						y ⁵ G ^o y ³ G ^o	z ⁵ H ^o z ³ H ^o	z ⁵ I ^o z ³ I ^o
3d ⁶ 4s(b ⁴ F)nx					v ⁵ D ^o v ³ D ^o	w ⁵ F ^o w ³ F ^o	x ⁵ G ^o x ³ G ^o	
3d ⁵ 4s ² (a ⁶ S)nx				y ⁷ P ^o w ⁵ P ^o				
3d ⁶ 4s(a ⁴ G)nx						v ⁵ F ^o v ³ F ^o	w ⁵ G ^o t ³ G ^o	x ⁵ H ^o x ³ H ^o
3d ⁶ 4s(b ² H)nx								u ³ H ^o x ³ I ^o
3d ⁶ 4s(b ⁴ D)nx				t ⁵ P ^o				
3d ⁷ (b ² F)nx		d ³ F a ¹ F						
	nd (n ≥ 4)							
3d ⁶ 4s(a ⁶ D)nx	{ e ⁷ S e ⁵ S	e ⁷ P e ⁵ P	f ⁷ D f ⁵ D	e ⁷ F f ⁵ F	e ⁷ G e ⁵ G			
3d ⁷ (a ⁴ F)nx		f ⁵ P e ³ P	h ⁵ D f ³ D	g ⁵ F f ³ F	f ⁵ G e ³ G	e ⁵ H e ³ H		
3d ⁶ 4s(a ⁴ D)nx			i ⁵ D	g ⁵ G				

*For predicted terms in the spectra of the Fe I isoelectronic sequence, see Vol. II, Introduction.

Fe II

(Mn I sequence; 25 electrons)

Z=26

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^6 4s^1 {}^6D_{4\frac{1}{2}}$ $a {}^6D_{4\frac{1}{2}}$ 130524 cm^{-1}

I. P. 16.18 volts

The early work on this spectrum by Meggers, Kiess, and Walters, and Russell was greatly extended by Dobbie, who in 1938 published a list of some 1,300 classified lines based on his observations covering the range between 2150 Å and 6627 Å. The listed terms are chiefly from this paper, supplemented by additional terms by Edlén and L. C. Green. A correction of 0.02 cm^{-1} has been subtracted from all of Dobbie's values (1938) in order to start with the ground state zero. In the course of his work on Fe III, Edlén has extended the analysis of Fe II and communicated 33 new terms in advance of publication. L. C. Green has observed the spectrum in the ultraviolet from 896 Å to 2495 Å, classified more than 450 lines, and published 57 levels. The arbitrary numbers he assigned to the miscellaneous levels are repeated here. For some of these levels two J -values are listed, denoting that the observed combinations do not indicate which of the two is correct. Edlén has been able to improve the values of some levels found by Green.

Two differences in interpretation have been handled as follows: Green's revision of Dobbie's incomplete $f {}^6D$ term as part of $e {}^6F$ has been adopted; Dobbie's $e {}^6P$ term, whose components are assigned numbers by Green, has been retained.

With four exceptions, the configurations are from Dobbie (1935 paper), Edlén, or Green. Russell in his investigation of the Fe I spectrum assigns $b {}^2F$ to $3d^7$ (instead of $3d^6(a {}^1F)4s$); $c {}^2F$ to $3d^6(a {}^1F)4s$; and $d {}^2D$ to $3d^7$. The writer has tentatively suggested the configuration of $e {}^4H$ as $3d^6(a {}^3H)5s$.

The limit from Dobbie's 1934 paper, 130978, is an average value based on the 4D and 6D terms from $3d^6(a {}^6D)ns$ ($n=4,5$). He used a Rydberg formula and corrected the results empirically to conform to a Ritz formula, as suggested by Russell. From a study of related spectra in which long series are known, Russell has suggested a further correction of -454 cm^{-1} , which has been adopted here.

The g -values have been determined by Miss Weeks from films of the Zeeman effect of Fe I automatically recorded from plates taken with the Bitter magnet at the Massachusetts Institute of Technology. Harrison generously supplied this observational material, and Miss Weeks derived the observed g -values especially for inclusion here.

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Fe II

Fe II

Config.	Desig.	J	Level	Interval	Obs. g	Config.	Desig.	J	Level	Interval	Obs. g		
$3d^6(a^5D)4s$	a^6D	$4\frac{1}{2}$	0.00	-384.77	1.58	$3d^6(a^1D)4s$	c^2D	$2\frac{1}{2}$	38164.24	-50.26	1.176		
		$3\frac{1}{2}$	384.77	-282.87	1.58			$1\frac{1}{2}$	38214.50		0.79		
		$2\frac{1}{2}$	667.64	-194.99	1.655			$3d^6(a^5D)4p$	z^6D^o	$4\frac{1}{2}$	38458.99	-201.05	1.542
		$1\frac{1}{2}$	862.63	-114.40	1.862					$3\frac{1}{2}$	38660.04	-198.92	1.584
		$0\frac{1}{2}$	977.03		3.31					$2\frac{1}{2}$	38858.96	-154.32	1.653
$3d^7$	a^4F	$4\frac{1}{2}$	1872.60	-557.48	1.33	$3d^6(a^5D)4p$	z^6F^o	$5\frac{1}{2}$	41968.11	-146.68			
		$3\frac{1}{2}$	2430.08	-407.86	1.223			$4\frac{1}{2}$	42114.79	-122.26	1.43		
		$2\frac{1}{2}$	2837.94	-279.54	1.02			$3\frac{1}{2}$	42237.05	-97.81	1.399		
		$1\frac{1}{2}$	3117.48		0.385			$2\frac{1}{2}$	42334.86	-66.47	1.304		
$3d^6(a^5D)4s$	a^4D	$3\frac{1}{2}$	7955.24	-436.68	1.419	$3d^6(a^5D)4p$	z^6P^o	$5\frac{1}{2}$	41968.11	-146.68			
		$2\frac{1}{2}$	8391.92	-288.55	1.365			$4\frac{1}{2}$	42114.79	-122.26	1.369		
		$1\frac{1}{2}$	8680.47	-166.29	1.200			$3\frac{1}{2}$	42237.05	-97.81	1.304		
		$0\frac{1}{2}$	8846.76		-0.05			$2\frac{1}{2}$	42334.86	-66.47	1.04		
$3d^7$	a^4P	$2\frac{1}{2}$	13474.43	-198.78	1.609	$3d^6(a^5D)4p$	z^6P^o	$3\frac{1}{2}$	42658.23	-580.38	1.702		
		$1\frac{1}{2}$	13673.21	-231.66	1.737			$2\frac{1}{2}$	43238.61	-382.37	2.398		
		$0\frac{1}{2}$	13904.87		2.67			$3d^6(a^5D)4p$	z^4F^o	$4\frac{1}{2}$	44232.51	-521.31	1.32
$3d^7$	a^2G	$4\frac{1}{2}$	15844.71	-524.68		$3\frac{1}{2}$	44753.82			-326.09	1.29		
		$3\frac{1}{2}$	16369.39			$2\frac{1}{2}$	45079.91			-209.93	1.069		
$3d^7$	a^2P	$1\frac{1}{2}$	18360.65	-526.10	1.28	$3d^6(a^5D)4p$	z^4D^o			$3\frac{1}{2}$	44446.89	-337.89	1.40
$0\frac{1}{2}$	18886.75			$2\frac{1}{2}$	44784.78			-259.43	1.35				
$3d^7$	a^2H	$5\frac{1}{2}$	20340.36	-465.47	0.92	$3d^6(a^5D)4p$	z^4D^o	$1\frac{1}{2}$	45044.21	-162.28	1.15		
		$4\frac{1}{2}$	20805.83					$0\frac{1}{2}$	45206.49		-0.021		
$3d^7$	a^2D	$2\frac{1}{2}$	20516.98	-791.10	1.22	$3d^6(a^1F)4s$	c^2F	$3\frac{1}{2}$	44915.07	-14.52			
$3d^6(a^3P)4s$	b^4P	$2\frac{1}{2}$	20830.52	-981.52	1.583	$3d^6(a^5D)4p$	z^4P^o	$2\frac{1}{2}$	44929.59				
		$1\frac{1}{2}$	21812.04	-597.78	1.720			$2\frac{1}{2}$	46967.47	-422.35	1.592		
		$0\frac{1}{2}$	22409.82		2.68			$1\frac{1}{2}$	47389.82	-236.32	1.717		
$3d^6(a^3H)4s$	a^4H	$6\frac{1}{2}$	21251.55	-178.84	1.20	$3d^7$	d^2D	$1\frac{1}{2}$	47674.78	364.45			
		$5\frac{1}{2}$	21430.39	-151.25	1.119			$2\frac{1}{2}$	48039.23				
		$4\frac{1}{2}$	21581.64	-130.25	0.951			$3d^6(b^3P)4s$	c^4P	$0\frac{1}{2}$	49101.09	405.90	
		$3\frac{1}{2}$	21711.89		0.661					$1\frac{1}{2}$	49506.99	705.94	
$3d^6(a^3F)4s$	b^4F	$4\frac{1}{2}$	22637.19	-173.14	1.307	$2\frac{1}{2}$	50212.93						
		$3\frac{1}{2}$	22810.33	-129.02	1.210	$3d^6(b^3F)4s$	c^4F	$1\frac{1}{2}$	50076.10	66.83			
		$2\frac{1}{2}$	22939.35	-91.95	1.019			$2\frac{1}{2}$	50142.93	45.02			
$1\frac{1}{2}$	23031.30		0.398	$3\frac{1}{2}$	50187.95			-30.34					
$0\frac{1}{2}$				$4\frac{1}{2}$	50157.61								
$3d^5 4s^2$	a^6S	$2\frac{1}{2}$	23317.60		1.996	$3d^6(b^3P)4s$	c^2P	$0\frac{1}{2}$	54063.53	838.89			
$3d^6(a^3G)4s$	a^4G	$5\frac{1}{2}$	25428.80	-376.52	1.237			$1\frac{1}{2}$	54902.42				
		$4\frac{1}{2}$	25805.32	-176.33	1.15			$3d^5 4s(a^7S)4p$	z^6P^o	$4\frac{1}{2}$			
		$3\frac{1}{2}$	25981.65	-73.75	0.98					$3\frac{1}{2}$			
$2\frac{1}{2}$	26055.40		0.574	$2\frac{1}{2}$	54490.2								
$3d^6(a^3P)4s$	b^2P	$1\frac{1}{2}$	25787.60	-1145.14	1.33	$3d^6(b^3F)4s$	d^2F	$2\frac{1}{2}$	54870.62	33.88			
		$0\frac{1}{2}$	26932.74		0.67			$3\frac{1}{2}$	54904.50				
$3d^6(a^3H)4s$	b^2H	$5\frac{1}{2}$	26170.19	-182.61	1.09	$3d^6(b^1G)4s$	d^2G	$4\frac{1}{2}$	58631.65	-34.71			
		$4\frac{1}{2}$	26352.80		0.927			$3\frac{1}{2}$	58666.36				
$3d^6(a^3F)4s$	a^2F	$3\frac{1}{2}$	27314.93	-305.46	1.129	$3d^6(a^3P)4p$	z^4S^o	$1\frac{1}{2}$	59663.45		1.89		
		$2\frac{1}{2}$	27620.39		0.851			$3d^5 4s^2$	c^4D	$3\frac{1}{2}$	60270.37	-174.91	
$3d^6(a^3G)4s$	b^2G	$4\frac{1}{2}$	30388.55	-375.91	1.10	$2\frac{1}{2}$	60445.28			4.23			
		$3\frac{1}{2}$	30764.46		0.898	$1\frac{1}{2}$	60441.05			56.59			
		$0\frac{1}{2}$	31368.45	-3.98		$0\frac{1}{2}$	60384.46						
$3d^6(a^3D)4s$	b^4D	$1\frac{1}{2}$	31364.47	23.51	1.327	$3d^6(a^3P)4p$	y^4P^o	$2\frac{1}{2}$	60402.38	-930.44	1.58		
		$2\frac{1}{2}$	31387.98	95.22	1.41			$1\frac{1}{2}$	61332.82	297.45	1.74		
		$3\frac{1}{2}$	31483.20					$0\frac{1}{2}$	61035.37		2.613		
		$3d^7$	b^2F	$2\frac{1}{2}$	31811.87			187.25	0.86	$3d^6(a^3H)4p$	z^4G^o	$5\frac{1}{2}$	60625.47
$3\frac{1}{2}$	31999.12				1.124	$4\frac{1}{2}$	60807.23	-149.54	1.155				
$3d^6(a^1I)4s$	a^2I	$6\frac{1}{2}$	32875.63	-34.24	1.062	$3d^6(a^3H)4p$	z^4H^o	$5\frac{1}{2}$	60837.66	-50.07	1.969		
		$5\frac{1}{2}$	32909.87		0.92			$4\frac{1}{2}$	60887.66	-101.82	0.799		
$3d^6(a^1G)4s$	c^2G	$4\frac{1}{2}$	33466.50	-34.82	1.099	$3d^6(a^3H)4p$	z^4H^o	$2\frac{1}{2}$	61041.78				
		$3\frac{1}{2}$	33501.32		0.88			$6\frac{1}{2}$	60837.59	-50.07			
$3d^6(a^3D)4s$	b^2D	$1\frac{1}{2}$	36126.41	126.55	0.799			$5\frac{1}{2}$	60887.66	-101.82			
		$2\frac{1}{2}$	36252.96		1.179			$4\frac{1}{2}$	60939.48	-167.42			
$3d^6(a^1S)4s$	a^2S	$0\frac{1}{2}$	37227.32		2.06	$3\frac{1}{2}$	61156.90		0.720				

Fe II—Continued

Fe II—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i> .	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>																																																																																																																																																																																																																																																																																																																																																																																																																																																																
$3d^6(a^3P)4p$	z^2D°	$2\frac{1}{2}$	61093. 44	-1032.22	1. 01	$3d^6(a^1I)4p$	z^2K°	$6\frac{1}{2}$	70986. 69	446. 06	1. 05																																																																																																																																																																																																																																																																																																																																																																																																																																																																
		$1\frac{1}{2}$	62125. 66		1. 019			$7\frac{1}{2}$	71432. 75			$3d^6(a^3H)4p$	z^4I°	$7\frac{1}{2}$	61347. 44	-180. 15	-59. 65	$3d^6(a^3D)4p$	w^4P°	$2\frac{1}{2}$	71964. 81	-78. 40	1. 66	$6\frac{1}{2}$	61527. 59	$1\frac{1}{2}$	72043. 21	$5\frac{1}{2}$	61587. 24	$0\frac{1}{2}$	72213. 10	$4\frac{1}{2}$	61512. 67			$3d^6(a^3P)4p$	y^4D°	$3\frac{1}{2}$	61726. 09	-963. 87	1. 411	$3d^6(a^3D)4p$	w^4F°	$4\frac{1}{2}$	72130. 44	131. 39	0. 91	$2\frac{1}{2}$	62639. 96	1. 349	$5\frac{1}{2}$	72261. 83	$1\frac{1}{2}$	62962. 26	1. 14	$1\frac{1}{2}$	72169. 15	$0\frac{1}{2}$	62829. 16		$2\frac{1}{2}$	72238. 63	$3d^6(a^7S)4p$	y^6P°	$1\frac{1}{2}$	61975. 11	74. 06	1. 68	$3d^6(a^3D)4p$	w^4D°	$0\frac{1}{2}$	72429. 74	94. 88	0. 91	$2\frac{1}{2}$	62049. 17	122. 59	$1\frac{1}{2}$	72524. 62	$3\frac{1}{2}$	62171. 76		$2\frac{1}{2}$	72619. 61	$3d^6(a^3F)4p$	y^4F°	$4\frac{1}{2}$	62158. 19	92. 62	1. 33	$3d^6(a^3D)4p?$	x^2F°	$3\frac{1}{2}$	73016. 33	-38. 64		$3\frac{1}{2}$	62065. 57	1. 198	$2\frac{1}{2}$	73054. 97	$2\frac{1}{2}$	62151. 61	-86. 04	1. 025			$1\frac{1}{2}$	62244. 57	-92. 96	0. 43			$3d^6(a^3H)4p$	z^2G°	$4\frac{1}{2}$	62083. 17	-239. 33	1. 097	$3d^6(a^1G)4p$	w^2G°	$4\frac{1}{2}$	73091. 70	-51. 78	0. 91	$3\frac{1}{2}$	62322. 50		$3\frac{1}{2}$	73143. 48	$3d^6(a^3H)4p$	z^2I°	$6\frac{1}{2}$	62293. 20	-369. 10	1. 069	$3d^6(a^3D)4p$	y^2P°	$0\frac{1}{2}$	73187. 46	1. 70		$5\frac{1}{2}$	62662. 30	0. 910	$1\frac{1}{2}$	73189. 16	$3d^6(a^3F)4p$	x^4D°	$3\frac{1}{2}$	62945. 12	-327. 91	1. 385	$3d^6(a^1I)4p$	y^2I°	$6\frac{1}{2}$	73966. 94	-2. 77		$2\frac{1}{2}$	63273. 03	1. 351	$5\frac{1}{2}$	73603. 53	$1\frac{1}{2}$	63465. 19	-192. 16	1. 21	$4\frac{1}{2}$	73751. 34	$0\frac{1}{2}$	63559. 56	-94. 37	0. 013			$3d^6(a^3F)4p$	y^4G°	$5\frac{1}{2}$	63876. 38	-72. 46	1. 24	$3d^6(a^3D)4p$	x^2D°	$1\frac{1}{2}$	74498. 15	108. 82		$4\frac{1}{2}$	63948. 84	1. 15	$2\frac{1}{2}$	74606. 97	$3\frac{1}{2}$	64040. 96	-92. 12	0. 975			$2\frac{1}{2}$	64087. 50	-46. 54	0. 617			$3d^6(a^3F)4p$	z^2F°	$3\frac{1}{2}$	64286. 44	-139. 02	1. 135	$3d^6(a^1S)4p$	x^2P°	$3\frac{1}{2}$	75600. 93	-314. 28	1. 125	$2\frac{1}{2}$	64425. 46	0. 82	$2\frac{1}{2}$	75915. 21	$3d^6(a^3P)4p$	z^2P°	$0\frac{1}{2}$	64806. 52	27. 60	1. 329	$3d^6(a^1D)4p$	v^2F°	$1\frac{1}{2}$	76129. 58	-447. 92	1. 34	$1\frac{1}{2}$	64834. 12		$0\frac{1}{2}$	76577. 50	$3d^6(a^3F)4p$	y^2G°	$4\frac{1}{2}$	64832. 00	-277. 71	1. 101	$3d^6(a^1D)4p$	w^2D°	$2\frac{1}{2}$	77742. 78	394. 75	1. 13	$3\frac{1}{2}$	65109. 71	0. 896	$3\frac{1}{2}$	78137. 53	$3d^6(a^3H)4p?$	z^2H°	$5\frac{1}{2}$	65363. 66	-192. 68	1. 066	$3d^6(a^5D)5s$	e^6D	$4\frac{1}{2}$	77861. 47	-376. 07		$4\frac{1}{2}$	65556. 34	0. 913	$3\frac{1}{2}$	78237. 54	$3d^6(a^3G)4p$	x^4G°	$5\frac{1}{2}$	65580. 09	-116. 02	1. 00	$3d^6(a^1D)4p$	w^2D°	$2\frac{1}{2}$	78525. 27	203. 73		$4\frac{1}{2}$	65696. 11	-235. 35	$1\frac{1}{2}$	78725. 61	$3\frac{1}{2}$	65931. 46	-146. 88	$0\frac{1}{2}$	78843. 72	$2\frac{1}{2}$	66078. 34	0. 62			$3d^6(a^3G)4p$	x^4F°	$4\frac{1}{2}$	66012. 83	-364. 54	1. 21	$3d^6(a^1D)4p$	w^2P°	$1\frac{1}{2}$	78487. 25	401. 76		$3\frac{1}{2}$	66377. 37	-144. 95	$2\frac{1}{2}$	78690. 98	$2\frac{1}{2}$	66522. 32	-90. 42	$0\frac{1}{2}$	78841. 97	$1\frac{1}{2}$	66612. 74	1. 02	$1\frac{1}{2}$	79243. 73	$3d^6(a^3P)4p$	z^2S°	$0\frac{1}{2}$	66248. 67			$3d^5 4s(a^5S)4p$	x^6P°	$1\frac{1}{2}$	79246. 13	38. 97					$2\frac{1}{2}$	79285. 10	$3d^6(a^3G)4p$	y^4H°	$6\frac{1}{2}$	66411. 70	-51. 89	1. 13	$3d^6(a^5D)5s$	e^4D	$3\frac{1}{2}$	79439. 30	-446. 02		$5\frac{1}{2}$	66463. 59	-125. 58	$2\frac{1}{2}$	79885. 32	$4\frac{1}{2}$	66589. 17	-83. 22	$1\frac{1}{2}$	80177. 87	$3\frac{1}{2}$	66672. 39	0. 69	$0\frac{1}{2}$	80345. 91	$3d^6(a^3F)4p$	y^2D°	$2\frac{1}{2}$	67000. 63	-273. 23	1. 16	$3d^6(a^5D)4d$	e^6F	$5\frac{1}{2}$	82853. 5	-125. 4		$1\frac{1}{2}$	67273. 86	0. 719	$4\frac{1}{2}$	82978. 9	$3d^6(a^3G)4p?$	y^2H°	$5\frac{1}{2}$	67516. 37	-484. 45	1. 07	$3d^6(a^5D)4d?$	v^2G°	$3\frac{1}{2}$	83136. 5	-171. 9		$4\frac{1}{2}$	68000. 82	0. 907	$2\frac{1}{2}$	83308. 4	$3d^5 4s(a^5S)4p$	x^4P°	$2\frac{1}{2}$	69102. 69	-199. 66	1. 11	$3d^6(a^1F)4p$	e^6P	$3\frac{1}{2}$	83459. 7	-98. 8		$1\frac{1}{2}$	69302. 35	-124. 92	$0\frac{1}{2}$	83558. 5	$0\frac{1}{2}$	69427. 27				$3d^6(a^3G)4p?$	y^2F°	$3\frac{1}{2}$	69606. 64	-43. 90	1. 13	$3d^6(a^5D)4d?$	$30(^6D)$	$3\frac{1}{2}$	83305. 37	565. 94		$2\frac{1}{2}$	69650. 54	0. 857	$4\frac{1}{2}$	83871. 31	$3d^6(a^3G)4p$	x^2G°	$4\frac{1}{2}$	70314. 74	-208. 99	1. 11	$3d^6(a^5D)4d$	e^6P	$3\frac{1}{2}$	83713. 3	-540. 24		$3\frac{1}{2}$	70523. 73	0. 87
$3d^6(a^3H)4p$	z^4I°	$7\frac{1}{2}$	61347. 44	-180. 15	-59. 65	$3d^6(a^3D)4p$	w^4P°	$2\frac{1}{2}$	71964. 81	-78. 40	1. 66																																																																																																																																																																																																																																																																																																																																																																																																																																																																
		$6\frac{1}{2}$	61527. 59					$1\frac{1}{2}$	72043. 21																																																																																																																																																																																																																																																																																																																																																																																																																																																																		
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		$4\frac{1}{2}$	61512. 67																																																																																																																																																																																																																																																																																																																																																																																																																																																																								
$3d^6(a^3P)4p$	y^4D°	$3\frac{1}{2}$	61726. 09	-963. 87	1. 411	$3d^6(a^3D)4p$	w^4F°	$4\frac{1}{2}$	72130. 44	131. 39	0. 91																																																																																																																																																																																																																																																																																																																																																																																																																																																																
		$2\frac{1}{2}$	62639. 96		1. 349			$5\frac{1}{2}$	72261. 83																																																																																																																																																																																																																																																																																																																																																																																																																																																																		
		$1\frac{1}{2}$	62962. 26		1. 14			$1\frac{1}{2}$	72169. 15																																																																																																																																																																																																																																																																																																																																																																																																																																																																		
		$0\frac{1}{2}$	62829. 16					$2\frac{1}{2}$	72238. 63																																																																																																																																																																																																																																																																																																																																																																																																																																																																		
$3d^6(a^7S)4p$	y^6P°	$1\frac{1}{2}$	61975. 11	74. 06	1. 68	$3d^6(a^3D)4p$	w^4D°	$0\frac{1}{2}$	72429. 74	94. 88	0. 91																																																																																																																																																																																																																																																																																																																																																																																																																																																																
		$2\frac{1}{2}$	62049. 17					122. 59	$1\frac{1}{2}$			72524. 62																																																																																																																																																																																																																																																																																																																																																																																																																																																															
		$3\frac{1}{2}$	62171. 76						$2\frac{1}{2}$			72619. 61																																																																																																																																																																																																																																																																																																																																																																																																																																																															
$3d^6(a^3F)4p$	y^4F°	$4\frac{1}{2}$	62158. 19	92. 62	1. 33	$3d^6(a^3D)4p?$	x^2F°	$3\frac{1}{2}$	73016. 33	-38. 64																																																																																																																																																																																																																																																																																																																																																																																																																																																																	
		$3\frac{1}{2}$	62065. 57		1. 198			$2\frac{1}{2}$	73054. 97																																																																																																																																																																																																																																																																																																																																																																																																																																																																		
		$2\frac{1}{2}$	62151. 61		-86. 04			1. 025																																																																																																																																																																																																																																																																																																																																																																																																																																																																			
		$1\frac{1}{2}$	62244. 57		-92. 96			0. 43																																																																																																																																																																																																																																																																																																																																																																																																																																																																			
$3d^6(a^3H)4p$	z^2G°	$4\frac{1}{2}$	62083. 17	-239. 33	1. 097	$3d^6(a^1G)4p$	w^2G°	$4\frac{1}{2}$	73091. 70	-51. 78	0. 91																																																																																																																																																																																																																																																																																																																																																																																																																																																																
		$3\frac{1}{2}$	62322. 50					$3\frac{1}{2}$	73143. 48																																																																																																																																																																																																																																																																																																																																																																																																																																																																		
$3d^6(a^3H)4p$	z^2I°	$6\frac{1}{2}$	62293. 20	-369. 10	1. 069	$3d^6(a^3D)4p$	y^2P°	$0\frac{1}{2}$	73187. 46	1. 70																																																																																																																																																																																																																																																																																																																																																																																																																																																																	
		$5\frac{1}{2}$	62662. 30		0. 910			$1\frac{1}{2}$	73189. 16																																																																																																																																																																																																																																																																																																																																																																																																																																																																		
$3d^6(a^3F)4p$	x^4D°	$3\frac{1}{2}$	62945. 12	-327. 91	1. 385	$3d^6(a^1I)4p$	y^2I°	$6\frac{1}{2}$	73966. 94	-2. 77																																																																																																																																																																																																																																																																																																																																																																																																																																																																	
		$2\frac{1}{2}$	63273. 03		1. 351			$5\frac{1}{2}$	73603. 53																																																																																																																																																																																																																																																																																																																																																																																																																																																																		
		$1\frac{1}{2}$	63465. 19		-192. 16			1. 21	$4\frac{1}{2}$			73751. 34																																																																																																																																																																																																																																																																																																																																																																																																																																																															
		$0\frac{1}{2}$	63559. 56		-94. 37			0. 013																																																																																																																																																																																																																																																																																																																																																																																																																																																																			
$3d^6(a^3F)4p$	y^4G°	$5\frac{1}{2}$	63876. 38	-72. 46	1. 24	$3d^6(a^3D)4p$	x^2D°	$1\frac{1}{2}$	74498. 15	108. 82																																																																																																																																																																																																																																																																																																																																																																																																																																																																	
		$4\frac{1}{2}$	63948. 84		1. 15			$2\frac{1}{2}$	74606. 97																																																																																																																																																																																																																																																																																																																																																																																																																																																																		
		$3\frac{1}{2}$	64040. 96		-92. 12			0. 975																																																																																																																																																																																																																																																																																																																																																																																																																																																																			
		$2\frac{1}{2}$	64087. 50		-46. 54			0. 617																																																																																																																																																																																																																																																																																																																																																																																																																																																																			
$3d^6(a^3F)4p$	z^2F°	$3\frac{1}{2}$	64286. 44	-139. 02	1. 135	$3d^6(a^1S)4p$	x^2P°	$3\frac{1}{2}$	75600. 93	-314. 28	1. 125																																																																																																																																																																																																																																																																																																																																																																																																																																																																
		$2\frac{1}{2}$	64425. 46		0. 82			$2\frac{1}{2}$	75915. 21																																																																																																																																																																																																																																																																																																																																																																																																																																																																		
$3d^6(a^3P)4p$	z^2P°	$0\frac{1}{2}$	64806. 52	27. 60	1. 329	$3d^6(a^1D)4p$	v^2F°	$1\frac{1}{2}$	76129. 58	-447. 92	1. 34																																																																																																																																																																																																																																																																																																																																																																																																																																																																
		$1\frac{1}{2}$	64834. 12					$0\frac{1}{2}$	76577. 50																																																																																																																																																																																																																																																																																																																																																																																																																																																																		
$3d^6(a^3F)4p$	y^2G°	$4\frac{1}{2}$	64832. 00	-277. 71	1. 101	$3d^6(a^1D)4p$	w^2D°	$2\frac{1}{2}$	77742. 78	394. 75	1. 13																																																																																																																																																																																																																																																																																																																																																																																																																																																																
		$3\frac{1}{2}$	65109. 71		0. 896			$3\frac{1}{2}$	78137. 53																																																																																																																																																																																																																																																																																																																																																																																																																																																																		
$3d^6(a^3H)4p?$	z^2H°	$5\frac{1}{2}$	65363. 66	-192. 68	1. 066	$3d^6(a^5D)5s$	e^6D	$4\frac{1}{2}$	77861. 47	-376. 07																																																																																																																																																																																																																																																																																																																																																																																																																																																																	
		$4\frac{1}{2}$	65556. 34		0. 913			$3\frac{1}{2}$	78237. 54																																																																																																																																																																																																																																																																																																																																																																																																																																																																		
$3d^6(a^3G)4p$	x^4G°	$5\frac{1}{2}$	65580. 09	-116. 02	1. 00	$3d^6(a^1D)4p$	w^2D°	$2\frac{1}{2}$	78525. 27	203. 73																																																																																																																																																																																																																																																																																																																																																																																																																																																																	
		$4\frac{1}{2}$	65696. 11					-235. 35	$1\frac{1}{2}$			78725. 61																																																																																																																																																																																																																																																																																																																																																																																																																																																															
		$3\frac{1}{2}$	65931. 46					-146. 88	$0\frac{1}{2}$			78843. 72																																																																																																																																																																																																																																																																																																																																																																																																																																																															
		$2\frac{1}{2}$	66078. 34					0. 62																																																																																																																																																																																																																																																																																																																																																																																																																																																																			
$3d^6(a^3G)4p$	x^4F°	$4\frac{1}{2}$	66012. 83	-364. 54	1. 21	$3d^6(a^1D)4p$	w^2P°	$1\frac{1}{2}$	78487. 25	401. 76																																																																																																																																																																																																																																																																																																																																																																																																																																																																	
		$3\frac{1}{2}$	66377. 37					-144. 95	$2\frac{1}{2}$			78690. 98																																																																																																																																																																																																																																																																																																																																																																																																																																																															
		$2\frac{1}{2}$	66522. 32					-90. 42	$0\frac{1}{2}$			78841. 97																																																																																																																																																																																																																																																																																																																																																																																																																																																															
		$1\frac{1}{2}$	66612. 74					1. 02	$1\frac{1}{2}$			79243. 73																																																																																																																																																																																																																																																																																																																																																																																																																																																															
$3d^6(a^3P)4p$	z^2S°	$0\frac{1}{2}$	66248. 67			$3d^5 4s(a^5S)4p$	x^6P°	$1\frac{1}{2}$	79246. 13	38. 97																																																																																																																																																																																																																																																																																																																																																																																																																																																																	
					$2\frac{1}{2}$			79285. 10																																																																																																																																																																																																																																																																																																																																																																																																																																																																			
$3d^6(a^3G)4p$	y^4H°	$6\frac{1}{2}$	66411. 70	-51. 89	1. 13	$3d^6(a^5D)5s$	e^4D	$3\frac{1}{2}$	79439. 30	-446. 02																																																																																																																																																																																																																																																																																																																																																																																																																																																																	
		$5\frac{1}{2}$	66463. 59					-125. 58	$2\frac{1}{2}$			79885. 32																																																																																																																																																																																																																																																																																																																																																																																																																																																															
		$4\frac{1}{2}$	66589. 17					-83. 22	$1\frac{1}{2}$			80177. 87																																																																																																																																																																																																																																																																																																																																																																																																																																																															
		$3\frac{1}{2}$	66672. 39					0. 69	$0\frac{1}{2}$			80345. 91																																																																																																																																																																																																																																																																																																																																																																																																																																																															
$3d^6(a^3F)4p$	y^2D°	$2\frac{1}{2}$	67000. 63	-273. 23	1. 16	$3d^6(a^5D)4d$	e^6F	$5\frac{1}{2}$	82853. 5	-125. 4																																																																																																																																																																																																																																																																																																																																																																																																																																																																	
		$1\frac{1}{2}$	67273. 86		0. 719			$4\frac{1}{2}$	82978. 9																																																																																																																																																																																																																																																																																																																																																																																																																																																																		
$3d^6(a^3G)4p?$	y^2H°	$5\frac{1}{2}$	67516. 37	-484. 45	1. 07	$3d^6(a^5D)4d?$	v^2G°	$3\frac{1}{2}$	83136. 5	-171. 9																																																																																																																																																																																																																																																																																																																																																																																																																																																																	
		$4\frac{1}{2}$	68000. 82		0. 907			$2\frac{1}{2}$	83308. 4																																																																																																																																																																																																																																																																																																																																																																																																																																																																		
$3d^5 4s(a^5S)4p$	x^4P°	$2\frac{1}{2}$	69102. 69	-199. 66	1. 11	$3d^6(a^1F)4p$	e^6P	$3\frac{1}{2}$	83459. 7	-98. 8																																																																																																																																																																																																																																																																																																																																																																																																																																																																	
		$1\frac{1}{2}$	69302. 35					-124. 92	$0\frac{1}{2}$			83558. 5																																																																																																																																																																																																																																																																																																																																																																																																																																																															
		$0\frac{1}{2}$	69427. 27																																																																																																																																																																																																																																																																																																																																																																																																																																																																								
$3d^6(a^3G)4p?$	y^2F°	$3\frac{1}{2}$	69606. 64	-43. 90	1. 13	$3d^6(a^5D)4d?$	$30(^6D)$	$3\frac{1}{2}$	83305. 37	565. 94																																																																																																																																																																																																																																																																																																																																																																																																																																																																	
		$2\frac{1}{2}$	69650. 54		0. 857			$4\frac{1}{2}$	83871. 31																																																																																																																																																																																																																																																																																																																																																																																																																																																																		
$3d^6(a^3G)4p$	x^2G°	$4\frac{1}{2}$	70314. 74	-208. 99	1. 11	$3d^6(a^5D)4d$	e^6P	$3\frac{1}{2}$	83713. 3	-540. 24																																																																																																																																																																																																																																																																																																																																																																																																																																																																	
		$3\frac{1}{2}$	70523. 73		0. 87			$2\frac{1}{2}$	84266. 48																																																																																																																																																																																																																																																																																																																																																																																																																																																																		
								$1\frac{1}{2}$	84326. 91																																																																																																																																																																																																																																																																																																																																																																																																																																																																		

Fe II—Continued

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Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g.</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g.</i>
$3d^6(a^5D)4d?$	$32(^6D)$	$2\frac{1}{2}$	83812. 1			$3d^6(b^3F)4p$	u^2D°	$1\frac{1}{2}$ $2\frac{1}{2}$	92216. 42 92695. 62	479. 20	
$3d^6(a^1F)4p$	v^2D°	$2\frac{1}{2}$ $1\frac{1}{2}$	83868. 65 84359. 90	-491. 25		$3d^6(b^3F)4p$	u^2G°	$4\frac{1}{2}$ $3\frac{1}{2}$	92427. 22 92602. 94	-175. 72	
$3d^6(a^5D)4d?$	$33(^6D)$	$1\frac{1}{2}$	83989. 7			$3d^6(b^3F)4p$	u^4F°	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	93328. 6 93395. 6 93487. 71 93484. 75	67. 0 92. 1 -2. 96	
$3d^6(a^5D)4d$	e^6G	$6\frac{1}{2}$ $5\frac{1}{2}$ $4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$	84035. 09 84296. 70 84527. 89 84710. 66 84844. 81 84938. 13	-261. 61 -231. 19 -182. 77 -134. 15 -93. 32	1. 33	$3d^5 4s(a^5G)4p$	w^4G°	$2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$ $5\frac{1}{2}$	93988. 30 94073. 46 94148. 71 94190. 02	85. 16 75. 25 41. 31	
$3d^6(a^5D)4d?$	$34(^6D)$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 0\frac{1}{2} \end{array} \right\}$	84131. 2				14°	$2\frac{1}{2}$	94210. 1		
$3d^6(a^5D)4d$	f^4D	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	84685. 16 84870. 84 85048. 56 85172. 77	-185. 68 -177. 72 -124. 21			15°	$3\frac{1}{2}$	94762. 3		
$3d^6(a^5D)4d$	e^4G	$5\frac{1}{2}$ $4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$	84863. 14 85184. 69 85462. 78 85679. 64	-321. 55 -278. 09 -216. 86	1. 27	$3d^5 4s(a^3G)4p$	v^2H°	$5\frac{1}{2}$ $4\frac{1}{2}$	96062. 18 96239. 1	-176. 9	
$3d^6(a^5D)4d$	e^4F	$4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$	86124. 14 86416. 22 86599. 66 86710. 84	-292. 08 -183. 44 -111. 18	1. 29	$3d^6(b^3F)4p$	t^2F°	$2\frac{1}{2}$ $3\frac{1}{2}$	96279. 65 96357. 07	77. 42	
$3d^6(b^3P)4p$	v^4D°	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	86388. 98 86544. 14 86767. 87 86929. 92	155. 16 223. 73 162. 05		$3d^6(a^3H)5s$	e^4H	$6\frac{1}{2}$ $5\frac{1}{2}$ $4\frac{1}{2}$ $3\frac{1}{2}$	98129. 98 98294. 60 98445. 24 98568. 75	-164. 62 -150. 64 -123. 51	
$3d^6(a^1F)4p$	u^2F°	$3\frac{1}{2}$ $2\frac{1}{2}$	86482. 75 86547. 55	-64. 80		$3d^6(a^3H)5s$	e^2H	$5\frac{1}{2}$ $4\frac{1}{2}$	99093. 29 99331. 95	-238. 66	
$3d^6(a^5D)5p$	y^6F°	$5\frac{1}{2}$ $4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	87340. 4 87470. 8 87536. 9 87571. 3 87601. 9 87635. 2	-130. 4 -66. 1 -34. 9 -30. 1 -33. 3		$3d^6(a^3F)5s$	f^4F	$4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$	99573. 11 99688. 20 99824. 04 99918. 30	-115. 09 -135. 84 -94. 26	
	$1^\circ(^6P^\circ)$	$3\frac{1}{2}$	88208. 6			$3d^6(a^3F)5s$	e^2F	$3\frac{1}{2}$ $2\frac{1}{2}$	100492. 04 100749. 75	-257. 71	
	$2^\circ(^6P^\circ)$	$3\frac{1}{2}$	89127. 7			$3d^6(a^3G)5s$	f^4G	$5\frac{1}{2}$ $4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$	102584. 81		
	$3^\circ(^6P^\circ)$	$2\frac{1}{2}$	89443. 7				e^2G	$4\frac{1}{2}$ $3\frac{1}{2}$	103608. 82 103983. 35	-374. 53	
	$4^\circ(^6P^\circ)$	$1\frac{1}{2}$	89625. 0				16°	$3\frac{1}{2}$	106863. 2		
	5°	$3\frac{1}{2}$	90300. 0				17°	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	107165. 6		
	6°	$\left\{ \begin{array}{l} 4\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	90335. 5				18°	$2\frac{1}{2}$	107196. 2		
	$8^\circ(^4P^\circ)$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 0\frac{1}{2} \end{array} \right\}$	90828. 4				$20^\circ(^6D^\circ)$	$3\frac{1}{2}$	107886. 6		
	$9^\circ(^4P^\circ)$	$1\frac{1}{2}$	90898. 2				$21^\circ(^6D^\circ)$	$\left\{ \begin{array}{l} 4\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	107964. 7		
	$10^\circ(^4P^\circ)$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 1\frac{1}{2} \end{array} \right\}$	90981. 5				$22^\circ(^6D^\circ)$	$2\frac{1}{2}$	108130. 6		
	11°	$2\frac{1}{2}$	91067. 1				$23^\circ(^6D^\circ)$	$1\frac{1}{2}$	108191. 6		
	w^6P°	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$	91167. 3 91574. 8 91843. 1	-407. 5 -268. 3			24°	$3\frac{1}{2}$	108239. 2		
$3d^5 4s(a^5G)4p$	x^4H°	$3\frac{1}{2}$ $4\frac{1}{2}$ $5\frac{1}{2}$ $6\frac{1}{2}$	92039. 41 92116. 98 92166. 81 92250. 27	27. 57 49. 83 83. 46		$3d^6(a^1I)5s$	e^2I	$5\frac{1}{2}$ $6\frac{1}{2}$	108630. 24 108648. 64	18. 40	
$3d^5 4s(a^5G)4p$	v^4F°	$4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$	92171. 95 92232. 76 92330. 1 92358. 8	-110. 81 -47. 3 -28. 7			27°	$1\frac{1}{2}$	108780. 0		
							28°	$1\frac{1}{2}$	109780. 0		
							29°	$2\frac{1}{2}$	111929. 0		
						Fe III(a^5D_4)	Limit		130524		

Observed Terms

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	a^6S	$ns (n \geq 4)$	$np (n \geq 4)$	$nd (n \geq 4)$
$3d^5 4s^2$	$\{ a^4D$			
$3d^7$	$\{ \{ a^4P$ a^2P	a^4F b^2F	a^2G	a^2H
$3d^6(a^5D)nx$	$\{ \{ a, e^6D$ a, e^4D		z^6P^o z^4P^o z^4D^o z^4F^o	e^6P f^4D e^6F e^4F e^6G e^4G
$3d^6(a^3P)nx$	$\{ \{ b^4P$ b^2P		z^4S^o z^2S^o y^4P^o z^2D^o	
$3d^6(a^3H)nx$	$\{ \{ a, e^4H$ b, e^2H		z^4G^o z^2G^o z^4H^o z^2H^o z^4I^o z^2I^o	
$3d^6(a^3F)nx$	$\{ \{ b, f^4F$ a, e^2F		x^4D^o y^2D^o y^4F^o z^2F^o y^2G^o	
$3d^6(a^3G)nx$	$\{ \{ a, f^4G$ b, e^2G		x^4F^o y^2F^o x^4G^o x^2G^o y^4H^o y^2H^o	
$3d^5 4s(a^7S)nx$	$\{ \{ a, e^2I$		z^8P^o y^6P^o	
$3d^6(a^1D)nx$	$\{ \{ b^4D$ b^2D		w^4P^o y^2P^o w^4D^o x^2D^o x^2F^o	y^2I^o z^2K^o
$3d^6(a^3D)nx$	$\{ \{ a^2S$		x^2P^o w^2P^o w^2D^o v^2F^o	
$3d^6(a^1G)nx$	$\{ \{ c^2G$		w^2G^o	
$3d^6(a^1S)nx$	$\{ \{ a^2S$		x^2P^o w^2P^o w^2D^o v^2F^o	
$3d^6(a^1D)nx$	$\{ \{ c^2D$		x^6P^o x^4P^o	
$3d^5 4s(a^5S)nx$	$\{ \{ c^3F$		v^2D^o v^4D^o u^2F^o v^2G^o	
$3d^6(a^1F)nx$	$\{ \{ c^4P$ c^2P		u^2D^o u^4D^o l^2F^o u^2G^o	
$3d^6(b^3P)nx$	$\{ \{ c^4F$ d^2F		u^4F^o l^2F^o u^2G^o	
$3d^6(b^3F)nx$	$\{ \{ d^2G$		v^4F^o w^4G^o x^4H^o v^2H^o	
$3d^5 4s(a^5G)nx$				
$3d^5 4s(b^3G)nx$				

*For predicted terms in the spectra of the Mn I isoelectronic sequence, see Vol. II, Introduction.

Fe III

(Cr I sequence; 24 electrons)

Z=26

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^6 {}^5D_4$ $a {}^5D_4$ 247200 cm^{-1}

I. P. 30.64 volts

The early work on this spectrum was by Bowen and L. C. Green, who found a few multiplets. Later Edlén and Swings published a detailed analysis which includes a remarkably complete array of terms from the $3d^6$, $3d^5 4s$, and $3d^5 4p$ configurations. Their observations extend from about 500 Å to 6500 Å and their final tables contain some 320 energy levels and approximately 1500 classified lines. A few revisions and one new level have been furnished by Glad and entered in page proof.

In the present work the prefixes a, b, \dots, z, y, \dots have been adopted throughout the spectrum, but the notation by Edlén and Swings is given in the first column for convenience in cross reference. A correction of 0.8 cm^{-1} has, also, here been added to their published values of all the levels in order to place the ground state at zero.

The terms of all four multiplicities recognized in this spectrum (1, 3, 5, 7), are connected by observed intersystem combinations.

REFERENCES

- B. Edlén and P. Swings, *Astroph. J.* **95**, 532 (1942). (I P) (T) (C L)
 S. Glad, unpublished material (Sept. 1951). (T)

Fe III

Fe III

Authors	Config.	Desig.	J	Level	Interval	Authors	Config.	Desig.	J	Level	Interval							
$3d {}^5D_4$ 5D_3 5D_2 5D_1 5D_0	$3d^6$	$a {}^5D$	4	0.0	-436.2	$4s {}^7S_3$ $3d {}^1I_6$ $3d {}^3D_1$ 3D_2 3D_3 $3d {}^1G_4$ $3d {}^1S_0$ $3d {}^1D_2$ $4s {}^6S_2$ $3d {}^1F_3$ $3d' {}^3P_0$ 3P_1 3P_2	$3d^5(a {}^6S)4s$ $3d^6$ $3d^6$ $3d^6$ $3d^6$ $3d^6$ $3d^6$ $3d^6$ $3d^6(a {}^6S)4s$ $3d^6$ $3d^6$	$a {}^7S$ $a {}^1I$ $a {}^3D$ $a {}^1G$ $a {}^1S$ $a {}^1D$ $a {}^6S$ $a {}^1F$ $b {}^3P$	3	30088.84	6	30356.2	1	30725.8	-9.6			
			3	436.2	-302.7											2	30716.2	141.6
			2	738.9	-193.5											3	30857.8	
			1	932.4	-94.9											4	30886.4	
			0	1027.3												0	34812.4	
			2	35803.7														
$3d {}^3P_2$ 3P_1 3P_0	$3d^6$	$a {}^3P$	2	19404.8	-1283.6	$3d {}^1G_4$ $3d {}^1S_0$ $3d {}^1D_2$ $4s {}^6S_2$ $3d {}^1F_3$ $3d' {}^3P_0$ 3P_1 3P_2	$3d^5(a {}^6S)4s$ $3d^6$ $3d^6$ $3d^6$ $3d^6$ $3d^6$ $3d^6$ $3d^6$	$a {}^1G$ $a {}^1S$ $a {}^1D$ $a {}^6S$ $a {}^1F$ $b {}^3P$	4	30886.4	0	34812.4	2	40999.87	3	42896.9		
			1	20688.4	-520.1												2	49148
			0	21208.5													1	49576.9
$3d {}^3H_6$ 3H_5 3H_4	$3d^6$	$a {}^3H$	6	20051.1	-249.7	$3d {}^1G_4$ $3d {}^1S_0$ $3d {}^1D_2$ $4s {}^6S_2$ $3d {}^1F_3$ $3d' {}^3P_0$ 3P_1 3P_2	$3d^5(a {}^6S)4s$ $3d^6$ $3d^6$ $3d^6$ $3d^6$ $3d^6$ $3d^6$ $3d^6$	$a {}^1G$ $a {}^1S$ $a {}^1D$ $a {}^6S$ $a {}^1F$ $b {}^3P$	4	30886.4	0	34812.4	2	40999.87	3	42896.9		
			5	20300.8	-181.1												2	49148
			4	20481.9													1	49576.9
$3d {}^3F_4$ 3F_3 3F_2	$3d^6$	$a {}^3F$	4	21462.2	-237.7	$3d {}^1G_4$ $3d {}^1S_0$ $3d {}^1D_2$ $4s {}^6S_2$ $3d {}^1F_3$ $3d' {}^3P_0$ 3P_1 3P_2	$3d^5(a {}^6S)4s$ $3d^6$ $3d^6$ $3d^6$ $3d^6$ $3d^6$ $3d^6$ $3d^6$	$a {}^1F$ $b {}^3P$	3	42896.9	0	49148	1	49576.9	2	50412.3		
			3	21699.9	-157.3												2	49148
			2	21857.2													1	49576.9
$3d {}^3G_5$ 3G_4 3G_3	$3d^6$	$a {}^3G$	5	24558.8	-382.1	$3d {}^1G_4$ $3d {}^1S_0$ $3d {}^1D_2$ $4s {}^6S_2$ $3d {}^1F_3$ $3d' {}^3P_0$ 3P_1 3P_2	$3d^5(a {}^6S)4s$ $3d^6$ $3d^6$ $3d^6$ $3d^6$ $3d^6$ $3d^6$ $3d^6$	$b {}^3P$	0	49148	1	49576.9	2	50412.3	429			
			4	24940.9	-201.5											1	49148	
			3	25142.4												1	49576.9	

Fe III—Continued

Fe III—Continued

Authors	Config.	Desig.	<i>J</i>	Level	Interval	Authors	Config.	Desig.	<i>J</i>	Level	Interval								
$3d' \ ^3F_2$ $\ ^3F_3$ $\ ^3F_4$	$3d^6$	$b \ ^3F$	2	50184. 9	110. 3	$b \ ^3G_3$ $\ ^3G_4$ $\ ^3G_5$	$3d^5(a \ ^2G)4s$	$c \ ^3G$	3	89697. 52	86. 07								
			3	50295. 2	-19. 1				4	89783. 59		124. 26							
			4	50276. 1					5	89907. 85									
$3d' \ ^1G_4$	$3d^6$	$b \ ^1G$	4	57221. 7		$b \ ^3F_2$ $\ ^3F_3$ $\ ^3F_4$	$3d^5(a \ ^4F)4s$	$d \ ^3F$	2	90423. 68	60. 26								
$a \ ^5G_6$ $\ ^5G_5$ $\ ^5G_4$ $\ ^5G_3$ $\ ^5G_2$	$3d^5(a \ ^4G)4s$	$a \ ^5G$	6	63425. 17	-41. 22	$a \ ^4H_5$	$3d^5(a \ ^2H)4s$	$a \ ^4H$	5	92523. 91		-3. 70							
			5	63466. 39	-20. 39				$e \ ^3F$	4			93388. 75	-20. 48					
			4	63486. 78	-7. 22					3	93392. 45								
			3	63494. 00	-0. 56					2	93412. 93								
			2	63494. 56															
$a \ ^5P_3$ $\ ^5P_2$ $\ ^5P_1$	$3d^5(a \ ^4P)4s$	$a \ ^5P$	3	66464. 64	-58. 31	$a \ ^1G_4$	$3d^5(a \ ^2G)4s$	$c \ ^1G$	4	93512. 64									
			2	66522. 95	-68. 73				$b \ ^4F_3$	$3d^5(b \ ^2F)4s$		$c \ ^1F$	3	97041. 38					
			1	66591. 68															
$a \ ^5D_4$ $\ ^5D_3$ $\ ^5D_2$ $\ ^5D_1$ $\ ^5D_0$	$3d^5(a \ ^4D)4s$	$b \ ^5D$	4	69695. 73	-141. 10	$c \ ^3D_1$ $\ ^3D_2$ $\ ^3D_3$	$3d^5(b \ ^2D)4s$	$d \ ^3D$	1	105895. 35	10. 88								
			3	69836. 83	-0. 93				2	105906. 23		22. 93							
			2	69837. 76	49. 57				3	105929. 16									
			1	69788. 19	40. 79														
			0	69747. 40															
$a \ ^3G_5$ $\ ^3G_4$ $\ ^3G_3$	$3d^5(a \ ^4G)4s$	$b \ ^3G$	5	70694. 03	-34. 72	$b \ ^1D_2$	$3d^5(b \ ^2D)4s$	$c \ ^1D$	2	109570. 84									
			4	70728. 75	3. 74				$z \ ^5G_2$ $\ ^5G_3$ $\ ^5G_4$ $\ ^5G_5$ $\ ^5G_5$	$3d^5(a \ ^4G)4p$		$z \ ^5G^\circ$	2	113584. 20	21. 17				
			3	70725. 01									3	113605. 37		29. 97			
$a \ ^3P_2$ $\ ^3P_1$ $\ ^3P_0$	$3d^5(a \ ^4P)4s$	$c \ ^3P$	2	73727. 64	-121. 46	$z \ ^5H_3$ $\ ^5H_4$ $\ ^5H_5$ $\ ^5H_6$ $\ ^5H_7$	$3d^5(a \ ^4G)4p$	$z \ ^5H^\circ$	3	114948. 55	162. 37								
			1	73849. 10	-86. 86				4	115110. 92		178. 99							
			0	73935. 96					5	115289. 91			184. 34						
			6	115474. 25	167. 98														
$a \ ^3D_3$ $\ ^3D_2$ $\ ^3D_1$	$3d^5(a \ ^4D)4s$	$b \ ^3D$	3	76956. 79		-145. 64	$c \ ^3G_5$ $\ ^3G_4$ $\ ^3G_3$	$3d^5(b \ ^2G)4s$	$d \ ^3G$	5	114325. 35		-14. 60						
			2	77102. 43	27. 13	4				114339. 95	-11. 97								
			3	77075. 30		3				114351. 92									
$a \ ^3I_7$ $\ ^3I_6$ $\ ^3I_5$	$3d^5(a \ ^2I)4s$	$a \ ^3I$	7	79840. 12	-4. 62	$z \ ^5F_5$ $\ ^5F_4$ $\ ^5F_3$ $\ ^5F_2$ $\ ^5F_1$	$3d^5(a \ ^4G)4p$	$z \ ^5F^\circ$	5	116316. 63	-150. 78								
			6	79844. 74	-15. 68				4	116467. 41		-601. 15							
			5	79860. 42					3	117068. 56			93. 51						
									2	116975. 05				37. 48					
$4p \ ^7P_2$ $\ ^7P_3$ $\ ^7P_4$	$3d^5(a \ ^6S)4p$	$z \ ^7P^\circ$	2	82001. 73	332. 19	$z \ ^5D_0$ $\ ^5D_1$ $\ ^5D_2$ $\ ^5D_3$ $\ ^5D_4$	$3d^5(a \ ^4P)4p$	$z \ ^5D^\circ$	0	116364. 76	15. 31								
			3	82333. 92	512. 67				1	116380. 07		39. 32							
			4	82846. 59					2	116419. 39			56. 05						
									3	116475. 44				1046. 47					
$b \ ^3D_3$ $\ ^3D_2$ $\ ^3D_1$	$3d^5(a \ ^2D)4s$	$c \ ^3D$	3	82382. 87	-28. 07	$z \ ^5S_2$	$3d^5(a \ ^4P)4p$	$z \ ^5S^\circ$	2	116898. 22									
			2	82410. 94	-83. 94				$b \ ^1G_4$	$3d^5(b \ ^2G)4s$		$d \ ^1G$	4	117950. 32					
			3	82494. 88									$z \ ^3F_2$ $\ ^3F_3$ $\ ^3F_4$	$3d^5(a \ ^4G)4p$	$z \ ^3F^\circ$	2	118163. 56	82. 96	
			1	82494. 88												3	118246. 52		103. 72
$a \ ^5F_5$ $\ ^5F_4$ $\ ^5F_3$ $\ ^5F_2$ $\ ^5F_1$	$3d^5(a \ ^4F)4s$	$a \ ^5F$	5	83138. 23	-23. 25	$z \ ^3H_6$ $\ ^3H_5$ $\ ^3H_4$	$3d^5(a \ ^4G)4p$	$z \ ^3H^\circ$	6	118355. 01	-202. 24								
			4	83161. 48	-76. 38				$z \ ^5P_3$ $\ ^5P_2$ $\ ^5P_1$	$3d^5(a \ ^4P)4p$		$y \ ^5P^\circ$	5	118557. 25	-129. 00				
			3	83237. 86	-121. 02								3	118442. 92		-278. 68			
			2	83358. 88	-288. 10												2	118721. 60	-146. 27
			1	83646. 98															
$a \ ^1I_6$	$3d^5(a \ ^2I)4s$	$b \ ^1I$	6	83429. 61															
			$a \ ^3F_4$ $\ ^3F_3$ $\ ^3F_2$	$3d^5(a \ ^2F)4s$	$c \ ^3F$	4	84159. 55	-512. 32											
						3	84671. 87	301. 95											
2	84369. 92																		
$a \ ^1D_2$	$3d^5(a \ ^2D)4s$	$b \ ^1D$	2	86847. 11															
			$a \ ^1F_3$	$3d^5(a \ ^2F)4s$	$b \ ^1F$	3	87901. 87												
$a \ ^3H_4$ $\ ^3H_5$ $\ ^3H_6$	$3d^5(a \ ^2H)4s$	$b \ ^3H$				4	88663. 87	30. 80											
			5	88694. 67	228. 40														
			6	88923. 07															
$4p \ ^5P_3$ $\ ^5P_2$ $\ ^5P_1$	$3d^5(a \ ^6S)4p$	$z \ ^5P^\circ$	3	89084. 79	-249. 80														
			2	89334. 59	-156. 85														
			1	89491. 44															

Fe III—Continued

Fe III—Continued

Authors	Config.	Desig.	J	Level	Interval	Authors	Config.	Desig.	J	Level	Interval
z^3P_2 3P_1 3P_0	$3d^5(a^4P)4p$	z^3P°	2 1 0	119697.64 119982.26 120179.95	-284.62 -197.69	y^3G_3 3G_4 3G_5	$3d^5(a^2F)4p$	y^3G°	3 4 5	134549.00 135554.41 135735.31	1005.41 180.90
y^5F_1 5F_2 5F_3 5F_4 5F_5	$3d^5(a^4D)4p$	y^5F°	1 2 3 4 5	120697.10 120826.17 121008.78 121241.67 121468.82	129.07 182.61 232.89 227.15	y^5G_2 5G_3 5G_4 5G_5 5G_6	$3d^5(a^4F)4p$	y^5G°	2 3 4 5 6	134937.84 135096.84 135239.74 135316.42 135739.47	159.00 142.90 76.68 423.05
z^3G_3 3G_4 3G_5	$3d^5(a^4G)4p$	z^3G°	3 4 5	121919.74 121941.29 121949.62	21.55 8.33	x^3D_3 3D_2 3D_1	$3d^5(a^2D)4p$	x^3D°	3 2 1	134976.22 135279.0 135217.1	-302.8 61.9
z^2D_3 3D_2 3D_1	$3d^5(a^4P)4p$	z^3D°	3 2 1	122346.61 122628.34 122843.03	-281.73 -214.69	z^1I_6	$3d^5(a^2I)4p$	z^1I°	6	135532.08	
y^5D_4 5D_3 5D_2 5D_1 5D_0	$3d^5(a^4D)4p$	y^5D°	4 3 2 1 0	122944.15 122829.55 122898.84 122921.37 123455.92	114.60 -69.29 -22.53 -534.55	w^3D_3 3D_2 3D_1	$3d^5(a^2F)4p$	w^3D°	3 2 1	135705.7 136793.8 136464.9	-1088.1 328.9
y^5P_1 5P_2 5P_3	$3d^5(a^4D)4p$	x^5P°	1 2 3	123552.95 123697.18 123750.39	144.23 53.21	x^5F_5 5F_4 5F_3 5F_2 5F_1	$3d^5(a^4F)4p$	x^5F°	5 4 3 2 1	136185.17 135990.62 136008.74 136117.94 136235.84	194.55 -18.12 -109.20 -117.90
y^3D_3 3D_2 3D_1	$3d^5(a^4D)4p$	y^3D°	3 2 1	124854.04 124903.92 124954.88	-49.88 -50.96	z^1F_3	$3d^5(a^2D)4p$	z^1F°	3	136200.13	
y^3F_4 3F_3 3F_2	$3d^5(a^4D)4p$	y^3F°	4 3 2	125443.58 125637.98 125672.83	-194.40 -34.85	w^3F_2 3F_3 3F_4	$3d^5(a^2F)4p$	w^3F°	2 3 4	136532.45 136797.05 136612.78	264.60 -184.27
z^3S_1	$3d^5(a^4P)4p$	z^3S°	1	126390.57		x^5D_4 5D_3 5D_2 5D_1 5D_0	$3d^5(a^4F)4p$	x^5D°	4 3 2 1 0	137209.73 137423.00 137544.60 137561.1 137573.2	-213.27 -121.60 -16.5 -12.1
y^3P_0 3P_1 3P_2	$3d^5(a^4D)4p$	y^3P°	0 1 2	128371.53 128605.65 128917.51	234.12 311.86	x^3H_4 3H_5 3H_6	$3d^5(a^2G)4p$	x^3H°	4 5 6	137527.92 137763.70 138264.47	235.78 500.77
z^3I_7 3I_6 3I_5	$3d^5(a^2I)4p$	z^3I°	7 6 5	130040.56 129854.80 130256.27	185.76 -401.47	x^3G_5 3G_4 3G_3	$3d^5(a^2H)4p$	x^3G°	5 4 3	138054.59 138103.12 138187.93	-48.53 -84.81
z^3K_6 3K_7 3K_8	$3d^5(a^2I)4p$	z^3K°	6 7 8	130756.84 131035.07 130852.25	278.23 -182.82	z^1P_1	$3d^5(a^2D)4p$	z^1P°	1	138691.81	
z^1D_2	$3d^5(a^2D)4p$	z^1D°	2	131445.03		w^3G_5 3G_4 3G_3	$3d^5(a^4F)4p$	w^3G°	5 4 3	139463.36 139625.17 139680.47	-161.81 -55.30
z^1H_5	$3d^5(a^2I)4p$	z^1H°	5	131710.79		y^3I_5 3I_6 3I_7	$3d^5(a^2H)4p$	y^3I°	5 6 7	139509.44 139846.18 140196.33	336.74 350.15
z^1K_7	$3d^5(a^2I)4p$	z^1K°	7	131991.58		y^1D_2	$3d^5(a^2F)4p$	y^1D°	2	139764.48	
x^3F_2 3F_3 3F_4	$3d^5(a^2D)4p$	x^3F°	2 3 4	132104.94 132079.91 132785.36	-25.03 705.45	y^1G_4	$3d^5(a^2G)4p$	y^1G°	4	139827.17	
y^3H_6 3H_5 3H_4	$3d^5(a^2I)4p$	y^3H°	6 5 4	132262.66 132564.71 132659.17	-302.05 -94.46	y^1F_3	$3d^5(a^2F)4p$	y^1F°	3	140453.10	
x^3P_2 3P_1 3P_0	$3d^5(a^2D)4p$	x^3P°	2 1 0	134265.42 134549.4 135088.1	-284.0 -538.7	v^3F_2 3F_3 3F_4	$3d^5(a^2G)4p$	v^3F°	2 3 4	140750.98 140693.36 141002.99	-57.62 309.63
z^1G_4	$3d^5(a^2F)4p$	z^1G°	4	134360.40		v^3D_3 3D_2 3D_1	$3d^5(a^4F)4p$	v^3D°	3 2 1	141466.53 141399.04 141469.45	67.49 -70.41
						y^1I_6	$3d^5(a^2H)4p$	y^1I°	6	141539.55	

Fe III—Continued

Fe III—Continued

Authors	Config.	Desig.	J	Level	Interval	Authors	Config.	Desig.	J	Level	Interval
u^3F_4	$3d^5(a^4F)4p$	u^3F°	4	142047. 0	-265. 9	v^1F_3	$3d^5(b^2D)4p$	v^1F°	3	159493. 0	
3F_3			3	142312. 90	-222. 17	v^3P_2	$3d^5(b^2D)4p$	v^3P°	2	160037. 9	
3F_2			2	142535. 07		3P_1		1			
w^3H_4	$3d^5(a^2H)4p$	w^3H°	4	142855. 59	52. 89	3P_0		0			
3H_5			5	142908. 48	412. 37	w^1D_2	$3d^5(b^2D)4p$	w^1D°	2	162084. 8?	
3H_5			6	143320. 85		v^3H_4	$3d^5(b^2G)4p$	v^3H°	4	165719. 1	220. 5
v^3G_5	$3d^5(a^2G)4p$	v^3G°	5	143883. 74	-202. 23	3H_5		5	165939. 6	247	
3G_4			4	144085. 97	-30. 67	3H_6		6	166187. ?		
3G_3			3	144116. 64		$5p^7P_2$	$3d^5(a^6S)5p$	y^7P°	2	166144. 66	108. 08
t^3F_4	$3d^5(b^2F)4p$	t^3F°	4	144332. 21	-238. 32	7P_3		3	166252. 74	168. 61	
3F_4			3	144570. 53	68. 79	7P_4		4	166421. 35		
3F_2			2	144501. 74		r^3F_4	$3d^5(b^2G)4p$	r^3F°	4	166222. 2	-276
y^1H_5	$3d^5(a^2G)4p$	y^1H°	5	144586. 83		3F_3		3	166498	-504	
x^1H_5	$3d^5(a^2H)4p$	x^1H°	5	144843. 24		3F_2		2	167002		
x^1G_4	$3d^5(a^2H)4p$	x^1G°	4	144968. 50		t^3G_3	$3d^5(b^2G)4p$	t^3G°	3	167085. 0	122. 3
x^1F_3	$3d^5(a^2G)4p$	x^1F°	3	145038. 61		3G_4		4	167207. 3	91. 6	
x^1D_2	$3d^5(b^2F)4p$	x^1D°	2	145618. 39		3G_5		5	167298. 9		
u^3G_3	$3d^5(b^2F)4p$	u^3G°	3	146891. 0	270. 4	$5p^5P_3$	$3d^5(a^6S)5p$	w^5P°	3	168329. 65	-91. 36
3G_4			4	147161. 36	244. 78	5P_2		2	168421. 01	-56. 35	
3G_5			5	147406. 14		5P_1		1	168477. 36		
$4d^7D_1$	$3d^5(a^6S)4d$	e^7D	1	147281. 72	9. 48	w^1H_5	$3d^5(b^2G)4p$	w^1H°	5	168780. 1	
7D_2			2	147291. 20	14. 77	v^1G_4	$3d^5(b^2G)4p$	v^1G°	4	169277. 6?	
7D_3			3	147305. 97	20. 87	u^1F_3	$3d^5(b^2G)4p$	u^1F°	3	170310. 6?	
7D_4			4	147326. 84	27. 86	$4f^7F_0$	$3d^5(a^6S)4f$	z^7F°	0		
7D_5			5	147354. 70		7F_1		1			
u^3D_1	$3d^5(b^2F)4p$	u^3D°	1	147556. 45	58. 20	7F_2		2	184247. 3	69. 3	
3D_2			2	147614. 65	21. 30	7F_3		3	184316. 6	58. 1	
3D_3			3	147635. 95		7F_4		4	184374. 7	42. 2	
w^3P_0	$3d^5(a^2S)4p$	w^3P°	0	148655	260	7F_5		5	184416. 9	30. 8	
3P_1			1	148915. 3	610. 3	7F_6		6	184447. 7		
3P_2			2	149525. 63		$4f^5F$	$3d^5(a^6S)4f$	w^5F°	1 to 5	184781 ±	
w^1G_4	$3d^5(b^2F)4p$	w^1G°	4	149013. 36		$5d^7D_1$	$3d^5(a^6S)5d$	f^7D	1	190393. 30	4. 44
$5s^7S_3$	$3d^5(a^6S)5s$	e^7S	3	149284. 94		7D_2		2	190397. 74	6. 60	
w^1F_3	$3d^5(b^2F)4p$	w^1F°	3	150654. 9		7D_3		3	190404. 34	9. 25	
$4d^5D_0$	$3d^5(a^6S)4d$	e^5D	0	151537. 83	-1. 21	7D_4		4	190413. 59	12. 17	
5D_1			1	151536. 62	-1. 80	7D_5		5	190425. 76		
5D_2			2	151534. 82	-0. 77	$6s^7S_3$	$3d^5(a^6S)6s$	f^7S	3	190918. 28	
5D_3			3	151534. 05	3. 65	$6s^5S_2$	$3d^5(a^6S)6s$	f^5S	2	192006. 99	
5D_4			4	151537. 70		$5d^5D_0$	$3d^5(a^6S)5d$	f^5D	0	193595. 3	4. 5
y^1P_1	$3d^5(a^2S)4p$	y^1P°	1	151637. 3?		5D_1		1	193599. 8	6. 3	
$5s^5S_2$	$3d^5(a^6S)5s$	e^5S	2	151757. 61		5D_2		2	193606. 1	4. 2	
s^3F_2	$3d^5(b^2D)4p$	s^3F°	2	157684. 3	297. 7	5D_3		3	193610. 3	4. 2	
3F_3			3	157982. 0	580. 7	5D_4		4	193610. 8	0. 5	
3F_4			4	158562. 7		$5g^7G$	$3d^5(a^6S)5g$	e^7G	1 to 7	207641	
t^3D_1	$3d^5(b^2D)4p$	t^3D°	1	158257. 3	159. 5	$5g^5G$	$3d^5(a^6S)5g$	e^5G	2 to 6	207643 ±	
3D_2			2	158416. 8	312. 5						
3D_3			3	158729. 3							
							Fe IV ($^6S_{2\frac{1}{2}}$)	Limit	-----	247200	

Fe III OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms				
	$ns (n \geq 4)$	$np (n \geq 4)$	$nd (n \geq 4)$	$nf (n \geq 4)$	$ng (n \geq 5)$
$3d^6$	a^1S a, b^3P a^3D a^1D a^3G a^3H a^1I				
$3d^5(a^6S)nx$	$\{ a, e, f^7S$ $(a, e, f^5S$	z, y^7P^o z, w^5P^o		e, f^7D e, f^5D	e^7G e^5G
$3d^5(a^4G)nx$		z^5S^o z^3S^o	z^5F^o z^3F^o z^5G^o z^3G^o z^5H^o z^3H^o		
$3d^5(a^4P)nx$	$\{ a^5P$ c^3P	z^5P^o z^3P^o z^5D^o z^3D^o			
$3d^5(a^4D)nx$	$\{ b^5D$ b^3D	x^5P^o y^3P^o y^5D^o y^3D^o y^5F^o y^3F^o		y^5H^o z^3I^o z^3I^o z^3K^o z^1I^o z^1K^o	
$3d^5(a^2I)nx$	$\{ a^3I$ b^1I				
$3d^5(a^2D)nx$	$\{ c^3D$ b^1D	x^3P^o z^1P^o x^3D^o z^1D^o x^3F^o z^1F^o			
$3d^5(a^4F)nx$	$\{ a^5F$ d^3F	x^5D^o v^3D^o u^3F^o w^3G^o y^5G^o w^3G^o			
$3d^5(a^2F)nx$	$\{ c^3F$ b^1F	w^3D^o y^1D^o w^3F^o y^1F^o y^3G^o z^1G^o			
$3d^5(a^2H)nx$	$\{ b^3H$ a^1H	x^3G^o x^1G^o w^3H^o y^1H^o			
$3d^5(a^2G)nx$	$\{ c^3G$ c^1G	v^3F^o x^1F^o y^3H^o x^3H^o y^1H^o			
$3d^5(b^2F)nx$	$\{ e^3F$ c^1F	u^3D^o x^1D^o w^3F^o w^1F^o u^3G^o w^1G^o			
$3d^5(b^2D)nx$	$\{ d^3D$ c^1D	v^3P^o w^1D^o s^3F^o v^1F^o			
$3d^5(b^2G)nx$	$\{ d^3G$ d^1G	r^3F^o u^1F^o t^3G^o v^1G^o w^3H^o w^1H^o			
$3d^5(a^2S)nx$	$\{ w^3P^o$ y^1P^o				

*For predicted terms in the spectra of the Cr I isoelectronic sequence, see Vol. II, Introduction.

Fe IV

(V I sequence; 23 electrons)

Z=26

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^5 {}^6S_{2\frac{1}{2}}$ $a {}^6S_{2\frac{1}{2}}$ cm⁻¹

I. P. volts

The analysis of the Fe IV spectrum is conspicuously incomplete. In 1931 Miss Gilroy published 11 terms, and 71 classified lines between 574 Å and 1825 Å. Later, Kruger and Gilroy reported that the three lines from the ground term, having the transition $3d^5 {}^6S - 4p {}^6P^\circ$, were near 526 Å, rather than between 574 Å and 587 Å as was stated in the earlier paper. Consequently, the relative term values published in 1931 need revision.

By analogy with Fe III, Edlén has estimated the positions of the lowest terms. These are entered in brackets in the table. The term $4p {}^6P^\circ$ has been taken from the observed wave numbers of the combination with the ground state.

REFERENCES

H. T. Gilroy, Phys. Rev. **38**, 2217 (1931). (T) (C L)P. G. Kruger and H. T. Gilroy, Phys. Rev. **48**, 720 (1935). (T) (C L)

B. Edlén, letter; Actualités Scientifiques et Industrielles, No. 895, Part I (Hermann et Cie, Paris, 1941) (T)

Fe IV

Config.	Desig.	J	Level	Interval
$3d^5$	$3d^5 {}^6S$	$2\frac{1}{2}$	0	
$3d^5$	$3d^5 {}^4G$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ \text{to} \\ 5\frac{1}{2} \end{array} \right\}$	[32000]	
$3d^5$	$3d^5 {}^4P$	$\left\{ \begin{array}{l} 0\frac{1}{2} \\ \text{to} \\ 2\frac{1}{2} \end{array} \right\}$	[35100]	
$3d^5$	$3d^5 {}^4D$	$\left\{ \begin{array}{l} 0\frac{1}{2} \\ \text{to} \\ 3\frac{1}{2} \end{array} \right\}$	[38500]	
$3d^5$	$3d^5 {}^4F$	$\left\{ \begin{array}{l} 0\frac{1}{2} \\ \text{to} \\ 4\frac{1}{2} \end{array} \right\}$	[52100]	
$3d^4({}^6D)4p$	$4p {}^6P^\circ$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	$\begin{array}{l} 189897 \\ 190013 \\ 190230 \end{array}$	$\begin{array}{l} 116 \\ 217 \end{array}$

January 1950.

Fe v

(Ti I sequence; 22 electrons)

 $Z=26$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^4 {}^5D_0$ $a {}^5D_0$ cm⁻¹

I. P. volts

The analysis is from Bowen who has extended White's early work and published 145 classified lines; 98 in the range from 364 Å to 432 Å, and 47 in the interval between 1302 Å and 1554 Å.

Intersystem combinations connecting the triplet and quintet systems of terms have been observed.

No series have been found and the spectrum needs further study. Bowen remarks that some ambiguity exists as to the assignment of some levels of the $z {}^5D^0$, $z {}^5F^0$ and $z {}^3D^0$ terms because of the overlapping of these terms. He states also that the term here labeled $x {}^3G^0$ may possibly be the ${}^3H^0$ term from the same limit, $a {}^2H$.

REFERENCE

I. S. Bowen, Phys. Rev. **52**, 1153 (1937). (T) (C L)

Fe v					Fe v				
Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3d^4$	$a {}^5D$	0	0	145	$3d^3(a {}^4F)4p$	$z {}^5D^0$	0		
		1	145	274			1	257746	
		2	419	385			2	258134	388
		3	804	481			3	258685	551
		4	1285			4	259349	664	
$3d^4$	$a {}^3P$	0	24054	919	$3d^3(a {}^4F)4p$	$z {}^5F^0$	1	258887	493
		1	24973	1493			2	259380	579
		2	26466				3	259959	569
$3d^4$	$a {}^3H$	4	24937	292			4	260528	526
		5	25229	298	5	261054			
		6	25527						
$3d^4$	$a {}^3F$	2	26765	81	$3d^3(a {}^4F)4p$	$z {}^3D^0$	1	259994	425
		3	26846	127			2	260419	763
		4	26973				3	261182	
$3d^4$	$a {}^3G$	3	29820	330	$3d^3(a {}^4F)4p$	$z {}^3G^0$	3	263911	534
		4	30150	279			4	264445	672
		5	30429				5	265117	
$3d^3(a {}^4F)4s$	$a {}^5F$	1	186437	299	$3d^3(a {}^4F)4p$	$z {}^3F^0$	2	266625	623
		2	186736	426			3	267248	684
		3	187162	563			4	267932	
		4	187725	676					
		5	188401						
$3d^3(a {}^4F)4s$	$b {}^3F$	2	195212	730	$3d^3(a {}^4P)4p$	$z {}^5P^0$	1	273643	493
		3	195942	903			2	274136	792
		4	196845				3	274928	
$3d^3(a {}^4F)4p$	$z {}^5G^0$	2	254805	601	$3d^3(a {}^2H)4p$	$x {}^3G^0$	5	292291	-146
		3	255406	773			4	292437	-81
		4	256179	963			3	292518	
		5	257142	1159					
		6	258301						

January 1949.

Fe v OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms	
$3d^4$	$\left\{ \begin{array}{l} a^3P \quad a^5D \quad a^3F \quad a^3G \quad a^3H \end{array} \right.$	
	$ns (n \geq 4)$	$np (n \geq 4)$
$3d^3(a^4F)nx$	$\left\{ \begin{array}{l} a^5F \\ b^3F \end{array} \right.$	$\begin{array}{lll} z^5D^\circ & z^5F^\circ & z^5G^\circ \\ z^3D^\circ & z^3F^\circ & z^3G^\circ \end{array}$
$3d^3(a^4P)nx$		z^5P°
$3d^3(a^2G)nx$		y^3G°
$3d^3(a^2H)nx$		x^3G°

*For predicted terms in the spectra of the Ti I isoelectronic sequence, see Vol. I, p. xxxvii.

Fe VI

(Sc I sequence; 21 electrons)

$Z=26$

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^3 ^4F_{1/2}$

$a^4F_{1/2} \quad \text{cm}^{-1}$

I. P. volts

The analysis is incomplete, but Bowen has classified about 100 lines in the region between 276 Å and 318 Å. Intersystem combinations connecting the doublet and quartet systems of terms have been observed.

REFERENCES

- I. S. Bowen, Phys. Rev. **47**, 924 (1935). (T) (C L)
 S. Pasternack, Astroph. J. **92**, 140 (1940). (T)

Fe VI

Fe VI

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval	
$3d^3$	a^4F	$1\frac{1}{2}$	0	510	$3d^2(a^3F)4p$	z^4F°	$2\frac{1}{2}$	342570	1030	
		$2\frac{1}{2}$	510	$3\frac{1}{2}$			343600			
		$3\frac{1}{2}$	1185	$3d^2(a^3F)4p$		z^4D°	$0\frac{1}{2}$	343619		-409
		$4\frac{1}{2}$	1994				$1\frac{1}{2}$	343210		1060
$3d^3$	a^4P	$0\frac{1}{2}$	18734	203	$3d^2(a^3F)4p$	z^2D°	$2\frac{1}{2}$	344270	1149	
		$1\frac{1}{2}$	18937	664			$3\frac{1}{2}$	345419		
		$2\frac{1}{2}$	19601							
$3d^3$	a^2G	$3\frac{1}{2}$	20609	696	$3d^2(a^3F)4p$	z^2D°	$1\frac{1}{2}$	344656	1252	
		$4\frac{1}{2}$	21305				$2\frac{1}{2}$	345908		
$3d^3$	a^2P	$1\frac{1}{2}$	26204	-279	$3d^2(a^3F)4p$	z^2G°	$3\frac{1}{2}$	348962	1054	
		$0\frac{1}{2}$	26483				$4\frac{1}{2}$	350016		
$3d^3$	a^2D	$2\frac{1}{2}$	28469	-138	$3d^2(a^3P)4p$	z^4S°	$1\frac{1}{2}$	355652		
		$1\frac{1}{2}$	28607							
$3d^3$	a^2H	$4\frac{1}{2}$	28723	473	$3d^2(a^3P)4p$	y^4D°	$0\frac{1}{2}$	359104	680	
		$5\frac{1}{2}$	29196				$1\frac{1}{2}$	359784		921
							$2\frac{1}{2}$	360705		1560
$3d^2(a^3F)4p$	z^4G°	$2\frac{1}{2}$	338256	1221	$3d^2(a^3P)4p$	z^4P°	$0\frac{1}{2}$	363944	446	
		$3\frac{1}{2}$	339477	1452			$1\frac{1}{2}$	364390		1102
		$4\frac{1}{2}$	340929	1794			$2\frac{1}{2}$	365492		
		$5\frac{1}{2}$	342723							
$3d^2(a^3F)4p$	z^4F°	$1\frac{1}{2}$	339538	803	$3d^2(a^1G)4p$	y^2G°	$3\frac{1}{2}$	365077	185	
		$2\frac{1}{2}$	340341	1020			$4\frac{1}{2}$	365262		
		$3\frac{1}{2}$	341361	1065						
		$4\frac{1}{2}$	342426							
					$3d^2(a^1G)4p$	z^2H°	$4\frac{1}{2}$	372096	1606	
				$5\frac{1}{2}$			373702			

December 1948.

Fe VI OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms
$3d^3$	a^4P a^4F a^2P a^2D a^2G a^2H
	np ($n \geq 4$)
$3d^2(a^3F)nx$	z^4D° z^4F° z^4G° z^2D° z^2F° z^2G°
$3d^2(a^3P)nx$	z^4S° z^4P° y^4D°
$3d^2(a^1G)nx$	y^2G° z^2H°

*For predicted terms in the spectra of the Sc I isoelectronic sequence, see Vol. I, p. xxxvi.

Fe VII

(Ca I sequence; 20 electrons)

 $Z=26$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^2 {}^3F_2$ $a {}^3F_2$ cm^{-1}

I. P. volts

The early analysis by Cady, who classified 33 lines between 231 Å and 245 Å, has been revised and extended by Edlén. The present term list is from Edlén's manuscript, kindly furnished by him for inclusion here in advance of publication.

The prefix "y" is assigned to the ${}^1P^\circ$ and ${}^3D^\circ$ terms from the $4f$ configuration on the assumption that the similar terms from the $4p$ configuration, which have not yet been found, will be lower and should, therefore, be assigned the prefix "z".

REFERENCES

- W. M. Cady, Phys. Rev. **43**, 324 (1933). (T) (C L)
 I. S. Bowen and B. Edlén, Nature **143**, 374 (1939). (T) (C L)
 B. Edlén, unpublished material (Feb. 1949). (T)

Fe VII

Fe VII

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3d^2$	$a {}^3F$	2	0	1047 1280	$3d({}^2D)4f$	$y {}^3F^\circ$	2	660022	338 816
		3	1047				3	660360	
		4	2327				4	661176	
$3d^2$	$a {}^1D$	2	17475		$3d({}^2D)4f$	$z {}^3G^\circ$	3	663104	849
$3d^2$	$a {}^3P$	0	20037	391 847	$3d({}^2D)4f$	$y {}^1D^\circ$	4	663953	530
		1	20428				5	664483	
		2	21275				2	663882	
$3d^2$	$a {}^1G$	4	28915		$3d({}^2D)4f$	$y {}^1F^\circ$	3	665425	
$3d({}^2D)4p$	$z {}^1D^\circ$	2	425388		$3d({}^2D)4f$	$y {}^3D^\circ$	1	665843	82
$3d({}^2D)4p$	$z {}^3F^\circ$	2		2264	$3d({}^2D)4f$	$y {}^3P^\circ$	2	665925	738
		3	431606				3	666663	
		4	433870				0	667903	
$3d({}^2D)4p$	$z {}^3P^\circ$	0	437010	-47 604	$3d({}^2D)4f$	$z {}^1H^\circ$	1	668265	-362
		1	436963				0	668497	-232
		2	437567				5	669978	
$3d({}^2D)4p$	$z {}^1F^\circ$	3	439812		$3d({}^2D)4f$	$y {}^1P^\circ$	1	671470	
$3d({}^2D)4f$	$z {}^1G^\circ$	4	659923						

March 1949.

Fe VII OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms						
$3d^2$	{ $a {}^3P$ $a {}^1D$ $a {}^3F$ $a {}^1G$						
	np ($n \geq 4$)			nf ($n \geq 4$)			
$3d({}^2D)nx$	{ $z {}^3P^\circ$ $z {}^1D^\circ$ $z {}^3F^\circ$ $z {}^1F^\circ$		$y {}^3P^\circ$	$y {}^3D^\circ$	$y {}^3F^\circ$	$z {}^3G^\circ$	$z {}^1H^\circ$
			$y {}^1P^\circ$	$y {}^1D^\circ$	$y {}^1F^\circ$	$z {}^1G^\circ$	

*For predicted terms in the spectra of the Ca I isoelectronic sequence, see Vol. I, p. xxxv.

Fe VIII

(K I sequence; 19 electrons)

Z=26

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^2 D_{1\frac{1}{2}}$ $3d^2 D_{1\frac{1}{2}}$ 1219360 cm^{-1}

I. P. 151 volts

The analysis is by Kruger and Weissberg, who have classified 15 lines in the interval 93 A to 370 A.

REFERENCE

P. G. Kruger and S. G. Weissberg, Phys. Rev. **52**, 316 (1937). (I P) (T) (C L)

Fe VIII

Config.	Desig.	J	Level	Interval
$3p^6(1S)3d$	$3d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	0 1875	1875
$3p^6(1S)4p$	$4p^2 P^\circ$	$0\frac{1}{2}$ $1\frac{1}{2}$	510084 513447	3363
$3p^6(1S)4f$	$4f^2 F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	763714 763821	107
$3p^6(1S)5s$	$5s^2 S$	$0\frac{1}{2}$	783402	
$3p^6(1S)5f$	$5f^2 F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	927024 927084	60
$3p^6(1S)6s$	$6s^2 S$	$0\frac{1}{2}$	960135	
$3p^6(1S)6f$	$6f^2 F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	1016828 1016877	49
$3p^6(1S)7f$	$7f^2 F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	1072765 1072875	110
-----	-----	---	-----	
Fe IX ($1S_0$)	<i>Limit</i>	---	1219360	

May 1948.

Fe IX

(A I sequence; 18 electrons)

Z=26

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 1S_0$ $3p^6 1S_0$ 1893000 cm^{-1}

I. P. 235 volts

Two lines, at 103 A and 105 A, respectively, are classified as combinations with the ground term. The tabular values have been rounded off in the last places. Apparently the limit and the 5s-level are extrapolated from isoelectronic data, as indicated by brackets in the table.

For convenience, the Paschen notation has been added by the writer in column one under the heading "A I". As for A I, the jl -coupling notation in the general form suggested by Racah is here introduced, although LS -designations indicated in column two under the heading "Authors" are perhaps preferable for the terms thus far identified.

REFERENCES

P. G. Kruger, S. G. Weissberg and L. W. Phillips, Phys. Rev. **51**, 1090 (1937). (I P) (T) (C L)
G. Racah, Phys. Rev. **61**, 537 (L) (1942).

Fe IX

A I	Authors	Config.	Desig.	J	Level
1p ₀	3p ⁶ 1S	3p ⁶	3p ⁶ 1S	0	0
1s ₄	3p ⁵ 4s 3P ^o	3p ⁵ (² P _{1/2})4s	4s [1½] ^o	2 1	950200
1s ₂	3p ⁵ 4s 1P ^o	3p ⁵ (² P _{3/2})4s	4s' [0½] ^o	0 1	965400
2s ₄	3p ⁵ 5s 3P ^o	3p ⁵ (² P _{1/2})5s	5s [1½] ^o	2 1	[1355000]
		-----	-----	---	-----
		Fe x (² P _{1/2})	<i>Limit</i>	---	[1893000]

May 1948.

Fe x

(Cl I sequence; 17 electrons)

Z=26

Ground state 1s² 2s² 2p⁶ 3s² 3p⁵ ²P_{1/2}^o3p⁵ ²P_{1/2}^o 2114000 cm⁻¹

I. P. 262 volts

Edlén has classified eight lines in the region between 94 Å and 97 Å, as combinations from the ground term. He has confirmed Grotrian's suggestion that the strong coronal line at 6374.51 Å (wave number 15683.2 cm⁻¹) may be due to the forbidden transition 3p⁵ ²P_{1/2}^o—3p⁵ ²P_{3/2}^o.

He has extrapolated the value of the limit along the isoelectronic sequence, as indicated by brackets in the table.

Edlén's unit, 10³ cm⁻¹, has here been changed to cm⁻¹.

REFERENCES

- B. Edlén, *Zeit. Phys.* **104**, 407 (1937). (I P) (T) (C L)
 W. Grotrian, *Naturwiss.* **27**, 214 (1939).
 B. Edlén, *Zeit. Astroph.* **22**, 30 (1942). (T) (C L)

Fe x

Config.	Desig.	J	Level	Interval
3s ² 3p ⁵	3p ⁵ 2P ^o	1½ 0½	0 15683	-15683
3s ² 3p ⁴ (³ P)4s	4s 4P	2½ 1½ 0½	1022100 1029630	-7530
3s ² 3p ⁴ (³ P)4s	4s 2P	1½ 0½	1040350 1048900	-8550
3s ² 3p ⁴ (1D)4s	4s' 2D	2½ 1½	1063690 1064190	-500
-----	-----	---	-----	-----
Fe xI (³ P ₂)	<i>Limit</i>	---	[2114000]	

January 1950.

Fe XI

(S I sequence; 16 electrons)

Z=26

Ground state $1s^2 2s^2 2p^6 3s^2 3p^4 \ ^3P_2$ $3p^4 \ ^3P_2$ 2342000 cm^{-1}

I. P. 290 volts

Edlén has classified 12 lines in the region from 86 Å to 90 Å. He has extrapolated the value of the limit from isoelectronic sequence data and also the value of $3p^4 \ ^1S_0$, entered in brackets in the table.

Following Grotrian's suggestion, Edlén has tentatively identified two coronal lines as due to the following forbidden transitions:

I. A.	Int.	Wave No.	Desig.
3986. 9	0. 7	25075	$3p^4 \ ^3P_1 - 3p^4 \ ^1D_2$
7891. 94	13	12667. 7	$3p^4 \ ^3P_2 - 3p^4 \ ^3P_1$

The singlet and triplet terms are connected by the intersystem combination furnished by the coronal line at 3986 Å.

Edlén's unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

- B. Edlén, Zeit. Phys. **104**, 188 (1937). (I P) (T) (C L)
 W. Grotrian, Naturwiss. **27**, 214 (1939).
 B. Edlén, Zeit. Astroph. **22**, 30 (1942). (C L)
 B. Edlén, letter (January 1950). (T)

Fe XI

Config.	Desig.	J	Level	Interval
$3s^2 3p^4$	$3p^4 \ ^3P$	2	0	-12668 -1772
		1	12668	
		0	14440	
$3s^2 3p^4$	$3p^4 \ ^1D$	2	37743	
$3s^2 3p^4$	$3p^4 \ ^1S$	0	[80300]	
$3s^2 3p^3(^4S^\circ)4s$	$4s \ ^3S^\circ$	1	1121270	
$3s^2 3p^3(^2D^\circ)4s$	$4s' \ ^3D^\circ$	1	1148650	450 3350
		2	1149100	
		3	1152450	
$3s^2 3p^3(^2D^\circ)4s$	$4s' \ ^1D^\circ$	2	1160020	
$3s^2 3p^3(^2P^\circ)4s$	$4s'' \ ^1P^\circ$	1	1193640	
-----	-----	---	-----	
Fe XII ($^4S_{3/2}$)	<i>Limit</i>	---	[2342000]	

January 1950.

Fe XIII

(Si 1 sequence; 14 electrons)

Z=26

Ground state $1s^2 2s^2 2p^6 3s^2 3p^2 {}^3P_0$ $3p^2 {}^3P_0$ cm⁻¹

I. P. 355 volts

In his classical work on the identification of the coronal lines Edlén suggests that three strong coronal lines may be attributed to [Fe XIII] as follows:

I. A.	Int.	Wave No.	Desig.
3388. 1	16	29507	$3p^2 {}^3P_2 - 3p^2 {}^1D_2$
10746. 80	55	9302. 5	$3p^2 {}^3P_0 - 3p^2 {}^3P_1$
10797. 95	35	9258. 5	$3p^2 {}^3P_1 - 3p^2 {}^3P_2$

The coronal wave numbers are in satisfactory agreement with the theoretical values predicted from isoelectronic sequence data. The term values in the table have been calculated from the coronal wave numbers, with the exception of $3p^2 {}^1S$ for which Edlén's estimated value is entered in brackets. Edlén's unit, 10^3 cm⁻¹, has here been changed to cm⁻¹.

REFERENCE

B. Edlén, *Zeit. Astroph.* **22**, 40 (1942). (I P) (T) (C L)

Fe XIII

Config.	Desig.	<i>J</i>	Level	Interval
$3s^2 3p^2$	$3p^2 {}^3P$	0	0. 0	9302. 5 9258. 5
		1	9302. 5	
		2	18561. 0	
$3s^2 3p^2$	$3p^2 {}^1D$	2	48068	
$3s^2 3p^2$	$3p^2 {}^1S$	0	[82000]	

October 1947.

Fe XIV

(Al I sequence; 13 electrons)

 $Z=26$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^2 P_{3/2}^{\circ}$ $3p^2 P_{1/2}^{\circ}$ cm⁻¹

I. P. 390 volts

Astrophysically, the spectrum of Fe XIV is one of the most significant, although only two lines have been classified in the laboratory spectrum. In his classical papers on the identification of the coronal lines Edlén has suggested that the conspicuous coronal line at 5302.86 Å, wave number 18852.5 cm⁻¹ and intensity 100, may be due to the forbidden transition that corresponds to the interval of the ground term of Fe XIV, i. e., $3p^2 P_{0/2}^{\circ} - 3p^2 P_{1/2}^{\circ}$.

He also classifies two lines as follows:

I. A.	Int.	Wave No.	Desig.
58. 963	2	1695980	} $3p^2 P^{\circ} - 4d^2 D$
59. 579	3	1678440	

His unit, 10³ cm⁻¹, is here changed to cm⁻¹.

REFERENCES

- B. Edlén, Zeit. Phys. **103**, 540 (1936). (C L)
 B. Edlén, Zeit. Astroph. **22**, 47 (1942). (I P) (T) (C L)

December 1947.

Fe XV

(Mg I sequence; 12 electrons)

 $Z=26$ Ground state $1s^2 2s^2 2p^6 3s^2^1 S_0$ $3s^2^1 S_0$ 3684000 cm⁻¹

I. P. 457 volts

Edlén has classified eight lines in the region between 55 Å and 70 Å. No singlet combinations have been observed and the triplet terms are not all connected by observed combinations. He has determined the relative positions of the terms and also the ionization potential by extrapolation along the isoelectronic sequence. Estimated values are denoted by brackets in the table.

Edlén's unit, 10³ cm⁻¹, has here been changed to cm⁻¹.

REFERENCE

- B. Edlén, Zeit. Phys. **103**, 536 (1936). (I P) (T) (C L)

Fe xv

Config.	Desig.	<i>J</i>	Level	Interval
$3s^2$	$3s^2\ ^1S$	0	0	
$3s(^2S)3p$	$3p\ ^3P^\circ$	0	[235690] + <i>x</i>	5800 14120
		1	241490 + <i>x</i>	
		2	255610 + <i>x</i>	
$3s(^2S)3d$	$3d\ ^3D$	1		
		2		
		3	687570 + <i>x</i>	
$3s(^2S)4p$	$4p\ ^1P^\circ$	1	[1889970]	
$3s(^2S)4d$	$4d\ ^3D$	1	2033120 + <i>x</i>	710 1140
		2	2033830 + <i>x</i>	
		3	2034970 + <i>x</i>	
$3s(^2S)4f$	$4f\ ^3F^\circ$	2		
		3		
		4	2115040 + <i>x</i>	
$3s(^2S)5f$	$5f\ ^3F^\circ$	2		
		3		
		4	2682780 + <i>x</i>	
-----	-----	---	-----	
Fe xvi ($^2S_{1/2}$)	<i>Limit</i>	---	[3684000]	

August 1947.

Fe xvi

(Na I sequence; 11 electrons)

 $Z=26$ Ground state $1s^2 2s^2 2p^6 3s\ ^2S_{1/2}$ $3s\ ^2S_{1/2}\ 3947840\ \text{cm}^{-1}$

I. P. 489 volts

Edlén has classified 13 lines in the interval 39 Å to 66 Å, and extrapolated the absolute value of the ground term from isoelectronic sequence data. His unit, $10^3\ \text{cm}^{-1}$, has here been changed to cm^{-1} .

REFERENCE

B. Edlén, *Zeit. Phys.* **100**, 621 (1936). (I P) (T) (C L)

Fe XVI

Config.	Desig.	J	Level	Interval
3s	3s 2S	$0\frac{1}{2}$	0	
3p	3p $^2P^\circ$	$0\frac{1}{2}$ $1\frac{1}{2}$	276500 297470	20970
3d	3d 2D	$1\frac{1}{2}$ $2\frac{1}{2}$	676930 679770	2840
4s	4s 2S	$0\frac{1}{2}$	1866860	
4p	4p $^2P^\circ$	$0\frac{1}{2}$ $1\frac{1}{2}$	1978040 1986100	8060
4d	4d 2D	$1\frac{1}{2}$ $2\frac{1}{2}$	2123500 2124690	1190
4f	4f $^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	2186070 2186570	500
5d	5d 2D	$1\frac{1}{2}$ $2\frac{1}{2}$	2787360 2787940	580
5f	5f $^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	2820050 2820270	220
-----	-----	---	-----	
Fe XVII (1S_0)	<i>Limit</i>	---	[3947840]	

June 1947.

Fe XVII

(Ne I sequence; 10 electrons)

 $Z=26$ Ground state $1s^2 2s^2 2p^6 ^1S_0$ $2p^6 ^1S_0$ 10211700 cm^{-1}

I. P. 1266 volts

Tyrén has classified nine lines in the region 12 Å to 17 Å, as combinations with the ground term. His absolute term values have been derived by extrapolation along the Ne I iso-electronic sequence.

By analogy with Ne I, the jl -coupling notation in the general form suggested by Racah is introduced.

The unit adopted by Tyrén, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

- F. Tyrén, Zeit. Phys. **111**, 314 (1938). (I P) (T) (C L)
 G. Racah, Phys. Rev. **61**, 537 (L) (1942).

Fe XVII

Fe XVII

Author	Config.	Desig.	<i>J</i>	Level	Author	Config.	Desig.	<i>J</i>	Level
$2p\ ^1S_0$	$2s^2\ 2p^6$	$2p^6\ ^1S$	0	0	$3p'\ ^3P_1$	$2s\ 2p^6(^2S)3p$	$3p\ ^2P^\circ$	2 1 0	7201000
$3s\ ^3P_1$	$2s^2\ 2p^5(^2P_{1/2})3s$	$3s\ [1\frac{1}{2}]^\circ$	2 1	5864800	$3p'\ ^1P_1$	$2s\ 2p^6(^2S)3p$	$3p\ ^1P^\circ$	1	7235900
$3s\ ^1P_1$	$2s^2\ 2p^5(^2P_{3/2})3s$	$3s'[0\frac{1}{2}]^\circ$	0 1	5961600	$4d\ ^1P_1$	$2s^2\ 2p^5(^2P_{1/2})4d$	$4d\ [1\frac{1}{2}]^\circ$	1	8155000
$3d\ ^3P_1$	$2s^2\ 2p^5(^2P_{1/2})3d$	$3d\ [0\frac{1}{2}]^\circ$	0 1	6471200	$4d\ ^3D_1$	$2s^2\ 2p^5(^2P_{3/2})4d$	$4d'[1\frac{1}{2}]^\circ$	1	8250000
$3d\ ^1P_1$	"	$3d\ [1\frac{1}{2}]^\circ$	1	6552700					
$3d\ ^3D_1$	$2s^2\ 2p^5(^2P_{3/2})3d$	$3d'[1\frac{1}{2}]^\circ$	1	6661300		Fe XVIII ($^2P_{1/2}$)	<i>Limit</i>	-----	[10211700]
						Fe XVIII ($^2P_{3/2}$)	<i>Limit</i>	-----	[10313800]

March 1947.

Fe XVII OBSERVED LEVELS*

Config. $1s^2+$	Observed Terms		
$2s^2\ 2p^6$	$2p^6\ ^1S$		
	<i>ns</i> ($n \geq 3$)	<i>np</i> ($n \geq 3$)	<i>nd</i> ($n \geq 3$)
$2s^2\ 2p^5(^2P^\circ)nx$	{ $3s\ ^3P^\circ$ $3s\ ^1P^\circ$	$3p\ ^3P^\circ$ $3p\ ^1P^\circ$	$3d\ ^3P^\circ$ $3, 4d\ ^1P^\circ$ $3, 4d\ ^3D^\circ$
$2s\ 2p^6(^2S)nx$			
<i>jl</i> -Coupling Notation			
Config. $1s^2+$	Observed Pairs		
	<i>ns</i> ($n \geq 3$)		<i>nd</i> ($n \geq 3$)
$2s^2\ 2p^5(^2P_{1/2})nx$	$3s\ [1\frac{1}{2}]^\circ$		$3d\ [0\frac{1}{2}]^\circ$ $3, 4d\ [1\frac{1}{2}]^\circ$
$2s^2\ 2p^5(^2P_{3/2})nx'$	$3s'[0\frac{1}{2}]^\circ$		$3, 4d'[1\frac{1}{2}]^\circ$

*For predicted levels in the spectra of the Ne I isoelectronic sequence, see Vol. I, p. xxxi.

COBALT

Co I

27 electrons

Z=27

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^7 4s^2 {}^4F_{4x}$ $a {}^4F_{4x}$ 63438 cm^{-1} I. P. 7.86 ± 0.02 volts

The number of terms known in this complex spectrum is 99; in addition, there are 43 miscellaneous levels which are not yet grouped into terms. Combinations among these give 2725 classified lines in the range between 1814 Å and 11894 Å, and include 768 multiplets. The terms of all multiplicities are connected by observed intersystem combinations.

In 1936 Catalán and Antunes published a detailed analysis, classifying 2076 lines. The terms in the table are from the 1940 paper by Russell, King, and Moore. Their analysis was begun without knowledge that Catalán and Antunes were investigating the spectrum. It was continued because further observations in the infrared and ultraviolet, together with extensive data on the Zeeman effect, made it possible to revise and extend the earlier work.

Some of the electron configurations of the low even terms were assigned on the basis of the theoretical results by Marvin. Racah suggests that two of these assignments should be interchanged, as follows:

Term	R K M	Racah
$a {}^4P$	$3d^8 4s$	$3d^7 4s^2$
$b {}^4P$	$3d^7 4s^2$	$3d^8 4s$

This change was subsequently confirmed by Shenstone, and is adopted in the table. Further changes suggested by Shenstone by analogy with Ni II and Cu III have also been adopted, namely, the configuration assignments published in the 1940 reference below have been interchanged for the following pairs of terms: $y {}^2H^\circ, x {}^2H^\circ$; $y {}^4S^\circ, x {}^4S^\circ$.

Roth and Bartunek observed Zeeman patterns of 151 lines of Co I between 3200 Å and 6500 Å. Later and more extensive observations by R. B. King and A. S. King are in good general agreement with their results. The two-place entries in the table are from the Kings' observations. Those given to three places are from unpublished material by Miss Weeks, who derived them from the films made at the Massachusetts Institute of Technology and furnished by G. R. Harrison.

The limit has been determined by estimating the Rydberg denominators by comparison with neighboring elements. Russell's value is 63536 cm^{-1} , while that of Catalán and Antunes, using the same method, is 63339 cm^{-1} . The mean of these two values, 63438 cm^{-1} , is given here.

Russell has remarked that the convergence of the components of selected terms of Co I to the components of the limit terms in Co II is of interest: "Catalán and Antunes have shown that among the terms which have $d^8 {}^3F$ as limit, the quartet and doublet components of lowest J go to 3F_2 , those of next higher J to 3F_3 , leaving two high J quartet components for 3F_4 ." This "inverted" convergence was found by Russell for the limit $d^9 {}^2D$ in Ni II (see text for Ni I).

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Co I

Co I

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$3d^7 4s^2$	<i>a</i> ⁴ F	4½	0. 00		1. 32	$3d^7 4s(b^3F)4p$	<i>z</i> ² F°	3½	31871. 15		1. 177
		3½	816. 00	-816. 00	1. 237			2½	32781. 71	-910. 56	0. 870
		2½	1406. 84	-590. 84	1. 028		<i>y</i> ⁴ D°	3½	32027. 50		1. 395
		1½	1809. 33	-402. 49	0. 402			2½	32654. 50	-627. 00	1. 366
$3d^8(a^3F)4s$	<i>b</i> ⁴ F	4½	3482. 82		1. 336	1½	33150. 68	-496. 18	1. 195		
		3½	4142. 66	-659. 84	1. 239	0½	33449. 18	-298. 50	0. 012		
		2½	4690. 18	-547. 52	1. 031	$3d^8(a^3F)4p$	<i>y</i> ⁴ G°	5½	32430. 59		1. 287
		1½	5075. 83	-385. 65	0. 404			4½	32464. 73	-34. 14	1. 154
<i>a</i> ² F	3½	7442. 41		1. 147	3½		33173. 36	-708. 63	1. 039		
	2½	8460. 81	-1018. 40	0. 862	2½		33674. 38	-501. 02	0. 704		
$3d^7 4s^2$	<i>a</i> ⁴ P	2½	13795. 52		1. 604	$3d^8(a^3F)4p$	<i>y</i> ⁴ F°	4½	32841. 99		1. 313
		1½	14036. 28	-240. 76	1. 722			3½	33466. 87	-624. 88	1. 155
		0½	14399. 28	-363. 00	2. 651		2½	33945. 90	-479. 03	0. 900	
$3d^8(^3P)4s$	<i>b</i> ⁴ P	2½	15184. 04		1. 515		1½	34196. 21	-250. 31	0. 430	
		1½	15774. 04	-590. 00	1. 476	<i>y</i> ² G°	4½	33439. 72		1. 165	
		0½	16195. 68	-421. 64	2. 682		3½	34133. 59	-693. 87	0. 917	
$3d^7 4s^2$	<i>a</i> ² G	4½	16467. 90		1. 109	$3d^7 4s(b^3F)4p$	<i>z</i> ² D°	2½	33462. 83		1. 186
		3½	17233. 68	-765. 78	0. 883			1½	34352. 42	-889. 59	0. 787
$3d^8(^1D)4s$	<i>a</i> ² D	1½	16470. 60		1. 101	$3d^8(a^3F)4p$	<i>y</i> ² F°	3½	35450. 56		1. 145
		2½	16778. 16	307. 56	1. 296			2½	36329. 86	-879. 30	0. 892
$3d^8(^3P)4s$	<i>a</i> ² P	1½	18389. 57		1. 300	$3d^8(a^3F)4p$	<i>y</i> ² D°	2½	36092. 44		1. 186
		0½	18775. 01	-385. 44	0. 695			1½	36875. 13	-782. 69	0. 794
$3d^7 4s^2$	<i>b</i> ² P	1½	20500. 71		1. 284	$3d^7 4s(b^3F)4p$	<i>x</i> ⁴ D°	3½	39649. 16		1. 428
		0½	21215. 90	-715. 19	0. 680			2½	40345. 95	-696. 79	1. 370
$3d^7 4s^2$	<i>a</i> ² H	5½	21780. 47		1. 100		1½	40827. 77	-481. 82	1. 240	
		4½	22475. 36	-694. 89	0. 921		0½	41101. 80	-274. 03	0. 026	
$3d^8(^1G)4s$	<i>b</i> ² D	2½	21920. 09		1. 180	$3d^7 4s(a^5P)4p$	<i>z</i> ⁴ S°	1½	40621. 62		2. 017
		1½	23152. 57	-1232. 48	0. 955			1°	3½	41041. 43	
$3d^7 4s^2$	<i>b</i> ² G	4½	23184. 23		1. 098	2°	2½	41104. 96		1. 863	
		3½	23207. 76	-23. 53	0. 883			$3d^7 4s(b^3F)4p$	<i>x</i> ⁴ F°	4½	41225. 76
$3d^7 4s(a^5F)4p$	<i>z</i> ⁶ F°	5½	23611. 78		1. 466	3½	41918. 41			-692. 65	1. 248
		4½	23855. 62	-243. 84	1. 481	2½	42434. 23		-515. 82	1. 024	
		3½	24326. 11	-470. 49	1. 436	1½	42796. 67		-362. 44	0. 406	
		2½	24733. 28	-407. 17	1. 336	$3d^7 4s(b^3F)4p$	<i>x</i> ⁴ G°	5½	41528. 53		1. 291
1½	25041. 16	-307. 88	1. 118	4½	42269. 32			-740. 79	1. 169		
0½	25232. 79	-191. 63	0. 622	3½	42811. 44			-542. 12	1. 004		
$3d^7 4s(a^5F)4p$	<i>z</i> ⁶ D°	4½	24627. 79		1. 569			2½	43199. 65	-388. 21	0. 649
		3½	25269. 25	-641. 46	1. 550	$3d^7 4s(a^5P)4p$	<i>z</i> ⁴ P°	2½	41968. 89		1. 627
		2½	25739. 93	-470. 68	1. 612			1½	41982. 66	-13. 77	1. 732
		1½	26063. 11	-323. 18	1. 812			0½	41969. 90	12. 76	2. 51
0½	26250. 49	-187. 38	3. 286	3°	3½			42988. 12			
$3d^7 4s(a^5F)4p$	<i>z</i> ⁶ G°	6½	25138. 88				1. 40	$3d^8(^1D)4p$	<i>z</i> ² P°	0½	43130. 24
		5½	25568. 68	-429. 80	1. 354	1½	43537. 71			407. 47	1. 120
		4½	25937. 59	-368. 91	1. 281	$3d^7 4s(a^5P)4p$	<i>w</i> ⁴ D°	3½	43398. 62		1. 334
		3½	26232. 05	-294. 46	1. 150			2½	43242. 95	155. 67	1. 101
		2½	26450. 02	-217. 97	0. 876			1½	43263. 57	-20. 62	1. 191
		1½	26597. 64	-147. 62	0. 006			0½	43435. 58	-172. 01	0. 169
$3d^9$	<i>c</i> ² D	2½	27497. 06		1. 200	$3d^7 4s(^2G)4p$	<i>w</i> ⁴ F°	4½	43295. 32		1. 295
		1½	28470. 51	-973. 45	0. 907			3½	43847. 98	-552. 66	1. 197
$3d^7 4s(a^5F)4p$	<i>z</i> ⁴ F°	4½	28345. 86		1. 330			2½	44201. 92	-353. 94	0. 950
		3½	28777. 27	-431. 41	1. 247			1½	44555. 71	-353. 79	0. 415
		2½	29216. 37	-439. 10	1. 033	$3d^7 4s(^3G)4p$	<i>x</i> ² F°	2½	43425. 71		1. 119
		1½	29563. 17	-346. 80	0. 410			3½	43555. 22	129. 51	1. 229
$3d^7 4s(a^5F)4p$	<i>z</i> ⁴ G°	5½	28845. 22		1. 276		<i>x</i> ² D°	1½	43911. 36		1. 127
		4½	29269. 73	-424. 51	1. 175			2½	43921. 89	10. 53	1. 230
$3d^7 4s(a^5F)4p$	<i>z</i> ⁴ D°	3½	29735. 18		0. 995	$3d^7 4s(^3G)4p$	<i>w</i> ⁴ G°	5½	43952. 06		1. 279
		2½	30102. 96	-465. 45	0. 577			4½	44183. 34	-231. 28	1. 163
		2½	29294. 52	-654. 24	1. 425			3½	44394. 47	-211. 13	1. 004
		2½	29948. 76	-494. 87	1. 359			2½	44568. 47	-174. 00	0. 676
$3d^7 4s(a^5F)4p$	<i>z</i> ² G°	1½	30443. 63		1. 192	4°	3½	43969. 90		1. 081	
		0½	30742. 65	-299. 02	0. 006						
		4½	31699. 69		1. 126						
$3d^7 4s(b^3F)4p$	<i>z</i> ² G°	3½	32733. 07		0. 899						

Co I—Continued

Co I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
	5°	1½	44381. 32:				<i>w</i> ² S°	0½	48837. 72		1. 50:
3 <i>d</i> ⁷ 4 <i>s</i> (³ P)4 <i>p</i> :	<i>z</i> ² S°	0½	44454. 51		2. 10		8°	1½	48851. 58:		
3 <i>d</i> ⁸ (³ P)4 <i>p</i>	<i>y</i> ⁴ P°	2½	44480. 14	-177. 89	1. 557	3 <i>d</i> ⁷ 4 <i>s</i> (³ P)4 <i>p</i> :	<i>w</i> ² P°	1½	49025. 42	-729. 31	1. 099
		1½	44658. 03	-199. 54	1. 674		0½	49754. 73			1. 365
		0½	44857. 57		2. 371						
3 <i>d</i> ⁸ (<i>a</i> ³ F)5 <i>s</i>	<i>e</i> ⁴ F	4½	44782. 13	-323. 46	1. 33		9°	{ 4½ } 49197. 74:			
		3½	45105. 59	-770. 99	1. 21		3½				
		2½	45876. 58	-498. 59	1. 01		10°	3½	49484. 05		1. 260
		1½	46375. 17		0. 44		11°	2½	49847. 08		1. 079
3 <i>d</i> ⁷ 4 <i>s</i> (³ G)4 <i>p</i>	<i>z</i> ² H°	4½	45111. 48	428. 80	0. 897		12°	1½	50105. 05		0. 569
		5½	45540. 28		1. 097						
3 <i>d</i> ⁷ 4 <i>s</i> (<i>a</i> ⁵ F)5 <i>s</i>	<i>e</i> ⁶ F	5½	45676. 00	-547. 01	1. 475	3 <i>d</i> ⁸ (¹ G)4 <i>p</i>	<i>y</i> ² H°	4½	50210. 80	165. 11	0. 899
		4½	46223. 01	-483. 82	1. 443		5½	50375. 91			1. 091
		3½	46706. 83	-383. 82	1. 396	3 <i>d</i> ⁸ (¹ G)4 <i>p</i>	<i>u</i> ² F°	3½	50578. 73	-133. 72	1. 125
		2½	47090. 65	-274. 08	1. 301		2½	50712. 45			0. 905
		1½	47364. 73	-163. 71	1. 054						
		0½	47528. 44		-0. 666	3 <i>d</i> ⁷ 4 <i>s</i> (³ H)4 <i>p</i> :	<i>w</i> ² G°	4½	50593. 38	-17. 84	1. 10
3 <i>d</i> ⁸ (³ P)4 <i>p</i> :	<i>w</i> ² D°	2½	45688. 15	-766. 80	1. 219		3½	50611. 22			0. 82
		1½	46454. 95		0. 869	3 <i>d</i> ⁷ 4 <i>s</i> (³ H)4 <i>p</i>	<i>x</i> ² H°	5½	50703. 08	-199. 53	1. 110
3 <i>d</i> ⁷ 4 <i>s</i> (³ G)4 <i>p</i>	<i>x</i> ² G°	3½	45766. 63	265. 47	0. 898		4½	50902. 61			0. 941
		4½	46032. 10		1. 131						
3 <i>d</i> ⁷ 4 <i>s</i> (³ P)4 <i>p</i>	<i>x</i> ⁴ P°	0½	45957. 29	-52. 61	2. 522		13°	{ 4½ } 50738. 20			
		1½	45904. 68	98. 15	1. 674		3½				
		2½	46002. 83		1. 543	3 <i>d</i> ⁷ 4 <i>s</i> (³ D)4 <i>p</i> :	<i>s</i> ⁴ D°	3½	50741. 66	-397. 72	1. 458
3 <i>d</i> ⁸ (<i>a</i> ³ F)5 <i>s</i>	<i>e</i> ² F	3½	45924. 98	-821. 02	1. 14		2½	51139. 38	-707. 89		
		2½	46746. 00		0. 49:		1½	51847. 27	-416. 74		
							0½	52264. 01:			
3 <i>d</i> ⁸ (³ P)4 <i>p</i>	<i>v</i> ⁴ D°	3½	45971. 19	-358. 44	1. 424		14°	2½	50806. 55		1. 136
		2½	46329. 63	69. 61	1. 365		<i>v</i> ² P°	1½	50925. 11	-20. 36	1. 340
		1½	46260. 02	-242. 13	1. 508		0½	50945. 47			0. 732
		0½	46502. 15		0. 161						
3 <i>d</i> ⁷ 4 <i>s</i> (³ P)4 <i>p</i> :	<i>v</i> ² D°	1½	46186. 41	485. 53	1. 218	3 <i>d</i> ⁸ (<i>a</i> ³ F)4 <i>d</i>	<i>e</i> ⁴ P	2½	51042. 26	-991. 00	1. 59
		2½	46671. 94		1. 233		1½	52033. 26	-882. 66		1. 40
3 <i>d</i> ⁷ 4 <i>s</i> (³ P)4 <i>p</i>	<i>y</i> ⁴ S°	1½	46562. 87		1. 273		0½	52915. 92:			
3 <i>d</i> ⁷ 4 <i>s</i> (³ P)4 <i>p</i> :	<i>y</i> ² P°	1½	46685. 43	-405. 71	1. 352	3 <i>d</i> ⁸ (<i>a</i> ³ F)4 <i>d</i>	<i>e</i> ⁴ D	3½	51052. 98	-507. 78	1. 402
		0½	47091. 14		0. 656		2½	51560. 76	-703. 73		1. 354
							1½	52264. 49:	-370. 13		1. 58:
							0½	52634. 62			
	<i>u</i> ⁴ D°	3½	46872. 74	-521. 19	1. 332	3 <i>d</i> ⁸ (<i>a</i> ³ F)4 <i>d</i>	<i>e</i> ⁴ H	6½	51142. 53	-31. 75	1. 224
		2½	47393. 93	-218. 25	1. 324		5½	51174. 28	-946. 93		1. 147
		1½	47612. 18	-293. 08	1. 122		4½	52121. 21	-595. 49		0. 96
		0½	47905. 26		0. 016		3½	52716. 70			0. 896
3 <i>d</i> ⁸ (¹ D)4 <i>p</i> :	<i>w</i> ² F°	2½	47128. 96	96. 15	0. 858	3 <i>d</i> ⁷ 4 <i>s</i> (³ D)4 <i>p</i> :	<i>w</i> ⁴ P°	2½	51160. 03	-854. 42	1. 578
		3½	47225. 11		1. 229		1½	52014. 45	-340. 67		1. 616
3 <i>d</i> ⁷ 4 <i>s</i> (<i>a</i> ⁵ F)5 <i>s</i>	<i>f</i> ⁴ F	4½	47524. 47	-677. 13	1. 328		0½	52355. 12			2. 304
		3½	48201. 60	-516. 97	1. 226						
		2½	48718. 57	-359. 86	1. 041	3 <i>d</i> ⁸ (<i>a</i> ³ F)4 <i>d</i>	<i>g</i> ⁴ F	4½	51170. 14	-29. 44	1. 337
		1½	49078. 43		0. 401		3½	51199. 58	-870. 42		1. 16
							2½	52070. 00	-632. 76		1. 08
	6°	3½?	47839. 15				1½	52702. 76			0. 76:
3 <i>d</i> ⁸ (³ P)4 <i>p</i> :	<i>y</i> ² S°	0½	47977. 94		2. 093		15°	3½	51184. 63		
3 <i>d</i> ⁷ 4 <i>s</i> (³ P)4 <i>p</i> :	<i>x</i> ² S°	0½	48026. 34		1. 699	3 <i>d</i> ⁸ (<i>a</i> ³ F)4 <i>d</i>	<i>e</i> ² P	1½	51200. 60	-840. 54	1. 368
							0½	52041. 14:			0. 48:
3 <i>d</i> ⁸ (³ P)4 <i>p</i> :	<i>x</i> ² P°	0½	48160. 43:	173. 94	1. 204	3 <i>d</i> ⁸ (<i>a</i> ³ F)4 <i>d</i>	<i>e</i> ⁴ G	5½	51203. 75	-64. 18	1. 21
		1½	48334. 37		1. 436		4½	51267. 93	-894. 09		1. 083
3 <i>d</i> ⁷ 4 <i>s</i> (³ P)4 <i>p</i>	<i>t</i> ⁴ D°	3½	48217. 32	-226. 44	1. 211		3½	52162. 02	-610. 28		1. 13
		2½	48443. 76	-102. 31	1. 340		2½	52772. 30			0. 74
		1½	48546. 07	-25. 70	1. 050						
		0½	48571. 77		0. 452						
	<i>v</i> ² F°	3½	48317. 17	-298. 39	1. 173		16°	2½	51863. 18		
		2½	48615. 56:		0. 619		<i>t</i> ² F°	3½	51896. 75:	-899. 38	1. 15:
3 <i>d</i> ⁸ (³ P)4 <i>p</i>	<i>x</i> ⁴ S°	1½	48753. 72		1. 728		2½	52796. 13			0. 883
	7°	2½	48828. 87				17°	2½	51989. 31:		

Co I—Continued

Co I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$3d^8(a^3F)4d$	f^2F	$3\frac{1}{2}$	52095. 00	-875. 62	1. 11	$3d^7 4s(a^5F)4d$	e^6P	$3\frac{1}{2}$	53789. 12	-656. 49	1. 635
		$2\frac{1}{2}$	52970. 62		1. 13			$2\frac{1}{2}$	54445. 61		1. 594
$3d^8(a^3F)4d$	e^2H	$5\frac{1}{2}$	52113. 91	-661. 56	1. 13	$3d^7 4s(a^5F)4d$	e^6H	$1\frac{1}{2}$	54949. 97	-504. 36	2. 134
		$4\frac{1}{2}$	52775. 47		0. 97			$7\frac{1}{2}$	53822. 08		1. 34
$3d^8(a^3F)4d$	e^2G	$4\frac{1}{2}$	52156. 46	-700. 22	1. 12	$3d^7 4s(a^5F)4d$	e^6H	$6\frac{1}{2}$	54452. 38	-630. 30	1. 289
		$3\frac{1}{2}$	52856. 68		0. 92			$5\frac{1}{2}$	54947. 68	-495. 30	1. 22
$3d^8(a^3F)4d$	e^2D	$2\frac{1}{2}$	52460. 10	-883. 17	0. 92	$3d^7 4s(a^5F)4d$	f^4P	$4\frac{1}{2}$	55312. 96	-365. 28	1. 108
		$1\frac{1}{2}$	53343. 27		0. 80			$3\frac{1}{2}$	55520. 64	-207. 68	0. 922
	18°	$3\frac{1}{2}$	52476. 64					$2\frac{1}{2}$	55555. 34	-34. 70	
	19°	$1\frac{1}{2}$	52498. 17					$1\frac{1}{2}$	53936. 68		1. 46
	20°	{ $2\frac{1}{2}$ $1\frac{1}{2}$ }	52526. 04					23°	$1\frac{1}{2}$	54165. 35	1. 353
$3d^7 4s(b^3F)5s$	g^2F	$3\frac{1}{2}$	52763. 68	-940. 46	0. 933	$3d^7 4s(a^5F)4d$	f^4P	$0\frac{1}{2}$			
		$2\frac{1}{2}$	53704. 14		0. 923			24°	$3\frac{1}{2}$	54398. 60	
$3d^7 4s(b^3F)5s$	h^4F	$4\frac{1}{2}$	52864. 41	-830. 16	1. 307	$3d^7 4s(a^5F)4d$	f^4P	1	$2\frac{1}{2}$	54561. 74	1. 323
		$3\frac{1}{2}$	53694. 57		1. 28:			v^4F^o	$4\frac{1}{2}$	54791. 2	-522. 8
		$2\frac{1}{2}$	54258. 75	-564. 18	0. 986			$3\frac{1}{2}$	55314. 04:	-370. 7	
		$1\frac{1}{2}$	54426. 64	-167. 89	0. 422			$2\frac{1}{2}$	55684. 7	61. 9	
	21°	{ $4\frac{1}{2}$ $3\frac{1}{2}$ }	53065. 96:					$1\frac{1}{2}$	55622. 84		
$3d^7 4s(b^3F)5s$	u^2D^o	$1\frac{1}{2}$	53074. 92	121. 06	0. 823	$3d^7 4s(a^5F)4d$	f^4G	25°	$4\frac{1}{2}$	54874. 08	
		$2\frac{1}{2}$	53195. 98		1. 206			26°	$3\frac{1}{2}$	54932. 32	
$3d^7 4s(b^3F)5s$	s^2F^o	$3\frac{1}{2}$	53103. 78	-43. 13	1. 136	$3d^7 4s(a^5F)4d$	f^4G	27°	$2\frac{1}{2}$	55061. 49	1. 539
		$2\frac{1}{2}$	53146. 91:					2	$1\frac{1}{2}$	55078. 76	1. 46:
$3d^8(1G)4p$	v^2G^o	$4\frac{1}{2}$	53276. 02	-97. 51	1. 124	$3d^7 4s(a^5F)4d$	f^4G	28°	$3\frac{1}{2}$	55120. 30:	
		$3\frac{1}{2}$	53373. 53		0. 888			3	$3\frac{1}{2}$	55223. 14	1. 14
	22°	$3\frac{1}{2}$	53463. 10					29°	$2\frac{1}{2}$	55387. 11:	1. 170
$3d^7 4s(a^5F)4d$	f^4G	$5\frac{1}{2}$	53511. 83	-646. 34	1. 274	$3d^7 4s(a^5F)4d$	f^4G	30°	$2\frac{1}{2}$	55508. 78	
		$4\frac{1}{2}$	54158. 17		1. 25:			4	$2\frac{1}{2}$	55598. 74	
		$3\frac{1}{2}$	54514. 67	-356. 50	1. 23:			31°	$1\frac{1}{2}$	55737. 87	
		$2\frac{1}{2}$	55165. 63:	-650. 96				5	{ $2\frac{1}{2}$ $1\frac{1}{2}$ }	55721. 01	
$3d^7 4s(a^5F)4d$	f^4H	$6\frac{1}{2}$	53618. 08	-697. 59	1. 227	$3d^7 4s(a^5F)4d$	f^4H	31°	$1\frac{1}{2}$	55737. 87	
		$5\frac{1}{2}$	54315. 67		1. 168			32°	$1\frac{1}{2}$	55818. 91	1. 31:
		$4\frac{1}{2}$	54860. 93	-545. 26	1. 10:			6	$0\frac{1}{2}$	55826. 81	
		$3\frac{1}{2}$	55268. 75	-407. 82	0. 857			33°	{ $2\frac{1}{2}$ $1\frac{1}{2}$ }	55922. 3	
$3d^7 4s(a^5F)4d$	f^6F	$5\frac{1}{2}$	53660. 37	-696. 08	1. 421	$3d^7 4s(a^5F)4d$	f^6F	34°	{ $2\frac{1}{2}$ $1\frac{1}{2}$ }	56101. 84:	
		$4\frac{1}{2}$	54356. 45		1. 403			35°	{ $2\frac{1}{2}$ $1\frac{1}{2}$ }	56222. 04:	
		$3\frac{1}{2}$	54896. 57	-540. 12	1. 27	$3d^8(3P)5s$	g^4P	$2\frac{1}{2}$	56545. 51:		
		$2\frac{1}{2}$	55283. 02	-386. 45	1. 17				$1\frac{1}{2}$		
		$1\frac{1}{2}$	55577. 28:	-294. 26	1. 07			$0\frac{1}{2}$			
$3d^7 4s(a^5F)4d$	f^4D	$3\frac{1}{2}$	53702. 13	-580. 60	1. 377	$3d^7 4s(a^5F)4d$	g^4H	$6\frac{1}{2}$	57922. 06	-518. 97	
		$2\frac{1}{2}$	54282. 73:					$5\frac{1}{2}$	58441. 03		-232. 70
		$1\frac{1}{2}$						$4\frac{1}{2}$	58673. 73	-641. 09	
		$0\frac{1}{2}$						$3\frac{1}{2}$	59314. 82:		
$3d^7 4s(a^5F)4d$	e^6D	$4\frac{1}{2}$	53725. 20	-627. 10	1. 387	$3d^7 4s(a^5F)4d$	g^4H	$3\frac{1}{2}$	59314. 82:		
		$3\frac{1}{2}$	54352. 30		1. 48			36°	{ $4\frac{1}{2}$ $3\frac{1}{2}$ }	58187. 39	0. 920
		$2\frac{1}{2}$	54946. 90	-594. 60	1. 47:			37°	{ $3\frac{1}{2}$ $2\frac{1}{2}$ }	59388. 89	
		$1\frac{1}{2}$	55407. 10:	-460. 20	2. 14			Limit	-----	63438	
		$0\frac{1}{2}$									
$3d^7 4s(a^5F)4d$	e^6G	$6\frac{1}{2}$	53728. 36:	-639. 07	1. 35	$3d^7 4s(a^5F)4d$	e^6G	$5\frac{1}{2}$	54367. 43	-315. 48	
		$5\frac{1}{2}$	54367. 43		1. 320			$4\frac{1}{2}$	54682. 91		-306. 71
		$4\frac{1}{2}$	54682. 91	-306. 71	1. 23			$3\frac{1}{2}$	54989. 62:	-460. 35	
		$3\frac{1}{2}$	54989. 62:	-460. 35	1. 23			$2\frac{1}{2}$	55449. 97	60. 24	
		$2\frac{1}{2}$	55449. 97	60. 24	1. 199			$1\frac{1}{2}$	55389. 73		
$3d^7 4s(a^5F)4d$	i^4F	$4\frac{1}{2}$	53788. 78	-688. 29	1. 316	$3d^7 4s(a^5F)4d$	i^4F	$3\frac{1}{2}$	54477. 07	-427. 92	
		$3\frac{1}{2}$	54477. 07		0. 85			$2\frac{1}{2}$	54904. 99		
		$2\frac{1}{2}$	54904. 99					$1\frac{1}{2}$			

March 1951.

843210°-52-8

Co I OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms			
	$ns (n \geq 4)$	$np (n \geq 4)$	$nd (n \geq 4)$	
$3d^7 4s^2$	a^4P b^2P b^2D c^2D	a^4F a^2G a^2H		
$3d^7 4s(a^3F)nx$	b, e^4F a, e^2F	y^4D° y^2D°	y^4F° y^2F°	e^4P e^2P e^4D e^2D
$3d^7 4s(a^5F)nx$	e^6F f^4F	z^6D° z^4D°	z^6F° z^4F°	e^6P f^4P e^6D f^4D
$3d^7 4s(b^3F)nx$	h^4F g^2F	x^4D° z^2D°	x^4F° z^2F°	e^4P e^2P e^4D e^2D
$3d^7 4s(a^5P)nx$		z^4S° z^4P° w^4D°	w^2F°	e^4P e^2P e^4D e^2D
$3d^7 4s(3P)nx$	b, g^4P a^2P	x^4S° y^2S° y^4P° x^2P°	v^4D° w^2D°	e^4P e^2P e^4D e^2D
$3d^7 4s(1D)nx$		z^2P° x^2D°	w^2F°	e^4P e^2P e^4D e^2D
$3d^7 4s(1G)nx$		a^2D	w^2F°	e^4P e^2P e^4D e^2D
$3d^7 4s(3G)nx$		b^2G	u^2F° v^2G°	e^4P e^2P e^4D e^2D
$3d^7 4s(3P)nx$		y^4S° z^2S° x^4P° y^2P°	w^4F° x^2F°	e^4P e^2P e^4D e^2D
$3d^7 4s(3H)nx$		x^2S° w^2P°	w^4G° x^2G°	e^4P e^2P e^4D e^2D
$3d^7 4s(3D)nx$		w^4P° s^4D°	w^2G° x^2H°	e^4P e^2P e^4D e^2D

*For predicted terms of the Co I isoelectronic sequence, see Vol. II, Introduction.

Limits with prefixes "a", "b" are known in Co II. The remaining limit terms are arranged in the order of their appearance in the term table.

Co II

(Fe I sequence; 26 electrons)

 $Z=27$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^8 {}^3F_4$ $a {}^3F_4$ 137572 cm^{-1}

I. P. 17.05 volts

In 1930 Findlay published 15 terms and 194 classified lines in the range between 1940.64 Å and 3621.22 Å. With the aid of observed Zeeman patterns he revised and extended the early work by Meggers, using Meggers' line list and some additional observations of his own. From a Rydberg formula he determined an ionization potential based on the $ns {}^{5,3}F$ series ($n=4,5$). Russell has recently improved this value by applying a revised Ritz correction to the Rydberg series. The revised value of the limit is quoted here.

A set of plates taken by Kiess at the National Bureau of Standards for wave lengths greater than 2000 Å, and a set taken in Shenstone's laboratory at Princeton for the region of wave length less than 2000 Å, have made it possible to study the spectrum further. With this material Hager has recently extended the analysis especially for inclusion here. He has provisionally submitted in manuscript 10 additional terms and nearly 100 newly classified lines from his own measurements combined with the earlier ones. His line list extends from 1226.970 Å to 5192.947 Å. This work is now in progress and will doubtless be extended appreciably in the near future. He has suggested one change in Findlay's paper, namely, that the level at 63616.0 be designated ${}^3D_2^{\circ}$ instead of ${}^3P_2^{\circ}$.

All g -values in the table except those for the $4p {}^5P^{\circ}$ term, for $z {}^5D_0^{\circ}$, and for the changed level, have been determined by Miss Weeks from films taken at the Massachusetts Institute of Technology. Her manuscript was submitted in advance of publication.

The triplet and quintet systems of terms are connected by observed intersystem combinations.

REFERENCES

- J. H. Findlay, Phys. Rev. **36**, 5 (1930). (I P) (T) (C L) (Z E)
 H. N. Russell, J. Opt. Soc. Am. **40**, 618 (1950). (I P)
 D. W. Weeks, unpublished material (March 1950). (Z E)
 N. E. Hager, Jr., unpublished material (April 1951). (T) (C L)

Co II

Co II

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$3d^8$	a^3F	4	0.0			$3d^7(a^4F)4p$	z^3G°	5	48555.9		1.19
		3	950.3	-950.3				4	49348.2	-792.3	1.111
		2	1597.2	-646.9				3	50035.9	-687.7	0.811
$3d^7(a^4F)4s$	a^5F	5	3350.5		1.406	$3d^7(a^4F)4p$	z^3F°	4	49697.5		1.197
		4	4028.9	-678.4	1.34			3	50381.6	-684.1	1.059
		3	4560.8	-531.9	1.26			2	50913.8	-532.2	0.689
		2	4950.0	-389.2	0.990						
		1	5204.5	-254.5	0.008						
$3d^7(a^4F)4s$	b^3F	4	9812.7		1.257	$3d^7(a^4F)4p$	z^3D°	3	51512.2		1.33
		3	10708.1	-895.4	1.087			2	52229.6	-717.4	1.167
		2	11321.5	-613.4	0.683			1	52684.5	-454.9	0.524
$3d^8$	a^3P	2	13261.1			$3d^7(a^4P)4p$	z^3S°	2	56010.6		2.00
		1	13404.6	-143.5							
		0	13593.5	-188.9							
$3d^7(a^4P)4s$	a^5P	3	17771.5		1.671	$3d^7(a^4P)4p$	y^5D°	4	61388.1		1.442
		2	18031.5	-260.0	1.82			3	61240.8	147.3	1.504
		1	18338.5	-307.0	2.50			2	61260.1	-19.3	1.490
$3d^7(a^4P)4s$	b^3P	2	24074.6			$3d^7(a^4P)4p$	z^3S°	1	61348.5?		0.000
		1	24267.8	-193.2				0	61457.9	-109.4	
		0	24411.5	-143.7							
$3d^7(a^2P)4s$	c^3P	2	24886.5			$3d^7(a^4P)4p$	z^3S°	1	62440.2		
		1									
		0									
$3d^7(a^2P)4s$	a^3D	3	27484.3			$3d^7(a^4P)4p$	y^3D°	3	63344.1		1.67
		2	28111.9	-627.6				2	63366.9	-22.8	1.86
		1	27585.3	526.6				1	63665.0	-298.1	2.62
$3d^6 4s^2$	a^5D	4	40694.9			$3d^6 4s(a^6D)4p$	x^5D°	4	63587.0		
		3	41313.9	-619.0				3	63616.0	-29.0	1.33
		2	41737.8	-423.9				2	63665.2	-249.2	
		1	42008.6	-270.8				1			
		0									
$3d^6 4s^2$	a^5D	4	40694.9			$3d^6 4s(a^6D)4p$	y^5F°	4	80299.2		-243.6
		3	41313.9	-619.0				3	80542.8	-245.9	
		2	41737.8	-423.9				2	80788.7	-182.7	
		1	42008.6	-270.8				1	80971.4	-195.3	
		0						0	81066.7?		
$3d^7(a^4F)4p$	z^5F°	5	45197.8		1.396	$3d^7(a^4F)4p$	e^5F	5	81970.9		-373.0
		4	45378.8	-181.0	1.407			4	82343.9	-284.3	
		3	45972.1	-593.3	1.303			3	82628.2		
		2	46452.6	-480.5	1.058			2			
		1	46786.3	-333.7	0.06			1			
$3d^7(a^4F)4p$	z^5D°	4	46320.8		1.442	$3d^7(a^4F)4p$	y^5P°	3	84012.3		-572.5
		3	47039.0	-718.2	1.43			4	84584.8	-580.5	1.34
		2	47537.1	-498.1	1.43			3	85165.3	-428.6	1.242
		1	47848.5	-311.4	1.42			2	85593.9	-280.2	
		0	47995.1	-146.6	0/0			1	85874.1		
$3d^7(a^4F)4p$	z^5G°	6	47078.2		1.350	$3d^6 4s(a^6D)4p$	e^3F	4	85044.3		-668.3
		5	47345.7	-267.5	1.260			2	85712.6	-421.8	
		4	47807.2	-461.5	1.154			1	86134.4		
		3	48150.7	-343.5	0.92						
		2	48388.1	-237.4	0.35						
						Co III ($^4F_{3/2}$)	Limit	-----	137572		

Co II OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms	
$3d^8$ $3d^6 4s^2$	a^3P	a^3F a^5D
	$ns (n \geq 4)$	$np (n \geq 4)$
$3d^7(a^4F)nx$	$\left\{ \begin{array}{l} a, e^5F \\ b, e^3F \end{array} \right.$	z^5D° z^5F° z^5G° z^3D° z^3F° z^3G°
$3d^7(a^4P)nx$		z^5S° z^5P° y^5D° z^3S° y^3D°
$3d^6 4s(a^6D)nx$		y^5P° x^5D° y^5F°
$3d^7(a^2P)nx$	c^3P	

*For predicted terms in the spectra of the Fe I isoelectronic sequence, see Vol. II, Introduction.

Co III

(Mn I sequence; 25 electrons)

$Z=27$

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^7 {}^4F_{4\frac{1}{2}}$

$a^4F_{4\frac{1}{2}}$ 270200 cm^{-1}

I. P. 33.49 volts

The analysis has been done by Shenstone especially for inclusion here. He has submitted, in manuscript form, the terms listed below. His observations extend from 650 Å to 2250 Å. There are at present more than 350 classified lines. Observed intersystem combinations connect the quartet and sextet systems of terms. His value of the ionization potential, determined from the $ns {}^6D$ series ($n=4, 5$) on the assumption that the Ritz correction, α , is the same for Co III as for Fe III, is 268963 cm^{-1} . From a study of isoelectronic data Catalán has derived an improved value, which is quoted here.

REFERENCES

- A. G. Shenstone, unpublished material (November 1951). (I P) (T) (C L)
 M. A. Catalán, unpublished material (April 1952). (I P)

Co III

Co III

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
3d ⁷	a ⁴ F	4½	0. 0		3d ⁶ (⁵ D)4p	z ⁶ F°	5½	103245. 0	
		3½	841. 2	-841. 2			4½	103386. 7	-141. 7
		2½	1451. 3	-610. 1			3½	103501. 7	-115. 0
		1½	1866. 8	-415. 5			2½	103593. 7	-92. 0
3d ⁷	a ⁴ P	2½	15201. 9		3d ⁶ (⁵ D)4p	z ⁶ P°	1½	103655. 7	-62. 0
		1½	15428. 2	-226. 3			0½	103690. 8	-35. 1
		0½	15811. 4	-383. 2			3½	105008. 7	-956. 4
							2½	105965. 1	-626. 8
3d ⁷	a ² G	4½	16977. 7		3d ⁶ (⁵ D)4p	z ⁴ D°	1½	106591. 9	
		3½	17766. 2	-788. 5			3½	106489. 1	-465. 6
3d ⁷	a ² H	5½	22720. 3		3d ⁶ (⁵ D)4p	z ⁴ D°	2½	106954. 7	-342. 3
		4½	23434. 3	-714. 0			1½	107297. 0	-210. 6
3d ⁷	a ² D	2½	23058. 8?		3d ⁶ (⁵ D)4p	z ⁴ F°	0½	107507. 6	
		1½					4½	106764. 9	-765. 2
3d ⁶ (⁵ D)4s	a ⁶ D	3½	46438. 3		3d ⁶ (⁵ D)4p	z ⁴ P°	3½	107530. 1	-522. 8
		3½	47003. 1	-564. 8			2½	108052. 9	-350. 5
		2½	47415. 4	-412. 3			1½	108403. 4	
		1½	47698. 6	-283. 2			2½	110371. 2	-590. 3
		0½	47864. 8	-166. 2			1½	110961. 5	-321. 6
3d ⁶ (⁵ D)4s	a ⁴ D	3½	55729. 2		3d ⁶ (³ P)4p	y ⁴ P°	0½	111283. 1	
		2½	56373. 8	-644. 6			2½	124597. 1?	
		1½	56794. 8	-421. 0			1½		
		0½	57036. 8	-242. 0			0½		
3d ⁶ (³ P)4s	b ⁴ P	2½	70933. 7		3d ⁶ (³ H)4p	z ⁴ G°	5½	124766. 1	-246. 3
		1½	72341. 6	-1407. 9			4½	125012. 4	-214. 6
		0½	73214. 7	-873. 1			3½	125227. 0	-141. 9
3d ⁶ (³ H)4s	a ⁴ H	6½	71623. 1		3d ⁶ (³ H)4p	z ⁴ H°	2½	125368. 9	
		5½	71873. 7	-250. 6			6½	125276. 2	-20. 0
		4½	72083. 3	-209. 6			5½	125296. 2	-125. 3
		3½	72270. 1	-186. 8			4½	125421. 5	-268. 9
3d ⁶ (³ F)4s	b ⁴ F	4½	73286. 0		3d ⁶ (³ P)4p	z ² D°	3½	125690. 4	
		3½	73540. 2	-254. 2			2½	125769. 8?	
		2½	73726. 6	-186. 4			1½		
		1½	73861. 3	-134. 7			7½	126119. 0	-356. 4
3d ⁶ (³ G)4s	a ⁴ G	5½	76518. 7		3d ⁶ (³ H)4p	z ⁴ I°	6½	126475. 4	-25. 8
		4½	77121. 4	-602. 7			5½	126501. 2	261. 8
		3½	77383. 1	-261. 7			4½	126239. 4	
		2½	77472. 0	-88. 9			3½	126549. 4	-1535. 8
3d ⁶ (³ H)4s	b ² H	5½	77411. 5		3d ⁶ (³ P)4p	y ⁴ D°	2½	128035. 2	-337. 9
		4½	77622. 9	-211. 4			1½	128423. 1	-113. 2
							0½	128536. 3	
3d ⁶ (³ F)4s	a ² F	3½	78927. 6		3d ⁶ (³ F)4p	y ⁴ F°	1½	126987. 4	-116. 7
		2½	79425. 0	-497. 4			2½	126870. 7	21. 4
3d ⁶ (³ G)4s	b ² G	4½	82363. 2		3d ⁶ (³ H)4p	z ² G°	3½	126892. 1	106. 0
		3½	82920. 4	-557. 2			4½	126998. 1	
3d ⁶ (¹ I)4s	a ² I	6½	85474. 1		3d ⁶ (³ H)4p	z ² G°	4½	127050. 8	-266. 8
		5½	85517. 3	-43. 2			3½	127317. 6	
3d ⁶ (¹ G)4s	c ² G	4½	86283. 7		3d ⁶ (³ H)4p	z ² I°	6½	127672. 8	-586. 1
		3½	86326. 9	-43. 2			5½	128258. 9	
3d ⁶ (⁵ D)4p	z ⁶ D°	4½	98290. 3		3d ⁶ (³ F)4p	x ⁴ D°	3½	128017. 8	-507. 2
		3½	98545. 6	-255. 3			2½	128525. 0	-279. 7
		2½	98823. 2	-277. 6			1½	128804. 7	-132. 2
		1½	99043. 5	-220. 3			0½	128936. 9	
		0½	99182. 2	-138. 7					

Co III—Continued

Co III—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval		
$3d^6(^3F)4p$	$y\ ^4G^\circ$	$5\frac{1}{2}$	129556. 0	-36. 3 -114. 6 -40. 0	$3d^6(^3G)4p$	$x\ ^2G^\circ$	$4\frac{1}{2}$	137661. 3	-151. 0		
		$4\frac{1}{2}$	129592. 3				$3\frac{1}{2}$	137812. 3			
		$3\frac{1}{2}$	129706. 9				$x\ ^2H^\circ$	$4\frac{1}{2}$		138920. 7	
		$2\frac{1}{2}$	129746. 9					$5\frac{1}{2}$		139137. 5	216. 8
$3d^6(^3F)4p$	$z\ ^2F^\circ$	$3\frac{1}{2}$	130183. 9	-223. 4	$3d^6(^1G)4p$	$w\ ^2G^\circ$	$4\frac{1}{2}$	140358. 3	-24. 6		
		$2\frac{1}{2}$	130407. 3				$3\frac{1}{2}$	140382. 9			
$3d^6(^3F)4p$	$y\ ^2G^\circ$	$4\frac{1}{2}$	130802. 0	-477. 4	$3d^6(^1G)4p$	$x\ ^2F^\circ$	$3\frac{1}{2}$	140646. 3	-141. 0		
		$3\frac{1}{2}$	131279. 4				$2\frac{1}{2}$	140787. 3			
$3d^6(^3H)4p$	$z\ ^2H^\circ$	$5\frac{1}{2}$	131054. 2	-483. 9	$3d^6(^1G)4p$	$w\ ^2H^\circ$	$5\frac{1}{2}$	141190. 8	-156. 4		
		$4\frac{1}{2}$	131538. 1				$4\frac{1}{2}$	141347. 2			
$3d^6(^3G)4p$	$x\ ^4G^\circ$	$5\frac{1}{2}$	131098. 1	-483. 5 -302. 2	$3d^6(^1I)4p$	$y\ ^2I^\circ$	$6\frac{1}{2}$	141868. 8	-5. 0		
		$4\frac{1}{2}$					$5\frac{1}{2}$	141873. 8			
		$3\frac{1}{2}$					131581. 6	$z\ ^4S^\circ$		$1\frac{1}{2}$	155702. 4
		$2\frac{1}{2}$					131883. 8			$x\ ^4P^\circ$	$2\frac{1}{2}$
$3d^6(^3G)4p$	$x\ ^4F^\circ$	$4\frac{1}{2}$	131887. 1	-390. 1 -211. 8 -103. 3	$3d^6(^3P')4p$	$e\ ^6D$	$4\frac{1}{2}$	169527. 7			
		$3\frac{1}{2}$	132277. 2				$3\frac{1}{2}$				
		$2\frac{1}{2}$	132489. 0				$2\frac{1}{2}$				
		$1\frac{1}{2}$	132592. 3				$1\frac{1}{2}$				
$3d^6(^3G)4p$	$y\ ^4H^\circ$	$6\frac{1}{2}$	132376. 6	-130. 0 -80. 5 -36. 4	$3d^6(^6D)5s$	$e\ ^6D$	$4\frac{1}{2}$	169527. 7			
		$5\frac{1}{2}$	132506. 6				$3\frac{1}{2}$				
		$4\frac{1}{2}$	132587. 1				$2\frac{1}{2}$				
		$3\frac{1}{2}$	132623. 5				$1\frac{1}{2}$				
$3d^6(^3G)4p$	$y\ ^2H^\circ$	$5\frac{1}{2}$	134696. 0	-708. 0	$3d^6(^3G)4p$	$y\ ^2F^\circ$	$2\frac{1}{2}$	136129. 3			
		$4\frac{1}{2}$	135404. 0				$3\frac{1}{2}$		136290. 1		
$3d^6(^3G)4p$	$y\ ^2F^\circ$	$2\frac{1}{2}$	136129. 3	160. 8	Co IV (6D_4)	Limit	-----	270200			
		$3\frac{1}{2}$	136290. 1								

November 1951.

Co III OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms	
$3d^7$	$\left\{ \begin{array}{l} a\ ^4P \\ a\ ^2D \\ a\ ^4F \\ a\ ^2G \\ a\ ^2H \end{array} \right.$	
	$ns\ (n \geq 4)$	$np\ (n \geq 4)$
$3d^6(^6D)nx$	$\left\{ \begin{array}{l} a, e\ ^6D \\ a\ ^4D \end{array} \right.$	$\left\{ \begin{array}{l} z\ ^6P^\circ \\ z\ ^4P^\circ \\ z\ ^6D^\circ \\ z\ ^4D^\circ \\ z\ ^6F^\circ \\ z\ ^4F^\circ \end{array} \right.$
$3d^6(^3P)nx$	$\left\{ b\ ^4P \right.$	$\left\{ \begin{array}{l} y\ ^4P^\circ \\ y\ ^4D^\circ \\ z\ ^2D^\circ \end{array} \right.$
$3d^6(^3H)nx$	$\left\{ \begin{array}{l} a\ ^4H \\ b\ ^2H \end{array} \right.$	$\left\{ \begin{array}{l} z\ ^4G^\circ \\ z\ ^2G^\circ \\ z\ ^4H^\circ \\ z\ ^2H^\circ \\ z\ ^4I^\circ \\ z\ ^2I^\circ \end{array} \right.$
$3d^6(^3F)nx$	$\left\{ \begin{array}{l} b\ ^4F \\ a\ ^2F \end{array} \right.$	$\left\{ \begin{array}{l} x\ ^4D^\circ \\ y\ ^4F^\circ \\ z\ ^2F^\circ \\ y\ ^4G^\circ \\ y\ ^2G^\circ \end{array} \right.$
$3d^6(^3G)nx$	$\left\{ \begin{array}{l} a\ ^4G \\ b\ ^2G \end{array} \right.$	$\left\{ \begin{array}{l} x\ ^4F^\circ \\ y\ ^2F^\circ \\ x\ ^4G^\circ \\ x\ ^2G^\circ \\ y\ ^4H^\circ \\ y\ ^2H^\circ \end{array} \right.$
$3d^6(^1I)nx$		$\left\{ \begin{array}{l} a\ ^2I \\ x\ ^2H^\circ \\ y\ ^2I^\circ \end{array} \right.$
$3d^6(^1G)nx$	$c\ ^2G$	$\left\{ \begin{array}{l} x\ ^2F^\circ \\ w\ ^2G^\circ \\ w\ ^2H^\circ \end{array} \right.$
$3d^6(^3P')nx$		$\left\{ \begin{array}{l} z\ ^4S^\circ \\ x\ ^4P^\circ \end{array} \right.$

*For predicted terms in the spectra of the Mn I isoelectronic sequence, see Vol. II, Introduction.

Co IV

(Cr I sequence; 24 electrons)

Z=27

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^5 {}^5D_4$ $a {}^5D_4$ cm^{-1}

I. P. volts

From a study of isoelectronic sequence data, and especially by analogy with Fe III, Morell, in 1928, classified 9 lines between 1349 Å and 1863 Å as combinations between $3d^5({}^6S)4s {}^7S$ and $3d^5({}^6S)4p {}^7P^o$. The present detailed analysis of Fe III indicates the urgent need of further investigation of the Co IV spectrum.

REFERENCE

L. Morell, Thesis (unpublished) Cornell University (1928). (T) (C L)

January 1950.

Co V

(V I sequence; 23 electrons)

Z=27

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^5 {}^6S_{2\frac{1}{2}}$ $a {}^6S_{2\frac{1}{2}}$ cm^{-1}

I. P. volts

This spectrum is very incompletely known. In 1931 Miss Gilroy published 9 terms, and 57 classified lines between 412 Å and 1488 Å. Later, Kruger and Gilroy reported that the three lines from the ground term having the transition $3d^5 {}^6S-4p {}^6P^o$ were near 355 Å, rather than near 414 Å, as was stated in the earlier paper. Consequently, the relative term values as published in 1931 need revision. The accepted classifications are as follows:

Int.	I. A.	Wave No.	Desig.
20	355. 523	281275	$3d^5 {}^6S_{2\frac{1}{2}}-4p {}^6P_{3\frac{1}{2}}^o$
18	355. 876	280996	$3d^5 {}^6S_{2\frac{1}{2}}-4p {}^6P_{2\frac{1}{2}}^o$
12	356. 060	280851	$3d^5 {}^6S_{2\frac{1}{2}}-4p {}^6P_{1\frac{1}{2}}^o$

REFERENCES

H. T. Gilroy, Phys. Rev. **38**, 2217 (1931). (T) (C L)P. G. Kruger and H. T. Gilroy, Phys. Rev. **48**, 720 (1935). (T) (C L)

January 1950.

Co VI

(Ti I sequence; 22 electrons)

Z=27

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^4 {}^5D_0$ $a {}^5D_0$ cm^{-1}

I. P. volts

The analysis is by Bowen who has classified 103 lines in the range between 266 Å and 306 Å. Intersystem combinations connecting the triplet and quintet systems of terms have been observed, but no series are known.

Phillips and Kruger find the line measured by Bowen at 278.184 Å as double, and classify the weaker component as $a {}^5D_3 - z {}^5F_4$. This places $z {}^5F_4$ at 360588 cm^{-1} , which is 102 cm^{-1} lower than Bowen's value. Further study of this spectrum is needed.

REFERENCES

I. S. Bowen, Phys. Rev. **53**, 889 (1938). (T) (C L)L. W. Phillips and P. G. Kruger, Phys. Rev. **54**, 839 (1938). (T) (C L)

Co VI

Co VI

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3d^4$	$a {}^3D$	0	0		$3d^3(a {}^4F)4p$	$z {}^5F^\circ$	1	358366	
		1	208	208			2	359036	670
		2	586	378			3	359826	790
		3	1129	543			4	360690	864
		4	1789	660			5	361263	573
$3d^4$	$a {}^3P$	0	27167		$3d^3(a {}^4F)4p$	$z {}^3D^\circ$	1	359903	
		1	28458	1291			2	360462	559
		2	30504	2046			3	361495	1033
$3d^4$	$a {}^3H$	4	28381		$3d^3(a {}^4F)4p$	$z {}^3G^\circ$	3	364343	
		5	28804	423			4	365083	740
		6	29233	429			5	366026	943
$3d^4$	$a {}^3F$	2	30545		$3d^3(a {}^4F)4p$	$z {}^3F^\circ$	2	367560	
		3	30635	90			3	368414	854
		4	30816	181			4	369316	902
$3d^4$	$a {}^3G$	3	34024		$3d^3(a {}^4P)4p$	$z {}^5P^\circ$	1	374556	
		4	34473	449			2	375250	694
		5	34852	379			3	376363	1113
$3d^3(a {}^4F)4p$	$z {}^5D^\circ$	0			$3d^3(a {}^2G)4p$	$y {}^2G^\circ$	3	380634	
		1	356754				4	381656	1022
		2	357279	525			5	382477	821
		3	358021	742					
		4	358928	907			$3d^3(a {}^3H)4p$	$x {}^2G^\circ$	5
			4	396420	-191				
			3	396579	-159				

January 1949.

Co VI OBSERVED TERMS*

Config $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms
$3d^4$	$\left\{ \begin{array}{ccccc} a^3P & a^5D & a^3F & a^3G & a^3H \end{array} \right.$
	$np \ (n \geq 4)$
$3d^3(a^4F)nx$	$\left\{ \begin{array}{ccc} z^5D^\circ & z^5F^\circ & \\ z^3D^\circ & z^3F^\circ & z^3G^\circ \end{array} \right.$
$3d^3(a^4P)nx$	z^5P°
$3d^3(a^2G)nx$	y^3G°
$3d^3(a^2H)nx$	x^3G°

*For predicted terms in the spectra of the Ti I isoelectronic sequence, see Vol. I, p. xxxvii.

Co VII

(Sc I sequence; 21 electrons)

$Z=27$

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^3 {}^4F_{1/2}$

$a^4F_{1/2}$

cm^{-1}

I. P.

volts

The analysis is by Anderson and Mack who have classified 133 lines in the region between 415 Å and 484 Å. They have been unable to locate with certainty the 2P term having the configuration $3d^3$, but they suggest tentatively for this term two possible pairs of levels whose reality is not yet established, as follows:

J	Level	
$1\frac{1}{2}$	28876?	29244?
$0\frac{1}{2}$	29324?	29712?

Intersystem combinations connecting the doublet and quartet systems of terms have been observed.

REFERENCE

E. E. Anderson and J. E. Mack, Phys. Rev. **59**, 717 (1941). (T) (C L)

Co VII

Co VII

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3d^3$	a^4F	$1\frac{1}{2}$	0		$3d^2(a^3F)4p$	z^4D°	$0\frac{1}{2}$	451559	
		$2\frac{1}{2}$	698	698			$1\frac{1}{2}$	451215	-344
		$3\frac{1}{2}$	1610	912			$2\frac{1}{2}$	452566	1351
		$4\frac{1}{2}$	2723	1113			$3\frac{1}{2}$	454157	1591
$3d^3$	a^4P	$0\frac{1}{2}$	21096		$3d^2(a^3F)4p$	z^3D°	$1\frac{1}{2}$	452709	
		$1\frac{1}{2}$	21304	208			$2\frac{1}{2}$	454482?	1773
		$2\frac{1}{2}$	22187	883					
$3d^3$	a^2G	$3\frac{1}{2}$	24151		$3d^2(a^3F)4p$	z^2G°	$3\frac{1}{2}$	459170	
		$4\frac{1}{2}$	25063	912			$4\frac{1}{2}$	460587	1417
$3d^3$	a^2D	$2\frac{1}{2}$	31348		$3d^2(a^3P)4p$	z^4S°	$1\frac{1}{2}$	465290	
		$1\frac{1}{2}$	31555?	-207			$3d^2(a^3P)4p$	y^4D°	$0\frac{1}{2}$
$3d^3$	a^2H	$4\frac{1}{2}$	33251						$1\frac{1}{2}$
		$5\frac{1}{2}$	33873	622			$2\frac{1}{2}$	471880	1505
							$3\frac{1}{2}$	474067	2187
$3d^2(a^3F)4p$	z^4G°	$2\frac{1}{2}$	445761		$3d^2(a^3P)4p$	z^4P°	$0\frac{1}{2}$	474575	
		$3\frac{1}{2}$	447314	1553			$1\frac{1}{2}$	475156	581
		$4\frac{1}{2}$	449305	1991			$2\frac{1}{2}$	476592	1436
		$5\frac{1}{2}$	451810?	2505					
$3d^2(a^3F)4p$	z^4F°	$1\frac{1}{2}$	446966		$3d^2(a^1G)4p$	y^2G°	$3\frac{1}{2}$	476054	
		$2\frac{1}{2}$	448051	1085			$4\frac{1}{2}$	476288	234
		$3\frac{1}{2}$	449440	1389	$3d^2(a^1G)4p$	z^2H°	$4\frac{1}{2}$	485616	
		$4\frac{1}{2}$	450989	1549			$5\frac{1}{2}$	487584	1968
$3d^2(a^3F)4p$	z^2F°	$2\frac{1}{2}$	450865						
		$3\frac{1}{2}$	452009	1144					

December 1948.

Co VII OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms
$3d^3$	a^4P a^2D a^4F a^2G a^2H
	np ($n \geq 4$)
$3d^2(a^3F)nx$	z^4D° z^4F° z^4G° z^2D° z^2F° z^2G°
	z^4S° z^4P° y^4D°
$3d^2(a^3P)nx$	
$3d^2(a^1G)nx$	y^2G° z^2H°

*For predicted terms in the spectra of the Sc I isoelectronic sequence, see Vol. I, p. xxxvi.

Co VIII

(Ca I sequence; 20 electrons)

 $Z=27$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^2 {}^3F_2$ a^3F_2 cm^{-1}

I. P. volts

Although 27 lines between 181 Å and 192 Å were reported as classified in 1933, later study of the spectra of this sequence by Edlén indicates that the early interpretation is incorrect.

REFERENCES

W. M. Cady, Phys. Rev. **43**, 325 (1933). (T) (C L)

B. Edlén, letter (Feb. 1949).

February 1949.

Co XI

(Cl I sequence; 17 electrons)

 $Z=27$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^5 {}^2P_{1/2}^{\circ}$ $3p^5 {}^2P_{1/2}^{\circ}$ **2462000** cm^{-1}

I. P. 305 volts

Edlén has classified six lines in the region between 81 Å and 84 Å, as combinations from the ground term. He has estimated the value of the limit by extrapolation along the isoelectronic sequence, as indicated by brackets in the table.

His unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCE

B. Edlén, Zeit. Phys. **104**, 407 (1937). (I P) (T) (C L)

Co XI

Config.	Desig.	J	Level	Interval
$3s^2 3p^5$	$3p^5 {}^2P^{\circ}$	$1\frac{1}{2}$ $0\frac{1}{2}$	0 19280	-19280
$3s^2 3p^4 ({}^3P) 4s$	$4s {}^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	1189920	
$3s^2 3p^4 ({}^3P) 4s$	$4s {}^2P$	$1\frac{1}{2}$ $0\frac{1}{2}$	1202070 1211730	-9660
$3s^2 3p^4 ({}^1D) 4s$	$4s' {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	1226890 1227610	-720
-----	-----	---	-----	
Co XII (3P_2)	<i>Limit</i>	---	[2462000]	

January 1948.

Co XIV

(Si I sequence; 14 electrons)

 $Z=27$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^2 \text{}^3\text{P}_0$ $3p^2 \text{}^3\text{P}_0$ cm^{-1}

I. P. volts

This spectrum has not been analyzed, but Edlén has predicted the interval between the two lowest levels ($3p^2 \text{}^3\text{P}_0 - 3p^2 \text{}^3\text{P}_1$) to be 8448 Å, or 11834 cm^{-1} .

REFERENCE

B. Edlén, *Zeit. Astroph.* **22**, 58 (1942). (C L)
October 1947.

Co XV

(Al I sequence; 13 electrons)

 $Z=27$ Ground state $1s^2 2s^2 2p^6 3s^2 3p \text{}^2\text{P}_{0\frac{1}{2}}^\circ$ $3p \text{}^2\text{P}_{0\frac{1}{2}}^\circ$ cm^{-1}

I. P. volts

This spectrum has not been analyzed, but Edlén has classified 2 lines as follows:

I. A.	Int.	Wave No.	Desig.
52. 583	1	1901750	} $3p \text{}^2\text{P}^\circ - 4d \text{}^2\text{D}$
53. 173	2	1880650	

His unit, 10^3 cm^{-1} , is here changed to cm^{-1} .

REFERENCE

B. Edlén, *Zeit. Phys.* **103**, 540 (1936). (C L)
December 1947.

Co XVI

(Mg I sequence; 12 electrons)

 $Z=27$ Ground state $1s^2 2s^2 2p^6 3s^2 {}^1S_0$ $3s^2 {}^1S_0$ 4133000 cm^{-1}

I. P. 512 volts

Edlén has classified 8 lines in the region between 49 Å and 61 Å. No singlet combinations have been observed and the triplet terms are not connected by observed combinations. He has determined the relative positions of the terms and also the ionization potential by extrapolation along the isoelectronic sequence. Estimated values are denoted by brackets in the table.

Edlén's unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCE

B. Edlén, Zeit. Phys. **103**, 536 (1936). (I P) (T) (C L)

Co XVI

Config.	Desig.	J	Level	Interval
$3s^2$	$3s^2 {}^1S$	0	0	
$3s({}^2S)3p$	$3p {}^3P^\circ$	0		17280
		1	[258310] + x	
		2	275590 + x	
$3s({}^2S)3d$	$3d {}^3D$	1		
		2		
		3	734890 + x	
$3s({}^2S)4p$	$4p {}^1P^\circ$	1	[2106020]	
$3s({}^2S)4d$	$4d {}^3D$	1	2259150 + x	840
		2	2259990 + x	
		3	2261410 + x	
$3s({}^2S)4f$	$4f {}^3F^\circ$	2		
		3		
		4	2348260 + x	
-----	-----	---	-----	
Co XVII (${}^2S_{0\frac{1}{2}}$)	<i>Limit</i>	---	[4133000]	

August 1947.

Co xvii

(Na I sequence; 11 electrons)

 $Z=27$ Ground state $1s^2 2s^2 2p^6 3s^2 S_{0\frac{1}{2}}$ $3s^2 S_{0\frac{1}{2}}$ 4410480 cm^{-1}

I. P. 547 volts

Edlén has classified 10 lines in the interval 41 Å to 58 Å, and extrapolated the absolute value of the ground term from isoelectronic sequence data. His unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCE

B. Edlén, Zeit. Phys. **100**, 621 (1936). (I P) (T) (C L)

Co xvii

Config.	Desig.	J	Level	Interval
3s	$3s^2 S$	$0\frac{1}{2}$	0	
3p	$3p^2 P^\circ$	$0\frac{1}{2}$	293550	25380
		$1\frac{1}{2}$	318930	
3d	$3d^2 D$	$1\frac{1}{2}$	721600	3630
		$2\frac{1}{2}$	725230	
4s	$4s^2 S$	$0\frac{1}{2}$	2078910	
4p	$4p^2 P^\circ$	$0\frac{1}{2}$	2196500	10080
		$1\frac{1}{2}$	2206580	
4d	$4d^2 D$	$1\frac{1}{2}$	2352650	1570
		$2\frac{1}{2}$	2354220	
4f	$4f^2 F^\circ$	$2\frac{1}{2}$	2421500	630
		$3\frac{1}{2}$	2422130	
5f	$5f^2 F^\circ$	$2\frac{1}{2}$		
		$3\frac{1}{2}$	3137080	
-----	-----	-----	-----	
Co xviii (1S_0)	Limit	-----	[4410480]	

June 1947.

Co XVIII

(Ne I sequence; 10 electrons)

 $Z=27$ Ground state $1s^2 2s^2 2p^6 {}^1S_0$ $2p^6 {}^1S_0$ 11316400 cm^{-1}

I P. 1403 volts

Tyrén has classified seven lines in the region 12 Å to 15 Å, as combinations with the ground term. His absolute term values have been derived by extrapolation along the Ne I isoelectronic sequence.

By analogy with Ne I the jl -coupling notation in the general form suggested by Racah is introduced.

The unit adopted by Tyrén, 10^3 cm^{-1} , has here been changed to cm^{-1} .

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Co XVIII

Co XVIII

Author	Config.	Desig.	J	Level	Author	Config.	Desig.	J	Level
$2p {}^1S_0$	$2s^2 2p^6$	$2p^6 {}^1S$	0	0	$3d {}^3D_1$	$2s^2 2p^5 ({}^2P_{0,1/2}) 3d$	$3d' [1\frac{1}{2}]^\circ$	1	7337300
$3s {}^3P_1$	$2s^2 2p^5 ({}^2P_{1/2}) 3s$	$3s [1\frac{1}{2}]^\circ$	2 1	6477900	$3p' {}^3P_1$	$2s 2p^6 ({}^2S) 3p$	$3p {}^3P^\circ$	2 1 0	7898300
$3s {}^1P_1$	$2s^2 2p^5 ({}^2P_{0,1/2}) 3s$	$3s' [0\frac{1}{2}]^\circ$	0 1	6592400	$3p' {}^1P_1$	$2s 2p^6 ({}^2S) 3p$	$3p {}^1P^\circ$	1	7937100
$3d {}^3P_1$	$2s^2 2p^5 ({}^2P_{1/2}) 3d$	$3d [0\frac{1}{2}]^\circ$	0 1	7124500		Co XIX (${}^2P_{1/2}$)	<i>Limit</i>	-----	[11316400]
$3d {}^1P_1$	"	$3d [1\frac{1}{2}]^\circ$	1	7214000		Co XIX (${}^2P_{0,1/2}$)	<i>Limit</i>	-----	11437600

April 1947.

NICKEL

Ni I

28 electrons

Z=28

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^8 4s^2 {}^3F_4$ $a {}^3F_4$ 61579 cm^{-1}

I. P. 7.633 volts

The analysis is by Russell, who revised and extended the earlier work on this spectrum and published 1071 classified lines in the interval between 1963 Å and 18040 Å. Subsequently, Meggers and Kiess observed the infrared region and classified 39 lines between 8580 Å and 10979 Å. Eighty-one terms are known, and observed intersystem combinations connect the terms of all three multiplicities.

Burns and Sullivan have made accurate measurements with the interferometer, of Ni I lines observed in the vacuum arc between 2173 Å and 8968 Å. Their observations confirm in detail Russell's analysis, as well as the great majority of Ni I lines identified in the solar spectrum from predicted wavelengths. They have "measured by interference some four hundred Ni I lines not hitherto observed". From these accurate wavelengths they have derived many improved values of the known energy levels. Their results are recorded in the table except for the levels $f {}^3P_1$, $e {}^5G_4$, and $g {}^5F_5$, for which the writer has derived the values quoted here. For all remaining levels, Russell's values are given.

Loyarte and Williams have observed the absorption spectrum of nickel and suggested the following rearrangement of nine levels given by Russell:

Loyarte and Williams	Russell	Level	Obs. g
$z {}^3D_3$	1_3°	40361. 24	
$z {}^3D_2$	$y {}^3P_2^{\circ}$	42653. 58	
$z {}^3D_1$	$w {}^3D_1^{\circ}$	45122. 37	0. 57
$y {}^3\bar{F}_4$	$x {}^3F_4^{\circ}$	42584. 87	1. 35
$y {}^3\bar{F}_3$	$x {}^3F_3^{\circ}?$	43654. 82	1. 24
$y {}^3\bar{F}_2$	$x {}^1D_2^{\circ}$	43933. 39	1. 48
$x {}^3\bar{F}_4$	$x {}^3D_3^{\circ}$	42620. 94	
$x {}^3\bar{F}_3$	$w {}^3D_3^{\circ}$	42767. 79	1. 22
$x {}^3\bar{F}_2$	$x {}^3D_2^{\circ}$	42954. 77	0. 840

The observed g -value for the level at 43933 fits neither designation. The observed combinations confirm $J=3$ for the level at 42620. The intensity behavior in the multiplets supports Russell's assignments, and his interpretation is quoted in the table.

The observed g -values are chiefly from the paper by Lindsley, but some are from the two other papers quoted for Zeeman effect. A theoretical discussion of some of the configuration assignments is given by Marvin.

Ni I—Continued

Russell derives the limit 61579 cm^{-1} from the 3D and 1D terms of the $3d^9(a^2D)ns$ series [$n=4,5,6$], by using a Ritz formula. He remarks that the 3D_3 and 3D_2 levels evidently converge to the lower component, $a^2D_{2\frac{1}{2}}$, of the ground term of Ni II; while 3D_1 and 1D_2 converge to $a^2D_{1\frac{1}{2}}$, which lies 1507 cm^{-1} higher. He notes that "The correlation of the term components and their limits is of the inverted type pointed out by Shenstone . . . All the terms arising from the a^2D limit of Ni II show the same inverted convergence."

In "normal" convergence only *one* level converges to that component of the limit term having the larger J -value. The *shifting* that occurs when *two* levels converge to this component, etc., as in this case, is referred to as "inverted" convergence.

Racah has pointed out that for levels from the $3d^9 nx$ configuration, the jl -coupling notation is preferable, as for the inert gases. This notation can be worked out for Ni I from the auxiliary table given in the text for Cu II.

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Ni I

Ni I

Config.	Desig.	J	Level	Interval	Obs. g	Config.	Desig.	J	Level	Interval	Obs. g														
$3d^8 4s^2$	a^3F	4	0. 000	-1332. 153 -884. 366	1. 250 1. 083 0. 671	$3d^9(a^2D)4p$	z^3P°	2	28569. 210	-931. 480 -691. 578	1. 485 1. 426														
		3	1332. 153					1	29500. 690																
		2	2216. 519					0	30192. 268																
$3d^9(a^2D)4s$	a^3D	3	204. 786	-675. 027 -833. 267	1. 332 1. 149 0. 497	$3d^9(a^2D)4p$	z^3F°	4	29481. 020	160. 238 -1298. 658	1. 287 1. 086 0. 740														
		2	879. 813					3	29320. 782																
		1	1713. 080					2	30619. 440																
$3d^9(a^2D)4s$	a^1D	2	3409. 925		1. 014	$3d^9(a^2D)4p$	z^3D°	3	29668. 918	-219. 587	1. 300														
								2	29888. 505	-1024. 333	1. 044														
								1	30912. 838		0. 552														
$3d^8 4s^2$	b^1D	2	13521. 352		1. 143	$3d^8 4s(a^2F)4p$	z^3G°	5	30922. 763	-57. 026	1. 21														
								4	30979. 789	-806. 421	1. 052														
								3	31786. 210		0. 761														
$3d^{10}$	a^1S	0	14728. 847																						
$3d^8 4s^2$	a^3P	2	15609. 861	-124. 157 -283. 299	1. 356 1. 497	$3d^9(a^2D)4p$	z^1F°	3	31031. 042		1. 048														
		1	15734. 018					$3d^9(a^2D)4p$	z^1D°	2	31441. 665	1. 060													
		0	16017. 317																						
$3d^8 4s^2$	a^1G	4	22102. 349		0. 99	$3d^8 4s(a^2F)4p$	y^3F°	4	32973. 414	-138. 954 -498. 548	1. 22 1. 193 0. 79														
		3	25753. 578	$3d^9(a^2D)4p$	z^1P°			1	32982. 280			1. 005													
		2	26665. 903										$3d^8 4s(a^2F)4p$	y^3D°	3	33500. 854	-662. 440 -245. 280	1. 198 1. 19 0. 511							
		1	27414. 893																$3d^8 4s(a^2F)4p$	z^1G°	4	33590. 159	1. 035		
		0	27943. 543																					$3d^8 4s(a^2F)4p$	y^1F°
	28212. 997	$3d^8 4s(a^2F)4p$	y^1D°			2	36600. 805			1. 013															
$3d^8 4s(a^4F)4p$	z^5G°			6	27260. 891			-319. 520 -487. 680 -509. 955 -435. 182	1. 32 1. 28 1. 171 0. 945 0. 364		$3d^8 4s(a^2F)4p$	z^1G°													
				5	27580. 411								$3d^8 4s(a^2F)4p$	y^1F°	3	35639. 148	1. 013								
				4	28068. 091													$3d^8 4s(a^2F)4p$	y^1D°	2	36600. 805	1. 013			
				3	28578. 046																		$3d^8 4s(a^2F)4p$	1°	3
		2	29013. 228	$3d^8 4s(a^2F)4p$	2°	2	40484. 282																		
$3d^8 4s(a^4F)4p$	z^5F°	5	28542. 113					-542. 365 -748. 332 -330. 330 -228. 912	1. 38 1. 288 1. 208 0. 985 0. 006		$3d^8 4s(a^2F)4p$	y^1D°													
		4	29084. 478										$3d^8 4s(a^2F)4p$	1°	3	40361. 254									
		3	29832. 810															$3d^8 4s(a^2F)4p$	2°	2	40484. 282				
		2	30163. 140																				$3d^8 4s(a^2F)4p$	1°	3
		1	30392. 052	$3d^8 4s(a^2F)4p$	2°	2	40484. 282																		

Ni I—Continued

Ni I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>			
$3d^8 4s(b^2D)4p$	x^3F°	4	42585. 296	-1069. 678	1. 35	$3d^9(a^2D)4d$	f^3D	3	49271. 578	-56. 267	1. 32			
		3	43654. 974?					2	49327. 845			-1389. 082	0. 45	
		2						1	50716. 927					
$3d^9(a^2D)5s$	e^3D	3	42605. 964	-184. 063	1. 34	$3d^9(a^2D)5p$	u^3D°	3	49327. 56	142. 41				
		2	42790. 027	-1322. 165	1. 085			2	49185. 146	-1666. 07				
		1	44112. 192					1	50851. 22					
$3d^8 4s(b^2D)4p$	x^3D°	3	42621. 048	-333. 186	0. 840	$3d^9(a^2D)4d$	e^3F	4	49332. 643	18. 792	-1520. 584			
		2	42954. 234					3	49313. 851					
		1						2	50834. 435					
$3d^8 4s(b^2D)4p$	y^3P°	2	42653. 723	-2. 594	1. 32	$3d^9(a^2D)5p$	w^1F°	3	50142. 8					
		1	42656. 317					$3d^9(a^2D)5p$	x^1P°	1	50458. 187			
		0												
$3d^8 4s(a^4F)4p$	w^3D°	3	42767. 900	-1707. 258	1. 22	$3d^8 4s(a^4F)5s$	f^3F	4	50466. 172	-839. 913	1. 27			
		2	44475. 153	-647. 302	1. 16			3	51306. 085	-734. 483	1. 08			
		1	45122. 460		0. 57			2	52040. 568					
$3d^8 4s(a^4F)4p$	y^3G°	5	43089. 636	-1225. 344	1. 23	$3d^9(a^2D)4d$	e^1P	1	50536. 742		1. 54			
		4	44314. 980	-966. 172	1. 18			$3d^9(a^2D)5p$	w^1D°	2	50689. 490			
		3	45281. 152		0. 78									
$3d^8 4s(a^4F)4p$	w^3F°	4	43258. 792	-1306. 31	1. 25	$3d^9(a^2D)4d$	e^1G	4	50706. 310		1. 02			
		3	44565. 10	-853. 76	1. 04			$3d^9(a^2D)4d$	f^1D	2	50754. 137			
		2	45418. 858		0. 68									
$3d^8 4s(b^2D)4p$	y^1P°	1	43464. 019		1. 39	$3d^8 4s(a^2G)4p$	u^3F°	4	50789. 5	-335. 3				
$3d^8 4s(b^2D)4p$	x^1D°	2	43933. 428		1. 48			3	51124. 8	-219. 0				
$3d^8 4s(b^2D)4p$	x^1F°	3	44206. 185					2	51343. 80					
$3d^9(a^2D)5s$	e^1D	2	44262. 619		1. 09	$3d^9(a^2D)4d$	e^1S	0	51457. 285					
		3°		44336. 10					$3d^9(a^2D)6s$	g^3D	3	52197. 482	-74. 234	
			2	46522. 965	-685. 263						2	52271. 716	-1432. 183	
$3d^8 4s(a^4P)4p$	x^3P°	1	47208. 228	-478. 397		$3d^9(a^2D)6s$	g^1D	2	53754. 036					
		0	47686. 625?					$3d^8 4s(a^2F)5s$	g^3F	4	54237. 136	-14. 217	1. 27	
					3					54251. 353	-1622. 43			
$3d^8 4s(a^4P)4p$	v^3D°	3	47030. 148	-109. 244	1. 331	$3d^9(a^2D)5d$	f^3S	1	54574. 64					
		2	47139. 392	-285. 438	1. 209			$3d^9(a^2D)5d$	f^3G	5	54659. 759	-8. 169	1. 03	
		1	47424. 830		0. 726					4	54667. 928	-1504. 776		
$3d^8 4s(a^4F)5s$	e^5F	5	48466. 530	-619. 500	1. 40	$3d^9(a^2D)5d$	h^3D	3	54699. 852	-32. 573				
		4	49086. 030	-691. 589	1. 33			2	54732. 425?					
		3	49777. 619	-568. 858	1. 23			1						
$3d^9(a^2D)5p$	v^3F°	4	48715. 2	43. 3		$3d^9(a^2D)5d$	h^3F	4	54761. 346	-11. 594				
		3	48671. 9	-1367. 3				3	54772. 940	-1501. 576				
		2	50039. 18					2	56274. 516					
$3d^9(a^2D)5p$	w^3P°	2	48735. 308	-668. 11		$3d^8 4s(a^2F)5s$	f^1F	3	55576. 905					
		1	49403. 42	-735. 11				$3d^9(a^2D)5d$	f^1G	4	56183. 51			
		0	50138. 53							3	56262. 92			
$3d^9(a^2D)4d$	e^3S	1	48953. 344		1. 92	$3d^8 4s(a^4F)4d$	e^3H	6	56624. 668	-1052. 981				
		6°		49032. 589					5	57677. 649	-840. 46			
									4	58518. 11				
$3d^9(a^2D)4d$	e^3G	5	49158. 529	-16. 282	1. 20	$3d^8 4s(a^4F)4d$	f^3P	2	56710. 889	-1056. 63				
		4	49174. 811	-1502. 788	1. 05			1	57767. 52	-681. 27				
		3	50677. 599		0. 77			0	58448. 79					
$3d^9(a^2D)4d$	e^3P	2	49159. 060	-12. 127	1. 43	$3d^8 4s(a^4F)4d$	i^3F	4	56766. 523	-1201. 56				
		1	49171. 187	-1105. 167	1. 00			3	57968. 08	-661. 76				
		0	50276. 354					2	58629. 84					

Ni I—Continued

Config.	Desig.	J	Level	Interval	Obs. g
3d ⁸ 4s(a ⁴ F)4d	g ³ G	5	56801. 654	—987. 957	
		4	57789. 611	—740. 74	
		3	58530. 35		
3d ⁸ 4s(a ⁴ F)4d	e ⁵ P	3	56821. 553	—765. 1	
		2	57586. 7	—938. 8	
		1	58525. 507		
3d ⁸ 4s(a ⁴ F)4d	e ⁵ D	4	56857. 933	—885. 663	
		3	57743. 596		
		2			
		1			
		0			
3d ⁸ 4s(a ⁴ F)4d	e ⁵ H	7	56885. 249	—876. 857	1. 26
		6	57762. 106	—758. 817	
		5	58520. 923	—518. 770	
		4	59039. 693	—149. 09	
		3	59188. 78		
3d ⁸ 4s(a ⁴ F)4d	e ⁵ G	6	56954. 167	—875. 238	
		5	57829. 405	—1043. 16	
		4	58872. 57	243. 02	
		3	58629. 55	—488. 51	
		2	59118. 06		
3d ⁸ 4s(a ⁴ F)4d	f ⁵ F	5	56973. 707	—836. 787	
		4	57810. 494	—777. 674	
		3	58588. 168	—404. 35	
		2	58992. 52	—233. 51	
		1	59226. 03		
3d ⁸ 4s(a ⁴ F)4d	i ³ D	3	57103. 946		
		1			
3d ⁸ 4s(a ⁴ F)6s	g ⁵ F	5	59862. 69		
		4			
		3			
		2			
		1			
Ni II (a ² D _{2½})	Limit		61579		
3d ⁸ 4s(a ² F)4d	j ³ F	4	61832. 47		
		3			
		2			
3d ⁸ 4s(a ² F)4d	h ³ G	5	61843. 28		
		4			
		3			
3d ⁸ 4s(a ² F)4d	f ³ H	6	61957. 517		
		5			
		4			
3d ⁸ 4s(a ⁴ F)5d	f ⁵ H	7	62782. 614		
		6			
		5			
		4			
		3			
3d ⁸ 4s(a ⁴ F)5d	f ⁵ G	6	62808. 03		
		5			
		4			
		3			
		2			
3d ⁸ 4s(a ⁴ F)5d	h ⁵ F	5	62815. 34?		
		4			
		3			
		2			
		1			

Ni I OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ +	Observed Terms		
	ns (n ≥ 4)	np (n ≥ 4)	nd (n ≥ 4)
3d ⁸ 4s ²	a ³ P		
3d ¹⁰	a ¹ S		
3d ⁸ (a ² D)nz	a ³ F	z, w ³ P ^o z, v ³ F ^o z, w ¹ F ^o	e, f ³ G e, f ¹ G
	b ¹ D	z ⁵ D ^o w ³ D ^o	f, h ³ D f ¹ D
3d ⁸ 4s(a ⁴ F)nz		z ⁵ F ^o w ³ F ^o	e, h ³ F e, g ¹ F
		y ³ D ^o y ¹ D ^o	f, h ⁵ F i ³ F
3d ⁸ 4s(a ² F)nz		x ³ P ^o y ¹ P ^o	e ⁵ D i ³ D
		x ³ D ^o x ¹ D ^o	e, f ⁵ H e ³ H
3d ⁸ 4s(b ² D)nz		y ³ P ^o y ¹ P ^o	f ³ F h ³ G
		x ³ P ^o v ³ D ^o	j ³ F f ³ H
3d ⁸ 4s(a ⁴ P)nz			
3d ⁸ 4s(a ² G)nz			

*For predicted terms in the spectra of the Ni I isoelectronic sequence, see Vol. II, Introduction.

Ni II

(Co I sequence; 27 electrons)

Z=28

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^9 {}^2D_{3/2}$ $a {}^2D_{3/2}$ 146408 cm^{-1}

I. P. 18.15 volts

The analysis is chiefly by Shenstone who has classified 242 lines in the interval from 1812 Å to 4362 Å, from measurements by Meggers between 3500 Å and 2150 Å, supplemented by his own observations in the short wave region. Further observations in the ultraviolet were needed to locate the ground term, $a {}^2D$. This term was found independently by both Lang and Menzies, who classified about 50 lines between 1252 Å and 1952 Å. The discovery of this term indicated that 8391 cm^{-1} (Lang) or 8392.9 (Menzies) should be added to all of Shenstone's values. An improved value by Shenstone, 8394.1 , has been adopted here.

Intersystem combinations connecting the doublet and quartet systems of terms, have been observed.

Shenstone has calculated the limit from the two levels $a, e {}^4F_{4/2}$, by assuming a Rydberg series. This is known to be too high. The correction derived from longer series in related spectra has been estimated by H. N. Russell as about -2792 cm^{-1} . This adjustment has been made in the limit quoted here.

The listed g -values are from Lindsley; they are derived from measurements of Zeeman patterns of 128 lines observed at the Massachusetts Institute of Technology.

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Ni II

Ni II

Shenstone	Config.	Desig.	J	Level	Interval	Obs. g	Shenstone	Config.	Desig.	J	Level	Interval	Obs. g
	$3d^9$	$a {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	0. 0 1506. 9	-1506. 9		$a {}^4D'_4$ $a {}^4D'_3$ $a {}^4D'_2$ $a {}^4D'_1$	$3d^8({}^3F)4p$	$z {}^4D^\circ$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	51558. 1 52738. 6 53635. 1 54176. 1	-1180. 5 -896. 5 -541. 0	1. 420 1. 356 1. 186 -0. 005
$a {}^4F'_5$ $a {}^4F'_4$ $a {}^4F'_3$ $a {}^4F'_2$	$3d^8({}^3F)4s$	$a {}^4F$	$4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$	8394. 1 9330. 5 10115. 9 10664. 2	-936. 4 -785. 4 -548. 3	1. 35 1. 24 1. 023 0. 397	$a {}^4G'_6$ $a {}^4G'_5$ $a {}^4G'_4$ $a {}^4G'_3$	$3d^8({}^3F)4p$	$z {}^4G^\circ$	$5\frac{1}{2}$ $4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$	53496. 8 53365. 2 54262. 7 55018. 8	131. 6 -897. 5 -756. 1	1. 305 1. 156 1. 02 0. 616
$a {}^2F'_4$ $a {}^2F'_3$	$3d^8({}^3F)4s$	$a {}^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	13550. 3 14995. 6	-1445. 3	1. 141 0. 866	$a {}^4F_5$ $a {}^4F_4$ $a {}^4F_3$ $a {}^4F_2$	$3d^8({}^3F)4p$	$z {}^4F^\circ$	$4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$	54557. 3 55417. 9 56075. 2 56424. 6	-860. 6 -657. 3 -349. 4	1. 26 1. 184 0. 985 0. 412
$a {}^2D_3$ $a {}^2D_2$	$3d^8({}^1D)4s$	$b {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	23108. 1 23796. 1	-688. 0	1. 428 1. 045	$a {}^2G'_5$ $a {}^2G'_4$	$3d^8({}^3F)4p$	$z {}^2G^\circ$	$4\frac{1}{2}$ $3\frac{1}{2}$	55300. 0 56371. 6	-1071. 6	1. 152 0. 940
$a {}^4P'_1$ $a {}^4P'_2$ $a {}^4P'_3$	$3d^8({}^3P)4s$	$a {}^4P$	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	24835. 9 24788. 1 25035. 8	-47. 8 247. 7	2. 667 1. 498 1. 368	$a {}^2F_4$ $a {}^2F_3$	$3d^8({}^3F)4p$	$z {}^2F^\circ$	$3\frac{1}{2}$ $2\frac{1}{2}$	57080. 3 58493. 0	-1412. 7	1. 154 0. 946
$a {}^2P'_2$ $a {}^2P'_1$	$3d^8({}^3P)4s$	$a {}^2P$	$1\frac{1}{2}$ $0\frac{1}{2}$	29070. 6 29593. 1	-522. 5	1. 322 0. 670	$a {}^2D'_3$ $a {}^2D'_2$	$3d^8({}^3F)4p$	$z {}^2D^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$	57419. 7 58705. 6	-1285. 9	1. 116 0. 795
$a {}^2G_5$ $a {}^2G_4$	$3d^8({}^1G)4s$	$a {}^2G$	$4\frac{1}{2}$ $3\frac{1}{2}$	32499. 4 32523. 4	-24. 0	1. 135 0. 895							

Ni II—Continued

Ni II—Continued

Shenstone	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Shenstone	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>				
<i>b</i> ⁴ P ₃	3 <i>d</i> ⁸ (³ P)4 <i>p</i>	<i>z</i> ⁴ P ^o	2½	66570.9	-8.6	1.48	<i>c</i> ² S' ₁	3 <i>d</i> ⁸ (³ P)4 <i>p</i>	<i>z</i> ² S ^o	0½	74283.0						
<i>b</i> ⁴ P ₂			1½	66579.5		1.550	<i>b</i> ⁴ S' ₂							<i>z</i> ⁴ S ^o	1½	74300.3	
<i>b</i> ⁴ P ₁			0½	67030.7		2.331											
<i>b</i> ² F ₃	3 <i>d</i> ⁸ (¹ D)4 <i>p</i>	<i>y</i> ² F ^o	2½	67694.2	436.8	0.960	<i>b</i> ² H ₅	3 <i>d</i> ⁸ (¹ G)4 <i>p</i>	<i>z</i> ² H ^o	4½	75149.6	572.0	0.903				
<i>b</i> ² F ₄			3½	68131.0		1.200	<i>b</i> ² H ₆							5½	75721.6	1.119	
<i>b</i> ² D' ₂	3 <i>d</i> ⁸ (¹ D)4 <i>p</i>	<i>y</i> ² D ^o	1½	68154.1	581.2	1.02	<i>c</i> ² F ₃	3 <i>d</i> ⁸ (¹ G)4 <i>p</i>	<i>x</i> ² F ^o	2½	75889.5?	27.8	1.16				
<i>b</i> ² D ₃			2½	68735.3		1.26	<i>c</i> ² F ₄							3½	75917.3?		
<i>b</i> ² P ₁	3 <i>d</i> ⁸ (¹ D)4 <i>p</i>	<i>z</i> ² P ^o	0½	68281.4	684.1	1.008	<i>b</i> ² G' ₄	3 <i>d</i> ⁸ (¹ G)4 <i>p</i>	<i>y</i> ² G ^o	3½	79822.7	101.1					
<i>b</i> ² P ₂			1½	68965.5		1.305	<i>b</i> ² G' ₅							4½	79923.8		
<i>b</i> ⁴ D' ₁	3 <i>d</i> ⁸ (³ P)4 <i>p</i>	<i>y</i> ⁴ D ^o	0½	70748.5	-41.9	1.190	<i>d</i> ⁴ F' ₅	3 <i>d</i> ⁸ (³ F)5 <i>s</i>	<i>e</i> ⁴ F	4½	91798.5	-525.4	1.350				
<i>b</i> ⁴ D ₂			1½	70706.6			-71.4							<i>d</i> ⁴ F' ₄	3½	92323.9	1.188
<i>b</i> ⁴ D ₃			2½	70635.2			142.2							<i>d</i> ⁴ F' ₃	2½	93388.0	1.02
<i>b</i> ⁴ D ₄			3½	70777.4			1.38							<i>d</i> ⁴ F' ₂	1½	94065.1	0.39
<i>c</i> ² D' ₃	3 <i>d</i> ⁸ (³ P)4 <i>p</i>	<i>x</i> ² D ^o	2½	71770.9	-604.2	1.240	<i>d</i> ² F' ₄	3 <i>d</i> ⁸ (³ F)5 <i>s</i>	<i>e</i> ² F	3½	93526.3	-1200.8	1.166				
<i>c</i> ² D ₂			1½	72375.1		0.844	<i>d</i> ² F' ₃							2½	94727.1	0.865	
<i>c</i> ² P ₂	3 <i>d</i> ⁸ (³ P)4 <i>p</i>	<i>y</i> ² P ^o	1½	72985.2	-917.5	1.326											
<i>c</i> ² P ₁			0½	73902.7		1.039				Ni III (³ F ₄)	<i>Limit</i>	---	146408				

October 1949.

Ni II OBSERVED TERMS*

Config. 1 <i>s</i> ² 2 <i>s</i> ² 2 <i>p</i> ⁶ 3 <i>s</i> ² 3 <i>p</i> ⁶ +	Observed Terms			
3 <i>d</i> ⁰	<i>a</i> ² D			
	<i>ns</i> (<i>n</i> ≥ 4)		<i>np</i> (<i>n</i> ≥ 4)	
	3 <i>d</i> ⁸ (³ F) <i>nx</i>	{ <i>a, e</i> ⁴ F <i>a, e</i> ² F	<i>z</i> ⁴ D ^o	<i>z</i> ⁴ F ^o
<i>z</i> ² D ^o			<i>z</i> ² F ^o	<i>z</i> ² G ^o
3 <i>d</i> ⁸ (³ P) <i>nx</i>	{ <i>a</i> ⁴ P <i>a</i> ² P	<i>z</i> ⁴ S ^o	<i>z</i> ⁴ P ^o	<i>y</i> ⁴ D ^o
		<i>z</i> ² S ^o	<i>y</i> ² P ^o	<i>x</i> ² D ^o
3 <i>d</i> ⁸ (¹ G) <i>nx</i>		<i>a</i> ² G	<i>x</i> ² F ^o	<i>y</i> ² G ^o
3 <i>d</i> ⁸ (¹ D) <i>nx</i>		<i>b</i> ² D	<i>z</i> ² P ^o	<i>y</i> ² D ^o
			<i>y</i> ² F ^o	<i>z</i> ² H ^o

*For predicted terms in the spectra of the Co I isoelectronic sequence, see Vol. II, Introduction.

Ni III

(Fe I sequence; 26 electrons)

Z=28

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^8 {}^3F_4$ $a {}^3F_4$ 283700 cm^{-1}

I. P. 36.16 volts

This spectrum has been analyzed by Shenstone especially for inclusion here. His observations cover the range 700 Å to 3000 Å. There are 165 classified lines, and the triplet and quintet terms are connected by observed intersystem combinations.

Shenstone has determined the limit from the $ns {}^5F$ series ($n=4,5$) by using a Ritz formula and assuming $\alpha=0.17 \times 10^{-6}$, to be 284400 cm^{-1} . From a study of isoelectronic data Catalán has derived an improved value, which is quoted here.

REFERENCES

A. G. Shenstone, unpublished material (May 1950). (I P) (T) (C L)

M. A. Catalán, unpublished material (April 1952). (I P).

Ni III

Ni III

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval				
$3d^8$	$a {}^3F$	4	0.0		$3d^7({}^4F)4p$	$z {}^3D^\circ$	3	118745.7					
		3	1360.7	-1360.7			2	119670.0	-924.3				
		2	2269.6	-908.9			1	120272.7	-602.7				
$3d^8$	$a {}^3P$	2	16661.6	-316.2	$3d^7({}^4P)4p$	$z {}^5S^\circ$	2	122282.3					
		1	16977.8	-252.9			$y {}^5D^\circ$	0					
		0	17230.7					1	129957.6	-44.8			
$3d^7({}^4F)4s$	$a {}^5F$	5	53703.7	-954.2	$3d^7 4p$	1°	5, 4	131500.3					
		4	54657.9	-748.2			2	129912.8	41.1				
		3	55406.1	-545.9			3	129953.9	358.4				
		2	55952.0	-356.2			4	130312.3					
		1	56308.2										
$3d^7({}^4F)4s$	$b {}^3F$	4	61339.3	-1267.0	$3d^7({}^2G)4p$	$y {}^3F^\circ$	4	131792.4	-1365.9				
		3	62606.3	-866.3			3	133158.3	-1074.4				
		2	63472.6				2	134232.7					
$3d^7({}^4P)4s$	$a {}^5P$	3	71067.0	-316.7	$3d^7 4p$	2°	4	132156.6					
		2	71383.7	-458.2			$3d^7 4p$	3°	3	133095.7			
		1	71841.9										
$3d^7({}^2G)4s$	$a {}^3G$	5	75123.8	-523.1	$3d^7 4p$	4°	2, 1	133276.8					
		4	75646.9	-590.9			$3d^7 4p$	5°	5, 4	133325.1			
		3	76237.8										
$3d^7({}^4P)4s$	$b {}^3P$	2	79143.3		$3d^7({}^4P)4p$	$z {}^5P^\circ$	3	133390.8	109.4				
		1					2	133500.2					
		0					1						
$3d^7({}^4F)4p$	$z {}^5F^\circ$	5	110212.4	-158.7	$3d^7({}^2G)4p$	$y {}^3G^\circ$	5	133691.8	-723.6				
		4	110371.1	-849.8			4	134415.4					
		3	111220.9	-693.2			3						
		2	111914.1	-487.4			$3d^7 4p$	6°	5, 4	134217.8			
		1	112401.5										
$3d^7({}^4F)4p$	$z {}^5D^\circ$	4	111898.5	-1036.8	$3d^7 4p$	7°	3, 2	134334.9					
		3	112935.3	-715.7			$3d^7 4p$	8°	3	136967.4			
		2	113651.0	-444.4					$3d^7 4p$	9°	3	138487.5	
		1	114095.4	-199.8									
		0	114295.2										
$3d^7({}^4F)4p$	$z {}^5G^\circ$	6	112787.4	-353.4	$3d^7 4p$	10°	2, 1	140885.5					
		5	113140.8	-564.2			$3d^7({}^4F)5s$	$e {}^5F$	5	181018.7			
		4	113705.0	-404.9					4				
		3	114109.9	-260.9					3				
		2	114370.8						2				
$3d^7({}^4F)4p$	$z {}^3G^\circ$	5	115272.5	-1402.0	$3d^7({}^4F)5s$	$e {}^3F$	4	183053.1					
		4	116674.5	-868.6			3						
		3	117543.1				2						
$3d^7({}^4F)4p$	$z {}^3F^\circ$	4	116192.1	-1058.9	Ni IV(${}^4F_{4\frac{1}{2}}$)	Limit	-----	283700					
		3	117251.0	-864.3									
		2	118115.3										

Ni III OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms					
$3d^8$	a^3P	a^3F				
	$ns (n \geq 4)$		$np (n \geq 4)$			
$3d^7(^4F)nx$	$\left\{ \begin{array}{l} a, e^5F \\ b, e^3F \end{array} \right.$		z^5D°	z^5F°	z^5G°	
$3d^7(^4P)nx$			z^3D°	z^3F°	z^3G°	
$3d^7(^2G)nx$	$\left\{ \begin{array}{l} a^5P \\ b^3P \end{array} \right.$	z^5S°	z^5P°	y^5D°		
		a^3G			y^3F°	y^3G°

*For predicted terms in the spectra of the Fe I isoelectronic sequence, see Vol. II, Introduction.

Ni V

(Cr I sequence; 24 electrons)

Z=28

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^6 ^5D_4$

a^5D_4 cm⁻¹

I. P. volts

From a study of isoelectronic sequence data, and especially by analogy with Fe III, Morell in 1928 classified 10 lines between 1123 Å and 1501 Å as combinations between $3d^5(^6S)4s^7.^5S$ and $3d^5(^6S)4p^7.^5P^\circ$. The present detailed analysis of Fe III indicates the urgent need of further investigation of the Ni V spectrum.

REFERENCE

L. Morell, Thesis (unpublished) Cornell University (1928). (T) (C L)
January 1950.

Ni VI

(V I sequence; 23 electrons)

Z=28

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^5 ^6S_{2\frac{1}{2}}$

$a^6S_{2\frac{1}{2}}$ cm⁻¹

I. P. volts

This spectrum is very incompletely known. In 1931 Miss Gilroy published 6 terms, and 25 classified lines between 844 Å and 1191 Å. Later, Kruger and Gilroy classified three lines as transitions from the ground term to $3d^4 4p ^6P^\circ$ as follows:

Int.	I. A.	Wave No.	Desig.
400	260.348	384101	$3d^5 ^6S_{2\frac{1}{2}} - 4p ^6P_{3\frac{1}{2}}$
300	260.591	383743	$3d^5 ^6S_{2\frac{1}{2}} - 4p ^6P_{2\frac{1}{2}}$
250	260.713	383564	$3d^5 ^6S_{2\frac{1}{2}} - 4p ^6P_{1\frac{1}{2}}$

The later work indicates that the terms published in 1931 need revision.

REFERENCES

H. T. Gilroy, Phys. Rev. **38**, 2217 (1931). (T) (CL)
P. G. Kruger and H. T. Gilroy, Phys. Rev. **48**, 720 (1935). (T) (C L)

January 1950.

Ni VII

(Ti I sequence; 22 electrons)

 $Z=28$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^4 {}^5D_0$ $a {}^5D_0$ cm^{-1}

I. P. volts

The analysis is incomplete. Phillips and Kruger have classified 92 lines in the range between 205 Å and 229 Å. Intersystem combinations connecting the triplet and quintet systems of terms have been observed. No series have been found.

REFERENCE

L. W. Phillips and P. G. Kruger, Phys. Rev. **54**, 839 (1938). (T) (C L)

Ni VII

Ni VII

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3d^4$	$a {}^5D$	0	0		$3d^3(a {}^4F)4p$	$z {}^5F^\circ$	1	469191	
		1	275	275			2	470075	884
		2	802	527			3	471115	1040
		3	1520	718			4	472074	959
		4	2379	859			5	472930	856
$3d^4$	$a {}^3P$	0	29334		$3d^3(a {}^4F)4p$	$z {}^3D^\circ$	1	470528	
		1	31100	1766			2	471258	730
		2	33706	2606			3	472514	1256
$3d^4$	$a {}^3H$	4	31052		$3d^3(a {}^4F)4p$	$z {}^3G^\circ$	3	475657	
		5	31668	616			4	476659	1002
		6	32267	599			5	477936	1277
$3d^4$	$a {}^3F$	2	33592		$3d^3(a {}^4F)4p$	$z {}^3F^\circ$	2	479348	
		3	33705	113			3	480494	1146
		4	33961	256			4	481639	1145
$3d^4$	$a {}^3G$	3	37551		$3d^3(a {}^4P)4p$	$z {}^5P^\circ$	1	486170	
		4	38141	590			2	487228	1058
		5	38638	497			3	488673	1445
$3d^4(a {}^4F)4p$	$z {}^5D^\circ$	0			$3d^3(a {}^2G)4p$	$y {}^3G^\circ$	3	493139	
		1	467048				4	494496	1357
		2	467730	682			5	495747	1251
		3	468717	987					
		4	469870	1153	$3d^3(a {}^2H)4p$	$x {}^3G^\circ$	5	510873	
			4	511101			-228		
			3	511402			-301		

January 1949.

Ni VII OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms
$3d^4$	$\left\{ \begin{array}{ccccc} a^5D & & & & \\ a^3P & & a^3F & a^3G & a^3H \end{array} \right.$
	$np \ (n \geq 4)$
$3d^3 (a^4F) nx$	$\left\{ \begin{array}{ccc} z^5D^{\circ} & z^5F^{\circ} & \\ z^3D^{\circ} & z^3F^{\circ} & z^3G^{\circ} \end{array} \right.$
$3d^3 (a^4P) nx$	z^5P°
$3d^3 (a^2G) nx$	y^3G°
$3d^3 (a^2H) nx$	x^3G°

*For predicted terms in the spectra of the Ti I isoelectronic sequence, see Vol. I, p. xxxvii.

Ni VIII

(Sc I sequence; 21 electrons)

$Z=28$

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^3 {}^4F_{1\frac{1}{2}}$

$a^4F_{1\frac{1}{2}} \quad \text{cm}^{-1}$

I. P. volts

The analysis is by Anderson and Mack, who have classified 132 lines in the region between 530 Å and 611 Å. They have been unable to locate with certainty the 2P term having the configuration $3d^3$, but suggest tentatively for this term two possible pairs of levels whose reality is not yet established, as follows:

J	Level	
$1\frac{1}{2}$	31515?	32261?
$0\frac{1}{2}$	32089?	32939?

Intersystem combinations connecting the doublet and quartet systems of terms have been observed.

REFERENCE

E. E. Anderson and J. E. Mack, Phys. Rev. **59**, 717 (1941). (T) (C L)

Ni VIII

Ni VIII

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3d^3$	a^4F	$1\frac{1}{2}$	0	1012	$3d^2(a^3F)4p$	z^2D°	$1\frac{1}{2}$	569667	2178
		$2\frac{1}{2}$	1012				$2\frac{1}{2}$	571845	
		$3\frac{1}{2}$	2184				$0\frac{1}{2}$	570353	
		$4\frac{1}{2}$	3721				$1\frac{1}{2}$	569839	
$3d^3$	a^4P	$0\frac{1}{2}$	23261	449	$3d^2(a^3F)4p$	z^4D°	$2\frac{1}{2}$	571517	-514
		$1\frac{1}{2}$	23710				$1\frac{1}{2}$	569839	
		$2\frac{1}{2}$	24669				$2\frac{1}{2}$	571517	
$3d^3$	a^2G	$3\frac{1}{2}$	26977	1091	$3d^2(a^3F)4p$	z^2F°	$3\frac{1}{2}$	570546	1258
		$4\frac{1}{2}$	28068				$3\frac{1}{2}$	571804	
$3d^3$	a^2D	$2\frac{1}{2}$	34689	-431	$3d^2(a^3F)4p$	z^2G°	$3\frac{1}{2}$	581337	1904
		$1\frac{1}{2}$	35120				$4\frac{1}{2}$	583241	
$3d^3$	a^2H	$4\frac{1}{2}$	36754	721	$3d^2(a^3P)4p$	z^4S°	$1\frac{1}{2}$	587305	
		$5\frac{1}{2}$	37475				$0\frac{1}{2}$	590764	
$3d^2(a^3F)4p$	z^4G°	$2\frac{1}{2}$	565124	1840	$3d^2(a^3P)4p$	y^4D°	$1\frac{1}{2}$	592175	1411
		$3\frac{1}{2}$	566964				$2\frac{1}{2}$	594068	
		$4\frac{1}{2}$	569564				$3\frac{1}{2}$	596908	
		$5\frac{1}{2}$	572969				$0\frac{1}{2}$	596770	
$3d^2(a^3F)4p$	z^4F°	$1\frac{1}{2}$	565388	1443	$3d^2(a^3P)4p$	z^4P°	$1\frac{1}{2}$	596905?	135
		$2\frac{1}{2}$	566831				$2\frac{1}{2}$	598570	
		$3\frac{1}{2}$	568746				$1\frac{1}{2}$	598638	
		$4\frac{1}{2}$	570960				$4\frac{1}{2}$	599079	
$3d^2(a^3F)4p$	z^4F°	$3\frac{1}{2}$	568746	1915	$3d^2(a^1G)4p$	y^2G°	$3\frac{1}{2}$	598638	441
		$4\frac{1}{2}$	570960				$4\frac{1}{2}$	599079	
$3d^2(a^3F)4p$	z^4F°	$4\frac{1}{2}$	570960	2214	$3d^2(a^1G)4p$	z^2H°	$4\frac{1}{2}$	613417	2308
		$5\frac{1}{2}$	570960				$5\frac{1}{2}$	615725	

December 1948.

Ni VIII OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms					
$3d^3$	{	a^4P	a^2D	a^4F	a^2G	a^2H
		$np (n \geq 4)$				
$3d^2(a^3F)nx$	{	z^4D°	z^4F°	z^4G°		
		z^2D°	z^2F°	z^2G°		
$3d^2(a^3P)nx$	{	z^4S°	z^4P°	y^4D°		
$3d^2(a^1G)nx$	{			y^2G°	z^2H°	

*For predicted terms in the spectra of the Sc I isoelectronic sequence, see Vol. I, p. xxxvi.

Ni IX

(Ca I sequence; 20 electrons)

Z=28

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^2 {}^3F_2$ $a {}^3F_2$ cm^{-1}

I. P. volts

Although 20 lines between 146 Å and 154 Å were reported as classified in 1933, later study of the spectra of this sequence by Edlén indicates that the early interpretation is incorrect.

REFERENCES

W. M. Cady, Phys. Rev. **43**, 325 (1933). (T) (C L)

B. Edlén, letter (Feb. 1949).

February 1949.

Ni XII

(Cl I sequence; 17 electrons)

Z=28

Ground state $1s^2 2s^2 2p^6 3s^2 3p^5 {}^2P_{1/2}$ $3p^5 {}^2P_{1/2}$ cm^{-1}

I. P. 350 volts

This spectrum has not been analyzed. Edlén has tentatively attributed the faint coronal line at 4231.4 Å (wave number 23626 cm^{-1}) to the transition $3p^5 {}^2P_{1/2} - 3p^5 {}^2P_{3/2}$ of [Ni XII], since this interval for the ground term is confirmed by extrapolation along the isoelectronic sequence.

REFERENCE

B. Edlén, Zeit. Astroph. **22**, 32 (1942). (I P) (T) (C L)

January 1948.

Ni XIII

(S I sequence; 16 electrons)

Z=28

Ground state $1s^2 2s^2 2p^6 3s^2 3p^4 {}^3P_2$ $3p^4 {}^3P_2$ cm^{-1}

I. P. volts

This spectrum has not been analyzed, but Edlén has calculated the relative positions of the five lowest levels, in his work on the identification of the coronal lines. He has tentatively attributed two coronal lines to [Ni XIII] as follows:

I. A.	Int.	Wave No.	Desig.
3642. 9		27443	$3p^4 {}^3P_1 - 3p^4 {}^1D_2$
5116. 03	4. 3	19541. 0	$3p^4 {}^3P_2 - 3p^4 {}^3P_1$

These lines have been used to calculate three levels listed in the table. The remaining two are Edlén's estimated values and are entered in brackets.

REFERENCE

B. Edlén, Zeit. Astroph. **22**, 47 (1942). (T) (C L)

Ni XIII

Config.	Desig.	J	Level	Interval
$3s^2 3p^4$	$3p^4 {}^3P$	2	0	-19541 [- 659]
		1	19541	
		0	[20200]	
$3s^2 3p^4$	$3p^4 {}^1D$	2	46984	
$3s^2 3p^4$	$3p^4 {}^1S$	0	[97000]	

January 1948.

Ni XV

(Si I sequence; 14 electrons)

 $Z=28$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^2 {}^3P_0$ $3p^2 {}^3P_0$ cm⁻¹

I. P. 455 volts

This spectrum has not been analyzed. Edlén has, however, suggested that two coronal lines may tentatively be ascribed to the [Ni xv] transitions corresponding to the intervals of the ground term. The coronal data are as follows:

I. A.	Int.	Wave No.	Desig.
6701. 83	5. 4	14917. 2	$3p^2 {}^3P_0 - 3p^2 {}^3P_1$
8024. 21	0. 5	12458. 9	$3p^2 {}^3P_1 - 3p^2 {}^3P_2$

The listed term values have been calculated from the coronal wave numbers, except for the last two, which are Edlén's estimated values and are entered in brackets.

Edlén's unit, 10³ cm⁻¹, has here been changed to cm⁻¹.

REFERENCE

B. Edlén, Zeit. Astroph. **22**, 40 (1942). (I P) (T) (C L)

Ni XV

Config.	Desig.	J	Level	Interval
$3s^2 3p^2$	$3p^2 {}^3P$	0	0. 0	14917. 2 12458. 9
		1	14917. 2	
		2	27376. 1	
$3s^2 3p^2$	$3p^2 {}^1D$	2	[62900]	
$3s^2 3p^2$	$3p^2 {}^1S$	0	[100000]	

October 1947.

Ni xvi

(Al I sequence; 13 electrons)

Z=28

Ground state $1s^2 2s^2 2p^6 3s^2 3p^2 P_{0\frac{1}{2}}^{\circ}$ $3p^2 P_{0\frac{1}{2}}^{\circ}$ cm⁻¹

I. P. volts

This spectrum has not been analyzed. Edlén has, however, suggested that the faint coronal line at 3601.0 Å, wave number 27762, may be classified as due to the forbidden transition that corresponds to the interval of the ground term of Ni xvi, i. e., $3p^2 P_{0\frac{1}{2}}^{\circ} - 3p^2 P_{1\frac{1}{2}}^{\circ}$.

REFERENCE

B. Edlén, *Zeit. Astroph.* **22**, 47 (1942). (T) (C L)

December 1947.

Ni xviii

(Na I sequence; 11 electrons)

Z=28

Ground state $1s^2 2s^2 2p^6 3s^2 S_{\frac{1}{2}}$ $3s^2 S_{0\frac{1}{2}}$ 4897400 cm⁻¹

I. P. 607 volts

Edlén has classified four lines in the interval 43 Å to 52 Å, and extrapolated the absolute value of the ground term from isoelectronic sequence data.

One term, $4d^2 D$, has been calculated from the observed combination with $3p^2 P^{\circ}$ listed by Edlén, and added to his published list.

The unit adopted by Edlén, 10³ cm⁻¹, has here been changed to cm⁻¹.

REFERENCE

B. Edlén, *Zeit. Phys.* **100**, 621 (1936). (I P) (T) (C L).

Ni xviii

Config.	Desig.	<i>J</i>	Level	Interval
3s	3s ² S	0½	0	
3p	3p ² P ^o	½	310600	30400
		1½	341000	
3d	3d ² D	1½	766700	4600
		2½	771300	
4d	4d ² D	1½	2593000	2900
		2½	2595900	
4f	4f ² F ^o	2½	2667400	800
		3½	2668200	
-----	-----	---	-----	
Ni xix (1S ₀)	<i>Limit</i>	---	[4897400]	

June 1947.

COPPER

Cu I

29 electrons

Z=29

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 S_{0\frac{1}{2}}$ $4s^2 S_{0\frac{1}{2}}$ 62317.2 cm^{-1}

I. P. 7.724 volts

The analysis of the arc spectrum of copper affords one of the finest examples of a beautifully complete interpretation of a fairly complex spectrum. Shenstone has verified all of the earlier observations, extended them to include the ultraviolet region, and published a detailed analysis in his 1948 Monograph. This spectrum now furnishes accurate ultraviolet wave lengths for use as standards in investigating other spectra in this region. The number of classified lines totals 732, and the observations extend from 1504 Å to 18229 Å. Many intersystem combinations have been observed, connecting the terms of both multiplicities.

Shenstone remarks that all of the predictable types of electron structure have been identified. To quote, "The spectrum is unique in having many more identified levels above the point of easiest ionization, than below. There are a great many series converging to complex limits and showing various degrees of perturbation". The above limit is from both the 2S and 2D doublet series fitted to an extended Ritz formula, the differences in the calculated values being less than 0.4 cm^{-1} .

Shenstone (August 1948) calls attention to the important fact that "The $3d^9 4s ns$ series is of special interest because it is the only known complex series in which individual series can be unambiguously assigned to the components of a limit of greater complexity than a doublet". Whenever he has designated the component of the 3D term which forms the limit of the separate components of Cu I terms, the J -value is entered in the configuration column of the table. (Lidén has recently reported a similar case in FI, Ark. Phys. (Stockholm) 1, No. 9, 229, 1949.)

In the term array the writer has tentatively assigned the limits 3D and 1D to the appropriate terms having the configuration $3d^9 4s 4p$, although Shenstone states that "it appears more reasonable to describe the electron coupling as $3d^9 ^2D+4s 4p ^3P, ^1P . . .$ " He adds that "the arrangement of levels in the next series member, $3d^9 4s 5p$, should approach much more closely to a grouping which can be correlated to $3d^9 4s ^3D, ^1D+p$." This is also discussed by Racah.

Shenstone has listed the levels in order of increasing numerical value. In order to conform to the general plan of the present volumes they are here grouped into terms on the basis of his assigned designations. Similarly, in column three under "Desig." the notation adopted for series spectra, and described in detail in Volume I, is entered, but Shenstone's notation is retained in column one.

Zeeman observations of Cu I are limited in both number and accuracy, except for the $4p ^2P^o-4d ^2D$ multiplet "which has been shown by Green to fit exceptionally well the formulae for the partial Paschen-Back effect developed by Darwin". The writer has derived the following tentative g -values from published and unpublished observations, using, for blended patterns, the formulae given by Russell, which are an extension of those by Shenstone and Blair. The g -values given by Sommer are included in the final means. A colon indicates that there is only one or a very weak mean determination of the g -value. Three place values are from measurements by Kiess who found selected Cu lines as impurities on plates of other spectra taken at Massachusetts Institute of Technology and at the National Bureau of Standards.

Cu I—Continued

Desig.	<i>J</i>	Obs. <i>g</i>	Desig.	<i>J</i>	Obs. <i>g</i>	Desig.	<i>J</i>	Obs. <i>g</i>
4s ² S	0½	2. 004	4p' ² P°	0½	0. 48	7p ² P°	0½	0. 69:
4s ² ² D	2½	1. 22		1½	1. 28	5s' ⁴ D	3½	1. 42
	1½	0. 806	4p' ² D°	1½	1. 00:	0½	0. 00	
4p ² P°	0½	0. 659		2½	1. 20	5s' ² D	2½	1. 27:
	1½	1. 338	5p ² P°	1½	1. 35:	4d' ⁴ S	1½	1. 74:
4p' ⁴ P°	2½	1. 60	4d ² D	1½	0. 82:	4d' ² F	3½	1. 09
	1½	1. 75		2½	1. 19:	4d' ⁴ G	5½	1. 27:
	0½	2. 63	6s ² S	0½	1. 99	4½	1. 18:	
4p' ⁴ F°	4½	1. 34	5d ² D	1½	0. 77:	2½	0. 53:	
	3½	1. 25		2½	1. 22	4d' ⁴ D	3½	1. 44
	2½	1. 01	4p'' ² F°	3½	1. 14:	2½	1. 36	
	1½	0. 43		2½	0. 84:	4d' ⁴ F	4½	1. 29:
4p' ⁴ D°	3½	1. 45	4p'' ² P°	1½	1. 27:	3½	1. 19	
	2½	1. 38		2½	1. 17:	2½	1. 05	
	1½	1. 18	4p'' ² D°	2½	1. 17:	5d' ⁴ G	5½	1. 27:
	0½	0. 10						
4p' ² F°	2½	0. 92:						
	3½	1. 21						

REFERENCES

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 E. Back, Ann. der Phys. [4] **70**, 361 (1923). (Z E)
 L. A. Sommer, Zeit. Phys. **39**, 711 (1926). (T) (C L) (Z E)
 C. S. Beals, Proc. Roy. Soc. London [A] **111**, 168 (1926). (T) (C L) (Z E)
 A. G. Shenstone, Phys. Rev. **28**, 449 (1926). (T) (C L) (Z E)
 A. G. Shenstone and H. A. Blair, Phil. Mag. **8**, 765 (1929).
 J. B. Green, Phys. Rev. **36**, 157 (1930). (Z E)
 H. N. Russell, Phys. Rev. **36**, 1590 (1930).
 G. Racah, Phys. Rev. **62**, 525 (1942).
 A. G. Shenstone, Phil. Trans. Roy. Soc. (London) [A] **241**, No. 832,297 (1948). (I P) (T) (C L) (Z E) (hfs)
 C. C. Kiess, unpublished material (January 1950). (Z E)

Cu I

Cu I

Author	Config.	Desig.	<i>J</i>	Level	Interval	Author	Config.	Desig.	<i>J</i>	Level	Interval
4s ² S _½	3d ¹⁰ (1S)4s	4s ² S	0½	0. 000		z ² D _½ _½	3d ⁹ 4s(3D)4p	4p' ² D°	1½	46172. 842	425. 50
m ² D _½ _½	3d ⁹ 4s ²	4s ² ² D	2½	11202. 565	—2042. 858	z ² D _½ _½		2½	46598. 34		
m ² D _½ _½		1½	13245. 423			5p ² P _½ _½	3d ¹⁰ (1S)5p	5p ² P°	1½	49332. 95	—0. 31
4p ² P _½ _½	3d ¹⁰ (1S)4p	4p ² P°	0½	30535. 302	248. 384	5p ² P _½ _½		0½	49333. 26		
4p ² P _½ _½		1½	30783. 686			4d ² D _½ _½	3d ¹⁰ (1S)4d	4d ² D	1½	49935. 200	6. 857
						4d ² D _½ _½		2½	49942. 057		
z ⁴ P _½ _½	3d ⁹ 4s(3D)4p	4p' ⁴ P°	2½	39018. 652	—1095. 34	6s ² S	3d ¹⁰ (1S)6s	6s ² S	0½	52848. 749	
z ⁴ P _½ _½		1½	40113. 99	—829. 74		6p ² P _½ _½	3d ¹⁰ (1S)6p	6p ² P°	1½	54784. 06	—243. 68
z ⁴ P _½ _½		0½	40943. 73			6p ² P _½ _½		0½	55027. 74		
z ⁴ F _½ _½	3d ⁹ 4s(3D)4p	4p' ⁴ F°	4½	40909. 133	—244. 295	5d ² D _½ _½	3d ¹⁰ (1S)5d	5d ² D	1½	55387. 668	3. 624
z ⁴ F _½ _½		3½	41153. 433	—409. 462		5d ² D _½ _½		2½	55391. 292		
z ⁴ F _½ _½		2½	41562. 895	—739. 57		4f ² F _¾ _¾	3d ¹⁰ (1S)4f	4f ² F°	3½	55426. 3	—3. 5
z ⁴ F _½ _½		1½	42302. 47			4f ² F _¾ _¾		2½	55429. 8		
5s ² S _½	3d ¹⁰ (1S)5s	5s ² S	0½	43137. 209		y ² F _¾ _¾	3d ⁹ 4s(1D)4p	4p'' ² F°	3½	56029. 95	—2089. 33
z ⁴ D _¾ _¾	3d ⁹ 4s(3D)4p	4p' ⁴ D°	3½	43513. 95	—892. 32	y ² F _¾ _¾		2½	58119. 28		
z ⁴ D _¾ _¾		2½	44406. 268	—137. 885		y ² F _¾ _¾		1½	56343. 74	—2020. 99	
z ⁴ D _¾ _¾		1½	44544. 153	—371. 46		y ² P _¾ _¾	3d ⁹ 4s(1D)4p	4p'' ² P°	0½	58364. 73	
z ⁴ D _¾ _¾		0½	44915. 61			y ² P _¾ _¾		1½	56651. 48	—2039. 38	
z ² F _¾ _¾	3d ⁹ 4s(3D)4p	4p' ² F°	2½	43726. 191	237. 032	y ² D _¾ _¾	3d ⁹ 4s(1D)4p	4p'' ² D°	2½	58690. 86	
z ² F _¾ _¾		3½	44963. 223			y ² D _¾ _¾		1½			
z ² P _¾ _¾	3d ⁹ 4s(3D)4p	4p' ² P°	0½	45821. 00	58. 31	y ² D _¾ _¾		1½			
z ² P _¾ _¾		1½	45879. 311			y ² D _¾ _¾		1½			

Cu I—Continued

Cu I—Continued

Author	Config.	Desig.	J	Level	Interval	Author	Config.	Desig.	J	Level	Interval	
7s	² S _{1/2}	3d ¹⁰ (1S) 7s	7s ² S	0 1/2	56671. 387	g	⁴ S _{1/2}	3d ⁹ 4s (3D ₃) 4d	4d' ⁴ S	1 1/2	70998. 12	
7p	² P _{1/2} ^o	3d ¹⁰ (1S) 7p	7p ² P ^o	0 1/2	57419. 31	g	² D _{2 1/2}	3d ⁹ 4s (3D ₃) 4d	4d' ² D	2 1/2	71098. 17	-1006. 6
7p	² P _{1/2}			1 1/2	57948. 71	g	² D _{1 1/2}	3d ⁹ 4s (3D ₂) 4d		1 1/2	72104. 8	
6d	² D _{1 1/2}	3d ¹⁰ (1S) 6d	6d ² D	1 1/2	57893. 05	g	² F _{3 1/2}	3d ⁹ 4s (3D ₃) 4d	4d' ² F	3 1/2	71127. 81	-1023. 37
6d	² D _{2 1/2}			2 1/2	57895. 10	g	² F _{2 1/2}	3d ⁹ 4s (3D ₂) 4d		2 1/2	72151. 18	
5f	² F _{3 1/2}	3d ¹⁰ (1S) 5f	5f ² F ^o	2 1/2	57905. 2	g	⁴ G _{5 1/2}	3d ⁹ 4s (3D ₃) 4d	4d' ⁴ G	5 1/2	71130. 69	-848. 01
5f	² F _{3 1/2}			3 1/2	57908. 7	g	⁴ G _{4 1/2}	3d ⁹ 4s (3D ₂) 4d		4 1/2	71978. 70	-1124. 04
8s	² S _{1/2}	3d ¹⁰ (1S) 8s	8s ² S	0 1/2	58568. 92	g	⁴ G _{3 1/2}	3d ⁹ 4s (3D ₁) 4d		3 1/2	73102. 74	-95. 97
7d	² D _{1 1/2}	3d ¹⁰ (1S) 7d	7d ² D	1 1/2	59249. 46	g	⁴ G _{2 1/2}	"		2 1/2	73198. 71	
7d	² D _{2 1/2}			2 1/2	59250. 72	g	⁴ P _{2 1/2}	3d ⁹ 4s (3D ₃) 4d	4d' ⁴ P	2 1/2	71178. 19	-749. 03
8p	² P _{1/2} ^o	3d ¹⁰ (1S) 8p	8p ² P ^o	1 1/2	59275. 33	g	⁴ P _{1 1/2}	3d ⁹ 4s (3D ₂) 4d		1 1/2	71927. 22	44. 26
8p	² P _{1/2}			0 1/2	59323. 17	g	⁴ P _{3/2}	"		0 1/2	71882. 96	
9s	² S _{1/2}	3d ¹⁰ (1S) 9s	9s ² S	0 1/2	59647. 88	g	⁴ D _{3 1/2}	3d ⁹ 4s (3D ₃) 4d	4d' ⁴ D	3 1/2	71268. 21	-798. 76
8d	² D _{1 1/2}	3d ¹⁰ (1S) 8d	8d ² D	1 1/2	60065. 51	g	⁴ D _{2 1/2}	3d ⁹ 4s (3D ₂) 4d		2 1/2	72066. 97	-1037. 91
8d	² D _{2 1/2}			2 1/2	60066. 33	g	⁴ D _{1 1/2}	3d ⁹ 4s (3D ₁) 4d		1 1/2	73104. 88	
9p	² P _{1/2} ^o	3d ¹⁰ (1S) 9p	9p ² P ^o	1 1/2	60070. 6	g	⁴ F _{4 1/2}	3d ⁹ 4s (3D ₃) 4d	4d' ⁴ F	4 1/2	71290. 54	-802. 54
9p	² P _{1/2}			0 1/2	60085. 2	g	⁴ F _{3 1/2}	3d ⁹ 4s (3D ₂) 4d		3 1/2	72093. 08	-1211. 59
9d	² D _{1 1/2}	3d ¹⁰ (1S) 9d	9d ² D	1 1/2	60594. 53	g	⁴ F _{2 1/2}	3d ⁹ 4s (3D ₁) 4d		2 1/2	73304. 67	-11. 79
9d	² D _{2 1/2}			2 1/2	60595. 05	x	⁴ F _{1 1/2}	"		1 1/2	73316. 46	
10p	² P _{1/2} ^o	3d ¹⁰ (1S) 10p	10p ² P ^o	1 1/2	60595. 0	x	² D _{2 1/2}	3d ⁹ 4s (3D) 5p	5p' ² D ^o	2 1/2	71745. 5	-278. 7
10p	² P _{1/2}			0 1/2	60601. 9	x	² D _{1 1/2}	"		1 1/2	72024. 2?	
10d	² D _{1 1/2}	3d ¹⁰ (1S) 10d	10d ² D	1 1/2	60956. 92	x	² P _{1 1/2}	3d ⁹ 4s (3D) 5p	5p' ² P ^o	0 1/2	71917. ?	
10d	² D _{2 1/2}			2 1/2	60957. 35	i	⁴ D _{3 1/2}	3d ⁹ 4s (3D ₃) 6s	6s' ⁴ D	3 1/2	73995. 15	-317. 76
11p	² P ^o	3d ¹⁰ (1S) 11p	11p ² P ^o	{0 1/2}	60958	i	⁴ D _{2 1/2}	"		2 1/2	74312. 91	-730. 70
11d	² D _{2 1/2}	3d ¹⁰ (1S) 11d	11d ² D	1 1/2	61215. 59	i	⁴ D _{1 1/2}	3d ⁹ 4s (3D ₂) 6s		1 1/2	75043. 61	-1020. 76
				2 1/2		i	⁴ D _{3/2}	3d ⁹ 4s (3D ₁) 6s		0 1/2	76064. 37	
	Cu II (1S ₀)	Limit	---		62317. 2	w	² P _{1 1/2}	3d ⁹ 4s (1D) 5p	5p'' ² P ^o	1 1/2	74259. 5?	-831. 1
e	⁴ D _{3 1/2}	3d ⁹ 4s (3D ₃) 5s	5s' ⁴ D	3 1/2	62403. 320	w	² P _{1 1/2}	"		0 1/2	75090. 6?	
e	⁴ D _{2 1/2}	"		2 1/2	62948. 29	w	² F _{3 1/2}	3d ⁹ 4s (1D) 5p	5p'' ² D ^o	2 1/2	74507. 6?	
e	⁴ D _{1 1/2}	3d ⁹ 4s (3D ₂) 5s		1 1/2	63584. 57	w	² F _{2 1/2}	"		1 1/2		
e	⁴ D _{3/2}	3d ⁹ 4s (3D ₁) 5s		0 1/2	64472. 300	h	² F _{2 1/2}	3d ⁹ 4s (1D) 5p	5p'' ² F ^o	3 1/2	74341. 9	-581. 8
e	² D _{2 1/2}	3d ⁹ 4s (3D ₂) 5s	5s' ² D	2 1/2	64657. 8	h	² F _{1 1/2}	"		2 1/2	74923. 7?	
e	² D _{1 1/2}	3d ⁹ 4s (3D ₁) 5s		1 1/2	65260. 1	w	² D ^o	3d ⁹ 4s (1D) 5p	5p'' ² D ^o	1 1/2		
f	² D _{2 1/2}	3d ⁹ 4s (1D) 5s	5s'' ² D	2 1/2	67142. 7	h	² P _{3/2}	3d ⁹ 4s (1D) 4d	4d'' ² P	0 1/2	75109. 46	153. 99
f	² D _{1 1/2}			1 1/2	67971. 94	h	² P _{1 1/2}	"		1 1/2	75263. 45	
x	⁴ P _{1 1/2} ^o	3d ⁹ 4s (3D) 5p	5p' ⁴ P ^o	2 1/2	70281. ?	i	² D _{2 1/2}	3d ⁹ 4s (3D ₂) 6s	6s' ² D	2 1/2	75170. 25	-1162. 0
x	⁴ P _{1 1/2}			1 1/2	71004. ?	i	² D _{1 1/2}	3d ⁹ 4s (3D ₁) 6s		1 1/2	76332. 3	
x	⁴ F _{3 1/2}	3d ⁹ 4s (3D) 5p	5p' ⁴ F ^o	4 1/2	70336. 5 ?	h	² G _{3 1/2}	3d ⁹ 4s (1D) 4d	4d'' ² G	3 1/2	75206. 4	139. 7
x	⁴ F _{3 1/2}			3 1/2	70414. 1	h	² G _{4 1/2}	"		4 1/2	75346. 1?	
x	⁴ D _{3 1/2}	3d ⁹ 4s (3D) 5p	5p' ⁴ D ^o	3 1/2	70441. 0 ?	h	² S _{1/2}	3d ⁹ 4s (1D) 4d	4d'' ² S	0 1/2	75386. 7	
x	⁴ D _{2 1/2}			2 1/2	70561. 2 ?	h	² D _{1 1/2}	3d ⁹ 4s (1D) 4d	4d'' ² D	1 1/2	75440. 1	6. 4
x	⁴ D _{1 1/2}			1 1/2	71029. 6	h	² D _{2 1/2}	"		2 1/2	75446. 5	
g	² P _{1 1/2}	3d ⁹ 4s (3D ₃) 4d	4d' ² P	1 1/2	70853. 39	h	² F _{2 1/2}	3d ⁹ 4s (1D) 4d	4d'' ² F	2 1/2	75536. 2	36. 6
g	² P _{3/2}	3d ⁹ 4s (3D ₂) 4d		0 1/2	72151. 49?	h	² F _{3 1/2}	"		3 1/2	75572. 85	
g	² S _{1/2}	3d ⁹ 4s (3D ₃) 4d	4d' ² S	0 1/2	70853. 9 ?	j	² G _{4 1/2}	3d ⁹ 4s (3D ₃) 5d	5d' ² G	4 1/2	76824. 3	-1074. 6
g	² G _{4 1/2}	3d ⁹ 4s (3D ₃) 4d	4d' ² G	4 1/2	70859. 53	j	² G _{3 1/2}	3d ⁹ 4s (3D ₂) 5d		3 1/2	77898. 9	
g	² G _{3 1/2}	3d ⁹ 4s (3D ₂) 4d		3 1/2	72016. 76	j	² P _{1 1/2}	3d ⁹ 4s (3D ₃) 5d	5d' ² P	1 1/2	76831. 31	
x	² F _{3 1/2}	3d ⁹ 4s (3D) 5p	5p' ² F ^o	2 1/2	70959. 7 ?	i	² D _{2 1/2}	3d ⁹ 4s (3D ₃) 5d	5d' ² D	2 1/2	76949. 2	-984. 1
x	² F _{3 1/2}			3 1/2	71613. 9 ?	j	² D _{1 1/2}	3d ⁹ 4s (3D ₂) 5d		1 1/2	77933. 3	
							⁴ S _{1 1/2}	3d ⁹ 4s (3D ₃) 5d	5d' ⁴ S	1 1/2	76959. 0	
						j	² F _{3 1/2}	3d ⁹ 4s (3D ₃) 5d	5d' ² F	3 1/2	76960. 2	-999. 1
						j	² F _{2 1/2}	3d ⁹ 4s (3D ₂) 5d		2 1/2	77959. 3	

Cu I—Continued

Author	Config.	Desig.	<i>J</i>	Level	Interval
<i>j</i>	$3d^9 4s(^3D_3)5d$	$5d' \ ^4G$	$5\frac{1}{2}$	77014. 1	—839. 9
<i>j</i>	$3d^9 4s(^3D_2)5d$		$4\frac{1}{2}$	77854. 0	—1134. 3
<i>j</i>	$3d^9 4s(^3D_1)5d$		$3\frac{1}{2}$	78988. 3	—65. 1
<i>j</i>	$3d^9 4s(^3D_1)5d$		$2\frac{1}{2}$	79053. 4	
<i>j</i>	$3d^9 4s(^3D_3)5d$	$5d' \ ^4P$	$2\frac{1}{2}$	77030. 59	—810. 3
<i>j</i>	$3d^9 4s(^3D_2)5d$		$1\frac{1}{2}$	77840. 9	26. 4
<i>j</i>	$3d^9 4s(^3D_1)5d$		$0\frac{1}{2}$	77814. 5	
<i>j</i>	$3d^9 4s(^3D_3)5d$	$5d' \ ^4D$	$3\frac{1}{2}$	77068. 2	—837. 3
<i>j</i>	$3d^9 4s(^3D_2)5d$		$2\frac{1}{2}$	77905. 5	—1097. 6
<i>j</i>	$3d^9 4s(^3D_1)5d$		$1\frac{1}{2}$	79003. 1	
<i>j</i>	$3d^9 4s(^3D_1)5d$		$0\frac{1}{2}$		
<i>j</i>	$3d^9 4s(^3D_3)5d$	$5d' \ ^4F$	$4\frac{1}{2}$	77080. 5	—838. 9
<i>j</i>	$3d^9 4s(^3D_2)5d$		$3\frac{1}{2}$	77919. 4	—1197. 1
<i>j</i>	$3d^9 4s(^3D_1)5d$		$2\frac{1}{2}$	79116. 5	—2. 8
<i>j</i>	$3d^9 4s(^3D_1)5d$		$1\frac{1}{2}$	79119. 3	
<i>k</i>	$3d^9 4s(^3D_3)7s$	$7s' \ ^4D$	$3\frac{1}{2}$	78261. 2	—225. 3
<i>k</i>	$3d^9 4s(^3D_2)7s$		$2\frac{1}{2}$	78486. 5	—771. 3
<i>k</i>	$3d^9 4s(^3D_2)7s$		$1\frac{1}{2}$	79257. 8	—1072. 6
<i>k</i>	$3d^9 4s(^3D_1)7s$		$0\frac{1}{2}$	80330. 4	
<i>p</i>	$3d^9 4s(^1D)6s$	$6s'' \ ^2D$	$2\frac{1}{2}$	78349. 6	—228. 4
<i>p</i>	$3d^9 4s(^1D)6s$		$1\frac{1}{2}$	78578. 0	
<i>k</i>	$3d^9 4s(^3D_2)7s$	$7s' \ ^2D$	$2\frac{1}{2}$	79268. 0	—1188. 4
<i>k</i>	$3d^9 4s(^3D_1)7s$		$1\frac{1}{2}$	80456. 4?	
<i>l</i>	$3d^9 4s(^3D_3)6d$	$6d' \ ^4S$	$1\frac{1}{2}$	79641. 4	
<i>l</i>	$3d^9 4s(^3D_3)6d$	$6d' \ ^4G$	$5\frac{1}{2}$	79667. 9	—837. 6
<i>l</i>	$3d^9 4s(^3D_2)6d$		$4\frac{1}{2}$	80505. 5	
			$3\frac{1}{2}$		
			$2\frac{1}{2}$		
<i>l</i>	$3d^9 4s(^3D_3)6d$	$6d' \ ^4P$	$2\frac{1}{2}$	79675. 1	
			$1\frac{1}{2}$		
			$0\frac{1}{2}$		
<i>l</i>	$3d^9 4s(^3D_3)6d$	$6d' \ ^4D$	$3\frac{1}{2}$	79694. 5	—847. 7
<i>l</i>	$3d^9 4s(^3D_2)6d$		$2\frac{1}{2}$	80542. 2	
			$1\frac{1}{2}$		
			$0\frac{1}{2}$		
<i>l</i>	$3d^9 4s(^3D_3)6d$	$6d' \ ^4F$	$4\frac{1}{2}$	79700. 5	—859. 5
<i>l</i>	$3d^9 4s(^3D_2)6d$		$3\frac{1}{2}$	80560. 0	
			$2\frac{1}{2}$		
			$1\frac{1}{2}$		
<i>n</i>	$3d^9 4s(^3D_3)8s$	$8s' \ ^4D$	$3\frac{1}{2}$	80318. 4	
			$2\frac{1}{2}$		
			$1\frac{1}{2}$		
			$0\frac{1}{2}$		
<i>l</i>	$3d^9 4s(^3D_3)6d$	$6d' \ ^2G$	$4\frac{1}{2}$	80553. 8	
<i>l</i>	$3d^9 4s(^3D_2)6d$		$3\frac{1}{2}$		
<i>l</i>	$3d^9 4s(^3D_3)6d$	$6d' \ ^2F$	$3\frac{1}{2}$	80574	
<i>l</i>	$3d^9 4s(^3D_2)6d$		$2\frac{1}{2}$		
<i>l</i>	$3d^9 4s(^3D_3)6d$	$6d' \ ^2D$	$2\frac{1}{2}$	80586. 7?	
<i>l</i>	$3d^9 4s(^3D_2)6d$		$1\frac{1}{2}$		
<i>o</i>	$3d^9 4s(^1D)5d$	$5d'' \ ^2D$	$2\frac{1}{2}$	81292. 5	—21. 2
<i>o</i>	$3d^9 4s(^1D)5d$		$1\frac{1}{2}$	81313. 7	
<i>o</i>	$3d^9 4s(^1D)5d$	$5d'' \ ^2F$	$2\frac{1}{2}$	81362. 7	13. 5
<i>o</i>	$3d^9 4s(^1D)5d$		$3\frac{1}{2}$	81376. 2	

December 1949.

Cu I OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms	$nf (n \geq 4)$	$nf (n \geq 4)$
$3d^{10} (1S)nx$	$4s^2 2D$	$4-11d' \ ^2D$	$4-11d' \ ^2D$
$3d^9 4s(^3D)nx'$	$ns (n \geq 4)$	$4-6d' \ ^4S$ $4d' \ ^2S$ $4d'' \ ^2S$	$4-6d' \ ^4S$ $4d' \ ^2S$ $4d'' \ ^2S$
$3d^9 4s(^1D)nx''$	$4-9s^2 S$	$4-6d' \ ^4P$ $4d' \ ^2P$ $4d'' \ ^2P$	$4-6d' \ ^4P$ $4d' \ ^2P$ $4d'' \ ^2P$
	$4-11p \ ^2P^o$	$4-6d' \ ^4D$ $4d' \ ^2D$ $4d'' \ ^2D$	$4-6d' \ ^4D$ $4d' \ ^2D$ $4d'' \ ^2D$
	$4-11p \ ^2P^o$	$4-6d' \ ^4F$ $4d' \ ^2F$ $4d'' \ ^2F$	$4-6d' \ ^4F$ $4d' \ ^2F$ $4d'' \ ^2F$
	$4-11p \ ^2P^o$	$4-6d' \ ^4G$ $4d' \ ^2G$ $4d'' \ ^2G$	$4-6d' \ ^4G$ $4d' \ ^2G$ $4d'' \ ^2G$

*For predicted terms in the spectra of the Cu I isoelectronic sequence, see Vol. II, Introduction.

Cu II

(Ni I sequence; 28 electrons)

 $Z=29$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} {}^1S_0$ $3d^{10} {}^1S_0$ 163665.6 cm^{-1}

I. P. 20.29 volts

The unique honor of having produced a practically complete analysis of a complex spectrum is due Shenstone for his investigation of the Cu II spectrum. This spectrum has been photographed from 600 Å to 10000 Å. Of the 1350 lines observed only a few faint lines remain unclassified. He has extended the earlier work to include over 40 series of three or more levels; and identified $d^9 f$ and $d^9 g$ structures completely for the first time. He remarks that "the most striking feature is the number of levels which show the effects of mixed configurations." A theoretical discussion of the spin-orbit interaction of the almost closed shell as exhibited by the Cu II spectrum, is given by Shortley and Fried.

Shenstone has determined the limit from the $3d^9 ns {}^3D {}^1D$ series ($n=4$ to 8), using an extended Ritz formula described in detail in his Monograph (1936). The series are so long, and the separation of the components of the 3D limit term in Cu III is so large, that for this limit it can be determined without ambiguity which component is the limit of the separate components of the Cu II terms. The J -values indicating the components of this limit term are included with the configuration in column two of the table. When the J -value of the limit is the same for all components of the Cu II term, only one entry is given in this column, as for example, for the term $4f {}^3P^o$ (135902 cm^{-1}). Shenstone remarks also that the intensities of the lines from the transition $4f-5g$ "indicate very strongly that each empirical level is in reality double . . .". This is completely explained by jl -coupling.

In addition to the analysis, the accurate ultraviolet wavelengths of Cu II provide excellent standards for the measurement of other spectra in this region.

In Shenstone's Monograph the levels are given in numerical order "because to group them as terms in such a spectrum as this would be quite misleading". In the present table, as in Volume I, those levels which he gives as components of a term are grouped together. Shenstone's notation, which appears in column one, has, similarly, been altered in column three to conform to the notation adopted for other series spectra.

Racah has also emphasized the similarity of coupling between levels from the $3d^9 nx$ configuration of Cu II and those of the inert gases from $p^5 nx$. He has furnished the writer with the following data indicating the jl -coupling notation for levels of Cu II.

Cu II—Continued

Author	Config.	Desig.	<i>J</i>	Author	Config.	Desig.	<i>J</i>
<i>ns</i> ³ D ₃ ³ D ₂	3 <i>d</i> ⁹ (² D _{2½}) <i>ns</i>	<i>ns</i> [2½]	3 2	<i>nf</i> ³ P ₂ ^o ¹ P ₀ ^o	3 <i>d</i> ⁹ (² D _{2½}) <i>nf</i>	<i>nf</i> [0½] ^o	1 0
<i>ns</i> ³ D ₁ ¹ D ₂	3 <i>d</i> ⁹ (² D _{1½}) <i>ns</i>	<i>ns</i> ' [1½]	1 2	<i>nf</i> ³ P ₂ ^o ¹ P ₁ ^o	"	<i>nf</i> [1½] ^o	2 1
<i>np</i> ³ P ₂ ^o ³ P ₁ ^o	3 <i>d</i> ⁹ (² D _{2½}) <i>np</i>	<i>np</i> [1½] ^o	2 1	<i>nf</i> ³ H ₆ ^o ³ H ₅ ^o	"	<i>nf</i> [5½] ^o	6 5
<i>np</i> ³ F ₃ ^o ³ F ₄ ^o	"	<i>np</i> [3½] ^o	3 4	<i>nf</i> ³ D ₃ ^o ³ D ₂ ^o	"	<i>nf</i> [2½] ^o	3 2
<i>np</i> ³ D ₃ ^o ³ D ₂ ^o	"	<i>np</i> [2½] ^o	3 2	<i>nf</i> ³ F ₃ ^o ³ F ₄ ^o	"	<i>nf</i> [3½] ^o	3 4
<i>np</i> ³ P ₀ ^o ¹ P ₁ ^o	3 <i>d</i> ⁹ (² D _{1½}) <i>np</i>	<i>np</i> ' [0½] ^o	0 1	<i>nf</i> ³ G ₅ ^o ³ G ₄ ^o	"	<i>nf</i> [4½] ^o	5 4
<i>np</i> ³ F ₂ ^o ¹ F ₃ ^o	"	<i>np</i> ' [2½] ^o	2 3	<i>nf</i> ¹ D ₂ ^o ³ D ₁ ^o	3 <i>d</i> ⁹ (² D _{1½}) <i>nf</i>	<i>nf</i> ' [1½] ^o	2 1
<i>np</i> ³ D ₁ ^o ¹ D ₂ ^o	"	<i>np</i> ' [1½] ^o	1 2	<i>nf</i> ¹ H ₅ ^o ³ H ₄ ^o	"	<i>nf</i> ' [4½] ^o	5 4
<i>nd</i> ³ S ₁	3 <i>d</i> ⁹ (² D _{2½}) <i>nd</i>	<i>nd</i> [0½]	1	<i>nf</i> ¹ F ₃ ^o ³ F ₂ ^o	"	<i>nf</i> ' [2½] ^o	3 2
<i>nd</i> ³ G ₅ ³ G ₄	"	<i>nd</i> [4½]	5 4	<i>nf</i> ¹ G ₄ ^o ³ G ₃ ^o	"	<i>nf</i> ' [3½] ^o	4 3
<i>nd</i> ³ P ₂ ³ P ₁	"	<i>nd</i> [1½]	2 1	<i>ng</i> ³ D _{1, 2}	3 <i>d</i> ⁹ (² D _{2½}) <i>ng</i>	<i>ng</i> [1½]	1, 2
<i>nd</i> ³ D ₃ ³ D ₂	"	<i>nd</i> [2½]	3 2	<i>ng</i> ³ I _{6, 7}	"	<i>ng</i> [6½]	6, 7
<i>nd</i> ³ F ₃ ³ F ₄	"	<i>nd</i> [3½]	3 4	<i>ng</i> ¹ D _{2, 3, ³D₃}	"	<i>ng</i> [2½]	2, 3
<i>nd</i> ³ P ₀	"	<i>nd</i> [0½]	0	<i>ng</i> ³ F _{3, 4}	"	<i>ng</i> [3½]	3, 4
<i>nd</i> ¹ P ₁	3 <i>d</i> ⁹ (² D _{1½}) <i>nd</i>	<i>nd</i> ' [0½]	1	<i>ng</i> ³ H _{5, 6}	"	<i>ng</i> [5½]	5, 6
<i>nd</i> ³ G ₃ ¹ G ₄	"	<i>nd</i> ' [3½]	3 4	<i>ng</i> ³ G _{4, 5}	"	<i>ng</i> [4½]	4, 5
<i>nd</i> ³ D ₁ ¹ D ₂	"	<i>nd</i> ' [1½]	1 2	<i>ng</i> ³ F _{2, 3} , ¹ F ₃	3 <i>d</i> ⁹ (² D _{1½}) <i>ng</i>	<i>ng</i> ' [2½]	2, 3
<i>nd</i> ¹ F ₃ ³ F ₂	"	<i>nd</i> ' [2½]	3 2	<i>ng</i> ³ I _{5, 6} , ¹ I ₆	"	<i>ng</i> ' [5½]	5, 6
<i>nd</i> ¹ S ₀	"	<i>nd</i> ' [0½]	0	<i>ng</i> ³ G _{3, 4} , ¹ G ₄	"	<i>ng</i> ' [3½]	3, 4
				<i>ng</i> ³ H _{4, 5} , ¹ H ₅	"	<i>ng</i> ' [4½]	4, 5

Intersystem combinations have been observed.

The *g*-values given below have been determined by the writer from the observed Zeeman patterns published in Shenstone's 1927 paper. The formulae for blended patterns as given by Russell were used in making these calculations. These formulas are an extension of those by Shenstone and Blair.

Desig.	<i>J</i>	Obs. <i>g</i>	Desig.	<i>J</i>	Obs. <i>g</i>
4 <i>s</i> ³ D	3	1.32	4 <i>p</i> ³ D ^o	2	1.08
	2	1.16		1	0.47
	1	0.48			
4 <i>s</i> ¹ D	2	1.00	4 <i>p</i> ¹ D ^o	2	0.99
			4 <i>p</i> ¹ P ^o	1	1.04
4 <i>p</i> ³ P ^o	2	1.49	5 <i>s</i> ³ D	3	1.31
	1	1.49		2	1.14
4 <i>p</i> ³ F ^o	4	1.23		1	0.49:
	3	1.06	5 <i>s</i> ¹ D	2	1.04
	2	0.67			

Cu II—Continued

Cu II—Continued

Author	Config.	Desig.	J	Level	Interval	Author	Config.	Desig.	J	Level	Interval
5p $^1D_{\frac{1}{2}}$	$3d^9(^2D_{1/2})5p$	$5p \ ^1D^\circ$	2	123556.94		4f 3F_4	$3d^9(^2D_{3/2})4f$	4f $^3F^\circ$	4	136133.06	97.42
y $^3D_{\frac{3}{2}}$	$3d^8 4s(^2D)4p$	$4p''^3D^\circ$	3	125230.6	-17.2	3F_3	$3d^9(^2D_{1/2})4f$		3	136035.64	-2141.58
$^3D_{\frac{5}{2}}$			2	125247.8		3F_2			2	138177.22	
3D_1			1	125563.7	-320.9	4f 3G_5	$3d^9(^2D_{3/2})4f$	4f $^3G^\circ$	5	136160.93	-109.33
y 1P_1	$3d^8 4s(^2D)4p$	$4p''^1P^\circ$	1	125399.8		3G_4	$3d^9(^2D_{1/2})4f$		4	136270.26	-1991.57
x 5S_2	$3d^8 4s(^4P)4p$	$4p^{IV}^5S^\circ$	2	128365.7		3G_3	$3d^9(^2D_{1/2})4f$		3	138261.83	
y 3F_2	$3d^8 4s(^2D)4p$	$4p''^3F^\circ$	2	128569.6	-10.7	5d 3S_1	$3d^9(^2D_{3/2})5d$	5d 3S	1	136336.88	
3F_3			3	128553.9	219.2		$3d^8 4s(^2G)4p$	$4p^{VI}^3H^\circ$	6		
3F_4			4	128778.1		u 3H_4			5	136694.1	
x 5P_3	$3d^8 4s(^4P)4p$	$4p^{IV}^5P^\circ$	3	129116.9	263.6	5d 3G_5	$3d^9(^2D_{3/2})5d$	5d 3G	5	136725.93	-39.45
5P_2			2	128853.3	-906.3	3G_4	$3d^9(^2D_{1/2})5d$		4	136765.38	-2053.79
5P_1			1	129759.6		3G_3			3	138819.17	
y 3P_2	$3d^8 4s(^2D)4p$	$4p''^3P^\circ$	2	130386.4		5d 3P_2	$3d^9(^2D_{3/2})5d$	5d 3P	2	136754.11	-19.1
			1			3P_1			1	136773.2	-841.1
			0			3P_0			0	137614.3	
y $^1D_{\frac{3}{2}}$	$3d^8 4s(^2D)4p$	$4p''^1D^\circ$	2	130632.4		5d 3D_3	$3d^9(^2D_{3/2})5d$	5d 3D	3	136919.35	-154.40
x 5D_0	$3d^8 4s(^4P)4p$	$4p^{IV}^5D^\circ$	0	131206.0	-261	3D_2	$3d^9(^2D_{1/2})5d$		2	137073.75	-1824.68
$^5D_{2,1}$			1	130945	368	3D_1			1	138898.43	
5D_3			2	131312.7		5d 3F_4	$3d^9(^2D_{3/2})5d$	5d 3F	4	137044.45	-9.48
			3			3F_3			3	137034.97	-2107.1
			4			3F_2	$3d^9(^2D_{1/2})5d$		2	139142.1	
y 1F_3	$3d^8 4s(^2D)4p$	$4p''^1F^\circ$	3	131044.1		u 1H_3	$3d^8 4s(^2G)4p$	$4p^{VI}^1H^\circ$	5	137082.5	
6s 3D_3	$3d^9(^2D_{3/2})6s$	$6s \ ^3D$	3	133594.23	-133.84	w 1P_1	$3d^8 4s(^2P)4p$	$4p^V^1P^\circ$	1	137212.7	
3D_2			2	133728.07	-1936.54	u 3F_4	$3d^8 4s(^2G)4p$	$4p^{VI}^3F^\circ$	4	137938.9	-463.3
3D_1	$3d^9(^2D_{1/2})6s$		1	135664.61		3F_3			3	138402.2	-626.0
w 3P_2	$3d^8 4s(^2P)4p$	$4p^V^3P^\circ$	2	133826.2	-533.7	3F_2			2	139028.2	
3P_1			1	134359.9	-1124.4	4f $^1D_{\frac{3}{2}}$	$3d^9(^2D_{1/2})4f$	4f $^1D^\circ$	2	138003.20	
3P_0			0	135484.3		4f $^1H_{\frac{3}{2}}$	$3d^9(^2D_{1/2})4f$	4f $^1H^\circ$	5	138064.32	
w $^3D_{\frac{3}{2}}$	$3d^8 4s(^2P)4p$	$4p^V^3D^\circ$	3	133984.6	-691.0	4f $^1F_{\frac{3}{2}}$	$3d^9(^2D_{1/2})4f$	4f $^1F^\circ$	3	138130.81	
$^3D_{\frac{5}{2}}$			2	134675.6	-460.0	4f 1G_4	$3d^9(^2D_{1/2})4f$	4f $^1G^\circ$	4	138220.07	
3D_1			1	135135.6		5d 1P_1	$3d^9(^2D_{1/2})5d$	5d 1P	1	138593.10	
v 3G_2	$3d^8 4s(^4F)4p$	$4p' \ ^3G^\circ$	5	134110.6	-1724.3	5d 1G_4	$3d^9(^2D_{1/2})5d$	5d 1G	4	138883.0	
3G_4			4	135834.9	-1242.7	5d 1D_2	$3d^9(^2D_{1/2})5d$	5d 1D	2	138981.3	
3G_3			3	137077.6		5d 1F_3	$3d^9(^2D_{1/2})5d$	5d 1F	3	139119.5	
v 3F_4	$3d^8 4s(^4F)4p$	$4p' \ ^3F^\circ$	4	134743.0	-1699.0	6p 3P_2	$3d^9(^2D_{3/2})6p$	6p $^3P^\circ$	2	139217.3	-24.3
3F_3			3	136442.05	-1207.1	3P_1			1	139241.6	
3F_2			2	137649.1		6p 3F_4	$3d^9(^2D_{3/2})6p$	6p $^3F^\circ$	4	139395.8	64.5
v $^3D_{\frac{3}{2}}$	$3d^8 4s(^4F)4p$	$4p' \ ^3D^\circ$	3	135733.6	-1065.7	3F_3	$3d^9(^2D_{1/2})6p$		3	139331.3	-2402.6
$^3D_{\frac{5}{2}}$			2	136799.3		3F_2			2	141733.9	
3D_1			1	137913.8	-1114.5	6p $^3D_{\frac{3}{2}}$	$3d^9(^2D_{3/2})6p$	6p $^3D^\circ$	3	139741.1	30.9
6s 1D_2	$3d^9(^2D_{1/2})6s$	$6s \ ^1D$	2	135760.27		$^3D_{\frac{5}{2}}$	$3d^9(^2D_{1/2})6p$		2	139710.2	-1534.4
4f 3P_0	$3d^9(^2D_{3/2})4f$	$4f \ ^3P^\circ$	0	135902.43	-38.50	3D_1	$3d^9(^2D_{1/2})6p$		1	141244.6	
3P_1			1	135863.93	47.06	1i	$3d^8 4s(^4P)4p$	$4p^{IV}^1^\circ$	1	140481.77	
3P_2			2	135910.99		5d 1S_0	$3d^9(^2D_{1/2})5d$	5d 1S	0	140589.4	
4f $^3H_{\frac{3}{2}}$	$3d^9(^2D_{3/2})4f$	$4f \ ^3H^\circ$	6	135931.27	-2.88	6p 1P_1	$3d^9(^2D_{1/2})6p$	6p $^1P^\circ$	1	140984.2	
$^3H_{\frac{5}{2}}$			5	135934.15	-2139.52	6p 1F_3	$3d^9(^2D_{1/2})6p$	6p $^1F^\circ$	3	141204.1	
3H_4	$3d^9(^2D_{1/2})4f$		4	138073.67							
w $^1D_{\frac{3}{2}}$	$3d^8 4s(^2P)4p$	$4p^V^1D^\circ$	2	135952.55							
4f 1P_1	$3d^9(^2D_{3/2})4f$	$4f \ ^1P^\circ$	1	135953.46							
4f $^3D_{\frac{3}{2}}$	$3d^9(^2D_{3/2})4f$	$4f \ ^3D^\circ$	3	135990.15	-24.11						
$^3D_{\frac{5}{2}}$			2	136014.26	-2014.44						
3D_1	$3d^9(^2D_{1/2})4f$		1	138028.70							

Cu II OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms		
	$4s^2 3P$	$4s^2 1D$	$4s^2 3F$ $4s^2 1G$
$3d^9 4s^2$			
$3d^{10} 1S$			
	$np (n \geq 4)$		
$3d^9 (2D) nx$	$4-6p$ $3P^{\circ}$ $1P^{\circ}$	$4-6p$ $3D^{\circ}$ $1D^{\circ}$	$4-6p$ $3F^{\circ}$ $1F^{\circ}$
$3d^8 4s (4F) nx'$	$4p'$ $3D^{\circ}$	$4p'$ $3D^{\circ}$	$4p'$ $3F^{\circ}$
$3d^8 4s (2F) nx''$	$4p''$ $3D^{\circ}$	$4p''$ $3D^{\circ}$	$4p''$ $3F^{\circ}$
$3d^8 4s (2D) nx'''$	$4p'''$ $3P^{\circ}$ $1P^{\circ}$	$4p'''$ $3D^{\circ}$ $1D^{\circ}$	$4p'''$ $3F^{\circ}$ $1F^{\circ}$
$3d^8 4s (4P) nx^{IV}$	$4p^{IV}$ $3P^{\circ}$ $1P^{\circ}$	$4p^{IV}$ $3D^{\circ}$ $1D^{\circ}$	$4p^{IV}$ $3F^{\circ}$ $1F^{\circ}$
$3d^8 4s (2P) nx^V$	$4p^V$ $3P^{\circ}$ $1P^{\circ}$	$4p^V$ $3D^{\circ}$ $1D^{\circ}$	$4p^V$ $3F^{\circ}$ $1F^{\circ}$
$3d^8 4s (2G) nx^{VI}$			$4p^{VI}$ $3F^{\circ}$
$3d^8 4s (2S) nx^{VII}$	$4p^{VII}$ $3P^{\circ}$		$4p^{VII}$ $3F^{\circ}$
	$nf (n \geq 4)$		
	$ng (n \geq 5)$		
$3d^9 (2D) nx$	$4-6f$ $3P^{\circ}$ $1P^{\circ}$	$4-6f$ $3D^{\circ}$ $1D^{\circ}$	$4-6f$ $3F^{\circ}$ $1F^{\circ}$
	$4, 5f$	$4, 5f$	$4, 5f$
	$4p^{VI}$ $3H^{\circ}$ $1H^{\circ}$		$4p^{VI}$ $3H^{\circ}$ $1H^{\circ}$
	$5g^3 D$ $5g^3 D$ $5g^3 D$	$5g^3 F$ $5g^3 F$ $5g^3 F$	$5g^3 G$ $5g^3 G$ $5g^3 G$

*For predicted terms in the spectra of the Ni I isoelectronic sequence, see Vol. II, Introduction.

Cu III

(Co I sequence; 27 electrons)

Z=29

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^9 {}^2D_{3/2}$ $a {}^2D_{3/2}$ 297100 cm^{-1}

I. P. 36.83 volts

The analysis of the Cu III spectrum was begun by Gibbs and Vieweg, who reported in 1929 that they had some important multiplets. It has been seriously incomplete until recently, when Shenstone and Wilets undertook a detailed study of the spectrum, especially for inclusion here.

There are at present approximately 325 classified lines in the interval 672 Å to 2822 Å. According to Shenstone, not a single line of any considerable intensity remains unclassified.

He has determined the limit from the 4F series, by using a Ritz formula, with the assumption that $\alpha=0.18 \times 10^{-6}$, as 299145 cm^{-1} . From a study of isoelectronic data Catalán has derived an improved value, which is quoted here.

The doublet and quartet systems of terms are connected by observed intersystem combinations.

REFERENCES

- R. C. Gibbs and A. M. Vieweg, Phys. Rev. **33**, 1092 (A) (1929).
 A. G. Shenstone and L. Wilets, Phys. Rev. **83**, 104 (1951). (I P) (T) (C L)
 M. A. Catalán, unpublished material (April 1952). (I P)

Cu III

Cu III

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval			
$3d^9$	$a {}^2D$	$2\frac{1}{2}$	0. 0	-2071. 8	$3d^8({}^3F)4p$	$z {}^4G^\circ$	$5\frac{1}{2}$	121698. 5	361. 3			
		$1\frac{1}{2}$	2071. 8				$4\frac{1}{2}$	121337. 2				
$3d^8({}^3F)4s$	$a {}^4F$	$4\frac{1}{2}$	60804. 9	-1260. 1	$3d^8({}^3F)4p$	$z {}^4F^\circ$	$4\frac{1}{2}$	123549. 9	-1007. 6			
		$3\frac{1}{2}$	62065. 0	-1078. 6			$3\frac{1}{2}$	124557. 5	-824. 3			
		$2\frac{1}{2}$	63143. 6	-742. 7			$2\frac{1}{2}$	125381. 8	-362. 8			
		$1\frac{1}{2}$	63886. 3				$1\frac{1}{2}$	125744. 6				
$3d^8({}^3F)4s$	$a {}^2F$	$3\frac{1}{2}$	67016. 6	-1947. 0	$3d^8({}^3F)4p$	$z {}^2G^\circ$	$4\frac{1}{2}$	124442. 5	-1651. 3			
		$2\frac{1}{2}$	68963. 6				$3\frac{1}{2}$	126093. 8				
$3d^8({}^1D)4s$	$b {}^2D$	$2\frac{1}{2}$	77967. 6	-811. 8	$3d^8({}^3F)4p$	$z {}^2F^\circ$	$3\frac{1}{2}$	126829. 4	-1850. 0			
		$1\frac{1}{2}$	78779. 4				$2\frac{1}{2}$	128679. 4				
$3d^8({}^3P)4s$	$a {}^4P$	$0\frac{1}{2}$	80422. 8	-117. 6	$3d^8({}^3F)4p$	$z {}^2D^\circ$	$2\frac{1}{2}$	126891. 9	-1543. 4			
		$1\frac{1}{2}$	80305. 2	246. 3			$1\frac{1}{2}$	128435. 3				
		$2\frac{1}{2}$	80551. 5									
$3d^8({}^3P)4s$	$a {}^2P$	$1\frac{1}{2}$	85446. 4	-686. 6	$3d^8({}^3P)4p$	$z {}^4P^\circ$	$2\frac{1}{2}$	136607. 3	124. 4			
		$0\frac{1}{2}$	86133. 0				$1\frac{1}{2}$	136482. 9	-558. 1			
$3d^8({}^1G)4s$	$a {}^2G$	$4\frac{1}{2}$	89017. 7	-28. 2	$3d^8({}^1D)4p$	$y {}^2F^\circ$	$2\frac{1}{2}$	138084. 0	898. 0			
		$3\frac{1}{2}$	89045. 9				$3\frac{1}{2}$	138982. 0				
$3d^8({}^3F)4p$	$z {}^4D^\circ$	$3\frac{1}{2}$	118864. 3	-1713. 2	$3d^8({}^1D)4p$	$y {}^2D^\circ$	$1\frac{1}{2}$	138988. 1	768. 5			
		$2\frac{1}{2}$	120577. 5				$2\frac{1}{2}$	139756. 6				
		$1\frac{1}{2}$	121863. 7				-1286. 2	$3d^8({}^1D)4p$		$z {}^2P^\circ$	$0\frac{1}{2}$	139260. 7
		$0\frac{1}{2}$	122637. 2								$1\frac{1}{2}$	140200. 9

Cu III—Continued

Cu III—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$3d^8(^3P)4p$	y^4D°	$0\frac{1}{2}$	142550. 1	-37. 8 -86. 0 393. 4	$3d^8(^3F)4d$	3	$3\frac{1}{2}, 4\frac{1}{2}$	194031. 6	
		$1\frac{1}{2}$	142512. 3						
		$2\frac{1}{2}$	142426. 3						
		$3\frac{1}{2}$	142819. 7						
$3d^8(^3P)4p$	x^2D°	$2\frac{1}{2}$	144194. 2	-681. 0	$3d^8(^3F)4d$	e^4G	$2\frac{1}{2}$	194330. 8	
		$1\frac{1}{2}$	144875. 2				$3\frac{1}{2}$		
$3d^8(^3P)4p$	y^2P°	$1\frac{1}{2}$	145353. 0	-1322. 9	$3d^8(^3F)4d$	4	$3\frac{1}{2}, 4\frac{1}{2}$	195060. 4	
		$0\frac{1}{2}$	146675. 9				$3\frac{1}{2}$		
$3d^8(^1G)4p$	z^2H°	$4\frac{1}{2}$	146533. 6	1113. 4	$3d^8(^3F)4d$	6	$3\frac{1}{2}, 4\frac{1}{2}$	195516. 9	
		$5\frac{1}{2}$	147647. 0				$3\frac{1}{2}$		
$3d^8(^3P)4p$	z^2S°	$0\frac{1}{2}$	147652. 5	-857. 0	$3d^8(^3F)5s$	e^2F	$3\frac{1}{2}$	195787. 1	-1611. 3
			148662. 9				$2\frac{1}{2}$		
$3d^8(^1G)4p$	x^2F°	$3\frac{1}{2}$	147805. 9	-857. 0	$3d^8(^3F)4d$	7	$2\frac{1}{2}$	196740. 2	
		$2\frac{1}{2}$	148662. 9				$2\frac{1}{2}, 3\frac{1}{2}$		
$3d^8(^3P)4p$	z^4S°	$1\frac{1}{2}$	147816. 4	199. 2	$3d^8(^3F)4d$	f^2F	$2\frac{1}{2}$	197373. 9	
			153808. 4				$3\frac{1}{2}$		
$3d^8(^1G)4p$	y^2G°	$3\frac{1}{2}$	153609. 2	199. 2	$3d^8(^3F)4d$	9	$1\frac{1}{2}, 2\frac{1}{2}$	197898. 7	
		$4\frac{1}{2}$	153808. 4				$3\frac{1}{2}$		
$3d^8(^3F)5s$	e^4F	$4\frac{1}{2}$	193369. 0	-746. 6 -1437. 6 -891. 7	$3d^8(^3F)4d$	10	$3\frac{1}{2}$	198299. 4	
		$3\frac{1}{2}$	194115. 6						
		$2\frac{1}{2}$	195553. 2						
		$1\frac{1}{2}$	196444. 9						
$3d^8(^3F)4d$	1	$3\frac{1}{2}, 4\frac{1}{2}$	193519. 2		Cu IV (3F_4)	Limit		297100	

March 1951.

Cu III OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms			
$3d^8$	a^2D			
	$ns (n \geq 4)$	$np (n \geq 4)$		$nd (n \geq 4)$
$3d^8(^3F)nx$	{ a, e^4F a, e^2F	z^4D°	z^4F°	z^4G°
$3d^8(^3P)nx$		z^2D°	z^2F°	z^2G°
$3d^8(^1G)nx$	{ a^4P a^2P	z^4S°	z^4P°	y^4D°
$3d^8(^1D)nx$		z^2S°	y^2P°	x^2D°
	a^2G		x^2F°	y^2G°
	b^2D	z^2P°	y^2D°	y^2F°
				z^2H°
				f^2F
				e^4G

*For predicted terms in the spectra of the Co I isoelectronic sequence, see Vol. II, Introduction.

Cu VII

(V I sequence; 23 electrons)

Z=29

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^5 {}^6S_{2\frac{1}{2}}$ $a {}^6S_{2\frac{1}{2}}$ cm⁻¹

I. P. volts

This spectrum is very incompletely known. Kruger and Gilroy have classified three lines as transitions from the ground term to $3d^4 4p {}^6P^\circ$, as follows:

Int.	I. A.	Wave No.	Desig.
200	200. 665	498343	$3d^5 {}^6S_{2\frac{1}{2}} - 4p {}^6P_{3\frac{1}{2}}^\circ$
150	200. 851	497881	$3d^5 {}^6S_{2\frac{1}{2}} - 4p {}^6P_{2\frac{1}{2}}^\circ$
100	200. 948	497640	$3d^5 {}^6S_{2\frac{1}{2}} - 4p {}^6P_{1\frac{1}{2}}^\circ$

REFERENCE

P. G. Kruger and H. T. Gilroy, Phys. Rev. **48**, 720 (1935). (T) (C L)

January 1950.

Cu XIX

(Na I sequence; 11 electrons)

Z=29

Ground state $1s^2 2s^2 2p^6 3s {}^2S_{0\frac{1}{2}}$ $3s {}^2S_{0\frac{1}{2}}$ 5410000 cm⁻¹

I. P. 671 volts

The terms are from Edlén, who lists one pair of classified lines at 47 Å. He has extrapolated the absolute value of the ground term from isoelectronic sequence data.

The unit adopted by Edlén, 10³ cm⁻¹, has here been changed to cm⁻¹.

REFERENCE

B. Edlén, Zeit. Phys. **100**, 621 (1936). (I P) (T) (C L)

Cu XIX

Config.	Desig.	J	Level	Interval
3s	3s ² S	0½	0	
3p	3p ² P°	0½	327700	36100
		1½	363800	
3d	3d ² D	1½	812300	5800
		2½	818100	
4f	4f ² F°	2½	2925200	1000
		3½	2926200	
-----	-----	---	-----	
Cu xx (¹ S ₀)	Limit	---	[5410000]	

June 1947.

ZINC

Zn I

30 electrons

Z=30

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 {}^1S_0$ $4s^2 {}^1S_0$ 75766.8 cm^{-1}

I. P. 9.391 volts

The well established series of Zn I are given by Paschen-Götze and by Fowler.

Hetzler, Boreman, and Burns observed the "*Spectrum of the Zinc Arc in Vacuum*" from 2138 Å to 7799 Å for the purpose of obtaining accurate wave lengths. From their prism, grating, and interferometric measurements they derived precise term values for the leading series terms. Their values are quoted in the table and are the only entries given to two or more decimal places. The limit, and remaining regular series terms are from Fowler. The term $13d {}^3D$ has been calculated from its combination with $4p {}^3P_2^o$, since the absolute value given by Fowler does not fit the observed wave number. Brackets denote that the entry for $5f {}^3F^o$ is calculated from the series formula. The 1D and 3P terms from the $4p^2$ configuration are from Sawyer's paper. These and Fowler's values have been adjusted slightly by the writer, to fit the scale of Hetzler, Boreman, and Burns.

Beutler and Guggenheimer have observed in absorption 24 lines between 713Å and 1109 Å which they classify as transitions from the ground term to terms above the ionization limit having the configurations $3d^9 4s^2({}^2D)np$ and $3d^9 4s^2({}^2D)nf$. The J -values of the limit term in Zn II, (2D), are quoted from their paper. Their values of $8p'' {}^3P^o$ and $9p'' {}^3P^o$, counted from zero, have here been decreased by 10 cm^{-1} to fit the observed wave numbers on which these levels are based. For triplet terms only the component with $J=1$ is indicated in the table, since only combinations with the ground term ($J=0$) are relevant.

Intersystem combinations connecting the singlet and triplet systems of terms have been observed.

Long before the development of the quantum theory of the Zeeman effect, the lines of the Zn I triplet $4p {}^3P^o - 5s {}^3S$, 4680 Å, 4722 Å, and 4810 Å, were among those used to test Preston's postulate that corresponding series lines of different elements are separated in the same manner in a magnetic field of the same strength. Although observed g -values have apparently not been published for Zn I, these lines serve excellently for the calibration of the magnetic field strength in determining the observed Zeeman patterns of various spectra. The quoted values are from Kiess, who furnished them especially for inclusion here.

REFERENCES

- C. Runge and F. Paschen, *Astroph. J.* **15**, 333 (1902). (Z E)
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Zn I

Zn I

Config.	Desig.	J	Level	Interval	Obs. g	Config.	Desig.	J	Level	Interval	Obs. g
$3d^{10} 4s^2$	$4s^2 \ ^1S$	0	0. 000			$3d^{10} 4s(2S) 9p$	$9p \ ^1P^\circ$	1	73469. 7		
$3d^{10} 4s(2S) 4p$	$4p \ ^3P^\circ$	0 1 2	32311. 308 32501. 390 32890. 317	190. 082 388. 927	1. 496 1. 505	$3d^{10} 4s(2S) 8d$	$8d \ ^3D$	1, 2, 3	73471. 1		
$3d^{10} 4s(2S) 4p$	$4p \ ^1P^\circ$	1	46745. 37			$3d^{10} 4s(2S) 10s$	$10s \ ^3S$	1	73698. 6		
$3d^{10} 4s(2S) 5s$	$5s \ ^3S$	1	53672. 241		2. 001	$3d^{10} 4s(2S) 10p$	$10p \ ^1P^\circ$	1	74012. 4		
$3d^{10} 4s(2S) 5s$	$5s \ ^1S$	0	55789. 220			$3d^{10} 4s(2S) 9d$	$9d \ ^3D$	1, 2, 3	74016. 7		
$3d^{10} 4s(2S) 5p$	$5p \ ^3P^\circ$	0 1 2	61247. 2 61273. 9 61330. 1	26. 7 56. 2		$3d^{10} 4s(2S) 11s$	$11s \ ^3S$	1	74169. 0		
$3d^{10} 4s(2S) 4d$	$4d \ ^1D$	2	62458. 51			$3d^{10} 4s(2S) 10d$	$10d \ ^3D$	1, 2, 3	74387. 7		
$3d^{10} 4s(2S) 4d$	$4d \ ^3D$	1 2 3	62768. 75 62772. 00 62776. 95	3. 25 4. 95		$3d^{10} 4s(2S) 12s$	$12s \ ^3S$	1	74495. 8		
$3d^{10} 4s(2S) 5p$	$5p \ ^1P^\circ$	1	62910. 0			$3d^{10} 4s(2S) 11d$	$11d \ ^3D$	1, 2, 3	74651. 3		
$3d^{10} 4s(2S) 6s$	$6s \ ^3S$	1	65432. 32			$3d^{10} 4s(2S) 12d$	$12d \ ^3D$	1, 2, 3	74855. 1		
$3d^{10} 4s(2S) 6s$	$6s \ ^1S$	0	66037. 60			$3d^{10} 4s(2S) 13d$	$13d \ ^3D$	1, 2, 3	74994. 0		
$3d^{10} 4s(2S) 6p$	$6p \ ^3P^\circ$	0 1 2	68070. 854 68080. 669 68101. 784	9. 815 21. 115		$3d^{10} 4s(2S) 14d$	$14d \ ^3D$	1, 2, 3	75112. 2		
$3d^{10} 4s(2S) 5d$	$5d \ ^1D$	2	68338. 48			Zn II ($^2S_{1/2}$)	<i>Limit</i>	-----	75766. 8		
$3d^{10} 4s(2S) 5d$	$5d \ ^3D$	1 2 3	68579. 13 68580. 60 68583. 03	1. 47 2. 43		$3d^{10} 4p^2$	$4p^2 \ ^3P$	0 1 2	80176 80395	219	
$3d^{10} 4s(2S) 6p$	$6p \ ^1P^\circ$	1	68607. 26			$3d^{10} 4p^2$	$4p^2 \ ^1D$	2	80795		
$3d^{10} 4s(2S) 4f$	$4f \ ^3F^\circ$	2, 3, 4	68834. 4			$3d^9 4s^2(2D_{3/2}) 4p$	$4p' \ ^1P^\circ$	1	90158		
$3d^{10} 4s(2S) 7s$	$7s \ ^3S$	1	69745. 94			$3d^9 4s^2(2D_{1/2}) 4p$	$4p' \ ^3P^\circ$	1	94717		
$3d^{10} 4s(2S) 7s$	$7s \ ^1S$	0	70003. 72			$3d^9 4s^2(2D_{1/2}) 4p$	$4p' \ ^3D^\circ$	1	95257		
$3d^{10} 4s(2S) 7p$	$7p \ ^3P^\circ$	0 1 2	70977. 14 70981. 97 70992. 16	4. 83 10. 19		$3d^9 4s^2(2D_{3/2}) 5p$	$5p' \ ^1P^\circ$	1	123470		
$3d^{10} 4s(2S) 6d$	$6d \ ^1D$	2	71050. 45			$3d^9 4s^2(2D_{1/2}) 5p$	$5p' \ ^3P^\circ$	1	125968		
$3d^{10} 4s(2S) 6d$	$6d \ ^3D$	1 2 3	71212. 13 71212. 90 71214. 24	0. 77 1. 34		$3d^9 4s^2(2D_{3/2}) 5p$	$5p' \ ^3D^\circ$	1	126255		
$3d^{10} 4s(2S) 7p$	$7p \ ^1P^\circ$	1	71219. 08			$3d^9 4s^2(2D_{3/2}) 6p$	$6p' \ ^1P^\circ$	1	130617		
$3d^{10} 4s(2S) 5f$	$5f \ ^3F^\circ$	2, 3, 4	[71323. 4]			$3d^9 4s^2(2D_{3/2}) 4f$	$4f' \ ^3P^\circ?$	1	131671		
$3d^{10} 4s(2S) 8s$	$8s \ ^3S$	1	71822. 5			$3d^9 4s^2(2D_{1/2}) 6p$	$6p' \ ^3P^\circ$	1	133214		
$3d^{10} 4s(2S) 8s$	$8s \ ^1S$	0	71956. 20			$3d^9 4s^2(2D_{1/2}) 6p$	$6p' \ ^3D^\circ$	1	133349		
$3d^{10} 4s(2S) 7d$	$7d \ ^1D$	2	72489. 13			$3d^9 4s^2(2D_{1/2}) 4f$	$4f' \ ^1P^\circ?$	1	133478		
$3d^{10} 4s(2S) 8p$	$8p \ ^3P^\circ$	0 1 2	72495. 8 72498. 56 72504. 20	2. 8 5. 64		$3d^9 4s^2(2D_{3/2}) 7p$	$7p' \ ^1P^\circ$	1	133638		
$3d^{10} 4s(2S) 8p$	$8p \ ^1P^\circ$	1	72626. 2			$3d^9 4s^2(2D_{3/2}) 5f$	$5f' \ ^3P^\circ?$	1	133989		
$3d^{10} 4s(2S) 7d$	$7d \ ^3D$	1, 2, 3	72627. 9			$3d^9 4s^2(2D_{3/2}) 8p$	$8p' \ ^1P^\circ$	1	135185		
$3d^{10} 4s(2S) 9s$	$9s \ ^3S$	1	72985. 4			$3d^9 4s^2(2D_{3/2}) 9p$	$9p' \ ^1P^\circ$	1	136066		
$3d^{10} 4s(2S) 9s$	$9s \ ^1S$	0	73060. 63			$3d^9 4s^2(2D_{1/2}) 7p$	$7p' \ ^3P^\circ$	1	136292		
$3d^{10} 4s(2S) 9p$	$9p \ ^3P^\circ$	0 1 2	73390. 7 73392. 27 73395. 86	1. 6 3. 59		$3d^9 4s^2(2D_{3/2}) 10p$	$10p' \ ^1P^\circ$	1	136656		
						$3d^9 4s^2(2D_{3/2}) 11p$	$11p' \ ^1P^\circ$	1	137060		
						$3d^9 4s^2(2D_{3/2}) 12p$	$12p' \ ^1P^\circ$	1	137344		
						$3d^9 4s^2(2D_{1/2}) 8p$	$8p' \ ^3P^\circ$	1	137883		
						$3d^9 4s^2(2D_{1/2}) 9p$	$9p' \ ^3P^\circ$	1	138813		
						$3d^9 4s^2(2D_{1/2}) 10p$	$10p' \ ^3P^\circ$	1	139377		
						$3d^9 4s^2(2D_{1/2}) 11p$	$11p' \ ^3P^\circ$	1	139762		
						$3d^9 4s^2(2D_{1/2}) 12p$	$12p' \ ^3P^\circ$	1	140059		

Zn I OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ +	Observed Terms			
3d ¹⁰ 4s ²	4s ² 1S			
3d ¹⁰ 4p ²	{ 4p ² 3P 4p ² 1D			
	ns (n ≥ 5)	np (n ≥ 4)	nd (n ≥ 4)	nf (n ≥ 4)
3d ¹⁰ 4s(2S)nx	{ 5-12s 3S 5- 9s 1S	4- 9p 3P° 4-10p 1P°	4-14d 3D 4- 7d 1D	4f 3F°
3d ⁹ 4s ² (2D)nx'		4-12p' 3P° 4-6p' 3D° 4-12p' 1P°		4, 5f' 3P° 4f' 1P°

*For predicted terms in the spectra of the Zn I isoelectronic sequence, see Vol. II, Introduction.

Zn II

(Cu I sequence; 29 electrons)

Z=30

Ground state 1s² 2s² 2p⁶ 3s² 3p⁶ 3d¹⁰ 4s 2S_{0½}

4s 2S_{0½} 144890.6 cm⁻¹

I. P. 17.96 volts

In 1925 von Salis published an analysis giving well established doublet series in Zn II, from which most of the terms in the table are taken. He derived the limit from the 2D-series, by using a Ritz formula.

From infrared observations Paschen and Ritschl resolved the 4f 2F° term and found the 5g 2G-term.

By analogy with Cu I, terms from 3d⁹ 4s nx are to be expected. Takahashi, Kalia, and Mazumder have published, respectively, 21, 8, and 26 levels, and grouped some of them into terms from this configuration. The first and last of these three papers are, however, not independent. Mazumder utilizes 6 levels given by Takahashi in making up the 6 terms he suggests.

More recently Shenstone and Gibson have investigated this spectrum and furnished their results in advance of publication, for inclusion here. They ascribe 8 terms to 3d⁹ 4s 4p and confirm 6 levels by Takahashi and 10 by Mazumder, of which 4 are common to both lists. Shenstone's interpretation of the terms from this configuration has been adopted here. He has also revised 7p 2P°, added 8p 2P°, and resolved 8f 2F°. His revised term values have been used throughout, for the present compilation. The remaining published levels have been omitted, awaiting further confirmation.

There are approximately 120 classified lines in the interval between 984 Å and 9950 Å, including the infrared data, Lang's ultraviolet material, and the lists by von Salis and Shenstone. Observed intersystem combinations connect the doublet and quartet systems of terms.

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Zn II

Zn II

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
3d ¹⁰ (1S)4s	4s ² S	0½	0. 0		3d ¹⁰ (1S)5g	5g ² G	4½ 3½	127310. 1 127310. 2	-0. 1
3d ¹⁰ (1S)4p	4p ² P°	0½ 1½	48480. 6 49354. 4	873. 8	3d ¹⁰ (1S)6d	6d ² D	1½ 2½	127629. 9 127643. 1	13. 2
3d ⁹ 4s ²	4s ² ² D	2½ 1½	62721. 9 65441. 1	-2719. 2	3d ¹⁰ (1S)7p	7p ² P°	1½ 0½	128343. 5 128517. 9	-174. 4
3d ¹⁰ (1S)5s	5s ² S	0½	88436. 8		3d ⁹ 4s(1D)4p	4p'' ² P°	1½ 0½	130371. 4	
3d ¹⁰ (1S)4d	4d ² D	1½ 2½	96909. 0 96959. 7	50. 7	3d ⁹ 4s(1D)4p	4p'' ² D°	2½ 1½	131651. 7	
3d ¹⁰ (1S)5p	5p ² P°	0½ 1½	101365. 3 101610. 8	245. 5	3d ¹⁰ (1S)8s	8s ² S	0½	131876. 0	
3d ⁹ 4s(3D)4p	4p' ⁴ P°	2½ 1½ 0½	103690. 2 105322. 5 106527. 3	-1632. 3 -1204. 8	3d ¹⁰ (1S)6f	6f ² F°	2½ 3½	132603. 3 132638. 7	35. 4
3d ⁹ 4s(3D)4p	4p' ⁴ F°	4½ 3½ 2½ 1½	106841. 9 107267. 6	-425. 7	3d ¹⁰ (1S)6g	6g ² G	{ 4½ 3½ }	132683. 4	
3d ⁹ 4s(3D)4p	4p' ⁴ D°	3½ 2½ 1½ 0½	111743. 2 111993. 7 112534. 7	-250. 5 -541. 0	3d ¹⁰ (1S)7d	7d ² D	1½ 2½	132879. 8 132887. 7	7. 9
3d ⁹ 4s(3D)4p	4p' ² P°	0½ 1½	113492. 7 113499. 0	6. 3	3d ¹⁰ (1S)8p	8p ² P°	{ 1½ 0½ }	133805	
3d ⁹ 4s(3D)4p	4p' ² D°	1½ 2½	114044. 1 114834. 0	789. 9	3d ¹⁰ (1S)9s	9s ² S	0½	135423. 0	
3d ⁹ 4s(3D)4p	4p' ² F°	3½ 2½	114208. 3		3d ¹⁰ (1S)7f	7f ² F°	3½ 2½	135889. 3 135892. 2	-2. 9
3d ¹⁰ (1S)6s	6s ² S	0½	114498. 5		3d ¹⁰ (1S)7g	7g ² G	{ 4½ 3½ }	135923. 3	
3d ¹⁰ (1S)4f	4f ² F°	3½ 2½	117262. 7 117263. 3	-0. 6	3d ¹⁰ (1S)8d	8d ² D	1½ 2½	136051. 2 136056. 3	5. 1
3d ¹⁰ (1S)5d	5d ² D	1½ 2½	117968. 7 117992. 9	24. 2	3d ¹⁰ (1S)8f	8f ² F°	3½ 2½	138001. 3 138002. 6	-1. 3
3d ¹⁰ (1S)6p	6p ² P°	0½ 1½	119887. 9 119959. 0	71. 1	3d ¹⁰ (1S)9d	9d ² D	1½ 2½	138113. 7 138117. 3	3. 6
3d ¹⁰ (1S)7s	7s ² S	0½	125879. 5		Zn III(1S ₀)	Limit		144890. 6	
3d ¹⁰ (1S)5f	5f ² F°	3½ 2½	127199. 0 127208. 6	-9. 6					

February 1950.

Zn II OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ +	Observed Terms					
3d ⁹ 4s ²	4s ² ² D					
	<i>ns</i> (<i>n</i> ≥ 4)	<i>np</i> (<i>n</i> ≥ 4)		<i>nd</i> (<i>n</i> ≥ 4)	<i>nf</i> (<i>n</i> ≥ 4)	<i>ng</i> (<i>n</i> ≥ 5)
3d ¹⁰ (1S) <i>nx</i>	4-9s ² S	4-8p ² P°		4-9d ² D	4-8f ² F°	5-7g ² G
3d ⁹ 4s(3D) <i>nx'</i>	{	4p' ⁴ P°	4p' ⁴ D°	4p' ⁴ F°		
		4p' ² P°	4p' ² D°	4p' ² F°		
3d ⁹ 4s(1D) <i>nx''</i>		4p'' ² P°	4p'' ² D°			

*For predicted terms of the Cu I isoelectronic sequence, see Vol. II, Introduction.

Zn III

(Ni I sequence; 28 electrons)

 $Z=30$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 1S_0$ $3d^{10} 1S_0$ 320300 cm^{-1}

I. P. 39.70 volts

The analysis is incomplete. In 1927 Laporte and Lang published 9 terms from the 2D limit in Zn IV, and classified 37 lines between 677 Å and 1839 Å. By extrapolation along the iso-electronic sequence they estimated the limit to be about 325000 cm^{-1} .

From more extensive observations, Mazumder has classified 219 more lines and found 24 new terms in the range between 497 Å and 3133 Å. His value of the limit, based on the $ns \ ^3D$ series ($n=4,5$) and derived from a Rydberg formula, is 324405 cm^{-1} . From a study of iso-electronic data Catalán has derived an improved value, which is quoted here.

The terms in the table are from these two papers. On the suggestion of Mack, Laporte, and Lang the two levels $4p \ ^3D_1^\circ$ and $4p \ ^1P_1^\circ$ have been interchanged, in order to agree better with the isoelectronic sequence data for Ga IV and Ge V.

The terms of all three multiplicities are connected by observed intersystem combinations.

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Zn III

Zn III

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3d^{10}$	$3d^{10} 1S$	0	0		$3d^9(^2D)4d$	$4d \ ^3G$	5	216617	144
$3d^9(^2D)4s$	$4s \ ^3D$	3	78105	-1178			4	216473	-609
		2	79283	-1576			3	217082	
		1	80859		$3d^9(^2D)4d$	$4d \ ^1D$	2	216903	
$3d^9(^2D)4s$	$4s \ ^1D$	2	83509		$3d^9(^2D)4d$	$4d \ ^3P$	2	217664	-1254
$3d^9(^2D)4p$	$4p \ ^3P^\circ$	2	137876	-2204			1	218918	-2445
		1	140080	-1321			0	221363	
		0	141401		$3d^9(^2D)4d$	$4d \ ^3D$	3	218048	-500
$3d^9(^2D)4p$	$4p \ ^3F^\circ$	4	141335	671			2	218548	-2802
		3	140664	-1827			1	221350	
		2	142491		$3d^9(^2D)4d$	$4d \ ^1F$	3	219352	
$3d^9(^2D)4p$	$4p \ ^3D^\circ$	3	144511	-741	$3d^9(^2D)4d$	$4d \ ^3F$	4	219368	-324
		2	145252	-2325			3	219692	-982
		1	147577				2	220674	
$3d^9(^2D)4p$	$4p \ ^1F^\circ$	3	145974		$3d^9(^2D)4d$	$4d \ ^1P$	1	221332	
$3d^9(^2D)4p$	$4p \ ^1P^\circ$	1	147505		$3d^9(^2D)5s$	$5s \ ^1D$	2	227854	
$3d^9(^2D)4p$	$4p \ ^1D^\circ$	2	147928		$3d^8 \ 4s(^4F)4p$	$4p' \ ^5G^\circ$	6		
$3d^9(^2D)4d$	$4d \ ^3S$	1	214362				5		
$3d^9(^2D)5s$	$5s \ ^3D$	3	214885	-462			4	252029	-551
		2	215347	-2322			3	252580	-3721
		1	217669		$3d^8 \ 4s(^4F)4p$	$4p' \ ^5F^\circ$	2	256301	
$3d^9(^2D)4d$	$4d \ ^1G$	4	215121				5		
$3d^9(^2D)4d$	$4d \ ^1S$	0	215476				4	252287	-128
							3	252115	-1288
							2	253703	-1932
							1	255635	

Zn III—Continued

Zn III—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval	
$3d^8 4s(4F)4p$	$4p' \ ^5D^\circ$	4	252311	-1821	$3d^8 4s(2F)4p$	$4p'' \ ^3G^\circ$	5	274154	-1113	
		3	254132				4			275267
		2	253739				3			
		1	255447							
		0	258072							
$3d^8 4s(4F)4p$	$4p' \ ^3G^\circ$	5	263829	306	$3d^8 4s(2F)4p$	$4p'' \ ^1D^\circ$	2	274339	-2111	
		4	263523				3	276128		
		3					2	278239		
$3d^8 4s(4F)4p$	$4p' \ ^3D^\circ$	3	265492	-4511	$3d^8 4s(2F)4p$	$4p'' \ ^1F^\circ$	3	276748	-1977	
		2	270003				4	277604		
		1	272104							
$3d^8 4s(4F)4p$	$4p' \ ^3F^\circ$	4	265538	-423	$3d^8 4s(2F)4p$	$4p'' \ ^3F^\circ$	2	277968	-644	
		3	265961				3	277324		
		2	272136				4	278139		
					Zn IV ($^2D_{3/2}$)	Limit		320300		

March 1951.

Zn III OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms				
$3d^{10}$	$3d^{10} \ ^1S$				
	$ns (n \geq 4)$	$np (n \geq 4)$			$nd (n \geq 4)$
$3d^9(2D)nx$	{ 4, 5s 3D 4, 5s 1D	$4p \ ^3P^\circ$ $4p \ ^1P^\circ$	$4p \ ^3D^\circ$ $4p \ ^1D^\circ$	$4p \ ^3F^\circ$ $4p \ ^1F^\circ$	$4d \ ^3S$ $4d \ ^3P$ $4d \ ^3D$ $4d \ ^3F$ $4d \ ^3G$ $4d \ ^1S$ $4d \ ^1P$ $4d \ ^1D$ $4d \ ^1F$ $4d \ ^1G$
$3d^8 4s(4F)nx'$	{		$4p' \ ^5D^\circ$ $4p' \ ^3D^\circ$	$4p' \ ^5F^\circ$ $4p' \ ^3F^\circ$	$4p' \ ^5G^\circ$ $4p' \ ^3G^\circ$
$3d^8 4s(2F)nx''$	{		$4p'' \ ^3D^\circ$ $4p'' \ ^1D^\circ$	$4p'' \ ^3F^\circ$ $4p'' \ ^1F^\circ$	$4p'' \ ^3G^\circ$ $4p'' \ ^1G^\circ$

*For predicted terms in the spectra of the Ni I isoelectronic sequence, see Vol. II, Introduction.

Zn IV

(Co I sequence; 27 electrons)

 $Z=30$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^9 \ ^2D_{3/2}$ $3d^9 \ ^2D_{3/2}$ cm⁻¹

I. P. volts

This spectrum has not been analyzed. Nine terms and 48 classified lines between 1029 Å and 1443 Å have been suggested by Subbaraya. Bloch and Bloch have also published 32 classified lines between 466 Å and 1449 Å but give no terms. A tolerance in the intervals of 35 cm⁻¹ in the former paper and 22 cm⁻¹ in the latter makes the analysis appear dubious in both cases, particularly since only one interval and possibly four lines are common to both papers.

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L. Bloch et E. Bloch, Ann. de Phys. [11] **5**, 348 (1936). (C L)

March 1950.

GALLIUM

Ga I

31 electrons

Z=31

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^2 P_{0\frac{1}{2}}^{\circ}$ $4p^2 P_{0\frac{1}{2}}^{\circ}$ 48380 cm^{-1}

I. P. 6.00 volts

Uhler and Tanch observed the spectrum from 2200 to 4172 Å, and extended two of the doublet series (2S and 2D) given by Fowler and by Paschen-Götze. Later Sawyer and Lang found the two terms $4p^2 \ ^2S$ and $4p^2 \ ^4P$ from their combination with $4p^2 \ ^2P^{\circ}$. Garton has since observed six intense absorption lines between 1500 and 1650 Å, which he classifies as combinations of the ground term with $4p^2 \ ^2S$ and $4p^2 \ ^2P$. His absolute values for these two terms, calculated from the limit 48380, have been used for the present compilation. His work indicates a revision of Sawyer's value of $4p^2 \ ^2S$, which, counted from zero, is 40265.

Meggers and Murphy have recently photographed the spectrum from 6396 to 12109 Å. They find the strong lines due to the transition $5s \ ^2S-5p \ ^2P^{\circ}$ at 11949.24 and 12109.93 Å, thus confirming the early predictions which placed them near 12000 Å. They have also extended the two series $5p \ ^2P^{\circ}-ns \ ^2S$, and $5p \ ^2P^{\circ}-nd \ ^2D$ to $n=12$ and 11, respectively, and have observed the $nf \ ^2F^{\circ}$ series ($n=4-7$).

The writer has recalculated all of the term values with the aid of the new data. Fowler's limit, quoted above, is confirmed. The number of classified lines is now approximately 70.

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Ga I

Ga I

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$4s^2(^1S)4p$	$4p \ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	0.00 326.24	826.24	$4s^2(^1S)6p$	$6p \ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	40376.46 40417.62	41.16
$4s^2(^1S)5s$	$5s \ ^2S$	$0\frac{1}{2}$	24788.58		$4s^2(^1S)5d$	$5d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	40802.72 40811.20	8.48
$4s^2(^1S)5p$	$5p \ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	33044.06 33155.03	110.97	$4s^2(^1S)7s$	$7s \ ^2S$	$0\frac{1}{2}$	42158.44	
$4s^2(^1S)4d$	$4d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	34781.67 34787.92	6.25	$4s^2(^1S)7p$	$7p \ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	43440.8 43461.7	20.9
$4s^2(^1S)6s$	$6s \ ^2S$	$0\frac{1}{2}$	37584.62		$4s^2(^1S)6d$	$6d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	43575.38 43580.59	5.21
$4s \ 4p^2$	$4p^2 \ ^4P$	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	37972 38338 38913	366 575	$4s^2(^1S)4f$	$4f \ ^2F^{\circ}$	$2\frac{1}{2}, 3\frac{1}{2}$	43954.71	
					$4s^2(^1S)8s$	$8s \ ^2S$	$0\frac{1}{2}$	44331.9	

Ga I—Continued

Ga I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
4s ² (¹ S)7 <i>d</i>	7 <i>d</i> ² D	1½ 2½	45072. 0 45076. 11	4. 1	4s ² (¹ S)10 <i>d</i>	10 <i>d</i> ² D	1½, 2½	46941. 6	
4s ² (¹ S)9 <i>s</i>	9 <i>s</i> ² S	0½	45536. 66		4s ² (¹ S)7 <i>f</i>	7 <i>f</i> ² F°	2½, 3½	47022. 0	
4s ² (¹ S)8 <i>d</i>	8 <i>d</i> ² D	1½ 2½	45969. 2 45972. 00	2. 8	4s ² (¹ S)12 <i>s</i>	12 <i>s</i> ² S	0½	47092. 52	
4s ² (¹ S)5 <i>f</i>	5 <i>f</i> ² F°	2½, 3½	46130. 7		4s ² (¹ S)11 <i>d</i>	11 <i>d</i> ² D	1½, 2½	47222. 10	
4s ² (¹ S)10 <i>s</i>	10 <i>s</i> ² S	0½	46273. 9	0. 6	Ga II(¹ S ₀)		<i>Limit</i>	48380	626
4s ² (¹ S)9 <i>d</i>	9 <i>d</i> ² D	1½ 2½	46548. 3 46548. 9		4s 4 <i>p</i> ²	4 <i>p</i> ² ² S	0½	62100	
4s ² (¹ S)6 <i>f</i>	6 <i>f</i> ² F°	2½, 3½	46660. 8		4s 4 <i>p</i> ³	4 <i>p</i> ² ² P	0½ 1½	65796 66422	
4s ² (¹ S)11 <i>s</i>	11 <i>s</i> ² S	0½	46758. 2						

November 1951.

Ga I OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 3d ¹⁰ +	Observed Terms			
4s ² (¹ S)4 <i>p</i>	4 <i>p</i> ² P°			
4s 4 <i>p</i> ²	{ 4 <i>p</i> ² ² S 4 <i>p</i> ² ⁴ P 4 <i>p</i> ² ² P			
	<i>ns</i> (<i>n</i> ≥ 5)	<i>np</i> (<i>n</i> ≥ 5)	<i>nd</i> (<i>n</i> ≥ 4)	<i>nf</i> (<i>n</i> ≥ 4)
4s ² (¹ S) <i>nx</i>	5–12 <i>s</i> ² S	5–7 <i>p</i> ² P°	4–11 <i>d</i> ² D	4–7 <i>f</i> ² F°

*For predicted terms in the spectra of the Ga I isoelectronic sequence, see Vol. II, Introduction.

Ga II

(Zn I sequence; 30 electrons)

Z = 31Ground state 1s² 2s² 2p⁶ 3s² 3p⁶ 3d¹⁰ 4s² ¹S₀4s² ¹S₀ 165458 cm⁻¹

I. P. 20.51 volts

The analysis has not been extended since 1929 when Sawyer and Lang classified about 90 lines in the region between 829 Å and 7793 Å. The terms are from their paper, predicted series members being entered in brackets. They have derived absolute values from the various series "in such a way as to make the deviations from true Rydberg type similar to those found in Al II." An evident misprint in their value of 7*p* ¹P° has here been corrected to fit the observed combinations.

The singlet and triplet terms are connected by observed intersystem combinations.

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- R. A. Sawyer and R. J. Lang, Phys. Rev. **34**, 712 (1929). (I P) (T) (C L)
J. E. Mack, Rev. Mod. Phys. **22**, No. 1, 64 (1950). (Summary hfs)

Ga II

Ga II

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
4s ²	4s ² 1S	0	0		4s(2S)4f	4f 1F ^o	3	137343	
4s(2S)4p	4p 3P ^o	0	47370	446	4s(2S)6p	6p 3P ^o	0, 1, 2	[139260]	
		1	47316	934	4s(2S)6p	6p 1P ^o	1	[139660]	
		2	48750		4s(2S)6d	6d 1D	2	139694	
4s(2S)4p	4p 1P ^o	1	70700		4s(2S)7s	7s 3S	1	145493	
4s(2S)5s	5s 3S	1	102943		4s(2S)7s	7s 1S	0	146013	
4s(2S)5s	5s 1S	0	106656		4s(2S)5f	5f 3F ^o	2, 3, 4	147483	
4s(2S)4d	4d 1D	2	107719		4s(2S)5f	5f 1F ^o	3	147492	
4s(2S)4d	4d 3D	1	113816	25	4s(2S)6d	6d 3D	1	147516	
		2	113841	34			2	147524	
		3	113875				3	147533	
4p ²	4p ² 3P	0	114701	524	4s(2S)7d	7d 1D	2	148437	
		1	115225	912	4s(2S)7p	7p 3P ^o	0, 1, 2	[148620]	
		2	116137		4s(2S)7p	7p 1P ^o	1	148707	
4s(2S)5p	5p 3P ^o	0	118427	89	4s(2S)8s	8s 3S	1	151919	
		1	118516	210	4s(2S)8s	8s 1S	0	152193	
		2	118726		4s(2S)6f	6f 3F ^o	2, 3, 4	153000	
4s(2S)5p	5p 1P ^o	1	120540		4s(2S)6f	6f 1F ^o	3	[153020]	
4s(2S)5d	5d 1D	2	[127190]		4s(2S)7d	7d 3D	1, 2, 3	153061	
4s(2S)6s	6s 3S	1	133012						
4s(2S)6s	6s 1S	0	[134310]						
4s(2S)5d	5d 3D	1	137154	12					
		2	137166	17					
		3	137183						
4s(2S)4f	4f 3F ^o	2, 3, 4	137333		Ga III (2S _{1/2})	Limit		165453	

8
9

March 1950.

Ga II OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 3d ¹⁰ +	Observed Terms			
4s ²	4s ² 1S			
4p ²	4p ² 3P			
	<i>ns</i> (<i>n</i> ≥ 4)	<i>np</i> (<i>n</i> ≥ 4)	<i>nd</i> (<i>n</i> ≥ 4)	<i>nf</i> (<i>n</i> ≥ 4)
4s(2S) <i>nx</i>	{ 5 - 8s 3S 5, 7, 8s 1S	4, 5p 3P ^o 4, 5, 7p 1P ^o	4-7d 3D 4, 6, 7d 1D	4-6f 3F ^o 4, 5f 1F ^o

*For predicted terms in the spectra of the Zn I isoelectronic sequence see Vol. II, Introduction.

Ga III

(Cu I sequence; 29 electrons)

 $Z=31$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 S_{0\frac{1}{2}}$ $4s^2 S_{0\frac{1}{2}}$ 247700 cm^{-1}

I. P. 30.70 volts

The early analysis was by Carroll, who reported six doublet series terms in 1925, and determined the limit as 247790 by assuming an absolute value of $5g^2G$ consistent with extrapolated isoelectronic sequence data.

From improved measurements Lang revised Carroll's absolute term values slightly and added the second member of the principal series. His classification of two faint lines at 632.92 Å and 635.02 Å, respectively, as $4s^2S-5p^2P^o$ was confirmed only by a rough extrapolation of the Rydberg formula. This places $5p^2P_{1\frac{1}{2}}^o$ at 157998 and $5p^2P_{0\frac{1}{2}}^o$ at 157475 (with $4s^2S=0$). The quoted value of the limit is an estimate by Catalán based on isoelectronic data.

Subsequently Rao extended the observations, added the terms $6s^2S$ and $5d^2D$, and revised Lang's value of $5p^2P^o$ on the basis of combinations of this term with $5s^2S$ and $4d^2D$. The revised value of $5p^2P^o$ has been adopted here. With this exception, Lang's values have been used in the present compilation, supplemented by the new terms from Rao. A correction of 10 cm^{-1} has here been added to the three terms from Rao's paper (from the ground state zero) to reduce them to Lang's scale. About 20 lines have been classified in the interval 1267 Å to 5992 Å.

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 R. J. Lang, Phys. Rev. **30**, 762 (1927). (I P) (T) (C L)
 K. R. Rao, A. L. Narayan, and A. S. Rao, Indian J. Phys. **2**, 483 (1928). (C L)
 M. A. Catalán, unpublished material (April 1952). (I P)

Ga III

Config.	Desig.	J	Level	Interval
4s	$4s^2S$	$0\frac{1}{2}$	0	
4p	$4p^2P^o$	$0\frac{1}{2}$ $1\frac{1}{2}$	65167 68885	1718
5s	$5s^2S$	$0\frac{1}{2}$	140744	
4d	$4d^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	144078 144195	117
5p	$5p^2P^o$	$0\frac{1}{2}$ $1\frac{1}{2}$	160761 161300	539
4f	$4f^2F^o$	$2\frac{1}{2}$ $3\frac{1}{2}$	185431 185438	7
6s	$6s^2S$	$0\frac{1}{2}$	187562	
5d	$5d^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	189183 189246	63
5g	$5g^2G$	{ $3\frac{1}{2}$ $4\frac{1}{2}$ }	208252	
-----	-----	-----	-----	-----
Ga IV (1S_0)	Limit	-----	247700	

March 1951.

Ga IV

(Ni I sequence; 28 electrons)

Z=31

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 1S_0$ $3d^{10} 1S_0$ 517600 cm^{-1}

I. P. 64.2 volts

Mack, Laporte, and Lang have classified 39 lines in the interval 422 Å to 1465 Å, as combinations among 9 terms. They have determined the limit by linear extrapolation of the Moseley diagram "slightly modified by a consideration of the first order screening numbers." The value of the limit calculated from their quoted ionization potential is entered in brackets in the table.

Intersystem combinations connecting the singlet and triplet terms, have been observed.

REFERENCE

J. E. Mack, O. Laporte, and R. J. Lang, Phys. Rev. **31**, 748 (1928). (I P) (T) (C L)

Ga IV

Config.	Desig.	<i>J</i>	Level	Interval
$3d^{10}$	$3d^{10} 1S$	0	0	
$3d^9 ({}^2D)4s$	$4s {}^3D$	3	149298	-1455
		2	150753	-2120
		1	152873	
$3d^9 ({}^2D)4s$	$4s {}^1D$	2	155810	
$3d^9 ({}^2D)4p$	$4p {}^3P^o$	2	224028	-3081
		1	227109	-1820
		0	228929	
$3d^9 ({}^2D)4p$	$4p {}^3F^o$	4	228742	1272
		3	227470	-2358
		2	229328	
$3d^9 ({}^2D)4p$	$4p {}^3D^o$	3	232976	-636
		2	233612	-3080
		1	236692	
$3d^9 ({}^2D)4p$	$4p {}^1F^o$	3	234724	
$3d^9 ({}^2D)4p$	$4p {}^1P^o$	1	236099	
$3d^9 ({}^2D)4p$	$4p {}^1D^o$	2	237243	
-----	-----	-----	-----	
Ga V (${}^2D_{2\frac{1}{2}}$)	<i>Limit</i>	-----	[517600]	

November 1949.

Ga IV OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms				
$3d^{10}$	$3d^{10} 1S$				
	$ns (n \geq 4)$		$np (n \geq 4)$		
$3d^9 ({}^2D)nx$	{	$4s {}^3D$	$4p {}^3P^o$	$4p {}^3D^o$	$4p {}^3F^o$
		$4s {}^1D$	$4p {}^1P^o$	$4p {}^1D^o$	$4p {}^1F^o$

*For predicted terms in the spectra of the Ni I isoelectronic sequence, see Vol. II, Introduction.

GERMANIUM

Ge I

32 electrons

 $Z=32$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^2 \ ^3P_0$ $4p^2 \ ^3P_0$ 63600 cm^{-1}

I. P. 7.88 volts

K. R. Rao, in 1929, revised and extended the early work of Gartlein and others, and published a list of 139 classified lines between 1639 Å and 4685 Å. Kiess, in 1940, added 7 terms, from the $5p$ and $6p$ configurations, and 25 classified lines from his observations between 5265 Å and 11144 Å. The terms in the table are from these two sources, although it is difficult to choose between the arrangement suggested by Gartlein and that by Rao for the terms which they interpret differently.

Andrew and Meissner have contributed new terms from the np configuration ($n=5$ to 7), and adjusted some other values. Their results are from unpublished material furnished especially for inclusion here.

The limit is from Gartlein, who derived it from the $ns \ ^1, \ ^3P^\circ$ series ($n=5$ to 7) and the $nd \ ^1P^\circ D^\circ F^\circ, \ ^3P^\circ D^\circ F^\circ$ series ($n=5, 6$) by the use of a Rydberg formula. Rao's value from the $ns \ ^1, \ ^3P^\circ$ series ($n=5$ to 7) is 63790 cm^{-1} .

Two miscellaneous levels given by Rao have been omitted from the table. With the ground state equal to zero they are as follows:

Desig.	Level
$^1X_1^\dagger$	59522.5
$^3X_2^\ddagger$	61250.9

The singlet and triplet systems of terms are connected by observed intersystem combinations.

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- C. W. Gartlein, *Phys. Rev.* **31**, 782 (1928). (I P) (T) (C L)
 K. R. Rao, *Proc. Roy. Soc. London [A]* **124**, 465 (1929). (I P) (T) (C L)
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 J. C. van den Bosch and P. F. A. Klinkenberg, *Proc. Ned. Akad. Wetensch. Amsterdam* **44**, No. 5, 559 (1941). (C L) (Z E)
 J. E. Mack, *Rev. Mod. Phys.* **22**, No. 1, 64 (1950). (Summary hfs)
 K. L. Andrew and K. W. Meissner, unpublished material (October 1951). (T) (C L)

Ge I

Ge I

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
4s ² 4p ²	4p ² ³ P	0	0. 00	557. 11 852. 80	1. 476 1. 514	4s 4p ³	4p ³ ¹ P ^o	1	55473. 6	1183. 89 260. 64	
		1	557. 10					0	55503. 10		
		2	1409. 90						1		
4s ² 4p ²	4p ² ¹ D	2	7125. 26		0. 989	4s ² 4p(2P ^o) 6p	6p ³ P	2	56947. 63		
4s ² 4p ²	4p ² ¹ S	0	16367. 14			4s ² 4p(2P ^o) 5d	5d ¹ D ^o	2	55717. 6		
4s ² 4p(2P ^o) 5s	5s ³ P ^o	0	37451. 53	250. 62 1415. 62	1. 442 1. 499	4s 4p ³	4p ³ ³ P ^o	2	56654. 9	-742. 2 -278. 0	
		1	37702. 15					1	57397. 1		
		2	39117. 77					0	57675. 1		
4s ² 4p(2P ^o) 5s	5s ¹ P ^o	1	40020. 43		1. 058	4s ² 4p(2P ^o) 6p	6p ¹ S	0	56772. 63		
4s ² 4p(2P ^o) 5p	5p ¹ P	1	45985. 45			4s ² 4p(2P ^o) 5d	5d ³ F ^o	2	56828. 3		
4s ² 4p(2P ^o) 5p	5p ³ D	1	46765. 12	69. 11 1269. 74		4s ² 4p(2P ^o) 6p	6p ³ S	1		57083. 07	
		2	46834. 23					3			
		3	48103. 97					4			
4s ² 4p(2P ^o) 5p	5p ³ P	0	47502. 50	585. 70 637. 76		4s ² 4p(2P ^o) 7s	7s ³ P ^o	0	57166. 5	-245. 9 2010. 8	
		1	48088. 20					1	56920. 6		
		2	48725. 96					2	58931. 4		
4s ² 4p(2P ^o) 4d	4d ¹ D ^o	2	48479. 7			4s ² 4p(2P ^o) 5d	5d ³ P ^o	2	57178. 5	-252. 2 -265. 4	
4s ² 4p(2P ^o) 4d	4d ³ D ^o	1	48962. 3	-80. 4 262. 2		4s ² 4p(2P ^o) 6p	6p ¹ D	2	57250. 84		
		2	48881. 9					1	57430. 7		
		3	49144. 1					0	57696. 1		
4s ² 4p(2P ^o) 5p	5p ³ S	1	49075. 76			4s ² 4p(2P ^o) 5d	5d ¹ P ^o	1	58056. 6		
4s ² 4p(2P ^o) 5p	5p ¹ D	2	49649. 47			4s 4p ³	4p ³ ¹ D ^o	2	58091. 3		
4s ² 4p(2P ^o) 4d	4d ³ F ^o	2	50068. 7	254. 1		4s ² 4p(2P ^o) 7p	7p ³ D	1	58560. 69	26. 07	
		3	50322. 8					2	58586. 76		
		4						3			
4s ² 4p(2P ^o) 5p	5p ¹ S	0	51011. 32			4s ² 4p(2P ^o) 5d	5d ¹ F ^o	3	58941. 6		
4s ² 4p(2P ^o) 4d	4d ³ P ^o	2	51437. 4	-267. 2 -275. 2	1. 50	4s ² 4p(2P ^o) 7s	7s ¹ P ^o	1	59113. 7		
		1	51704. 6					3	59655. 1		
		0	51979. 8					2	59688. 1		
4s ² 4p(2P ^o) 6s	6s ³ P ^o	0	52170. 3	-22. 1 1762. 3		4s 4p ³	4p ³ ³ D ^o	1	59724. 7	-33. 0 -36. 6	
		1	52148. 2					2	60885. 2		
		2	53910. 5					1	61151. 9		
4s ² 4p(2P ^o) 4d	4d ¹ F ^o	3	52592. 0			4s ² 4p(2P ^o) 6d	6d ¹ P ^o	1	61151. 9		
4s ² 4p(2P ^o) 4d	4d ¹ P ^o	1	52847. 0		1. 03	4s ² 4p(2P ^o) 6d	6d ¹ F ^o	3	61266. 8		
4s ² 4p(2P ^o) 6s	6s ¹ P ^o	1	54174. 6			4s ² 4p(2P ^o) 8s	8s ¹ P ^o	1	61342. 8		
4s ² 4p(2P ^o) 6p	6p ¹ P	1	54935. 74			Ge II (2P ^o _{0½})		<i>Limit</i>		63600	
4s ² 4p(2P ^o) 6p	6p ³ D	1	55235. 68	30. 30 1501. 57							
		2	55265. 98								
		3	56767. 55?								
4s ² 4p(2P ^o) 5d	5d ³ D ^o	1	55469. 7	-97. 3 313. 2							
		2	55372. 4								
		3	55685. 6								

Ge I OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 3d ¹⁰ +	Observed Terms		
4s ² 4p ²	{ 4p ² 1S 4p ² 3P 4p ² 1D		
4s 4p ³	{ 4p ³ 3P ^o 4p ³ 3D ^o 4p ³ 1P ^o 4p ³ 1D ^o		
	<i>ns</i> (<i>n</i> ≥ 5)	<i>np</i> (<i>n</i> ≥ 5)	<i>nd</i> (<i>n</i> ≥ 4)
4s ² 4p(2P ^o) <i>nx</i>	{ 5-7s 3P ^o 5-8s 1P ^o	5, 6p 3S 5, 6p 3P 5-7p 3D 5, 6p 1S 5, 6p 1P 5, 6p 1D	4, 5d 3P ^o 4, 5d 3D ^o 4, 5d 3F ^o 4-6d 1P ^o 4-6d 1D ^o 4-6d 1F ^o

*For predicted terms in the spectra of the Ge I isoelectronic sequence, see Vol. II, Introduction.

Ge II

(Ga I sequence; 31 electrons)

Z=32

Ground state 1s² 2s² 2p⁶ 3s² 3p⁶ 3d¹⁰ 4s² 4p 2P_{0/2}^o

4p 2P_{0/2}^o 128518 cm⁻¹

I. P. 15.93 volts

The analysis of this spectrum was begun by Lang, who classified 33 lines and found 10 terms. Rao and Narayan also reported 8 terms and an estimated value of the limit. It has been greatly extended by Gartlein, who has furnished his unpublished data especially for inclusion here. He has observed the spectrum from 813 Å to 6484 Å and now has 131 classified lines.

From three series, *ns* 2S (*n*=5-9), *nd* 2D (*n*=4-10), and *nf* 2F^o (*n*=4-11) he determines the limit quoted here. It is based on a Hicks formula and is derived by the method suggested by Shenstone. Meissner and Andrew have made further observations between 4741 and 7145 Å that have enabled the writer to derive improved term values for inclusion here. They have resolved the 4-7f 2F^o terms by combining their data with those of Gartlein. The three place entries in the table are from their interferometer observations.

The quartet terms are connected with the doublet terms by observed intersystem combinations.

The observed *g*-values have been determined by the writer from the Zeeman patterns published in the 1941 reference below.

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- R. J. Lang, Proc. Nat. Acad. Sci. **14**, 32 (1928). (I P) (T) (C L)
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 C. C. Kiess, J. Research Nat. Bur. Std. **24**, 5, RP1266 (1940). (C L)
 J. C. van den Bosch and P. F. A. Klinkenberg, Proc. Ned. Akad. Wetensch. Amsterdam **44**, No. 5, 560 (1941). (C L) (Z E)
 J. E. Mack, Rev. Mod. Phys. **22**, No. 1, 64 (1950). (Summary hfs)
 C. W. Gartlein, unpublished material (August 1950). (I P) (T) (C L)
 K. W. Meissner and K. L. Andrew, unpublished material (November 1951). (T) (C L)

Ge II

Ge II

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
4s ² (1S)4p	4p ² P ^o	0½ 1½	0. 0 1767. 1	1767. 1		4s ² (1S)6g	6g ² G	{ 3½ 4½	116265. 94		
4s 4p ²	4p ² ⁴ P	0½ 1½ 2½	51575. 5 52290. 5 53366. 7	715. 0 1076. 2		4s ² (1S)9s	9s ² S	0½	118522. 4		
4s ² (1S)5s	5s ² S	0½	62402. 360		2. 00	4s ² (1S)8d	8d ² D	1½ 2½	119337. 8 119345. 2	7. 4	
4s 4p ²	4p ² ² D	1½ 2½	65015. 0 65184. 070	169. 1		4s ² (1S)7f	7f ² F ^o	3½ 2½	119373. 27 119373. 72	-0. 45	
4s ² (1S)5p	5p ² P ^o	0½ 1½	79006. 205 79365. 847	359. 642	0. 67 1. 33	4s ² (1S)7g	7g ² G	{ 3½ 4½	119521. 03		
4s ² (1S)4d	4d ² D	1½ 2½	80836. 05 81011. 80	175. 75		4s ² (1S)10s	10s ² S	0½	120975. 9		
4s 4p ²	4p ² ² S	0½	85889. 9			4s ² (1S)9d	9d ² D	{ 1½ 2½	121512. 7		
4s 4p ²	4p ² ² P	0½ 1½	91014. 8 92122. 1	1107. 3		4s ² (1S)8f	8f ² F ^o	{ 2½ 3½	121531. 4		
4s ² (1S)6s	6s ² S	0½	94783. 758			4s 4p(² P ^o)5s:	5s ² P ^o	0½ 1½	121914. 1 123099. 8	1185. 7	
4s ² (1S)5d	5d ² D	1½ 2½	100089. 32 100130. 18	40. 86	0. 77 1. 20	4s 4p(³ P ^o)5s:	5s ⁴ P ^o	0½ 1½ 2½	122693. 7? 123401. 4 124730. 5	707. 7 329. 1	
4s ² (1S)4f	4f ² F ^o	3½ 2½	100316. 50 100317. 14	-0. 64		4s ² (1S)10d	10d ² D	{ 1½ 2½	122996. 0		
4s ² (1S)6p	6p ² P ^o	{ 0½ 1½	101242. 8			4s ² (1S)9f	9f ² F ^o	{ 2½ 3½	123007. 0		
4s ² (1S)7s	7s ² S	0½	107935. 40			4s ² (1S)11d	11d ² D	{ 1½ 2½	124053. 7		
4s ² (1S)6d	6d ² D	1½ 2½	110377. 39 110396. 93	19. 54		4s ² (1S)10f	10f ² F ^o	{ 2½ 3½	124061. 6		
4s ² (1S)5f	5f ² F ^o	3½ 2½	110503. 69 110504. 43	-0. 74		4s ² (1S)11f	11f ² F ^o	{ 2½ 3½	124840. 3		
4s ² (1S)7p	7p ² P ^o	{ 0½ 1½	111016. 2			4s ² (1S)12f	12f ² F ^o	{ 2½ 3½	125431. 5		
4s ² (1S)8s	8s ² S	0½	114637. 84			-----					
4s ² (1S)7d	7d ² D	1½ 2½	115977. 10 115987. 14	10. 04		Ge III(¹ S ₀)	Limit	-----	128518		
4s ² (1S)6f	6f ² F ^o	3½ 2½	116040. 62 116041. 16	-0. 54		4p ³	4p ³ ⁴ S ^o	1½	136237. 3		

November 1951.

Ge II OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ² 3d ¹⁰ +	Observed Terms				
4s ² (1S)4p	4p ² P ^o				
4s 4p ²	{ 4p ² ² S 4p ² ⁴ P 4p ² ² P 4p ² ² D				
4p ³	4p ³ ⁴ S ^o :				
	ns (n ≥ 5)	np (n ≥ 5)	nd (n ≥ 4)	nf (n ≥ 4)	ng (n ≥ 5)
4s ² (1S)n _x	5-10s ² S	5-7p ² P ^o	4-11d ² D	4-12f ² F ^o	6, 7g ² G
4s 4p(³ P ^o)n _x	{ 5s ⁴ P ^o : 5s ² P ^o :				

*For predicted terms in the spectra of the Ga I isoelectronic sequence, see Vol. II, Introduction.

Ge III

(Zn I sequence; 30 electrons)

 $Z=32$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 {}^1S_0$ $4s^2 {}^1S_0$ 276036 cm^{-1}

I. P. 34.21 volts

The analysis is from Lang, who has revised and extended the earlier work on this spectrum, and published a list of 86 classified lines between 542 Å and 5256 Å. His term values "depend upon an arbitrary choice of 65500 cm^{-1} " for the absolute value of $4f {}^3F_4$. This choice is based on isoelectronic sequence data and "has been made with reasonable accuracy."

Rao and Narayan list in addition 6 "even" miscellaneous levels that are not included here.

The writer has derived the tabulated g -values from the observed Zeeman patterns given in the 1941 reference below.

The singlet and triplet systems of terms are connected by observed intersystem combinations.

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K. R. Rao and A. L. Narayan, Proc. Roy. Soc. London [A], **119**, 616 (1928). (I P) (T) (C L)

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J. C. van den Bosch and P. F. A. Klinkenberg, Proc. Ned. Akad. v. Wetensch. Amsterdam **44**, No. 5, 560 (1941). (Z E)

Ge III

Ge III

Config.	Desig.	J	Level	Interval	Obs. g	Config.	Desig.	J	Level	Interval	Obs. g	
$4s^2$	$4s^2 {}^1S$	0	0			$4p({}^2P^\circ)4d$	$4d {}^1D^\circ$	2	197122			
$4s({}^2S)4p$	$4p {}^3P^\circ$	0	61733	763 1642		$4s({}^2S)4f$	$4f {}^3F^\circ$	2	210455.3	17.4 63		
		1	62496						3		210472.7	
		2	64138						4		210536	
$4s({}^2S)4p$	$4p {}^1P^\circ$	1	91873			$4s({}^2S)4f$	$4f {}^1F^\circ$	3	210531			
$4s({}^2S)4d$	$4d {}^1D$	2	144972			$4s({}^2S)6s$	$6s {}^3S$	1	211149		1.94	
$4p^2$	$4p^2 {}^3P$	0	147691	953 1729		$4p({}^2P^\circ)4d$	$4d {}^1P^\circ$	1	212359			
		1	148644				$4s({}^2S)5d$	$5d {}^3D$	1	213133.2	30.6 44.2	0.50 1.16 1.32
		2	150373						2	213163.8		
$4p^2$	$4p^2 {}^1D$	2	148776					3	213208.0			
$4s({}^2S)5s$	$5s {}^3S$	1	158576.2		1.99	$4p({}^2P^\circ)4d$	$4d {}^1F^\circ$	3	213579			
$4s({}^2S)4d$	$4d {}^3D$	1	162851.0	70.8 106.5		$4s({}^2S)5g$	$5g {}^1G$	4	234910??			
		2	162921.8				$4s({}^2S)5g$	$5g {}^3G$	3, 4, 5	234928?		
		3	163028.3				$4d^2?$	$4d^2 {}^1D$	2	235908		
$4s({}^2S)5s$	$5s {}^1S$	0	167450									
$4s({}^2S)5p$	$5p {}^3P^\circ$	0	181871	168 459								
		1	182039			1.48						
		2	182498			1.49	Ge IV (${}^2S_{1/2}$)	Limit		276036		
$4s({}^2S)5p$	$5p {}^1P^\circ$	1	184308.8									

March 1950.

Ge III OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 3d ¹⁰ +	Observed Terms					
4s ²	4s ² ¹ S					
4p ²	{	4p ² ³ P		4p ² ¹ D		
4d ²		4d ² ¹ D?				
		ns (n ≥ 5)	np (n ≥ 4)	nd (n ≥ 4)	nf (n ≥ 4)	ng (n ≥ 5)
4s(² S)nx	{	5, 6s ³ S	4, 5p ³ P ^o	4, 5d ³ D	4f ³ F ^o	5g ³ G?
4p(² P ^o)nx		5s ¹ S	4, 5p ¹ P ^o	4d ¹ D	4f ¹ F ^o	5g ¹ G?
				4d ¹ P ^o	4d ¹ D ^o	4d ¹ F ^o

*For predicted terms of the Zn I isoelectronic sequence, see Vol. II, Introduction.

Ge IV

(Cu I sequence; 29 electrons)

Z=32

Ground state 1s² 2s² 2p⁶ 3s² 3p⁶ 3d¹⁰ 4s ²S_{0½}

4s ²S_{0½} **368701** cm⁻¹

I. P. 45.7 volts

Carroll, in 1925, reported five doublet series terms, and determined the limit by assuming an absolute value of 4f ²F^o (111200 cm⁻¹) consistent with extrapolated isoelectronic sequence data.

In 1929 Lang remeasured the spectrum, and revised and extended the earlier term values, retaining the same assumed absolute value of 4f ²F^o. He concludes from a study of the isoelectronic sequence that this value is as nearly accurate as can be hoped for at present. Lang's values have been used in the present compilation. Two probable misprints in his term list have been corrected, namely, the terms 4s² ²D and 4f ²F^o are here entered as inverted.

The observed *g*-values have been derived by the writer from the observed Zeeman patterns given in the 1941 reference below.

There are 34 classified lines between 440 Å and 3676 Å.

REFERENCES

- J. A. Carroll, Phil. Trans. Roy. Soc. (London) [A] **225**, 374 (1925). (I P) (T) (C L)
 R. J. Lang, Phys. Rev. **34**, 696 (1929). (I P) (T) (C L)
 J. C. van den Bosch and P. F. A. Klinkenberg, Proc. Ned. Akad. v. Wet. Amsterdam **44**, No. 5, 560 (1941). (Z E)

Ge IV

Ge V

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
3d ¹⁰ (¹ S)4s	4s ² S	0½	0			3d ¹⁰ (¹ S)5d	5d ² D	1½ 2½	266637 266717	80	0.80 1.19
3d ¹⁰ (¹ S)4p	4p ² P ^o	0½ 1½	81315 84103	2788		3d ¹⁰ (¹ S)6s	6s ² S	0½	270058		
3d ¹⁰ (¹ S)4d	4d ² D	1½ 2½	190607 190861	254	0.82 1.26	3d ¹⁰ (¹ S)6p	6p ² P ^o	0½ 1½	283621 283763	142	
3d ¹⁰ (¹ S)5s	5s ² S	0½	199269		2.04	3d ¹⁰ (¹ S)5g	5g ² G	{ 3½ 4½ }	298379		
3d ¹⁰ (¹ S)5p	5p ² P ^o	0½ 1½	226459 227397	938	0.68 1.34						
3d ¹⁰ (¹ S)4f	4f ² F ^o	3½ 2½	257496 257501	-5		Ge v (¹ S ₀)	<i>Limit</i>		368701		
3d ⁹ 4s ²	4s ² ² D	2½ 1½	259942 264445	-4503							

Ge v

(Ni I sequence; 28 electrons)

 $Z=32$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 1S_0$ $3d^{10} 1S_0$ 753800 cm^{-1}

I. P. 93.4 volts

Mack, Laporte, and Lang have found 8 terms and classified 33 lines between 942 Å and 1222 Å by using the observations by Carroll. From three lines at 294 Å, 295 Å, and 304 Å, the ground term was found by Kruger and Shoupp; who also determined the limit from an empirical formula based on extrapolation of isoelectronic sequence data. Their value of the limit, rounded off, is entered in brackets in the table. The observations in the far ultraviolet indicate a correction of $+234231 \text{ cm}^{-1}$ to the terms by Mack, Laporte, and Lang, in order to reduce them to the ground term zero. This correction has been made in compiling the table below.

The singlet and triplet terms are connected by observed intersystem combinations.

REFERENCES

- J. A. Carroll, Phil. Trans. Roy. Soc. (London) [A] **225**, 357 (1925).
 J. E. Mack, O. Laporte, and R. J. Lang, Phys. Rev. **31**, 748 (1928). (T) (C L)
 P. G. Kruger and W. E. Shoupp, Phys. Rev. **46**, 124 (1934). (I P) (T) (C L)

Ge v

Ge v

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3d^{10}$	$3d^{10} 1S$	0	0		$3d^9(2D)4p$	$4p \ ^3D^\circ$	3	335187	-362
$3d^9(2D)4s$	$4s \ ^3D$	3	234231	-1740			2	335549	-4005
		2	235971	-2796			1	339554	
		1	238767		$3d^9(2D)4p$	$4p \ ^1F^\circ$	3	337192	
$3d^9(2D)4s$	$4s \ ^1D$	2	241947		$3d^9(2D)4p$	$4p \ ^1P^\circ$	1	338255	
$3d^9(2D)4p$	$4p \ ^3P^\circ$	2	323772	-4083	$3d^9(2D)4p$	$4p \ ^1D^\circ$	2	340311	
		1	327855	-2487					
		0	330342						
$3d^9(2D)4p$	$4p \ ^3F^\circ$	4	329879	2113	Ge VI ($^2D_{3/2}$)	Limit		[753800]	
		3	327766	-3039					
		2	330805						

November 1949.

Ge v OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms				
$3d^{10}$	$3d^{10} 1S$				
	$ns (n \geq 4)$		$np (n \geq 4)$		
$3d^9(2D)nx$	{	$4s \ ^3D$ $4s \ ^1D$	$4p \ ^3P^\circ$ $4p \ ^1P^\circ$	$4p \ ^3D^\circ$ $4p \ ^1D^\circ$	$4p \ ^3F^\circ$ $4p \ ^1F^\circ$

* For predicted terms in the spectra of the Ni I isoelectronic sequence, see Vol. II, Introduction.

ARSENIC

As I

33 electrons

 $Z=33$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^3 \ ^4S_{1/2}^{\circ}$ $4p^3 \ ^4S_{1/2}^{\circ}$ 79165 cm^{-1}

I. P. 9.81 volts

Meggers has extended and revised the earlier work by himself and others. His observations extend from 1407 Å to 11679 Å, those of wavelength shorter than 2000 Å being from arc spectrograms made by Shenstone.

The present analysis has been submitted in manuscript especially for inclusion here. There are 246 classified lines, and the doublet and quartet terms are connected by observed intersystem combinations. More than 74 percent of the total number and 97 percent of the total intensity of observed lines have been explained as combinations of 30 odd energy levels arising from $4s^2 4p^3$ and $4s^2 4p^2 np$ electrons, and 58 even levels from $4s 4p^4$, $4s^2 4p^2 ns$, and $4s^2 4p^2 nd$.

The limit is derived from the $ns \ ^4P$ series ($n=5, 6$), and is based on a Ritz formula with the assumption that $\alpha=2.5 \times 10^{-6}$, as suggested by Shenstone.

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- K. R. Rao, Proc. Roy. Soc. (London) [A] **125**, 238 (1929). (T) (C L)
 W. F. Meggers and T. L. deBruin, Bur. Std. J. Research **3**, 765, RP 116 (1929). (T) (C L)
 A. S. Rao, Proc. Phys. Soc. London **44**, 594 (1932). (I P) (T) (C L)
 J. B. Green and W. M. Barrows Jr., Phys. Rev. **47**, 132 (1935). (Z E)
 J. E. Mack, Rev. Mod. Phys. **22**, No. 1, 64 (1950). (Summary hfs)
 W. F. Meggers, A. G. Shenstone, and C. E. Moore, Bur. Std. J. Research **45**, 346, RP2144 (1950). (I P) (T) (C L)

As I

As I

Config.	Desig.	J	Level	Interval	Obs. g	Config.	Desig.	J	Level	Interval	Obs. g
$4s^2 4p^3$	$4p^3 \ ^4S^{\circ}$	$1\frac{1}{2}$	0.0			$4s^2 4p^2(^3P)5p$	$5p \ ^4D^{\circ}$	$0\frac{1}{2}$	60791.5		
$4s^2 4p^3$	$4p^3 \ ^2D^{\circ}$	$1\frac{1}{2}$	10592.5	322.1	0.813			$1\frac{1}{2}$	60860.0	68.5	
		$2\frac{1}{2}$	10914.6		1.208			$2\frac{1}{2}$	61685.0	825.0	
								$3\frac{1}{2}$	62871.2	1186.2	
$4s^2 4p^3$	$4p^3 \ ^2P^{\circ}$	$0\frac{1}{2}$	18186.1	461.4	0.687	$4s^2 4p^2(^1D)5s$	$5s' \ ^2D$	$2\frac{1}{2}$	60815.0	-19.8	
		$1\frac{1}{2}$	18647.5		1.313			$1\frac{1}{2}$	60834.8		0.84
$4s^2 4p^2(^3P)5s$	$5s \ ^4P$	$0\frac{1}{2}$	50693.8	916.4	2.617	$4s^2 4p^2(^3P)5p$	$5p \ ^4P^{\circ}$	$0\frac{1}{2}$	62026.3	371.7	
		$1\frac{1}{2}$	51610.2	1287.7	1.706			$1\frac{1}{2}$	62398.0	884.8	
		$2\frac{1}{2}$	52897.9					$2\frac{1}{2}$	63282.8		
$4s^2 4p^2(^3P)5s$	$5s \ ^2P$	$0\frac{1}{2}$	53135.6	1469.7	0.736	$4s^2 4p^2(^3P)5p$	$5p \ ^2D^{\circ}$	$1\frac{1}{2}$	62554.4	1614.8	
		$1\frac{1}{2}$	54605.3		1.327			$2\frac{1}{2}$	64169.2		
$4s 4p^4$	$4p^4 \ ^4P$	$2\frac{1}{2}$	55366.4	-1497.2		$4s^2 4p^2(^3P)4d$	$4d \ ^4P$	$2\frac{1}{2}$	62751.1	-752.5	
		$1\frac{1}{2}$	56863.6	-624.5				$1\frac{1}{2}$	63503.6	-478.1	
		$0\frac{1}{2}$	57488.1					$0\frac{1}{2}$	63981.7		

As I—Continued

As I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
4s ² 4p ² (³ P) 5p	5p ⁴ S°	1½	63647. 0			4s ² 4p ² (³ P) 6p	6p ² S°	0½	71199. 1		
4s ² 4p ² (³ P) 5p	5p ² S°	0½	64059. 0			4s ² 4p ² (³ P) 6p	6p ² D°	1½ 2½	71621. 4 72944. 4	1323. 0	
4s ² 4p ² (³ P) 4d	4d ² P	0½ 1½	64122. 0 66485. 2	2363. 2		4s ² 4p ² (¹ D) 5p	5p' ² P°	0½ 1½	71796. 2 71931. 7	135. 5	
4s ² 4p ² (³ P) 5p	5p ² P°	0½ 1½	64323. 8 64687. 6	363. 8		4s ² 4p ² (¹ D) 4d	4d' ² P	0½ 1½	71924. 4 72125. 0	200. 6	
4s ² 4p ² (³ P) 4d	4d ² D	1½ 2½	64342. 4 66725. 4	2383. 0		4s ² 4p ² (¹ S) 5s	5s'' ² S	0½	72399. 4		
4s ² 4p ² (³ P) 4d	4d ² F	2½ 3½	64639. 5 66785. 2	2145. 7			4	1½	72519. 4		
4s 4p ⁴	4p ⁴ ² P	1½ 0½	64811. 5 65497. 4	-685. 9		4s ² 4p ² (³ P) 6p	6p ⁴ S°	1½	72620. 3		
4s 4p ⁴	4p ⁴ ² S	0½	66588. 3				5	1½	72779. 0		
4s ² 4p ² (³ P) 6s	6s ⁴ P	0½ 1½ 2½	67008. 7 67920. 2 69314. 8	911. 5 1394. 6		4s ² 4p ² (³ P) 6p	6p ² P°	0½ 1½	72800. 6 73244. 8	444. 2	
4s 4p ⁴	4p ⁴ ² D	1½ 2½	68300. 6 68402. 7	102. 1			6	1½, 2½	73566. 0		
4s ² 4p ² (³ P) 6s	6s ² P	0½ 1½	68314. 5 69696. 2	1381. 7			7	1½	73635. 0		
	1	0½, 1½	69644. 8				8	1½, 2½	74049. 7		
	2	1½, 2½	69698. 3				9	0½, 1½	74143. 8		
4s ² 4p ² (³ P) 6p	6p ⁴ P°	0½ 1½ 2½	69881. 5 70849. 7 72501. 9	968. 2 1652. 2		4s ² 4p ² (¹ S) 4d	4d'' ² D	2½ 1½	74453. 1 74810. 5	-357. 4	
4s ² 4p ² (¹ D) 5p	5p' ² D°	1½ 2½	70108. 4 70376. 6	268. 2			10	0½, 1½	75086. 0		
4s ² 4p ² (³ P) 6p	6p ⁴ D°	0½ 1½ 2½ 3½	70274. 9 70516. 6 71131. 0 72341. 6	241. 7 614. 4 1210. 6			11	1½, 2½	75176. 4		
4s ² 4p ² (¹ D) 5p	5p' ² F°	2½ 3½	70821. 4 70927. 5	106. 1			12	1½, 2½	75578. 7		
	3	1½	71055. 2				13	0½, 1½	75799. 4		
							14	1½	76076. 2		
							15	1½, 2½	76835. 6		
							16	1½, 2½	77121. 7		
							17	1½, 2½	77272. 5		
						As II (³ P ₀)	Limit		79165		

May 1950.

AS I OBSERVED TERMS*

1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 3d ¹⁰ +	Observed Terms		
4s ² 4p ²	{ 4p ³ ⁴ S°	4p ³ ² P°	4p ³ ² D°
4s 4p ⁴	{ 4p ⁴ ⁴ P	4p ⁴ ² P	4p ⁴ ² D
	ns (n ≥ 5)		np (n ≥ 5)
4s ² 4p ² (³ P) nx	{ 5, 6s ⁴ P	5, 6p ⁴ S°	5, 6p ⁴ P°
4s ² 4p ² (¹ D) nx'	5, 6s ² P	5, 6p ² S°	5, 6p ² P°
4s ² 4p ² (¹ S) nx''	5s'' ² S	5p' ² P°	5p' ² D°
		5p' ² F°	5p' ² F°
			4d ⁴ P
			4d ² P
			4d ² D
			4d ² F
			4d' ² P
			4d'' ² D

*For predicted terms in the spectra of the As I isoelectronic sequence, see Introduction.

(Ge I sequence; 32 electrons)

Z=33

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^2 \ ^3P_0$ $4p^2 \ ^3P_0$ 163000 cm^{-1}

I. P. 20.2 volts

A. S. Rao has published 187 classified lines in the interval 802 Å to 6528 Å, 22 terms, and 14 miscellaneous levels. C. W. Gartlein has also studied this spectrum and confirmed 15 of these terms and four of the miscellaneous levels, with the conclusion that the spectrum needs careful observation over the entire range for three reasons: existing wavelengths need to be improved in precision, the lines due only to As II need to be carefully selected, and the series should be extended.

The writer has adopted Gartlein's term values when available, supplemented by those of Rao, adding $5d \ ^3F_2^\circ$ and $5d \ ^3D_2^\circ$ from the paper by Green and Barrows on Zeeman Effect. Decimals are omitted throughout, since the discordances between the two term lists exceed 1 cm^{-1} in many cases. All g -values are from the 1935 paper, that for $4i$ having been derived by the writer. All others are either the g -values adopted by these authors, or means taken by the writer from the individual g -values they derive.

Miscellaneous levels have here been assigned numbers as follows: those labeled by Rao "a to g" are here called 1° to 7° respectively; and, similarly those called "A to G" are here designated 1 to 7, respectively. The combinations do not indicate which of the two J -values entered in the table is correct for the levels 1, 2, 3, 5, and 6.

Rao has determined the limit 162788 cm^{-1} by applying a Rydberg formula to the lines designated as $5s \ ^1P_1^\circ - 4p^2 \ ^1D_2$ and $5s \ ^1P_1^\circ - 5p \ ^1D_2$. Similarly, the writer obtains the value 163040 cm^{-1} from the $ns \ ^3P_2^\circ$ series ($n=5,6$). The round figure 163000 is adopted in the table pending further observations.

Observed intersystem combinations connect the singlet and triplet systems of terms.

REFERENCES

- A. S. Rao, Ind. J. Phys. **7**, 561 (1932). (I P) (T) (C L)
 J. B. Green and W. M. Barrows, Jr., Phys. Rev. **47**, 133 (1935). (C L) (Z E)
 J. E. Mack, Rev. Mod. Phys. **22**, No. 1, 64 (1950). (Summary hfs)
 C. W. Gartlein, unpublished material, December 1950. (T)

As II

As II

Config.	Desig.	J	Level	Interval	Obs. g	Config.	Desig.	J	Level	Interval	Obs. g
$4s^2 4p^2$	$4p^2 \ ^3P$	0	0	1061 1479		$4s^2 4p(2P^\circ) 5s$	$5s \ ^1P^\circ$	1	82818		1. 075
		1	1061								
		2	2540								
$4s^2 4p^2$	$4p^2 \ ^1D$	2	10093			$4s \ 4p^3$	$4p^3 \ ^1P^\circ$	1	83099		
$4s^2 4p^2$	$4p^2 \ ^1S$	0	22593			$4s \ 4p^3$	$4p^3 \ ^3P^\circ$	0	84449	189 467	
$4s \ 4p^3$	$4p^3 \ ^3D^\circ$	1	73748	112 382	0. 508 1. 176 1. 335			1	84638		
		2	73860					2	88827		
$4s^2 4p(2P^\circ) 5s$	$5s \ ^3P^\circ$	3	74242			$4s^2 4p(2P^\circ) 4d$	$4d \ ^3F^\circ$	3	89545	718	0. 58 1. 07
		0	78729	4							
		1	79128								
$4s^2 4p(2P^\circ) 5s$	$5s \ ^3P^\circ$	2	81506	399 2378	0/0 1. 424 1. 492	$4s^2 4p(2P^\circ) 5p$	$5p \ ^1P$	1	95327		0. 812

As II—Continued

As II—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>				
$4s^2 4p(^2P^\circ)5p$	$5p \ ^3D$	1	96912	202 2081	0. 895	$4s^2 4p(^2P^\circ)4f$	4°	1	120769	-55 1064	1. 60				
		2	97114		1. 174			5°	1		121355	0. 726			
		3	99195		1. 325		1		2, 3		121519	$4f \ ^3F$	2	121623	0. 88
$4s^2 4p(^2P^\circ)5p$	$5p \ ^3P$	0	98235	944 1081	0/0		3	121568	3		121632		1. 06		
		1	99179		1. 389		2	2, 3	121625		$4s^2 4p(^2P^\circ)5d$		$5d \ ^3D^\circ$	1	123190
		2	100260		1. 433							2		3	
$4s^2 4p(^2P^\circ)4d$	$4d \ ^3D^\circ$	1	99063	482 845	0. 895		$6s \ ^3P^\circ$	0	1		123223	1. 14			
		2	99545		1. 27								6°	1	123784
		3	100390		1. 896		3	1, 2	124065		$4s^2 4p(^2P^\circ)6s$	$6s \ ^3P^\circ$			
$4s^2 4p(^2P^\circ)5p$	$5p \ ^3S$	1	101088	1. 046	0/0		1	2	124206				124547	1. 11	
		2	102388								0. 94	6			2, 3
$4s^2 4p(^2P^\circ)4d$	$4d \ ^1P^\circ$	1	101486	-475 80	1. 11		7°	1	125243		163000	As III ($^2P^\circ_{3/2}$)	Limit		
$4s^2 4p(^2P^\circ)5p$	$5p \ ^1D$	2	102388			0. 698									
$4s^2 4p(^2P^\circ)4d$	$4d \ ^1D^\circ$	2	102486	118887	0. 698	7°	1	125243	163000	As III ($^2P^\circ_{3/2}$)	Limit				
$4s^2 4p(^2P^\circ)4d$	$4d \ ^1D^\circ$	1	102602												
$4s^2 4p(^2P^\circ)5p$	$5p \ ^1S$	0	105786												
$4s^2 4p(^2P^\circ)4d$	$4d \ ^1F^\circ$	3	109872	118887	0. 698	7°	1	125243	163000	As III ($^2P^\circ_{3/2}$)	Limit				
$4s^2 4p(^2P^\circ)4d$	$4d \ ^3P^\circ$	2	111846												
$4s^2 4p(^2P^\circ)4d$	$4d \ ^3P^\circ$	1	112321												
$4s^2 4p(^2P^\circ)4d$	$4d \ ^3P^\circ$	0	112241	118887	0. 698	7°	1	125243	163000	As III ($^2P^\circ_{3/2}$)	Limit				
$4s^2 4p(^2P^\circ)4d$	$4d \ ^3P^\circ$	1	117741												
$4s^2 4p(^2P^\circ)4d$	$4d \ ^3P^\circ$	1	118884?												
$4s^2 4p(^2P^\circ)5d$	$5d \ ^3F^\circ$	2	118887	118887	0. 698	7°	1	125243	163000	As III ($^2P^\circ_{3/2}$)	Limit				
$4s^2 4p(^2P^\circ)5d$	$5d \ ^3F^\circ$	3													
$4s^2 4p(^2P^\circ)5d$	$5d \ ^3F^\circ$	4													

January 1951.

AS II OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10}+$	Observed Terms			
$4s^2 4p^2$	$\left\{ \begin{array}{l} 4p^2 \ ^1S \\ 4p^2 \ ^3P \\ 4p^2 \ ^1D \end{array} \right.$			
$4s 4p^3$	$\left\{ \begin{array}{l} 4p^3 \ ^3P^\circ \\ 4p^3 \ ^1P^\circ \\ 4p^3 \ ^3D^\circ \end{array} \right.$			
	$ns (n \geq 5)$	$np (n \geq 5)$	$nd (n \geq 4)$	$nf (n \geq 4)$
$4s^2 4p(^2P^\circ)5s$	$\left\{ \begin{array}{l} 5, 6s \ ^3P^\circ \\ 5s \ ^1P^\circ \end{array} \right.$	$\begin{array}{l} 5p \ ^3S \\ 5p \ ^1S \end{array} \quad \begin{array}{l} 5p \ ^3P \\ 5p \ ^1P \end{array} \quad \begin{array}{l} 5p \ ^3D \\ 5p \ ^1D \end{array}$	$\begin{array}{l} 4d \ ^3P^\circ \\ 4d \ ^1P^\circ \end{array} \quad \begin{array}{l} 4, 5d \ ^3D^\circ \\ 4d \ ^1D^\circ \end{array} \quad \begin{array}{l} 4, 5d \ ^3F^\circ \\ 4d \ ^1F^\circ \end{array}$	$4f \ ^3F$

*For predicted terms in the spectra of the Ge I isoelectronic sequence, see Vol. II, Introduction.

As III

(Ga I sequence; 31 electrons)

 $Z=33$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^2 P_{0\frac{1}{2}}^{\circ}$ $4p^2 P_{0\frac{1}{2}}^{\circ}$ 228400 cm^{-1}

I. P. 28.3 volts

This spectrum needs further study. In 1928 Lang published 10 terms, and 35 classified lines in the interval between 603 Å and 4226 Å. He estimated the limit from isoelectronic sequence data to be $226500 \pm 1500 \text{ cm}^{-1}$.

K. R. Rao, in 1931, extended and revised Lang's analysis slightly, and from an assumed absolute value of $5g^2 G = 39500 \text{ cm}^{-1}$ derived the limit 228406 cm^{-1} . The latter value in round numbers is entered in brackets in the table. The terms in the table are also from this paper.

Six additional series members and five multiplets involving suggested quartet combinations are given in the 1929 reference below. They are omitted here awaiting confirmation.

REFERENCES

- P. Pattabhiramiah and A. S. Rao, Indian J. Phys. **3**, 437 (1928). (I P) (T) (C L)
 R. J. Lang, Phys. Rev. **32**, 737 (1928). (I P) (T) (C L)
 A. S. Rao and A. L. Narayan, Zeit. Phys. **57**, 865 (1929). (C L)
 K. R. Rao, Proc. Phys. Soc. London **43**, 68 (1931). (I P) (T) (C L)
 J. B. Green and W. M. Barrows, Jr., Phys. Rev. **47**, 135 (1935). (Z E)

As III

As III

Config.	Desig.	J	Level	Interval	Obs. g	Config.	Desig.	J	Level	Interval	Obs. g
$4s^2(1S)4p$	$4p^2 P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	0 2940	2940		$4s^2(1S)5p$	$5p^2 P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	131458 132181	723	1.333 0.658
$4s 4p^2$	$4p^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	85309 85630	321		$4s^2(1S)6s$	$6s^2 S$	$0\frac{1}{2}$	162889		
$4s^2(1S)5s$	$5s^2 S$	$0\frac{1}{2}$	106694		1.996	$4s^2(1S)4f$	$4f^2 F^{\circ}$	$3\frac{1}{2}$ $2\frac{1}{2}$	164105 164114		-9
$4s 4p^2$	$4p^2 S$	$0\frac{1}{2}$	107806			$4s^2(1S)5d$	$5d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	165623 165708		85
$4s 4p^2$	$4p^2 P$	$0\frac{1}{2}$ $1\frac{1}{2}$	113938 115425	1487		$4s^2(1S)5g$	$5g^2 G$	$3\frac{1}{2}, 4\frac{1}{2}$	188906		
$4s^2(1S)4d$	$4d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	117651 117742	91							
						As IV (1S ₀)	Limit		[228400]		

April 1950.

As IV

(Zn I sequence; 30 electrons)

Z=33

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 {}^1S_0$ $4s^2 {}^1S_0$ 404369 cm^{-1}

I. P. 50.1 volts

Three triplet terms $4p {}^3P^\circ$, $4d {}^3D$, and $5s {}^3S$, and also the combinations $4p {}^3P^\circ - 4p^2 {}^3P$ and $4s^2 {}^1S_0 - 4p {}^1P_1^\circ$, were found by Sawyer and Humphreys, who classified 16 lines between 692 Å and 980 Å. Later Rao extended the analysis from observations in the interval 530 Å to 4533 Å, and added about 70 newly classified lines including intersystem combinations connecting the singlet and triplet systems of terms. He determined the limit from the two series $5p {}^3P^\circ - nd {}^3D$ ($n=4, 5$) and $5p {}^3P^\circ - ns {}^3S$ ($n=5, 6$) by using a Rydberg formula.

Three miscellaneous levels called, respectively, α , β , γ by Rao have been omitted here.

REFERENCES

R. A. Sawyer and C. J. Humphreys, Phys. Rev. **32**, 583 (1928). (T) (C L)K. R. Rao, Proc. Roy. Soc. (London) [A] **134**, 604 (1931). (I P) (T) (C L)J. E. Mack, Rev. Mod. Phys. **22**, No. 1, 64 (1950). (Summary hfs)

As IV

As IV

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$4s^2$	$4s^2 {}^1S$	0	0		$4s(2S)5p$	$5p {}^3P^\circ$	0	251212.0	
$4s(2S)4p$	$4p {}^3P^\circ$	0	75812	1150			1	251466.9	254.9
		1	76962	2530			2	252285.0	818.1
		2	79492		$4s(2S)5p$	$5p {}^1P^\circ$	1	254039.9	
$4s(2S)4p$	$4p {}^1P^\circ$	1	112022		$4s(2S)4f$	$4f {}^3F^\circ$	2	284456	274
$4s(2S)4d$	$4d {}^1D$	2	179563				3	284730	413
							4	285143	
$4p^3$	$4p^2 {}^3P$	0	179934	1534	$4s(2S)4f$	$4f {}^1F^\circ$	3	285730	
		1	181468	2926	$4s(2S)5d$	$5d {}^1D$	2	293834.4	
		2	184394		$4s(2S)5d$	$5d {}^3D$	1	295580.8	58.2
$4p^2$	$4p^2 {}^1D$	2	204658				2	295639.0	91.2
$4s(2S)4d$	$4d {}^3D$	1	210597.0	132.8			3	295730.2	
		2	210729.8	202.8	$4s(2S)6s$	$6s {}^3S$	1	298875.1	
		3	210932.6		$4s(2S)6s$	$6s {}^1S$	0	300698.9	
$4p^2$	$4p^2 {}^1S$	0	211683						
$4s(2S)5s$	$5s {}^3S$	1	220127.6						
$4s(2S)5s$	$5s {}^1S$	0	229420.5						
					As v ($2S_{0\frac{1}{2}}$)	Limit		404369	

March 1950.

AS IV OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 3d ¹⁰ +	Observed Terms			
4s ²	4s ² ¹ S			
	$\left\{ \begin{array}{l} 4p^2 \text{ } ^3\text{P} \\ 4p^2 \text{ } ^1\text{S} \quad 4p^2 \text{ } ^1\text{D} \end{array} \right.$			
	<i>ns</i> (<i>n</i> ≥ 5)	<i>np</i> (<i>n</i> ≥ 4)	<i>nd</i> (<i>n</i> ≥ 4)	<i>nf</i> (<i>n</i> ≥ 4)
4s(² S) <i>nx</i>	$\left\{ \begin{array}{l} 5,6s \text{ } ^3\text{S} \\ 5,6s \text{ } ^1\text{S} \end{array} \right.$	$\left\{ \begin{array}{l} 4,5p \text{ } ^3\text{P}^\circ \\ 4,5p \text{ } ^1\text{P}^\circ \end{array} \right.$	$\left\{ \begin{array}{l} 4,5d \text{ } ^3\text{D} \\ 4,5d \text{ } ^1\text{D} \end{array} \right.$	$\left\{ \begin{array}{l} 4f \text{ } ^3\text{F}^\circ \\ 4f \text{ } ^1\text{F}^\circ \end{array} \right.$

*For predicted terms in the spectra of the Zn I isoelectronic sequence, see Vol. II, Introduction.

As V

(Cu I sequence; 29 electrons)

Z = 33

Ground state 1s² 2s² 2p⁶ 3s² 3p⁶ 3d¹⁰ 4s ²S_{0½}

4s ²S_{0½} 505136 cm⁻¹

I. P. 62.6 volts

This spectrum is incompletely analyzed. By extrapolation along the isoelectronic sequence Sawyer and Humphreys have classified 9 lines between 600 Å and 1056 Å and determined the limit quoted here.

REFERENCE

R. A. Sawyer and C. J. Humphreys, Phys. Rev. **32**, 583 (1928). (I P) (T) (C L).

As V

Config.	Desig.	<i>J</i>	Level	Interval
4s	4s ² S	0½	0	
4p	4p ² P ^o	0½	97135	4110
		1½	101245	
4d	4d ² D	1½	236897	445
		2½	237342	
5s	5s ² S	0½	263596	
4f	4f ² F ^o	$\left\{ \begin{array}{l} 3½ \\ 2½ \end{array} \right.$	331987	
----- As VI (¹ S ₀)	<i>Limit</i>	-----	505136	

February 1950.

As VI

(Ni I sequence; 28 electrons)

Z=33

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} {}^1S_0$ $3d^{10} {}^1S_0$ 1028800 cm^{-1}

I. P. 127.5 volts

Mack and Borg have found 8 terms and classified about 30 lines between 803 Å and 1050 Å. Kruger and Shoupp subsequently observed 4 lines in the interval 219 Å to 232 Å, and classified 3 as combinations from the ground term. These later observations indicate a correction of $+331180 \text{ cm}^{-1}$ to Mack's values, to reduce them to the ground term zero. In compiling the terms in the table this correction has been made, and the last figure has been rounded off throughout.

Kruger and Shoupp have determined the limit from an empirical formula based on extrapolation of isoelectronic sequence data. Their limit is entered in brackets in the table.

The singlet and triplet terms are connected by observed intersystem combinations.

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- J. E. Mack and D. Borg, Phys. Rev. **37**, 470 (A) (1931) and unpublished material (see 1934 ref. below). (T)
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As VI					As VI				
Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$3d^{10}$	$3d^{10} {}^1S$	0	0		$3d^9({}^2D)4p$	$4p {}^3D^\circ$	3	449470	10 -5110
$3d^9({}^2D)4s$	$4s {}^3D$	3	331180	-2050			2	449460	
		2	333230	-3640			1	454570	
		1	336870		$3d^9({}^2D)4p$	$4p {}^1F^\circ$	3	451800	
$3d^9({}^2D)4s$	$4s {}^1D$	2	340230		$3d^9({}^2D)4p$	$4p {}^1P^\circ$	1	452400	
$3d^9({}^2D)4p$	$4p {}^3P^\circ$	2	435430	-5460	$3d^9({}^2D)4p$	$4p {}^1D^\circ$	2	455560	
		1	440890	-3170					
		0	444060						
$3d^9({}^2D)4p$	$4p {}^3F^\circ$	4	443070	3160					
		3	439910	-3890	As VII (${}^2D_{2\frac{1}{2}}$)	Limit		[1028800]	
		2	443800						

December 1949.

AS VI OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 +$	Observed Terms			
$3d^{10}$	$3d^{10} {}^1S$			
	$ns (n \geq 4)$		$np (n \geq 4)$	
$3d^9({}^2D)nx$	{ $4s {}^3D$ $4s {}^1D$	$4p {}^3P^\circ$ $4p {}^1P^\circ$	$4p {}^3D^\circ$ $4p {}^1D^\circ$	$4p {}^3F^\circ$ $4p {}^1F^\circ$

*For predicted terms in the spectra of the Ni I isoelectronic sequence, see Vol. II, Introduction.

SELENIUM

Se I

34 electrons

Z=34

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^4 \ ^3P_2$ $4p^4 \ ^3P_2$ 78658.22 cm^{-1}

I. P. 9.75 volts

The analysis is by Ruedy and Gibbs. They photographed the spectrum, as excited by means of a positive column discharge in helium, from 300 A to 11000 A. Out of a total of 510 observed lines they have classified 391. Their limit is well determined from two long series: $5p \ ^5P - ns \ ^5S^\circ (n=5 \text{ to } 11)$, and $5p \ ^5P_3 - nd \ ^5D_4^\circ (n=4 \text{ to } 11)$. Observed intersystem combinations connect the terms of all three multiplicities.

An analysis of the Se I spectrum has also been published independently by Meissner, Bartelt, and Eckstein, who observed it from 3588 A to 9665 by A, using a condensed discharge through selenium vapor as the source.

The two published line lists are surprisingly discordant with regard to the wavelengths of a number of lines common to both, and also with regard to the number of lines recorded independently from the two sets of observations, i. e., not common to both lists. This raises two questions: (1) the proper source of excitation to be used in observing the spectrum, and (2) the accuracy of the measured wavelengths. Further observations are needed to clarify this puzzling situation.

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Se I

Se I

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval		
$4p^4$	$4p^4 \ ^3P$	2	0. 00	-1989. 49 -544. 86	$4p^3(^4S^\circ)4d$	$4d \ ^5D^\circ$	4	63370. 07	-17. 39 14. 27 -9. 55 1. 90		
		1	1989. 49				3	63387. 46			
		0	2534. 35				2	63373. 19			
$4p^4$	$4p^4 \ ^1D$	2	9576. 08		$4p^3(^2D^\circ)5s$	$5s' \ ^1D^\circ$	1	63382. 74			
		0					0	63380. 84			
$4p^4$	$4p^4 \ ^1S$	0	22446. 03				2	63479. 31			
$4p^3(^4S^\circ)5s$	$5s \ ^5S^\circ$	2	48182. 19		$4p^3(^4S^\circ)4d$	$4d \ ^3D^\circ$	1	65299. 44	-21. 48 61. 76		
$4p^3(^4S^\circ)5s$	$5s \ ^3S^\circ$	1	50996. 93				2	65277. 96			
							3	65339. 72			
$4p^3(^4S^\circ)5p$	$5p \ ^5P$	1	59242. 88	45. 03 103. 47	$4p^3(^4S^\circ)6s$	$6s \ ^5S^\circ$	2	65989. 12			
		2	59287. 91				$4p^3(^4S^\circ)6s$	$6s \ ^3S^\circ$		1	66623. 12
		3	59391. 38								
$4p^3(^4S^\circ)5p$	$5p \ ^3P$	2	60677. 46	55. 09 -73. 70	$4p^3(^4S^\circ)6p$	$6p \ ^5P$	1	69263. 45	14. 33 36. 80		
		1	60622. 37				2	69277. 78			
		0	60696. 07				3	69314. 58			
$4p^3(^2D^\circ)5s$	$5s' \ ^3D^\circ$	1	61681. 31	147. 21 419. 10	$4p^3(^4S^\circ)6p$	$6p \ ^3P$	2	69614. 07	14. 58 -29. 98		
		2	61828. 52				1	69599. 49			
		3	62247. 62				0	69629. 47			

Se I—Continued

Se I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval	
$4p^3(4S^\circ)5d$	$5d \ ^5D^\circ$	0			$4p^3(2D^\circ)4d$	$4d' \ ^2P^\circ$	1	75258.47		
		1	70388.56	0.50			2	75309.88		
		2	70389.06	2.66			3	75312.00		
		3	70391.72	-0.68			2	75312.84		
$4p^3(2P^\circ)5s$	$5s'' \ ^3P^\circ$	0	70995.0		$4p^3(4S^\circ)7d$	$7d \ ^3D^\circ$	3	75364.92	-0.84	
		1	71199.6	204.6			2	75364.92	-52.08	
		2	71659.02	459.4			1	75574.14		
$4p^3(4S^\circ)5d$	$5d \ ^3D^\circ$	1	71106.25	-10.01	$4p^3(4S^\circ)9s$	$9s \ ^5S^\circ$	2	75587.25		
		2	71096.24	58.72			1 to 5	75588.92		
		3	71154.96				2 to 4	75638.12		
$4p^3(4S^\circ)7s$	$7s \ ^5S^\circ$	2	71638.28		$4p^3(4S^\circ)9s$	$9s \ ^3S^\circ$	1	75638.12		
$4p^3(2D^\circ)5p$	$5p' \ ^3D$	1	71856.09	571.22	$4p^3(2D^\circ)4d$	$4d' \ ^3P^\circ$	2	75973.6		
		2	72427.31	603.62			1	76013.93		
		3	73030.93				2	76016.64		
$4p^3(4S^\circ)7s$	$7s \ ^3S^\circ$	1	71890.33		$4p^3(4S^\circ)9p$	$9p \ ^5P$	3	76021.52	2.71	
		2					1	76016.64	4.88	
		3					2	76021.52		
$4p^3(2P^\circ)5s$	$5s'' \ ^1P^\circ$	1	72568.56		$4p^3(4S^\circ)9p$	$9p \ ^3P$	0			
$4p^3(2D^\circ)5p$	$5p' \ ^3F$	2	72618.29	-1.67	$4p^3(4S^\circ)8d$	$8d \ ^3D^\circ$	1	76104.53	8.83	
		3	72716.62	537.48			2	76113.36		
		4	73254.10				3	76128.78		
		1	72866.21				2	76248.69		
$4p^3(2D^\circ)5p$	$5p' \ ^1P$	1	72866.21		$4p^3(4S^\circ)8d$	$8d \ ^5D^\circ$	1	76216.28	-19.91	
		2					2	76159.61	32.41	
		3					3	76161.59		
$4p^3(4S^\circ)7p$	$7p \ ^3P$	0	73042.05	0.99	$4p^3(4S^\circ)8d$	$8d \ ^5D^\circ$	2	76160.45	-1.98	
		1	73043.04	9.51			1	76161.59	1.14	
		2	73052.55				0	76160.95	-0.50	
$4p^3(4S^\circ)7p$	$7p \ ^5P$	1	73083.23	11.19	$4p^3(4S^\circ)10s$	$10s \ ^5S^\circ$	2	76397.26		
		2	73094.42	6.78			3	76400.41		
		3	73101.20				1 to 5	76403.68		
$4p^3(2D^\circ)5p$	$5p' \ ^1F$	3	73264.03		$4p^3(2D^\circ)4d$	$4d' \ ^4P^\circ$	3	76400.41		
$4p^3(4S^\circ)6d$	$6d \ ^5D^\circ$	0			$4p^3(4S^\circ)8f$	$8f \ ^5F$	2 to 4	76404.89		
		1	73543.68	-2.26			$4p^3(4S^\circ)8f$	$8f \ ^3F$	1	76435.27
		2	73541.42	6.14					0	
		3	73547.56	0.1					1	76766.58
4	73547.7		2	76770.41						
$4p^3(4S^\circ)6d$	$6d \ ^3D^\circ$	1	73863.88	-43.89	$4p^3(4S^\circ)9d$	$9d \ ^5D^\circ$	3	76771.74	3.83	
		2	73819.99	40.69			4	76767.39	1.33	
		3	73860.68				1	76777.65	-4.35	
$4p^3(2D^\circ)5p$	$5p' \ ^3P$	0	74034.06	49.51	$4p^3(4S^\circ)9d$	$9d \ ^3D^\circ$	2	76781.85	4.20	
		1	74083.57	8.53			3	76783.34	1.49	
		2	74092.10				2	76929.68		
$4p^3(2D^\circ)4d$	$4d' \ ^1P^\circ$	2	74161.70		$4p^3(4S^\circ)11s$	$11s \ ^5S^\circ$	2	76929.68		
$4p^3(4S^\circ)8s$	$8s \ ^5S^\circ$	2	74210.06		$4p^3(4S^\circ)9f$	$9f \ ^5F$	1 to 5	76933.25		
$4p^3(4S^\circ)6f$	$6f \ ^5F$	1 to 5	74232.47		$4p^3(4S^\circ)10d$	$10d \ ^5D^\circ$	0			
$4p^3(4S^\circ)6f$	$6f \ ^3F$	2 to 4	74234.58		$4p^3(4S^\circ)10f$	$10f \ ^5F$	1	77184.09	-0.30	
$4p^3(4S^\circ)8s$	$8s \ ^3S^\circ$	1	74318.20				2	77183.79	-0.31	
$4p^3(4S^\circ)8p$	$8p \ ^5P$	1	74948.23	3.69			3	77183.48	3.31	
		2	74951.92	8.38			4	77186.79		
		3	74960.30		1 to 5	77295.98				
$4p^3(4S^\circ)8p$	$8p \ ^3P$	0	75137.89	4.43	$4p^3(4S^\circ)11d$	$11d \ ^5D^\circ$	0			
		1	75142.32	4.39			1	77471.5	2.2	
		2	75146.71				2	77473.70	0.48	
$4p^3(4S^\circ)7d$	$7d \ ^5D^\circ$	0			$4p^3(4S^\circ)11f$	$11f \ ^5F$	3	77474.18		
		1	75154.12	42.91			1 to 5	77555.41		
		2	75197.03	-1.26						
		3	75195.77	-0.1						
		4	75195.7		Se II ($4S_{1/2}^\circ$)	Limit		78658.22		

Se I OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 3d ¹⁰ 4s ² +	Observed Terms			
4p ⁴	{ 4p ⁴ ¹ S 4p ⁴ ³ P 4p ⁴ ¹ D			
	ns (n ≥ 5)	np (n ≥ 5)	nd (n ≥ 4)	nf (n ≥ 4)
4p ³ (⁴ S ^o)nx	{ 5-11s ⁶ S ^o 5-10s ³ S ^o	5-9p ⁵ P 5-9p ³ P	4-11d ⁵ D ^o 4- 9d ³ D ^o	6-11f ⁵ F 6- 8f ³ F
4p ³ (² D ^o)nx'	{	5p' ³ P 5p' ³ D 5p' ³ F 5p' ¹ P 5p' ¹ D 5p' ¹ F		
4p ³ (² P ^o)nx''	{	5s'' ³ P ^o 5s'' ¹ P ^o		

*For predicted terms in the spectra of the Se I isoelectronic sequence, see Vol. II, Introduction.

Se II

(As I sequence; 33 electrons)

Z=34

Ground state 1s² 2s² 2p⁶ 3s² 3p⁶ 3d¹⁰ 4s² 4p³ ⁴S_{1/2}^o

4p³ ⁴S_{1/2}^o 173557 cm⁻¹

I. P. 21.5 volts

Three different term lists having a number of levels in common, have been published for Se II. Martin gives 192 classified lines in the range between 694 Å and 9816 Å, and discusses the other investigations in detail, noting which levels were found by Krishnamurty and Rao. His analysis is quoted in the table. The observed *g*-values are from van den Bosch, who suggests the following additions and corrections:

Level	Martin	K and R*	v d B*	Obs. <i>g</i>
98182. 1		4d ⁴ F _{4/2}	4d ⁴ F _{4/2}	1. 27
127984. 3	21 _{3/2}		5p' ² F _{3/2}	1. 24
131165. 4	23 _{3/2} (5p' ² F _{3/2})	d	5p' ² F _{1/2}	1. 26
132189. 6	5p' ² P _{1/2}		5p'' ² P _{1/2}	1. 16
138365. 7	5d ⁴ F _{1/2} ?	5d ⁴ D _{1/2}	5d ⁴ D _{1/2} ?	1. 44
140946. 0	31 _{2/2}	5d ⁴ D _{2/2}	5d ⁴ D _{2/2}	1. 25
142128. 7		5d ⁴ D _{3/2}	5d ⁴ D _{3/2}	1. 33
142302. 2	33 _{1/2}		33 _{2/2}	0. 87

*Letters refer to authors listed under references.

The corrected *J*-value of the level 131165 has been adopted. The other changes are noted separately because some lines acquire two designations if any two sets of terms are combined. Further study will doubtless clarify some of the present differences in this respect. An evident misprint in level "27" has also been corrected.

The limit is also from Martin. He has determined it from the 5,6s ⁴P and 5,6s ²P series, by using a Rydberg formula.

The doublet and quartet systems of terms are connected by observed intersystem combinations.

Se II—Continued

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Se II

Se II

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
4s ² 4p ³	4p ³ ⁴ S ^o	1½	0. 0			4s ² 4p ² (³ P) 5p	5p ² P ^o	0½ 1½	121273. 2 121381. 9	108. 7	0. 64 1. 27
4s ² 4p ³	4p ³ ² D ^o	1½ 2½	13168. 2 13784. 4	616. 2			15	2½	121730		
4s ² 4p ³	4p ³ ² P ^o	0½ 1½	23038. 3 23894. 8	856. 5			16	2½	122720		
4s 4p ⁴	4p ⁴ ⁴ P	2½ 1½ 0½	83876. 7 85579. 5 86437. 7	-1702. 8 -858. 2	1. 57 1. 67 2. 60		17	1½	123323		
4s ² 4p ² (³ P) 5s	5s ⁴ P	0½ 1½ 2½	95270. 0 96753. 3 98674. 4	1483. 3 1921. 1	2. 51 1. 68 1. 56	4s ² 4p ² (¹ D) 5p	5p' ² D ^o	2½ 1½	127415. 7 127921. 4	-505. 7	0. 97 0. 8
	1	0½	96517. 4				20	0½	127867. 5		
	2	2½	96655. 3		1. 30		21°	2½	127984. 3		1. 24
	3	0½	98118. 2				22	1½	129010. 4		
4s ² 4p ² (³ P) 5s	5s ² P	0½ 1½	98896. 1 101356. 0	2459. 9	0. 77 1. 37		23°	1½	131165. 4		1. 26
	4	1½	99368. 5		0. 82	4s ² 4p ² (¹ D) 5p	5p' ² P ^o	0½ 1½	132189. 6		1. 16
4s 4p ⁴	4p ⁴ ² P	1½ 0½	100295. 1		1. 14		24	2½	133867. 1		
	5	2½	101631. 5		1. 12	4s ² 4p ² (³ P) 6s	6s ⁴ P	0½ 1½ 2½	134042. 9 135635. 6 137669. 1	1592. 7 2033. 5	2. 35 1. 67 1. 60
	6	1½	104694. 4				25	1½	136188. 2		1. 29
	7	1½	104873. 7			4s ² 4p ² (³ P) 5d	5d ⁴ F	1½ 2½ 3½ 4½	138365. 7? 138923. 1 140011. 8 141710. 1	557. 4 1088. 7 1698. 3	1. 44 1. 14 1. 27 1. 27
	8	0½	105258. 0			4s ² 4p ² (³ P) 6s	6s ² P	0½ 1½	138525. 7 140939. 9	2414. 2	0. 90
	9	0½	105973. 8				26	2½	138535. 9		
4s ² 4p ² (¹ D) 5s	5s' ² D	2½ 1½	108355. 7 108449. 8	-94. 1	1. 15 0. 81		27	1½	138701. 0		
	10	2½	108833. 9				28	1½	140131. 3		1. 85
	11	2½	110297. 4				29	1½	140745. 8		
	12	1½	112403. 4				30	2½	140930. 1		
4s ² 4p ² (³ P) 5p	5p ⁴ P ^o	0½ 1½ 2½	113048. 7 116776. 6 118398. 0	3727. 9 1621. 4	1. 64 1. 45 1. 40		31	2½	140946. 0		1. 25
4s ² 4p ² (³ P) 5p	5p ⁴ D ^o	0½ 1½ 2½ 3½	114299. 0 114711. 7 116068. 1 117798. 7	412. 7 1356. 4 1730. 6	0. 62 1. 28 1. 39 1. 38		32	2½	142171. 0		
4s ² 4p ² (³ P) 5p	5p ² S ^o	0½	117406. 0		2. 49		33	1½	142302. 2		
4s ² 4p ² (³ P) 5p	5p ² D ^o	1½ 2½	117739. 6 120387. 1	2647. 5	1. 10 1. 31		34	1½	142374. 1		
4s ² 4p ² (³ P) 5p	5p ⁴ S ^o	1½	119308. 5		1. 78		35	2½	143341. 7		1. 24
	13	0½	119343				36	2½	143919. 5		
	14°	1½	121051. 5			Se III (³ P ₀)	Limit	-----	173557		

SE II OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} +$	Observed Terms		
$4s^2 4p^3$	{ $4p^3 \ ^4S^\circ$ $4p^3 \ ^2P^\circ$ $4p^3 \ ^2D^\circ$		
$4s 4p^4$	{ $4p^4 \ ^4P$ $4p^4 \ ^2P$		
	$ns (n \geq 5)$	$np (n \geq 5)$	$nd (n \geq 4)$
$4s^2 4p^2(^3P)nx$	{ $5,6s \ ^4P$ $5,6s \ ^2P$	$5p \ ^4S^\circ$ $5p \ ^4P^\circ$ $5p \ ^4D^\circ$ $5p \ ^2S^\circ$ $5p \ ^2P^\circ$ $5p \ ^2D^\circ$	$5d \ ^4F$
$4s^2 4p^2(^1D)nx'$		$5s' \ ^2D$ $5p' \ ^2P^\circ$ $5p' \ ^2D^\circ$	

*For predicted terms in the spectra of the As I isoelectronic sequence, see Vol. II, Introduction.

Se III

(Ge I sequence; 32 electrons)

Z=34

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^2 \ ^3P_0$

$4p^2 \ ^3P_0$ 258000 cm^{-1}

I. P. 32.0 volts

The analysis is incomplete. The authors have observed the spectrum from 517 Å to 6613 Å and classified 218 lines. They give the limit 274924 cm^{-1} , derived from the combinations $4p^2 \ ^3P - 5s \ ^3P^\circ$; $5s \ ^3P^\circ - 5p \ ^3P$; and $4p^2 \ ^1D - 5s \ ^1P^\circ$; $5s \ ^1P^\circ - 5p \ ^1D$. The writer has recalculated the limit from the series $5,6s \ ^1,^3P^\circ$; $4,5d \ ^1,^3F^\circ$; $4,5d \ ^1D_2^\circ$; $4,5d \ ^3D_3^\circ$; $4,5d \ ^3P_{2,1}^\circ$ by using a Rydberg formula and assuming that $5,6s \ ^3P_2$; $5,6s \ ^1P_1^\circ$; $4,5d \ ^3F_{4,3,2}^\circ$ have as a limit the component $4p \ ^2P_{1/2}^\circ$ in Se IV. This component is 4376 cm^{-1} above the ground state of the ion, $4p \ ^2P_{0/2}^\circ = 0$. Consequently, this correction has been taken into account in determining the final value of the limit quoted in the table.

Observed intersystem combinations connect the singlet and triplet systems of terms.

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Se III

Se III

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval		
$4s^2 4p^2$	$4p^2 \ ^3P$	0	0	1741 2196	$4s^2 4p(^2P^\circ)5p$	$5p \ ^1P$	1	150761.1			
		1	1741				$4s^2 4p(^2P^\circ)5p$	$5p \ ^3D$		1	153211.4
		2	3937							2	153520.9
									3172.0		
$4s^2 4p^2$	$4p^2 \ ^1D$	2	13032								
$4s 4p^3$	$4p^3 \ ^3D^\circ$	1	91091	1632 3825	$4s^2 4p(^2P^\circ)5p$	$5p \ ^3P$	0	154783.0	1577.4 1513.8		
		2	92723				1	156360.4			
		3	96548				2	157874.2			
$4s 4p^3$	$4p^3 \ ^3P^\circ$	0	106482	109 -76	$4s^2 4p(^2P^\circ)5p$	$5p \ ^3S$	1	159302.9			
		1	106591				$4s^2 4p(^2P^\circ)5p$	$5p \ ^1D$		2	161170.2
		2	106515								
$4s 4p^3$	$4p^3 \ ^1D^\circ$	2	112565		$4s^2 4p(^2P^\circ)6s$	$6s \ ^3P^\circ$	0	187168.7	257.8 4096.5		
				1			187426.5				
				2			191523.0				
$4s^2 4p(^2P^\circ)4d$	$4d \ ^3F^\circ$	2	124051.9	1257.9 2099.7	$4s^2 4p(^2P^\circ)5d$	$5d \ ^3F^\circ$	2	188429.2	1218.7 1945.7		
		3	125309.8				3	189647.9			
		4	127409.5				4	191593.6			
$4s^2 4p(^2P^\circ)5s$	$5s \ ^3P^\circ$	0	126276.9	504.5 3609.6	$4s^2 4p(^2P^\circ)5d$	$5d \ ^3D^\circ$	1	190841.2	-821.0 3894.8		
		1	126781.4				2	190020.2			
		2	130391.0				3	193915.0			
$4s^2 4p(^2P^\circ)5s$	$5s \ ^1P^\circ$	1	131655.9		$4s^2 4p(^2P^\circ)6s$	$6s \ ^1P^\circ$	1	192161.8			
$4s^2 4p(^2P^\circ)4d$	$4d \ ^1P^\circ$	1	136946.5	$4s^2 4p(^2P^\circ)5d$			$5d \ ^1D^\circ$	2	193304.4		
$4s^2 4p(^2P^\circ)4d$	$4d \ ^1D^\circ$	2	139203.7								
$4s^2 4p(^2P^\circ)4d$	$4d \ ^3D^\circ$	1	140639.7		-1230.0 2604.9	$4s^2 4p(^2P^\circ)5d$		$5d \ ^3P^\circ$	0	194950.5	-223.5
		2	139409.7	1			194727.0				
		3	142014.6	2							
$4s^2 4p(^2P^\circ)4d$	$4d \ ^3P^\circ$	0	142315.8	442.4 -52.1	$4s^2 4p(^2P^\circ)5d$	$5d \ ^1F^\circ$	3	196845.4			
		1	142758.2								
		2	142706.1								
$4s^2 4p(^2P^\circ)4d$	$4d \ ^1F^\circ$	3	148675		Se IV ($^2P_{3/2}$)	<i>Limit</i>			258000		

August 1950.

Se III OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} +$	Observed Terms					
$4s^2 4p^2$	$\left\{ \begin{array}{l} 4p^2 \ ^3P \\ 4p^2 \ ^1D \end{array} \right.$					
$4s 4p^3$	$\left\{ \begin{array}{l} 4p^3 \ ^3P^\circ \\ 4p^3 \ ^3D^\circ \\ 4p^3 \ ^1D^\circ \end{array} \right.$					
	<i>ns</i> ($n \geq 5$)		<i>np</i> ($n \geq 5$)		<i>nd</i> ($n \geq 4$)	
$4s^2 4p(^2P^\circ)nx$	$\left\{ \begin{array}{l} 5,6s \ ^3P^\circ \\ 5,6s \ ^1P^\circ \end{array} \right.$		$5p \ ^3S$	$5p \ ^3P$ $5p \ ^1P$	$5p \ ^3D$ $5p \ ^1D$	$4,5d \ ^3P^\circ$ $4d \ ^1P^\circ$ $4,5d \ ^3D^\circ$ $4,5d \ ^1D^\circ$ $4,4d \ ^3F^\circ$ $4,5d \ ^1F^\circ$

*For predicted terms in the spectra of the Ge I isoelectronic sequence, see Vol. II, Introduction.

Se IV

(Ga I sequence; 31 electrons)

 $Z=34$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^2 P_{0\frac{1}{2}}^{\circ}$ $4p^2 P_{0\frac{1}{2}}^{\circ}$ 346094 cm^{-1}

I. P. 42.9 volts

The analysis is by Rao and Badami who have published 35 classified lines between 635 Å and 3059 Å. Their limit "depends upon a choice of 70240 cm^{-1} for the term $5g^2 G$, which is assumed to be very nearly hydrogenic." It is entered in brackets in the table.

REFERENCE

K. R. Rao and J. S. Badami, Proc. Roy. Soc. London [A] **131**, 159 (1931). (I P) (T) (C L)

Se IV					Se IV				
Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$4s^2 (1S) 4p$	$4p^2 P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	0 4376	4376	$4s^2 (1S) 4f$	$4f^2 F^{\circ}$	$3\frac{1}{2}$ $2\frac{1}{2}$	229684 229714	-30
$4s 4p^2$	$4p^2^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	104211 104706	495	$4s^2 (1S) 5d$	$5d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	237747 237899	152
$4s 4p^2$	$4p^2^2 S$	$0\frac{1}{2}$	128787		$4s^2 (1S) 6s$	$6s^2 S$	$0\frac{1}{2}$	240745	
$4s 4p^2$	$4p^2^2 P$	$0\frac{1}{2}$ $1\frac{1}{2}$	136134 138354	2220	$4s^2 (1S) 5g$	$5g^2 G$	$3\frac{1}{2}, 4\frac{1}{2}$	275854	
$4s^2 (1S) 4d$	$4d^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	153217 153606	389	$4s^2 (1S) 7s$	$7s^2 S$	$0\frac{1}{2}$	288146?	
$4s^2 (1S) 5s$	$5s^2 S$	$0\frac{1}{2}$	157241		Se v ($1S_0$)	<i>Limit</i>		[346094]	
$4s^2 (1S) 5p$	$5p^2 P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	189913 191111	1198					

May 1950.

Se V

(Zn I sequence; 30 electrons)

 $Z=34$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 1S_0$ $4s^2 1S_0$ 589781 cm^{-1}

I. P. 73.1 volts

This spectrum is incompletely analyzed. Sawyer and Humphreys found the three triplet terms $4p^3 P^{\circ}$, $4d^3 D$, and $5s^3 S$, and also the combinations $4s^2 1S - 4p^1 P^{\circ}$ and $4p^3 P^{\circ} - 4p^2 3P$, from 16 lines between 505 Å and 839 Å. They estimated the absolute term values by an extrapolation of the Moseley diagram.

Subsequently Rao and Badami classified 6 more lines from observations extending to 1227 Å. They added the singlet term $4d^1 D$ together with 5 intersystem combinations connecting the singlet and triplet systems of terms. Their limit is based on these observations and the absolute values of Sawyer and Humphreys.

REFERENCES

R. A. Sawyer and C. J. Humphreys, Phys. Rev. **32**, 583 (1928). (T) (C L)
K. R. Rao and J. S. Badami, Proc. Roy. Soc. London [A] **131**, 166 (1931). (I P) (T) (C L)

Se v

Config.	Desig.	<i>J</i>	Level	Interval
4s ²	4s ² ¹ S	0	0	
4s(2S)4p	4p ³ P ^o	0	89756	1595 3610
		1	91351	
		2	94961	
4s(2S)4p	4p ¹ P ^o	1	131733	
4p ²	4p ² ³ P	0	211789	2298 2528
		1	214087	
		2	218615	
4s(2S)4d	4d ¹ D	2	213194	
4s(2S)4d	4d ³ D	1	257534	214 335
		2	257748	
		3	258083	
4s(2S)5s	5s ³ S	1	287426	
-----	-----	-----	-----	
Se vi (² S _{1/2})	<i>Limit</i>	-----	589781	

March 1950.

Se VI

(Cu I sequence; 29 electrons)

Z=34

Ground state 1s² 2s² 2p⁶ 3s² 3p⁶ 3d¹⁰ 4s ²S_{0/2}4s ²S_{0/2} 658994 cm⁻¹

I. P. 81.7 volts

By extrapolation along the isoelectronic sequence Sawyer and Humphreys have classified seven lines between 452 Å and 886 Å, and determined the limit quoted here.

REFERENCE

R. A. Sawyer and C. J. Humphreys, Phys. Rev. **32**, 583 (1928). (I P) (T) (C L)

Se VI

Config.	Desig.	<i>J</i>	Level	Interval
4s	4s ² S	0½	0	
4p	4p ² P ^o	0½	112762	5700
		1½	118462	
4d	4d ² D	1½	282830	679
		2½	283509	
5s	5s ² S	0½	333594	
-----	-----	-----	-----	
Se VII (¹ S ₀)	<i>Limit</i>	-----	[658994]	

February 1950.

Se VII

(Ni I sequence; 28 electrons)

Z=34

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} {}^1S_0$ $3d^{10} {}^1S_0$ 1253300 cm^{-1}

I. P. 155 volts

Kruger and Shoupp have observed four lines between 171 Å and 180 Å and classified three of them as combinations of the ground term with terms from $3d^9 4p$.

From preliminary measurements, Edlén has confirmed these results, extended the analysis, and derived provisional absolute term values from the Rydberg nf -series. All the values in the table except those from $3d^9 4p$ are from his unpublished manuscript. His unit, 10^3cm^{-1} has here been changed to cm^{-1} .

Rao and Murti attribute some 40 lines in the interval 560 Å to 860 Å to Se VII and suggest tentative classifications of five of them. Further study of these assignments is needed.

REFERENCES

P. G. Kruger and W. E. Shoupp, Phys. Rev. **46**, 124 (1934). (I P) (T) (C L)K. R. Rao and S. G. K. Murti, Proc. Roy. Soc. London [A] **145**, 696 (1934).

B. Edlén, letter (July 1950). (I P) (T)

Se VII				Se VII			
Config.	Desig.	<i>J</i>	Level	Config.	Desig.	<i>J</i>	Level
$3d^{10}$	$3d^{10} {}^1S$	0	0	$3d^9({}^2D)5f$	$5f {}^1P^\circ$	1	1033500
$3d^9({}^2D)4p$	$4p {}^3P^\circ$	2 1 0	565700	$3d^9({}^2D)5f$	$5f {}^3D^\circ$	3 2 1	1040900
$3d^9({}^2D)4p$	$4p {}^1P^\circ$	1	578300	$3d^9({}^2D)6f$	$6f {}^1P^\circ$	1	1101000
$3d^9({}^2D)4p$	$4p {}^3D^\circ$	3 2 1	581500	$3d^9({}^2D)6f$	$6f {}^3D^\circ$	3 2 1	1108000
$3d^9({}^2D)5p$	$5p {}^1P^\circ$	1	889200	$3d^9({}^2D)7f$	$7f {}^1P^\circ$	1	1141700
$3d^9({}^2D)5p$	$5p {}^3D^\circ$	3 2 1	894100	$3d^9({}^2D)7f$	$7f {}^3D^\circ$	3 2 1	1148400
$3d^9({}^2D)4f$	$4f {}^3P^\circ$	2 1 0	904500	$3d^9({}^2D)8f$	$8f {}^1P^\circ$	1	1167800
$3d^9({}^2D)4f$	$4f {}^1P^\circ$	1	909100	$3d^9({}^2D)8f$	$8f {}^3D^\circ$	3 2 1	1174700
$3d^9({}^2D)4f$	$4f {}^3D^\circ$	3 2 1	917300	Se VIII (${}^2D_{3/2}$)		Limit	1253300

August 1950.

BROMINE

Br I

35 electrons

Z=35

Ground States $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^5 {}^2P_{1/2}^{\circ}$ $4p^5 {}^2P_{1/2}^{\circ}$ 95550 cm^{-1}

I. P. 11.84 volts

This spectrum has been analyzed by Kiess and de Bruin, who have observed more than 300 lines between 3735 Å and 9320 Å. By utilizing the ultraviolet observations by Turner in the interval 1232 Å to 1633 Å, they have classified a total of 208 lines as combinations among terms of the doublet and quartet systems. The terms of these two systems are connected by observed intersystem combinations.

They have determined the limit from a Rydberg formula representing 11 sets of terms of the ns , np , and nd configurations. This limit depends on series of two members each.

Two changes have been made in the published analysis. Revised tables have recently been prepared, which give the maximum binding energies of electrons in the first and second spectra of the elements H through Nb. From an examination of these data Russell has pointed out that the known nd terms are from $6d$ and $7d$, and that the terms from $4d$ and $5d$ are not yet known because the observed lines lie beyond the range of the present observations.

In compiling the table the writer has assumed that the 1S limit in Br II is higher than the 1D limit, although the $5s {}^2S$ and $5s {}^2D$ terms of Br I from these respective limits are in the reverse order. Further analysis of Br II is required to settle this question; the term $4p^4 {}^1S$ in Br II has not been found.

REFERENCES

- L. A. Turner, Phys. Rev. **27**, 400 (1926).
 C. C. Kiess, and T. L. de Bruin, Bur. Std. J. Research **4**, 667, RP 172 (1930). (I P) (T) (C L) (G D)
 P. Lacroute, Ann. de Phys. [11] **3**, 59 [1934]. (Z E).
 J. E. Mack, Rev. Mod. Phys. **22**, No. 1, 64 (1950). (Summary hfs)

Br I

Br I

Config.	Desig.	J	Level	Interval	Obs. g	Config.	Desig.	J	Level	Interval	Obs. g
$4s^2 4p^5$	$4p^5 {}^2P^{\circ}$	$1\frac{1}{2}$ $0\frac{1}{2}$	0 3685	-3685		$4s^2 4p^4 ({}^1D) 5s$	$5s' {}^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	77305. 93 77324. 11	18. 18	
$4s^2 4p^4 ({}^3P) 5s$	$5s {}^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	63429. 82 64900. 50 66877. 16	-1470. 68 -1976. 66	1. 60 1. 50 2. 60	$4s^2 4p^4 ({}^3P) 5p$	$5p {}^2D^{\circ}$	$2\frac{1}{2}$ $1\frac{1}{2}$	78504. 88 79689. 16	-1184. 28	
$4s^2 4p^4 ({}^3P) 5s$	$5s {}^2P$	$1\frac{1}{2}$ $0\frac{1}{2}$	67176. 87 68963. 52	-1786. 65	1. 49 0. 75:	$4s^2 4p^4 ({}^3P) 5p$	$5p {}^4S^{\circ}$	$1\frac{1}{2}$	78669. 92		
$4s^2 4p^4 ({}^3P) 5p$	$5p {}^4P^{\circ}$	$2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	74665. 67 75002. 52 75807. 33	-336. 85 -804. 81		$4s^2 4p^4 ({}^3P) 5p$	$5p {}^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	78858. 98 79171. 64	312. 66	
$4s^2 4p^4 ({}^3P) 5p$	$5p {}^4D^{\circ}$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	75514. 82 75690. 44 76736. 45 78069. 29	-175. 62 -1046. 01 -1332. 84		$4s^2 4p^4 ({}^3P) 5p$	$5p {}^2S^{\circ}$	$0\frac{1}{2}$	79861. 30		
$4s^2 4p^4 ({}^3P) 5p$	$5p {}^4D^{\circ}$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	75514. 82 75690. 44 76736. 45 78069. 29	-175. 62 -1046. 01 -1332. 84		$4s^2 4p^4 ({}^3P) 6p$	$6p {}^4P^{\circ}$	$2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	85520. 14 85579. 60 85792. 56	-59. 46 -212. 96	1. 46 1. 54
$4s^2 4p^4 ({}^1S) 5s$	$5s'' {}^2S$	$0\frac{1}{2}$	75901. 89			$4s^2 4p^4 ({}^3P) 6p$	$6p {}^4D^{\circ}$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	85756. 22 85813. 82 85937. 22 86429. 88	-57. 60 -123. 40 -492. 66	1. 43 1. 23 1. 62 2. 26

Br I—Continued

Br I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$4s^2 4p^4(^3P)6p$	$6p \ ^2P^\circ$	$1\frac{1}{2}$	87252.57	-239.79	1.29 1.06	$4s^2 4p^4(^3P)6d$	$6d \ ^4P$	$0\frac{1}{2}$	90186.50	372.62 223.08	
		$0\frac{1}{2}$	87492.36					$1\frac{1}{2}$	90559.12		
$4s^2 4p^4(^3P)7s$	$7s \ ^2P$	$1\frac{1}{2}$	87748.30	-2354.63		$4s^2 4p^4(^3P)6d$	$6d \ ^4F$	$4\frac{1}{2}$	90238.19	-22.97 -87.17	
		$0\frac{1}{2}$	90102.93					$3\frac{1}{2}$	90261.16		
$4s^2 4p^4(^3P)7s$	$7s \ ^4P$	$2\frac{1}{2}$	88128.33	-140.44		$4s^2 4p^4(^3P)6d$	$6d \ ^2F$	$2\frac{1}{2}$	91432.00	30.44	
		$1\frac{1}{2}$	88268.77					$3\frac{1}{2}$	91462.44		
		$0\frac{1}{2}$	90285.62	-2016.85							
$4s^2 4p^4(^1D)5p$	$5p' \ ^2P^\circ$	$1\frac{1}{2}$	88476.73	-76.19		$4s^2 4p^4(^3P)6d$	$6d \ ^2D$	$2\frac{1}{2}$	91646.58	-165.06	
		$0\frac{1}{2}$	88552.92					$1\frac{1}{2}$	91811.64		
$4s^2 4p^4(^1D)5p$	$5p' \ ^2F^\circ$	$3\frac{1}{2}$	88658.93	-15.99		$4s^2 4p^4(^3P)7d$	$7d \ ^4D$	$0\frac{1}{2}$	91733.24	60.16 10.22	
		$2\frac{1}{2}$	88674.92					$1\frac{1}{2}$	91793.40		
$4s^2 4p^4(^3P)6p$	$6p \ ^2D^\circ$	$1\frac{1}{2}$	88841.29	183.76		$4s^2 4p^4(^3P)6d$	$6d \ ^2P$	$1\frac{1}{2}$	91785.35		
		$2\frac{1}{2}$	89025.05					$0\frac{1}{2}$	91785.35		
$4s^2 4p^4(^3P)6p$	$6p \ ^4S^\circ$	$1\frac{1}{2}$	88947.12		1.36	$4s^2 4p^4(^3P)6d$	$6d \ ^2P$	$1\frac{1}{2}$	91785.35		
$4s^2 4p^4(^3P)6p$	$6p \ ^2S^\circ$	$0\frac{1}{2}$	89140.50		0.47			$0\frac{1}{2}$	91785.35		
$4s^2 4p^4(^3P)7p$	$7p \ ^4P^\circ$	$2\frac{1}{2}$	89541.96	-237.52	1.40:	$4s^2 4p^4(^3P)7d$	$7d \ ^4F$	$4\frac{1}{2}$	91844.52	-29.89	
		$1\frac{1}{2}$	89779.48					$3\frac{1}{2}$	91874.41		
		$0\frac{1}{2}$	89941.14	-161.66	1.56:						
$4s^2 4p^4(^3P)6d$	$6d \ ^4D$	$0\frac{1}{2}$	89722.68	138.45 80.16 -87.27		$4s^2 4p^4(^3P)7d$	$7d \ ^4P$	$2\frac{1}{2}$	92722.87	-23.97 10.54	
		$1\frac{1}{2}$	89861.13					$1\frac{1}{2}$	92746.84		
		$2\frac{1}{2}$	89941.29					$0\frac{1}{2}$	92736.30		
		$3\frac{1}{2}$	89854.02								
$4s^2 4p^4(^1S)5p$	$5p'' \ ^2P^\circ$	$0\frac{1}{2}$	89758.04	141.12		Br II (3P_2)	<i>Limit</i>	-----	95550		
		$1\frac{1}{2}$	89899.16								
$4s^2 4p^4(^1D)5p$	$5p' \ ^2D^\circ$	$2\frac{1}{2}$	89786.97	-67.37							
		$1\frac{1}{2}$	89854.34								

January 1950.

Br I OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} +$	Observed Terms					
$4s^2 4p^5$	$4p^5 \ ^2P^\circ$					
	$ns \ (n \geq 5)$		$np \ (n \geq 5)$			$nd \ (n \geq 4)$
$4s^2 4p^4(^3P)nx$	{	$5,7s \ ^4P$	$5,6p \ ^4S^\circ$	$5,6,7p \ ^4P^\circ$	$5,6p \ ^4D^\circ$	$6,7d \ ^4P$
		$5,7s \ ^2P$	$5,6p \ ^2S^\circ$	$5,6p \ ^2P^\circ$	$5,6p \ ^2D^\circ$	$6d \ ^2P$
$4s^2 4p^4(^1D)nx'$				$5p' \ ^2P^\circ$	$5p' \ ^2D^\circ$	$6,7d \ ^4D$
		$5s' \ ^2D$				$6d \ ^2D$
$4s^2 4p^4(^1S)nx''$				$5p'' \ ^2P^\circ$		$6,7d \ ^4F$
						$6d \ ^2F$

*For predicted terms in the spectra of the Br I isoelectronic sequence, see Vol. II, Introduction.

Br II

(Se I sequence; 34 electrons)

Z=35

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^4 \ ^3P_2$ $4p^4 \ ^3P_2$ 174119 cm^{-1}

I. P. 21.6 volts

This spectrum would well repay further observation; the analysis is far from complete. L. Bloch, E. Bloch, and P. Lacroute have given absolute values for 23 terms from the two limits $^4S^\circ$ and $^2D^\circ$ in Br III. The limit quoted here is from their paper, which gives no details about its derivation. The writer derives the limit 175870 (I P 21.8) from the $ns \ ^5S^\circ$ and $ns \ ^3S^\circ$ series ($n=5,6$) by using a Rydberg formula. This value is probably too high.

R. Ramanadham and K. R. Rao have extended the earlier analysis and revised the $4d \ ^3D^\circ$, $5d \ ^3D^\circ$, and $5s' \ ^1D^\circ$ terms. In general their terms have been used in preparing the table; although the writer has made adjustments in the values of the $6s \ ^5S^\circ$, $6s \ ^3S^\circ$, $5d \ ^5D^\circ$, and $5d \ ^3D^\circ$ terms by utilizing unpublished wave lengths by Kiess. The level they list as $6s' \ ^3D_2^\circ$ has been corrected and entered in the table as 2_2° . The two levels listed as $4d \ ^3D_{2,1}^\circ$ by the Blochs and Lacroute, and rejected by Ramanadham and Rao, are included in the table as $4p^5 \ ^3P^\circ$. This designation is tentatively suggested by the writer.

The observed g -values are from the 1934 paper except for the following, which have been determined by the writer from Lacroute's observed Zeeman patterns:

$4p^5 \ ^3P^\circ$, $5p' \ ^3D_{2,3}$, $5p' \ ^3F_4$, $5d' \ ^1D^\circ$, $5d' \ ^1P^\circ$, $5d' \ ^1G^\circ$, $6s' \ ^1D^\circ$, $5d' \ ^3F_3^\circ$, $5d' \ ^3G^\circ$, $5d' \ ^3P^\circ$, $5d' \ ^3S^\circ$, $5d' \ ^1F^\circ$.

Approximately 260 lines are classified in the interval between 711 A and 6352 A. Observed intersystem combinations connect the terms of all multiplicities.

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- P. Lacroute, Ann. de Phys. [11] **3**, 5-96 (1935). (T) (C L) (Z E)
 L. Bloch, E. Bloch, et P. Lacroute, Compt. Rend. **199**, 41 (1934). (I P) (T) (C L) (Z E)
 R. Ramanadham and K. R. Rao, Ind. J. Phys. **18**, 317 (1944). (T) (C L)
 C. C. Kiess, unpublished material (1940).
 J. D. Ranade, Phil. Mag. **42**, No. 326, 284 (1951). (hfs)

Br II

Br II

Config.	Desig.	J	Level	Interval	Obs. g	Config.	Desig.	J	Level	Interval	Obs. g
$4p^4$	$4p^4 \ ^3P$	2	0	-3139 -701		$4p^3(^4S^\circ)4d$	$4d \ ^3D^\circ$	1	112395.86	-865.34 930.03	0.52
		1	3139					2	111530.52		1.10
		0	3840					3	112460.55		1.29
$4p^4$	$4p^4 \ ^1D$	2	11409			$4p^3(^2D^\circ)5s$	$5s' \ ^1D^\circ$	2	112939.46		1.06
$4p^3(^4S^\circ)5s$	$5s \ ^5S^\circ$	2	93927.48		2.02	$4p^3(^4S^\circ)5p$	$5p \ ^5P$	1	114682.76	135.36 358.05	2.51
$4s \ 4p^5$	$4p^5 \ ^3P^\circ$	1	96439.39	-2367.96	1.53			2	114818.12		1.83
		0	98807.35		1.62			3	115176.17		1.69
$4p^3(^4S^\circ)5s$	$5s \ ^3S^\circ$	1	98476.38		1.99	$4p^3(^2P^\circ)5s$	$5s'' \ ^1P^\circ$	1	116786.0		
$4p^3(^2D^\circ)5s$	$5s' \ ^3D^\circ$	1	109428.2	154.3 695.7	0.55	$4p^3(^4S^\circ)5p$	$5p \ ^3P$	2	117767.56	206.06 -272.84	1.51
		2	109682.5		1.14			1	117561.50		1.49
		3	110378.2		1.29			0	117834.34		0/0

Br II—Continued

Br II—Continued

Config.	Desig.	J	Level	Interval	Obs. g	Config.	Desig.	J	Level	Interval	Obs. g
4p ³ (² P°)5s	5s'' ³ P°	0	121154.0			4p ³ (² D°)5d	5d' ³ D°	1	151357.8		1.30
		1	121558.0	404.0	2			151502.27	144.5	1.28	
		2	122629.0	1071.0	3			152380.58	878.31		
4p ³ (² D°)5p	5p' ³ D	1	128676.26	1289.60	0.67	4p ³ (² D°)5d	5d' ¹ D°	2	151459.30		1.06
		2	129965.86	688.02	1.19			2	152832.3		
		3	130653.88		1.31			1	154088.7		
4p ³ (² D°)5p	5p' ³ F	2	130609.88	698.87	0.86	4p ³ (² D°)5d	5d' ¹ G°	1	154984.6		1.20
		3	131308.75	437.0	1.18			2	155001.5		
		4	131745.7		1.31			2	155591.61		
4p ³ (² D°)5p	5p' ¹ P	1	130823.82		1.08	4p ³ (² D°)5d	5d' ³ F°	3	156036.3	444.7	0.71
		3	131688.75		1.06			4	156684.39	648.1	1.11
		2	133278.13	-323.01	1.42			1	156025.0		1.20
4p ³ (² D°)5p	5p' ³ P	1	133601.14	26.51	1.48	4p ³ (² D°)6s	6s' ³ D°	2			
		0	133574.63		0/0			1			
		2	135794.2		2.02			2			
4p ³ (⁴ S°)6s	6s ⁵ S°	2	135794.2		2.02	4p ³ (² D°)5d	5d' ³ G°	3	156116.1	36.2	0.88
4p ³ (² D°)5p	5p' ¹ D	2	136607.73		1.01			4	156152.3	123.3	1.13
		1	137608.1					5	156275.6		1.23
		1	138126.9			2	158234.3		1.52		
4p ³ (⁴ S°)5d	5d ³ D°	3	139257.5	-866.1		4p ³ (² D°)5d	5d' ³ P°	1	157633.7	600.6	1.40
		2	140123.6	673.4	1.50			0	159257.9	-1624.2	
		1	139450.2					1	157725.9		1.64
4p ³ (⁴ S°)5d	5d ⁵ D°	0	140223.09	-0.43	0/0	4p ³ (² D°)5d	5d' ¹ F°	3	160887.7		1.07
		1	140222.66		1.49						
		2	140506.63		1.51						
		3	140294.19	-212.44	1.50						
		4	140291.94	-2.25	1.51						
					Br III (⁴ S _{1/2})	Limit		174119			

October 1950.

Br II OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 3d ¹⁰ +	Observed Terms									
4s ² 4p ⁴	{ 4p ⁴ ³ P · 4p ⁴ ¹ D									
4s 4p ⁵	4p ⁵ ³ P°									
	ns(n ≥ 5)			np(n ≥ 5)			nd(n ≥ 4)			
4s ² 4p ³ (⁴ S°)nx	{ 5, 6s ⁵ S° 5, 6s ³ S°			5p ⁵ P 5p ³ P			5d ⁵ D° 4,5d ³ D°			
4s ² 4p ³ (² D°)nx'	{ 5, 6s' ³ D° 5, 6s' ¹ D°			5p' ³ P 5p' ³ D 5p' ³ F 5p' ¹ P 5p' ¹ D 5p' ¹ F			5d' ³ S° 5d' ³ P° 5d' ³ D° 5d' ³ F° 5d' ³ G° 5d' ¹ P° 5d' ¹ D° 5d' ¹ F° 5d' ¹ G°			
4s ² 4p ³ (² P°)nx''	{ 5s'' ³ P° 5s'' ¹ P°									

*For predicted terms in the spectra of the Se I isoelectronic sequence, see Vol. II, Introduction.

Br III

(As I sequence; 33 electrons)

 $Z=35$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^3 {}^4S_{1/2}^{\circ}$ $4p^3 {}^4S_{1/2}^{\circ}$ 289529 cm^{-1}

I. P. 35.9 volts

The analysis is from K. R. Rao and Krishnamurty, who revised the early work of Deb, and published 128 classified lines between 665 Å and 4519 Å. Rao has since (1944) reported that "a further investigation of the spectrum has led to the identification of many intercombination lines . . ." He states that the interval between $4p^3 {}^4S_{1/2}^{\circ}$ and $4p^3 {}^2D_{1/2}^{\circ}$ is 15042 cm^{-1} . This correction has been added to the doublet terms in the 1937 paper, starting with $4p^3 {}^2D_{1/2}^{\circ}$ as zero. The six miscellaneous levels of the odd set, labeled a, b, \dots, f by the authors, are here called $1^{\circ}, 2^{\circ}, \dots, 6^{\circ}$.

The limit is from the two series terms 5, 6s 4P .

REFERENCES

K. R. Rao and S. G. Krishnamurty, Proc. Roy. Soc. London [A] **161**, 38 (1937). (I P) (T) (C L)
K. R. Rao, Current Sci. **13**, 72 (L) (1944). (T)

Br III

Br III

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$4s^2 4p^3$	$4p^3 {}^4S^{\circ}$	$1\frac{1}{2}$	0. 0		$4s^2 4p^2({}^3P)5p$	$5p {}^2D^{\circ}$	$1\frac{1}{2}$ $2\frac{1}{2}$	180695. 7 184178. 9	3483. 2
$4s^2 4p^3$	$4p^3 {}^2D^{\circ}$	$1\frac{1}{2}$ $2\frac{1}{2}$	15042. 0 16301. 0	1259. 0	$4s^2 4p^2({}^3P)5p$	$5p {}^4S^{\circ}$	$1\frac{1}{2}$	182017. 8	
$4s^2 4p^3$	$4p^3 {}^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	26915. 0 28579. 0	1664. 0	$4s^2 4p^2({}^3P)5p$	$5p {}^2S^{\circ}$	$0\frac{1}{2}$	182871. 5	
$4s^2 4p^2(?)4d$	$4d 1$	$1\frac{1}{2}$	137483. 2		$4s^2 4p^2(?)5p$	$5p 1^{\circ}$		184627. 4	
$4s^2 4p^2(?)4d$	$4d 2$	$1\frac{1}{2}, 2\frac{1}{2}$	137531. 1		$4s^2 4p^2(?)5p$	$5p 2^{\circ}$		185422. 2	
$4s^2 4p^2(?)4d$	$4d 3$	$1\frac{1}{2}$	139792. 0		$4s^2 4p^2({}^3P)5p$	$5p {}^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	185432. 4	
$4s^2 4p^2(?)4d$	$4d 4$	$2\frac{1}{2}$	141996. 6		$4s^2 4p^2(?)5p$	$5p 3^{\circ}$		187768. 1	
$4s^2 4p^2(?)4d$	$4d 5$	$0\frac{1}{2}, 1\frac{1}{2}$	143994. 9		$4s^2 4p^2(?)5p$	$5p 4^{\circ}$		188043. 1	
$4s^2 4p^2(?)4d$	$4d 6$	$1\frac{1}{2}$	145174. 6		$4s^2 4p^2(?)5p$	$5p 5^{\circ}$		191903. 3	
$4s^2 4p^2({}^3P)5s$	$5s {}^4P$	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	145416. 9 147670. 4 150259. 8	2253. 5 2589. 4	$4s^2 4p^2(?)5p$	$5p 6^{\circ}$		192109. 7	
$4s^2 4p^2(?)4d$	$4d 7$	$1\frac{1}{2}$	147557. 5		$4s^2 4p^2(?)5d$	$5d 10$	$2\frac{1}{2}$	209006. 1	
$4s^2 4p^2({}^3P)5s$	$5s {}^2P$	$0\frac{1}{2}$ $1\frac{1}{2}$	149199. 5 152109. 8	2910. 3	$4s^2 4p^2({}^3P)6s$	$6s {}^4P$	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	210256. 5 213015. 3 215924. 1	2758. 8 2908. 8
$4s^2 4p^2(?)4d$	$4d 8$	$2\frac{1}{2}$	150917. 8		$4s^2 4p^2(?)5d$	$5d 11$	$1\frac{1}{2}$	212451. 6	
$4s^2 4p^2(?)4d$	$4d 9$	$1\frac{1}{2}$	151063. 2		$4s^2 4p^2(?)5d$	$5d 12$	$1\frac{1}{2}$	212586. 2	
$4s^2 4p^2(1D)5s$	$5s' {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	157746. 7 158512. 4	-765. 7	$4s^2 4p^2(?)5d$	$5d 13$	$1\frac{1}{2}$	214882. 6	
$4s^2 4p^2({}^3P)5p$	$5p {}^4D^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	173181. 8 173840. 0 175910. 4 178322. 5	658. 2 2070. 4 2412. 1	$4s^2 4p^2(?)5d$	$5d 14$	$2\frac{1}{2}$	215470. 3	
$4s^2 4p^2({}^3P)5p$	$5p {}^4P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	176670. 6 178130. 4 180253. 8	1459. 8 2123. 4	$4s^2 4p^2(?)5d$	$5d 15$	$1\frac{1}{2}, 2\frac{1}{2}$	219987. 2	
					$4s^2 4p^2(?)5d$	$5d 16$	$1\frac{1}{2}$	222868. 3	
					-----	-----	-----	-----	-----
					Br IV (3P_0)	Limit		289529	

OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 3d ¹⁰ +	Observed Terms		
4s ² 4p ³	{ 4p ³ ⁴ S ^o	4p ³ ² P ^o	4p ³ ² D ^o
	ns (n ≥ 5)		np (n ≥ 5)
4s ² 4p ² (³ P) nx	{ 5,6s ⁴ P		5p ⁴ S ^o 5p ⁴ P ^o 5p ⁴ D ^o
	5s ² P		5p ² S ^o 5p ² P ^o 5p ² D ^o
4s ² 4p ² (¹ D) nx'		5s' ² D	

*For predicted terms in the spectra of the As I isoelectronic sequence, see Vol. II, Introduction.

Br IV

(Ge I sequence; 32 electrons)

Z=35

Ground state 1s² 2s² 2p⁶ 3s² 3p⁶ 3d¹⁰ 4s² 4p² ³P₀

4p² ³P₀

cm⁻¹

I. P.

volts

This spectrum seriously needs further investigation. The authors have revised Deb's early work and published 47 classified lines in the interval 538 Å to 3041 Å. From combinations of 5s ¹³P with terms of the 4p² and 5p configurations, they derive absolute term values with the limit equal to 404890 cm⁻¹, giving an ionization potential of about 50 volts. No regular series have been observed and this estimate requires further confirmation.

Observed intersystem combinations connect the singlet and triplet systems of terms.

REFERENCE

A. S. Rao and S. G. Krishnamurty, Proc. Phys. Soc. London **46**, 531 (1934). (IP) (T) (CL)

Br IV

Br IV

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
4s ² 4p ²	4p ² ³ P	0	0		4s ² 4p(² P ^o)4d	4d ³ P ^o	2	176686	-1892 1625
		1	3247	1			180578		
		2	6237	0			178953		
4s ² 4p ²	4p ² ¹ D	2	18115		4s ² 4p(² P ^o)5s	5s ³ P ^o	0	180619	1281 4688
		a ^o	0, 1	139179			1	181900	
4s 4p ³	4p ³ ³ D ^o	1	153897	8053 10608	4s ² 4p(² P ^o)5s	5s ¹ P ^o	1	189034	2677 2567
		2	161950				0	216786	
		3	172558				1	219463	
4s 4p ³	4p ³ ³ P ^o	2	167568	-419 -348	4s ² 4p(² P ^o)5p	5p ³ P	0	216786	2677 2567
		1	167987				1	219463	
		0	168335				2	222030	
4s ² 4p(² P ^o)4d	4d ³ D ^o	1	168910	1115 2856	4s ² 4p(² P ^o)5p	5p ¹ D	2	227763	
		2	170025						
		3	172881						

Br IV OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 3d ¹⁰ +	Observed Terms		
4s ² 4p ²	$\left\{ \begin{array}{l} 4p^2 \ ^3P \\ \quad \quad 4p^2 \ ^1D \\ 4p^3 \ ^3P^\circ \quad 4p^3 \ ^3D^\circ \end{array} \right.$		
4s 4p ³			
	<i>ns</i> (<i>n</i> ≥ 5)	<i>np</i> (<i>n</i> ≥ 5)	<i>nd</i> (<i>n</i> ≥ 4)
4s ² 4p(² P ^o) <i>nz</i>	$\left\{ \begin{array}{l} 5s \ ^3P^\circ \\ 5s \ ^1P^\circ \end{array} \right.$	5p ³ P 5p ¹ D	4d ³ P ^o 4d ³ D ^o

*For predicted terms in the spectra of the Ge I isoelectronic sequence, see Vol. II, Introduction.

Br V

(Ga I sequence; 31 electrons)

Z=35

Ground state 1s² 2s² 2p⁶ 3s² 3p⁶ 3d¹⁰ 4s² 4p ²P₀^o4p ²P_{0½} cm⁻¹

I. P. volts

This spectrum needs investigation. From a study of isoelectronic sequence data, A. S. Rao and K. R. Rao have reported five terms and 12 classified lines in the range between 468 Å and 855 Å. Their term values are quoted in the table. The term 4p² ²D has been added by the writer with the value derived from its combination with 4p ²P^o, which they give.

REFERENCE

A. S. Rao and K. R. Rao, Proc. Phys. Soc. London **46**, 164 (1934). (T) (C L)

Br V

Config.	Desig.	<i>J</i>	Level	Interval
4s ² (¹ S)4p	4p ² P ^o	0½ 1½	0 6090	6090
4s 4p ²	4p ² ² D	1½ 2½	122941 123628	687
4s 4p ²	4p ² ² P	0½ 1½	158158 161011	2853
4s ² (¹ S)4d	4d ² D	1½ 2½	187970 188592	622
4s ² (¹ S)5s	5s ³ S	0½	213510	

May 1950.

Br VI

(Zn I sequence; 30 electrons)

 $Z=35$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 {}^1S_0$ $4s^2 {}^1S_0$ cm^{-1}

I. P. volts

Little is known about this spectrum. Rao and Rao have classified 14 lines between 499 Å and 939 Å. They list one dubious intersystem combination connecting the singlet and triplet terms. The singlet term $4p {}^1P_1^{\circ}$ needs further confirmation since it is based upon only one combination.

REFERENCE

A. S. Rao and K. R. Rao, Proc. Phys. Soc. London **46**, 166 (1934). (T) (C L)

Br VI

Config.	Desig.	J	Level	Interval
$4s^2$	$4s^2 {}^1S$	0	0	
$4s(2S)4p$	$4p {}^3P^{\circ}$	0	104316	2115 4905
		1	106431	
		2	111336	
$4s(2S)4p$	$4p {}^1P^{\circ}$	1	151274	
$4p^2$	$4p^2 {}^3P$	0	244136	3270 6486
		1	247406	
		2	253892	
$4s(2S)4d$	$4d {}^3D$	1	304648	319 484
		2	304967	
		3	305451	

March 1950.

Br VII

(Cu I sequence; 29 electrons)

 $Z=35$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 S_{0\frac{1}{2}}$ $4s^2 S_{0\frac{1}{2}}$ cm⁻¹

I. P. volts

This spectrum is incompletely known. In 1931 Mack reported that the lines of the principal doublet were very intense on his plates. Subsequently Rao and Rao classified five lines between 502 Å and 779 Å. The terms given in the table are from their paper.

REFERENCES

J. E. Mack, Phys. Rev. **38**, 194 (L) (1931).A. S. Rao and K. R. Rao, Proc. Phys. Soc. London **46**, 167 (1934). (T) (C L)

Br VII

Config.	Desig.	J	Level	Interval
4s	4s ² S	0½	0	
4p	4p ² P°	0½ 1½	128274 135854	7580
4d	4d ² D	1½ 2½	327205 328066	861

March 1950.

Br VIII

(Ni I sequence; 28 electrons)

 $Z=35$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} {}^1S_0$ $3d^{10} {}^1S_0$ 1554700 cm⁻¹

I. P. 193 volts

Kruger and Shoupp have observed two lines near 139 Å, and classified them as combinations of the ground term with terms from $3d^9 4p$.

From preliminary measurements Edlén has confirmed these results, extended the analysis, and derived provisional absolute term values from the Rydberg nf -series. All of the values in the table except those from $3d^9 4p$ are from his unpublished manuscript. His unit, 10^3 cm⁻¹, has here been changed to cm⁻¹.

REFERENCES

P. G. Kruger and W. E. Shoupp, Phys. Rev. **46**, 124 (1934). (I P) (T) (C L)

B. Edlén, letter (July 1950). (I P) (T)

Br VIII

Br VIII

Config.	Desig.	<i>J</i>	Level	Config.	Desig.	<i>J</i>	Level
$3d^{10}$	$3d^{10} 1S$	0	0	$3d^9(2D)4f$	$4f \ ^3D^\circ$	3 2 1	
$3d^9(2D)4p$	$4p \ ^3P^\circ$	2 1 0	701900	$3d^9(2D)5f$	$5f \ ^1P^\circ$	1	1115900
$3d^9(2D)4p$	$4p \ ^1P^\circ$	1	715300	$3d^9(2D)5f$	$5f \ ^3D^\circ$	3 2 1	1266900
$3d^9(2D)4p$	$4p \ ^3D^\circ$	3 2 1	719800	$3d^9(2D)6f$	$6f \ ^1P^\circ$	1	1276700
$3d^9(2D)4f$	$4f \ ^3P^\circ$	2 1 0	1098300	$3d^9(2D)6f$	$6f \ ^3D^\circ$	3 2 1	1355400
$3d^9(2D)4f$	$4f \ ^1P^\circ$	1	1104500	Br IX ($^2D_{3/2}$)	Limit		1554700

August 1950.

Br IX

(Co I sequence; 27 electrons)

 $Z=35$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^9 \ ^2D_{3/2}$ $3d^9 \ ^2D_{3/2} \text{ cm}^{-1}$

I. P. volts

Edlén has observed three lines due to the transition $3p^6 3d^9 \ ^2D - 3p^5 3d^{10} \ ^2P^\circ$. From preliminary unpublished measurements he has furnished the provisional term values quoted in the table. His unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCE

B. Edlén, letter (July 1950). (T)

Br IX

Config.	Desig.	<i>J</i>	Level	Interval
$3p^6 3d^9$	$3d^9 \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	0 8600	-8600
$3p^5 3d^{10}$	$3d^{10} \ ^2P^\circ$	$1\frac{1}{2}$ $0\frac{1}{2}$	912500 962300	-49800

July 1950.

KRYPTON

Kr I

36 electrons

Z=36

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 {}^1S_0$ $4p^6 {}^1S_0$ 112915.2 cm^{-1}

I. P. 13.996 volts

The Kr I spectrum has now been observed from 850A to 22000A. The observations in the infrared beyond the photographic limit are listed in the 1947 and 1949 references below, by Humphreys and Plyler, and Sittner and Peck, respectively.

The present list of energy levels has been compiled from an unpublished manuscript kindly furnished by Edlén, who has made a careful study of the spectrum and interpreted it with the aid of present atomic theory. Many of the levels are from the papers published at the National Bureau of Standards, which, in turn, contain revisions and extensions of the earlier work on Kr I. Three place entries are from interferometer measurements. Edlén's predicted value for one level of the $6f[4\frac{1}{2}]$ pair is entered in brackets in the table.

The observed g -values are from the 1940 and 1943 references by J. B. Green and others. Those for the $4f[1\frac{1}{2}]_1$, and $5f[1\frac{1}{2}]_2$ have been added by the writer from Table 1 in the 1940 paper. Pogány has also published a shorter list of g -values of Kr I.

White has discussed the effect of autoionization on the spectra of the inert gases, as an explanation of the lack of higher series members having the higher limit (${}^2P_{0\frac{1}{2}}^{\circ}$). Beutler has observed a number of these lines. The levels $ns'[0\frac{1}{2}]_1^{\circ}$, ($n=8$ to 12) and $nd'[1\frac{1}{2}]_1^{\circ}$, ($n=6$ to 14) have been determined by the writer from Beutler's observations of absorption series in the region 850 A to 900 A, and added to Edlén's list.

Edlén has determined the revised series limits quoted here. Boyce's value, 112915.7, has been used for the compilation of the energy levels. This value is 1.53 cm^{-1} higher than the earlier value adopted in the 1931 paper by Meggers, de Bruin, and Humphreys.

The Paschen notation is entered in column one of the table in the same form as for Ne I and Ar I. The letters X, Z, U, T, Y, and W, adopted when configurations involving f -electrons were found, are also entered in this column.

Edlén suggested that a pair-coupling notation be adopted for Ne I-like spectra to take into account the departure from LS -coupling. According to Shortley, LS -designations can be significantly assigned in only a few cases. The writer suggests them provisionally for the following groups of levels:

Paschen	Desig.	Paschen	Desig.	Paschen	Desig.	Paschen	Desig.	Paschen	Desig.
$(n-4)s_5$	$ns {}^3P_2^{\circ}$	$2p_{10}$	$5p {}^3S_1$	$2p_5$	$5p {}^3P_0$	$4d_6$	$5d {}^3P_0^{\circ}$	$4d_1''$	$5d {}^3F_2^{\circ}$
$(n-4)s_4$	$ns {}^3P_1^{\circ}$	$2p_9$	$5p {}^3D_3$	$2p_4$	$5p {}^1P_1$	$4d_5$	$5d {}^3P_1^{\circ}$	$4d_1'$	$5d {}^1F_3^{\circ}$
$(n-4)s_3$	$ns {}^3P_0^{\circ}$	$2p_8$	$5p {}^3D_2$	$2p_3$	$5p {}^3P_1$	$4, 5d_4'$	$5, 6d {}^3F_2^{\circ}$	$4s_1''''$	$5d {}^1D_2^{\circ}$
$(n-4)s_2$	$ns {}^1P_1^{\circ}$	$2p_7$	$5p {}^3D_1?$	$2p_2$	$5p {}^3P_2$	$4, 5d_4$	$5, 6d {}^3F_3^{\circ}$	$4s_1'''$	$5d {}^3D_3^{\circ}$
		$2p_6$	$5p {}^1D_2?$	$2p_1$	$5p {}^1S_0$	$4d_3$	$5d {}^3P_2^{\circ}$	$4s_1''$	$5d {}^3D_2^{\circ}$
						$4d_2$	$5d {}^1P_1^{\circ}$	$4s_1'$	$5d {}^3D_1^{\circ}$

Consequently, the jI -coupling notation in the general form suggested by Racah is here introduced, as for Ne I and Ar I. The present arrangement has been suggested by Shortley, who has made a detailed study of the theoretical arrangement of the "pairs", to be used as a guide in preparing the present table.

Twenty lines of Kr I between 4273 A and 6456 A were recommended by the International Astronomical Union (1935) as secondary standards of wavelength. All but two are regarded as accurate to eight figures. This Union (1935) listed, also, 25 lines between 3424 A and 3845 A measured to four decimal places. Further observations are required before these can be adopted as secondary standards.

Kr I—Continued

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Kr I

Kr I

Authors	Config.	Desig.	<i>J</i>	Level	Obs. <i>g</i>	Authors	Config.	Desig.	<i>J</i>	Level	Obs. <i>g</i>
$1p_0$	$4p^6$	$4p^6 \ ^1S$	0	0. 0		$3d_1''$ $3d_1'$	$4p^5(^2P_{1/2})4d$	$4d \ [2\frac{1}{2}]^\circ$	2 3	98868. 22 99080. 19	
$1s_6$ $1s_4$	$4p^5(^2P_{1/2})5s$	$5s \ [1\frac{1}{2}]^\circ$	2 1	79972. 535 80917. 561	1. 502 1. 242	$3s_1''''$ $3s_1''''$	$4p^5(^2P_{3/2})4d$	$4d' \ [2\frac{1}{2}]^\circ$	2 3	103443. 46 103702. 239	
$1s_3$ $1s_2$	$4p^5(^2P_{3/2})5s$	$5s' \ [0\frac{1}{2}]^\circ$	0 1	85192. 414 85847. 501	1. 259	$3s_1''$ $3s_1'$	"	$4d' \ [1\frac{1}{2}]^\circ$	2 1	103267. 12 104888. 114	1. 018
$2p_{10}$	$4p^5(^2P_{1/2})5p$	$5p \ [0\frac{1}{2}]$	1	91169. 313	1. 898	$2s_5$ $2s_4$	$4p^5(^2P_{1/2})6s$	$6s \ [1\frac{1}{2}]^\circ$	2 1	99627. 67 99894. 83	
$2p_9$ $2p_8$	"	$5p \ [2\frac{1}{2}]$	3 2	92295. 199 92308. 177	1. 336 1. 099	$2s_3$ $2s_2$	$4p^5(^2P_{3/2})6s$	$6s' \ [0\frac{1}{2}]^\circ$	0 1	105092. 14 105147. 13	
$2p_7$ $2p_6$	"	$5p \ [1\frac{1}{2}]$	1 2	92965. 194 93124. 140	1. 004 1. 388	$3p_{10}$	$4p^5(^2P_{1/2})6p$	$6p \ [0\frac{1}{2}]$	1	102888. 002	1. 834
$2p_5$	"	$5p \ [0\frac{1}{2}]$	0	94093. 662		$3p_9$ $3p_8$	"	$6p \ [2\frac{1}{2}]$	3 2	103116. 449 103121. 953	1. 333 1. 107
$2p_4$ $2p_2$	$4p^5(^2P_{3/2})5p$	$5p' \ [1\frac{1}{2}]$	1 2	97596. 718 97945. 970	0. 647 1. 181	$3p_7$ $3p_6$	"	$6p \ [1\frac{1}{2}]$	1 2	103314. 284 103363. 425	1. 034 1. 403
$2p_3$ $2p_1$	"	$5p' \ [0\frac{1}{2}]$	1 0	97919. 951 98855. 871	1. 452	$3p_5$	"	$6p \ [0\frac{1}{2}]$	0	103762. 446	
$3d_6$ $3d_5$	$4p^5(^2P_{1/2})4d$	$4d \ [0\frac{1}{2}]^\circ$	0 1	96772. 31 97086. 00		$3p_4$ $3p_2$	$4p^5(^2P_{3/2})6p$	$6p' \ [1\frac{1}{2}]$	1 2	108439. 074 108568. 583	0. 648 1. 158
$3d_4'$ $3d_4$	"	$4d \ [3\frac{1}{2}]^\circ$	4 3	97798. 15 98227. 08		$3p_3$ $3p_1$	"	$6p' \ [0\frac{1}{2}]$	1 0	108514. 999 108822. 382	1. 401
$3d_3$ $3d_2$	"	$4d \ [1\frac{1}{2}]^\circ$	2 1	97688. 58 99647. 02		$4d_6$ $4d_5$	$4p^5(^2P_{1/2})5d$	$5d \ [0\frac{1}{2}]^\circ$	0 1	104074. 276 103802. 598	1. 098

Kr I—Continued

Kr I—Continued

Authors	Config.	Desig.	<i>J</i>	Level	Obs. <i>g</i>	Authors	Config.	Desig.	<i>J</i>	Level	Obs. <i>g</i>
4 <i>d</i> ' ₄ 4 <i>d</i> ₄	4 <i>p</i> ⁵ (² P _{1/2})5 <i>d</i>	5 <i>d</i> [3½]°	4 3	104631. 37 104917. 282	1. 050	5U	4 <i>p</i> ⁵ (² P _{1/2})5 <i>f</i>	5 <i>f</i> [4½]	5 4	108487. 73 108487. 88	
4 <i>d</i> ₃ 4 <i>d</i> ₂	"	5 <i>d</i> [1½]°	2 1	105008. 054 105649. 25	1. 295 0. 935	5T 5Y	"	5 <i>f</i> [2½]	3 2	108504. 045 108504. 62	
4 <i>d</i> ' ₁ 4 <i>d</i> ' ₁	"	5 <i>d</i> [2½]°	2 3	105164. 302 105209. 278	1. 006 1. 243	5W	"	5 <i>f</i> [3½]	3, 4	108517. 77	
4 <i>s</i> ' ₁ 4 <i>s</i> ' ₁	4 <i>p</i> ⁵ (² P _{0/2})5 <i>d</i>	5 <i>d</i> ' [2½]°	2 3	110122. 88 110238. 22	0. 899 1. 140	5 <i>p</i> ₁₀	4 <i>p</i> ⁵ (² P _{1/2})8 <i>p</i>	8 <i>p</i> [0½]	1	109083. 580	1. 795
4 <i>s</i> ' ₁ 4 <i>s</i> ' ₁	"	5 <i>d</i> ' [1½]°	2 1	110104. 05 110734. 07	1. 169	5 <i>p</i> ₉ 5 <i>p</i> ₈	"	8 <i>p</i> [2½]	3 2	109104. 106 109106. 58	
3 <i>s</i> ₅ 3 <i>s</i> ₄	4 <i>p</i> ⁵ (² P _{1/2})7 <i>s</i>	7 <i>s</i> [1½]°	2 1	105648. 254 105771. 50	1. 496 1. 097	5 <i>p</i> ₇ 5 <i>p</i> ₆	"	8 <i>p</i> [1½]	1 2	109150. 501 109161. 762	1. 014 1. 411
3 <i>s</i> ₂	4 <i>p</i> ⁵ (² P _{0/2})7 <i>s</i>	7 <i>s</i> ' [0½]°	0 1	111003. 79	1. 208	5 <i>p</i> ₅	"	8 <i>p</i> [0½]	0	109296. 997	
4X 4Z	4 <i>p</i> ⁵ (² P _{1/2})4 <i>f</i>	4 <i>f</i> [1½]	1 2	105965. 258 105966. 379	0. 52	6 <i>d</i> ₅ 6 <i>d</i> ₄	4 <i>p</i> ⁵ (² P _{1/2})7 <i>d</i>	7 <i>d</i> [0½]°	0 1	109331. 75 109343. 75	1. 355
4U	"	4 <i>f</i> [4½]	5	105989. 60		6 <i>d</i> ' ₄ 6 <i>d</i> ' ₄	"	7 <i>d</i> [3½]°	4 3	109434. 716 109472. 27	1. 228 1. 094
4T 4Y	"	4 <i>f</i> [2½]	3 2	106021. 657 106022. 425		6 <i>d</i> ₃ 6 <i>d</i> ₂	"	7 <i>d</i> [1½]°	2 1	109376. 14 109689. 56	1. 315 0. 797
4W	"	4 <i>f</i> [3½]	3, 4	106048. 19		6 <i>d</i> ' ₁ 6 <i>d</i> ' ₁	"	7 <i>d</i> [2½]°	2 3	109528. 36 109579. 82	0. 954 1. 231
	4 <i>p</i> ⁵ (² P _{0/2})4 <i>f</i>	4 <i>f</i> ' [3½]	4 3	111378. 7 111379. 21		6 <i>s</i> ' ₁	4 <i>p</i> ⁵ (² P _{0/2})7 <i>d</i>	7 <i>d</i> ' [1½]°	2 1	115010	
	"	4 <i>f</i> ' [2½]	3 2	111381. 10 111381. 98		5 <i>s</i> ₅ 5 <i>s</i> ₄	4 <i>p</i> ⁵ (² P _{1/2})9 <i>s</i>	9 <i>s</i> [1½]°	2 1	109752. 74 109780. 11	1. 495 1. 174
4 <i>p</i> ₁₀	4 <i>p</i> ⁵ (² P _{1/2})7 <i>p</i>	7 <i>p</i> [0½]	1	107006. 184	1. 795	5 <i>s</i> ₂	4 <i>p</i> ⁵ (² P _{0/2})9 <i>s</i>	9 <i>s</i> ' [0½]°	0 1	115128	
4 <i>p</i> ₉ 4 <i>p</i> ₈	"	7 <i>p</i> [2½]	3 2	107141. 983 107141. 611		6X 6Z	4 <i>p</i> ⁵ (² P _{1/2})6 <i>f</i>	6 <i>f</i> [1½]	1 2	109836. 94 109837. 56	
4 <i>p</i> ₇ 4 <i>p</i> ₆	"	7 <i>p</i> [1½]	1 2	107222. 145 107247. 499	1. 041 1. 403	6U	"	6 <i>f</i> [4½]	5 4	[109843. 8] 109843. 91	
4 <i>p</i> ₅	"	7 <i>p</i> [0½]	0	107411. 191		6T 6Y	"	6 <i>f</i> [2½]	3 2	109853. 01 109853. 55	
5 <i>d</i> ₅ 5 <i>d</i> ₅	4 <i>p</i> ⁵ (² P _{1/2})6 <i>d</i>	6 <i>d</i> [0½]°	0 1	107604. 43 107676. 953	1. 348	6W	"	6 <i>f</i> [3½]	3, 4	109861. 12	
5 <i>d</i> ' ₄ 5 <i>d</i> ' ₄	"	6 <i>d</i> [3½]°	4 3	107779. 701 107877. 712	1. 231 1. 073	6 <i>p</i> ₁₀	4 <i>p</i> ⁵ (² P _{1/2})9 <i>p</i>	9 <i>p</i> [0½]	1	110180. 87	
5 <i>d</i> ₃ 5 <i>d</i> ₂	"	6 <i>d</i> [1½]°	2 1	107797. 681 108259. 59	1. 318 0. 823	6 <i>p</i> ₉ 6 <i>p</i> ₈	"	9 <i>p</i> [2½]	3 2	110210. 35 110210. 64	
5 <i>d</i> ' ₁ 5 <i>d</i> ' ₁	"	6 <i>d</i> [2½]°	2 3	107993. 585 108047. 121	0. 965 1. 254	6 <i>p</i> ₇ 6 <i>p</i> ₆	"	9 <i>p</i> [1½]	1 2	110235. 64 110243. 62	
5 <i>s</i> ' ₁	4 <i>p</i> ⁵ (² P _{0/2})6 <i>d</i>	6 <i>d</i> ' [1½]°	2 1	113500		6 <i>p</i> ₅	"	9 <i>p</i> [0½]	0	110308. 93	
4 <i>s</i> ₅ 4 <i>s</i> ₄	4 <i>p</i> ⁵ (² P _{1/2})8 <i>s</i>	8 <i>s</i> [1½]°	2 1	108325. 81 108373. 87	1. 506 1. 171	7 <i>d</i> ₅ 7 <i>d</i> ₅	4 <i>p</i> ⁵ (² P _{1/2})8 <i>d</i>	8 <i>d</i> [0½]°	0 1	110336. 47 110291. 15	1. 294
4 <i>s</i> ₃ 4 <i>s</i> ₂	4 <i>p</i> ⁵ (² P _{0/2})8 <i>s</i>	8 <i>s</i> ' [0½]°	0 1	113711		7 <i>d</i> ' ₄ 7 <i>d</i> ' ₄	"	8 <i>d</i> [3½]°	4 3	110404. 42 110471. 70	1. 236 1. 037
5X 5Z	4 <i>p</i> ⁵ (² P _{1/2})5 <i>f</i>	5 <i>f</i> [1½]	1 2	108481. 562 108471. 937	0. 61 1. 09	7 <i>d</i> ₃ 7 <i>d</i> ₂	"	8 <i>d</i> [1½]°	2 1	110513. 57 110514. 84	

Kr I—Continued

Kr I—Continued

Authors	Config.	Desig.	<i>J</i>	Level	Obs. <i>g</i>	Authors	Config.	Desig.	<i>J</i>	Level	Obs. <i>g</i>
$7d_1''$ $7d_1'$	$4p^5(^2P_{1/2})8d$	$8d [2\frac{1}{2}]^\circ$	2 3	110497. 47 110508. 89	1. 005 1. 227	$9d_3$	$4p^5(^2P_{1/2})10d$	$10d [1\frac{1}{2}]^\circ$	2 1	111446. 18	
$7s_1'$	$4p^5(^2P_{3/2}^\circ)8d$	$8d' [1\frac{1}{2}]^\circ$	2 1	115910		$9d_1''$ $9d_1'$	"	$10d [2\frac{1}{2}]^\circ$	2 3	111468. 14 111474. 87	
$6s_5$ $6s_4$	$4p^5(^2P_{1/2})10s$	$10s [1\frac{1}{2}]^\circ$	2 1	110609. 13 110619. 78	1. 161	$9s_1'$	$4p^5(^2P_{3/2}^\circ)10d$	$10d' [1\frac{1}{2}]^\circ$	2 1	116870	
$6s_2$	$4p^5(^2P_{3/2}^\circ)10s$	$10s' [0\frac{1}{2}]^\circ$	0 1	115961		$8s_5$ $8s_4$	$4p^5(^2P_{1/2})12s$	$12s [1\frac{1}{2}]^\circ$	2 1	111528. 62 111537. 42	
$7X$ $7Z$	$4p^5(^2P_{1/2})7f$	$7f [1\frac{1}{2}]$	1 2	110656. 24 110656. 80		$8s_2$	$4p^5(^2P_{3/2}^\circ)12s$	$12s' [0\frac{1}{2}]^\circ$	0 1	116904	
$7U$	"	$7f [4\frac{1}{2}]$	4, 5	110660. 69		$9X$	$4p^5(^2P_{1/2})9f$	$9f [1\frac{1}{2}]$	1 2	111551. 3	
$7T$ $7Y$	"	$7f [2\frac{1}{2}]$	3 2	110666. 24 110666. 54		$9U$	"	$9f [4\frac{1}{2}]$	5 4	111553. 16	
$7W$	"	$7f [3\frac{1}{2}]$	3, 4	110671. 46		$9T$	"	$9f [2\frac{1}{2}]$	3	111556. 56	
$7p_{10}$	$4p^5(^2P_{1/2})10p$	$10p [0\frac{1}{2}]$	1	110873. 22		$10d_6$ $10d_5$	$4p^5(^2P_{1/2})11d$	$11d [0\frac{1}{2}]^\circ$	0 1	111709. 11 111718. 95	
$7p_6$	"	$10p [1\frac{1}{2}]$	2	110916. 86		$10d_4'$ $10d_4$	"	$11d [3\frac{1}{2}]^\circ$	4 3	111726. 00 111737. 65	
$7p_5$	"	$10p [0\frac{1}{2}]$	0	110957. 03		$10d_3$	"	$11d [1\frac{1}{2}]^\circ$	2 1	111731. 93	
$8d_6$	$4p^5(^2P_{1/2})9d$	$9d [0\frac{1}{2}]^\circ$	0	110934. 16		$10d_1'$	"	$11d [2\frac{1}{2}]^\circ$	2 3	111755. 14	
$8d_4'$ $8d_4$	"	$9d [3\frac{1}{2}]^\circ$	4 3	111019. 65 111047. 90		$10s_1'$	$4p^5(^2P_{3/2}^\circ)11d$	$11d' [1\frac{1}{2}]^\circ$	2 1	117145	
$8d_3$ $8d_2$	"	$9d [1\frac{1}{2}]^\circ$	2 1	111047. 86 111143		$10X$	$4p^5(^2P_{1/2})10f$	$10f [1\frac{1}{2}]$	1 2	111810. 4	
$8d_1''$ $8d_1'$	"	$9d [2\frac{1}{2}]^\circ$	2 3	111072. 24 111079. 85		$10U$	"	$10f [4\frac{1}{2}]$	5 4	111812. 3	
$8s_1'$	$4p^5(^2P_{3/2}^\circ)9d$	$9d' [1\frac{1}{2}]^\circ$	2 1	116480		$10T$	"	$10f [2\frac{1}{2}]$	3 2	111814. 2	
$7s_5$ $7s_4$	$4p^5(^2P_{1/2})11s$	$11s [1\frac{1}{2}]^\circ$	2 1	111155. 19 111171. 62		$11d_1'$ $11d_4$	$4p^5(^2P_{1/2})12d$	$12d [3\frac{1}{2}]^\circ$	4 3	111939. 50 111947. 70	
$7s_2$	$4p^5(^2P_{3/2}^\circ)11s$	$11s' [0\frac{1}{2}]^\circ$	0 1	116527		$11s_1'$	$4p^5(^2P_{3/2}^\circ)12d$	$12d' [1\frac{1}{2}]^\circ$	2 1	117345	
$8X$	$4p^5(^2P_{1/2})8f$	$8f [1\frac{1}{2}]$	1 2	111187. 33		$12d_4'$	$4p^5(^2P_{1/2})13d$	$13d [3\frac{1}{2}]^\circ$	4 3	112100. 54	
$8U$	"	$8f [4\frac{1}{2}]$	4, 5	111190. 29		$12s_1'$	$4p^5(^2P_{3/2}^\circ)13d$	$13d' [1\frac{1}{2}]^\circ$	2 1	117497	
$8T$ $8Y$	"	$8f [2\frac{1}{2}]$	3 2	111193. 45 111193. 78			Kr II ($^2P_{1/2}^\circ$)	<i>Limit</i>	-----	112915. 2	
$8p_5$	$4p^5(^2P_{1/2})11p$	$11p [0\frac{1}{2}]$	0	111391. 1		$13s_1'$	$4p^5(^2P_{3/2}^\circ)14d$	$14d' [1\frac{1}{2}]^\circ$	2 1	117625	
$9d_6$ $9d_5$	$4p^5(^2P_{1/2})10d$	$10d [0\frac{1}{2}]^\circ$	0 1	111413. 23 111429. 36			Kr II ($^2P_{3/2}^\circ$)	<i>Limit</i>	-----	118284. 7	
$9d_4'$ $9d_4$	"	$10d [3\frac{1}{2}]^\circ$	4 3	111433. 88 111451. 22							

Kr I OBSERVED LEVELS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 +$	Observed Terms			
$4p^6$	$4p^6 \ ^1S$			
	$ns (n \geq 5)$	$np (n \geq 5)$		$nd (n \geq 4)$
$4p^5(^2P^\circ)nx$	{ 5-12s $^3P^\circ$ 5-12s $^1P^\circ$	$5p \ ^3S$ $5p \ ^1S$	$5p \ ^3P$ $5p \ ^1P$	$5p \ ^3D$ $5p \ ^1D$
		$5d \ ^3P^\circ$ $5d \ ^1P^\circ$	$5d \ ^3D^\circ$ $5d \ ^1D^\circ$	$5, 6d \ ^3F^\circ$ $5, 6d \ ^1F^\circ$
<i>jl</i> -Coupling Notation				
	Observed Pairs			
	$ns (n \geq 5)$	$np (n \geq 5)$	$nd (n \geq 4)$	$nf (n \geq 4)$
$4p^6(^2P_{1/2})nx$	5-12s $[1\frac{1}{2}]^\circ$	5-11p $[0\frac{1}{2}]$ 5-9p $[2\frac{1}{2}]$ 5-10p $[1\frac{1}{2}]$	4-11d $[0\frac{1}{2}]^\circ$ 4-13d $[3\frac{1}{2}]^\circ$ 4-11d $[1\frac{1}{2}]^\circ$ 4-11d $[2\frac{1}{2}]^\circ$	4-10f $[1\frac{1}{2}]$ 4-10f $[4\frac{1}{2}]$ 4-10f $[2\frac{1}{2}]$ 4-7f $[3\frac{1}{2}]$
$4p^5(^2P_{0/2}^\circ)nx'$	5-12s' $[0\frac{1}{2}]^\circ$	5-6p' $[1\frac{1}{2}]$ 5-6p' $[0\frac{1}{2}]$	4-5d' $[2\frac{1}{2}]^\circ$ 4-14d' $[1\frac{1}{2}]^\circ$	4f' $[3\frac{1}{2}]$ 4f' $[2\frac{1}{2}]$

*For predicted levels in the spectra of the Kr I isoelectronic sequence, see Introduction.

Kr II

(Br I sequence; 35 electrons)

$Z=36$

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^5 \ ^2P_{1/2}^\circ$

$4p^5 \ ^2P_{1/2}^\circ$ **198182.00** cm^{-1}

I. P. 24.56 volts

A detailed analysis of this spectrum has been published by de Bruin, Humphreys, and Meggers, who have observed 1050 lines between 2080 Å and 10659 Å. Of these, 750, or 71 percent, are classified. In the extreme ultraviolet, Boyce has extended the work of Abbink and Dorgelo to include 82 classified lines between 559 Å and 964 Å, of which 31 are newly classified. No new terms are added from this ultraviolet work.

The doublet and quartet systems of terms are connected by observed intersystem combinations.

The limit is derived by applying the Rydberg series formula to the first two members of the $ns \ ^4P_{2\frac{1}{2}}$ series ($n=5,6$).

Most of the observed g -values are from the 1931 paper listed below. A few additional values, based on unpublished observations furnished by Bakker, Zeeman, and de Bruin are also included in the 1933 paper containing the analysis of Kr II.

REFERENCES

- J. H. Abbink and H. B. Dorgelo, *Zeit. Phys.* **47**, 221 (1928).
 C. J. Bakker and T. L. de Bruin, *Zeit. Phys.* **69**, 36 (1931). (Z E)
 T. L. de Bruin, C. J. Humphreys, and W. F. Meggers, *Bur. Std. J. Research* **11**, 409, RP 599 (1933). (I P)
 (T) (C L) (Z E)
 J. C. Boyce, *Phys. Rev.* **47**, 718 (1935). (C L)

Kr II

Kr II

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
$4s^2 4p^5$	$4p^5 \ ^2P^\circ$	$1\frac{1}{2}$ $0\frac{1}{2}$	0. 00 5371. 00	-5371. 00		$4s^2 4p^4(^1D)4d$	$4d' \ ^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	149860. 27? 150814. 24?	-953. 97	
$4s 4p^6$	$4p^6 \ ^2S$	$0\frac{1}{2}$	109002. 06			$4s^2 4p^4(^1D)5p$	$5p' \ ^2P^\circ$	$1\frac{1}{2}$ $0\frac{1}{2}$	150203. 48 152240. 97	-2037. 49	1. 33 0. 70
$4s^2 4p^4(^3P)5s$	$5s \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	112830. 00 115093. 71 117604. 73	-2263. 71 -2511. 02	1. 60 1. 54 2. 64	$4s^2 4p^4(^1D)4d$	$4d' \ ^2P$	$1\frac{1}{2}$ $0\frac{1}{2}$	151828. 00 152186. 70	-358. 70	
$4s^2 4p^4(^3P)5s$	$5s \ ^2P$	$1\frac{1}{2}$ $0\frac{1}{2}$	118476. 07 121003. 87	-2527. 80	1. 52 0. 70	$4s^2 4p^4(^1D)5p$	$5p' \ ^2D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	152191. 86 152316. 20	124. 34	0. 80 1. 20
$4s^2 4p^4(^3P)4d$	$4d \ ^4D$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	120211. 59 120423. 65 121002. 10 121781. 28	-217. 06 -573. 45 -779. 18	0. 00	$4s^2 4p^4(^3P)6s$	$6s \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	157079. 07 157885. 41 161877. 37	-806. 34 -3991. 96	1. 60 1. 39 2. 34
$4s^2 4p^4(^3P)4d$	$4d \ ^4F$	$4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$	126002. 57 127931. 20 129698. 98 129743. 56	-1928. 63 -1767. 78 -44. 58		$4s^2 4p^4(^3P)5d$	$5d \ ^4P$	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	160796. 68 161409. 29 161013. 56	612. 61 -395. 73	2. 07 2. 47
$4s^2 4p^4(^1D)5s$	$5s' \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	127599. 19 127863. 23	264. 04	0. 80 1. 20	$4s^2 4p^4(^3P)5d$	$5d \ ^4D$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	161285. 31 161451. 83 161801. 92 163033. 48	-166. 52 -350. 09 -1231. 56	1. 40 1. 37 0. 88
$4s^2 4p^4(^3P)4d$	$4d \ ^4P$	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	130514. 59 130895. 28 131377. 24	380. 69 481. 96		$4s^2 4p^4(^3P)6s$	$6s \ ^2P$	$1\frac{1}{2}$ $0\frac{1}{2}$	162059. 03 162566. 16	-507. 13	1. 33 0. 92
$4s^2 4p^4(^3P)4d$	$4d \ ^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	131633. 86 137099. 98	-5466. 12		$4s^2 4p^4(^3P)5d$	$5d \ ^4F$	$4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$	162208. 84 162531. 91 165077. 35 167047. 13	-323. 07 -2545. 44 -1969. 78	1. 33 1. 17 1. 12 0. 52
$4s^2 4p^4(^3P)4d$	$4d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	132967. 28 134568. 82	1601. 54		$4s^2 4p^4(^1S)6s$	$6s'' \ ^2S$	$0\frac{1}{2}$	164439. 16?		1. 94
	1	$2\frac{1}{2}$	132972. 50			$4s^2 4p^4(^3P)5d$	$5d \ ^2P$	$1\frac{1}{2}$ $0\frac{1}{2}$	165141. 91 166953. 31	-1811. 40	1. 40 0. 51
$4s^2 4p^4(^3P)5p$	$5p \ ^4P^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	133925. 65 134288. 44 135783. 03	-362. 79 -1494. 59	1. 58 1. 67 1. 98	$4s^2 4p^4(^3P)5d$	$5d \ ^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	165397. 72 167518. 91	-2121. 19	1. 24 1. 04
$4s^2 4p^4(^3P)4d$	$4d \ ^2P$	$0\frac{1}{2}$ $1\frac{1}{2}$	134623. 17				1°	$1\frac{1}{2}$	166155. 50?		
$4s^2 4p^4(^3P)5p$	$5p \ ^4D^\circ$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	135783. 18 136071. 00 138381. 35 140163. 25	-287. 82 -2310. 35 -1781. 90	1. 43 1. 23 1. 26 0. 00	$4s^2 4p^4(^3P)5d$	$5d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	167913. 11 169704. 85	1791. 74	1. 18 1. 15
	2	$1\frac{1}{2}$	138495. 02			$4s^2 4p^4(^3P)5f$	$5f \ ^4F^\circ$	$4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$	168085. 49 168118. 00 168183. 20 168385. 12	-32. 51 -65. 20 -201. 92	
$4s^2 4p^4(^3P)5p$	$5p \ ^2P^\circ$	$0\frac{1}{2}$ $1\frac{1}{2}$	139103. 36 140137. 15	1033. 79	1. 78 1. 26	$4s^2 4p^4(^3P)5f$	$5f \ ^2F^\circ$	$3\frac{1}{2}$ $2\frac{1}{2}$	168260. 43 168462. 53	-202. 10	
$4s^2 4p^4(^3P)5p$	$5p \ ^2D^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$	140118. 99 141995. 68	-1876. 69	1. 34 1. 33	$4s^2 4p^4(^1S)5p$	$5p'' \ ^2P^\circ$	$0\frac{1}{2}$ $1\frac{1}{2}$	168262. 90 168939. 35	676. 45	1. 24 0. 90
$4s^2 4p^4(^3P)5p$	$5p \ ^4S^\circ$	$1\frac{1}{2}$	141722. 72		1. 54		2°	$3\frac{1}{2}$	168490. 67		
$4s^2 4p^4(^3P)5p$	$5p \ ^2S^\circ$	$0\frac{1}{2}$	142363. 55		1. 50	$4s^2 4p^4(^3P)5f$	$5f \ ^2D^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$	168630. 36		1. 02
$4s^2 4p^4(^1S)5s$	$5s'' \ ^2S$	$0\frac{1}{2}$	145813. 61		2. 00	$4s^2 4p^4(^3P)5f$	$5f \ ^4D^\circ$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	168718. 90		
$4s^2 4p^4(^1D)4d$	$4d' \ ^2G$	$4\frac{1}{2}$ $3\frac{1}{2}$	146892. 32?			$4s^2 4p^4(^1D)6s$	$6s' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	170571. 12		1. 00
$4s^2 4p^4(^1D)5p$	$5p' \ ^2F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	149173. 42 149704. 55	531. 13	0. 86 1. 14						
$4s^2 4p^4(^1D)4d$	$4d' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	149515. 90 150179. 87	-663. 97							

Kr II—Continued

Kr II—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>		
$4s^2 4p^4(1S)4d$	3°	1½	170887. 53			$4s^2 4p^4(1D)5d$	5d' ²P	1½	177683. 83	-1103. 80	1. 18		
	4°	2½	171295. 70				0½	178787. 63					
	5°	2½	171386. 32			$4s^2 4p^4(1D)5d$	5d' ²F	2½	177710. 28	198. 72	0. 89		
	6°	3½	171560. 00				3½	177909. 00					
$4s^2 4p^4(1S)4d$	4d'' ²D	2½	171970. 61	-81. 28	1. 20	$4s^2 4p^4(³P)7s$	7s ²P	1½	177956. 88	-550. 57			
		1½	172051. 89				0½	178507. 45					
$4s^2 4p^4(1D)5f$	5f' ²D°	1½	172714. 72	58. 65		$4s^2 4p^4(³P)5g$	5g ⁴G	5½	178842. 39	31. 09			
		2½	172773. 37				4½	178811. 30					
$4s^2 4p^4(1D)5f$	5f' ²F°	3½	173131. 03	-25. 39			8°	1½	182464. 85	-4. 21	-45. 37		
		2½	173156. 42				9°	0½?	182479. 27				
$4s^2 4p^4(³P)7s$	7s ⁴P	2½	173309. 68	-330. 34			10°	1½	182811. 90	11°	2½	182841. 19	
		1½	173640. 02				-4414. 77	11°	2½				182841. 19
		0½	178054. 79				-815. 12	12°	1½				182965. 59
$4s^2 4p^4(1D)5f$	5f' ²P°	1½	173687. 88	-815. 12			13°	2½	183308. 55				
		0½	174503. 00										
$4s^2 4p^4(1D)5d$	5d' ²G	3½	175891. 74	701. 23	0. 89		-----	-----	-----	198182. 00			
		4½	176592. 97				1. 11						
$4s^2 4p^4(1D)5d$	5d' ²D	1½	176110. 96	2209. 67	1. 20		Kr III (³P₂)	Limit	-----	-----	-----		
		2½	178320. 63										

January 1951.

Kr II OBSERVED TERMS*

Config. 1s² 2s² 2p⁶ 3s² 3p⁶ 3d¹⁰ +	Observed Terms		
$4s^2 4p^5$	4p⁵ ²P°		
4s 4p⁶	4p⁶ ²S		
	ns (n ≥ 5)	np (n ≥ 5)	
$4s^2 4p^4(³P)nx$	{ 5,6,7s ⁴P 5,6,7s ²P	5p ⁴S° 5p ⁴P° 5p ⁴D° 5p ²S° 5p ²P° 5p ²D°	
$4s^2 4p^4(1D)nx'$	5, 6s' ²D	5p' ²P° 5p' ²D°	5p' ²F°
$4s^2 4p^4(1S)nx''$	5, 6s'' ²S	5p'' ²P°	
	nd (n ≥ 4)	nf (n ≥ 4)	ng (n ≥ 5)
$4s^2 4p^4(³P)nx$	{ 4, 5d ⁴P 4,5d ⁴D 4,5d ⁴F 4, 5d ²P 4,5d ²D 4,5d ²F	5f ⁴D° 5f ⁴F° 5f ²D° 5f ²F°	5g ⁴G
$4s^2 4p^4(1D)nx'$	4, 5d' ²P 4,5d' ²D 4,5d' ²F 4,5d' ²G	5f' ²P° 5f' ²D° 5f' ²F°	
$4s^2 4p^4(1S)nx''$	4d'' ²D		

*For predicted terms in the spectra of the Br I isoelectronic sequence, see Vol. II, Introduction.

Kr III

(Se I sequence; 34 electrons)

Z=36

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^4 \ ^3P_2$ $4p^4 \ ^3P_2$ 298020 cm^{-1}

I. P. 36.9 volts

The analysis is from Humphreys, who has discussed the earlier work on this spectrum. He lists 369 classified lines in the interval between 2116 Å and 7353 Å. In addition, Boyce has classified 138 lines between 516 Å and 1923 Å. Observed intersystem combinations connect the terms of all multiplicities.

The limit is based on the $nd \ ^5D_3$, $nd \ ^3D_3$ ($n=4,5$), and $ns \ ^3S_1$ ($n=5,6$) series.

The Zeeman effect for two lines, 3507.42 Å and 3564.23 Å has been observed by Bakker and de Bruin. The resulting g -values are:

Desig.	Obs. g
$5s \ ^3S_1$	1.99
$5p \ ^3P_2$	1.50
$5p \ ^3P_1$	1.55

REFERENCES

C. J. Humphreys, Phys. Rev. **47**, 712 (1935). (I P) (T) (C L)J. C. Boyce, Phys. Rev. **47**, 718 (1935). (C L)T. L. de Bruin, *Zeeman Verhandelingen*, p. 416 (Martinus Nijhoff, The Hague, 1935). (T) (C L) (Z E)

Kr III

Kr III

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval				
$4s^2 4p^4$	$4p^4 \ ^3P$	2	0	-4548 -765	$4s^2 4p^3(2D^\circ)4d$	$4d' \ ^1D^\circ$	2	165464.45					
		1	4548					$4s^2 4p^3(2D^\circ)4d$		$4d' \ ^3D^\circ$	1	170203.38	
		0	5313								2	172466.51	
$4s^2 4p^4$	$4p^4 \ ^1D$	2	14644		$4s^2 4p^3(2D^\circ)4d$	$4d' \ ^3D^\circ$	3	174452.05	2263.13 1985.54				
								$4s^2 4p^3(2D^\circ)5s$	$5s' \ ^1D^\circ$	2	170900.02		
											$4s^2 4p^3(2D^\circ)4d$		$4d' \ ^1P^\circ?$
$4s^2 4p^3(2D^\circ)4d$	$4d' \ ^3S^\circ?$	1	172984.33										
			$4s^2 4p^3(4S^\circ)4d$	$4d \ ^5D^\circ$	0	138447.68	24.44	$4s^2 4p^3(2P^\circ)4d$	$4d'' \ ^3F^\circ$	4		175043.98	
						1					138472.12	9.80	3
2	138481.92	11.86				2					175212.28		
3	138493.78	156.49											
$4s 4p^5$	$4p^5 \ ^3P^\circ$	2	115932	-3449 -2163	$4s^2 4p^3(4S^\circ)5p$	$5p \ ^5P$	1	175545.05	234.69 741.44				
			1					119381		2	175779.74		
			0					121544		3	176521.18		
$4s^2 4p^3(4S^\circ)5s$	$5s \ ^5S^\circ$	2	145720.00	930.74 -1267.29	$4s^2 4p^3(2P^\circ)4d$	$4d'' \ ^3D^\circ$	1	182266.00	-5474.17 8101.02				
			$4s^3 4p^3(4S^\circ)4d$					$4d \ ^3D^\circ$		3	148736.44	2	176791.83
											2		147805.70
1	149072.99												
$4s^2 4p^3(4S^\circ)5s$	$5s \ ^3S^\circ$	1	151581.26	1136.87 1381.91	$4s^2 4p^3(2P^\circ)5s$	$5s'' \ ^3P^\circ$	0	178244.46	15.66 1987.98				
			$4s^2 4p^3(2D^\circ)4d$					$4d' \ ^3F^\circ$		2	153564.18	1	178260.12
											3		154701.05
4	156082.96												
$4s^2 4p^3(2D^\circ)4d$	$4d' \ ^3G^\circ$	3	159997.52	418.36 693.67	$4s^2 4p^3(4S^\circ)5p$	$5p \ ^3P$	0	179629.83	454.29				
			$4s^2 4p^3(2D^\circ)4d$					$4d' \ ^1G^\circ$		4	160415.88	1	179629.83
											4		160415.88
5	161109.55												
$4s^2 4p^3(2D^\circ)4d$	$4d' \ ^1G^\circ$	4	162842.07	366.82 1417.72	$4s^2 4p^3(2P^\circ)5s$	$5s'' \ ^1P^\circ$	1	181264.53					
			$4s^2 4p^3(2D^\circ)5s$					$5s' \ ^3D^\circ$		1	163270.01	3	182968.27
											2		163636.83
3	165054.55	1		188234.28									
$4s^2 4p^3(2D^\circ)5s$	$5s' \ ^3D^\circ$	1	163270.01	418.36 693.67	$4s^2 4p^3(2P^\circ)4d$	$4d'' \ ^1P^\circ$	1	188234.28					
			$4s^2 4p^3(2D^\circ)4d$					$4d' \ ^1G^\circ$	4	160415.88	1	188234.28	
										4		160415.88	
5	161109.55												

Kr III—Continued

Config.	Desig.	<i>J</i>	Level	Interval
4s ² 4p ³ (² P°) 4d	4d'' ¹ D°	2	188570. 30	
4s ² 4p ³ (² P°) 4d	4d'' ³ P°	0 1 2	190227. 24 193652. 80	3425. 56
4s ² 4p ³ (² D°) 5p	5p' ³ D	1 2 3	190724. 64 192702. 93 193826. 24	1978. 29 1123. 31
4s ² 4p ³ (² D°) 5p	5p' ³ F	2 3 4	193856. 49 194963. 87 195675. 44	1107. 38 711. 57
4s ² 4p ³ (² D°) 5p	5p' ¹ P	1	194121. 20	
4s ² 4p ³ (² D°) 5p	5p' ¹ F	3	195478. 99	
4s ² 4p ³ (² D°) 5p	5p' ³ P	2 1 0	198108. 88 198825. 49 198790. 11	-716. 61 35. 38
4s ² 4p ³ (² D°) 5p	5p' ¹ D	2	202897. 08	
4s ² 4p ³ (² P°) 5p	5p'' ³ D	1 2 3	207248. 18 208511. 04 209869. 57	1262. 86 1358. 53
4s ² 4p ³ (² D°) 5p	5p'' ³ S	1	208610. 74	
4s ² 4p ³ (² P°) 5p	5p'' ¹ P	1	209285. 43	
4s ² 4p ³ (² P°) 5p	5p'' ³ P	0 1 2	209787. 45 212264. 81 213058. 56	2477. 36 793. 75
4s ² 4p ³ (² P°) 5p	5p'' ¹ D	2	212124. 52	
4s ² 4p ³ (⁴ S°) 6s	6s ⁵ S°	2	215522. 57	
4s ² 4p ³ (⁴ S°) 5d	5d ⁵ D°	0 1 2 3 4	216501. 29 216515. 30 216529. 32 216545. 64 216605. 16	14. 01 14. 02 16. 32 59. 52
4s ² 4p ³ (⁴ S°) 5d	5d ³ D°	1 2 3	219295. 44 217376. 66 220759. 93	-1918. 78 3383. 27
4s ² 4p ³ (⁴ S°) 6s	6s ³ S°	1	221843. 40	
4s ² 4p ³ (² D°) 6s	6s' ³ D°	1 2 3	233111. 63 234567. 20	1455. 57
4s ² 4p ³ (² D°) 5d	5d' ¹ D°?	2	233347. 38	
4s ² 4p ³ (² D°) 5d	5d' ¹ G°	4	235357. 36	
4s ² 4p ³ (² D°) 6s	6s' ¹ D°?	2	236183. 67	
4s ² 4p ³ (² D°) 5d	5d' ¹ °	1	237971. 13	
4s ² 4p ³ (² D°) 5d	5d' ² °	2	238608. 32	
4s ² 4p ³ (² P°) 6s	6s'' ³ P°	0 1 2	249167. 98 250911. 83	1743. 85
4s ² 4p ³ (² P°) 6s	6s'' ¹ P°	1	252461. 12	
Kr IV (⁴ S _{1½})	Limit	-----	298020	

November 1950.

843210°—52—14

Kr III OBSERVED TERMS*

Config.	Observed Terms
1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 3d ¹⁰ +	
4s ² 4p ⁴	{ 4p ⁴ ¹ S
4s 4p ⁵	{ 4p ⁴ ³ P 4p ⁵ ³ P° 4p ⁵ ¹ P°
4s ² 4p ³ (⁴ S°) nx	{ 5,6s ⁵ S° 5,6s ³ S°
4s ² 4p ³ (² D°) nx'	{ 5,6s' ³ D° 5,6s' ¹ D°
4s ² 4p ³ (² P°) nx''	{ 5,6s'' ³ P° 5,6s'' ¹ P°
	ns (n ≥ 5)
	np (n ≥ 5)
	nd (n ≥ 4)

*For predicted terms in the spectra of the Se I isoelectronic sequence, see Vol. II, Introduction.

(As I sequence; 33 electrons)

Z=36

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^3 {}^4S_{1/2}^{\circ}$ $4p^3 {}^4S_{1/2}^{\circ}$ cm⁻¹

I. P. volts

This spectrum has not been satisfactorily analyzed. Boyce has observed the three ultimate lines, 805 Å to 842 Å, having the transition $4p^3 {}^4S^{\circ} - 4p^4 {}^4P$. These observations have been used to calculate the recorded value of $4p^4 {}^4P$.

Rao and Krishnamurty have reported 61 additional lines in the interval from 2237 Å to 3934 Å, as combinations among four terms and a number of miscellaneous levels. Their terms start with $5s {}^4P_{0/2}=0$. By comparison with As I and Se II the writer has estimated that this level is approximately 166100 cm^{-1} above the ground state zero. This value, entered in brackets in the table, has been added to all levels taken from the 1939 reference, x denoting that they are not connected with the rest by observed combinations. The writer has also assigned numbers to the miscellaneous levels. The levels here designated as 1, 2, and 3 have been corrected to fit the observed combinations, because they are incorrectly tabulated in the published paper.

All but the two lowest terms in the table need confirmation.

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Kr IV

Kr IV

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$4s^2 4p^3$	$4p^3 {}^4S^{\circ}$	$1\frac{1}{2}$	0. 0		$4s^2 4p^2({}^3P)4d$	9	$0\frac{1}{2}, 1\frac{1}{2}$	172999 +x	
$4s 4p^4$	$4p^4 {}^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	118760 122426 124108	-3666 -1682	$4s^2 4p^2({}^3P)4d$	10	$1\frac{1}{2}, 2\frac{1}{2}$	173552 +x	
$4s^2 4p^2({}^3P)5s$	$5s {}^4P$	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	[166100] +x 169200 +x 172423 +x	3100 3223	$4s^2 4p^2({}^3P)4d$	11	$1\frac{1}{2}, 2\frac{1}{2}$	177800 +x	
$4s^2 4p^2({}^3P)4d$	1	$1\frac{1}{2}, 2\frac{1}{2}$	164416 +x		$4s^2 4p^2({}^3P)4d$	12	$1\frac{1}{2}$	182553. 5+x	
$4s^2 4p^2({}^3P)4d$	2	$1\frac{1}{2}, 2\frac{1}{2}$	165881 +x		$4s^2 4p^2({}^3P)5p$	$5p {}^4D^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	202711. 8+x 204240. 9+x 207424. 5+x 210732 +x	1529. 1 3183. 6 3308
$4s^2 4p^2({}^3P)4d$	3	$1\frac{1}{2}, 2\frac{1}{2}$	166062 +x		$4s^2 4p^2({}^3P)5p$	$5p {}^4P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	207962 +x 208450 +x 208798. 5+x	488 348
$4s^2 4p^2({}^3P)4d$	4	$2\frac{1}{2}$	167708 +x		$4s^2 4p^2({}^3P)5p$	$5p {}^4S^{\circ}$	$1\frac{1}{2}$	208881 +x	
$4s^2 4p^2({}^3P)4d$	5	$2\frac{1}{2}, 3\frac{1}{2}$	168025 +x		$4s^2 4p^2({}^3P)5p$	1°	$1\frac{1}{2}, 2\frac{1}{2}$	210126. 9+x	
$4s^2 4p^2({}^3P)4d$	6	$2\frac{1}{2}, 3\frac{1}{2}$	168159 +x		$4s^2 4p^2({}^3P)5p$	2°	$1\frac{1}{2}, 2\frac{1}{2}$	210563. 4+x	
$4s^2 4p^2({}^3P)4d$	7	$1\frac{1}{2}, 2\frac{1}{2}$	170862 +x		$4s^2 4p^2({}^3P)5p$	3°	$1\frac{1}{2}, 2\frac{1}{2}$	212831. 3+x	
$4s^2 4p^2({}^3P)4d$	8	$2\frac{1}{2}$	171482 +x						

Kr IX

(Ni I sequence; 28 electrons)

Z=36

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} {}^1S_0$ $3d^{10} {}^1S_0$ cm⁻¹

I. P. volts

This spectrum has not been analyzed, but Paul has observed three lines as follows:

I A	Int.	Wave No.
114. 948	75	869958
115. 741	100	863998
117. 710	20	849546

By analogy with the related isoelectronic spectra Se VII to Y XII, he has classified these lines as combinations of the ground term with terms from the $3d^9 4p$ configuration, as given in the table. He has examined this spectrum and furnished these results especially for inclusion here.

REFERENCE

F. W. Paul (letter, November 1950). (T) (C L)

Kr IX

Config.	Desig.	J	Level
$3d^{10}$	$3d^{10} {}^1S$	0	0
$3d^9({}^2D)4p$	$4p {}^3P^\circ$	2	849546
		1	
		0	
$3d^9({}^2D)4p$	$4p {}^1P^\circ$	1	863998
$3d^9({}^2D)4p$	$4p {}^3D^\circ$	3	869958
		2	
		1	

November 1950.

RUBIDIUM

Rb I

37 electrons

Z=37

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 5s^2 S_{0\frac{1}{2}}$ $5s^2 S_{0\frac{1}{2}}$ Rb⁸⁵ 33691.02 ± 0.03 cm⁻¹ I. P. 4.176 volts $5s^2 S_{0\frac{1}{2}}$ Rb⁸⁷ 33691.10 ± 0.03 cm⁻¹ I. P. 4.176 volts

The well known series of Rb I reported in the classical publications of Paschen-Götze and Fowler have recently been extended. Kratz has observed in absorption the principal series (np^2P°) to $n=77$.

Mack has furnished in advance of publication his observations of forbidden absorption lines, all of which are combinations of the ground term, $5s^2S$, with higher series members nd^2D and ns^2S . The series called nf^2F° in the table is labeled by Mack n^2Q to indicate that the terms are nf^2F° , ng^2G , nh^2H° , etc.

Ramb has observed with the interferometer 33 lines between 5169.651 Å and 10075.711 Å. The inverted terms of the Bergmann series have been discussed in detail by Meissner and Masaki. From these and other papers Mack has prepared the list of terms as far as $12s^2S$, and furnished his manuscript for inclusion here. He notes that the values of the terms $6s^2S$, $4d^2D$, 4 to $8f^2F^\circ$, 5, $6g^2G$, and $6h^2H^\circ$ need further adjustment from more accurate wave lengths. He suggests, also, the following improved values of intervals:

Term	Interval	Term	Interval
$5p^2P^\circ$	237.598	$7d^2D$	1.507
$5d^2D$	2.960	$8d^2D$	1.013
$6d^2D$	2.262	$9d^2D$	0.697

Brackets denote that the tabular entries have been derived from series calculations.

Kratz derives the limit 33690.96 ± 0.03 cm⁻¹, from the long np^2P° series, and calculates the limit of the lowest hyperfine structure sublevel as 33691.02 ± 0.03 . Mack confirms this calculation for Rb⁸⁵ and adds the value quoted above for Rb⁸⁷.

Beutler has observed in absorption 39 lines between 594.61 Å and 809.72 Å, and classified all but 3 as transitions from the ground term to levels above the ionization limit. His rounded-off wave numbers have been entered in the table. These levels have as limits, two higher terms $^3P^\circ$ and $^1P^\circ$ in Rb II. The miscellaneous levels have been assigned numbers by the writer. The double entries of J for these levels (unlike those for unresolved terms) indicate that the existing data are insufficient to distinguish which value of J is correct.

The Quadratic Zeeman Effect in the principal series of Rb I is discussed by Harting and Klinkenberg.

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Rb I

Rb I

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$4p^6(^1S)5s$	$5s\ ^2S$	$0\frac{1}{2}$	0. 00		$4p^6(^1S)11d$	$11d\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	32514. 63 32514. 93	0. 30
$4p^6(^1S)5p$	$5p\ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	12578. 96 12816. 56	237. 60	$4p^6(^1S)13p$	$13p\ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	32665. 03 32667. 63	2. 60
$4p^6(^1S)4d$	$4d\ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	19355. 01 19355. 45	-0. 44	$4p^6(^1S)12d$	$12d\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	32724. 68 32724. 93	0. 25
$4p^6(^1S)6s$	$6s\ ^2S$	$0\frac{1}{2}$	20133. 6		$4p^6(^1S)14p$	$14p\ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	32838. 02 32840. 02	2. 00
$4p^6(^1S)6p$	$6p\ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	23715. 19 23792. 69	77. 50	$4p^6(^1S)13d$	$13d\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	32883. 21 32883. 41	0. 20
$4p^6(^1S)5d$	$5d\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	25700. 56 25703. 52	2. 96	$4p^6(^1S)15p$	$15p\ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	32970. 66 32972. 19	1. 53
$4p^6(^1S)7s$	$7s\ ^2S$	$0\frac{1}{2}$	26311. 46		$4p^6(^1S)14d$	$14d\ ^2D$	$1\frac{1}{2}, 2\frac{1}{2}$	33005. 84	
$4p^6(^1S)4f$	$4f\ ^2F^{\circ}$	$2\frac{1}{2}, 3\frac{1}{2}$	26791. 98		$4p^6(^1S)16p$	$16p\ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	33074. 59 33075. 83	1. 24
$4p^6(^1S)7p$	$7p\ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	27835. 05 27870. 14	35. 09	$4p^6(^1S)15d$	$15d\ ^2D$	$1\frac{1}{2}, 2\frac{1}{2}$	33102. 39	
$4p^6(^1S)6d$	$6d\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	28687. 15 28689. 41	2. 26	$4p^6(^1S)17p$	$17p\ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	33157. 54 33158. 54	1. 00
$4p^6(^1S)8s$	$8s\ ^2S$	$0\frac{1}{2}$	29046. 84		$4p^6(^1S)16d$	$16d\ ^2D$	$1\frac{1}{2}, 2\frac{1}{2}$	33180. 03	
$4p^6(^1S)5f$	$5f\ ^2F^{\circ}$	$3\frac{1}{2}$ $2\frac{1}{2}$	29277. 58 29277. 59	-0. 01	$4p^6(^1S)18p$	$18p\ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	33224. 83 33225. 67	0. 84
$4p^6(^1S)5g$	$5g\ ^2G$	$3\frac{1}{2}, 4\frac{1}{2}$	29297. 6		$4p^6(^1S)17d$	$17d\ ^2D$	$1\frac{1}{2}, 2\frac{1}{2}$	33244. 23	
$4p^6(^1S)8p$	$8p\ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	29834. 96 29853. 82	18. 86	$4p^6(^1S)19p$	$19p\ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	33280. 13 33280. 81	0. 68
$4p^6(^1S)7d$	$7d\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	30280. 18 30281. 69	1. 51	$4p^6(^1S)18d$	$18d\ ^2D$	$1\frac{1}{2}, 2\frac{1}{2}$	33295. 28	
$4p^6(^1S)9s$	$9s\ ^2S$	$0\frac{1}{2}$	30499. 06		$4p^6(^1S)20p$	$20p\ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	33326. 13 33326. 70	0. 57
$4p^6(^1S)6f$	$6f\ ^2F^{\circ}$	$3\frac{1}{2}$ $2\frac{1}{2}$	30627. 77 30627. 78	-0. 01	$4p^6(^1S)19d$	$19d\ ^2D$	$1\frac{1}{2}, 2\frac{1}{2}$	33338. 80	
$4p^6(^1S)6g$	$6g\ ^2G$	$3\frac{1}{2}, 4\frac{1}{2}$	30636. 9		$4p^6(^1S)21p$	$21p\ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	33364. 81 33365. 29	0. 48
$4p^6(^1S)6h$	$6h\ ^2H^{\circ}$	$4\frac{1}{2}, 5\frac{1}{2}$	30643. 5		$4p^6(^1S)20d$	$20d\ ^2D$	$1\frac{1}{2}, 2\frac{1}{2}$	33375. 4	
$4p^6(^1S)9p$	$9p\ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	30958. 94 30970. 22	11. 28	$4p^6(^1S)22p$	$22p\ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	33397. 66 33398. 09	0. 43
$4p^6(^1S)8d$	$8d\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	31221. 47 31222. 48	1. 01	$4p^6(^1S)21d$	$21d\ ^2D$	$1\frac{1}{2}, 2\frac{1}{2}$	33406. 86	
$4p^6(^1S)10s$	$10s\ ^2S$	$0\frac{1}{2}$	31362. 36		$4p^6(^1S)23p$	$23p\ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	33425. 76 33426. 15	0. 39
$4p^6(^1S)7f$	$7f\ ^2F^{\circ}$	$3\frac{1}{2}$ $2\frac{1}{2}$	31441. 52 31441. 53	-0. 01	$4p^6(^1S)22d$	$22d\ ^2D$	$1\frac{1}{2}, 2\frac{1}{2}$	33433. 88	
$4p^6(^1S)10p$	$10p\ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	31653. 88 31661. 19	7. 31	$4p^6(^1S)24p$	$24p\ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	33450. 03 33450. 33	0. 35
$4p^6(^1S)9d$	$9d\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	31821. 88 31822. 58	0. 70	$4p^6(^1S)23d$	$23d\ ^2D$	$1\frac{1}{2}, 2\frac{1}{2}$	33457. 06	
$4p^6(^1S)11s$	$11s\ ^2S$	$0\frac{1}{2}$	31917. 25		$4p^6(^1S)25p$	$25p\ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	33471. 11 33471. 43	0. 32
$4p^6(^1S)8f$	$8f\ ^2F^{\circ}$	$2\frac{1}{2}, 3\frac{1}{2}$	31969. 43		$4p^6(^1S)24d$	$24d\ ^2D$	$1\frac{1}{2}, 2\frac{1}{2}$	33477. 0	
$4p^6(^1S)11p$	$11p\ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	32113. 58 32118. 55	4. 97	$4p^6(^1S)26p$	$26p\ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	33489. 53 33489. 79	0. 26
$4p^6(^1S)10d$	$10d\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	[32227. 68] 32228. 16	[0. 48]	$4p^6(^1S)25d$	$25d\ ^2D$	$1\frac{1}{2}, 2\frac{1}{2}$	33494. 8	
$4p^6(^1S)12s$	$12s\ ^2S$	$0\frac{1}{2}$	[32295. 04]		$4p^6(^1S)27p$	$27p\ ^2P^{\circ}$	$0\frac{1}{2}, 1\frac{1}{2}$	33505. 92	
$4p^6(^1S)12p$	$12p\ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	32433. 50 32437. 04	3. 54	$4p^6(^1S)26d$	$26d\ ^2D$	$1\frac{1}{2}, 2\frac{1}{2}$	33510. 1	
					$4p^6(^1S)28p$	$28p\ ^2P^{\circ}$	$0\frac{1}{2}, 1\frac{1}{2}$	33520. 22	

Rb I—Continued

Rb I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
4p ^o (1S)27d	27d ² D	1½, 2½	33524. 01		4p ^o (1S)40p	40p ² P ^o	0½, 1½	33612. 34	
4p ^o (1S)29p	29p ² P ^o	0½, 1½	33532. 91		4p ^o (1S)39d	39d ² D	1½, 2½	33613. 38	
4p ^o (1S)28d	28d ² D	1½, 2½	33535. 9		4p ^o (1S)41s	41s ² S	0½	33614. 23	
4p ^o (1S)30p	30p ² P ^o	0½, 1½	33544. 26		4p ^o (1S)38f	38f ² F ^o	2½, etc.	33614. 55	
4p ^o (1S)29d	29d ² D	1½, 2½	33547. 3		4p ^o (1S)41p	41p ² P ^o	0½, 1½	33616. 38	
4p ^o (1S)31p	31p ² P ^o	0½, 1½	33554. 47		4p ^o (1S)40d	40d ² D	1½, 2½	33617. 34	
4p ^o (1S)30d	30d ² D	1½, 2½	33557. 02		4p ^o (1S)42s	42s ² S	0½	33618. 26	
4p ^o (1S)32s	32s ² S	0½	33559. 2		4p ^o (1S)39f	39f ² F ^o	2½, etc.	33619. 5	
4p ^o (1S)29f	29f ² F ^o	2½, etc.	33559. 9		4p ^o (1S)42p	42p ² P ^o	0½, 1½	33620. 15	
4p ^o (1S)32p	32p ² P ^o	0½, 1½	33563. 57		4p ^o (1S)41d	41d ² D	1½, 2½	33621. 00	
4p ^o (1S)31d	31d ² D	1½, 2½	33565. 95		4p ^o (1S)43s	43s ² S	0½	33621. 62	
4p ^o (1S)33s	33s ² S	0½	33567. 7		4p ^o (1S)40f	40f ² F ^o	2½, etc.	33622. 05	
4p ^o (1S)30f	30f ² F ^o	2½, etc.	33568. 5		4p ^o (1S)43p	43p ² P ^o	0½, 1½	33623. 61	
4p ^o (1S)33p	33p ² P ^o	0½, 1½	33571. 85		4p ^o (1S)42d	42d ² D	1½, 2½	33624. 41	
4p ^o (1S)32d	32d ² D	1½, 2½	33574. 06		4p ^o (1S)44s	44s ² S	0½	33625. 08	
4p ^o (1S)34s	34s ² S	0½	[33575. 6]		4p ^o (1S)41f	41f ² F ^o	2½, etc.	33625. 56	
4p ^o (1S)31f	31f ² F ^o	2½, etc.	[33576. 7]		4p ^o (1S)44p	44p ² P ^o	0½, 1½	33626. 83	
4p ^o (1S)34p	34p ² P ^o	0½, 1½	33579. 34		4p ^o (1S)43d	43d ² D	1½, 2½	33627. 55	
4p ^o (1S)33d	33d ² D	1½, 2½	33581. 21		4p ^o (1S)45s	45s ² S	0½	33628. 23	
4p ^o (1S)35s	35s ² S	0½	33582. 7		4p ^o (1S)42f	42f ² F ^o	2½, etc.	33628. 76	
4p ^o (1S)32f	32f ² F ^o	2½, etc.	33583. 8		4p ^o (1S)45p	45p ² P ^o	0½, 1½	33629. 80	
4p ^o (1S)35p	35p ² P ^o	0½, 1½	33586. 11		4p ^o (1S)44d	44d ² D	1½, 2½	33630. 51	
4p ^o (1S)34d	34d ² D	1½, 2½	33587. 87		4p ^o (1S)46s	46s ² S	0½	33631. 13	
4p ^o (1S)36s	36s ² S	0½	33589. 09		4p ^o (1S)43f	43f ² F ^o	2½, etc.	[33631. 6]	
4p ^o (1S)33f	33f ² F ^o	2½, etc.	33589. 93		4p ^o (1S)46p	46p ² P ^o	0½, 1½	33632. 60	
4p ^o (1S)36p	36p ² P ^o	0½, 1½	33592. 31		4p ^o (1S)45d	45d ² D	1½, 2½	33633. 23	
4p ^o (1S)35d	35d ² D	1½, 2½	33593. 83		4p ^o (1S)47s	47s ² S	0½	33633. 93	
4p ^o (1S)37s	37s ² S	0½	33595. 02		4p ^o (1S)44f	44f ² F ^o	2½, etc.	[33634. 28]	
4p ^o (1S)34f	34f ² F ^o	2½, etc.	33595. 9		4p ^o (1S)47p	47p ² P ^o	0½, 1½	33635. 21	
4p ^o (1S)37p	37p ² P ^o	0½, 1½	33597. 96		4p ^o (1S)46d	46d ² D	1½, 2½	33635. 83	
4p ^o (1S)36d	36d ² D	1½, 2½	33599. 42		4p ^o (1S)48s	48s ² S	0½	33636. 33	
4p ^o (1S)38s	38s ² S	0½	33600. 57		4p ^o (1S)45f	45f ² F ^o	2½, etc.	33636. 86	
4p ^o (1S)35f	35f ² F ^o	2½, etc.	33601. 44		4p ^o (1S)48p	48p ² P ^o	0½, 1½	33637. 63	
4p ^o (1S)38p	38p ² P ^o	0½, 1½	33603. 14		4p ^o (1S)47d	47d ² D	1½, 2½	33638. 17	
4p ^o (1S)37d	37d ² D	1½, 2½	33604. 43		4p ^o (1S)49s	49s ² S	0½	33638. 80	
4p ^o (1S)39s	39s ² S	0½	33605. 26		4p ^o (1S)46f	46f ² F ^o	2½, etc.	[33639. 10]	
4p ^o (1S)36f	36f ² F ^o	2½, etc.	33605. 97		4p ^o (1S)49p	49p ² P ^o	0½, 1½	33639. 89	
4p ^o (1S)39p	39p ² P ^o	0½, 1½	33607. 96		4p ^o (1S)48d	48d ² D	1½, 2½	33640. 46	
4p ^o (1S)38d	38d ² D	1½, 2½	33609. 08		4p ^o (1S)50s	50s ² S	0½	33640. 95	
4p ^o (1S)40s	40s ² S	0½	33610. 09		4p ^o (1S)47f	47f ² F ^o	2½, etc.	33641. 32	
4p ^o (1S)37f	37f ² F ^o	2½, etc.	33610. 41		4p ^o (1S)50p	50p ² P ^o	0½, 1½	33642. 01	

Rb I—Continued

Rb I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
4p ⁶ (¹ S)49d	49d ² D	1½, 2½	33642. 7		4p ⁵ 5s(³ P ₁)4d	4d 2°	0½, 1½	139910	
4p ⁶ (¹ S)51s	51s ² S	0½	33643. 98		4p ⁵ 5s(³ P ₂)4d	4d 3°	0½, 1½	145540	
4p ⁶ (¹ S)51p	51p ² P°	0½, 1½	33644. 02		4p ⁵ 5s(³ P ₃)4d	4d 4°	0½, 1½	146050	
4p ⁶ (¹ S)50d	50d ² D	1½, 2½	33644. 5		4p ⁵ 5s(³ P ₁)4d	4d 5°	0½, 1½	146780	
4p ⁶ (¹ S)52s	52s ² S	0½	33644. 95		4p ⁵ 5s(³ P ₂)4d	4d 6°	0½, 1½	147330	
4p ⁶ (¹ S)52p	52p ² P°	0½, 1½	33645. 96		4p ⁵ 5s(³ P ₃)4d	4d 7°	0½, 1½	147580	
4p ⁶ (¹ S)51d	51d ² D	1½, 2½	33646. 54		4p ⁵ 5s(³ P ₁)4d	4d 8°	0½, 1½	148910	
4p ⁶ (¹ S)53s	53s ² S	0½	33646. 84?		4p ⁵ 5s(³ P ₃)6s	6s ² P°	1½	151760	— 1220
4p ⁶ (¹ S)53p	53p ² P°	0½, 1½	33647. 72		4p ⁵ 5s(³ P ₁)6s		0½?	152980	
4p ⁶ (¹ S)52d	52d ² D	1½, 2½	33648. 29		4p ⁵ 5s(³ P°)6s	6s ⁴ P°	2½		
4p ⁶ (¹ S)54p	54p ² P°	0½, 1½	33649. 33		4p ⁵ 5s(³ P ₁)6s		1½?	152380	— 6140
4p ⁶ (¹ S)53d	53d ² D	1½, 2½	33650. 03		4p ⁵ 5s(³ P ₃)6s		0½?	158520	
4p ⁶ (¹ S)55p	55p ² P°	0½, 1½	33650. 95		4p ⁵ 5s(¹ P ₁)4d	4d' 9°	0½, 1½	152690	
4p ⁶ (¹ S)54d	54d ² D	1½, 2½	33651. 36		4p ⁵ 5s(¹ P ₁)4d	4d' 10°	0½, 1½	154040	
4p ⁶ (¹ S)56p	56p ² P°	0½, 1½	33652. 41		4p ⁵ 5s(³ P ₂)5d	5d 11°	0½, 1½	154260	
4p ⁶ (¹ S)57p	57p ² P°	0½, 1½	33653. 80		4p ⁵ 5s(¹ P ₁)4d?	4d' 12°	0½, 1½	154950	
4p ⁶ (¹ S)58p	58p ² P°	0½, 1½	33655. 11		4p ⁵ 5s(³ P ₁)5d	5d 13°	0½, 1½	156310	
4p ⁶ (¹ S)59p	59p ² P°	0½, 1½	33656. 39		4p ⁵ 5s(³ P ₂)5d	5d 14°	0½, 1½	156980	
4p ⁶ (¹ S)60p	60p ² P°	0½, 1½	33657. 55		4p ⁵ 5s(³ P ₃)5d	5d 15°	0½, 1½	157380	
4p ⁶ (¹ S)61p	61p ² P°	0½, 1½	33658. 69		4p ⁵ 5s(³ P ₁)5d	5d 16°	0½, 1½	157750	
4p ⁶ (¹ S)62p	62p ² P°	0½, 1½	33659. 80		4p ⁵ 5s(³ P ₁)5d	5d 17°	0½, 1½	159070	
4p ⁶ (¹ S)63p	63p ² P°	0½, 1½	33660. 82		4p ⁵ 5s(¹ P ₁)6s	6s' ² P°	0½	159210	1780
4p ⁶ (¹ S)64p	64p ² P°	0½, 1½	33661. 81				1½	160990	
4p ⁶ (¹ S)65p	65p ² P°	0½, 1½	33662. 73		4p ⁵ 5s(³ P ₂)7s	7s ² P°	1½	159210	
4p ⁶ (¹ S)66p	66p ² P°	0½, 1½	33663. 63				0½		
4p ⁶ (¹ S)67p	67p ² P°	0½, 1½	33664. 48		4p ⁵ 5s(³ P°)7s	7s ⁴ P°	2½		
4p ⁶ (¹ S)68p	68p ² P°	0½, 1½	33665. 28		4p ⁵ 5s(³ P ₁)7s?		1½?	160060	— 6770
4p ⁶ (¹ S)69p	69p ² P°	0½, 1½	33666. 04		4p ⁵ 5s(³ P ₃)7s?		0½?	166830	
4p ⁶ (¹ S)70p	70p ² P°	0½, 1½	33666. 78		4p ⁵ 5s(³ P ₂)6d	6d 18°	0½, 1½	161210	
4p ⁶ (¹ S)71p	71p ² P°	0½, 1½	33667. 49		4p ⁵ 5s(³ P ₃)6d	6d 19°	0½, 1½	161450	
4p ⁶ (¹ S)72p	72p ² P°	0½, 1½	33668. 11		4p ⁵ 5s(³ P ₁)6d	6d 20°	0½, 1½	162820	
4p ⁶ (¹ S)73p	73p ² P°	0½, 1½	33668. 79		4p ⁵ 5s(³ P ₂)7d	7d 21°	0½, 1½	162820	
4p ⁶ (¹ S)74p	74p ² P°	0½, 1½	33669. 86		4p ⁵ 5s(³ P ₃)5d	5d 22°	0½, 1½	163380	
4p ⁶ (¹ S)75p	75p ² P°	0½, 1½	33670. 01		4p ⁵ 5s(³ P°)8s	8s ⁴ P°	2½		
4p ⁶ (¹ S)76p	76p ² P°	0½, 1½	33670. 61		4p ⁵ 5s(³ P ₁)8s		1½?	163500	
4p ⁶ (¹ S)77p	77p ² P°	0½, 1½	33671. 07				0½		
Rb II (¹ S ₀)	<i>Limit</i>	-----	33691. 02		4p ⁵ 5s(¹ P ₁)5d	5d' 23°	0½, 1½	164700	
4p ⁵ (³ P ₂)5s ²	5s ² ² P°	1½	123500	— 6820	4p ⁵ 5s(¹ P ₁)5d	5d' 24°	0½, 1½	165080	
4p ⁵ (¹ P ₁)5s ²		0½	130320		4p ⁵ 5s(¹ P ₁)5d	5d' 25°	0½, 1½	166060	
4p ⁵ 5s(³ P ₂)4d	4d 1°	0½, 1½	138780		4p ⁵ 5s(¹ P ₁)7s	7s' ² P°	0½?	168180	
							1½		
					Rb II (5s ³ P ₂)	<i>Limit</i>	-----	167040	
					Rb II (5s ³ P ₁)	<i>Limit</i>	-----	168570	
					Rb II (5s ³ P ₃)	<i>Limit</i>	-----	175570	
					Rb II (5s ¹ P ₁)	<i>Limit</i>	-----	177160	

Rb II

(Kr I sequence; 36 electrons)

Z=37

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 {}^1S_0$ $4p^6 {}^1S_0$ 221852 cm^{-1}

I. P. 27.5 volts

Laporte, Miller, and Sawyer have published a list of 102 classified lines extending from 697.04 Å to 7698.57 Å. They have utilized earlier data by Reinheimer and Otsuka together with their own additional observations. Their limit is from the $ns {}^3P_{2,1}^{\circ}$ series ($n=5,6$) fitted to a Rydberg formula. The higher limit given in the table has been obtained by adding to their ionization limit the interval of the ground term of Rb III.

The author's notation is entered in column one of the table. The writer has provisionally assigned to the miscellaneous levels the pair-coupling notation, used in these volumes for spectra of the inert gases. These assignments require further confirmation.

REFERENCES

O. Laporte, G. R. Miller, and R. A. Sawyer, Phys. Rev. **38**, 843 (1931). (I P) (T) (C L)
 J. E. Mack, Rev. Mod. Phys. **22**, No. 1, 64 (1950). (Summary hfs)

Rb II

Rb II

Authors	Config.	Desig.	J	Level	Authors	Config.	Desig.	J	Level
$4p^6 {}^1S_0$	$4p^6$	$4p^6 {}^1S$	0	0. 00	$5p \ 7_1$ $5p \ 8_2$	$4p^5({}^2P_{0,2}^{\circ})5p$	$5p' [1\frac{1}{2}]$	1 2	163929. 47 164972. 81
$4d \ 1_1^{\circ}$ $4d \ 4_0^{\circ}$	$4p^5({}^2P_{1,3}^{\circ})4d$	$4d [0\frac{1}{2}]^{\circ}$	1 0	126453. 53 138799. 56	$5p \ 9_1$ $5p \ 10_0$	"	$5p' [0\frac{1}{2}]$	1 0	165094. 55 167637. 30
$4d \ 2_2^{\circ}$ $4d \ 5_1^{\circ}$	"	$4d [1\frac{1}{2}]^{\circ}$	2 1	128693 70 140615. 18	$6s \ 3P_2^{\circ}$ $6s \ 3P_1^{\circ}$	$4p^5({}^2P_{1,3}^{\circ})6s$	$6s [1\frac{1}{2}]^{\circ}$	2 1	179740. 11 180173. 33
$4d \ 3_2^{\circ}$	"	$4d [2\frac{1}{2}]^{\circ}$	2 3	136665. 21	$6s \ 1P_1^{\circ}$	$4p^5({}^2P_{0,2}^{\circ})6s$	$6s' [0\frac{1}{2}]^{\circ}$	0 1	188622. 28
$4d \ 6i_{1,2}$	$4p^5({}^2P_{0,2}^{\circ})4d$	$4d' [2\frac{1}{2}]^{\circ}$	2 3	143027. 39	$5d \ 1i$	$4p^5({}^2P_{1,3}^{\circ})5d$	$5d [0\frac{1}{2}]^{\circ}$	1 0	184205. 27
$4d \ 7_2^{\circ}$ $4d \ 8i_{1,2}$	"	$4d' [1\frac{1}{2}]^{\circ}$	2 1	143960. 94 145630. 06	$5d \ 3_3$	"	$5d [3\frac{1}{2}]^{\circ}$	3 4	185131. 62
$5s \ 3P_2^{\circ}$ $5s \ 3P_1^{\circ}$	$4p^5({}^2P_{1,3}^{\circ})5s$	$5s [1\frac{1}{2}]^{\circ}$	2 1	133347. 29 134875. 14	$5d \ 2_2^{\circ}$ $5d \ 6i$	"	$5d [1\frac{1}{2}]^{\circ}$	2 1	184841. 65 187340. 37
$5s \ 3P_0^{\circ}$ $5s \ 1P_1^{\circ}$	$4p^5({}^2P_{0,2}^{\circ})5s$	$5s' [0\frac{1}{2}]^{\circ}$	0 1	141879. 24 143467. 00	$5d \ 4_2^{\circ}$ $5d \ 5_3^{\circ}$	"	$5d [2\frac{1}{2}]^{\circ}$	2 3	185622. 53 186010. 87
$5p \ 1_1$	$4p^5({}^2P_{1,3}^{\circ})5p$	$5p [0\frac{1}{2}]$	1	154279. 25	$5d \ 7_1$ $5d \ 8i_{1,2}$	$4p^5({}^2P_{0,2}^{\circ})5d$	$5d' [1\frac{1}{2}]^{\circ}$	1 2	189006. 27 192380. 15
$5p \ 2_2$ $5p \ 3_3$	"	$5p [2\frac{1}{2}]$	2 3	156742. 20 156900. 72					
$5p \ 4_1$ $5p \ 5_2$	"	$5p [1\frac{1}{2}]$	1 2	158156. 66 158717. 04		Rb III (${}^2P_{1,3}^{\circ}$)	Limit	-----	221852
$5p \ 6_0$	"	$5p [0\frac{1}{2}]$	0	161205. 13		Rb III (${}^2P_{0,2}^{\circ}$)	Limit	-----	229232

May 1951.

Rb II OBSERVED LEVELS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 3d ¹⁰ 4s ² +	Observed Terms		
4p ⁶	4p ⁶ 1S		
	ns (n ≥ 5)		
4p ⁵ (² P ^o)nx	{ 5, 6s ³ P ^o 5, 6s ¹ P ^o		
jI-Coupling Notation			
	Observed Pairs		
	ns (n ≥ 5)	np (n ≥ 5)	nd (n ≥ 4)
4p ⁵ (² P _{1/2} ^o)nx	5, 6s [1½] ^o	5p [0½] 5p [2½] 5p [1½]	4,5d [0½] ^o 5d [3½] ^o 4,5d [1½] ^o 4,5d [2½] ^o
4p ⁵ (² P _{3/2} ^o)nx'	5, 6s' [0½] ^o	5p' [1½] 5p' [0½]	4d' [2½] ^o 4,5d' [1½] ^o

*For predicted levels in the spectra of the Kr I isoelectronic sequence, see Introduction.

Rb III

(Br I sequence; 35 electrons)

Z=37

Ground state 1s² 2s² 2p⁶ 3s² 3p⁶ 3d¹⁰ 4s² 4p⁵ ²P_{1/2}^o

4p⁵ ²P_{1/2}^o 320000 cm⁻¹

I. P. 40 volts

The analysis is incomplete. Tomboulian has classified 30 lines of Rb III in the region between 482 Å and 815 Å as due to transitions from the ground term to 11 higher terms.

He has estimated the value of the limit by extrapolation of isoelectronic sequence data.

REFERENCE

D. H. Tomboulian, Phys. Rev. **54**, 350 (1938). (I P) (T) (C L)

Rb III

Rb III

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
4s ² 4p ⁵	4p ⁵ ² P ^o	1½ 0½	0 7380	-7380	4s ² 4p ⁴ (³ P)4d	4d ² D	1½ 2½	173475	
4s 4p ⁶	4p ⁶ ² S	0½	130036		4s ² 4p ⁴ (³ P)4d	4d ² P	0½ 1½	177044	
4s ² 4p ⁴ (³ P)4d	4d ⁴ D	3½ 2½ 1½ 0½	152353 154403	-2050	4s ² 4p ⁴ (¹ D)5s	5s' ² D	1½ 2½	182963	
4s ² 4p ⁴ (³ P)5s	5s ⁴ P	2½ 1½ 0½	164507 168085	-3578	4s ² 4p ⁴ (¹ D)4d	4d' ² D	2½ 1½	200883	
4s ² 4p ⁴ (³ P)4d	4d ⁴ P	0½ 1½ 2½	170423 170847	424	4s ² 4p ⁴ (¹ D)4d	4d' ² P	1½ 0½	204104	
4s ² 4p ⁴ (³ P)5s	5s ² P	1½ 0½	172045 175396	-3351	4s ² 4p ⁴ (¹ S)5s	5s'' ² S	0½	207277	
					Rb IV (³ P ₂)	Limit	-----	[320000]	

Rb III OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} +$	Observed Terms	
$4s^2 4p^6$	$4p^6 \ ^2P^o$	
$4s 4p^6$	$4p^6 \ ^2S$	
	$ns (n \geq 5)$	$nd (n \geq 4)$
$4s^2 4p^4(^3P)nx$	$\left\{ \begin{array}{l} 5s \ ^4P \\ 5s \ ^2P \end{array} \right.$	$4d \ ^4P \quad 4d \ ^4D$
$4s^2 4p^4(^1D)nx'$		$4d \ ^2P \quad 4d \ ^2D$
$4s^2 4p^4(^1S)nx''$	$5s' \ ^2D$	$4d' \ ^2P \quad 4d' \ ^2D$
	$5s'' \ ^2S$	

*For predicted terms in the spectra of the Br I isoelectronic sequence, see Vol. II, Introduction.

Rb IX

(Cu I sequence; 29 electrons)

$Z=37$

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s \ ^2S_{0\frac{1}{2}}$

$4s \ ^2S_{0\frac{1}{2}} \quad \text{cm}^{-1}$

I. P. volts

This spectrum has not been analyzed, but Mack has observed the principal doublet, as follows:

I A	Wave No.	Desig.
583. 37	171417	$4s \ ^2S_{0\frac{1}{2}} - 4p \ ^2P_{1\frac{1}{2}}$
628. 62	159078	$4s \ ^2S_{0\frac{1}{2}} - 4p \ ^2P_{0\frac{1}{2}}$

REFERENCE

J. E. Mack, Phys. Rev. **38**, 193 (L) (1931). (I P) (T) (C L)
March 1950.

Rb X

(Ni I sequence; 28 electrons)

$Z=37$

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} \ ^1S_0$

$3d^{10} \ ^1S_0 \quad 2235100 \text{ cm}^{-1}$

I. P. 277 volts

Edlén has observed lines from the $3d^{10} - 3d^9 np (n=4,5)$ and $3d^{10} - 3d^9 nf (n=4-6)$ configurations. In figure 2 of his paper on the spectra of highly ionized atoms, some lines from each group are indicated on the photograph of vacuum spark spectra of Rb observed between 50 Å and 100 Å.

From preliminary unpublished measurements he has derived provisional absolute term values from the Rydberg nf -series. The terms in the table are from his manuscript. His unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

B. Edlén, Physica **13**, No. 9, 547 (1947).
B. Edlén, letter (July 1950). (I P) (T)

Rb x

Rb x

Config.	Desig.	<i>J</i>	Level	Config.	Desig.	<i>J</i>	Level
$3d^{10}$	$3d^{10} 1S$	0	0	$3d^9(^2D)5p$	$5p \ ^3D^\circ$	$\begin{matrix} 3 \\ 2 \\ 1 \end{matrix}$	
$3d^9(^2D)4p$	$4p \ ^1P^\circ$	1	1023400				1568400
$3d^9(^2D)4p$	$4p \ ^3D^\circ$	$\begin{matrix} 3 \\ 2 \\ 1 \end{matrix}$	1031100	$3d^9(^2D)5f$	$5f \ ^1P^\circ$	1	1733800
				$3d^9(^2D)5f$	$5f \ ^3D^\circ$	$\begin{matrix} 3 \\ 2 \\ 1 \end{matrix}$	1799200
$3d^9(^2D)4f$	$4f \ ^3P^\circ$	$\begin{matrix} 2 \\ 1 \\ 0 \end{matrix}$	1517200	$3d^9(^2D)6f$	$6f \ ^1P^\circ$	1	1922300
$3d^9(^2D)4f$	$4f \ ^1P^\circ$	1	1527700	$3d^9(^2D)6f$	$4f \ ^3D^\circ$	$\begin{matrix} 3 \\ 2 \\ 1 \end{matrix}$	1935400
$3d^9(^2D)4f$	$4f \ ^3D^\circ$	$\begin{matrix} 3 \\ 2 \\ 1 \end{matrix}$	1548000	-----			
$3d^9(^2D)5p$	$5p \ ^1P^\circ$	1	1559600	Rb XI ($^2D_{2\frac{1}{2}}$)	Limit	-----	2235100

August 1950.

Rb xi

(Co I sequence; 27 electrons)

 $Z=37$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^9 \ ^2D_{2\frac{1}{2}}$ $3d^9 \ ^2D_{2\frac{1}{2}} \quad \text{cm}^{-1}$

I. P. volts

Edlén has observed three lines due to the transition $3p^6 3d^9 \ ^2D - 3p^5 3d^{10} \ ^2P^\circ$. From preliminary unpublished measurements he has furnished the provisional term values quoted in the table. In figure 2 of his paper on the spectra of highly ionized atoms, the lines are indicated on the photograph of the vacuum spark spectra of Rb observed between 50 Å and 100 Å. Another group of lines from $3d^9 - 3d^8 4p$ is also indicated in the figure.

Edlén's unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

- B. Edlén, *Physica*, **13**, No. 9, 547 (1947).
 B. Edlén, letter (July 1950). (T)

Rb xi

Config.	Desig.	<i>J</i>	Level	Interval
$3p^6 3d^9$	$3d^9 \ ^2D$	$2\frac{1}{2}$	0	-12300
		$1\frac{1}{2}$	12300	
$3p^5 3d^{10}$	$3d^{10} \ ^2P^\circ$	$1\frac{1}{2}$	1028700	-72700
		$0\frac{1}{2}$	1101400	

July 1950.

Rb XII

(Fe I sequence; 26 electrons)

 $Z=37$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^8 {}^3F_4$ $3d^8 {}^3F_4$ cm^{-1}

I. P. volts

This spectrum has not been analyzed, but Edlén has observed a group of lines due to terms from the $3d^8-3d^7 4p$ configurations. In figure 2 of his paper on the spectra of highly ionized atoms, the lines are indicated on the photograph of vacuum spark spectra of Rb observed between 50 Å and 100 Å.

The writer has assumed the ground state indicated above, by analogy with Ni III.

REFERENCE

B. Edlén, *Physica* **13**, No. 9, 547 (1947).

February 1950.

Rb XIII

(Mn I sequence; 25 electrons)

 $Z=37$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^7 {}^4F_{4\frac{1}{2}}$ $3d^7 {}^4F_{4\frac{1}{2}}$ cm^{-1}

I. P. volts

This spectrum has not been analyzed, but Edlén has observed a group of lines due to terms from the $3d^7-3d^6 4p$ configurations. In figure 2 of his paper on the spectra of highly ionized atoms, the lines are indicated on the photograph of vacuum spark spectra of Rb observed between 50 Å and 100 Å.

The writer has assumed the ground state indicated above, by analogy with Co III.

REFERENCE

B. Edlén, *Physica* **13**, No. 9, 547 (1947).

February 1950.

Rb XIV

(Cr I sequence; 24 electrons)

 $Z=37$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^6 {}^5D_4$ $3d^6 {}^5D_4$ cm^{-1}

I. P. volts

This spectrum has not been analyzed, but Edlén has observed a group of lines due to terms from the $3d^6-3d^5 4p$ configurations. In figure 2 of his paper on the spectra of highly ionized atoms the lines are indicated on the photograph of vacuum spark spectra of Rb observed between 50 Å and 100 Å.

The writer has assumed the ground state indicated above, by analogy with Fe III.

REFERENCE

B. Edlén, *Physica* **13**, No. 9, 547 (1947).

February 1950.

STRONTIUM

Sr I

38 electrons

Z=38

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 5s^2 {}^1S_0$ $5s^2 {}^1S_0$ 45925.6 cm^{-1}

I. P. 5.692 volts

The Sr I spectrum, like Ca I, has played an important part in the development of a satisfactory theory for atomic spectra. In addition to the "regular" series, terms involving two excited electrons were found in Sr I as well as in Ca I, and were first discussed in the classical paper by Russell and Saunders in 1925. White discusses these terms further in his paper on "Auto-Ionization in the Alkaline Earth Metals and the Inert Gases."

The regular series terms, i. e., those having the limit 2S in Sr II, have been taken from the paper by Saunders, except for $5d {}^1D_2$. Other terms are from the papers by Russell and Saunders and by Meggers. The writer has also had at her disposal Russell's unpublished line list and multiplet arrays.

Sullivan has observed 102 Sr I lines with the interferometer. From these observations he has improved a number of the term values. All of the three-place entries in the table, except that of $6p {}^3P_0^{\circ}$; and the two-place values for the terms 6, 7, $10p {}^1P^{\circ}$; 4, $5f {}^1F^{\circ}$; 7, $8d {}^3D$; and $4d^2 {}^3P$, are from his paper. The rest have been adjusted to these by the writer with the aid of the best available line list. The resolution of the terms 6, $7f {}^3F^{\circ}$ is dubious, owing to the inaccuracy of the existing wavelengths.

Although long series are well known, and all of the terms are well connected by observed intersystem combinations, a number of Sr I lines remain unclassified, and many of the observed lines should be remeasured. Humphreys has recently remeasured two lines, 20764.19 Å and 20778.14 Å, which are designated as $6s {}^3S_1 - 6p {}^3P_0^{\circ}$ and $5d {}^1D_2 - 4f {}^1F_3^{\circ}$, respectively. He expects to extend the analysis from further observations in the far infrared. A monograph containing a homogeneous list of lines and term values is seriously needed as soon as his work is completed.

REFERENCES

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 H. N. Russell and F. A. Saunders, *Astroph. J.* **61**, 38 (1925). (I P) (T) (C L)
 H. E. White, *Phys. Rev.* **33**, 2016 (1931)
 W. F. Meggers, *Bur. Std. J. Research* **10**, 679, RP558 (1933). (I P) (T) (C L)
 F. J. Sullivan, *Univ. Pittsburgh Bull.* **35**, No. 1, 1 (1938). (T) (C L)
 J. F. Mack, *Rev. Mod. Phys.* **22**, No. 1, 64 (1950). (Summary hfs)
 C. J. Humphreys, unpublished material (April 1951). (C L)

Sr I

Sr I

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
5s ²	5s ² 1S	0	0. 000		5s(2S)8p	8p 1P°	1	41172. 15	
5s(2S)5p	5p 3P°	0	14317. 520	186. 831	5s(2S)5f	5f 3F°	2	41364. 614	0. 918
		1	14504. 351	394. 212			3	41365. 532	1. 163
		2	14898. 563				4	41366. 695	
5s(2S)4d	4d 3D	1	18159. 056	59. 739	5s(2S)5f	5f 1F°	3	41518. 91	
		2	18218. 795	100. 472	5s(2S)7d	7d 1D	2	41831. 7	
		3	18319. 267		5s(2S)7d	7d 3D	1	41864. 40	4. 92
5s(2S)4d	4d 1D	2	20149. 700				2	41869. 32	5. 58
5s(2S)5p	5p 1P°	1	21698. 482				3	41874. 90	
5s(2S)6s	6s 3S	1	29038. 795		5s(2S)9s	9s 3S	1	42451. 2	
5s(2S)6s	6s 1S	0	30591. 8		5s(2S)9p	9p 1P°	1	42462. 36	
4d(2D)5p	5p' 3F°	2	33266. 872	322. 852	5s(2S)9s	9s 1S	0	42596. 0	
		3	33589. 724	329. 606	5s(2S)6f	6f 3F°	2	42776. 2?	0. 4?
		4	33919. 330				3	42776. 6?	1. 6?
4d(2D)5p	5p' 1D°	2	33826. 927				4	42778. 155	
5s(2S)6p	6p 3P°	0	33853. 516	14. 817	5s(2S)6f	6f 1F°	3	42839. 4	
		1	33868. 333	104. 748	5s(2S)8d	8d 1D	2	43020. 9	
		2	33973. 081		5s(2S)8d	8d 3D	1	43066. 7	3. 6
5s(2S)6p	6p 1P°	1	34098. 44				2	43070. 31	4. 46
5s(2S)5d	5d 1D	2	34727. 483				3	43074. 77	
5s(2S)5d	5d 3D	1	35006. 943	15. 072	5s(2S)10p	10p 1P°	1	43327. 94	
		2	35022. 015	23. 040	5s(2S)10s	10s 3S	1	43427. 6	
		3	35045. 055		5s(2S)10s	10s 1S	0	43512. 6	
5p ²	5p ² 3P	0	35193. 47	206. 67	5s(2S)7f	7f 3F°	2, 3	43624. 0	0. 5?
		1	35400. 138	274. 530			4	43624. 513	
		2	35674. 668		5s(2S)7f	7f 1F°	3	43656. 1	
4d(2D)5p	5p' 3D°	1	36264. 181	117. 588	5s(2S)9d	9d 1D	2	43780. 6	
		2	36381. 769	177. 745	5s(2S)9d	9d 3D	1, 2	43807. 8	4. 8
		3	36559. 514				3	43812. 6	
5p ²	5p ² 1D	2	36960. 881		5s(2S)11p	11p 1P°	1	43938. 26	
5p ²	5p ² 1S	0	37160. 278		5s(2S)11s	11s 3S	1	44043. 4	
4d(2D)5p	5p' 3P°	0	37292. 106	10. 654	5s(2S)11s	11s 1S	0	44097. 1	
		1	37302. 760	33. 856	5s(2S)8f	8f 3F°	2, 3, 4	44171. 8	
		2	37336. 616		5s(2S)8f	8f 1F°	3	44190. 1	
5s(2S)7s	7s 3S	1	37424. 713		5s(2S)10d	10d 3D	1	44283. 8	3. 3?
4d(2D)5p	5p' 1F°	3	38008. 0				2	44287. 1	12. 6
5s(2S)7s	7s 1S	0	38444. 054				3	44299. 7	
5s(2S)4f	4f 3F°	2	38750. 454	1. 986	5s(2S)12p	12p 1P°	1	44365. 9	
		3	38752. 440	2. 759	5s(2S)12s	12s 3S	1	44457. 3	
		4	38755. 199		4d ²	4d ² 3P	0	44525. 88	70. 09
5s(2S)7p	7p 1P°	1	38906. 90				1	44595. 97	133. 70
5s(2S)7p	7p 3P°	0	39411. 703	14. 768	5s(2S)9f	9f 3F°	2, 3, 4	44544. 6	
		1	39426. 471	30. 938	5s(2S)9f	9f 1F°	3	44556. 1	
		2	39457. 409		5s(2S)11d	11d 3D	1	44618. 4	6. 7
5s(2S)4f	4f 1F°	3	39539. 04				2	44625. 1	
5s(2S)6d	6d 3D	1	39685. 903	4. 946	5s(2S)13p	13p 1P°	1	44675. 1	
		2	39690. 849	12. 305					
		3	39703. 154						
5s(2S)6d	6d 1D	2	39733. 114						
5s(2S)8s	8s 3S	1	40761. 440						
5s(2S)8s	8s 1S	0	41052. 5						

Sr I—Continued

Sr I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval	
5s(2S)10f	10f 3F°	2, 3, 4	44809.8	4.9	5s(2S)12f	12f 1F°	3	45160.4		
5s(2S)10f	10f 1F°	3	44819.1		5s(2S)13f	13f 3F°	2, 3, 4	45274.1		
5s(2S)12d	12d 3D	1	44860.0		5s(2S)13f	13f 1F°	3	45277.9		
		2	44864.9		5s(2S)15d	15d 3D	1, 2, 3	45287.4		
5s(2S)14p	14p 1P°	1	44903.5		5s(2S)16d	16d 3D	1, 2, 3	45372.6		
5s(2S)11f	11f 3F°	2, 3, 4	45005.6							
5s(2S)11f	11f 1F°	3	45012.0		Sr II(2S _{0½})	Limit		45925.6		
5s(2S)13d	13d 3D	1, 2, 3	45043.9		4d(2D)6p	6p' 3F°	2	49409.9		148.4 107.5
5s(2S)12f	12f 3F°	2, 3, 4	45155.6				3	49558.3		
							4	49665.8		

April 1951.

Sr I OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 3d ¹⁰ 4s ² 4p ⁶ +	Observed Terms			
5s ²	5s ² 1S			
4d ²	4d ² 3P			
5p ²	{	5p ² 1S	5p ² 3P	5p ² 1D
		<i>ns</i> (<i>n</i> ≥ 5)	<i>np</i> (<i>n</i> ≥ 5)	<i>nd</i> (<i>n</i> ≥ 4)
5s(2S) <i>nx</i>	{	6-12s 3S 6-11s 1S	5-7p 3P° 5-14p 1P°	4-13, 15, 16d 3D 4-9d 1D
4d(2D) <i>nx</i> '	{		5p' 3P° 5p' 3D° 5, 6p' 3F° 5p' 1D° 5p' 1F°	4-13f 3F° 4-13f 1F°

*For predicted terms in the spectra of the Sr I isoelectronic sequence, see Vol. II, Introduction.

Sr II

(Rb I sequence; 37 electrons)

Z = 38Ground state 1s² 2s² 2p⁶ 3s² 3p⁶ 3d¹⁰ 4s² 4p⁶ 5s 2S_{0½}5s 2S_{0½} 88964.0 cm⁻¹

I. P. 11.027 volts

The analysis is from the paper by Saunders, Schneider, and Buckingham, who list 93 classified lines between 1482 Å and 13124 Å, including the infrared observations by Meggers and by Randall, in addition to their own observations. Sullivan has observed 14 of these lines with the interferometer, and derived improved term values for the six terms given to two decimal places in the table. Brackets denote calculated values.

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 J. E. Mack, Rev. Mod. Phys. **22**, No. 1, 64 (1950). (Summary hfs)

Sr II

Sr II

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$4p^6(^1S)5s$	$5s^2S$	$0\frac{1}{2}$	0. 00		$4p^6(^1S)10s$	$10s^2S$	$0\frac{1}{2}$	80701. 8	
$4p^6(^1S)4d$	$4d^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	14555. 90 14836. 24	280. 34	$4p^6(^1S)9d$	$9d^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	81240. 2 [81249. 0]	[8. 8]
$4p^6(^1S)5p$	$5p^2P^\circ$	$0\frac{1}{2}$ $1\frac{1}{2}$	23715. 19 24516. 65	801. 46	$4p^6(^1S)8f$	$8f^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	82005. 9	
$4p^6(^1S)6s$	$6s^2S$	$0\frac{1}{2}$	47736. 53		$4p^6(^1S)7g$	$7g^2G$	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 4\frac{1}{2} \end{array} \right\}$	82090. 4	
$4p^6(^1S)5d$	$5d^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	53286. 31 53372. 97	86. 66	$4p^6(^1S)11s$	$11s^2S$	$0\frac{1}{2}$	82576. 1	
$4p^6(^1S)6p$	$6p^2P^\circ$	$0\frac{1}{2}$ $1\frac{1}{2}$	55769. 7 56057. 9	288. 2	$4p^6(^1S)10d$	$10d^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	82951. 5 82953. 9	-2. 4
$4p^6(^1S)4f$	$4f^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	60991. 7		$4p^6(^1S)9f$	$9f^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	83472. 7	
$4p^6(^1S)7s$	$7s^2S$	$0\frac{1}{2}$	64964. 10		$4p^6(^1S)8g$	$8g^2G$	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 4\frac{1}{2} \end{array} \right\}$	[83533. 8]	
$4p^6(^1S)6d$	$6d^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	67522. 87 67563. 15	40. 28	$4p^6(^1S)12s$	$12s^2S$	$0\frac{1}{2}$	83879. 9	
$4p^6(^1S)7p$	$7p^2P^\circ$	$0\frac{1}{2}$ $1\frac{1}{2}$	68645. 2 68793. 6	148. 4	$4p^6(^1S)11d$	$11d^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	84142. 0 84146. 9	4. 9
$4p^6(^1S)5f$	$5f^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	71065. 8		$4p^6(^1S)10f$	$10f^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	84521. 0	
$4p^6(^1S)4g$	$4g^2G$	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 4\frac{1}{2} \end{array} \right\}$	71357. 8		$4p^6(^1S)9g$	$9g^2G$	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 4\frac{1}{2} \end{array} \right\}$	84567. 0	
$4p^6(^1S)8s$	$8s^2S$	$0\frac{1}{2}$	73237. 1		$4p^6(^1S)13s$	$13s^2S$	$0\frac{1}{2}$	84819. 1	
$4p^6(^1S)7d$	$7d^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	74621. 3 74643. 1	21. 8	$4p^6(^1S)12d$	$12d^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	85015. 5	
$4p^6(^1S)8p$	$8p^2P^\circ$	$0\frac{1}{2}$ $1\frac{1}{2}$	75311. 8		$4p^6(^1S)11f$	$11f^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	85294. 8	
$4p^6(^1S)6f$	$6f^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	76553. 4		$4p^6(^1S)10g$	$10g^2G$	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 4\frac{1}{2} \end{array} \right\}$	85330. 3	
$4p^6(^1S)5g$	$5g^2G$	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 4\frac{1}{2} \end{array} \right\}$	76737. 7		$4p^6(^1S)13d$	$13d^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	85668. 8	
$4p^6(^1S)9s$	$9s^2S$	$0\frac{1}{2}$	77857. 6		$4p^6(^1S)12f$	$12f^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	85883. 9	
$4p^6(^1S)8d$	$8d^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	78688. 8 78702. 4	13. 6	$4p^6(^1S)14d$	$14d^2D$	$\left\{ \begin{array}{l} 1\frac{1}{2} \\ 2\frac{1}{2} \end{array} \right\}$	86174. 6	
$4p^6(^1S)7f$	$7f^2F^\circ$	$\left\{ \begin{array}{l} 2\frac{1}{2} \\ 3\frac{1}{2} \end{array} \right\}$	79861. 3						
$4p^6(^1S)6g$	$6g^2G$	$\left\{ \begin{array}{l} 3\frac{1}{2} \\ 4\frac{1}{2} \end{array} \right\}$	79984. 3		Sr III(1S_0)	<i>Limit</i>		88964. 0	

March 1951.

Sr IV

(Br I sequence; 35 electrons)

Z=38

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^5 \ ^2P_{1/2}^{\circ}$

$4p^5 \ ^2P_{1/2}^{\circ}$ 460000 cm^{-1}

I. P. 57 volts

The analysis is incomplete. Tomboulian has classified 29 lines of Sr IV in the region between 367 Å and 710 Å as due to transitions from the ground term to 11 higher terms.

He has estimated the value of the limit by extrapolation of isoelectronic sequence data.

REFERENCE

D. H. Tomboulian, Phys. Rev. 54, 350 (1938). (I P) (T) (C L)

Sr IV					Sr IV				
Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$4s^2 4p^5$	$4p^5 \ ^2P^{\circ}$	$1\frac{1}{2}$ $0\frac{1}{2}$	0 9731	-9731	$4s^2 4p^4(^3P)5s$	$5s \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	221149 225853	-4704
$4s 4p^6$	$4p^6 \ ^2S$	$0\frac{1}{2}$	150505		$4s^2 4p^4(^3P)5s$	$5s \ ^2P$	$1\frac{1}{2}$ $0\frac{1}{2}$	232970 238203	-5233
$4s^2 4p^4(^3P)4d$	$4d \ ^4D$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	183305 186065	-2760	$4s^2 4p^4(^1D)5s$	$5s' \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	247616	
$4s^2 4p^4(^3P)4d$	$4d \ ^4P$	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	209214 209529	315	$4s^2 4p^4(^1D)4d$	$4d' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	249646	
$4s^2 4p^4(^3P)4d$	$4d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	214345		$4s^2 4p^4(^1D)4d$	$4d' \ ^2P$	$1\frac{1}{2}$ $0\frac{1}{2}$	253233	
$4s^2 4p^4(^3P)4d$	$4d \ ^2P$	$0\frac{1}{2}$ $1\frac{1}{2}$	218939		$4s^2 4p^4(^1S)5s$	$5s'' \ ^2S$	$0\frac{1}{2}$	271774	
					Sr v (3P_2)	Limit		460000	

February 1951.

Sr IV OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} +$	Observed Terms	
$4s^2 4p^5$	$4p^5 \ ^2P^{\circ}$	
$4s 4p^6$	$4p^6 \ ^2S$	
	$ns (n \geq 5)$	$nd (n \geq 4)$
$4s^2 4p^4(^3P)nx$	{ $5s \ ^4P$ $5s \ ^2P$	$4d \ ^4P$ $4d \ ^4D$ $4d \ ^2P$ $4d \ ^2D$
$4s^2 4p^4(^1D)nx'$	$5s' \ ^2D$	$4d' \ ^2P$ $4d' \ ^2D$
$4s^2 4p^4(^1S)nx''$	$5s'' \ ^2S$	

*For predicted terms in the spectra of the Br I isoelectronic sequence, see Vol. II, Introduction.

Sr XI

(Ni I sequence; 28 electrons)

Z=38

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} {}^1S_0$ $3d^{10} {}^1S_0$ 2613800 cm^{-1}

I. P. 324 volts

Edlén has observed lines from the $3d^{10}-3d^9 np$ ($n=4,5$) and $3d^{10}-3d^9 nf$ ($n=4-6$) configurations. In figure 2 of his paper on the spectra of highly ionized atoms, some lines from each group are indicated on the photograph of vacuum spark spectra of Sr observed between 50 A and 100 A.

From preliminary unpublished measurements he has derived provisional absolute term values from the Rydberg nf -series. The terms in the table are from his manuscript. His unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

B. Edlén, *Physica* **13**, No. 9, 547 (1947).

B. Edlén, letter (July 1950). (I P) (T)

Sr XI

Sr XI

Config.	Desig.	J	Level	Config.	Desig.	J	Level
$3d^{10}$	$3d^{10} {}^1S$	0	0	$3d^9({}^2D)5p$	$5p {}^3D^\circ$	3 2 1	1824800
$3d^9({}^2D)4p$	$4p {}^1P^\circ$	1	1193900	$3d^9({}^2D)5f$	$5f {}^1P^\circ$	1	2067000
$3d^9({}^2D)4p$	$4p {}^3D^\circ$	3 2 1	1203500	$3d^9({}^2D)5f$	$5f {}^3D^\circ$	3 2 1	2086400
$3d^9({}^2D)4f$	$4f {}^3P^\circ$	2 1 0	1743100	$3d^9({}^2D)6f$	$6f {}^1P^\circ$	1	2236600
$3d^9({}^2D)4f$	$4f {}^1P^\circ$	1	1755900	$3d^9({}^2D)6f$	$6f {}^3D^\circ$	3 2 1	2251700
$3d^9({}^2D)4f$	$4f {}^3D^\circ$	3 2 1	1781900				
$3d^9({}^2D)5p$	$5p {}^1P^\circ$	1	1814900	Sr XII (${}^2D_{3/2}$)	Limit	-----	2613800

August 1950.

Sr XII

(Co I sequence; 27 electrons)

Z=38

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^9 {}^2D_{3/2}$ $3d^9 {}^2D_{3/2}$ cm^{-1}

I. P. volts

Edlén has observed three lines due to the transition $3p^6 3d^9 {}^2D-3p^5 3d^{10} {}^2P^\circ$. From preliminary unpublished measurements he has furnished the provisional term values quoted in the table. In figure 2 of his paper on the spectra of highly ionized atoms, the lines are indicated on the photograph of the vacuum spark spectra of Sr observed between 50 A and 100 A. Another group of lines from $3d^9-3d^8 4p$ is also indicated in the figure.

Edlén's unit, 10^3 cm^{-1} , has here been changed to cm^{-1} .

REFERENCES

B. Edlén, *Physica* **13**, No. 9, 547 (1947).

B. Edlén, letter (July 1950). (T)

Sr XII

Config.	Desig.	J	Level	Interval
$3p^6 3d^9$	$3d^9 \ ^2D$	$2\frac{1}{2}$	0	-14500
		$1\frac{1}{2}$	14500	
$3p^5 3d^{10}$	$3d^{10} \ ^2P^o$	$1\frac{1}{2}$	1086400	-85400
		$0\frac{1}{2}$	1171800	

July 1950.

Sr XIII

(Fe I sequence; 26 electrons)

 $Z=38$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^8 \ ^3F_4$ $3d^8 \ ^3F_4$ cm^{-1}

I. P. volts

This spectrum has not been analyzed, but Edlén has observed a group of lines due to terms from the $3d^8-3d^7 4p$ configurations. In figure 2 of his paper on the spectra of highly ionized atoms, the lines are indicated on the photograph of vacuum spark spectra of Sr observed between 50 Å and 100 Å.

The writer has assumed the ground state indicated above, by analogy with Ni III.

REFERENCE

B. Edlén, *Physica* **13**, No. 9, 547 (1947).

February 1950.

Sr XIV

(Mn I sequence; 25 electrons)

 $Z=38$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^7 \ ^4F_{4\frac{1}{2}}$ $3d^7 \ ^4F_{4\frac{1}{2}}$ cm^{-1}

I. P. volts

This spectrum has not been analyzed, but Edlén has observed a group of lines due to terms from the $3d^7-3d^6 4p$ configurations. In figure 2 of his paper on the spectra of highly ionized atoms, the lines are indicated on the photograph of vacuum spark spectra of Sr observed between 50 Å and 100 Å.

The writer has assumed the ground state indicated above, by analogy with Co III.

REFERENCE

B. Edlén, *Physica* **13**, No. 9, 547 (1947).

February 1950.

YTTRIUM

Y I

39 electrons

Z=39

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 5s^2 4d^2 D_{1\frac{1}{2}}$ $4d^2 D_{1\frac{1}{2}}$ 52650 cm^{-1}

I. P. 6.5 volts

The analysis is by Meggers and Russell, who have classified 448 lines in the interval 2332.58 Å to 9494.81 Å, based chiefly on observations of wave length by Meggers and of Zeeman effect by Meggers and B. E. Moore. Observed intersystem combinations connect the doublet and quartet systems of terms. Their limit is from the $ns^2 D(n=5,6)$, $ns^4 D(n=6,7)$, $ns^4 F(n=5,6)$, and $ns^4 P(n=5,6)$ series represented by a Rydberg formula, with a Ritz correction applied to take into account the error in limits derived from series of only two members.

The writer has changed the notation to agree with that used in the present volumes for complex spectra. The notation of Meggers and Russell is given in the first column. In their term table the J -values for $z^2 H$ should be interchanged. This correction has been made in column one.

McNally and Harrison have observed the Zeeman Effect of Y I at the Massachusetts Institute of Technology and derived the g -values listed in the table from their own data combined with the earlier observations. In the course of this work they have adjusted the decimals of known levels to agree with their observed wave lengths, but have added no new analysis. These revisions have been adopted in the table.

REFERENCES

- W. F. Meggers and H. N. Russell, Bur. Std. J. Research **2**, 745, RP55 (1929). (I P) (T) (C L) (Z E) (G D)
 J. R. McNally, Jr., and G. R. Harrison, J. Opt. Soc. Am. **35**, 584 (1945). (Z E) (T) (C L)
 J. E. Mack, Rev. Mod. Phys. **22**, No. 1, 64 (1950). (Summary hfs)

Y I

Y I

Au- thors	Config.	Desig.	J	Level	Interval	Obs. g	Au- thors	Config.	Desig.	J	Level	Interval	Obs. g
$a^2 D_2$ $a^2 D_3$	$4d 5s^2$	$a^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	0.00 530.36	530.36	0.798 1.196	$z^4 D'_1$ $z^4 D'_2$ $z^4 D'_3$ $z^4 D'_4$	$4d 5s(a^3 D)5p$	$z^4 D^\circ$	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$	16435.80 16597.30 16816.60 17116.30	161.50 219.30 299.70	0.01 1.22 1.38 1.42
$z^2 P_1$ $z^2 P_2$	$5s^2(a^1 S)5p$	$z^2 P^\circ$	$0\frac{1}{2}$ $1\frac{1}{2}$	10529.20 11359.70	830.50	0.63 1.34	$a^2 G_3$ $a^2 G_4$	$4d^2(a^1 G)5s$	$a^2 G$	$4\frac{1}{2}$ $3\frac{1}{2}$	18499.30 18512.46	-13.16	1.12 0.90
$a^4 F'_2$ $a^4 F'_3$ $a^4 F'_4$ $a^4 F'_5$	$4d^2(a^3 F)5s$	$a^4 F$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	10937.35 11078.61 11277.96 11532.13	141.26 199.35 254.17	0.402 1.033 1.242 1.337	$z^4 P_1$ $z^4 P_2$ $z^4 P_3$	$4d 5s(a^3 D)5p$	$z^4 P^\circ$	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	18976.30 19027.50 19148.00	51.20 120.50	1.49
$z^4 F_2$ $z^4 F_3$ $z^4 F_4$ $z^4 F_5$	$4d 5s(a^3 D)5p$	$z^4 F^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$ $3\frac{1}{2}$ $4\frac{1}{2}$	14949.00 15245.80 15712.40 16234.50	296.80 466.60 522.10	0.47 1.08 1.26 1.33	$a^2 P'_1$ $a^2 P'_2$	$4d^2(a^3 P)5s$	$a^2 P$	$0\frac{1}{2}$ $1\frac{1}{2}$	19237.65 19406.19	168.54	1.18? 1.28?
$a^4 P'_1$ $a^4 P'_2$ $a^4 P'_3$	$4d^2(a^3 P)5s$	$a^4 P$	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	15221.81 15329.04 15476.69	107.23 147.65	2.61 1.62 1.27	$z^2 F_3$ $z^2 F_4$	$4d 5s(a^3 D)5p$	$z^2 F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	21528.60 21915.40	386.80	0.854 1.148
$a^2 F'_3$ $a^2 F'_4$	$4d^2(a^3 F)5s$	$a^2 F$	$2\frac{1}{2}$ $3\frac{1}{2}$	15326.83 15864.35	537.52	1.15 1.17	$y^2 D'_2$ $y^2 D'_3$	$4d 5s(a^1 D)5p$	$y^2 D^\circ$	$1\frac{1}{2}$ $2\frac{1}{2}$	24131.20 24746.60	615.40	0.823 1.088
$b^2 D_2$ $b^2 D_3$	$4d^2(b^1 D)5s$	$b^2 D$	$1\frac{1}{2}$ $2\frac{1}{2}$	15994.11 16158.9	164.8	0.86 1.22	$y^2 P_2$ $y^2 P_1$	$4d 5s(a^3 D)5p$	$y^2 P^\circ$	$1\frac{1}{2}$ $0\frac{1}{2}$	24480.60 24698.80	-218.20	1.302 0.674
$z^2 D'_3$ $z^2 D'_2$	$4d 5s(a^3 D)5p$	$z^2 D^\circ$	$2\frac{1}{2}$ $1\frac{1}{2}$	16066.00 16146.10	-80.10	1.203 0.797	$y^2 F_3$ $y^2 F_4$	$4d 5s(a^1 D)5p$	$y^2 F^\circ$	$2\frac{1}{2}$ $3\frac{1}{2}$	24518.80 24899.50	380.70	0.964 1.147

Y I—Continued

Y I—Continued

Au- thors	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Au- thors	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>					
<i>x</i> ² P ₁	4 <i>d</i> 5 <i>s</i> (<i>a</i> ¹ D) 5 <i>p</i>	<i>x</i> ² P°	0½	27824. 50	315. 10	0. 667	<i>z</i> ² H ₅	4 <i>d</i> ² (<i>a</i> ¹ G) 5 <i>p</i>	<i>z</i> ² H°	4½	37588. 20	379. 00	0. 91					
<i>x</i> ² P ₂			1½	28139. 60		1. 324	<i>z</i> ² H ₆			5½	37967. 20		1. 11					
<i>z</i> ⁴ G ₃	4 <i>d</i> ² (<i>a</i> ³ F) 5 <i>p</i>	<i>z</i> ⁴ G°	2½	28694. 00	294. 90	0. 56	<i>b</i> ⁴ D ₁	4 <i>d</i> 5 <i>s</i> (<i>a</i> ³ D) 5 <i>d</i>	<i>f</i> ⁴ D	0½	38469. 9	73. 8	1. 22					
<i>z</i> ⁴ G ₄			3½	28988. 90		1. 00	<i>b</i> ⁴ D ₂			1½	38543. 7			131. 7				
<i>z</i> ⁴ G ₅			4½	29364. 20		1. 18	<i>b</i> ⁴ D ₃			2½	38675. 4			190. 2				
<i>z</i> ⁴ G ₆			5½	29820. 40		1. 27	<i>b</i> ⁴ D ₄			3½	38865. 6			1. 34				
<i>b</i> ⁴ F ₂	4 <i>d</i> ³	<i>b</i> ⁴ F	1½	29272. 0	148. 4		<i>y</i> ² G ₄	4 <i>d</i> ² (<i>a</i> ¹ G) 5 <i>p</i>	<i>y</i> ² G°	3½	38479. 00	117. 70	0. 89					
<i>b</i> ⁴ F ₃			2½	29420. 4			193. 4			<i>y</i> ² G ₅	4½			38596. 70	1. 11			
<i>b</i> ⁴ F ₄			3½	29613. 8			228. 9			<i>a</i> ⁴ G ₃	4 <i>d</i> 5 <i>s</i> (<i>a</i> ³ D) 5 <i>d</i>			<i>c</i> ⁴ G	2½	38635. 5	126. 6	0. 73
<i>b</i> ⁴ F ₅			4½	29842. 7						<i>a</i> ⁴ G ₄					3½	38762. 1	187. 5	
				<i>a</i> ⁴ G ₅	4½	38949. 6	273. 4											
<i>y</i> ⁴ F ₂	4 <i>d</i> ² (<i>a</i> ³ F) 5 <i>p</i>	<i>y</i> ⁴ F°	1½	31508. 40	171. 60	0. 402	<i>a</i> ⁴ G ₆	4 <i>d</i> 5 <i>s</i> (<i>a</i> ³ D) 5 <i>d</i>	<i>e</i> ⁴ F	5½	39223. 0	118. 8	0. 57					
<i>y</i> ⁴ F ₃			2½	31680. 00		1. 033	<i>c</i> ⁴ F ₂			1½	39446. 3			112. 7				
<i>y</i> ⁴ F ₄			3½	31909. 20		1. 242	<i>c</i> ⁴ F ₃			2½	39565. 1			192. 7				
<i>y</i> ⁴ F ₅			4½	32188. 10		1. 337	<i>c</i> ⁴ F ₄			3½	39757. 8			205. 9				
<i>a</i> ² S ₁	5 <i>s</i> ² (<i>a</i> ¹ S) 6 <i>s</i>	<i>e</i> ² S	0½	31671. 5		2. 03	<i>c</i> ⁴ F ₅	4½	39963. 7									
<i>b</i> ⁴ P ₁	4 <i>d</i> ³	<i>b</i> ⁴ P	0½	31977. 5	113. 5		<i>d</i> ⁴ P ₁	4 <i>d</i> 5 <i>s</i> (<i>a</i> ³ D) 5 <i>d</i>	<i>f</i> ⁴ P	0½		62. 0	1. 52					
<i>b</i> ⁴ P ₂			1½	32091. 0			275. 3			<i>d</i> ⁴ P ₂	1½			40455. 1				
<i>b</i> ⁴ P ₃			2½	32366. 3						<i>d</i> ⁴ P ₃	2½			40517. 1				
<i>a</i> ⁴ D ₁	4 <i>d</i> 5 <i>s</i> (<i>a</i> ³ D) 6 <i>s</i>	<i>e</i> ⁴ D	0½	33148. 3	89. 6	0. 01	<i>v</i> ² D ₂	4 <i>d</i> ² (<i>a</i> ³ P) 5 <i>p</i>	<i>v</i> ² D°	1½	40636. 30	36. 10	1. 25					
<i>a</i> ⁴ D ₂			1½	33237. 9		173. 7	<i>v</i> ² D ₃			2½	40672. 40							
<i>a</i> ⁴ D ₃			2½	33411. 6		341. 2	<i>e</i> ² D ₂			1½	42655. 6			414. 4				
<i>a</i> ⁴ D ₄			3½	33752. 8			<i>e</i> ² D ₃			2½	43070. 0							
<i>y</i> ⁴ D ₁	4 <i>d</i> ² (<i>a</i> ³ F) 5 <i>p</i>	<i>y</i> ⁴ D°	0½	33215. 40	49. 90	-0. 04	<i>b</i> ² S ₁	5 <i>p</i> ² (¹ S) 5 <i>s</i> ?	<i>f</i> ² S	0½	42685. 7	136. 90						
<i>y</i> ⁴ D ₂			1½	33265. 30		1. 20	<i>v</i> ² F ₃			2½	42857. 90							
<i>y</i> ⁴ D ₃			2½	33357. 60		1. 38	<i>v</i> ² F ₄			3½	42994. 80							
<i>y</i> ⁴ D ₄			3½	33614. 50		1. 40	<i>d</i> ⁴ F ₂			4 <i>d</i> ² (<i>a</i> ³ F) 5 <i>d</i> ?	<i>f</i> ⁴ F			1½	43095. 7	241. 9		
<i>z</i> ² G ₄	4 <i>d</i> ² (<i>a</i> ³ F) 5 <i>p</i>	<i>z</i> ² G°	3½	33432. 30	356. 50	0. 91	<i>d</i> ⁴ F ₃	2½	43337. 6			366. 9						
<i>z</i> ² G ₅			4½	33788. 80			1. 15	<i>d</i> ⁴ F ₄	3½				43704. 5	485. 8				
<i>x</i> ² F ₃	4 <i>d</i> ² (<i>a</i> ³ F) 5 <i>p</i>	<i>x</i> ² F°	2½	33608. 20	421. 60	0. 985	<i>d</i> ⁴ F ₅	4½	44190. 3			1. 26						
<i>x</i> ² F ₄			3½	34029. 80			1. 17	<i>c</i> ² S ₁	0½	43643. 6	313. 3							
<i>b</i> ² P ₁	5 <i>p</i> ² (<i>b</i> ³ P) 5 <i>s</i> ?	<i>e</i> ² P	0½	33613. 2	229. 2	0. 93	<i>e</i> ⁴ F ₂	4 <i>d</i> ² (<i>a</i> ³ F) 6 <i>s</i>	<i>g</i> ⁴ F	1½			267	0. 73				
<i>b</i> ² P ₂			1½	33842. 4			1. 36			<i>e</i> ⁴ F ₃	2½	44052. 8?						
<i>x</i> ² D ₂	4 <i>d</i> ² (<i>a</i> ³ F) 5 <i>p</i>	<i>x</i> ² D°	1½	33906. 80	340. 90	0. 826	<i>e</i> ⁴ F ₄			3½	44366. 1	393. 5						
<i>x</i> ² D ₃			2½	34247. 70			1. 186			<i>e</i> ⁴ F ₅	4½				44759. 6			
<i>c</i> ⁴ P ₁	5 <i>p</i> ² (<i>b</i> ³ P) 5 <i>s</i>	<i>e</i> ⁴ P	0½	33911. 5	244. 4	2. 58	<i>c</i> ⁴ D ₁	4 <i>d</i> 5 <i>s</i> (<i>a</i> ³ D) 7 <i>s</i>	<i>g</i> ⁴ D	0½		267						
<i>c</i> ⁴ P ₂			1½	34155. 9			1. 47			<i>c</i> ⁴ D ₂	1½			44654. 8				
<i>c</i> ⁴ P ₃			2½	34521. 3			1. 46			<i>c</i> ⁴ D ₃	2½			44654. 8				
<i>c</i> ² D ₂	5 <i>p</i> ² (<i>f</i> ¹ D) 5 <i>s</i> ?	<i>e</i> ² D	1½	34231. 6	25. 8	0. 94?	<i>c</i> ⁴ D ₄			3½	44922			267				
<i>c</i> ² D ₃			2½	34257. 4			1. 34	<i>d</i> ⁴ D ₁	0½	44660. 1	88. 2							
<i>z</i> ² S ₁	4 <i>d</i> ² (<i>a</i> ³ P) 5 <i>p</i>	<i>z</i> ² S°	0½	34438. 20		1. 70	<i>d</i> ⁴ D ₂	1½	44748. 3	260. 1								
<i>x</i> ⁴ D ₁	4 <i>d</i> ² (<i>a</i> ³ P) 5 <i>p</i>	<i>x</i> ⁴ D°	0½	35816. 70	319. 00	0. 18	<i>d</i> ⁴ D ₃	2½	45008. 4		195. 6							
<i>x</i> ⁴ D ₂			1½	36135. 70			1. 94?	<i>d</i> ⁴ D ₄	3½	45204. 0								
<i>x</i> ⁴ D ₃			2½	36061. 00			1. 34	<i>f</i> ⁴ F ₂	5 <i>p</i> ² (<i>b</i> ³ P) 4 <i>d</i>	<i>h</i> ⁴ F		1½	44742. 7?	326. 6?				
<i>x</i> ⁴ D ₄			3½	36361. 30			1. 38	<i>f</i> ⁴ F ₃				2½	45069. 3		408. 0			
<i>d</i> ² D ₂	4 <i>d</i> 5 <i>s</i> (<i>a</i> ³ D) 6 <i>s</i>	<i>f</i> ² D	1½	36420. 6	10. 5	0. 90	<i>f</i> ⁴ F ₄	3½			45388. 5							
<i>d</i> ² D ₃			2½	36431. 1			1. 21	<i>f</i> ⁴ F ₅			4½	45796. 5						
<i>w</i> ² D ₂	4 <i>d</i> ² (<i>b</i> ¹ D) 5 <i>p</i>	<i>w</i> ² D°	1½	36452. 30	166. 20	0. 86	2	2	{0½ 1½}	44984. 1								
<i>w</i> ² D ₃			2½	36618. 50			1. 32											
<i>z</i> ⁴ S ₂	4 <i>d</i> ² (<i>a</i> ³ P) 5 <i>p</i>	<i>z</i> ⁴ S°	1½	36750. 80			3	3	3½	45663. 5								
<i>y</i> ⁴ P ₁	4 <i>d</i> ² (<i>a</i> ³ P) 5 <i>p</i>	<i>y</i> ⁴ P°	0½	37039. 50	109. 00	2. 28	<i>c</i> ² P ₁	5 <i>p</i> ² (<i>b</i> ³ P) 4 <i>d</i> ?	<i>f</i> ² P	0½	45947. 5	46. 7						
<i>y</i> ⁴ P ₂			1½	37148. 50			1. 58			<i>c</i> ² P ₂	1½			45994. 2				
<i>y</i> ⁴ P ₃			2½	37476. 00			1. 45			<i>e</i> ⁴ P ₁	4 <i>d</i> ² (<i>a</i> ³ P) 6 <i>s</i>			<i>g</i> ⁴ P	0½			
1	2½	37074. 2		<i>e</i> ⁴ P ₂	1½													
<i>w</i> ² P ₂	4 <i>d</i> ² (<i>b</i> ¹ D) 5 <i>p</i>	<i>w</i> ² P°	1½	37243. 89	-35. 41	1. 32	<i>e</i> ⁴ P ₃	2½	50254. 0									
<i>w</i> ² P ₁			0½	37279. 30			0. 60											
<i>w</i> ² F ₃	4 <i>d</i> ² (<i>b</i> ¹ D) 5 <i>p</i>	<i>w</i> ² F°	2½	37412. 90	206. 90	0. 86												
<i>w</i> ² F ₄			3½	37619. 80			1. 14											
							Y II (¹ S ₀)	Limit			52650							

Y I OBSERVED TERMS*

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 +$	Observed Terms		
	$ns (n \geq 5)$	$np (n \geq 5)$	$nd (n \geq 4)$
$4d 5s^2$	a^2D		
$4d^3$	b^4P	b^4F	
$5s^2(a^1S)nx$	e, g^3S		
$4d 5s(a^3D)nx$	e, g^4D f^2D	z^2P^o z^4P^o y^2P^o x^2P^o	g^2D f^4D e^4F e^4G
$4d 5s(a^1D)nx$		y^2D^o y^4D^o x^2D^o	$f^4F?$
$4d^2(a^3F)nx$	a, g^4F a^2F	y^4F^o x^2F^o z^4G^o z^2G^o	
$4d^2(a^3P)nx$	a, g^4P a^2P	x^4D^o v^2D^o	
$4d^2(b^1D)nx$	b^2D	w^2P^o w^2D^o	
$4d^2(a^1G)nx$	a^2G	v^2F^o y^2G^o z^2H^o	
$5p^2(b^3P)nx$	e^4P $e^2P?$		
$5p^2(f^1D)nx$	$e^2D?$		
$5p^2(^1S)nx$	$f^2S?$		$f^2P?$ h^4D h^4F

*For predicted terms in the spectra of the Y I isoelectronic sequence, see Vol. II, Introduction.

Y II

(Sr I sequence; 38 electrons)

Z=39

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 5s^2 {}^1S_0$ $a {}^1S_0$ 100000 cm^{-1}

I. P. 12.4 volts

The analysis is by Meggers and Russell, who greatly extended the earlier work on Y II. They have published 223 classified lines in the interval 2243.06 Å to 8835.85 Å, based on Meggers' observations of wavelength and Zeeman effect. Observed intersystem combinations connect the systems of terms of different multiplicity.

Their limit has been determined by applying a Rydberg formula to the ns -series terms $a, e {}^1S$ and $a, f {}^1S$, ($n=5,6$); and an empirical correction to compensate for the error resulting in limits from series of only two members.

Subsequently McNally and Harrison observed the Zeeman effect of Y II at the Massachusetts Institute of Technology, and determined g -values for all but one of the 61 known energy levels. Their work confirms in great detail the 1929 analysis. In the course of this work they have adjusted the decimals of the known levels to agree with their observations, but have added no new levels. Their revised level values and g -values are quoted in the table.

In 1937 Miss Ho observed the Y spectra with a hollow cathode discharge, and provisionally suggested 39 miscellaneous levels of Y II based on the additional lines observed with this source. Since these levels are not fitted into the term array of Y II, they are not included in the table, pending further confirmation of the new lines as observed with higher dispersion.

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 J. R. McNally, Jr. and G. R. Harrison, *J. Opt. Soc. Am.* **35**, 584 (1945). (Z E) (T)

Y II

Y II

Au- thors	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Au- thors	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>			
<i>a</i> ¹ S	5 <i>s</i> ²	<i>a</i> ¹ S	0	0.00		0/0	<i>a</i> ³ S	5 <i>s</i> (² S)6 <i>s</i>	<i>e</i> ³ S	1	58262.1		1.99			
<i>a</i> ³ D	4 <i>d</i> (² D)5 <i>s</i>	<i>a</i> ³ D	1	840.18	204.89 404.68	0.495	<i>a</i> ¹ F'	4 <i>d</i> (² D)5 <i>d</i>	<i>e</i> ¹ F	3	58532.8		1.02			
			2	1045.07		1.162										
			3	1449.75		1.332										
<i>a</i> ¹ D	4 <i>d</i> (² D)5 <i>s</i>	<i>a</i> ¹ D	2	3296.20		1.000	<i>c</i> ³ D	4 <i>d</i> (² D)5 <i>d</i>	<i>f</i> ³ D	1 2 3	58719.6 58946.6 59326.7	227.0 380.1	0.58 1.17 1.17			
<i>a</i> ³ F'	4 <i>d</i> ²	<i>a</i> ³ F	2	8003.12	324.91 415.30	0.668	<i>b</i> ³ P'	5 <i>p</i> ²	<i>e</i> ³ P	0 1 2	58776.3 59147.4 59669.3	371.1 521.9	0/0 1.49 1.38			
			3	8328.03		1.087										
			4	8743.33		1.256										
<i>a</i> ³ P'	4 <i>d</i> ²	<i>a</i> ³ P	0	13883.44	134.86 79.84	0/0	<i>a</i> ³ G	4 <i>d</i> (² D)5 <i>d</i>	<i>e</i> ³ G	3 4 5	59179.4 59472.1 59900.5	292.7 428.4	0.90 1.06 1.22			
			1	14018.24		1.497										
			2	14098.08		1.41										
<i>b</i> ¹ D	4 <i>d</i> ²	<i>b</i> ¹ D	2	14832.89		1.10	<i>b</i> ¹ S	4 <i>d</i> (² D)5 <i>d</i>	<i>e</i> ¹ S	0	59616.0?					
<i>a</i> ¹ G	4 <i>d</i> ²	<i>a</i> ¹ G	4	15682.91		0.999	<i>a</i> ¹ P'	4 <i>d</i> (² D)5 <i>d</i>	<i>e</i> ¹ P	1	59715.8		0.99			
<i>z</i> ³ P	5 <i>s</i> (² S)5 <i>p</i>	<i>z</i> ³ P°	0	23445.05	331.20 870.89	0/0	<i>d</i> ¹ D	4 <i>d</i> (² D)5 <i>d</i>	<i>f</i> ¹ D	2	60535.1		1.09			
			1	23776.25		1.489										
			2	24647.14		1.493										
<i>z</i> ¹ D'	4 <i>d</i> (² D)5 <i>p</i>	<i>z</i> ¹ D°	2	26147.27		0.912	<i>b</i> ³ F'	4 <i>d</i> (² D)5 <i>d</i>	<i>e</i> ³ F	2 3 4	61336.6 61650.8 61934.5	314.2 283.7	0.69 1.09 1.26			
<i>z</i> ³ F	4 <i>d</i> (² D)5 <i>p</i>	<i>z</i> ³ F°	2	27227.00	305.34 861.83	0.764										
			3	27532.34		1.082										
			4	28394.17		1.254										
<i>z</i> ¹ P	4 <i>d</i> (² D)5 <i>p</i>	<i>z</i> ¹ P°	1	27516.70		0.910	<i>c</i> ¹ S	5 <i>s</i> (² S)6 <i>s</i>	<i>f</i> ¹ S	0	61367.4		0/0			
<i>z</i> ³ D'	4 <i>d</i> (² D)5 <i>p</i>	<i>z</i> ³ D°	1	28595.27	134.75 483.93	0.596	<i>e</i> ¹ D	5 <i>s</i> (² S)5 <i>d</i>	<i>g</i> ¹ D	2	62495.2		0.99			
			2	28730.02		1.165										
			3	29213.95		1.333										
<i>y</i> ³ P	4 <i>d</i> (² D)5 <i>p</i>	<i>y</i> ³ P°	0	32048.76	75.31 159.36	0/0	<i>b</i> ¹ G	4 <i>d</i> (² D)5 <i>d</i>	<i>e</i> ¹ G	4	63350.3		1.00			
			1	32124.07		1.496										
			2	32283.43		1.494										
<i>z</i> ¹ F	4 <i>d</i> (² D)5 <i>p</i>	<i>z</i> ¹ F°	3	33336.74		0.997	<i>c</i> ³ P'	4 <i>d</i> (² D)5 <i>d</i>	<i>f</i> ³ P	0 1 2	64102.8 64263.2 64596.9	160.4 333.7	0/0 1.50 1.48			
			<i>y</i> ¹ P	5 <i>s</i> (² S)5 <i>p</i>	<i>y</i> ¹ P°	1								44568.63		0.998
						<i>b</i> ³ D								4 <i>d</i> (² D)6 <i>s</i>	<i>e</i> ³ D	1
2	55032.2															
3	55645.2															
<i>c</i> ¹ D	4 <i>d</i> (² D)6 <i>s</i>	<i>e</i> ¹ D	2	55724.9		1.05	Y III (² D) _{1/2}	<i>Limit</i>	-----	100000						

May 1951.

Y II OBSERVED TERMS*

$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 \dagger$	Observed Terms		
$5s^2$	a^1S		
$4d^2$	{ a^3P b^1D a^3F a^1G		
$5p^2$	{ e^3P h^1D		
	ns ($n \geq 5$)		np ($n \geq 5$)
$4d(2D)nx$	{ a, e^3D a, e^1D		y^3P^o z^3D^o z^3F^o z^1P^o z^1D^o z^1F^o
$5s(2S)nx$	{ e^3S f^1S		z^3P^o y^1P^o
	nd ($n \geq 4$)		
$4d(2D)nx$	{ f^3S e^1S	f^3P f^3D e^1P f^1D	e^3F e^3G e^1F e^1G
$5s(2S)nx$	{ g^3D g^1D		

*For predicted terms in the spectrum of the Sr I isoelectronic sequence, see Vol. II, Introduction.

Y III

(Rb I sequence; 37 electrons)

$Z=39$

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^2 D_{1/2}$

$4d^2 D_{1/2}$ 165289 cm^{-1}

I. P. 20.5 volts

The analysis is incomplete. Twelve lines have been classified, in the interval between 989.21 Å and 2945.92 Å. The term values in the table are quoted from the paper by Meggers and Russell. They are based on the measurements by Meggers, except for the two ultraviolet lines 989.21 Å and 996.37 Å from Bowen and Millikan.

Gibbs and White published Moseley diagrams for the $5s$ and $5p$ electrons of this isoelectronic sequence. Bowen and Millikan extended these diagrams to include $4d$, $5d$, and $4f$ electrons and showed that the $4d$ electron was the most firmly bound in Y III. By comparison with Rb I and Sr II they estimated the value of the $4f^2F^o$ term and from this derived the limit quoted here.

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 J. R. McNally, Jr., and G. R. Harrison, J. Opt. Soc. Am. **35**, 584 (1945). (Z E)
 J. E. Mack, Rev. Mod. Phys. **22**, No. 1, 64 (1950). (Summary hfs)

Y III

Config.	Desig.	J	Level	Interval	Obs. g
$4p^6(^1S)4d$	$4d\ ^2D$	$1\frac{1}{2}$	0. 0	724. 8	0. 79
		$2\frac{1}{2}$	724. 8		1. 22
$4p^6(^1S)5s$	$5s\ ^2S$	$0\frac{1}{2}$	7466. 2		2. 00
$4p^6(^1S)5p$	$5p\ ^2P^o$	$0\frac{1}{2}$	41401. 2	1553. 5	0. 66
		$1\frac{1}{2}$	42954. 7		1. 33
$4p^6(^1S)6s$	$6s\ ^2S$	$0\frac{1}{2}$	86713. 9		
$4p^6(^1S)5d$	$5d\ ^2D$	$1\frac{1}{2}$	88378. 8	198. 3	
		$2\frac{1}{2}$	88577. 1		
$4p^6(^1S)4f$	$4f\ ^2F^o$	$2\frac{1}{2}$	101090. 0		
		$3\frac{1}{2}$			
-----	-----	---	-----		
Y IV (1S_0)	<i>Limit</i>	---	[165289]		

March 1951.

Y v

(Br I sequence; 35 electrons)

 $Z=39$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^5\ ^2P_{1\frac{1}{2}}^o$ $4p^5\ ^2P_{1\frac{1}{2}}^o$ 620000 cm^{-1}

I. P. 77 volts

The analysis is incomplete. Paul and Rense have classified 41 lines between 313 A and 629 A, as due to transitions from the ground term to 13 higher terms.

They have estimated the limit by extrapolation of isoelectronic sequence data.

REFERENCE

F. W. Paul and W. A. Rense, Phys. Rev. **56**, 1110 (1939). (I P) (T) (C L)

Y v

Y v

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$4s^2 4p^5$	$4p^5\ ^2P^o$	$1\frac{1}{2}$	0	-12068	$4s^2 4p^4(^1D)4d$	$4d'\ ^2D$	$2\frac{1}{2}$	289836	-7236
		$0\frac{1}{2}$	12068				$1\frac{1}{2}$	297072	
$4s 4p^6$	$4p^6\ ^2S$	$0\frac{1}{2}$	170936		$4s^2 4p^4(^1D)4d$	$4d'\ ^2F$	$3\frac{1}{2}$	291052	
$4s^2 4p^4(^3P)4d$	$4d\ ^4D$	$3\frac{1}{2}$	202902	-10352	$4s^2 4p^4(^3P)5s$	$5s\ ^2P$	$1\frac{1}{2}$	296745	-9604
		$2\frac{1}{2}$	213254				$0\frac{1}{2}$	306349	
		$1\frac{1}{2}$	219116						
$4s^2 4p^4(^3P)4d$	$4d\ ^4P$	$0\frac{1}{2}$	247473	879	$4s^2 4p^4(^1D)4d$	$4d'\ ^2P$	$0\frac{1}{2}$	299567	650
		$1\frac{1}{2}$	248352				$1\frac{1}{2}$	300217	
		$2\frac{1}{2}$	250406				2054		
$4s^3 4p^4(^3P)4d$	$4d\ ^2D$	$1\frac{1}{2}$	253678	9846	$4s^2 4p^4(^1D)5s$	$5s'\ ^2D$	$2\frac{1}{2}$	315430	-1762
		$2\frac{1}{2}$	263524				$1\frac{1}{2}$	317192	
$4s^2 4p^4(^3P)4d$	$4d\ ^2P$	$0\frac{1}{2}$	258518	49	$4s^2 4p^4(^1S)4d$	$4d''\ ^2D$	$2\frac{1}{2}$	318885	
		$1\frac{1}{2}$	258567				$1\frac{1}{2}$		
$4s^2 4p^4(^3P)4d$	$4d\ ^2F$	$3\frac{1}{2}$	274254		Y VI (3P_2)	[Limit]	-----	[620000]	
$4s^2 4p^4(^3P)5s$	$5s\ ^4P$	$2\frac{1}{2}$	280932	-6273					
		$1\frac{1}{2}$	287205						
		$0\frac{1}{2}$	290911		-3706				

February 1951.

Y V OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 3d ¹⁰ +	Observed Terms	
4s ² 4p ⁵	4p ⁵ ² P ^o	
4s 4p ⁶	4p ⁶ ² S	
	ns (n ≥ 5)	nd (n ≥ 4)
4s ² 4p ⁴ (³ P)nx	{ 5s ⁴ P 5s ² P	4d ⁴ P 4d ⁴ D 4d ² P 4d ² D 4d ² F
4s ² 4p ⁴ (¹ D)nx'		5s' ² D 4d' ² P 4d' ² D 4d' ² F
4s ² 4p ⁴ (¹ S)nx''		4d'' ² D

*For predicted terms in the spectra of the Br I isoelectronic sequence, see Vol. 11, Introduction.

Y XII

(Ni I sequence; 28 electrons)

Z=39

Ground state 1s² 2s² 2p⁶ 3s² 3p⁶ 3d¹⁰ ¹S₀

3d¹⁰ ¹S₀ 3016800 cm⁻¹

I. P. 374 volts

Edlén has observed three lines from the 3d¹⁰—3d⁹ np and 3d¹⁰—3d⁹ nf (n=4, 5) configurations. In figure 2 of his paper on the spectra of highly ionized atoms, two lines of the former group are indicated on the photograph of vacuum spark spectra of Y observed between 50 Å and 100 Å.

From preliminary unpublished measurements he has derived provisional absolute term values from the Rydberg nf-series. The terms in the table are from his manuscript. His unit, 10³ cm⁻¹, has here been changed to cm⁻¹.

REFERENCES

- B. Edlén, *Physica* **13**, No. 9, 547 (1947).
B. Edlén, letter (July 1950). (I P (T))

Y XII

Y XII

Config.	Desig.	J	Level	Config.	Desig.	J	Level
3d ¹⁰	3d ¹⁰ ¹ S	0	0	3d ⁹ (² D)5p	5p ³ D ^o	3 2 1	2099500
3d ⁹ (² D)4p	4p ¹ P ^o	1	1374600	3d ⁹ (² D)5f	5f ³ D ^o	3 2 1	2388900
3d ⁹ (² D)4p	4p ³ D ^o	3 2 1	1386800				
3d ⁹ (² D)4f	4f ¹ P ^o	1	1994800	Y XIII (² D _{3/2})		Limit	3016800
3d ⁹ (² D)4f	4f ³ D ^o	3 2 1	2026800				
3d ⁹ (² D)5p	5p ¹ P ^o	1	2088600				

August 1950.

Y XIII

(Co I sequence; 27 electrons)

Z=39

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^9 {}^2D_{3/2}$ $3d^9 {}^2D_{3/2}$ cm⁻¹ I. P. volts

Edlén has observed three lines due to the transition $3p^6 3d^9 {}^2D - 3p^5 3d^{10} {}^2P^o$. From preliminary unpublished measurements he has furnished the provisional term values quoted in the table. In figure 2 of his paper on the spectra of highly ionized atoms, the lines are indicated on the photograph of the vacuum spark spectra of Y observed between 50 Å and 100 Å. Another group of lines from $3d^9 - 3d^8 4p$ is also indicated in the figure.

Edlén's unit, 10^3 cm⁻¹, has here been changed to cm⁻¹.

REFERENCES

B. Edlén, *Physica* **13**, No. 9, 547 (1947).

B. Edlén, letter (July 1950). (T)

Y XIII

Config.	Desig.	J	Level	Interval
$3p^6 3d^9$	$3d^9 {}^2D$	$2\frac{1}{2}$	0	-17200
		$1\frac{1}{2}$	17200	
$3p^5 3d^{10}$	$3d^{10} {}^2P^o$	$1\frac{1}{2}$	1143600	-98800
		$0\frac{1}{2}$	1242400	

July 1950.

Y XIV

(Fe I sequence; 26 electrons)

Z=39

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^8 {}^3F_4$ $3d^8 {}^3F_4$ cm⁻¹ I. P. volts

This spectrum has not been analyzed, but Edlén has observed a group of lines due to terms from the $3d^8 - 3d^7 4p$ configurations. In figure 2 of his paper on the spectra of highly ionized atoms, the lines are indicated on the photograph of vacuum spark spectra of Y observed between 50 Å and 100 Å.

The writer has assumed the ground state indicated above, by analogy with Ni III.

REFERENCE

B. Edlén, *Physica* **13**, No. 9, 547 (1947).

February 1950.

Y XV

(Mn I sequence; 25 electrons)

Z=39

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^7 {}^4F_{4/2}$ $3d^7 {}^4F_{4/2}$ cm⁻¹ I. P. volts

This spectrum has not been analyzed, but Edlén has observed a group of lines due to terms from the $3d^7 - 3d^6 4p$ configurations. In figure 2 of his paper on the spectra of highly ionized atoms, the lines are indicated on the photograph of vacuum spark spectra of Y observed between 50 Å and 100 Å.

The writer has assumed the ground state indicated above by analogy with Co III.

REFERENCE

B. Edlén, *Physica* **13**, No. 9, 547 (1947).

February 1950.

ZIRCONIUM

Zr I

40 electrons

Z=40

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^2 5s^2 {}^3F_2$ $a {}^3F_2$ 56077 cm^{-1}

I. P. 6.95 volts

The analysis is by Kiess and Mrs. Kiess, who have observed nearly 1600 lines in this spectrum between 2088.89 Å and 9276.94 Å, and classified approximately 80 percent of them. Meggers and Kiess have since extended the observations to 10738.94 Å. Many of the terms are confirmed by the Zeeman patterns, for which the observations by B. E. Moore as well as those of the authors were utilized.

Their limit is from a Rydberg representation of the $ns {}^3F$ and $ns {}^5F$ series ($n=5,6$), the former having the configuration $4d^2 5s(a {}^4F)ns$, and the latter $4d^3(b {}^4F)ns$.

The singlet, triplet, and quintet systems of terms are connected by observed intersystem combinations.

Kiess has suggested that his published configuration assignment of the $c {}^2D$ term in Zr II be changed from $4d^3$ to $4d 5s^2$, and that the term $d {}^2D$ remain unassigned. These changes have been adopted here. Consequently, the three Zr I terms having the limit term $d {}^2D$ in Zr II, $q {}^3F^\circ$, $o {}^3D^\circ$, and $t {}^3P^\circ$, do not appear in the array of observed terms following the table.

The observed g -values are those given by Sancho except for seven pairs of levels affected by g -sharing, which are taken from Table 5 of the Kiess' paper.

The detailed analysis of this complex spectrum presents another beautiful confirmation of Hund's theory of line spectra and atomic structure.

REFERENCES

- C. C. Kiess and H. K. Kiess, Bur. Std. J. Research **6**, 621, RP296 (1931). (I P) (T) (C L) (Z E) (E D)
 W. F. Meggers and C. C. Kiess, Bur. Std. J. Research **9**, 324, RP473 (1932). (C L)
 P. M. Sancho, Anal. Soc. Esp. de Fisica y Quimica (Madrid) **30**, 867 (1932). (Z E)
 J. E. Mack, Rev. Mod. Phys. **22**, No. 1, 64 (1950). (Summary hfs)

Zr I

Zr I

Config.	Desig.	J	Level	Interval	Obs. g	Config.	Desig.	J	Level	Interval	Obs. g
$4d^2 5s^2$	$a {}^3F$	2	0.00		0.66	$4d^3(a {}^2G)5s$	$a {}^3G$	3	12503.44		0.78
		3	570.41	570.41	1.06			4	12760.66	257.22	1.15
		4	1240.84	670.43	1.24			5	12772.78	12.12	1.20
$4d^2 5s^2$	$a {}^3P$	2	4186.11		1.25	$4d^2 5s^2$	$a {}^1S$	0	13141.76		
		1	4376.28	190.17	1.48						
		0	4196.85	179.43	0.00						
$4d^3(b {}^4F)5s$	$a {}^5F$	1	4870.53		0.00	$4d^3(b {}^2D)5s$	$a {}^3D$	1	14123.01		0.61
		2	5023.41	152.88	0.98			2	14348.78	225.77	1.17
		3	5249.07	225.66	1.28			3	14697.03	348.25	1.35
		4	5540.54	291.47	1.31	$4d^2 5s(a {}^4F)5p$	$z {}^5G^\circ$	2	14783.54	417.72	0.43
		5	5888.93	348.39	1.37			3	15201.26	519.10	1.09
								4	15720.36	596.60	1.12
								5	16316.96	661.33	1.28
								6	16978.29		1.36
$4d^2 5s^2$	$a {}^1D$	2	5101.68		1.25						
$4d^2 5s^2$	$a {}^1G$	4	8057.30		1.00						
$4d^3(b {}^4F)5s$	$a {}^5P$	1	10885.36		2.50	$4d^3(a {}^3H)5s$	$a {}^3H$	4	14791.28		0.77
		2	11016.65	131.29	1.82			5	14988.51	197.23	1.03
		3	11258.38	241.73	1.66			6	15119.66	131.15	1.15
$4d^3(b {}^4F)5s$	$b {}^3F$	2	11640.72		0.75	$4d^3(b {}^2F)5s$	$c {}^3F$	2	15146.48		
		3	11956.33	315.61	1.05			3	15457.40	310.92	
		4	12342.37	386.04	1.15			4	15699.86	242.46	

Zr I—Continued

Zr I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>		
$4d^3(b^4P)5p$	w^5D°	0	33349.56		0.00	$4d^2 5s(b^2G)5p$	u^3G°	3	39389.29				
		1	33444.87	95.31	1.52			4	39934.14	544.85	1.05		
		2	33632.48	187.61	1.51			5	40178.44	244.30	1.13		
		3	33912.09	279.61	1.49								
		4	34287.49	375.40	1.51								
$4d^2 5s(a^2F)5p$	u^3D°	1	33486.82		0.62	$4d^3(b^2D)5p$	r^3D°	1					
		2	33764.12	277.30	1.10			2	39766.47				
		3	34239.82	475.70	1.35			3	40346.35	579.88	1.29		
$4d^2 5s(b^2G)5p$	z^1H°	5	33839.20		1.03	$4d^3(a^2G)5p$	v^1F°	3	39803.73				
$4d^3(a^2H)5p$	y^3H°	4	34450.60		0.80	$4d^3(a^2G)5p$	y^1H°	5	39855.22		1.02		
		5	34705.90	255.30	1.02								
		6	35135.07	429.17	1.16								
$4d^3(b^4P)5p$	y^5P°	1	34617.00		2.46	$4d^2 5s(a^4F)5d$	e^5H	3					
		2	34761.52	144.52	1.42			4	39936.70	700.35	1.07		
		3	35090.90	329.38	1.66			5	40637.05	806.48			
$4d 5s^2(c^2D)5p$	w^1D°	2	34850.96		1.42			$4d 5s^2(d^2D)5p$	t^3P°	0	40536.38		
		1			1					40973.94	437.56		
$4d^2 5s(a^4F)6s$	e^5F	1	35046.95					$4d^3(b^2F)5p$	v^1D°	2	40557.65		
		2	35210.30	163.35	1.26								
		3	35476.07	265.77	1.23								
		4	35860.83	384.76	1.35								
		5	36360.20	499.37	1.40								
$4d^3(b^4P)5p$	w^3P°	0				$4d^3(b^4F)6s$	f^5F	1					
		1	35205.52					2					
		2	35456.25	250.73	1.45			3	40653.41	196.29			
$4d^2 5s(b^2G)5p$	t^3F°	2	35514.53		0.69			$4d^2 5s(a^4F)5d$	e^5G	2	40660.65		
		3	35805.63	291.10	1.00					3	40887.61	226.96	1.09
		4	36001.35	195.72	1.22	4	41179.30			291.69	1.07		
$4d^3(a^2H)5p$	z^3I°	5	35781.67		0.82	5	41538.23			358.93			
		6	36173.03	391.36	1.02	6	41940.86			402.63	1.41		
		7	36152.85	-20.18	1.13								
$4d^3(b^4P)5p$	y^5S°	2	35990.21		2.01	$4d 5s^2(c^2D)5p$	w^1P°	1	40931.60				
$4d^3(b^2D)5p$	v^3P°	0	36034.54			$4d^3(a^2H)5p$	w^1G°	4	41319.96		0.99		
		1	36489.10	454.56		$4d^3(a^2H)5p$	t^3G°	3	42102.56	169.85			
		2	37008.40	519.30	1.55			4	42272.41	562.55			
5				5	42834.96								
$4d^3(b^4P)5p$	t^3D°	1	36125.16		0.45	$4d 5s^2(c^2D)5p$	q^3D°	1	42296.80	136.85			
		2	36294.87	169.71	1.16			2	42433.65	365.55			
		3	36220.45	-74.42	1.28			3	42799.20		1.35		
$4d^2 5s(b^2G)5p$	x^1G°	4	36336.48		1.15	$4d^3(a^2H)5p$	x^1H°	5	42309.29				
$4d 5s^2(c^2D)5p$	u^3P°	0	36538.27			$4d 5s^2(d^2D)5p$	q^3F°	2	42706.00	562.24			
		1	36970.65	432.38				3	43268.24	7.76			
		2	37450.23	479.58				4	43276.00				
$4d^2 5s(b^2G)5p$	x^3H°	4	36608.41		0.82	$4d^3(b^2P)5p$	w^3S°	1	43182.96				
		5	36597.48	-10.93	1.04								
		6	36840.59	243.11	1.13			$4d^3(b^2P)5p$	s^3P°	0			
$4d 5s^2(c^2D)5p$	w^1F°	3	36759.90		0.87	$4d^3(b^2P)5p$	p^3D°	1	44882.30	134.83			
					0.87			2	45017.13				
								1	45405.30	182.32			
$4d^3(a^2G)5p$	v^3G°	3	36941.65		1.02	2	45537.62	122.67					
		4	37229.54	287.89	1.18	3	45710.29						
		5	37422.36	192.82									
$4d^3(a^2G)5p$	s^3F°	2	37123.42		0.71	$4d^2 5p^2$	f^5G	2	45798.48				
		3	37468.87	345.45	1.11			3	46195.15	396.67			
		4	37920.96	452.09	1.26			4	46641.42	446.27			
								5	47134.50	493.08			
$4d^2 5s(a^4F)6s$	e^3F	2	37459.60		1.02	6	47698.29	563.79					
		3	37701.08	241.48	1.25								
		4	38101.09	400.01		$4d^3(b^2F)5p$	u^1F°	3	46328.16				
$4d^3(b^2F)5p$	s^3D°	1	38270.81			$4d 5s^2(d^2D)5p$	o^3D°	1	47765.56	368.08			
		2	38326.72	55.91	1.41			2	48133.64	579.80			
		3	38435.88	109.16	1.00			3	48713.44				
$4d^3(a^2H)5p$	z^1I°	6	38475.82										
$4d 5s^2(c^2D)5p$	r^3F°	2	38566.00			$4d^2 5s(a^2S)5p$	v^1P°	1	51899.40				
		3	38881.80	315.80	1.26								
		4	39174.44	292.64									
					Zr II ($^4F_{1/2}$)	Limit		56077					

Zr I OBSERVED TERMS *

Config. $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 +$	Observed Terms		$nd (n \geq 5)$
$4d^2 5s^2$	$\left\{ \begin{array}{l} a^3P \\ a^1S \end{array} \right.$	$a^3F \quad a^1G$	
$4d^4$	a^5D		
$4d^2 5p^2$		f^5G	
		$ns (n \geq 5)$	$np (n \geq 5)$
$4d^2 5s(a^4F)nx$	$\left\{ \begin{array}{l} e^5F \\ e^3F \end{array} \right.$		$z^5D^\circ \quad z^5F^\circ \quad z^5G^\circ$ $z^3D^\circ \quad z^3F^\circ \quad z^3G^\circ$
$4d^2(b^4F)nx$	$\left\{ \begin{array}{l} a, f^5F \\ b^3F \end{array} \right.$		$x^5D^\circ \quad y^5F^\circ \quad y^5G^\circ$ $w^3D^\circ \quad y^3F^\circ \quad y^3G^\circ$
$4d^2 5s(a^2D)nx$			$x^3P^\circ \quad v^3D^\circ \quad x^3F^\circ$ $z^1P^\circ \quad z^1D^\circ \quad z^1F^\circ$
$4d^2 5s(a^2P)nx$			$z^3S^\circ \quad x^3D^\circ \quad x^3G^\circ$ $y^1P^\circ \quad y^1D^\circ \quad z^1G^\circ$
$4d^2 5s(a^2F)nx$			$u^3D^\circ \quad w^3F^\circ \quad x^3G^\circ$ $y^1F^\circ \quad y^1G^\circ \quad z^1G^\circ$
$4d^2 5s(a^4P)nx$	$\left\{ \begin{array}{l} z^5S^\circ \\ y^3S^\circ \end{array} \right.$		$z^5D^\circ \quad y^5D^\circ \quad v^3G^\circ$ $y^3D^\circ \quad y^3F^\circ \quad y^1G^\circ$
$4d^2(a^2G)nx$		a^3G b^1G	$s^3F^\circ \quad v^3G^\circ \quad z^3H^\circ$ $v^1F^\circ \quad y^1G^\circ \quad y^1H^\circ$
$4d^2(b^4P)nx$	$\left\{ \begin{array}{l} a^5P \\ b^3P \end{array} \right.$		$s^3F^\circ \quad v^3G^\circ \quad z^3H^\circ$ $v^1F^\circ \quad y^1G^\circ \quad y^1H^\circ$
$4d^2(a^2H)nx$		a^3H a^1H	$l^3G^\circ \quad y^3H^\circ \quad z^3I^\circ$ $w^1G^\circ \quad x^1H^\circ \quad z^1I^\circ$
$4d^2(b^2D)nx$	$\left\{ \begin{array}{l} a^3D \\ b^1D \end{array} \right.$		$l^3G^\circ \quad y^3H^\circ \quad z^3I^\circ$ $w^1G^\circ \quad x^1H^\circ \quad z^1I^\circ$
$4d^2 5s(b^2G)nx$			$l^3F^\circ \quad u^3G^\circ \quad x^3H^\circ$ $x^1G^\circ \quad x^1H^\circ \quad z^1H^\circ$
$4d 5s^2(c^2D)nx$			$u^3P^\circ \quad q^3D^\circ \quad r^3F^\circ$ $x^1P^\circ \quad w^1D^\circ \quad w^1F^\circ$
$4d^2(b^2F)nx$		c^3F	$u^3P^\circ \quad q^3D^\circ \quad r^3F^\circ$ $w^1P^\circ \quad w^1D^\circ \quad w^1F^\circ$
$4d^2(b^2P)nx$			$s^3D^\circ \quad v^3F^\circ \quad w^3G^\circ$ $v^1D^\circ \quad u^1F^\circ \quad w^1G^\circ$
$4d^2 5s(a^2S)nx$	c^3P		$w^3S^\circ \quad s^3P^\circ \quad p^3D^\circ$ v^1P°

* For predicted terms in the spectra of the Zr I isoelectronic sequence, see Vol. II, Introduction.

Zr II

(Y I sequence; 39 electrons)

Z=40

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^2 5s^4 F_{1/2}$ $a^4 F_{1/2}$ 113175 cm^{-1}

I. P. 14.03 volts

The analysis is by Kiess and Mrs. Kiess, whose line list extends from 1743.53 to 6787.15 Å, includes 735 classified lines, and gives observed and theoretical Zeeman patterns. The terms of different multiplicities are connected by observed intersystem combinations.

Their value of the limit is based on the four pairs of series terms involving 5s and 6s electrons, and is derived by using a Rydberg formula.

C. C. Kiess has recently revised the configuration assignments of $b^2 D$, $c^2 D$, and $d^2 D$ as given in the table. On theoretical grounds, Ufford suggests that these configurations be interchanged as follows:

Term	C. C. Kiess	Ufford
$b^2 D$	$4d^3$	$4d 5s^2$
$c^2 D$	$4d 5s^2$	$4d^3$
$d^2 D$	Unassigned	Unassigned

The observed g -values are from three sources: those for $z^4 F_{1/2}^o$ and $y^2 F_{23/2}^o$ are from the Kiess paper. All other three-place entries are from unpublished material by Miss Weeks furnished especially for inclusion here. These are from films of spectrograms taken with the Bitter magnet at the Massachusetts Institute of Technology. The rest are quoted from the paper by Sancho.

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- C. C. Kiess and H. K. Kiess, Bur. Std. J. Research **5**, 1205, RP255 (1930). (I P) (T) (C L) (Z E) (G D)
 P. M. Sancho, An. Soc. Española Fis. y Quim. **30**, 874 (1932). (Z E)
 C. W. Ufford, Phys. Rev. **44**, 732 (1933).
 D. W. Weeks, unpublished material (February 1950). (Z E)

Zr II

Zr II

Config.	Desig.	J	Level	Interval	Obs. g	Config.	Desig.	J	Level	Interval	Obs. g			
$4d^2(a^3 F)5s$	$a^4 F$	$1\frac{1}{2}$	0. 00	314. 67	0. 398	$4d^3$	$a^2 G$	$3\frac{1}{2}$	7837. 74	315. 06	0. 887			
		$2\frac{1}{2}$	314. 67		1. 023			$4\frac{1}{2}$	8152. 80		1. 107			
		$3\frac{1}{2}$	763. 44		1. 235			$4d^3$	$b^4 P$		$0\frac{1}{2}$	9553. 10	189. 70	2. 649
		$4\frac{1}{2}$	1322. 91		1. 324						$1\frac{1}{2}$	9742. 80	225. 85	1. 721
$4d^3$	$b^4 F$	$1\frac{1}{2}$	2572. 21	322. 84	0. 413	$4d^3$	$a^2 H$	$2\frac{1}{2}$	9968. 65	375. 20	1. 593			
		$2\frac{1}{2}$	2895. 05		1. 025			$4\frac{1}{2}$	11984. 46		0. 910			
		$3\frac{1}{2}$	3299. 64		1. 227			$5\frac{1}{2}$	12359. 66		1. 091			
		$4\frac{1}{2}$	3757. 66		1. 326			$4d^3$	$b^2 D$		$1\frac{1}{2}$	13428. 50	734. 40	0. 800
$2\frac{1}{2}$	4505. 50	1. 172	$2\frac{1}{2}$	14162. 90	1. 209									
$4d^2(a^1 D)5s$	$a^2 D$	$1\frac{1}{2}$	4248. 30	257. 20	0. 812	$4d^2(^1 G)5s$	$b^2 G$	$3\frac{1}{2}$	14059. 76	130. 69	0. 890			
		$2\frac{1}{2}$	4505. 50		1. 172			$4\frac{1}{2}$	14190. 45		1. 103			
$4d^2(a^3 P)5s$	$a^2 P$	$0\frac{1}{2}$	5724. 38	387. 32	0. 690	$4d 5s^2$	$c^2 D$	$1\frac{1}{2}$	14298. 64	434. 73	0. 807			
		$1\frac{1}{2}$	6111. 70		1. 304			$2\frac{1}{2}$	14733. 37		1. 188			
$4d^2(a^3 F)5s$	$a^2 F$	$2\frac{1}{2}$	5752. 92	714. 69	0. 883	$4d^2(^1 G)5s$	$d^2 D$	$1\frac{1}{2}$	17614. 00	782. 54	1. 188			
		$3\frac{1}{2}$	6467. 61		1. 144			$2\frac{1}{2}$	18396. 54					
$4d^2(a^3 P)5s$	$a^4 P$	$0\frac{1}{2}$	7512. 67	223. 35	2. 656	$4d^2(^1 G)5s$	$d^2 D$	$1\frac{1}{2}$	17614. 00	782. 54	1. 188			
		$1\frac{1}{2}$	7736. 02		1. 720			$2\frac{1}{2}$	18396. 54					
		$2\frac{1}{2}$	8058. 16		322. 14			1. 585						

Zr II—Continued

Zr II—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>
4d ³	b ² F	3½ 2½	19433. 24 19514. 84	81. 60	1. 153 0. 855	4d ² (a ¹ D)5p	x ² F°	3½ 2½	42504. 11 42860. 72	-356. 61	1. 134 0. 887
4d ³	b ² P	0½ 1½	19613. 54 20080. 30	466. 76	0. 514 1. 326	4d ² (a ³ P)5p	y ⁴ P°	0½ 1½ 2½	42789. 24 42893. 54 43202. 45	104. 30 308. 91	2. 632 1. 710 1. 561
4d ² (¹ S)5s	a ² S	0½	25201. 57		1. 990	4d 5s(a ¹ D)5p	w ² D°	1½ 2½	45054. 87 45186. 05	131. 18	0. 870 1. 226
4d ² (a ³ F)5p	z ⁴ G°	2½ 3½ 4½ 5½	27983. 83 28909. 04 29839. 87 30795. 74	925. 21 930. 83 955. 87	0. 664 0. 998 1. 164 1. 275	4d ² (¹ S)5p	x ² P°	1½ 0½	45568. 21 45944. 00	-375. 79	1. 14 0. 724
4d ² (a ³ F)5p	z ² F°	2½ 3½	29504. 97 30561. 75	1056. 78	0. 841 1. 132	4d 5s(a ¹ D)5p	w ² F°	2½ 3½	47881. 88 48344. 91	463. 03	0. 871 1. 142
4d ² (a ³ F)5p	z ⁴ F°	1½ 2½ 3½ 4½	29777. 60 30551. 48 31249. 28 31866. 49	773. 88 697. 80 617. 21	0. 700 1. 046 1. 238 1. 321	4d 5s(a ¹ D)5p	w ² P°	0½ 1½	52585. 80 52876. 80	291. 00	0. 659 1. 318
4d ² (a ³ F)5p	z ² D°	1½ 2½	30435. 38 31160. 04	724. 66	0. 589 1. 117	4d 5s(a ³ D)5p	v ² D°	1½ 2½	55835. 53 56569. 44	733. 91	1. 16
4d ² (a ³ F)5p	z ⁴ D°	0½ 1½ 2½ 3½	31981. 25 32256. 71 32614. 71 32899. 46	275. 46 358. 00 284. 75	0. 016 1. 166 1. 342 1. 408	4d 5s(a ³ D)5p	v ² P°	0½ 1½	60814. 50 61861. 90	1047. 40	
4d ² (a ³ P)5p	y ² D°	1½ 2½	32983. 73 33419. 45	435. 72	0. 810 1. 195	4d ² (a ³ F)6s	e ⁴ F	1½ 2½ 3½ 4½	63602. 64 63868. 45 64368. 28 64901. 71	265. 81 499. 83 533. 43	1. 11 1. 205 1. 274
4d ² (a ³ F)5p	z ² G°	3½ 4½	34485. 42 35185. 64	700. 22	0. 889 1. 109	4d ² (a ³ F)6s	e ² F	2½ 3½	65872. 41 66192. 68	320. 27	1. 18
4d ² (a ³ P)5p	z ² S°	0½	34810. 03		1. 956	4d ² (a ¹ D)6s	e ² D	1½ 2½	66686. 25 66868. 35	182. 10	1. 14
4d ² (a ¹ D)5p	z ² P°	1½ 0½	35914. 81 36196. 57	-281. 76	1. 340 0. 610	4d ² (¹ G)6s	e ² G	3½ 4½	69116. 70 69283. 38	166. 68	
4d ² (a ³ P)5p	y ⁴ D°	0½ 1½ 2½ 3½	36237. 04 36638. 50 37171. 22 38041. 49	401. 46 532. 72 870. 27	0. 144 1. 038 1. 140 1. 306	4d ² (a ³ F)5d	f ² F	2½ 3½	73852. 95 74496. 80	643. 85	
4d 5s(a ³ D)5p	y ⁴ F°	1½ 2½ 3½ 4½	36451. 79 36869. 00 37429. 76 38644. 12	417. 21 560. 76 1214. 36	0. 579 1. 091 1. 266 1. 321	4d ² (a ³ F)5d	f ⁴ F	1½ 2½ 3½ 4½	74611. 28 75343. 57 76009. 05 76593. 58	732. 29 665. 48 584. 53	
4d ² (¹ G)5p	y ² F°	2½ 3½	37346. 31 37787. 59	441. 28	0. 975 1. 212	4d ² (¹ G)5d	e ² I	5½ 6½	76395. 50 76838. 70	443. 20	0. 999 1. 081
4d ² (a ³ P)5p	z ⁴ S°	1½	37681. 75		1. 908	4d ² (¹ G)5d	e ² H	4½ 5½	77743. 00 78280. 90	537. 90	1. 073
4d 5s(a ³ D)5p	z ⁴ P°	0½ 1½ 2½	38063. 40 38133. 50 38482. 64	70. 10 349. 14	2. 448 1. 734 1. 606	4d ² (a ³ F)5d	e ⁴ H	3½ 4½ 5½ 6½	78577. 85 78847. 67 79198. 35 79280. 30	269. 82 350. 68 81. 95	
4d 5s(a ³ D)5p	x ⁴ D°	0½ 1½ 2½ 3½	38934. 37 39192. 35 39640. 08 40238. 55	257. 98 447. 73 598. 47	0. 055 1. 209 1. 370 1. 408	4d ² (¹ G)5d	f ² G	3½ 4½	79624. 60 80311. 54	686. 94	
4d ² (a ³ P)5p	y ² P°	0½ 1½	40727. 26 41337. 36	610. 10	0. 677 1. 326	4d ² (a ³ F)5d	g ² G	3½ 4½	83221. 45 83547. 45	326. 00	
4d ² (¹ G)5p	y ² G°	3½ 4½	40852. 74 40878. 25	25. 51	0. 915 1. 083	4d ² (a ³ F)5d	f ² H	4½ 5½	90986. 50 91737. 40	750. 90	
4d ² (a ¹ D)5p	x ² D°	1½ 2½	41467. 72 41676. 82	209. 10	0. 821 1. 184						
4d ² (¹ G)5p	z ² H°	4½ 5½	41738. 21 42409. 93	671. 72	0. 954 1. 080	Zr III (³ F ₂)	Limit		113175		

Zr II OBSERVED TERMS*

Config.	Observed Terms		
$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 +$			
$4d 5s^2$	c^2D		
$4d^3$	b^4P b^2D	b^4F b^2F	a^2G a^2H
	$ns (n \geq 5)$		$np (n \geq 5)$
$4d^2(a^3F)nx$	a, e^4F a, e^2F		z^4D^0 z^4F^0 z^4G^0 z^2D^0 z^2F^0 z^2G^0
$4d^2(a^1D)nx$	a, e^2D		z^2P^0 x^2D^0 x^2F^0
$4d^2(a^3P)nx$	a^4P a^2P		z^4S^0 y^4P^0 y^4D^0 z^3S^0 y^2P^0 y^2D^0
$4d^2(1G)nx$		b, e^2G	y^2F^0 y^2G^0 z^3H^0
$4d^2(1S)nx$	a^3S		x^2P^0
$4d 5s(a^3D)nx$			z^4P^0 x^4D^0 y^4F^0 v^2P^0 v^2D^0 v^2F^0
$4d 5s(a^1D)nx$			w^2P^0 w^2D^0 w^2F^0
			f^2G e^2H e^2I
			f^4F f^2G e^4H f^2F g^2G f^2H
			$nd (n \geq 4)$

*For predicted terms of the Y I isoelectronic sequence, see Vol. II, Introduction.

(Sr I sequence; 38 electrons)

Z=40

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^2 {}^3F_2$ $a {}^3F_2$ 200000 \pm cm⁻¹

I. P. 24.8 volts

Kiess has revised the earlier work by himself and Lang, especially for inclusion here, and he has further analysis in progress. The observations extend from 732 Å to 3497 Å, and the total number of classified lines is about 200. Observed intersystem combinations connect the singlet and triplet series of terms.

His value of the limit quoted here is provisional, since it has been derived by the application of a Rydberg formula to the $nd {}^3F$ and $nd {}^1D$ series ($n=4,5$).

The observed g -values are from plates and films furnished by Harrison at the Massachusetts Institute of Technology. Kiess has examined the plates and Miss Weeks the films; both have contributed to the values quoted here.

REFERENCES

C. C. Kiess and R. J. Lang, Bur. Std. J. Research 5, 309, RP 202 (1930). (I P) (T) (C L) (G D)

D. W. Weeks, unpublished material (March 1951). (Z E)

C. C. Kiess, unpublished material (1951). (I P) (T) (C L) (Z E)

Zr III

Zr III

Config.	Desig.	J	Level	Interval	Obs. g	Config.	Desig.	J	Level	Interval	Obs. g
$4d^2$	$a {}^3F$	2	0. 00			$4d(2D)5d$	$e {}^3D$	1	105006. 9		
		3	681. 0	681. 0				2	105320. 9	314. 0	
		4	1486. 4	805. 4				3	106300. 8	979. 9	
$4d^2$	$a {}^1D$	2	5741. 55			$4d(2D)5d$	$e {}^1P$	1	105550. 2		
$4d^2$	$a {}^3P$	0	8062. 07			$4d(2D)5d$	$e {}^1D$	2	106499. 6		
		1	8325. 65	263. 58				2	107253. 0	567. 6	
		2	8838. 21	512. 56				3	107820. 6	493. 8	
$4d^2$	$a {}^1G$	4	11048. 70			$4d(2D)5d$	$e {}^3F$	2	107253. 0		
$4d^2$	$a {}^1S$	0	13832. 0?			$4d(2D)5d$	$e {}^1G$	4	107456. 2		
$4d(2D)5s$	$a {}^3D$	1	18398. 87			$4d(2D)5d$	$e {}^1S$	0	107650. 5		
		2	18802. 79	403. 92	0. 511			1	109073. 3		
		3	19533. 35	730. 56	1. 173			2	109268. 7	-329. 9	
$4d(2D)5s$	$b {}^1D$	2	25066. 25		1. 000	$4d(2D)5d$	$e {}^3P$	2	109268. 7		
$5s^2$	$b {}^1S$	0	36257. 8??					0	109598. 6		
$4d(2D)5p$	$z {}^1D^\circ$	2	53647. 21		0. 927	$4d(2D)4f$	$y {}^3F^\circ$	4	117813		-1713
$4d(2D)5p$	$z {}^3F^\circ$	2	55555. 63			$4d(2D)4f$	$z {}^3G^\circ$	3	119526		-1892
		3	56075. 33	519. 70	0. 789			4	121923	677	
		4	57681. 40	1606. 07	1. 112			5	123500	900	
$4d(2D)5p$	$z {}^3D^\circ$	1	55614. 42			$4d(2D)4f$	$x {}^3P^\circ$	0	123390		203
		2	56435. 65	821. 23	0. 519			1	123593	362	
		3	57346. 83	911. 18	1. 119			2	123955		
$4d(2D)5p$	$z {}^3P^\circ$	0	59944. 79			$4d(2D)4f$	$y {}^1F^\circ$	3	127378		
$4d(2D)5p$	$z {}^1P^\circ$	1	62115. 66		1. 044	$4d(2D)4f$	$y {}^1D^\circ$	2	128877		
$4d(2D)5p$	$z {}^1F^\circ$	3	62589. 13		1. 000	$4d(2D)4f$	$y {}^3D^\circ$	1	131010		499
$5s(2S)5p$	$y {}^3P^\circ$	0	79436. 7			$4d(2D)4f$	$x {}^1P^\circ$	1	131509		678
		1	80103. 0	666. 3				2	132187		
		2	81553. 6	1450. 6				3	132187		
$5s(2S)5p$	$y {}^1P^\circ$	1	86512. 3??			$4d(2D)4f$	$z {}^1G^\circ$	4	134975?		
$4d(2D)5d$	$e {}^1F$	3	103581. 5			$4d(2D)4f$	$z {}^1H^\circ$	5	138517?		
$4d(2D)5d$	$e {}^3G$	3	104633. 4			-----	-----	-----	-----	-----	-----
		4	105189. 6	556. 2							
		5	105720. 1?	530. 5?							
						Zr VI ($2D_{1/2}$)	Limit		200000		

Zr III OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 3d ¹⁰ 4s ² 4p ⁶ +	Observed Terms	
4d ²	{ a ³ P	a ³ F
5s ²	{ a ¹ S	a ¹ D
		a ¹ G
	b ¹ S?	
	ns (n ≥ 5)	np (n ≥ 5)
4d(2D)nx	{ a ³ D	z ³ P ^o z ³ D ^o z ³ F ^o
5s(2S)nx	{ b ¹ D	z ¹ P ^o z ¹ D ^o z ¹ F ^o
		y ³ P ^o
		y ¹ P ^o ?
	nd (n ≥ 5)	nf (n ≥ 4)
4d(2D)nx	{ e ³ S	x ³ P ^o y ³ D ^o y ³ F ^o z ³ G ^o
	{ e ³ P	x ¹ P ^o y ¹ D ^o y ¹ F ^o z ¹ G ^o
	{ e ³ D	
	{ e ³ F	
	{ e ³ G	
	{ e ¹ S	
	{ e ¹ P	
	{ e ¹ D	
	{ e ¹ F	
	{ e ¹ G	
		z ¹ H ^o

*For predicted terms in the spectra of the Sr I isoelectronic sequence, see Vol. II, Introduction.

Zr IV

(Rb I sequence; 37 electrons)

Z=40

Ground state 1s² 2s² 2p⁶ 3s² 3p⁶ 3d¹⁰ 4s² 4p⁶ 4d²D_{1/2}

4d²D_{1/2} 274067 cm⁻¹

I. P. 33.97 volts

Fourteen lines have been classified as transitions among seven terms. The analysis is from the paper by Kiess and Lang who reobserved the spectrum from 628.66 Å to 2286.66 Å, and extended the earlier work of Bowen and Millikan and of Gibbs and White.

From a Ritz formula applied to the ns²S series (n=5 to 7) Kiess and Lang have determined the limit quoted here.

REFERENCE

C. C. Kiess and R. J. Lang, Bur. Std. J. Research 5, 307, RP202 (1930). (I P) (T) (C L) (G D)

Zr IV

Config.	Desig.	J	Level	Interval
4p ⁶ (1S)4d	4d ² D	1½ 2½	0 1250	1250
4p ⁶ (1S)5s	5s ² S	0½	38258	
4p ⁶ (1S)5p	5p ² P ^o	0½ 1½	81976 84462	2486
4p ⁶ (1S)5d	5d ² D	1½ 2½	146650 147001	351
4p ⁶ (1S)6s	6s ² S	0½	152509	
4p ⁶ (1S)4f	4f ² F ^o	2½ 3½	159068 159088	20
4p ⁶ (1S)7s	7s ² S	0½	198840	
-----	-----	-----	-----	-----
Zr v (1S ₀)	Limit	---	274067	

March 1951.

Zr VI

(Br I sequence; 35 electrons)

Z=40

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^5 \ ^2P_{1\frac{1}{2}}^{\circ}$ $4p^5 \ ^2P_{1\frac{1}{2}}^{\circ}$ 798000 cm^{-1}

I. P. 99 volts

The analysis is incomplete. Paul and Rense have classified 46 lines between 236 Å and 568 Å as due to transitions from the ground term to 15 higher terms.

They have estimated the limit by extrapolation of isoelectronic sequence data.

REFERENCE

F. W. Paul and W. A. Rense, Phys. Rev. **56**, 1110 (1939). (I P) (T) (C L)

Zr VI

Zr VI

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$4s^2 4p^5$	$4p^5 \ ^2P^{\circ}$	$1\frac{1}{2}$ $0\frac{1}{2}$	0 15600	-15600	$4s^2 4p^4(^3P)4d$	$4d \ ^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	319496	
$4s 4p^6$	$4p^6 \ ^2S$	$0\frac{1}{2}$	191570		$4s^2 4p^4(^1D)4d$	$4d' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	335527 346344	-10817
$4s^2 4p^4(^3P)4d$	$4d \ ^4D$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	241099 246007 248938	-4908 -2931	$4s^2 4p^4(^1D)4d$	$4d' \ ^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	336345	
$4s^2 4p^4(^3P)4d$	$4d \ ^4F$	$4\frac{1}{2}$ $3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$	288606 288680	-74	$4s^2 4p^4(^1D)4d$	$4d' \ ^2P$	$0\frac{1}{2}$ $1\frac{1}{2}$	355280 355604	324
$4s^2 4p^4(^3P)4d$	$4d \ ^4P$	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	289323 291660 295919	2337 4259	$4s^2 4p^4(^3P)5s$	$5s \ ^2P$	$1\frac{1}{2}$ $0\frac{1}{2}$	369716 379780	-10064
$4s^2 4p^4(^3P)4d$	$4d \ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	296155 306483	10328	$4s^2 4p^4(^1D)5s$	$5s' \ ^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	380383 384790	-4407
$4s^2 4p^4(^3P)4d$	$4d \ ^2P$	$0\frac{1}{2}$ $1\frac{1}{2}$	298923 299453	530	$4s^2 4p^4(^1S)5s$	$5s'' \ ^2S$	$0\frac{1}{2}$	423215	
$4s^2 4p^4(^3P)5s$	$5s \ ^4P$	$2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	314483 334692 346616	-20209 -11924	Zr VII (3P_2)	Limit		798000	

February 1951.

Zr VI OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 3d ¹⁰ +	Observed Terms				
4s ² 4p ⁵	4p ⁵ ² P ^o				
4s 4p ⁶	4p ⁶ ² S				
	<i>ns</i> (<i>n</i> ≥ 5)	<i>nd</i> (<i>n</i> ≥ 4)			
4s ² 4p ⁴ (³ P) <i>nx</i>	{ 5s ⁴ P 5s ² P	4d ⁴ P	4d ⁴ D	4d ⁴ F	
		4d ² P	4d ² D	4d ² F	
4s ² 4p ⁴ (¹ D) <i>nx</i> '		5s' ² D	4d' ² P	4d' ² D	4d' ² F
4s ² 4p ⁴ (¹ S) <i>nx</i> ''	5s'' ² S		4d'' ² D		

*For predicted terms in the spectra of the Br I isoelectronic sequence, see Vol. II, Introduction.

Zr XIV

(Co I sequence; 27 electrons)

Z=40

Ground state 1s² 2s² 2p⁶ 3s² 3p⁶ 3d⁹ ²D_{3/2}

3d⁹ ²D_{3/2} cm⁻¹

I. P. volts

Edlén has observed three lines due to the transition 3p⁶ 3d⁹ ²D—3p⁵ 3d¹⁰ ²P^o. From preliminary unpublished measurements he has furnished the provisional term values quoted in the table. In figure 4 of his paper on the spectra of highly ionized atoms, the observed wave numbers are plotted against atomic numbers for this combination in the Co I-like spectra Rb XI to Mo XVI, and compared with the closely related X-ray transition of the same elements.

Edlén's unit, 10³ cm⁻¹, has here been changed to cm⁻¹.

REFERENCES

- B. Edlén, *Physica* **13**, No. 9, 550 (1947).
 B. Edlén, letter (July 1950). (T)

Zr XIV

Config.	Desig.	<i>J</i>	Level	Interval
3p ⁶ 3d ⁹	3d ⁹ ² D	2½ 1½	0 20300	-20300
3p ⁵ 3d ¹⁰	3d ¹⁰ ² P ^o	1½ 0½	1201800 1314500	-112700

July 1950.

NIOBIUM

Nb I

41 electrons

Z=41

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^4 5s^1 {}^6D_{0\frac{1}{2}}$ $a {}^6D_{0\frac{1}{2}}$ 54600 cm^{-1}

I. P. 6.77 volts

Humphreys and Meggers have observed the Nb I spectrum from 2161.54 Å to 10920.7 Å. Of the 3313 lines recorded in this range, they have classified 2836 as combinations among 364 atomic energy levels "representing 58 doublet, 55 quartet, and 13 sextet spectral terms." The systems of terms of different multiplicity are connected by observed intersystem combinations. Zeeman patterns measured for 911 Nb I lines have been utilized in this analysis.

The limit is derived from four two-member series, namely: $4d^4 ns({}^6D)$ and $4d^3 5s np({}^4D^{\circ} F^{\circ} G^{\circ})$ ($n=5,6$), by means of a Rydberg formula.

This element is familiarly known as Columbium (Cb). The International Union of Chemistry has recently adopted Niobium as the official name of element 41. In compliance with this agreement, the revised name and symbol are adopted here.

REFERENCES

- C. J. Humphreys and W. F. Meggers, J. Research Nat. Bur. Std. **34**, 515, RP1656 (1945).
J. E. Mack, Rev. Mod. Phys. **22**, No. 1, 64 (1950). (Summary hfs)

Nb I

Nb I

Config.	Desig.	J	Level	Interval	Obs. g	Config.	Desig.	J	Level	Interval	Obs. g
$4d^4(a {}^5D)5s$	$a {}^6D$	$0\frac{1}{2}$	0. 00	154. 19	3. 323	$4d^5$	$a {}^6S$	$2\frac{1}{2}$	11344. 70		1. 99
		$1\frac{1}{2}$	154. 19	237. 80	1. 863	$4d^4(a {}^3G)5s$	$a {}^4G$	$2\frac{1}{2}$	12018. 25	118. 61	0. 742
		$2\frac{1}{2}$	391. 99	303. 26	1. 652			$3\frac{1}{2}$	12136. 86	220. 84	1. 081
		$3\frac{1}{2}$	695. 25	355. 01	1. 582			$4\frac{1}{2}$	12357. 70	654. 50	1. 23
$4\frac{1}{2}$	1050. 26		1. 549	$5\frac{1}{2}$	13012. 20				1. 26		
$4d^3 5s^2$	$a {}^4F$	$1\frac{1}{2}$	1142. 79	444. 11	0. 402	$4d^3 5s^2$	$a {}^2H$	$4\frac{1}{2}$	12102. 12	400. 85	0. 93
		$2\frac{1}{2}$	1586. 90	567. 21	1. 029			$5\frac{1}{2}$	12502. 97		1. 10
		$3\frac{1}{2}$	2154. 11	651. 25	1. 235	$4d^4(b {}^3F)5s$	$b {}^4F$	$1\frac{1}{2}$	12288. 25	403. 87	0. 402
		$4\frac{1}{2}$	2805. 36		1. 330			$2\frac{1}{2}$	12692. 12	290. 26	0. 852
$4d^3 5s^2$	$a {}^4P$	$0\frac{1}{2}$	4998. 17	299. 75	2. 650	$4d^4(a {}^3P)5s$	$b {}^4P$	$3\frac{1}{2}$	12982. 38	163. 33	1. 120
		$1\frac{1}{2}$	5297. 92	667. 53	1. 721			$4\frac{1}{2}$	13145. 71		1. 224
		$2\frac{1}{2}$	5965. 45		1. 596						
$4d^4(a {}^5D)5s$	$a {}^4D$	$0\frac{1}{2}$	8410. 90	294. 42	0. 06	$4d^3 5s^2$	$a {}^2F$	$2\frac{1}{2}$	13404. 77	110. 43	0. 860
		$1\frac{1}{2}$	8705. 32	337. 82	1. 197			$3\frac{1}{2}$	13515. 20		1. 130
		$2\frac{1}{2}$	9043. 14	454. 38	1. 360	$4d^4(a {}^3P)5s$	$b {}^4P$	$0\frac{1}{2}$	13629. 15	592. 15	2. 64
		$3\frac{1}{2}$	9497. 52		1. 420			$1\frac{1}{2}$	14211. 30	687. 96	1. 71
$4d^3 5s^2$	$a {}^2G$	$3\frac{1}{2}$	8827. 00	501. 88	0. 885	$4d^4(a {}^3D)5s$	$b {}^4D$	$2\frac{1}{2}$	14899. 26		1. 54
		$4\frac{1}{2}$	9328. 88		1. 103			$3\frac{1}{2}$	15282. 35	-184. 73	1. 43
$4d^3 5s^2$	$a {}^2D$	$1\frac{1}{2}$	9439. 08	798. 43	0. 953	$4d^3 5s^2$	$z {}^6G^{\circ}$	$2\frac{1}{2}$	15467. 08	27. 83	1. 42
		$2\frac{1}{2}$	10237. 51		1. 206			$1\frac{1}{2}$	15439. 25	-21. 52	1. 21
$4d^3 5s^2$	$a {}^2P$	$0\frac{1}{2}$	10126. 06	1192. 03	0. 66	$4d^3 5s(a {}^5F)5p$	$z {}^6G^{\circ}$	$0\frac{1}{2}$	15460. 77		0. 04
		$1\frac{1}{2}$	11318. 09		1. 175			$1\frac{1}{2}$	16672. 00	309. 01	0. 00
$4d^4(a {}^3H)5s$	$a {}^4H$	$2\frac{1}{2}$	11318. 09		1. 175	$4d^3 5s(a {}^5F)5p$	$z {}^6G^{\circ}$	$2\frac{1}{2}$	16981. 01	322. 95	
		$3\frac{1}{2}$	10922. 74	121. 34	0. 690			$3\frac{1}{2}$	17303. 96	633. 30	
		$4\frac{1}{2}$	11044. 08	203. 80	0. 984			$4\frac{1}{2}$	17937. 26	497. 88	
		$5\frac{1}{2}$	11247. 88	276. 77	1. 12			$5\frac{1}{2}$	18435. 14	441. 32	
		$6\frac{1}{2}$	11524. 65		1. 22			$6\frac{1}{2}$	18876. 46		

Nb I—Continued

Nb I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>			
$4d^4(a^3H)5s$	b^2H	$4\frac{1}{2}$	16828. 52	647. 70	1. 04	$4d^4(a^5D)5p$	y^4D°	$0\frac{1}{2}$	26717. 73	219. 13	1. 141			
		$5\frac{1}{2}$	17476. 22		1. 01			$1\frac{1}{2}$	26936. 86		1. 292			
$4d^4(a^3G)5s$	b^2G	$3\frac{1}{2}$	16918. 78	1117. 19	0. 88	$4d^4(a^3H)5p$	z^2G°	$2\frac{1}{2}$	27359. 70	422. 84	1. 32			
		$4\frac{1}{2}$	18035. 97		0. 97			$3\frac{1}{2}$	27596. 74		1. 422			
$4d^3 5s(a^3F)5p$	z^6F°	$5\frac{1}{2}$	18332. 04	245. 46	—0. 373	$4d^3 5s(c^3P)5p$	y^4P°	$3\frac{1}{2}$	26896. 68	283. 63	1. 013			
		$0\frac{1}{2}$	18791. 09		1. 145			$4\frac{1}{2}$	27331. 80		1. 078			
		$1\frac{1}{2}$	19036. 55		1. 35			276°	$0\frac{1}{2}$		27498. 94	2. 467		
		$2\frac{1}{2}$	19427. 90		1. 39				$1\frac{1}{2}$		27782. 57	1. 660		
		$3\frac{1}{2}$	19916. 69		1. 44				$2\frac{1}{2}$		28445. 33	1. 606		
$4\frac{1}{2}$	20432. 11	3. 01	$4d^3 5s(c^3P)5p$	x^4D°	$2\frac{1}{2}$	27614. 10	1. 370							
$5\frac{1}{2}$		1. 72			$0\frac{1}{2}$	27666. 46	412. 63	0. 222						
		228. 60			$1\frac{1}{2}$	28079. 09	470. 33	1. 443						
$4d^3 5s(a^3F)5p$	z^4D°	$2\frac{1}{2}$	19993. 78	141. 22	1. 56	$4d^3 5s(a^3F)5p$	z^2F°	$2\frac{1}{2}$	28549. 42	737. 92	1. 472			
		$3\frac{1}{2}$	20315. 74		1. 55			$3\frac{1}{2}$	28209. 42		1. 241			
		$4\frac{1}{2}$	20733. 88		1. 54			$3\frac{1}{2}$	27797. 44		1. 16			
									28535. 36		1. 12			
$4d^3 5s(a^3F)5p$	z^4D°	$0\frac{1}{2}$	20107. 36	276. 26	—0. 02	$4d^3 5s(a^3F)5p$	y^2G°	$3\frac{1}{2}$	27855. 13	578. 61	0. 92			
		$1\frac{1}{2}$	20383. 62		1. 26			$4\frac{1}{2}$	28433. 74		1. 12			
		$2\frac{1}{2}$	20837. 98		1. 39			z^2P°	$1\frac{1}{2}$		27918. 85	—523. 31		
		$3\frac{1}{2}$	21512. 18		1. 44				$0\frac{1}{2}$		28442. 16	1. 450		
$4d^4(a^3D)5p$	z^4P°	$0\frac{1}{2}$	22006. 74	1000. 12	2. 47	$4d^4(a^3P)5p$	z^4S°	$1\frac{1}{2}$	28208. 48	1. 794	0. 772			
		$1\frac{1}{2}$	23006. 86		1. 61			$4d^3 5s(a^3P)5p$	y^6P°		$1\frac{1}{2}$	28278. 25	1. 981	
		$2\frac{1}{2}$	23684. 44		1. 477						$2\frac{1}{2}$	28652. 66	374. 41	1. 768
$4d^3 5s(a^3F)5p$	z^4G°	$2\frac{1}{2}$	22647. 03	375. 53	0. 578	$4d^3 5s(a^3P)5p$	z^4H°	$3\frac{1}{2}$	28973. 12	320. 46	1. 701			
		$3\frac{1}{2}$	23022. 56		0. 98			$4d^3 5s(b^3G)5p$	z^4H°		$4\frac{1}{2}$	29271. 99	247. 06	
		$4\frac{1}{2}$	23536. 77		1. 15						$5\frac{1}{2}$	29519. 05	327. 45	1. 01
		$5\frac{1}{2}$	24203. 05		1. 25			$6\frac{1}{2}$	29846. 50		344. 75	1. 15		
$4d^3 5s(a^3F)5p$	z^4F°	$1\frac{1}{2}$	23243. 87	330. 27	0. 416	$4d^4(a^3H)5p$	y^4G°	$2\frac{1}{2}$	29359. 58	403. 12	0. 693			
		$2\frac{1}{2}$	23574. 14		1. 061			$3\frac{1}{2}$	29762. 70		354. 62	0. 999		
		$3\frac{1}{2}$	24015. 11		1. 243			$4\frac{1}{2}$	30117. 32		540. 28	1. 276		
		$4\frac{1}{2}$	24506. 53		1. 336			$5\frac{1}{2}$	30657. 60		1. 25	1. 25		
$4d^4(a^3P)5p$	z^2D°	$1\frac{1}{2}$	23525. 80	1247. 23	0. 898	$4d^3 5s(a^3F)5p$	y^2D°	$1\frac{1}{2}$	29622. 73	153. 07	0. 81			
		$2\frac{1}{2}$	24773. 03		1. 30			$2\frac{1}{2}$	29775. 80		1. 348			
$4d^4(a^3P)5p$	z^2S°	$0\frac{1}{2}$	23910. 90		2. 123	$4d^4(b^3F)5p$	x^4F°	$1\frac{1}{2}$	29779. 44	208. 01	0. 42			
$4d^4(a^5D)5p$	y^6F°	$0\frac{1}{2}$	23934. 87	179. 92	—0. 601			$2\frac{1}{2}$	29987. 45		174. 11	1. 006		
		$1\frac{1}{2}$	24164. 79		1. 060			$3\frac{1}{2}$	30161. 56		117. 67	1. 18		
		$2\frac{1}{2}$	24396. 80		1. 306			$4\frac{1}{2}$	30279. 23		1. 20	1. 20		
		$3\frac{1}{2}$	24769. 91		1. 380			$4^3 5s(a^5P)5p$	z^6S°		$2\frac{1}{2}$	30059. 60		
		$4\frac{1}{2}$	25199. 81		1. 427	$3\frac{1}{2}$	30716. 50			309. 02	0. 891			
$5\frac{1}{2}$	25680. 36	1. 450	$4d^4(a^1D)5p$	y^2F°	$2\frac{1}{2}$	31025. 52		1. 139						
$4d^4(a^5D)5p$	z^6P°	$1\frac{1}{2}$	24283. 34	259. 79	2. 382	$4d^3 5s(a^3F)5p$	x^4G°	$2\frac{1}{2}$	31056. 60	428. 60	0. 630			
		$2\frac{1}{2}$	24543. 13		1. 820			$3\frac{1}{2}$	31485. 20		519. 43	1. 012		
		$3\frac{1}{2}$	24904. 86		1. 610			$4\frac{1}{2}$	32004. 63		568. 09	1. 160		
					446. 07			$5\frac{1}{2}$	32572. 72		1. 24	1. 24		
$4d^3 5s(a^5P)5p$	y^6D°	$0\frac{1}{2}$	25879. 81	187. 25	3. 22	$4d^3 5s(c^3P)5p$	y^4S°	$1\frac{1}{2}$	31174. 65	1. 957				
		$1\frac{1}{2}$	26067. 06		1. 820			$4d^3 5s(a^3F)5p$	w^4F°		$1\frac{1}{2}$	31551. 46	461. 94	0. 501
		$2\frac{1}{2}$	26386. 36		1. 610						$2\frac{1}{2}$	32013. 40	319. 78	1. 01
		$3\frac{1}{2}$	26832. 43		1. 248			$3\frac{1}{2}$	32333. 18		590. 69	1. 199		
$4\frac{1}{2}$	27419. 62	587. 19	1. 422	$4\frac{1}{2}$	32923. 87		1. 24							
$4d^4(a^5D)5p$	y^4F°	$1\frac{1}{2}$	25930. 01	130. 64	0. 467	$4d^3 5s(a^3F)5p$	x^2F°	$2\frac{1}{2}$	31687. 53	1. 20				
		$2\frac{1}{2}$	26060. 65		1. 085									
		$3\frac{1}{2}$	26165. 79		1. 245									
		$4\frac{1}{2}$	26440. 33		274. 54			1. 334						
$4d^4(a^5D)5p$	x^6D°	$0\frac{1}{2}$	26552. 40	160. 92	2. 441	$4d^3 5s(b^1G)5p$	x^2F°	$2\frac{1}{2}$						
		$1\frac{1}{2}$	26713. 32		1. 45									
		$2\frac{1}{2}$	26933. 34		1. 618									
		$3\frac{1}{2}$	27427. 07		443. 73									
$4\frac{1}{2}$	27974. 87	547. 80	1. 542											

Nb I—Continued

Nb I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>		
$4d^4(a^3D)5p$	v^4F°	$1\frac{1}{2}$	31707.94	99.61	0.80	$4d^4(b^3F)5p$	u^4D°	$0\frac{1}{2}$	35920.45	95.81	0.019		
		$2\frac{1}{2}$	31807.55		1.048			$1\frac{1}{2}$	36016.26		1.195		
		$3\frac{1}{2}$	31973.24		1.343			$2\frac{1}{2}$	36180.13		1.316		
		$4\frac{1}{2}$	32605.39		1.216			$3\frac{1}{2}$	36334.21		1.378		
$4d^3 5s(b^1G)5p$	x^2G°	$3\frac{1}{2}$	31800.74	413.20	0.906	$4d^3 5s(b^3G)5p$	v^2G°	$3\frac{1}{2}$	36048.10	285.60	0.925		
		$4\frac{1}{2}$	32218.94		1.092			$4\frac{1}{2}$	36333.70		1.086		
$4d^4(a^3G)5p$	u^4F°	$1\frac{1}{2}$	31907.74	232.04	0.791	$4d^4(a^1G)5p$	x^2H°	$4\frac{1}{2}$	35275.77		1.14		
		$2\frac{1}{2}$	32139.78		1.035								
		$3\frac{1}{2}$	32451.99		1.115								
		$4\frac{1}{2}$	33136.30		1.240								
$4d^4(a^1G)5p$	w^2F°	$2\frac{1}{2}$	31933.68	153.90	0.982	$4d^3 5s(a^3P)5p$	w^4S°	$1\frac{1}{2}$	36371.05		1.948		
		$3\frac{1}{2}$	32087.58		1.074								
$4d^4(a^3P)5p$	w^4D°	$0\frac{1}{2}$	32066.06	182.63	0.046	$4d^4(a^3H)5p$	y^4H°	$3\frac{1}{2}$	36460.34	256.77	0.691		
		$1\frac{1}{2}$	32248.69		1.184			$4\frac{1}{2}$	36717.11		0.970		
		$2\frac{1}{2}$	32545.52		1.320			$5\frac{1}{2}$	36976.10		1.15		
		$3\frac{1}{2}$	33003.89		1.341			$6\frac{1}{2}$	37254.41		1.23		
$4d^4(a^3H)5p$	z^4I°	$4\frac{1}{2}$	32156.00	226.24	0.835	$4d^4(a^3D)5p$	s^2F°	$2\frac{1}{2}$	36866.60	112.60	0.846		
		$5\frac{1}{2}$	32382.24		0.993			$3\frac{1}{2}$	36979.20		1.14		
		$6\frac{1}{2}$	32672.39		1.08			$4d^4(c^3F)5p$	t^4F°		$1\frac{1}{2}$	37111.67	0.526
		$7\frac{1}{2}$	33116.36		1.19						$2\frac{1}{2}$	37286.62	0.964
$4d^4(b^3F)5p$	w^4G°	$2\frac{1}{2}$	32501.33	301.11	1.06	$4d^4(c^3F)5p$	u^4G°	$2\frac{1}{2}$	37188.28	155.22	0.840		
		$3\frac{1}{2}$			1.21			$3\frac{1}{2}$	37343.50		1.058		
		$4\frac{1}{2}$			625.76			$4\frac{1}{2}$	37523.53		1.190		
		$5\frac{1}{2}$			1.27			$5\frac{1}{2}$	37760.60		1.25		
$4d^3 5s(a^1P)5p$	x^2D°	$1\frac{1}{2}$	32623.02		1.00	$4d^4(a^5D)6s$	e^6D	$0\frac{1}{2}$	37410.17	168.55			
		$2\frac{1}{2}$			1.17			$1\frac{1}{2}$	37578.72			263.64	
$4d^4(a^3G)5p$	v^2F°	$2\frac{1}{2}$	32654.48	244.60	0.830	$4d^4(a^3D)5p$	t^4D°	$0\frac{1}{2}$	37536.56	418.43	0.10		
		$3\frac{1}{2}$	32899.08		1.17			$1\frac{1}{2}$	37954.99		1.051		
		$0\frac{1}{2}$	33011.45		705.56			$2\frac{1}{2}$	38393.49		1.371		
		$1\frac{1}{2}$	33717.01		1.230			$3\frac{1}{2}$	38854.14		1.305		
$4d^3 5s(a^5F)5p$	v^4D°	$1\frac{1}{2}$	33872.18	296.76	1.350	$4d^4(a^3G)5p$	x^4H°	$3\frac{1}{2}$	37624.53	241.53	0.706		
		$2\frac{1}{2}$	34168.94		1.390			$4\frac{1}{2}$	37866.06		0.992		
		$3\frac{1}{2}$						$5\frac{1}{2}$	38143.76		1.089		
		$4\frac{1}{2}$						$6\frac{1}{2}$	38513.85		1.21		
$4d^3 5s(b^1D)5p$	w^2D°	$1\frac{1}{2}$	33086.98	302.89	1.058	$4d^4(b^3F)5p$	r^2F°	$2\frac{1}{2}$	37814.64	417.21	0.876		
		$2\frac{1}{2}$	33389.87		1.212			$3\frac{1}{2}$	38231.85		1.111		
$4d^3 5s(c^3P)5p$	y^2P°	$0\frac{1}{2}$	33902.24	350.72	0.442	$4d^3 5s(b^3D)5p$	t^2D°	$1\frac{1}{2}$	37865.42	314.90	0.90		
		$1\frac{1}{2}$	34252.96					$2\frac{1}{2}$	38180.32		1.077		
$4d^3 5s(a^1H)5p$	z^2I°	$5\frac{1}{2}$	34004.08	319.12	0.946	$4d^3 5s(a^5F)6s$	e^6F	$0\frac{1}{2}$	37871.30	150.11			
		$6\frac{1}{2}$	34323.20		1.08			$1\frac{1}{2}$	38021.41			255.18	
$4d^3 5s(a^1H)5p$	w^2G°	$4\frac{1}{2}$	34235.04	-84.05	1.10	$4d^4(a^3D)5p$	x^2P°	$0\frac{1}{2}$	38182.96	263.82	0.564		
		$3\frac{1}{2}$	34319.09		0.87			$1\frac{1}{2}$	38446.78		1.29		
$4d^4(a^3H)5p$	z^2H°	$4\frac{1}{2}$	34415.52	422.81	0.819	$4d^4(a^3H)5p$	y^2I°	$5\frac{1}{2}$	38251.28	331.76	0.97		
		$5\frac{1}{2}$	34838.33		1.01			$6\frac{1}{2}$	38583.04		1.09		
$4d^3 5s(a^5P)5p$	x^4P°	$0\frac{1}{2}$	34644.22	223.46	2.05	$4d^3 5s(b^1G)5p$	w^2H°	$4\frac{1}{2}$	38448.77	572.04	0.936		
		$1\frac{1}{2}$	34867.68		1.587			$5\frac{1}{2}$	39020.81		1.084		
		$2\frac{1}{2}$	34703.70		-163.98			1.55	$4d^4(a^3D)5p$		w^4P°	$0\frac{1}{2}$	38730.16
$3\frac{1}{2}$	34703.70			$1\frac{1}{2}$	38709.66	53.68	1.660						
$4d^4(a^3G)5p$	v^4G°	$2\frac{1}{2}$	34654.79	198.71	0.627	$4d^3 5s(b^3G)5p$	s^4F°	$1\frac{1}{2}$	38903.00	345.30	0.448		
		$3\frac{1}{2}$	34853.50		1.000			$2\frac{1}{2}$	39248.30		371.83		
		$4\frac{1}{2}$	35156.94		1.073			$3\frac{1}{2}$	39620.13		1.094		
		$5\frac{1}{2}$	35630.62		1.160			$4\frac{1}{2}$	40008.52		1.108		
$4d^3 5s(b^3P)5p$	v^2D°	$1\frac{1}{2}$	34752.70	744.78	0.948	$4d^3 5s(c^3P)5p$	y^2S°	$0\frac{1}{2}$	34807.57				
		$2\frac{1}{2}$	35497.48		1.160								
$4d^3 5s(c^3P)5p$	y^2S°	$0\frac{1}{2}$	34807.57		2.080	$4d^3 5s(b^1D)5p$	u^2F°	$2\frac{1}{2}$	35099.86	78.96	0.868		
		$1\frac{1}{2}$	35178.82		1.117								
$4d^3 5s(b^1D)5p$	u^2F°	$2\frac{1}{2}$	35099.86		0.868	$4d^4(a^3D)5p$	w^4P°	$1\frac{1}{2}$	38709.66	-20.50	2.55		
		$3\frac{1}{2}$	35178.82		1.117			$2\frac{1}{2}$	38763.34			53.68	1.588
$4d^3 5s(b^3P)5p$	x^4S°	$1\frac{1}{2}$	35119.65		1.806	$4d^4(a^3G)5p$	y^2H°	$5\frac{1}{2}$	35344.86	-151.53	1.125		
		$2\frac{1}{2}$	35928.35		1.17			$4\frac{1}{2}$	35496.39			0.954	
$4d^4(b^3F)5p$	u^2D°	$1\frac{1}{2}$	35829.46	98.89	0.834	$4d^3 5s(c^3P)5p$	y^2S°	$0\frac{1}{2}$	34807.57				
		$2\frac{1}{2}$	35928.35		1.17								

Nb I—Continued

Nb I—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	
$4d^3 5s(b^3H)5p$	u^2G°	$3\frac{1}{2}$	38982.20	398.57	1.010	$4d^3 5s(b^3D)5p$	p^4D°	$0\frac{1}{2}$	42801.67	98.40		
		$4\frac{1}{2}$	39380.77		1.10			$1\frac{1}{2}$	42900.07			286.93
$4d^3 5s(d^3F)5p$	t^2G°	$3\frac{1}{2}$	39426.84	210.99	0.91	$4d^3 5s(b^3D)5p$	o^2F°	$2\frac{1}{2}$	43187.00	227.44		
		$4\frac{1}{2}$	39637.83		1.04			$3\frac{1}{2}$	43414.44			
$4d^3 5s(b^3G)5p$	t^4G°	$2\frac{1}{2}$	39530.46	355.20	0.61	$4d^3 5s(d^3F)5p$	p^4F°	$1\frac{1}{2}$	42894.24	128.36		
		$3\frac{1}{2}$	39885.66		1.085			$2\frac{1}{2}$	43022.60			150.40
		$4\frac{1}{2}$	40421.93		536.27			$3\frac{1}{2}$	43173.00			222.80
		$5\frac{1}{2}$	40481.96		60.03			1.18	$4\frac{1}{2}$			43395.80
$4d^4(a^1I)5p$	v^2H°	$4\frac{1}{2}$	39845.51	157.49	1.152	$4d^3 5s(d^3F)5p$	n^2F°	$2\frac{1}{2}$	43342.88	210.29	1.22	
$5\frac{1}{2}$	40003.00	1.228	$3\frac{1}{2}$					43553.17	1.28			
$4d^3 5s(b^3D)5p$	r^4F°	$1\frac{1}{2}$	39981.20	228.52	0.45	$4d^3 5s(b^3D)5p$	q^2G°	$3\frac{1}{2}$	43880.22			
		$2\frac{1}{2}$	40209.72		0.886			$4\frac{1}{2}$				
		$3\frac{1}{2}$	40362.00		152.28							
		$4\frac{1}{2}$	40853.89		491.89			1.129				
$4d^3 5s(b^3H)5p$	s^4G°	$2\frac{1}{2}$	40009.84	475.98	0.929	$4d^3 5s(b^3D)5p$	v^2P°	$1\frac{1}{2}$	44063.49	-1940.80	1.27	
		$3\frac{1}{2}$	40555.28		1.089			$0\frac{1}{2}$	46004.29			
$4d^3 5s(b^3H)5p$	s^4G°	$4\frac{1}{2}$	40921.15	365.87	1.243	$4d^3 5s(b^3D)5p$	o^4D°	$0\frac{1}{2}$	44939.08	-33.54		
		$5\frac{1}{2}$	41371.17		1.25			$1\frac{1}{2}$	44905.54			265.36
					450.02							
$4d^3 5s(b^1F)5p$	q^2F°	$3\frac{1}{2}$	40469.71	-265.47	1.17	$4d^3 5s(b^3G)5p$	u^2H°	$4\frac{1}{2}$	45110.13	148.90	1.09	
$2\frac{1}{2}$	40735.18	0.91	$5\frac{1}{2}$					45259.03				
$4d^4(c^3F)5p$	s^4D°	$0\frac{1}{2}$		479.46	1.29	$4d^4(c^3F)5p$	m^2F°	$2\frac{1}{2}$	45297.30		0.94	
		$1\frac{1}{2}$			1.367			$3\frac{1}{2}$				
		$2\frac{1}{2}$	40473.90									
$4d^3 5s(b^1D)5p$	w^2P°	$0\frac{1}{2}$		1.38	1.38	$4d^4(c^3F)5p$	p^2G°	$3\frac{1}{2}$	45719.07	263.49	0.84	
		$1\frac{1}{2}$	41082.24					$4\frac{1}{2}$	45982.56			
		$3\frac{1}{2}$	41139.95		0.88							
		$4\frac{1}{2}$	41460.97		1.42							
$4d^3 5s(b^1D)5p$	q^4F°	$1\frac{1}{2}$	41554.86	191.48	0.556	$4d^3 5s(a^3F)6p$	r^4G°	$2\frac{1}{2}$	46011.48	459.23		
		$2\frac{1}{2}$	41746.34		1.25			$3\frac{1}{2}$	46470.71			253.49
		$3\frac{1}{2}$	41930.87		184.53			$4\frac{1}{2}$	46724.20			356.63
		$4\frac{1}{2}$	41873.91		-56.96			1.186	$5\frac{1}{2}$			47080.83
$4d^4(a^3G)5p$	s^2G°	$3\frac{1}{2}$	41571.61	324.06		$4d^3 5s(a^3F)6p$	o^4F°	$1\frac{1}{2}$	46170.04	373.52		
		$4\frac{1}{2}$	41895.67						$2\frac{1}{2}$			46543.56
$4d^3 5s(b^3P)5p$	r^4D°	$2\frac{1}{2}$	41615.40			$4d^3 5s(b^3H)5p$	t^2H°	$3\frac{1}{2}$	47022.78	657.81		
		$4\frac{1}{2}$							$4\frac{1}{2}$			47680.59
		$0\frac{1}{2}$	41676.81					662.75				
		$1\frac{1}{2}$	42339.56					212.00				
$4d^4(a^1F)5p$	p^2F°	$3\frac{1}{2}$	42551.56	167.62	1.45	$4d^3 5s(b^3H)5p$	o^2G°	$3\frac{1}{2}$	46932.06	422.26		
		$2\frac{1}{2}$	42719.18		1.39			$4\frac{1}{2}$	47146.06			214.00
$4d^4(a^1F)5p$	p^2F°	$2\frac{1}{2}$	41829.11	158.24	0.86	$4d^3 5s(b^3H)5p$	t^2H°	$4\frac{1}{2}$	46954.40	200.82	1.00	
		$3\frac{1}{2}$	41987.35		1.135			$5\frac{1}{2}$	47155.22			
$4d^4(b^3F)5p$	r^2G°	$3\frac{1}{2}$				$4d^3 5s(b^3H)5p$	o^2G°	$4\frac{1}{2}$	48242.44	713.95	0.92	
		$4\frac{1}{2}$	42193.70					1.15				
$4d^4(c^3F)5p$	s^2D°	$1\frac{1}{2}$				$4d^3 5s(b^3H)5p$	475°	$1\frac{1}{2}$	47537.67			
		$2\frac{1}{2}$	42316.90					1.21				
$4d^3 5s(a^5P)5p$	q^4D°	$0\frac{1}{2}$	42324.31	148.90		$4d^3 5s(b^3H)5p$	n^2G°	$3\frac{1}{2}$	51788.30			
		$1\frac{1}{2}$	42473.21					169.69				
		$2\frac{1}{2}$	42642.90					689.76				
		$3\frac{1}{2}$	43332.66									
						Nb II (5D_0)	Limit	54600				

Nb I OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 3d ¹⁰ 4s ² 4p ⁶ +	Observed Terms	
4d ⁵	a ⁶ S	
4d ³ 5s ²	$\left\{ \begin{array}{l} a \ ^4P \\ a \ ^2P \end{array} \right. \quad a \ ^2D \quad \left\{ \begin{array}{l} a \ ^4F \\ a \ ^2F \end{array} \right. \quad a \ ^2G \quad a \ ^2H$	
	ns (n ≥ 5)	np (n ≥ 5)
4d ⁴ (a ⁵ D)nx	$\left\{ \begin{array}{l} a, c \ ^6D \\ a \ ^4D \end{array} \right.$	$\left\{ \begin{array}{l} z \ ^6P^\circ \quad x \ ^6D^\circ \quad y \ ^6F^\circ \\ z \ ^4P^\circ \quad y \ ^4D^\circ \quad y \ ^4F^\circ \end{array} \right.$
4d ³ 5s(a ⁵ F)nx	$\left\{ \begin{array}{l} e \ ^6F \end{array} \right.$	$\left\{ \begin{array}{l} z \ ^6D^\circ \quad z \ ^6F^\circ \quad z \ ^6G^\circ \\ v \ ^4D^\circ \quad w \ ^4F^\circ \quad x \ ^4G^\circ \end{array} \right.$
4d ⁴ (a ³ P)nx	$\left\{ \begin{array}{l} b \ ^4P \end{array} \right.$	$\left\{ \begin{array}{l} z \ ^4S^\circ \quad w \ ^4D^\circ \\ z \ ^2S^\circ \quad z \ ^2D^\circ \end{array} \right.$
4d ³ 5s(a ³ F)nx	$\left\{ \begin{array}{l} b \ ^4P \end{array} \right.$	$\left\{ \begin{array}{l} z, n \ ^4D^\circ \quad z, o \ ^4F^\circ \quad z, r \ ^4G^\circ \\ y \ ^2D^\circ \quad z \ ^2F^\circ \quad y \ ^2G^\circ \end{array} \right.$
4d ⁴ (a ³ H)nx	$\left\{ \begin{array}{l} a \ ^4H \\ b \ ^2H \end{array} \right.$	$\left\{ \begin{array}{l} y \ ^4G^\circ \quad y \ ^4H^\circ \quad z \ ^4I^\circ \\ z \ ^2G^\circ \quad z \ ^2H^\circ \quad y \ ^2I^\circ \end{array} \right.$
4d ⁴ (a ³ G)nx	$\left\{ \begin{array}{l} a \ ^4G \\ b \ ^2G \end{array} \right.$	$\left\{ \begin{array}{l} u \ ^4F^\circ \quad v \ ^4G^\circ \quad \tau \ ^4H^\circ \\ v \ ^2F^\circ \quad s \ ^2G^\circ \quad y \ ^2H^\circ \end{array} \right.$
4d ³ 5s(a ⁵ P)nx	$\left\{ \begin{array}{l} z \ ^6S^\circ \quad y \ ^6P^\circ \quad y \ ^6D^\circ \\ w \ ^4S^\circ \quad x \ ^4P^\circ \quad q \ ^4D^\circ \end{array} \right.$	
4d ⁴ (a ¹ D)nx		$\left\{ \begin{array}{l} y \ ^2F^\circ \end{array} \right.$
4d ⁴ (b ³ F)nx	$\left\{ \begin{array}{l} b \ ^4F \end{array} \right.$	$\left\{ \begin{array}{l} u \ ^4D^\circ \quad x \ ^4F^\circ \quad w \ ^4G^\circ \\ u \ ^2D^\circ \quad r \ ^2F^\circ \quad r \ ^2G^\circ \end{array} \right.$
4d ⁴ (a ³ D)nx	$\left\{ \begin{array}{l} b \ ^4D \end{array} \right.$	$\left\{ \begin{array}{l} w \ ^4P^\circ \quad t \ ^4D^\circ \quad v \ ^4F^\circ \\ x \ ^2P^\circ \quad s \ ^2F^\circ \end{array} \right.$
4d ³ 5s(b ³ P)nx	$\left\{ \begin{array}{l} b \ ^4D \end{array} \right.$	$\left\{ \begin{array}{l} x \ ^4S^\circ \quad r \ ^4D^\circ \\ v \ ^2D^\circ \end{array} \right.$
4d ⁴ (a ¹ G)nx		$\left\{ \begin{array}{l} w \ ^2F^\circ \quad x \ ^2H^\circ \end{array} \right.$
4d ⁴ (a ¹ I)nx		$\left\{ \begin{array}{l} v \ ^2H^\circ \end{array} \right.$
4d ³ 5s(b ³ G)nx	$\left\{ \begin{array}{l} b \ ^4D \end{array} \right.$	$\left\{ \begin{array}{l} s \ ^4F^\circ \quad t \ ^4G^\circ \quad z \ ^4H^\circ \\ t \ ^2F^\circ \quad v \ ^2G^\circ \quad u \ ^2H^\circ \end{array} \right.$
4d ³ 5s(b ¹ G)nx	$\left\{ \begin{array}{l} b \ ^4D \end{array} \right.$	$\left\{ \begin{array}{l} x \ ^2F^\circ \quad x \ ^2G^\circ \quad w \ ^2H^\circ \end{array} \right.$
4d ³ 5s(b ³ H)nx	$\left\{ \begin{array}{l} b \ ^4D \end{array} \right.$	$\left\{ \begin{array}{l} s \ ^4G^\circ \\ u \ ^2G^\circ \quad t \ ^2H^\circ \end{array} \right.$
4d ⁴ (a ¹ F)nx	$\left\{ \begin{array}{l} b \ ^4D \end{array} \right.$	$\left\{ \begin{array}{l} p \ ^2F^\circ \end{array} \right.$
4d ³ 5s(b ³ D)nx	$\left\{ \begin{array}{l} b \ ^4D \end{array} \right.$	$\left\{ \begin{array}{l} v \ ^2P^\circ \quad p \ ^4D^\circ \quad r \ ^4F^\circ \\ t \ ^2D^\circ \quad o \ ^2F^\circ \end{array} \right.$
4d ³ 5s(c ³ P)nx	$\left\{ \begin{array}{l} b \ ^4D \end{array} \right.$	$\left\{ \begin{array}{l} y \ ^4S^\circ \quad y \ ^4P^\circ \quad x \ ^4D^\circ \\ y \ ^2S^\circ \quad y \ ^2P^\circ \end{array} \right.$
4d ³ 5s(a ¹ P)nx	$\left\{ \begin{array}{l} b \ ^4D \end{array} \right.$	$\left\{ \begin{array}{l} x \ ^2D^\circ \end{array} \right.$
4d ⁴ (c ³ F)nx	$\left\{ \begin{array}{l} b \ ^4D \end{array} \right.$	$\left\{ \begin{array}{l} s \ ^4D^\circ \quad t \ ^4F^\circ \quad u \ ^4G^\circ \\ s \ ^2D^\circ \quad m \ ^2F^\circ \quad p \ ^2G^\circ \end{array} \right.$
4d ³ 5s(a ¹ H)nx	$\left\{ \begin{array}{l} b \ ^4D \end{array} \right.$	$\left\{ \begin{array}{l} w \ ^2G^\circ \quad z \ ^2I^\circ \end{array} \right.$
4d ³ 5s(b ¹ D)nx	$\left\{ \begin{array}{l} b \ ^4D \end{array} \right.$	$\left\{ \begin{array}{l} w \ ^2P^\circ \quad w \ ^2D^\circ \quad u \ ^2F^\circ \end{array} \right.$
4d ³ 5s(d ³ F)nx	$\left\{ \begin{array}{l} b \ ^4D \end{array} \right.$	$\left\{ \begin{array}{l} p \ ^4F^\circ \\ n \ ^2F^\circ \quad t \ ^2G^\circ \end{array} \right.$
4d ³ 5s(b ¹ F)nx	$\left\{ \begin{array}{l} b \ ^4D \end{array} \right.$	$\left\{ \begin{array}{l} q \ ^2F^\circ \end{array} \right.$

*For predicted terms in the spectra of the Nb I isoelectronic sequence, see Vol. II, Introduction.

Nb II—Continued

Nb II—Continued

Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>	Config.	Desig.	<i>J</i>	Level	Interval	Obs. <i>g</i>						
$4d^3(a^4F)5p$	z^3D°	1	34886.33	634.50 1032.44	0.420	$4d^3(^2H)5p$	z^3I°	5	48130.50	486.99 772.24	0.870						
		2	35520.83		1.138			6	48617.49		1.051						
		3	36553.27		1.311			7	49389.73		1.15						
$4d^3(a^4F)5p$	z^5F°	1	36731.79	230.97 414.15 151.48 495.93	0.122	$4d^3(^2G)5p$	z^1G°	4	48253.44		1.083						
		2	36962.76		1.023			$4d^3(^2P)5p$	z^1P°		1	48520.41	0.936				
		3	37376.91		1.266						$4d^3(^2H)5p$	y^1G°	4	49158.16	0.950		
		4	37528.39		1.325								$4d^3(^2P)5p$	x^3D°	1	49687.72	0.820
		5	38024.32		1.357										2	49245.55	1.257
$4d^3(a^4F)5p$	z^5D°	0	37298.20	181.83 317.26 419.08 74.88	0/0	$4d^3(^2P)5p$	x^3D°	3	49759.19	513.64	1.288						
		1	37480.03		1.460			$4d^2\ 5s()5p$	495°	2	49536.62						
		2	37797.29		1.469					$4d^3(^2D)5p$	w^3D°		1	49733.45	38.85		
		3	38216.37		1.466								2	49772.30	91.95		
4	38291.25	1.473	3	49864.25	1.308												
$4d^3(a^4F)5p$	z^3G°	3	38684.96	650.34 768.31	0.764	$4d^3(^2D)5p$	w^3D°	1	49733.45	38.85 91.95	0.616						
		4	39335.30		1.072			2	49772.30		1.167						
		5	40103.61		1.234			3	49864.25		1.308						
$4d^3(a^4F)5p$	z^3F°	2	38984.40	795.55 781.05	0.691	$4d^2\ 5s()5p$	500°	3	50068.70		1.245						
		3	39779.95		1.088			$4d^3(^2D)5p$	x^3P°		0	50295.20	0/0				
		4	40561.00		1.260						1	50474.92	1.202				
		$4d^3(^2P)5p$	z^1S°		0			39791.90			0/0	$4d^3(^2D)5p$	x^3P°	2	50447.36	179.72 -27.56	1.360
2	41710.15			1.312	$4d^3(^2H)5p$	x^3G°	5	50497.90		-353.80							
$4d^3(^4P)5p$	y^5D°	0	42596.58	-463.88 1485.65 268.65 1083.73			0/0	$4d^3(^2H)5p$	x^3G°	4	50851.70	266.39	1.171				
		1	42132.70				1.385			3	50585.31	0.811					
		2	43618.35		1.347	$4d^3(^2D)5p$	y^1D°			2	51182.16	0.965					
		3	43887.00		1.447					$4d^3(^2H)5p$	z^1I°	6	51707.49	1.010			
4	44970.73	1.480	$4d^3(^2D)5p$	y^1P°	1	51787.87	1.13										
$4d^3(^2G)5p$	z^3H°	4			42868.97	698.96 964.47	0.810	$4d^3(^2F)5p$	v^3F°	2	51927.28	9.11 343.25	0.80				
		5			43567.93		1.040			3	51936.39		1.085				
		6	44532.40	1.167	4		52279.64			1.261							
$4d^3(^4P)5p$	y^3D°	1	43649.19	-358.85 1348.43	0.980	$4d^2\ 5s()5p?$	w^3P°	0	52129.80	85.05 173.74	0/0						
		2	43290.34		1.227			1	52214.85		1.43						
		3	44638.77		1.035			2	52388.59		1.44						
$4d^3(^4P)5p$	z^5P°	1	43450.00	776.81 544.66	2.255	$4d^3(^4P)5p$	y^3S°	1	52553.90		1.820						
		2	44226.81		1.745			$4d^2\ 5s()5p$	527°		3	52714.88	1.415				
		3	44771.47		1.562						$4d^3(^2H)5p$	y^1H°	5	52788.10	1.006		
$4d^3(^4P)5p$	z^3P°	0	44286.00	-219.35 857.94	0/0	$4d^3(^2D)5p$	y^1F°	3	53035.95				1.036				
		1	44066.65		1.230			$4d^3(^2F)5p$	w^3G°				3	53702.02	0.88		
		2	44924.59		1.253						4	54428.57	726.55				
$4d^3(^2P)5p$	y^3P°	0	45206.59	168.37 1170.32	0/0	$4d^3(^2F)5p$	w^3G°	5	55021.20	592.63	1.180						
		1	45374.96		1.765			$4d^3(^2G)5p$	v^3D°	3	54009.50	-115.30					
		2	46545.28		1.506					2	54124.80	1.164					
$4d^3(^2G)5p$	z^1H°	5	45342.25		1.018	$4d^3(^2F)5p$	v^3D°	1	54172.60	-47.80	0.527						
		$4d^3(^2G)5p$	y^3G°		3			45919.08	1.014	$4d^3(^2F)5p$	x^1F°	3	55460.54	1.06			
					4			45621.97	1.126			$4d^3(^2F)5p$	x^1D°	2	55721.74	1.000	
5	46499.62			1.171	$4d^3(^2F)5p$	x^1G°	4	57145.25	1.000								
$4d^3(^2G)5p$	y^3F°	2	45655.84	772.79 -133.02			0.740	$4d^3(^2F)5p$	x^1F°	3	55460.54	1.06					
		3	46428.63		1.123	$4d^3(^2F)5p$	x^1D°			2	55721.74	1.000					
		4	46295.61		1.111					$4d^3(^2F)5p$	x^1G°	4	57145.25	1.000			
$4d^3(^2G)5p$	z^1F°	3	45802.49		1.085	$4d^2\ 5s(b^4F)5p$	u^3F°	2	57808.35			694.08 897.37	0.98				
		$4d^3(^2D)5p$	x^3F°		2			46343.10	0.43	3	58502.43		1.25				
					3			46949.47	1.00	4	59399.80		1.34				
$4d^3(^2D)5p$	x^3F°	4	50552.26	3602.79	1.22	$4d^2\ 5s(b^4F)5p$	u^3D°	1	59622.26	-125.24	0.64						
		$4d^3(^2P)5p$	z^3S°	1	46358.94			1.821	$4d^2\ 5s(b^4F)5p$	u^3D°	2	59497.02	208.48	1.19			
				2	47072.92			1.940			3	59705.50	1.33				
$4d^3(^2D)5p$	x^3F°	2	46343.10	606.37 3602.79	0.43	$4d^2\ 5s(b^4F)5p$	v^3G°	3	59509.23	585.27 345.25	1.19						
		3	46949.47		1.00			4	60094.50								
		4	50552.26		1.22			5	60439.75								
$4d^3(^2P)5p$	z^3S°	1	46358.94	1155.62 270.10	0.963	$4d^2\ 5s(b^4F)5p$	v^3G°	3	59509.23	585.27 345.25	1.19						
		5	48500.80		1.038			4	60094.50								
		6	48770.90		1.132			5	60439.75								
$4d^3(^2H)5p$	y^3H°	4	47345.18	1155.62 270.10	0.963	$4d^2\ 5s(b^4F)5p$	v^3G°	3	59509.23	585.27 345.25	1.19						
		5	48500.80		1.038			4	60094.50								
		6	48770.90		1.132			5	60439.75								
$4d^3(^2D)5p^2\ S$	w^3F°	2	47755.76	321.91 549.53	0.63	$4d^2\ 5s(b^4F)5p$	v^3G°	3	59509.23	585.27 345.25	1.19						
		3	48077.67		1.065			4	60094.50								
		4	48627.20		1.100			5	60439.75								
						Nb III ($^4F_{1/2}$)	Limit	113000									

Nb II OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 3d ¹⁰ 4s ² 4p ⁶ +	Observed Terms	
4d ⁴	$\left\{ \begin{array}{l} a^3P \quad a^5D \\ d^3P \quad a^3D \quad b^3F \quad a^3G \quad a^3H \\ a^1S \quad a^1D \quad a^1F \quad a^1G \\ c^1D \quad c^1G \quad a^1I \end{array} \right.$	
	ns (n ≥ 5)	np (n ≥ 5)
4d ³ (a ⁴ F)nx	$\left\{ \begin{array}{l} a^5F \\ a^3F \end{array} \right.$	$\left\{ \begin{array}{l} z^5D^\circ \quad z^5F^\circ \quad z^5G^\circ \\ z^3D^\circ \quad z^3F^\circ \quad z^3G^\circ \end{array} \right.$
4d ³ (⁴ P)nx	$\left\{ \begin{array}{l} a^5P \\ c^3P \end{array} \right.$	$\left\{ \begin{array}{l} z^5S^\circ \quad z^5P^\circ \quad y^5D^\circ \\ y^3S^\circ \quad z^3P^\circ \quad y^3D^\circ \end{array} \right.$
4d ³ (² P)nx	$\left\{ \begin{array}{l} b^3P \\ a^1P \end{array} \right.$	$\left\{ \begin{array}{l} z^3S^\circ \quad y^3P^\circ \quad x^3D^\circ \\ z^1S^\circ \quad z^1P^\circ \quad z^1D^\circ \end{array} \right.$
4d ³ (² G)nx	$\left\{ \begin{array}{l} b^3G \\ b^1G \end{array} \right.$	$\left\{ \begin{array}{l} y^3F^\circ \quad y^3G^\circ \quad z^3H^\circ \\ z^1F^\circ \quad z^1G^\circ \quad z^1H^\circ \end{array} \right.$
4d ³ (² H)nx	$\left\{ \begin{array}{l} b^3H \\ a^1H \end{array} \right.$	$\left\{ \begin{array}{l} x^3G^\circ \quad y^3H^\circ \quad z^3I^\circ \\ y^1G^\circ \quad y^1H^\circ \quad z^1I^\circ \end{array} \right.$
4d ³ (² D)nx	$\left\{ \begin{array}{l} b^3D \\ b^1D \end{array} \right.$	$\left\{ \begin{array}{l} x^3P^\circ \quad w^3D^\circ \quad x^3F^\circ \\ y^1P^\circ \quad y^1D^\circ \quad y^1F^\circ \end{array} \right.$
4d ³ (² F)nx	$\left\{ \begin{array}{l} d^3F \\ b^1F \end{array} \right.$	$\left\{ \begin{array}{l} v^3D^\circ \quad v^3F^\circ \quad w^3G^\circ \\ x^1D^\circ \quad x^1F^\circ \quad x^1G^\circ \end{array} \right.$
4d ³ (² D)nx§		$\left\{ \begin{array}{l} w^3F^\circ \end{array} \right.$
4d ² 5s(b ⁴ F)nx		$\left\{ \begin{array}{l} u^3D^\circ \quad u^3F^\circ \quad v^3G^\circ \end{array} \right.$

*For predicted terms in the spectra of the Zr I isoelectronic sequence, see Vol. II, Introduction.

§ This entry denotes the higher of the two ²D limit terms for this configuration.

Nb III

(Y I sequence; 39 electrons)

Z=41

Ground state 1s² 2s² 2p⁶ 3s² 3p⁶ 3d¹⁰ 4s² 4p⁶ 4d³ ⁴F_{1½}

4d³ ⁴F_{1½} 227005 cm⁻¹

I. P. 28.1 volts

In 1928 Gibbs and White classified 27 lines between 2218.61 Å and 2685.75 Å as combinations between 5s ⁴F and 5p ⁴(DFG)^o. Subsequently Eliason classified 26 additional lines, in the range from 1422.87 Å to 1599.72 Å, as combinations between the ground term, 4d³ ⁴F, and the same triad of odd terms. He confirmed the earlier work in detail, except for the level 5p ⁴F_{1½} at 66499.5, which he rejected.

Eliason derived an improved value of the limit by a linear extrapolation of isoelectronic sequence data. His value in round figures is entered in brackets in the table.

REFERENCES

- R. C. Gibbs and H. E. White, Phys. Rev. **31**, 520 (1928). (T) (C L)
 A. Y. Eliason, Phys. Rev. **43**, 745 (1933). (I P) (T) (C L)

Nb III

Nb III

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
4 <i>d</i> ³	4 <i>d</i> ³ ⁴ F	1½	0. 0	516. 9 659. 7 762. 7	4 <i>d</i> ² (³ F)5 <i>p</i>	5 <i>p</i> ⁴ F°	1½	67095. 1 67868. 7 68787. 8	773. 6 919. 1
		2½	516. 9				2½		
		3½	1176. 6				3½		
		4½	1939. 3				4½		
4 <i>d</i> ² (³ F)5 <i>s</i>	5 <i>s</i> ⁴ F	1½	25221. 2	515. 3 728. 4 909. 8	4 <i>d</i> ² (³ F)5 <i>p</i>	5 <i>p</i> ⁴ D°	0½	69184. 2	490. 1 606. 4 307. 3
		2½	25736. 5				1½	69674. 3	
		3½	26464. 9				2½	70280. 7	
		4½	27374. 7				3½	70588. 0	
4 <i>d</i> ² (³ F)5 <i>p</i>	5 <i>p</i> ⁴ G°	2½	63687. 4	1319. 7 1449. 4 1605. 2	----- Nb IV (³ F ₂) -----	<i>Limit</i>	-----	[227005]	-----
		3½	65007. 1				-----		-----
		4½	66456. 5				-----		-----
		5½	68061. 7				-----		-----

April 1951.

Nb III OBSERVED TERMS*

Config. 1 <i>s</i> ² 2 <i>s</i> ² 2 <i>p</i> ⁶ 3 <i>s</i> ² 3 <i>p</i> ⁶ 3 <i>d</i> ¹⁰ 4 <i>s</i> ² 4 <i>p</i> ⁶ +	Observed Terms			
4 <i>d</i> ³	4 <i>d</i> ³ ⁴ F			
	<i>ns</i> (<i>n</i> ≥ 5)	<i>np</i> (<i>n</i> ≥ 5)		
4 <i>d</i> ² (³ F) <i>nx</i>	5 <i>s</i> ⁴ F	5 <i>p</i> ⁴ D°	5 <i>p</i> ⁴ F°	5 <i>p</i> ⁴ G°

*For predicted terms in the spectra of the Y I isoelectronic sequence, see Vol. II, Introduction.

Nb IV

(Sr I sequence; 38 electrons)

Z = 41Ground state 1*s*² 2*s*² 2*p*⁶ 3*s*² 3*p*⁶ 3*d*¹⁰ 4*s*² 4*p*⁶ 4*d*² ³F₂4*d*² ³F₂ 308600 cm⁻¹

I. P. 38.3 volts

The analysis is from Lang, who has extended the earlier work on this spectrum. He lists 113 classified lines in the interval from 542.38 Å to 2249.98 Å. His limit is derived by a Rydberg formula from the 5, 6*s* ^{1,3}D series.

The values of the 4*d* ¹G and 5*p* ¹F° terms are tentative. "These values must be nearly correct but . . . lack confirmation which can only be supplied by the discovery of the 4*f* singlets."

Observed intersystem combinations connect the singlet and triplet terms.

REFERENCE

R. J. Lang, *Zeeman Verhandelingen* p. 44 (Martinus Nijhoff, The Hague, 1935). (I P) (T) (C L)

Nb IV

Nb IV

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval		
4d ²	4d ² ³ F	2	0.0	1086.4 1258.2	4d(² D)5d	5d ³ P	0	165272.1	717.7 973.9		
		3	1086.4				1	165939.8			
		4	2344.6				2	166963.7			
4d ²	4d ² ¹ G	4	3054.6?		4d(² D)5d	5d ³ F	2	169199	358 1306		
4d ²	4d ² ¹ D	2	7163.3				3	169557			
							4	170863			
4d ²	4d ² ³ P	0	9692.1	431.4 998.0	4d(² D)5d	5d ¹ D	2	172552.0			
		1	10123.5								
		2	11121.5								
4d(² D)5s	5s ¹ D	2	46509.3		4d(² D)6s	6s ¹ D	2	172867.3			
4d(² D)5s	5s ³ D	1	52062.8	565.9 1180.4	4d(² D)6s	6s ³ D	1	174484.1	299.8 1535.9		
		2	52628.7				2	174783.9			
		3	53809.1				3	176319.8			
4d(² D)5p	5p ¹ F ^o	3	92338.7?		4d(² D)4f	4f ³ F ^o	2	174870	2826 3231		
4d(² D)5p	5p ¹ D ^o	2	95814.3				3	177696			
							4	180927			
4d(² D)5p	5p ¹ P ^o	1	97059.8		4d(² D)4f	4f ³ G ^o	3	184372	846 542		
4d(² D)5p	5p ³ D ^o	1	98991.5	1397.2 1388.0						4	185218
		2	100388.7							5	185760
		3	101776.7			¹ D	2	183861.7			
4d(² D)5p	5p ³ F ^o	2	99723.8	661.0 2608.5	Nb v (² D _{1/2})	<i>Limit</i>	-----		308600		
		3	100384.8								
		4	102993.3								
4d(² D)5p	5p ³ P ^o	0	105273.5	-308.0 862.0							
		1	104965.5								
		2	105827.5								

April 1951.

Nb IV OBSERVED TERMS*

Config. 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 3d ¹⁰ 4s ² 4p ⁶ +	Observed Terms			
4d ²	{ 4d ² ³ P 4d ² ¹ D 4d ² ³ F 4d ² ¹ G?			
	ns (n ≥ 5)		np (n ≥ 5)	
4d(² D)nx	{ 5, 6s ³ D 5, 6s ¹ D		{ 5p ³ P ^o 5p ³ D ^o 5p ³ F ^o 5p ¹ P ^o 5p ¹ D ^o 5p ¹ F ^o ?	
			{ 5d ³ P 5d ¹ D 5d ³ F 5d ¹ P 5d ¹ D	
			{ 4f ³ F ^o 4f ³ G ^o	

*For predicted terms in the spectra of the Sr I isoelectronic sequence, see Vol. II, Introduction.

Nb v

(Rb I sequence; 37 electrons)

Z=41

Ground state 1s² 2s² 2p⁶ 3s² 3p⁶ 3d¹⁰ 4s² 4p⁶ 4d ²D_{1/2}4d ²D_{1/2} 400000 cm⁻¹

I. P. 50 volts

Thirteen lines have been classified between 464 Å and 1877 Å as combinations among six terms. The analysis is by Trawick, who has extrapolated the limit (entered in brackets in the table) from isoelectronic sequence data. Charles has recently observed six lines between 464 Å and 774 Å, which agree well with the earlier measurements. He has resolved the 4d ²D—4f ²F^o group. Of the three terms common to the two lists, an average value is used in the table for 4d ²D; 5p ²P^o is the same in both; and Charles' value is entered for 4f ²F^o; but the differences are insignificant.

Nb v—Continued

REFERENCES

- M. W. Trawick, Phys. Rev. **46**, 63 (1934). (I P) (T) (C L)
 G. W. Charles, Phys. Rev. **77**, 120 (1950). (T) (C L)

Nb v

Nb v

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
$4p^6(^1S)4d$	$4d\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	0 1870	1870	$4p^6(^1S)4f$	$4f\ ^2F^{\circ}$	$2\frac{1}{2}$ $3\frac{1}{2}$	215260 215394	134
$4p^6(^1S)5s$	$5s\ ^2S$	$0\frac{1}{2}$	75929		$4p^6(^1S)6s$	$6s\ ^2S$	$0\frac{1}{2}$	228500	
$4p^6(^1S)5p$	$5p\ ^2P^{\circ}$	$0\frac{1}{2}$ $1\frac{1}{2}$	129196 132800	3604					
$4p^6(^1S)5d$	$5d\ ^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	211690 212237	547	Nb vi (1S_0)	<i>Limit</i>		[400000]	

March 1951.

Nb vi

(Kr I sequence; 36 electrons)

 $Z=41$ Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6\ ^1S_0$ $4p^6\ ^1S_0$ 829750 cm^{-1}

I. P. 103 volts

The analysis is far from complete. Charles has classified seven lines between 164 Å and 325 Å as due to transitions from the ground term to levels produced by $4d$, $5d$, $5s$, and $6s$ electrons. The level values in the table are the rounded off wave numbers of the observed lines. He has estimated the limit quoted above by applying a Rydberg formula to the $ns\ s_{\frac{1}{2}}$ levels ($n=5,6$). The higher limit in the table has been determined by the writer by adding the interval of the ground term of Nb vii to the lower limit.

As for Kr I the writer has introduced the $j\ell$ -coupling notation in the general form suggested by Racah. Charles has noted that the $4d$ and $5d$ assignments are open to some question. Consequently, the pair-coupling notation for these levels may need revision.

REFERENCES

- G. Racah, Phys. Rev. **61**, 537 (L) (1942).
 G. W. Charles, Phys. Rev. **77**, 120 (1950). (I P) (T) (C L)

Nb vi

Nb vi

Author	Config.	Desig.	J	Level	Author	Config.	Desig.	J	Level
$4p^6\ p_0$	$4p^6$	$4p^6\ ^1S$	0	0					
	$4p^5(^2P_{1/2})4d$	$4d\ [1\frac{1}{2}]^{\circ}$	2		$5d\ d_{3/2}^?$	$4p^5(^2P_{1/2})5d$	$5d\ [1\frac{1}{2}]^{\circ}$	2	
$4d\ d_{3/2}$			1	306940		$4p^5(^2P_{1/2})6s$	$6s\ [1\frac{1}{2}]^{\circ}$	2	563220
	$4p^5(^2P_{1/2})5s$	$5s\ [1\frac{1}{2}]^{\circ}$	2		$6s\ s_{1/2}^{\circ}$			1	587610
$5s\ s_{1/2}^{\circ}$			1	402050		$4p^5(^2P_{3/2}^{\circ})6s$	$6s'\ [0\frac{1}{2}]^{\circ}$	0	
	$4p^5(^2P_{3/2}^{\circ})5s$	$5s'\ [0\frac{1}{2}]^{\circ}$	0		$6s\ s_{3/2}^{\circ}$			1	606140
$5s\ s_{3/2}^{\circ}$			1	419860					
	$4p^5(^2P_{1/2})5d$	$5d\ [0\frac{1}{2}]^{\circ}$	0			Nb vii ($^2P_{1/2}$)	<i>Limit</i>		829750
$5d\ d_{3/2}^?$			1	553870		Nb vii ($^2P_{3/2}^{\circ}$)	<i>Limit</i>		848930

February 1951.

Nb VII

(Br I sequence; 35 electrons)

Z=41

Ground state $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^5 {}^2P_{1/2}^{\circ}$ $4p^5 {}^2P_{1/2}^{\circ}$ 1005000 cm^{-1}

I. P. 125 volts

The analysis is incomplete. Charles has classified 41 lines between 197 Å and 517 Å as due to transitions from the ground term to 13 higher terms.

He has extrapolated the value of the limit from isoelectronic sequence data.

REFERENCE

G. W. Charles, Phys. Rev. **77**, 120 (1950). (I P) (T) (C L)

Nb VII

Nb VII

Config.	Desig.	<i>J</i>	Level	Interval	Config.	Desig.	<i>J</i>	Level	Interval
$4s^2 4p^5$	$4p^5 {}^2P^{\circ}$	$1\frac{1}{2}$ $0\frac{1}{2}$	0 19185	-19185	$4s^2 4p^4(^1D)4d$	$4d' {}^2P$	$1\frac{1}{2}$ $0\frac{1}{2}?$	398817 401699	-2882
$4s 4p^6$	$4p^6 {}^2S$	$0\frac{1}{2}$	212501		$4s^2 4p^4(^1S)4d$	$4d'' {}^2D$	$2\frac{1}{2}?$ $1\frac{1}{2}?$	403260 407544	-4284
$4s^2 4p^4(^3P)4d$	$4d {}^4D$	$3\frac{1}{2}$ $2\frac{1}{2}$ $1\frac{1}{2}$ $0\frac{1}{2}$	270858 277702 283097	-6844 -5395	$4s^2 4p^4(^3P)5s$	$5s {}^4P$	$2\frac{1}{2}$ $1\frac{1}{2}?$ $0\frac{1}{2}$	416055 426310	-10255
$4s^2 4p^4(^3P)4d$	$4d {}^4P$	$0\frac{1}{2}$ $1\frac{1}{2}$ $2\frac{1}{2}$	326509 331170 336071	4661 4901	$4s^2 4p^4(^3P)5s$	$5s {}^2P$	$1\frac{1}{2}$ $0\frac{1}{2}$	446064 458824	-12760
$4s^2 4p^4(^3P)4d$	$4d {}^2D$	$1\frac{1}{2}$ $2\frac{1}{2}$	335593 349314	13721	$4s^2 4p^4(^1D)5s$	$5s' {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	455071 464153	-9082
$4s^2 4p^4(^3P)4d$	$4d {}^2P$	$0\frac{1}{2}$ $1\frac{1}{2}?$	339687		$4s^2 4p^4(^1S)5s$	$5s'' {}^2S$	$0\frac{1}{2}$	506552	
$4s^2 4p^4(^3P)4d$	$4d {}^2F$	$3\frac{1}{2}$ $2\frac{1}{2}$	364672		Nb VIII (3P_2)		<i>Limit</i>	[1005000]	
$4s^2 4p^4(^1D)4d$	$4d' {}^2D$	$2\frac{1}{2}$ $1\frac{1}{2}$	381021 396124	-15103					

February 1951.

Nb VII OBSERVED TERMS*

$1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} +$	Observed Terms	
$4s^2 4p^5$	$4p^5 {}^2P^{\circ}$	
$4s 4p^6$	$4p^6 {}^2S$	
	<i>ns</i> ($n \geq 5$)	<i>nd</i> ($n \geq 4$)
$4s^2 4p^4(^3P)nx$	{ $5s {}^4P$ $5s {}^2P$	$4d {}^4P$ $4d {}^4D$ $4d {}^2P$ $4d {}^2D$ $4d {}^2F$
$4s^2 4p^4(^1D)nx'$	$5s' {}^2D$	$4d' {}^2P$ $4d' {}^2D$
$4s^2 4p^4(^1S)nx''$	$5s'' {}^2S$	$4d'' {}^2D$

*For predicted terms in the spectra of the Br I isoelectronic sequence, see Vol. II, Introduction.

Additions and Corrections to Volume I

Since the publication of Volume I, in June 1949, some investigations have been carried out that indicate important revisions and additions to data on atomic energy levels included in that work. One survey of all the material has been made quite incidentally, because W. E. Forsythe has requested a revision of the tables of binding energies of electrons for inclusion in the forthcoming revised edition of the Smithsonian Physical Tables. H. N. Russell and the writer¹ have, accordingly, prepared these revised tables of binding energies for the first and second spectra of the elements H to Nb. These calculations are based on the material contained in Volumes I and II of AEL.

In the course of this work Russell² has also revised the ionization potentials of the second spectra of the elements of the iron group to take into account the Ritz correction to limits derived from series of only two members. The idea is not a new one, but the excellent series recently observed in some of these spectra now make the revisions fairly definitive. Catalán³ has similarly corrected the third spectra of the elements Sc through Ga,

but from more meager data. These revised limits and ionization potentials are listed below for Sc II, Sc III; Ti II, Ti III; and V II, V III.

The tables of binding energies indicate three other corrections that should be mentioned. (1) The ionization potential of P I estimated by Finkelnburg and Stern⁴ from a study of the regularities in the run of screening constants, appears to be preferable to the value published in Volume I. (2) The P I term $5s^4P$ at 75064, etc., should be rejected. (3) The run of the binding energies indicates that in Si I the two $3D^o$ terms 45276, etc. and 48399, etc. should possibly have their configurations interchanged, but further observations are needed to settle this question.

Humphreys'⁵ recent work on infrared atomic spectra has led to the discovery of the long awaited $3d^2\ ^3F$ term of Ca I. In addition, he has observed and classified new infrared lines in Si I and V I; and has provisionally located the new $3d^5\ ^4D$ term in V I.

The leading revisions for spectra in Volume I are listed below.

¹ C. E. Moore and H. N. Russell, J. Research Nat. Bur. Std. 48, 61, RP2285 (1952).

² H. N. Russell, J. Opt. Soc. Am. 40, 618 (1950).

³ M. A. Catalán, unpublished material (April 1952).

⁴ W. Finkelnburg and F. Stern, Phys. Rev. 77, 303 (1950).

⁵ C. J. Humphreys, J. Research Nat. Bur. Std. 47, 262, RP2252 (1951).

Page	Spectrum	Remarks
4	He I	Observations on He I sequence, see B. Edlén, Ark. f Fys. (Stockholm) 4, No. 28, 441 (1952).
6	He II	Line 2 of text, p. e. of Λ should read ± 0.0009 .
9	Li I	$10d\ ^2D$ should read 42389 instead of 41489, and should be inserted after $10p\ ^2P^o$.
60	F I	Add reference: K. Lidén, Ark. för Fys. 1, No. 9, 229 (Stockholm) (1949). (I P) (T) (C L) (hfs); and revise terms accordingly. Revised limit should read 140524.5 I. P. 17.418.
76	Ne I	Add reference: K. Burns, K. B. Adams, and J. Longwell, J. Opt. Soc. Am. 40, 339 (1950). (T) (C L) (revised level values). See also; C. J. Humphreys and H. J. Kostkowski, J. Research Nat. Bur. Std. 49, RP2345 (1952).
124	Al I	Add reference: W. R. S. Garton, Nature 165, 322 (1950). (T) (C L) (confirm $3p^2\ ^2S$ term).
154	Si VII	Level $2p^5\ ^3P_2^o$ should read 363170.
163	P I	See discussion above. Revised limit should read 85115 , I. P. 10.55 Reject $5s^4P$ term, 75064.6, etc. Add reference: W. Finkelnburg and F. Stern, Phys. Rev. 77, 303 (1950). (I P)
174	P IX	H. A. Robinson (April 1951) in private conversation has stated that the terms having the configuration $2s\ 2p^4$ need revision.
176	P X	H. A. Robinson (April 1951) in private conversation has stated that the terms having the configuration $2s\ 2p^3$ need revision.
211	A I	Line 8 of the text should read "between 781 A and 786 A." Last reference should read: W. R. Sittner and E. R. Peck, J. Opt. Soc. Am. 39, 474 (1949). (C L)

Page	Spectrum	Remarks																	
243 245	Ca I	Add Humphreys new term discussed above: <table border="1" style="margin-left: auto; margin-right: auto;"> <thead> <tr> <th>Config.</th> <th>Desig.</th> <th><i>J</i></th> <th>Level</th> <th>Interval</th> </tr> </thead> <tbody> <tr> <td rowspan="3">3d²</td> <td rowspan="3">3d² 3F</td> <td>2</td> <td>43474. 87</td> <td rowspan="2">14. 32</td> </tr> <tr> <td>3</td> <td>43489. 19</td> </tr> <tr> <td>4</td> <td>43508. 10</td> <td>18. 91</td> </tr> </tbody> </table>	Config.	Desig.	<i>J</i>	Level	Interval	3d ²	3d ² 3F	2	43474. 87	14. 32	3	43489. 19	4	43508. 10	18. 91		
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3d ²	3d ² 3F	2	43474. 87	14. 32															
		3	43489. 19																
		4	43508. 10	18. 91															
260	Sc I	Reference: C. J. Humphreys, J. Research Nat. Bur. Std. 47 , 262, RP2252 (1951). (T) (C L) Racah (1951) in private conversation has suggested that the configuration of the term x^2P^o																	
261		should read $4s^2(a^1S)4p$ instead of $3d^2(a^1S)4p$.																	
262	Sc II	See discussion above. Revised limit should read 103240 , I. P. 12.80.																	
263	Sc III	Add reference: H. N. Russell, J. Opt. Soc. Am. 40 , 618 (1950). (I P). See discussion above. Revised limit should read 199700 , I. P. 24.75.																	
279	Ti II	Add references: J. E. Mack, Rev. Mod. Phys. 22 , No. 1, 64 (1950). (Summary hfs.) M. A. Catalán, unpublished material (April 1952). (I P). See discussion above.																	
280		Revised limit should read 109506 , I. P. 13.57.																	
281	Ti III	Add reference: H. N. Russell, J. Opt. Soc. Am. 40 , 618 (1950). (I P). See discussion above.																	
282		Revised limit should read 221600 , I. P. 27.47.																	
291	V I	Add reference: M. A. Catalán, unpublished material (April 1952). (I P). Add provisional new term by Humphreys and Kostkowski discussed above: <table border="1" style="margin-left: auto; margin-right: auto;"> <thead> <tr> <th>Config.</th> <th>Desig.</th> <th><i>J</i></th> <th>Level</th> <th>Interval</th> </tr> </thead> <tbody> <tr> <td rowspan="4">3d⁵</td> <td rowspan="4">c 4D</td> <td>0½</td> <td>34329. 05</td> <td rowspan="2">14. 17</td> </tr> <tr> <td>1½</td> <td>34343. 22</td> </tr> <tr> <td>2½</td> <td>34359. 26</td> <td rowspan="2">16. 04</td> </tr> <tr> <td>3½</td> <td>34366. 84</td> </tr> </tbody> </table>	Config.	Desig.	<i>J</i>	Level	Interval	3d ⁵	c 4D	0½	34329. 05	14. 17	1½	34343. 22	2½	34359. 26	16. 04	3½	34366. 84
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293																			
297																			
298	V II	Racah (1951 in private conversation) has suggested that the configuration of the term e^4F should read $3d^4(c^3F)4s$ instead of $3d^5$. Add reference: C. J. Humphreys and H. J. Kostkowski. J. Opt. Soc. Am. 40 , 801 (A) (1950) and unpublished material (May 1951) (T) (C L) See discussion above.																	
300		Revised limit should read [118200] I. P. 14.65.																	
301	V III	Add reference: H. N. Russell, J. Opt. Soc. Am. 40 , 618 (1950). (I P). See discussion above.																	
302		Revised limit should read [236500] I. P. 29.31. Add reference: M. A. Catalán, unpublished material (April 1952): (I P).																	

The writer hopes that the users of these Volumes will report further additions and corrections that come to their attention. The present list is included here in order that readers may have the benefit of these corrections while this program is still in progress.

